Biogeosciences, 11, 3299–3305, 2014 www.biogeosciences.net/11/3299/2014/ doi:10.5194/bg-11-3299-2014 © Author(s) 2014. CC Attribution 3.0 License.





# Composition changes of eroded carbon at different spatial scales in a tropical watershed suggest enrichment of degraded material during transport

C. Rumpel<sup>1</sup>, V. Chaplot<sup>2</sup>, P. Ciais<sup>3</sup>, A. Chabbi<sup>4</sup>, B. Bouahom<sup>5</sup>, and C. Valentin<sup>6</sup>

<sup>1</sup>CNRS, Institut d'Ecologie et de l'Environnement de Paris (IEES), Centre INRA Versailles-Grignon, Bâtiment EGER, Thiverval-Grignon, France

<sup>2</sup>Institut de Recherche pour le Développement (IRD), Laboratoire d'Océanographie et du Climat (LOCEAN), Université Pierre et Marie Curie, Paris, France

<sup>3</sup>Laboratoire des Sciences du Climat et de l'Environnement (LSCE), UMR8212, CEA CNRS UVSQ, Gif-sur-Yvette, France

<sup>4</sup>INRA, UR P3F, Centre INRA Poitou-Charentes, Lusignan, France

<sup>5</sup>SSLCC/NAFRI, Vientiane, LAO-PDR

<sup>6</sup>Centre IRD d'île de France, 32, avenue Henri Varagnat, 93143 Bondy Cedex, France

Correspondence to: C. Rumpel (cornelia.rumpel@grignon.inra.fr)

Received: 1 October 2013 – Published in Biogeosciences Discuss.: 22 November 2013

Revised: 1 April 2014 – Accepted: 3 April 2014 – Published: 20 June 2014

Abstract. In order to assess whether eroded carbon is a net source or sink of atmospheric CO2, characterisation of the chemical composition and residence time of eroded organic matter (EOM) at the landscape level is needed. This information is crucial to evaluate (1) how fast EOM can be decomposed by soil microbes during its lateral transport and (2) its impact at deposition sites. This study considers a continuum of scales to measure the composition of EOM across a steep hillslope landscape of the Mekong basin with intense erosion. We sampled suspended sediments eroded during rainfall events from runoff plots (1 and 2.5 m<sup>2</sup>) and the outlets of four nested watersheds  $(0.6 \times 10^4 \text{ to } 1 \times 10^7 \text{ m}^2)$ . Here we show that changes in the chemical composition of EOM (measured by nuclear magnetic resonance spectroscopy) and in its <sup>13</sup>C and <sup>15</sup>N isotope composition from plot scale through to landscape scale provide consistent evidence for enrichment of more decomposed EOM across distances of 10 km. Between individual soil units (1 m<sup>2</sup>) to a small watershed ( $10^7 \,\mathrm{m}^2$ ), the observed 28 % decrease of the C/N ratio, the enrichment of  $^{13}\mathrm{C}$  and  $^{15}\mathrm{N}$  isotopes as well as O-alkyl C in EOM is of similar magnitude as changes recorded with depth in soil profiles due to soil organic matter "vertical" decomposition. Radiocarbon measurements indicated ageing of EOM from the plot to the watershed scale.

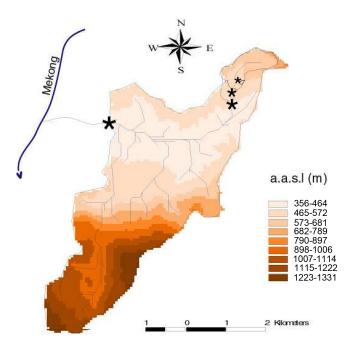
Therefore transport of EOM may lead to enrichment of stabilised soil organic matter compounds, eventually being subject to export from the watershed.

### 1 Introduction

Decomposition of soil organic matter is one of the most important processes controlling the response of the global carbon cycle to climate and land use change (e.g. Lal et al., 2004). Decomposition generally slows down with depth in the soil profile and stable organic C compounds are generally found well below the soil surface in subsoil horizons (Rumpel and Kögel-Knabner, 2011). Organic matter in subsoils most probably experienced intensive degradation as well as vertical transport (Kalbitz and Kaiser, 2012). The paradigm of one-dimensional microbial decomposition occurring only at depth in the soil profile, as adopted by all global models, is however highly questionable because of lateral transport of soil organic matter during hydrologic erosion. In this context, it remains controversial whether eroded organic matter (EOM) is a source or a sink of carbon (van Oost et al., 2007, 2008; Lal and Pimentel, 2008; Kuhn et al., 2009). Organic matter erosion has long been considered to represent a major source of atmospheric CO<sub>2</sub> (Jacinthe and Lal, 2001), but recent modelling evidence suggests that agricultural erosion could lead to a removal of 0.06–0.27 Pg of C from the atmosphere per year (van Oost et al., 2007; Berthe et al., 2007) because erosion transports otherwise labile carbon into landscape elements where its decomposition is slowed down. Transport processes may lead to EOM fractionation as well as to its transformation thereby changing its impact at the site of deposition. In order to obtain detailed understanding of changes occurring during transport, the nature of EOM has to be evaluated at different scales. Compositional changes of organic matter transported in the horizontal direction during soil erosion and in the vertical direction within the soil profile have never been compared. Such a comparison may provide an indication about the effect of the decomposition and transport processes operating in either direction.

In general, erosion leads to preferential detachment and transport of light particulate organic matter (Gregorich et al., 1998; Rodriguez-Rodriguez et al., 2004). This fraction is usually composed of structurally labile plant litter compounds, such as proteins and polysaccharides, with minor contribution of more recalcitrant structures, such as lignin and cutin (Kölbl and Kögel-Knabner, 2004). Therefore, the potential carbon mineralisation from EOM collected in eroded sediments is usually enhanced compared to bulk soil (Jacinthe et al., 2002; Mora et al., 2007; Juarez et al., 2011). However, contribution of highly stable organic matter compounds, such as black carbon, found to be eroded preferentially from fire-affected soils under slash and burn agriculture (Rumpel et al., 2006a), may lead to reduced respiration rates of EOM. Ultimately, it is the relative contribution of labile versus other stable organic matter types to the eroded carbon flux, and their evolution during transport through the landscape, which determines the net C source or sink balance of EOM as well as their impact on soil organic matter mineralisation at the sites of their ultimate deposition.

In this study, we tracked EOM compositional changes of sediments collected at different nested scales in a tropical sub-watershed of the Mekong River, one of the most biochemically active regions of the world because of high erosion rates and tropical climate conditions. Specifically, we measured the elemental, isotopic and bulk chemical composition of sediments eroded from six nested scales of observation, going from pedon units of one square metre to the watershed (10<sup>7</sup> m<sup>2</sup>). Data for every scale were recorded during 34 rainfall events of the rainy season of 2003. Depending on the rainfall intensity EOM transport can be fast during thunderstorms leading to massive soil carbon loss as previously demonstrated by Chaplot and Poesen (2012), but so far, little information exists on the quality of EOM and its fate during its downslope and downstream transport. The aim of the study was to assess the changes occurring in elemental, isotopic and biochemical composition of EOM recovered from different scales (horizontal direction) and to compare them to changes occurring during soil organic matter stabilisation



**Figure 1.** Topography of the study area in northern Laos and location of the sampling sites at the outlet of the  $0.6 \times 10^4$  m<sup>2</sup>,  $30 \times 10^4$  m<sup>2</sup>,  $60 \times 10^4$  m<sup>2</sup> and  $1 \times 10^7$  m<sup>2</sup> catchments.

from topsoil to subsoil horizons (vertical direction) for five major rainfall events occurring throughout the rainy season.

# 2 Material and methods

## 2.1 Study area (geology and climate)

This study was performed in an easily accessible typical sloping land area of northern Laos under traditional slash-and-burn agriculture (Fig. 1). The study area is a watershed of  $10 \,\mathrm{km^2}$  (i.e.  $1 \times 10^7 \,\mathrm{m^2}$ ) forming part of the Mekong River Basin. It shows high relief formed within silty to sandy shales (Fig. 1). Altitudes (Z) range from 280 m at the watershed outlet to 1331 m in the southern part characterised by limestone cliffs. The median Z is 521 m and the coefficient of variation of Z is 78%. Hillslopes exhibit on average a gradient of 32% and are marked by an asymmetry with short gentle hillslopes in the northern part and long and steep slopes in the southern part. Alfisols (Soil Survey Staff, 1999) or Luvisols (WRB) developed from shales are the most common soils in the watershed (Chaplot et al., 2005).

The area is under traditional shifting cultivation and slash-and-burn agriculture. This practice consists of clearing patches of secondary forest and woody/bushy fallows and burning of ground residues to enable cultivation for a short period before the land is allowed to revert to fallow re-growth. Secondary forests cover about 15 % of the whole surface area, mostly on the crest tops.

The study area has a tropical climate with two distinct seasons. The 30-year average annual rainfall is 1403 mm and the mean annual temperature is 25 °C. Rainfall exceeds evapotranspiration throughout the rainy season. The months from November to April are the driest. During the study period in 2003, 34 rainstorms occurred from 25 May–2 October with a cumulative rainfall amount of 1044mm. These events occurred with rainfall between 4 and 127 mm and eight events had a cumulative rainfall over 40 mm.

### 2.2 Plot establishment

In a tributary of the Mekong River, we selected six nested scales from 1, 2.5,  $0.6 \times 10^4$ ,  $30 \times 10^4$ ,  $60 \times 10^4$  and  $1 \times 10^7$  m<sup>2</sup> in an attempt to investigate the different erosion mechanisms of detachment, transport and sedimentation and those involved in the decomposition of the EOM (Fig. 1).

At the hillslope level, 12 enclosed micro-plots of 1 m² and eight plots of  $2.5\,\mathrm{m}^2$  were installed at four positions from the back-slope to the upslope. These micro-plots are part of the micro-catchment of  $0.6\times10^4\,\mathrm{m}^2$  under the third year of rice production following a 4-year fallow period. The weir of the  $0.6\times10^4\,\mathrm{m}^2$  micro-catchment was constructed within the hillslope, at the back-slope position and collected surface runoff and sediments. It exited in the main Houay Pano flume with a permanent flow to the watershed outlet. In this paper the term "micro-plot" is used for the  $1\,\mathrm{m}^2$  areas, and "plot" for the  $2.5\,\mathrm{m}^2$ . The term "micro-catchment" is used for an area of  $0.6\times10^4\,\mathrm{m}^2$ , "catchment" for both the  $30\times10^4\,\mathrm{m}^2$  and  $64\times10^4\,\mathrm{m}^2$  surfaces and "watershed" for the  $1\times10^7\,\mathrm{m}^2$  surface area.

Field measurements were carried out from May immediately after the sowing of rain-fed rice to November following the harvest. The plots were weeded in mid-June, late July and at the end of August. Weeding was performed by shallow tilling (0–2 cm) with a hoe.

### 2.3 Soil sampling

Soil surface samples (0–5 cm) of the bare soil were collected in twelve 1 m<sup>2</sup> plots and eight 2.5 m<sup>2</sup> plots. Three randomly chosen sampling locations were selected in the vicinity of each plot. The samples were collected manually, mixed, airdried at room temperature, and passed through a 2 mm sieve for further analysis of the soil organic carbon.

## 2.4 Sediment sampling

The sediment sampling was performed in situ and at all the spatial scales investigated from 15 May 2003, which corresponded to crop sowing to the end of the 2003 rainy season. A total of 34 rainstorm events with a cumulative rain of 800 mm occurred during that period and were considered. On micro-plots and plots (1 and 2.5 m<sup>2</sup> plots), the total runoff volume from each micro-plot replicate was measured after each rainfall event and aliquots of water and suspended

sediments were collected using buckets. At the catchment level  $(0.6 \times 10^4 \text{ to } 1 \times 10^7 \text{ m}^2)$  automatic water samplers and recorders were used. All the 34 events were used to estimate the runoff, sediment losses by water erosion, while the information on organic matter quality and losses was computed from five rainstorms, randomly selected but stratified by storm size and occurrence during the rainy season to investigate events of different intensities.

All the water samples were oven dried to estimate sediment concentration and sediment discharge. These samples were later acid fumigated to remove inorganic C (Harris et al., 2001) and analysed for total organic carbon (OC), and elemental, chemical and isotopic compositions.

# 2.5 Elemental and stable isotope analysis of carbon and nitrogen

OC and N contents were determined by the dry combustion method using a CHN auto-analyser (CHN NA 1500, Carlo Erba) coupled to an isotopic ratio mass spectrometer (VG Sira 10) yielding the ratio of stable OC isotopes ( $\delta^{13}$ C). Stable N isotope ratios ( $\delta^{15}$ N) were determined with a CHN analyser coupled to an Isochrom III Isotopic mass spectrometer (Micromass-GVI Optima). Results for isotope abundance were reported in per mil (‰) relative to the Pee Dee Belemnite standard (PDB) and relative to air N<sub>2</sub> for  $\delta^{13}$ C and  $\delta^{15}$ N, respectively. Analytical precision was  $\pm 0.1$  mg g<sup>-1</sup> for OC and  $\pm 0.05$  mg g<sup>-1</sup> for N content. Analytical precision for isotope measurements was  $\pm 0.3$  ‰. C/N ratios were calculated on a weight/weight basis.

## 2.6 <sup>14</sup>C activity measurements

Measurements of <sup>14</sup>C activity were performed to assess the mean residence time of C within the system. The <sup>14</sup>C activity was measured on CO<sub>2</sub> obtained by combustion of solid samples at the accelerated mass spectrometer "AMS" facility "Artemis" in Saclay, France.

### 2.7 Chemical composition

The chemical composition of EOM was analysed by <sup>13</sup>C CP-MAS NMR spectroscopy after demineralisation with 10 % hydrofluoric acid. This treatment was found not to alter the chemical composition of organic matter as seen by NMR spectroscopy (Rumpel et al., 2006b). The spectra were recorded on a Bruker DSX-200 NMR spectrometer. Crosspolarisation with magic angle spinning (CPMAS) (Schaefer and Stejskal, 1976) was applied at 6.8 kHz. The <sup>13</sup>C chemical shifts were referenced to tetramethylsilane. A contact time of 1 m was used and the pulse delay was 400 ms. Solid-state <sup>13</sup>C NMR signals were recorded as free induction decay (FID) and Fourier transformed to yield the NMR spectra. The spectra were integrated using the integration routine of the spectrometer. The chemical shift regions 0–45, 45–110, 110–140, 140–160 and 160–220 ppm corresponded to alkyl C, O-alkyl

| Scale           | Size<br>ha                     | <sup>14</sup> C activity pmC | alkyl C/<br>O-alkyl C | O-alkyl C/<br>aryl C |
|-----------------|--------------------------------|------------------------------|-----------------------|----------------------|
| Soil            |                                | $112.1 \pm 0.6$              | 0.69                  | 1.14                 |
| Micro-plot      | $1 \mathrm{m}^2$               | n.d.                         | 0.74                  | 1.03                 |
| Plot            | $2.5 \mathrm{m}^2$             | n.d.                         | 0.71                  | 1.35                 |
| Micro-catchment | $0.6 \times 10^4 \mathrm{m}^2$ | $108.1 \pm 0.6$              | 0.63                  | 2.05                 |
| Catchment       | $30 \times 10^4  \text{m}^2$   | $108.3 \pm 0.6$              | 0.57                  | 2.10                 |
|                 | $64 \times 10^4 \mathrm{m}^2$  | $109.0 \pm 0.6$              | 0.56                  | 2.44                 |
| Watershed       | $1 \times 10^7 \mathrm{m}^2$   | $104.5 \pm 0.6$              | 0.60                  | 2.69                 |

**Table 1.** <sup>14</sup>C activity and ratios of chemical groups as seen by <sup>13</sup>C CPMAS NMR spectroscopy determined for soil and eroded sediments at different scales.

C, C substituted aryl C, O substituted aryl C and carboxylic C, respectively (Wilson, 1987).

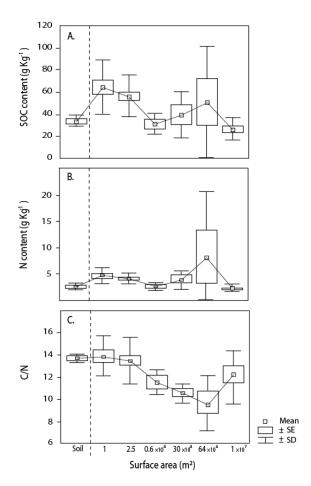
### 2.8 Statistical analyses

Data sets were tested for significance using the Mann and Whitney test. Significant difference was declared at p < 0.05. These analyses were done with Microsoft Excel.

### 3 Results and discussion

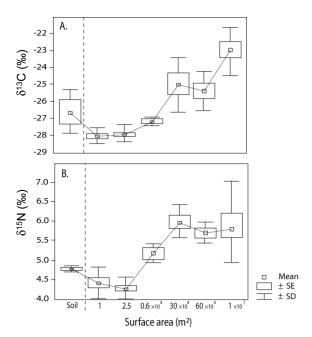
The amount of EOM, and its isotopic and chemical composition, collected across a continuum of spatial scales across the watershed suggest changes in its biogeochemical properties during transport (Figs. 2, 3 and 4, Table 1). Firstly, we observed an increase in the carbon content of eroded sediments from the 1 to the  $2.5 \,\mathrm{m}^2$  scale, followed by a decrease at larger scales (Fig. 2). The C/N ratio of EOM decreased significantly towards larger scales, going from C/N = 14 at the  $1 \,\mathrm{m}^2$  scale down to C/N = 10 at the  $10^7 \,\mathrm{m}^2$  scale. This suggests either preferential removal of C relative to N by microbial decomposition, or enrichment in nitrogen during EOM transport. Stable carbon and nitrogen isotopic ratios increased towards larger scales by 4–5 ‰ for  $^{13}$ C and  $1-2 \,\mathrm{\%}$  for  $^{15}$ N (Fig. 3).

While both sediment detachment and runoff generation are certainly point phenomena and may be assessed on microplots, sedimentation or EOM decomposition is only operative at a certain distance from the "source" (Bloschl and Sivalapan, 1995; Chaplot and Poesen, 2012). Thus different observation points associated with the various nested scales along the downstream transport of sediments allowed the localisation and the quantification of the sedimentation together with an assessment of the potential fate of EOM during its downstream transfer. The action of microbial decomposition on EOM at larger scales is evidenced by changes in the stable carbon and nitrogen isotopic ratios, which both increase towards larger horizontal scales and therefore longer transport (Fig. 3). An isotopic enrichment of similar magnitude is usually observed within soil profiles during organic



**Figure 2.** Organic carbon (**A**), nitrogen content (**B**) and C/N ratio (**C**) for the bulk soil and sediments collected at the different spatial scales under study. Mean values, standard errors and standard deviations were computed from the different plot repetitions and from the 34 rainfall events under study.

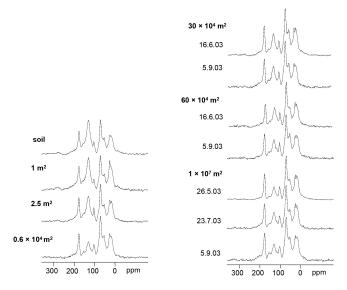
matter decomposition and stabilisation due to "vertical" microbial processing of labile carbon compounds (Rumpel and Kögel-Knabner, 2011). The biggest increase in <sup>13</sup>C and <sup>15</sup>N content of soil organic carbon is found at the scale of 30



**Figure 3.** Box plot of  $\delta^{13}$ C (**A**),  $\delta^{15}$ N (**B**) for the bulk soil and for sediments collected at the different spatial scales under study. Mean values, standard errors and standard deviations were computed from the different plot repetitions and from the 34 rainfall events under study.

 $\times$  10<sup>4</sup> m<sup>2</sup>. Further, changes in the radiocarbon content of the samples also occur in the horizontal direction, i.e. following transport. The <sup>14</sup>C of EOM tends to continue to decrease from 108 to 104 pMC between soil pedon and watershed scale (Table 1). Overall, the <sup>14</sup>C data indicate that EOM is recent (less than 50 years old).

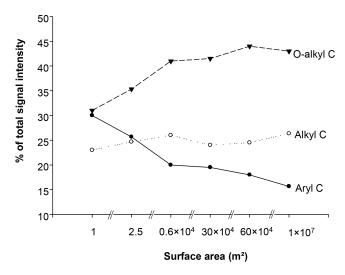
Chemical composition of EOM was analysed using solidstate <sup>13</sup>C nuclear magnetic resonance spectroscopy. This method gives a good overview of the bulk chemical composition of organic matter. In particular, quantitative information on the contribution of alkyl, O-alkyl, aromatic and carboxyl functional groups can be obtained. Figure 4 presents the <sup>13</sup>C CPMAS NMR spectra of an A horizon of soil and EOM collected as suspended sediment at the outlet of the erosion plots at different scales and weirs of the micro-catchment, catchment and watershed. For larger scales, NMR spectra of two or three sampling dates are presented. Data show that the O-alkyl C (45–105 ppm) contribution represented with 31 and 44 % the largest contribution to organic matter in soils and EOM in sediments (Fig. 3). These signals are most likely related to the presence of polysaccharides (Kögel-Knabner, 1997). The peaks between 0 and 45 ppm correspond to the presence of lipids, cutin, suberin and other aliphatic bio-macromolecules, all grouped as alkyl C components. The main signal in this region at 32 ppm corresponds to long-chain methylene structures whereas that at 23 ppm corresponded to short-chain or branched structures synthe-



**Figure 4.** Chemical composition of OC in the upper mineral soil as well as EOM collected after rainfall events at different scales as seen by 13C CPMAS NMR spectroscopy. Note that at three scales sediments sampled at several dates throughout the rainy season were analysed.

sised by micro-organisms during biodegradation (Baldock et al., 1989; Golchin et al., 1996). Signals between 110 and 160 ppm corresponded to aryl C, mainly lignin-derived phenols: protonated, C-substituted and O-substituted aromatic C (Knicker, 1993). The main signal at 130 ppm represented C-substituted aromatic C, which may be derived from stable aromatic compounds, such as black carbon (e.g. Skjemstad et al., 1996). No distinct signal was found between 130 and 160 ppm, in the spectrum area corresponding to tannins and tannin-like structures.

<sup>13</sup>C CPMAS NMR spectroscopy indicated that at 1 and 2.5 m<sup>2</sup> scale, EOM is dominated by an aromatic signal between 110 and 160 ppm most likely related to the presence of fire-derived black carbon produced during slash-and-burn agriculture (Rumpel et al., 2006b). Towards larger scales this signal is decreasing in favour of O-alkyl C structures, indicators of easily degradable polysaccharide material (Fig. 5). This is illustrated by an increasing O-alkyl/aryl ratio (Table 1). The alkyl/O-alkyl ratio of EOM, which is usually increasing upon organic matter decomposition (Baldock et al., 1997), is decreasing towards larger scales (Table 1). This may be related to the clayey texture of the studied soils, which stabilises preferentially O-alkyl material (Rumpel et al., 2008). Consistent with the elemental and stable isotope data, the Oalkyl and alkyl compounds may be microbial derived. Chemical composition at the different scales was not dependent on environmental parameters, as EOM sampled at the same scale but different sampling dates showed similar NMR spectra (Fig. 4).



**Figure 5.** Proportion of signal intensity of <sup>13</sup>C CPMAS NMR spectra recorded for EOM collected at different spatial scales assigned to aryl C, alkyl C and O-alkyl C.

It is interesting to note that changes in chemical and stable isotope composition observed during erosion are very similar to those occurring with depth in soil profiles established at the site of the micro-catchment, where the erosion plots were located (Rumpel et al., 2008). Quantitative evaluation of carbon erosion showed that only 2% of the initial EOM reaches the outlet of the watershed (Chaplot et al., 2005). Estimates on the carbon remaining in soil following humification and stabilisation processes are in a similar range (Rasse et al., 2006). However, the timescales on which these changes are occurring are quite different for the vertical and horizontal direction (> 500 yr in the vertical and < 50 yr in the horizontal direction). Our data suggest in accordance with other work on OM compounds stabilised by pedogenic processes (e.g. Kiem and Kögel-Knabner, 2003; Spielvogel et al., 2008; Rumpel et al., 2010) that EOM compounds are mainly microbial-derived carbohydrates, which are stabilised due to interaction with the mineral phase. Desorption and in turn mineralisation of these labile compounds when reaching fresh and/or saltwater systems may be limited (Butman et al., 2007). We conclude that OM stabilisation is greatly accelerated during the erosion process. Our results do however only provide qualitative information on these changes. To address quantitatively the source and sink functions of EOM it would be necessary to assess the actual CO2 loss due to its degradation. Such an assessment is extremely difficult due to the high probability of EOM to be re-deposited within the watershed (Chaplot et al., 2005). In fact, EOM recovered at the watershed scale may have been subject to continuous deposition and re-mobilisation and during these processes it may have been subject to decomposition and stabilisation processes, leading to enrichment of stabilised organic matter and microbial compounds.

Acknowledgements. This research is part of the Management Soil Erosion Consortium (MSEC) aimed at giving Southeast Asia a better understanding of water erosion within catchments. The authors gratefully acknowledge the Institut de Recherche pour le Développement (IRD, former ORSTOM), the Asian Development Bank (ADB) and the International Water Management Institute (IWMI) for funding this project and the Soil Survey and Land Classification Center for providing data and hosting the foreign research in Laos. We also thank the SSLCC GIS laboratory staff for assistance. We acknowledge CNRS/INSU for funding, the Lehrstuhl für Bodenkunde (TUM, Munich) for providing the NMR spectrometer and the Artemis facility (Saclay, France) for the radiocarbon analyses.

Edited by: S. Bouillon

### References

Baldock, J. A., Oades, J. M., Vassallo, A. M., and Wilson, M. A.: Incorporation of uniformly labeled <sup>13</sup>C-glucose carbon into the organic fraction of a soil, Carbon balance and CP/MAS 13C NMR measurements, Austr. J. Soil Sci., 27, 725–746. 1989.

Baldock, J. A., Oades, J. M., Nelson, P. N., Skene, T. M., Golchin, A., and Clarke, P.: Assessing the extent of decomposition of natural organic materials using solid-state <sup>13</sup>C NMR spectroscopy, Austr. J. Soil Res., 35, 1061–1083, 1997.

Berhe, A. A., Harte, J., Harden, J. W., and Torn, M. S.: The significance of the erosion-induced terrestrial carbon sink, Bioscience, 57, 337–346, 2007.

Bloschl, G. and Sivapalan, M.: Scale issues in hydrological modelling – a review, Hydrol. Proc., 9, 251–290, 1995.

Butman, D., Raymond, P., Oh, N. H., and Mull, K.: Quantity, <sup>14</sup>C age and lability of desorbed soil organic carbon in fresh water and seawater, Org. Geochem., 38, 1547–1557, 2007.

Chaplot, V., Rumpel, C., and Valentin, C.: Water erosion impact on soil and carbon redistributions within uplands of South-East Asia, Global Biogeochem. Cy., 19, GB4004, doi:10.1029/2005GB002493, 2005

Chaplot V. and Poesen J.: Sediment, soil organic carbon and runoff delivery at various spatial scales, Catena, 88, 46–56, 2012.

Golchin, A., Clarke, P., and Oades, J. M.: The heterogeneous nature of microbial products as shown by solid-state 13C CP/MAS NMR spectroscopy, Biogeochemistry, 24, 71–94, 1996.

Gregorich, E. G., Geer, K. J., Anderson, D. W., and Liang, B. C.: Carbon distribution and losses: erosion and deposition effects, Soil Till. Res., 47, 291–302, 1998.

Harris, D., Horwath, W. R., an van Kesseld, C.: Acid fumigation of soils to remove carbonates prior to total organic carbon or carbon-13 isotopic analysis, Soil Sci. Soc. Am. J., 65, 1853– 1856, 2001.

Jacinthe, P. A. and Lal, R.: A mass balance approach to assess carbon dioxide evolution during erosional events, Land Degrad. Dev., 12, 329–339, 2001.

Jacinthe, P. A., Lal, R., and Kimble, J. M.: Carbon dioxide evolution in runoff from simulated rainfall on long-term no-till and plowed soils in southwestern Ohio, Soil Till. Res., 66, 23–33, 2002.

Juarez, S., Rumpel, C., Nontokozo-MChunu, C., and Chaplot, V.: Carbon mineralisation and lignin contribution to sediments de-

- tached from soils by water erosion in a south-african grassland watershed, Geoderma, 167/168, 247–253, 2011.
- Kiem, R. and Kögel-Knabner, I.: Contribution of lignin and polysaccharides to the refractory carbon pool as studied in Cdepleted arable soils, Soil Biol. Biochem., 35, 101–118, 2003.
- Kölbl, A. and Kögel-Knabner, I.: Content and composition of particulate organic matter in arable cambisols of different texture, J. Plant Nutr. Soil Sci., 167, 45–53, 2004.
- Kögel-Knabner, I.: <sup>13</sup>C and 1<sup>5</sup>N NMR spectroscopy as a tool in soil organic matter studies, Geoderma, 80, 243–270, 1997.
- Knicker, H.: Quantitative <sup>15</sup>N- und <sup>13</sup>C-CPMAS-Festkörper- und <sup>15</sup>N-Flüssigkeits-NMR Spektroskopie an Pflanzenkomposten und natürlichen Böden, Dissertation, 1993.
- Kuhn, N. J., Hoffmann, T., Schwanghart, W., and Dotterweich, M.: Agricultural soil erosion and global carbon cycle: controversy over?, Earth Surf. Proc. Landforms, 34, 1033–1038, 2009.
- Lal, R.: Soil carbon sequestration impacts on global climate change and food security, Science, 304, 1623–1627, 2004.
- Lal, R. and Pimentel, D.: Letter on "Soil Erosion: A carbon sink or source?", Science, 319, 1040–1041, 2008.
- Mora, J. L., Guerra, J. A., Armas, C. M., Rodriguez-Rodriguez, A., Arbelo, C. D., and Notario, J. S.: Mineralization rate of eroded organic C in Andosols of the Canary Islands, Sci. Total Environ., 378, 143–146, 2007.
- Rasse, D. P., Dignac, M.-F., Bahri, H., Rumpel, C., Mariotti, A., and Chenu, C.: Assessing lignin turnover in an agricultural field: from plant residues to soil-protected fractions, Europ. J. Soil Sci., 57, 530–538, 2006.
- Rodriguez Rodriguez, A., Guerra, A., Arbelo, C., Mora, J. L., Gorrin, S. P., and Armas, C.: Forms of eroded soil organic carbon in andosols of the Canary Islands (Spain), Geoderma, 121, 205–219, 2004.
- Rumpel, C., Chaplot, V., Planchon, O., Bernadoux, J., Valentin, C., and Mariotti, A.: Preferential erosion of black carbon on steep slopes with slash and burn agriculture, Catena, 65, 30–40, 2006a.

- Rumpel, C., Alexis, M., Chabbi, A., Chaplot, V., Rasse, D. P., Valentin, C., and Mariotti, A.: Black carbon contribution to soil organic matter composition in tropical sloping land under slash and burn agriculture, Geoderma, 130, 35–46, 2006b.
- Rumpel, C., Chaplot, V., Chabbi, A., Largeau, C., and Valentin, C.: Stabilisation of HF soluble and HCl resistant organic matter in tropical sloping soils under slash and burn agriculture, Geoderma, 145, 347–354, 2008.
- Rumpel, C., Eusterhues, K., and Kögel-Knabner, I.: Non-cellulosic neutral sugar contribution to mineral associated organic matter in top-and subsoil horizons of two acid forest soils, Soil Biol. Biochem., 42, 379–382, 2010.
- Rumpel, C. and Kögel-Knabner, I.: Deep soil organic matter a key but poorly understood component of terrestrial C cycle, Plant Soil, 338, 143–158, 2011.
- Schaefer, J. and Stejskal, E. O.: Carbon-13 nuclear magnetic resonance of polymers spinning at magic angle, J. Am. Chem. Soc., 98, 1031–1032, 1976.
- Skjemstad, J. O., Clarke, P., Taylor, J. A., Oades, J. M., and Mc-Clure, S. G.: The chemistry and nature of protected carbon in soil, Austr. J. Soil Res., 34, 251–71, 1996.
- Soil Survey Staff: Soil Taxonomy: A basic system of soil classification for making and interpreting soil surveys, 2nd Edn., USDA-SCS Agric, Handb. 436, US Gov. Print, Office, Washington, DC, 1999.
- Spielvogel, S., Prietzel, J., and Kögel-Knabner, I.: Soil organic matter stabilization in acidic forest soils is preferential and soil type specific, Europ. J. Soil Sci., 59, 674–692, 2008.
- Van Oost, K., Quine, T. A., Govers, G., De Gryze, S., Six, J., Harden, J. W., Ritchie, J. C., McCarty, G. W., Heckrath, G., Kosmas, C., Giraldez, J. V., da Silva, J. R. M., and Merckx, R.: The impact of agricultural soil erosion on the global carbon cycle, Science, 318, 626–629, 2007.
- Van Oost, K., Six, J., Govers, G., Quine, T., and de Gryze, S.: Reply to letter on "Soil Erosion: A carbon sink or source?" by R. Lal and D. Pimentel, Science, 319, 1041–1042, 2008.