



Supplement of

Changes in dissolved iron deposition to the oceans driven by human activity: a 3-D global modelling study

S. Myriokefalitakis et al.

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Table S1. Emissions used in TM4-ECPL for PRESENT (year 2008), PAST (year 1850) and FUTURE (year 2100) simulations. All emissions are in Tg yr⁻¹, except for NOx (Tg-N yr⁻¹), NH₃ (Tg-N yr⁻¹), SO_x (Tg-S yr⁻¹) and DMS (Tg-S yr⁻¹).

| Species | Year | Biomass Burning | Anthropogenic | Aircrafts | Ships | Biogenic | Soils | Oceans |
|--------------------------------|------|-----------------|---------------|-----------|----------|----------|-------|----------|
| ACETONE | 1850 | 6.46E-02 | 4.23E-01 | | | 3.59E+01 | | |
| | 2008 | 4.23E+00 | 2.64E+00 | 3.03E-07 | | | | |
| | 2100 | 5.09E+00 | 2.73E+00 | 2.80E-07 | | | | |
| BC | 1850 | 2.62E-01 | 1.05E+00 | | 8.05E-04 | | | |
| | 2008 | 2.62E+00 | 5.39E+00 | 8.39E-06 | 1.24E-01 | | | |
| | 2100 | 3.17E+00 | 1.11E+00 | 7.32E-06 | 1.43E-02 | | | |
| BENZENE | 1850 | 4.83E-01 | 7.51E-01 | | 8.05E-04 | | | |
| | 2008 | 4.83E+00 | 6.31E+00 | 4.90E-06 | 1.16E-01 | | | |
| | 2100 | 5.89E+00 | 2.15E+00 | 4.52E-06 | 4.80E-02 | | | |
| C ₂ H ₄ | 1850 | 6.35E-01 | 1.24E+00 | | 2.21E-03 | 1.75E+01 | | 1.20E+00 |
| | 2008 | 6.35E+00 | 6.87E+00 | 3.04E-06 | 2.57E-01 | | | |
| | 2100 | 7.76E+00 | 2.24E+00 | 2.80E-06 | 1.06E-01 | | | |
| C ₂ H ₆ | 1850 | 3.17E-01 | 5.93E-01 | | 1.01E-03 | 3.01E-01 | | 8.00E-01 |
| | 2008 | 3.17E+00 | 2.87E+00 | 4.37E-06 | 1.50E-01 | | | |
| | 2100 | 3.96E+00 | 8.09E-01 | 4.03E-06 | 6.22E-02 | | | |
| C ₃ H ₆ | 1850 | 6.26E-01 | 9.14E-01 | | 2.56E-03 | 1.66E+01 | | 1.30E+00 |
| | 2008 | 6.25E+00 | 6.41E+00 | 4.64E-06 | 2.98E-01 | | | |
| | 2100 | 7.68E+00 | 2.27E+00 | 4.29E-06 | 1.23E-01 | | | |
| C ₃ H ₈ | 1850 | 1.96E-01 | 2.42E-01 | | 1.96E-01 | 3.01E-02 | | 1.00E+00 |
| | 2008 | 1.96E+00 | 3.47E+00 | 4.87E-06 | 4.17E-01 | | | |
| | 2100 | 2.54E+00 | 1.26E+00 | 4.50E-06 | 1.73E-01 | | | |
| C ₄ H ₁₀ | 1850 | 9.07E-02 | 4.56E-01 | | 6.27E-03 | 6.09E-02 | | 6.80E+00 |
| | 2008 | 9.07E-01 | 3.32E+01 | 4.01E-05 | 8.79E-01 | | | |
| | 2100 | 1.11E+00 | 1.96E+01 | 3.70E-05 | 3.64E-01 | | | |
| CH ₂ O | 1850 | 5.87E-01 | 1.45E-01 | | | 4.33E+00 | | |
| | 2008 | 5.86E+00 | 3.39E+00 | 7.54E-07 | | | | |
| | 2100 | 7.21E+00 | 1.43E+00 | 6.97E-07 | | | | |
| CH ₃ CHO | 1850 | 4.86E-01 | 3.16E-01 | | | 2.04E+01 | | |
| | 2008 | 4.85E+00 | 1.80E+00 | 2.07E-06 | | | | |
| | 2100 | 6.00E+00 | 5.95E-01 | 1.91E-06 | | | | |
| CH ₃ OH | 1850 | 1.17E+00 | 6.14E-01 | | | 1.38E+02 | | |
| | 2008 | 1.17E+01 | 4.87E+00 | 2.90E-06 | | | | |
| | 2100 | 1.42E+01 | 2.85E+00 | 2.68E-06 | | | | |

| | | | | | | | | |
|---------------------------|------|----------|----------|----------|----------|----------|----------|----------|
| CO | 1850 | 4.61E+01 | 6.30E+01 | | 7.74E-03 | 8.37E+01 | | 2.00E+01 |
| | 2008 | 4.60E+02 | 5.99E+02 | 6.55E-04 | 1.15E+00 | | | |
| | 2100 | 5.49E+02 | 2.34E+02 | 5.11E-04 | 2.29E-01 | | | |
| MEK | 1850 | 0.00E+00 | 0.00E+00 | 0.00E+00 | 0.00E+00 | 6.98E-01 | | |
| | 2008 | | | 1.04E-06 | | | | |
| | 2100 | | | 9.58E-07 | | | | |
| NOx | 1850 | 7.74E-03 | 7.74E-03 | | 7.74E-03 | 8.00E+00 | | |
| | 2008 | 5.48E+00 | 2.67E+01 | 1.44E-03 | 5.14E+00 | | | |
| | 2100 | 6.39E+00 | 7.48E+00 | 1.84E-03 | 9.47E-01 | | | |
| POA | 1850 | 2.33E+00 | 4.63E+00 | | 9.05E-04 | | | |
| | 2008 | 2.33E+01 | 1.39E+01 | 2.52E-06 | 1.29E-01 | | | |
| | 2100 | 2.82E+01 | 3.46E+00 | 2.20E-06 | 1.42E-02 | | | |
| SOx | 1850 | 1.93E-01 | 1.01E+00 | | 3.60E-02 | | | |
| | 2008 | 1.93E+00 | 4.69E+01 | 1.14E-04 | 5.22E+00 | | | |
| | 2100 | 2.21E+00 | 7.70E+00 | 1.00E-04 | 5.04E-01 | | | |
| TOLUENE | 1850 | 1.42E-01 | 3.55E-01 | | 5.71E-04 | 1.41E+00 | | |
| | 2008 | 1.42E+00 | 7.22E+00 | 7.36E-06 | 6.71E-02 | | | |
| | 2100 | 1.73E+00 | 4.70E+00 | 6.80E-06 | 2.78E-02 | | | |
| XYLENE | 1850 | 3.71E-02 | 2.57E-01 | | 1.56E-03 | | | |
| | 2008 | 3.70E-01 | 1.18E+01 | 6.69E-06 | 2.06E-01 | | | |
| | 2100 | 4.51E-01 | 7.66E+00 | 8.52E-02 | 6.18E-06 | | | |
| CH₃COOH | 1850 | 1.73E+00 | 1.45E+00 | | | 3.25E+00 | | |
| | 2008 | 1.73E+01 | 7.46E+00 | | | | | |
| | 2100 | 2.08E+01 | 2.33E+00 | | | | | |
| HCOOH | 1850 | 4.33E-01 | 3.62E-01 | | | 3.25E+00 | | |
| | 2008 | 4.32E+00 | 1.87E+00 | | | | | |
| | 2100 | 5.21E+00 | 5.83E-01 | | | | | |
| DMS | 1850 | 1.47E-04 | | | | | 1.90E+01 | |
| | 2008 | 1.47E-03 | | | | | | |
| | 2100 | 1.76E-03 | | | | | | |
| GLY | 1850 | 5.87E-01 | | | | | 2.00E+01 | |
| | 2008 | 5.86E+00 | | | | | | |
| | 2100 | 7.21E+00 | | | | | | |
| NH3 | 1850 | 9.20E-01 | 5.40E+00 | | | | 8.00E+00 | |
| | 2008 | 9.20E+00 | 3.41E+01 | | | | | |
| | 2100 | 1.00E+01 | 5.15E+01 | | | | | |
| ISOPRENE | 2008 | | | | | 5.20E+02 | | 1.00E+00 |
| TERPENE | 2008 | | | | | 1.04E+02 | | 2.00E-04 |

Table S2. Aqueous-phase mechanism for Fe chemistry used in the TM4-ECPL model.

| Reactions | K_{298} (M ⁻ⁿ⁺¹ s ⁻¹) | Ea/R (K) | References |
|--|---|-------------|--------------------------------|
| $\text{Fe}^{2+} + \text{H}_2\text{O}_2 \rightarrow \text{Fe}^{3+} + \text{OH} + \cdot\text{OH}$ | 5.24×10^1 | 5050 | Kremer, 2003 |
| $\text{Fe}^{2+} + \text{O}_2^- + 2\text{H}^+ \rightarrow \text{Fe}^{3+} + \text{H}_2\text{O}_2$ | 1.00×10^7 | 5050 | Rush and Bielski, 1985 |
| $\text{Fe}^{2+} + \text{HO}_2 + \text{H}^+ \rightarrow \text{Fe}^{3+} + \text{H}_2\text{O}_2$ | 1.20×10^6 | 5050 | Jayson et al., 1973b |
| $\text{Fe}^{2+} + \text{OH} \rightarrow \text{Fe(OH)}^{2+}$ | 4.60×10^8 | 1100 | Christensen and Sehested, 1981 |
| $\text{Fe}^{2+} + \text{NO}_3 \rightarrow \text{Fe}^{3+} + \text{NO}_3^-$ | 8.00×10^6 | | Pikaev et al., 1974 |
| $\text{Fe}^{2+} + \text{NO}_2 + \text{H}^+ \rightarrow \text{Fe}^{3+} + \text{HONO}$ | 3.10×10^4 | | Epstein et al., 1982 |
| $\text{Fe}^{2+} + \text{O}_3 (+ \text{H}_2\text{O}) \rightarrow \text{Fe(OH)}^{2+} + \text{OH} + \text{O}_2 (*)$ | 8.20×10^5 | | Logager et al., 1992 |
| | | | |
| $\text{Fe}^{3+} + h\nu (+ \text{H}_2\text{O}) \rightarrow \text{Fe}^{2+} + \text{OH} + \text{H}^+$ | 6.41×10^{-6} | | Benkelberg and Warneck, 1995 |
| $\text{Fe(OH)}^{2+} + h\nu \rightarrow \text{Fe}^{2+} + \text{OH}$ | 4.51×10^{-3} | | Benkelberg and Warneck, 1995 |
| $\text{Fe(OH)}_2^+ + h\nu \rightarrow \text{Fe}^{2+} + \text{OH} + \cdot\text{OH}$ | 5.77×10^{-3} | | Benkelberg et al., 1991 |
| $\text{Fe(OH)}^{2+} + \text{O}_2^- \rightarrow \text{Fe}^{2+} + \text{O}_2 + \cdot\text{OH}$ | 1.50×10^8 | | Rush and Bielski, 1985 |
| $\text{Fe(OH)}^{2+} + \text{HO}_2 \rightarrow \text{Fe}^{2+} + \text{O}_2 + \text{H}_2\text{O}$ | 1.30×10^5 | | Zajka et al., 1994 |
| | | | |
| $\text{Fe}^{3+} + \text{SO}_4^{2-} \rightarrow \text{Fe}(\text{SO}_4)^+$ | 3.20×10^3 | | Jayson et al., 1973b |
| $\text{Fe}(\text{SO}_4)^+ \rightarrow \text{Fe}^{3+} + \text{SO}_4^{2-}$ | 2.70×10^1 | | Jayson et al., 1973b |
| $\text{Fe}(\text{SO}_4)^+ + h\nu \rightarrow \text{Fe}^{2+} + \text{SO}_4^-$ | 6.43×10^{-3} | | Lin et al., 2014 |
| | | | |
| $\text{Fe}^{3+} + \text{C}_2\text{O}_4^{2-} \rightarrow \text{Fe}(\text{C}_2\text{O}_4)^+$ | 7.50×10^6 | | Lin et al., 2014 |
| $\text{Fe}(\text{C}_2\text{O}_4)^+ \rightarrow \text{Fe}^{3+} + \text{C}_2\text{O}_4^{2-}$ | 3.00×10^{-3} | | Lin et al., 2014 |
| $\text{Fe}(\text{C}_2\text{O}_4)^+ + \text{C}_2\text{O}_4^{2-} \rightarrow \text{Fe}(\text{C}_2\text{O}_4)_2^-$ | 1.89×10^4 | | Lin et al., 2014 |
| $\text{Fe}(\text{C}_2\text{O}_4)_2^- \rightarrow \text{Fe}(\text{C}_2\text{O}_4)^+ + \text{C}_2\text{O}_4^{2-}$ | 3.30×10^{-3} | | Lin et al., 2014 |
| $\text{Fe}(\text{C}_2\text{O}_4)^+ + \text{O}_2^- \rightarrow \text{Fe}(\text{C}_2\text{O}_4) + \text{O}_2$ | 1.00×10^6 | | Sedlak and Hoigne, 1993 |
| $\text{Fe}(\text{C}_2\text{O}_4)^+ + \text{HO}_2 \rightarrow \text{Fe}(\text{C}_2\text{O}_4) + \text{O}_2 + \text{H}^+$ | 1.20×10^5 | | Sedlak and Hoigne, 1993 |
| $\text{Fe}(\text{C}_2\text{O}_4)_2^- + \text{O}_2^- \rightarrow \text{Fe}(\text{C}_2\text{O}_4)_2^{2-} + \text{O}_2$ | 1.00×10^6 | | Sedlak and Hoigne, 1993 |
| $\text{Fe}(\text{C}_2\text{O}_4)_2^- + \text{HO}_2 \rightarrow \text{Fe}(\text{C}_2\text{O}_4)_2^{2-} + \text{O}_2 + \text{H}^+$ | 1.20×10^5 | | Sedlak and Hoigne, 1993 |
| $\text{Fe}(\text{C}_2\text{O}_4)^+ + \text{H}_2\text{O}_2 \rightarrow \text{Fe}(\text{C}_2\text{O}_4)^+ + \text{OH} + \cdot\text{OH}$ | 5.24×10^4 | | Sedlak and Hoigne, 1993 |
| $\text{Fe}(\text{C}_2\text{O}_4)_2^- + h\nu (+ \text{O}_2) \rightarrow \text{Fe}^{2+} + \text{C}_2\text{O}_4^{2-} + 2\text{CO}_2 + \text{O}_2^-$ | 2.47×10^{-2} | | Lin et al., 2014 |
| | | | |
| Equilibrium | K_{eq} (mol kg ⁻¹) | | References |
| $\text{Fe}^{3+} + \text{H}_2\text{O} \leftrightarrow \text{Fe}(\text{OH})^{2+} + \text{H}^+$ | 1.10×10^{-4} | | Ervens et al., 2003 |
| $\text{Fe}(\text{OH})^{2+} + \text{H}_2\text{O} \leftrightarrow \text{Fe}(\text{OH})_2^+ + \text{H}^+$ | 1.40×10^{-7} | | Ervens et al., 2003 |

(*) For simplicity FeO^{2+} formation is neglected and formation of $\text{Fe}(\text{OH})^{2+}$ is here considered.

Table S3. Number of measurements (N), mean correlation coefficient (r), normalised mean bias (NMB), normalised mean error (NME) and root mean square error (RMSE) values calculated for total Iron (TFe) and dissolved Iron (DFe) as presented in Figure 5.

| Figure | Observations | N | R | NMB (%) | NME (%) | RMSE |
|-----------------|---|-----|------|---------|---------|------|
| Fig. 5a | TFe (ng-Fe m⁻³) | 412 | 0.07 | 37 | 147 | 619 |
| Fig. 5b | DFe (ng-Fe m⁻³) | 412 | 0.07 | 67 | 84 | 29 |
| Fig. 5c | SFe (%) | 412 | 0.07 | 47 | 20 | 174 |
| Fig. 5d | TFe (ng-Fe m⁻³) | 452 | 0.4 | -26 | 82 | 77 |
| Fig. S5i | OXL-with Fe-Oxalato Chemistry (ng m⁻³) | 130 | 0.47 | -0.396 | 69 | 233 |
| Fig. S5i | OXL-without Fe-Oxalato Chemistry (ng m⁻³) | 130 | 0.47 | 2.110 | 68 | 231 |

$$R = \left[\frac{\frac{1}{N} \sum_{i=1}^N (O_i - \bar{O})(P_i - \bar{P})}{\sigma_O \sigma_P} \right] \quad (\text{eq. S1})$$

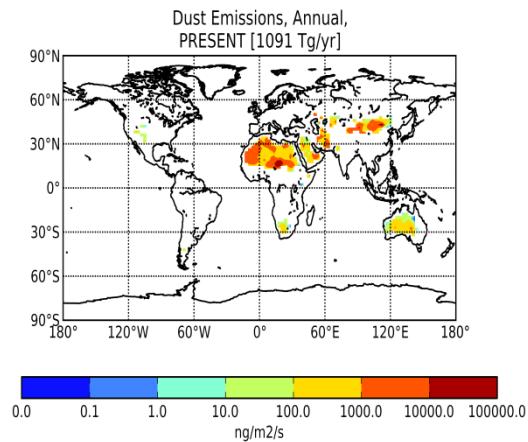
$$NMB = \frac{\sum_{i=1}^N (M_i - O_i)}{\sum_{i=1}^N O_i} \times 100 \quad (\text{eq. S2})$$

$$RMSE = \sqrt{\frac{1}{N} \sum_{i=1}^N (P_i - O_i)^2} \quad (\text{eq. S3})$$

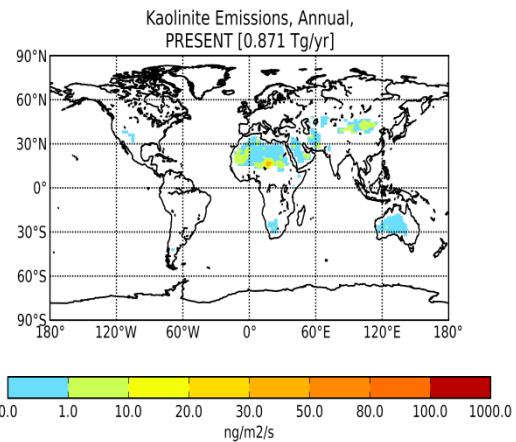
$$NME = \frac{\sum_{i=1}^N |M_i - O_i|}{\sum_{i=1}^N O_i} \times 100 \quad (\text{eq. S4})$$

Figure S1. Annual emissions ($\text{ng m}^{-2} \text{s}^{-1}$) as taken into account by TM4-ECPL for the present atmosphere of a) dust and of Fe contained in: b) kaolinite, c) illite, d) smectite, e) hematite, f) feldspars g) in fossil fuels (including shipping oil combustion aerosols) and h) in biomass burning aerosols.

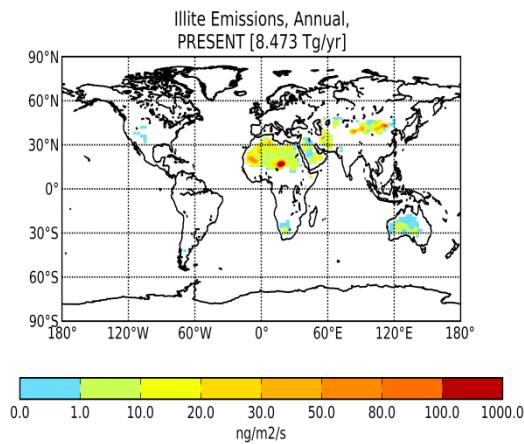
a)



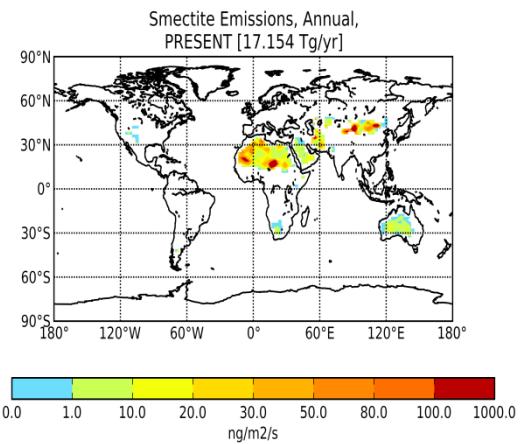
b)



c)



d)



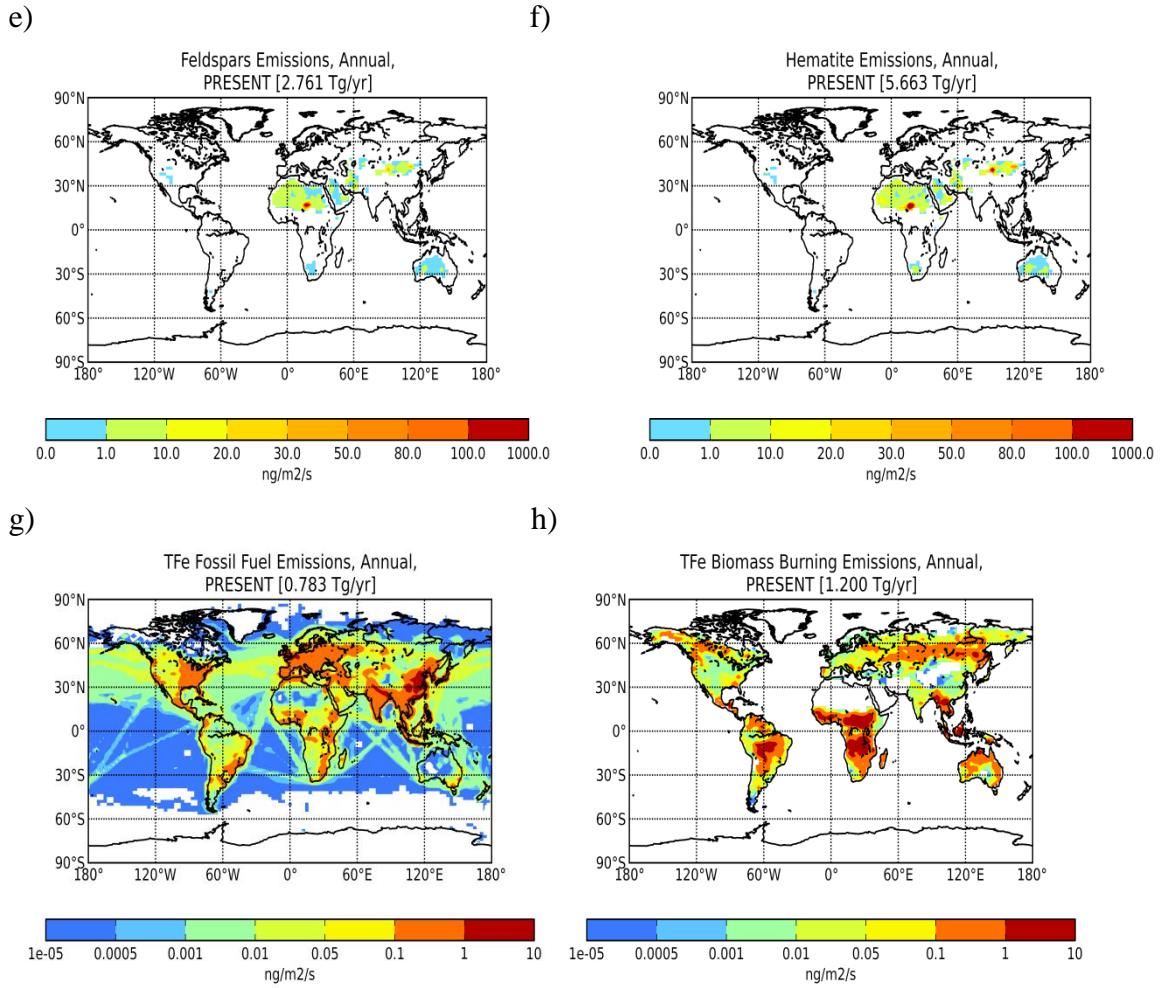
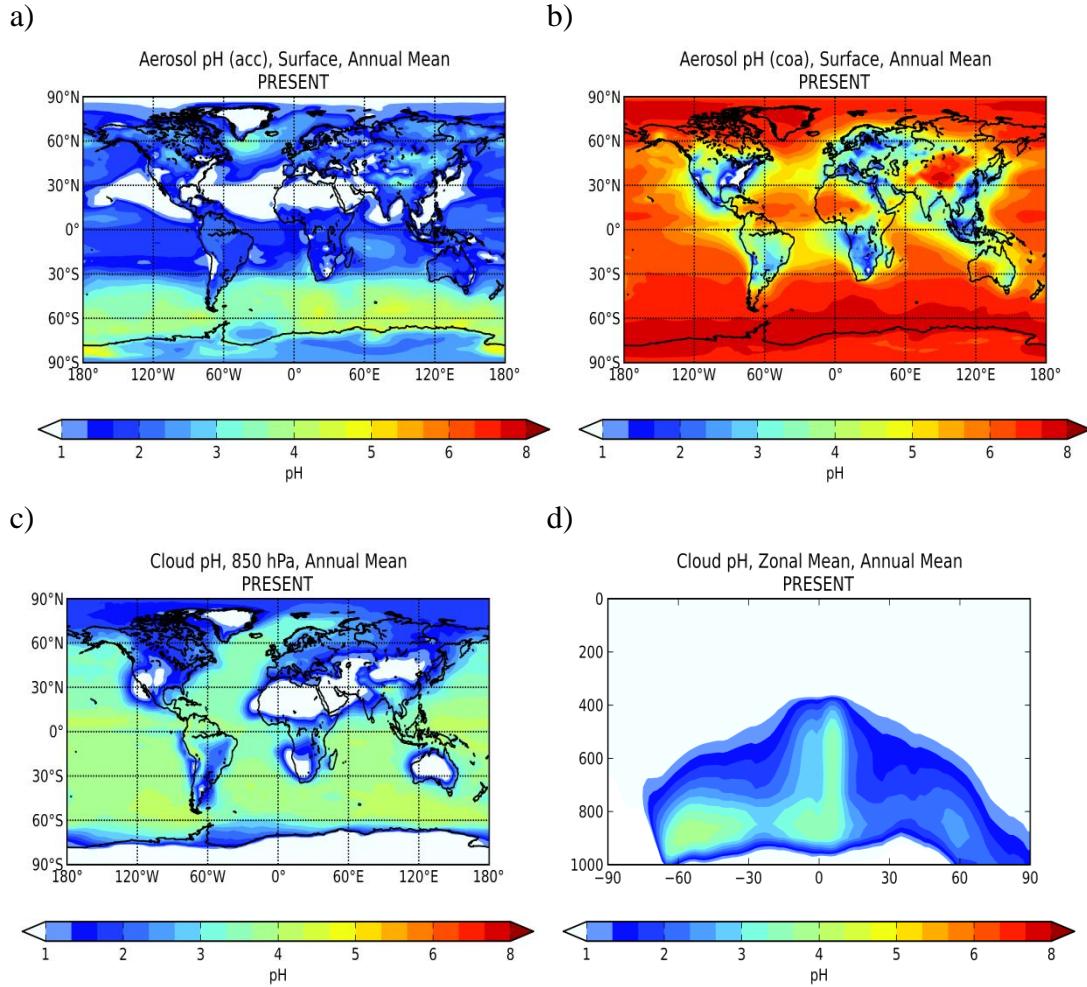


Figure S2. pH values as calculated by TM4-ECPL for surface aerosol water for fine (a) and coarse (b) particles and for cloud water at 850hPa pressure level (c) and zonal mean (d) and the corresponding percentage changes (%) using PAST (left panels: e, g, i) and FUTURE (right panels: f, h, j) emissions with respect to the present-day (PRESENT) simulations.



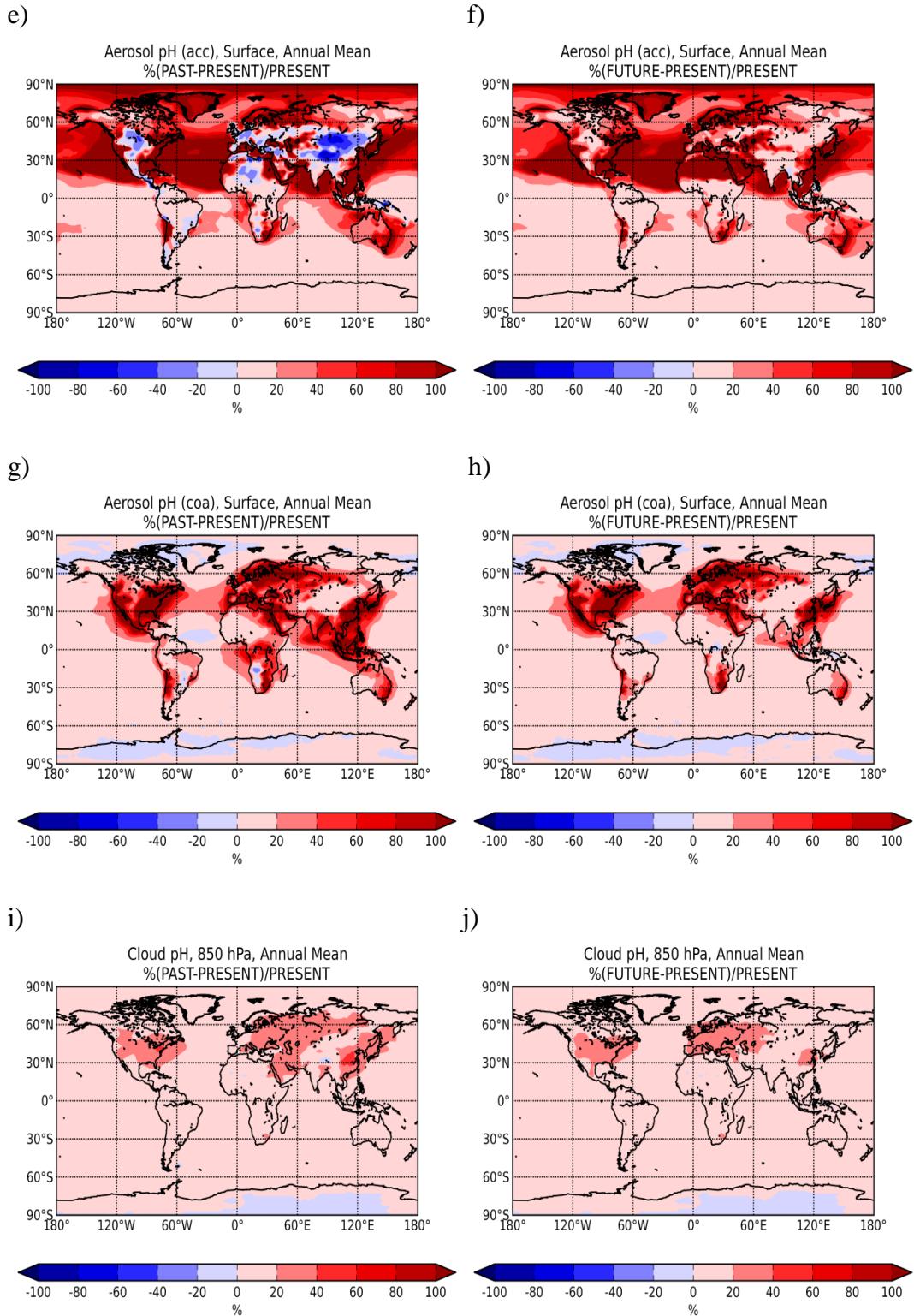


Figure S3. Comparison of OXL surface concentrations with (+) and without (o) taking into account Fe-oxalato complexes in the aqueous-phase chemical scheme for measurements (red colour for urban sites, green colour for rural sites and blue colour for marine sites) as compiled by Myriokefalitakis et al. (2011).

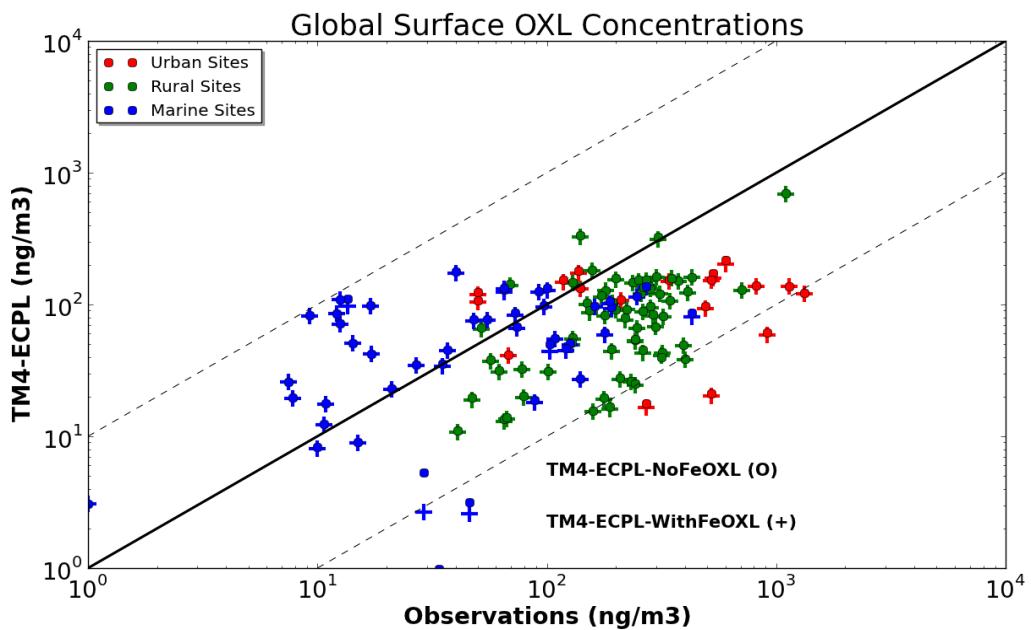


Figure S4. The percentage (%) iron solubility (SFe) predicted by the box model using the same chemical scheme with the global model, for dust concentrations 1 mg L^{-1} (assuming at time = 0 h initial SFe equals to zero), constant pH values of 1.5 (green line), 4.5 (blue line) and 8.5 (red line) and for initial (time = 0 h) oxalic acid (OXL) concentrations of a,b,c) 0 μM , d,e,f) 4.5 μM and g,h,i) 8 μM . Left column shows $\text{SFe} = 100x(\text{Fe(II)} + \text{Fe(III)})/\text{TFe}$, middle column shows $\text{SFe(II)} = 100x\text{Fe(II)}/\text{TFe}$, right column shows $\text{SFe(III)} = 100x\text{Fe(III)}/\text{TFe}$.

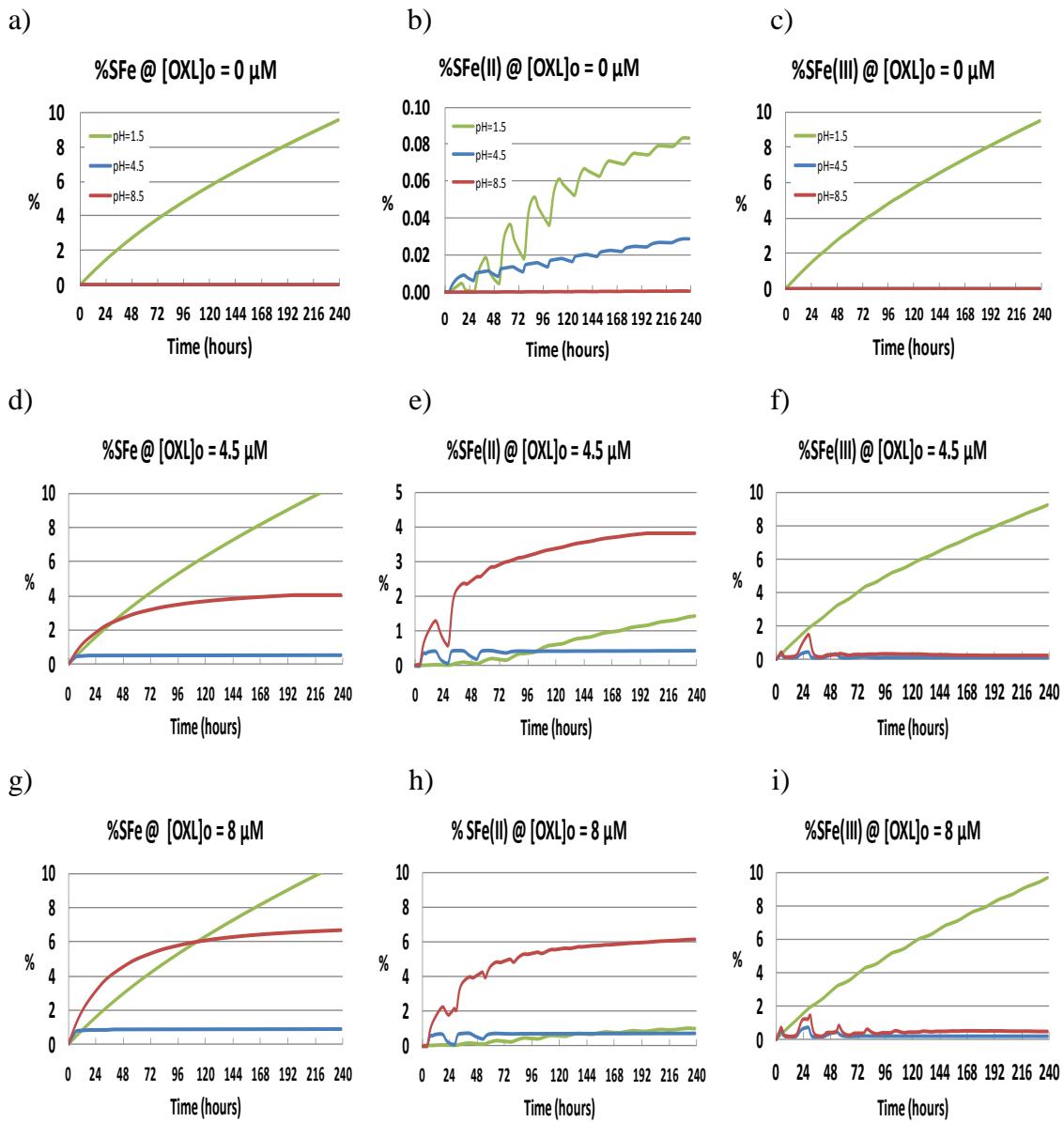


Figure S5. Calculated annual mean concentrations for the present atmosphere (in ng-Fe m⁻³) for Fe(III) (sum of Fe³⁺ and ferric-oxalato complexes) (a,b) and Fe(II) (sum of Fe²⁺ and ferrous-oxalato complexes) (c,d) for surface (left panels; a,c) and zonal mean up to ~200 hPa (right panels; b,d). Note difference in scales.

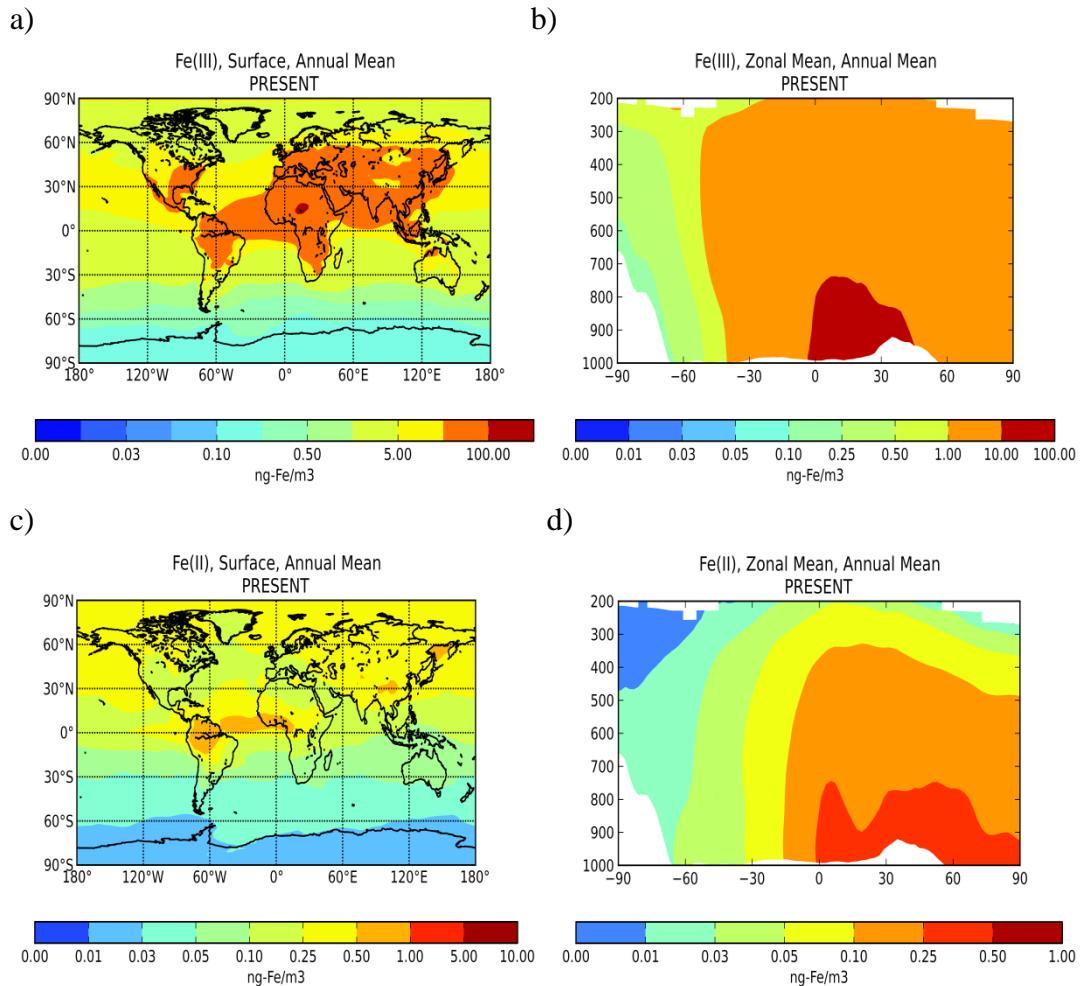


Figure S6. Annual mean Iron Solubility (SFe) calculated for the present atmosphere for a) all Fe sources; b) SFe in dust; c) SFe due to fossil fuel combustion emissions and d) SFe due to biomass burning emissions.

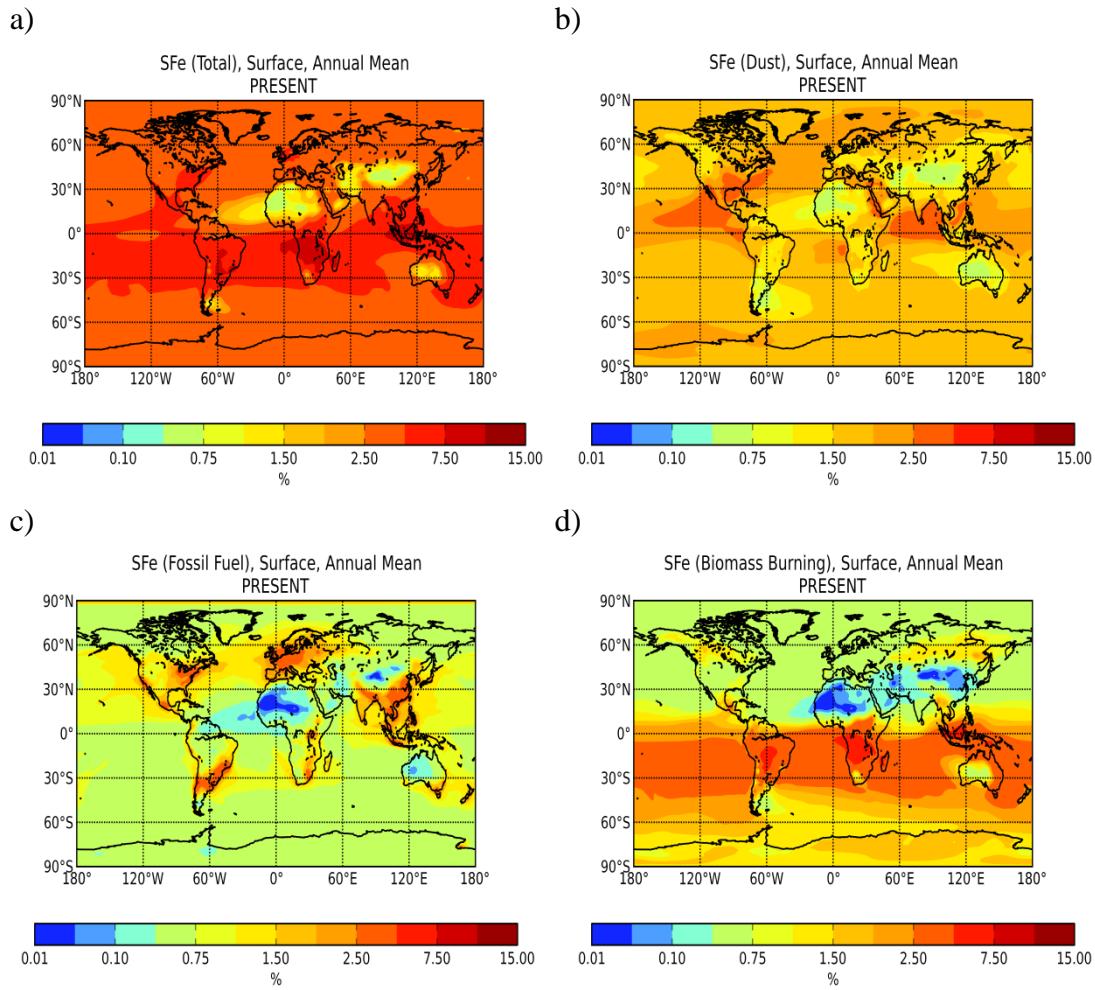


Figure S7. Oceanic regions (red colour) used for a-d) Fig. 7 deposition calculations (as described by Baker et al. 2013) and e) HNLC regions used for the Fig. 9 calculations, see text for definition.

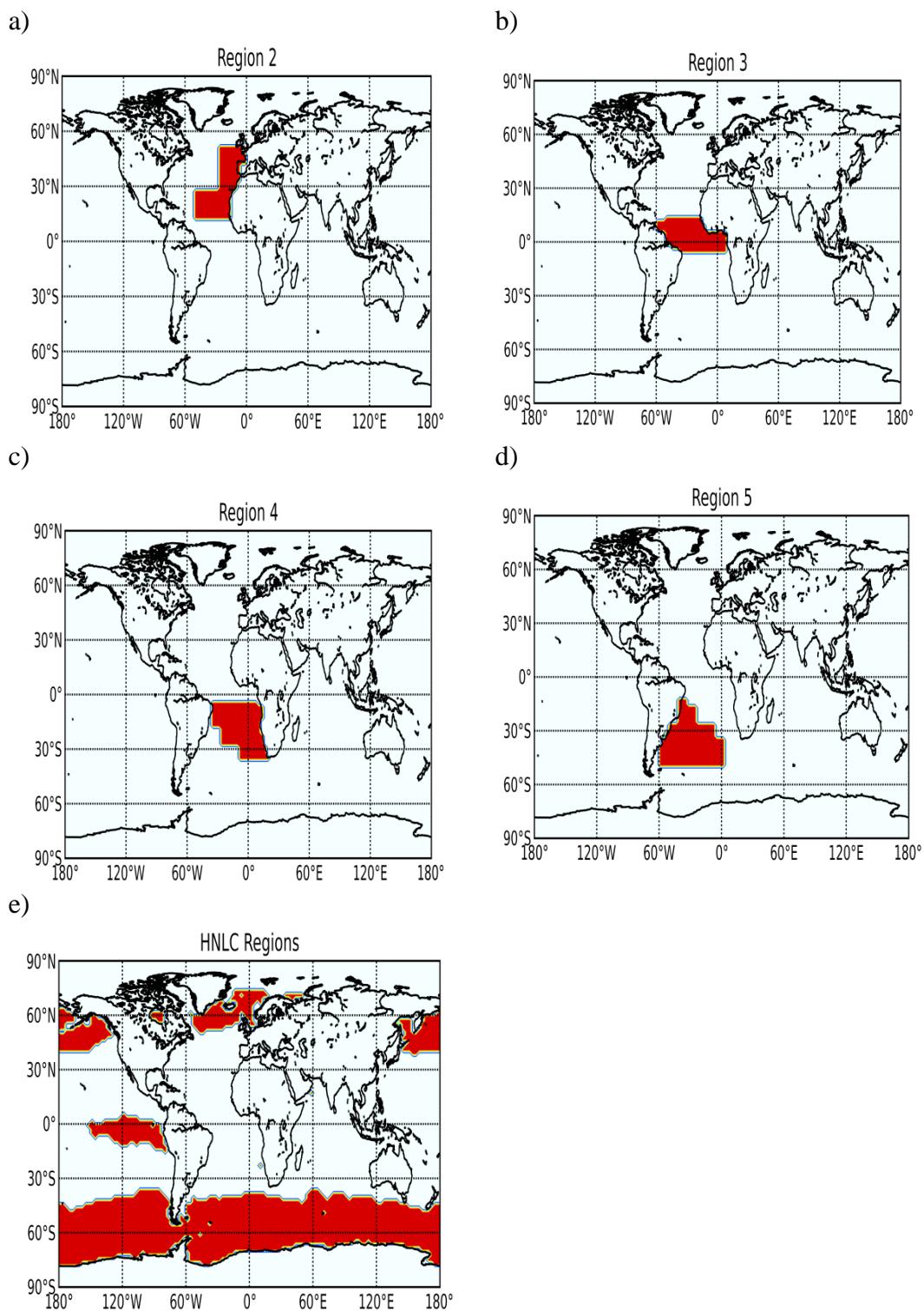
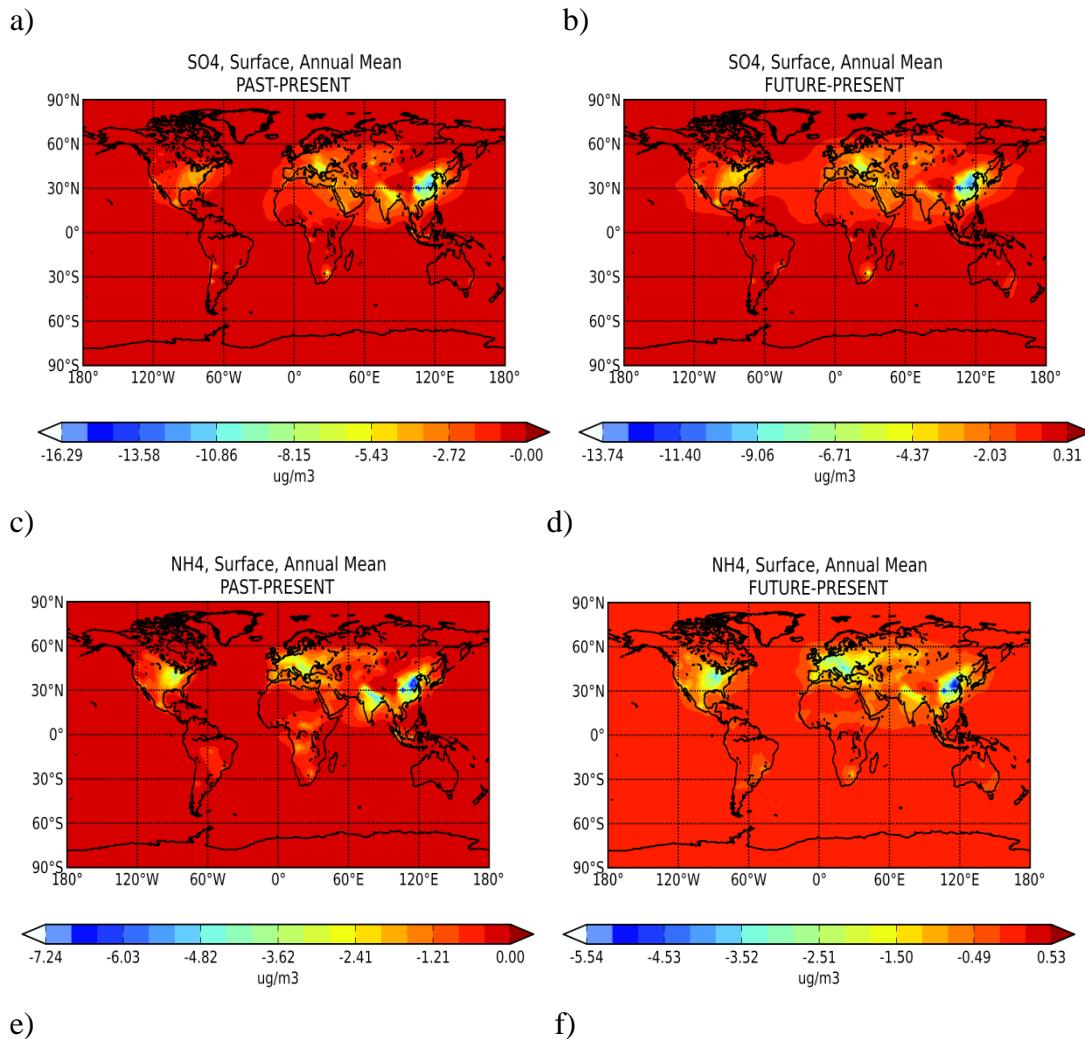
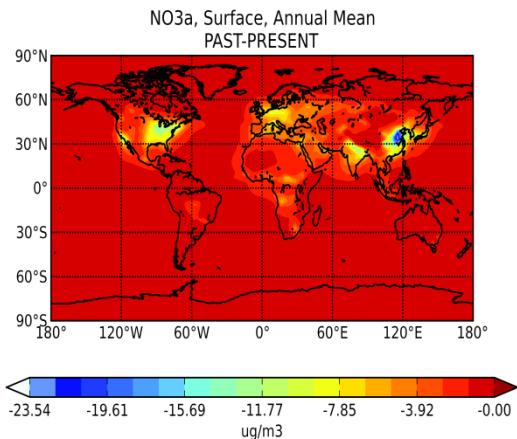
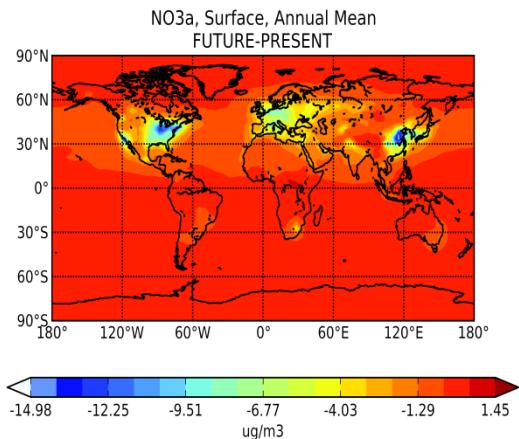


Figure S8. Absolute differences (in $\mu\text{g m}^{-3}$) of surface concentrations calculated by TM4-ECPL using the emissions of PAST (left panels: a, c, e, g) and FUTURE (right panels: b, d, f, h) from those calculated using the present-day emission scenario (PRESENT) for a, b) SO_4^{2-} ; c, d) NH_4^+ ; e, f) NO_3^- ; g, h) for OXL.





g)



h)

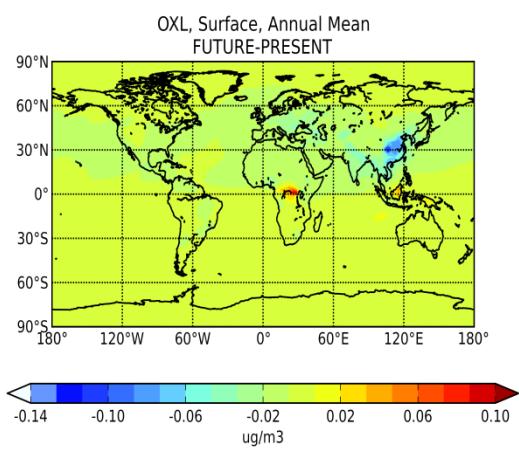
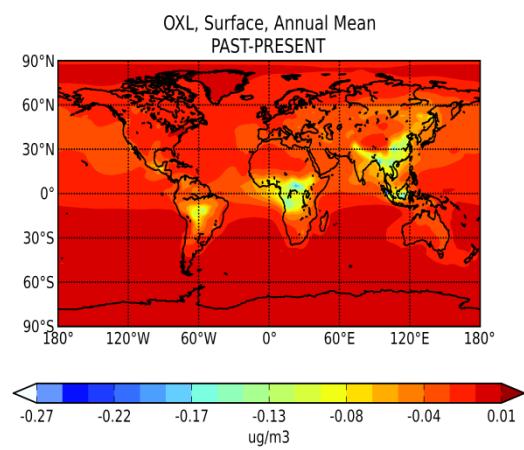
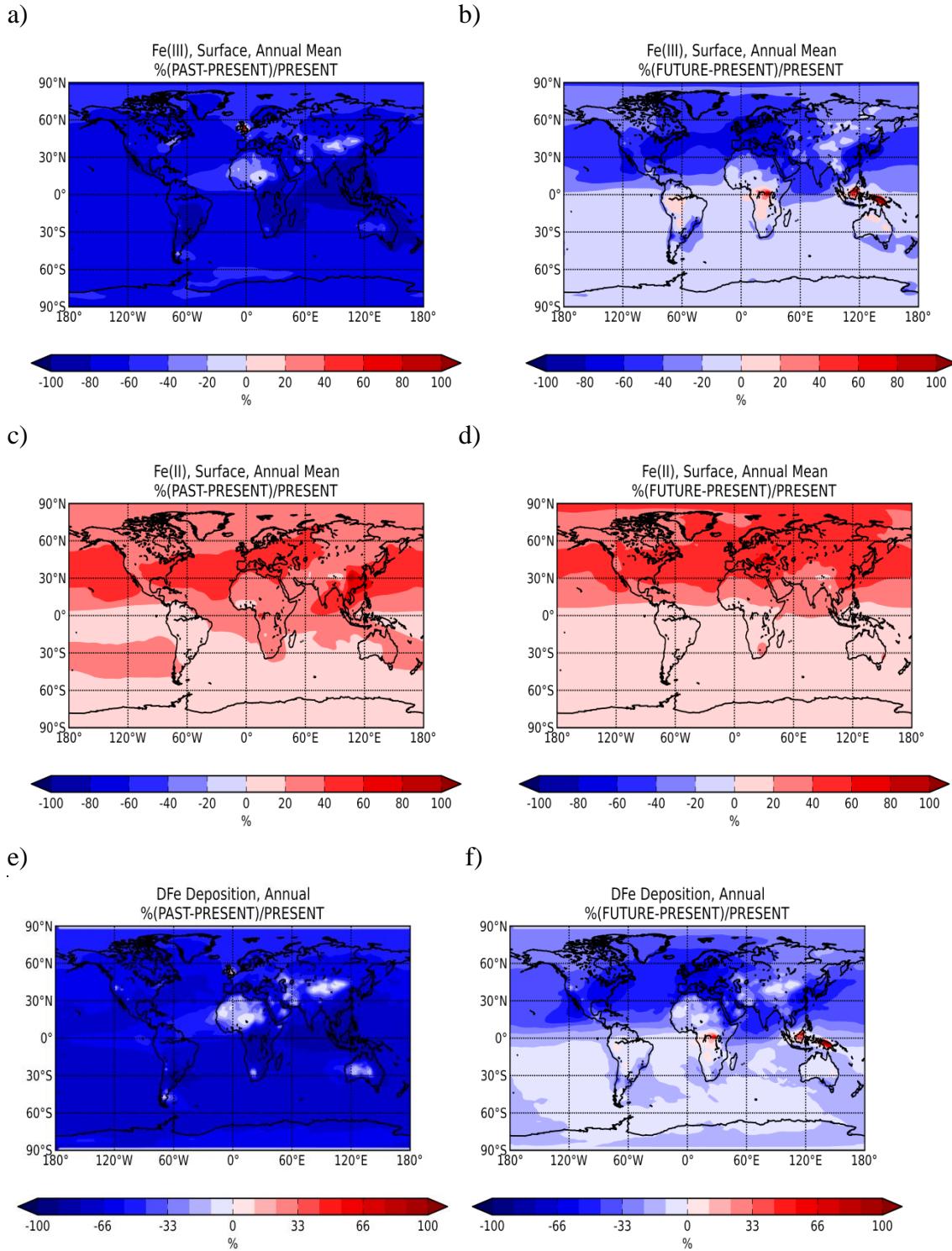


Figure S9. Percentage differences (%) of PAST (left panels: a, c, f) and FUTURE (right panels: b, d, e) simulations from the PRESENT day simulation for annual mean surface concentrations of a, b) Fe(III); c, d) Fe(II) and e, f) DFe deposition as calculated by TM4-ECPL.



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