

Effect of soil saturation on denitrification in a grassland soil

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Abstract. Nitrous oxide (N₂O) is of major importance as a greenhouse gas and precursor of ozone (O_3) destruction in the stratosphere mostly produced in soils. The soil-emitted N₂O is generally predominantly derived from denitrification and, to a smaller extent, nitrification, both processes controlled by environmental factors and their interactions, and are influenced by agricultural management. Soil water content expressed as water-filled pore space (WFPS) is a major controlling factor of emissions and its interaction with compaction, has not been studied at the micropore scale. A laboratory incubation was carried out at different saturation levels for a grassland soil and emissions of N₂O and N₂ were measured as well as the isotopocules of N₂O. We found that flux variability was larger in the less saturated soils probably due to nutrient distribution heterogeneity created from soil cracks and consequently nutrient hot spots. The results agreed with denitrification as the main source of fluxes at the highest saturations, but nitrification could have occurred at the lower saturation, even though moisture was still high (71 % WFSP). The isotopocules data indicated isotopic similarities in the wettest treatments vs. the two drier ones. The results agreed with previous findings where it is clear there are two N pools with different dynamics: added N producing intense denitrification vs. soil N resulting in less isotopic fractionation.

1 Introduction

Nitrous oxide (N_2O) is of major importance as a greenhouse gas and precursor of ozone (O_3) destruction in the stratosphere (Crutzen, 1970). Agriculture is a major source of greenhouse gases (GHGs), such as carbon dioxide (CO₂), methane (CH₄) and also N₂O (IPCC, 2006). The application of organic and inorganic fertiliser N to agricultural soils enhances the production of N₂O (Baggs et al., 2000). This soil-emitted N₂O is predominantly derived from denitrification and to a smaller extent, nitrification in soils (Davidson and Verchot, 2000). Denitrification is a microbial process in which reduction of nitrate (NO_3^-) occurs to produce N_2O_3 , and N2 is the final product of this process, benign for the environment, but represents a loss of N in agricultural systems. Nitrification is an oxidative process in which ammonium (NH_4^+) is converted to NO_3^- (Davidson and Verchot, 2000). Both processes are controlled by environmental factors and their interactions, and are influenced by agricultural management (Firestone and Davidson, 1989). It is well recognised that soil water content expressed as water-filled pore space (WFPS) is a major controlling factor, and as Davidson (1991) illustrated, nitrification is a source of N2O until WFPS values reach about 70 %, after which denitrification dominates. In fact, Firestone and Davidson (1989) gave oxygen supply a ranking of 1 in importance as a controlling factor in fertilised soils, above C and N. At WFPS between 45 and 75 % a mixture of nitrification and denitrification act as N2O sources. Davidson also suggested that at WFPS values above 90% only N₂ is produced. Several studies have later proposed models to relate WFPS with emissions (Schmidt et al., 2000; Dobbie and Smith, 2001; Parton et al., 2001; del Prado et al., 2006; Castellano et al., 2010) but the "optimum" WFPS for N₂O emissions varies from soil to soil (Davidson, 1991). Soil structure could be influencing this effect and it has been identified to strongly interact with soil moisture (Ball et al., 1999; van Groenigen et al., 2005) through changes in WFPS. Particularly soil compaction due to livestock treading and the use of heavy machinery affect soil structure and emissions as reported by studies relating bulk density to fluxes (Klefoth et al., 2014) and degrees of tillage to emissions (Ludwig et al., 2011).

Compaction is known to affect the size of the larger pores (macropores) thereby reducing the soil air volume and therefore increasing the WFPS (for the same moisture content) (van der Weerden et al., 2012). However, little is known about the effect of compaction on the smaller soil pores (micropores), and this could provide valuable information for understanding the simultaneous behaviour of the dynamics of water in the various pore sizes in soil. Such an understanding would lead to the development of better N₂O mitigation strategies via dealing with soil compaction issues.

The role of water in soils is closely linked to microbial activity but also relates to the degree of aeration and gas diffusivity in soils (Morley and Baggs, 2010). Water facilitates nutrient supply to microbes and restricts gas diffusion, thereby increasing the residence time of gases in soil, and the chance of further N2O reduction before it can be released to the atmosphere. This is further aided by the restriction of the diffusion of atmospheric O₂ (Dobbie and Smith, 2001), increasing the potential for denitrification. In consequence, counteracting effects (high microbial activity vs. low diffusion) occur simultaneously, making it difficult to predict net processes and corresponding outputs (Davidson, 1991). Detailed understanding of the sources of N2O and the influence of physical factors, i.e. soil structure and its interaction with moisture, is a powerful basis for developing effective mitigation strategies.

Isotopocules of N₂O represent the isotopic substitution of the O and/or the two N atoms within the N₂O molecule. The isotopomers of N₂O, are those differing in the peripheral (β) and central N positions (α) of the linear molecule (Toyoda and Yoshida, 1999) with the intramolecular ¹⁵N site preference (SP; the difference between $\delta^{15}N^{\alpha}$ and $\delta^{15}N^{\beta}$) used to identify production processes at the level of microbial species or enzymes involved (Toyoda et al., 2005; Ostrom and Ostrom, 2011). Moreover, δ^{18} O, δ^{15} N and SP of emitted N₂O depend on the denitrification product ratio (N₂O / (N₂+N₂O)) and hence provide insight into the dynamics of N₂O reduction (Well and Flessa, 2009; Lewicka-Szczebak et al., 2014, 2015). Koster et al. (2013), for example, recently reported δ^{15} N^{bulk} values of N₂O between -36.8 and -31.9% under the conditions of their experiment, which are indicative of denitrification according to Perez et al. (2006) and Well and Flessa (2009), who proposed the range -54 to -10% relative to the substrate. Baggs (2008) summarised that values between -90 and -40% are indicative of nitrification. Determination of these values is normally carried out in pure culture studies or in conditions favouring either production or reduction of N₂O (Well and Flessa, 2009). The SP is, however, considered a better predictor of the N₂O source due to its independence from the substrate signature (Ostrom and Ostrom, 2011).

Simultaneous occurrence, production and reduction of N₂O as in natural conditions present a challenge for isotopic factors determination due to uncertainty on N2O reduction and the co-existence of different microbial communities producing N₂O (Lewicka-Szczebak et al., 2014). Recently, using data from the experiment reported here, where soil was incubated under aerobic atmosphere and the complete denitrification process occurs, Lewicka-Szczebak et al. (2015) determined fractionation factors associated with N₂O production and reduction using a modelling approach. The analysis comprised measurements of the N2O and N2 fluxes combined with isotopocule data. Net isotope effects (η values) are variable to a certain extent as they result from a combination of several processes causing isotopic fractionation (Well et al., 2012). The results generally confirmed the range of values of η (net isotope effects) and $\eta^{18}O / \eta^{15}N$ ratios reported by previous studies for N₂O reduction for that part of the soil volume were denitrification was enhanced by the N+C amendment. This did not apply for the other part of the soil volume not reached by the N+C amendment, showing that the validity of published net isotope effects for soil conditions with low denitrification activity still needs to be evaluated.

Lewicka-Szczebak et al. (2015) observed a clear relationship between ¹⁵N and ¹⁸O isotope effects during N₂O production and denitrification rates. For N₂O reduction, differential isotope effects were observed for two distinct soil pools characterised by different product ratios $N_2O / (N_2 + N_2O)$. For moderate product ratios (from 0.1 to 1.0) the range of isotope effects given by previous studies was confirmed and refined, whereas for very low product ratios (below 0.1) the net isotope effects were much smaller. In this paper, we present the results from the gas emissions measurements from soils collected from a long-term permanent grassland soil to assess the impact of different levels of soil saturation on N2O and N2 and CO2 emissions after compaction. CO₂ emissions were measured in addition as an estimate of aerobic respiration and thus of O2 consumption, which indicates denitrification is promoted. The measurements included the soil isotopomer $({}^{15}N_{\alpha}, {}^{15}N_{\beta}$ and site preference) analysis of emitted N₂O, which in combination with the bulk ¹⁵N and ¹⁸O was used to distinguish between N₂O from bacterial denitrification and other processes (e.g. nitrification and fungal denitrification) (Lewicka-Szczebak et al., 2017).

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We conducted measurements at defined saturation of pores size fractions as a prerequisite to model denitrification as a function of water status (Butterbach Bahl et al., 2013; Müller and Clough, 2014). We have under controlled conditions created a single compaction stress of 200 kPa (typical of soils compacted after grazing) in incremental layers using a uniaxial pneumatic piston to simulate a grazing pressure. We hypothesised that at high water saturation, spatial heterogeneity of N emissions decreases due to more homogeneous distribution of the soil nutrients and/or anaerobic microsites. We also hypothesised that even at high soil moisture a mixture of nitrification and denitrification can occur. We base this on the creation of pockets of aerobicity as well of anaerobicity at high soil moisture, mainly driven by soil respiration after application of N and C (using up O₂) and further recovery after nutrients are used becoming limiting (increasing aeration). We also aimed to assess how these effects (spatial heterogeneity and source processes) occur in a relatively narrow range of moisture (70–100%). As far as we know there no other studies going to this level of detail. They mostly rely on the knowledge of the effect of moisture on soil processes, whilst in our study we combined direct measurements of both N₂O and N₂ with isotopomers of N₂O to verify the source processes. In addition, the packing of the cores in our study was of great precision, increasing our potential to achieve reproducibility in the replicates where a mixture of aerobic/anaerobic pores might have occurred. We aimed to understand changes in the ratio $N_2O / (N_2O + N_2)$ at the different moisture levels studied in a controlled manner on soil micro- and macropores. The N2 emissions were based on direct measurements from the incubated soils, avoiding methodologies that rely on inhibitors such as acetylene with limitations in diffusion in soil and causing oxidation of NO (Nadeem et al., 2013). Moreover, we used isotopocule values of N2O to evaluate whether the contribution of bacterial denitrification to the total N2O flux was affected by moisture status.

2 Materials and methods

2.1 Soil used in the study

An agricultural soil, under grassland management since at least 1838 (Barré et al., 2010), was collected from a location adjacent to a long-term ley-arable experiment at Rothamsted Research in Hertfordshire (Highfield; see soil properties in Table 1 and further details in Rothamsted Research, 2006; Gregory et al., 2010). The mixed sward is dominated by *Lolium* and *Trifolium* species and is cut two–three times a year. The soil was sampled as described in Gregory et al. (2010). Briefly, it was sampled from the upper 150 mm of the profile, air-dried in the laboratory, crumbled and sieved (< 4 mm), mixed to make a bulk sample, and equilibrated at

a pre-determined water content (37 g 100 g^{-1} ; Gregory et al., 2010) in airtight containers at 4 °C for at least 48 h.

2.2 Preparation of soil blocks

The equilibrated soil was then packed into 12 stainless-steel blocks (145 mm diameter; h: 100 mm), each of which contained three cylindrical holes (i.d: 50 mm; h: 100 mm each). The cores were packed to a single compaction stress of 200 kPa in incremental layers using a uniaxial pneumatic piston. The three hole blocks were used to facilitate the compression of the cores. The 200 kPa stress was analogous to a severe compaction event by a tractor (Gregory et al., 2010) or livestock (Scholefield et al., 1985). The total area of the upper surface of soil in each block was therefore 58.9 cm² $(3 \times 19.6 \text{ cm}^2)$ and the target volume of soil was set to be 544.28 cm^3 (3 × 181.43 cm³) with the objective of leaving a headspace of approximately 45 cm^3 (3 × 15 cm³) for the subsequent experiment. The precise height of the soil (and hence the volume) was measured using the displacement measurement system of a DN10 Test Frame (Davenport-Nene, Wigston, Leicester, UK) with a precision of 0.001 mm.

2.3 Equilibration of soil cores at different saturations

The soil was equilibrated to four different initial saturation conditions or treatments (t0) which were based on the likely distribution of water between macropores and micropores. The first treatment was where both the macro- and micropores (and hence the total soil) were fully saturated; the second treatment was where the macropores were half-saturated and the micropores remained fully saturated; the third treatment was where the macropores were fully unsaturated and the micropores again remained fully saturated; and the fourth treatment was where the macropores were fully unsaturated and the micropores were half-saturated. These four treatments are hereafter referred to as SAT/sat, HALFSAT/sat, UNSAT/sat and UNSAT/halfsat, respectively, where upper-case refers to the saturation condition of the macropores and lower-case refers to the saturation condition of the micropores. In order to set these initial saturation conditions, we referred to the gravimetric soil water release characteristic for the soil, as given in Gregory et al. (2010) (see Supplement). To achieve target water contents during the incubation, the amount of liquid added with the C / N amendment (15 mL) was considered in the total volume of water added. For the SAT/sat and HALFSAT/sat conditions, two sets of three replicate blocks were placed on two fine-grade sand tension tables connected to a water reservoir. For the UNSAT/sat condition a set of three replicate blocks was placed on a tension plate connected to a water reservoir, and the final set of three replicate blocks was placed in pressure plate chambers connected to high-pressure air. All blocks were saturated on their respective apparatus for 24 h, and were then equilibrated for 7 days at the adjusted target matric potentials which were achieved Table 1. Highfield soil properties.

Property	Units	Highfield
Location		Rothamsted Research Herts.
Grid reference	GB National Grid	TL129130
	Longitude	00°21′48″ W
	Latitude	51°48′18″ N
Soil type	SSEW ^a group ^c	Palaeo-argillic brown earth
	SSEW ^a series ^d	Batcombe
	FAO ^{bc}	Chromic Luvisol
Land use		Grass; unfertilised; cut
pH		5.63
Sand (2000–63 µm)	g g ⁻¹ dry soil	0.179
Silt (63–2 µm)	$g g^{-1} dry soil$	0.487
Clay (< $2 \mu m$)	$g g^{-1} dry soil$	0.333
Texture	SSEW ^a class ^c	Silty clay loam
Particle density	$\rm gcm^{-3}$	2.436
Organic matter	$g g^{-1} dry soil$	0.089
Water content for packing	$g g^{-1} dry soil$	0.37

^a Soil Survey of England and Wales classification system. ^b United Nations Food and Agriculture Organization World Reference Base for Soil Resources classification system (approximation). ^c Avery (1980). ^d Clayden and Hollis (1984).

by either lowering the water level in the reservoir (sand tables and tension plate) or by increasing the air pressure (pressure chambers). At the end of equilibration period, the blocks were removed carefully from the apparatus, wrapped in airtight film, and maintained at 4 °C until the subsequent incubation.

2.4 Incubation

The study was carried out under controlled laboratory conditions, using a specialised laboratory denitrification (DENIS) incubation system (Cardenas et al., 2003). Each block containing three cores was placed in an individual incubation vessel of the automated laboratory system in a randomised block design to avoid effect of vessel. The lids for the vessels containing three holes were lined with the cores in the block to ensure that the solution to be applied later would fall on top of each soil core. Stainless steel bulkheads fitted (size for 1/4 in. tubing) on the lids had a three-layered Teflon coated silicone septum (4 mm thick \times 7 mm diameter) for supplying the amendment solution by using a gas tight hypodermic syringe. The bulkheads were covered with a stainless-steel nut and only open when amendment was applied. The incubation experiment lasted 13 days from the time the cores started to be flushed until the end of the incubation. The incubation vessels with the soils were contained in a temperature controlled cabinet and the temperature set at 20 °C. The incubation vessels were flushed from the bottom at a rate of 30 mL min^{-1} with a He / O₂ mixture (21 % O₂, natural atmospheric concentration) for 24 h, or until the system and the soil atmosphere were emitting low background levels of both N_2 and N_2O (N_2 can get down to levels of 280 ppm much smaller than atmospheric values). Subsequently, the He $/ O_2$ supply was reduced to 10 mL min^{-1} and directed across the soil surface and measurements of N₂O and N₂ carried out at approximately 2-hourly cycles to sample from all the 12 vessels. Emissions of CO₂ were simultaneously measured.

2.5 Application of amendment

An amendment solution equivalent to 75 kg N ha^{-1} and 400 kg C ha⁻¹ was applied as a 5 mL aliquot a solution containing KNO₃ and glucose to each of the three cores in each vessel on day 0 of the incubation. Glucose is added to optimise conditions for denitrification to occur (Morley and Baggs, 2010). The aliquot was placed in a stainless-steel container (volume 1.2 L) which had three holes drilled with bulkheads fitted: two to connect stainless-steel tubing for flushing the vessel and the third one to place a septum on a bulkhead to withdraw solution. Flushing was carried out with He for half an hour before the solution was required for application to the soil cores and continued during the application process to avoid atmospheric N₂ contamination (a total of 1.5 h). The amendment solution was manually withdrawn from the container with a glass syringe fitted with a three-way valve onto the soil surface; care was taken to minimise contamination from atmospheric N₂ entering the system. The syringe content was injected to the soil cores via the inlets on the lids consecutively in each lid (three cores) and all vessels, completing a total of 36 applications that lasted about 45 min. Incubation continued for 12 days, and the evolution of N₂O, N₂ and CO₂ was measured continuously. At the end of each incubation experiment, the soils were removed from the incubation vessels for further analysis. The three cores in each incubation vessel were pooled in one sample and subsamples

taken and analysed for mineral N, total N and C, and moisture status.

2.6 Gas measurements

Gas samples were directed to the relevant analysers via an automated injection valve fitted with two loops to direct the sample to two gas chromatographs. Emissions of N_2O and CO₂ were measured by gas chromatographs (GCs), fitted with an electron capture detector (ECD) and separation achieved by a stainless-steel packed column (2 m long, 4 mm bore) filled with "Porapak Q" (80-100 mesh) and using N2 as the carrier gas. The detection limit for N₂O was equivalent to $2.3 \text{ g N} \text{ ha}^{-1} \text{ d}^{-1}$. The N₂ was measured by GC with a He ionisation detector (HID) and separation was achieved by a PLOT column (30 m long 0.53 mm i.d.), with He as the carrier gas. The detection limit was $9.6 \text{ g N} \text{ ha}^{-1} \text{ d}^{-1}$. The response of the two GCs was assessed by measuring a range of concentrations for N2O, CO2 and N2. Parent standards of the mixtures $10\,133\,\text{ppm}\,\text{N}_2\text{O} + 1015.8\,\text{ppm}\,\text{N}_2$, 501 ppm $N_2O + 253 \text{ ppm } N_2$, and 49.5 ppm $N_2O + 100.6 \text{ ppm } N_2$ were diluted by means of mass flow controllers with He to give a range of concentrations of up to 750 ppm for N_2O and 1015 ppm for N₂. For CO₂, a parent standard of 30 100 ppm was diluted down to 1136 ppm (all standards were in He as the balance gas). Daily calibrations were carried out for N₂O and N₂ by using the low standard and doing repeated measurements. The temperature inside the refrigeration cabinet containing the incubation vessels was logged on an hourly basis and checked at the end of the incubation. The gas outflow rates were also measured and recorded daily, and subsequently used to calculate the flux.

2.7 Measurement of N₂O isotopic signatures

Gas samples for isotopocule analysis were collected in 115 mL serum bottles sealed with grey butyl crimp-cap septa (part no. 611012, Altmann, Holzkirchen, Germany). The bottles were connected by a Teflon tube to the end of the chamber vents and were vented to the atmosphere through a needle in order to maintain flow through the experimental system. Dual isotope and isotopocule signatures of N₂O, i.e. δ^{18} O of N₂O (δ^{18} O-N₂O), average δ^{15} N (δ^{15} N^{bulk}) and δ^{15} N from the central N position ($\delta^{15}N^{\alpha}$) were analysed after cryofocussing by isotope ratio mass spectrometry as described previously (Well et al., 2008). ¹⁵N site preference (SP) was obtained as SP = $2 \times (\delta^{15} N^{\alpha} - \delta^{15} N^{bulk})$. Dual isotope and isotopocule ratios of a sample (R_{sample}) were expressed as per mille deviation from ${}^{15}N/{}^{14}N$ and ${}^{18}O/{}^{16}O$ ratios of the reference standard materials (R_{std}), atmospheric N₂ and standard mean ocean water (SMOW), respectively:

$$\delta X = (R_{\text{sample}}/R_{\text{std}} - 1) \times 1000, \qquad (1$$

)

where $X = {}^{15}N^{\text{bulk}}, {}^{15}N^{\alpha}, {}^{15}N^{\beta}$, or ${}^{18}O$.

2.8 Data analysis and additional measurements undertaken

The areas under the curves for the N₂O, CO₂ and N₂ data were calculated by using GenStat 11 (VSN International Ltd, Hemel Hempstead, Herts, UK). The resulting areas for the different treatments were analysed by applying analysis of variance (ANOVA). The isotopic ($^{15}N^{bulk}$, ^{18}O), and site preference (SP) differences between the four treatment for the different sampling dates were analysed by two-way ANOVA. We also used the Student's *t* test to check for changes in soil water content over the course of the experiments.

Calculation of the relative contribution of the N₂O derived from bacterial denitrification ($\% B_{\text{DEN}}$) was done according to Lewicka-Szczebak et al. (2015). The isotopic value of initially produced N₂O, i.e. prior to its partial reduction (δ_0) , was determined using a Rayleigh model (Mariotti et al., 1982), were δ_0 is calculated using the fractionation factor of N₂O reduction ($\eta_{N_2O-N_2}$) for SP and the fraction of residual N₂O (r_{N_2O}) which is equal to the N₂O / (N₂+N₂O) product ratio obtained from direct measurements of N2 and N2O flux. An endmember mixing model was then used to calculate the percentage of bacterial N₂O in the total N₂O flux (% B_{DEN}) from calculated δ_0 values and the SP and δ^{18} O endmember values of bacterial denitrification and fungal denitrification/nitrification. The range in endmember and $\eta_{N_2Q-N_2}$ values assumed (adopted from Lewicka-Szczebak, 2017) to calculated maximum and minimum estimates of $\% B_{\text{DEN}}$ is given in Table 4. We also fitted three functions through these data (SP vs. $N_2O / (N_2 + N_2O)$) including a second-degree polynomial and a linear and logarithmic function.

Because both, endmember values and $\eta_{N_2O-N_2}$ values are not constant but subject to the given ranges, we calculated here several scenarios using combinations of maximum, minimum and average endmember and $\eta_{N_2O-N_2}$ values (Table 4) to illustrate the possible range of $\% B_{\text{DEN}}$ for each sample. For occasional cases where $\% B_{\text{DEN}} > 100\%$ the values were set to 100 %.

At the same time as preparing the main soil blocks, a set of replicate samples was prepared in exactly the same manner, but in smaller cores (i.d: 50 mm; h: 25 mm). On these samples, we analysed soil mineral N, total N and C and moisture at the start of the incubation. The same parameters were measured after incubation by doing destructive sampling from the cores. Mineral N (NO_3^- , NO_2^- and NH_4^+) was analysed after extraction with KCl by means of a segmented flow analyser using a colorimetric technique (Searle, 1984). Total C and N in the air-dried soil were determined using a thermal conductivity detector (TCD, Carlo Erba, model NA2000). Soil moisture was determined by gravimetric analysis after drying at 105 °C.

3 Results

3.1 Soil composition

The results after moisture adjustment at the start of the experiment resulted in a range of WFPS of 100 to 71 % for the four treatments (Table 2). The results from the end of the incubation also confirmed that there remained significant differences in soil moisture between the high-moisture treatments (SAT/sat and HALFSAT/sat) and the two lowermoisture treatments (Table 3; one-way ANOVA, p < 0.05). Soils in the two wettest states lost statistically significant amounts of water (10 % (p = 0.006) and 4.4 % (p < 0.001) for SAT/sat and HALFSAT/sat, respectively) over the course of the 13-day incubation experiment. This was inevitable as there was no way to hold a high (near-saturation) matric potential once the soil was inside the DENIS assembly, and water would have begun to drain by gravitational forces out of the largest macropores (> $30 \,\mu$ m). An additional factor was the continuous He/O2 delivery over the soil surface which would have caused some drying. We accepted these as unavoidable features of the experimental set-up, but we assume that the main response of the gaseous emissions occurred under the initial conditions, prior to the loss of water over subsequent days. Soil in the two drier conditions had no significant change in their water content over the experimental period (p = 0.153 and 0.051 for UNSAT/sat and UN-SAT/halfsat, respectively). The results of the initial soil composition for mineral N were $85.5 \text{ mg NO}_3^- \text{-N kg}^{-1}$ dry soil and 136.2 mg NH_4^+ -N kg⁻¹ dry soil. The mineral N contents of the soils at the end of the incubation are reported in Table 3, showing that NO₃⁻ was very low in treatments SAT/sat and HALFSAT/sat (~ 1 mg N kg^{-1} dry soil) compared to UN-SAT/sat and UNSAT/halfsat (50–100 mg N kg⁻¹ dry soil) at the end of the incubation. Therefore, there was a significant difference in soil NO₃ between the former highmoisture treatments and the latter drier (UNSAT) treatments, which were also significantly different between themselves (p < 0.001 for both). The NH₄⁺ content was similar in treatments SAT/sat, HALFSAT/sat and UNSAT/sat $(\sim 100 \,\mathrm{mg}\,\mathrm{N}\,\mathrm{kg}^{-1}\,\mathrm{dry}\,\mathrm{soil})$, but slightly lower in treatment UNSAT/halfsat (71.3 mg N kg⁻¹ dry soil); however, overall differences were not significant, probably due to the large variability on the driest treatment (p > 0.05).

3.2 Gaseous emissions of N₂O, CO₂ and N₂

All datasets of N₂O and N₂ emissions showed normal distribution (Fpr. < 0.001). The treatments SAT/sat and HALF-SAT/sat for all three gases (N₂O, CO₂ and N₂) showed fluxes that were well replicated for all the vessels (see Fig. 1); in contrast, for UNSAT/sat and UNSAT/halfsat the emissions between the various replicated vessel in each treatment was not as consistent, leading to a larger within-treatment variability in the magnitude and shape of the GHG fluxes measured. The cumulative fluxes also resulted in larger variability for the drier treatments (Table 3).

3.2.1 Nitrous oxide and nitrogen gas

The general trend was that the N₂O concentrations in the headspace increased shortly after the application of the amendment (Fig. 1). The duration of the N₂O peak for each replicate soil samples was about 3 days, except for UN-SAT/halfsat, in which one of the replicate soils exhibits a peak which lasted for about 5 days. The N₂O maximum in the SAT/sat and HALFSAT/sat treatments was of similar magnitude (means of 5.5 and $6.5 \text{ kg N ha}^{-1} \text{ d}^{-1}$, respectively) but not those of UNSAT/sat and UNSAT/halfsat (means of 7.1 and 11.9 kg N ha⁻¹ d⁻¹, respectively). The N₂ concentrations always increased before the soil-emitted N2O reached the maximum. The lag between both N₂O and N₂ peak for all samples was only a few hours. Peaks of N₂ generally lasted just over 4 days, except in UNSAT/halfsat, where one replicate lasted about 6 days (Fig. 1). Unlike in the N₂O data, there was larger within treatment variability in the replicates for all four treatments. The standard deviations of each mean (Table 3) also indicate the large variability in treatments UNSAT/sat and UNSAT/halfsat for both N2O and N₂.

The product ratios, i.e. $N_2O / (N_2O + N_2)$, resulted in a peak just after amendment addition by ca. 0.73 (at 0.49 days), 0.65 (at 0.48 days), 0.99 (at 0.35 days) and 0.88 (at 0.42 days) for SAT/sat, HALFSAT/sat, UNSAT/sat and UNSAT/halfsat, respectively, and then decreases gradually until day 3, where it becomes nearly zero for the two wettest treatments and stays stable for the driest treatments between 0.1 and 0.2 (see Table 5, where the daily means of these ratios are presented).

The cumulative areas of the N₂O and N₂ peaks analysed by one-way ANOVA resulted in no significant differences between treatments for both N₂O and N₂ (Table 3). Due to the large variation in treatments UNSAT/sat and UNSAT/halfsat we carried out a pair wise analysis by using a weighted *t* test (Cochran and Cox, 1957). This analysis resulted in treatment differences between SAT/sat and HALFSAT/sat, HALFSAT/sat and UNSAT/sat, and SAT/sat and UNSAT/sat, but only at the 10 % significance level (P < 0.1 for both N₂O and N₂).

The results showed that total N emission $(N_2O + N_2)$ (Table 3) decreased between the highest and lowest soil moistures, i.e. from 63.4 for SAT/sat (100% WFPS) to 34.1 kg N ha⁻¹ (71% WFPS) for UNSAT/halfsat. The maximum cumulative N₂O occurred at around 80% WFPS (Fig. 2), whereas the total N₂O + N₂ was largest at about 95% and for N₂ it was our upper treatment at 100% WFPS.

3.2.2 Carbon dioxide

The background CO_2 fluxes (before amendment application, i.e. day -1 to day 0) were high at around $30 \text{ kg C ha}^{-1} \text{ d}^{-1}$

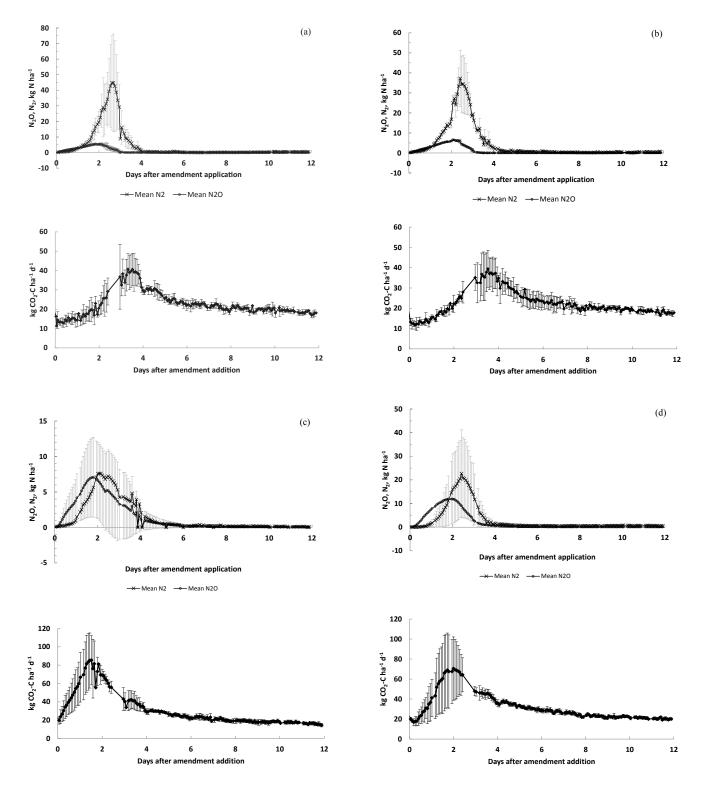


Figure 1. Mean of the three replicates for N_2O , N_2 and CO_2 emissions from (a) SAT/sat treatment, (b) HALFSAT/sat, (c) UNSAT/sat and (d) UNSAT/halfsat. Grey lines correspond to the standard error of the means (in c and d only errors for the N_2 are shown to avoid overlapping of bars).

Table 2.	The f	our	saturation	conditions	set for th	e Highfield soil.

Saturation condition	SAT/sat	HALFSAT/sat	UNSAT/sat	UNSAT/halfsat
Macropores	Saturated	Half-saturated	Unsaturated	Unsaturated
Micropores	Saturated	Saturated	Saturated	Half-saturated
As prepared:				
Matric potential, kPa	4.1	12.3	27.3	136.9
Water content, g 100 g^{-1}	47.7	42.5	37.2	29.4
Water content, $cm^{-3} 100 cm^{-3}$	61.1	54.4	47.7	37.3
Water-filled pore space, %	98	91	82	68
Threshold pore size saturated, µm	73	24	11	2
Final, following amendment:				
Matric potential, kPa	0	8.6	20.0	78.1
Water content, g 100 g^{-1}	49.8	44.6	39.3	31.5
Water content, cm^{-3} 100 cm^{-3}	63.8	57.1	50.4	40.0
Water-filled pore space, %	100	94	85	71
Threshold pore size saturated, µm	all	35	15	4

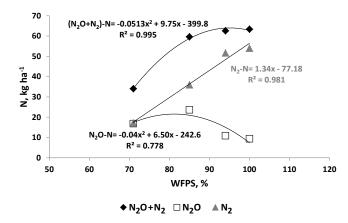


Figure 2. Total N emissions $(N_2O + N_2)$ -N, N_2O and N_2 vs. WFPS. Fitted functions through each dataset are also shown.

and variable (not shown). The CO₂ concentrations in the headspace increased within a few hours after amendment application. The maximum CO₂ flux was reached earlier in the drier treatments (about 1–2 days; ~70 kg Cha⁻¹ d⁻¹) compared to the wettest (3 days; ~40 kg Cha⁻¹ d⁻¹) and former peaks were also sharper (Fig. 1). The cumulative CO₂ fluxes were significantly larger in the two drier unsaturated treatments (ca. 400–420 kg Cha⁻¹) when compared to the wetter more saturated treatment (ca. 280–290 kg Cha⁻¹, P < 0.05) (Table 3).

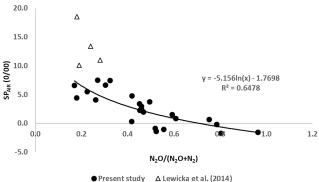
3.2.3 Isotopocules of N₂O

The δ^{15} N^{bulk} of the soil-emitted N₂O in our study differed significantly among the four treatments and between the seven sampling dates (p < 0.001 for both); there was also a significant treatment × sampling date interaction (p < 0.001). The maximum δ^{15} N^{bulk} generally occurred on day 3, except for SAT/sat on day 4 (Table 6).

The maximum δ^{18} O-N₂O values were also found on day 3, except for SAT/sat, which peaked at day 2 (Table 6). Overall, the δ^{18} O-N₂O values varied significantly between treatment and sampling dates (p < 0.001 for both), but there was no significant treatment × time interaction (p > 0.05).

The site preference (SP) for the SAT/sat treatment had an initial maximum value on day 2 (6.3%), which decreased thereafter in the period from day 3 to 5 to mean SP values of the emitted N₂O of 2.0% on day 5, subsequently rising to 8.4 % on day 12 of the experiment (Table 6). The HALF-SAT/sat treatment had the highest initial SP values on day 2 and 3 (both 6.4%), decreasing again to a value of 2.0%, but now on day 4 followed by subsequent higher SP values of up to 9.2 % on day 7 (Table 6). The two driest treatments (UN-SAT/sat and UNSAT/halfsat) both had an initial maximum on day 3 (11.9 and 5.9%, respectively), and in UNSAT/sat the SP value then decreased to day 7 (3.9%). However, in the UNSAT/halfsat treatment, after a marginal decrease on day 4 (5.4%), it then increased throughout the experiment, reaching 11.8% on day 12 (Table 6). The lowest SP values were generally on day 1 in all treatments. Overall, for all parameters, there was more similarity between the more saturated treatments SAT/sat and HALFSAT/sat, and between the two more dry and aerobic treatments UNSAT/sat and UN-SAT/halfsat.

The N₂O / (N₂O + N₂) ratios vs. SP for all treatments in the first 2 days (when N₂O was increasing and the N₂O / (N₂O + N₂) ratio was decreasing) shows a significant negative response of the SP when the ratio increased (Fig. 3). This behaviour suggests that when the emitted gaseous N is dominated by N₂O (ratio close to 1) the SP values will be slightly negative with an intercept of -2% (Fig. 3), i.e. within the SP range of bacterial denitrification. With decreasing N₂O / (N₂O + N₂) ratio the SP values of soil-emitted N₂O were increasing to values up to 8%. This is in juxtaposition with the situation when the N emissions are dominated



• Present study Δ Lewicka et al. (2014)

Figure 3. Ratio $N_2O / (N_2O + N_2)$ vs. site preference (SP) for all for treatments in the first 2 days. A logarithmic function was fitted through the data; the corresponding equation and correlation coefficient are given. Also shown data from Lewicka et al. (2014).

by N_2 , or N_2O is low, where the SP values of soil-emitted N_2O were much higher (Fig. 3), pointing to an overall product ratio related to an "isotopic shift" of 10 to 12.5%. The fitted logarithmic function in Fig. 3 is in almost perfect agreement with Lewicka-Szczebak et al. (2014).

It has been reported that the combination of the isotopic signatures of N₂O potentially identifies the contribution of processes other than bacterial denitrification (Köster et al., 2015; Wu et al., 2016; Deppe et al., 2017). The question arises of to what extent the relationships between the $\delta^{18}O$ and $\delta^{15} N^{\text{bulk}}$ and between $\delta^{18} O$ and SP within the individual treatments denitrification dynamics are related. We checked this to evaluate the robustness of isotope effects during N₂O reduction as a prerequisite to calculate the percentage of bacterial denitrification in N2O production. In our data, maximum δ^{18} O and SP values were generally observed at or near the peak of N₂ emissions on days 2-3, independent of the moisture treatment (Table 6 and Fig. 3). δ^{15} N^{bulk} values of all treatments were mostly negative when N₂O fluxes started to increase (day 1, Fig. 1, Table 6), except for UNSAT/halfsat, in which the lowest value was before amendment application, reaching their highest values between days 3 and 4 for when N₂O fluxes were back to the low initial values, and then decreased during the remaining period. δ^{18} O values increased about 10-20% after day 1, reaching maximum values on days 2 or 3 in all treatments, while SP increased in parallel, at least by 3 % (SAT/sat) and up to 12 % (UNSAT/sat). While δ^{18} O exhibited a steady decreasing trend after day 3, SP behaved oppositely to $\delta^{15}N^{\text{bulk}}$ with decreasing values, while $\delta^{15} N^{\text{bulk}}$ was rising again after days 4 or 5.

We further explored the data by looking at the relationships between the δ^{18} O and δ^{15} N^{bulk} for all the treatments. The δ^{18} O vs. δ^{15} N^{bulk} for all treatments is presented separating the data in three periods (see Fig. 4). There was a strong and significant relationship (P < 0.001 and 0.05, respectively) between δ^{18} O vs. δ^{15} N^{bulk} for the high-moisture treatments ($R^2 = 0.973$ and 0.923 for SAT/sat and HALF- SAT/sat, respectively) at the beginning of the incubation ("1– 2") when the N₂O emissions are still increasing, in contrast to those of the lower soil moisture treatments that were lower and not significant ($R^2 = 0.294$ and 0.622, for UNSAT/sat and UNSAT/halfsat, respectively). The relationships between δ^{18} O vs. δ^{15} N^{bulk} of emitted N₂O for the "3–12" period were significant for SAT/sat and HALFSAT/sat with R^2 values between 0.549 and 0.896 and P values < 0.05 and 0.001, respectively (Fig. 4). Regressions were also significant for this period for the driest treatments (P < 0.001). Interestingly, with decreasing soil moisture content (Fig. 4a to d) the regression lines of "1-2" and "3-12" day period got closer together in the graphs. Overall, the $\delta^{15} N^{\text{bulk}}$ isotopic distances between the two lines was larger for a given δ^{18} O-N₂O value for SAT/sat and HALFSAT/sat (ca. 20%) when compared to the UNSAT/sat and UNSAT/halfsat treatments (ca. 13%) (Fig. 4). Therefore, it seems the $\delta^{15}N^{\text{bulk}} / \delta^{18}O-N_2O$ signatures are more similar for the drier soils than the two wettest treatments. In addition, Fig. 4 exactly reflects the two-pool dynamics with increasing δ^{15} N and δ^{18} O while the product ratio goes down (days 2, 3), then only $\delta^{15}N$ continues increasing due to fractionation of the NO_3^- during exhaustion of pool 1 in the wet soil (days 3, 4, 5). Finally, as pool 1 is depleted and more and more comes from pool 2, the product ratio increases somewhat, and $\delta^{15}N$ decreases somewhat since pool 2 is less fractionated and also δ^{18} O decreases due to slightly increasing product ratio. Note that the turning points of δ^{18} O and product ratio (Tables 3 and 4) for the wetter soils almost coincide.

Similarly to Fig. 4, δ^{18} O vs. the SP (Fig. 5) was analysed for the different phases of the experiment. Generally, the slopes (Table 7) for days 1-2 for the three wettest treatments were similar (\sim 0.2–0.3) following the range of known reduction slopes and also had high and significant (P <0.05) regression coefficients ($R^2 = 0.65$, 0.90 and 0.87 for SAT/sat, HALFSAT/Sat and UNSAT/sat, respectively). The slopes on days 3–5 were variable but slightly similar on days 7-12 (between 41 and 0.68) for the same three treatments. They were only significant for the two driest treatments (P < 0.05). On days 7–12 SAT/sat and UNSAT/sat gave significant correlations (P < 0.001 and 0.05, respectively). Figure 5 also shows the "map" for the values of SP and δ^{18} O from all treatments. Reduction lines (vectors) represent minimum and maximum routes of isotopocules values with increasing N₂O reduction to N₂ based on the reported range in the ratio between the isotope fractionation factors of N2O reduction for SP and δ^{18} O (Lewicka-Szczebak et al. (2017). Most samples are located within the vectors (from Lewicka-Szczebak et al., 2017) area of N₂O production by bacterial denitrification with partial N2O reduction to N2 (within uppermost and lowermost N₂O reduction vectors representing the extreme values for the bacterial endmember and reduction slopes). Only a few values of the UNSAT/sat and UN-SAT/halfsat treatments are located above that vector area and

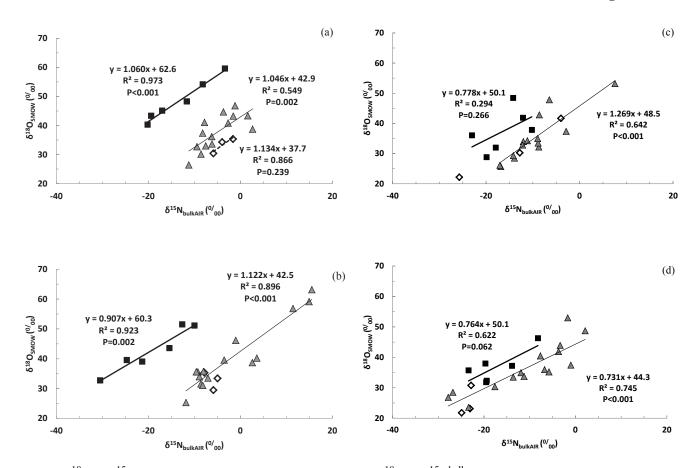


Figure 4. δ^{18} O vs. δ^{15} N_{bulk} in all treatments for three periods: "-1", with δ^{18} O vs. δ^{15} N^{bulk} values 1 day prior to the moisture adjustment (and N and C application); "1–2", with values in the first 2 days after the addition of water, N and C, and N₂O emissions were generally increasing in all treatments; and "3–12", the period in days after moisture adjustment and N and C addition when N₂O emissions generally decreased back to baseline soil emissions (day –1 in diamond symbols, days 1–2 in square symbols and days 3–12 in triangle symbols) in the experiment. (a) SAT/sat treatment, (b) HALFSAT/sat, (c) UNSAT/sat and (d) UNSAT/halfsat. Equations of fitted functions and correlation coefficients are shown. Correlations are unadjusted; the *P* value tests whether the slope is different from zero.

more close or within the vector area of mixing between bacterial denitrification and fungal denitrification/nitrification.

The estimated ranges of the proportion of emitted N₂O resulting from bacterial denitrification ($\% B_{DEN}$) were on day 1 and 2 after the amendment comparable in all four moisture treatments (Table 6). However, during day 3 to 12 the $\% B_{DEN}$ ranged from 78 to 100 % in SAT/sat and 79 to 100 % HALF-SAT/Sat, which was generally higher than that estimated at 54–86 % for UNSAT/halfsat treatment. The $\% B_{DEN}$ of UN-SAT/halfsat in that period was intermediate between SAT/sat and UNSAT/sat with range of range 60–100 % (Table 6). The final values were similar to those on day -1, except for the UNSAT/sat treatment.

4 Discussion

4.1 N₂O and N₂ fluxes

4.1.1 Effect of soil moisture

The observed decrease in total N emissions with decreasing initial soil moisture reflects the effect of soil moisture as reported in previous studies (Well et al., 2006). The differences when comparing the cumulative fluxes, however, were only marginally (p < 0.1) significant (Table 3) mostly due to large variability within replicates in the drier treatments (see Fig. 1b). Davidson et al. (1991) provided a WFPS threshold for determination of source process, with a value of 60 % WFPS as the borderline between nitrification and denitrification as source processes for N₂O production. The WFPS in all treatments in our study was larger than 70 %, above this 60 % threshold, and referred to as the "optimum water content" for N₂O by Scheer et al. (2009), so we can be confident that denitrification was likely to have been the main

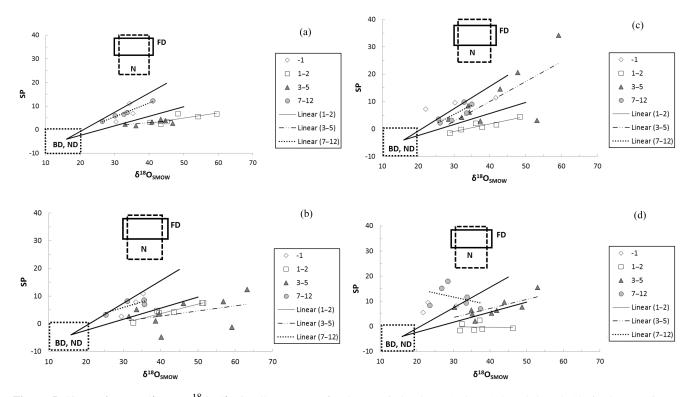


Figure 5. Site preference (‰) vs. δ^{18} O (‰) in all treatments for three periods (day -1, days 1-2 and days 3-12) in the experiment: (a) SAT/sat treatment; (b) HALFSAT/sat; (c) UNSAT/sat; (d) UNSAT/halfsat. Equations of fitted functions and correlation coefficients are in Table 7 for 1-2, 3-5 and 7-12 (5-12 for c). Endmember areas for nitrification, N; bacterial denitrification, D; fungal denitrification, FD; and nitrifier denitrification, ND, as well as corresponding vectors or reduction lines (black solid lines) are from Lewicka-Szczebak et al. (2017) and represent minimum and maximum routes of isotopocule values with increasing N₂O reduction to N₂ based on the reported range in the ratio between the isotope fractionation factors of NO reduction for SP and δ^{18} O (Lewicka-Szczebak et al., 2017).

source process in our experiment. In addition, Bateman et al. (2004) observed the largest N2O fluxes at 70 % WFPS on a silty loam soil, lower than the 80% value for the largest fluxes from the clay soil in our study (Fig. 2), suggesting that this optimum value could change with soil type. Further, the maximum total measured N lost $(N_2O + N_2)$ in our study occurred at about 95 % WFPS (Fig. 2), but not many studies report N₂ fluxes for comparison and we are still missing measurements of nitric oxide (NO) (Davidson et al., 2000) and ammonia (NH₃) to account for the total N losses. It is, however, possible that the $N_2O + N_2$ fluxes in the SAT/sat treatment were underestimated due to low diffusivity in the water-filled pores (Well et al., 2001). Gases would have been trapped (particularly in the higher saturation treatments) due to low diffusion and thus possibly masked differences in N₂ and N₂O production since this fraction of gases was not detected (Harter et al., 2016). It is worth mentioning that there was some drying during the incubation. The flow of the gas is very slow $(10 \,\mathrm{mL}\,\mathrm{min}^{-1})$, simulating a low wind speed, so normally this would dry the soil in field conditions too. It would represent a rainfall event where the initial moisture differs between treatments but some drying occurs due to the wind flow. We believe, however, that the effect of drying will be more relevant (and significant relative to the initial moisture) later in the incubation.

The smaller standard errors in both N₂O and N₂ data for the larger soil moisture levels (Table 3 and Fig. 1) could suggest that, at high moisture contents, nutrient distribution (N and C) on the top of the core is more homogeneous, causing replicate cores to behave similarly. At the lower soil moisture for both N₂O and N₂, it is possible that some cracks appear on the soil surface, causing downwards nutrient movement, resulting in heterogeneity in nutrient distribution on the surface and increasing variability between replicates, reflected in the larger standard errors of the fluxes. Laudone et al. (2011) studied, using a biophysical model, the positioning of the hot-spot zones away from the critical percolation path (described as "where air first breaks through the structure as water is removed at increasing tensions") and found it slowed the increase and decline in emission of CO₂, N₂O and N₂. They found that hot-spot zones further away from the critical percolation path would reach the anaerobic conditions required for denitrification in shorter time, and the products of the denitrification reactions take longer to migrate from the hot-spot zones to the critical percolation path and to reach the surface of the system. The model and its parameters can be

Treatment	% mean moisture	NO ₃ ⁻¹ , mg N kg ⁻¹ dry soil	NH4 ⁺ , mgNkg ⁻¹ dry soil	Total C, %	Total N, %	N ₂ O, kg N ha ⁻¹	N ₂ O, mg N kg ⁻¹ dry soil	N_2 , $kg N ha^{-1}$	N ₂ , mgNkg ⁻¹ dry soil	Total emitted N, kg N ha ⁻¹	CO ₂ , kg C ha
SAT/sat		1.1 (0.4)	104.3 (1.1)	3.61 (0.04)	0.35 (0.004)	9.4 (1.1)	7.8 (0.9)	54.0 (14.0)	44.8 (11.6)	63.4	289.2 (30.4
HALFSAT/sat	40.2 (0.2)	0.8(1.0)	104.2 (6.8)	3.64(0.08)	0.36 (0.004)	10.9 (0.4)	9.0(0.3)	51.7 (9.0)	42.8 (7.4)	62.6	283.0 (35.5)
UNSAT/sat		51.2 (37.4)	100.8 (5.7)	3.64(0.10)	0.36 (0.007)	23.7 (11.0)	20.0 (9.5)	36.0 (28.5)	30.2 (23.7)	59.7	417.6 (57.1)
UNSAT/halfsat		100.6 (16.1)	71.3 (33.6)	3.53(0.08)	0.36 (0.01)	16.8 (15.8)	14.0 (13.1)	17.2 (19.4)	14.3 (16.1)	34.1	399.7 (40.6)

standard deviation of the mean of three values (emissions are expressed per area an	Table 3. Contents of soil moisture, NO_3^- , NH_4^+
missions are expressed per area and soil weight basis).	of soil moisture, NO_3^- , NH_4^+ and C:N ratio, and cumulative fluxes of N_2O and N_2 and CO_2 from all treatments at the end of the incubation. Values in brackets are

it remains to be determined whether soil structure, or simply
saturation, is the determining factor when the biological pa-
rameters are constrained. Furthermore, Clough et al. (2013)
indicate that microbial-scale models need to be included in
larger models linking microbial processes and nutrient cy-
cling in order to consider spatial and temporal variation.
Kulkarni et al. (2008) refer to "hot spots" and "hot moments"
of denitrification as scale-dependant and highlight the limi-
tations for extrapolating fluxes to larger scales due to these
inherent variabilities. In addition, in order to understand het-
erogeneity of added amendment, we assumed (for modelling
purposes) multiple pools after N and glucose amendment. In
Bergstermann et al. (2011), for example, we presumed they
occupied 10 % of the pore volume of the core (pool 1), be-
cause this resulted in a good fit for measured and modelled
N_2 and N_2O fluxes as well as $\delta^{15}N^{bulk}$ values. In the current
study, we could assume that in the wettest treatment this (pro-
portional) volume was smaller, i.e. similar to the pore volume
displaced by the added 5 mL of amendment, since pores were
almost completely filled with water. Furthermore, we could
assume that it would have been the largest in the driest treat-
ment, where the amendment solution was also able to infil-
trate air-filled pores in the partly saturated pore space and
thereby increase the water content in the infiltrated volume.
With regards to leaching, it was minimal (< 0.5 mL water in
the core) and so significant leaching of amendment can thus
be excluded. Other techniques such as X-ray and MRI could
help determine the distribution of added nutrients in the soil
matrix.
412 Deletionship with soil remembers to determine
4.1.2 Relationship with soil parameters to determine
processes
Our results, for the two highest water contents (SAT/sat and
HALFSAT/sat), indicated that N_2O only contributed 20% of
the total N emissions, as compared to $40-50\%$ at the low-
est water contents (UNSAT/sat and UNSAT/halfsat, Table 3).
This was due to reduction to N_2 at the high moisture level,
confirmed by the larger N_2 fluxes, favoured by low gas diffu-

used for modelling the effect of soil compaction and satura-

tion on the emission of N2O. They suggest that, having determined biophysical parameters influencing N₂O production,

C sion, which increased the N2O residence time and the chance of further transformation (Klefoth et al., 2014). We should also consider the potential underestimation of the fluxes in the highest saturation treatment due to restricted diffusion in the water-filled pores (Well et al., 2001). A total of 99 % of the soil NO_3^- was consumed in the two high-water treatments, whereas in the drier UNSAT/sat and UNSAT/halfsat treatments there still was 35 and 70% of the initial amount of NO_3^- left in the soil, at the end of the incubation, respectively (Table 3). The total amount of gas lost compared to the NO_3^- consumed was almost 3 times greater for the wetter treatments, and less than twice for the two drier ones. This agrees with denitrification as the dominant process source for

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Table 4. Scenarios with different combinations of δ^{18} O and site preference (SP) endmember values and $\eta_{N_2O-N_2}$ values to calculate maximum and minimum estimates of $\%B_{den}$ (minimum, maximum and average values adopted from Lewicka-Szczabak et al., 2017).

	SP0BD	SP0FDN	ηSP	$\eta^{18}\mathrm{O}$
Model (min endmember plus η)	-11	30	-2	-12
Model (max endmember plus η)	0	37	-8	-12
Model (max endmember)	0	37	-5.4	-12
Model (min endmember)	-11	30	-5.4	-12
Model (max η)	-5	33	-8	-12
Model (min η)	-5	33	-2	-12

Table 5. Ratios $N_2O / (N_2O + N_2)$ for all treatments.

	SAT	[/sat	HALFS	SAT/sat	UNSAT	7/halfsat	UNSA	AT/sat
Days	mean	SE	mean	SE	mean	SE	mean	SE
-1	0.276	0.043	0.222	0.009	0.849	0.043	0.408	0.076
0	0.630	0.022	0.538	0.038	0.763	0.053	0.861	0.043
1	0.371	0.025	0.360	0.019	0.622	0.018	0.644	0.031
2	0.096	0.016	0.139	0.015	0.425	0.005	0.296	0.020
3	0.004	0.002	0.015	0.006	0.439	0.052	0.256	0.025
4	0.017	0.002	0.008	0.001	0.475	0.049	0.232	0.012
5	0.019	0.003	0.012	0.001	0.503	0.037	0.174	0.010
6	0.068	0.008	0.020	0.001	0.459	0.052	0.135	0.010
7	0.085	0.008	0.047	0.003	0.333	0.057	0.127	0.003
8	0.106	0.004	0.066	0.002	0.277	0.006	0.122	0.002
9	0.089	0.003	0.053	0.005	0.265	0.006	0.122	0.005
10	0.060	0.003	0.090	0.014	0.428	0.086	0.118	0.006
11	0.063	0.002	0.053	0.002	0.414	0.051	0.125	0.005

 N_2O with larger consumption of NO_3^- at the higher moisture and larger N_2 to N_2O ratios (5.7, 4.7 for SAT/sat and HALF-SAT/sat, respectively), whereas at the lower moisture, ratios were lower (1.5 and 1.0 for UNSAT/sat and UNSAT/halfsat, respectively) (Davidson, 1991). This also indicates that with WFPS above the 60% threshold for N₂O production from denitrification, there was an increasing proportion of anaerobic microsites with increase in saturation controlling NO_2^{-1} consumption and N2/N2O ratios in an almost linear manner. With WFPS values between 71 and 100 % and N_2 / N_2O between 1.0 and 5.7, a regression can be estimated: Y = 0.1723X - 11.82 ($R^2 = 0.8585$), where Y is N₂ / N₂O and X is %WFPS. In summary, we propose that heterogeneous distribution of anaerobic microsites could have been the limiting factor for complete depletion of NO_3^- and conversion to N_2O in the two drier treatments. In addition, in the UNSAT/halfsat treatment there was a decrease in soil NH_4^+ at the end of the incubation (almost 50%; Table 3), suggesting nitrification could have been occurring at this water content which also agrees with the increase in NO₃, even though WFPS was relatively high (> 71 %) (Table 3). It is important to note that as we did not assess gross nitrification, the observed net nitrification based on lowering in NH₄⁺ could underestimate gross nitrification since there might have been substantial N

mineralisation during the incubation. However, under conditions favouring denitrification at high soil moisture the typical N₂O produced from nitrification is much lower compared to that from denitrification (Lewicka-Szczebak et al., 2017) with the maximum reported values for the N₂O yield of nitrification of 1-3% (e.g. Deppe et al., 2017). If this is the case, nitrification fluxes could not have exceeded 1 kg N with NH_4^+ loss of $< 30 \text{ kg} \times 3 \% \sim 1 \text{ kg}$ N. This would have represented for the driest treatment, if conditions were suitable only for one day, that nitrification-derived N2O would have been 6 % of the total N₂O produced. Loss of NH₃ was not probable at such low pH (5.6). The corresponding rate of NO_3^- production using the initial and final soil contents and assuming other processes were less important in magnitude would have been $< 1 \text{ mg NO}_3^-$ -N kg dry soil⁻¹ d⁻¹, which is a reasonable rate (Hatch et al., 2002). The other three treatments lost similar amounts of soil NH_4^+ during the incubation (23–26%), which could have been due to some degree of nitrification at the start of the incubation before O₂ was depleted in the soil microsites or due to NH_4^+ immobilisation (Table 3) (Geisseler et al., 2010).

A mass N balance, considering the initial and final soil NO_3^- , NH_4^+ , added NO_3^- and the emitted N (as N₂O and N₂), results in unaccounted-for N loss of 177.2, 177.6, 130.6 and

		15				
Day	SAT/sat	δ ¹³ N _{bul} HALFSAT/sat	kAIR (%) UNSAT/sat	UNSAT/halfsat		
-						
-1	-3.8 (2.1)	-6.2 (1.5)	-14.2 (10.9)	-23.6 (1.1)		
1	-18.9 (1.6)	-25.5 (4.6)	-20.3 (2.6)	-20.8 (2.3)		
2	-7.7 (4.2)	-12.7 (2.7)	-12.2 (2.0)	-13.9 (5.7)		
3	-2.4 (1.8)	14.0 (2.2)	-1.1 (7.6)	-4.4(3.0)		
4	-0.9(2.2)	-0.3 (3.6)	-7.8 (4.6)	-9.3(3.7)		
5	-6.9 (0.9)	-4.3 (6.1)	-11.3 (3.7)	-8.9 (7.7)		
7	-9.6 (1.5)	-10.0 (1.6)	-14.3 (4.7)	-13.4 (13.5)		
12	-7.5 (1.2)	-8.6 (0.9)	-11.8 (2.6)	-21.3 (6.9)		
		$\delta^{18}O_{SM}$	10W(%)			
	SAT/sat	HALFSAT/sat	UNSAT/sat	UNSAT/halfsat		
-1	33.3 (2.6)	32.7 (3.0)	31.4 (9.8)	25.2 (4.9)		
1	42.9 (2.4)	37.1 (3.8)	32.3 (3.6)	33.3 (2.1)		
2	54.0 (5.7)	48.7 (4.5)	42.7 (5.3)	40.5 (5.0)		
3	45.7 (1.5)	59.7 (3.2)	53.4 (5.7)	41.2 (1.0)		
4	42.5 (1.4)	42.0 (3.7)	38.1 (4.5)	39.9 (7.7)		
5	36.0 (2.9)	34.6 (3.7)	30.4 (2.6)	36.5 (6.9)		
7	32.2 (5.5)	31.6 (5.5)	28.4 (4.4)	32.7 (5.4)		
12	34.9 (5.6)	34.1 (2.7)	32.4 (2.9)	28.5 (5.0)		
		$\delta^{15} \mathrm{N} lpha_{\mathrm{AIR}}(\% c)$				
	SAT/sat	HALFSAT/sat	UNSAT/sat	UNSAT/halfsat		
-1	-0.3 (3.4)	-2.6 (1.8)	-9.5 (12.0)	-19.7 (2.1)		
1	-17.4 (1.8)	-24.0(5.8)	-20.2(2.0)	-21.1 (2.6)		
2	-4.6 (4.2)	-9.5 (3.6)	-11.1 (1.1)	-13.8 (5.9)		
3	-0.8(1.3)	17.2 (4.0)	7.6 (4.7)	-2.7 (3.2)		
4	1.0 (2.5)	0.7 (2.2)	-3.5(3.7)	-2.8 (7.7)		
5	-5.9(0.7)	-2.9(5.4)	-9.4(3.9)	-5.2 (7.9)		
7	-7.8 (2.3)	-5.3 (4.2)	-12.3 (5.6)	-7.7 (11.5)		
12	-3.3(2.1)	-4.6(0.6)	-8.1(4.2)	-15.3 (5.5)		
		SPAI	R (%)			
	SAT/sat	HALFSAT/sat	UNSAT/sat	UNSAT/halfsat		
-1	7.0 (3.9)	7.1 (4.2)	9.4 (2.1)	7.7 (1.9)		
1	2.9 (0.6)	3.0 (2.3)	0.1 (1.8)	-0.7 (1.4)		
2	6.3 (0.64)	6.4 (1.9)	2.2 (2.0)	0.2 (1.9)		
3	3.3 (1.0)	6.4 (6.9)	11.9 (12.4)	5.9 (0.8)		
4	3.7 (0.6)	2.0 (6.2)	8.7 (5.9)	5.4 (3.0)		
5	2.0 (0.4)	3.0 (2.1)	3.9 (0.5)	7.4 (2.3)		
7	5.0 (2.1)	9.2 (5.2)	3.9 (1.8)	11.2 (4.1)		
12	8.4 (3.3)	7.9 (0.8)	7.3 (3.7)	11.8 (5.3)		
		Estimated rar	nge of % B _{DEN}			
	SAT/sat	HALFSAT/sat	UNSAT/sat	UNSAT/halfsat		
-1	63–100	60–100	53-85	56-84		
1-2	68–100	67-100	73–100	77–100		
3-12	78-100	79-100	60-100	54-86		

Table 6. The temporal trends in $\delta^{15}N_{\text{bulk}}$, $\delta^{18}O$, $\delta^{15}N_{\alpha}$, site preference (SP) and $\%B_{\text{DEN}}$ for all experimental treatments (values in brackets are the standard deviation of the mean).

Table 7. Equations of fitted functions and correlation coefficients corresponding to Fig. 5 for site preference (SP) (y axis) vs. $\delta^{18}O(x axis)$
in all treatments for three periods. Correlations are unadjusted; the P value tests whether the slope is different from zero.

Treatment	Days 1–2	Days 3–5	Days 7–12
SAT/sat	$y = 0.2151x - 5.8386, R^2 = 0.6529$ P = 0.05	$y = 0.1204x - 1.848, R^2 = 0.397$ P = 0.129	$y = 0.5872x - 12.223, R^2 = 0.985$ P < 0.001
HALFSAT/sat	$y = 0.3447x - 10.129, R^2 = 0.9048$ P = 0.004	$y = 0.18x - 4.5966, R^2 = 0.1728$ $P = 0.266$	$y = 0.4063x - 6.2632, R^2 = 0.6876$ P = 0.171
UNSAT/sat	$y = 0.2709x - 8.9968, R^2 = 0.8664$ P = 0.007	$y = 0.7248x - 18.874, R^2 = 0.507$ P = 0.031	$y = 0.6848x - 15.236, R^2 = 0.7156$ $P = 0.034$
UNSAT/halfsat	$y = -0.0146x + 0.2506, R^2 = 0.0024$ P = 0.927	$y = 0.3589x - 7.2194, R^2 = 0.4839$ P = 0.037	$y = -0.318x + 21.261, R^2 = 0.1491$ P = 0.450

110.8 mg N kg⁻¹ for SAT/sat, HALFSAT/sat, UNSAT/sat and UNSAT/halfsat, respectively, that could have been emitted as other N gases (such as NO), and some immobilised in the microbial biomass. NO fluxes reported by Loick et al. (2016), for example, result in a ratio N_2O / NO of 0.4. In summary, unaccounted-for N loss is 2 to 3 times the total measured gas loss (Table 3). In addition, in the SAT/sat treatment there was probably an underestimation of the produced N_2 and N_2O due to restricted diffusion at the high WFPS (e.g. Well et al., 2001).

4.1.3 Implications for field distribution of fluxes

Well et al. (2003) found that under saturated conditions there was good agreement between laboratory and field measurements of denitrification, and attributed deviations, under unsaturated conditions, to spatial variability of anaerobic microsites and redox potential. Dealing with spatial variability when measuring N₂O fluxes in the field remains a challenge, but the uncertainty could be potentially reduced if water distribution is known. Our laboratory study suggests that soil N₂O and N₂ emission for higher moisture levels would be less variable than for drier soils and suggests that for the former a smaller number of spatially defined samples will be needed to get an accurate field estimate. This applied to a lesser extent to the CO₂ fluxes.

4.2 Isotopocule trends

Trends of isotopocule values of emitted N_2O coincided with those of N_2 and N_2O fluxes. The results from the isotopocule data (Table 6 and Fig. 3) also indicated that generally there were more isotopic similarities between the two wettest treatments when compared to the two contrasting drier soil moisture treatments.

Isotopocule values of emitted N₂O reflect multiple processes where all signatures are affected by the admixture of several microbial processes, the extent of N2O reduction to N₂ and the variability of the associated isotope effects (Lewicka-Szczebak et al., 2015). Moreover, for δ^{18} O and $\delta^{15} N^{\text{bulk}}$ the precursor signatures are variable (Decock and Six, 2013), and for δ^{18} O the O exchange with water can be also variable (Lewicka-Szczebak et al., 2017). Since the number of influencing factors clearly exceeds the number of isotopocule values, unequivocal results can only be obtained if certain processes can be excluded or be determined independently (Lewicka-Szczebak et al., 2015, 2017). The two latter conditions were fulfilled in this study, i.e. N₂O fluxes were high and several orders of magnitude above possible nitrification fluxes, since the $N_2O - to - NO_3^-$ ratio yield of nitrification products rarely exceeds 1 % (Well et al., 2008). Moreover, N₂ fluxes and thus N₂O reduction rates were exactly quantified.

The estimated values of $\% B_{DEN}$ indicate that, in the period immediately after amendment application, all moisture treatments were similar, reflecting that the microbial response to N and C added was the same and denitrification dominated. This was the same for the rest of the period for the wetter treatments. In the drier treatments, proportions decreased afterwards and were similar to values before amendment application, possibly due to recovery of more aerobic conditions that could have encouraged other processes to contribute. As N₂ was still produced in the driest treatment (but in smaller amounts), this indicated ongoing denitrifying conditions and thus large contributions to the total N₂O flux from nitrification were not probable, but some occurred as suggested by NH⁺₄ consumption.

The trends observed reflect the dynamics resulting from the simultaneous application of NO_3^- and labile C (glucose) on the soil surface as described in previous studies (Meijide et al., 2010; Bergstermann et al., 2011) where the same soil was used, resulting in two locally distinct NO_3^- pools with differing denitrification dynamics. In the soil volume reached by the NO_3^- / glucose amendment, denitrification was initially intense with high N₂ and N₂O fluxes and rapid isotopic enrichment of the NO_3^- -N. When the NO_3^- and/or glucose of this first pool was exhausted, N₂ and N₂O fluxes were much lower and dominated by the initial NO_3^- pool that was not reached by the glucose / NO_3^- amendment and that is less fractionated due to its lower exhaustion by denitrification, causing decreasing trends in $\delta^{15}N^{\text{bulk}}$ of emitted N₂O.

This is also reflected in Fig. 4, where N₂O fluxes from both pools exhibited correlations (and mostly significant) between $\delta^{15}N^{\text{bulk}}$ and $\delta^{18}O$ due to varying N₂O reduction, but $\delta^{15}N^{\text{bulk}}$ values in days 1 and 2 – i.e. the phase when pool 1 dominated – were distinct from the previous and later phase.

The fit of ¹⁵N^{bulk} / ¹⁸O data to two distinct and distant regression lines can be attributed to two facts: firstly, in the wet treatment (Fig. 4a, b) pool 1 was probably completely exhausted and there was little NO₃⁻ formation from nitrification (indicated by final NO₃⁻ values close to 0, Table 3), whereas the drier treatment exhibited substantial NO₃⁻ formation and high residual NO₃⁻. Hence, there was probably still some N₂O from pool 1 after day 2 in the dry treatment but not in the wetter ones. Secondly, the product ratios after day 2 of the drier treatments (0.001 to 0.09). Thus the isotope effect of N₂O reduction was smaller in the drier treatments, leading to a smaller upshift of δ^{15} N^{bulk} and thus more negative values after day 2, i.e. with values closer to days 1+2.

This finding further confirms that δ^{15} N / δ^{18} O patterns are useful to identify the presence of several N pools, e.g. typically occurring after application of liquid organic fertilisers, which has been previously demonstrated using isotopocule patterns (Koster et al., 2015).

Interestingly, the highest $\delta^{15}N^{\text{bulk}}$ and $\delta^{18}O$ values of the emitted N₂O were found in the soils of the HALF-SAT/sat treatment, although it may have been expected that the highest isotope values from the N₂O would be found in the wettest soil (SAT/sat) because N2O reduction to N2 is favoured under water-saturated conditions due to extended residence time of produced N₂O (Well et al., 2012). However, N₂O / (N₂+N₂O) ratios of the SAT/sat and SAT/halfsat treatments were not different (Table 5). Bol et al. (2004) also found that some estuarine soils under flooded conditions (akin to our SAT/sat) showed some strong simultaneous depletions (rather than enrichments) of the emitted N2O δ^{15} N^{bulk} and δ^{18} O values. These authors suggested that this observation may have resulted from a flux contribution of an "isotopically" unidentified N₂O production pathway. Another explanation could be complete consumption of some of the produced N₂O in isolated micro-niches in the SAT/sat treatment due to inhibited diffusivity in the fully saturated pore space. N2 formation in these isolated domains would not affect the isotopocule values of emitted N_2O and this would thus result in lower apparent isotope effects of N_2O reduction in water saturated environments as suggested by Well et al. (2012).

The SP values obtained were generally below 12%, in agreement with reported ranges attributed to bacterial denitrification: -2.5 to 1.8% (Sutka et al., 2006), 3.1 to 8.9%(Well and Flessa, 2009) and -12.5 to 17.6% (Ostrom and Ostrom, 2011). The SP, believed to be a better predictor of the N₂O source as it is independent of the substrate isotopic signature (Ostrom and Ostrom, 2011), has been suggested as it can be used to estimate N_2O reduction to N_2 in cases when bacterial denitrification can be assumed to dominate N₂O fluxes (Koster et al., 2013; Lewicka-Szczebak et al., 2015). There was a strong correlation between the SP and $N_2O / (N_2O + N_2)$ ratios on the first 2 days of the incubation for all treatments up until the N2O reached its maximum (Fig. 3), which reflects the accumulation of $\delta^{15}N$ at the alpha position during ongoing N₂O reduction to N₂. Later on in the experiment, beyond day 3, this was not observed probably because in that period the product ratio remained almost unchanged and very low (Table 6). Similar observations have been reported by Meijide et al. (2010) and Bergstermann et al. (2011), as they also found a decrease in SP during the peak flux period in total N_2+N_2O emissions, but only when the soil had been kept wet prior to the start of the experiment (Bergstermann et al., 2011). These results confirm from two independent studies (Lewicka-Szczebak et al., 2014) that there is a relationship between the product ratios and isotopic signatures of the N₂O emitted. The δ^{18} O vs. SP regressions indicate more similarity between the three wettest treatments as well as high regression coefficients, suggesting this SP/ δ^{18} O ratio could also be used to help identify patterns for emissions and their sources.

4.3 Link to modelling approaches.

Since isotopocule data could be compared to N₂ and N₂O fluxes, the variability of isotope effects of N₂O production and reduction to N₂ by denitrification could be determined from this dataset (Lewicka-Szczebak et al., 2015), and this included modelling the two pool dynamics discussed above. It was demonstrated that net isotope effects of N₂O reduction $(\eta_{N_2O-N_2})$ determined for both NO₃⁻ pools differed. Pool 1 representing amended soil and resulting in high fluxes but moderate product ratio, exhibited $\eta_{N_2O-N_2}$ values and the characteristic η^{18} O / η^{15} N ratios similar to those previously reported, whereas for pool 2 (amendment-free soil), characterised by lower fluxes and very low product ratio, the net isotope effects were much smaller and the $\eta^{18}O / \eta^{15}N$ ratios, previously accepted as typical for N₂O reduction processes (i.e. higher than 2), were not valid. The question arises of whether the poor coincidence of pool 2 isotopologue fluxes with previous N2O reduction studies reflects the variability of isotope effects of N₂O reduction or whether the contribution of other processes like fungal denitrification could explain this (Lewicka-Szczabak et al., 2017). The latter explanation is evaluated in Sect. 4.4.

Liu et al. (2016) noted that on the catchment-scale potential N₂O emission rates were related to hydroxylamine and NO₃⁻, but not NH₄⁺ content in soil. Zou et al. (2014) found high SP (15.0 to 20.1 %) values at WFPS of 73 to 89 %, suggesting that fungal denitrification and bacterial nitrification contributed to N₂O production to a degree equivalent to that of bacterial denitrification.

To verify the contribution of fungal denitrification and/or hydroxylamine oxidation we can first look at the $\eta SP_{N_2O-NO_3}$ values calculated in the previous modelling study applied on the same dataset (Table 1, the final modelling step; Lewicka-Szczebak et al., 2015). For pool 1 there are no significant differences between the values of various treatments, SP₀ ranges from (-1.8 ± 4.9) to $(+0.1 \pm 2.5)$. Pool 1 emission was mostly active in days 1-2; hence, these values confirm the bacterial dominance in the emission at the beginning of incubation, which originates mainly from the amendment addition and represents a similar pathway for all treatments. However, for the pool 2 emission we could observe a significant difference when compared the two wet treatments (SAT/sat and HALFSAT/sat: (-5.6 ± 7.0)) with the UNSAT/sat treatment $(+3.8 \pm 5.8)$. This represents the emission from unamended soil which was dominating after the third day of the incubation and indicates higher nitrification contribution for the drier treatment.

4.4 Contribution of bacterial denitrification

An endmember mixing approach has been previously used to estimate the fraction of bacterial N_2O (% B_{DEN}), but without independent estimates of N_2O reduction (Zou et al., 2014), but due to the unknown isotopic shift by N_2O reduction, the ranges of minimum and maximum estimates were large, showing that limited information is obtained without N_2 flux measurement.

In an incubation study with two arable soils, Koster et al. (2013) used $N_2O / (N_2 + N_2O)$ ratios and isotopocule values of gaseous fluxes to calculate SP of N2O production (referred to as SP_0), which is equivalent to SP_0 using the Rayleigh model and published values of $\eta_{N_2O-N_2}$. The endmember mixing approach based on SP₀ was then used to estimate fungal denitrification and/or hydroxylamine oxidation, giving indications for a substantial contribution in a clay soil, but not in a loamy soil. Here we presented for the first time an extensive dataset with large range in product ratios and moisture to calculate the contribution of bacterial denitrification $(\% B_{\text{DEN}})$ of emitted N₂O from SP₀. The uncertainty of this approach arises from three factors: (i) from the range of SP_0 endmember values for bacterial denitrification of -11 to 0%and 30 to 37 for hydroxylamine oxidation/fungal denitrification, (ii) from the range of net isotope effect values of N₂O reduction $(\eta_{N_2O-N_2})$ for SP which vary from -2 to -8%

(Lewicka-Szczebak et al., 2015), and (iii) system condition (open vs. closed) taken to estimate the net isotope effect (Wu et al., 2016).

The observation that $\% B_{\text{DEN}}$ of emitted N₂O was generally high (63-100%) in the wettest treatment (SAT/sat) was not unexpected. However, interestingly, $\% B_{\text{DEN}}$ in the HALFSAT/sat treatment was very similar (71-98%), pointing to the role of the wetter areas of the soil microaggregates contributing to high $\% B_{\text{DEN}}$ values. The slightly lower values, i.e. down 60 % in UNSAT/sat % BDEN range of 60-100 %, suggest that the majority of N₂O derived from bacterial denitrification still results from the wetter microaggregates of the soils, despite the fact that the macropores are now more aerobic. Only when the micropores become partially wet, as in the UNSAT/halfsat treatment, do the more aerobic soil conditions allow a higher contribution of nitrification/fungal denitrification, ranging from 0 to 46 % (1 -%B_{DEN}, Table 6) on days 3–12 (Zhu et al., 2013). Differences in the contribution of nitrification/fungal denitrification between the flux phases when different NO_3^- pools were presumably dominating are only indicated in the driest treatment, since $1 - \% B_{\text{DEN}}$ was higher after day 2 (14 to 46 %) compared to days 1+2 (0 to 33%). This larger share of nitrification/fungal denitrification can be attributed to the increasing contribution from pool 2 to the total flux as indicated by the modelling of higher SP₀ for pool 2 (see previous section and Lewicka-Szczebak et al. (2015). In addition, indications for elevated contribution of processes other than bacterial denitrification were only evident in the drier treatments during phases before and after N2, and N2O fluxes were strongly enhanced by glucose amendment. The data supply no clue whether the other processes were suppressed during the anoxia induced by glucose decomposition or just masked by the vast glucose-induced bacterial N₂O fluxes.

5 Conclusions

This study combined direct measurements of N₂ as an indicator of denitrification with isotopomers, providing a measurement approach that verifies the source processes of N₂O emissions. The results from this study demonstrated that at high soil moisture levels, there was less variability in N fluxes between replicates, potentially decreasing the importance of soil hot spots in emissions at these moisture levels. At high moisture there was also complete depletion of nitrate, confirming denitrification as the main pathway for N₂O emissions, and due to less diffusion of the produced N₂O, the potential for further reduction to N₂ increased. Under less saturation, but still relatively high soil moisture, nitrification occurred. Isotopic similarities were observed between similar saturation levels and patterns of $\delta^{15}N / \delta^{18}O$ and SP / $\delta^{18}O$ are suggested as indicators of source processes.

Data availability. Data are available on request from Laura Cardenas at laura.cardenas@rothamsted.ac.uk.

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Competing interests. The authors declare that they have no conflict of interest.

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