



Effect of temperature rise and ocean acidification on growth of calcifying tubeworm shells (*Spirorbis spirorbis*): an in situ benthocosm approach

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Abstract. The calcareous tubeworm *Spirorbis spirorbis* is a widespread serpulid species in the Baltic Sea, where it commonly grows as an epibiont on brown macroalgae (genus *Fucus*). It lives within a Mg-calcite shell and could be affected by ocean acidification and temperature rise induced by the predicted future atmospheric CO₂ increase. However, *Spirorbis* tubes grow in a chemically modified boundary layer around the algae, which may mitigate acidification. In order to investigate how increasing temperature and rising pCO₂ may influence *S. spirorbis* shell growth we carried out four seasonal experiments in the Kiel Outdoor Benthocosms at elevated pCO₂ and temperature conditions. Compared to laboratory batch culture experiments the benthocosm approach provides a better representation of natural conditions for physical and biological ecosystem parameters, including seasonal variations. We find that growth rates of *S. spirorbis* are significantly controlled by ontogenetic and seasonal effects. The length of the newly grown tube is inversely related to the initial diameter of the shell. Our study showed no significant difference of the growth rates between ambient atmospheric and elevated (1100 ppm) pCO₂ conditions. No influence of daily average CaCO₃ saturation state on the growth rates of *S. spirorbis* was observed. We found, however, net growth of the shells even in temporarily undersaturated bulk solutions, under conditions that concurrently favoured selective shell surface dissolution. The results suggest an overall resistance of *S. spirorbis* growth to acidification levels predicted for the year 2100 in the Baltic Sea. In

contrast, *S. spirorbis* did not survive at mean seasonal temperatures exceeding 24 °C during the summer experiments. In the autumn experiments at ambient pCO₂, the growth rates of juvenile *S. spirorbis* were higher under elevated temperature conditions. The results reveal that *S. spirorbis* may prefer moderately warmer conditions during their early life stages but will suffer from an excessive temperature increase and from increasing shell corrosion as a consequence of progressing ocean acidification.

1 Introduction

Atmospheric carbon dioxide (CO₂) is a primary substrate for life on Earth but is also a major driver of global-scale environmental change, causing ocean acidification (Greene et al., 2012), controlling climate variability (Retallack, 2002; Galeotti et al., 2016) and initiating mass extinctions (Jaraula et al., 2013; Veron et al., 2009). The recent rapid CO₂ rise from anthropogenic emissions is a source of ocean acidification including pH reductions and alterations in fundamental chemical balances (Doney et al., 2009). Since the beginning of the industrial era, atmospheric pCO₂ rose from about 280 to 405 µatm (NOAA-ESRL, 2017¹) due to human activities such as fossil fuel combustion, cement production and deforestation. At the same time surface seawater pH decreased

¹www.esrl.noaa.gov/gmd/ccgg/trends

by 0.1 units, corresponding to 30 % increase in the hydrogen ion concentration (Raven et al., 2005; Cao and Caldeira, 2008). It is predicted to further decrease by 0.3 to 0.4 pH units until the year 2100 when atmospheric pCO₂ levels may reach 950 µatm (IPCC, 2013). By the end of this century, the average surface ocean pH could be lower than it has been for more than 50 Myr (Caldeira and Wickett, 2003) with severe consequences for marine calcifying organisms (Orr et al., 2005; Andersson et al., 2008; Erez et al., 2011).

The CO₂ rise also caused an increase of sea surface temperatures (SSTs) of about 1 °C on a global scale (IPCC, 2013). However, mid- and high-latitude SSTs are more variable and increase more rapidly than the global average. For instance, the Baltic Sea annual mean SST warmed by up to 1 °C per decade between 1990 and 2008 (Elken et al., 2015). Warming of up to 6 °C and prolonged summer heatwaves are expected by the end of 21st century (HELCOM, 2007; Gräwe et al., 2013). Rising temperatures and summer heatwaves may increasingly affect mid/high-latitude marine ecosystems in the future, e.g. through microalgae/macroalgae ecological functions, impacts on food-web structures or reduced reproduction (Knight-Jones et al., 1972; Graiff et al., 2015a; Werner et al., 2016). Stress from elevated temperatures can cause a depletion of organisms' energy supplies resulting in energy deficiencies and increased mortality (Ivanina et al., 2013).

Coastal water pCO₂ and pH can be much more variable than that of the open ocean due to the effects of run-off, upwelling, eutrophication, atmospheric deposition and remineralisation (Doney et al., 2007). The Baltic Sea is an intra-continental non-tidal brackish water environment with highly variable seasonal dynamics of pCO₂ and pH. Annual pH ranges vary from 8.1–8.4 in the Kattegat area to 7.4–8.4 in the less saline eastern Baltic (Havenhand, 2012). Kiel Fjord and Eckernförde Bay are narrow coastal embayments in the western Baltic Sea. Surface water data from Kiel Fjord show a seasonal pH range from 7.3 to 8.5 (NBS scale) with pCO₂ varying from 385 to 2500 µatm (Thomsen et al., 2010, 2013; Wahl et al., 2015). Significant variations in pH and pCO₂ were observed along the coast line of the Kiel Bight (Winde et al., 2017). In the *Fucus* meadows of Eckernförde Bay diurnal pH variations from 7.3 to 7.8 were found during an upwelling episode, while during normal summer conditions pH varied between 8.0 and 8.4 (Saderne et al., 2013). These observed ranges of pCO₂ and pH by far exceed the predicted levels at the end of the 21st century. Therefore, the following question arose: are calcifying organisms living under such dynamic conditions better adapted for future ocean acidification?

Consequences of ocean warming and acidification for marine organisms have been investigated in many studies (e.g. Reynaud et al., 2003; Marshall and Clode, 2004; Veron et al., 2009; Saderne and Wahl, 2013; Wisshak et al., 2013; Cornwall et al., 2016; Wahl et al., 2016). However, only few studies investigated combined effects of simultaneously in-

creased temperature and CO₂ on entire ecosystems (Wahl et al., 2015). To study the combined impact of temperature rise and elevated CO₂ on typical marine calcifiers from the Baltic Sea, we carried out experiments in the Kiel Outdoor Benthocosms (KOB, Wahl et al., 2015) to investigate calcification of the serpulid tubeworm *Spirorbis spirorbis* under near-natural habitat conditions as sessile epibionts on the thalli of *Fucus* seaweeds.

The brown algae *Fucus vesiculosus* and *Fucus serratus* are among the most widespread brown seaweed found on the coasts of the Baltic Sea. The pH in the seaweed ecosystems shows significant diurnal variations due to photosynthesis (high pH during the day) and respiration (low pH during the night; Saderne et al., 2013). A diffusive boundary layer (DBL) of typically 50 µm to 2 mm thickness surrounds the algal thalli depending primarily on the flow conditions (Larkum et al., 2003; Spilling et al., 2010; Hurd and Pilditch, 2011; Wahl et al., 2016). Microepibionts and macroepibionts living in the DBL are affected by conditions with variable concentrations of chemical compounds (e.g. O₂, DIC and pH) that are created by algal bioprocesses (Larkum et al., 2003). In the DBL of *F. vesiculosus*, pH was found to increase by up to 1.5 units from dark conditions to bright daylight (Spilling et al., 2010; Wahl et al., 2016). Consequently, this surface boundary layer of the algae can potentially provide a shelter from ocean acidification during daylight (Hendriks et al., 2014; Pettit et al., 2015).

Water temperature significantly influences growth, photosynthesis and metabolism of algae. Optimal temperature for growth of Baltic *F. vesiculosus* is in the range of 15 to 20 °C, but growth decreases rapidly when the water temperature exceeds 27 °C for several days (Graiff et al., 2015a). High temperatures may therefore have indirect adverse effects on epibionts, like *S. spirorbis*, because the ecological functions of their host algae may be reduced or damaged.

Spirorbis spirorbis (Linnaeus, 1758) is a millimetre-sized, coiled calcareous tubeworm which belongs to the family Serpulidae, subfamily Spirorbinae (class Polychaeta). The Spirorbinae originated in the later Mesozoic and became common during the latest Cretaceous (Ippolitov and Rzhavsky, 2014). The tube of *S. spirorbis* is sinistral, planospiral, unsculptured, commonly with a small, peripheral flange increasing the area attached to the substrate (Fig. 1; Ippolitov and Rzhavsky, 2015). *S. spirorbis* usually lives attached to seaweed and eel grass in shallow sublittoral and intertidal marine environments (Ippolitov and Rzhavsky, 2015). It favours toothed wrack (*Fucus serratus*), bladder wrack (*Fucus vesiculosus*, Fig. 1) and kelp (*Laminaria* spp.), and rarely grows on other substrates like rocks or other algae (De Silva, 1962; O'Connor and Lamont, 1978; Qian, 1999). It is a common species in the Baltic Sea, where it lives in coastal macrophyte meadows characterised by large pH variations (> 1 pH unit) and frequent aragonite undersaturation ($\Omega_{\text{arag}} > 0.6$, Saderne et al., 2013).

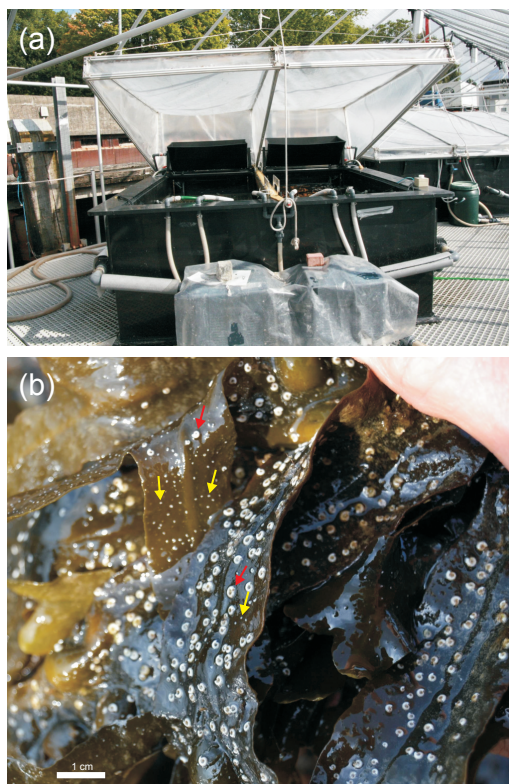


Figure 1. (a) Two subunits of the Kiel Outdoor Benthocosm with open hood. Subunits with closed hoods are visible in the background on the right. (b) *Spirorbis spirorbis* specimens attached to living brown alga *Fucus vesiculosus*. Juvenile (white dots, yellow arrows) and adult (white spires, red arrows) specimens of *S. spirorbis* are visible.

S. spirorbis shells are purely calcitic. No or only questionable indications of aragonite have been reported (Ippolito and Rzhavsky, 2015). The tubes consist of Mg-calcite with about 10 mol% MgCO₃ (Bornhold and Milliman, 1973; Ni et al., 2018), which has a similar solubility as aragonite (Plummer and Mackenzie, 1974; Walter and Morse, 1984; Morse and Mackenzie, 1990). Obviously, *S. spirorbis* is able to prosper in temporarily CaCO₃ undersaturated water. Other serpulid worms have even been reported to calcify in abyssal waters below the calcium carbonate compensation depth (Kupriyanova et al., 2014).

Previous work on Baltic *S. spirorbis* in laboratory experiments (Saderne and Wahl, 2013) found significantly reduced growth only at pH values lower than 7.7 ($\Omega_{\text{arag}} < 0.8$). The study confirmed that the tubeworms were able to calcify in aragonite undersaturated water ($\Omega_{\text{arag}} < 1$). This points to a high short-term tolerance for ocean acidification for at least some of the serpulid worm species.

Several recent ocean acidification experiments have included serpulid worms of a variety of species. Most studies have focused on the tropical species *Hydroides elegans* (Lane et al., 2013; Chan et al., 2012, 2013; Mukherjee et al.,

2013; Li et al., 2014). The results indicated reduced growth, increased porosity and reduced mechanical strength of the worm tubes, as well as increased mortality of larvae at lowered pH (< 7.9).

Field experiments in subtropical settings show reduced serpulid population counts at lowered pH (Campbell and Fourqurean, 2014; Peck et al., 2015). In a Mediterranean seagrass meadow, naturally acidified by volcanic CO₂ seeps, calcareous serpulids were absent at sites with high pCO₂ (pH = 7.1; $\Omega_{\text{arag}} = 0.6$; Cigliano et al., 2010). In this area, the specialised tubeworm *Simplaria* spp. dominates serpulid populations in intermediate-pH habitats (pH ~ 7.4, $\Omega \sim 1.1$; Lucey et al., 2016). Experiments with *Hydroides crucigera* in a temperate setting, on the other hand, showed only moderate impacts of acidification on serpulids even in undersaturated water ($\Omega_{\text{arag}} < 0.8$), including a shift in tube mineralogy (Ries et al., 2009; Ries, 2011).

In the present study, we compare growth rates and corrosion features of *Spirorbis spirorbis* grown under ambient and elevated pCO₂ and temperature conditions in four seasonal experiments to test their sensitivity to ocean acidification and warming. Our results also provide new information about the life cycle and shell microstructure of *S. spirorbis*. The growth experiments were carried out in the Kiel Outdoor Benthocosms under near-natural conditions, exposed to the weather and water conditions of the Kiel Fjord, by using a flow-through setup with water pumped directly from the fjord (Wahl et al., 2015).

2 Material and methods

2.1 Sampling

Healthy *F. vesiculosus* plants bearing intermediate amounts of live *S. spirorbis* were collected for four seasonal experiments in less than 1.5 m water depth in Eckernförde Bay (54°27' N, 9°53' E; western Baltic Sea, Germany) in March 2013, June 2013, October 2013 and January 2014. The location is described in detail by Saderne et al. (2013) and Winde et al. (2017). Individual *Fucus* plants were selected by visual inspection to contain approximately the same volume of blades and similar amounts of *S. spirorbis*. The typical density of *S. spirorbis* tubes at the start of the experiments is shown in Fig. 1. The collected plants were transported in a cool box to GEOMAR (Kiel, Germany) for subsequent treatments.

2.2 Culturing

The samples were stained outdoor at the quay at GEOMAR in a closed 10 L transparent plastic box for 3 days in Kiel Fjord seawater with ~ 50 mg L⁻¹ calcein. The box was continuously bubbled with ambient air. At the start of the staining *S. spirorbis* were fed with *Rhodomonas* algae. The staining box was placed in a flow-through water trough with sea-

water pumped from the Kiel Fjord to keep the temperature close to ambient conditions in the Fjord. The absorption of the dye into newly grown tubes provides a well-defined starting point for growth under the experimental conditions. After 3 days of staining, 12 individual *Fucus* plants were transplanted into the 12 subunits of the Kiel Outdoor Benthocosms (Wahl et al., 2015), fixed on a plastic grid at the bottom of the basins under 0.4 m of water. The incubations started immediately after staining. Average *S. spirorbis* starting populations were on the order of 100–200 specimens per subunit.

The 12 benthocosm subunits were assigned to four treatments and each treatment had three replicates (Wahl et al., 2015): “control treatment” with ambient pCO₂ (380–400 µatm) and water temperature, “+CO₂ treatment” with 1100 µatm pCO₂ in the headspace of the subunit, “+T treatment” with water temperature elevated by 5 °C over ambient conditions and “+CO₂ + T treatment” as a combination of both elevated pCO₂ and temperature. These conditions are considered as representative for acidification and temperature changes at the end of 21st century (Wahl et al., 2015). Each benthocosm subunit had a volume of 1500 L and was continuously flushed with ambient fjord water, pumped from 1 m below the surface at a flow rate of about 65 L h⁻¹. Water in the subunits was additionally mixed by artificial waves with a frequency of 30 waves per hour. Four seasonal experiments were carried out: “spring” (4 April–19 June 2013), “summer” (4 July–17 September 2013), “autumn” (10 October–17 December 2013) and “winter” (16 January–1 April 2014). In total, each subunit contained 21 *Fucus* plants, but only one plant with attached *Spirorbis*, and a fauna of mollusks, arthropods and echinoderms. Details of the KOB setup and experimental parameters are described in Wahl et al. (2015), Graiff et al. (2015b) and Werner et al. (2016). After 10–11 weeks of incubation, the 12 algal plants with *S. spirorbis* were collected from the benthocosms for freeze drying and further analysis.

2.3 Measurements and statistics of *S. spirorbis* growth

S. spirorbis specimens were peeled off from the algal surfaces and photographed under an epifluorescence microscope. The initial and final diameter (in millimetres) of *S. spirorbis* shells and the length of the newly grown tube segments (mm) were measured after observing the position of the staining line (Fig. 2). The absolute tube length increase (mm) was measured as the length of the newly formed external arc of the tube between the staining front and the terminal tube edge, following Saderne and Wahl (2013).

From the spring, summer and autumn experiments *S. spirorbis* tubes were collected from some basins for chemical analysis (Ni et al., 2018). From each basin the newly grown tube parts of up to 20 specimens were cut off at the stain line, pooled, bleached, washed, dried and weighed. Bleaching was carried out using sodium hypochlorite with 1 % active chlorine.

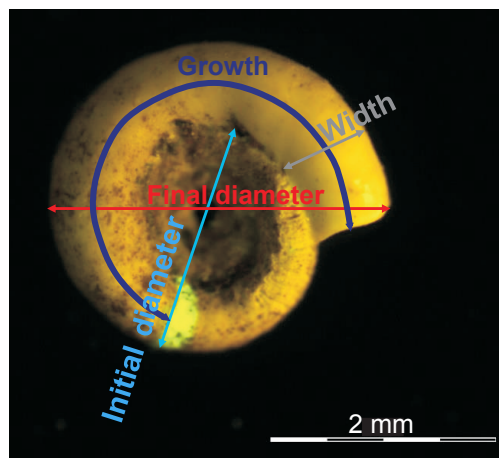


Figure 2. Fluorescence microscope view of *S. spirorbis* shell with indicators for measured size parameters used in this study (initial diameter, final diameter, growth length, tube width). Brightly fluorescent yellowish calcein stain line (lower left) marks beginning of shell part grown during experiment.

The measured length increase and final diameters were normalised by the initial diameter. In our analysis we compared the resulting five growth parameters: (1) initial diameter, D_i ; (2) final diameter, D_f ; (3) growth, Gr ; (4) growth/initial diameter, Gr/D_i ; and (5) final diameter/initial diameter, D_f/D_i . In order to test the robustness of the different parameters we measured Gr and D_f of specimens with similar initial diameters in the autumn, winter and spring populations. The results showed that Gr measurements were more sensitive in detecting growth differences than the D_f measurements.

Normalisation to the initial diameter was applied because growth of *S. spirorbis* tubes is strongly size-dependent. However, as the dependence is not strictly linear (see Sect. 3.5) we based all growth rate comparisons on the condition that the initial diameters of the starting populations were in the same range. The clearly bimodal populations in the autumn experiment were treated separately (autumn-big and autumn-small). The summer populations and autumn-small populations, which both were dominated by juveniles, differed significantly from the autumn-big, winter and spring populations, dominated by adults. Therefore no comparisons were carried out between these two sets of populations, because there was very little overlap in the initial sizes (compare Results Section, Fig. 5). Initial diameters of the summer and autumn-small populations overlapped to a high degree, but the medians differed significantly (two-way ANOVA, $p = 0.02$). In order to derive comparable populations with similar D_i in the summer and autumn-small data we selected sub-populations that had similar D_i ranges and similar median D_i values. Tubes outside this D_i range were not used in the statistical analysis. For the autumn-big, winter and spring

populations the initial diameters were not significantly different, as verified by Tukey's HSD tests.

Three-, two- and one-way ANOVA and Tukey's HSD tests were used for testing statistical significance of differences between the median values from different treatments and seasons. Each treatment had three replicates but, with a total duration of 1 year, seasonal experiments were not replicated. Median values were calculated for each of the treatment replicates based on the measured values, resulting in 12 basin medians for every seasonal experiment. In the three-way ANOVA, the three factors were temperature, pCO₂ and season. The temperature and pCO₂ factors had two levels, elevated and ambient. It should be kept in mind that the season factor here is a multiple factor which includes a range of parameters/conditions such as fjord temperature, pH, saturation state, nutrients and ontogenetic effects of *S. spirorbis*. Only differences caused by the temperature and pCO₂ offsets between the treatments were tested for statistical significance. The "seasonal" factor had no independent (multi-annual) replicates. Differences between seasonal experiments may consequently arise from any of the above mentioned seasonal factors as well as from other unknown factors.

Assumption of normality of the models' residuals and homogeneity of residual variances were tested with Shapiro–Wilk tests and box plots respectively. Statistical analyses were conducted with R (version 3.2, <http://cran.r-project.org>), PAST (version 3.13; Hammer et al., 2001) and Microsoft Excel (Data Analysis Tool). A probability value of < 0.05 was considered significant.

2.4 Microstructures

For localisation of the calcein stain line *S. spirorbis* specimens were photographed with an epifluorescence microscope (AxioScope A1, Carl Zeiss, Germany). Polished longitudinal and cross sections were used for electron microscopy. Samples were wet polished with grinding paper followed by polishing solutions of 9, 3 and 1 µm grain size until no more scratches were visible on the polished surface. Backscatter electron images (BEI) and element concentration maps of calcium were taken with a JEOL JXA 8200 "Superprobe" electron microprobe (EMP) at GEOMAR Kiel, Germany. High-resolution (2–3 µm per pixel) maps of calcium were recorded with 50 nA beam intensity at 15 kV, eight accumulations and 100 ms dwell time. Internal structures of stained skeletons were imaged on polished cross sections with a Zeiss Axio Imager.M2 microscope using white field and differential interference contrast. The Cy3 filter set was applied for detection of calcein. Images were acquired with a resolution of 1360 × 1024 pixels. Excitation wavelength was 495 nm. Emission from calcein (517 nm) was recorded.

2.5 Seawater chemistry

Temperature and pH_{NBS} in all benthocosm treatments and the fjord water inflow were logged at 2-hour intervals by GHL temperature sensors (PT1000) and pH glass electrodes respectively. Air pCO₂ in the head space of the +CO₂ treatment subunits was monitored using infrared spectroscopy and kept at a constant level as described by Wahl et al. (2015).

Additionally, pH_{NBS} values were measured daily using a Seven MultiInLab Expert Pro (pH, Mettler Toledo GmbH, Giessen, Germany). The pH electrode was calibrated with NBS pH-buffer solutions (4.001, 6.865) kept at in situ temperature (Winde et al., 2014, 2017). The pH 9/10 buffer was avoided to prevent impact of possible CO₂ contamination under field conditions. The stability of the electrodes' Nernst slope and the applicability of the two-point calibration to higher pH was previously tested by the measurement of a pH 10 calibration solution. Independently calculated pH values (using CO2SYS) based on measured dissolved inorganic carbon (DIC) and total alkalinity (TA) values showed good agreement with the measured pH in the range of 8 to 9 (Wahl et al., 2015). Discrete water samples were taken as described by Wahl et al. (2015) and analysed for TA two times a week, as well as for DIC on a monthly base. Water samples for DIC analysis were filled bubble-free into 50 mL Winkler bottles, poisoned by the addition of one drop of saturated mercury chloride (HgCl₂) solution and measured via coulometric titration (Johnson et al., 1993). To remove microbes and particles TA samples were filtered through 0.45 µm Minisart syringe filters (Sartorius SFCA, Sartorius) and measured by potentiometric titration using 0.01 M HCl with a Schott titriplus and an Ioline electrode A157. NaCl was added to avoid changes in ionic strength during the analysis. The titration cell was kept at 25 °C. Measurements were calibrated using certified seawater standards for DIC and TA (Dickson et al., 2003, 2007).

The speciation in the dissolved carbonate system, including the carbonate ion concentration, was calculated from pH_{NBS}, TA, temperature and salinity using the code of the CO2SYS software package for MATLAB, version 1.1 (Lewis and Wallace, 1998; van Heuven et al., 2011), with constants recommended for best practice (Dickson et al., 2007; Orr et al., 2015), i.e. K_1 and K_2 from Lueker et al. (2000), K_S and K_B from Dickson (1990), K_F from Dickson and Riley (1979), K_W , K_{1-3P} and K_{Si} from Millero (1995) and the total boron–salinity relationship from Upström (1974). The K_1 and K_2 constants from Lueker et al. (2000) are defined for a salinity range from 19 to 43, while the brackish Kiel Fjord water ranged from about 10 to 20 PSU during the experiments. An alternative set of equations for K_1 and K_2 is available from Millero (2010) for salinities as low as 1. However, as discussed in Orr et al. (2015) applications of the latter showed discrepancies on different pH scales. Therefore, we used the Lueker et al. (2000)

constants. Using the latter to calculate carbonate ion concentrations at salinities as low as 10 PSU resulted in offsets of less than 0.5 % compared to the Millero (2010) constants, which is negligible for our interpretations.

Salinity and concentrations of Ca, Si and P were measured in all benthocosm treatments 2 times a week. Dissolved Si, P and Ca were analysed by inductively coupled plasma optical emission spectrometry (iCAP 6300 DUO, Thermo Fisher Scientific) after appropriate dilution. The accuracy and precision was routinely checked with the certified seawater standard CASS-5 as previously described (Kowlaski et al., 2012). PO₄ was also measured by spectrophotometry using a QuAatro nutrient analyser (SEAL Analytical; Winde et al., 2014). Accuracy and precision checked by replicate analyses of a solution from powdered phosphate salts were better than 8 % RSD. Si and P concentrations were usually too low to have a substantial impact on alkalinity. Alkalinity and salinity behaved conservatively in our experiments and showed no significant systematic variability on diurnal timescales (Wahl et al., 2015). Calcium concentrations ranged from about 3.5 to 6 mM and were closely coupled to salinity ($R^2 > 0.9$).

The saturation state in the benthocosm treatments with respect to the calcium carbonate of *S. spirorbis* tubes was calculated considering the shell composition. It has been shown that the thermodynamic stability of biogenic Mg-calcites differs from pure calcite (Plummer and Mackenzie, 1974; Busenberg and Plummer, 1989) and varies with the Mg content. The MgCO₃ content of *S. spirorbis* tubes is about 10 ± 1 mole% (Ni et al., 2018; Bornhold and Milliman, 1973). Unfortunately, the solubility of *S. spirorbis* has not yet been explicitly determined. According to different experimental studies biogenic Mg-calcite with about 10 mol% MgCO₃ has a solubility which is thermodynamically equivalent to aragonite (Walter and Morse, 1984; Morse and Mackenzie, 1990; Andersson et al., 2008). Therefore, the saturation state with respect to aragonite (Ω) was taken as an estimate for the Mg-calcite forming the *S. spirorbis* shell. It should, however, be kept in mind that the solubility of biogenic Mg-calcites may not only differ with shell composition, but may also depend on crystal ordering, trace element impurities and other mineralogical factors (Mackenzie et al., 1983). Most of these factors increase the solubility of Mg-calcite.

Saturation states in the benthocosms at the measured in situ temperatures and salinities were calculated from carbonate ion concentrations, calcium ion concentrations and the apparent solubility constant (K_{sp}^*) of aragonite (Mucci, 1983):

$$\Omega = [\text{Ca}^{2+}] \cdot [\text{CO}_3^{2-}] / K_{sp}^* \quad (1)$$

Only pH and temperature were measured with 2-hourly resolution (Wahl et al., 2015). All other parameters ([Ca²⁺], TA, salinity, Si, P) were interpolated to calculate diurnal variations of Ω (Fig. S1 in the Supplement). Linear interpolation

is justified by the conservative behaviour of these properties. The resulting 2-hourly resolved time series of Ω were used to estimate the mean saturation state and the percentage of time when treatments were undersaturated with respect to *S. spirorbis* tube Mg-calcite.

Average diurnal amplitudes of saturation state, pH and temperature were calculated as follows: the pH and temperature time series from Wahl et al. (2015) and the resulting saturation values have an interpolated resolution of 10 min. We averaged all values of the period of interest into 24 bins of 1 h length, resulting in a mean value for each hour of the day. The minimum and maximum values of the resulting mean diurnal cycle define the mean diurnal amplitude. The resulting values were averaged for each of the four different treatments. Each experimental period was subdivided into four sub-periods with durations of 17–19 days and mean diurnal amplitudes of each sub-period were calculated as explained above.

We use the daily insolation sum measured at the GEOMAR meteorological observatory², situated close to the benthocosms, to compare with the diurnal pH variations. Average daily insolation ranged from 0.2 kWh m⁻² in December to 6.6 kWh m⁻² in July (Fig. 3).

3 Results

3.1 Seawater carbonate chemistry and saturation state

Variations of physical and chemical parameters (TA, pH, temperature, salinity, etc.) in the control treatments of the four seasonal experiments are shown in Fig. S2. The calcium carbonate saturation state of the seawater (Ω) in all basins was dominantly controlled by the pH. Average diurnal cycles showed a minimum in pH and Ω around sunrise followed by a late afternoon maximum (Fig. S1). The pH values showed strong diurnal fluctuations in all treatments. Average day/night pH differences were smallest (< 0.05) in December 2013 and largest (up to 0.6) in June, July and August 2013 and February and March 2014 (Fig. 3). The pH amplitudes clearly follow insolation showing saturation behaviour at high insolation values, most pronounced in the ambient CO₂ treatments (Fig. 4). Generally, pH values declined from the spring to the autumn experiment and reached a minimum in November and December (Fig. 3).

Saturation states closely followed the pH dynamics. Average saturation was highest during the spring experiment when all treatments were generally oversaturated with respect to aragonite and Mg-calcite ($\Omega > 1$). Basin waters were undersaturated ($\Omega < 1$) only 6 to 51 % of the time during the spring experiment (Table 1). The lowest saturation states occurred during the autumn experiment with $\Omega < 1$ during 81 to 100 % of the experiment. Average autumn saturation ranged from 0.6 to 0.8 (Table 1). It was only slightly elevated during

²www.geomar.de/service/wetter

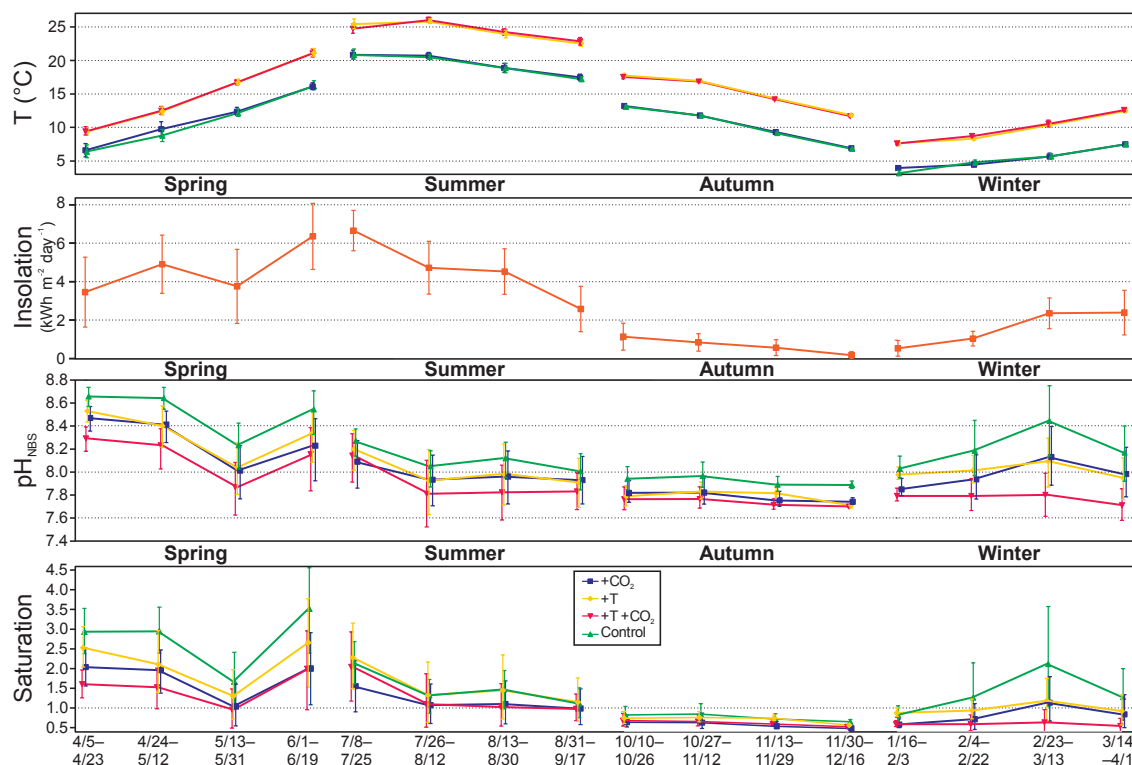


Figure 3. Average water temperature, daily insolation, pH and saturation state with respect to aragonite (as proxy for *S. spirorbis* Mg-calcite) in the four different treatments. Each of the four seasonal experiments is divided into four sub-periods lasting 17–19 days (start and end dates indicated at x axis). Error bars indicate minimum and maximum values of the mean diurnal cycle during the sub-periods, except for insolation where they indicate day-to-day variability (standard deviation). Insolation was measured at the GEOMAR meteorological observatory (www.geomar.de/service/wetter), about 100 m from the benthocosms.

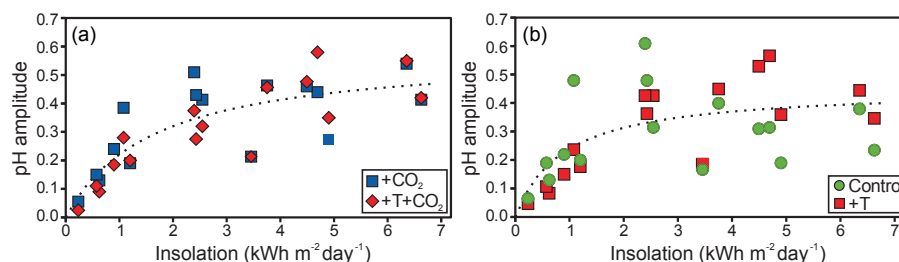


Figure 4. Light dependence of diurnal pH cycles. Average diurnal pH amplitudes in the benthocosm basins for CO₂-enriched (a) and ambient (b) treatments plotted versus the average daily insolation (as in Fig. 3) for the sub-periods of the four seasonal experiments. Dotted lines are Michaelis–Menten fits to the data, $y = A \cdot x / (B + x)$, with rate constants (A) of 0.5 and 0.6 and half-saturation constants (B) of 0.9 and 1.6, for ambient and CO₂-enriched treatments respectively.

daytime (Ω_{\max} of 0.6 to 1.1, Fig. 3). Average day–night differences in Ω largely tracked the diurnal pH amplitudes with smallest differences during the autumn experiment (< 0.1 in December 2013) and large fluctuations in February and March, June and July (up to 2.2, Fig. 3).

3.2 *Spirorbis spirorbis* tube size and ontogenetic cycle

The sizes that the *S. spirorbis* shells reached before the experiments in their natural environment are indicated by the initial diameters. They reflect the size distributions under natural conditions. In contrast, the final diameters of our specimens reflect changes from the initial sizes under experimental conditions. Note that only stained specimens were included in the analysis. Therefore, juveniles that settled during

Table 1. Average water data of the four treatments in the four seasonal experiments.

		Spring (April–June 2013)				Summer (July–September 2013)			
		A/B	C/D	E/F	Mean	A/B	C/D	E/F	Mean
+T + CO ₂	<i>T</i> (°C)	14.7 ± 4.6	14.5 ± 4.6	14.4 ± 4.7	14.5 ± 0.2	24.3 ± 1.9	24.2 ± 2.0	24.6 ± 2.0	24.4 ± 0.2
	pH _{NBS}	7.98 ± 0.27	8.23 ± 0.30	8.22 ± 0.28	8.1 ± 0.1	7.87 ± 0.24	7.91 ± 0.23	7.94 ± 0.27	7.9 ± 0.04
	Ω	1.05 ± 0.54	1.78 ± 0.98	1.68 ± 0.73	1.5 ± 0.4	1.19 ± 0.63	1.28 ± 0.63	1.43 ± 0.92	1.3 ± 0.1
	<i>t</i> _{Ω<1} (%)	51	24	20	32 ± 17	47	42	39	42 ± 4
+T	<i>T</i> (°C)	14.6 ± 4.7		14.5 ± 4.7	14.6 ± 0.1	24.3 ± 1.9	24.2 ± 2.0	24.3 ± 2.0	24.3 ± 0.1
	pH _{NBS}	8.22 ± 0.24	no data ^a	8.43 ± 0.31	8.3 ± 0.2	7.99 ± 0.19	8.00 ± 0.21	8.04 ± 0.27	8.0 ± 0.03
	Ω	1.69 ± 0.66		2.58 ± 1.19	2.1 ± 0.6	1.46 ± 0.62	1.49 ± 0.64	1.73 ± 0.98	1.6 ± 0.1
	<i>t</i> _{Ω<1} (%)	16		11	13 ± 4	31	29	30	30 ± 1
+CO ₂	<i>T</i> (°C)	11.1 ± 3.7	11.1 ± 3.8	10.6 ± 3.9	10.9 ± 0.3	19.3 ± 1.8	19.4 ± 1.8	19.3 ± 1.8	19.3 ± 0.1
	pH _{NBS}	8.24 ± 0.26	8.28 ± 0.29	8.34 ± 0.35	8.3 ± 0.1	7.96 ± 0.20	8.01 ± 0.17	7.97 ± 0.22	8.0 ± 0.02
	Ω	1.56 ± 0.78	1.73 ± 0.83	1.99 ± 1.06	1.8 ± 0.2	1.13 ± 0.51	1.23 ± 0.46	1.20 ± 0.57	1.2 ± 0.1
	<i>t</i> _{Ω<1} (%)	27	24	24	25 ± 2	46	36	46	43 ± 6
Control	<i>T</i> (°C)	10.7 ± 3.8	10.0 ± 4.3	10.9 ± 3.7	10.5 ± 0.5	19.2 ± 1.8	19.5 ± 1.8		19.4 ± 0.2
	pH _{NBS}	8.36 ± 0.22	8.61 ± 0.27	8.59 ± 0.35	8.5 ± 0.1	8.11 ± 0.18	8.14 ± 0.14	no data ^b	8.1 ± 0.02
	Ω	1.93 ± 0.73	3.06 ± 1.36	3.22 ± 1.71	2.7 ± 0.7	1.54 ± 0.64	1.58 ± 0.50		1.6 ± 0.03
	<i>t</i> _{Ω<1} (%)	11	6	11	9 ± 3	24	13		19 ± 8
		Autumn (October–December 2013)				Winter (January–March 2014)			
		A/B	C/D	E/F	Mean	A/B	C/D	E/F	Mean
+T + CO ₂	<i>T</i> (°C)	15.1 ± 2.5	15.0 ± 2.6		15.1 ± 0.1	9.8 ± 1.9	9.8 ± 2.1	<i>11.7 ± 1.1^d</i>	9.8 ± 0.0
	pH _{NBS}	7.71 ± 0.08	7.76 ± 0.09	no data ^c	7.7 ± 0.04	7.70 ± 0.11	7.84 ± 0.15	<i>7.84 ± 0.14^d</i>	7.8 ± 0.1
	Ω	0.58 ± 0.12	0.64 ± 0.14		0.6 ± 0.04	0.49 ± 0.12	0.68 ± 0.22	<i>0.68 ± 0.22^d</i>	0.6 ± 0.1
	<i>t</i> _{Ω<1} (%)	100	97		98 ± 2	100	89	<i>87^d</i>	95 ± 8
+T	<i>T</i> (°C)	15.1 ± 2.5	15.3 ± 2.5	15.2 ± 2.6	15.2 ± 0.1	9.6 ± 2.0	9.8 ± 2.1	9.4 ± 2.0	9.6 ± 0.2
	pH _{NBS}	7.83 ± 0.09	7.80 ± 0.20	7.72 ± 0.10	7.8 ± 0.06	8.05 ± 0.15	7.99 ± 0.17	7.98 ± 0.11	8.0 ± 0.04
	Ω	0.76 ± 0.13	0.75 ± 0.27	0.58 ± 0.11	0.7 ± 0.1	1.08 ± 0.37	0.96 ± 0.36	0.90 ± 0.22	1.0 ± 0.1
	<i>t</i> _{Ω<1} (%)	95	86	100	94 ± 7	53	68	76	66 ± 12
+CO ₂	<i>T</i> (°C)	10.3 ± 2.7	10.6 ± 2.5	<i>7.8 ± 1.2^d</i>	10.5 ± 0.2		5.2 ± 1.7	5.4 ± 1.5	5.3 ± 0.1
	pH _{NBS}	7.76 ± 0.05	7.81 ± 0.12	<i>7.79 ± 0.07^d</i>	7.8 ± 0.04	no data ^b	7.95 ± 0.22	7.99 ± 0.18	8.0 ± 0.03
	Ω	0.53 ± 0.07	0.61 ± 0.20	<i>0.55 ± 0.11^d</i>	0.6 ± 0.06		0.79 ± 0.47	0.84 ± 0.36	0.8 ± 0.04
	<i>t</i> _{Ω<1} (%)	100	93	<i>99^d</i>	97 ± 5		78	74	76 ± 3
Control	<i>T</i> (°C)	10.2 ± 2.6	10.4 ± 2.6		10.3 ± 0.1		5.2 ± 1.6	<i>6.9 ± 1.1^d</i>	5.2
	pH _{NBS}	7.88 ± 0.06	7.95 ± 0.11	no data ^c	7.9 ± 0.05	no data ^b	8.20 ± 0.24	<i>8.30 ± 0.24^d</i>	8.2
	Ω	0.69 ± 0.10	0.83 ± 0.23		0.8 ± 0.1		1.34 ± 0.82	<i>1.65 ± 0.80^d</i>	1.3
	<i>t</i> _{Ω<1} (%)	99	81		90 ± 13		48	<i>28^d</i>	48

Mean values for temperature, pH, saturation (Ω) and percent of experimental time when basins were undersaturated with respect to aragonite and Mg-calcite (*t*_{Ω<1}). Columns show mean values for single basins (A1, A2, B1, B2, etc.) and averages for each treatment with ±1 SD ranges. ^a pH data only for final 2 weeks of the experiment; ^b no pH data recorded; ^c no data recorded; ^d data only for final 4 weeks of the experiment. Data shown in italic font were not used for calculation of means.

the experiments and specimens that did not calcify during the staining were not included.

The final and initial diameters of 2782 stained and photographed *S. spirorbis* tubes from all four seasonal experiments were in a range of 0.2 to 4.0 mm (Fig. 5). The tube with the biggest final diameter (~4.0 mm) was found in the winter experiment. The smallest measured shell diameters (0.2 mm) occurred in summer and autumn (Fig. 5b, c). The size distributions of the shells indicate distinct populations that, in summer and autumn, were separated by a minimum in shell counts at a diameter of about 1.3 mm (Fig. 5b, c). Accordingly we classified *S. spirorbis* specimens into two general populations: “small” (diameter < 1.3 mm) and “large” (diameter > 1.3 mm). Seed et al. (1981) observed reproduction of

S. spirorbis at shell diameters > 1.9 mm. Therefore, our small populations consist of juveniles, while the large populations are mostly adults but may include immature specimens.

Large specimens were observed in the starting populations of all seasons (Fig. 5b–e), including spring and summer, which is in accordance with the maximum life span of *S. spirorbis* of about 1.5 years (Seed et al., 1981). Most small specimens grew to large sizes during the ~10 weeks of the experiments (Fig. 5f–i). The majority of *S. spirorbis* in the maximum size range (> 3 mm) were seen at the end of the winter and spring experiments (March, June; Fig. 5a, h, i). The initial shell diameters of the *S. spirorbis* autumn population showed a clear bimodal distribution (Fig. 5c). A juvenile population with a modal diameter of 0.6 mm (autumn-small)

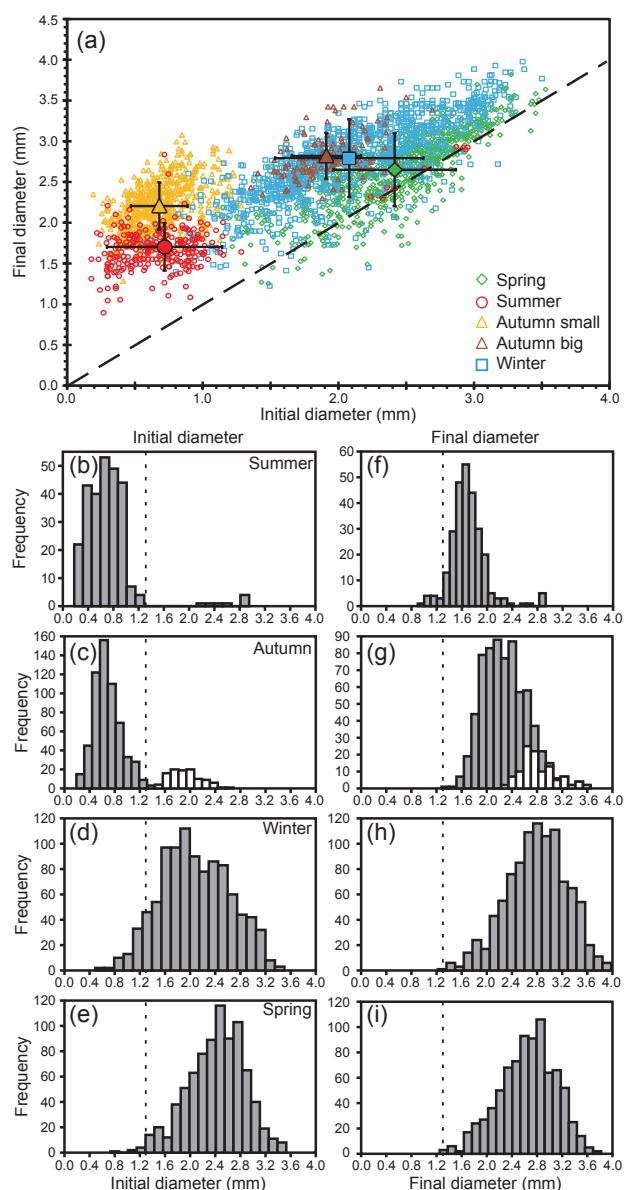


Figure 5. (a) Cross-plot of initial and final diameters of individual *S. spirorbis* tubes from all treatments and seasons. Note that only specimens that grew during the initial staining were included. Big symbols are seasonal mean diameters (\pm SD). The population of the autumn experiment was subdivided into a small and a big sub-population as described in the text. The tubes that plot above the dashed diagonal line of non-growth grew during the experiments. Note that most specimens of the spring and winter experiments plot close to this line indicating little growth. Specimens below the line showed deformations and irregular growth. (b–e, f–i) Size distributions of *S. spirorbis* at the start and end of the four seasonal experiments respectively. Vertical dashed lines at 1.3 mm indicate threshold diameter between small and large populations. White bars in (c, g) represent the autumn-big population. Frequency indicates number of specimens in each size class.

was clearly separated from an “adolescent/adult” population with a modal diameter of 1.8 mm (autumn-big). Initial diameters in the intermediate range of 1.4–1.5 mm were scarce. A similar size distribution was found in the summer experiment. However, the large population had very few specimens in summer (Fig. 5b).

Juveniles occurred in all four seasons but were rarely observed in winter and spring. The proportion of juveniles in the initial populations decreased from July (Fig. 5b) to April (Fig. 5e). Accordingly, the majority of the *S. spirorbis* specimens at the start of the summer and autumn experiments were in the juvenile stage (< 1.3 mm), while the winter and spring experiments were dominated by large specimens (Fig. 5). The modal initial diameter increased systematically with the sequence of the seasons from July (~ 0.7 mm, Fig. 5a, b) until April (~ 2.4 mm, Fig. 5a, e). The spring, winter and autumn-big populations started with similar initial diameters (modes of 1.8 to 2.5 mm, Fig. 5c–e) and all grew into a typical final diameter range (modes of 2.5 to 2.8 mm, Fig. 5g–i) representing the most common size of adult *S. spirorbis*.

As visible in Fig. 5, the diameter increase of *S. spirorbis* tubes during the experiments strongly depended on the season and the initial size distribution of the populations. Diameter increases ranged from $4 \mu\text{m day}^{-1}$ for the adult-dominated population in spring to $20 \mu\text{m day}^{-1}$ for the juvenile population in autumn. Modal diameter increases of the summer, autumn-big and winter populations were similar (Fig. 5) with values of about $10 \mu\text{m day}^{-1}$. This ontogenetic influence has to be taken into account when interpreting growth rates in terms of temperature and saturation state.

3.3 Tube microstructure

SEM pictures of *S. spirorbis* sections (Fig. 6) show a relatively rough and irregular outer tube wall surface whereas the inner surface is smooth. The internal wall structures consist of convex-forward lamellae or chevrons (Fig. 6b). New lamellae were laid down by the worm on the anterior tube surface, forming curved convex-forward layers, wrapping the end of the tube wall to completely cover the end of the anterior tube wall with a new layer. Thin crescent pores exist in the wall interior between the chevrons (Fig. 6b). These pores taper towards the inner and outer rims of the tube wall where the chevron lamellae fuse into a dense, calcium-rich wall (Fig. 6). The high calcium concentrations indicate that strongly calcified dense layers, not organic layers, armour the inner and outer tube wall surfaces.

A comparison of a cross section (Fig. 6a) and a longitudinal section (Fig. 6d) through *S. spirorbis* shells reveals the complex shape of the growth lamellae. The convex-forward layers are additionally curved upward, forming convex-upward lamellae in longitudinal sections. The convex-upward lamellae were built upward successively from the bottom on both sides of the tube and then converge

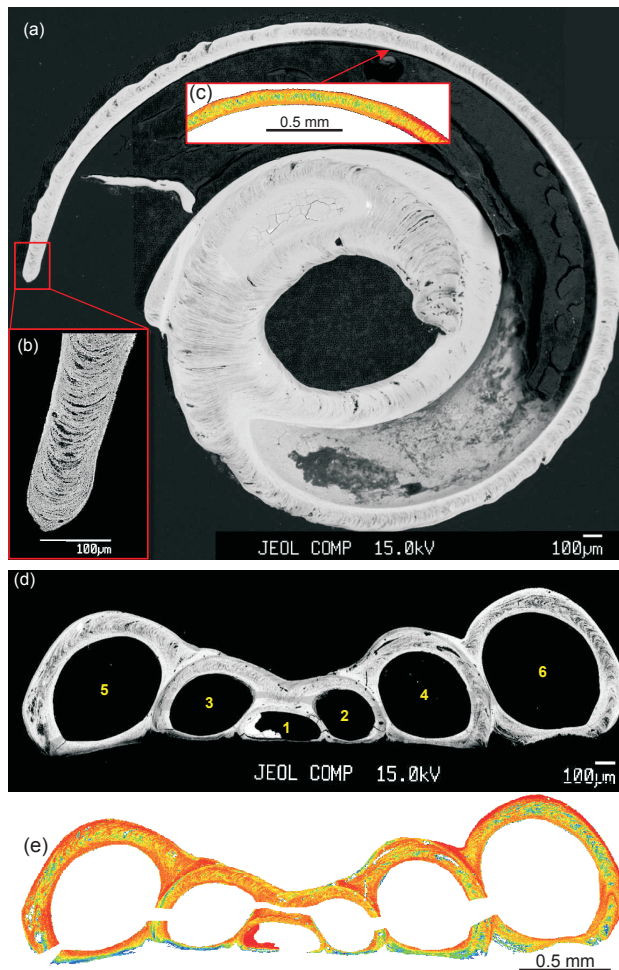


Figure 6. Backscatter SEM images (BEI) and electron microprobe (EMP) calcium maps of embedded and polished *S. spirorbis* specimens from the winter control experiment. (a) Cross section viewed from shell bottom. (b) Detail from area in red frame, outer wall at the tube mouth showing convex forward lamellae (chevron structure). (c) EMP calcium map of upper-right tube wall showing densely calcified outer layers along the inner and outer rim (red, high calcium concentration). Inner parts of tube wall are laminated and less calcified (yellow-green, low calcium concentration). (d) Longitudinal cross section. Numbers indicate order of tube whorls. 1: juvenile tube, partly filled with secondary dense material (left). 6: final whorl, added during the experiment. (e) Calcium EMP map showing densely calcified wall rims and laminated less calcified wall fillings. Later whorls partly coat the older tube with a thick calcium-rich shell layer. White areas cutting the tube walls are artefacts of data acquisition.

at the tube top. The growth direction is indicated by the convex layering.

In addition, the inner and outer sides of each convex-forward layer of the tube walls are asymmetric (Fig. 7). The fluorescent, stained skeleton outlines the pattern of lamellae which were accreted during the 3-day staining period. The

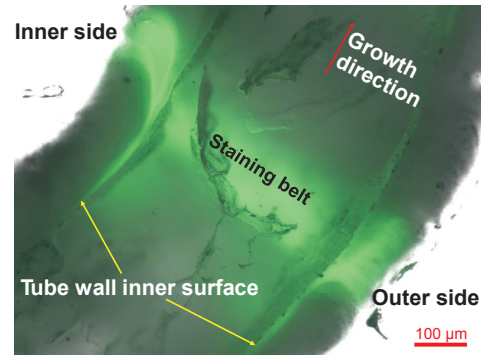


Figure 7. Fluorescent microscope image of *S. spirorbis* tube cross section with staining line showing green fluorescence. Sample from spring +CO₂ + *T* experiment. The green belt is the stained part of the shell that formed during the 3 days of calcein staining before the start of the experiment. Newly grown shell forms a lining along the inner tube wall surface (yellow arrows). Red arrow indicates the growth direction.

newly grown lamellae cover a large area along the inner tube wall surface, while little new material is attached to the outer tube wall surface.

The bottom of the tube, which was attached to the substrate, is relatively thin and characterised by parallel planar lamellae. An idealised sketch of the *S. spirorbis* tube structures is shown in Fig. 8. Where the wall of a new whorl attached to an older whorl it formed a thickened wedge-like structure partly filling the gap between the old and new whorl (Figs. 6d, 8). These wedges are usually calcium-rich, densely calcified (Fig. 6e), increasing the stability of the shell. The tube diameter of the whorls and the tube wall thickness generally increased as the *S. spirorbis* shell grew (Fig. 6d). The wall thickness ranges from about 30 to 180 µm. It is thicker in the fully developed shell parts and tapers towards the tube opening (Fig. 6a, b).

3.4 Shell corrosion

Shell corrosion (Fig. 9) occurred in all treatments during all seasons, but was most commonly observed in the high pCO₂ treatments of the autumn and winter experiments (Fig. 10, Table 2). In the basins of these treatments up to 75 % of the specimens showed corroded shells. On average, the proportion of corroded samples (P_{corr}) was highest in the autumn +CO₂ treatment and in the winter +CO₂ + *T* treatments, with treatment averages of 58 and 62 % respectively. In contrast, corrosion was nearly absent in the control treatments, where in all four seasons P_{corr} values were lower than 1.5 %. Additionally, corroded specimens were nearly absent in all spring treatments, except for the +CO₂ + *T* treatment.

The percentage of corroded samples was clearly related to the saturation state (Fig. 10). Except for one basin from the spring +CO₂ + *T* experiment P_{corr} was below 10 % when

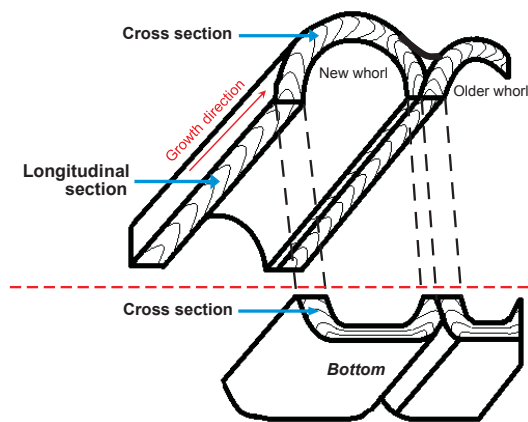


Figure 8. Schematic line drawing of the shell structures of *S. spirorbis* showing the orientations of cross and longitudinal sections (blue arrows) and the respective orientation of the chevron lamellae. Red arrows indicate growth direction. The parts below the red dashed line are only visible in the longitudinal sections. Tube wall thickness is about 100 μm .

Table 2. Corroded sample percentages (%) and number of corroded specimens (*n*).

Basin	Spring		Summer		Autumn		Winter	
	%	<i>n</i>	%	<i>n</i>	%	<i>n</i>	%	<i>n</i>
A1 (+ <i>T</i> + CO ₂)	36.8	7	–	–	60.8	31	74.3	75
A2 (+ <i>T</i>)	0.0	0	–	–	3.5	2	0.0	0
B1 (+CO ₂)	0.0	0	0.0	0	58.9	33	23.0	23
B2 (control)	0.0	0	2.0	2	2.7	2	1.0	1
C1 (+ <i>T</i> + CO ₂)	14.5	16	–	–	16.1	9	64.5	20
C2 (+ <i>T</i>)	0.0	0	–	–	5.8	3	6.4	5
D1 (+CO ₂)	2.0	1	0.0	0	57.4	31	35.8	24
D2 (control)	0.0	0	0.0	0	0.0	0	0.0	0
E1 (+ <i>T</i> + CO ₂)	5.0	2	–	–	28.6	16	42.5	31
E2 (+ <i>T</i>)	0.0	0	–	–	22.0	9	6.7	6
F1 (+CO ₂)	2.2	1	10.0	1	58.1	25	17.9	17
F2 (control)	0.0	0	1.3	1	1.2	1	0.7	1

No specimens were recovered from elevated temperature treatments in summer.

average saturation (Ω) was above 1. For average saturation $\Omega > 2$ corroded shells were completely absent. On the other hand, although $P_{\text{corr}} = 0\%$ was observed in basins with an average saturation as low as 0.8 (basin D2, autumn control), corrosion frequencies generally increased in undersaturated basins. For $\Omega < 1$ we observed a significant inverse correlation between P_{corr} and saturation state:

$$P_{\text{corr}}(\%) = -143 \pm 72 \cdot \Omega + 131 \pm 51; \quad R^2 = 0.54; \\ n = 17; \quad p < 0.0008. \quad (2)$$

Notably, shells grew significantly even in undersaturated waters. Thus corrosion selectively affected the previously grown parts of the shell (Fig. 9b).

P_{corr} was independent of temperature in autumn, winter and spring ($R^2 = 0.03$, $n = 42$, $p = 0.28$), but temperature

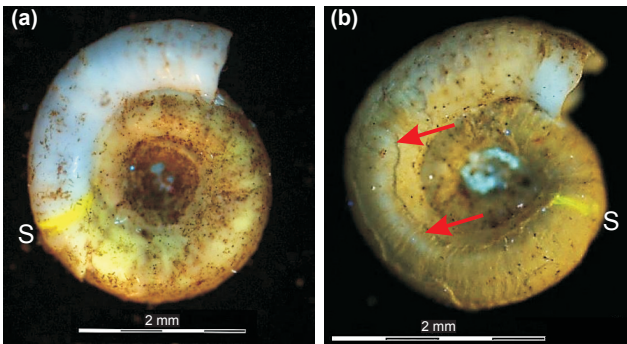


Figure 9. Pristine and corroded *S. spirorbis* shells. (a) Pristine smooth surface without visible corrosion. Specimen from winter control experiment. (b) Corroded surface of a tube from winter +CO₂ + *T* experiment. The outermost shell layer was removed by corrosion, exposing the ring structure of the underlying shell layer (arrows). “S” indicates position of the stain line (start of experiment). Note that the specimen in (a) had a larger initial diameter than the specimen in (b), but grew a shorter new tube segment during the experiment.

may have fostered corrosion and bioerosion in the summer experiments. Ambient temperature treatments of the summer experiments showed very low P_{corr} values (Fig. 10). However, the few recovered samples from the elevated temperature experiments were highly corroded and showed very little net growth. Unfortunately, because very few specimens were recovered from these treatments of the summer experiment, P_{corr} values could not be determined.

Strong bioerosion by microborers was observed in a cross section of a summer control specimen. Numerous microborings of about 5 to 45 μm diameter affected the outer tube wall (Fig. 11). The microborings penetrated the whole tube wall. This is in contrast to the shell corrosion of the other seasons, which mostly affected the outermost layer of the tube wall (Fig. 9b).

3.5 Growth rate

The length of new tube segments that grew during an experiment (Fig. 2: “growth”, Gr) varied considerably between populations and seasons, ranging from less than 0.1 up to 7.3 mm. This corresponds to a range in growth rates of 1 to 100 $\mu\text{m day}^{-1}$. The longest newly grown tube in all experiments (7.3 mm) occurred in the autumn-small population. Growth was found to be inversely correlated with the initial diameter of the shells (D_i), i.e. smaller tubeworms generally grew faster than bigger ones (Fig. 12). The correlation is highly significant:

$$\text{Gr (mm)} = -1.1 \pm 0.05 \cdot D_i \text{ (mm)} + 5.17 \pm 0.09; \\ R^2 = 0.41; \quad n = 2783; \quad p = 0; \quad (3)$$

for D_i ranging from 0.2 to 3.5 mm.

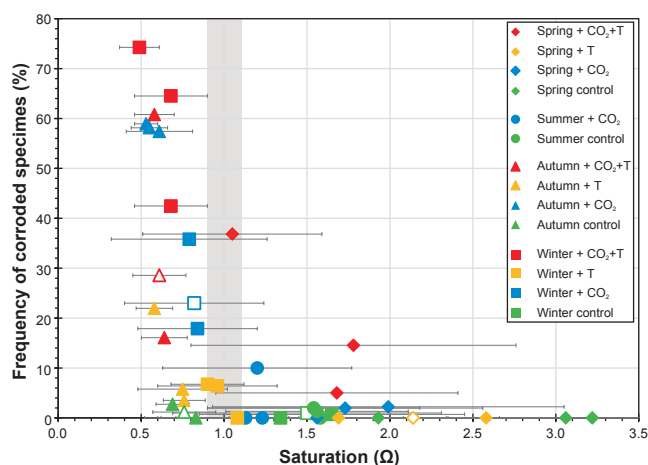


Figure 10. Proportion of corroded samples as a function of the calcium carbonate saturation state of seawater. Each data point represents one basin. Grey bar indicates saturated water ($\Omega = 1.0 \pm 0.1$). Error bars are standard deviations of saturation data for each basin (Table 1). For basins without available saturation data the treatment averages were used (open symbols).

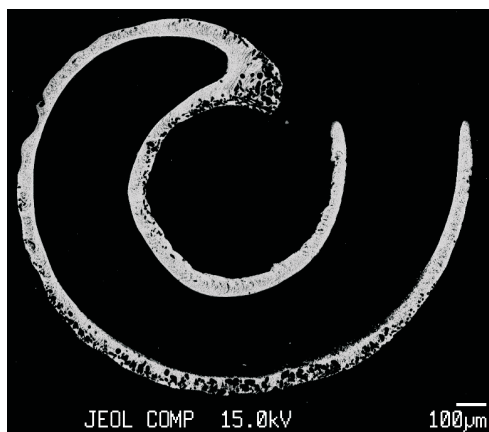


Figure 11. SEM image (BEI) of polished cross section of *S. spirorbis* shell from summer control experiment. Dark spots are microborings mostly affecting the outer tube wall.

Growth of the winter populations showed the highest variability of all treatments, ranging from 0.4 to 6.3 mm (Fig. 12). Growth rates and initial sizes in winter were similar to those of the autumn-big populations. This indicates that the tubeworms from these two experiments were in the same developing stage, although they represented different generations of *S. spirorbis* populations (Sect. 3.2, Fig. 5).

In a subset of specimens from the spring and autumn (control, +CO₂ + T) and the summer (control, +CO₂) experiments average weights of newly grown tube segments, W_t , were determined (Table S1 in the Supplement). The results show similar weight increases in spring and summer of 0.1–0.9 and 0.2–0.6 mg shell⁻¹ respectively. In contrast, W_t val-

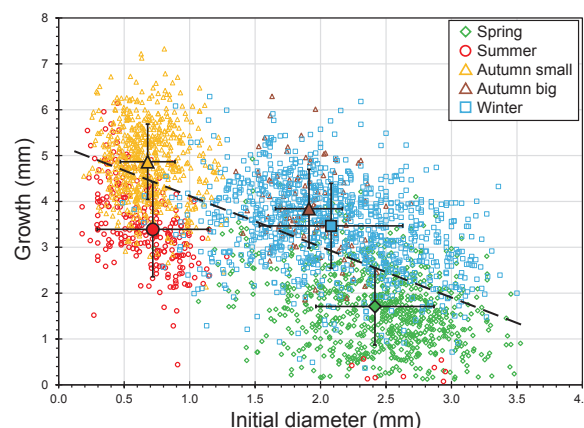


Figure 12. Length of new tube growth during the experiments plotted against initial diameters of all measured worm tubes. The dashed line is a linear fit to the data ($R^2 = 0.41$, $n = 2783$, $p = 0$). Data are from all experiments and treatments. Small symbols indicate individual *S. spirorbis* specimens, while the larger symbols show the seasonal mean values (± 1 SD). Autumn-small and autumn-big populations are plotted separately.

ues in autumn were significantly larger, ranging from 1.2 to 2.1 mg shell⁻¹. As visible in Fig. 12, Gr varied seasonally (Fig. 12). For the weighed specimens mean Gr ranged from 2.2 to 3.8 mm and 3.8 to 5.4 mm in summer and autumn respectively. It was only 1.0 to 2.3 mm in spring. We accordingly normalised W_t by Gr. This resulted in overlapping W_t /Gr ranges for spring and autumn of 0.1–0.4 and 0.3–0.4 mg mm⁻¹ of tube respectively (Fig. 13). The summer shells increased their weights by only 0.1–0.2 mg mm⁻¹ of tube.

Generally, this is in agreement with smaller final shell sizes in summer (Fig. 5) and consequently smaller tube widths (Fig. 2). Assuming a cylindrical tube geometry and a constant wall thickness of 0.1 mm the measured tube width values (Table 3) allow for the estimation of average shell densities: 1.1 ± 0.3 and 1.8 ± 0.3 g cm⁻³ (± 1 SD) for the summer and autumn specimens respectively. This indicates that density and/or tube wall thickness of the summer tubes was 38 ± 13 % lower compared to the autumn tubes. The difference is significant (t test, $p = 0.005$).

3.5.1 Treatment effects

Only very few *S. spirorbis* specimens, most of which were broken and strongly damaged, could be recovered from the elevated temperature treatments (+T, +CO₂ + T) of the summer experiment. Growth of broken and damaged tubes was not measured (indicated by “no data” in Fig. 14). In these experiments a temperature-driven collapse of the grazer community had caused epiphytic overgrowth of *Fucus* thalli and *S. spirorbis* tubes leading to an increased mortality (Werner et al., 2016). Except for these elevated temperature sum-

Table 3. Mean size and growth parameters of juvenile *S. spirorbis* populations in summer and autumn.

Season	Basin	Count	D_i (mm)	D_f (mm)	Gr (mm)	TbWd (mm)
Summer control	B2	18	0.44 ± 0.07	1.70 ± 0.23	4.57 ± 0.82	0.55 ± 0.07
	D2	19	0.40 ± 0.08	1.53 ± 0.20	3.53 ± 0.64	0.51 ± 0.08
	F2	43	0.42 ± 0.20	1.73 ± 0.20	4.36 ± 1.03	0.55 ± 0.06
Summer +CO ₂	B1*	6	0.37 ± 1.06	1.54 ± 0.58	2.45 ± 1.66	0.51 ± 0.09
	D1*	3	0.18 ± 0.07	1.46 ± 0.12	3.88 ± 0.44	0.45 ± 0.08
	F1*	10	0.33 ± 0.21	1.27 ± 0.67	2.95 ± 0.56	0.44 ± 0.09
Autumn control	B2	13	0.60 ± 0.07	2.09 ± 0.19	4.97 ± 0.51	0.73 ± 0.08
	D2	19	0.48 ± 0.09	1.93 ± 0.21	4.62 ± 0.67	0.69 ± 0.09
	F2	20	0.58 ± 0.07	2.05 ± 0.22	4.67 ± 0.80	0.69 ± 0.08
Autumn +CO ₂	B1	28	0.54 ± 0.09	2.06 ± 0.18	4.41 ± 0.49	0.76 ± 0.07
	D1	28	0.54 ± 0.11	1.98 ± 0.22	4.53 ± 0.43	0.73 ± 0.08
	F1	30	0.54 ± 0.11	2.19 ± 0.21	5.20 ± 0.55	0.76 ± 0.07

Selected sub-populations with homogenous initial diameter range (similar median D_i) from summer and autumn control and +CO₂ treatments. Values are averages and standard deviations. D_i : initial diameter, D_f : final diameter, Gr: growth, TbWd: tube width. * Insufficient data to select sub-population with D_i similar to other treatments.

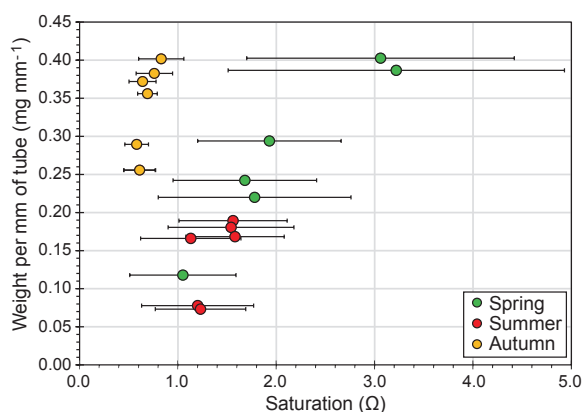


Figure 13. Weights of *S. spirorbis* tube segments that grew during the spring, summer and autumn experiments. Data points are average weights per millimetre of tube of selected basins (Table S1), plotted against average saturation state. Vertical error bars ($\pm 1\%$ of measured weight) are smaller than the symbols. Horizontal error bars (± 1 SD) represent the variability of saturation in each basin during the experiments.

mer treatments there was no significant treatment influence (pCO₂ or T) on growth in spring, summer, winter or autumn-big populations. Notably, elevated pCO₂ had no detectable influence on growth in any of the four seasonal experiments (Fig. 14).

In the autumn-small population, temperature caused a significant increase of growth, but only under ambient pCO₂ conditions (Fig. 14; three-way ANOVA and Tukey's HSD tests, $p = 0.03$ for Gr/ D_i , $p = 0.04$ for D_f/D_i). There were marginally significant interactions ($p < 0.1$) among the factors temperature, pCO₂ and season. Each factor influenced

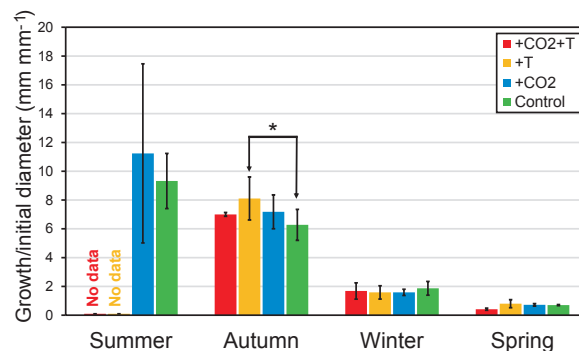


Figure 14. Average growth (Gr/ D_i) in different treatments during seasonal experiments. In autumn, growth differed significantly between the + T and control treatments. The effect is only significant in the small sub-population, while the autumn-big sub-population showed no significant temperature effect. However, the small sub-population dominates the autumn population. Thus the total population shows a significant temperature effect. In summer, no tubes were recovered from the elevated temperature treatments. Results from three-way ANOVA and Tukey's HSD tests; * significant difference ($p < 0.05$).

the growth parameters in each experiment differently due to the effects of the other two factors.

No significant correlation between saturation (Ω) and growth parameters (Gr, Gr/ D_i , D_f/D_i) was found ($p > 0.14$ to 1.0, $R^2 = 0.00$ to 0.40, seasonal basin data in Tables 1 and S2). The extension rates of *S. spirorbis* tubes were not negatively impacted by the saturation state of seawater. In contrast, weight increase (W_t /Gr, Table S1, Fig. 13) showed a significant positive correlation with saturation state in spring ($R^2 = 0.94$, $n = 6$, $p = 0.002$) and autumn ($R^2 = 0.68$, $n = 6$, $p = 0.04$). Weight increases of the autumn specimens

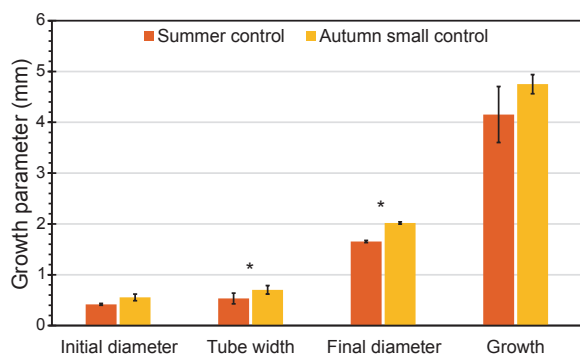


Figure 15. Size and growth of juvenile populations in control treatments of summer and autumn experiments. Two-way ANOVA and Tukey's HSD tests; * significant difference ($p < 0.05$).

were similar to spring, although the tubes formed in under-saturated water. In the summer experiment no significant correlation was observed ($R^2 = 0.48$, $n = 6$, $p = 0.13$), but the data lie close to the spring trend line ($R^2 = 0.88$, $p < 10^{-5}$ for summer and spring combined, Fig. 13).

3.5.2 Differences between seasonal experiments

The term “seasonal” in this study collectively describes differences between the four seasonal experiments which are not influenced by treatment effects. They may comprise truly seasonal variations but also variability on different timescales.

As described in Sect. 3.2 we observed a significant seasonal variation in the proportion of juvenile specimens (Fig. 5), indicating limited reproductive activity during the cold seasons. In spring and winter less than 10 % of the stained specimens were juveniles ($D_i < 1.3$ mm) while there were more than 84 % juveniles in summer and autumn. As a consequence D_i values were seasonally biased, which can explain at least some of the seasonal variations of Gr (Fig. 12).

In order to detect additional seasonal impacts on *S. spirorbis* tube growth we compared the juvenile populations in the control treatments of the summer and autumn experiments. Populations with similar mean D_i were selected (Table 3). No significant seasonal impact on growth (Gr) was found (Fig. 15). However, the final diameters of the autumn-small population were significantly larger than those of the summer experiment (Tukey's HSD test, $p < 0.01$, Fig. 5f–g). Additionally, the width of the newly grown tubes (Fig. 2) differed significantly between the two seasons (Tukey's HSD test, $p < 0.001$). The tubes that formed in autumn were wider than the summer tubes. Two-way ANOVA of tube width values from the control and +CO₂ treatments of the two seasons (Table 3) indicated no treatment effect ($p = 0.21$) but a significant seasonal impact ($p < 0.0001$).

There was no significant difference in growth of the large populations between the winter and autumn experiments (Fig. 12). There was no influence of temperature, pCO₂ or season on Gr/ D_i of these populations (three-way ANOVA, $p > 0.67$). All populations that had large sizes at the start of the experiments (spring, autumn-big, winter) grew to a similar final size distribution at the end of the experiments (Fig. 5g–i). Consequently, because the initial diameters of the winter and autumn-big populations were generally smaller compared to spring (Fig. 5c–e), average growth was higher in autumn and winter than in spring (Fig. 12).

4 Discussion

As shown in Figs. 3 and S2 there was strong intra- and inter-experimental variability in several environmental parameters, most prominently temperature, insolation, pH and saturation state, but also salinity and nutrient availability. Further, food supply and faunal/floral composition varied during the experiments as discussed below (Sect. 4.5) and shown in Werner et al. (2016). This natural variability is an intentional part of the benthic mesocosm set-up as it allows one to consider the dynamics of benthic communities reacting to environmental changes under near-natural boundary conditions (Wahl et al., 2015, 2016). On the other hand, not controlling for several environmental parameters also has drawbacks for the interpretation, comparability and reproducibility of results from different seasonal experiments. As described in Sect. 2.3, we use the term “seasonal factors” to collectively describe variations of experimental conditions between the four experiments, including environmental parameters and the ontogenetic development of *S. spirorbis*. While some of these factors are clearly dominated by seasonal change (e.g. light, temperature), others may vary on different timescales. Without multi-annual replicates we can not prove the seasonal nature of the observed changes in *S. spirorbis* growth between the four experiments. We therefore use the term “seasonal” as a simplifying descriptor of inter-experimental changes, although their seasonal nature needs to be verified in future multi-annual experiments.

4.1 Water chemistry

The aim of the study was to detect influences of elevated pCO₂ and temperature on growth and destruction of calcareous tubeworm shells under near-natural conditions in different seasons. The temperature manipulations produced consistent offsets of 4–5 °C between the respective treatments (Fig. 3). However, the basin water acidification (pH, saturation state Ω) induced by elevated pCO₂ was more complex. The average pH and Ω values were highest in the control treatments and lowest in the +CO₂ + T treatments. Intermediate values occurred in the +CO₂ and + T treatments. At the same pCO₂ level, pH was lower in the elevated temperature

treatments (Table 1; Wahl et al., 2015). This was probably caused by biological activity or nutrient cycling. It cannot be explained by the carbonate chemistry, which would result in higher pH at elevated temperatures under otherwise constant conditions (Lewis and Wallace, 1998). The mean pH difference between the +CO₂ + *T* and the control treatments was 0.2 units in summer and autumn and 0.4 units in spring and winter (Table 1). This simulated pH change is in good agreement with the predicted pH decrease at the end of this century (Omstedt et al., 2012; IPCC, 2014).

The seasonal fluctuations of pH (0.4 to 0.6) and Ω (0.9 to 1.9) exceeded the respective differences between treatments (pH: 0.2 to 0.4, Ω : 0.2 to 1.2). This has to be considered when comparing data from different seasons (e.g. Fig. 13). In addition, the strong diurnal cycles of pH (≤ 0.6) and saturation (≤ 2.2) complicate interpretations of carbonate chemistry impacts on tube growth and corrosion (e.g. Fig. 10).

Such interpretations are further hampered by potential impacts from the diffusive boundary layer (DBL) forming at the surface of *Fucus*, the substrate of *S. spirorbis* tubes (Spilling et al., 2010; Wahl et al., 2016). Photosynthetic activity during the day can elevate pH and saturation state in the algal DBL compared to the bulk fluid. Average saturation of the autumn experiment was as low as $\Omega = 0.6$ in the bulk fluids of some treatments. To elevate saturation from this value to slight oversaturation ($\Omega = 1.1$) pH has to be increased by $\log(1.1/0.6)$, i.e. by about 0.3 pH units. A pH elevation of this magnitude was reported by Wahl et al. (2016) at a DBL thickness corresponding to the height of *S. spirorbis* tubes. However, these observations were made in stagnant water while conditions in the benthocosms were quite turbulent due to artificial waves generated every 2 min (Wahl et al., 2015). Additionally, considering that insolation was reduced during the autumn experiment (Fig. 3), it appears unlikely that photosynthesis-driven daytime pH elevation (Fig. 4) was sufficient to overcome undersaturated water conditions in the DBL. This means that *S. spirorbis* was able to build tubes with above-average rates (Gr of ~ 5 mm, Table S2; Fig. 12) in spite of constant undersaturation ($t_{\Omega < 1} = 100\%$) in the autumn treatments A1, B1 and E2 (Table 1). *S. spirorbis* tube growth in undersaturated water was previously observed by Saderne and Wahl (2013).

4.2 Reproduction and life cycle

S. spirorbis reproduces and releases larvae predominantly during the warm seasons (Knight-Jones and Knight-Jones, 1977; Seed et al., 1981). Larvae settle in episodic pulses that may be coupled to fortnightly lunar or tidal cycles (De Silva, 1967; Daly, 1978). The episodic larval settlement provides an explanation for the presence of distinct populations in our experiments (Fig. 5b–e). In line with previous studies, we found living (actively calcifying) juveniles at the beginning of all four seasonal experiments, i.e. in January, April, July and October. This indicates that the Eckernförde Bay

S. spirorbis population reproduces throughout the year although juveniles were very rare in January and April. At the end of the summer and autumn experiments (September and December respectively) we found numerous unstained living *S. spirorbis* specimens on the *Fucus* thalli which had shell diameters < 1.3 mm. These juveniles obviously had settled during the experiments, indicating continuous reproduction in the benthocosms from July to December.

In addition to temperature, fecundity of *S. spirorbis* is affected by salinity and food supply and increases with individual size and age (Daly, 1978; Kupriyanova et al., 2001). Salinity fluctuated strongly during the experiments (from ~ 10 PSU in June to ~ 20 PSU in November–January, Fig. S2), but our data do not allow one to draw conclusions about salinity impacts on reproduction. Food supply for the filter-feeding Baltic tubeworms is generally lower in winter and increases when increased light availability promotes phytoplankton growth in spring, summer and autumn. Juveniles were rare in the initial populations in April, when the water temperature was still $< 10^\circ\text{C}$ and in January when temperatures had decreased to $< 5^\circ\text{C}$ (Fig. 3). In April, however, phytoplankton biomass is already high in the Kiel Bay area (Rheinheimer, 1996). Therefore, temperature probably dominates over food availability in controlling *S. spirorbis* reproduction in Eckernförde Bay.

4.3 Microstructures

The *S. spirorbis* investigated in this study displays the typical chevron lamellae microstructure (Fig. 6) that has been reported for *Spirorbis spirorbis* (Ippolitov and Rzhavsky, 2015) and for many other serpulid species (e.g. Wrigley, 1950; Hedley, 1958; Burchette and Riding, 1977; Weedon, 1994; Buckman, 2015). We observed a complex three-dimensional shape of the *S. spirorbis* chevron lamellae with convex-forward curving layers that show convex-upward curving substructures (Fig. 8).

S. spirorbis tube walls are purely calcitic and two-layered with an irregularly oriented prismatic (IOP) ultrastructure in the chevrons of the wall's core and a spherulitic prismatic ultrastructure (SPHP) of the thin outer wall region (Ippolitov and Rzhavsky, 2015; Vinn et al., 2008). The IOP chevrons and the SPHP structure are also common in a range of other serpulid genera (*Crucigera*, *Floriprotis*, *Pyrgopolon*, *Spiraserpula*; Gee and Knight-Jones, 1962; Vinn, 2011). Weedon (1994) pointed out that this complex internal tube architecture is difficult to explain with simple pasting models for serpulid calcification, i.e. secretion of calcium carbonate granules or of a mucus paste with small calcite crystals that are molded into the calcitic tube. It is likely that extracellular organic matrices and scaffolds play a role in tubeworm biocalcification (Tanur et al., 2010). Thin layers of organic matrix could be secreted onto the surface of the growing shell, as indicated by the chevron-like pores between growth lamellae (Fig. 6b).

Chevron-like accretion of new tube lamellae is indicated by the shape of the shell's stain line (Fig. 7). The figure additionally shows that synchronously with the accretion of new chevron lamellae new material was added in a thin layer to the inner tube wall. This wall thickening is in agreement with the observed tapering of the tube walls near the tube mouth (Fig. 6a, b). The asymmetric chevron lamellae structure of the *S. spirorbis* shells reported here (Fig. 7) has not been recorded previously in serpulid tubes. It shows that the inner and outer tube wall linings are differently constructed. *S. spirorbis* prefers to consolidate the inner surface of the tube while constructing new chevron layers.

In many *S. spirorbis* specimens the chevron lamellae of the central tube wall became visible as ring structures when the outer tube wall layer broke off or dissolved (Fig. 9). The outer tube wall layer appears to be susceptible to corrosion in spite of its massive densely calcified nature (Fig. 6).

4.4 Shell corrosion

In a recent study Saderne and Wahl (2013) incubated *S. spirorbis* in a laboratory experiment for 30 days at three different pCO₂ levels (450 µatm, $\Omega = 1.8$; 1200 µatm, $\Omega = 0.8$; 3150 µatm, $\Omega = 0.4$). They used specimens from the same site as in the current study, i.e. Eckernförde Bay. The tubes exhibited substantial dissolution at the highest pCO₂ conditions (3150 µatm, $\Omega = 0.4$), but not in experiments with lower pCO₂, even though waters were slightly undersaturated with respect to *S. spirorbis* calcite (1200 µatm, $\Omega = 0.8$). In contrast, in the current study corrosion of *S. spirorbis* shell surfaces was common (> 10 % of the shells) when average seawater saturation state was below ~ 0.9 , indicating corrosion starting even under mildly undersaturated conditions. Shell corrosion increased with decreasing saturation when the seawater was undersaturated ($\Omega < 1$; Fig. 10), but occurred in only a few experiments for $\Omega > 1$. It was completely absent at $\Omega > 2$.

The *S. spirorbis* tubes in our experiments may have been more susceptible to shell corrosion compared to those of Saderne and Wahl (2013) for several reasons. First, the duration of the benthocosm experiments was much longer (> 70 days) than the laboratory experiments (30 days). Second, during our experiments saturation state in the benthocosms fluctuated strongly between day and night (Fig. 3). Even with a mean saturation of 1 the shells may have been exposed to strongly undersaturated water during nighttime. Saderne and Wahl (2013) did not record pH or Ω on diurnal timescales, but the low biomass (1 g per 0.6 L flask) and constant vigorous gas bubbling most likely prohibited strong diurnal fluctuations of saturation states in their experiments. Third, with the more natural conditions in the benthocosms (unfiltered seawater, presence of natural fauna and flora) bioerosion by microbes may have fostered corrosion of the shells. Therefore, our experiments indicate that under natural conditions *S. spirorbis* can be significantly affected

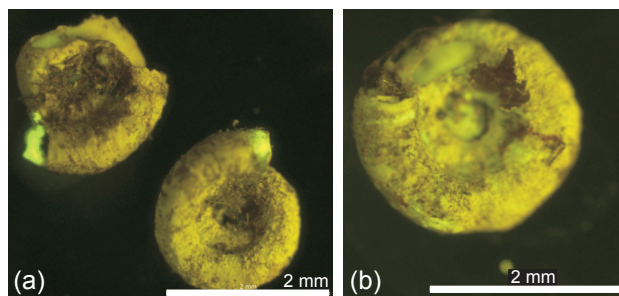


Figure 16. Strongly bio-eroded and broken *S. spirorbis* tubes from summer +*T* experiment (a: basin A2; b: basin C2). Tubes are partly covered by filamentous algae. Note spongy appearance of tubes due to intense microboring. Stain lines indicating the start of the experiment are visible at the tube mouths in the left picture. Scale bars are 2 mm.

by shell corrosion at acidification levels expected for the end of the century.

Corrosion in the mostly undersaturated waters of all autumn and the high-CO₂ winter experiments ($t_{\Omega < 1} > 70\%$; Table 1, Fig. 10) was likely induced by mineral dissolution. In contrast, during the summer experiment when waters were mostly oversaturated ($t_{\Omega < 1} < 50\%$) the tubes were affected by bioerosion. Boring organisms play an important role in the ecology of many marine habitats (Warme, 1977). Microborings were observed in a *S. spirorbis* shell from the summer control experiment. They probably affected the stability of the worm tube (Fig. 11). The few tubes recovered from residual *Fucus* thalli in the summer experiments with elevated temperatures (+*T* and +CO₂ + *T*) were mostly broken and strongly corroded (Fig. 16). These observations hint at a detrimental influence of elevated summer temperatures on *S. spirorbis* shells, either directly by affecting the worm's metabolism or indirectly through the reduction of grazing organisms (Werner et al., 2016) and increased anti-fouling activities of the *Fucus* host plants (Raddatz et al., 2017). Additionally, irreversible damage of *Fucus* algae at high summer temperatures (> 27 °C; Graiff et al., 2015a) leads to substratum loss for *S. spirorbis*, which preferentially settle on *Fucus* (De Silva, 1962; O'Connor and Lamont, 1978).

S. spirorbis tubes that grow during the warm season might be especially susceptible to mechanical stress and bioerosion. As shown in Sect. 3.5, summer tubes were significantly lighter than expected for their size. This indicates thinner and/or less dense tube walls compared to autumn and spring specimens. With the increasing frequency and duration of summer heatwaves in central Europe predicted for the 21st century (Beniston et al., 2007; Gräwe et al., 2013), increased bioerosion and loss of substratum could severely affect future *S. spirorbis* populations in the Baltic Sea.

4.5 Growth rate

Tube growth rates of *S. spirorbis* in our experiments were strongly controlled by the ontogenetic development. Growth rates were highest for juveniles and decreased when the worms got older and the tubes reached the maximum diameter range (Fig. 12). Similar growth–age relationships were found previously for *S. spirorbis* and other serpulid worms (O'Connor and Lamont, 1978; Kupriyanova et al., 2001; Riedi, 2012).

However, if only large specimens of similar initial sizes are considered, *S. spirorbis* tubes grew more rapidly in autumn and winter (October–March) than in spring (April–June), with the slowest growth occurring in the large summer population (July–September, Fig. 12). We know of no comparable published data. Reports of more rapid tube growth in summer compared to winter for several temperate serpulid species (Riedi, 2012; Kupriyanova et al., 2001) usually refer to the ontogenetic effect described above, i.e. enhanced growth of juvenile serpulids. The enhanced cold season growth of large *S. spirorbis* in our experiments was quite unexpected. Water was frequently undersaturated with respect to Mg-calcite during autumn and winter (48 to 98 % of experimental time, compared to 9 to 43 % in spring and summer, Table 1, Fig. 3). Other factors may play a role, like food availability and salinity. Food supply is generally lower in winter than during the warm seasons when increased light availability promotes phytoplankton growth. It consequently provides no explanation for the observed enhanced cold season growth of large *S. spirorbis*. Salinity was high during November–January and lowest in June. Spring salinities between 10 and 15 PSU contrasted with autumn and winter values between 16 and 21 PSU (Fig. S2). Enhanced calcification at higher salinities was previously observed in Baltic bivalves (Hiebenthal et al., 2012) and may potentially provide an explanation for enhanced growth of large *S. spirorbis* during the autumn and winter experiments. No significant treatment effects on growth were detected (see below). Consequently, the cold season growth enhancement of large *S. spirorbis* is not an artefact of increased temperatures and pCO₂ levels in the benthocosms. We suggest that, in addition to possible salinity effects, the reduced growth of adult specimens during the warm seasons reflects enhanced reproductive activity during this time (Knight-Jones and Knight-Jones, 1977; Seed et al., 1981), re-allocating energy resources from calcification to reproduction and thus reducing tube growth capacities.

The influence of increased pCO₂ on the growth of *S. spirorbis* worm tubes was previously studied in the experiments of Saderne and Wahl (2013). A significant growth rate reduction was only observed for large specimens at the highest pCO₂ (3150 µatm, $\Omega = 0.4$). No significant growth rate reduction was found at the intermediate pCO₂ level of 1200 µatm. In agreement with these results we found no significant change in tube growth parameters (Gr, Gr/D_i, D_f/D_i) when elevating pCO₂ from ambient levels to

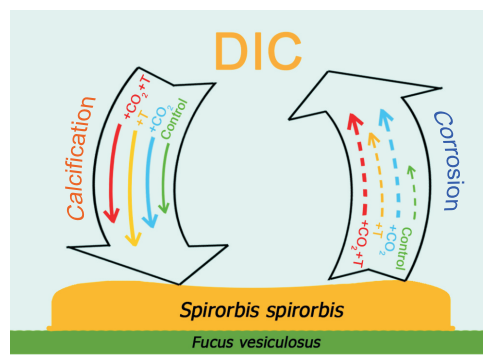


Figure 17. Synchronous calcification and shell corrosion in the undersaturated waters (mean $\Omega < 0.8$) of the autumn experiment. Length of arrows indicates relative magnitudes of treatment influences on shell calcification (left, solid arrows) and shell corrosion (right, dashed arrows) respectively. Shell growth showed little variability between treatments, except for increased growth in the +T treatment. Corrosion was strongly increased in the high-CO₂ treatments and slightly enhanced in the +T treatment. DIC: dissolved inorganic carbon.

1100 µatm (Fig. 14), corresponding to average saturation values as low as $\Omega = 0.6$ (Table 1). We detected no significant impact of saturation state on growth (tube length or diameter) in any season. However, the average weights of newly grown tubes correlated with saturation states in the spring and autumn experiments (Fig. 13).

Apparently, the Baltic *S. spirorbis* worms are able to build their tubes with little changes in extension rates at pCO₂ levels as high as 1100–1200 µatm. Notably, these pCO₂ values are in the range of their natural habitats (385 to 2500 µatm; Thomsen et al., 2010, 2013; Wahl et al., 2015). However, the tubes that are formed at lower CaCO₃ saturation may be more fragile. This is in line with results from cultured juvenile worm tubes of the tropical serpulid species *Hydroides elegans*, which showed reductions in shell hardness and wall thickness at lowered pH and CaCO₃ saturation states (Chan et al., 2012, 2013; Li et al., 2014).

As discussed in Sect. 4.4 high temperatures in the +T and +CO₂ + T treatments of the summer experiment (average T of 24 °C, Table 1) led to high mortality and strongly reduced growth of *S. spirorbis* tubes (Fig. 16). The only other significant temperature influence on growth was found in the juvenile populations of the autumn experiment. The higher temperature in the +T treatment induced higher growth rates of the juvenile populations compared to the control treatment (Fig. 14). There was, however, no significant temperature influence on the growth parameters at elevated pCO₂ (+CO₂ + T treatment), possibly indicating interactions between the effects of temperature and pCO₂ on growth.

5 Conclusions

The results of our benthic mesocosm experiments clearly demonstrate that the growth of *S. spirorbis* tubes is predominantly controlled by ontogenesis. Elevated pCO₂ levels, lowered pH and calcium carbonate saturation states expected for the end of the 21st century had no significant impact on tube extension rates. Rather, *S. spirorbis* is capable of calcifying in undersaturated water with respect to the Mg-calcite of its shell. New tube parts were observed to be formed in undersaturated water when at the same time parts of the older tube were being corroded (Fig. 17). This is clear evidence for a strict biological biomineralisation control of *S. spirorbis*.

Opposed to the batch culture experiments of Saderne and Wahl (2013) significant shell corrosion occurred in our experiments at a pCO₂ of 1100 µatm. While acidification had no impact on shell extension, shell corrosion increased with progressing acidification and undersaturation. Additionally, increased bioerosion, reduced growth and loss of substratum occurred at high summer temperatures. Most *S. spirorbis* were not able to survive at a mean temperature of 24 °C in the benthocosms. On the other hand, among the juvenile populations of the autumn experiment, elevated temperatures (15 °C) increased tube growth rates, but only under ambient pCO₂ conditions.

We conclude that under continued warming and ocean acidification, with conditions expected for the end of the 21st century, *S. spirorbis* in the Baltic Sea could be seriously affected by high summer temperatures and by enhanced dissolution and bioerosion in increasingly warmer, acidified seawater. These results contrast with previous batch culture experiments, indicating the need for experiments simulating near-natural conditions in climate change research.

Data availability. Data are archived on the PANGAEA database; Böhm et al. (2018) (<https://doi.pangaea.de/10.1594/PANGAEA.886884>).

The Supplement related to this article is available online at <https://doi.org/10.5194/bg-15-1425-2018-supplement>.

Competing interests. The authors declare that they have no conflict of interest.

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