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## Supplement of

# Decoupling salinity and carbonate chemistry: low calcium ion concentration rather than salinity limits calcification in Baltic Sea mussels

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#### **Datasets**

Dataset #1

**Product name** 

BALTIC SEA PHYSICS ANALYSIS AND FORECAST

**Product identifier** 

BALTICSEA\_ANALYSIS\_FORECAST\_PHY\_003\_006

#### Link

https://resources.marine.copernicus.eu/?option=com\_csw&task=results?option=com\_csw&view=details&product\_id=BALTICSEA\_ANALYSIS\_FORECAST\_PHY\_003\_006 last accessed 29/08/2017

#### **Short description**

This Baltic Sea physical model product provides forecasts for the physical conditions in the Baltic Sea. The Baltic forecast is updated twice daily providing a new six days forecast with hourly data for sea level variations, ice concentration and thickness at the surface, and temperature, salinity and horizontal velocities for the 3D field. The product is produced by the 3D ocean model code HBM developed within the Baltic ocean community. The product grid has a resolution of 1 nautical mile in the horizontal, and up to 25 vertical depth levels. The area covers the Baltic Sea including the transition area towards the North Sea (i.e. the Danish Belts, the Kattegat and Skagerrak).

#### **Spatial resolution**

2 km x 2 km

#### Vertical coverage

From -400 to 0 (25 levels)

#### **Temporal resolution**

Hourly – instantaneous, daily – mean, monthly - mean

#### **Update frequency**

2 x daily

#### **Production unit**

BAL-DMI-COPENHAGEN-DK; BAL-BSH-HAMBURG-GE

## **Supplementary Tables**

Table S1: Constituent chemicals added to solution to make calcium free artificial seawater (CFASW) after Kester, 1967. Mass of each chemical added was adjusted for the

			g/L at respec	ctive salinity	
Salt	Molecular weight	35	16	11	6
NaCl	58.44	23.9260	10.9376	7.5196	4.1016
Na <sub>2</sub> SO <sub>4</sub>	142.04	4.0080	1.8322	1.2597	0.6871
$MgCL_2$	95.21	2.37	1.0834	0.7449	0.4063
KCl	74.56	0.6770	0.3095	0.2128	0.1161
NaHCO <sub>3</sub>	84.00	0.1960	0.0896	0.0616	0.0336
KBr	119.01	0.0980	0.0448	0.0308	0.0168
$H_3BO_3$	61.83	0.0260	0.0119	0.0082	0.0045
SrCl <sub>2</sub>	158.51	0.0085	0.0039	0.0027	0.0015
NaF	41.99	0.0030	0.0014	0.0009	0.0005

experimental salinities in this study (16, 11 and 6). All stock solutions also contained 5 % filtered seawater from the collections site to ensure presence of trace elements.

Table S2: Mean experimental biomass densities and food availability over the duration of both experiments. Metrics presented include: aquarium volumes, mean animal no. per tank, mean mortality rates (no mortality was observed in the calcium experiment), ml of seawater per animal, body dry mass (BM) per animal, total BM per tank, BM L<sup>-1</sup> seawater, mean no. of feeds day<sup>-1</sup>, final phytoplankton concentration in experimental vessels after feeding, total no. of phytoplankton cells day<sup>-1</sup> and the no. of cells mg<sup>-1</sup> BM day<sup>-1</sup>. Mean values are presented across treatments (n = 4 replicates) over the duration of both experiments. The two most important metrics, BM litre<sup>-1</sup> and no. of phytoplankton cells mg<sup>-1</sup> BM day<sup>-1</sup> were not identical, but within the same order of magnitude for both experiments. For comparison, inter-treatment range values are given below experimental means within the bicarbonate ion manipulation experiment, illustrating BM litre<sup>-1</sup> and phytoplankton cell no. mg<sup>-1</sup> BM day<sup>-1</sup> varied by a larger degree within the bicarbonate experiment than between both experiments, further supporting comparability between both experiments.

Experiment	Aquarium volume (ml)	no. animals per tank	mortality rate end of exp.	ml per animal	BM per animal (mg)	total BM per tank (mg)	BM per litre (mg l <sup>-1</sup> )	no. of feeds per day	Phytoplankton cell conc. in tanks (cells ml <sup>-1</sup> )	total no. of cells per day	no. of cells mg BM
Calcium manipulation	50	2	N/A	25.0	0.329	0.658	13.2	1.0	10 000	5 x 10 <sup>5</sup>	7.6 x 10 <sup>5</sup>
Bicarbonate manipulation	2000	752 (1440 - 400)	53 % (10 - 75 %)	1.3 (1.4 - 5)	0.064 (0.163 - 0.746)	48.128 (65.2 – 1074.2)	24.06 (32.6 – 537.1)	2.5	10 000	5 x 10 <sup>7</sup>	$4.9 \times 10^{5}$ $(4.7 \times 10^{4}$ $- 7.7 \times 10^{5})$

Table S3: List of all R packages used for data analysis complete with versions, years, authors and intended use.

Package	Complete name	Version	Year	Author	Use
fitdistrplus	Fit distribution	1.0-14	2019	Marie Laure	Fitting data
				Delignette-Muller,	distributions
				Christophe Dutang	
				(2015)	
ggplot2	Elegant	3.3.0	2016	Wickham H (2016)	Graphics
	graphics for				
	data analysis				
nlme	Linear and non-	3.1-147	2020	Pinheiro J, Bates D,	
	linar mixed			Debroy S, Sarkar D,	models
	effect models			R Core Team	
	_			(2020)	
drc	Dose response	3.0-1	2016	Ritz, C., Baty, F.,	•
	analysis			Streibig, J. C.,	response curves
104	T	4.45	2010	Gerhard, D. (2015)	<b>D</b> 1 1 0
ncdf4	Interface to	1.17	2019	David Pierce (2010)	
	Unidata				netCDF files
	netCDF files	0.2	2010	771 77	C 1 C
cmocean	cmocean	0.2	2019	• •	Colour maps for
				*	oceanography
				Krylov, I.,	

Table S4: Model parameters for the power relationship between shell length and  $CaCO_3$  mass with statistical results; standard error, T value and p value (Fig. S3). These separate relationships were used to calculate calcification rates in the field from shell length measurements for each population.

Parameter	Site	value	Std. Error	<i>T</i> -value	<i>p</i> -value
intercept	Usedom	0.171	0.032	5.36	< 0.001
	Ahrenshoop	0.050	0.012	4.29	< 0.001
	Kiel	0.036	0.004	8.17	< 0.001
power	Usedom	2.163	0.092	23.64	< 0.001
	Ahrenshoop	2.716	0.107	25.43	< 0.001
	Kiel	3.000	0.046	65.24	< 0.001

Table S5: Model comparisons using AIC for selection of the most parsimonious negative exponential decay model fit to laboratory calcification rates across experimental  $[HCO_3^-]$  in the bicarbonate ion manipulation experiment. Model names and equations are shown with fixed parameters allowing direct comparison between salinity treatments. The lowest AIC value (in bold) represents the model which best explains the experimental data and was chosen for statistical analysis.

			Estimated	
Model	Equation	Fixed parameters	parameters	AIC
	$y \sim C_{max} * (1 - e(-K * (x -$			
Von Bertalanffy	$s_0)))$	$s_0 = 380$	$C_{\max}, K$	115.58
	$y \sim C_{max} * (1 + s_0 * (e(-$			
Gompertz	$(K^*x)))^{-1}$	$s_0 = 380$	$C_{\max}, K$	116.49
	$y \sim C_{max} * (1 + s_0 * (e(-$			
Logistic	$(K^*x))^{-1}$	$s_0 = 380$	$C_{\max}, K$	115.79
Negative				
exponential	$y \sim C_{\text{max}} * (1 - e(x/K))$		$C_{\max}, K$	121.82
Michaelis-Menten	$y \sim C_{max} * x/(K+x)$		$C_{\max}, K$	122.21

Table S6: Statistical test results of all analyses graphically depicted in Figs 2 - 7. Pairwise comparisons (*post-hoc* tests) between treatments for statistically significant factors are listed on the right with significant *P*-values being shown in bold.

ANCOVA – laborar	tory calcifica	ation rates [Ca <sup>2</sup>	+] manipulation	experiment (Fig. 5	b)		
factor	df	Sum Sq.	Mean Sq.	F-value	P-value	pair-wise comparisons	P-value
Calcium	1	7160	7160	106.9	< 0.001	11-16	0.988
Salinity	2	110.9	55.5	0.83	0.442	6-16	0.559
Calcium:Salinity	2	452.4	226.2	3.38	0.041	6-11	0.47
Residuals	54	3616	67				
Negative exponentia	al decay mod	del – Laborator	y calcification a	and $\Omega_{aragonite}$ (Fig. 6a	a)		
parameter	estimate	Std. Error	T-value	P-value			
$C_{\max}$	31.2315	5.244	5.956	< 0.001			
K	0.374	0.1598	2.341	0.027			
Negative exponenti	al decay mod	del – Labor <mark>ator</mark>	y calcification a	and ESIR (Fig. 6b)			
parameter	estimate	Std. Error	T-value	P-value			
$C_{\max}$	33.7696	5.9676	5.659	< 0.001			
K	0.2752	0.1117	2.463	0.020			
ANCOVA – Field	calcification	rates (Fig. 7)					
factor	df	Sum Sq.	Mean Sq.	F-value	P-value	pair-wise comparisons	P-value
time	1	952304	952304	517.7	< 0.001	Kie:Ahp	< 0.001
population	2	2096837	1048419	570.0	< 0.001	Use:Ahp	0.99
time:population	2	1769631	884815	491.0	< 0.001	Use:Kie	< 0.001
Residuals	12	30873	2573				
ANOVA – Mean sa	linity at field	d monitoring si	tes (Fig. 2)				
factor	df	Sum Sq.	Mean Sq.	F-value	P-value	pair-wise comparisons	P-value
site	2	1248	624.2	38518	< 0.001	Ahp-Kie	< 0.001
residuals	17253	279.6	0.02			Ahp-Use	< 0.001
						Kie-Use	< 0.001
ANOVA – Mean [H	ICO <sub>3</sub> ] at fie	ld monitoring s	sites (Fig. 3c)				
factor	df	Sum Sq.	Mean Sq.	F-value	P-value	pair-wise comparisons	P-value

site	2	414716	207358	38.80	< 0.001	Ahp-Kie	< 0.001
residuals	55	293930	5344			Ahp-Use Kie-Use	0.722 < <b>0.001</b>
ANOVA – Mea	n pH at field	monitoring sites	s (Fig. 3a)			Rie-Osc	< 0.001
factor	df	Sum Sq.	Mean Sq.	F-value	P-value		
site	2	0.092	0.046	1.217	0.304		
residuals	55	2.070	0.038				
ANOVA – Sali	nity-A <sub>T</sub> relation	onship across all	field monitoring	g sites (Fig. 4)			
factor	df	Sum Sq.	Mean Sq.	F-value	P-value		
salinity	1	571206	571206	86.6	< 0.001		
residuals	56	369521	6599				
ANOVA – Mea	n $\Omega_{ m aragonite}$ at $1$	field monitoring	g sites (Fig. 3e)				
factor	df	Sum Sq.	Mean Sq.	F-value	P-value	pair-wise comparisons	P-value
site	2	1.89	0.94	7.22	0.002	Ahp-Kie	0.044
residuals	55	7.18	0.13			Ahp-Use	0.002
						Kie-Use	0.654
ANOVA – Mea	n ESIR at fiel	ld monitoring si	tes (Fig. 3g)				
factor	df	Sum Sq.	Mean Sq.	F-value	P-value	pair-wise comparisons	P-value
site	2	3.98	1.99	10.88	< 0.001	Ahp-Kie	0.020
residuals	55	10.07	0.18			Ahp-Use	0.363
						Kie-Use	< 0.001
ANOVA - Mean	n Chl-a value	s at field monito	oring sites (Fig. 3	3i)			
factor	df	Sum Sq.	Mean Sq.	F-value	P-value	pair-wise comparisons	P-value
site	2	11.41	5.7	13.8	< 0.001	Ahp-Kie	< 0.001
residuals	77	31.82	0.41			Ahp-Use	< 0.001
						Kie-Use	0.35

Table S7: Model parameters and  $R^2$  values for the linear relationships: Salinity- $A_T$  and  $\Omega_{aragonite}$ -ESIR. Also shown are the parameter estimates and residual sum of squares (RSS) for the negative exponential decay model fit to laboratory calcification rates in both experiments and substrate inhibitor ratio (SIR) not including [Ca<sup>2+</sup>].

Linear model – Salinity-A <sub>T</sub> (Fig. 4)									
factor	estimate	Std. Error	T-value	P-value	$R^2$				
intercept	1679.9	32.4	51.9	< 0.001	0.6				
slope	22.13	2.4	9.3	< 0.001					
Linear model – Field Ω <sub>aragonite</sub> -ESIR relationship (Fig. S8)									
factor	estimate	Std. Error	T-value	P-value	$R^2$				
intercept	0.337	0.049	6.89	< 0.001	0.805				
slope	0.781	0.051	15.39	< 0.001					
Negative exponent	ial decay model	- Laboratory c	alcification a	and SIR (Fig	. S5)				
parameter	estimate	Std. Error	T-value	P-value	RSS				
$C_{\max}$	22.66	2.75	8.23	< 0.001	4631				
K	0.02	0.01	1.45	0.157					

Table S8: Parameters for the linear model fit to  $[Ca^{2+}]$  and calcification rates in the calcium ion manipulation experiment and the negative exponential decay model fit to  $[HCO_3^-]$  and calcification rates in the bicarbonate manipulation experiment. Significant p-values are shown in bold and graphical representations of these models are depicted in Fig. 5.

Linear mode	l parameters – [Ca	<sup>2+</sup> ]-calcification	l		
salinity	treatment	value	Std. Error	T-value	P-value
6	intercept	7.21	3.84	1.88	0.065
11		2.42	5.44	-0.88	0.382
16		-2.69	2.33	1.66	0.102
6	slope	1.67	1.67	3.9	< 0.001
11		5.56	2.33	-1.78	0.081
16		2.37	2.37	2.57	0.013
Negative exp	onential decay mod	del parameters	- [HCO3 <sup>-</sup> ]-calc	ification	
salinity	treatment	value	Std. Error	T-value	P-value
6	$C_{\max}$	4.68	2.03	2.31	< 0.001
11		34.00	2.07	16.39	< 0.001
16		35.25	1.78	19.79	< 0.001
6	K	0.007	0.008	0.843	< 0.001
11		0.007	0.001	5.966	< 0.001
16		0.011	0.002	6.106	< 0.001

Table S9: Parameters for negative exponential decay models fit to calcification rates ( $\mu g$  CaCO<sub>3</sub> d<sup>-1</sup>) and [Ca<sup>2+</sup>] and [HCO<sub>3</sub>-] predictors across both laboratory experiments. Parameters with significant estimates have *P*-values in bold.

Negative exponenti	Negative exponential model – [Ca <sup>2+</sup> ] – calcification (Fig. S4b)								
parameter	value	Std. Error	T-value	P-value					
$C_{\max}$	47.29	30.18	1.57	0.128					
K	5.10	4.89	1.04	0.306					
Negative exponenti	al model – [HCO	3 <sup>-</sup> ] – calcification	(Fig. S4a)						
parameter	value	Std. Error	T-value	P-value					
$C_{\max}$	25.00	6.49	3.85	< 0.001					
K	740.38	618.65	1.20	0.241					

Table S10: Parameters and statistical results for the linear model fit to field calcification over time during the first 7 months. The linear slopes of these models express the calcification rates for each of the 3 populations (Table 1).

Linear model - Field calcification rates over first 7 months.							
salinity	parameter	value	Std. Error	T-value	P-value		
Usedom	intercept	-495.0	7657.9	-0.07	0.953		
Ahrenshoop		-457.5	7657.9	-0.08	0.938		
Kiel		-116675.0	5414.9	-15.24	< 0.001		
Usedom	slope	18.6	53.2	-0.33	0.761		
Ahrenshoop		36.4	53.2	0.97	0.405		
Kiel		2202.9	37.6	40.70	< 0.001		

Table S11: Statistical results for the Kruskal-Wallis test comparing temperatures (°C) at each site during the monitoring period.

Kruskal-Wallis test – Mean temperatures at field monitoring sites (Fig. S7)					
Kruskal-Wallis chi squared	df	P-value	Dunn test	P-value	
122.73	2	< 0.001	Ahp-Kie	< 0.001	
			Ahp-Use	0.042	
			Kie-Use	< 0.001	

## **Supplementary Figures**

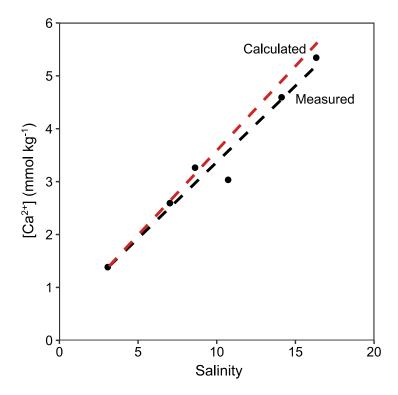


Figure S1: Comparison of the linear relationship between salinity and  $[Ca^{2+}]$  in the Southwest Baltic Sea from calculated  $[Ca^{2+}]$  values (red dashed line) and measured  $[Ca^{2+}]$  values (blue dashed line). Methodologies for calculated and measured  $[Ca^{2+}]$  values are given in the text (Sections 2.1 and .5). Samples for the lowest salinity (3.15) were taken from the Achterwasser in Usedom (54° 0' 5" N, 14° 2' 47" E).

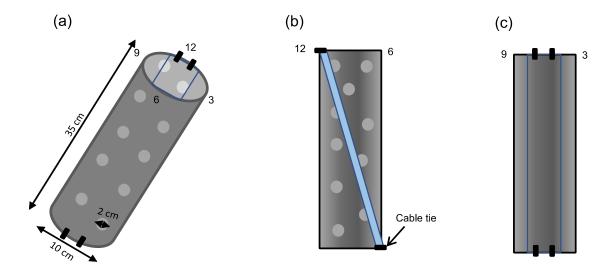


Figure S2: Settlement structures deployed at all three sites (Kiel, Ahrenshoop and Usedom) in March 2016 (a). A cross section of the settlement structures from 12 o'clock to 6 o'clock showing the orientation of the mesh net inside the cylinder (b). Cross section of the settlement structures showing the width of the mesh net inside the cylinder (c). A total of twenty x 2 cm diameter holes were drilled into each cylinder and a 0.2 cm pore size cotton/nylon spat sock was diagonally positioned across the inside of the cylinders using cable ties. Numbers at the top of each figure represent the orientation of the cross sections in panels b and c.

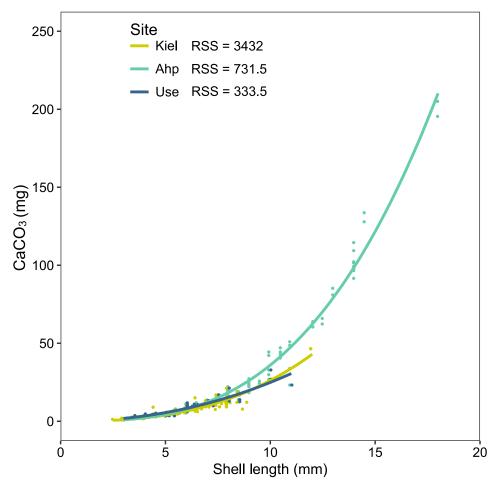


Figure S3: SL-CaCO<sub>3</sub> mass (mg) relationships specific to each of the three study populations: Kiel, Ahrenshoop and Usedom (data from Sanders et al., 2018). This was used to calculate initial CaCO<sub>3</sub> mass in laboratory experiments and field calcification at the three monitoring sites in this study. Power model parameters are given in Table S4 with the insets listing the residual sum of squares (RSS) for each population.

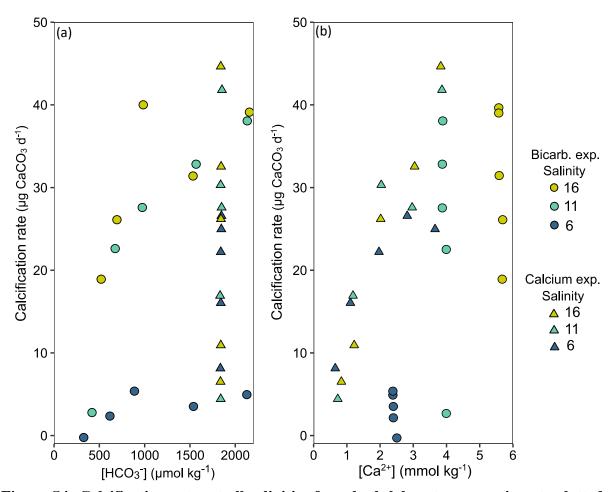


Figure S4: Calcification rates at all salinities from both laboratory experiments plotted across all  $[HCO_3^-]$  (a) and all  $[Ca^{2+}]$  (b). No significant relationship was found for either variable.

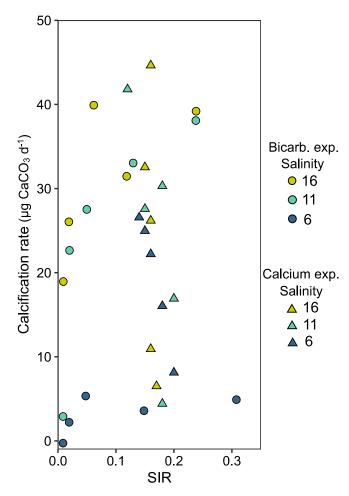


Figure S5. Calcification rates at all salinities from both laboratory experiments plotted across substrate inhibitor ratio (SIR). This measure excludes the effect of  $[Ca^{2+}]$  and no statistically significant model could be fit (Table S7).

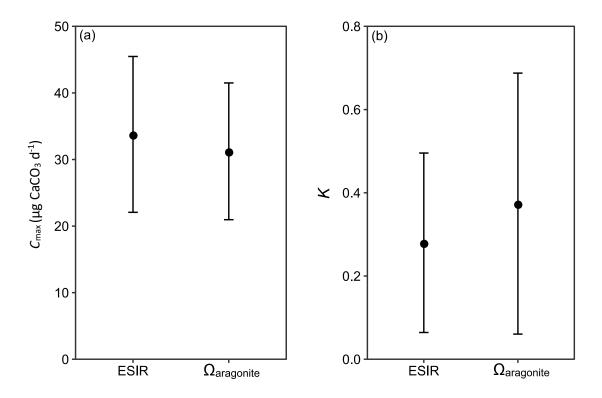


Figure S6: Graphical comparisons of parameters  $C_{max}$  (µg CaCO<sub>3</sub> d<sup>-1</sup>) and K in the negative exponential decay model depicted in Fig. 6. Values are shown  $\pm$  95 % confidence intervals (CI). Overlaps in the CI indicates the two parameters in both models are not significantly different.

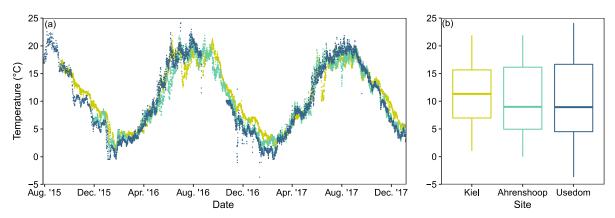


Figure S7: Temperature (°C) data from deployed CTD's at the three monitoring sites from Aug. 2015 - Dec. 2017 (a). Box pots are shown on the right depicting median temperatures and interquartile ranges (b).

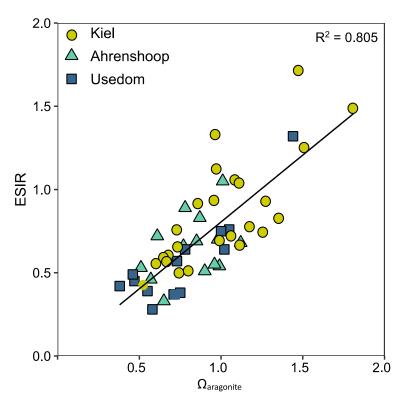


Figure S8: Field extended substrate inhibitor ratio (ESIR) plotted against aragonite saturation state ( $\Omega_{aragonite}$ ) over the monitoring period at all three Baltic Sea sites. Parameters for the linear model are given in Table S7.

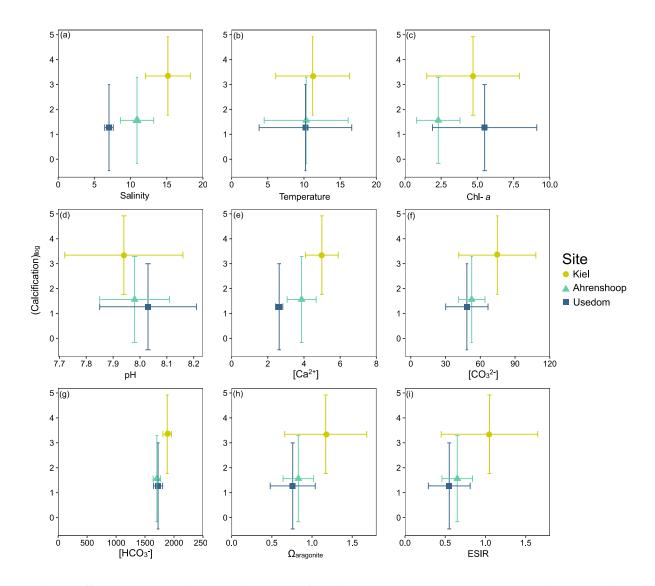


Figure S9: Log transformed field calcification rates (y-axes) plotted against multiple environmental parameters; Salinity (a), Temperature (°C) (b), chl-a ( $\mu$ g L-1) (c), pH<sub>total</sub> (d), [Ca<sup>2+</sup>] (mmol kg-1) (e), [CO<sub>3</sub><sup>2-</sup>] ( $\mu$ mol kg-1) (f), [HCO<sub>3</sub>-] ( $\mu$ mol kg-1) (g),  $\Omega_{aragonite}$  (h), ESIR ( [Ca<sup>2+</sup>][HCO<sub>3</sub>-] / [H<sup>+</sup>]) (i). Values are shown  $\pm$  standard deviation.

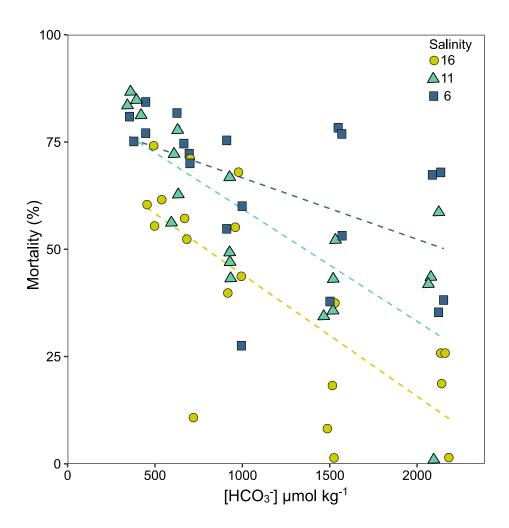


Figure S10: Total mortality rate in each treatment (n=4 replicates) at the end of the bicarbonate ion manipulation experiment from an initial number of 1600 animals per 2L aquaria.

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