

1 Supplemental material submitted for the paper

2 **The importance of ocean transport in the fate of anthropogenic CO₂**

3 L. Cao, M. Eby, A. Ridgwell, K. Caldeira, D. Archer, A. Ishida, F. Joos, K. Matsumoto, U.
4 Mikolajewicz, A. Mouchet, J. C. Orr, G.-K. Plattner, R. Schlitzer, K. Tokos, I. Totterdell, T.
5 Tschumi, Y. Yamanaka , A. Yool

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7 **Introduction:**

8 This supplemental material includes one table and two figures. Table S1 compares parameter
9 values used in three versions of the GENIE-1 model. Figure S1 shows the effect of vertical
10 diffusivity, vertical resolution, and seasonality on modeled oceanic uptake of anthropogenic CO₂.
11 Figure S2 shows the effect of marine biology on modeled oceanic uptake of anthropogenic CO₂.

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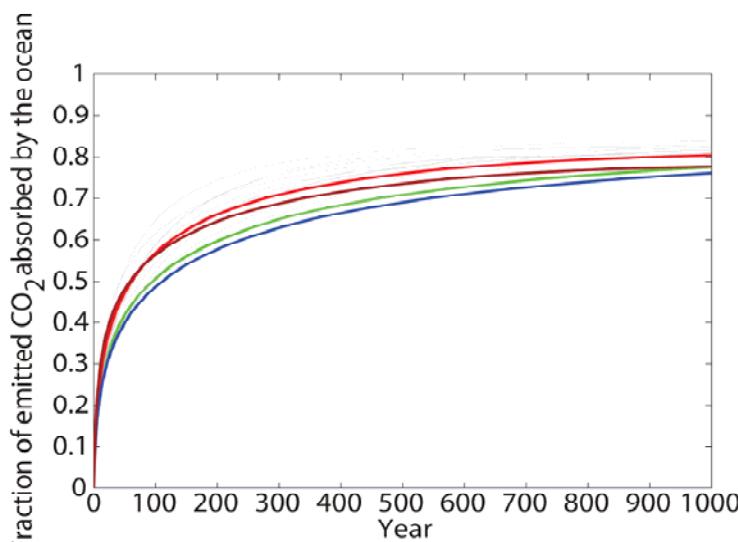
26 **Table S1.** Controlling parameters in different versions of the GENIE-1 model.

Parameter Name	GENIE8	GENIE16	MESMO	Parameter description and units
Ocean physics ^a				
W	1.93	1.531	2.208	Wind-scale
κ_h	4489	1494	4467	Isopycnal diffusion ($m^2 s^{-1}$)
κ_v	0.27	0.25	0.1 to 1.2	Diapycnal diffusion ($cm^2 s^{-1}$)
λ	2.94	2.71	2.21	1/friction (days)
Atmosphere physics ^a				
k_T	4.67×10^6	5.20×10^6	3.27×10^6	T diffusion amplitude ($m^2 s^{-1}$)
l_d	1.08	1.41	0.979	T diffusion width (Radians)
s_d	0.06	0.09	0.1700	T diffusion slope
κ_q	1.10×10^6	1.17×10^6	1.70×10^6	Q diffusion ($m^2 s^{-1}$)
β_T	0.11	0.0010	0.0023	T advection coefficient
β_q	0.23	0.165	0.23	Q diffusion coefficient
F_a	0.23	0.73	0.36	FW flux factor (Sv)
Sea-ice physics ^a				
κ_{hi}	6200	3574	5579	Sea-ice diffusion ($m^2 s^{-1}$)
Ocean biogeochemistry ^b				
u_0^{POC}	1.96	8.99	1.91	maximum PO ₄ uptake rate ($\mu\text{mol kg}^{-1} \text{yr}^{-1}$)
K^{POC}	0.22	0.89	0.21	PO ₄ half-saturation concentration ($\mu\text{mol kg}^{-1}$)
r^{POC}	0.065	0.056	0.055	partitioning of POC export into fraction #2
l^{POC}	550	590	variable	e-folding depth of POC fraction #1 (m)
l_2^{POC}	∞	∞	∞	e-folding depth of POC fraction #2 (m)
$r_0^{CaCO_3:POC}$	0.044	0.048	0.046	CaCO ₃ :POC export ‘rain ratio’ scalar ^c
η	0.81	0.77	1.28	calcification rate power
r^{CaCO_3}	0.4325 ^d	0.45 ^d	0.49	partitioning of CaCO ₃ export into fraction #2
l^{CaCO_3}	1083	1890	variable	e-folding depth of CaCO ₃ fraction #1 (m)
$l_2^{CaCO_3}$	∞	∞	∞	e-folding depth of CaCO ₃ fraction #2 (m)

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28 ^a See: Edwards and Marsh (2005); Hargreaves et al. (2004); Ridgwell et al. (2007a); Singarayer et al.
29 (2008), Matsumoto (2008)30 ^b See: Ridgwell et al. (2007a,b); Ridgwell and Hargreaves (2007).31 ^c Note that the rain ratio scalar parameter is not the same as the actual CaCO₃:POC export rain ratio
32 because it is multiplied by $(\Omega - 1)^{\eta}$ where Ω is the surface ocean saturation state (with respect to calcite),
33 as described in Ridgwell et al. (2007a,b). Pre-industrial mean ocean surface Ω is ~5.2 in the GENIE-1
34 model, so that the global CaCO₃:POC export rain ratio can be estimated using the 8-parameter
35 assimilation^d as being equal to $(5.2 - 1)^{0.81} \times 0.044 = 0.14$.36 ^d Adjusted compared to formal calibration in order to achieve an improved prediction of mean sediment
37 surface wt% CaCO₃ compared to observations (Ridgwell and Hargreaves, 2007).

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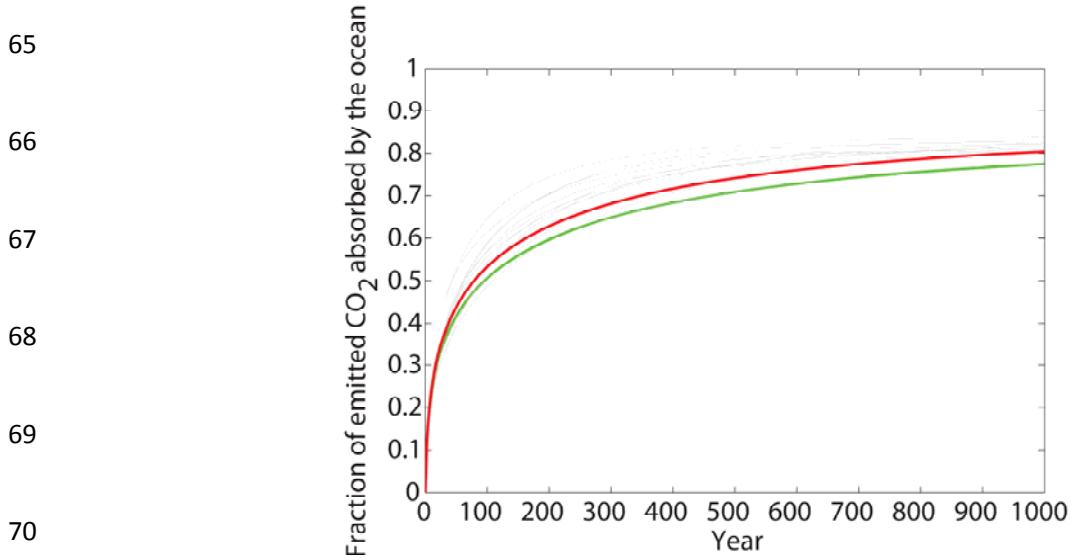
Fig. S1. Model-simulated oceanic uptake of CO₂ in response to a CO₂ pulse emission of 590.2 PgC (corresponding to an instantaneous doubling of atmospheric CO₂ from 278 to 556 ppm). Results from different runs using the GENIE16 model are shown: GENIE16 base run as shown in Fig. 1 (green), GENIE16 run with vertical diffusivity doubled from 0.25 to 0.5 cm⁻² s⁻¹ (red), GENIE16 run with vertical resolution reduced from 16 to 8 levels (brown), GENIE16 run with the seasonal cycle removed (blue). Other model results as shown in Fig. 1 are presented here by grey lines.

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73 **Fig. S2.** Model-simulated oceanic uptake of CO₂ in response to a CO₂ pulse emission of 590.2
 74 PgC (corresponding to an instantaneous doubling of atmospheric CO₂ from 278 to 556 ppm).
 75 Results from two different runs using the GENIE16 model are shown: GENIE16 base run as
 76 shown in Fig. 1 (green), GENIE16 run without the inclusion of marine biology (red). Other
 77 model results as shown in Fig. 1 are presented here by grey lines. The abiotic run absorbs more
 78 excess CO₂ than the biotic run because during model spinup the removal of marine biology leads
 79 to higher surface alkalinity (a global mean value of 2361.9 μmole/kg in abiotic run compared
 80 with 2271.5 μmole/kg in biotic run). Higher surface alkalinity leads to greater buffering capacity
 81 of the ocean to absorb excess CO₂.

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- 87 **References:**
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