

# 1 **Supplementary material: MATERIALS AND METHODS**

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### 8 **S1.1 Model description:**

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10 To identify the factors controlling chemical weathering of carbonates and to quantify their  
11 effect in given environments, we used an updated version of the WITCH box-model  
12 (Goddéris et al. 2006; Roelandt et al. 2010; Goddéris et al. 2010). WITCH simulates the time  
13 evolution of the chemical composition of belowground waters and vertical drainage. In its  
14 original design, WITCH includes the mathematical description of the dissolution/precipitation  
15 of mineral phases in the various horizons of the weathering profile, from the surface down to  
16 the impervious bedrock. Laboratory kinetic laws derived from the transition state theory  
17 (Eyring 1935) are used to describe the interaction between water and minerals.

18 Here, the code has been adapted to the specific conditions of the studied area. The main  
19 improvement of the WITCH model is the computation of a budget equation for carbon,  
20 accounting for both diffusion and ventilation. The only lithology considered in the simulations  
21 is the carbonate rock lithology, assumed to be 100% calcitic according to the mineralogical  
22 analysis at the El Llano de los Juanes site. Calcite dissolution/precipitation rate  $F_{cal}$  ( $\text{mol m}^{-2} \text{s}^{-1}$ )  
23 is described kinetically by the following equation (Goddéris et al. 2010):

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$$25 \quad F_{cal} = \left( k_H^{cal} \cdot a_H + \frac{k_o}{1 \cdot 10^{-5} + a_{CO_3}} \right) \cdot (1 - \Omega_{cal}) \quad (Eq. S1)$$

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27 where  $k_H^{cal}$  equals  $10^{-0.659} \text{ mol m}^{-2} \text{ s}^{-1}$  and  $k_o$   $10^{-11} \text{ mol m}^{-2} \text{ s}^{-1}$  at  $25^\circ\text{C}$  (Wollast 1990). The  
28 activation energies for rate constants are respectively set to  $8.5 \text{ kJ mol}^{-1}$  and  $30 \text{ kJ mol}^{-1}$   
29 (Alkattan et al. 1998; Pokrovsky et al. 2009).  $a_H$  and  $a_{CO_3}$  stand for the activity of aqueous  
30 protons and carbonate ion respectively, and  $\Omega_{cal}$  is the solution saturation state with respect to  
31 calcite. Dissolution or precipitation occurs when reactions depart from equilibrium (Goddéris

32 et al. 2006) (e.g. when  $\Omega_{cal}$  is respectively lower or greater than 1). The dependence on  
33 temperature of calcite solubility product is given by (Drever 1997):  
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$$35 \quad K_{cal} = 10^{-8.48} \exp \left[ -\frac{9610}{R} \cdot \left( \frac{1}{298.15} - \frac{1}{T} \right) \right] \quad (Eq.S2)$$

36  
37 where R is the gas constant and T the temperature in Kelvin.

38 The mass balance is solved for each water reservoir (corresponding to a given soil and  
39 weathering profile layer) for every time step. The outputs of these budget equations - carbon  
40 content, dissolved calcium, and total alkalinity in each modeled layer - are injected at each  
41 time step into the speciation module that calculates the complete carbonate speciation  
42 accounting for the environmental conditions (such as the fluctuating temperature and water  
43 volumetric content).

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45 We designed the model simulations to accord with the conditions of the Mediterranean study  
46 area (see below S1.2: Site description and geological context). The weathering profile was  
47 prescribed as a ten layer structure with a total thickness of 100 m, the two upper layers having  
48 soil properties (porosity  $0.59 \text{ m}^3\text{m}^{-3}$  -measured on site- and thickness 15 cm each) and the  
49 others layers having bedrock properties (porosity of  $0.03 \text{ m}^3\text{m}^{-3}$  -measured on site- and  
50 thickness resp. 4 m, 4 m, 4 m, 10 m, 10 m, 10 m, 10 m and 47.7 m with increasing depth).  
51 Half hourly meteorological data (see below: S1.3 Measurements) from 2009 were used to  
52 force the model, these include: the soil water content (which defines the volumetric size of  
53 each reservoir and the fraction of the total reactive mineral surface that is available for  
54 weathering), rainfall, soil temperature and inputs of  $\text{CO}_2$  from biological production. The  
55 latter was estimated by means of a  $Q_{10}$  function using the nighttime  $\text{CO}_2$  fluxes from eddy  
56 covariance measurements during biologically active periods when geochemical fluxes were  
57 negligible. This production was prescribed only in the two upper layers (soil) and was  
58 assumed to be zero during the drought period, during which all plants are senescent and  $\text{CO}_2$   
59 efflux was typically zero in the absence of high turbulence. The measured daytime effluxes in  
60 this season could therefore not be attributed to respiration, making this the appropriate period  
61 to study ventilation.

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63 To account for the effect of ventilation on the weathering rates, we introduced an equation in  
64 the WITCH model to account for the  $\text{CO}_2$  efflux due to ventilation ( $F_{Vent}$ ):

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$$F_{Vent} = k \cdot u_* \cdot \Delta[CO_2] \cdot I_r \cdot u_{*0} \quad (Eq.S3)$$

Where  $u_*$  is the friction velocity, a parameter strongly controlling ventilation as can be seen in Figure S1 (Supplementary Materials: Figure S1).  $\Delta[CO_2]$  is the difference in  $CO_2$  concentration between the layer where ventilation occurs and the atmosphere. Ventilation ceases if soil  $CO_2$  concentrations are depleted.  $I_r$  is the solar radiation deduced from measurements of PPFD (photosynthetic photon flux density). It accounts for surface heating, which induces mass transport through convective flows.  $k$  is a constant accounting for the site specific texture of belowground system. The more fractioned it is and the more caves occur in it, the higher the value of  $k$ . Ventilation is promoted in highly fractioned Karst systems such as our study area.  $u_{*0}$  is a threshold depending on the ratio between the amplitude of atmospheric turbulence and the amount of water in the soil  $\left(\frac{u_*}{VWC}\right)$ , and on the maximum water content for enabling ventilation ( $VWC_{Vmax}$ ). In these simulations we allowed ventilation only when  $u_{*0} > 2$  ( $VWC$  expressed in  $m^3m^{-3}$  and  $u_*$  in  $m \cdot s^{-1}$ ) and  $VWC_{Vmax} \leq 0.1$ . By adjusting this equation to meet the observed fluxes, we achieved a realistic estimate for the amount of  $CO_2$  that is extracted from the soil by ventilation. In this simulation, we prescribed ventilation to occur only in the upper 30 cm of the surface. We emphasize the fact that this ventilation equation (Eq. S3) has not been validated and therefore cannot be easily extrapolated to other areas. Estimations for  $k$  and  $u_{*0}$  are site-specific. In spite of this, the equation functions well enough for the purpose of this study, which is to evaluate the effect of extracting  $CO_2$  from the soil on carbonate weathering rates.

87 **S1.2 Site description and geological context:**

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89 The Sierra de Gádor is a mountain range in the South East of Spain (province of Almería)  
90 which reaches 2246 meters above sea level (Li et al. 2007, 2008). The Sierra consists of an up  
91 to 1000 meter thick series of Triassic carbonate rocks (limestone and dolomite) that are highly  
92 permeable and fractured. The carbonates are intercalated with calcoschists of low  
93 permeability and underlain by impermeable metapelites of Permian age (Pulido-Bosch et al.  
94 2000; Contreras et al. 2008). The Sierra de Gádor is part of the Betic Cordillera, the  
95 westernmost alpine mountain belt in Europe. It was collisionally generated during  
96 convergence between African and Iberian plates in Tertiary times. The Betic Cordillera is  
97 composed of rocks ranging from Paleozoic or even Precambrian age up to present day. Two  
98 main tectonic domains can be distinguished in the Betic Cordilera: the External Zones,  
99 located to the west and north of the Granada Basin, where Mesozoic and Tertiary carbonate  
100 rocks are very abundant (limestones and dolomites) and the Internal Zones, located to the  
101 south and east of the Granada Basin, which are composed mainly of siliceous rocks (e.g. mica  
102 schists, phyllites, quartzites) and carbonates rocks as limestones and dolomites. Calcareous  
103 accumulations are very common in the area and are formed by leaching of carbonates from  
104 red soils (pedogenic) or during later diagenesis. Karst landforms are widely represented in the  
105 calcareous areas of the Betic Cordillera, with good examples of specific karst formations,  
106 such as dolines, karren and caves. The Sierra de Gádor is an uplifted zone located to the  
107 south-east of the Sierra Nevada, the area where the most noticeable vertical movements in the  
108 Betic Internal Zones has occurred, and is separated from it by the narrow Alpujarra Corridor  
109 (Sanz de Galdeano and Alfaro 2004). The karstic landscape of the Sierra de Gádor is  
110 characterized by a mosaic of rock outcrops, bare soil and vegetation patches.

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112 The study site “El Llano de los Juanes”, with an elevation of about 1660 m above sea level is  
113 a relatively flat shrub-land area corresponding to a well developed karstic plateau (Serrano-  
114 Ortiz et al. 2007). The carbonate rocks here are mainly dark limestone, with 98% calcite (X-  
115 ray diffraction analysis) (Were et al. 2010). The site is characterized by a semiarid montane  
116 Mediterranean climate, with a mean annual temperature of 12 °C and mean annual  
117 precipitation of ca. 475 mm, falling mostly during autumn and winter, and by a very dry  
118 season in summer (Serrano-Ortiz et al. 2007; Kowalski et al. 2008). Thickness of the soil  
119 overlaying the bedrock ranges from 0 to 0.5 m. The vegetation, so called “Macchia” or  
120 “Matorral”, at this study site is sparse and only around 0.5 m in height, but nonetheless bio-

121 diverse. The two most predominant species are *Festuca scariosa* (Lag.) Hackel (19.0 %  
122 ground cover) and *Genista pumila* (Vierh) ssp. *pumila* (11.5 % ground cover) Other common  
123 species are *Hormathophylla spinosa* (L.) P. Küpfer, *Thymus serpylloides* Bory, *Phlomis*  
124 *lychnitis* L., *Lavandula lanata* Boiss, *Salvia lavandulifolia* Vahl, and *Eryngium campestre* L.  
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126        **S1.3        Measurements:**

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128    In 2009, the fluxes of CO<sub>2</sub> and H<sub>2</sub>O were estimated from 10 Hz eddy covariance  
129    measurements at 2.5 m height. The extent of fetch, i.e. the distance upwind over the  
130    homogeneous surface, is several hundreds of meters from the tower in every direction. An  
131    open-path infrared gas analyzer (*Li-Cor 7500*, Lincoln, NE, USA) measured densities of CO<sub>2</sub>  
132    and H<sub>2</sub>O; it was calibrated monthly using an N<sub>2</sub> standard for zero and a 500.1 μmol(CO<sub>2</sub>)  
133    mol<sup>-1</sup> gas standard as a span. Wind speed and sonic temperature were measured by a three-  
134    axis sonic anemometer (*CSAT-3*, *Campbell Scientific*, Logan, UT, USA).

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136    A thermohygrometer (*HMP 35C*, *CSI*, USA) at 1.5 m above the surface was used to measure  
137    air temperature and humidity. Soil water content (SWC) was measured by three water content  
138    reflectometers (*CS615*, *CSI*) at 4 cm depth. Soil temperature was determined as the mean of  
139    two pairs of thermocouples (TCAV, *CSI*) at 2 cm and 6 cm. Rainfall was measured by a  
140    tipping bucket (0.2 mm) rain gauge (model 785 *M*, *Davis Instruments Corp.*, Hayward, CA,  
141    USA). Fluxes of incident and reflected photons in photosynthetic wavelengths, measured by  
142    two quantum sensors (*Li-190*, *Li-Cor*, Lincoln, NE, USA) at 1.5 m over a representative  
143    ground surface were used to determine F<sub>p</sub> (photosynthetically active photon flux density).

144    The measurement system centers on a datalogger (*CR3000*, *CSI*) that calculated and stored  
145    means, variances and co-variances of 10 Hz data every 30 min. Eddy fluxes calculated from  
146    density fluctuations (Webb et al. 1980) and coordinate rotations (McMillen 1988) were  
147    carried out in post-processing. Measurements of nighttime CO<sub>2</sub> fluxes with friction velocity  
148    lower than 0.2 m s<sup>-1</sup> were eliminated from the analysis to avoid possible underestimation due  
149    to low turbulence (Serrano-Ortiz et al. 2009). For further details see Serrano-Ortiz et al.  
150    (Serrano-Ortiz et al. 2009).

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152    Deep CO<sub>2</sub> molar fraction was measured in a borehole (7 m depth and 0.1 m diameter) through  
153    the bedrock outcropping using a GMP-343 (*Vaisala, Inc.*, *Finland*). These measurements  
154    were made every 30s and stored as 5 min averages in a datalogger (*CR23X*, *CSI*).

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