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2 Table 1: Overview of all measurements and sampling stations (PW = Pore water analyses including ex situ  
3 methane concentration measurements; AOM = anaerobic oxidation of methane; SR = sulphate  
4 reduction; AODC =Acridine Orange Direct Counts) of this study and their PANGAEA event label.  
5 Sampling and measurements were performed in 2006 during the M70-2 expedition with the RV Meteor  
6 and in 2009 during the MSM 13-3 expedition with the RV Maria S. Merian. The event label are named  
7 accordingly. All data of this study is submitted to the PANGAEA data base (<http://www.pangaea.de/>)  
8 and available under the doi:10.1594/PANGAEA.804779.  
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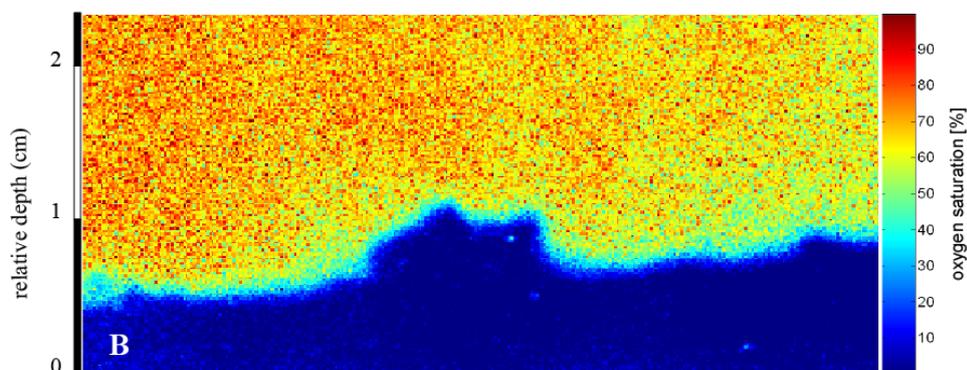
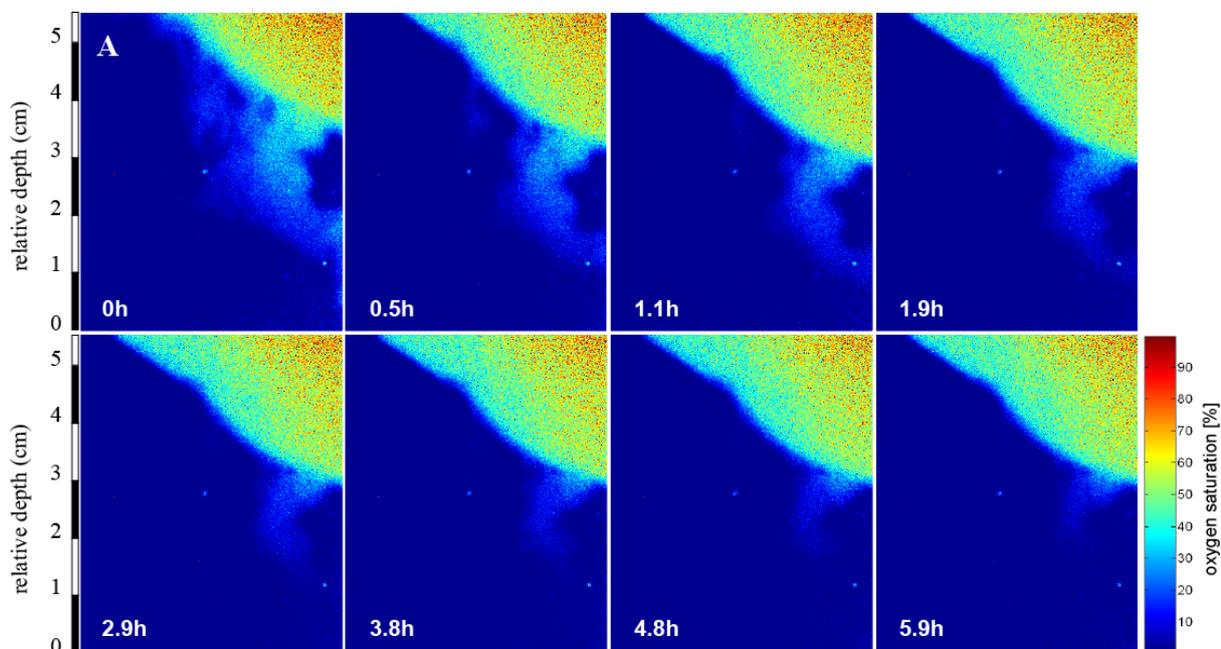
Habitat	measurements	PANGAEA database event labels
<b>central dome (I)</b>	Benthic chamber	MSM13/3_929-1_CHAM1; MSM13/3_937-1_CHAM5; MSM13/3_971-1_CHAM12
	Microprofiler	M70/2b_825_PROF-2; MSM13/3_971-1_MICP11
	PW	M70/2a_757_PUC-22; M70/2a_760_PUC-45; M70/2a_760_PUC-48; M70/2b_791; M70/2b_804
	SR	M70/2a_760_PUC-25; M70/2a_760_PUC-53; MSM13/3_933-1_PUC2; MSM13/3_933-1_PUC24
	AOM	M70/2a_760_PUC-15; M70/2a_760_PUC-25
	AODC	M70/2a_760_PUC-33
<b>northern transition (Ia)</b>	PW	M70/2a_757_PUC-10; M70/2a_757_PUC-22
	SR	M70/2a_757_PUC-11
	AODC	M70/2a_757_PUC-52
<b>bacterial mats (II)</b>	Benthic chamber	M70/2b_825_CALMAR-1; MSM13/3_929-1_CHAM3; MSM13/3_933-1_CHAM4; MSM13/3_971-1_CHAM13
	Microprofiler	M70/2b_825_PROF-1; MSM13/3_971-1_MICP12
	Planar Optode	(M70/2b_825_PO-2)
	PW	M70/2a_760_PUC-39; M70/2a_760_PUC-41
	SR	M70/2a_760_PUC-12; M70/2b_825_PUC-9; MSM13/3_929-1_PUC1; MSM13/3_929-1_PUC9
	AOM	M70/2a_760_PUC-44; M70/2b_825_PUC-9; M70/2b_825_PUC-11; M70/2b_825_PUC-12
	AODC	M70/2a_760_PUC-40; M70/2b_825_PUC-9
<b>biogenic mounds (III)</b>	Benthic chamber	M70/2b_825_CALMAR-2
	Planar Optode	M70/2b_825_PO-1
	PW	M70/2a_765_PUC-14; M70/2a_765_PUC-52
	SR	M70/2a_765_PUC-26; M70/2a_765_PUC-42
	AOM	M70/2a_765_PUC-6; M70/2a_765_PUC-38
	AODC	M70/2a_765_PUC-26
<b>sulfur band (IV)</b>	Benthic chamber	M70/2b_790_CALMAR-1; M70/2b_798_CALMAR-1; MSM13/3_953-1_CHAM6
	Microprofiler	M70/2b_790_MICP-1; M70/2b_825_PROF-1; M70/2b_825_PROF-2; MSM13/3_944-1_MICP3; MSM13/3_947-1_MICP7
	PW	M70/2a_765_PUC-46; M70/2a_765_PUC-47
<b>sulfur band (IV)</b>	SR	M70/2a_765_PUC-8; M70/2a_765_PUC-49; M70/2a_765_PUC-68;

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Habitat	measurements	PANGAEA database event labels
		MSM13/3_947-1_PUC27; MSM13/3_947-1_PUC32; MSM13/3_968-1_PUC8; MSM13/3_968-1_PUC110; MSM13/3_968-1_PUC117; MSM13/3_968- 1_PUC19; MSM13/3_968-1_PUC23
	AOM	M70/2a_765_PUC-8; M70/2a_765_PUC-49; M70/2a_765_PUC-68
	AODC	M70/2a_765_PUC-46
<b>reference</b>	SR	M70/2a_763
	AOM	M70/2a_763
	AODC	M70/2a_763

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Figure 1: Two-dimensional oxygen distribution in sediments of a (A) biogenic mound and (B) a bacterial mat (scale-oxygen saturation %) recorded in situ by a planar optode. The (A) During the deployment at a mud shrimp burrow, the oxygen content was decreasing to background values within 6 hrs. (B) Oxygen penetration in the bacterial mat is limited to 0.17 mm. The picture results from averaging nine single oxygen pictures taking during one deployment.



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## 19 **Material & Methods Supplements**

### 20 **Planar Optode**

21 The two dimensional oxygen distribution in the sediment was measured in 2006 with an  
22 autonomous in situ planar optode module (Glud et al., 2001; Glud et al., 2005). The planar  
23 optode foil was produced according to the method described by Precht and coworkers (2004).  
24 The oxygen sensitive layer is based on a platinum(II)mesotetra-(pentafluorophenyl)-porphyrin  
25 (Frontier Scientific, Inc.) embedded in polystyrene and immobilized on a transparent  
26 polyester support foil by a knife-coating procedure. The sensing layer had a thickness of 15-  
27 20  $\mu\text{m}$ . The optode foil was placed in front of the inverse periscope of the planar optode  
28 module (detailed description see Glud et al., 2001; Wenzhöfer and Glud, 2004; Glud et al.,  
29 2005). The oxygen-quenched fluorescence emitted by the fluorochrome was recorded using a  
30 modulated digital charge coupled device (CCD) camera (Sensi Cam, PCO Computer Optics).  
31 The planar optode was calibrated on board with oxygen saturated and anoxic seawater at in  
32 situ temperature. The calibrated oxygen image covered an area of 6 x 8 cm with a spatial pixel  
33 resolution of 128  $\mu\text{m}$  (CCD camera chip size, 1280 x 1024 pixel, two pixels horizontal and  
34 vertical binning). Oxygen images were calculated based on a modified version of the Stern-  
35 Vollmer equation (Holst et al., 1998). The image analysis and the calculations of lifetime and  
36 oxygen images were done with homemade software using the program MATLAB 6.5  
37 (MatchWorks Inc.)

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### 39 **Benthic chamber**

40 Total benthic oxygen uptake (TOU) and total methane efflux (only measured in 2009) were  
41 determined with a benthic chamber module equipped with two Clark-type oxygen mini-

42 electrodes or oxygen optodes (AADI, Norway) by the ROV QUEST to ensure targeted  
43 measurements at selected spots. A valve in the chamber lid releases the enclosed water during  
44 the careful placement of the chamber on the seafloor by the ROV. The centrally stirred  
45 chamber had a radius of 9.5 cm and hence enclosed an area of 284 cm<sup>2</sup> together with 10-20  
46 cm (equivalent to 4-6 L) of overlying bottom water. During each deployment the exact water  
47 height inside the chamber was determined by visual inspection with the ROV camera system.  
48 The oxygen concentration in the enclosed water volume was continuously recorded with  
49 either two Clark-type oxygen mini-electrodes or oxygen optodes (AADI, Norway) that were  
50 mounted to the chamber lid (Glud et al., 2009). The sensors were calibrated against bottom  
51 water samples and a zero reading was recorded at in situ temperature on board. During the  
52 chamber deployment up to 5 water samples with 50 mL syringes were taken in pre-  
53 programmed time intervals. After recovery in 2009, the water samples were injected into  
54 airtight glass vials containing NaOH pellets to release the dissolved methane into the  
55 headspace. Methane concentrations were determined by injecting 100 µL of the glass bottle  
56 headspace into a gas chromatograph (5890A, Hewlett Packard). At the “sulfur band” habitat  
57 sampled in 2006, water samples were preserved already at the seafloor in 5 mL 5% ZnAc  
58 solution, which were added to the syringes before the deployment in order to measure in situ  
59 sulfide flux from the sediment into the water column.

60 TOU, methane, and sulphide fluxes (mmol m<sup>-2</sup> d<sup>-1</sup>) were calculated from the linear regression  
61 of the respective concentration versus time (Wenzhöfer and Glud, 2002):

$$62 \quad (3) \quad \text{TOU/CH}_4/\text{H}_2\text{S - flux} = \frac{dC}{dt} \times \frac{V_{\text{chamber}}}{A_{\text{chamber}}} \quad \text{TOU/CH}_4/\text{H}_2\text{S - flux} = \frac{dC}{dt} \times \frac{V_{\text{chamber}}}{A_{\text{chamber}}}$$

63 where dC/dt is the change of the concentration over the incubation time, V<sub>chamber</sub> is the volume  
64 of the enclosed water, and A<sub>chamber</sub> is the area of the sediment enclosed by the chamber.

65 **Microprofiler**

66 Bases on the microsensor measurements diffusive oxygen uptake (DOU) and sulphide flux  
67 were calculated using Fick's first law of diffusion as described previously (Revsbech et al.,  
68 1983):

69 (4)  $J = D \times \frac{\delta C}{\delta z}$

70 where J is the diffusive flux ( $\text{mmol m}^{-2} \text{d}^{-1}$ ), D is the diffusion coefficient of oxygen in water  
71 ( $D = 1.64 \times 10^{-9} \text{ m}^2 \text{ s}^{-1}$ ) corrected for temperature and salinity (Li and Gregory, 1974) and  
72  $\delta C / \delta z$  is the concentration gradient ( $\delta C$  ( $\text{mmol m}^{-3}$ );  $\delta z$  (m)). To calculate fluxes in the  
73 sediment, the diffusion coefficient (D) had to be corrected for the porosity ( $\varphi$ ) of the sediment  
74 with the following equation (Iversen and Jørgensen, 1993):

75 (5)  $D_{\text{Sed}} = \frac{D}{1 + 3 \times (1 - \varphi)}$

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77 Sulphide fluxes were calculated from the highest concentration gradients in the profile  
78 (Iversen and Jørgensen, 1993) with the specific diffusion coefficient for sulphide ( $D_{\text{Sed}} = 0.66$   
79  $\times 10^{-9} \text{ m}^2 \text{ d}^{-1}$ , (Schulz and Zabel, 1999)) and the following equation:

80 (6)  $J = \varphi \times D_{\text{Sed}} \times \frac{\delta C}{\delta z}$

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