

1 **A laboratory experiment on the behavior of soil-**
2 **derived core and intact polar GDGTs in aquatic**
3 **environments**

4

5 **Francien Peterse^{1,*}, Christopher M. Moy², Timothy I. Eglinton¹**

6 [1] {ETH Zürich, Geological Institute, Sonneggstrasse 5, 8092 Zürich, Switzerland}

7 [2] {University of Otago, Geology Department, Dunedin, New Zealand}

8 [*] {Now at: Utrecht University, Department of Earth Sciences, Utrecht, the
9 Netherlands}

10 Correspondence to: F. Peterse (f.peterse@uu.nl)

11

12 **Abstract**

13 We have performed incubation experiments in order to examine the behavior of soil-
14 derived branched glycerol dialkyl glycerol tetraether (brGDGT) membrane lipids
15 upon entering an aquatic environment and to evaluate the processes that potentially
16 take place during their fluvial transport from land to sea. We incubated a soil from the
17 Rakaia River catchment on the South Island of New Zealand using Rakaia River
18 water and ocean water collected near the river mouth as inocula for a period of up to
19 152 days. The concentrations, as well as the relative distribution of brGDGTs derived
20 from intact polar ('living'; IPL) lipids and core ('fossil'; CL) lipids remained
21 unaltered over the course of the experiment. **Although the stability of the brGDGTs**
22 **may be a consequence of the higher than natural soil:water ratio used in the laboratory**
23 **experiment,** the substantial increase (27-72%) in the total pool of isoprenoid GDGTs
24 (isoGDGTs) in all incubation setups, **including the control using distilled water,**
25 **indicates that entering an aquatic environment does influence the behavior of soil-**
26 **derived GDGTs. However, the availability of water appears to be more important than**
27 **its properties.** As a consequence of increasing isoGDGT concentrations, a decrease in
28 Branched and Isoprenoid Tetraether (BIT) index values - a proxy for the relative input
29 of fluvially discharged soil material into a marine system - became evident after an
30 incubation period of 30 days, with a maximum final decrease of 0.88 to 0.74 in the

1 experiment with river water. The relative distribution within the isoGDGT pool shows
2 changes with time, suggesting that isoGDGT producers may either have different
3 rates of membrane adaptation or production/degradation, or that preferential release
4 from the soil matrix or a shift in source organism(s) may take place. While the
5 apparent stability of soil brGDGTs during this incubation experiment reinforces their
6 potential as tracers for land-sea transport of soil organic carbon and their use in
7 paleoclimate reconstructions, the distributional differences between GDGTs in river
8 water and nearby soil, as well as in river and ocean water, indicate that further
9 research is needed to pinpoint the sources of GDGTs that are ultimately discharged to
10 the oceans and are subsequently archived in continental margin sediments.

11

12 **1 Introduction**

13 The global carbon cycle encompasses a myriad of biogeochemical processes that
14 influence our climate and link all carbon reservoirs on the Earth surface. Soils are
15 considered to play a very active and fundamental role in this cycle, as their
16 mobilization from land and subsequent deposition in marine sediments serves as a
17 long-term sink of atmospheric CO₂. However, the exact magnitude and mechanisms
18 of soil organic carbon (OC) transfer from terrestrial source to marine sink remain
19 elusive (e.g. Cole et al., 2007; Weyhenmeyer et al., 2012), which partly stems from
20 the lack of a suitable tracer of the soil OC pool. Our current insights in land-sea soil
21 OC transport dynamics are primarily based on bulk properties of river suspended
22 particulate matter (SPM). However, next to soil OC, river SPM also comprises carbon
23 derived from aquatic production and ‘fossil’ carbon from rock erosion, each of which
24 influences the bulk properties of SPM (Blair et al., 2010).

25 Analytical advances in the field of organic biogeochemistry have yielded a growing
26 number of powerful tools with the potential to exclusively target components of
27 specific pools, including soil OC. In this context, branched glycerol dialkyl glycerol
28 tetraethers (brGDGTs; Fig. 1) have been put forward as tracers for soil-derived OC in
29 carbon transport studies (e.g. Hopmans et al., 2004; Kim et al., 2006). Since their
30 discovery (Sinninghe Damsté et al., 2000; Schouten et al., 2000), brGDGTs have been
31 found in soils and peats worldwide (Weijers et al., 2007b, 2009; Liu et al., 2010). The
32 exact organism(s) that produce(s) these compounds have yet to be identified, but
33 current evidence points towards an origin from heterotrophic soil bacteria (Pancost

1 and Sinninghe Damsté, 2003; Oppermann et al., 2010; Weijers et al., 2010) from the
2 phylum of *Acidobacteria* (Weijers et al., 2009; Peterse et al., 2010; Sinninghe Damsté
3 et al., 2011). BrGDGTs have been used as a proxy for the relative input of fluvially
4 transported soil OC in marine systems based on their abundance in coastal marine
5 sediments relative to that of crenarchaeol (Fig. 1), an isoprenoid GDGT (isoGDGTs)
6 produced by marine *Thaumarchaeota* (Sinninghe Damsté et al., 2002), parameterized
7 as the Branched and Isoprenoid Tetraether (BIT) index (Hopmans et al., 2004).
8 Moreover, subtle variations in the molecular distribution of the brGDGTs have been
9 found to relate to mean annual air temperature and soil pH (Weijers et al., 2007b),
10 indicating their potential for utility in paleoclimate studies. Indeed, using a
11 combination of the Methylation of Branched Tetraethers (MBT) and Cyclisation of
12 Branched Tetraether (CBT) indices (the so-called “MBT-CBT” proxy; Weijers et al.,
13 2007b, recently revised as MBT'-CBT by Peterse et al., 2012), down-core variations
14 in brGDGT distribution in fluvially-dominated continental margin sediments have
15 been used to obtain an integrated climate history of the adjacent drainage basin (e.g.
16 Weijers et al., 2007a).

17 Although GDGT-based proxies are finding increase use, our understanding of
18 mobilization and transfer of soil OC and associated GDGT signals to fluvial
19 environments, and the processes acting upon these components during transport are
20 still poorly understood. There is emerging evidence, for example, for *in situ*
21 production of brGDGTs in aquatic environments including high-latitude fjord systems
22 (Peterse et al., 2009), open shelf sediments (Zhu et al., 2011), lakes (e.g. Sinninghe
23 Damsté et al., 2009; Tierney and Russell, 2009), and most recently, rivers (De Jonge
24 et al., 2014; Kim et al., 2012; Zell et al., 2013b; Zhang et al., 2012). The presence of
25 intact polar lipid (IPL) precursors of brGDGTs in SPM from the Amazon (Zell et al.,
26 2013b), Pearl (Zhang et al., 2012), and Yenisei rivers (De Jonge et al., 2014) provides
27 strong evidence for aquatic brGDGT production, as the IPL head groups are thought
28 to be rapidly lost upon cell death (Harvey et al., 1986; White et al., 1979), although
29 recent studies suggest that IPLs with ether bound headgroups may actually have very
30 slow turnover rates in marine sediments (e.g. Logemann et al., 2011; Xie et al., 2013).
31 Regardless, the core lipids (CLs) that are released after IPL degradation are
32 considered to represent ‘dead’, or fossil material - the fraction that is considered to be
33 stored in sedimentary archives - and targeted for paleoclimate reconstruction. Hence,

1 during fluvial transport from land to sea, the initial soil-derived brGDGT distribution
2 entering a river may be modified by the addition of aquatic produced brGDGTs with a
3 potentially different signature, but also by microbial degradation and transformation
4 of the soil brGDGTs.

5 In order to explore the latter process, i.e. the behavior of soil brGDGTs upon entering
6 an aquatic system, we performed a laboratory experiment in which we incubated soil
7 using river water or ocean water from near the mouth of the same river as microbial
8 inocula. In the frame of a larger study on fluvial transport of soil organic carbon from
9 land to sea on the South Island of New Zealand, we used soil and river water from the
10 Rakaia River catchment for the experiment. In addition, the braided character of the
11 Rakaia River may reduce the potential contribution of in situ produced brGDGTs due
12 to the generally harsh and unstable conditions in such river systems. This should
13 enable the monitoring in a controlled setting of what could also be an actual soil-
14 derived brGDGT signal in the river. For that matter, concentration and distributional
15 changes in IPL-derived and CL-brGDGTs were determined at different time intervals
16 during the incubation experiment. Although isoGDGTs (Fig. 1) are primarily
17 associated with marine archaea, they are also produced by soil *Thaumarchaeota*
18 (Sinninghe Damsté et al., 2012), albeit in small amounts relative to the brGDGTs
19 (Weijers et al., 2006b). In the marine realm, changes in isoGDGT distributions form
20 the basis for the TEX₈₆ index, which is used as a proxy for sea surface temperature
21 (SST; Schouten et al., 2002; Kim et al., 2010). Large contributions from soil
22 isoGDGTs to the total isoGDGTs pool in marine sediments may consequently
23 complicate the interpretation of TEX₈₆-derived SST records from near continental
24 margins (Weijers et al., 2006b). Therefore, all soil-derived GDGTs, i.e. both branched
25 and isoprenoidal, were monitored during the experiment.

26

27 **2 Materials and methods**

28 **2.1 Sample collection and incubation setup**

29 The Rakaia River is one of the largest braided rivers on the South Island of New
30 Zealand and has a mean annual discharge of 203 m³/s. The Rakaia is estimated to
31 contribute 4.15 Mt of sediment a year, accounting for approximately 5% of the total
32 South Island sediment yield to the adjacent continental shelf (Hicks et al., 2011). The

1 river originates in the Southern Alps and flows through the Canterbury Plains before
2 reaching the Pacific Ocean approximately 150 km from its source. Fresh water from
3 the river was sampled in jerry cans close to the town Rakaia (S 43°44'57.9", E
4 172°01'52.9"), about 20 km upstream from the river mouth and outside of the tidal
5 influence. The ocean water used as inoculum was collected from the shoreline (S
6 43°54'15.1", E 172°11'41.9"), close to the river mouth (Fig. 2). Based on pH
7 measurements, this water (pH 7.6) was strongly fluviually influenced (pH 7.5). Surface
8 soil (0-10 cm; pH 5.8) was collected on the Canterbury Plains within the Rakaia River
9 catchment (S 43°52'00.8", E 172°09'10.8"). All soil material and incubation water
10 was collected in January 2012 and directly transported to the University of Otago,
11 where the incubation experiment was set up the next day. SPM was collected by
12 filtration of river and ocean water (100L onto a 0.2 µm polyethersulfone membrane
13 and 10L onto a pre-combusted 0.7 µm GF/F, respectively) in the field, after which the
14 filters were stored frozen.

15 In the laboratory at the University of Otago, the soil was sieved over a 2 mm mesh to
16 remove roots and homogenize the sample, after which two 10 g subsamples of the soil
17 were directly frozen at -20°C for subsequent determination of initial GDGT
18 composition. For the incubations, twenty-eight 250 ml bottles were filled with 10 g
19 soil subsamples, after which 100 ml of river water or ocean water, untreated to
20 preserve the natural microbial community, was added to twelve bottles each. Distilled
21 water was added to the remaining four bottles to serve as control setup. The
22 incubation bottles were loosely plugged with cotton wool, creating quasi-aerobic
23 conditions, and placed on a shaker table from January to June 2012. Based on IPL
24 degradation rates observed in previous studies (Harvey et al., 1986; Logemann et al.,
25 2011), two bottles with river water and two bottles with ocean water were taken from
26 the shaker table and immediately placed into a freezer (-20°C) at 1, 7, 14, 30, 91, and
27 152 days after the start of the experiment. Duplicate control samples were frozen after
28 91 and 152 days. At the end of the experiment all bottles were shipped on ice to ETH
29 Zürich, Switzerland, where they were briefly thawed and the supernatant was pipetted
30 off. The soils were then freeze dried, and stored frozen at -20°C until subsequent
31 sample work-up and analysis.

32 **2.2 Total organic carbon and nitrogen analysis**

1 The freeze-dried, homogenized soils were weighed in silver capsules and fumigated
2 with 1M HCl for three days and neutralized with NaOH for two days in a desiccator
3 to remove carbonates prior to total organic carbon (TOC) and total nitrogen (TN)
4 analysis. The measurements were performed on a vario MICRO elemental analyzer at
5 ETH Zürich.

6 **2.3 GDGT extraction and analysis**

7 Freeze-dried soil samples (3-4 g) were solvent-extracted three times using a Bligh and
8 Dyer technique modified from Sturt et al. (2004). In short, a solvent mixture of
9 MeOH:dichloromethane (DCM): phosphate buffer at pH 7.4 (2:1:0.8, v/v/v) was
10 added to the soils and ultrasonically extracted for 10 min. The extract was collected
11 each time after centrifuging at 1000 rpm for 5 mins. DCM and phosphate buffer were
12 added to the combined extracts to a volume ratio of 1:1:0.9 to obtain phase separation.
13 The DCM phase, containing the GDGTs, was collected after centrifuging, and the
14 remaining solvent was rinsed twice with DCM. The combined DCM phases were
15 dried under N₂ and passed over a silica column (deactivated with 1 weight% water) to
16 separate CLs and IPLs according to (Pitcher et al., 2009), with the exception that
17 hexane:ethyl acetate 1:1 (v/v) was used to elute the CLs. An aliquot of the IPL
18 fraction was analyzed directly for CLs in order to assess potential carry over. The
19 remainder of IPL fraction was dissolved in 6N HCl in MeOH and heated at 100°C for
20 at least 3h to release IPL-bound CLs. A known amount of C₄₆ GDGT standard
21 (Huguet et al., 2006) was added to all fractions prior to analysis.

22 The filters were freeze dried and extracted with a MARS Xpress microwave
23 extraction system, using DCM:MeOH 9:1 (v/v). After centrifugation (5 min at 400
24 rpm), the total lipid extract (TLE) was pipetted off and the residues were rinsed twice
25 with DCM:MeOH 9:1 (v/v). The combined extracts were dried under N₂ with a
26 known amount of C₄₆ GDGT standard, after which the TLEs were separated into an
27 apolar and a polar (GDGT) fraction by passing them over a silica (1% water
28 deactivated) column using hexane:DCM 9:1 (v/v) and DCM:MeOH 1:1 (v/v),
29 respectively.

30 All GDGT-fractions were dissolved in hexane:isopropanol 99:1 (v/v), filtered over a
31 0.45 µm PTFE filter, and analyzed using high performance liquid
32 chromatography/atmospheric pressure chemical ionization – mass spectrometry

1 (HPLC/APCI-MS) with an Agilent 1260 Infinity series LC/MS at ETH Zürich
2 according to Schouten et al. (2007a). Separation of the GDGTs was achieved with a
3 Grace Prevail Cyano column (3µm, 150×2,1mm) after passing through a guard
4 column of the same material (5µm, 7,5×2,1mm) with hexane:isopropanol (99:1, v/v)
5 as an eluent at a flow rate of 0.2 ml/min. The GDGT-fractions eluted isocratically
6 with 90% A and 10% B for 5 min, and then with a linear gradient to 18%B for 34
7 min, where A=hexane and B=hexane:isopropanol 9:1 (v/v). Selective ion monitoring
8 of the $[M+H]^+$ was used to detect the different GDGTs. Although brGDGTs appear to
9 have a higher response factor than isoGDGTs (Schouten et al., 2013a), quantification
10 was done assuming similar response factors for all GDGTs and the internal standard.
11 Hence, reported brGDGT concentrations are likely overestimated compared to those
12 of the isoGDGTs. The amounts of all GDGTs are given as the average and the range
13 of variation of the duplicate incubation samples. The amount of GDGTs in each
14 sample has been corrected for carry over of CLs into the IPL fraction, which was on
15 average <8% for brGDGTs and <1% for isoGDGTs.

16 **2.4 GDGT-based index calculations**

17 The CBT and MBT' indices based on brGDGTs were calculated according to
18 (Weijers et al., 2007b):

$$19 \text{ CBT} = -\log ((Ib + IIb) / (Ia + IIa)) \quad (\text{Eq. 1})$$

20 and (Peterse et al., 2012):

$$21 \text{ MBT}' = (Ia + Ib + Ic) / (Ia + Ib + Ic + IIa + IIb + IIc + IIIa) \quad (\text{Eq. 2})$$

22 The BIT index was calculated following (Hopmans et al., 2004):

$$23 \text{ BIT} = (Ia + IIa + IIIa) / (\text{cren} + Ia + IIa + IIIa) \quad (\text{Eq. 3})$$

24 and for the TEX₈₆ index based on isoGDGTs the equation of (Schouten et al., 2002)
25 was used:

$$26 \text{ TEX}_{86} = (\text{GDGT-2} + \text{GDGT-3} + \text{cren}') / (\text{GDGT-1} + \text{GDGT-2} + \text{GDGT-3} + \text{cren}') \quad (\text{Eq. 4})$$

28 Roman numerals and GDGT names refer to the molecular structures in Fig. 1.
29 Regular reruns of selected samples on the HPLC-MS at ETH show that the analytical
30 error on the indices is <0.01.

1

2 **3 Results and discussion**

3 **3.1 TOC and TN concentrations over time**

4 The TOC concentration in the soil is 1.6-1.7% at $t=0$, and varies in a range from 1.2-
5 1.7%, whereas the TN concentration is 0.15% at $t=0$ and remains practically stable
6 (0.12-0.15) over the course of the experiment (Fig. 3). Given the limited range of
7 variation in TOC and TN concentrations, no direct influence of TOC or TN on the
8 GDGTs in this experiment is assumed during the following discussion.

9 **3.2 BrGDGT concentrations over time**

10 The total brGDGT pool (IPLs + CLs) initially present in the soil is 239 ± 13 ng/g. In
11 the different incubation experiments these concentrations varied between 187-325
12 ng/g (river water), 186-322 ng/g (ocean water), and 252-407 ng/g (distilled water
13 control) (Fig. 4). Although the abundance of CL-brGDGTs seems to increase with
14 time in the control setup, the trend is only weak and not significant ($r^2 = 0.34$, $p =$
15 0.122) due to the large spread in concentration between the two replicate samples at
16 $t=152$ days (Fig. 4). Overall, the CL concentrations are within the same range
17 between different experimental setups and remain essentially constant through time.
18 Since the soils were incubated under natural light conditions, this suggests that soil-
19 derived brGDGTs might not be sensitive to photodegradation during the incubation.

20 The contribution of IPLs to the total amount of brGDGTs in the soil at $t=0$ is
21 $12.4 \pm 0.3\%$ (Fig. 5), which is in the same range as has previously been reported for
22 temperate soils from the Netherlands, Scotland, and the UK (Peterse et al., 2010;
23 Weijers et al., 2011). The contribution of IPL-derived brGDGTs to the total pool
24 varies mostly within the same range in the river water (7-16%), ocean water (9-15%),
25 and control (9-17%) setups, and like the total amount of CL-brGDGTs, IPL-brGDGT
26 concentrations show no strong trends or changes over time (river water: $r^2 = 0.34$, $p =$
27 0.014 ; ocean water: $r^2 = 0.00$, $p = 0.944$; distilled water: $r^2 = 0.20$, $p = 0.268$). This
28 suggests that brGDGTs are either produced at the same rate that they are degraded, or
29 that production and/or degradation of brGDGTs occurs at such low rates that these
30 processes are not detected within the timeframe of this experiment. The apparent
31 stability of the IPL-derived, or 'living' brGDGT signature (Fig. 4) during the

1 experiment is surprising, given the perceived lability of IPLs in general and their
2 susceptibility to loss of their headgroups within a few days after cell lysis (Harvey et
3 al., 1986; White et al., 1979). However, previous studies have indicated that the type
4 of headgroup and the bond through which it is attached to a core lipid may influence
5 the rate of degradation, offering a potential explanation for the absence of changes in
6 IPL-brGDGT concentration in our experiment. For example, after a 96 hr aerobic
7 incubation of a mixture of IPLs with different headgroups in beach sediment, 70% of
8 the phospholipids had degraded, compared to only 3% of the glycolipids (Harvey et
9 al., 1986). Although a primarily glycosidic headgroup composition would be expected
10 based on these former results, the majority of IPL-brGDGTs in soils, peat, and lake
11 sediments are thus far identified as phospholipids (Weijers et al., 2011; Peterse et al.,
12 2011; Tierney et al., 2012), implying that they should be sensitive to degradation,
13 counter to our observations. That this may not be the case may potentially be
14 explained by the findings of Logemann et al. (2011), who recently showed that the
15 bond type within the CL, rather than the type of headgroup and its connection to that
16 CL, determines IPL sensitivity to degradation. In their experiment, membrane lipids
17 in which the side chains were ester-bound, as is common for bacteria, started to
18 rapidly degrade within the first 5 days of the incubation period, whereas the
19 abundance of lipids with ether-bound chains remained invariant during the entire 97-
20 day experiment. BrGDGTs are thought to be produced by bacteria, but they also
21 possess archaeal traits, including their tetraether structure (Weijers et al., 2006a). As a
22 consequence, the presence of at least four ether bonds in each brGDGT may thus not
23 only protect the IPL, but also the CL-brGDGTs from degradation during our
24 experiment, even when the IPLs possess a phospho head group.

25 Weijers et al. (2010) determined a turnover rate for the total pool of CL brGDGTs in a
26 soil of about two decades, which implies that any changes in brGDGT concentration
27 or distribution may indeed not yet be detectable after 152 days of incubation. On the
28 other hand, a recent study by Huguet et al. (2013) indicated that the brGDGT
29 signature in a French peat bog had completely adapted to the 2°C maximal daytime
30 temperature increase induced by the placement of open top mini-greenhouses within a
31 period of less than 26 months. Although these results suggest that brGDGTs are
32 turned over at a substantially faster rate, significant changes were only observed in the
33 latter phase of the warming experiment, and were primarily distributional, as their

1 concentration remained the same (Huguet et al., 2013). The unaltered IPL-brGDGT
2 concentrations in our experiment may thus be an indication that the time frame of our
3 incubation has been too short to reveal any degradation processes based on IPL-
4 brGDGT abundance. However, since the type of headgroup attached to the GDGTs
5 has not been identified, our data does not allow to fully exclude equal production and
6 degradation rates as possible explanation for the apparently stable IPL-brGDGT
7 concentrations during the experiment.

8 While nutrient conditions in the river and ocean water added to the soil are
9 presumably different, this does not seem to have had any effect on brGDGT
10 concentrations. Moreover, given the soil:water ratio (10 g in 100 ml) used in this
11 experiment, the amounts of nutrients that are released from the soil after the addition
12 of water is likely substantially higher than the concentrations initially present in the
13 river and ocean water. Nevertheless, the effect of nutrient availability on brGDGTs in
14 soils has not yet received much attention, although recently was shown that variations
15 in the amount and distribution of brGDGTs in lake sediments did not relate to water
16 column nutrient concentrations (Loomis et al., 2014).

17 **3.3 IsoGDGT concentrations over time**

18 Crenarchaeol is the most abundant isoGDGT in the New Zealand soil at the start of
19 the incubation, although its total amount (CLs + IPLs; 63 ± 7 ng/g), as well as that of
20 all other isoGDGTs (50 ± 5 ng/g) is low compared to the total pool of brGDGTs
21 (239 ± 13 ng/g). In contrast, the average fraction of isoGDGTs initially present in the
22 soil as IPL is 54% for crenarchaeol to 76% for isoGDGT-2, which is substantially
23 higher than proportion of brGDGTs occurring as IPLs (12%) at the commencement of
24 the experiment. Similar proportions of IPL-derived crenarchaeol have been found in
25 soils from the Amazon (~50%; Zell et al., 2013b). The total concentration (IPLs +
26 CLs) of all isoGDGTs and increases substantially during the experiment (Fig. 4), and
27 that of isoGDGT-0 in the ocean water setup has even doubled by the end of the
28 incubation (24 ± 2 to 48 ± 5 ng/g, $r^2 = 0.58$, $p = 0.000$).

29 The concentrations of the individual IPL-derived isoGDGTs vary between the
30 different incubation setups, but are overall highest in ocean water (Fig. 4;
31 Supplementary Table). The difference in IPL-isoGDGT concentration between fresh
32 and saline water incubations may potentially be a result of the addition of varying

1 amounts of aquatic isoGDGTs to the soil-derived isoGDGT pool. However, 100 ml of
2 ocean water only contained ~0.5 ng isoGDGTs, which is less than 1% of their initial
3 concentration in the soil. Alternatively, the isoGDGTs could be increasingly released
4 from the soil matrix upon mixing with (saline) water, although the absence of a
5 simultaneous increase in brGDGTs suggests that this is not the most likely scenario.
6 Finally, since *Thaumarchaeota* play an important role in the nitrogen cycle in soils
7 and the marine environment (Leininger et al., 2006; Wuchter et al., 2006; Zhang et al.,
8 2010), nutrient availability may provide another explanation for the concentration
9 differences between experiments. So far, crenarchaeol has been found in all
10 (enrichment) cultures of ammonia-oxidizing *Thaumarchaeota*, and is thus considered
11 as a biomarker lipid for this group (de la Torre et al., 2008; Pitcher et al., 2010, 2011;
12 Sinninghe Damsté et al., 2012), although a recent study suggested that Marine Group
13 II Euryarchaeota may also produce crenarchaeol (Lincoln et al., 2014). Nevertheless,
14 the relatively high crenarchaeol concentration in the incubation setup using ocean
15 water could then be a result of ammonium release due to the higher abundance of
16 exchangeable cations (e.g. Na^+ , Mg^+) in ocean water compared to fresh water. This is
17 in agreement with the common method to determine available NH_4^+ in soils, which
18 involves shaking soil in a KCl or Na_2SO_4 solution to extract ammonium from the soil
19 matrix (e.g. Mehlich, 1953). The concentration profiles of the other isoGDGTs are
20 comparable to that of crenarchaeol (Fig. 4), suggesting that they are most likely also
21 primarily derived from soil *Thaumarchaeota*.

22 Despite the significant increase of the total pools of crenarchaeol and isoGDGTs 0-3
23 in the river ($r^2 = 0.74$, $p = 0.000$ for crenarchaeol; $r^2 = 0.56$, $p = 0.000$ for isoGDGTs),
24 ocean water ($r^2 = 0.30$, $p = 0.014$ for crenarchaeol; $r^2 = 0.40$, $p = 0.003$ for
25 isoGDGTs), as well as control setups ($r^2 = 0.52$, $p = 0.034$ for crenarchaeol; the
26 increase of isoGDGTs is not significant $r^2 = 0.24$, $p = 0.215$), there is no statistically
27 significant increase in the concentration of individual IPL-derived isoGDGTs during
28 the course of the incubation experiment (e.g. max. $r^2 = 0.18$ for IPL-derived
29 crenarchaeol in ocean water; Fig. 4). We assume therefore that the production rate of
30 IPL-isoGDGTs is approximately constant during this time. The proportional decrease
31 of IPL-isoGDGTs with time (Fig. 5) subsequently indicates that the increase in the
32 overall isoGDGT pool is a result of CL accumulation. This in turn confirms the
33 general finding that the turnover of the IPL pool, and thus the release of CL-

1 isoGDGTs, is faster than the degradation of these CLs. Regardless, the increase in
2 isoGDGT concentrations in all incubation setups suggests that availability of water is
3 likely more important than the type of water that is added to the soil.

4 **3.4 GDGT distribution changes**

5 To evaluate if and how soil GDGT signatures are modified during the incubations, we
6 calculated brGDGT and isoGDGT-based indices commonly used in paleoclimate and
7 carbon cycle studies (i.e. the MBT', CBT, BIT, TEX₈₆; Supplementary Table).

8 Next to the absence of changes in brGDGT concentration, also the MBT' and CBT
9 indices for the ocean and river water experiments exhibit minimal variation over time,
10 remaining within a range of 0.04 for the MBT' index, and 0.02 for the CBT (Fig. 6).
11 Surprisingly, the largest changes are observed in the control experiment using distilled
12 water. The MBT' index for the IPL-derived brGDGT fraction in the control setup
13 increases from 0.23 ± 0.00 to 0.29 ± 0.04 ($r^2 = 0.57$, $p = 0.023$; Fig. 6), and is primarily
14 caused by an increase in the concentration of brGDGT-Ia (from 23 ± 0.3 to $29 \pm 3.9\%$).
15 A closer look at the relative distributions of each of the brGDGT types at $t=0$ and
16 $t=152$ days reveals a subtle shift in the contribution of IPL-derived brGDGT-III
17 (decreasing from 23 ± 1.3 to $19 \pm 0.1\%$) to brGDGT-II (increasing from 54 ± 0.4 to
18 $58 \pm 0.4\%$) in river water (data not shown). Since both brGDGTs-II and III in the
19 denominator of the MBT' index (Eq. 2), and the fraction of IPL-derived brGDGT-I
20 remains stable, this change is not reflected by the index values. Given that the
21 absolute amounts of IPL-derived brGDGTs do not significantly change with time, it is
22 hard to determine the exact processes that cause the distributional changes, as both *in*
23 *situ* production and preferential degradation of specific brGDGTs could influence the
24 total concentration. Since brGDGTs with cyclopentane moieties were only present
25 below detection limit in most of the IPL-derived fractions, it was not possible to
26 reliably calculate CBT index values for any of these fractions.

27 The BIT index is clearly influenced by the substantial increase in crenarchaeol during
28 the experiment, and starts to decrease between $t=30$ and $t=91$ days in all water types
29 (Fig. 6). Due to the large percentage of crenarchaeol that is present as IPL compared
30 to that of the brGDGTs, the BIT index of the IPL-derived fraction is much (0.48)
31 lower than that of the CL fraction (0.88) at $t=0$. The lowest BIT index value for the
32 IPL-derived fraction is 0.35 at $t=152$ days in ocean water (Fig. 6), and stems from the

1 increase in crenarchaeol (and isoGDGTs in general) in this experiment. The trend of a
2 lower BIT value for the IPL versus the CL fraction has also been found for river SPM
3 from the Yenisei (De Jonge et al., 2014) and Amazon (Zell et al., 2013b), where the
4 difference was explained by a lower degradation rate of soil-derived crenarchaeol
5 IPLs compared to brGDGTs and/or a contribution of *in situ* produced crenarchaeol
6 IPLs in the river (Zell et al., 2013b). Our incubation results suggest that the
7 degradation of IPLs from soil-derived crenarchaeol is faster than that of brGDGTs,
8 and that the difference in BIT index between the IPL-derived and CL fractions is thus
9 mainly caused by the higher production rate of crenarchaeol. This is in agreement
10 with the recent finding that the BIT index of river SPM primarily tracks the seasonal
11 aquatic production of crenarchaeol rather than that of soil input (Yang et al., 2013;
12 Zell et al., 2013a). Our results furthermore indicate that care should be taken with
13 using the BIT index to constrain the input of fluvially transported soil OC into a
14 marine system. However, the stability of the soil brGDGT pool supports earlier
15 observations that tracing the absolute amount (as opposed to the relative abundance)
16 of brGDGTs in rivers during land-sea transport may provide a more reliable tracer of
17 soil OC at our study site (c.f. Fietz et al., 2011; Smith et al., 2012; Zell et al., 2013b).
18 Moreover, the spread in GDGT response factors between laboratories resulting in a
19 range of BIT values for the same sample provides another, independent argument to
20 use absolute concentration measurements (Schouten et al., 2013a).

21 TEX_{86} index values start to change between 30 and 91 days of incubation for the IPL-
22 fraction in river water and the CL-fraction in ocean water experiments (Fig. 6). We
23 find no clear explanation for the diverging trends for the different water types or IPL
24 versus CL fractions, but there are several factors that may contribute to the observed
25 changes. For example, shifts in the initial isoGDGT-community due to water contact,
26 different membrane adaptation rates, or preferential release from the soil matrix may
27 all have occurred. Alternatively, the relative distribution of isoGDGTs may differ per
28 type of headgroup in both marine (Schouten et al., 2008; Pitcher et al., 2011; Lengger
29 et al., 2012) and soil *Thaumarchaeota* (Sinninghe Damsté et al., 2012), so that the
30 deviation in TEX_{86} values may also be explained by different turnover rates per
31 headgroup. However, even though the total variation in TEX_{86} corresponds with a
32 4°C SST change in the experiment, given the low absolute amounts of terrestrial
33 isoGDGTs that are finally discharged to the ocean in this case, this will have a minor

1 impact on TEX₈₆ records from the marine environment where their abundances are
2 much higher.

3 **3.5 Experimental setup evaluation**

4 Despite prior evidence for in situ production of brGDGTs in several large rivers (e.g.
5 Kim et al., 2012; Zhang et al., 2012; Zell et al., 2013b; Yang et al., 2013; De Jonge et
6 al., 2014), we have not observed this in our experiment. The absence of an in situ
7 production signal in our data may be a consequence of the soil:water ratio of 1:10 that
8 we used in our experimental setup, which is likely more concentrated than in most
9 natural systems, so that the relatively large amount of soil-brGDGTs may have
10 overprinted any evidence of aquatic brGDGT production during the experiment. An
11 indication that in situ production in the Rakaia River may take place after all is
12 reflected by the offset between brGDGTs in river SPM and that in the soil (Fig. 6).
13 Although this soil-SPM offset is based on a single sample location in the entire river
14 basin and brGDGT distributions in soils may show substantial variation within a small
15 area (e.g. Weijers et al., 2007b; Naeher et al., 2014), it may also support the idea that
16 brGDGTs in soils and river SPM may have different sources. Notably, also the
17 brGDGT distributions in river and ocean SPM show an offset (Fig. 6), despite the
18 ocean water sample location close to the river mouth. Differences in biomarker
19 distributions in SPM from the fresh-saline water transition have previously been
20 attributed to hydrological sorting (e.g. Goñi et al., 1998), or, more specifically in case
21 of brGDGTs, additional marine production (e.g. Zhu et al., 2011; Zell et al., 2014). In
22 our case, this offset may have also been introduced by the use of filters with different
23 pore size to obtain the SPM from the river and ocean water (0.2 µm and 0.7 µm,
24 respectively). This has resulted in a comparison of ‘free living’ and ‘suspended’ lipid
25 fractions, which at least in marine water may have a different composition (Ingalls et
26 al., 2012; Close et al., 2014).

27 Furthermore, 152 days of incubation appears not to have been long enough to
28 determine the sensitivity of brGDGTs to degradation in aquatic environments based
29 on changes in their distribution or abundance. The unexpected apparent stability of
30 the IPL-derived brGDGTs suggests that the presence of IPL-brGDGTs in aquatic
31 systems does not necessarily indicate that they are produced in situ, and rather support
32 the earlier finding that IPLs containing ether lipids are relatively resistant against

1 degradation, even on a longer time scale than has previously been shown (t=97 days;
2 Logemann et al., 2011). Alternatively, the water-saturated conditions may have
3 inhibited both growth and degradation of brGDGTs to take place. However, without
4 analyzing the exact headgroup composition of the brGDGTs in our experiment it is
5 not possible to distinguish between these processes.

6 For this study, it was assumed that the soil-derived (IPL-)GDGTs would be degraded
7 in aquatic environments, yet our data indicate that the concentration of isoGDGTs
8 actually increased over time in all setups (Fig. 4). This suggests that isoGDGTs are
9 either produced, or that they are (preferentially) released from the soil matrix during
10 the experiment. In order to get a better control on the exact processes taking place, the
11 control setup could have benefitted from the addition of a chemical agent to prevent
12 any microbial activity, and thus to avoid additional GDGT-production during the
13 experiment.

14 To summarize, follow-up experiments could benefit from a more natural soil:water
15 ratio, so that a potential contribution of aquatic GDGTs can be detected by analyzing
16 the soils. The control setup should be treated to prevent any microbial activity. In
17 addition, determining the abundance and distribution of GDGTs present in the
18 incubation water could likely provide further information on water column processes,
19 provided that the amount of water added to the soil is sufficient to obtain a large
20 enough sample upon filtering for GDGT analysis. Finally, to better understand the
21 apparent stability of IPL-brGDGTs in aquatic environment, the type of headgroup
22 should be monitored during the incubation experiment, which should run longer than
23 the 152 days of this study.

24

25 **4 Conclusions**

26 Laboratory incubation experiments involving admixture of soil and fresh/ocean water
27 indicate that soil-derived brGDGTs appear to be surprisingly stable in aquatic
28 environments in both IPL- and CL configurations. Our observations suggest that soil
29 brGDGT signatures will likely be unaltered during fluvial transport from land to sea,
30 although any influence of aquatic degradation and/or production processes may be
31 obscured due to the higher-than-natural soil:water ratio used in our experimental
32 setup. Indeed, distributional offsets between brGDGTs in river SPM and catchment

1 soil, as well as between brGDGTs in river SPM and ocean SPM indicate that
2 additional sources may contribute to the final distribution in which brGDGTs are
3 delivered to the ocean. This means that the exact sources of brGDGTs in a river
4 system need to be well constrained before these compounds can be used as reliable
5 tracers for land-sea transport of soil organic carbon, as well as for paleoclimate
6 reconstructions.

7 As opposed to brGDGTs, the concentration of isoGDGTs increased during all
8 incubation experiments, altering their initial relative distribution. Although the
9 changes in TEX₈₆ index values correspond with a maximum change of about 4°C in
10 reconstructed SSTs in the ocean water setup, the low abundance of isoGDGTs in soils
11 relative to in marine settings suggests that their land-sea transport would have
12 minimal impact on TEX₈₆ recorded in the marine sedimentary record. Nevertheless,
13 after 30 days of incubation, the increase in isoGDGTs was sufficient to affect BIT
14 index values. **The observed increase in isoGDGT concentration in the control setup
15 using distilled water indicates that the availability of water is more important for the
16 behavior of isoGDGTs than the properties of the aquatic system in which the soil is
17 introduced.** Our findings may have consequences for the interpretation of TEX₈₆-
18 based SST records for coastal marine settings with a BIT index close to the cutoff
19 value of 0.3 (Weijers et al., 2006b), as the actual input of soil-derived GDGTs upon
20 delivery to the ocean may be larger than anticipated.

21

22 **Acknowledgements**

23 Three anonymous reviewers are thanked for their comments that have improved this
24 manuscript. This study received funding from ETH Fellowship (FEL-36 11-1) and
25 ESF-MOLTER Exchange Grant (nr. 3695) awarded to FP. We thank John Williams
26 (University of Otago) for sampling the incubation experiment, Daniel Montluçon and
27 Negar Haghypour (ETH) for laboratory support, and Lukas Jonkers (Universidad
28 Autònoma de Barcelona/Cardiff University) for help in the field.

29

30 **References**

31 Blair, N.E., Leithold, E.L., Brackley, H., Trustrum, N., Page, M., Childress, L., 2010.
32 Terrestrial sources and export of particulate organic carbon in the Waipaoa
33 sedimentary system: Problems, progress and processes. *Marine Geology* 270, 108-

1 118.

2 Close, H.G., Wakeham, S.G., Pearson, A., 2014. Lipid and ¹³C signatures of
3 submicron and suspended particulate organic matter in the Eastern Tropical North
4 Pacific: Implications for the contribution of Bacteria. *Deep-Sea Research I* 85, 15-
5 34.

6 Cole, J.J., Prairie, Y.T., Caraco, N.F., McDowell, W.H., Tranvik, L.J., Striegl, R.G.,
7 Duarte, C.M., Kortelainen, P., Downing, J.A., Middelburg, J.J., Melack, J., 2007.
8 Plumbing the global carbon cycle: Integrating inland waters into the terrestrial
9 carbon budget. *Ecosystems* 10, 171-184.

10 De Jonge, C., Stadnitskaia, A., Hopmans, E.C., Cherkashov, G., Fedotov, A.,
11 Sinninghe Damsté, J.S., 2014. In-situ produced branched glycerol dialkyl glycerol
12 tetraethers in suspended particulate matter from the Yenisei River, Eastern Siberia.
13 *Geochimica Et Cosmochimica Acta* 125, 476-491.

14 De la Torre, J.R., Walker, C.B., Ingalls, A.E., Könneke, M., Stahl, D.A., 2008.
15 Cultivation of a thermophilic ammonia oxidizing archaeon synthesizing
16 crenarchaeol. *Environmental Microbiology* 10, 810-818.

17 Fietz, S., Martínez-García, A., Huguet, C., Rueda, G., Rosell-Melé, A., 2011.
18 Constraints in the application of the Branched and Isoprenoid Tetraether index as
19 a terrestrial input proxy. *Journal of Geophysical Research* 116, C10032.

20 Goñi, M.A., Ruttenger, K.C., Eglinton, T.I., 1998. A reassessment of the sources and
21 importance of land-derived organic matter in surface sediments from the Gulf of
22 Mexico. *Geochimica Et Cosmochimica Acta* 62, 3055-3075.

23 Harvey, H.R., Fallon, R.D., Patton, J.S., 1986. The effect of organic matter and
24 oxygen on the degradation of bacterial membrane lipids in marine sediments.
25 *Geochimica Et Cosmochimica Acta* 50, 795-804.

26 Hicks, D.M., Shankar, U., Mckerchar, A.I., Basher, L., Lynn, I., Page, M., Jessen, M.,
27 2011. Suspended sediment yields from New Zealand rivers. *Journal of Hydrology*
28 *New Zealand* 50, 81-142.

29 Hopmans, E.C., Weijers, J.W.H., Schefuss, E., Herfort, L., Sinninghe Damsté, J.S.,
30 Schouten, S., 2004. A novel proxy for terrestrial organic matter in sediments based
31 on branched and isoprenoid tetraether lipids. *Earth and Planetary Science Letters*
32 224, 107-116.

33 Huguet, A., Fosse, C., Laggoun-Défarge, F., Delarue, F., Derenne, S., 2013. Effects of
34 a short-term experimental microclimate warming on the abundance and
35 distribution of branched GDGTs in a French peatland. *Geochimica Et*
36 *Cosmochimica Acta* 105, 294-315.

37 Huguet, C., Hopmans, E.C., Febo-Ayala, W., Thompson, D.H., Sinninghe Damsté,
38 J.S., Schouten, S., 2006. An improved method to determine the absolute
39 abundance of glycerol dibiphytanyl glycerol tetraether lipids. *Organic*
40 *Geochemistry* 37, 1036-1041.

41 Ingalls, A.E., Huguet, C., Truxal, L.T., 2012. Lipids of archaeal glycerol dialkyl
42 glycerol tetraethers among size-fractionated particulate organic matter in Hood
43 Canal, Puget Sound. *Applied and Environmental Microbiology* 78, 1480-1490.

44 Kim, J.-H., Schouten, S., Buscail, R., Ludwig, W., Bonnin, J., Sinninghe Damsté,
45 J.S., Bourrin, F., 2006. Origin and distribution of terrestrial organic matter in the
46 NW Mediterranean (Gulf of Lions): Exploring the newly developed BIT index.
47 *Geochemistry Geophysics Geosystems* 7, Q11017, doi:10.1029/2006GC001306.

48 Kim, J.-H., van der Meer, J., Schouten, S., helmke, P., Willmott, V., Sangiorgi, F.,
49 Koç, N., Hopmans, E.C., Sinninghe Damsté, J.S., 2010. New indices and
50 calibrations derived from the distribution of crenarchaeol isoprenoid tetraether

- 1 lipids: Implications for past sea surface temperature reconstructions. *Geochimica*
2 *Et Cosmochimica Acta* 74, 4639-4654.
- 3 Kim, J.-H., Zell, C., Moreira-Turcq, P., Pérez, M.A.P., Abril, G., Mortillaro, J.-M.,
4 Weijers, J.W.H., Meziane, T., Sinninghe Damsté, J.S., 2012. Tracing soil organic
5 carbon in the lower Amazon River and its tributaries using GDGT distributions
6 and bulk organic matter properties. *Geochimica Et Cosmochimica Acta* 90, 163-
7 180.
- 8 Leininger, S., Urich, T., Schloter, M., Schwark, L., Qi, J., Nicol, G.W., Prosser, J.I.,
9 Schuster, S.C., Schleper, C., 2006. Archaea predominate among ammonia-
10 oxidizing prokaryotes in soils. *Nature* 442, 806-809.
- 11 Lengger, S.K., Hopmans, E.C., Reichart, G.-J., Nierop, K.G.J., Sinninghe Damsté,
12 J.S., Schouten, S., 2012. Distribution of core and intact polar glycerol dibiphytanyl
13 glycerol tetraether lipids in the Arabian Sea Oxygen Minimum Zone. II: evidence
14 for selective preservation and degradation in sediments and consequences for the
15 TEX₈₆. *Geochimica et Cosmochimica Acta* 98, 244-258.
- 16 Lincoln, S.A., Wai, B., Eppley, J.M., Church, M.J., Summons, R.E., DeLong, E.F.,
17 2014. Planktonic Euryarchaeota are a significant source of archaeal tetraether
18 lipids in the ocean. *Proceedings of the National Academy of Sciences USA* 111,
19 9858-9863.
- 20 Liu, X.-L., Leider, A., Gillespie, A., Gröger, J., Versteegh, G.J.M., Hinrichs, K.-U.,
21 2010. Identification of polar lipid precursors of the ubiquitous branched GDGT
22 orphan lipids in a peat bog in Northern Germany. *Organic Geochemistry* 41, 653-
23 660.
- 24 Logemann, J., Graue, J., Köster, J., Engelen, B., Rullkötter, J., Cypionka, H., 2011. A
25 laboratory experiment of intact polar lipid degradation in sandy sediments.
26 *Biogeosciences* 8, 2547-2560.
- 27 Loomis, S.E., Russell, J.M., Eggermont, H., Verschuren, D., Sinninghe Damsté, J.S.,
28 2014. Effects of temperature, pH, and nutrient concentration on branched GDGT
29 distributions in East African lakes: implications for paleoenvironmental
30 reconstruction. *Organic Geochemistry* 66, 25-37.
- 31 Mehlich, A., 1953. Determination of P, Ca, Mg, K, Na, and NH₄. Pub. No. 1-53, NC
32 Soil Testing Division, Raleigh, NC.
- 33 Naeher, S., Peterse, F., Smittenberg, R.H., Niemann, H., Zigah, P.K., Schubert, C.J.,
34 2014. Sources of glycerol dialkyl glycerol tetraethers (GDGTs) in catchment soils,
35 water column and sediments of Lake Rotsee (Switzerland) – Implications for the
36 application of GDGT-based proxies for lakes. *Organic Geochemistry* 66, 164-173.
- 37 Oppermann, B.I., Michaelis, W., Blumenberg, M., Frerichs, J., Schulz, H.M.,
38 Schippers, A., Beaubien, S.E., Krüger, M., 2010. Soil microbial community
39 changes as a result of long-term exposure to a natural CO₂ vent. *Geochimica Et*
40 *Cosmochimica Acta* 74, 2697-2716.
- 41 Pancost, R.D. and Sinninghe Damsté, J.S., 2003. Carbon isotopic compositions of
42 prokaryotic lipids as tracers of carbon cycling in diverse settings. *Chemical*
43 *Geology* 195, 29-58.
- 44 Peterse, F., Kim, J.-H., Schouten, S., Klitgaard Kristensen, D., Koç, N., Sinninghe
45 Damsté, J.S., 2009. Constraints on the application of the MBT/CBT
46 palaeothermometer at high latitude environments (Svalbard, Norway). *Organic*
47 *Geochemistry* 40, 692-699.
- 48 Peterse, F., Nicol, G.W., Schouten, S., Sinninghe Damsté, J.S., 2010. Influence of soil
49 pH on the abundance and distribution of core and intact polar lipid-derived
50 branched GDGTs in soil. *Organic Geochemistry* 41, 1171-1175.

- 1 Peterse, F., Hopmans, E.C., Schouten, S., Mets, A., Rijpstra, W.I.C., Sinninghe
2 Damsté, J.S., 2011. Identification and distribution of intact polar branched
3 tetraether lipids in peat and soil. *Organic Geochemistry* 42, 1007-1015.
- 4 Peterse, F., van der Meer, J., Schouten, S., Weijers, J.W.H., Fierer, N., Jackson, R.B.,
5 Kim, J.-H., Sinninghe Damsté, J.S., 2012. Revised calibration of the MBT-CBT
6 paleotemperature proxy based on branched tetraether membrane lipids in surface
7 soils. *Geochimica Et Cosmochimica Acta* 96, 215-229.
- 8 Pitcher, A., Hopmans, E.C., Schouten, S., Sinninghe Damsté, J.S., 2009. Separation
9 of core and intact polar archaeal tetraether lipids using silica columns: Insights
10 into living and fossil biomass contributions. *Organic Geochemistry* 40, 12-19.
- 11 Pitcher, A., Rychlik, N., Hopmans, E.C., Spieck, E., Rijpstra, W.I.C., Ossebaar, J.,
12 Schouten, S., Wagner, M., Sinninghe Damsté, J.S., 2010. Crenarchaeol and its
13 regioisomer dominate the membrane lipids of “*Candidatus Nitrososphaera*
14 *gargensis*”, a thermophilic Group I.1b Crenarchaeote. *The ISME Journal* 4, 542–
15 552.
- 16 Pitcher, A., Hopmans, E.C., Mosier, A.C., Francis, C.A., Reese, S.K., Schouten, S.,
17 Sinninghe Damsté, J.S., 2011. Distribution of core and intact polar tetraether lipids
18 in enrichment cultures of Thaumarchaeota from marine sediments. *Applied and*
19 *Environmental Microbiology* 77, 3468–3477.
- 20 Schouten, S., Hopmans, E.C., Pancost, R.D., Sinninghe Damsté, J.S., 2000.
21 Widespread occurrence of structurally diverse tetraether membrane lipids:
22 Evidence for the ubiquitous presence of low-temperature relatives of
23 hyperthermophiles. *Proceedings of the National Academy of Sciences of the*
24 *United States of America* 97, 14421-14426.
- 25 Schouten, S., Hopmans, E.C., Schefuss, E., Sinninghe Damsté, J.S., 2002.
26 Distributional variations in marine crenarchaeotal membrane lipids: a new tool for
27 reconstructing ancient sea water temperatures? *Earth and Planetary Science*
28 *Letters* 204, 265-274.
- 29 Schouten, S., Huguet, C., Hopmans, E.C., Kienhuis, M.V.M., Sinninghe Damsté, J.S.,
30 2007. Analytical methodology for TEX86 paleothermometry by high-performance
31 liquid chromatography/atmospheric pressure chemical ionization-mass
32 spectrometry. *Analytical Chemistry* 79, 2940-2944.
- 33 Schouten, S., Hopmans, e.C., Baas, M., Boumann, H., Standfest, S., Könneke, M.,
34 Stahl, D.A., Sinninghe Damsté, J.S., 2008. Intact membrane lipids of “*Candidatus*
35 *Nitrosopumilus maritimus*”, a cultivated representative of the cosmopolitan
36 mesophilic group I crenarchaeota. *Applied and Environmental Microbiology* 74,
37 2433-2440.
- 38 Schouten, S., Hopmans, E.C., Rosell-Melé, A., Pearson, A., Adam, P., Bauersachs, T.,
39 Bard, E., Bernasconi, S.M., Bianchi, T.S., Brocks, J.J., Carlson, L.T., Castañeda,
40 I.S., Derenne, S., Selver, A.D., Dutta, K., Eglinton, T., Fosse, C., Galy, V., Grice,
41 K., Hinrichs, K.U., Huang, Y., Huguet, A., Huguet, C., Hurley, S., Ingalls, A., Jia,
42 G., Keely, B., Knappy, C., Kondo, M., Krishnan, S., Lincoln, S., Lipp, J.,
43 Mangelsdorf, K., Martínez-García, A., Ménot, G., Mets, A., Mollenhauer, G.,
44 Ohkouchi, N., Ossebaar, J., Pagani, M., Pancost, R.D., Pearson, E.J., Peterse, F.,
45 Reichart, G.-J., Schaeffer, P., Schmitt, G., Schwark, L., Shah, S.R., Smith, R.W.,
46 Smittenberg, R.H., Summons, R.E., Takano, Y., Talbot, H.M., Taylor, K.W.R.,
47 Tarozo, R., Uchida, M., Dongen, B.E. van, Van Mooy, B.A.S., Wang, J., Warren,
48 C., Weijers, J.W.H., Werne, J.P., Woltering, M., Xie, S., Yamamoto, M., Zhang,
49 C.L., Zhang, Y., Zhao, M. & Sinninghe Damsté, J.S., 2013a. An interlaboratory

1 study of TEX₈₆ and BIT analysis of sediments, extracts, and standard mixtures.
2 *Geochemistry Geophysics Geosystems* 14, 5263-5285.

3 Schouten, S., Hopmans, E.C., Sinninghe Damsté, J.S., 2013b. The organic
4 geochemistry of glycerol dialkyl glycerol tetraether lipids: A review. *Organic*
5 *Geochemistry* 54, 19-61.

6 Sinninghe Damsté, J.S., Hopmans, E.C., Pancost, R.D., Schouten, S., Geenevasen,
7 J.A.J., 2000. Newly discovered non-isoprenoid glycerol dialkyl glycerol tetraether
8 lipids in sediments. *Chemical Communications* 17, 1683-1684.

9 Sinninghe Damsté, J.S., Rijpstra, W.I.C., Hopmans, E.C., Prahl, F.G., Wakeham,
10 S.G., Schouten, S., 2002. Distribution of membrane lipids of planktonic
11 Crenarchaeota in the Arabian Sea. *Applied and Environmental Microbiology* 68,
12 2997-3002.

13 Sinninghe Damsté, J.S., Ossebaar, J., Abbas, B., Schouten, S., Verschuren, D., 2009.
14 Fluxes and distribution of tetraether lipids in an equatorial African lake:
15 Constraints on the application of the TEX(86) palaeothermometer and BIT index
16 in lacustrine settings. *Geochimica Et Cosmochimica Acta* 73, 4232-4249.

17 Sinninghe Damsté, J.S., Rijpstra, W.I.C., Hopmans, E.C., Weijers, J.W.H., Foesel,
18 B.U., Overmann, J., Dedysh, S.N., 2011. 13,16-Dimethyl Octacosanedioic Acid
19 (iso-Diabolic Acid), a Common Membrane-Spanning Lipid of Acidobacteria
20 Subdivisions 1 and 3. *Applied and Environmental Microbiology* 77, 4147-4154.

21 Sinninghe Damsté, J.S., Rijpstra, W.I.C., Hopmans, E.C., Jung, M.Y., Kim, J.G.,
22 Rhee, S.K., Stieglmeier, M., Schleper, C., 2012. Intact polar and core glycerol
23 dibiphytanyl glycerol tetraether lipids of group I.1a and I.1b Thaumarchaeota in
24 soil. *Applied and Environmental Microbiology* 78, 6866-6874.

25 Smith, R.W., Bianchi, T.S., Li, X., 2012. A re-evaluation of the use of branched
26 GDGTs as terrestrial biomarkers: Implications for the BIT index. *Geochimica et*
27 *Cosmochimica Acta* 80, 14-29.

28 Sturt, H.F., Summons, R.E., Smith, K., Elvert, M., Hinrichs, K.-U., 2004. Intact polar
29 membrane lipids in prokaryotes and sediments deciphered by high-performance
30 liquid chromatography/electrospray ionization multistage mass spectrometry –
31 new biomarkers for biogeochemistry and microbial ecology. *Rapid*
32 *Communications in Mass Spectrometry* 18, 617-628.

33 Tierney, J.E., Russell, J.M., 2009. Distributions of branched GDGTs in a tropical lake
34 system: Implications for lacustrine application of the MBT/CBT paleoproxy.
35 *Organic Geochemistry* 40, 1032-1036.

36 Tierney, J.E., Schouten, S., Pitcher, A., Hopmans, E.C., Sinninghe Damsté, J.S.,
37 2012. Core and intact polar glycerol dialkyl glycerol tetraethers (GDGTs) in Sand
38 Pond, Warwick, Rhode Island (USA): Insights into the origin of lacustrine
39 GDGTs. *Geochimica et Cosmochimica Acta* 77, 561-581.

40 Weijers, J.W.H., Schouten, S., Hopmans, E.C., Geenevasen, J.A.J., David, O.R.P.,
41 Coleman, J.M., Pancost, R.D., Sinninghe Damsté, J.S., 2006a. Membrane lipids of
42 mesophilic anaerobic bacteria thriving in peats have typical archaeal traits.
43 *Environmental Microbiology* 8, 648-657.

44 Weijers, J.W.H., Schouten, S., Spaargaren, O.C., Sinninghe Damsté, J.S., 2006b.
45 Occurrence and distribution of tetraether membrane lipids in soils: Implications
46 for the use of the TEX(86) proxy and the BIT index. *Organic Geochemistry* 37,
47 1680-1693.

48 Weijers, J.W.H., Schefuss, E., Schouten, S., Sinninghe Damsté, J.S., 2007a. Coupled
49 thermal and hydrological evolution of tropical Africa over the last deglaciation.
50 *Science* 315, 1701-1704.

- 1 Weijers, J.W.H., Schouten, S., van den Donker, J.C., Hopmans, E.C., Sinninghe
2 Damsté, J.S., 2007b. Environmental controls on bacterial tetraether membrane
3 lipid distribution in soils. *Geochimica Et Cosmochimica Acta* 71, 703-713.
- 4 Weijers, J.W.H., Panoto, E., van Bleijswijk, J., Schouten, S., Rijpstra, W.I.C., Balk,
5 M., Stams, A.J.M., Sinninghe Damsté, J.S., 2009. Constraints on the biological
6 source(s) of the orphan branched tetraether membrane lipids. *Geomicrobiology*
7 *Journal* 26, 402-414.
- 8 Weijers, J.W.H., Wiesenberg, G.L.B., Bol, R., Hopmans, E.C., Pancost, R.D., 2010.
9 Carbon isotopic composition of branched tetraether membrane lipids in soils
10 suggest a rapid turnover and a heterotrophic lifestyle of their source organism(s).
11 *Biogeosciences* 7, 2959-2973.
- 12 Weijers, J.W.H., Bernhardt, B., Peterse, F., Werne, J.P., Dungait, J.A.J., Schouten, S.,
13 Sinninghe Damsté, J.S., 2011. Absence of seasonal patterns in MBT-CBT indices
14 in mid-latitude soils. *Geochimica Et Cosmochimica Acta* 75, 3179-3190.
- 15 Weyhenmeyer, G.A., Fröberg, M., Karlton, E., Khalili, M., Kothawala, D., Temnerud
16 J., Tranvik, L.J., 2012. Selective decay of terrestrial organic carbon during
17 transport from land to sea. *Global Change biology* 18, 349-355.
- 18 White, D.C., Davis, W.M., Nickels, J.S., King, J.D., Bobbie, R.J., 1979.
19 Determination of the sedimentary microbial biomass by extractible lipid
20 phosphate. *Oecologia* 40, 51-62.
- 21 Wuchter, C., Abbas, B., Coolen, M.J.L., Herfort, L., Timmers, P., Strous, M., van
22 Bleijswijk, J., Teira, E., Herndl, G.J., Middelburg, J.J., Schouten S., Sinninghe
23 Damsté, J.S., 2006. Archaeal nitrification in the ocean. *Proceedings of the*
24 *National Academy of Sciences USA* 103, 12317-12322.
- 25 Xie, S., Lipp, J.S., Wegener, G., Ferdelman, T.G., Hinrichs, K.-U., 2013. Turnover of
26 microbial lipids in the deep biosphere and growth of benthic archaeal populations.
27 *Proceedings of the National Academy of Sciences USA* 110, 6010-1014.
- 28 Yang, G., Zhang, C.L., Xie, S., Chen, Z., Gao, M., Ge, Z., Yang, Z. 2013. Microbial
29 glycerol dialkyl glycerol tetraethers from river water and soil near the Three
30 Gorges Dam on the Yangtze River. *Organic Geochemistry* 56, 40-50.
- 31 Zell, C., Kim, J.-H., Abril, G., Lima Sobrinho, R., Dorhout, D., Moreira-Turcq, P.,
32 Sinninghe Damsté, J.S., 2013. Impact of seasonal hydrological variation on the
33 distributions of tetraether lipids along the Amazon River in the central Amazon
34 basin: Implications for the MBT/CBT paleothermometer and the BIT index.
35 *Frontiers in Terrestrial Microbiology* 4, doi:10.3389/fmicb.2013.00228.
- 36 Zell, C., Kim, J.-H., Moreira-Turcq, P., Abril, G., Hopmans, E.C., Bonnet, M.-P.,
37 Lima Sobrinho, R., Sinninghe Damsté, J.S., 2013b. Disentangling the origins of
38 branched tetraether lipids and crenarchaeol in the lower Amazon River:
39 Implications for GDGT-based proxies. *Limnology and Oceanography* 58, 343-
40 353.
- 41 Zell, C., Kim, J.-H., Hollander, D., Lorenzoni, L., Baker, P., Guizan Silva, C.,
42 Nittrouer, C., 2014. Sources and distributions of branched and isoprenoid
43 tetraether lipids on the Amazon shelf and fan: Implications for the use of GDGT-
44 based proxies in marine sediments. *Geochimica et Gosmochimica Acta* 139, 293-
45 312.
- 46 Zhang, C.L., Wang, J., Wei, Y., Zhu, C., Huang, L., Dong, H., 2012. Production of
47 branched tetraether lipids in the lower Pearl River and estuary: Effects of
48 extraction methods and impact on bGDGT proxies. *Frontiers in Terrestrial*
49 *Microbiology* 2, doi:10.3389/fmicb.2011.00274.

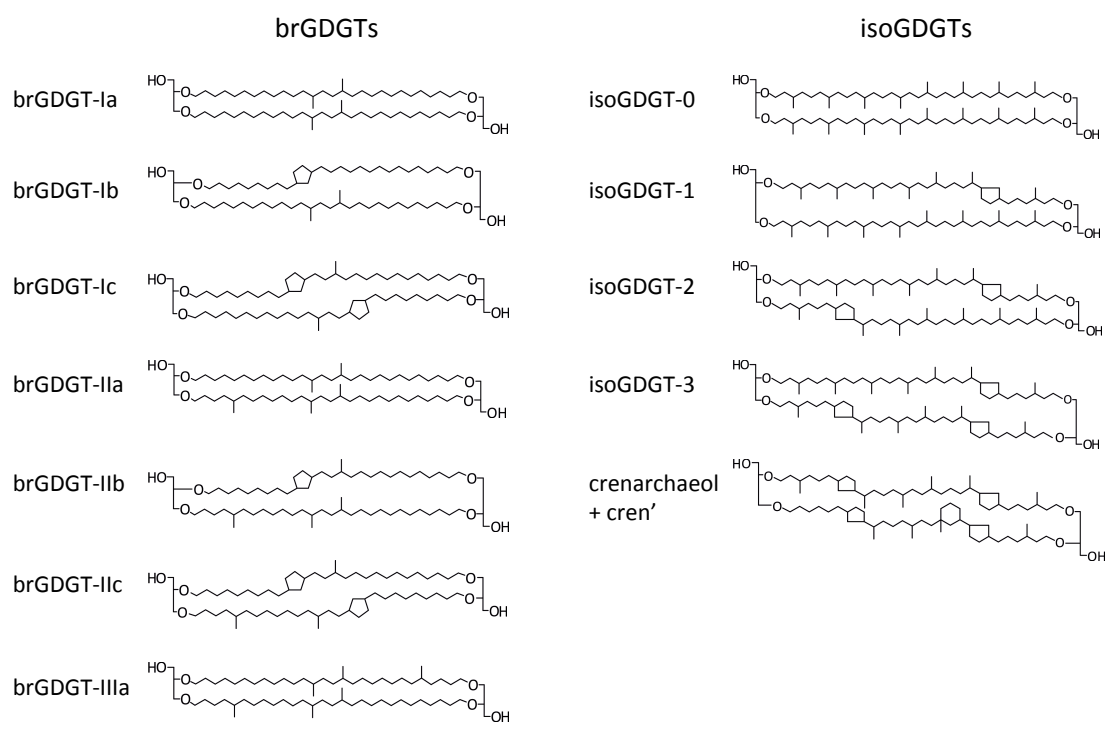
1 Zhang, L.-M., Offre, P.R., he, J.-Z., Verhamme, D.T., Nicol, G.W., Prosser, J.I.,
2 2010. Autotrophic ammonia oxidation by soil Thaumarchaeota. Proceedings of the
3 National Academy of Sciences of the United States of America 107, 17240-17245.
4 Zhu, C., Weijers, J.W.H., Wagner, T., Pan, J.-M., Chen, J.-F., Pancost, R.D., 2011.
5 Sources and distributions of tetraether lipids in surface sediments across a large
6 river-dominated continental margin. Organic Geochemistry 42, 376-386.
7

1

2 Figure 1. Molecular structures of the brGDGTs and isoGDGTs monitored during the
3 incubation experiment.

4

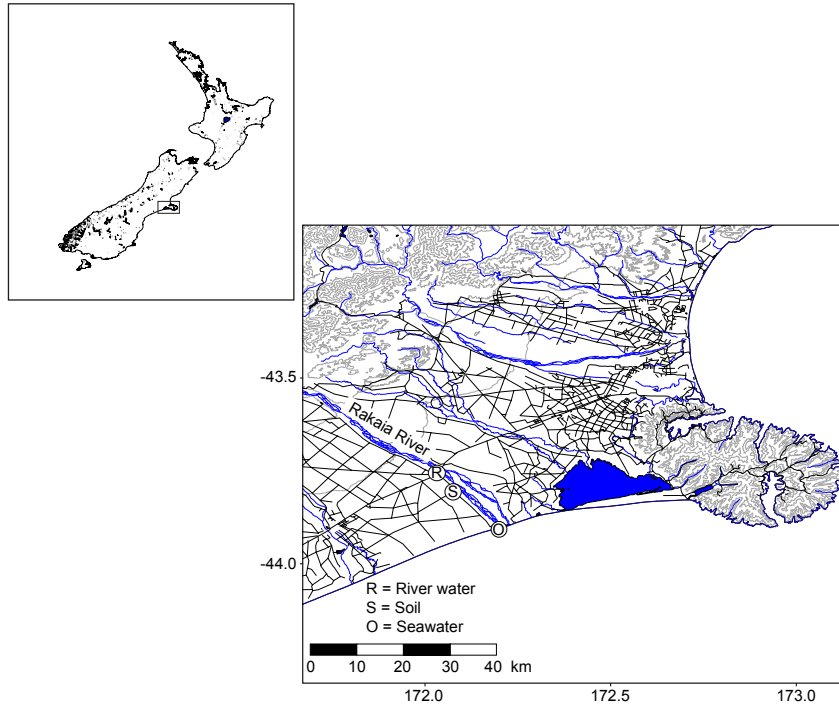
5



6

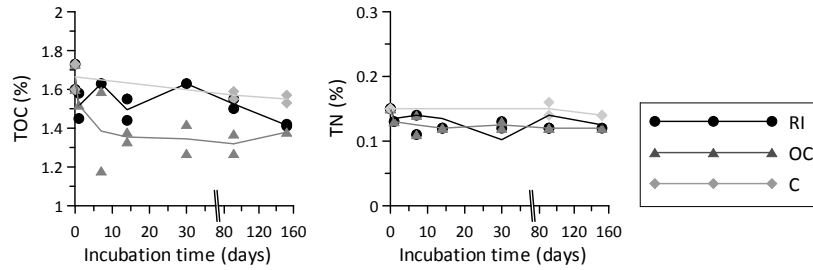
7

- 1 Figure 2. Overview map of the Rakaia River, South Island, New Zealand, indicating
- 2 the sampling locations.
- 3
- 4



5

1 Figure 3. Total organic carbon (TOC) and total nitrogen (TN) concentrations in a
2 sandy loam soil from the Raiaka River catchment, New Zealand, during incubation (t
3 = 152 days) in river (RI), ocean (OC), and distilled (C) water.
4
5

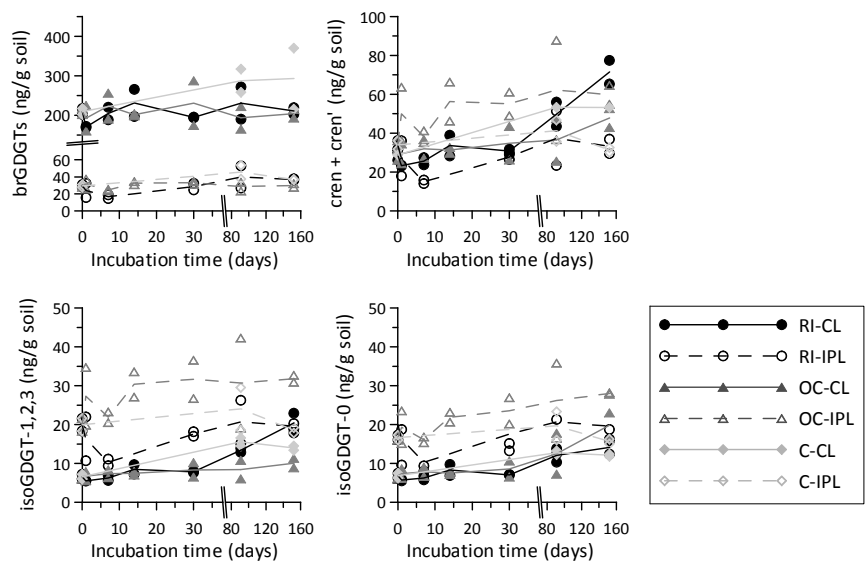


6

1

2 Figure 4. Concentrations of core lipid (CL) and intact polar lipid (IPL)-derived a)
3 brGDGTs, b) crenarchaeol + cren', c) isoGDGT-1,2,3, and d) isoGDGT-0 in a sandy
4 loam soil from the Raiaka River catchment, New Zealand, during incubation (t = 152
5 days) in river (RI), ocean (OC), and distilled (C) water. Average concentrations are
6 indicated by solid (CLs) and dashed (IPL-derived) lines.

7

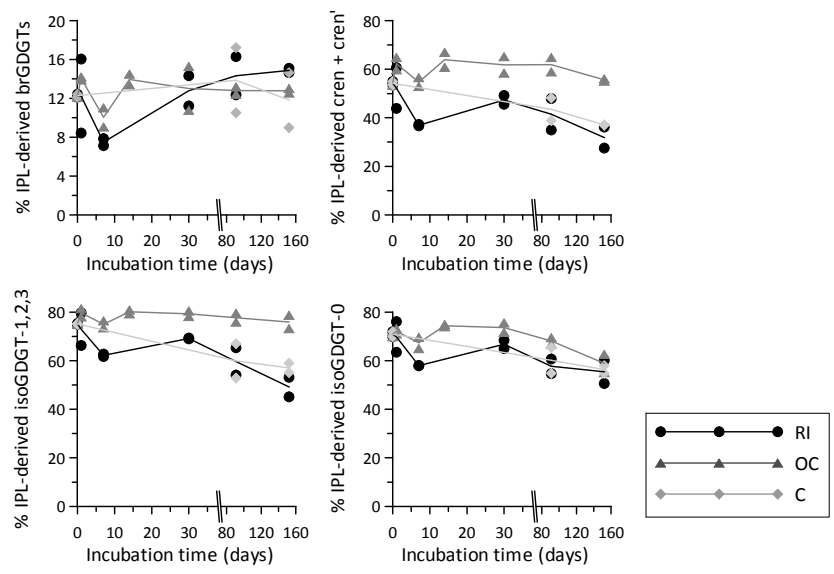


8

1

2 Figure 5. Percentages of the total pool of a) brGDGTs, b) crenarchaeol + cren', c)
3 isoGDGTs-1,2,3, and d) isoGDGT-0 present in 'living', or intact polar lipid (IPL)-
4 derived form. Average concentrations are indicated by solid lines.

5



6

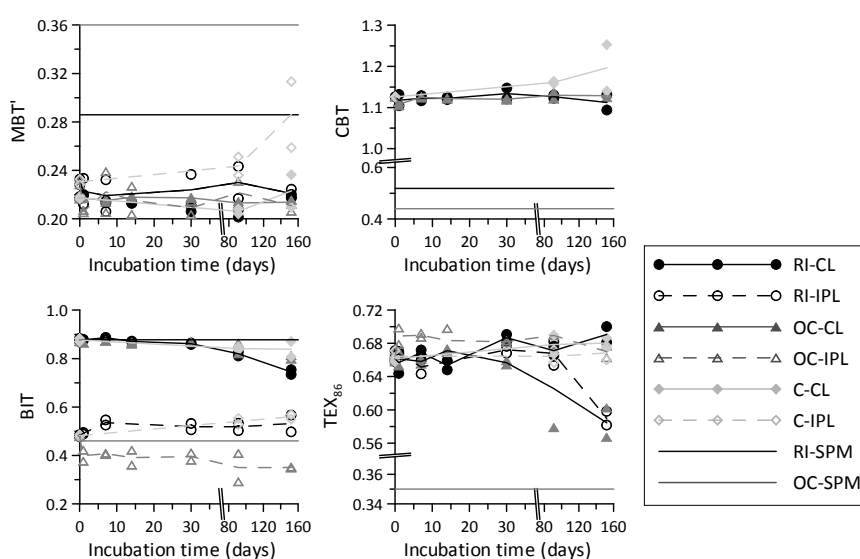
7

8

1

2 Figure 6. Distribution of core lipid (CL) and intact polar lipid (IPL)-derived GDGTs
3 in a sandy loam soil from the Rakaia River catchment, New Zealand, as reflected by
4 the a) MBT' index, b) CBT index, c) BIT index, and d) TEX₈₆ index during
5 incubation (t = 152 days) in river (RI), ocean (OC), and distilled (C) water under
6 quasi-aerobic conditions. Horizontal straight lines represent GDGT composition in
7 river (black) and ocean water (grey) suspended particulate material (SPM). Average
8 concentrations are indicated by solid (CLs) and dashed (IPL-derived) lines.

9



10