

Title: Nitrous oxide emission and nitrogen use efficiency in response to nitrophosphate, N-(n-butyl) thiophosphoric triamide and dicyandiamide of a wheat cultivated soil under sub-humid monsoon conditions

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1 **Nitrous oxide emission and nitrogen use efficiency in response to nitrophosphate,**
2 **N-(n-butyl) thiophosphoric triamide and dicyandiamide of a wheat cultivated**
3 **soil under sub-humid monsoon conditions**

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11

12 **Abstract**

13 A field experiment was designed to study the effects of nitrogen (N) source and
14 urease inhibitor N-(n-butyl) thiophosphoric triamide (NBPT) or nitrification inhibitor
15 dicyandiamide (DCD) on nitrous oxide (N₂O) emission and N use efficiency (NUE)
16 in a sandy loam soil. Six treatments including no N fertilizer (control), N fertilizer
17 urea alone (U), urea plus NBPT (NBPT), (4) urea plus DCD (DCD), urea plus NBPT
18 and DCD (NBPT + DCD), and nitrate-based fertilizer nitrophosphate (NP) were
19 designed and implemented separately during the wheat growth period. Seasonal
20 cumulative N₂O emissions with urea alone amounted to 0.49 ± 0.12 kg N₂O-N ha⁻¹
21 and were significantly ($P < 0.05$) reduced to 0.28 ± 0.03 , 0.31 ± 0.01 and 0.26 ± 0.01
22 kg N₂O-N ha⁻¹ by application of DCD, NBPT and NBPT + DCD, respectively.

23 Cumulative N₂O emissions from NP were 0.28 ± 0.01 kg N₂O-N ha⁻¹. A single N₂O
24 flux peak was identified following basal fertilization, and DCD and/or NBPT
25 inhibition effects mainly occurred during the peak emission period. The NP
26 application significantly ($P < 0.05$) increased wheat yield by 12.3 % and NUE from
27 28.8 % (urea alone) to 35.9 %, while urease and/or nitrification inhibitors showed a
28 slight increase effect. Our results clearly indicated that the application of urea as basal
29 fertilizer, but not as supplemental fertilizer, together with DCD and NBPT is an
30 effective practice to reduce N₂O emissions. The application of NP instead of urea
31 would be an optimum agricultural strategy for reducing N₂O emissions and increasing
32 crop yield and NUE for wheat cultivation in soils of the North China Plain.

33 **Key words:** Ammonium-based fertilizer; Inhibitors; Nitrate-based fertilizer; Nitrogen
34 use efficiency; Nitrous oxide; Wheat yield

35

36 **1 Introduction**

37 Nitrous oxide (N₂O) is a potent and long-lived atmospheric greenhouse gas, with an
38 annual increasing rate of 0.26 % over the past decades and a contribution of 7 % to
39 the annual increase in radiative forcing (IPCC, 2007). Agricultural soils are identified
40 as the major source of atmospheric N₂O, contributing 4.1 Tg N yr⁻¹ (IPCC, 2013) to
41 the global atmospheric N₂O budget of ~ 14 Tg N yr⁻¹ (Fowler et al., 2009). Field
42 management practices along with soil and climatic factors are recognized as being
43 determinants of N₂O emissions from agricultural soils (Stehfest and Bouwman, 2006;
44 Gagnon et al., 2011). Among management practices, the large inputs of industrially

45 fixed N in agriculture are a major perturbation to terrestrial N cycling and a major
46 contribution to accelerating N₂O emissions (Galloway et al., 2008). During the period
47 1990–2005, agricultural N₂O emissions were globally estimated to have increased by
48 17 % (USEPA, 2006), and are projected to increase by 35–60 % by 2030 due to the
49 continuous increase of global N fertilizer consumption and animal manure production
50 (FAO, 2003).

51 China is a major agricultural producer (West et al., 2014) and the amount of
52 applied N fertilizer has increased from 7.07 to 26.21 Tg N yr⁻¹ over the period from
53 1977 to 2005 (Ju et al., 2009). The North China Plain, primarily containing low
54 organic carbon (C) calcareous soils (6.40 vs. 9.60 g C kg⁻¹ for national upland soils)
55 (Xie et al., 2007), is an intensive agricultural region. It covers ~ 300 000 km² and
56 produces up to one-fourth of the total annual grain yield in China (Liu et al., 2001). A
57 winter wheat and summer maize rotation is a commonly used cropping system, and
58 the annual application rates of synthetic N fertilizers have amounted to 600 kg N ha⁻¹
59 or more (Zhao et al., 2006; Ju et al., 2009). However, a low proportion of fertilizer N
60 is taken up by crops (< 30 %) and it is estimated that up to 41 % of N applied during
61 the growth season is subject to losses via leaching, nitrification, denitrification and
62 ammonia volatilization (Cai et al., 2002). At present, up to 33 % of fertilizer N was
63 over-applied to the fields in China, resulting in China's contribution to 28 % of the
64 global annual N₂O emissions from croplands (West et al., 2014). Hence, it is urgent to
65 develop optimum methods for enhancing the recycling of N in the agricultural
66 ecosystem and reducing the fertilizer N-induced N₂O emissions.

67 In the past decade, a number of field measurements of N₂O emissions have been
68 conducted in the North China Plain (Meng et al., 2005; Ding et al., 2007; Ju et al.,
69 2011). A 3 year field measurement showed that the direct N₂O emission factors of
70 synthetic N applied to the wheat-maize cropping systems was 0.82% (Cai et al., 2013),
71 which was higher than the 0.6% reported from fertilizer N-treated upland soils in
72 China (Xing, 1998). Nitrification is found to be the main process for the N₂O
73 emission because low availability of easily degradable organic C limits denitrification
74 in this region (Ding et al., 2007; Ju et al., 2011). Nitrification inhibitors such as DCD
75 help to retard the oxidation of NH₄⁺ to NO₃⁻ by inhibiting the activities of
76 *Nitrosomonas* bacteria in soil (Prasad and Power, 1995), resulting in the reduction of
77 N₂O emissions directly by decreasing nitrification or indirectly by reducing the
78 availability of NO₃⁻ for denitrification and leaching. As a consequence, DCD can
79 increase NUE by increasing plant growth and N uptake (Asing et al., 2008). Similarly,
80 a urease inhibitor like NBPT can slow the conversion of urea to NH₄⁺, thereby
81 reducing N losses by NH₃ volatilization (Manunza et al., 1999; Zaman et al., 2009)
82 and potentially reducing nitrification and subsequent denitrification rates. As such the
83 use of NBPT with urea-based fertilizers may be a potential management strategy to
84 mitigate N₂O emissions (Menéndez et al., 2009). A combined application of
85 nitrification inhibitor and urease inhibitor with urea can maintain N as NH₄⁺ for a
86 longer time with more chance of the fertilizer-derived N being taken up by the crops
87 or immobilized by the organic or mineral component of the soil, thereby reducing the
88 gaseous loss (Xu et al., 2002). Though application of inhibitors to reduce N₂O

89 emissions has attracted more attention recently and has already been investigated in
90 many areas (Menéndez et al., 2009; Zaman et al., 2009), their effect on N₂O emissions
91 in the North China Plain has not been fully investigated.

92 Soil N₂O emissions are also influenced by the source of fertilizer N. Gagnon et al.
93 (2011) found that N₂O emissions from urea ammonium nitrate were drastically greater
94 than those from anhydrous ammonia during the maize growth season in a poorly
95 drained clay soil of Canada. In contrast, Venterea et al. (2005, 2010) reported N₂O
96 emissions from soils amended with anhydrous NH₃ to be 2- to 4-fold greater than that
97 from soils receiving urea ammonium nitrate in a silt loam of the United States. Based
98 on the analysis of published data in the literature, Stehfest and Bouwman (2006)
99 concluded that the N₂O emissions from nitrate-based fertilizers were on average lower
100 than those from ammonium-based fertilizers. During the winter wheat growth season
101 in the North China Plain, limited precipitation occurs. Therefore, it is likely that
102 applying nitrate-based fertilizer instead of urea will not accelerate the leaching of
103 NO₃⁻ but reduce N₂O emissions and increase NUE.

104 In this study, we hypothesize that application of urease inhibitor and/or
105 nitrification inhibitor with urea will lower N₂O emission and increase wheat yield by
106 suppressing the nitrification rate and increasing NUE in the North China Plain. We
107 also hypothesize that use of a nitrate-based fertilizer nitrophosphate by replacing urea
108 will have similar effects. The objectives of this study were: (1) to evaluate the
109 influence of application of urea with NBPT, DCD and NBPT + DCD on N₂O
110 emissions and (2) to investigate whether the use of nitrophosphate instead of urea

111 reduces N₂O emissions from an intensively cultivated calcareous soil during the wheat
112 growth season.

113

114 **2 Materials and methods**

115 **2.1 Experimental site and soil characteristics**

116 The field experiment was conducted at the Fengqiu State Key Agro-ecological
117 Experimental Station, Chinese Academy of Sciences, Henan Province, China
118 (35°00'N, 114°24'E), a typical region of the North China Plain. The region has a
119 sub-humid temperate continental monsoon climate with dry cold winters and wet hot
120 summers. A winter wheat (*Triticum aestivum* L.) and summer maize (*Zea mays* L.)
121 rotation is selected as an intensively managed double-cropping system. The 30 year
122 mean annual temperature was 13.9 °C, with a range varying from -1.0 °C in January
123 to 27.2 °C in July. The mean annual precipitation is 615 mm, two thirds of which falls
124 between June and September. The soil is derived from alluvial sediments of the
125 Yellow River and is classified as aquic inceptisol (Soil Survey Staff, 1994). The
126 physicochemical properties of the soil are summarized in Table 1.

127

128 **2.2 Treatment and crop management**

129 The field experiment was carried out during the winter wheat growth season and
130 included six fertilization treatments: (1) no N fertilizer (control), (2) N fertilizer urea
131 alone (U), (3) urea plus N-(n-butyl) thiophosphoric triamide (NBPT), (4) urea plus
132 dicyandiamide (DCD), (5) urea plus NBPT and DCD (NBPT + DCD), and (6)

133 nitrate-based fertilizer nitrophosphate (NP). The plots were arranged in a randomized
134 complete block with three replicates and the plot size was 5 m × 5 m. Urea and
135 nitrophosphate (Jinkai chemical, Kaifeng, China), totaling 200 kg N ha⁻¹, were added
136 in two applications: 120 kg N ha⁻¹ as basal fertilizer and 80 kg N ha⁻¹ as supplemental
137 fertilizer. Calcium superphosphate was applied as basal fertilizer at a rate of 125 kg
138 P₂O₅ ha⁻¹ for all treatments. For the NP treatment, calcium superphosphate was added
139 as the basal fertilizer to ensure the same application rate of phosphate between the
140 treatments. The NBPT (Hengshuo Chemical, Wuhan, China) and DCD (Sunnyfield
141 Chemicals, Ningxia, China) were applied at a rate of 0.2 % and 10 % of the applied N
142 (w/w), respectively. Inhibitor(s) and urea were thoroughly mixed. All basal fertilizers
143 were evenly spread onto the soil surface by hand and immediately incorporated into
144 the surface soil (0–20 cm) by plowing before sowing on 15 October 2009. The
145 supplemental urea and inhibitor(s) or nitrophosphate were surface applied by hand,
146 then integrated into the plowed layer with irrigation water (40 mm) on 6 March 2010.
147 The mature wheat was harvested on 10 June 2010.

148

149 **2.3 N₂O flux measurement**

150 In situ soil-surface fluxes of N₂O were measured using the static chamber-gas
151 chromatograph (GC) method. Flux measurements were taken over the period from 16
152 October 2009 to 8 June 2010 (235 days) during the wheat growth season. Immediately
153 after sowing, a PVC chamber base (30 cm × 30 cm × 10 cm) was inserted into the soil
154 about 5 cm deep between wheat rows in the center of each plot. The PVC chamber

155 (30 cm × 30 cm × 15 cm) was tightly fitted to the top of the base by inserting the
156 flange of the chamber into the water trough at the upper end of the chamber base. The
157 chamber was equipped with two ports: a small, silicon-sealed vent for sampling and a
158 second port for measuring chamber temperature. Gas samples were initially taken
159 twice a week and later reduced weekly then fortnightly over the winter. Sampling was
160 done in the morning between 09:00 LT and 12:00 LT in order to minimize diurnal
161 variation in flux patterns. Each time, four samples of the chamber air were manually
162 pulled into 50 mL syringes at 0, 10, 20 and 30 min after closure, injected into 20 mL
163 pre-evacuated vials fitted with butyl rubber stoppers and taken to our laboratory for
164 analysis. The air temperature inside the chamber was simultaneously measured with a
165 mercury thermometer.

166 N₂O concentrations were analyzed on a gas chromatograph (Agilent 7890, Santa
167 Clara, CA, USA) equipped with an electron capture detector. The interfering oxygen
168 contained in the injected gas sample (1.0 mL) was separated by a pre-column (1 m) in
169 combination with an analytical column (3 m). Both columns, packed with Porapak Q
170 (80/100 mesh), were attached directly to the 6-port valve to control the backflush. The
171 temperatures of column oven, injector and detector were 40 °C, 100 °C and 300 °C,
172 respectively. The flow rate of carrier gas (95 % Argon + 5 % CH₄) was 40 mL min⁻¹.
173 The standard N₂O gas was provided by the National Institute for Agro-Environmental
174 Sciences, Japan. The N₂O fluxes were calculated using the following equation:

$$175 \quad F = \rho \times (P/760) \times (V/A) \times (\Delta C/\Delta t) \times [273/(273+T)] \quad (1)$$

176 where F is the N₂O flux (μg N₂O-N m⁻² h⁻¹), ρ is the density of N₂O at 0 °C and 760

177 mm Hg (kg m^{-3}), V is the chamber volume (m^3), A is the area from which N_2O was
178 emitted into the chamber (m^2), $\Delta C/\Delta t$ is the rate of N_2O accumulation in the chamber
179 ($\text{ppbv N}_2\text{O-N h}^{-1}$), T is the chamber air temperature in Celsius, and P is the air
180 pressure of the experimental site (mm Hg). The altitude of the experimental site for
181 this study is very close to sea level, so $P/760 \approx 1$. Few sample sets were discarded
182 when they yielded a linear regression value of R^2 greater than 0.90.

183

184 **2.4 Grain yield and aboveground N uptake**

185 After crops reached physiological maturity (10 June 2011), grain and straw were
186 manually harvested from each plot. Grain and straw were air-dried, then further dried
187 for 3 days at 65°C and weighed to obtain dry matter yields. Subsamples were ground
188 with a ball mill and analyzed for N concentration with an elemental N analyzer
189 (VarioMax, Elementar, Hanau, Germany). Total N content in aboveground biomass
190 was calculated from the sum of N masses harvested in grain and straw from each plot.

191

192 **2.5 Auxiliary variables**

193 Soil temperatures were measured, simultaneously with gas sampling, at vertical
194 depths of 5, 10 and 15 cm with a digital thermometer (Model 2455, Yokogawa, Japan).

195 Soil moisture was measured at 5 cm depth at three different positions in the vicinity of
196 each chamber using time domain reflectometry probes and was expressed as
197 water-filled pore space (WFPS) by the equation:

$$198 \text{WFPS [\%]} = (\text{volumetric water content [\%]} / \text{total soil porosity [\%]}) \times 100 \quad (2)$$

199 where total soil porosity = $1 - (\text{soil bulk density}/2.65)$, with $2.65 \text{ [g cm}^{-3}\text{]}$ being the
200 assumed particle density of the soil. The precipitation and air temperature were
201 monitored at a neighboring meteorological station 100 m away from the experimental
202 field.

203 During the growth season, six soil samples were taken from the 0–20 cm soil
204 layer at different positions in each plot just after flux measurement using a 5 cm
205 diameter stainless steel soil sampler and then all samples from each plot were
206 thoroughly mixed to form a composite. After visible roots and litter materials were
207 removed, soil samples were passed through a 2 mm sieve and then extracted with 2 M
208 KCl (soil/KCl suspension of 30:100 ratio) for 1 h on a rotary shaker. The extracted
209 solutions were filtered and stored in a deep freezer ($-18 \text{ }^\circ\text{C}$) until analysis. The
210 NH_4^+ -N and NO_3^- -N concentrations were measured using a colorimetric method on a
211 Skalar segmented flow analyzer (SAN⁺⁺, the Netherlands).

212

213 **2.6 Data analysis and statistics**

214 Average fluxes and standard errors of the N_2O fluxes were calculated from triplicate
215 plots. Seasonal cumulative N_2O emissions were calculated using the following
216 equation:

$$217 \text{ Cumulative } \text{N}_2\text{O} \text{ emission} = \sum_{i=1}^n (F_i + F_{i+1})/2 \times (t_{i+1} - t_i) \times 24 \quad (3)$$

218 where F is the N_2O flux ($\mu\text{g N}_2\text{O-N m}^{-2} \text{ h}^{-1}$), i is the i th measurement, the term of
219 $(t_{i+1} - t_i)$ is the number of days between two measurements, and n is the total number
220 of the measurements. The N_2O direct emission factor (%) of fertilizer N applied to the

221 soil with background adjustment was calculated as follows:

$$222 \text{ Emission factor} = ((\text{N}_2\text{O}-\text{N}_{\text{fertilizer}} - \text{N}_2\text{O}-\text{N}_{\text{control}})/\text{N}_{\text{fertilizer}}) \times 100 \quad (4)$$

223 where $\text{N}_2\text{O}-\text{N}_{\text{fertilizer}}$ and $\text{N}_2\text{O}-\text{N}_{\text{control}}$ are the cumulative N_2O emissions ($\text{kg N}_2\text{O}-\text{N}$
224 ha^{-1}) in the N-fertilized treatment and the control treatment, respectively, and $\text{N}_{\text{fertilizer}}$
225 is the amount of fertilizer N applied (kg N ha^{-1}). Yield-scaled N_2O emissions were
226 calculated by dividing cumulative N_2O emission by grain yield for each plot. NUE
227 was calculated by dividing differences of the N amount in the aboveground biomass
228 between N-fertilized plots and control plots within the same block by the N
229 application rate (200 kg N ha^{-1}). Soil inorganic N intensities were calculated
230 separately for NH_4^+ (NH_4I), NO_3^- (NO_3I) and the sum of $\text{NO}_3^- + \text{NH}_4^+$ (IONI) as the
231 summation of daily NH_4^+-N , NO_3^--N or $(\text{NO}_3^- + \text{NH}_4^+)-\text{N}$ concentrations in the 0–20
232 cm layer over the same period as for cumulative N_2O emissions using linear
233 interpolation between sampling dates, and presented in units of g d kg^{-1} , the index
234 being what is commonly reported (Zebarth et al., 2008; Engel et al., 2010).

235 All data were statistically analyzed using the SPSS software package for
236 Windows (Version 13.0, SPSS inc, Chicago, IL, USA). The effects of fertilization
237 management on N_2O emissions, emission factor, and grain yields were evaluated
238 using one-way ANOVA, followed by the least significant difference (LSD) test at $P <$
239 0.05 . All dependent variables were evaluated for normality using the
240 Kolmogorov-Smirnov test and were log-transformed to normalize the distributions if
241 necessary prior to statistical analysis. Correlation and nonlinear regression analyses
242 were used to test relationships between N_2O fluxes and other factors.

243

244 **3 Results**

245 **3.1 Wheat yield and nitrogen use efficiency**

246 Grain yield in the urea alone treatment was 4652 kg ha⁻¹ and this was increased by 1.3
247 %, 1.8 % or 1.8 % when NBPT, DCD or both were added with urea fertilizer (Table
248 2). Compared with the urea alone, the increase in the grain yield in the NP treatment
249 was 12.3 % (Table 2). The N process inhibitors, NBPT, DCD or the combination of
250 both, slightly increased the amount of N uptake by wheat plants. However, this
251 increase was not statistically significant ($P > 0.05$). A significant increase in the
252 plant N uptake was observed in the NP treatment compared with urea only treatment
253 ($P < 0.05$). The NUE was calculated at 28.8 % for the urea alone treatment and this
254 was slightly increased to 29.2–31.2 % when urea was applied with NBPT, DCD or the
255 combination of NBPT and DCD. However, the NUE was increased to 35.9% for the
256 NP treatment, which was significantly higher than that for all the urea treatments ($P <$
257 0.05).

258

259 **3.2 Soil temperature and moisture**

260 Temporal variations of air temperature, precipitation, and soil WFPS and soil
261 temperature at 5 cm depth over the experimental period are presented in Fig. 1. The
262 cumulative rainfall over the wheat growth season was 97.6 mm which was lower than
263 the long-term average. Soil moisture levels were highly variable, with WFPS values
264 varying from 10.7 % to 80.4 %. Periods with high soil moisture (> 75 % WFPS) were

265 observed following heavy rainfall or irrigation events. Soil temperature at 5 cm depth
266 was below zero in early January and increased to 23 °C in early June.

267

268 **3.3 N₂O emissions**

269 Variations of the N₂O fluxes over the wheat growth season are illustrated in Fig. 2.
270 N₂O fluxes from all fertilizer-incorporated treatments showed almost the same pattern
271 with peak N₂O fluxes being observed soon after application of basal fertilizers. The
272 fluxes from all N fertilizer treatments were generally low on most of the other
273 sampling dates and these fluxes were not significantly different from those from the
274 control. On several occasions in the winter the fluxes were negative. No significant
275 increases in the N₂O fluxes were found following the supplemental fertilization
276 coupled with irrigation. N₂O fluxes did not increase after heavy rainfall events either.

277 The peak N₂O flux was 120.4 μg N₂O-N m⁻² h⁻¹ in the urea alone treatment.
278 Compared with the urea only, application of NBPT, DCD or the combination of both
279 reduced the peak fluxes by 41.1 %, 75.0 % and 61.2 %, respectively. Application of
280 NP reduced peak fluxes by 69.1 % compared with application of urea alone. Analysis
281 showed that the natural logarithms of the N₂O fluxes were weakly, but not
282 significantly, correlated with soil WFPS in all treatments except the DCD treatment,
283 but significantly ($P < 0.05$) correlated with soil temperature in all treatments except
284 the NBPT + DCD treatment (Table 3).

285 Cumulative N₂O emissions from the different treatments are listed in Table 4.
286 Total N₂O emissions from the control, urea alone, urea + NBPT, urea + DCD, urea +

287 NBPT + DCD and NP treatments were 0.16 ± 0.02 , 0.49 ± 0.12 , 0.31 ± 0.01 , $0.28 \pm$
288 0.01 , 0.26 ± 0.01 and 0.28 ± 0.03 kg N₂O-N ha⁻¹, respectively, over the wheat growth
289 season. The highest total N₂O emission was found from the plot which only received
290 urea. These emissions mainly occurred during the 18 day peak emission period
291 following basal fertilizer application from 16 October to 3 November. Application of
292 NBPT, DCD or the combination of both significantly reduced the seasonal N₂O
293 emissions from urea by 36.7 %, 42.9 % or 46.9 %, respectively ($P < 0.05$). Compared
294 with the emissions from the urea alone treatment, significantly lower N₂O emissions
295 were also observed from the NP treatment ($P < 0.05$) (42.9 % less than those from the
296 urea alone treatment). The direct N₂O emission factor for urea application alone was
297 0.17 %, and the addition of NBPT, DCD or the combination of both reduced the
298 emission factor for urea to 0.05–0.08 %. These reductions were statistically
299 significant ($P < 0.05$). The direct N₂O emission factor for NP was 0.06 %, which was
300 also significantly lower than that for urea application alone ($P < 0.05$).

301 The grain yield-scaled N₂O emission from the NP treatment was significantly
302 lower than that from the urea alone treatment ($P < 0.05$), but not different from those
303 from the NBPT, DCD or NBPT + DCD treatments during the wheat growth season
304 (Table 4).

305

306 **3.4 Soil NH₄⁺ and NO₃⁻ concentrations**

307 Soil NH₄⁺ and NO₃⁻ concentrations drastically increased after application of basal N
308 fertilizers compared with the control. The levels of NO₃⁻ in the NBPT and NBPT +

309 DCD treatments were relatively low for one week after basal fertilizer application
310 compared with those in the other treatments. However, the levels of NO_3^- in the
311 NBPT and NBPT + DCD treatments gradually increased, this was probably due to
312 degradation of NBPT and its subsequent loss of effectiveness. Following application
313 of supplemental fertilizer urea, no apparent increase in soil NO_3^- levels was observed,
314 and NO_3^- concentration kept at a relatively constant level. In contrast, soil NO_3^-
315 concentration following application of supplemental fertilizer NP showed a rapidly
316 decreasing trend. In all urea-added treatments, soil NO_3^- concentration sharply
317 decreased to less than 10 mg N kg^{-1} from 15 April onwards.

318 Soil NH_4^+ concentration increased from 2 to 10 mg N kg^{-1} after application of
319 basal fertilizer; however it sharply decreased soon afterwards. Application of DCD or
320 NBPT + DCD sustained soil NH_4^+ concentrations at higher levels compared with urea
321 application alone. In the NP treatment, soil NH_4^+ concentrations were always at low
322 levels. The natural logarithms of the N_2O fluxes were more correlated with NH_4^+
323 concentrations than with NO_3^- concentrations in the soil, despite the fact that a
324 significant relationship was only observed in the urea alone treatment.

325 Mean soil NH_4I levels in the NP treatment were the lowest among all N-added
326 treatments and tended to be higher in the DCD and NBPT + DCD treatments
327 compared with urea alone (Table 5). Mean soil NO_3I levels showed a similar trend
328 among the treatments and were ranked in the order of NBPT, NBPT + DCD > DCD >
329 urea alone, NP > control. Mean soil IONI levels were also similar among the
330 treatments and were ranked as NBPT + DCD > NBPT, DCD > urea alone, NP >

331 control.

332

333 **4 Discussion**

334 **4.1 Nitrous oxide emissions as affected by nitrogen sources**

335 Compared with the urea alone, application of NP significantly reduced N₂O emissions
336 by 42.9 % during the wheat growth season (Table 4), and increased wheat yield by
337 12.3 % and NUE by 24.7 % (Table 2). Ju et al. (2011) obtained a similar result in the
338 North China Plain, finding that emissions of N₂O derived from Ca(NO₃)₂ were lower
339 than those from NH₄(SO₄)₂ during the maize growth season (0.38–0.81 vs. 1.31–3.52
340 kg N₂O-N ha⁻¹). A lower N₂O emission for urea ammonium nitrate than for
341 anhydrous ammonia was also reported in a silt loam of the United States (Venterea et
342 al., 2005). In contrast, Gagnon et al. (2011) measured a significantly higher N₂O
343 emission following application of urea ammonium nitrate or calcium ammonium
344 nitrate compared with anhydrous ammonia in a poorly drained clay soil of Canada. In
345 a German grassland ecosystem, Müller and Sherlock (2004) found that the emissions
346 for ammonium-based fertilizer were lower than those for nitrate-based fertilizer.
347 These researchers suggested that higher emissions from nitrate-based fertilizers were
348 because of the propensity of the fine-textured clay soil to become anaerobic following
349 rainfall and a strong fixation of NH₄⁺ in clay lattices reducing NH₄⁺ available for N₂O
350 production (Chantigny et al., 2004). In this study, mean soil NO₃⁻ intensities (NO₃I)
351 were not significantly different between the NP and urea alone treatments (Table 5),
352 and mean NO₃⁻ concentrations (Fig. 3) were higher than the suggested threshold value

353 for denitrification of 5 mg N kg⁻¹ (Dobbie and Smith, 2003) during the growth season
354 except for the period from 24 April to 10 June. These results imply that soil NO₃⁻
355 concentration was not the only limiting factor affecting denitrification and N₂O
356 emission in the test soil.

357 The notable difference in the seasonal N₂O emissions between the NP and urea
358 alone treatments occurred mainly during the 18 day peak emission period following
359 the basal fertilizer application and concurrent irrigation from 16 October to 3
360 November. It has been reported that application of ammonium-based fertilizers
361 emitted more N₂O than nitrate-based fertilizers under aerobic soil conditions, while
362 application of nitrate-based fertilizers induced a greater increase in N₂O production
363 when soil conditions were anoxic (Pathak and Nedwell, 2001; Tenuta and Beauchamp,
364 2003). For cultivated soils, the primary mechanism of N₂O production is generally
365 believed to be the nitrification process when soil WFPS levels are between 30 % and
366 70 % and the denitrification process when soil WFPS levels were between 70 % and
367 90 % (Granli and Bøckman, 1994). Some other studies also suggest that
368 denitrification could in general produce more N₂O compared with nitrification (eg.
369 Dobbie et al., 1999). According to the studies of Ding et al. (2007) and Wan et al.
370 (2009), N₂O in sandy loam soils of the North China Plain was primarily produced by
371 nitrification unless soil WFPS reached 75 % or more. Pihlatie et al. (2004) reported
372 that even at 100 % WFPS in a loamy sand soil with 24 g organic C kg⁻¹, nitrification
373 was still the dominant N₂O production process. In this study, the highest soil WFPS
374 measured during the peak emission period was ~65 %; thus we suggest that low soil

375 moisture limited denitrification and N₂O production from the nitrate-based fertilizer in
376 the test soil.

377 In the North China Plain, the addition of starch to soil treated with nitrate-based
378 fertilizers in the field stimulated N₂O production through denitrification, but wheat
379 straw amendment did not do so (Wan et al., 2009; Ju et al., 2011). Previous studies
380 demonstrated that denitrification was not only controlled by soil moisture and nitrate,
381 but also by organic C supply, and increasing organic C availability could reduce the
382 minimum soil moisture threshold for denitrification (van Groenigen et al., 2004;
383 Chantigny et al., 2013). Yu et al. (2012) found that the mass proportion of
384 macroaggregates in a NPK-treated soil with 6.0 g organic C kg⁻¹ only accounted for
385 8.8 %, while this proportion amounted to 30.8 % in an 18 year compost-added soil
386 with 10.0 g organic C kg⁻¹ in the North China Plain. This change significantly
387 increased the proportion of pores with a neck diameter < 4 μm by reducing the
388 proportion of pores with a neck diameter of 15–60 μm, which in turn lowered the
389 effective diffusion coefficient of oxygen in the soils and the ratio of monounsaturated
390 to branched phospholipid fatty acids (PLFAs) i.e. aerobic to anaerobic
391 microorganisms (Zhang et al., 2014a). According to results found by Myrold and
392 Tiedje (1984), only large aggregates have anaerobic microsites. Thus, it is likely that
393 the relatively low organic C concentration in the test soil retards macroaggregation
394 and slows formation of anaerobic microsites, which in turn results in rise of the
395 minimum moisture threshold required for denitrification. Consequently, the
396 denitrification process is of much less importance than nitrification for N₂O

397 production and emissions in soils of the North China Plain.

398 In a German silt loam soil, similar to that tested in this study, Rover et al. (1998)
399 reported that winter was a key period for N₂O emissions from arable crops in the
400 temperate climate zone, contributing ~ 70 % of the annual N₂O losses during the
401 thawing from December to February. Wolf et al. (2010) also verified that N₂O pulses
402 due to spring thaw dominated total annual N₂O emission in a steppe grassland of
403 Inner Mongolia, China. At our site, spring thawing of the soil at the fertilized plots
404 only caused minor N₂O emission pulses, which were considerably lower than those
405 reported earlier for other arable soils (Syväsalo et al., 2004; Teepe et al., 2000). It is
406 suggested that reduced oxygen supply through alteration of pore structure during
407 thawing, and high soil water contents in the winter, would promote microbial
408 denitrification (Edwards and Killham, 1986; Mørkved et al., 2006). Our present study,
409 together with previous measurements (Ding et al., 2007; Ju et al., 2010; Cui et al.,
410 2012; Cai et al., 2013), showed that the highest soil WFPS was no more than 70 %
411 during the spring thawing period, a value that was lower than the threshold value of
412 80 % for thawing N₂O pulses in a silt loam found by Rover et al. (1998). The
413 cumulative rainfall during the winter period from December 2009 to February 2010
414 was only 4.4 mm and no apparent snow cover was observed at our study site. So the
415 warm temperate monsoon zone, with cold and dry winter in the North China Plain,
416 which is distinctly different from other climatic zones such as western Europe
417 (Dobbie and Smith 2003) and Inner Mongolia of China (Wolf et al., 2010), would not
418 induce thawing N₂O pulses from arable soils, as found in our study. However, Zhang

419 et al. (2014b) found that the North China Plain is a large agricultural N₂O source in
420 China, contributing 36.3 % of the total annual N₂O emission from the China's
421 croplands. To make a global comparison, we compiled the literature data of N₂O
422 emissions from the temperate uplands under inorganic N fertilizer application in some
423 countries of Asia, Europe and North America with similar latitudes to the studied
424 region (Table 6). The emission factors of N applied in the North China Plain are
425 generally lower than in the other countries, indicating a lower capacity of N applied
426 being converted into N₂O in the test soil. This is probably because N₂O is
427 predominantly produced from nitrification and denitrification is organic C-limited as
428 discussed above. In contrast, the total N₂O emissions from the studied region are
429 obviously higher due to the greater N fertilizer loading (Ju et al., 2009), suggesting
430 that agricultural practices for reducing N₂O emission are urgently required. Our
431 results confirm that N fertilizer sources influence soil N₂O emissions, but that this
432 effect probably depends on soil properties and especially climate conditions. Our
433 study also suggests that, compared with urea or ammonium-based fertilizer, applying
434 nitrate-based fertilizer is an effective management strategy to mitigate N₂O emissions
435 and to increase NUE and wheat yield in the North China Plain.

436

437 **4.2 Nitrous oxide emissions as affected by inhibitors**

438 The application of basal fertilizer urea followed by flooding irrigation resulted in N₂O
439 emission pulses for 18 days. This finding is in agreement with those of other studies
440 for arable fields (Bouwman et al., 2002; Ding et al., 2007; Cui et al., 2012). The
441 presence of inhibitors NBPT and/or DCD significantly lowered N₂O peak fluxes, and

442 cumulative N₂O emissions during the 18 day peak emission period were reduced by
443 50.0 % by NBPT, 78.6 % by DCD and 67.9 % by NBPT + DCD, compared with
444 application of urea alone. Our results indicate that the addition of DCD alone or in
445 combination with NBTP effectively reduced N₂O emissions from application of urea.
446 In other sites of the North China Plain, Liu et al. (2013) also reported that nitrification
447 inhibitors DCD and DMPP (3,4-dimethylpyrazole phosphate) could reduce N₂O
448 emissions from application of N fertilizers by 30 % and 21 %, respectively, during the
449 wheat growth season. Ju et al. (2011) observed no apparent differences in cumulative
450 N₂O emissions between zero N control and urea with DMPP during the maize growth
451 season, suggesting strong nitrification inhibition effectiveness of DMPP.

452 N₂O emission is directly related to the amount of mineral N available in the soil
453 and application of inhibitors with urea can effectively regulate the NO₃⁻ and NH₄⁺
454 concentrations (Li et al., 2009; Zaman et al., 2009). Recently, Maharjan and Venterea
455 (2013) demonstrated that N₂O emissions were more correlated with soil NO₂⁻
456 intensity rather than NO₃⁻ or NH₄⁺ intensity, and that inhibitors controlled N₂O
457 production by adjusting soil NO₂⁻ intensity. In this study, soil NH₄⁺ concentration
458 slightly increased in the presence of DCD and, in contrast, relatively low NH₄⁺
459 concentration was found after NBPT application following application of basal
460 fertilizer. NBPT delays urea hydrolysis, thereby lowering soil pH elevation and NH₄⁺
461 production, which can in turn reduce NH₃ toxicity effects on nitrite-oxidizing bacteria
462 (NOB). DCD slows oxidation of NH₄⁺ to NO₂⁻ mainly by inhibiting activities of
463 ammonia-oxidizing bacteria (AOB), which allows NOB to use NO₂⁻ at the rate closely

464 matched to its production rate (Zaman et al., 2008; Maharjan and Venterea, 2013).
465 Both NBPT and DCD could additively attenuate formation of N₂O from urea in the
466 soil. Thus, the reduction of N₂O emissions by inhibitors is probably due to both low
467 oxidation rate of NH₄⁺ and low NO₂⁻ concentration, thereby reducing N₂O “leaking”
468 as a by-product of nitrification (Firestone and Davidson, 1989).

469 Following supplemental fertilization with or without inhibitors, no distinct N₂O
470 flux peaks were found in our study. This may be attributable to no significant increase
471 of soil NH₄⁺ and NO₃⁻ concentrations after urea top-dressing. Cui et al. (2012)
472 ascribed low increases in mineral N concentrations to large losses of urea-derived
473 ammonia via volatilization. However, a field measurement at our study site showed
474 that < 1 % of the N applied was lost via volatilization following urea top-dressing in
475 March (Ni et al., 2009); so a large amount of NH₃ loss would not occur at our site.
476 Milchunas et al. (1988) suggested that urea hydrolysis is primarily affected by soil
477 moisture. An incubation at 13 °C demonstrated that lowering soil moisture level from
478 60 % to below 40 % water holding capacity produced a longer lag before ammonia
479 evolution and considerably retarded urea hydrolysis (Foster et al., 1980). The range of
480 soil WFPS between 40 and 60 % during the period following urea top-dressing with
481 subsequent irrigation suggested that soil moisture could partly have affected N₂O
482 production. Suter et al. (2011) observed that lowering incubation temperature from 25
483 °C to 5 °C greatly retarded the hydrolysis of urea when WFPS was below 60%,
484 especially for an alkaline soil with low urease activity. In contrast, the temperature
485 decrease increased the inhibitory effectiveness of NBPT on urea hydrolysis. In this

486 study, soil temperature measured in the field after urea top-dressing varied from 2 °C
487 to 9 °C, close to or just above the thresholds for nitrification (above 5.0 °C)
488 (Anderson et al., 1971) and urea hydrolysis (~ 2 °C) (Xu et al., 1993; Yadav et al.,
489 1987). It is obvious that low soil temperature led to the absence of fertilizer N-induced
490 N₂O flux peaks following the supplemental fertilization and urease or nitrification
491 inhibitors should not necessarily be applied with supplemental fertilizers during the
492 wheat growth season.

493 Application of urea with NBPT and/or DCD compared with urea alone slightly
494 increased wheat yields, which differs from application of NP. Similar results were
495 also obtained by Ju et al. (2011) and Liu et al. (2013) in the North China Plain. A
496 meta-analysis of data measured in Germany showed that N fertilizers with
497 nitrification inhibitors did not significantly influence the yields of all investigated
498 crops (Hu et al., 2014). The absence of inhibitor effects on crop yields might have
499 been ascribed to the following three reasons. Firstly, low precipitation during the
500 wheat growth season reduced the risk of N leaching and resulted in low N losses. This
501 is evidenced by the significant stimulation of NP on wheat yields. Secondly, it is well
502 known that the application rate of N fertilizers is far above optimum for crops (West
503 et al., 2014). The overloading of N fertilizer might mask the influence on crop yields
504 of increased mineral N in soils caused by inhibitors. The result of Sharma and Prasad
505 (1996) supported the hypothesis that application of DCD significantly increased
506 maize yield when the application rate of fertilizer N was as low as 60 kg N ha⁻¹. It
507 should be noted that the increase in NH₄⁺ concentration in the test soil due to DCD

508 application alone following the basal fertilization may stimulate NH_3 volatilization,
509 resulting in higher N losses compared with urea alone, since NH_3 volatilization
510 accounted for ~ 13 % of N applied (Ni et al., 2009). Mahmood et al. (2011)
511 demonstrated that application of DCD to an alkaline calcareous soil increased
512 fertilizer N losses. Finally, also more importantly, application of DCD with
513 supplemental fertilizer slowed the nitrification rate and then lowered NO_3^- supply for
514 wheat growth when it was at the rapid growth stage. A lower soil NO_3^- concentration
515 in the NP treatment than in the urea-added treatments following the supplemental
516 fertilizer in this study supports this speculation. Based on this study, it is not necessary
517 to apply DCD with supplemental fertilizer urea and a combination of urease and
518 nitrification inhibitors would be a better approach to reduce N_2O emission than urease
519 or nitrification inhibitor application alone with basal fertilizer urea for wheat
520 cultivation.

521

522 **5 Conclusions**

523 The present field study provided an insight into N_2O emissions from a calcareous soil
524 during the wheat growth season in the North China Plain, as affected by application of
525 urease or nitrification inhibitors and nitrate-based fertilizer nitrophosphate. A single
526 N_2O flux peak was found following basal fertilization during the wheat growth period.
527 Application of urea with NBPT, DCD or NBPT + DCD significantly reduced N_2O
528 emissions from urea by 36.7 %, 42.9 % or 46.9 %, respectively. Application of
529 nitrophosphate also resulted in reduction of total N_2O emissions by 42.9 %, compared

530 with application of urea alone. NBPT and/or DCD were effective in reducing N₂O
531 emissions following basal fertilization. Compared with urea application alone,
532 application of inhibitors with urea, either individually or combined together, slightly
533 increased wheat yield and NUE, while nitrophosphate significantly increased wheat
534 yield by 12.3 % and increased NUE from 28.8 % (urea alone) to 35.9 %. N₂O flux
535 was primarily affected by soil temperature and low temperature at the study site
536 minimized fertilizer N-induced N₂O peaks following application of supplemental
537 fertilizer. Based on our findings, the combination of NBPT and DCD with basal
538 fertilizer urea would be an effective practice for reducing N₂O emission. As well, this
539 study suggests that application of nitrophosphate, instead of urea, is an optimum
540 agricultural strategy for reducing N₂O emission and for increasing crop yield and
541 NUE for wheat cultivation in the soils of the North China Plain.

542

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547

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785 **Table 1.** Soil properties.

Soil depth (cm)	pH (H ₂ O)	Bulk density (Mg m ⁻³)	Organic C (g C kg ⁻¹)	Total N (g N kg ⁻¹)	C/N	NO ₃ ⁻ -N (mg N kg ⁻¹)	NH ₄ ⁺ -N (mg N kg ⁻¹)	Particle size distribution (%)		
								Sand	Silt	Clay
0-20	8.60	1.40	12.0	1.50	8.0	14.70	2.48	17.0	72.0	11.0

786

787 **Table 2.** Effects of urease and/or nitrification inhibitors and nitrophosphate on wheat biomass, amount of N uptake by crops and N use
 788 efficiency.

Treatment	Biomass (kg ha ⁻¹)			Amount of N uptake (kg N ha ⁻¹)			N use efficiency (%)
	Grain	Straw	Total	Grain	Straw	Total	
Control	2297±150 c	2215±134 b	4513±283 c	24.4±1.6 c	11.9±0.7 b	36.4±2.3 c	–
U	4652±11 b	4075±81 a	8727±85 b	59.6±0.1 b	34.2±0.7 a	93.9±0.7 b	28.8 ±0.8 b
NBPT	4711±126 b	4098±356 a	8809±472 b	60.4±1.6 b	34.4±3.0 a	94.8±4.5 b	29.2 ±1.0 b
DCD	4736±103 ab	4080±52 a	8816±86 b	60.7±1.3 b	34.3±0.4 a	95.0±1.1 b	29.3 ±0.9 b
NBPT + DCD	4735±290 ab	4535±503 a	9271±764 ab	60.7±3.7 b	38.1±4.2 a	98.8±7.6 ab	31.2 ±1.2 b
NP	5225±142 a	4906±251 a	10131±370 a	67.0±1.8 a	41.2±2.1 a	108.2±3.7 a	35.9 ±1.1 a

789 Mean ± standard error ($n = 3$).

790 Different letters within the column indicate significantly difference between treatments at $P < 0.05$.

791 **Table 3.** Correlation between $\ln [\text{N}_2\text{O flux} + 1]$ and soil WFPS, soil temperature at depths of 5 ($T_{5\text{cm}}$), 10 ($T_{10\text{cm}}$) and 15 cm ($T_{15\text{cm}}$), ammonium
 792 ($\text{NH}_4^+\text{-N}$), nitrate ($\text{NO}_3^-\text{-N}$) or inorganic nitrogen ($\text{NH}_4^+\text{-N}$ plus $\text{NO}_3^-\text{-N}$) concentration.

Treatment	WFPS	$T_{5\text{cm}}$	$T_{10\text{cm}}$	$T_{15\text{cm}}$	$\text{NH}_4^+\text{-N}$	$\text{NO}_3^-\text{-N}$	Inorganic N
Control	0.095	0.413**	0.376*	0.392*	-0.153	0.140	0.109
U	0.023	0.381**	0.340*	0.346*	0.274*	0.365**	0.380**
NBPT	0.118	0.275*	0.264	0.274	0.215	0.206	0.222
DCD	0.323*	0.282*	0.180	0.189	0.104	-0.127	-0.092
NBPT + DCD	0.021	0.216	0.252	0.272	0.074	0.155	0.156
NP	0.084	0.301*	0.370*	0.403**	0.105	-0.056	-0.037

793 * $P < 0.05$, ** $P < 0.01$.

794 **Table 4.** Effects of urease and/or nitrification inhibitors and nitrophosphate on cumulative N₂O emissions, fertilizer N-induced N₂O emission
 795 factors and yield-scaled N₂O emissions.

Treatment	Cumulative N ₂ O emission (kg N ₂ O-N ha ⁻¹)		Ratio of peak to total emissions (%)	Emission factor (% of applied N)	Yield-scaled N ₂ O emission (g N ₂ O-N kg ⁻¹ grain)
	Total	Peak			
Control	0.16±0.02 c	0.03±0.00 d	18.8±2.3 d	–	0.068±0.006 b
U	0.49±0.12 a	0.28±0.10 a	57.1±4.2 a	0.17±0.05 a	0.105±0.026 a
NBPT	0.31±0.01 b	0.14±0.01 b	45.2±3.9 b	0.08±0.00 b	0.065±0.003 b
DCD	0.28±0.01 b	0.06±0.01 c	21.4±2.1 d	0.06±0.00 b	0.060±0.004 b
NBPT + DCD	0.26±0.01 b	0.09±0.00 bc	34.6±3.2 c	0.05±0.00 b	0.056±0.003 b
NP	0.28±0.03 b	0.11±0.03 bc	39.3±3.7 c	0.06±0.01 b	0.053±0.008 b

796 Mean ± standard error (*n* = 3).

797 Peak emission denotes cumulative emissions during the 18 days' period following the basal fertilizer application from 16 October to 3
 798 November.

799 Different letters within the column indicate significantly difference between treatments at *P* < 0.05.

800 **Table 5.** Effects of urease and/or nitrification inhibitors and nitrophosphate on soil
 801 ammonium (NH₄I), nitrate (NO₃I) and inorganic N (IONI) intensities.

Treatment	NH ₄ I (g N d kg ⁻¹)	NO ₃ I (g N d kg ⁻¹)	IONI (g N d kg ⁻¹)
Control	0.24±0.01 e	2.58±0.01 d	2.82±0.00 d
U	0.40±0.03 d	4.75±0.13 c	5.15±0.16 c
NBPT	0.61±0.02 c	6.18±0.08 a	6.79±0.08 b
DCD	0.96±0.01 b	5.74±0.01 b	6.70±0.01 b
NBPT + DCD	1.07±0.01 a	6.11±0.16 a	7.17±0.16 a
NP	0.36±0.02 d	4.69±0.09 c	5.05±0.07 c

802 Mean±standard error (*n* = 3). Different letters within the column indicate significantly
 803 difference between treatments at *P* < 0.05.

804 **Table 6.** Summary of N₂O emissions from uplands under inorganic fertilizer application in the countries with temperate climate.

Site	MAT (°C)	MAP (mm)	SOC (g C kg ⁻¹)	pH	Warm season			Cold season			Whole year			Reference		
					Crop	Applied N (kg N ha ⁻¹)	N ₂ O emission (kg N ha ⁻¹)	N ₂ O EF (%)	Crop	Applied N (kg N ha ⁻¹)	N ₂ O emission (kg N ha ⁻¹)	N ₂ O EF (%)	Applied N (kg N ha ⁻¹)		N ₂ O emission (kg N ha ⁻¹)	N ₂ O EF (%)
Fengqiu, China	14	615	7	8.7	Maize	250	3.8	1.3	Wheat	250	0.6	0.3	500	4.5	0.8	Ding et al. (2007)
Huantai, China	13	586	10	8.3	Maize	330	1.6	0.4	Wheat	270	2.4	0.8	600	4.0	0.6	Cui et al. (2012)
Baoding, China	12	555	9	8.1	Maize	173	4.5	2.2	Wheat	165	3.3	1.3	338	7.7	1.8	Zhang et al. (2014c)
Tsukuba, Japan	16	1460	19	5.7	Soybean	20	2.7	13	Wheat	100	0.5	0.5	120	3.2	2.7	Nishimura et al. (2005)
Fukushima, Japan	14	1207	14	7.4	Barley	150	3.2	2.0	–	–	–	–	–	–	–	Shoji et al. (2001)
Madrid, Spain	13	430	8	7.3	Onion	110	0.8	0.6	Fallow	0	0.25	–	110	1.2	0.7	Mejjide et al. (2009)
Lavesum, Germany	10	887	18	5.3	Wheat	220	0.6	0.2	Fallow	0	1.0	–	220	1.9	0.5	Lebender et al. (2014)
Turin, Italy	12	734	10	8.1	Maize	130	0.0	0.0	Fallow	0	2.9	–	130	2.9	3.4	Alluvione et al. (2010)
Boone, USA	9	825	33	7.2	Maize	168	2.9	1.0	–	–	–	–	–	–	–	Parkin and Hatfield (2014)
Michigan, USA	8	628	20	7.0	Maize	225	3.9	1.4	–	–	–	–	–	–	–	Hoben et al. (2011)
Michigan, USA	8	628	20	7.0	Maize	180	2.5	1.2	–	–	–	–	–	–	–	Hoben et al. (2011)
Michigan, USA	8	628	20	7.0	Maize	135	1.7	0.9	–	–	–	–	–	–	–	Hoben et al. (2011)
Michigan, USA	8	628	20	7.0	Maize	90	1.1	0.7	–	–	–	–	–	–	–	Hoben et al. (2011)
Michigan, USA	8	628	20	7.0	Maize	45	0.9	1.1	–	–	–	–	–	–	–	Hoben et al. (2011)
Morris, USA	6	645	32	7.2	Maize	78	–	–	Fallow	0	–	–	78	5.2	3.0	Johnson et al. (2012)
Morris, USA	6	645	32	7.2	Wheat	78	–	–	Fallow	0	–	–	78	4.2	2.8	Johnson et al. (2012)

805 MAT, mean annual temperature; MAP, mean annual precipitation; EF, the N₂O emission factor of applied N.

806 **Figure caption**

807 **Figure 1.** Temporal variation of daily precipitation and air temperature, and mean soil
808 moisture and water-filled pore space (WFPS) at time of N₂O sampling in the control,
809 urea alone (U), urea with NBPT (NBPT), urea with DCD (DCD), urea with NBPT +
810 DCD (NBPT + DCD) and nitrophosphate (NP) treatments during the wheat growth
811 season. The standard errors of soil temperature and moisture were not shown for
812 figure clarity.

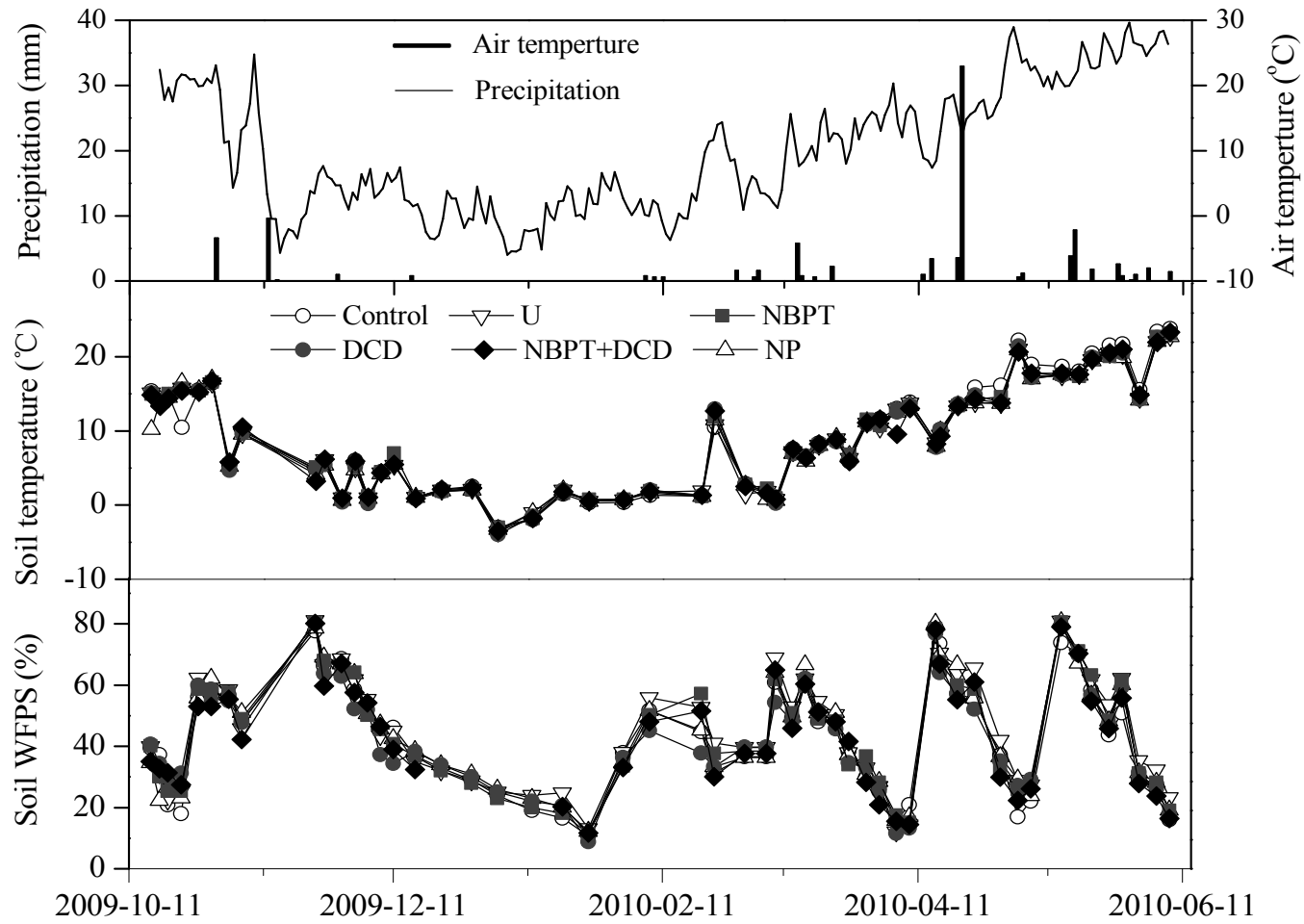
813

814 **Figure 2.** Temporal variation of nitrous oxide fluxes in the control, urea alone (U),
815 urea with NBPT (NBPT), urea with DCD (DCD), urea with NBPT + DCD (NBPT +
816 DCD) and nitrophosphate (NP) treatments during the wheat growth season. Flux
817 values are mean values ± standard errors for three replicates. Arrows indicate date of
818 fertilizer application.

819

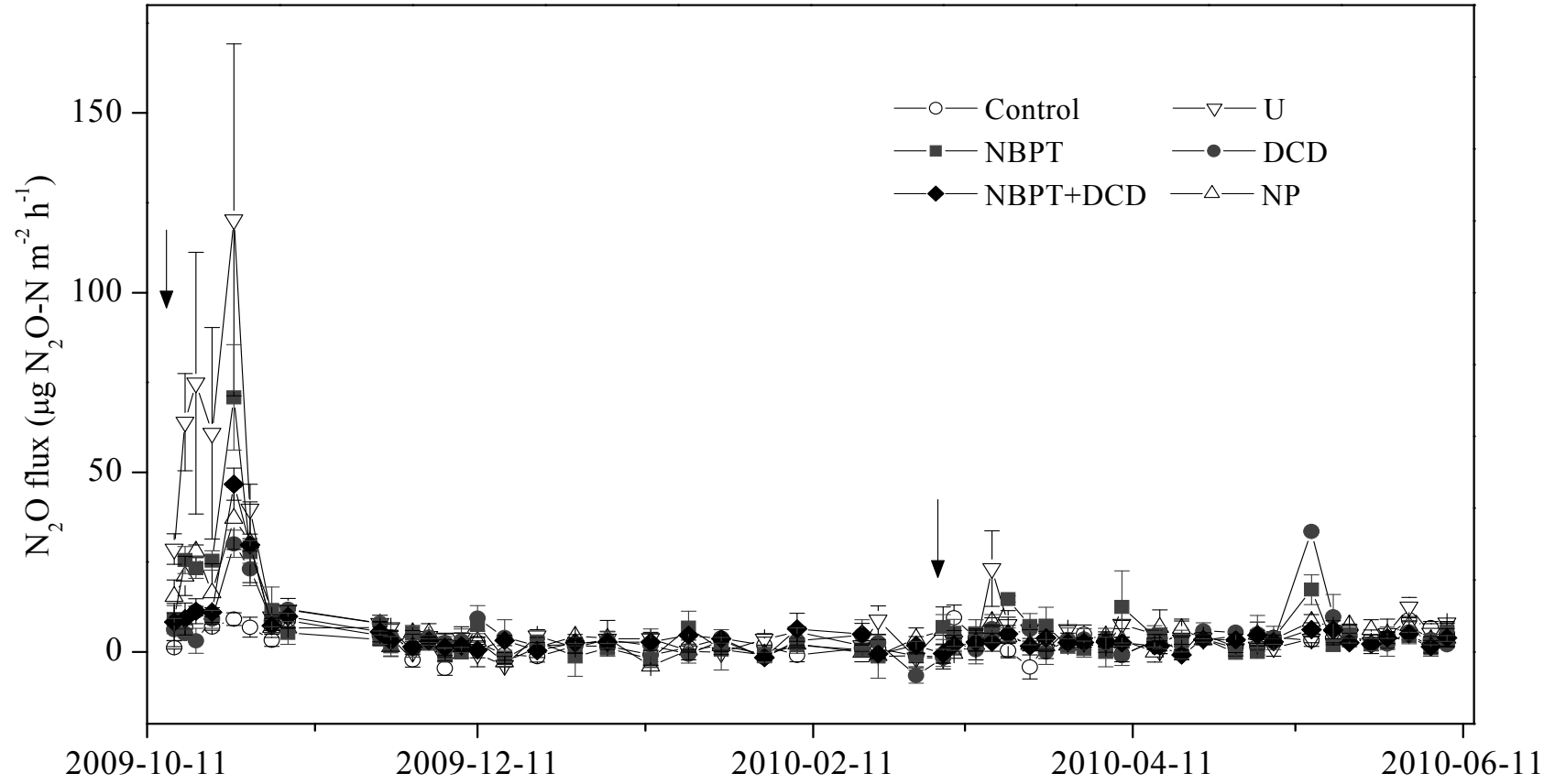
820 **Figure 3.** Temporal variation of ammonium and nitrate concentrations in samples
821 from the 0–20 cm depth in the control, urea alone (U), urea with NBPT (NBPT), urea
822 with DCD (DCD), urea with NBPT + DCD (NBPT + DCD) and nitrophosphate (NP)
823 treatments during the wheat growth season. Vertical bars denote the standard error of
824 the means ($n = 3$).

825 **Figure 1**



826

827 **Figure 2**



828

