

The response to reviewers' constructive comments and suggestions

Dear Editor,

The manuscript has been revised carefully according to constructive comments and suggestions of two reviewers. We hope this revised manuscript can fit with the acceptable standard for Biogeosciences. We would like to express our heartfelt gratitude to two anonymous reviewers and the editor for their constructive comments and suggestions that really improved the manuscript greatly. Our point-by-point responses are provided below, and text that has been added or modified from the original text is shown in the marked-up manuscript **in red font**.

Reviewer 1#:

Comments: This is a well-written paper. The research topic is scientifically sound and interesting, related to the influence of inhibitors and nitrate-based fertilizer on the N₂O emissions, crop yield and N use efficiency in the North China Plain. The authors found that the inhibitors of NBPT and DCD together with urea as basal fertilizer rather than supplemental fertilizer, and the nitrate-based fertilizer instead of urea could greatly reduce N₂O emissions during the winter wheat growth season. Meanwhile, nitrate-based fertilizer significantly increased wheat yield by 12.3% and N use efficiency from 28.8% to 35.9%. These are very interesting and valuable findings for readers and policy-makers. The experiment design is robust, data presentation is clear and the discussion section is written fully. Therefore, I would recommend its publication with a minor revision. It certainly falls within the remit of Biogeosciences.

Answer: Thank you very much for so nice comments.

Comments: A few smaller issues: 1. P13572, L10, add “kg N₂O-N ha⁻¹” following “0.49-0.12”.

2. P13574, L21, change “...were drastically than...” into “...were drastically higher than...”.

Answer: We have revised the sentences according to above suggestions.

Comments: 3. P13575, L22, show soil taxonomy such as UAS or FAO.

4. P13580, L3, I suggest that the analysis method for data normality test should be added.

Answer: We have added the missing information in the revised manuscript as suggested by the reviewer.

Comments: 5. P13581, L7, please delete “the”.

6. P135813, L22, Change “ammonia-based” into “ammonium-based”.

7. P13584, L16, Change “denitrification in general could produce more N_2O ” into “denitrification could in general produce more N_2O ”.

8. P13586, L5, “inner Mongolia” should be “Inner Mongolia”.

Answer: Thanks greatly. We improved the sentences considerably according to above suggestions.

Comments: 9. P13588 L9, “Application of urea with NBPT and/or DCD slightly increased wheat yields”, I suggest adding “compared with urea alone” to this sentence.

Answer: We have revised the sentences as above suggestions.

Reviewer #2

Comments:

This manuscript reports on the changes in N_2O fluxes and NUE along with yield by the application of a nitrification inhibitor and different types of N fertilizers. The topic is within the range of the scope and the manuscript is overall well written. The authors employed a field-scale manipulation experiment and the measurements covered whole growing season. The authors employed a proper statistical test and analytical methodology is well established one. One reservation of mine is the global importance of the study to the general readers of the journal. It is obvious that Northern China is

one of the biggest agricultural sources of N₂O, but I would like to see more generalization of the results. Are there any other reports on N₂O emissions from agricultural fields in other countries with the similar climate zone? Any regional comparison or even simple literature review would benefit the quality of the manuscript.

Answer: Thank you very much for so nice comments. According to above suggestion, we have compiled data on N₂O emissions from uplands in other countries with the similar climate zone in the literature (by adding Table 6, please see below table) and make a comparison. The section of Discussion has been revised as follows:

P13586 L7, "...study. However, Zhang et al. (2014b) found that the North China Plain is a large agricultural N₂O source in China, contributing 36.3 % of the total annual N₂O emission from the China's croplands. To make a global comparison, we compiled the literature data of N₂O emissions from the temperate uplands under inorganic N fertilizer application in some countries of Asia, Europe and North America with similar latitudes to the studied region (Table 6). The emission factors of N applied in the North China Plain are generally lower than in the other countries, indicating a lower capacity of N applied being converted into N₂O in the test soil. This is probably because N₂O is predominantly produced from nitrification and denitrification is organic C-limited as discussed above. In contrast, the total N₂O emissions from the studied region are obviously higher due to the greater N fertilizer loading (Ju et al., 2009), suggesting that agricultural practices for reducing N₂O emission are urgently required. Our results confirm that N fertilizer sources ...".

Comments: Other minor comments;

P13573 L1. Practice along with soil and climatic factors

P13574 L11. So -> As such

Answer: We revised the sentences according to above suggestions.

Comments: P13574L22. Drastically greater or smaller? than (something is missing)

Answer: We have added the missing word “greater”.

Comments: P13582L12. concentrations

Table 5. Pls use subscripts for chemicals (ammonium, nitrate)

Answer: We have revised the Table 5 according to above suggestions.

Table 6. Summary of N₂O emissions from uplands under inorganic fertilizer application in the countries with temperate climate.

Site	MAT (°C)	MAP (mm)	SOC (g C kg ⁻¹)	pH	Warm season			Cold season				Whole year			Reference	
					Crop	Applied N	N ₂ O	N ₂ O	Crop	Applied N	N ₂ O	N ₂ O	Applied N	N ₂ O		N ₂ O
						(kg N ha ⁻¹)	emission	EF		(kg N ha ⁻¹)	emission	EF	(kg N	emission		EF
						(kg N ha ⁻¹)	(%)	(%)		ha ⁻¹)	(kg N ha ⁻¹)	(%)	ha ⁻¹)	(kg N ha ⁻¹)		(%)
Fengqiu, China	14	615	7	8.7	Maize	250	3.8	1.3	Wheat	250	0.6	0.3	500	4.5	0.8	Ding et al. (2007)
Huantai, China	13	586	10	8.3	Maize	330	1.6	0.4	Wheat	270	2.4	0.8	600	4.0	0.6	Cui et al. (2012)
Baoding, China	12	555	9	8.1	Maize	173	4.5	2.2	Wheat	165	3.3	1.3	338	7.7	1.8	Zhang et al. (2014c)
Tsukuba, Japan	16	1460	19	5.7	Soybean	20	2.7	13	Wheat	100	0.5	0.5	120	3.2	2.7	Nishimura et al. (2005)
Fukushima, Japan	14	1207	14	7.4	Barley	150	3.2	2.0	–	–	–	–	–	–	–	Shoji et al. (2001)
Madrid, Spain	13	430	8	7.3	Onion	110	0.8	0.6	Fallow	0	0.25	–	110	1.2	0.7	Mejjide et al. (2009)
Lavesum, Germany	10	887	18	5.3	Wheat	220	0.6	0.2	Fallow	0	1.0	–	220	1.9	0.5	Lebender et al. (2014)
Turin, Italy	12	734	10	8.1	Maize	130	0.0	0.0	Fallow	0	2.9	–	130	2.9	3.4	Alluvione et al. (2010)
Boone, USA	9	825	33	7.2	Maize	168	2.9	1.0	–	–	–	–	–	–	–	Parkin and Hatfield (2014)
Michigan, USA	8	628	20	7.0	Maize	225	3.9	1.4	–	–	–	–	–	–	–	Hoben et al. (2011)
Michigan, USA	8	628	20	7.0	Maize	180	2.5	1.2	–	–	–	–	–	–	–	Hoben et al. (2011)
Michigan, USA	8	628	20	7.0	Maize	135	1.7	0.9	–	–	–	–	–	–	–	Hoben et al. (2011)
Michigan, USA	8	628	20	7.0	Maize	90	1.1	0.7	–	–	–	–	–	–	–	Hoben et al. (2011)
Michigan, USA	8	628	20	7.0	Maize	45	0.9	1.1	–	–	–	–	–	–	–	Hoben et al. (2011)
Morris, USA	6	645	32	7.2	Maize	78	–	–	Fallow	0	–	–	78	5.2	3.0	Johnson et al. (2012)
Morris, USA	6	645	32	7.2	Wheat	78	–	–	Fallow	0	–	–	78	4.2	2.8	Johnson et al. (2012)

MAT, mean annual temperature; MAP, mean annual precipitation; EF, the N₂O emission factor of applied N.

Nitrous oxide emission and nitrogen use efficiency in response to nitrophosphate, N-(n-butyl) thiophosphoric triamide and dicyandiamide of a wheat cultivated soil under sub-humid monsoon conditions

W. X. Ding¹, Z. M. Chen¹, H. Y. Yu¹, J. F. Luo², G. Y. Yoo³, J. Xiang¹, H. J. Zhang¹ and J. J. Yuan¹

¹ State Key Laboratory of Soil and Sustainable Agriculture, Institute of Soil Science, Chinese Academy of Sciences, Nanjing 210008, China

² Land and Environment, AgResearch, Hamilton 3240, New Zealand

³ College of Engineering, Kyung Hee University, Yongin-si 446-701, Republic of Korea

Abstract

A field experiment was designed to study the effects of nitrogen (N) source and urease inhibitor N-(n-butyl) thiophosphoric triamide (NBPT) or nitrification inhibitor dicyandiamide (DCD) on nitrous oxide (N₂O) emission and N use efficiency (NUE) in a sandy loam soil. Six treatments including no N fertilizer (control), N fertilizer urea alone (U), urea plus NBPT (NBPT), (4) urea plus DCD (DCD), urea plus NBPT and DCD (NBPT + DCD), and nitrate-based fertilizer nitrophosphate (NP) were designed and implemented separately during the wheat growth period. Seasonal cumulative N₂O emissions with urea alone amounted to $0.49 \pm 0.12 \text{ kg N}_2\text{O-N ha}^{-1}$

and were significantly ($P < 0.05$) reduced to 0.28 ± 0.03 , 0.31 ± 0.01 and 0.26 ± 0.01 kg N₂O-N ha⁻¹ by application of DCD, NBPT and NBPT + DCD, respectively. Cumulative N₂O emissions from NP were 0.28 ± 0.01 kg N₂O-N ha⁻¹. A single N₂O flux peak was identified following basal fertilization, and DCD and/or NBPT inhibition effects mainly occurred during the peak emission period. The NP application significantly ($P < 0.05$) increased wheat yield by 12.3 % and NUE from 28.8 % (urea alone) to 35.9 %, while urease and/or nitrification inhibitors showed a slight increase effect. Our results clearly indicated that the application of urea as basal fertilizer, but not as supplemental fertilizer, together with DCD and NBPT is an effective practice to reduce N₂O emissions. The application of NP instead of urea would be an optimum agricultural strategy for reducing N₂O emissions and increasing crop yield and NUE for wheat cultivation in soils of the North China Plain.

Key words: Ammonium-based fertilizer; Inhibitors; Nitrate-based fertilizer; Nitrogen use efficiency; Nitrous oxide; Wheat yield

1 Introduction

Nitrous oxide (N₂O) is a potent and long-lived atmospheric greenhouse gas, with an annual increasing rate of 0.26 % over the past decades and a contribution of 7 % to the annual increase in radiative forcing (IPCC, 2007). Agricultural soils are identified as the major source of atmospheric N₂O, contributing 4.1 Tg N yr⁻¹ (IPCC, 2013) to the global atmospheric N₂O budget of ~ 14 Tg N yr⁻¹ (Fowler et al., 2009). Field management practices ~~and~~ along with soil and climatic factors are recognized as being

determinants of N₂O emissions from agricultural soils (Stehfest and Bouwman, 2006; Gagnon et al., 2011). Among management practices, the large inputs of industrially fixed N in agriculture are a major perturbation to terrestrial N cycling and a major contribution to accelerating N₂O emissions (Galloway et al., 2008). During the period 1990–2005, agricultural N₂O emissions were globally estimated to have increased by 17 % (USEPA, 2006), and are projected to increase by 35–60 % by 2030 due to the continuous increase of global N fertilizer consumption and animal manure production (FAO, 2003).

China is a major agricultural producer (West et al., 2014) and the amount of applied N fertilizer has increased from 7.07 to 26.21 Tg N yr⁻¹ over the period from 1977 to 2005 (Ju et al., 2009). The North China Plain, primarily containing low organic carbon (C) calcareous soils (6.40 vs. 9.60 g C kg⁻¹ for national upland soils) (Xie et al., 2007), is an intensive agricultural region. It covers ~ 300 000 km² and produces up to one-fourth of the total annual grain yield in China (Liu et al., 2001). A winter wheat and summer maize rotation is a commonly used cropping system, and the annual application rates of synthetic N fertilizers have amounted to 600 kg N ha⁻¹ or more (Zhao et al., 2006; Ju et al., 2009). However, a low proportion of fertilizer N is taken up by crops (< 30 %) and it is estimated that up to 41 % of N applied during the growth season is subject to losses via leaching, nitrification, denitrification and ammonia volatilization (Cai et al., 2002). At present, up to 33 % of fertilizer N was over-applied to the fields in China, resulting in China's contribution to 28 % of the global annual N₂O emissions from croplands (West et al., 2014). Hence, it is urgent to

develop optimum methods for enhancing the recycling of N in the agricultural ecosystem and reducing the fertilizer N-induced N₂O emissions.

In the past decade, a number of field measurements of N₂O emissions have been conducted in the North China Plain (Meng et al., 2005; Ding et al., 2007; Ju et al., 2011). A 3 year field measurement showed that the direct N₂O emission factors of synthetic N applied to the wheat-maize cropping systems was 0.82% (Cai et al., 2013), which was higher than the 0.6% reported from fertilizer N-treated upland soils in China (Xing, 1998). Nitrification is found to be the main process for the N₂O emission because low availability of easily degradable organic C limits denitrification in this region (Ding et al., 2007; Ju et al., 2011). Nitrification inhibitors such as DCD help to retard the oxidation of NH₄⁺ to NO₃⁻ by inhibiting the activities of *Nitrosomonas* bacteria in soil (Prasad and Power, 1995), resulting in the reduction of N₂O emissions directly by decreasing nitrification or indirectly by reducing the availability of NO₃⁻ for denitrification and leaching. As a consequence, DCD can increase NUE by increasing plant growth and N uptake (Asing et al., 2008). Similarly, a urease inhibitor like NBPT can slow the conversion of urea to NH₄⁺, thereby reducing N losses by NH₃ volatilization (Manunza et al., 1999; Zaman et al., 2009) and potentially reducing nitrification and subsequent denitrification rates. ~~SoAs~~ such the use of NBPT with urea-based fertilizers may be a potential management strategy to mitigate N₂O emissions (Menéndez et al., 2009). A combined application of nitrification inhibitor and urease inhibitor with urea can maintain N as NH₄⁺ for a longer time with more chance of the fertilizer-derived N being taken up by the crops

or immobilized by the organic or mineral component of the soil, thereby reducing the gaseous loss (Xu et al., 2002). Though application of inhibitors to reduce N₂O emissions has attracted more attention recently and has already been investigated in many areas (Menéndez et al., 2009; Zaman et al., 2009), their effect on N₂O emissions in the North China Plain has not been fully investigated.

Soil N₂O emissions are also influenced by the source of fertilizer N. Gagnon et al. (2011) found that N₂O emissions from urea ammonium nitrate were drastically greater than those from anhydrous ammonia during the maize growth season in a poorly drained clay soil of Canada. In contrast, Venterea et al. (2005, 2010) reported N₂O emissions from soils amended with anhydrous NH₃ to be 2- to 4-fold greater than that from soils receiving urea ammonium nitrate in a silt loam of the United States. Based on the analysis of published data in the literature, Stehfest and Bouwman (2006) concluded that the N₂O emissions from nitrate-based fertilizers were on average lower than those from ammonium-based fertilizers. During the winter wheat growth season in the North China Plain, limited precipitation occurs. Therefore, it is likely that applying nitrate-based fertilizer instead of urea will not accelerate the leaching of NO₃⁻ but reduce N₂O emissions and increase NUE.

In this study, we hypothesize that application of urease inhibitor and/or nitrification inhibitor with urea will lower N₂O emission and increase wheat yield by suppressing the nitrification rate and increasing NUE in the North China Plain. We also hypothesize that use of a nitrate-based fertilizer nitrophosphate by replacing urea will have similar effects. The objectives of this study were: (1) to evaluate the

influence of application of urea with NBPT, DCD and NBPT + DCD on N₂O emissions and (2) to investigate whether the use of nitrophosphate instead of urea reduces N₂O emissions from an intensively cultivated calcareous soil during the wheat growth season.

2 Materials and methods

2.1 Experimental site and soil characteristics

The field experiment was conducted at the Fengqiu State Key Agro-ecological Experimental Station, Chinese Academy of Sciences, Henan Province, China (35°00'N, 114°24'E), a typical region of the North China Plain. The region has a sub-humid temperate continental monsoon climate with dry cold winters and wet hot summers. A winter wheat (*Triticum aestivum* L.) and summer maize (*Zea mays* L.) rotation is selected as an intensively managed double-cropping system. The 30 year mean annual temperature was 13.9 °C, with a range varying from –1.0 °C in January to 27.2 °C in July. The mean annual precipitation is 615 mm, two thirds of which falls between June and September. The soil is derived from alluvial sediments of the Yellow River and is classified as aquic inceptisol (Soil Survey Staff, 1994). The physicochemical properties of the soil are summarized in Table 1.

2.2 Treatment and crop management

The field experiment was carried out during the winter wheat growth season and included six fertilization treatments: (1) no N fertilizer (control), (2) N fertilizer urea

alone (U), (3) urea plus N-(n-butyl) thiophosphoric triamide (NBPT), (4) urea plus dicyandiamide (DCD), (5) urea plus NBPT and DCD (NBPT + DCD), and (6) nitrate-based fertilizer nitrophosphate (NP). The plots were arranged in a randomized complete block with three replicates and the plot size was 5 m × 5 m. Urea and nitrophosphate (Jinkai chemical, Kaifeng, China), totaling 200 kg N ha⁻¹, were added in two applications: 120 kg N ha⁻¹ as basal fertilizer and 80 kg N ha⁻¹ as supplemental fertilizer. Calcium superphosphate was applied as basal fertilizer at a rate of 125 kg P₂O₅ ha⁻¹ for all treatments. For the NP treatment, calcium superphosphate was added as the basal fertilizer to ensure the same application rate of phosphate between the treatments. The NBPT (Hengshuo Chemical, Wuhan, China) and DCD (Sunnyfield Chemicals, Ningxia, China) were applied at a rate of 0.2 % and 10 % of the applied N (w/w), respectively. Inhibitor(s) and urea were thoroughly mixed. All basal fertilizers were evenly spread onto the soil surface by hand and immediately incorporated into the surface soil (0–20 cm) by plowing before sowing on 15 October 2009. The supplemental urea and inhibitor(s) or nitrophosphate were surface applied by hand, then integrated into the plowed layer with irrigation water (40 mm) on 6 March 2010. The mature wheat was harvested on 10 June 2010.

2.3 N₂O flux measurement

In situ soil-surface fluxes of N₂O were measured using the static chamber-gas chromatograph (GC) method. Flux measurements were taken over the period from 16 October 2009 to 8 June 2010 (235 days) during the wheat growth season. Immediately

after sowing, a PVC chamber base (30 cm × 30 cm × 10 cm) was inserted into the soil about 5 cm deep between wheat rows in the center of each plot. The PVC chamber (30 cm × 30 cm × 15 cm) was tightly fitted to the top of the base by inserting the flange of the chamber into the water trough at the upper end of the chamber base. The chamber was equipped with two ports: a small, silicon-sealed vent for sampling and a second port for measuring chamber temperature. Gas samples were initially taken twice a week and later reduced weekly then fortnightly over the winter. Sampling was done in the morning between 09:00 LT and 12:00 LT in order to minimize diurnal variation in flux patterns. Each time, four samples of the chamber air were manually pulled into 50 mL syringes at 0, 10, 20 and 30 min after closure, injected into 20 mL pre-evacuated vials fitted with butyl rubber stoppers and taken to our laboratory for analysis. The air temperature inside the chamber was simultaneously measured with a mercury thermometer.

N₂O concentrations were analyzed on a gas chromatograph (Agilent 7890, Santa Clara, CA, USA) equipped with an electron capture detector. The interfering oxygen contained in the injected gas sample (1.0 mL) was separated by a pre-column (1 m) in combination with an analytical column (3 m). Both columns, packed with Porapak Q (80/100 mesh), were attached directly to the 6-port valve to control the backflush. The temperatures of column oven, injector and detector were 40 °C, 100 °C and 300 °C, respectively. The flow rate of carrier gas (95 % Argon + 5 % CH₄) was 40 mL min⁻¹. The standard N₂O gas was provided by the National Institute for Agro-Environmental Sciences, Japan. The N₂O fluxes were calculated using the following equation:

$$F = \rho \times (P/760) \times (V/A) \times (\Delta C/\Delta t) \times [273/(273+T)] \quad (1)$$

where F is the N_2O flux ($\mu\text{g N}_2\text{O-N m}^{-2} \text{ h}^{-1}$), ρ is the density of N_2O at 0 °C and 760 mm Hg (kg m^{-3}), V is the chamber volume (m^3), A is the area from which N_2O was emitted into the chamber (m^2), $\Delta C/\Delta t$ is the rate of N_2O accumulation in the chamber ($\text{ppbv N}_2\text{O-N h}^{-1}$), T is the chamber air temperature in Celsius, and P is the air pressure of the experimental site (mm Hg). The altitude of the experimental site for this study is very close to sea level, so $P/760 \approx 1$. Few sample sets were discarded when they yielded a linear regression value of R^2 greater than 0.90.

2.4 Grain yield and aboveground N uptake

After crops reached physiological maturity (10 June 2011), grain and straw were manually harvested from each plot. Grain and straw were air-dried, then further dried for 3 days at 65 °C and weighed to obtain dry matter yields. Subsamples were ground with a ball mill and analyzed for N concentration with an elemental N analyzer (VarioMax, Elementar, Hanau, Germany). Total N content in aboveground biomass was calculated from the sum of N masses harvested in grain and straw from each plot.

2.5 Auxiliary variables

Soil temperatures were measured, simultaneously with gas sampling, at vertical depths of 5, 10 and 15 cm with a digital thermometer (Model 2455, Yokogawa, Japan). Soil moisture was measured at 5 cm depth at three different positions in the vicinity of each chamber using time domain reflectometry probes and was expressed as

water-filled pore space (WFPS) by the equation:

$$\text{WFPS [\%]} = (\text{volumetric water content [\%]} / \text{total soil porosity [\%]}) \times 100 \quad (2)$$

where total soil porosity = $1 - (\text{soil bulk density} / 2.65)$, with $2.65 \text{ [g cm}^{-3}\text{]}$ being the assumed particle density of the soil. The precipitation and air temperature were monitored at a neighboring meteorological station 100 m away from the experimental field.

During the growth season, six soil samples were taken from the 0–20 cm soil layer at different positions in each plot just after flux measurement using a 5-cm diameter stainless steel soil sampler and then all samples from each plot were thoroughly mixed to form a composite. After visible roots and litter materials were removed, soil samples were passed through a 2 mm sieve and then extracted with 2 M KCl (soil/KCl suspension of 30:100 ratio) for 1 h on a rotary shaker. The extracted solutions were filtered and stored in a deep freezer ($-18 \text{ }^{\circ}\text{C}$) until analysis. The NH_4^+ -N and NO_3^- -N concentrations were measured using a colorimetric method on a Skalar segmented flow analyzer (SAN^{++} , the Netherlands).

2.6 Data analysis and statistics

Average fluxes and standard errors of the N_2O fluxes were calculated from triplicate plots. Seasonal cumulative N_2O emissions were calculated using the following equation:

$$\text{Cumulative } \text{N}_2\text{O emission} = \sum_{i=1}^n (F_i + F_{i+1}) / 2 \times (t_{i+1} - t_i) \times 24 \quad (3)$$

where F is the N_2O flux ($\mu\text{g N}_2\text{O-N m}^{-2} \text{ h}^{-1}$), i is the i th measurement, the term of

$(t_{i+1} - t_i)$ is the number of days between two measurements, and n is the total number of the measurements. The N_2O direct emission factor (%) of fertilizer N applied to the soil with background adjustment was calculated as follows:

$$\text{Emission factor} = ((N_2O-N_{\text{fertilizer}} - N_2O-N_{\text{control}})/N_{\text{fertilizer}}) \times 100 \quad (4)$$

where $N_2O-N_{\text{fertilizer}}$ and N_2O-N_{control} are the cumulative N_2O emissions ($\text{kg } N_2O-N \text{ ha}^{-1}$) in the N-fertilized treatment and the control treatment, respectively, and $N_{\text{fertilizer}}$ is the amount of fertilizer N applied (kg N ha^{-1}). Yield-scaled N_2O emissions were calculated by dividing cumulative N_2O emission by grain yield for each plot. NUE was calculated by dividing differences of the N amount in the aboveground biomass between N-fertilized plots and control plots within the same block by the N application rate (200 kg N ha^{-1}). Soil inorganic N intensities were calculated separately for NH_4^+ (NH_4I), NO_3^- (NO_3I) and the sum of $NO_3^- + NH_4^+$ (IONI) as the summation of daily NH_4^+-N , NO_3^--N or $(NO_3^- + NH_4^+)-N$ concentrations in the 0–20 cm layer over the same period as for cumulative N_2O emissions using linear interpolation between sampling dates, and presented in units of g d kg^{-1} , the index being what is commonly reported (Zebarth et al., 2008; Engel et al., 2010).

All data were statistically analyzed using the SPSS software package for Windows (Version 13.0, SPSS inc, Chicago, IL, USA). The effects of fertilization management on N_2O emissions, emission factor, and grain yields were evaluated using one-way ANOVA, followed by the least significant difference (LSD) test at $P < 0.05$. All dependent variables were evaluated for normality using the Kolmogorov-Smirnov test and were log-transformed to normalize the distributions if

necessary prior to statistical analysis. Correlation and nonlinear regression analyses were used to test relationships between N₂O fluxes and other factors.

3 Results

3.1 Wheat yield and nitrogen use efficiency

Grain yield in the urea alone treatment was 4652 kg ha⁻¹ and this was increased by 1.3 %, 1.8 % or 1.8 % when NBPT, DCD or both were added with urea fertilizer (Table 2). Compared with the urea alone, the increase in the grain yield in the NP treatment was 12.3 % (Table 2). The N process inhibitors, NBPT, DCD or the combination of both, slightly increased the amount of N uptake by wheat plants. However, this increase was not statistically significant ($P > 0.05$). A significant increase in the plant N uptake was observed in the NP treatment compared with urea only treatment ($P < 0.05$). The NUE was calculated at 28.8 % for the urea alone treatment and this was slightly increased to 29.2–31.2 % when urea was applied with NBPT, DCD or the combination of NBPT and DCD. However, the NUE was increased to 35.9% for the NP treatment, which was significantly higher than that for all the urea treatments ($P < 0.05$).

3.2 Soil temperature and moisture

Temporal variations of air temperature, precipitation, and soil WFPS and soil temperature at 5 cm depth over the experimental period are presented in Fig. 1. The cumulative rainfall over the wheat growth season was 97.6 mm which was lower than

the long-term average. Soil moisture levels were highly variable, with WFPS values varying from 10.7 % to 80.4 %. Periods with high soil moisture (> 75 % WFPS) were observed following heavy rainfall or irrigation events. Soil temperature at 5 cm depth was below zero in early January and increased to 23 °C in early June.

3.3 N₂O emissions

Variations of the N₂O fluxes over the wheat growth season are illustrated in Fig. 2. N₂O fluxes from all fertilizer-incorporated treatments showed almost the same pattern with peak N₂O fluxes being observed soon after application of basal fertilizers. The fluxes from all ~~the~~-N fertilizer treatments were generally low on most of the other sampling dates and these fluxes were not significantly different from those from the control. On several occasions in the winter the fluxes were negative. No significant increases in the N₂O fluxes were found following the supplemental fertilization coupled with irrigation. N₂O fluxes did not increase after heavy rainfall events either.

The peak N₂O flux was 120.4 µg N₂O-N m⁻² h⁻¹ in the urea alone treatment. Compared with the urea only, application of NBPT, DCD or the combination of both reduced the peak fluxes by 41.1 %, 75.0 % and 61.2 %, respectively. Application of NP reduced peak fluxes by 69.1 % compared with application of urea alone. Analysis showed that the natural logarithms of the N₂O fluxes were weakly, but not significantly, correlated with soil WFPS in all treatments except the DCD treatment, but significantly ($P < 0.05$) correlated with soil temperature in all treatments except the NBPT + DCD treatment (Table 3).

Cumulative N₂O emissions from the different treatments are listed in Table 4. Total N₂O emissions from the control, urea alone, urea + NBPT, urea + DCD, urea + NBPT + DCD and NP treatments were 0.16 ± 0.02 , 0.49 ± 0.12 , 0.31 ± 0.01 , 0.28 ± 0.01 , 0.26 ± 0.01 and 0.28 ± 0.03 kg N₂O-N ha⁻¹, respectively, over the wheat growth season. The highest total N₂O emission was found from the plot which only received urea. These emissions mainly occurred during the 18 day peak emission period following basal fertilizer application from 16 October to 3 November. Application of NBPT, DCD or the combination of both significantly reduced the seasonal N₂O emissions from urea by 36.7 %, 42.9 % or 46.9 %, respectively ($P < 0.05$). Compared with the emissions from the urea alone treatment, significantly lower N₂O emissions were also observed from the NP treatment ($P < 0.05$) (42.9 % less than those from the urea alone treatment). The direct N₂O emission factor for urea application alone was 0.17 %, and the addition of NBPT, DCD or the combination of both reduced the emission factor for urea to 0.05–0.08 %. These reductions were statistically significant ($P < 0.05$). The direct N₂O emission factor for NP was 0.06 %, which was also significantly lower than that for urea application alone ($P < 0.05$).

The grain yield-scaled N₂O emission from the NP treatment was significantly lower than that from the urea alone treatment ($P < 0.05$), but not different from those from the NBPT, DCD or NBPT + DCD treatments during the wheat growth season (Table 4).

3.4 Soil NH₄⁺ and NO₃⁻ concentrations

309 Soil NH_4^+ and NO_3^- concentrations drastically increased after application of basal N
310 fertilizers compared with the control. The levels of NO_3^- in the NBPT and NBPT +
311 DCD treatments were relatively low for one week after basal fertilizer application
312 compared with those in the other treatments. However, the levels of NO_3^- in the
313 NBPT and NBPT + DCD treatments gradually increased, this was probably due to
314 degradation of NBPT and its subsequent loss of effectiveness. Following application
315 of supplemental fertilizer urea, no apparent increase in soil NO_3^- levels was observed,
316 and NO_3^- concentration kept at a relatively constant level. In contrast, soil NO_3^-
317 concentration following application of supplemental fertilizer NP showed a rapidly
318 decreasing trend. In all urea-added treatments, soil NO_3^- concentration sharply
319 decreased to less than 10 mg N kg^{-1} from 15 April onwards.

320 Soil NH_4^+ concentration increased from 2 to 10 mg N kg^{-1} after application of
321 basal fertilizer; however it sharply decreased soon afterwards. Application of DCD or
322 NBPT + DCD sustained soil NH_4^+ concentrations at higher levels compared with urea
323 application alone. In the NP treatment, soil NH_4^+ concentrations were always at low
324 levels. The natural logarithms of the N_2O fluxes were more correlated with NH_4^+
325 concentrations than with NO_3^- concentrations in the soil, despite the fact that a
326 significant relationship was only observed in the urea alone treatment.

327 Mean soil NH_4 I levels in the NP treatment were the lowest among all N-added
328 treatments and tended to be higher in the DCD and NBPT + DCD treatments
329 compared with urea alone (Table 5). Mean soil NO_3 I levels showed a similar trend
330 among the treatments and were ranked in the order of NBPT, NBPT + DCD > DCD >

urea alone, NP > control. Mean soil IONI levels were also similar among the treatments and were ranked as NBPT + DCD > NBPT, DCD > urea alone, NP > control.

4 Discussion

4.1 Nitrous oxide emissions as affected by nitrogen sources

Compared with the urea alone, application of NP significantly reduced N₂O emissions by 42.9 % during the wheat growth season (Table 4), and increased wheat yield by 12.3 % and NUE by 24.7 % (Table 2). Ju et al. (2011) obtained a similar result in the North China Plain, finding that emissions of N₂O derived from Ca(NO₃)₂ were lower than those from NH₄(SO₄)₂ during the maize growth season (0.38–0.81 vs. 1.31–3.52 kg N₂O-N ha⁻¹). A lower N₂O emission for urea ammonium nitrate than for anhydrous ammonia was also reported in a silt loam of the United States (Venterea et al., 2005). In contrast, Gagnon et al. (2011) measured a significantly higher N₂O emission following application of urea ammonium nitrate or calcium ammonium nitrate compared with anhydrous ammonia in a poorly drained clay soil of Canada. In a German grassland ecosystem, Müller and Sherlock (2004) found that the emissions for ammonium-based fertilizer were lower than those for nitrate-based fertilizer. These researchers suggested that higher emissions from nitrate-based fertilizers were because of the propensity of the fine-textured clay soil to become anaerobic following rainfall and a strong fixation of NH₄⁺ in clay lattices reducing NH₄⁺ available for N₂O production (Chantigny et al., 2004). In this study, mean soil NO₃⁻ intensities (NO₃I)

were not significantly different between the NP and urea alone treatments (Table 5), and mean NO_3^- concentrations (Fig. 3) were higher than the suggested threshold value for denitrification of 5 mg N kg^{-1} (Dobbie and Smith, 2003) during the growth season except for the period from 24 April to 10 June. These results imply that soil NO_3^- concentration was not the only limiting factor affecting denitrification and N_2O emission in the test soil.

The notable difference in the seasonal N_2O emissions between the NP and urea alone treatments occurred mainly during the 18 day peak emission period following the basal fertilizer application and concurrent irrigation from 16 October to 3 November. It has been reported that application of ammonium-based fertilizers emitted more N_2O than nitrate-based fertilizers under aerobic soil conditions, while application of nitrate-based fertilizers induced a greater increase in N_2O production when soil conditions were anoxic (Pathak and Nedwell, 2001; Tenuta and Beauchamp, 2003). For cultivated soils, the primary mechanism of N_2O production is generally believed to be the nitrification process when soil WFPS levels are between 30 % and 70 % and the denitrification process when soil WFPS levels were between 70 % and 90 % (Granli and Bøckman, 1994). Some other studies also suggest that denitrification could in general produce more N_2O compared with nitrification (eg. Dobbie et al., 1999). According to the studies of Ding et al. (2007) and Wan et al. (2009), N_2O in sandy loam soils of the North China Plain was primarily produced by nitrification unless soil WFPS reached 75 % or more. Pihlatie et al. (2004) reported that even at 100 % WFPS in a loamy sand soil with $24 \text{ g organic C kg}^{-1}$, nitrification

was still the dominant N₂O production process. In this study, the highest soil WFPS measured during the peak emission period was ~65 %; thus we suggest that low soil moisture limited denitrification and N₂O production from the nitrate-based fertilizer in the test soil.

In the North China Plain, the addition of starch to soil treated with nitrate-based fertilizers in the field stimulated N₂O production through denitrification, but wheat straw amendment did not do so (Wan et al., 2009; Ju et al., 2011). Previous studies demonstrated that denitrification was not only controlled by soil moisture and nitrate, but also by organic C supply, and increasing organic C availability could reduce the minimum soil moisture threshold for denitrification (van Groenigen et al., 2004; Chantigny et al., 2013). Yu et al. (2012) found that the mass proportion of macroaggregates in a NPK-treated soil with 6.0 g organic C kg⁻¹ only accounted for 8.8 %, while this proportion amounted to 30.8 % in an 18 year compost-added soil with 10.0 g organic C kg⁻¹ in the North China Plain. This change significantly increased the proportion of pores with a neck diameter < 4 µm by reducing the proportion of pores with a neck diameter of 15–60 µm, which in turn lowered the effective diffusion coefficient of oxygen in the soils and the ratio of monounsaturated to branched phospholipid fatty acids (PLFAs) i.e. aerobic to anaerobic microorganisms (Zhang et al., 2014a). According to results found by Myrold and Tiedje (1984), only large aggregates have anaerobic microsites. Thus, it is likely that the relatively low organic C concentration in the test soil retards macroaggregation and slows formation of anaerobic microsites, which in turn results in rise of the

minimum moisture threshold required for denitrification. Consequently, the denitrification process is of much less importance than nitrification for N₂O production and emissions in soils of the North China Plain.

In a German silt loam soil, similar to that tested in this study, Rover et al. (1998) reported that winter was a key period for N₂O emissions from arable crops in the temperate climate zone, contributing ~ 70 % of the annual N₂O losses during the thawing from December to February. Wolf et al. (2010) also verified that N₂O pulses due to spring thaw dominated total annual N₂O emission in a steppe grassland of Inner Mongolia, China. At our site, spring thawing of the soil at the fertilized plots only caused minor N₂O emission pulses, which were considerably lower than those reported earlier for other arable soils (Syväsalo et al., 2004; Teepe et al., 2000). It is suggested that reduced oxygen supply through alteration of pore structure during thawing, and high soil water contents in the winter, would promote microbial denitrification (Edwards and Killham, 1986; Mørkved et al., 2006). Our present study, together with previous measurements (Ding et al., 2007; Ju et al., 2010; Cui et al., 2012; Cai et al., 2012; Cui et al., 2012), showed that the highest soil WFPS was no more than 70 % during the spring thawing period, a value that was lower than the threshold value of 80 % for thawing N₂O pulses in a silt loam found by Rover et al. (1998). The cumulative rainfall during the winter period from December 2009 to February 2010 was only 4.4 mm and no apparent snow cover was observed at our study site. So the warm temperate monsoon zone, with cold and dry winter in the North China Plain, which is distinctly different from other climatic zones such as

western Europe (Dobbie and Smith 2003) and Inner Mongolia of China (Wolf et al., 2010), would not induce thawing N₂O pulses from arable soils, as found in our study. However, Zhang et al. (2014b) found that the North China Plain is a large agricultural N₂O source in China, contributing 36.3 % of the total annual N₂O emission from the China's croplands. To make a global comparison, we compiled the literature data of N₂O emissions from the temperate uplands under inorganic N fertilizer application in some countries of Asia, Europe and North America with similar latitudes to the studied region (Table 6). The emission factors of N applied in the North China Plain are generally lower than in the other countries, indicating a lower capacity of N applied being converted into N₂O in the test soil. This is probably because N₂O is predominantly produced from nitrification and denitrification is organic C-limited as discussed above. In contrast, the total N₂O emissions from the studied region are obviously higher due to the greater N fertilizer loading (Ju et al., 2009), suggesting that agricultural practices for reducing N₂O emission are urgently required. Our results confirm that N fertilizer sources influence soil N₂O emissions, but that this effect probably depends on soil properties and especially climate conditions. Our study also suggests that, compared with urea or ammonium-based fertilizer, applying nitrate-based fertilizer is an effective management strategy to mitigate N₂O emissions and to increase NUE and wheat yield in the North China Plain.

4.2 Nitrous oxide emissions as affected by inhibitors

The application of basal fertilizer urea followed by flooding irrigation resulted in N₂O emission pulses for 18 days. This finding is in agreement with those of other studies

for arable fields (Bouwman et al., 2002; Ding et al., 2007; Cui et al., 2012). The presence of inhibitors NBPT and/or DCD significantly lowered N₂O peak fluxes, and cumulative N₂O emissions during the 18 day peak emission period were reduced by 50.0 % by NBPT, 78.6 % by DCD and 67.9 % by NBPT + DCD, compared with application of urea alone. Our results indicate that the addition of DCD alone or in combination with NBTP effectively reduced N₂O emissions from application of urea. In other sites of the North China Plain, Liu et al. (2013) also reported that nitrification inhibitors DCD and DMPP (3,4-dimethylpyrazole phosphate) could reduce N₂O emissions from application of N fertilizers by 30 % and 21 %, respectively, during the wheat growth season. Ju et al. (2011) observed no apparent differences in cumulative N₂O emissions between zero N control and urea with DMPP during the maize growth season, suggesting strong nitrification inhibition effectiveness of DMPP.

N₂O emission is directly related to the amount of mineral N available in the soil and application of inhibitors with urea can effectively regulate the NO₃⁻ and NH₄⁺ concentrations (Li et al., 2009; Zaman et al., 2009). Recently, Maharjan and Venterea (2013) demonstrated that N₂O emissions were more correlated with soil NO₂⁻ intensity rather than NO₃⁻ or NH₄⁺ intensity, and that inhibitors controlled N₂O production by adjusting soil NO₂⁻ intensity. In this study, soil NH₄⁺ concentration slightly increased in the presence of DCD and, in contrast, relatively low NH₄⁺ concentration was found after NBPT application following application of basal fertilizer. NBPT delays urea hydrolysis, thereby lowering soil pH elevation and NH₄⁺ production, which can in turn reduce NH₃ toxicity effects on nitrite-oxidizing bacteria

(NOB). DCD slows oxidation of NH_4^+ to NO_2^- mainly by inhibiting activities of ammonia-oxidizing bacteria (AOB), which allows NOB to use NO_2^- at the rate closely matched to its production rate (Zaman et al., 2008; Maharjan and Venterea, 2013). Both NBPT and DCD could additively attenuate formation of N_2O from urea in the soil. Thus, the reduction of N_2O emissions by inhibitors is probably due to both low oxidation rate of NH_4^+ and low NO_2^- concentration, thereby reducing N_2O “leaking” as a by-product of nitrification (Firestone and Davidson, 1989).

Following supplemental fertilization with or without inhibitors, no distinct N_2O flux peaks were found in our study. This may be attributable to no significant increase of soil NH_4^+ and NO_3^- concentrations after urea top-dressing. Cui et al. (2012) ascribed low increases in mineral N concentrations to large losses of urea-derived ammonia via volatilization. However, a field measurement at our study site showed that $< 1\%$ of the N applied was lost via volatilization following urea top-dressing in March (Ni et al., 2009); so a large amount of NH_3 loss would not occur at our site. Milchunas et al. (1988) suggested that urea hydrolysis is primarily affected by soil moisture. An incubation at $13\text{ }^\circ\text{C}$ demonstrated that lowering soil moisture level from 60 % to below 40 % water holding capacity produced a longer lag before ammonia evolution and considerably retarded urea hydrolysis (Foster et al., 1980). The range of soil WFPS between 40 and 60 % during the period following urea top-dressing with subsequent irrigation suggested that soil moisture could partly have affected N_2O production. Suter et al. (2011) observed that lowering incubation temperature from $25\text{ }^\circ\text{C}$ to $5\text{ }^\circ\text{C}$ greatly retarded the hydrolysis of urea when WFPS was below 60%,

especially for an alkaline soil with low urease activity. In contrast, the temperature decrease increased the inhibitory effectiveness of NBPT on urea hydrolysis. In this study, soil temperature measured in the field after urea top-dressing varied from 2 °C to 9 °C, close to or just above the thresholds for nitrification (above 5.0 °C) (Anderson et al., 1971) and urea hydrolysis (~ 2 °C) (Xu et al., 1993; Yadav et al., 1987). It is obvious that low soil temperature led to the absence of fertilizer N-induced N₂O flux peaks following the supplemental fertilization and urease or nitrification inhibitors should not necessarily be applied with supplemental fertilizers during the wheat growth season.

Application of urea with NBPT and/or DCD compared with urea alone slightly increased wheat yields, which differs from application of NP. Similar results were also obtained by Ju et al. (2011) and Liu et al. (2013) in the North China Plain. A meta-analysis of data measured in Germany showed that N fertilizers with nitrification inhibitors did not significantly influence the yields of all investigated crops (Hu et al., 2014). The absence of inhibitor effects on crop yields might have been ascribed to the following three reasons. Firstly, low precipitation during the wheat growth season reduced the risk of N leaching and resulted in low N losses. This is evidenced by the significant stimulation of NP on wheat yields. Secondly, it is well known that the application rate of N fertilizers is far above optimum for crops (West et al., 2014). The overloading of N fertilizer might mask the influence on crop yields of increased mineral N in soils caused by inhibitors. The result of Sharma and Prasad (1996) supported the hypothesis that application of DCD significantly increased

maize yield when the application rate of fertilizer N was as low as 60 kg N ha⁻¹. It should be noted that the increase in NH₄⁺ concentration in the test soil due to DCD application alone following the basal fertilization may stimulate NH₃ volatilization, resulting in higher N losses compared with urea alone, since NH₃ volatilization accounted for ~ 13 % of N applied (Ni et al., 2009). Mahmood et al. (2011) demonstrated that application of DCD to an alkaline calcareous soil increased fertilizer N losses. Finally, also more importantly, application of DCD with supplemental fertilizer slowed the nitrification rate and then lowered NO₃⁻ supply for wheat growth when it was at the rapid growth stage. A lower soil NO₃⁻ concentration in the NP treatment than in the urea-added treatments following the supplemental fertilizer in this study supports this speculation. Based on this study, it is not necessary to apply DCD with supplemental fertilizer urea and a combination of urease and nitrification inhibitors would be a better approach to reduce N₂O emission than urease or nitrification inhibitor application alone with basal fertilizer urea for wheat cultivation.

5 Conclusions

The present field study provided an insight into N₂O emissions from a calcareous soil during the wheat growth season in the North China Plain, as affected by application of urease or nitrification inhibitors and nitrate-based fertilizer nitrophosphate. A single N₂O flux peak was found following basal fertilization during the wheat growth period. Application of urea with NBPT, DCD or NBPT + DCD significantly reduced N₂O

emissions from urea by 36.7 %, 42.9 % or 46.9 %, respectively. Application of nitrophosphate also resulted in reduction of total N₂O emissions by 42.9 %, compared with application of urea alone. NBPT and/or DCD were effective in reducing N₂O emissions following basal fertilization. Compared with urea application alone, application of inhibitors with urea, either individually or combined together, slightly increased wheat yield and NUE, while nitrophosphate significantly increased wheat yield by 12.3 % and increased NUE from 28.8 % (urea alone) to 35.9 %. N₂O flux was primarily affected by soil temperature and low temperature at the study site minimized fertilizer N-induced N₂O peaks following application of supplemental fertilizer. Based on our findings, the combination of NBPT and DCD with basal fertilizer urea would be an effective practice for reducing N₂O emission. As well, this study suggests that application of nitrophosphate, instead of urea, is an optimum agricultural strategy for reducing N₂O emission and for increasing crop yield and NUE for wheat cultivation in the soils of the North China Plain.

Acknowledgments: This work was supported by the National Basic Research Program of China (2012CB417102), Strategic Priority Research Program of the Chinese Academy of Sciences (XDB15020101) and National Natural Science Foundation of China (41171191, 41471207).

Reference

Alluvione, F., Bertora, C., Zavattaro, L., and Grignani, C.: Nitrous oxide and carbon

552 dioxide emissions following green manure and compost fertilization in corn, *Soil*
553 *Sci. Soc. Am. J.*, 74, 384–395, 2010.

554 Anderson, O. E., Boswell, F. C., and Harrison, R. M.: Variations in low temperature
555 adaptability of nitrifiers in acid soils, *Soil Sci. Soc. Am. Pro.*, 35, 68–71, 1971.

556 Asing, J., Saggar, S., Singh, J., and Bolan, N. S.: Assessment of nitrogen losses from
557 urea and organic manure with and without nitrification inhibitor, dicyandiamide,
558 applied to lettuce under glasshouse conditions, *Aust. J. Soil Res.*, 46, 535–541,
559 2008.

560 Bouwman, A. F., Boumans, L. J. M., and Batjes, N. H.: Emissions of N₂O and NO
561 from fertilized fields: summary of available measurement data, *Global*
562 *Biogeochem. Cy.*, 16, 1058, 2002..

563 Cai, G., Chen, D., White, R. E., Fan, X. H., Pacholski, A., Zhu, Z. L., and Ding, H.:
564 Gaseous nitrogen losses from urea applied to maize on a calcareous fluvo-aquic
565 soil in the North China Plain, *Aust. J. Soil Res.*, 40, 737–748, 2002.

566 Cai, Y. J., Ding, W. X., and Luo, J. F.: Nitrous oxide emissions from Chinese
567 maize-wheat rotation systems: a 3 year field measurement, *Atmos. Environ.*,
568 65:112–122, 2013.

569 Chantigny, M. H., Angers, D. A., Morvan, T., and Pomar, C.: Dynamics of pig slurry
570 nitrogen in soil and plant as determined with ¹⁵N, *Soil Sci. Soc. Am. J.* 68,
571 637–643, 2004.

572 Chantigny, M. H., Pelster, D. E., Perron, M. H., Rochette, P., Angers, D. A., Parent, L.
573 E., Massé, D., and Ziadi, N.: Nitrous oxide emissions from clayey soils amended

574 with paper sludges and biosolids of separated pig slurry, *J. Environ. Qual.*, 42,
575 30–39, 2013.

576 Cui, F., Yan, G. X., Zhou, Z. X., Zheng, X. H., and Deng, J.: Annual emissions of
577 nitrous oxide and nitric oxide from a wheat-maize cropping system on a silt loam
578 calcareous soil in the North China Plain, *Soil Biol. Biochem.*, 48, 10–19, 2012.

579 Ding, W. X., Cai, Y., Cai, Z. C., Yagi, K., and Zheng, X. H.: Nitrous oxide emissions
580 from an intensively cultivated maize-wheat rotation soil in the North China Plain,
581 *Sci. Total Environ.*, 373, 501–511, 2007.

582 Dobbie, K. E., and Smith, K. A.: Impact of different forms of N fertilizer on N₂O
583 emissions from intensive grassland, *Nutr. Cycl. Agroecosys.*, 67, 37–46, 2003.

584 Dobbie, K. E., McTaggart, I. P., and Smith, K. A.: Nitrous oxide emissions from
585 intensive agricultural systems: variations between crops and seasons, key driving
586 variables and mean emission factors, *J. Geophys. Res.* 104, 26891–26899, 1999.

587 Edwards, A. C., and Killham, K.: The effect of freeze/thaw on gaseous nitrogen loss
588 from upland soils. *Soil Use Manage.*, 3, 86–91, 1986.

589 Engel, R., Liang, D. L., Wallander, R., and Bembenek, A.: Influence of urea fertilizer
590 placement on nitrous oxide production from a silt loam soil, *J. Environ. Qual.*, 39,
591 115–125, 2010.

592 FAO: World Agriculture: Towards 2015/2030. An FAO Perspective. FAO, Rome. 97,
593 2003.

594 Firestone, M. K., and Davidson, E. A.: Microbiological basis of NO and N₂O
595 production and consumption in soil, in: *Exchange of trace gases between*

596 terrestrial ecosystems and the atmosphere, edited by: Andreae, M. O., and
 597 Schimel, D. S., John Wiley & Sons, New York, 7–21, 1989.

598 Foster, N. W., Beauchamp, E. G., and Corke, C. T.: The influence of soil moisture on
 599 urea hydrolysis and microbial respiration in jack pine humus, *Can. J. Soil Sci.*,
 600 60, 675–684, 1980.

601 Fowler, D., Pilegaard, K., Sutton, M., Ambus, P., Raivonen, M., Duyzer, J., Simpson,
 602 D., Fagerli, H., Fuzzi, S., and Schjrring, J. K.: Atmospheric composition change:
 603 ecosystems – atmosphere interactions, *Atmos. Environ.*, 43, 5193–5267, 2009.

604 Gagnon, B., Ziadi, N., Rochette, P., Chantigny, M. H., and Angers, D. A.: Fertilizer
 605 source influenced nitrous oxide emissions from a clay soil under corn, *Soil Sci.*
 606 *Soc. Am. J.*, 75, 595–604, 2011.

607 Galloway, J. N., Townsend, A. R., Erisman, J. W., Bekunda, M., Cai, Z. C., Freney, J.
 608 R., Martinelli, L. A., Seitzinger, S. P., and Sutton, M. A.: Transformation of the
 609 nitrogen cycle: recent trends, questions, and potential solutions, *Science*, 320,
 610 889–892, 2008.

611 Granli, T., and Bøckman, O.C.: Nitrous oxide from agriculture, *Nor. J. Agric. Sci.*, 12,
 612 1–128, 1994.

613 Hoben, J. P., Gehl, R. J., Millar, N., Grace, P. R., and Robertson, G. P.: Nonlinear
 614 nitrous oxide (N₂O) response to nitrogen fertilizer in on-farm corn crops of the
 615 US Midwest, *Glob. Change Biol.*, 17, 1140–1152, 2011.

616 Hu, Y., Schraml, M., von Tucher, S., Li, F., and Schmidhalter, U.: Influence of
 617 nitrification inhibitors on yields of arable crops: A meta-analysis of recent

618 studies in Germany, *Int. J. Plant Prod.*, 8, 1735–6814, 2014.

619 IPCC: Climate Change 2007, The Physical Science Basis, Contribution of Working
620 Group I to the Fourth Assessment Report of the Intergovernmental Panel on
621 Climate Change, Cambridge University Press, Cambridge, UK and New York,
622 NY, USA, 2007.

623 IPCC: Climate Change 2013, The Physical Science Basis, Working Group I
624 Contribution to the Fifth Assessment Report of the Intergovernmental Panel on
625 Climate Change, Cambridge University Press, 2013.

626 Johnson, J. M. F., Weyers, S. L., Archer, D. W., and Barbour, N. W.: Nitrous oxide,
627 methane emission, and yield-scaled emission from organically and
628 conventionally managed systems, *Soil Sci. Soc. Am. J.*, 76, 1347–1357, 2012.

629 Ju, X. T., Xing, G. X., Chen, X. P., Zhang, S. L., Zhang, L. J., Liu, X. J., Cui, Z. L.,
630 Yin, B., Christie, P., Zhu, Z. L., and Zhang, F. S.: Reducing environmental risk
631 by improving N management in intensive Chinese agricultural systems, *P. Natl.*
632 *Acad. Sci. USA*, 106, 3041–3046, 2009.

633 Ju, X. T., Lu, X., Gao, Z. L., Chen, X. P., Su, F., Kogge, M., Römheld, V., Christie, P.,
634 and Zhang, F. S.: Processes and factors controlling N₂O production in an
635 intensively managed low carbon calcareous soil under sub-humid monsoon
636 conditions, *Environ. Pollut.*, 159, 1007–1016, 2011.

637 Lebender, U., Senbayram, M., Lammel, J., and Kuhlmann, H.: Effect of mineral
638 nitrogen fertilizer forms on N₂O emissions from arable soils in winter wheat
639 production, *J. Plant Nutr. Soil Sci.*, 177, 722–732, 2014.

640 Li, X. L., Zhang, G. B., Xu, H., Cai, Z. C., and Yagi, K.: Effect of timing of joint
641 application of hydroquinone and dicyandiamide on nitrous oxide emission from
642 irrigated lowland rice paddy field, *Chemosphere*, 75, 1417–1422, 2009.

643 Liu, C., Yu, J., and Kendy, E.: Groundwater exploitation and its impact on the
644 environment in the North China Plain. *Water Int.*, 26, 265–272, 2001.

645 Liu, C., Wang, K., and Zheng, X.: Effects of nitrification inhibitors (DCD and DMPP)
646 on nitrous oxide emission, crop yield and nitrogen uptake in a wheat–maize
647 cropping system, *Biogeosciences*, 10, 2427–2437, doi:10.5194/bg-10-2427-2013,
648 2013.

649 Maharjan, B., and Venterea, R. T.: Nitrite intensity explains N management effects on
650 N₂O emissions in maize, *Soil Biol. Biochem.* 66, 229–238, 2013.

651 Mahmood, T., Ali, R., Latif, Z., and Ishaque, W.: Dicyandiamide increases the
652 fertilizer N loss from an alkaline calcareous soil treated with ¹⁵N-labelled urea
653 under warm climate and under different crops, *Biol. Fert. Soils*, 47, 619–631,
654 2011.

655 Manunza, B., Deiana, S., Pintore, M., and Gessa, C.: The binding mechanism of urea,
656 hydroxamic acid and N-(n-butyl)-phosphoric triamide to the urease active site: a
657 comparative molecular dynamics study, *Soil Biol. Biochem.*, 31, 789–796, 1999.

658 Meijide, A., Garcia-Torres, L., Arce, A., and Vallejo, A.: Nitrogen oxide emissions
659 affected by organic fertilization in a non-irrigated Mediterranean barley field, *Agric.*
660 *Ecosyst. Environ.*, 132, 106–115, 2009.

661 Menéndez, S., Merino, P., Pinto, M., González-Murua, G., and Estavillo, J. M.: Effect

662 of N-(n-butyl) thiophosphoric triamide and 3, 4-dimethylpyrazole phosphate on
 663 gaseous emissions from grasslands under different soil water contents, *J. Environ.*
 664 *Qual.* 38, 27–35, 2009.

665 Meng, L., Ding, W. X., and Cai, Z. C.: Long-term application of organic manure and
 666 nitrogen fertilizer on N₂O emissions, soil quality and crop production in a sandy
 667 loam soil, *Soil Biol. Biochem.*, 37, 2037–2045, 2005.

668 Milchunas, D. G., Parton, W. J., Bigelow, D. S., and Schimel, D. S.: Factors
 669 influencing ammonia volatilization from urea in soils of the shortgrass steppe, *J.*
 670 *Atmos. Chem.*, 6, 323–340, 1988.

671 Mørkved, P. T., Dörsch, P., Henriksen, T. M., and Bakken, L. R.: N₂O emissions and
 672 product ratios of nitrification and denitrification as affected by freezing and
 673 thawing, *Soil Biol. Biochem.*, 38, 3411–3420, 2006.

674 Müller, C., and Sherlock, R. R.: Nitrous oxide emissions from temperate grassland
 675 ecosystems in the Northern and Southern Hemispheres, *Global Biogeochem. Cy.*,
 676 18, GB1045, doi:10.1029/2003GB002175, 2004.

677 Myrold, D. D., and Tiedje, J. M.: Diffusional constraints on denitrification in soil,
 678 *Soil Sci. Soc. Am. J.*, 49, 651–657, 1984.

679 Ni, K., Ding, W. X., and Cai, Z. C.: Ammonia volatilization from soil as affected by
 680 long-term application of organic manure and chemical fertilizers during wheat
 681 growing season, *J. Agro-Environ. Sci.*, 28, 2614–2622, 2009.

682 Nishimura, S., Sawamoto, T., Akiyama, H., Sudo, S., Cheng, W. G., and Yagi, K.:
 683 Continuous, automated nitrous oxide measurements from paddy soils converted

684 to upland crops, *Soil Sci. Soc. Am. J.*, 69, 1977–1986, 2005.

685 Parkin, T. B., and Hatfield, J. L.: Enhanced efficiency fertilizers: effect on nitrous
686 oxide emissions in Iowa, *Agron. J.*, 106, 694–702, 2014.

687 Pathak, H., and Nedwell, D. B.: Nitrous oxide emission from soil with different
688 fertilizers, water levels and nitrification inhibitors, *Water Air Soil Poll.*, 129,
689 217–228, 2001.

690 Pihlatie, M., Syväsalo, E., Simojoki, A., Esala, M., and Regina, K.: Contribution of
691 nitrification and denitrification to N₂O production in peat, clay and loamy sand
692 soils under different soil moisture conditions, *Nutr. Cycl. Agroecosys.*, 70,
693 35–141, 2004.

694 Prasad, R., and Power, J. F.: Nitrification inhibitors for agriculture, health and the
695 environment, *Adv. Agron.*, 54, 233–281, 1995.

696 Rover, M., Heinemeyer, O., and Kaiser, E. A.: Microbial induced nitrous oxide
697 emissions from an arable soil during winter, *Soil Biol. Biochem.*, 30, 1859–1865,
698 1998.

699 Sharma, S. N., and Prasad, R.: Use of nitrification inhibitors (neem and DCD) to
700 increase N efficiency in maize-wheat cropping system, *Fertil. Res.*, 44, 169–175,
701 1996.

702 Shoji, S., Delgado, J., Mosier, A., and Miura, Y.: Use of controlled release fertilizers
703 and nitrification inhibitors to increase nitrogen use efficiency and to conserve air
704 and water quality, *Commun. Soil Sci. Plant Anal.*, 32, 1051–1070, 2001.

705 Soil Survey Staff: Keys to soil taxonomy, 6th edn. United States Department of

706 **Agriculture, Natural Resources Conservation Service, Washington, DC, 1994.**

707 Stehfest, E., and Bouwman, L.: N₂O and NO emission from agricultural fields and
708 soils under natural vegetation: summarizing available measurement data and
709 modeling of global annual emissions, *Nutr. Cycl. Agroecosys.*, 74, 207–228,
710 2006.

711 Suter, H. C., Pengthamkeerati, P., Walker, C., and Chen, D.: Influence of temperature
712 and soil type on inhibition of urea hydrolysis by N-(n-butyl) thiophosphoric
713 triamide in wheat and pasture soils in south-eastern Australia, *Soil Res.*, 49,
714 315–319, 2011.

715 Syväsalo, E., Regina, K., Pihlatie, M., and Esala, M.: Emissions of nitrous oxide from
716 boreal agricultural clay and loamy sand soils, *Nutr. Cycl. Agroecosys.*, 69,
717 155–165, 2004.

718 Teepe, R., Brumme, R., and Beese, F.: Nitrous oxide emissions from frozen soils
719 under agricultural, fallow and forest land, *Soil Biol. Biochem.*, 32, 1807–1810,
720 2000.

721 Tenuta, M., and Beauchamp, E. G.: Nitrous oxide production from granular nitrogen
722 fertilizers applied to a silt loam soil, *Can. J. Soil Sci.*, 83, 521–532, 2003.

723 United States Environmental Protection Agency (USEPA): Global Anthropogenic
724 Non-CO₂ Greenhouse Gas Emissions: 1990–2020, EPA 430-R-06-003, United
725 States Environmental Protection Agency, Washington, DC, 2006.

726 Van Groenigen, J. W., Kasper, G. J., Velthof, G. L., Dasselaar, Pol-van van den A.,
727 and Kuikman, P. J.: Nitrous oxide emissions from silage maize fields under

728 different mineral nitrogen fertilizer and slurry application, *Plant Soil*, 263,
 729 101–111, 2004.

730 Venterea, R. T., Burger, M., and Spokas, K. A.: Nitrogen oxide and methane
 731 emissions under varying tillage and fertilizer management, *J. Environ. Qual.*, 34,
 732 1467–1477, 2005.

733 Venterea, R. T., Dolan, M. S., and Ochsner, T. E.: Urea decreases nitrous oxide
 734 emissions compared with anhydrous ammonia in a Minnesota corn cropping system,
 735 *Soil Sci. Soc. Am. J.*, 74, 407–418, 2010.

736 Wan, Y. J., Ju, X. T., Ingwersen, J., Schwarz, U., Stange, C. F., Zhang, F. S., and
 737 Streck, T.: Gross nitrogen transformations and related nitrous oxide emissions in
 738 an intensively used calcareous soil, *Soil Sci. Soc. Am. J.*, 73, 102–112, 2009.

739 West, P. C., Gerber, J. S., Engstrom, P. M., Mueller, N. D., Brauman, K. A., Carlson,
 740 K. M., Cassidy, E. S., Johnston, M., MacDonald, G. R. D. K., and Siebert, S.:
 741 Leverage points for improving global food security and the environment, *Science*,
 742 345, 325–327, 2014..

743 Wolf, B., Zheng, X., Brüggemann, N., Chen, W., Dannenmann, M., Han, X., Sutton,
 744 M. A., Wu, H., Yao, Z., and Butterbach-Bahl, K.: Grazing-induced reduction of
 745 natural nitrous oxide release from continental steppe, *Nature*, 464, 881–884,
 746 2010.

747 Xie, Z. B., Zhu, J. G., Liu, G., Georg, C., Toshihiro, H., Chen, C. M., Sun, H. F.,
 748 Tang, H. Y., and Zeng, Q.: Soil organic carbon stocks in China and changes from
 749 1980s to 2000s, *Glob. Change Biol.*, 13, 1989–2007, 2007.

750 Xing, G. X.: N₂O emission from cropland in China, *Nutr. Cycl. Agroecosys.*, 52,
 751 249–254, 1998.

752 Xu, J. G., Heeraman, D. A., and Wang, Y.: Fertilizer and temperature effects on urea
 753 hydrolysis in undisturbed soil, *Biol. Fert. Soils*, 16, 63–65, 1993.

754 Xu, X. K., Boeckx, P., van Cleemput, O., and Zhou, L. K.: Urease and nitrification
 755 inhibitors to reduce emissions of CH₄ and N₂O in rice production, *Nutr. Cycl.*
 756 *Agroecosys.*, 64, 203–211, 2002..

757 Yadav, D. S., Kumar, V., Singh, M., and Relan, P. S.: Effect of temperature and
 758 moisture on kinetics of urea hydrolysis and nitrification, *Aust. J. Soil Res.*, 25,
 759 185–191, 1987.

760 Yu, H. Y., Ding, W. X., Luo, J. F., Geng, R. L., and Cai, Z. C.: Long-term application
 761 of compost and mineral fertilizers on aggregation and aggregate-associated
 762 carbon in a sandy loam soil, *Soil Till. Res.*, 124, 170–177, 2012.

763 Zaman, M., Nguyen, M. L., Blennerhassett, J. D., and Quin, B. F.: Reducing NH₃,
 764 N₂O and NO₃-N losses from a pasture soil with urease or nitrification inhibitors
 765 and elemental S-amended nitrogenous fertilizers, *Biol. Fert. Soils*, 44, 693–705,
 766 2008.

767 Zaman, M., Saggar, S., Blennerhassett, J. D., and Singh, J.: Effect of urea and
 768 nitrification inhibitors on N transformation, gaseous emissions of ammonia and
 769 nitrous oxide, pasture yield and N uptake in grazed pasture system, *Soil Biol.*
 770 *Biochem.*, 41, 1270–1280, 2009.

771 Zebarth, B. J., Rochette, P., and Burton, D. L.: N₂O emissions from spring barley

772 production as influenced by fertilizer nitrogen rate, *Can. J. Soil Sci.*, 88, 197–205,
 773 2008.

774 Zhang, H. J., Ding, W. X., He, X. H., Yu, H. Y., Fan, J. L., and Liu, D. Y.: Influence of
 775 20 year organic and inorganic fertilization on organic carbon accumulation and
 776 microbial community structure of aggregates in an intensively cultivated sandy
 777 loam soil, *Plos One*, 9, e92733, doi:10.1371/journal.pone.0092733, 2014a.

778 Zhang, W., Yu, Y. Q., Li, T. T., Sun, W. J., and Huang, Y.: Net greenhouse gas balance
 779 in China's croplands over the last three decades and its mitigation potential,
 780 *Environ. Sci. Technol.*, 48, 2589–2597, 2014b.

781 Zhang, Y., Mu, Y., Zhou, Y., Liu, J., and Zhang, C.: Nitrous oxide emissions from
 782 maize-wheat field during 4 successive years in the North China Plain,
 783 *Biogeosciences*, 11, 1717–1726, 2014c.

784 Zhao, R. F., Chen, X. P., Zhang, F. S., Zhang, H., Schroder, J., And Romheld. V.:
 785 Fertilization and nitrogen balance in a wheat maize rotation system in North
 786 China, *Agron. J.*, 98, 938–945, 2006.

787 **Table 1.** Soil properties.

Soil depth	pH	Bulk density	Organic C	Total N	C/N	NO ₃ ⁻ -N	NH ₄ ⁺ -N	Particle size distrubution (%)		
(cm)	(H ₂ O)	(Mg m ⁻³)	(g C kg ⁻¹)	(g N kg ⁻¹)		(mg N kg ⁻¹)	(mg N kg ⁻¹)	Sand	Silt	Clay
0-20	8.60	1.40	12.0	1.50	8.0	14.70	2.48	17.0	72.0	11.0

788

789 **Table 2.** Effects of urease and/or nitrification inhibitors and nitrophosphate on wheat biomass, amount of N uptake by crops and N use
790 efficiency.

Treatment	Biomass (kg ha ⁻¹)			Amount of N uptake (kg N ha ⁻¹)			N use efficiency (%)
	Grain	Straw	Total	Grain	Straw	Total	
Control	2297±150 c	2215±134 b	4513±283 c	24.4±1.6 c	11.9±0.7 b	36.4±2.3 c	–
U	4652±11 b	4075±81 a	8727±85 b	59.6±0.1 b	34.2±0.7 a	93.9±0.7 b	28.8 ±0.8 b
NBPT	4711±126 b	4098±356 a	8809±472 b	60.4±1.6 b	34.4±3.0 a	94.8±4.5 b	29.2 ±1.0 b
DCD	4736±103 ab	4080±52 a	8816±86 b	60.7±1.3 b	34.3±0.4 a	95.0±1.1 b	29.3 ±0.9 b
NBPT + DCD	4735±290 ab	4535±503 a	9271±764 ab	60.7±3.7 b	38.1±4.2 a	98.8±7.6 ab	31.2 ±1.2 b
NP	5225±142 a	4906±251 a	10131±370 a	67.0±1.8 a	41.2±2.1 a	108.2±3.7 a	35.9 ±1.1 a

791 Mean ± standard error (*n* = 3).

792 Different letters within the column indicate significantly difference between treatments at *P* < 0.05.

793 **Table 3.** Correlation between $\ln [\text{N}_2\text{O flux} + 1]$ and soil WFPS, soil temperature at depths of 5 ($T_{5\text{cm}}$), 10 ($T_{10\text{cm}}$) and 15 cm ($T_{15\text{cm}}$), ammonium
794 ($\text{NH}_4^+\text{-N}$), nitrate ($\text{NO}_3^-\text{-N}$) or inorganic nitrogen ($\text{NH}_4^+\text{-N}$ plus $\text{NO}_3^-\text{-N}$) concentration.

Treatment	WFPS	$T_{5\text{cm}}$	$T_{10\text{cm}}$	$T_{15\text{cm}}$	$\text{NH}_4^+\text{-N}$	$\text{NO}_3^-\text{-N}$	Inorganic N
Control	0.095	0.413 ^{**}	0.376 [*]	0.392 [*]	-0.153	0.140	0.109
U	0.023	0.381 ^{**}	0.340 [*]	0.346 [*]	0.274 [*]	0.365 ^{**}	0.380 ^{**}
NBPT	0.118	0.275 [*]	0.264	0.274	0.215	0.206	0.222
DCD	0.323 [*]	0.282 [*]	0.180	0.189	0.104	-0.127	-0.092
NBPT + DCD	0.021	0.216	0.252	0.272	0.074	0.155	0.156
NP	0.084	0.301 [*]	0.370 [*]	0.403 ^{**}	0.105	-0.056	-0.037

795 ^{*} $P < 0.05$, ^{**} $P < 0.01$.

Table 4. Effects of urease and/or nitrification inhibitors and nitrophosphate on cumulative N₂O emissions, fertilizer N-induced N₂O emission factors and yield-scaled N₂O emissions.

Treatment	Cumulative N ₂ O emission (kg N ₂ O-N ha ⁻¹)		Ratio of peak to total emissions (%)	Emission factor (% of applied N)	Yield-scaled N ₂ O emission (g N ₂ O-N kg ⁻¹ grain)
	Total	Peak			
Control	0.16±0.02 c	0.03±0.00 d	18.8±2.3 d	—	0.068±0.006 b
U	0.49±0.12 a	0.28±0.10 a	57.1±4.2 a	0.17±0.05 a	0.105±0.026 a
NBPT	0.31±0.01 b	0.14±0.01 b	45.2±3.9 b	0.08±0.00 b	0.065±0.003 b
DCD	0.28±0.01 b	0.06±0.01 c	21.4±2.1 d	0.06±0.00 b	0.060±0.004 b
NBPT + DCD	0.26±0.01 b	0.09±0.00 bc	34.6±3.2 c	0.05±0.00 b	0.056±0.003 b
NP	0.28±0.03 b	0.11±0.03 bc	39.3±3.7 c	0.06±0.01 b	0.053±0.008 b

Mean ± standard error ($n = 3$).

Peak emission denotes cumulative emissions during the 18 days' period following the basal fertilizer application from 16 October to 3 November.

Different letters within the column indicate significantly difference between treatments at $P < 0.05$.

Table 5. Effects of urease and/or nitrification inhibitors and nitrophosphate on soil ammonium (NH₄I), nitrate (NO₃I) and inorganic N (IONI) intensities.

Treatment	NH ₄ I (g N d kg ⁻¹)	NO ₃ I (g N d kg ⁻¹)	IONI (g N d kg ⁻¹)
Control	0.24±0.01 e	2.58±0.01 d	2.82±0.00 d
U	0.40±0.03 d	4.75±0.13 c	5.15±0.16 c
NBPT	0.61±0.02 c	6.18±0.08 a	6.79±0.08 b
DCD	0.96±0.01 b	5.74±0.01 b	6.70±0.01 b
NBPT + DCD	1.07±0.01 a	6.11±0.16 a	7.17±0.16 a
NP	0.36±0.02 d	4.69±0.09 c	5.05±0.07 c

Mean±standard error ($n = 3$). Different letters within the column indicate significantly difference between treatments at $P < 0.05$.

806 **Table 6.** Summary of N₂O emissions from uplands under inorganic fertilizer application in the countries with temperate climate.

Site	MAT (°C)	MAP (mm)	SOC (g C kg ⁻¹)	pH	Warm season			Cold season			Whole year			Reference		
					Crop	Applied N	N ₂ O	N ₂ O	Crop	Applied N	N ₂ O	N ₂ O	Applied N		N ₂ O	N ₂ O
						(kg N ha ⁻¹)	emission (kg N ha ⁻¹)	EF (%)		(kg N ha ⁻¹)	emission (kg N ha ⁻¹)	EF (%)			(kg N ha ⁻¹)	emission (kg N ha ⁻¹)
Fengqiu, China	14	615	7	8.7	Maize	250	3.8	1.3	Wheat	250	0.6	0.3	500	4.5	0.8	Ding et al. (2007)
Huantai, China	13	586	10	8.3	Maize	330	1.6	0.4	Wheat	270	2.4	0.8	600	4.0	0.6	Cui et al. (2012)
Baoding, China	12	555	9	8.1	Maize	173	4.5	2.2	Wheat	165	3.3	1.3	338	7.7	1.8	Zhang et al. (2014c)
Tsukuba, Japan	16	1460	19	5.7	Soybean	20	2.7	13	Wheat	100	0.5	0.5	120	3.2	2.7	Nishimura et al. (2005)
Fukushima, Japan	14	1207	14	7.4	Barley	150	3.2	2.0	–	–	–	–	–	–	–	Shoji et al. (2001)
Madrid, Spain	13	430	8	7.3	Onion	110	0.8	0.6	Fallow	0	0.25	–	110	1.2	0.7	Meijide et al. (2009)
Lavesum, Germany	10	887	18	5.3	Wheat	220	0.6	0.2	Fallow	0	1.0	–	220	1.9	0.5	Lebender et al. (2014)
Turin, Italy	12	734	10	8.1	Maize	130	0.0	0.0	Fallow	0	2.9	–	130	2.9	3.4	Alluvione et al. (2010)
Boone, USA	9	825	33	7.2	Maize	168	2.9	1.0	–	–	–	–	–	–	–	Parkin and Hatfield (2014)
Michigan, USA	8	628	20	7.0	Maize	225	3.9	1.4	–	–	–	–	–	–	–	Hoben et al. (2011)
Michigan, USA	8	628	20	7.0	Maize	180	2.5	1.2	–	–	–	–	–	–	–	Hoben et al. (2011)
Michigan, USA	8	628	20	7.0	Maize	135	1.7	0.9	–	–	–	–	–	–	–	Hoben et al. (2011)
Michigan, USA	8	628	20	7.0	Maize	90	1.1	0.7	–	–	–	–	–	–	–	Hoben et al. (2011)
Michigan, USA	8	628	20	7.0	Maize	45	0.9	1.1	–	–	–	–	–	–	–	Hoben et al. (2011)
Morris, USA	6	645	32	7.2	Maize	78	–	–	Fallow	0	–	–	78	5.2	3.0	Johnson et al. (2012)
Morris, USA	6	645	32	7.2	Wheat	78	–	–	Fallow	0	–	–	78	4.2	2.8	Johnson et al. (2012)

807 MAT, mean annual temperature; MAP, mean annual precipitation; EF, the N₂O emission factor of applied N.

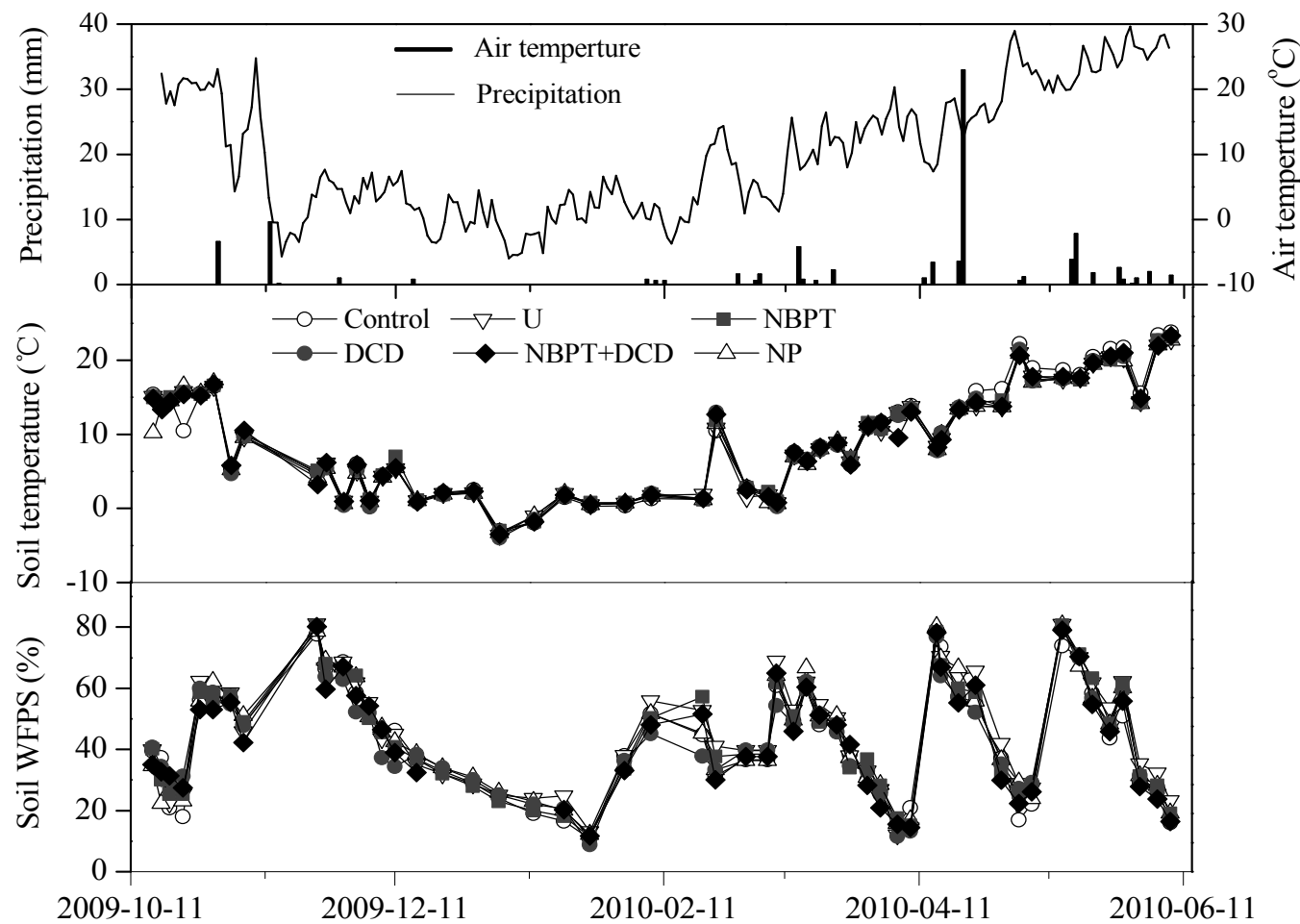
Figure caption

Figure 1. Temporal variation of daily precipitation and air temperature, and mean soil moisture and water-filled pore space (WFPS) at time of N₂O sampling in the control, urea alone (U), urea with NBPT (NBPT), urea with DCD (DCD), urea with NBPT + DCD (NBPT + DCD) and nitrophosphate (NP) treatments during the wheat growth season. The standard errors of soil temperature and moisture were not shown for figure clarity.

Figure 2. Temporal variation of nitrous oxide fluxes in the control, urea alone (U), urea with NBPT (NBPT), urea with DCD (DCD), urea with NBPT + DCD (NBPT + DCD) and nitrophosphate (NP) treatments during the wheat growth season. Flux values are mean values \pm standard errors for three replicates. Arrows indicate date of fertilizer application.

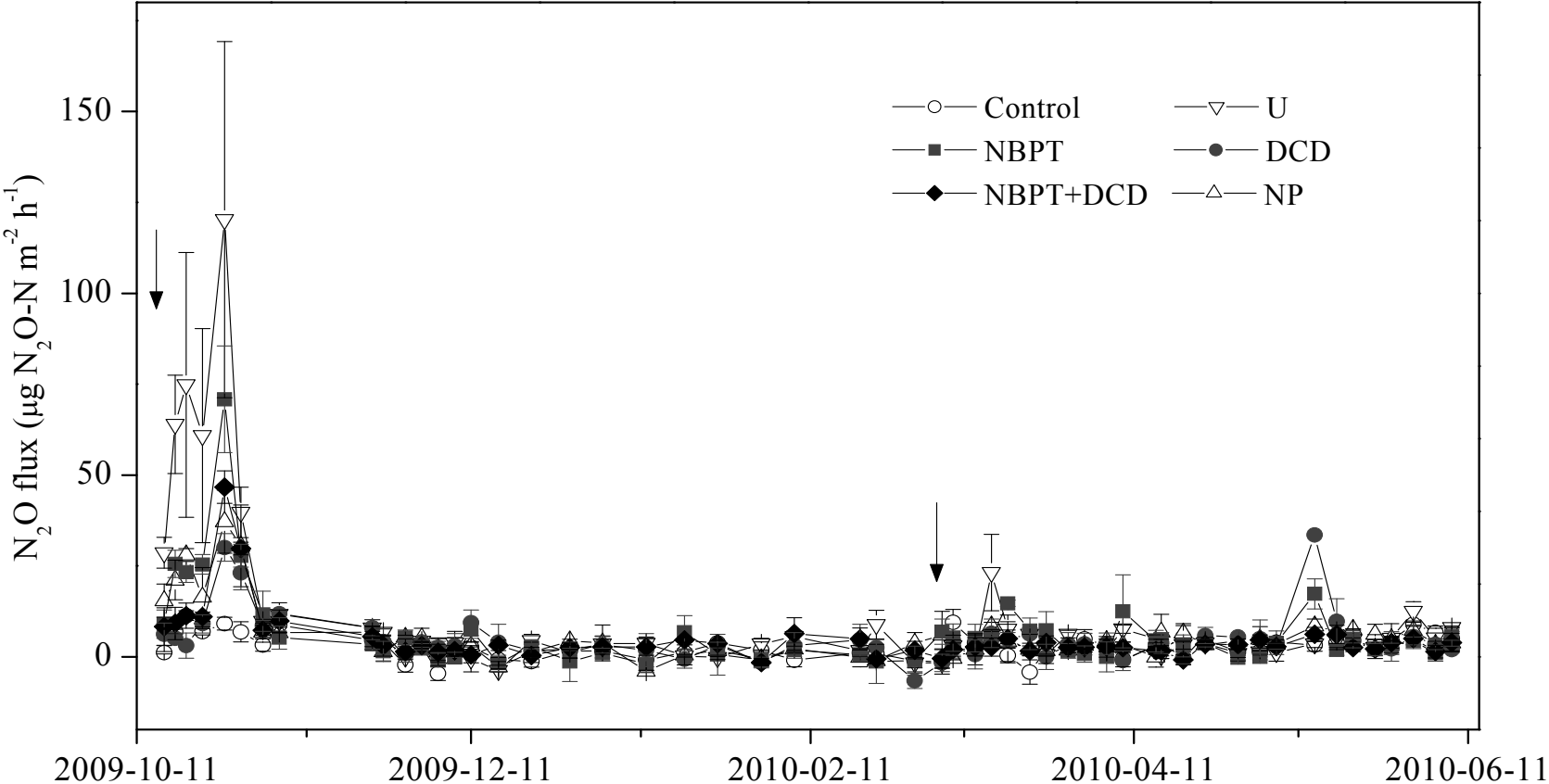
Figure 3. Temporal variation of ammonium and nitrate concentrations in samples from the 0–20 cm depth in the control, urea alone (U), urea with NBPT (NBPT), urea with DCD (DCD), urea with NBPT + DCD (NBPT + DCD) and nitrophosphate (NP) treatments during the wheat growth season. Vertical bars denote the standard error of the means ($n = 3$).

827 **Figure 1**

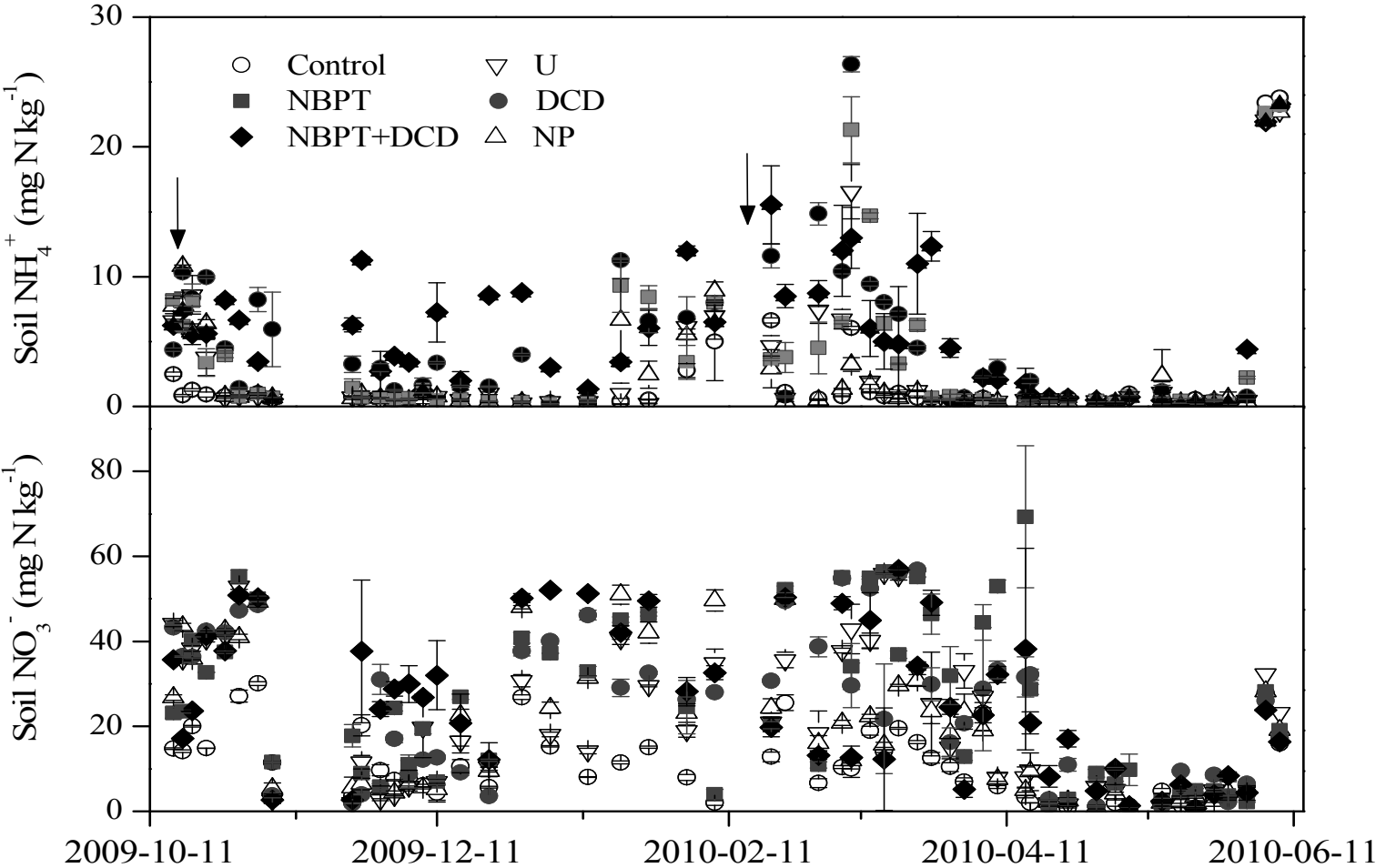


828

829 **Figure 2**



831 **Figure 3**



832