

# Cover letter regarding manuscript Bg-2014-411 “Organic N and P in eutrophic fjord sediments - rates of mineralization and consequences for internal nutrient loading”

Dear editor,

We have carefully considered the comments provided by the two reviewers and have edited the manuscript where appropriate. In the text below we have written our response to the individual reviewer comments. By the end of this document we have attached the revised manuscript with changes highlighted by track changes – here it is indicated which changes were made in response to specific comments.

We hope that with these changes, the revised version of the manuscript can be accepted for publication in Biogeosciences.

Sincerely,

Thomas Valdemarsen

## **RESPONSE TO REVIEWER#1, Susanna Hietanen**

*The authors thank Susanna Hietanen (Reviewer#1) for her constructive criticism of the manuscript. In the text below, the Reviewers original comment is indicated by the headline “REVIEWER COMMENT” and author’s responses are indicated by the headlines “REPLY” and “ACTION”*

### REVIEWER#1, COMMENT 1:

This MS discusses a topical, important and controversial aspect in aquatic ecosystem protection and restoration – the effect internal, independently of external, nutrient loading has on the water quality. While internal nutrient loading has traditionally been seen only as phosphorus release from iron hydroxides under oxygen stress, these authors define it wider, as the release of inorganic nutrients by mineralization of organic matter. To quantify such a release they performed a two-year experiment in which they incubated different types of sediments from different areas of a eutrophied fjord in both oxic and anoxic conditions and followed the fluxes of inorganic nutrients in and out of the sediments, production of ammonium and phosphate in anoxic conditions and the concentrations of total nutrients in the sediments. The same authors have recently published the results of organic carbon mineralization in this same experiment in Marine Ecology Progress Series (503: 41-58; 2014), stating that organic carbon accumulates in sediments and it’s degradation is a slow process that delays the recovery of the water ecosystem. In this discussion paper the authors present data on nitrogen and phosphorus mineralization, concluding that internal nitrogen loading ceases much faster than that of phosphorus, which can be seen as good news for nitrogen limited

systems (although the authors fail to mention e.g. that low N:P ratio favors the growth of cyanobacteria that, in turn, may release plenty of freshly fixed nitrogen to the system).

**REPLY:**

*The same point was addressed by reviewer#2. The reviewers are right that we did not mention that decreasing internal nitrogen loading could lead to stimulation of cyanobacteria, which could potentially buffer the decreasing internal nitrogen loading through N-fixation. This was omitted because we thought it was too speculative. In our experiment we only looked at N coming from the sediment and neglected the external N-loading, which is still the dominating source of N in the studied system. Decreasing internal nutrient loading may therefore not necessarily lead to reduced DIN concentrations as long as external N-loading remains high. Therefore major changes in phytoplankton composition towards increased dominance of cyanobacteria will probably not occur.*

**ACTION:**

*In the revised manuscript we mention that the decreasing internal N-loading and stable internal P-loading could lead to increased dominance of cyanobacteria. However, a major shift in phytoplankton community can only occur in systems where decreased internal nutrient loading results in markedly lower DIN-concentrations, i.e. in systems where (1) internal nutrient sources dominate, or where (2) external N-loading is significantly reduced.*

**REVIEWER#1, COMMENT 2:**

The experiment has been carefully conducted and the paper is well written, with good quality tables and figures. The authors show that they are quite well aware of the shortcomings of the experimental setup used and discuss the results mainly accordingly. While the paper does not present especially novel ideas, it strengthens theories on shallow-water ecosystem recovery processes. I have very few specific comments: Page 15114 row 5: The selected habitat types are said to have covered the whole fjord (100%) – still, for example the highest porosity found in these samples was 0.8 which is surprisingly low – are there no high-organic muddy areas in this estuary, with porosities well over 0.9?

**REPLY:**

*The 3 stations located in the outer fjord (St 6-8) were chosen to represent the most organic rich, muddy areas in the fjord. A previous survey of sediment characteristics in Odense fjord covering >100 stations, showed that the most fine grained sediments had porosities around 0.8. The absence of extremely muddy sites in Odense Fjord is probably due to wind driven resuspension events, which may affect even the deep parts of this shallow system (down to 10 m depth) and prevent accumulation of the finest sediment fractions.*

**ACTION:**

*This comment will not lead to changes in the manuscript.*

REVIEWER#1, COMMENT2b:

Page 15114 row 20 on; removing macrofauna from naturally permanently oxic system makes the interpretation and generalization of the results dubious. The authors mention this shortcoming briefly in discussion as a possible source of error, and probably this is the only way to study slow processes in laboratory conditions. However, the role of macrofauna in benthic mineralisation in shallow, oxic waterbodies is very large and excluding them from the experimental setup a very drastic manipulation of the system. This might merit a bit longer discussion about the reliability of the results from this point of view.

REPLY:

*A similar point was raised by Reviewer#2. We agree with the reviewers that removing macrofauna was a major manipulation, which undoubtedly changed/stimulated mineralization rates and temporal patterns. However, the main goal of the experiment was to assess how much nutrients could be released from the organic matter that had accumulated in the sediments during eutrophication, and this could not have been accomplished if macrofauna had been included in the study. To keep macrofauna alive in a 2 year laboratory experiment is virtually impossible. Macrofauna would need a stable food source (addition of some source of organic matter, including organic N and P), and then we would not know how much of the generated nutrients were coming from the sediment and added organic matter, respectively. Furthermore, most benthic infauna has a life span of less than 2 years, meaning that they would eventually die out in the lab experiment since there was no recruitment. Finally, the infauna composition is highly variable at the different stations (i.e. some stations have large bioturbators and others not), and this variability would have disturbed the major patterns in sediment nutrient generation at the different stations.*

ACTION:

*In the revised manuscript we will strengthen our discussion of the consequences of removing infauna at the beginning of the experiment. We will emphasize that our estimates of nutrient regeneration are probably conservative, since macrofauna would have stimulated the remineralization of organic matter and nutrients.*

REVIEWER#1, COMMENT 3:

Similarly, some mention could be made about the annual oxygen and temperature conditions these sediments might be exposed to in contrast to fully oxic, 15 degree incubation used in the experiment.

REPLY:

*A similar comment was made by Reviewer#2. The incubation temperature of 15°C was chosen to reflect the average annual temperature in Odense Fjord. Hypoxia/anoxia is no longer occurring in Odense Fjord, so the incubation condition of fully oxygenated water is fully representative for Odense Fjord. It is true that annual temperature variations occur in the system and that this will influence microbial reaction rates. However, the main goal of the experiment was to provide an estimate of the total nutrient release from Odense Fjord sediments and not to simulate seasonal changes in microbial reaction rates. We therefore*

*choose to omit any temperature driven variation in microbial reaction rates, which would complicate the interpretation of results.*

**ACTION:**

*The influence of temperature variations were briefly mentioned in the original manuscript (Page 15129, Line 23). In the revised manuscript we have strengthened the discussion of consequences of constant contra variable temperature for internal nutrient loading.*

**REVIEWER#1, COMMENT 4:**

Page 15117 Jar experiments. These experiments were run fully anoxic to prevent oxidation of end products, and homogenized probably for minimizing variation between samples. Both cutting the contact with oxygen for the surface sediments and homogenizing across redox zones heavily changes conditions compared to core incubations. These effects are not discussed at all. Was oxygen penetration to the sediments so shallow that it merited the anoxic incubation in the top layer?

**REPLY:**

*It is true that the jar technique may underestimate the actual mineralization rates in oxidized sediments where O<sub>2</sub> sensitive organic matter is degraded. However, most coastal and estuarine sediments are predominantly anoxic beneath a shallow (1-3 mm deep) oxidized sediment layer, i.e. conditions similar to the beginning of this experiment. We (and probably the Reviewer) expected the sediments to become more and more oxic as the experiment progressed and labile organic matter was depleted, but this did not occur. This was for instance shown by hydrogen sulfide (data not shown), which was present at all sediment depths at all stations throughout the experiment, suggesting that sediments remained predominantly anoxic except for the upper few mm. Anoxic sediment incubations was therefore an appropriate technique to measure microbial reaction rates throughout the experiment.*

**ACTION:**

*A paragraph has been added where it is mentioned that the sediments remained mostly anoxic throughout the experiment and that closed anoxic sediment incubations therefore was an appropriate technique to measure microbial reaction rates throughout the experiment.*

**REVIEWER#1, COMMENT 5:**

Page 15117 row 25; you probably mean that the concentrations of Fe (III) were compared at the beginning (initial) and end (final) of the experiments using pairwise t-tests? the results are really surprising – according to table 3 there are no significant differences, despite increase or decrease by 2.5-3.4 times.

**REPLY:**

*True – we compared total pools of Fe(III) at the beginning and end by pairwise t-tests. We were also very surprised that 2 year incubation with no addition of new organic matter did*

*not lead to significant changes in Fe(III)-pools, owing to large spatial heterogeneity between sediment cores. Unfortunately, intercore variability is one of the drawbacks of using non-manipulated sediment cores.*

**ACTION:**

*We rephrased the lines (Page 15117 row 25-26) according to the reviewer's suggestion.*

**REVIEWER#1, COMMENT 6:**

Page 15120 rows 18-25; there seems to be some words missing and some in excess in the text (check grammar).

**ACTION:**

*The indicated section was rephrased to: "Surface NH<sub>4</sub> production decreased rapidly over time in sediments from shallow locations in the inner and outer fjord, by 96% of initial rates on St 1 and by 61–82% on St 2–5. The surface NH<sub>4</sub><sup>+</sup> production in the sediments sampled in the deep outer basin (St 6–8) decreased by 8–67% during the experiment. NH<sub>4</sub><sup>+</sup> production at 4–6 cm depth was initially 18–60 nmol cm<sup>-3</sup> d<sup>-1</sup> on all stations and temporal changes were also observed in this layer, especially in shallow silty sediments from the inner basin where NH<sub>4</sub><sup>+</sup> production decreased by 75–96% to 1.4–12 nmol cm<sup>-3</sup> d<sup>-1</sup> by the end (Fig. 3)."*

**REVIEWER#1, COMMENT 7:**

Page 15123 general comment on all the discussion about the sandy site: Sandy sediments are usually permeable, which means they operate by advection, not by diffusion. Enclosing sand in a core, out of reach of advective flow, changes conditions in the porewater dramatically. Recent research has indicated sandy sediments as areas of extremely high mineralization despite the low organic content. The authors mention that the sandy sediment came from area affected by with wind driven waves, deeply burying macrofauna and intense microphytobenthic production. None of this could be reproduced in the experimental setup, which questions the interpretation of the results on this site.

**REPLY:**

*A similar comment was made by reviewer#2. We are aware that shallow sandy sediments may be highly reactive due to a very dynamic environment. As mentioned in previous comments the main purpose of the experiment was not to simulate in situ degradation rates, but to measure how much nutrients could actually be made available by degradation and efflux to the overlying water column in different sediment types. In this respect the omission of waves, light and microphytobenthos is critical to be able to compare the reactivity of sedimentary organic matter and nutrients at different sites.*

**ACTION:**

*In the revised manuscript we will underline that shallow sandy sediments are highly dynamic environments supporting a rapid cycling of nutrients and organic matter. We will also emphasize that this might have lead to a bias in our estimates of nutrient release.*

REVIEWER#1, COMMENT 8:

Page 15124 row 25 typos- subscripts instead of superscripts in Mkg.

*ACTION:*

*This was corrected.*

REVIEWER#1, COMMENT 9:

Page 15128 Ecological implications – the first half of the chapter is very repetitive and could be pruned heavily.

*ACTION:*

*We will shorten this paragraph.*

REVIEWER#1, COMMENT 10:

Page 15138 Table 2 should show “average TN:TP ratios” but it does not.

*ACTION:*

*TN:TP ratios were added to Table 2.*

REVIEWER#1, COMMENT 11:

Page 15146 Figure 5 DIN= NH<sub>4</sub><sup>+</sup> + NO<sub>x</sub><sup>-</sup> in figure but DIN= NH<sub>4</sub><sup>+</sup> + NO<sub>3</sub><sup>-</sup> in legend.

*ACTION:*

*The text in the figure legend was changed to “DIN = NH<sub>4</sub><sup>+</sup> + NO<sub>x</sub><sup>-</sup>”*

REVIEWER#1, COMMENT 12:

Page 15147 Figure 6 I am not sure this figure is really needed, although it is nice.

*REPLY:*

*Reviewer#2 had the same comment. The temporal trends in NH<sub>4</sub><sup>+</sup> and NO<sub>x</sub><sup>-</sup> are described sufficiently in the text.*

*ACTION:*

*The figure was deleted.*

## **RESPONSE TO REVIEWER#2, Anonymous**

*The authors thank the reviewer for his/her constructive criticism of the manuscript. In the text below, the Reviewers original comments are indicated by the headline “REVIEWER COMMENT” and author responses are indicated by the headlines “REPLY” and “ACTION”*

### **REVIEWER#2, COMMENT 1:**

The manuscript discusses the importance of internal nutrient loading in the recovery of eutrophic estuaries by examining the mineralisation of organic nitrogen (ON) and phosphorus (OP) buried in the sediment. The magnitude of the internal loading of ON and OP was investigated in a two year experiment spanning a number of different sediment types found within the Odense Fjord. Cores were kept oxic throughout the experimental phase in which sediment porewater profiles, dissolved nutrient fluxes and anaerobic jar experiments (anaerobic mineralisation rates) were periodically measured over the two year period. The authors concluded that internal nitrogen loading is of minor importance (6% external N) and DIN fluxes stabilised after 50-200 days, alternatively internal phosphorus loading is potentially a source of P (36% external P) and PO<sub>4</sub><sup>3-</sup> fluxes throughout the two year period were variable and P efflux could be sustained over years which may be important for the management of P.

The principles within the manuscript are not necessarily novel, however, the authors do provide further understanding of internal nutrient loading which is an important measure in the management of these systems and the potential for a delayed response after reducing external inputs. The manuscript is generally well written some areas could be shortened. There are, however, some limitations in the experimental design, possibly unavoidable due to the required length of the experiment but they do raise the question as to how relevant long term laboratory experiments are to the natural environment. Some of these limitations have been discussed briefly in the discussion and more thoroughly in the partner paper on carbon mineralisation published in the Marine Ecology Progress Series (Vol. 503: 41–58, 2014) however these need to be addressed more thoroughly.

### **REPLY:**

*Both reviewers indicate that sources of errors and uncertainties regarding experimental setup and estimates of internal nutrient loading should be emphasized more clearly in the text. We agree with both reviewers that variation in temperature, macrofauna and hydrodynamics/advection will impact nutrient regeneration and hence the magnitude of internal nutrient loading compared to our estimates.*

### **ACTION:**

*Where appropriate we will emphasize sources of errors and consequences of omitting different environmental variables in the experiment set-up. See our reply to specific comments for more details.*

REVIEWER#2, COMMENT 2:

Page 15114 Line 1 – Station 5 represents sandy sediment which has been collected in a closed core. Measuring fluxes in sediment within closed cores is based on diffusion principles, however in the natural environment processes within sandy sediments are not only governed by diffusion but also advective flow. This will likely alter the estimate of nutrient effluxes from the sediment depending on residence time of the porewaters etc. The difference of this site to the others is briefly touched on in the discussion (15124 – line 14) however it is not discussed in terms of an experimental limitation and how this influences the interpretation of the results at this site and requires further comment.

REPLY:

*A similar comment was made by reviewer#1. We are aware that shallow sandy sediments may be highly reactive due to a very dynamic environment. However, the main purpose of the experiment was not to simulate in situ degradation rates, but to measure how much nutrients could actually be made available by degradation and efflux to the overlying water column in different sediment types. In this respect the omission of waves, light and microphytobenthos is critical to be able to compare the reactivity of sedimentary organic matter and nutrients at different sites.*

ACTION:

*In the revised manuscript we will underline that shallow sandy sediments are highly dynamic environments supporting a rapid cycling of nutrients and organic matter. We will also emphasize that this might have lead to a bias in our estimates of nutrient release.*

REVIEWER#2, COMMENT 3:

Page 15114 Line 20 – What was the density of the macrofauna at each of the sites? In the literature macrofauna play an important role in nutrient cycling in the Odense Fjord turning over large volumes of sediment annually if this also applies to these sites excluding them from your experiments (even though it would be challenging to include them in a long term experiment such as this) could drastically change your estimates ON and OP released from the sediment. This also changes the redox conditions at the sediment surface where burrowing macrofauna can increase the depth of oxygen penetration and nutrient transport into the sediment and in inturn potentially influence your mineralisation rates. I am uncertain here how relevant to natural conditions your estimates of internal nutrient loadings are having excluded all macrofauna from your experiments. This requires further discussion.

REPLY:

*A similar point was raised by Reviewer#1. We agree with the reviewers that removing macrofauna was a major manipulation, which undoubtedly changed mineralization rates and temporal patterns. However, the main goal of the experiment was to assess how much nutrients could be released from the organic matter that had accumulated in the sediments during eutrophication, and this could not have been accomplished if macrofauna had been included in the study. To keep macrofauna alive in a 2 year laboratory experiment is virtually impossible. Macrofauna would need a stable food source (addition of some source of organic matter, including organic N and P), and then we would*



*not know how much of the generated nutrients were coming from the sediment and added organic matter, respectively. Furthermore, most benthic infauna has a life span of less than 2 years, meaning that they would eventually die out in the lab experiment since there was no recruitment. Finally, the infauna composition is highly variable at the different stations (i.e. some stations have large bioturbators and others not), and this variability would have disturbed the major patterns in sediment nutrient generation at the different stations.*

**ACTION:**

*In the revised manuscript we will strengthen our discussion of the consequences of removing infauna at the beginning of the experiment. We will emphasize that our estimates of nutrient regeneration are probably conservative, since macrofauna would have stimulated the remineralization of organic matter and nutrients.*

**REVIEWER#2, COMMENT 4:**

Page 15114 Line 11/25 - The authors state that at the time of sampling the in situ temperature was 10 – 12°C, however the experimental conditions were set to 15°C I am uncertain why this temperature was chosen. Furthermore the authors then extrapolate annual ON and OP release under these conditions where on an annual cycle temperatures in the Odense Fjord range from 3 - 19°C. Bacterial populations could also vary throughout this time based on temperature differences. The influence of temperature on mineralisation rates should be something that is considered and discussed further particularly when temperature regulation is quite easy to manipulate in an experimental setup.

**REPLY:**

*A similar comment was made by Reviewer#1. The incubation temperature of 15°C was chosen to reflect the average annual temperature in Odense Fjord. It is true that annual temperature variations occur in the system and that this will influence microbial reaction rates. However, the main goal of the experiment was to provide an estimate of the total potential nutrient release from Odense Fjord sediments and not to simulate seasonal changes in microbial reaction rates. We therefore choose to omit any temperature driven variation in microbial reaction rates, which would complicate the interpretation of results.*

**ACTION:**

*The influence of temperature variations were briefly mentioned in the original manuscript (Page 15129, Line 23). In the revised manuscript we have strengthened the discussion of consequences of constant versus variable temperature for internal nutrient loading.*

**REVIEWER#2, COMMENT 5:**

Page 15117 Line 6 – Was there a control to compare the rate of aerobic and anaerobic mineralisation to confirm that anaerobic mineralisation measurements were representative of all mineralisation within the fjord? This would not be an issue for deep anaerobic sediments but in aerobic surface sediments an aerobic mineralisation measurement

should have been considered.

*REPLY:*

*We have evidence (hydrogen sulfide concentrations in porewater) suggesting that sediments from all stations were predominantly anoxic throughout the experiment, indicating that anaerobic degradation was most important for organic matter degradation in this experiment. Furthermore, results obtained in anoxic jar experiments are evaluated against flux measurements, which include both anaerobic and aerobic processes. Hence both anaerobic and aerobic processes are actually accounted for in the manuscript.*

*ACTION:*

*A paragraph was added where it is mentioned that sediments remained mostly anoxic throughout the experiment and that anaerobic degradation processes were most important in the present experiment.*

REVIEWER#2, COMMENT 6:

Page 15126 Line 15 – The authors mention that the missing  $\text{NH}_4^+$  could be lost due to nitrification-denitrification coupling. I do agree the missing  $\text{NH}_4^+$  in the oxic fluxes is most likely accounted for by nitrification however in Figure 6 it does not appear that nitrification and denitrification are strongly coupled with most of the  $\text{NO}_3^-$  released from the sediment after ~20 days. Possibly revise this statement.

*REPLY:*

*We are not sure we understand the reviewer's argument. We measure much higher  $\text{NH}_4^+$  production in the sediments than is released as DIN-efflux or accumulates as  $\text{NH}_4^+$  in porewater. We argue that the deficit between  $\text{NH}_4^+$  production and DIN efflux must be caused by closely coupled nitrification-denitrification (i.e. we do not see the produced  $\text{NO}_3^-$  as it is consumed as fast as it is produced). This is the only explanation for the observed deficit and we see no reason to revise this statement.*

*ACTION:*

*This comment will not lead to major changes in the text. However, Fig. 6 will be deleted from the revised manuscript as requested by both reviewers.*

REVIEWER#2, COMMENT 7:

Page 15128 Line 3 – The authors state in the discussion that the sandy sediments are the most important for the total fjord release of N (39%) I would note that this estimate probably has the most uncertainty out of all sites due to the absence of advective flow in these estimates. This statement is also made about P release (Line 16).

*ACTION:*

*As also mentioned for other comments, we will emphasize in the revised manuscript that our estimates of nutrient release should be extrapolated with caution because we have omitted important factors in our experimental setup (temperature variation, macrofauna*

*and hydrodynamics/advection). The error by omitting these factors is probably largest for shallow sandy sediments, which are impacted by e.g. the largest temperature fluctuations and more intense hydrodynamics compared to the other environments.*

REVIEWER#2, COMMENT 8:

Page 15129 Line 17 – As mentioned by the previous referee (S. Hietanen) – the authors need to explore further how a change in the N:P ratio from high internal P loading could potentially result in N<sub>2</sub> fixers dominating the system and be a long term management issue.

*REPLY:*

*The same point was addressed by reviewer#1. The reviewers are right that we did not mention that decreasing internal nitrogen loading could lead to stimulation of cyanobacteria, which could potentially buffer the decreasing internal nitrogen loading through N-fixation. This was omitted because we thought it was too speculative. In our experiment we only looked at N coming from the sediment and neglected the external N-loading, which is still the dominating source of N in the studied system. Decreasing internal nutrient loading may therefore not necessarily lead to reduced DIN concentrations as long as external N-loading remains high. Therefore major changes in phytoplankton composition towards increased dominance of cyanobacteria will probably not occur.*

*ACTION:*

*In the revised manuscript we mention that the decreasing internal N-loading and stable internal P-loading could lead to increased dominance of cyanobacteria. However, a major shift in phytoplankton community can only occur in systems where decreased internal nutrient loading results in markedly lower DIN-concentrations, i.e. in systems where (1) internal nutrient sources dominate, or where (2) external N-loading is significantly reduced.*

REVIEWER#2, COMMENT 9:

Page 15120 Line 6 – This line is not essential. Why Fe(III) was measured is already mentioned in the introduction.

*ACTION:*

*Line was deleted.*

REVIEWER#2, COMMENT 10:

Page 15124 Line 25 subscript on numbers instead of superscript

*ACTION:*

*This was corrected.*

REVIEWER#2, COMMENT 11:

Page 15138 - Table 2 is missing the average TP:TN ratios

*ACTION:*

*TN:TP ratios were added to Table 2.*

REVIEWER#2, COMMENT 12:

Page 15147 - Figure 6 is not essential it is explained well in results.

*REPLY:*

*Reviewer#2 had the same comment. The temporal trends in  $\text{NH}_4^+$  and  $\text{NO}_x^-$  are described sufficiently in the text.*

*ACTION:*

*The figure was deleted.*

REVIEWER#2, COMMENT 13:

Page 15125 Lines 1-2 are not necessary this is clear in the previous sentence and was set out in the introduction

*ACTION:*

*The indicated section was deleted.*

REVIEWER#2, COMMENT 14:

Page 15141 - Table 5 should ON and OP degradation or any of the other measurements here be time integrated? the units are in  $\text{mmol m}^{-2}$

*REPLY:*

*Values in table 5 are time integrated – hence the unit  $\text{mol m}^{-2}$ . This is clearly indicated in the table legend: “Total  $\text{NH}_4^+$ ,  $\text{NO}_x^-$  and  $\text{PO}_4^{3-}$  effluxes were calculated by time integration of effluxes over the entire experimental period.”*

REVIEWER#2, COMMENT 15:

Page 15127 Line 13 – Normal text for N-1 should be superscript

*ACTION:*

*This was corrected.*

1 **Organic N and P in eutrophic fjord sediments - rates of mineralization**  
2 **and consequences for internal nutrient loading**

3  
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12 **Key words:** nutrient release, fluxes, reactivity, ammonium production, phosphate production,  
13 mineralization, oligotrophication, recovery, adsorption

14 **Abstract**

15 Nutrient release from the sediments in shallow eutrophic estuaries may counteract reductions of the  
16 external nutrient load and prevent or prolong ecosystem recovery. The magnitude and temporal  
17 dynamics of this potential source, termed internal nutrient loading, is poorly understood. We  
18 quantified the internal nutrient loading driven by microbial mineralization of accumulated organic  
19 N (ON) and P (OP) in sediments from a shallow eutrophic estuary (Odense Fjord, Denmark).  
20 Sediments were collected from 8 stations within the system and nutrient production and effluxes  
21 were measured over a period of ~2 years. DIN effluxes were high initially but quickly faded to low  
22 and stable levels after 50-200 d, whereas  $\text{PO}_4^{3-}$  effluxes were highly variable in the different  
23 sediments. Mineralization patterns suggested that internal N-loading would quickly (<200 days)  
24 fade to insignificant levels whereas internal  $\text{PO}_4^{3-}$  loading could be sustained for extended time  
25 (years). When results from all stations were combined, internal N-loading and P-loading from the  
26 fjord bottom was up to  $121 \cdot 10^3 \text{ kg N yr}^{-1}$  ( $20 \text{ kg N ha}^{-1} \text{ yr}^{-1}$ ) and  $22 \cdot 10^3 \text{ kg P yr}^{-1}$  ( $3.6 \text{ kg P ha}^{-1} \text{ yr}^{-1}$ )  
27 <sup>1</sup>) corresponding to 6% (N) and 36% (P) of the external nutrient loading to the system. We conclude  
28 that the internal N-loading resulting from degradation of accumulated ON is low in shallow  
29 eutrophic estuaries, whereas microbial mineralization of accumulated OP is a potential source of P.  
30 Overall it appears that in N-limited eutrophic systems, internal nutrient resulting from  
31 mineralization of ON and OP in sediments is of minor importance.

32       **1. Introduction**

33       The nutrient loading of coastal ecosystems is often divided into internal and external sources, i.e.  
34       release from sediments resulting from organic N (ON) and P (OP) mineralization, and natural and  
35       anthropogenic supplies via the water shed and atmospheric deposition, respectively. The external  
36       nutrient loading can be quantified by summing up the external sources (e.g. Petersen et al. 2009). It  
37       is difficult, however, to use a mass balance approach to obtain reliable estimates of internal nutrient  
38       loading, since release from sediments and export to adjacent water bodies are difficult to quantify  
39       with sufficient temporal and spatial precision in large and dynamic estuaries with extensive spatial  
40       variability and open boundaries.

41               To complicate matters more, the internal nutrient loading can be divided into two  
42       fractions with different temporal dynamics. The first is rapid nutrient release from mineralization of  
43       fresh and newly deposited labile organic material, and the second is slow and continued nutrient  
44       release from mineralization of buried organic material with lower reactivity. High turnover of labile  
45       ON and OP deposited at the sediment-water interface ensures a rapid recycling of inorganic  
46       nutrients to the water column (Kelly & Nixon 1984; Valdemarsen et al. 2009). The primary  
47       productivity in many shallow estuaries is therefore partially controlled by nutrients released from  
48       the sediments (Cowan & Boynton 1996; Fullweiler et al. 2010; Mortazavi et al. 2012; Bukaveckas  
49       & Isenberg 2013). The contribution from mineralization of low reactivity and often deeply buried  
50       ON and OP to total sediment nutrient release, however, remains largely unknown. Nutrient release  
51       reported in most published studies is dominated by the nutrients generated by labile ON and OP  
52       mineralization due to the short time-scale applied for measurements. It is nonetheless important to  
53       obtain reliable estimates of the nutrient generation and efflux resulting from mineralization of low  
54       reactivity ON and OP. In many instances the recovery of eutrophic ecosystems after reductions of  
55       the external nutrient loading does not occur or only occurs after considerable delay (Kronvang et al.

56 2005). This may be caused by substantial release of nutrients, which have accumulated to high  
57 concentrations over time in the sediments exposed to eutrophication (Pitkanen et al. 2001;  
58 Carstensen et al. 2006). Such delayed nutrient release is thought to counteract reductions in the  
59 external nutrient load and cause delayed recovery.

60 Determining the magnitude and temporal dynamics of the internal nutrient loading  
61 originating from ON and OP accumulated in sediments requires detailed biogeochemical studies.  
62 Organic matter degradation in sediments follow exponential decay kinetics (Westrich & Berner,  
63 1984; Burdige 1991; Valdemarsen et al. 2014) and inorganic nutrient production from ON and OP  
64 is therefore expected to decrease exponentially with time. Not all produced inorganic nutrients will  
65 result in internal nutrient loading, however, since chemical and biological processes within  
66 sediments lead to nutrient retention or transformation before efflux to the overlying water.  $\text{NH}_4^+$ , for  
67 instance, can be adsorbed to the sediment matrix (Mackin and Aller 1984), assimilated by microbes  
68 or benthic microalgae or microbially transformed to other nitrogenous compounds (Christensen et  
69 al. 2000; Tyler & McGlathery 2003; Hulth et al. 2005). Coupled nitrification-denitrification in the  
70 oxic-anoxic transition of surface sediments, whereby  $\text{NH}_4^+$  is converted to inert  $\text{N}_2$ -gas, is for  
71 instance an ecologically important process which reduces the amount of bioavailable N (Seitzinger  
72 1988; Burgin & Hamilton 2007). Due to adsorption and denitrification, the efflux of dissolved  
73 inorganic nitrogen ( $\text{DIN} = \text{NH}_4^+ + \text{NO}_3^- + \text{NO}_2^-$ ) is generally much lower than anticipated from total  
74 ON mineralization in the sediment (Mackin and Swider 1989). As for  $\text{NH}_4^+$ ,  $\text{PO}_4^{3-}$  may adsorb to  
75 the sediment matrix; mainly to Fe-minerals in oxidized surface sediment (Sundby et al. 1992).  $\text{PO}_4^{3-}$   
76 efflux is therefore generally low in marine sediments lined with an oxic surface layer (Sundby et al.  
77 1992; Jensen et al. 1995; Viktorsson et al. 2013).

78 In this study an experimental approach was used to determine the internal nutrient  
79 loading resulting from long-term mineralization of accumulated ON and OP in various sediment



80 types of a large shallow, eutrophic estuary (Odense Fjord, Denmark). The goals of the study were  
81 two-fold; (1) to quantify the magnitude and temporal dynamics of internal nutrient loading resulting  
82 from mineralization of ON and OP accumulated in sediments and (2) to evaluate the role of internal  
83 nutrient loading for the recovery of eutrophic ecosystems. Sediment cores were collected from  
84 various locations representing the dominating sediment types and environments in the estuary.  
85 These were maintained in experiments lasting ~2 years, during which the mineralization of ON and  
86 OP and resulting effluxes of inorganic nutrients were measured with high spatial and temporal  
87 resolution. By comparing total inorganic nutrient production to effluxes, the fate of inorganic  
88 nutrients was elucidated. The total internal nutrient loading of the entire system was estimated  
89 based on the measured nutrient effluxes and the areal distribution of dominating sediment types.  
90 Finally, the importance of internal nutrient loading in shallow eutrophic ecosystems is evaluated.

91

## 92 **2. Materials and methods**

### 93 *2.1 Study area*

94 Odense Fjord is a shallow eutrophic estuary located on the island of Fyn, Denmark. It is divided  
95 into a 16 km<sup>2</sup> shallow inner basin and a 45 km<sup>2</sup> deeper outer basin, with average depths of 0.8 and  
96 2.7 m, respectively (Fig. 1). The fjord is connected to Kattegat through a narrow opening in the  
97 northeast. The main external nutrient source to Odense Fjord is Odense River, which has a  
98 catchment area of 1095 km<sup>2</sup>, consisting mainly of farmland and urban areas (Petersen et al. 2009).  
99 Odense Fjord was critically eutrophic in the past due to high external nutrient loading exceeding  
100 3000\*10<sup>3</sup> kg N y<sup>-1</sup> and 300\*10<sup>3</sup> kg P y<sup>-1</sup> before 1990 (Petersen et al. 2009). The massive nutrient  
101 loading caused extensive problems with high pelagic primary production, low water transparency,  
102 hypoxic events and blooms of opportunistic macroalgae. Implementation of several water action  
103 plans has reduced the external nutrient loading considerably to current levels of about 2000\*10<sup>3</sup> kg

104 N y<sup>-1</sup> and 60\*10<sup>3</sup> kg P y<sup>-1</sup>. This has improved the ecological quality of the system, since hypoxia is  
105 now rare and levels of opportunistic macroalgae have decreased. Nonetheless, excessive nutrient  
106 levels and high primary production are still a problem in Odense Fjord, which may be due to high  
107 and sustaining internal nutrient loading.

108

### 109 *2.2 Sampling of sediment and water*

110 Intact sediment cores were collected on 8 stations from 4 habitat types in Odense Fjord during  
111 October and November 2009 (Fig. 1). The stations were chosen to cover all major sediment types in  
112 the fjord; 3 stations (St 1-3) represented shallow silty sediments in the inner fjord, St 4 and 5  
113 represented shallow (< 1 m) silty and sandy sediments in the outer fjord, respectively, and finally, 3  
114 stations (St 6-8) represented deep (2-6 m) silty sediments in the outer fjord. A detailed survey of  
115 sediment characteristics conducted in 2009 (partially presented in Valdemarsen et al. 2014)  
116 revealed that the four selected habitat types (shallow silty inner fjord, shallow silty outer fjord,  
117 shallow sandy outer fjord and deep silty outer fjord) represented 21, 11, 29 and 39% of the fjord  
118 area, respectively. Fifteen sediment cores were sampled from each station with 30 cm long, 8 cm  
119 internal diameter Plexiglas core liners. The shallow stations (St 1-5) were sampled from a dinghy  
120 using a hand operated coring device. Cores from the deeper stations (St 6-8) were subsampled from  
121 a 'HAPS' box corer on board a larger vessel ("Liv II", Danish Nature Agency). Water temperatures  
122 were 10-12°C at the time of sampling.

123                 Seawater used for the experiment was collected at Kerteminde Harbor at various times  
124 during 2009-2011. The seawater was GF/C-filtered and adjusted to the appropriate salinity (10 or  
125 20) before it was used for experiments.

### 126 *2.3 Experimental setup*

127 Sediment cores were pre-treated before the experiment to assure that they had equal sediment height  
128 and were free of macrofauna. The sediment cores were adjusted to 20 cm depth by removing the  
129 bottom stopper and carefully removing excess sediment from below. After reinserting the bottom  
130 stopper, the overlying water was purged with N<sub>2</sub> for 30 min to induce anoxia and the top stopper  
131 was reinserted. Asphyxiated macrofauna was removed from the sediment surface after ~48 h in  
132 darkness.

133 The pre-treatment was completed 2-4 days after sampling and sediment cores were  
134 then transferred to the experimental setup consisting of eight ~70 L water tanks located in a  
135 temperature controlled room at 15°C. The incubation temperature of 15°C approximately  
136 corresponds to the average annual water temperature in Odense Fjord. Each tank contained all  
137 sediment cores from one station, and was filled with filtered seawater with salinity 10 for St1-3 and  
138 salinity 20 for St 4-8, corresponding to the average salinity in the inner and outer basins of Odense  
139 Fjord (Fyns Amt, 2006). The water reservoir in each tank was vigorously mixed and aerated by air  
140 pumps, and kept at a level 0.5 cm above the upper rim of the open core liners to assure mixing of  
141 the headspace. The tanks were kept in darkness and about 1/3 of the water was renewed with fresh  
142 seawater every 2 weeks.

143 The sediment cores were maintained in this setup for the entire experiment, which  
144 lasted 589-635 days, depending on station. The time when cores were first transferred to the  
145 incubation tanks is referred to as  $t = 0$ . At selected times, 3 random sediment cores from each  
146 station were temporarily removed for flux measurements, and at other times 3 sediment cores were  
147 removed permanently for porewater and solid phase analysis as well as anoxic sediment incubations  
148 (see detailed sections below).

149

150 *2.4 Flux measurements*

Kommentar [TBV1]: Line was added in response to REVIEWER#2, COMMENT 4

151 The net exchange of nutrients (DIN and  $\text{PO}_4^{3-}$ ) between sediment and water was determined in flux  
152 experiments with 3 random sediment cores from each station. Flux experiments were conducted  
153 weekly during the first 30 days, monthly until day 180 and every 2-3 months to the end. One day  
154 prior to flux measurements, the inside headspace wall of the cores designated for flux  
155 measurements were cleaned with a Q-tip to avoid biased flux measurements resulting from bacterial  
156 biofilms on the inner surface of core liners (Valdemarsen and Kristensen 2005). These cores were  
157 removed from the incubation tanks the next day, equipped with 4 cm long magnetic stirring bars a  
158 few cm above the sediment surface and placed around a central magnet rotating at 60 rpm. Initial  
159 water samples were taken from all cores, before they were closed with rubber stoppers. The cores  
160 were incubated in darkness for 4 hours initially and up to 24 hours at the end of the experiment,  
161 before the rubber stoppers were removed and final water samples were taken. Nutrient samples  
162 were stored frozen ( $-20^\circ\text{C}$ ) until analyzed for  $\text{NH}_4^+$ ,  $\text{NO}_x^-$  ( $\text{NO}_3^- + \text{NO}_2^-$ ) and  $\text{PO}_4^{3-}$  on a Lachat  
163 Quickchem 8500 Flow injection Analyzer.

164

### 165 *2.5 Core sectioning*

166 Three sediment cores from each station were sectioned into 2 cm intervals to 16 cm depth at various  
167 times (after 1 day and 1, 7-8, 16-17 and 20-21 months). Core sectioning and subsequent sediment  
168 and porewater handling was done inside a  $\text{N}_2$ -filled glovebag. Individual sediment slices were  
169 homogenized and porewater for nutrient analysis was obtained after centrifugation of sediment  
170 subsamples in double centrifuge tubes (10 min,  $\sim 500$  g) and GF/C-filtration. Samples for  $\text{NH}_4^+$  and  
171  $\text{PO}_4^{3-}$  were stored frozen ( $-20^\circ\text{C}$ ) until analysis as described above.

172 Sediment characteristics were determined on subsamples from every depth interval  
173 during the core sectioning on day 1. Grain size composition, loss on ignition (LOI), total organic C  
174 (TOC) content, density and porosity was determined as described in Valdemarsen et al. (2014).

175 Total N (TN) was measured by elemental analysis on dried sediment subsamples on a Carlo Erba  
176 CHN EA1108 Elemental Analyzer. Total P (TP) was extracted by boiling combusted sediment  
177 subsamples for 1 h in 1 M HCl. After centrifugation (10 min, 500 g) the supernatants were stored  
178 until analyzed for  $\text{PO}_4^{3-}$  by colorimetric analysis (Koroleff 1983).

179           During initial and final core sectionings, reactive Fe was extracted from ~0.2 g  
180 sediment subsamples with 0.5 M HCl. After 30 min extraction on a shaking table and centrifugation  
181 (10 min, 500 g) the supernatants were stored in 4 mL plastic vials at room temperature until  
182 analysis. Supernatants were analysed for reduced Fe (FeII) and total Fe by the ferrozine method  
183 before and after reduction with hydroxylamine (Stookey 1970; Lovley and Phillips 1987). Oxidized  
184 iron (FeIII) was determined as the difference between Total Fe and FeII.

185           Linear dimensionless  $\text{NH}_4^+$  adsorption coefficients were determined during the initial  
186 core sectioning on wet sediment subsamples from 0-2, 4-6 and 8-10 cm depth intervals in  $\text{NH}_4^+$ -  
187 adsorption experiments as described in Holmboe and Kristensen (2002). Sediment subsamples were  
188 incubated for 2 d in slurries with different  $\text{NH}_4^+$ -concentrations (0, 1, 2 and 3 mM) and 10 mg/L  
189 allylthiourea to inhibit nitrification. After centrifugation (10 min, 500 g) the supernatant was  
190 decanted and adsorbed  $\text{NH}_4^+$  was extracted from the sediment pellet in 2 M KCl (Mackin and Aller,  
191 1984). Supernatants from slurries and KCl-extractions was stored frozen (-20°C) and analyzed for  
192  $\text{NH}_4^+$  by the salicylate-hypochlorite method (Bower and Holm-Hansen 1980).

193

#### 194 *2.6 Jar experiments*

195 Closed anoxic sediment incubations ('jar experiments') were performed with sediment from  
196 different depths (0-2, 4-6 and 8-10 cm) right after core sectionings. Jar experiments measure the  
197 total anaerobic mineralization rates of ON and OP from temporal accumulation of metabolic end-  
198 products ( $\text{NH}_4^+$  and  $\text{PO}_4^{3-}$ ) in the porewater and yields solid results under a wide range of

199 environmental and experimental conditions (Kristensen and Hansen 1995; Kristensen et al. 2011;  
200 Valdemarsen et al. 2012; Quintana et al. 2013). Sediment from different depths was homogenized  
201 and fully packed into 6-8 glass scintillation vials ('jars'), leaving no headspace. The jars were  
202 closed with screw caps and buried in anoxic sediment at 15°C. Two jars were sacrificed at 3-5 day  
203 intervals for porewater extraction by centrifugation. The jars were fitted with a perforated lid  
204 containing a GF/C-filter inside before centrifugation and were then centrifuged head-down in a  
205 centrifuge tube (10 min, ~500 g). Extracted porewater was stored frozen (-20°C) and analyzed for  
206  $\text{NH}_4^+$  and  $\text{PO}_4^{3-}$  by colorimetric analysis as described above.

207

## 208 2.7 Calculations and statistics

209 Initial area specific pools of TN and TP were calculated by depth integration (0-20 cm) of TN and  
210 TP content in individual sediment layers. Differences in area specific pools of TN and TP between  
211 stations were detected by one-way ANOVA followed by Tuckey's post hoc test. Data were log-  
212 transformed before statistical analysis when assumptions of homoscedasticity were not met (only

213 TN). ~~Area specific pools of FeIII were calculated by depth integration at the beginning (initial) and~~  
214 ~~end (final) and compared by pairwise t-tests.~~

215  $\text{NH}_4^+$  adsorption coefficients ( $K_{\text{NH}}$ ) in individual sediment layers were determined  
216 based on  $\text{NH}_4^+$ -adsorption experiments. Extracted  $\text{NH}_4^+$  ( $\mu\text{mol g dw sediment}$ ) was plotted against  
217  $\text{NH}_4^+$ -concentration ( $\mu\text{mol cm}^{-3}$ ) and the linear slope,  $K'$ , was determined by least squares  
218 regression.  $K_{\text{NH}}$  could hereafter be determined from the relationship  $K_{\text{NH}} = ((1-\phi)/\phi) * \rho_{\text{ds}} * K'$ , where  
219  $\phi$  is sediment porosity and  $\rho_{\text{ds}}$  is dry sediment density (Holmboe and Kristensen 2002).

220 Rates of microbial ON and OP mineralization in discrete depth intervals (0-2, 4-6 and  
221 8-10 cm) were obtained from jar experiments by fitting the time dependent linear concentration  
222 change of  $\text{NH}_4^+$  and  $\text{PO}_4^{3-}$  by least-squares regression (Aller and Yingst 1980). When slopes were

Slettet: were also

Slettet: and

Kommentar [TBV2]: Sentence was rephrased in response to REVIEWER#1, COMMENT 5

Slettet: temporal changes in area specific FeIII pools over the whole experimental period were detected by

228 significant ( $p < 0.05$ ) the volume specific reaction rates ( $\text{nmol cm}^{-3} \text{ d}^{-1}$ ) in individual depth layers  
229 were calculated from the slopes and corrected for sediment porosity and adsorption (Kristensen and  
230 Hansen 1995). The mineralization rates at 10-20 cm depth were calculated from exponential  
231 regressions based on ON and OP mineralization rates in the top 10 cm. Total area specific ON and  
232 OP mineralization were calculated by depth integration (0-20 cm) of measured  $\text{NH}_4^+$  and  $\text{PO}_4^{3-}$   
233 production at different depths. The temporal patterns of total area specific ON and OP  
234 mineralization were fitted to a double exponential decay regression model of the form  $y = C_L \cdot \exp(-$   
235  $k_L \cdot t) + C_R \cdot \exp(-k_R \cdot t)$ , where  $t$  is time,  $C_L$  and  $C_R$  are constants and  $k_L$  and  $k_R$  denote the first order  
236 decay constants for labile and refractory ON and OP, respectively. We hereby assume that  
237 considerations based on organic C degradation kinetics (Westrich and Berner 1984) are also valid  
238 for ON and OP mineralization. Half lives of labile and refractory ON and OP could hereafter be  
239 calculated from the formula  $T_{0.5} = \ln(2)/k'$ , where  $k'$  denote  $k_L$  and  $k_R$ .

240

### 241 **3. Results**

#### 242 *3.1 Sediment characteristics*

243 Detailed sediment characteristics of the 8 stations in Odense Fjord were previously described in  
244 Valdemarsen et al. (2014) and only a brief summary is given here. The sediments from all stations  
245 had high sand content and variable silt-clay content with wet densities ranging from  $1.2\text{-}1.8 \text{ g cm}^{-3}$   
246 and porosities of 0.3-0.8. The medium grain size varied from 87 to 397  $\mu\text{m}$  among stations. The  
247 sediments from the innermost stations (St 1-3) and most of the stations in the outer basin (St 4 and  
248 6-8) contained a high proportion of silt-clay particles (13-63%). Furthermore, the stations rich in  
249 silt-clay particles were organic rich with 0.6-5.2% POC compared to the more sandy St 5 (0.1-0.2%  
250 POC).

251  $\text{NH}_4^+$ -adsorption coefficients varied erratically among stations and sediment depths  
252 (Table 1).  $K_{\text{NH}}$  ranged from 0.14 in the 8-10 cm deep sediment on St 7 to 1.06 in the surface  
253 sediment on St 2.

254 St 1 and St 3 from the inner basin had similar TN content ranging between 57-156  
255  $\mu\text{mol cm}^{-3}$  (Fig. 2). St 2 had slightly higher TN (103-227  $\mu\text{mol cm}^{-3}$ ) with a pronounced subsurface  
256 peak occurring at 3 cm depth. In the outer basin the shallow and deep silty stations (St 4 and 6-8)  
257 had similar TN-content (92-154  $\mu\text{mol cm}^{-3}$ ), except at the surface where TN was lower at St 4 (38-  
258 60  $\mu\text{mol cm}^{-3}$ ). The sandy St 5 contained exceptionally low TN (8-16  $\mu\text{mol cm}^{-3}$ ). Depth integrated  
259 TN was therefore lowest on St 5 ( $4.5 \pm 0.1 \text{ mol N m}^{-2}$ ), intermediate at St 1 ( $13.5 \pm 0.4 \text{ mol N m}^{-2}$ )  
260 and similarly high on the remaining stations (16.0 to 21.4  $\text{mol N m}^{-2}$ , Table 2).

261 Two of the stations in the inner basin (St 1 and 2) had similar TP profiles, with 10-11  
262  $\mu\text{mol cm}^{-3}$  at the sediment surface and a gradual decrease to 5.1-5.8  $\mu\text{mol cm}^{-3}$  at 15 cm depth (Fig.  
263 2). St 3 had the lowest TP content of the stations in the inner basin. The shallow silty sediments in  
264 the outer basin (St 4) were similar to St 1-2 with respect to TP, whereas the shallow sandy sediment  
265 (St 5) was similar to St 3. The deep silty sediments in the outer basin (St 6-8) were characterized by  
266 constant TP with depth (9.6-13.5  $\mu\text{mol cm}^{-3}$ ). Depth integration showed that the highest area  
267 specific TP content was found on the deep outer fjord stations (1.8-1.9  $\text{mol P m}^{-2}$ ), whereas shallow  
268 silty sediments in the inner and outer fjord contained intermediate TP content (1.2-1.3  $\text{mol P m}^{-2}$ ; St  
269 1, 2 and 4; Table 2). The lowest TP content ( $\sim 0.7 \text{ mol P m}^{-2}$ ) was found on the silty St 3 and sandy  
270 St 5 in inner and outer fjord, respectively.

271 Initial FeIII pools varied 30-fold between stations (6-243  $\text{mmol m}^{-2}$ ; Table 3), with the  
272 lowest FeIII content found in shallow sandy sediment from the outer basin (St 5). FeIII only  
273 constituted a minor fraction (2-10%) of total Fe on all stations. No statistically significant

**Kommentar [TBV3]:** Line was deleted in response to REVIEWER#2, COMMENT 9

**Slettet:** Oxidized FeIII binds  $\text{PO}_4^{3-}$  in marine sediments (Sundby et al. 1992) and it was therefore essential to know the temporal behavior of FeIII in the studied sediments.



279 differences were detected between initial and final FeIII-pools ( $p > 0.17$ ), but there were trends  
280 towards higher final FeIII content, except on St 1 and 5.

281

### 282 3.2 ON and OP mineralization

283 Mineralization rates obtained in the fully anoxic jar experiments might have underestimated  
284 mineralization rates at the sediment surface, where  $O_2$  can stimulate mineralization of  $O_2$ -sensitive  
285 organic matter (Hulthe et al. 1998). In coastal and estuarine sediments  $O_2$  only penetrates to 1-3 mm  
286 depth, suggesting a minor importance of this artefact at the beginning of the experiment.  
287 Surprisingly the sediments did not become significantly more oxidized during the long term  
288 incubations as indicated by a modest build-up of oxidized FeIII and continuous presence of  
289 hydrogen sulfide in the porewater of surface sediment from all stations (data not shown). Hence we  
290 assume that mineralization rates in the sediment cores underlying an oxic water phase were closely  
291 approximated by the rates obtained in jar experiments.

**Kommentar [TBV4]:** This paragraph was added in response to REVIEWER#1, COMMENT 4 and REVIEWER#2, COMMENT 5

292  $NH_4^+$  production in jar experiments was significant throughout the experiment, except  
293 for St 1, 8-10 cm depth after 607 d. Initially  $NH_4^+$  production was highest in the surface 0-2 cm  
294 sediment from the silty St 1-2 in the inner fjord and the sandy St 5 in the outer fjord (159-338 nmol  
295  $cm^{-3} d^{-1}$ ) and was similar on remaining stations (63-101 nmol  $cm^{-3} d^{-1}$ ; Fig. 3). Surface  $NH_4^+$   
296 production decreased rapidly over time in sediments from shallow locations in the inner and outer  
297 fjord, by 96% of initial rates on St 1 and by 61–82% on St 2–5. The surface  $NH_4^+$  production in the  
298 sediments sampled in the deep outer basin (St 6–8) decreased by 8–67% during the experiment.  
299  $NH_4^+$  production at 4–6 cm depth was initially 18–60 nmol  $cm^{-3} d^{-1}$  on all stations and temporal  
300 changes were also observed in this layer, especially in shallow silty sediments from the inner basin  
301 where  $NH_4^+$  production decreased by 75–96% to 1.4–12 nmol  $cm^{-3} d^{-1}$  by the end (Fig. 3). In  
302 sediments from the outer basin  $NH_4^+$  production at 4-6 cm depth only decreased by 19-58%. At 8-

**Kommentar [TBV5]:** Section was edited in response to REVIEWER#1, COMMENT 6

**Slettet:** Surface  $NH_4^+$  production decreased rapidly over time in sediments sampled from shallow locations in the inner and outer, by 96% on St 1 and by 61-82 on St 2-5 of initial rates. The surface  $NH_4^+$  production in the sediments sampled on deep locations (St 6-8) in the outer basin only decreased by 8-67% during the experiment.  $NH_4^+$  production at 4-6 cm depth was initially 18-60 nmol  $cm^{-3} d^{-1}$  on all stations and temporal changes were also observed in this layer. Especially in shallow silty sediments from the inner basin where  $NH_4^+$  production had decreased by 75-96% to 1.4-12 nmol  $cm^{-3} d^{-1}$  by the end (Fig. 3).

318 10 cm depth  $\text{NH}_4^+$  production at all stations occurred at similar rates and showed similar temporal  
319 trends as observed at 4-6 cm depth (Fig. 3).

320 Significant  $\text{PO}_4^{3-}$  production was measured in the surface sediment from all stations  
321 throughout the experiment (Fig. 4). Initial rates were highest ( $30\text{-}35 \text{ nmol cm}^{-3} \text{ d}^{-1}$ ) on  
322 St 1 and 2 from the shallow inner basin and considerably lower ( $7\text{-}18 \text{ nmol cm}^{-3} \text{ d}^{-1}$ ) on the  
323 remaining stations.  $\text{PO}_4^{3-}$  production initially decreased rapidly in the surface sediment from St 1  
324 and 2 and stabilized at relatively low and stable levels after  $\sim 200$  d ( $0.7\text{-}6.0 \text{ nmol cm}^{-3} \text{ d}^{-1}$ ). Surface  
325  $\text{PO}_4^{3-}$  production also decreased over time on the other stations, but temporal trends were more  
326 erratic.  $\text{PO}_4^{3-}$  production in deeper sediment was generally lower than at the surface, and with less  
327 variability among stations (Fig. 4).  $\text{PO}_4^{3-}$  production at 4-6 cm depth was  $0\text{-}6 \text{ nmol cm}^{-3} \text{ d}^{-1}$  and  
328 remained quite stable throughout the experiment on all stations. The only significant decrease ( $p =$   
329  $0.01\text{-}0.03$ ) occurred in silty sediments from the inner basin (St 1-3) and St 6 and 8 from the deep  
330 outer basin.  $\text{PO}_4^{3-}$  production varied between  $0\text{-}5 \text{ nmol cm}^{-3} \text{ d}^{-1}$  at 8-10 cm depth and was stable  
331 throughout the experiment.

332 Area-specific ON mineralization was calculated by depth integration of  $\text{NH}_4^+$   
333 production rates (Fig. 3). The sediments from the inner basin (St 1-3) showed high initial ON  
334 mineralization ( $6\text{-}11 \text{ mmol m}^{-2} \text{ d}^{-1}$ ) in the same range as the shallow silty and sandy sediments from  
335 the outer basin ( $6$  and  $10 \text{ mmol m}^{-2} \text{ d}^{-1}$  on St 4 and 5, respectively). The deep silty sediments from  
336 the outer basin showed the lowest initial ON mineralization (St 6-8;  $3\text{-}5 \text{ mmol m}^{-2} \text{ d}^{-1}$ ). Area  
337 specific ON mineralization decreased during the experiment on all stations, by 82-93% for the silty  
338 inner fjord and 34-71% on remaining stations. The temporal decrease was mainly driven by  
339 successively lower ON mineralization in surface sediment during the first  $\sim 200$  d and area-specific  
340 ON mineralization was fairly constant hereafter. Initial area-specific OP mineralization was  $0.2\text{-}1.0$   
341  $\text{mmol m}^{-2} \text{ d}^{-1}$  (Fig. 4) and decreased (59-70%) over time on several of the stations (St 1-3 and St 6).

342 As for ON mineralization, the successively lower OP mineralization was mainly due to decreased  
343 OP mineralization in surface sediment. On the other stations area-specific OP mineralization  
344 remained relatively high and did not show clear temporal trends.

345 Double exponential decay models fitted the ON mineralization kinetics on St 1-6 and  
346 the OP mineralization kinetics on St 1-3 and 6. Erratic mineralization patterns prevented the use of  
347 exponential decay models on remaining stations (see Fig. 3-4). Decay constants for labile and  
348 refractory ON and OP in were fairly similar at all stations, with  $k_L$ 's of 0.02-0.06 d<sup>-1</sup> (except for 10  
349 times higher values for ON at St 6 and for OP at St 2) and  $k_R$ 's of 0.0003-0.0015 (Table 4). The half  
350 lives for ON and OP were in the range of 0.01-0.11 and 1.3-6.3 years for labile and refractory  
351 fractions, respectively.

352

### 353 3.3 DIN- and DIP-fluxes

354 DIN fluxes followed a similar exponentially decreasing pattern for all stations (Fig. 5), and ranged  
355 from 1.1-3.7 mmol m<sup>-2</sup> d<sup>-1</sup> initially (t=0-90 d) to 0.09-0.5 mmol m<sup>-2</sup> d<sup>-1</sup> by the end. The main form  
356 of DIN released initially was NH<sub>4</sub><sup>+</sup>, which contributed 59-100% of DIN-release. Subsequently the  
357 NH<sub>4</sub><sup>+</sup> efflux decreased while NO<sub>x</sub><sup>-</sup> switched from uptake to release and after 0.5-1 y to the end of  
358 the experiment, 68-100% of the DIN was released as NO<sub>x</sub><sup>-</sup>.

359 The 8 stations showed different patterns of PO<sub>4</sub><sup>3-</sup> fluxes. The stations from the shallow  
360 inner basin, St 1-3, showed exponentially decreasing PO<sub>4</sub><sup>3-</sup> fluxes over time (initial fluxes of 0.1-0.2  
361 mmol m<sup>-2</sup> d<sup>-1</sup> decreasing to 0.01-0.05 mmol m<sup>-2</sup> d<sup>-1</sup> by the end; Fig. 5). Initial (day 0-90) PO<sub>4</sub><sup>3-</sup>  
362 fluxes on the shallow silty St 4 was around zero, but increased to 0.07-0.14 mmol m<sup>-2</sup> d<sup>-1</sup> during d  
363 90-360 of the experiment. The highest PO<sub>4</sub><sup>3-</sup> fluxes (0.07-0.21 mmol m<sup>-2</sup> d<sup>-1</sup>) were observed on the  
364 TP-poor sandy St 5, particularly towards the end of the experiment, while the TP-rich outer fjord

**Kommentar [TBV6]:** Figure was deleted in response to REVIEWER#1, COMMENT 12 and REVIEWER#2, COMMENT 12

**Slettet:** (Fig. 6)

366 stations 6-8 had the lowest and most irregular  $\text{PO}_4^{3-}$  fluxes ranging from slightly negative to 0.1  
367  $\text{mmol m}^{-2} \text{d}^{-1}$ .

368

### 369 *3.4 $\text{PO}_4^{3-}$ and $\text{NH}_4^+$ in porewater*

370 Porewater nutrient concentrations increased gradually at all depths during the experiment (data not  
371 shown).  $\text{NH}_4^+$  and  $\text{PO}_4^{3-}$  only increased moderately in the upper 2 cm, but accumulated to high  
372 levels in the deeper diffusion limited sediment. Depth-averaged initial porewater  $\text{NH}_4^+$   
373 concentration varied between 171-407  $\mu\text{M}$  on the stations. The sandy St 5 showed the highest  $\text{NH}_4^+$   
374 accumulation over time with a depth-average of 1473  $\mu\text{M}$  in porewater by the end. On the  
375 remaining stations,  $\text{NH}_4^+$  only accumulated to 259-587  $\mu\text{M}$ . Depth-averaged  $\text{PO}_4^{3-}$  concentrations at  
376 the beginning varied between 17-71  $\mu\text{M}$  depending on station. As for  $\text{NH}_4^+$ , the nutrient-poor sandy  
377 St 5 showed the highest  $\text{PO}_4^{3-}$  accumulation to 368  $\mu\text{M}$  compared with 43-170  $\mu\text{M}$  on the other  
378 stations.

379

### 380 *3.5 N- and P-budgets*

381 Area-specific nutrient mineralization obtained in jar-experiments was used to calculate total ON and  
382 OP mineralization during the experiment. ON mineralization was fairly constant for all stations  
383 except St 5 (1.4 to 1.9  $\text{mol m}^{-2}$ ) corresponding to 8-10% of initial TN (Table 5). St 5, on the other  
384 hand, had 3-fold higher ON mineralization that accounted for 80% of the initial ON. A 3-fold range  
385 among stations was also evident for OP mineralization, but with lowest rates of 0.12-0.18  $\text{mol m}^{-2}$  at  
386 St 1-4 and the highest rates of 0.22-0.33  $\text{mol m}^{-2}$  at St 5-8 (8-48% of initial TP). Interestingly, there  
387 was no apparent relationship between sediment TN and TP content and mineralization activity as  
388 some of the highest N- and P-mineralization rates were observed on the organic-poor St 5 (Table 4).  
389 DIN-effluxes, porewater accumulation and adsorption only accounted for 18-32% of total ON

390 mineralization, indicating that most of the generated  $\text{NH}_4^+$  was not accounted for by our  
391 measurements. For P, the sum of  $\text{PO}_4^{3-}$  efflux and porewater accumulation only accounted for 10-  
392 48% of total OP mineralization.

393

#### 394 **4. Discussion**

##### 395 *4.1 Sediment nutrient content*

396 TN and TP in sediments from Odense Fjord were in the same range or higher than reported for  
397 other eutrophic systems (e.g. Boynton and Kemp 1985; Cowan & Boynton 1996; Lomstein et al.  
398 1998; Coelho et al. 2004; Viktorsson et al. 2013) emphasizing the history of intense eutrophication  
399 in Odense Fjord. TN and TP in the silty sediments of Odense Fjord (all stations except St 5) were  
400 remarkably similar and only varied ~1.5 (TN) and ~3 (TP) times among stations. Despite these  
401 overall similarities, the silty sediments from the shallow inner basin showed higher initial ON- and  
402 OP-mineralization and nutrient effluxes than silty sediments from the outer fjord. This could be due  
403 to higher availability of labile ON and OP in the sediments from the inner basin, reflecting the  
404 nutrient rich conditions in the inner compared to the outer basin (Petersen et al.2009).

405           The sandy St 5 was markedly different from the other stations. It had the lowest total  
406 nutrient content and yet exhibited some of the highest rates of ON and OP mineralization. The  
407 frequent erosion by wind driven waves in this area (Valdemarsen et al. 2010) and deep (>20 cm)  
408 reworking by lugworms (*Arenicola marina*) (Riisgaard & Banta 1998; Valdemarsen et al. 2011)  
409 may remove fine particles and refractory organic matter from St 5 sediments (Wendelboe et al.  
410 2012) and prevent organic matter accumulation, hence explaining the low organic content on this  
411 station. On the other hand, intense growth and burial of microphytobenthos and other reactive  
412 detritus by the strong physical disturbance and vertical mixing, can explain the unexpected high TN  
413 and TP reactivity of St 5 sediment.

414 A rough areal estimate based on the measured TN and TP content on the examined  
415 stations (Table 2) suggest that  $12.6 \cdot 10^6 \text{ kg N}$  and  $3.7 \cdot 10^6 \text{ kg P}$  are stored in the upper 20 cm of  
416 Odense Fjord sediments, corresponding to ~6 (N) and ~62 (P) years of the current annual external  
417 nutrient loading to the system. ↓

#### 418 419 4.2 Organic N and P mineralization

420 Microbial mineralization of ON and OP in Odense Fjord sediments led to marked release of  
421 inorganic nutrients, especially in the initial phase of the experiment. Initially there were strong  
422 vertical gradients of ON and OP mineralization in silty and sandy sediments from shallow  
423 environments, indicating that newly deposited and relatively labile organic matter was being  
424 degraded near the sediment surface, with the depth gradient reflecting a gradual and time-dependent  
425 depletion of labile ON and OP (Westrich & Berner 1984; Mackin and Swider 1989; Valdemarsen et  
426 al. 2014). It was expected that ON and OP mineralization would decrease with time at all depths  
427 due to diminishing reactivity of the organic pools. However, significant temporal decreases were  
428 only observed in surface sediments from shallow locations, whereas mineralization rates were  
429 surprisingly stable in the underlying sediment and the entire sediment column in the deep outer  
430 fjord. Assuming that organic matter degradation follows an exponential decay pattern, the lack of a  
431 detectable attenuation in mineralization rates over a ~2 yr period indicates very low initial reactivity  
432 of ON and OP in the deeper layers (Westrich & Berner 1984). Nevertheless, since ON and OP of  
433 low reactivity was present at high concentrations, it remained a significant source for inorganic  
434 nutrients.

435 Total jar-based microbial ON and OP mineralization over the ~2 years experimental  
436 period (Table 5) only accounted for a minor fraction of initial TN and TP in sediments from Odense  
437 Fjord suggesting that the standing stock of organic N and P will be a source of nutrients for

**Kommentar [TBV7]:** Changes in response to REVIEWER#1, COMMENT 8

**Slettet:** <sup>3</sup>

**Slettet:** T

**Slettet:** <sup>3</sup>

**Slettet:** T

**Kommentar [TBV8]:** Line deleted in response to REVIEWER#2, COMMENT 13

**Slettet:** The main goal of this experiment was to evaluate the extent to which these accumulated nutrients can be recycled to the overlying water as internal loading.

446 extended time. Decay constants from the exponential decay model suggested that labile ON and OP  
447 was rapidly degraded on all stations within 10-240 d, whereas depletion of more refractory ON and  
448 OP will only occur on decadal time-scales (8-40 years), indicating that depletion of buried and  
449 degradable ON and OP in eutrophic ecosystems will take considerable time.

450

#### 451 *4.3 Fate of inorganic nutrients*

452  $\text{NH}_4^+$  and  $\text{PO}_4^{3-}$  produced by microbial mineralization accumulated in porewater of all sediments  
453 within the first 1-6 months and only changed slightly hereafter. However, over the whole  
454 experiment, porewater accumulation explained only a minor fraction of the jar-based total ON and  
455 OP mineralization (0.8-8.1%). It was also investigated if  $\text{NH}_4^+$  adsorption to mineral surfaces was  
456 an important N sink. Despite the large spatial heterogeneity of  $\text{NH}_4^+$  adsorption, this process never  
457 accounted for more than 1% of the total produced  $\text{NH}_4^+$  over the whole experiment and was  
458 therefore not quantitatively important.

459 Nutrient release to the overlying water was the most important route for inorganic  
460 nutrients produced by microbial mineralization. We could not account for all the produced  
461 nutrients, since nutrient mineralization in jar experiments exceeded DIN and  $\text{PO}_4^{3-}$  effluxes by 70-  
462 84% and 62-93%, respectively. The missing  $\text{NH}_4^+$  may have been lost through coupled nitrification-  
463 denitrification (e.g. Mackin and Swider 1989; Quintana et al. 2013). The conspicuous shift from  
464  $\text{NH}_4^+$  to  $\text{NO}_3^-$  release indicated that nitrification was an active process in all sediment types, and  
465 denitrifying bacteria probably proliferated in the  $\text{NO}_3^-$ -rich surface sediment. In the present case,  
466 coupled nitrification-denitrification rates of 1-2  $\text{mmol m}^{-2} \text{d}^{-1}$  are required to account for the missing  
467  $\text{NH}_4^+$ , which is within the range reported in previous studies (e.g. Nielsen et al. 1995; Christensen et  
468 al. 2000; Tobias et al. 2003). On the other hand, the missing  $\text{PO}_4^{3-}$  must have been retained within  
469 the sediments. Several studies suggest almost complete  $\text{PO}_4^{3-}$  retention in marine sediments with an

470 oxic sediment surface (Rozan et al. 2002; Viktorsson et al. 2013) where  $\text{PO}_4^{3-}$  adsorbs to oxidized  
471 FeIII-minerals preventing  $\text{PO}_4^{3-}$  efflux (Sundby et al.1992). Experimental studies suggest that every  
472 FeIII molecule can retain more than 0.5  $\text{PO}_4^{3-}$  molecules (Gunnars & Blomqvist, 1997; Rozan et al.  
473 2002). Hence the FeIII levels on all the silty stations were sufficient to retain the missing  $\text{PO}_4^{3-}$ ,  
474 especially when considering that 0.5 M HCl extractions only extracts a fraction of the available  
475 FeIII. On the sandy St 5 the FeIII levels were too low to account for the missing  $\text{PO}_4^{3-}$ , indicating  
476 that there were other  $\text{PO}_4^{3-}$  sinks.  $\text{PO}_4^{3-}$  adsorption in the anoxic sediment (Krom & Berner, 1980)  
477 or precipitation of  $\text{PO}_4^{3-}$ - $\text{CaCO}_3$  complexes (Coelho et al. 2004) are possible sinks that were not  
478 quantified in this experiment.

479

#### 480 4.4 Internal nutrient loading

481 We calculated the potential internal nutrient loading in Odense Fjord resulting from microbial  
482 mineralization of ON and OP for a 2 y period based on the measured nutrient effluxes. Average  
483 nutrient fluxes were calculated for each sediment type, i.e. shallow inner fjord (St 1-3), shallow silty  
484 outer fjord (St 4), sandy outer fjord (St 5) and deep outer fjord (St 6-8). The monthly time-weighted  
485 DIN and  $\text{PO}_4^{3-}$  fluxes and the total areal distribution of the different sediment types in Odense Fjord  
486 were then used to calculate the total internal nutrient loading ( $10^3 \text{ kg N and P mo}^{-1}$ ) for each  
487 sediment type and for the whole ecosystem. Evidently these calculations do not represent the *in situ*  
488 internal nutrient loading, since effects of the otherwise continuous deposition of organic matter  
489 were omitted by the experimental setup. It can also be debated if all the released nutrients can be  
490 considered internal nutrient loading, since the mineralization of recently deposited organic matter in  
491 surface sediments drove the majority of nutrient release during the first ~200 d. This nutrient release  
492 is largely determined by the ecosystem primary productivity extending only a few years back, and is  
493 therefore closely coupled to the recent levels of external nutrient loading. In any case the

Kommentar [TBV9]: Change in  
response to REVIEWER#2, COMMENT  
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Slettet: t



495 calculations represent the nutrient release resulting from the mineralization of slowly reacting ON  
496 and OP, which have accumulated in the sediments.

497 The calculations show the magnitude of nutrient release driven by microbial  
498 mineralization of sediment-bound ON and OP in eutrophic ecosystems (Fig. 6). Total DIN release  
499 from the whole fjord bottom is equivalent to  $121 \cdot 10^3 \text{ kg N y}^{-1}$  ( $\sim 20 \text{ kg N ha}^{-1} \text{ y}^{-1}$ ) the first year after  
500 sedimentation of new organic matter has ceased, but only  $38 \cdot 10^3 \text{ kg N y}^{-1}$  ( $\sim 6.2 \text{ kg N ha}^{-1} \text{ y}^{-1}$ ) the  
501 second year, since ON effluxes decreased exponentially on all stations. The shallow sandy  
502 sediments in the outer fjord were most important for the total fjord-wide N-release (39%), whereas  
503 the remaining 3 sediment types contributed equally (16-23%). The numbers for internal N-loading  
504 are impressive at first, but only correspond to maximum 2-6% (N) of the current external N-loading  
505 to Odense Fjord (about  $2000 \cdot 10^3 \text{ kg N y}^{-1}$ ; Petersen et al. 2009). In the shallow N-limited Odense  
506 Fjord the internal N-loading can therefore only have minor effects for overall ecosystem  
507 productivity. In any case the external N-loading is far more important for the overall primary  
508 productivity and ecological status.

509 The internal P-loading showed different temporal dynamics than internal N-loading.  
510 Total P-release from the whole fjord bottom was stable over time at rates of  $21\text{-}22 \cdot 10^3 \text{ kg P y}^{-1}$   
511 ( $\sim 3.4\text{-}3.6 \text{ kg P ha}^{-1} \text{ y}^{-1}$ ; Fig. 6) while internal N-loading decreased exponentially. The stability was  
512 driven by the increasing P release in shallow sandy outer fjord sediment and constant P release in  
513 deep outer fjord sediment. As for N, the shallow sandy sediments in the outer fjord was most  
514 important for total internal P-loading (57%) and the remaining 3 sediment types contributed equally  
515 (14-15%). The internal P-loading corresponded to 35-36% (P) of the current external P-loading to  
516 Odense Fjord ( $60 \cdot 10^3 \text{ kg P y}^{-1}$ ; Petersen et al. 2009) and thus potentially constitutes a stable and  
517 significant P-source in the system. However, since Odense Fjord and most other temperate coastal

Kommentar [TBV10]: The original  
Fig. 7 is now Fig. 6

Slettet: 7

Kommentar [TBV11]: The original  
Fig. 7 is now Fig. 6

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520 ecosystems are mostly N-limited (Howarth et al. 2011) it is uncertain to which degree this excess P  
521 will affect ecosystem productivity.

522

#### 523 4.5 Ecological implications

524 In many shallow eutrophic estuaries the external nutrient loading has been reduced to induce  
525 oligotrophication, but lower nutrient concentrations in the recipient estuary often occurs after  
526 considerable delay and rarely corresponds proportionally to the reductions (Kronvang et al. 2005;  
527 Carstensen 2006). This indicates that a transient phase occurs, where accumulated nutrients are  
528 being released from the soils and sediments in the water shed and receiving estuary, respectively,  
529 while the system equilibrates to a new level of external nutrient loading. Our study shows the  
530 magnitude and temporal dynamics of the internal nutrient loading that can be expected in shallow  
531 estuaries recovering from eutrophication. It appears that internal N-loading will be insignificant  
532 during recovery since it only corresponded to 2-6% of the external N-loading in our example and  
533 decreased rapidly. Internal N-loading will therefore only lead to marginally elevated N-availability  
534 and have minor effects on primary productivity and eutrophication status. The results are different  
535 with respect to  $\text{PO}_4^{3-}$ , since the internal P-loading was stable and corresponded to  $>1/3$  of the  
536 external P-loading. Internal P-loading may therefore be a significant source of dissolved  $\text{PO}_4^{3-}$  for  
537 extended time in shallow eutrophic estuaries, and at a sufficiently high level to counteract  
538 reductions in the external P-loading. Most shallow estuaries are N-limited (Conley et al. 2000;  
539 Howarth & Marinho 2006; Howarth et al. 2011) so a high internal P-loading might only exacerbate  
540 N-limitation while having no further consequences for ecological quality. Decreasing internal N-  
541 loading and stable internal P-loading could also lead to increased dominance of cyanobacteria,  
542 which have low requirements for dissolved N. However, major shifts in phytoplankton communities  
543 would only occur in systems where decreased internal nutrient loading results in markedly lower

**Slettet:** management efforts have been implemented to reduce

**Slettet:** and

**Slettet:** (e.g Carstensen et al. 2006). This generally results in

**Slettet:** , but

**Slettet:** in the external nutrient loading

**Slettet:** during recovery from eutrophication

**Kommentar [TBV12]:** The first half of this paragraph was shortened in response to REVIEWER#1, COMMENT 9

**Slettet:** in our example

**Slettet:** in all sediments types

**Slettet:** in eutrophic estuaries

**Slettet:** However, since m

557 DIN-concentrations in the water phase, i.e. in systems where N-loading is low and internal nutrient  
558 sources dominate.

559 The estimates of internal nutrient loading presented here provide an illustrative  
560 example, but the exact values are only valid for the experimental conditions and must be  
561 extrapolated with caution. Microbial reaction rates and DIN and  $\text{PO}_4^{3-}$  release from sediments are  
562 strongly influenced by ambient conditions. For instance, sediment macrofauna may stimulate the  
563 rates of organic matter degradation and sediment nutrient release through bioturbation (e.g.  
564 Kristensen et al. 2012; 2014) leading to higher internal nutrient loading than estimated from  
565 defaunated sediment cores in this experiment. Similarly microbial mineralization processes and  
566 hence sediment DIN and  $\text{PO}_4^{3-}$  release are strongly temperature dependent (Westrich and Berner  
567 1988; Sanz-Lazaro et al. 2011) and the magnitude of internal nutrient loading will therefore vary  
568 seasonally compared to our estimates based on a constant temperature experiment. Finally, in our  
569 experimental setup we also omitted hydrodynamics and porewater advection which are known to  
570 stimulate nutrient cycling in shallow permeable sediments (Cook et al. 2007; Huettel et al. 2014).  
571 This will especially affect the estimated nutrient release from the sandy sediments from this study.  
572 Given the multitude of factors influencing nutrient mineralization rates, the actual magnitude of  
573 internal nutrient loading and related consequences for primary productivity will therefore follow a  
574 seasonal pattern driven by e.g. temperature, hydrodynamics and composition and activity of benthic  
575 fauna. Other environmental variables such as hypoxia in the water column may also influence the  
576 magnitude of internal nutrient loading, since it hampers  $\text{PO}_4^{3-}$  retention by Fe-oxides (Azzoni et al.  
577 2005; Mort et al. 2010; Viktorsson et al. 2013) and limits coupled nitrification-denitrification while  
578 stimulating dissimilatory nitrate reduction to  $\text{NH}_4^+$  (Christensen et al. 2000; Jäntti & Hietanen  
579 2012). Ecosystems suffering from hypoxia may therefore experience a much higher internal nutrient  
580 loading than measured in this experiment. A comparison between total ON and OP mineralization

**Kommentar [TBV13]:** This section was edited according to REVIEWER#1, COMMENT 1 and REVIEWER#2, COMMENT 8

**Slettet:** a high internal P-loading will probably only exacerbate N-limitation while having no further consequences for ecological quality.

**Kommentar [TBV14]:** This line was edited in response to REVIEWER#2, COMMENT 7

**Kommentar [TBV15]:** This line was edited in response to REVIEWER#1, COMMENT 2b and REVIEWER#2, COMMENT 3

**Kommentar [TBV16]:** This line was edited in response to REVIEWER#1, COMMENT 3 and REVIEWER#2, COMMENT 4

**Kommentar [TBV17]:** This line was added in response to REVIEWER#1, COMMENT 7, REVIEWER#2, COMMENT 2, REVIEWER#2, COMMENT 7

**Slettet:** and temperature changes can lead to several fold variation in DIN and  $\text{PO}_4^{3-}$  release from sediments (Sanz-Lazaro et al. 2011). T

**Slettet:** ,

**Slettet:** and benthic primary production

**Slettet:** During warm summer periods, for instance, where nutrient concentrations in the water column are kept low by primary producers, even moderate nutrient release from sediments can have a high impact on ecosystem functioning and may even shift the balance between limiting nutrients (Kristensen et al. *submitted*).

599 and effluxes from this experiment, suggests that nutrient effluxes could potentially increase 3-6  
600 (DIN) and 2-10 ( $\text{PO}_4^{3-}$ ) times if there are no mechanisms to transform or retain inorganic nutrients  
601 at the sediment surface.

602

#### 603 *4.6 Conclusions*

604 In this study we investigated the mineralization of organic N and P buried in the sediments from a  
605 shallow eutrophic estuary and obtained estimates of the magnitude and temporal dynamics of  
606 internal nutrient loading. Total internal N-loading, which attenuated rapidly, corresponded to only a  
607 minor fraction of the external N-loading and was therefore not important for the ecological state in  
608 the studied ecosystem. Total internal P-loading showed no temporal attenuation and was  
609 quantitatively more important as it corresponded to  $>1/3$  of the external P-loading. However, the  
610 studied ecosystem was N-limited, and it is therefore uncertain if high internal P-loading will result  
611 in negative ecological effects. This study indicates that internal nutrient loading, and especially  
612 internal N-loading, is a transient phenomena that can only temporarily influence the recovery  
613 trajectory of ecosystems recovering from eutrophication. In turn, internal nutrient loading driven by  
614 mineralization of organic N and P in sediments, cannot explain the lack of recovery in shallow  
615 estuaries where external nutrient loading has been reduced.

616

#### 617 **5. Acknowledgements**

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622

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785 **Tables**

786 Table 1. Dimensionless linear  $\text{NH}_4^+$  adsorption coefficients,  $K_{\text{NH}}$ , for different sediment depths at St  
787 1-8.

	<b>St 1</b>	<b>St 2</b>	<b>St 3</b>	<b>St 4</b>	<b>St 5</b>	<b>St 6</b>	<b>St 7</b>	<b>St 8</b>
<b>0-2 cm</b>	0.26	1.06	0.33	0.46	0.64	0.31	0.57	0.48
<b>6-8 cm</b>	0.52	0.76	0.49	0.45	0.82	0.51	0.62	0.36
<b>8-10 cm</b>	0.40	0.82	0.20	0.79	0.55	0.66	0.14	0.45

788

789 Table 2. Depth integrated (0-16 cm) area specific TN and TP content  $\pm$  SE (n=3) on St 1-8.

790 Superscript capital letters indicate the grouping of data obtained by ANOVA and subsequent post

791 hoc analysis. Average TN:TP ratios are also shown.

**Kommentar [TBV18]:** TN:TP ratios were added to the table in response to REVIEWER#1, COMMENT 10 and REVIEWER#2, COMMENT 11

	<u>TN</u> (mol m <sup>-2</sup> )	<u>TP</u> (mol m <sup>-2</sup> )	<u>TN:TP</u> =
<u>St 1</u>	<u>13.5 <math>\pm</math> 0.4<sup>A</sup></u>	<u>1.34 <math>\pm</math> 0.04<sup>A</sup></u>	<u>10.1</u>
<u>St 2</u>	<u>21.5 <math>\pm</math> 0.5<sup>B</sup></u>	<u>1.31 <math>\pm</math> 0.02<sup>A</sup></u>	<u>16.4</u>
<u>St 3</u>	<u>16.0 <math>\pm</math> 0.2<sup>B</sup></u>	<u>0.70 <math>\pm</math> 0.06<sup>B</sup></u>	<u>22.9</u>
<u>St 4</u>	<u>16.6 <math>\pm</math> 1.1<sup>B</sup></u>	<u>1.18 <math>\pm</math> 0.06<sup>A</sup></u>	<u>14.1</u>
<u>St 5</u>	<u>4.5 <math>\pm</math> 0.1<sup>C</sup></u>	<u>0.73 <math>\pm</math> 0.04<sup>B</sup></u>	<u>6.2</u>
<u>St 6</u>	<u>17.1 <math>\pm</math> 0.1<sup>B</sup></u>	<u>1.94 <math>\pm</math> 0.03<sup>C</sup></u>	<u>8.8</u>
<u>St 7</u>	<u>18.1 <math>\pm</math> 0.0<sup>B</sup></u>	<u>1.86 <math>\pm</math> 0.05<sup>C</sup></u>	<u>9.7</u>
<u>St 8</u>	<u>19.5 <math>\pm</math> 0.2<sup>B</sup></u>	<u>1.83 <math>\pm</math> 0.03<sup>C</sup></u>	<u>10.7</u>

792

793

794 Table 3. Initial and final depth integrated pools (0-20 cm) of FeIII  $\pm$  SE (n=3) on St 1-8. t-tests  
 795 showed no significant difference between initial and final FeIII pools on any station.

	Initial		Final	
	FeII (mmol m <sup>-2</sup> )	FeIII (mmol m <sup>-2</sup> )	FeII (mmol m <sup>-2</sup> )	FeIII (mmol m <sup>-2</sup> )
<b>St 1</b>	2390 $\pm$ 34	243 $\pm$ 24	2294 $\pm$ 153	92 $\pm$ 22
<b>St 2</b>	2302 $\pm$ 160	157 $\pm$ 32	2399 $\pm$ 189	271 $\pm$ 161
<b>St 3</b>	1356 $\pm$ 155	62 $\pm$ 25	1358 $\pm$ 154	109 $\pm$ 40
<b>St 4</b>	1054 $\pm$ 86	28 $\pm$ 20	996 $\pm$ 23	97 $\pm$ 37
<b>St 5</b>	258 $\pm$ 2	6.3 $\pm$ 1.0	274 $\pm$ 39	6.4 $\pm$ 1.2
<b>St 6</b>	1887 $\pm$ 37	75 $\pm$ 12	1813 $\pm$ 43	141 $\pm$ 40
<b>St 7</b>	2464 $\pm$ 105	52 $\pm$ 2.0	2142 $\pm$ 60	137 $\pm$ 48
<b>St 8</b>	1697 $\pm$ 63	156 $\pm$ 8.0	1813 $\pm$ 43	210 $\pm$ 89

796

797

798 Table 4. Double exponential regression statistics for the temporal trends of total ON and OP  
799 degradation in jar experiments. Total organic N (ON) and P (OP) degradation were fitted to the  
800 exponential decay function  $y = C_L \cdot \exp(-k_L \cdot x) + C_R \cdot \exp(-k_R \cdot x)$ , where  $C_L$  and  $C_R$  denote constants  
801 and  $k_L$  and  $k_R$  denote decay constants for labile and refractory organic ON and OP, respectively.  
802 Statistics were not calculated for St 7-8 (ON) and for St 4-5 and 7-8 (OP), since the temporal  
803 degradation patterns did not fit the double exponential decay model.  $T_{L, 0.5, L}$  and  $T_{R, 0.5, R}$  denote  
804 the half life (y) of labile and refractory ON and OP, respectively.

<b>ON</b>						
	$k_L$	$k_R$	$C_L$	$C_R$	$T_{L, 0.5}$	$T_{R, 0.5}$
St 1	$4.6 \cdot 10^{-2}$	$1.1 \cdot 10^{-3}$	7.7	2.4	0.04	1.73
St 2	$2.3 \cdot 10^{-2}$	$1.0 \cdot 10^{-3}$	3.1	2.9	0.08	1.90
St 3	$5.3 \cdot 10^{-2}$	$1.1 \cdot 10^{-3}$	8.6	2.8	0.04	1.73
St 4	$4.3 \cdot 10^{-2}$	$0.4 \cdot 10^{-3}$	4.0	1.8	0.04	4.75
St 5	$5.7 \cdot 10^{-2}$	$0.6 \cdot 10^{-3}$	2.7	7.2	0.03	3.17
St 6	$52.4 \cdot 10^{-2}$	$0.3 \cdot 10^{-3}$	3.2	2.9	0.01	6.33
St 7	-	-	-	-	-	-
St 8	-	-	-	-	-	-
<b>OP</b>						
	$k_L$	$k_R$	$C_L$	$C_R$	$T_{L, 0.5}$	$T_{R, 0.5}$
St 1	$3.9 \cdot 10^{-2}$	$0.4 \cdot 10^{-3}$	0.6	0.3	0.05	4.75
St 2	$56.0 \cdot 10^{-2}$	$1.5 \cdot 10^{-3}$	1.1	0.3	0.01	1.27
St 3	$2.2 \cdot 10^{-2}$	$1.3 \cdot 10^{-3}$	0.1	0.3	0.08	1.46
St 4	-	-	-	-	-	-
St 5	-	-	-	-	-	-
St 6	$1.7 \cdot 10^{-2}$	$0.9 \cdot 10^{-3}$	0.4	0.5	0.11	2.11
St 7	-	-	-	-	-	-
St 8	-	-	-	-	-	-

805

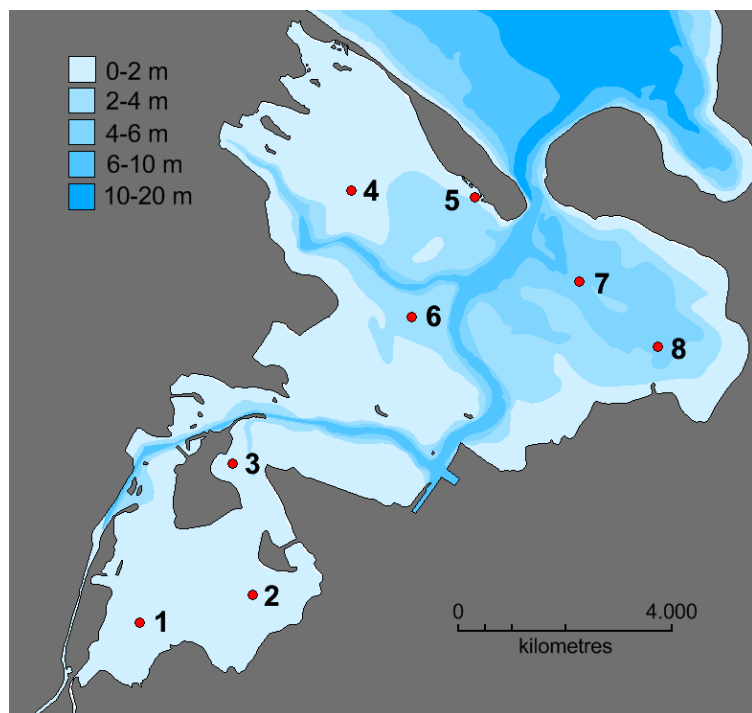
806 Table 5. N and P budgets for the experiment. Initial TN and TP are the depth integrated values  
807 based on initial measurements. ON and OP degradation were calculated based on area specific rates  
808 obtained from jar experiments. Total  $\text{NH}_4^+$ -,  $\text{NO}_x^-$ - and  $\text{PO}_4^{3-}$ -effluxes were calculated by time  
809 integration of effluxes over the entire experimental period.  $\text{NH}_4^+$ - and  $\text{PO}_4^{3-}$ -accumulation in  
810 porewater (pw) was calculated from the difference between initial and final pw profiles.  $\text{NH}_4^+$   
811 adsorbtion was calculated from initial and final pw-inventories of  $\text{NH}_4^+$  and the average  $\text{NH}_4^+$ -  
812 adsorbtion coefficient for each station. Values in parentheses marked with \* or \*\* represent  
813 percentage relative to initial TN and TP or total N and P mineralization, respectively.

<i><b>N-mineralization</b></i>	<b>St 1</b>	<b>St 2</b>	<b>St 3</b>	<b>St 4</b>	<b>St 5</b>	<b>St 6</b>	<b>St 7</b>	<b>St 8</b>
Initial TN ( $\text{mol m}^{-2}$ )	13.5	21.5	16.0	16.6	4.5	17.1	18.1	19.5
ON degradation, jars ( $\text{mol m}^{-2}$ )*	1.38 (10.2)	1.62 (7.5)	1.56 (9.8)	1.44 (8.6)	3.62 (80.1)	1.77 (10.1)	1.61 (8.9)	1.86 (9.6)
$\text{NH}_4^+$ efflux ( $\text{mol m}^{-2}$ )**	0.26 (19.1)	0.10 (6.2)	0.23 (14.6)	0.15 (10.7)	0.38 (10.6)	0.09 (5.1)	0.27 (16.6)	0.12 (6.2)
$\text{NO}_x^-$ efflux ( $\text{mol m}^{-2}$ )**	0.15 (11.2)	0.21 (13.0)	0.19 (12.1)	0.11 (7.3)	0.18 (5.0)	0.22 (12.9)	0.17 (10.8)	0.20 (10.9)
$\text{NH}_4^+$ accumulation, pw ( $\text{mol m}^{-2}$ )**	0.02 (1.6)	0.01 (0.7)	0.00 (0.0)	0.08 (5.9)	0.06 (1.6)	0.03 (1.8)	0.01 (0.7)	0.02 (1.3)
$\text{NH}_4^+$ adsorbtion ( $\text{mol m}^{-2}$ )**	0.01 (0.7)	0.01 (0.6)	0.00 (0.0)	0.01 (0.4)	0.04 (1.0)	0.02 (1.0)	0.00 (0.2)	0.01 (0.6)
<i><b>P-mineralization</b></i>	<b>St 1</b>	<b>St 2</b>	<b>St 3</b>	<b>St 4</b>	<b>St 5</b>	<b>St 6</b>	<b>St 7</b>	<b>St 8</b>
Initial TP ( $\text{mol m}^{-2}$ )	1.34	1.31	0.70	1.18	0.73	1.94	1.86	1.83
OP degradation, jars ( $\text{mol m}^{-2}$ )*	0.16 (12.6)	0.12 (7.9)	0.18 (19.6)	0.13 (11.2)	0.29 (47.7)	0.28 (14.5)	0.22 (11.5)	0.33 (17.4)
$\text{PO}_4^{3-}$ efflux ( $\text{mol m}^{-2}$ )**	0.02 (12.6)	0.04 (38.0)	0.02 (16.8)	0.02 (15.8)	0.10 (27.8)	0.02 (7.4)	0.03 (12.8)	0.02 (6.5)
$\text{PO}_4^{3-}$ accumulation, pw ( $\text{mol m}^{-2}$ )**	0.01 (4.8)	0.01 (9.6)	0.00 (1.6)	0.00 (0.8)	0.02 (5.1)	0.01 (3.0)	0.00 (1.7)	0.01 (3.1)

814

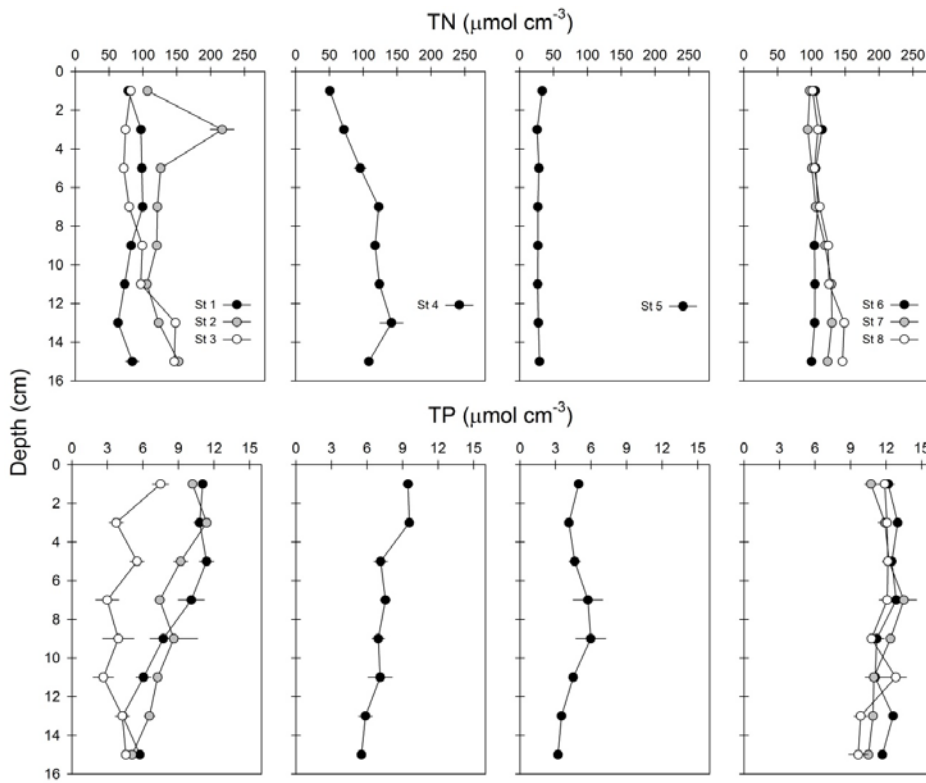


815 **Figures**



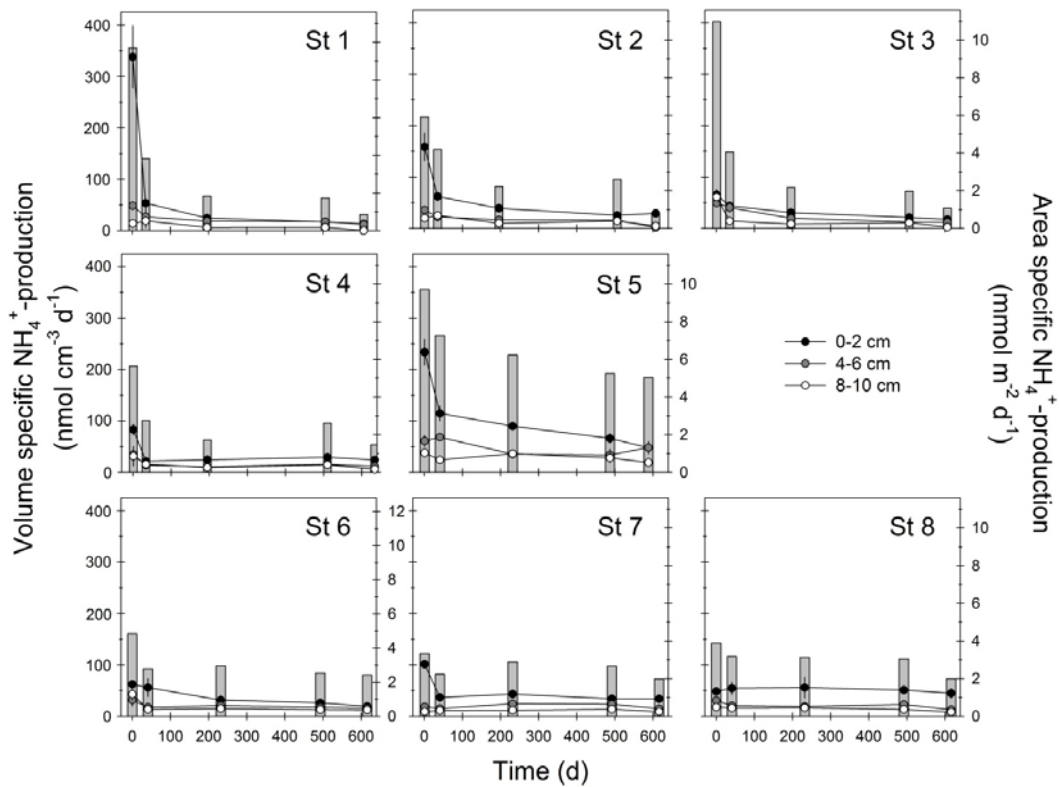
816

817 Figure 1. Map of Odense Fjord (55°29'15" N; 10°31'09") showing the 8 stations, where sediments  
818 were sampled for the long term degradation experiment. Gray color indicates land and different  
819 shades of blue indicate water depth.



820

821 Figure 2. Total nitrogen (TN) and total phosphorus (TP) in sediments from Odense Fjord. Left  
 822 panels show stations from the shallow inner fjord (St 1, 2 and 3), middle panels show shallow silty  
 823 and sandy sediments in the outer fjord (St 4 and 5, respectively) and right panels show deep silty  
 824 sediments in the outer fjord (St 6, 7 and 8). Error bars indicate standard error ( $n = 3$ ).



825

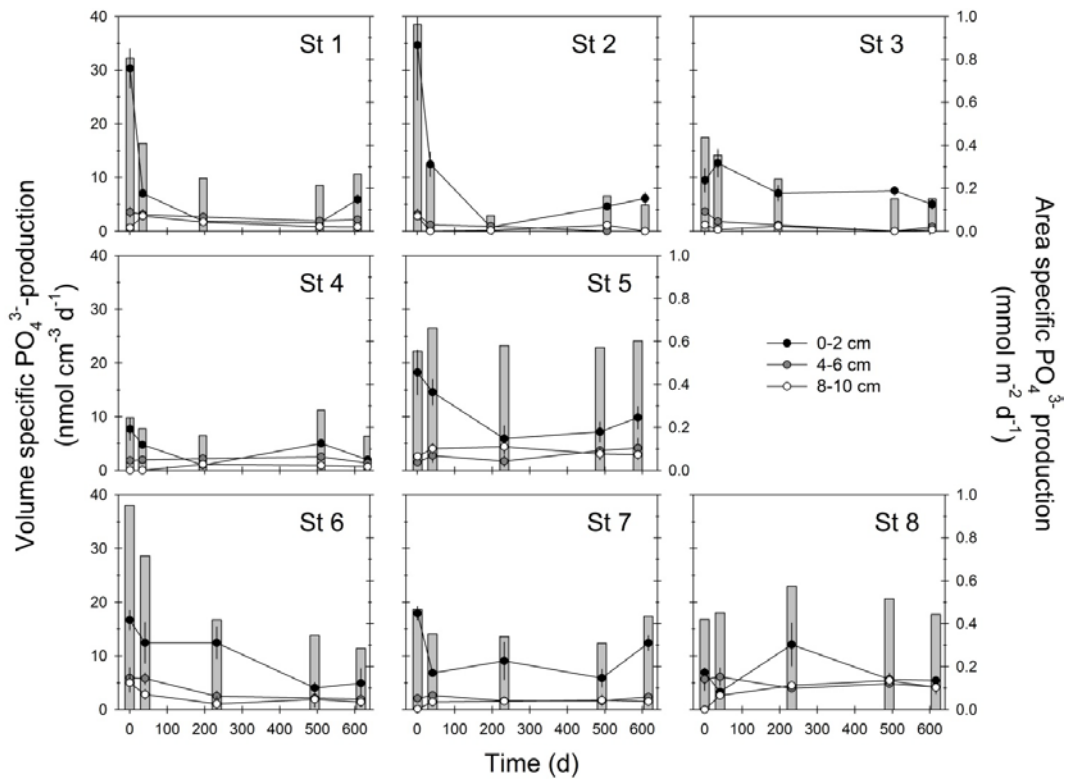
826 Figure 3.  $\text{NH}_4^+$  production measured in jar experiments with sediment from shallow inner basin

827 (upper panels), shallow silty and sandy outer basin (middle panels) and deep silty outer basin (lower

828 panels). Black, gray and white symbols indicate volume specific  $\text{NH}_4^+$  production in sediment from

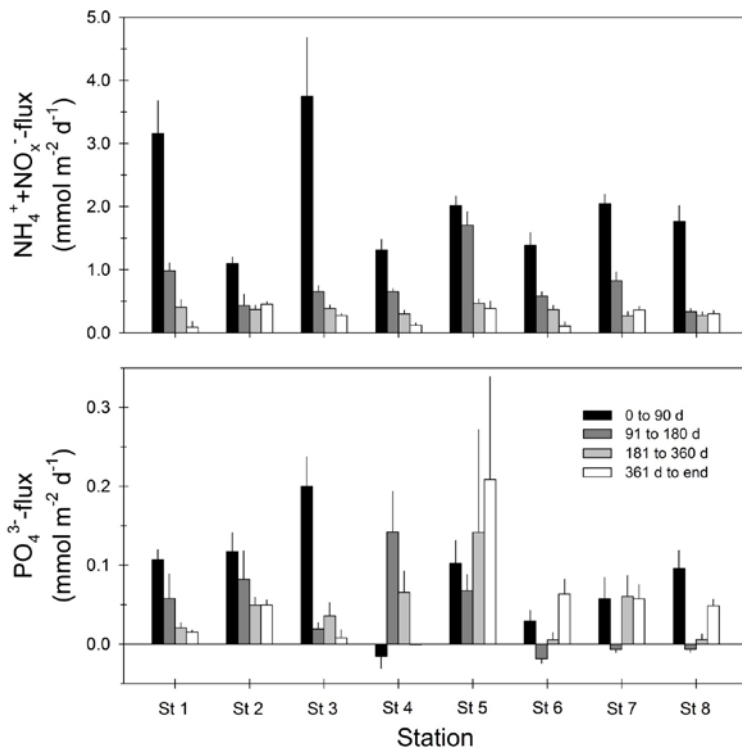
829 0-2, 4-6 and 8-10 cm depth, respectively (left y-axis). Bars indicate depth integrated (0-20 cm)

830  $\text{NH}_4^+$  production based on volume specific production rates (right y-axis).



831

832 Figure 4.  $\text{PO}_4^{3-}$  production measured in jar experiments performed with sediment from shallow  
 833 inner fjord (upper panels), shallow silty and sandy outer fjord (middle panels) and deep silty outer  
 834 fjord (lower panels). Black, gray and white symbols indicate volume specific  $\text{PO}_4^{3-}$  production in  
 835 sediment from 0-2, 4-6 and 8-10 cm depth, respectively (left y-axis). Bars indicate depth integrated  
 836 (0-20 cm)  $\text{PO}_4^{3-}$  production based on volume specific production rates (right y-axis).



837

838 Figure 5. Fluxes of dissolved inorganic nitrogen (DIN =  $\text{NH}_4^+ + \text{NO}_x^-$ ) and  $\text{PO}_4^{3-}$  at various times  
 839 during the experiment. Error bars represent standard error (n = 6-24).

840

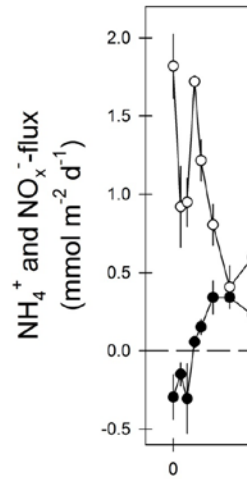
**Kommentar [TBV19]:** Change in response to REVIEWER#1, COMMENT 11

**Slettet:**  $\text{NO}_3^-$

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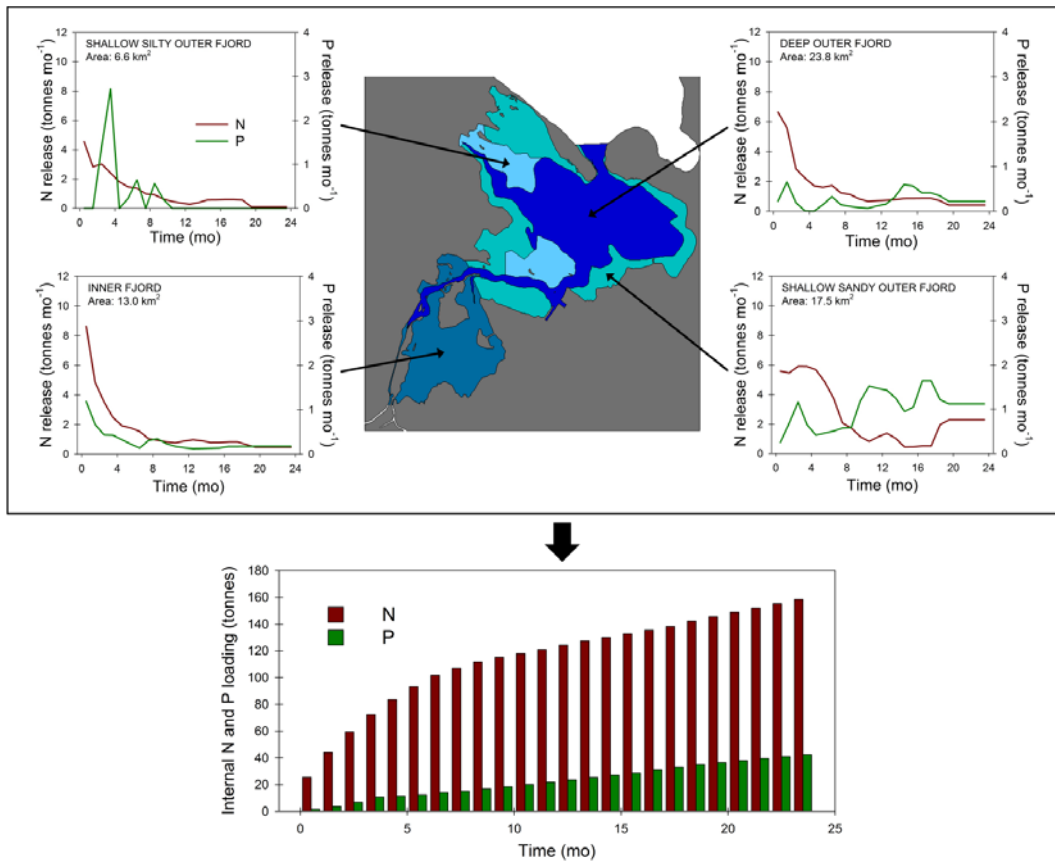
843

**Kommentar [TBV20]:** Figure was deleted in response to REVIEWER#1, COMMENT 12 and REVIEWER#2, COMMENT 12



**Slettet:**  
Figure 6. Example of changes in NH<sub>4</sub><sup>+</sup> and NO<sub>3</sub><sup>-</sup> fluxes measured during 600 d on sediment from the silty inner fjord (St 2). NH<sub>4</sub><sup>+</sup> was the main form of N released in the beginning of the experiment and NO<sub>3</sub><sup>-</sup> was the main form released at the end. A similar pattern was observed on all stations. Error bars indicate standard error (n=3) and the dashed horizontal line indicate y=0

**Slettet: .**



855

856 | Figure 6. Estimated internal nutrient loading in Odense Fjord. The upper figure shows a schematic  
 857 overview of Odense Fjord with the distribution of sediment types included in this study and their  
 858 nutrient release over a 24 month period. The lower figure shows the cumulated nutrient release from  
 859 the entire fjord bottom.

Slettet: 7