

This discussion paper is/has been under review for the journal Biogeosciences (BG). Please refer to the corresponding final paper in BG if available.

Predicting the denitrification capacity of sandy aquifers from in situ measurements using push-pull ¹⁵N tracer tests

W. Eschenbach¹, R. Well¹, and W. Walther^{2,*}

retired

Received: 21 October 2014 - Accepted: 3 November 2014 - Published: 2 December 2014

Correspondence to: W. Eschenbach (wolfram.eschenbach@ti.bund.de)

Published by Copernicus Publications on behalf of the European Geosciences Union.

)iscussior

cussion Pap

Discussion Paper

Discussion Paper

Discussion Paper

BGD

11, 16527–16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page

Abstract Introduction

Conclusions

References

Tables Figures

ı **▶**I

•

Back Close

Full Screen / Esc

Printer-friendly Version



¹Johann Heinrich von Thünen-Institut, Federal Research Institute for Rural Areas, Forestry and Fisheries, Institute of Climate-Smart Agriculture, Bundesallee 50, 38116 Braunschweig, Germany

²formerly at: Institute for Groundwater Management, Dresden University of Technology, 01062 Dresden, Germany

Knowledge about the spatial variability of in situ denitrification rates (D_r (in situ)) and their relation to the denitrification capacity in nitrate-contaminated aquifers is crucial to predict the development of groundwater quality. Therefore, 28 push-pull ¹⁵N tracer tests for the measurement of in situ denitrification rates were conducted in two sandy Pleistocene aquifers in Northern Germany.

The 15 N analysis of denitrification derived 15 N labelled N $_2$ and N $_2$ O dissolved in water samples collected during the push-pull 15 N tracer tests was performed by isotope ratio mass spectrometry (IRMS) in the lab and additionally for some tracer tests online in the field with a quadrupole membrane inlet mass spectrometer (MIMS), in order to test the feasibility of on-site real-time 15 N analysis. Aquifer material from the same locations and depths as the push-pull injection points was incubated and the initial and cumulative denitrification after one year of incubation ($D_{\text{cum}}(365)$) as well as the stock of reduced compounds (SRC) was compared with in situ measurements of denitrification. This was done to derive transfer functions suitable to predict $D_{\text{cum}}(365)$ and SRC from $D_{\text{r}}(\text{in situ})$.

 $D_{\rm r}$ (in situ) ranged from 0 to 51.5 $\mu{\rm g}\,{\rm N}\,{\rm kg}^{-1}\,{\rm d}^{-1}$. Denitrification rates derived from onsite isotope analysis using membrane-inlet mass spectrometry satisfactorily coincided with laboratory analysis by conventional isotope ratio mass spectrometry, thus proving the feasibility of in situ analysis. $D_{\rm r}$ (in situ) was significantly higher in the sulphidic zone of both aquifers compared to the zone of non-sulphidic aquifer material. Overall, regressions between the $D_{\rm cum}$ (365) and SRC of the tested aquifer material with $D_{\rm r}$ (in situ) exhibited only a modest linear correlation for the full data set. But the predictability of $D_{\rm cum}$ (365) and SRC from $D_{\rm r}$ (in situ) data clearly increased for aquifer samples from the zone of NO_3^- -bearing groundwater.

In the NO_3^- -free aquifer zone a lag phase of denitrification after NO_3^- injections was observed, which confounded the relationship between reactive compounds and in situ denitrification activity. This finding was attributed to adaptation processes in the mi-

BGD

Discussion Paper

Discussion Paper

Discussion Pape

11, 16527-16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page

Abstract Introduction

Conclusions References

Tables Figures

I∢ ≯I

→

Back Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



crobial community after NO₃ injections. Exemplarily, it was demonstrated that the microbial community in the NO₃-free zone close below the NO₃-bearing zone can be adapted to denitrification by amending wells with NO3-injections for an extended period. In situ denitrification rates were 30 to 65 % higher after pre-conditioning with NO₃. Results from this study suggest that such pre-conditioning is crucial for the measurement of D_r (in situ) in deeper aquifer material from the NO_3^- -free groundwater zone and thus for the prediction of $D_{cum}(365)$ and SRC from $D_r(in situ)$.

Introduction

Denitrification, the microbial mediated reduction of nitrate (NO₂) and nitrite (NO₂) to the nitrogen gasses nitric oxide (NO), nitrous oxide (N₂O) and dinitrogen (N₂) is important to water quality and chemistry at landscape, regional and global scales (Groffman et al., 2006). NO₃ is quantitatively the most abundant reactive nitrogen (Nr¹) species. Diffuse NO₃ emissions from the agricultural sector are the dominant source of Nr fluxes to aquifers. Denitrification in aquifers, reviewed e.g. by Korom (1992), Hiscock et al. (1991), Burgin and Hamilton, (2007) and Rivett et al. (2008), ranges from 0 to 100% of total NO₃ input with a high spatial variability (Seitzinger et al., 2006). This leads to the question, how individual aquifers will respond to the anthropogenic NO₃ pollution of groundwater. Not only the questions how rates of denitrification will respond to Nr loading (Seitzinger et al., 2006) and where and how long denitrification in aquifers can remediate NO₃ pollution (Kölle et al., 1985) are of importance. Since continuous NO₃ input via seepage waters leads to ongoing exhaustion of the reductive capacity of aquifers. This can be a problem for keeping the standard for NO₃ in drinking water

BGD

11, 16527-16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page

Abstract Introduction

Conclusions References

> **Tables Figures**

Close

Printer-friendly Version

¹The term reactive nitrogen is used in accordance with Galloway et al. (2004) in this work, and includes all biologically or chemically active N compounds like reduced forms (e.g. NH₂, NH₄⁺), oxidized forms (e.g. NO_x, HNO₃, N₂O, NO₃⁻) and organic compounds (e.g. urea, amines, proteins ...).

(< $50\,\mathrm{mg\,L^{-1}}$, Drinking Water Directive $98/83/\mathrm{EC}$) and due to possible eutrophication of water bodies (Vitousek et al., 1997). But NO_3^- can also mobilise unforeseen species such as uranium (U) which can be mobilised if NO_3^- reaches reduced aquifer zones (Senko et al., 2002; Istok et al., 2004). Therefore, knowledge about the denitrification capacity of aquifers is needed to predict the possible development of groundwater quality.

The presented study continues previous research on denitrification rates measured in two sandy Pleistocene aguifers in Northern Germany (Fuhrberger Feld aguifer (FFA) and the aguifer of Großenkneten (GKA)). Frind et al. (1990) reported that due to lithotrophic denitrification, NO₃ has a half-life of 1 to 2 years in the deeper zone (below 5 to 10 m) of the well investigated Fuhrberger Feld aguifer. Weymann et al. (2010) reported very low denitrification rates with values as low as 4 μg N kg⁻¹ d⁻¹ in the surface near groundwater, in the organotrophic denitrification zone of the same aquifer. In a recent study, Eschenbach and Well (2013) measured median denitrification rates of 15.1 and 9.6 mg N kg⁻¹ yr⁻¹ during one year anaerobic incubations of FFA and GKA aquifer samples, with significantly higher denitrification rates in the deeper parts of both aquifers. This study showed that the cumulative denitrification after prolonged incubation of aquifer samples is correlated to the stock of reduced compounds (SRC). Similar results had been earlier obtained for another aquifers in Northern Germany (Konrad, 2007). While we found close correlations between initial laboratory denitrification rates and the SRC in aquifer zones were NO₃ is present in groundwater, samples from NO_3^- -free groundwater zone showed a lag time of denitrification of several weeks during incubations (Eschenbach and Well, 2013) possibly due to the initial absence of denitrifying enzymes. These findings demonstrate, that the SRC can be estimated from denitrification rates in case the microbial community is adapted to denitrification (Eschenbach and Well, 2013).

In situ denitrification rates can be measured using single well push-pull tests where a test solution containing solutes of interest is rapidly injected into a well (push-phase) and process information is obtained from analysing groundwater collected during the BGD

11, 16527-16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page

Abstract Introduction

Conclusions References

Tables Figures

4 ▶|

→

Back Close

Full Screen / Esc

Printer-friendly Version



Abstract

Introduction

Conclusions

BGD

11, 16527–16572, 2014

Predicting the

denitrification

capacity of sandy

aquifers

W. Eschenbach and

R. Well

Title Page

References

Tables

Figures

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



subsequent pull-phase. These tests, perhaps first used for in situ measurement of denitrification rates by Trudell et al. (1986), have proven to be a relatively low-cost instrument to obtain quantitative information about several aquifer properties. This method was applied in a variety of studies to derive in situ denitrification rates indirectly by the ₅ measurement of NO₃ depletion during push-pull tests (Trudell et al., 1986; Istok et al., 1997, 2004; Schroth et al., 2001; McGuire et al., 2002; Harris et al., 2006). In comparison only a limited number of studies directly measured denitrification rates from the gaseous denitrification products (Sanchez-Perez et al., 2003; Kneeshaw et al., 2007; Well and Myrold, 2002, 1999; Addy et al., 2002; Well et al., 2003; Addy et al., 2005; Kellogg et al., 2005; Konrad, 2007). Beside the study of Konrad (2007) these push-pull tests were only conducted in surface near groundwater.

Well et al. (2005) showed that in situ denitrification rates measured with the pushpull ¹⁵N tracer method in the saturated zone of hydromorphic soils agreed relatively well with denitrification rates measured in parallel soil samples. Konrad (2007) reported a close correlation between in situ denitrification rates and the cumulative denitrification after at least one year of incubation based on a small number of only 5 comparisons, so the data set was relatively small to derive robust transfer functions.

Since denitrification is a microbial mediated redox reaction, the composition, activity and amount of microbes in aquifers should directly influence the measured denitrification rates during single well push-pull tests. It is known that steep gradients in the composition of microbial communities occur in aquifers resulting from the distribution and availability of electron donors and acceptors in aguifers (Kölbelboelke et al., 1988; Griebler and Lueders, 2009; Santoro et al., 2006). Law et al. (2010) reported substantial changes in the microbial community composition after the begin of denitrification and the transition from denitrification to Fe(III)-reduction within incubated aguifer material. Higher microbial activities after bio stimulation of indigenous microorganisms by the injection of electron donors into aquifers was reported by Istok et al. (2004), Kim et al. (2005) and Kim et al. (2004). Compared with preceding push-pull tests at the same groundwater monitoring wells, the multiple injection of electron donors increased

the reduction rates of NO_3^- , pertechnetate (Tc(VII)) and U(VI) measured during subsequent push-pull tests in a shallow unconfined silty-clayey aquifer (Istok et al., 2004). Trudell et al. (1986) found increasing denitrification rates during a 12 day push-pull test in NO_3^- -free groundwater suggesting that the microbial community needed a certain time to adapt to the electron acceptor NO_3^- before denitrification could proceed at a rate equivalent to the availability of reduced compounds. At all the possible influence of the location of push-pull tests within aquifers regarding the presence or absence of NO_3^- on measured in situ denitrification rates in groundwater has not been addressed so far.

Overall, the performance of previous push-pull studies suggests that this approach may be suitable to deliver in situ denitrification data that reflect the reduction capacity of the aquifer, i.e. it might be used to estimate SRC without the need for collecting aquifer material.

To test if ¹⁵N push-pull tests could be evaluated during the course of experiments directly in the field, a membrane inlet mass spectrometer (MIMS) was used during 5 push-pull tests at two monitoring wells for direct field measurements of ¹⁵N labelled denitrification products (see Supplement). The main advantages of MIMS with respect to the conventional IRMS approach is that MIMS is low-priced compared to IRMS and results can be obtained in the course of experiments directly in the field. Sampling intervals can thus be adapted to get more precise rates. Moreover, the length of the pull phase can be limited to the duration of clearly increasing (N₂+N₂O)_{den} concentrations to save working time. Finally, the relatively low cost and simple handling of the MIMS system are favourable to enable extensive application of the ¹⁵N push-pull approach to explore denitrification capacities of aquifers.

This study is the second part of a combined approach (a) to quantify exhaustibility of the denitrification capacity in aquifers, (b) to investigate controlling factors and derive predictive models during incubation experiments, and (c) to check if the cumulative denitrification measured after one year of incubation ($D_{\text{cum}}(365)$) (Eschenbach and Well, 2013) can be derived from in situ denitrification rates measured with push-pull

BGD

11, 16527-16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page

Abstract Introduction

Conclusions

Tables Figures

l∢ ⊳i

•

Back Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



Printer-friendly Version

tracer tests. Here a study on objective (c) is presented. The specific objectives of this study are (i) to measure in situ denitrification rates with push-pull ¹⁵N tracer tests at groundwater monitoring wells, (ii) to develop regression models to predict D_{cum} (365) as well as the stock of reduced compounds from in situ denitrification rates (iii) to test an approach to adapt the microbial community in NO₃-free aquifer zones to NO₃ as a new available electron donor during experiments as a means of conditioning prior to subsequent push-pull ¹⁵N tracer tests. Additionally (vi) the suitability of MIMS for online field analysis during ¹⁵N tracer tests was tested (Supplement).

Materials and methods

2.1 Study sites

In situ measurements of denitrification were conducted in the Fuhrberger Feld aguifer (FFA) and the Großenkneten aquifer (GKA). Both aquifers are located in drinking water catchment areas in the North of Germany. The FFA is situated about 30 km NE of the city of Hanover and the GKA about 30 km SW of the city of Bremen. Both aquifers consist of carbonate free, Quaternary sands and the deeper parts of the GKA additionally of carbonate free marine sands (Pliocene). The thickness of the FFA and GKA is 20 to 40 and 60 to 100 m, respectively. Both aguifers are unconfined and contain unevenly distributed amounts of microbial available sulphides and organic carbon. Intense agricultural land use leads to considerable NO₃ inputs to the groundwater of both aquifers (Böttcher et al., 1989; van Berk et al., 2005; Schuchert, 2007). Groundwater recharge is 250 mm yr⁻¹ in the FFA (Wessolek et al., 1985) and 200 to 300 mm yr⁻¹ in the GKA (Schuchert, 2007).

Evidence of an intense ongoing denitrification within the FFA is given by NO₃ and redox gradients (Böttcher et al., 1992) as well as excess-N₂ measurements (Weymann et al., 2008). The FFA can be divided into two hydro-geochemical zones, the zone of organotrophic denitrification near the groundwater surface with organic carbon (C_{org})

BGD

11, 16527-16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page

Abstract Introduction

Conclusions References

> **Tables Figures**

Back Close

Full Screen / Esc

Interactive Discussion

as electron donor and a deeper zone of predominantly lithotrophic denitrification with pyrite as electron donor (Böttcher et al., 1991, 1992). Detailed information about the FFA is given by Strebel et al. (1992), Frind et al. (1990) and von der Heide et al. (2008). The geological structure of the GKA is described in Howar (2005) and Wirth (1990). Extended zones with oxidizing and reducing conditions in the groundwater are evident in the GKA (van Berk et al., 2005) but their distribution within the aquifer is more complex as in the FFA and denitrification is known to occur in the zone of reduced groundwater (van Berk et al., 2005). Own excess-N₂ measurements (Well et al., 2012) at monitoring wells prove intense denitrification within the GKA. But there are no studies on the type of denitrification in this aquifer.

2.2 Single well push-pull ¹⁵N tracer tests

2.2.1 Well types and sampling procedure

To quantify in situ denitrification rates (D_r (in situ)), a total of 28 single well push-pull ¹⁵N tracer tests, afterwards referred to as push-pull tests, were performed in the FFA and GKA (Table 1) by injecting ¹⁵N labelled NO_3^- tracer solution into groundwater monitoring wells. In the FFA, push-pull tests were conducted at multilevel wells consisting of PE tubings (4 mm ID) (Böttcher et al., 1985). Each of these tubings were connected to a filter element at the respective depth. In the GKA, two types were used, (i) conventional groundwater monitoring wells (101 mm ID) with 1 to 4 m long filter screens and (ii) multilevel wells (CMT multilevel system, Soilinst, Georgetown, Canada) consisting of PE pipes with 3 individual channels (13 mm ID) with 25 cm long filter screens at the end. Each channel ended in a different depth. To allow a direct comparison with a previous laboratory incubation study (Eschenbach and Well, 2013), wells from the same locations and with filter screens at the same depth where the aquifer samples had been collected were selected in the FFA and GKA. In situ experiments were conducted principally as described in previous studies (Addy et al., 2002; Trudell et al., 1986; Well et al., 2003).

BGD

11, 16527–16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page

Abstract Introduction

Conclusions References

Tables Figures

l**⊲** ▶l

→

Back Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



For sampling multilevel wells, groundwater and tracer solution were extracted with a peristaltic pump (Masterflex COLE-PARMER, Vernon Hills, USA). A submersible pump (GRUNDFOS MP1, Bjerringbro, Denmark) was used for common groundwater monitoring wells. During sampling, an outflow tubing with the extracted groundwater or tracer solution was placed at the bottom of 26 or 120 mL serum bottles (multilevel wells and common groundwater monitoring wells, respectively). After an overflow of at least three times the volume of these bottles, the tubing was removed and the bottles were immediately sealed air tight with grey butyl rubber septa (ALTMANN, Holzkirchen, Germany) and aluminium crimp caps. 4 replications were collected per sampling. Groundwater was sampled from the injection depth prior to each push-pull test.

2.2.2 Push-pull tests

A single well push-pull test consists of the injection of a tracer solution into a monitoring well (push-phase) and the extraction of the mixture of test solution and groundwater from the same well (pull-phase).

5 Push-phase

To prepare the tracer solution, 50 L of groundwater were extracted from multilevel wells (FFA and GKA) or 220 L at common groundwater monitoring wells (GKA) for each push-pull test (Fig. 1). The groundwater was pumped to a stainless steel storage container (Type BO 220 L, SPEIDEL, Ofterdingen, Germany), which was equipped with a floating lid to avoid gas exchange with the atmosphere and thus maintain the dissolved gas composition of the extracted groundwater. After extraction, a stock solution of deionised water (100 mL) with dissolved ¹⁵N labelled potassium nitrate (KNO₃ with 60 atom % ¹⁵N) and potassium bromide (KBr) was added to attain a concentration of 10 mg ¹⁵N labelled NO₃-N L⁻¹ and 10 mg Br⁻ L⁻¹, respectively. The mixture of the stock solution and the extracted groundwater is hereinafter referred to as tracer solution. The tracer solution was mixed for 1 h with a submersible pump (Gigant, Eijkelkamp,

BGD

11, 16527-16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page

Abstract Introduction

Conclusions

Tables Figures

l∢ ≯l

Close

•

Full Screen / Esc

Back

Printer-friendly Version

Interactive Discussion



Giesbeek, the Netherlands) within the stainless steel storage container. The extracted groundwater from the NO_3^- -bearing groundwater zone (NO_3^- -bearing zone) contained varying concentrations of NO_3^- (Table 2). Consequently, the NO_3^- in the tracer solution of these push-pull tests was a mixture of natural and ^{15}N enriched NO_3^- and NO_3^- concentrations in these tracer solutions were > 10 mg NO_3^- -N L^{-1} (see discussion about influence of NO_3^- concentrations on denitrification rates in Sect. 4.2 and in Eschenbach and Well, 2013).

During injection, the outflow of the stainless steel storage container was connected with Tygon[®] tubings to the selected depths of the multilevel wells. For common ground-water monitoring wells the submersible pump was connected with a pump riser pipe and an inflatable packer (Packer set, UIT Umwelt- und Ingenieurtechnik GmbH, Dresden, Germany). The packer was installed within the groundwater monitoring well to prevent mixing of the injected tracer solution with the water column in the groundwater monitoring well (Fig. 1). The packer was inflated with air to a pressure of 1 bar above the pressure of the overlying water column. The inflated packer and the pump riser pipe remained during the entire tracer test within the groundwater monitoring well. The pump riser pipe was connected with a PVC hose (13 mm ID) to the stainless steel container. For both types of monitoring wells, the tracer solution was injected gravimetrically.

Pull-phase

The common groundwater monitoring wells in the GKA were constantly sampled at 12 h intervals. The multilevel wells in the FFA were sampled every 12 h during night and every 3 to 4 h during day to investigate temporal patterns more detailed. The multilevel wells were more suitable for this, due to their smaller dead volumes and lower extraction rates. The pull phase of the conducted tracer test lasted maximal 72 h. The first sampling was performed immediately after injection. Prior to each sampling, an amount of tracer solution sufficient to replace the dead volume of the groundwater monitoring well was extracted. In total, 4 and 30 to 60 L were extracted per sampling form multilevel

BGD

11, 16527-16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page

Abstract Introduction

Conclusions References

Tables Figures

· ·

→

Back Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



and groundwater monitoring wells, respectively. For common groundwater monitoring wells the sampling volume differed because of different lengths of filter screens and resulting different dead volumes. During extraction, groundwater temperature, dissolved oxygen, pH and electrical conductivity were measured with sensors (pH/Oxi 340i and pH/Cond 340i, WTW Wissenschaftlich-Technische Werkstätten GmbH, Weilheim, Germany) installed in a flow-through chamber.

2.3 Incubation of aquifer material

Laboratory experiments were performed to compare denitrification rates measured during laboratory anaerobic incubation ($D_{\rm r}(365)$) with in situ denitrification rates. The incubated aquifer material originated from the same location and depths as the filter screens of the push-pull test injection points. The aquifer material was sampled and incubated as described in detail in Eschenbach and Well (2013).

Briefly, aquifer material from both aquifers were collected between 2 to 68 m below soil surface. The aquifer samples were incubated in transfusion bottles, in 3 to 4 replications. 15 N labelled KNO $_3$ solution was added and the transfusion bottles were sealed airtight. To ensure anaerobic conditions during incubation, the headspaces of the transfusion bottles were evacuated and flushed with pure N_2 . Afterwards, the samples were incubated for one year in the dark at $10\,^{\circ}$ C, which is approximately the groundwater temperature in both aquifers. The transfusion bottles were shaken manually two times a week to mix sediment and batch solution. The headspace and the supernatant batch solution in the transfusion bottles were sampled at days 1, 2, 7, 84, 168 and 365 of incubation.

2.4 Pre-conditioning of wells in the NO₃⁻-free zone of the FFA

To stimulate denitrification in the NO_3^- -free zone with suspected lack of active denitrifiers (Eschenbach and Well, 2013) groundwater monitoring wells were amended by repeated injections of groundwater with added NO_3^- of natural ^{15}N abundance. Injec-

BGD

11, 16527–16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page

Abstract Introduction

Conclusions References

Tables Figures

I4 ≯I

•

Close

Full Screen / Esc

Back

Printer-friendly Version

Interactive Discussion



Printer-friendly Version

Interactive Discussion



tions were designed to maintain elevated NO₃ levels in the vicinity of the filter screens during a period of several weeks. This was done to test if in situ denitrification rates measured in these wells after pre-conditioning would reflect the average denitrification rates measured during one year of incubation of corresponding aguifer samples (Eschenbach and Well, 2013).

Pre-conditioning was performed at 4 depths in the NO₃-free groundwater zone at multilevel well B4 in the FFA, that had been previously tested without pre-conditioning. Therefore 800 L of NO₃-free reduced groundwater were extracted from a groundwater monitoring well, with a filter screen at 7 to 8 m depth below soil surface, which is located 30 m west of multilevel well B4, into a 800 L tank container (IBC Tank Wassertank Container 800 L, Barrel Trading GmbH & Co. KG, Gaildorf, Germany) using a drill pump (Wolfcraft Bohrmaschinenpumpe 8 mm Schaft, Wolfcraft GmbH, Kempenich, Germany). The drill pump was connected with a PVC hose (13 mm ID) to the groundwater monitoring well and to the 800 L tank container. The extracted groundwater was supplemented with KNO₃ of natural ¹⁵N abundance to a concentration of 10 mg NO₃-NL⁻¹. Approximately 40 L of this mixture were injected weekly into each of the depths 7, 8, 9 and 10 m below soil surface, respectively, at multilevel well B4. The injection rate was approx. 1 Lmin⁻¹. For 7 and 8 m depth the peristaltic pump and for 9 and 10 m depth the drill pump were used for injection and both pumps were connected with Tvgon® tubings to the selected depths of the multilevel well. The first injection took place on 22 February 2011 and the last on 22 March 2011. In total, 5 pre-conditioning injections were conducted at the 4 depths. Subsequently, 4 push-pull tests were performed in the previously pre-conditioned injection depths as described above between 29 March and 1 April 2011.

BGD

11, 16527–16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page **Abstract** Introduction

Conclusions References

> Tables **Figures**

Back Close Full Screen / Esc

2.5.1 Isotope analysis of dissolved N₂ and N₂O

Water samples sampled during push-pull tests were adjusted to 25 °C and a headspace was generated within the serum bottles by the injection of 15 or 40 mL of ambient air into the 26 and 115 mL serum bottles, respectively, replacing the same volume of sample solution. The replaced solution was directly transferred into 20 mL PE vials and frozen for later NO_3^- and SO_4^{2-} analysis. After headspace generation the serum bottles were agitated for 3 h on a horizontal shaker at a constant temperature of 25 °C to equilibrate the dissolved gases with the headspace gas. Finally, 13 mL of the headspace gas of each serum bottle were extracted with a plastic syringe and then transferred to an evacuated 12 mL sampling vial (Exetainer® Labco, High Wycombe, UK), giving a slight positive pressure within the sampling vial. The sampled nitrogen gases in the 12 mL vials were then a mixture of N_2 and N_2O gained from atmosphere and from denitrification, respectively.

The 15 N analysis of gas samples was performed by isotope ratio mass spectrometry (IRMS) at the Centre for Stable Isotope Research and Analysis in Göttingen, Germany using a Delta V advantage IRMS (Thermo Scientific, Bremen, Germany) following the method described in Well et al. (2003). Analysis included reduction of N_2O to N_2 prior to IRMS entrance. The sum of N_2 and N_2O isotopologues was thus detected as N_2 in the mass spectrometer. In the following, the sum of denitrification derived N_2 and N_2O is referred to as $(N_2+N_2O)_{den}$. The 15 N abundance of (N_2+N_2O) was derived from the measured 29/28 molecular ion mass ratio. We analysed replicate samples, one was equilibrated by electrodeless discharge and the other untreated (Well et al., 1998). This allowed calculating $(N_2+N_2O)_{den}$ as well as the 15 N abundance in NO_3^- undergoing denitrification. N_2O was measured using a gas chromatograph (Fisons GC 8000, Milan Italy) equipped with a split-injector and an electron capture detector and a HP-Plot Q column (50 m length × 0.32 mm ID; Agilent Technologies, Santa Clara, USA) kept at

BGD

Discussion Paper

Discussion Paper

Discussion Pape

11, 16527-16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page

Abstract Introduction

Conclusions References

Tables Figures

I∢ ≯I

•

Back Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



38°C. Gas analysis was completed within two weeks after the respective push-pull tests. The concentrations of denitrification derived ¹⁵N labelled N₂ and N₂O in the gas samples were calculated as described by Well and Myrold (1999) and Well et al. (2003), respectively. The concentration of N₂O in the added atmospheric air was taken into 5 account when calculating denitrification derived N2O in the sample. The measured molar concentrations of N₂ and N₂O in the headspace samples were converted into dissolved gas concentrations using gas solubilities given by Weiss (1970) and Weiss and Price (1980) and taking into account the temperature, headspace pressure and the liquid-to-headspace volume ratio during equilibration of dissolved gases with the headspace gases in the serum bottles.

Additionally to the standard IRMS analysis of (N₂+N₂O)_{den} ¹⁵N labelled denitrification products were measured with a membrane inlet mass spectrometer (MIMS) during 5 push-pull tracer tests directly in the field (see Supplement).

2.5.2 Analysis of NO_3^- , SO_4^{2-} and Br^-

NO₃ concentrations in the water samples were determined photometrically in a continuous flow analyser (Skalar, Erkelenz, Germany). SO_4^{2-} concentrations were analysed by potentiometric back-titration of excess Ba2 ions remaining in the solution after addition of a defined amount of BaCl₂ in excess to SO_4^{2-} . SO_4^{2-} precipitated as BaSO₄²⁻. The original SO₄²⁻ concentration was then analysed by potentiometric back-titration of the excess Ba₂ ions remaining in the solution using EDTA as titrant. Possible interfering metal cations were removed from the samples prior to this analysis by cation exchange. Bromide (Br⁻) was analysed with an Inductively Coupled Plasma-Atomic Emission Spectrometer (ICP-AES, Spectro Analytical Instruments, Kleve, Germany) after stabilizing the aliquot of the analysed water samples with 10% HNO₃.

BGD

11, 16527–16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page **Abstract** Introduction

Conclusions References

Figures Tables

Close

Back Full Screen / Esc

Printer-friendly Version

Interactive Discussion



Measured concentrations of (N₂+N₂O)_{den} were converted from the unit (μg N L⁻¹) to (μg N kg⁻¹) under the following assumptions: (i) the average density of the solid aquifer material is 2.65 g cm⁻³ and (ii) the effective porosity of the aquifer material was estimated to be 0.3 from literature values for sediments of similar grain size distribution (Kollmann, 1986), with a range of uncertainty of 0.2 to 0.4, respectively.

The concentrations of $(N_2+N_2O)_{den}$ measured during the push-pull tests were corrected for dilution caused by dispersion, diffusion and the tortuosity of the pores. To do this the dilution factor $(F_{dil}(ti))$ (Eq. 1) was derived from the concentration changes of the conservative tracer Br⁻ during the push-pull tests as proposed by Sanches-Perez et al. (2003):

$$F_{\text{dil}}(ti) = \frac{[Br^{-}]_{t0}}{[Br^{-}]_{ti}}$$
 (1)

where $\mathrm{Br}^-{}_{t0}$ and $\mathrm{Br}^-{}_{ti}$ are the Br^- concentrations of the injected tracer solution and the sampled tracer solution at sampling time ti, respectively. The corrected concentrations of $(\mathrm{N_2} + \mathrm{N_2O})_{\mathrm{den}}$ are then obtained by multiplying the uncorrected concentrations of $(\mathrm{N_2} + \mathrm{N_2O})_{\mathrm{den}}$ at time ti with $F_{\mathrm{dil}}(ti)$. Denitrification rates were calculated from the tangent of dilution corrected time courses of $(\mathrm{N_2} + \mathrm{N_2O})_{\mathrm{den}}$ concentrations at time intervals with the steepest increase during the respective push-pull test (Sanchez-Perez et al., 2003; Istok et al., 2004).

2.7 Detection limit and precision of $(N_2 + N_2O)_{den}$ measurements

The detection limit of 15 N analysis was calculated as the minimum amount of 15 N labelled $(N_2+N_2O)_{den}$ mixed with the given background of headspace N_2 of natural 15 N abundance necessary to increase the measured $^{29}N_2/^{28}N_2$ ratio to fulfil the following

BGD

Paper

Paper

Discussion Paper

Discussion Pape

11, 16527-16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page

Abstract Introduction

Conclusions References

Tables Figures

l∢ ≯l

- ★

Back Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



$$r_{\rm sa} - r_{\rm st} \ge 3 \times \rm sdr_{\rm st} \tag{2}$$

where $r_{\rm sa}$ and $r_{\rm st}$ are the $^{29}{\rm N}_2/^{28}{\rm N}_2$ ratios in sample and standard, respectively and sdr_{st} is the SD of repeated $r_{\rm st}$ measurements. The $r_{\rm st}$ values were analysed with IRMS by measuring repeated air samples. Under the experimental conditions, the detection limit for the amount of $({\rm N}_2 + {\rm N}_2 {\rm O})_{\rm den}$ was 5 and 1 μ g N L⁻¹ for samples in 26 and 115 mL serum bottles, respectively, depending on the different ratio of liquid sample to headspace in the respective serum bottles.

The mean coefficient of variation (CV) of concentration measurements of $(N_2+N_2O)_{den}$ (µg NL⁻¹) in 3 replicates per sampling event during all push-pull tests was 0.18%. The conversion of concentration data from the unit (µg NL⁻¹) to (µg Nkg⁻¹) increased the mean CV significantly to 0.49%. (The mean CV after conversion to (µg Nkg⁻¹) was calculated from the 3 concentrations resulting from the range of effective porosity values (see Supplement).)

2.8 Statistical analysis and modelling

Statistical analysis and regression modelling was conducted with WinSTAT for MS Excel Version 2000.1 (R. Fitch Software, Bad Krozingen, Germany). Experimental data (x) was converted into Box–Cox transformed data $(f^{B-C}(x))$ according to Eq. (3) using different lambda coefficients (λ) to achieve a normal like distribution of experimental data within the different data sets.

$$f^{\text{B-C}}(x) = \frac{(x^{\lambda} - 1)}{\lambda} \tag{3}$$

Box-Cox transformations were performed with the statistic software STATISTICA 8 (StatSoft, Tulsa, USA). Simple linear regression analysis was conducted to evaluate

BGD

11, 16527–16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page

Discussion Paper

Discussion Paper

Discussion Pape

Abstract Introduction

Conclusions References

Tables Figures

I**4** ►I

→

Back Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



Conclusions

Introduction References

Figures

BGD

11, 16527–16572, 2014

Predicting the

denitrification

capacity of sandy

aquifers

W. Eschenbach and

R. Well

Title Page

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

quantitative relations between in situ denitrification rates (D_r (in situ)) and various sediment parameters of corresponding aquifer material measured in the laboratory (Eschenbach and Well, 2013). Normal distribution of the measured parameters within the different data sets and the residuals of linear regressions were tested with the Kolmogorov–Smirnov-Test, normal distribution was assumed at the P > 0.05 level, with the null hypothesis that the tested parameter was normal distributed. The uniform distribution of residuals of regressions was checked with scatter plots of residuals vs. independent variables of the respective regression analysis. This was done to ensure homoscedasticity during regression analysis, i.e. to ensure that the least-squares method yielded best linear estimators for the modelled parameter. To use the regression functions given in the result section with own data, the experimental values have to be transformed according to Eq. (3) with the lambda coefficients given in Table S2 in the Supplement.

Differences between partial data sets were considered significant at the P < 0.05level (Kruskal-Wallis test (kw) with the null hypothesis that both partial data sets belong to the same population).

Model sediment properties using regression functions with D_r (in situ)

In situ denitrification rates (D_r (in situ)) measured during push-pull tests were used to model parameters of the investigated aquifer samples measured in the laboratory. These parameters were: (i) the cumulative denitrification after one year of incubation $(D_{cum}(365))$, (ii) the stock of reduced compounds (SRC) and (iii) several sediment parameters like water soluble organic carbon (C_{hws}), the fraction of KMnO₄ labile organic carbon (C_I), total sulphur (total-S) and total organic carbon (C_{org}). D_{cum} (365) is the cumulative amount of denitrification products per kg dry weight of incubated aquifer material at the end of one year of anaerobic incubation (mg N kg⁻¹). The SRC is the amount of sulphides and Cora converted into N equivalents (mgNkg-1) according to their potential ability to reduce NO₃ to N₂ (Eschenbach and Well, 2013). These sedi-

16543

Tables

Abstract









ment parameters and denitrification rates were analysed during a laboratory incubation study with aquifer samples from the FFA and GKA (Eschenbach and Well, 2013).

The aquifer samples were collected from drilled material obtained during well construction of groundwater monitoring and multilevel wells in the FFA and GKA. The analysed aquifer samples originated from depth intervals of approximately 1 m above to 1 m below filter screens or filter elements of respective groundwater monitoring or multilevel wells, used for push-pull tests (Table 1).

Results

Grouping of push-pull test measuring points

Wells were grouped according to the redox state of groundwater and aquifer properties into sub data sets of in situ denitrification rates (D_r(in situ)) measured in the NO₃bearing and NO₃-free groundwater zone (NO₃-bearing and NO₃-free zone, respectively) and into D_r (in situ) measured in the zone of non-sulphidic, sulphidic and transition zone aquifer material (Tables 1 and 2). Sulphidic and non-sulphidic aquifer material was distinguished using the sulphate formation capacity (SFC (mgSkg⁻¹yr⁻¹)) of incubated aquifer material (Eschenbach and Well, 2013). Samples with a SFC> 1 mg SO_{4}^{2} -S kg⁻¹ yr⁻¹ during incubation were assigned sulphidic. The transition zone was defined as aquifer material from the region where sulphides were present, but groundwater still contained NO₃. It follows, that the NO₃-bearing groundwater zone comprises the zone of sulphidic aquifer material and the transition zone. For a detailed description of the classification of aquifer material see Eschenbach and Well (2013).

BGD

11, 16527–16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page **Abstract** Introduction

Conclusions References

> Tables **Figures**

Back Close

Full Screen / Esc

Printer-friendly Version



The non-sulphidic zone of both aquifers exhibited the lowest mean D_r (in situ) (1.04 μ g N kg⁻¹ d⁻¹) of all partial data sets (Table 4) and statistical significant differences (kw: P < 0.05) occurred with the full and all partial data sets except D_r (in situ) measured in the GKA and in the NO $_3^-$ -bearing zone of both aquifers. The other partial data sets exhibited no significant differences between one another. Mean D_r (in situ) of the transition zone (9.32 μ g N kg⁻¹ d⁻¹) was slightly higher than in the sulphidic zone of both aquifers.

Except for the multilevel well B6 in 6 m depth, all push-pull injection points with O_2 concentrations above 1 mg O_2 L $^{-1}$ in the groundwater exhibited D_r (in situ) below 0.75 μ g N kg $^{-1}$ d $^{-1}$ (Tables 2 and 3) and aquifer material from this locations were assigned to non-sulphidic aquifer material during laboratory incubations (Eschenbach and Well, 2013).

 $D_{\rm r}$ (in situ) measured after pre-conditioning of push-pull injection points at multiple well B4 (FFA) (67.83 to 152.70 μ g N kg⁻¹ d⁻¹) were 30 to 65 times higher than $D_{\rm r}$ (in situ) measured one year before without pre-conditioning (2.76 and 2.28 μ g N kg⁻¹ d⁻¹) (Table 3).

Among the total of 28 push-pull tests, 24 were conducted without pre-conditioning from which twelve were located in the NO_3^- -bearing and twelve in the NO_3^- -free zone of both aquifers, respectively. Among the 12 push-pull tests in the NO_3^- -free zone all of the 5 FFA locations showed an exponential increase of $(N_2+N_2O)_{den}$ during push-pull tests, whereas in the GKA this was only the case in two to three of the 7 GKA locations. In contrast to this, only 2 out of 12 push-pull tests in the NO_3^- -bearing zone of both aquifers exhibited exponential increases and these push-pull tests were located in the transition zone of multilevel well B2. The two push-pull tests at multilevel well

BGD

Discussion Paper

Discussion Paper

Discussion Paper

11, 16527–16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page

Abstract Introduction

Conclusions References

Tables Figures

▼ ►|

→

Back Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



B4 (NO_3^- -free zone of the FFA) showed an exponential increase of (N_2+N_2O)_{den}. All other push-pull tests in the NO_3^- -bearing zone exhibited almost linear trends. After preconditioning at the same depths of multilevel well B4 in the NO_3^- -free zone, the time course of denitrification products was drastically different compared to the initial tests with a much steeper and initially almost linear trend (Fig. 4).

3.3 Relationship between D_r (in situ), D_{cum} (365) and aquifer parameters

3.3.1 Comparison of D_r (in situ) and D_{cum} (365)

 $D_{\rm r}({\rm in~situ})$ was compared with mean denitrification rates during 365 days of laboratory incubation ($D_{\rm r}(365)$) (Eschenbach and Well, 2013) with aquifer material collected from the locations of the monitoring wells (see Sect. 2.3). $D_{\rm r}(365)$ was obtained by dividing cumulative ($N_2+N_2O_{\rm den}$ production ($D_{\rm cum}(365)$) by incubation time (365 d). $D_{\rm r}$ (in situ) was generally lower than $D_{\rm r}(365)$ (Fig. 3 and Table S1 Supplement). The means of the $D_{\rm r}({\rm in~situ})$ -to- $D_{\rm r}(365)$ ratio were calculated for the different partial data sets giving a range of 0.05 to 0.47, with the lowest and highest ratios for the data sets of GKA and transition zone push-pull tests, respectively (Table 4). In the transition zone, $D_{\rm r}({\rm in~situ})$ -to- $D_{\rm r}(365)$ ratios were significantly higher compared to the other data sets (kw: P < 0.05). $D_{\rm r}$ (in situ) of FFA aquifer material was statistical significant closer related to $D_{\rm r}(365)$ than $D_{\rm r}({\rm in~situ})$ measured in the GKA. The mean $D_{\rm r}({\rm in~situ})$ -to- $D_{\rm r}(365)$ ratio from the NO_3^- -bearing zone of both aquifers (0.23) was significantly larger than in the NO_3^- -free zone of both aquifers (0.1) (Table 4).

 $D_{\rm r}$ (in situ) after pre-conditioning (well B4, FFA) was comparable or higher than $D_{\rm r}$ (365) with $D_{\rm r}$ (in situ)-to- $D_{\rm r}$ (365) ratios of 0.73 to 2.76 (Fig. 3 and Table 4). $D_{\rm r}$ (in situ) was 30 to 65 times higher compared to values obtained without pre-conditioning at the same wells (Fig. 5 and Table 3).

BGD

11, 16527–16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page

Abstract Introduction

Conclusions

Tables Figures

I∢ ⊳I

•

Back Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



3.3.2 Regression models to predict D_{cum} (365), SRC and denitrification relevant aguifer parameters from D_r (in situ)

Simple linear regression analysis was applied to obtain regression models for the prediction of $D_{\text{cum}}(365)$ from $D_{\text{r}}(\text{in situ})$ for the full and partial data sets. The goodness of fit of the regression models given by the correlation coefficient (R) and the average ratio of calculated $D_{\text{cum}}(365)$ to measured $D_{\text{cum}}(365)$ for the full and partial data sets.

The goodness of fit of regression models to predict $D_{\text{cum}}(365)$ by $D_{\text{r}}(\text{in situ})$ varied for the various sub data sets from no fit in the sulphidic zone and a good approximation of $D_{\text{cum}}(365)$ by $D_{\text{r}}(\text{in situ})$ in the NO₃-bearing zone (R = 0.04 and R = 0.84, respectively, Table 5). For the full data set, the goodness of fit was modest (R = 0.62) resulting in a wide range of deviations between calculated and measured D_{cum} (365) from -49.1 to 18.1 mg N kg⁻¹ in the different sub data sets. Linear relationships between D_r (in situ) and $D_{\text{cum}}(365)$ were better for GKA in comparison to FFA aguifer material. Aguifer material which was not jet in contact with NO₃ bearing groundwater (NO₃-free zone and most of sulphidic zone material) exhibited D_r (in situ) values which were clearly less correlated with D_{cum} (365) than aquifer material which was already in contact with NO₃-bearing groundwater (non-sulphidic zone, transition zone and NO₃-bearing zone) (Table 5).

The goodness of fit of regression models to calculate the SRC from D_r (in situ) was on average slightly worse than the one of regression models to predict $D_{\text{cum}}(365)$ from D_r (in situ). As for the prediction of D_{cum} (365) the best goodness of fit of regression models was obtained for the sub data sets of GKA, the transition zone and the NO₃bearing zone with coefficients of determination of R = 0.75, 0.77 and 0.50 (Table 5). Like D_{cum} (365) also for SRC the prediction was best for zones of both aquifers were the aquifer material was already in contact with NO₃-bearing groundwater. Contrary to other partial data sets, the sub-set of D_r(in situ) measured in sulphidic aquifer material exhibited a clearly better goodness of fit between D_r (in situ) and SRC than between $D_{\text{cum}}(365)$ and $D_{\text{r}}(\text{in situ})$, R = 0.41 and R = 0.04, respectively.

References

BGD

11, 16527–16572, 2014

Predicting the

denitrification

capacity of sandy

aquifers

W. Eschenbach and

R. Well

Title Page

Conclusions

Abstract

Figures

Close

Introduction

Tables

Printer-friendly Version

Regression analysis between several denitrification relevant parameters of aquifer material (Eschenbach and Well, 2013) and $D_{\rm r}$ (in situ) revealed that for some partial data sets, the linear regressions between some of these parameters and $D_{\rm r}$ (in situ) were even better than the regression between $D_{\rm r}$ (in situ) and $D_{\rm cum}$ (365) (Table S3 Supplement in comparison to Table 5). For GKA aquifer material, $D_{\rm cum}$ (365) was in closest linear correlation with $D_{\rm r}$ (in situ). Contrary to this, for FFA aquifer material $D_{\rm r}$ (in situ) was closer related to ${\rm SO}_4^{2-}_{\rm extr}$ and ${\rm C}_{\rm hws}$ than to $D_{\rm cum}$ (365) or SRC. For sub data sets grouped according to the sulphate formation capacity of the incubated aquifer material, several parameters were in better or the same linear correlation to $D_{\rm r}$ (in situ) than $D_{\rm cum}$ (365). These parameters were ${\rm C}_{\rm org}$ and total-S in the non-sulphidic zone, ${\rm SO}_4^{2-}_{\rm extr}$ and total-S in the sulphidic zone, ${\rm C}_{\rm org}$ and total-S in the transition zone, ${\rm C}_{\rm org}$ and ${\rm SO}_4^{2-}_{\rm extr}$ in the NO $_3^{-}$ -bearing zone, and ${\rm SO}_4^{2-}_{\rm extr}$ and ${\rm C}_{\rm l}$ in the NO $_3^{-}$ -free zone.

4 Discussion

4.1 Quantifying D_r (in situ) with push-pull tests

4.1.1 Ranges and comparison with previous studies

To compare previous D_r (in situ) data with our measurements, all data were converted to the dimension $\mu g \, N \, k g^{-1} \, d^{-1}$ assuming an effective pore space of 0.3 and an average density of dry aquifer solids of 2.65 g cm⁻³. D_r (in situ) values measured in the FFA and GKA (Table 3) are comparable with D_r (in situ) (2.3–27.1 $\mu g \, N \, k g^{-1} \, d^{-1}$) measured by Konrad (2007) in two Pleistocene sandy aquifers in Northern Germany (aquifers of Thülsfelde and Sulingen, about 40 km west and 30 km south of the city of Bremen, re-

Discussion Paper

Discussion Paper

Discussion Pape

BGD

11, 16527–16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page

Abstract Introduction

Conclusions References

Tables Figures

►I

< 1

Close

Back

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



spectively). Also D_r (in situ) reported by (Addy et al., 2002) and (Addy et al., 2005) show a similar range of denitrification rates with 2.1–121.2 and 0.5–87.9 μ g N kg⁻¹ d⁻¹, respectively. Those values were measured in two riparian sites and a site with marsh sediments on Rhode Island USA. Somewhat larger spans of D_r (in situ) were reported by Well et al. (2003) for water-saturated mineral sub-soils from various locations in Northern Germany and by Konrad (2007) for the sandy to silty aquifer of Wehnsen (about 30 km southeast of the FFA) with D_r (in situ) from 0–300 and 45–339 μ g N kg⁻¹ d⁻¹, respectively. These larger spans cover also the magnitude of D_r (in situ) values measured at multilevel well B4 in the FFA after pre-conditioning (Table 3). Sanches-Perez (2003) measured D_r (in situ) from 22.1 to 7646.4 μ g N kg⁻¹ d⁻¹ with the acetylene inhibition method in 2 shallow sandy aquifers in France and Spain. Overall, there is a wide range of reported D_r (in situ) in aquifers.

But there is not only a strong local variability in D_r (in situ) of aquifers also D_r (in situ) can change substantially during push-pull tests itself. During a push-pull test conducted by Trudell et al. (1986) in situ denitrification rates increased strongly. During the 12 day lasting pull-phase of this tracer test in the O_2 and NO_3^- -free groundwater zone of a shallow sandy aquifer in south western Ontario Canada D_r (in situ) increased from 30.3 to 504.6 μ g N kg⁻¹ d⁻¹ (Trudell et al., 1986).

4.1.2 Time course of $(N_2 + N_2O)_{den}$ and pre-conditioning

In the NO_3^- -free zone, an exponential increase of $(N_2+N_2O)_{den}$ was observed during most of the push-pull tests. Sections of exponential time courses of dilution corrected denitrification products observed during tracer tests were also previously reported (Eschenbach and Well, 2011; Konrad, 2007). In the study of Konrad (2007), 5 out of 13 push-pull tests showed an exponential increase of dilution corrected denitrification products. 4 of these 5 push-pull tests were located in the NO_3^- -free groundwater zone. As a result, it is concluded, that the exponential increase of denitrification products observed during push-pull tests in our study and previous studies can probably be at-

BGD

11, 16527–16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page

Abstract Introduction

Conclusions References

Tables Figures

I4 ►I

•

Back Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



tributed to growth and stimulation of denitrifiers by the injection of NO_3^- into aquifer zones that have previously not been in contact with NO_3^- . Trudell et al. (1986) found an increase of denitrifying bacteria species during the 12 day lasting tracer test which was accompanied by a 17-fold increase of measured denitrification rates. Several other investigations showed increasing microbial activity after bio stimulation of aquifer sediments by the injection of electron donors to monitoring wells (Istok et al., 2004; Kim et al., 2004, 2005). Istok et al. (2004) reported that the viable biomass on solid samplers installed in monitoring wells more than doubled compared with samplers installed in monitoring wells without electron donor addition.

To our knowledge, pre-conditioning of aquifer material prior to a push-pull ¹⁵N tracer test by the injection of only NO₃ as new available electron acceptor was firstly used in the presented study. This was done to establish a denitrifying microbial community in the strict anaerobic zone of an aquifer. Pre-conditioning at multilevel well B4 (see Sect. 2.4) in the FFA resulted in a 30- to 65-fold increase in measured in situ denitrification rates compared with push-pull tests without pre-conditioning at the same depths of multilevel well B4 (Table 3 and Fig. 5). It can be concluded that pre-conditioning in the NO₃-free zone of the FFA led to growth of the community of active denitrifiers in the aquifer material in the vicinity of the respective injection points. The increase of D_r(in situ) due to pre-conditioning might be a combined effect from the increase of active denitrifiers and a higher denitrification rate per microbial cell due to synthesis of enzymes for denitrification. Pre-conditioning resulted not only in higher denitrification rates but also the time course of $(N_2+N_2O)_{den}$ did not show a section of a distinct exponential increase compared with prior measurements without pre-conditioning (Fig. 4). This might show that denitrifiers in the tested aguifer material after pre-conditioning were ready to denitrify and that there was a stable denitrifying community, see also Sect. 4.2.

Higher average denitrification rates during one year of incubation (D_r (365)) in comparison to denitrification rates derived with push pull tests (D_r (in situ)) may have resulted from several factors including the stimulation of denitrification in the lab due to

BGD

11, 16527–16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page

Abstract Introduction

Conclusions References

Tables Figures

l**∢** ⊳l

→

Back Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



Back

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



disturbance of aquifer material, establishment of strictly anaerobic conditions and the adaptation of the microbial community over time. The ratio between D_r (in situ) and $D_r(365)$ was highly variable within the data set. Interestingly, it was lowest in the nonsulphidic and NO₃-free zones of both aquifers (Table 4). Non-sulphidic aquifer material 5 exhibited low denitrification rates during the push-pull tests also because dissolved O₂ inhibited NO₃ reduction. Dissolved O₂ concentrations in the ambient groundwater and therefore also in the injected test solutions were $> 1 \text{ mg O}_2 \text{ L}^{-1}$ at 6 out of 8 injection points in the non-sulphidic zone of both aquifers (Table 2). $D_r(365)$ of non-sulphidic aquifer material measured during anaerobic incubation in the laboratory (Eschenbach and Well, 2013) can therefore be seen as a potential activity which is only partly effective under in situ conditions due to low consumption of dissolved O₂ in groundwater. This is also reflected by the low D_r (in situ) to D_r (365) ratio in the non-sulphidic wells (Table 4).

The mean D_r (in situ)-to- D_r (365) ratio in the NO_3^- bearing zone were twice as high compared to the NO₃-free zone (Table 4 and Fig. 3). This probably reflects the need for microbial adaptation to NO_3^- discussed in the previous section. Mean D_r (in situ) and the ratio of D_r (in situ)-to- D_r (365) of 0.47 were highest in the transition zone, showing that in the transition zone D_r (in situ) and D_r (365) were in closer agreement compared with other zones. During the push-pull tests in the transition zone the ambient concentration of dissolved O₂ were always below 0.13 mg L⁻¹ and NO₃ was always detectable in the ambient groundwater at the 5 injection points in the transition zone (Table 2). Denitrification was therefore presumably not inhibited by dissolved O2 and the microbial population already adapted to NO₃ as an available electron acceptor. Hence, denitrifying conditions during push-pull tests and during laboratory incubation were similar, resulting in closer agreement in denitrification rates.

BGD

11, 16527–16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page **Abstract** Introduction Conclusions References

> **Tables Figures**

Close

Figure 6 sums up our interpretation of the results from push-pull tests in the NO_3^- -free zone. Immediately after the injection of the ^{15}N tracer in the NO_3^- -free zone of both aquifers there seems to follow a time interval with no or negligible production of ^{15}N labelled $(N_2+N_2O)_{den}$ (= lag-phase) (compare with Figs. 2 and 4). During this time, denitrifiers might still have to synthesise enzymes for denitrification and are not yet ready to denitrify.

After the lag-phase follows a phase of exponential increase of (N₂+N₂O)_{den} during which the amount of active denitrifiers and or their activity might adapt to the new available electron acceptor NO₃. The growth of denitrifiers might depend on the available stock of reduced compounds (SRC), i.e. on the surface area of reactive compounds (saRC) present in the aquifer material. If the denitrifying community is adapted to NO₃ and had colonized the saRC, denitrification rates should be relatively constant. Hence a zero order reaction model should fit the measured data during the relatively short duration of a push-pull test (Fig. 6, linear response phase). It is suspected that these conditions apply to the NO₃-bearing zone but not to the NO₃-free zone. After preconditioning at multilevel well B4, $(N_2+N_2O)_{den}$ was initially high and there was no subsequent exponential increase, while the opposite was the case during previous tests at the same well without pre-conditioning (Fig. 4). This probably reflects a more constant activity of denitrifiers during the push-pull tests after pre-conditioning (Fig. 6, linear response phase). Similar adaptation effects have been reported previously, where bio-stimulation by injecting electron donors like ethanol, glucose, propane or fumarate resulted in constant activity, thus allowing the use of zero-order reaction models to derive reduction rates during push-pull tests (Istok et al., 2004; Kim et al., 2004, 2005). This supports our interpretation that pre-conditioning leads to a kind of equilibrium between the denitrifying community, the injected NO₃ and the saRC present in the aquifer material, ultimately resulting in relatively constant reaction rates while NO3 is not limiting (Fig. 6, linear response phase). In our experiments, the latter condition was

BGD

11, 16527-16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Discussion Paper

Discussion

Discussion Paper

Title Page

Abstract Introduction

Conclusions References

Tables Figures

4 ▶|

. →

Back Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



fulfilled, because NO_3^- concentrations during the pull-phase were always clearly above $1.0 \, \text{mg} \, NO_3^- \cdot N \, L^{-1}$, which is assumed to be the threshold of NO_3^- concentrations limiting denitrification rates reported by Wall et al. (2005).

From the dynamics of microbial adaptation outlined above it follows, that preconditioning prior to push-pull tests in the zone of NO₃⁻ free groundwater is needed to allow estimating the stock of reduced compounds from in situ denitrification rates.

4.3 Predicting D_{cum} (365) and SRC of aquifer sediments from D_{r} (in situ)

The main objective of this study is to predict the cumulative denitrification measured during one year of laboratory incubation of aquifer samples ($D_{\rm cum}(365)$) and the stock of reactive compounds (SRC) from in situ denitrification rates ($D_{\rm r}({\rm in~situ})$). In comparison to costly drilling of aquifer material and laboratory measurement of $D_{\rm cum}(365)$ and SRC, $D_{\rm r}({\rm in~situ})$ can be measured with relatively low cost push-pull tests at existing groundwater monitoring wells, which would thus allow spatial mapping of denitrification activity within aquifers.

There are only scarce data comparing the stock of reduced compounds (SRC) or longer-term denitrification rates (e.g. $D_{\rm r}(365)$) with $D_{\rm r}({\rm in~situ})$). Well et al. (2003) showed for denitrification in the saturated zone of hydromorphic soils that laboratory derived denitrification rates after 24 h of anaerobic incubation were in good agreement with in situ denitrification rates, but the study was limited to near-surface groundwater. Konrad (2007) tested this approach in deeper aquifer zones with a small data set of pairs of $D_{\rm r}({\rm in~situ})$ vs. $D_{\rm cum}$ (4 push-pull $^{15}{\rm N}$ tracer tests and incubations of corresponding aquifer material) and found that both quantities were related (spearman rank correlation coefficients of $R \ge 0.8$).

In this study, transfer functions were developed to predict $D_{\text{cum}}(365)$ from $D_{\text{r}}(\text{in situ})$ measurements with a larger data set in different redox zones typically present in aquifers. Moreover, pre-conditioning was evaluated by addition of NO_3^- to aquifer material and the subsequent measurement of in situ denitrification rates.

BGD

11, 16527–16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page

Abstract Introduction

.... oudous.

Conclusions References

Tables Figures

l∢ ⊁l

→

Back Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



11, 16527–16572, 2014

BGD

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page **Abstract** Introduction Conclusions References

Tables

Figures

Back Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

Only a modest goodness of fit (R = 0.62) was found using linear regression between D_r (in situ) and D_{cum} (365) for the full data set (Table 5). Without Box–Cox transformations of input data the correlation coefficient was even lower (R = 0.1). This shows that it was necessary to transform the input data to approach normal distribution and ho-5 moscedasticity for regression analysis. Otherwise the ordinary least squares method did not find the best or efficient linear estimators for regression coefficients.

Like in the previous laboratory study (Eschenbach and Well, 2013) grouping of D_r (in situ) measuring points by locality or according to hydro-geochemical zones increased the predictive power of D_r (in situ) with respect to the measured D_{cum} (365) and SRC of aguifer material for some partial data sets. Altogether, D_r (in situ) was the best predictor for D_{cum} (365) and SRC of the partial data set of GKA aquifer material with correlation coefficients of 0.82 and 0.75, respectively. For the FFA the predictive power of D_r(in situ) for D_{cum} (365) and SRC was significantly lower compared to the GKA (Table 5). This finding mirrors results of laboratory incubations with FFA and GKA material reported by Eschenbach and Well (2013) (Table 4 of the cited study), in which initial denitrification rates $(D_r(7))$ of GKA material were a better predictor of $D_{cum}(365)$ than in case of FFA material. Contrary to GKA aquifer samples, the SRC of the FFA samples was not predictable by D_r (in situ). One reason might be a different microbial availability of organic carbon (Corg), which is one major constituent of SRC in both aquifers (Eschenbach and Well, 2013). The ratio of KMnO₄ labile organic carbon (C₁) to C_{org} was almost twice as high in GKA material compared to FFA material (Eschenbach and Well, 2013), suggesting that the proportion of C_{org} available for microbes is higher in GKA aquifer material and on the other hand that a significant proportion of Corg is unavailable for denitrification in the FFA.

Grouping of aquifer material according to hydro-geochemical zones or sediment parameters resulted in better regressions between D_r (in situ) and D_{cum} (365) and SRC for partial data sets were NO₃⁻ is still present in the groundwater, i.e. in the transition and NO_3^- -bearing zone (Table 5). Konrad (2007) reported similar relationship between D_r (in situ) and D_{cum} (365) under comparable conditions. Relatively weak fits were obtained

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

for data sets with push-pull measuring points located completely or mostly in the zone of NO₃ free groundwater (NO₃-free zone and sulphidic aquifer material, respectively) and in the non-sulphidic zone (Table 5). For the NO₃-free zone this is attributed to a missing adaptation of the microbial community to NO₃ as electron acceptor as discussed above. In the study of Trudell et al. (1986) it took at least 8 days until measured denitrification rates stopped to increase during the push-pull test. In our study, such long pull-periods were not possible because of comparatively higher groundwater velocities in both aquifers. At some injection points in the FFA, the tracer plume moved away with groundwater already within 35 h after injection.

The goodness of fit in the modelling of D_r (in situ) and SRC using linear regression functions was highly variable among partial data-sets. The mean ratios of calculated $D_{\text{cum}}(365)$ to measured $D_{\text{cum}}(365)$ and calculated SRC and measured SRC were best for the transition zone with ratios near 1 and worst for the sulphidic and NO3-free zone (Table 5). $D_{\rm r}$ (in situ) underestimated especially $D_{\rm cum}$ (365) and SRC of deeper aquifer samples with high values of $D_{cum}(365)$ and SRC to a large extent (Table 5) apparently because of the lack of adaptation of the microbial community to NO3, as already discussed above.

Pre-conditioning at multilevel well B4 led to a clearly better fit of D_r (in situ) and D_r (365) (Table 4). This indicates that pre-conditioning should increase the predictability of $D_{\text{cum}}(365)$ and probably also of SRC from $D_{\text{r}}(\text{in situ})$ measurements in the NO₃-free zone.

Conclusions

The possibility to predict the capacity of aquifer zones to remove NO₃ inputs over extended time periods based on in situ measurement of denitrification rates (D_r (in situ)) was evaluated in two Pleistocene aquifers in Northern Germany. Direct comparison of in situ push-pull tests and laboratory incubations to determine the cumulative denitrification measured during one year of incubation (D_{cum} (365)) and the stock of reduced

BGD

11, 16527–16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page

Abstract Introduction

Conclusions References

> **Figures Tables**

compounds (SRC) of aquifer material proved to be a suitable approach to calibrate linear regression models.

Prediction of $D_{\rm cum}(365)$ and SRC from $D_{\rm r}({\rm in~situ})$ for data sets containing data from both aquifers was only satisfactory in the aquifer zones were NO_3^- was present. This type of in situ tests might thus be suitable for mapping $D_{\rm cum}(365)$ and SRC in NO_3^- bearing zones of Pleistocene sandy aquifers using existing monitoring wells. It is thus a promising and low-cost method to estimate $D_{\rm cum}(365)$ of aquifer material from aquifer zones were NO_3^- is still present in the groundwater. Moreover, future routine applications of this approach could be facilitated by online field analysis using membrane inlet mass spectrometry, which we demonstrated to be feasible and precise (see Supplement).

In the NO₂-free aquifer zone increasing denitrification rates were observed during the conducted push-pull tests, which was interpreted as the result of adaptation processes of the denitrifying communities following NO_3^- injections. Also D_r (in situ) without pre-conditioning were generally lower than average denitrification rates after one year of incubation $(D_r(365))$ in the laboratory. This was especially the case for $D_r(in situ)$ measurements in the NO_3^- free groundwater zone. In this study it was demonstrated exemplarily that the microbial community in the NO₃-free zone close below the NO₃bearing zone can be adapted to denitrification by amending wells with NO₃ injections for an extended period. In situ denitrification rates measured after this pre-conditioning reflected the $D_{\text{cum}}(365)$ and SRC more satisfactorily. From this findings it is assumed that microbial adaptation after NO₃ injection confounded the relationship between reactive compounds present in the tested aquifer material and D_r (in situ) measured during push-pull tests, which resulted in poor prediction of $D_{\text{cum}}(365)$ and SRC from $D_r(\text{in})$ situ). Therefore we assume that pre-conditioning is a prerequisite for the measurement of in situ denitrification rates using push-pull tracer tests in the NO₃-free groundwater zone. Further research is needed to check if this microbial adaptation would also work in deeper layers far below the NO₃-bearing zone.

BGD

11, 16527–16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page

Abstract Introduction

Conclusions References

Tables Figures

l∢ ≯l

•

Back Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



- Addy, K., Kellogg, D. Q., Gold, A. J., Groffman, P. M., Ferendo, G., and Sawyer, C.: In situ push-pull method to determine ground water denitrification in riparian zones, J. Environ. Qual., 31, 1017–1024, 2002.
- Addy, K., Gold, A., Nowicki, B., McKenna, J., Stolt, M., and Groffman, P.: Denitrification capacity in a subterranean estuary below a Rhode Island fringing salt marsh, Estuaries, 28, 896–908, 2005.
- Böttcher, J., Strebel, O., and Duijnisveld, W. H. M.: Vertikale Stoffkonzentrationsprofile im Grundwasser eines Lockergesteins-Aquifers und deren Interpretation (Beispiel Fuhrberger Feld), Z. dt. Geol. Ges., 136, 543–552, 1985.
 - Böttcher, J., Strebel, O., and Duijnisveld, W. H. M.: Kinetik und Modellierung gekoppelter Stoffumsetzungen im Grundwasser eines Lockergesteins-Aquifers., Geoll. Jahrb. Reihe C, 51, 3–40, 1989.
 - Böttcher, J., Strebel, O., and Duijnisveld, W. H. M.: Reply (to a comment of Scott, F. Korom), Water Resour. Res., 27, 3275–3278, 1991.
 - Böttcher, J., Strebel, O., and Kölle, W.: Redox conditions and microbial sulfur reactions in the Fuhrberger Feld sandy aquifer, in: Progress in hydrogeochemistry, edited by: Matthess, G. et al., Springer, Berlin, 219–226, 1992.
 - Burgin, A. J. and Hamilton, S. K.: Have we overemphasized the role of denitrification in aquatic ecosystems? A review of nitrate removal pathways, Front. Ecol. Environ., 5, 89–96, doi:10.1890/1540-9295(2007)5[89:hwotro]2.0.co;2, 2007.
 - Eschenbach, W. and Well, R.: Online measurement of denitrification rates in aquifer samples by an approach coupling an automated sampling and calibration unit to a membrane inlet mass spectrometry system, Rapid Commun. Mass Sp., 25, 1993–2006, doi:10.1002/rcm.5066, 2011.
 - Eschenbach, W. and Well, R.: Predicting the denitrification capacity of sandy aquifers from shorter-term incubation experiments and sediment properties, Biogeosciences, 10, 1013–1035, doi:10.5194/bg-10-1013-2013, 2013.

Paper

Discussion Paper

Discussion Paper

Discussion Pape

11, 16527-16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page

Abstract Introduction

Conclusions References

Tables Figures

I**∢** ►I

•

Back Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



16557

20

25

BGD

11, 16527-16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

- Title Page

 Abstract Introduction

 Conclusions References
 - nciusions
 - Tables Figures
 - I⊀ ≻I
- •
- Back Close
 Full Screen / Esc
- Printer-friendly Version
- Interactive Discussion
 - © BY

- Frind, E. O., Duynisveld, W. H. M., Strebel, O., and Boettcher, J.: Modeling of multicomponent transport with microbial transformation in groundwater the Fuhrberg case, Water Resour. Res., 26, 1707–1719, 1990.
- Griebler, C. and Lueders, T.: Microbial biodiversity in groundwater ecosystems, Freshwater Biol., 54, 649–677, doi:10.1111/j.1365-2427.2008.02013.x, 2009.
- Groffman, P. M., Altabet, M. A., Bohlke, J. K., Butterbach-Bahl, K., David, M. B., Firestone, M. K., Giblin, A. E., Kana, T. M., Nielsen, L. P., and Voytek, M. A.: Methods for measuring denitrification: diverse approaches to a difficult problem, Ecol. Appl., 16, 2091–2122, 2006.
- Harris, S. H., Istok, J. D., and Suflita, J. M.: Changes in organic matter biodegradability influencing sulfate reduction in an aquifer contaminated by landfill leachate, Microb. Ecol., 51, 535–542, doi:10.1007/s00248-006-9043-y, 2006.
- Hiscock, K. M., Lloyd, J. W., and Lerner, D. N.: Review of natural and artificial denitrification of groundwater, Water Res., 25, 1099–1111, 1991.
- Howar, M.: Geologische 3D-Untergrundmodellierung im Bereich Großenkneten/Ahlhorn, unpubl. Expertise: INSIGHT, Geologische Softwaresysteme GmbH, Köln, Germany, 11 pp., 2005.
- Istok, J. D., Humphrey, M. D., Schroth, M. H., Hyman, M. R., and Oreilly, K. T.: Single-well, "push-pull" test for in situ determination of microbial activities, Ground Water, 35, 619–631, 1997.
- Istok, J. D., Senko, J. M., Krumholz, L. R., Watson, D., Bogle, M. A., Peacock, A., Chang, Y. J., and White, D. C.: In situ bioreduction of technetium and uranium in a nitrate-contaminated aquifer, Environ. Sci. Technol., 38, 468–475, doi:10.1021/es034639p, 2004.
 - Kellogg, D. Q., Gold, A. J., Groffman, P. M., Addy, K., Stolt, M. H., and Blazejewski, G.: In situ ground water denitrification in stratified, permeable soils underlying riparian wetlands, J. Environ. Qual., 34, 524–533, 2005.
 - Kim, Y., Istok, J. D., and Semprini, L.: Push-pull tests for assessing in situ aerobic cometabolism, Ground Water, 42, 329–337, doi:10.1111/j.1745-6584.2004.tb02681.x, 2004.
 - Kim, Y., Kim, J. H., Son, B. H., and Oa, S. W.: A single well push-pull test method for in situ determination of denitrification rates in a nitrate-contaminated groundwater aquifer, Water Sci. Technol., 52, 77–86, 2005.
 - Kneeshaw, T. A., McGuire, J. T., Smith, E. W., and Cozzarelli, I. M.: Evaluation of sulfate reduction at experimentally induced mixing interfaces using small-

scale push-pull tests in an aquifer-wetland system, Appl. Geochem., 22, 2618-2629, doi:10.1016/j.apgeochem.2007.06.006, 2007.

Kölbelboelke, J., Anders, E. M., and Nehrkorn, A.: Microbial communities in the saturated groundwater environment. 2. Diversity of bacterial communities in a Pleistocene sand aquifer and their invitro activeties, Microb. Ecol., 16, 31-48, doi:10.1007/bf02097403, 1988.

Kölle, W., Strebel, O., and Böttcher, J.: Formation of sulfate by microbial denitrification in a reducing aguifer, Water Supp., 3, 35-40, 1985.

Kollmann, W.: Die Bestimmung des durchflußwirksamen Porenvolumens von Sedimenten und seine Bedeutung für den Grundwasserschutz, Mitt. österr. geol. Ges., 79, 63-76, 1986.

Konrad, C.: Methoden zur Bestimmung des Umsatzes von Stickstoff für drei pleistozäne Grundwasserleiter Norddeutschlands, Ph.D. Thesis, Univ. of Tech. Dresden, Dresden, Germany, 161 pp., 2007.

Korom, S. F.: Natural denitrification in the saturated zone – a review, Water Resour, Res., 28. 1657–1668, 1992,

Law, G. T. W., Geissler, A., Boothman, C., Burke, I. T., Livens, F. R., Llovd, J. R., and Morris, K.: Role of nitrate in conditioning aguifer sediments for technetium bioreduction, Environ. Sci. Technol., 44, 150-155, doi:10.1021/es9010866, 2010.

McGuire, J. T., Long, D. T., Klug, M. J., Haack, S. K., and Hyndman, D. W.: Evaluating behavior of oxygen, nitrate, and sulfate during recharge and quantifying reduction rates in a contaminated aquifer, Environ. Sci. Technol., 36, 2693–2700, doi:10.1021/es015615q, 2002.

Rivett, M. O., Buss, S. R., Morgan, P., Smith, J. W. N., and Bemment, C. D.: Nitrate attenuation in groundwater: a review of biogeochemical controlling processes, Water Res., 42, 4215–4232, doi:10.1016/j.watres.2008.07.020, 2008.

Sánchez-Pérez, J. M., Bouey, C., Sauvage, S., Teissier, S., Antiquedad, I., and Vervier, P.: A standardised method for measuring in situ denitrification in shallow aquifers: numerical validation and measurements in riparian wetlands, Hydrol. Earth Syst. Sci., 7, 87-96, doi:10.5194/hess-7-87-2003. 2003.

Santoro, A. E., Boehm, A. B., and Francis, C. A.: Denitrifier community composition along a nitrate and salinity gradient in a coastal aquifer, Appl. Environ. Microbiol., 72, 2102-2109, doi:10.1128/aem.72.3.2102-2109.2006. 2006.

Schroth, M. H., Kleikemper, J., Bolliger, C., Bernasconi, S. M., and Zeyer, J.: In situ assessment of microbial sulfate reduction in a petroleum-contaminated aquifer using push-pull tests and stable sulfur isotope analyses, J. Contam. Hydrol., 51, 179–195, 2001.

BGD

11, 16527–16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page Abstract Introduction Conclusions References **Tables Figures**

Back Close Full Screen / Esc

Printer-friendly Version

Interactive Discussion



BGD

11, 16527-16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

- Title Page

 Abstract Introduction

 Conclusions References
 - Tables Figures
 - Id ≯I
- - Back Close
 - Full Screen / Esc
- Printer-friendly Version
- Interactive Discussion
 - © (1)

- Schuchert, A.: Zielflächenidentifikation für Grundwasserschutzmaßnahmen. Eine GIS-Datenanalyse im Wasserschutzgebiet Großenkneten, Landkreis Oldenburg, Diploma thesis, Institute for Geography, University of Bremen, Germany, 2007.
- Seitzinger, S., Harrison, J. A., Bohlke, J. K., Bouwman, A. F., Lowrance, R., Peterson, B., Tobias, C., and Van Drecht, G.: Denitrification across landscapes and waterscapes: a synthesis, Ecol. Appl., 16, 2064–2090, 2006.
- Senko, J. M., Istok, J. D., Suflita, J. M., and Krumholz, L. R.: In-situ evidence for uranium immobilization and remobilization, Environ. Sci. Technol., 36, 1491–1496, doi:10.1021/es011240x, 2002.
- Strebel, O., Böttcher, J., and Duijnisveld, W. H. M.: Identifizierung und Quantifizierung von Stoffumsetzungen in einem Sand-Aquifer (Beispiel Fuhrberger Feld), DVGW Schriftenreihe Wasser, 73, 55–73, 1992.
 - Trudell, M. R., Gillham, R. W., and Cherry, J. A.: An insitu study of the occurence and rate of denitrification in a shallow unconfined sand aquifer, J. Hydrol., 83, 251–268, 1986.
- van Berk, W., Kübeck, C., Steding, T., van Straaten, L., and Wilde, S.: Vorstudie zur Hydrogeologie im Wassergewinnungsgebiet Großenkneten, 55 pp., 2005.
 - Vitousek, P. M., Aber, J. D., Howarth, R. W., Likens, G. E., Matson, P. A., Schindler, D. W., Schlesinger, W. H., and Tilman, G. D.: Human alteration of the global nitrogen cycle: sources and consequences, Ecol. Appl., 7, 737–750, 1997.
- von der Heide, C., Bottcher, J., Deurer, M., Weymann, D., Well, R., and Duijnisveld, W. H. M.: Spatial variability of N₂O concentrations and of denitrification-related factors in the surficial groundwater of a catchment in Northern Germany, J. Hydrol., 360, 230–241, doi:10.1016/j.jhydrol.2008.07.034, 2008.
 - Weiss, R. F.: Solubility of nitrogen, oxygen and argon in water and seawater, Deep-Sea Res., 17, 721–735, 1970.
 - Weiss, R. F. and Price, B. A.: Nitrous-oxide solubility in water and seawater, Mar. Chem., 8, 347–359, 1980.
 - Well, R. and Myrold, D. D.: Laboratory evaluation of a new method for in situ measurement of denitrification in water-saturated soils, Soil Biol. Biochem., 31, 1109–1119, 1999.
- Well, R. and Myrold, D. D.: A proposed method for measuring subsoil denitrification in situ, Soil Sci. Soc. Am. J., 66, 507–518, 2002.

- Well, R., Becker, K. W., Langel, R., Meyer, B., and Reineking, A.: Continuous flow equilibration for mass spectrometric analysis of dinitrogen emissions, Soil Sci. Soc. Am. J., 62, 906–910, 1998.
- Well, R., Augustin, J., Meyer, K., and Myrold, D. D.: Comparison of field and laboratory measurement of denitrification and N₂O production in the saturated zone of hydromorphic soils, Soil Biol. Biochem., 35, 783–799, doi:10.1016/s0038-0717(03)00106-8, 2003.
- Well, R., Höper, H., Mehranfar, O., and Meyer, K.: Denitrification in the saturated zone of hydromorphic soils-laboratory measurement, regulating factors and stochastic modeling, Soil Biol. Biochem., 37, 1822–1836, doi:10.1016/j.soilbio.2005.02.014, 2005.
- Well, R., Eschenbach, W., Flessa, H., von der Heide, C., and Weymann, D.: Are dual isotope and isotopomer ratios of N₂O useful indicators for N₂O turnover during denitrification in nitrate-contaminated aquifers?, Geochim. Cosmochim. Ac., 90, 265–282, doi:10.1016/j.gca.2012.04.045, 2012.
- Wessolek, G., Renger, M., Strebel, O., and Sponagel, H.: Einfluß von Boden und Grundwasserflurabstand auf die jährliche Grundwasserneubildung unter Acker, Grünland und Nadelwald, Z. Kulturtech. Flurber., 26, 130–137, 1985.
- Weymann, D., Well, R., Flessa, H., von der Heide, C., Deurer, M., Meyer, K., Konrad, C., and Walther, W.: Groundwater N₂O emission factors of nitrate-contaminated aquifers as derived from denitrification progress and N₂O accumulation, Biogeosciences, 5, 1215–1226, doi:10.5194/bg-5-1215-2008, 2008.
- Weymann, D., Geistlinger, H., Well, R., von der Heide, C., and Flessa, H.: Kinetics of N₂O production and reduction in a nitrate-contaminated aquifer inferred from laboratory incubation experiments, Biogeosciences, 7, 1953–1972, doi:10.5194/bg-7-1953-2010, 2010.
- Wirth, K.: Hydrogeologisches Gutachten zur Bemessung und Gliederung der Trinkwasserschutzgebiete für die Fassungen Hagel, Sage und Baumweg, Wasserwerk Großenkneten (OOWV), Beratungsbüro für Hydrogeologie (Hrsg.), Göttingen, Germany, 18 pp., 1990.

11, 16527–16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page

Abstract Introduction

Conclusions References

Tables Figures

•

Close

Full Screen / Esc

Back

Printer-friendly Version

Interactive Discussion



Table 1. Overview of the conducted push-pull ¹⁵N tracer tests in both aquifers and depth position of their filter screens. Push-pull tests with and without pre-conditioning were conducted at multilevel well B4 in 7, 8, 9 and 10 m below soil surface.

| | Fuhrberg (multilevel wells) | | | | Großenkneten (conventional monitoring and multilevel wells | | | | | wells) | |
|---|--------------------------------|--------------|-----------------------------|----|---|------------|-----------------------|-------|---------|----------------------|-------------|
| Monitoring well | B1 | B2 | B4 | B6 | N10 | Gro 326 | Gro 327 surface | S1 | S2 1 | CMT 2 | CMT |
| non-sulphidic zone (NO ₃ -bearing zone) | | 3 4.2 | | 3 | | 8-10 | | | | | 8.3 22.8 |
| Transition zone (NO ₃ -bearing zone) | | 8 9 10 | | | 5 8 | | | | | | |
| Sulphidic zone (NO ₃ ⁻ -free zone) | 7 8 | 14 | 7* 8, 8* 9, 9* 10* | | | | 35–39 | 66–67 | 26–27 | 29.3 31.3 33.5 | 26.8 |

^{*} Push-pull tests with pre-conditioning.

11, 16527-16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page

Abstract Introduction

Conclusions References

Tables Figures

15 21

Back Close

Full Screen / Esc

Printer-friendly Version



Table 2. Background conditions of groundwater at the locations of push-pull ¹⁵N tracer tests.

| Location | inj. depth ^a | aquifer | O ₂ | NO ₃ | N ₂ O | SO ₄ ²⁻ | рН | redox | con ^b |
|----------|----------------------------|-----------------|------------------|-----------------|----------------------|-------------------------------|------|-------|------------------|
| | m m | zone | ${\rm mgL}^{-1}$ | $mgNL^{-1}$ | μg N L ⁻¹ | $mgSL^{-1}$ | | mV | μScm^{-1} |
| FFA B1 | 7.0 | sulphidic | 0.67 | < 0.25 | n.d. | 27.64 | 6.00 | -171 | 473 |
| FFA B1 | 8.0 | sulphidic | 0.76 | < 0.25 | n.d. | 24.73 | 6.04 | -175 | 440 |
| FFA B2 | 3.0 | non-sulphidic | 3.66 | 41.47 | 1.59 | 15.07 | 4.66 | 273 | 563 |
| FFA B2 | 4.2 | non-sulphidic | 0.96 | 27.59 | 68.31 | 36.94 | 4.83 | 209 | 564 |
| FFA B2 | 8.0 | transition zone | 0.16 | 12.58 | 0.03 | 32.52 | 4.48 | 341 | 553 |
| FFA B2 | 9.0 | transition zone | 0.13 | 7.09 | 0.05 | 38.41 | 4.65 | 367 | 488 |
| FFA B2 | 10.0 | transition zone | 0.06 | 1.0 | n.d. | 43.30 | 4.75 | 374 | 458 |
| FFA B2 | 14.0 | sulphidic | 0.40 | 0.63 | n.d. | 42.51 | 6.75 | 117 | 453 |
| FFA B4 | 8.0 | sulphidic | 0.22 | < 0.25 | 1.14 | 42.30 | 5.28 | -38 | 432 |
| FFA B4 | 9.0 | sulphidic | 0.12 | < 0.25 | 0.70 | 51.19 | 5.43 | _ | _ |
| FFA B6 | 3.0 | non-sulphidic | 9.51 | 6.10 | 0.02 | 13.95 | 5.70 | 365 | 255 |
| FFA B6 | 6.0 | non-sulphidic | 1.28 | 19.55 | 10.66 | 22.45 | 5.18 | 349 | 441 |
| FFA N10 | 5.0 | transition zone | 0.12 | 13.12 | 184.8 | 59.87 | 4.61 | 341 | 660 |
| FFA N10 | 8.0 | transition zone | 0.16 | 0.4 | 1.03 | 52.03 | 5.60 | 3 | 463 |
| GKA 326 | 8-10 | non-sulphidic | 6.30 | 3.06 | 0.12 | 4.67 | 4.10 | 374 | 105 |
| GKA CMT2 | 8.3 | non-sulphidic | 6.10 | 3.14 | 0.12 | 5.06 | 4.40 | 387 | 100 |
| GKA CMT2 | 22.8 | non-sulphidic | 5.70 | 3.98 | 0.56 | 12.09 | 5.10 | 276 | 163 |
| GKA CMT2 | 26.8 | sulphidic | 0.10 | < 0.25 | 0.01 | 18.57 | 5.40 | 30 | 221 |
| GKA S2 | 26-27 | sulphidic | 0.30 | < 0.25 | n.d. | 17.85 | 5.30 | 161 | 217 |
| GKA CMT1 | 29.3 | sulphidic | 0.20 | < 0.25 | n.d. | 18.16 | 5.50 | -24 | 240 |
| GKA CMT1 | 31.3 | sulphidic | 0.14 | < 0.25 | n.d. | 17.91 | 5.20 | 134 | 195 |
| GKA CMT1 | 33.5 | sulphidic | 0.20 | < 0.25 | n.d. | 18.60 | 5.10 | 122 | 272 |
| GKA 327 | 35-39 | sulphidic | 0.10 | < 0.25 | 0.13 | 10.85 | 5.30 | 26 | 275 |
| GKA S1 | 66–67 | non-sulphidic | 0.13 | < 0.25 | 0.02 | 5.10 | 5.72 | -54 | 103 |

FFA Fuhrberger Feld aquifer. GKA Großenkneten aquifer.

BGD

11, 16527-16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page

Abstract Introduction

Conclusions References

Tables Figures

I◀

•

Close

Full Screen / Esc

Back

Printer-friendly Version



a Injection depth.

^b Conductivity.

Table 3. In situ denitrification rates (D_r (in situ)) and minimum and maximum values of D_r (in situ) in dependence of the range of estimated effective porosities (0.2 to 0.4). D_r (in situ) were calculated from a regression line through the $(N_2+N_2O)_{den}$ concentrations at time intervals with the steepest increase of $(N_2+N_2O)_{den}$ during the respective push-pull test. Tracer tests after pre-conditioning are marked with *.

| Location | Injection depth M | Aquifer zone | D _r (in situ) | $D_{\rm r}$ (in situ) Max μ g N kg ⁻¹ d ⁻ | D _r (in situ) min | Rª |
|----------|-------------------------|------------------------------|--------------------------|---|---------------------------------|------|
| | | | | | | |
| FFA B1 | 7.0 | sulphidic | 17.59 | 27.361 | 10.261 | 0.94 |
| FFA B1 | 8.0 | sulphidic | 1.512 | 2.352 | 0.882 | 0.92 |
| FFA B2 | 3.0 | non-sulphidic ^b | 0.120 | 0.186 | 0.070 | 0.14 |
| FFA B2 | 4.2 | non-sulphidic ^b | 0.065 | 0.102 | 0.038 | 0.0 |
| FFA B2 | 8.0 | transition zone ^b | 0.429 | 0.667 | 0.250 | 0.95 |
| FFA B2 | 9.0 | transition zone ^b | 1.415 | 2.201 | 0.825 | 0.90 |
| FFA B2 | 10.0 | transition zone ^b | 8.650 | 13.456 | 5.046 | 0.99 |
| FFA B2 | 14.0 | sulphidic ^c | 51.47 | 80.078 | 30.029 | 0.82 |
| FFA B4 | 8.0 | sulphidic | 2.755 | 4.286 | 1.607 | 0.98 |
| FFA B4 | 9.0 | sulphidic ^c | 2.278 | 3.544 | 1.329 | 0.86 |
| FFA B6 | 3.0 | non-sulphidic ^b | 0.057 | 0.089 | 0.033 | 0.02 |
| FFA B6 | 6.0 | non-sulphidic ^b | 4.998 | 7.774 | 2.915 | 0.96 |
| FFA N10 | 5.0 | transition zoneb | 12.89 | 20.052 | 7.520 | 0.95 |
| FFA N10 | 8.0 | transition zone ^b | 23.19 | 36.074 | 13.528 | 0.99 |
| FFA B4* | 7.0 | sulphidic ^c | 152.6 | 237.527 | 89.073 | 0.94 |
| FFA B4* | 8.0 | sulphidic ^c | 67.83 | 105.514 | 39.568 | 0.99 |
| FFA B4* | 9.0 | sulphidic ^c | 145.5 | 226.481 | 84.930 | 0.98 |
| FFA B4* | 10.0 | sulphidic ^c | 150.7 | 234.530 | 87.949 | 1.00 |
| GKA 326 | 8-10 | non-sulphidic ^b | 0.747 | 1.162 | 0.436 | 0.96 |
| GKA CMT2 | 8.3 | non-sulphidic ^b | 0.051 | 0.079 | 0.030 | 0.02 |
| GKA CMT2 | 22.8 | non-sulphidic ^b | 0.009 | 0.013 | 0.005 | 0.00 |
| GKA CMT2 | 26.8 | sulphidic ^c | 1.233 | 1.918 | 0.719 | 0.70 |
| GKA S2 | 26-27 | sulphidic ^c | 0.860 | 1.338 | 0.502 | 0.99 |
| GKA CMT1 | 29.3 | sulphidic | 4.427 | 6.886 | 2.582 | 0.78 |
| GKA CMT1 | 31.3 | sulphidic ^c | 0.504 | 0.784 | 0.294 | 0.63 |
| GKA CMT1 | 33.5 | sulphidic ^c | 2.002 | 3.114 | 1.168 | 0.77 |
| GKA 327 | 35-39 | sulphidic | 6.192 | 9.632 | 3.612 | 0.99 |
| GKA S1 | 66-67 | non-sulphidic ^c | 2.271 | 3.533 | 1.325 | 1.00 |

FFA Fuhrberger Feld aquifer; GKA Großenkneten aquifer.

11, 16527–16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page

Abstract Introduction

Conclusions References

Tables Figures

▶I

Back Close

Full Screen / Esc

Printer-friendly Version



^a Correlation coefficient of the regression line.

^b NO₃ -bearing zone.

c NO3-free zone.

Discussion Paper

Table 4. Means, SD and ranges of D_r (in situ) of the data sets. Statistical significant differences (kw: P < 0.05) between D_r (in situ) values measured in the various sub data sets occurred only between D_r (in situ) measured in the non-sulphidic zone and some other partial data sets.

| | | | D _r (in situ) ^a (μg kg ⁻¹ N d ⁻¹) | | | | $D_{\rm r}({\rm in~situ})/D_{\rm r}~(365)^{\rm b}$ | | | |
|-------------------------------|----|------------------|--|----------------------------|----|-----------------|--|--|--|--|
| Data set | N° | means | range | non-sulphidic ^d | Ne | means | Range | | | |
| Whole data set | 24 | 6.07 ± 11.36 | 0.00-51.48 | s ¹ | 34 | 0.15 ± 0.20 | 0.00-0.60 | | | |
| FFA | 14 | 9.10 ± 14.20 | 0.06-51.48 | s ¹ | 16 | 0.26 ± 0.24 | 0.01-0.60 | | | |
| GKA | 10 | 1.83 ± 2.02 | 0.00-6.19 | Ns | 18 | 0.06 ± 0.06 | 0.00-0.20 | | | |
| non-sulphidic zone | 8 | 1.04 ± 1.78 | 0.00 - 5.00 | _ | 11 | 0.05 ± 0.08 | 0.00-0.23 | | | |
| sulphidic zone | 14 | 8.59 ± 13.67 | 0.43-51.48 | s^2 | 23 | 0.20 ± 0.22 | 0.01-0.60 | | | |
| transition zone | 5 | 9.32 ± 9.32 | 0.43-23.19 | s ¹ | 8 | 0.47 ± 0.14 | 0.25-0.60 | | | |
| NO ₃ -bearing zone | 12 | 4.38 ± 7.24 | 0.00-23.19 | Ns | 17 | 0.23 ± 0.24 | 0.00-0.60 | | | |
| NO ₃ -free zone | 16 | 7.76 ± 14.53 | 0.50-51.48 | s ¹ | 17 | 0.10 ± 0.10 | 0.01-0.37 | | | |
| B4 pre-conditioned | 4 | 128.1 ± 43.4 | 67.8-152.7 | _ | 4 | 1.87 ± 0.84 | 0.72-2.76 | | | |
| B4 un-conditioned | 2 | 2.52 ± 0.34 | 2.28-2.76 | _ | 2 | 0.04 ± 0.02 | 0.02-0.05 | | | |

^a All D_r(in situ) measurements.

BGD

11, 16527-16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page

Introduction **Abstract**

Conclusions References

> **Tables Figures**

Back Close

Full Screen / Esc

Printer-friendly Version



^b Only *D*_r(in situ) measurements with corresponding incubated aquifer samples.

^c Number of D_r (in situ) measurements.

^d Statistical differences between non-sulphidic and other data sets (s significant differences; ns not significant differences.

¹ Differences significant at the 0.05 probability level.

² Differences significant at the 0.1 probability level.

^e Number of comparisons between D_r (in situ) and corresponding incubated aquifer samples.

Table 5. Simple regressions between D_r (in situ) and D_{cum} (365) and SRC from anaerobic incubations with corresponding aquifer material. $f^{B-C}(X) = A + B \times f^{B-C}(D_r(\text{in situ}))$.

| | | | | | | calculated/measured | | Deviation (mg kg ⁻¹ yr ⁻¹) | |
|------------------------------|-----------------------|----|--------|--------|----------------|---------------------|-------------|--|--------------|
| Data set | Xª | Nb | Α | В | R ^c | mean | range | mean | range |
| Whole data set | $D_{\rm cum}(365)$ | 34 | 2.878 | 0.603 | 0.62 | 2.29 ± 4.19 | 0.16-22.96 | -3.07 ± 14.67 | -47.2-12.8 |
| Whole data set | SRC | 34 | 6.123 | 0.152 | 0.40 | 1.51 ± 1.31 | 0.12-5.19 | -671.2 ± 2091 | -7734-1379 |
| FFA | $D_{\text{cum}}(365)$ | 16 | 2.640 | 0.578 | 0.52 | 2.83 ± 4.90 | 0.13-19.18 | -3.08 ± 14.71 | -49.1 - 7.0 |
| FFA | SRC | 16 | 3.772 | 0.006 | 0.07 | 1.22 ± 0.82 | 0.11 - 2.92 | -377.8 ± 1375 | -5317 -413.7 |
| GKA | $D_{\rm cum}(365)$ | 18 | 3.046 | 0.818 | 0.82 | 1.34 ± 0.92 | 0.26 - 3.85 | -2.25 ± 12.28 | -30.8-15.5 |
| GKA | SRC | 18 | 8.024 | 0.613 | 0.75 | 1.43 ± 1.23 | 0.178-4.47 | -617.0 ± 2179 | -5780-2390 |
| non-sulphidic | $D_{\rm cum}(365)$ | 11 | 1.050 | 0.156 | 0.40 | 2.25 ± 3.20 | 0.26-10.65 | -0.10 ± 2.41 | -5.2-1.8 |
| non-sulphidic | SRC | 11 | 8407 | 752.8 | 0.43 | 1.50 ± 0.84 | 0.46 - 3.19 | 31.54 ± 240.7 | -553-272.6 |
| sulphidic | $D_{\text{cum}}(365)$ | 23 | 4.185 | -0.033 | 0.04 | 1.33 ± 0.90 | 0.30-4.19 | -3.32 ± 15.13 | -39.4-13.1 |
| sulphidic | SRC | 23 | 21.40 | -1.372 | 0.41 | 0.30 ± 0.18 | 0.03-0.61 | -1823 ± 2313 | -8564144 |
| transition zone | $D_{\rm cum}(365)$ | 8 | 1.109 | 0.581 | 0.53 | 1.03 ± 0.26 | 0.74-1.43 | -0.36 ± 2.84 | -4.5 - 3.3 |
| transition zone | SRC | 8 | 5.349 | -0.602 | 0.77 | 1.05 ± 0.41 | 0.58 - 1.92 | -50.11 ± 340.6 | -518.7-561 |
| NO ₃ -bearing | $D_{\text{cum}}(365)$ | 17 | 2.132 | 0.454 | 0.84 | 2.21 ± 3.76 | 0.13-15.17 | -0.67 ± 2.52 | -6.3-2.7 |
| NO3-bearing | SRC | 17 | 193.3 | 16.32 | 0.55 | 1.36 ± 0.75 | 0.41 - 2.76 | -19.35 ± 365.2 | -929-462.6 |
| NO3-free | $D_{\rm cum}(365)$ | 17 | 7.774 | 2.036 | 0.36 | 1.47 ± 0.88 | 0.31-3.00 | -1.69 ± 16.23 | -38.7-18.1 |
| NO₃-free | SRC | 17 | 77.61 | 8.421 | 0.21 | 1.78 ± 1.46 | 0.27-4.47 | -485.4 ± 2494 | -6077-2095 |
| pre-conditioned ¹ | $D_{\text{cum}}(365)$ | 4 | 14.402 | 0.099 | 0.54 | 1.06 ± 0.35 | 0.62-1.47 | $0.12 \pm 9.49.79$ | -12.95-9.41 |
| pre-conditioned ¹ | SRC | 4 | 319.5 | 4.895 | 0.53 | 1.12 ± 0.52 | 0.51-1.77 | 5.5 ± 462 | -638.0-464 |

Experimental data of pre-conditioned push-pull tracer tests was not Box-Cox transformed before regression analysis, because of the small number of data pairs. For these data pairs the following equation applies: $X = A + B \times D_r$ (in situ).

Abstract

Introduction

Conclusions

References

Tables

Figures

BGD

11, 16527-16572, 2014

Predicting the

denitrification capacity of sandy

aquifers

W. Eschenbach and R. Well

Title Page

Close

Back

a Independent sediment parameter.

^b Number of samples.

^c Correlation coefficient.

Back

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



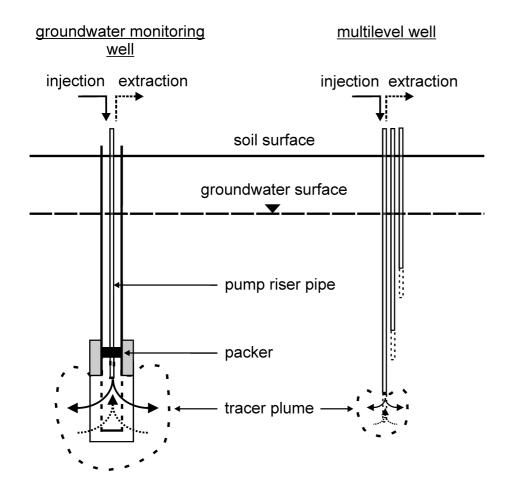


Figure 1. Schematic of push-pull ¹⁵N tracer tests at groundwater monitoring and multilevel wells.

BGD

11, 16527-16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page

Introduction **Abstract**

Conclusions References

> **Tables Figures**

Close

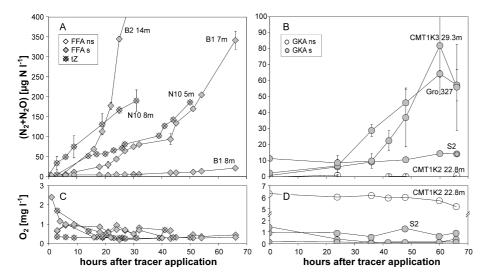


Figure 2. Time courses of denitrification derived $(N_2+N_2O)_{den}$ and dissolved O_2 during ¹⁵N push-pull tests in the FFA (**a** and **c**) and GKA (**b** and **d**). FFA = Fuhrberger Feld aquifer; GKA = Großenkneten aquifer; ns non-sulphidic; s sulphidic and tZ transition zone aquifer material.

11, 16527-16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page

Abstract Introduction

Conclusions References

Tables Figures

I∢ ≯I

■ Back Close

Full Screen / Esc

Printer-friendly Version



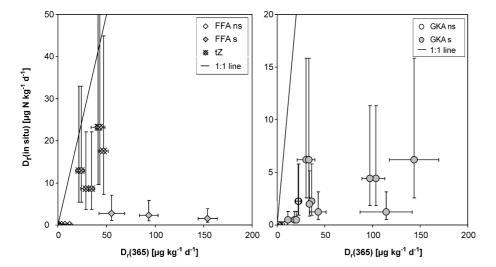


Figure 3. Relation between in situ denitrification rates determined by ¹⁵N push-pull tracer tests and average denitrification rates during one year of incubation (Eschenbach and Well, 2013). FFA Fuhrberger Feld aquifer; GKA Großenkneten aquifer; ns non-sulphidic; s sulphidic and tZ transition zone aquifer material.

11, 16527-16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page

Abstract Introduction

Conclusions References

Tables Figures

I ◀ ▶I

Back Close

Full Screen / Esc

Printer-friendly Version



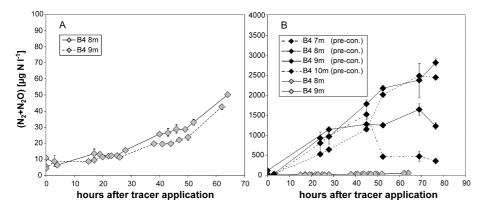


Figure 4. Time courses of $(N_2+N_2O)_{den}$ during push-pull tests without pre-conditioning **(a)** (grey diamonds) and with pre-conditioning **(b)** (black diamonds) at multilevel well B4 in the FFA. The push-pull tests without pre-conditioning at B4 were conducted in April 2010. One year later in April 2011 the aquifer material of the respective depths was conditioned over 5 weeks with NO_3^- amended groundwater of natural ^{15}N abundance prior to the ^{15}N push-pull tests.

11, 16527-16572, 2014

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page

Abstract Introduction

Conclusions References

Tables Figures

•

Back Close
Full Screen / Esc

Printer-friendly Version





Predicting the denitrification capacity of sandy aquifers

BGD

11, 16527-16572, 2014

W. Eschenbach and R. Well





Full Screen / Esc

Printer-friendly Version



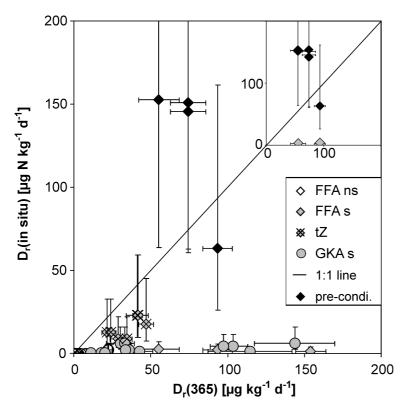


Figure 5. D_r(in situ) after 5 weeks of pre-conditioning of aquifer material (black diamonds) in comparison to D_r (in situ) without pre-conditioning. The small diagram shows the difference between D_r (in situ) after pre-conditioning and unconditioned D_r (in situ) at multilevel well B4 in the FFA.



Back Full Screen / Esc

Printer-friendly Version

Interactive Discussion



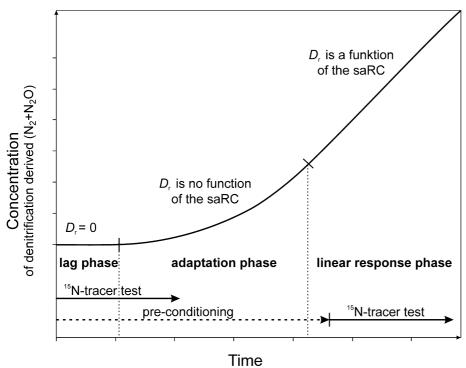


Figure 6. Schematic time courses of denitrification during push-pull tests in the NO₃-free groundwater zone. (D_r = measured in situ denitrification rates, saRC = surface area of reduced compounds present in the testet aquifer.)

11, 16527-16572, 2014

BGD

Predicting the denitrification capacity of sandy aquifers

W. Eschenbach and R. Well

Title Page

Introduction **Abstract**

Conclusions References

> **Tables Figures**

Close