1 Abstract

 $\mathbf{2}$ Iron (Fe) can limit phytoplankton productivity in approximately 40% of the global 3 ocean, including high-nutrient, low-chlorophyll (HNLC) waters. However, there is little 4 information available on the impact of CO₂-induced seawater acidification on natural phytoplankton assemblages in HNLC regions. We therefore conducted an on-deck $\mathbf{5}$ 6 experiment manipulating CO₂ and Fe using Fe-deficient Bering Sea waters during the $\mathbf{7}$ summer of 2009. The concentrations of CO_2 in the incubation bottles were set at 380 8 and 600 ppm in the non-Fe-added (control) bottles and 180, 380, 600, and 1000 ppm in 9 the Fe-added bottles. The phytoplankton assemblages were primarily composed of 10 diatoms followed by haptophytes in all incubation bottles as estimated by pigment 11 signatures throughout the 5 (controls) or 6 (Fe-added treatments) days incubation period. 12At the end of incubation, the relative contribution of diatoms to chlorophyll a biomass 13was significantly higher in the 380 ppm CO₂ treatment than in the 600 ppm treatment in 14the controls, whereas minimal changes were found in the Fe-added treatments. These 15results indicate that, under Fe-deficient conditions, the growth of diatoms could be 16negatively affected by the increase in CO₂ availability. To further support this finding, 17we estimated the expression and phylogeny of *rbcL* (which encodes the large subunit of 18 RubisCO) mRNA in diatoms by quantitative reverse transcription PCR and clone 19library techniques, respectively. Interestingly, regardless of Fe availability, the 20transcript abundance of *rbcL* decreased in the high CO₂ treatments (600 and 1000 ppm). 21The present study suggests that the projected future increase in seawater pCO_2 could 22reduce the RubisCO transcription of diatoms, resulting in a decrease in primary 23productivity and a shift in the food web structure of the Bering Sea.

24

25 **1. Introduction**

The atmospheric CO_2 concentration has risen from a pre-industrial level of approximately 280 ppm to the present level of approximately 400 ppm (WMO, 2013). Since the industrial revolution, the ocean has absorbed about one-third of CO_2 emitted by human activity (Sabine et al., 2004). It is predicted that the atmospheric CO_2 concentration could reach more than 700 ppm by the end of the 21st century (Meehl et al, 2007), driving a surface seawater pH decrease of 0.3–0.4, the so-called "ocean acidification" (Caldeira and Wickett, 2003). Such a rapid decrease in seawater pH has
most likely not occurred for at least millions of years in the earth's history (Pearson and
Palmer, 2000). Therefore, it has been suggested that these predicted changes in seawater
carbonate chemistry would have enormous impacts on the health and function of marine
organisms (Raven et al., 2005).

37 In the last decade, numerous studies have been performed to evaluate the impacts of 38ocean acidification on marine phytoplankton. In laboratory incubation experiments 39 using individual species (a single strain), the response of phytoplankton to increased CO₂ levels differed among phytoplankton species, possibly depending on their ability to 40 41 assimilate carbon (Riebesell and Tortell, 2011; Collins et al., 2014). In the natural 42environment, these taxon-specific differences in CO₂ response can cause a shift in the 43phytoplankton community composition (Engel et al., 2008; Meakin and Wyman, 2011; 44 Endo et al., 2013) and subsequent changes in ocean trophic structures and 45 biogeochemical cycles (Riebesell et al., 2007; Yoshimura et al., 2013). However, the 46 current understanding of the effects of elevated CO₂ on marine phytoplankton is still 47insufficient at the community level.

48In terms of physiology, CO_2 is fixed by the carboxylation enzyme ribulose 49bisphosphate carboxylase/oxygenase (RubisCO) in the Calvin-Benson-Bassham (CBB) 50cycle. In general, the half-saturation constant of the enzyme ranges between 20 and 70 μ mol kg⁻¹ CO₂ (Badger et al., 1998), whereas the ambient seawater CO₂ levels are 51between 10 and 25 μ mol kg⁻¹. Therefore, the present CO₂ concentration could be 5253insufficient to ensure effective RubisCO carboxylation. The progression of ocean 54acidification could enhance photosynthetic carbon fixation in marine phytoplankton by increasing CO₂ availability. 5556Recent advances in molecular biology techniques have enabled us to examine the

taxon-specific responses to environmental changes by quantifying functional gene
expression in natural phytoplankton assemblages. For example, John et al. (2007a)
developed a suite of quantitative reverse transcription PCR (qRT-PCR) assays to
quantify *rbcL* (gene encoding the large subunit of RubisCO) mRNA in *Synechococcus*,
haptophytes, and heterokonts including diatoms. John et al. (2007b) demonstrated a
strong negative correlation between diatom-specific *rbcL* mRNA abundance and

63 seawater pCO_2 in the Mississippi River plume, suggesting that diatoms were

64 responsible for the greatest drawdown in seawater pCO_2 . In addition, positive

65 correlations between diatom-specific *rbcL* transcripts and light-saturated photosynthetic

rates (P_{max}) in seawater were reported (Corredor et al., 2004; John et al., 2007b). These

67 results suggest that *rbcL* expression in diatoms could be used to estimate the

68 photosynthetic carbon-fixation capacity of natural phytoplankton assemblages.

69 Therefore, quantification of clade-specific *rbcL* transcripts can be used to assess the

70 physiological photosynthetic responses of individual phytoplankton taxa to

71 environmental changes.

72The oceanic Bering Sea investigated in this study is an HNLC region (Banse and 73English, 1999), where low iron (Fe) availability limits phytoplankton growth and nitrate 74utilization, so surface chlorophyll a (Chl a) concentrations usually remain low in the 75summer (Suzuki et al., 2002). Despite the low phytoplankton biomass, the oceanic 76domain has the greatest amount of total primary and secondary production in the Bering Sea (Springer et al., 1996). Suzuki et al. (2002) reported that diatoms were the dominant 7778phytoplankton group in the oceanic regions of the Bering Sea in the summer. In 79addition, Takahashi et al. (2002) showed that diatoms had the greatest contribution in 80 the sinking particles in the area. However, less is known about the combined effects of 81 ocean acidification and Fe enrichment on diatoms in such HNLC regions. In addition, 82 there are no reports on the effects of CO₂ and Fe availability on *rbcL* transcription of 83 natural diatom community in HNLC regions.

84 The purpose of this study is to clarify the responses of phytoplankton, especially 85 diatoms, to CO₂ enrichment under Fe-depleted and Fe-replete conditions in the Bering Sea basin using on-deck bottle incubation. Recently, Sugie et al. (2013) reported 86 87 changes in phytoplankton biomass and nutrient stoichiometry in this experiment. They 88 showed that Chl *a* biomass decreased with increased CO₂ levels only in Fe-depleted 89 treatments, suggesting that Fe deficiency and increased CO₂ synergistically reduced the 90 growth of phytoplankton in the study area. In addition, Yoshimura et al. (2014) 91 demonstrated that the net production of particulate organic carbon (POC) and total 92organic carbon (TOC) decreased under high CO₂ levels only in the Fe-limited 93 treatments, whereas those in the Fe-replete treatments were insignificantly different.

- 94 These studies suggest that the increase in CO₂ could have negative impacts on
- 95 phytoplankton growth and/or organic-matter production especially under Fe-depleted
- 96 conditions. However, the molecular mechanisms of photosynthetic carbon assimilation
- 97 in phytoplankton assemblages were not mentioned in the previous studies. Therefore, in
- 98 the present paper, we primarily focused on changes in *rbcL* transcripts in diatoms with
- 99 different CO₂ and/or Fe availability.
- 100

101 **2. Materials and Methods**

102 **2.1 Experimental setup**

103 The study was carried out aboard the R/V Hakuho Maru (JAMSTEC) during the 104 KH-09-4 cruise in September 2009. The water samples for incubation were collected 105from 10 m depth at a station (53° 05' N, 177° 00' W) in the Bering Sea on 9 September 106 with acid-cleaned Niskin-X bottles attached to a CTD-CMS system. A total of 300 L of 107 seawater was poured into six 50 L polypropylene carboys through acid-clean silicon 108 tubing with a 197 µm mesh Teflon net to remove large particles. Subsamples were taken 109 from each carboy and poured into triplicate acid-cleaned 12 L polycarbonate bottles 110 (total 18 bottles) for incubation. Initial samples were collected from each carboy. All 111 sampling was carried out using a trace-metal clean technique to avoid any trace metal contamination. Prior to incubation, $FeCl_3$ solutions (5 nmol L^{-1} in final concentration) 112113were added to 12 bottles in order to reduce Fe limitation for the phytoplankton 114communities. The CO₂ levels in the incubation bottles were manipulated by injecting 115CO₂ controlled dry air purchased from a commercial gas supply company 116 (Nissan-Tanaka Co., Japan). The air mixtures were passed through 47 mm PTFE filters 117 (0.2 µm pore size, Millipore) before being added to the incubation bottles. The detailed 118 procedures for trace metal clean techniques were described in Yoshimura et al. (2013). 119 The CO₂ concentrations were set at 380 and 600 ppm for the non-Fe-added (control) 120bottles (hereafter referred to as 'C-380' and 'C-600', respectively), and 180, 380, 600, 121and 1000 ppm for the Fe-added bottles (hereafter referred to as 'Fe-180', 'Fe-380', 122'Fe-600', and 'Fe-1000', respectively). Incubation was performed on deck in 123temperature-controlled water-circulating tanks for 5 (controls) or 6 (Fe-added 124treatments) days at the in situ temperature (8.2°C) and 50% surface irradiance adjusted

by natural density screens. The sampling opportunities for each parameter are shown inTable S1.

127

128 **2.2** Carbonate chemistry, nutrients, and Chl *a*

129The detailed methodology and basic chemical and biological parameters were 130 reported in Sugie et al. (2013). In brief, during the incubation experiment, samples were 131 collected from the incubation bottles for dissolved inorganic carbon (DIC), total 132 alkalinity (TA), nutrients, and Chl a determination. DIC and TA concentrations were 133 measured with a total alkalinity analyzer using the potentiometric Gran plot method 134 (Kimoto Electric) following Edmond (1970). The levels of pCO_2 and pH were 135calculated from the DIC and TA using the CO2SYS program (Lewis and Wallance, 136 1998). Concentrations of nitrate plus nitrite, nitrite, phosphate, and silicic acid were 137 measured using a QuAATro-2 continuous-flow analyzer (Bran+Luebbe). The 138 concentration of total dissolved Fe (TD-Fe) was determined by a flow-injection method 139 with chemiluminescence detection (Obata et al., 1993). Chl a concentrations were 140 determined with a Turner Design fluorometer (model 10-AU) with the non-acidification

- determined with a runner Design motorineter (moder 10-700) with the
- 141 method (Welschmeyer, 1994).
- 142

143 **2.3 HPLC and CHEMTAX analyses**

Samples for high-performance liquid chromatography (HPLC) pigment analysis were collected on days 3 and 5 for the control treatments and on days 2, 4, and 6 for the Fe-added treatments. Water samples (400–1000 mL) were filtered onto GF/F filters under gentle vacuum (< 0.013 MPa) and stored in liquid nitrogen or a deep freezer (-80°C) until analysis. HPLC pigment analysis was performed following the method of Endo et al. (2013).

To estimate the temporal changes in phytoplankton community structure during incubation, the CHEMTAX program (MacKey et al., 1996) was used following Endo et al. (2013). Briefly, optimal initial ratios were obtained following the method of Latasa (2007). Matrix A was obtained from Suzuki et al. (2002) (Table S2), who examined phytoplankton community compositions in the Bering Sea. Matrices B, C, and D were also prepared to determine the optimal pigment/Chl *a* ratios (Table S2). The pigment

156ratios of Matrices B and C were double and half the Matrix A ratio, respectively. For 157Matrix D, values of 0.75, 0.5, and 0.25 for dominant (rank in high pigment/Chl a ratio: 1581-5), secondary (rank: 6-10), and minor (rank: 11-15) pigments, respectively, were 159multiplied by each pigment ratio of Matrix A. We averaged the successive convergent 160 ratios after the 10 runs among the 4 matrices to identify the most promising initial 161 pigment ratios. The calculated final pigment/Chl a ratios in both the control and 162Fe-added treatments (Table S3) were within the range of values reported in Mackey et 163 al. (1996), Wright and van den Enden (2000), and Suzuki et al. (2002).

164

165 **2.4 qPCR and qRT-PCR**

Water samples for DNA and RNA analyses were collected on days 3 and 5 for the control treatments and on days 2, 4, and 6 for the Fe-added treatments. DNA samples (400–500 mL) were collected onto 25 mm, 0.2 μ m pore size polycarbonate Nuclepore filters (Whatman) with gentle vacuum (< 0.013 MPa) and stored in liquid nitrogen or a deep freezer at –80°C until analysis. DNA extraction was performed following the method of Endo et al. (2013). Extracted DNA pellets were resuspended in 100 μ L of 10 mM Tris-HCl buffer (pH 8.5).

173 For RNA analysis, seawater samples (400-500 mL) were filtered onto 25 mm, 0.2 174 μ m pore size polycarbonate Nuclepore filters (Whatman) with gentle vacuum (< 0.013) 175MPa) and stored in 1.5 mL cryotubes previously filled with 0.2 g of muffled 0.1 mm glass beads and 600 μ L RLT buffer (Oiagen) with 10 μ L mL⁻¹ β -mercaptoethanol 176 177(Sigma, St Louis, USA). RNA samples were stored in liquid nitrogen or a deep freezer 178at -80°C until analysis. Extraction and purification of RNA samples were performed 179 using the RNeasy extraction kit (Qiagen) on a vacuum manifold with on-column DNA 180 digestion using RNase-free DNase (Qiagen) according to the manufacturer's protocol. 181 RNA was eluted using 50 µL of RNase-free H₂O. Total RNA was then reverse transcribed into complementary DNA (cDNA) using the PrimeScriptTM RT regent Kit 182183 with gDNA Eraser (TaKaRa) following the manufacturer's specifications. 184 Following Smith et al. (2006), we used double-stranded DNA and single-stranded 185cDNA standards for DNA and cDNA quantification, respectively. Standard curves for 186 rbcL DNA were generated from plasmid DNA (pUC18, TaKaRa) containing an 187 artificial gene fragment (113 bp in size) of *rbcL* from the diatom *Thalassiosira*

188 *weissflogii* (CCMP1336). The plasmid DNA was linearized with *Hind*III (TaKaRa) and

- 189 quantified using a Thermo NanoDrop spectrophotometer (ND-1000). On the other hand,
- 190 to produce the cDNA standard, a PCR-amplified *rbcL* gene fragment of *T. weissflogii*
- 191 (CCMP1336) was inserted into a plasmid DNA (pCR2.1, Invitrogen). The plasmid
- 192 DNA was purified using the Plasmid maxi kit (Qiagen) and linearized with BamHI
- 193 (TaKaRa), and in vitro transcription was performed using T7 RNA polymerase
- 194 (Invitrogen) for 2 hours at 37°C with Recombinant RNase Inhibitor (TaKaRa). To
- 195 eliminate DNA contamination, RNA was digested for 2 min at 42°C using gDNA
- 196 Eraser (TaKaRa). RNA was purified using an RNeasy column (Qiagen) following the
- 197 manufacturer's instructions and quantified with a Ribogreen RNA quantification kit
- 198 (Molecular Probes) using the manufacturer's standard. RNA was reverse transcribed
- 199 into cDNA using the PrimeScriptTM RT regent Kit with gDNA Eraser (TaKaRa).
- 200 Copy numbers of DNA and cDNA standards were calculated using the equation of 201 Smith et al. (2006), where the molecular mass of each nucleotide (or nucleotide pair) in 202 double- and single-stranded DNA is assumed to be 660 and 330 Da, respectively. Serial 203 dilutions of DNA and cDNA standards were prepared using sterilized Milli-Q water.
- To amplify the *rbcL* gene and cDNA fragments from diatoms, the following specific primer set designed by John et al. (2007a) was used.
- 206 Forward primer: 5'-GATGATGARAAYATTAACTC-3'
- 207 Reverse primer: 5'-TAWGAACCTTTWACTTCWCC-3'.
- 208 Real-time PCR amplification was performed using SYBR Premix Ex Taq II (Perfect
- 209 Real Time, TaKaRa) with primer concentrations of 0.4 µM each and a Thermal Cycler
- 210 Dice Real Time System (TP800, TaKaRa). Diluted nucleic acid standards were then
- added to the PCR mixture. The thermal cycling conditions were 95°C for 60 s, then 40
- 212 cycles of 95°C 5 s and 52°C 60 s. The fluorescence intensity of the complex formed by
- 213 SYBR green and the double-stranded PCR product was continuously monitored from
- 214 cycle 1 to 40. Quantification was achieved by the second-derivative maximum method
- (Luu-The et al., 2005), and the copy number for each sample was determined by the
- standard curves generated by serial dilutions of the standards.
- 217
- 218 **2.5 Clone libraries**

219Clone libraries of rbcL cDNA were constructed for the C-380 and C-600 samples on 220 day 3, and Fe-380 and Fe-600 samples on day 2. The cDNA samples were PCR 221amplified with the diatom-specific primer set and thermal cycling condition described 222above using the TaKaRa Ex Taq Hot Start Version (TaKaRa). Triplicate PCR products 223were mixed and then purified with agarose gel electrophoresis and the PureLink Quick 224Gel Extraction Kit (Invitrogen). Purified amplicons from cDNA samples were then 225cloned into the pCR2.1 vector using the TOPO TA cloning kit (Invitrogen) following 226 the manufacturer's instructions. Thirty-five to 50 colonies were randomly picked from 227 each clone library. Correct cDNA insertions were identified by PCR amplification using 228the M13 forward (5'-GTAAAACGACGGCCAG-3') and reverse 229(5'-CAGGAAACAGCTATGA-3') primers flanking the cloning site. Plasmid DNA 230containing the inserts was cycle-sequenced using the Big Dye Terminator v3.1 Kit 231(Applied Biosystems) with the M13 forward primer. The cycle sequencing products 232were cleaned by isopropanol precipitation. Sequencing was performed with a 3130 233Genetic Analyzer (Life Technologies). The obtained sequences were compared with 234rbcL sequences deposited in GenBank database (http://www.ncbi.nlm.nih.gov) using 235the BLAST query engine. Our rbcL cDNA sequences were deposited in the DDBJ 236database with the following accession numbers: AB985799-AB986033.

237

238 **2.6 Phylogenetic and diversity analyses**

239The *rbcL* sequences obtained were assembled into operational taxonomic units 240(OTUs) with > 95% sequence identity, and rarefaction curves were plotted for each 241clone library with the software mothur v. 1.27 (Schloss et al., 2009). To estimate OTU 242richness, chao1 index (Chao, 1984) values were calculated using the number of 243singleton sequences obtained in this study. Genetic diversity was assessed based on the 244Shannon-Wiener index (H', Shannon, 1948) and Simpson's index (D, Simpson, 1949). 245The statistical significance of differences in the compositions of pairs of *rbcL* sequences 246in the libraries was tested using LIBSHUFF (Singleton et al., 2001). The LIBSHUFF 247program determined the integral form of the Cramer-von Mises statistic for each pair of 248communities using 10,000 randomizations. Any two libraries were considered to be 249significantly different from each other if the lower of the significance values generated

250 by the software was < 0.025 (p < 0.05).

251

252 2.7 Statistical analysis

Statistical analyses were performed with the program R (http://www.r-project.org). To assess the statistically significant differences between pCO_2 levels in the control treatments or between control and Fe treatments, Welch's *t*-test was used. Differences among pCO_2 levels in the Fe-added treatments were evaluated with Kruskal-Wallis one-way analysis of variance (ANOVA). Holm's test for multiple comparisons was used to identify the source of the variance. For all of the analyses, the confidence level was set at 95% (p < 0.05).

260

261 3 Results

262 **3.1 Experimental conditions**

263The bubbling of CO₂-controlled air succeeded in creating significant gradients in 264 pCO_2 , pH, and DIC in the different CO₂ treatments except on day 4 in the Fe-added 265treatments, when those values did not significantly differ between Fe-380 and Fe-600 266(Table 1; Fig. S1). The initial concentrations of nitrate, phosphate, and silicic acid were 18.06 ± 0.10 , 1.47 ± 0.01 , and 16.90 ± 0.12 µmol L⁻¹, respectively (Table 1). In the 267268control bottles, these macronutrients remained until the end of the incubation in both 269CO₂ treatments except for silicic acid, which was almost depleted on day 5 in the C-380 270treatment (Fig. S2). In the Fe-added bottles, macronutrients were depleted on days 4 or 5 in all CO₂ treatments (Fig. S2). The TD-Fe concentration was 1.35 nmol L^{-1} in the 271272initial seawater, and it remained low throughout the experiment in the control treatments (Table 1). In the Fe-added treatments, the TD-Fe concentrations were 5.50 ± 0.10 nmol 273 L^{-1} in the initial bottles and remained above 4 nmol L^{-1} until the end of incubation 274(Table 1). The initial Chl *a* concentration was $1.96 \pm 0.14 \ \mu g \ L^{-1}$ (Table 1). In the 275control bottles, the Chl a concentration increased until the end of the incubation and 276reached $10.22 \pm 0.89 \text{ \mug L}^{-1}$ in the C-380 and $6.28 \pm 0.64 \text{ \mug L}^{-1}$ in the C-600 277treatments (Fig. S3). In the Fe-added bottles, the Chl a concentration increased rapidly 278279and reached the maximum on day 4 in the Fe-180 and Fe-380 treatments (27.51 ± 0.71) μ g L⁻¹ and 28.45 ± 3.40 μ g L⁻¹, respectively) and on day 5 in the Fe-600 and Fe-1000 280

treatments (27.68 \pm 0.44 µg L⁻¹ and 27.32 \pm 3.05 µg L⁻¹, respectively), then declined toward the end of the incubation (Fig. S3).

283

284 **3.2 Phytoplankton pigments**

285Throughout the experiment, the concentrations of fucoxanthin (Fuco), mainly a biomarker for diatoms (Ondrusek et al., 1991; Suzuki et al., 2011), and 286287 19'-hexanoyloxyfucoxanthin (19'-Hex), an indicator of haptophytes (Jeffrey and Wright, 2881994), were relatively high among the phytoplankton pigments. In the control bottles, 289the concentrations of Fuco and 19'-Hex increased over time and reached the maximum 290values on day 5 in both the C-380 and C-600 treatments (Figs. 1a and c). After day 3, 291the concentrations of Fuco and 19'-Hex were higher in the C-380 treatment than in the 292C-600 treatment (day 5: Welch's *t*-test C-380 > C-600, p < 0.05), although no statistical 293significance was assessed on day 3 because samples were collected from each single bottle. In the Fe-added bottles, Fuco concentrations increased throughout the incubation 294295and reached the maximum values on day 6, whereas 19'-Hex concentrations decreased 296 after day 4 (Figs. 1b and d). The concentrations of Fuco were significantly different 297 among CO₂ treatments on day 6 (Kruskal-Wallis ANOVA, p < 0.05), although these 298differences were not supported by multiple comparisons (Holm's test, p > 0.05). 299Significant differences among CO₂ treatments were also found for the 19'-Hex 300 concentration on day 6 (Kruskal-Wallis ANOVA, p < 0.05), and the values in the 301 Fe-180 treatment was significantly higher than those in the Fe-1000 treatment (Holm's 302 test, p < 0.05).

303

304 3.3 CHEMTAX outputs

In the initial phytoplankton community, diatoms and haptophytes were the predominant numbers of the phytoplankton groups (i.e., they contributed 45% and 17% of the Chl *a* concentration, respectively). The initial contributions of chlorophytes, cryptophytes, peridinin-containing dinoflagellates, pelagophytes, prasinophytes, and cyanobacteria to the Chl *a* biomass were 10%, 9%, 8%, 5%, 4%, and 2%, respectively. In the control bottles, the contributions of diatoms to the Chl *a* biomass increased with

311 time, and their contributions reached the maximum (70% at the C-380 and 60% at the

C-600 treatments) on day 5 (Fig. 2a). On day 5, the contribution of diatoms in the 313 C-380 treatment was significantly higher than that in the C-600 treatment (Welch's 314*t*-test, p < 0.05). However, the contribution of haptophytes to the Chl *a* biomass was 315higher in the C-600 treatment (21%) than in the C-380 treatment (14%) on day 5 316 (Welch's *t*-test, p < 0.05). Increases in the contributions of diatoms were also observed 317 in the Fe-added treatment, and the contributions reached the maximum (82–85%) on 318 day 4 in all CO₂ treatments (Fig. 2b). In terms of diatom contribution, a significant 319 difference among CO_2 treatments was not detected with Kruskal-Wallis ANOVA (p >320 0.05) in the Fe-added bottles. The contributions of haptophytes to Chl a biomass did not 321 differ significantly among CO₂ levels in the Fe-added bottles (Kruskal-Wallis ANOVA, 322 p > 0.05).

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312

3243.4 Expression of diatom *rbcL* gene

325 A significant linear relationship between the Fuco concentration and the diatom-specific *rbcL* gene copy number was found (regression analysis: $r^2 = 0.677$, p < 100326 3270.001, n = 28) in our experiment (Fig. 3). In the control bottles, the transcript abundance 328 normalized to gene abundance (i.e., cDNA/DNA) of the diatom-specific rbcL gene 329fragment for the C-380 treatment was significantly higher than that of the C-600 330 treatment on day 3 (Fig. 4; Welch's *t*-test, p < 0.05). In the Fe-added bottles, the 331 cDNA/DNA ratio of the diatom rbcL fragment in the lower CO₂ treatments (Fe-180 and 332 Fe-380) was higher than that in the Fe-600 treatment on day 2 (Fig. 4; Holm's test, p < p333 0.05).

334

335 3.5 Clone libraries of diatom rbcL cDNA

336 Rarefaction curves were plotted for the *rbcL* cDNA libraries (Fig. 5). In terms of 337 unique taxa, the highest number of OTUs was found in the C-380 treatment (Table 2). 338 The highest chaol value was found in the C-600 treatment, whereas the lowest value 339 was found in the Fe-600 treatment. Shannon-Wiener and Simpson diversity indices 340 revealed that the cDNA libraries in the 380 ppm CO₂ bottles were more diverse than those in the 600 ppm CO₂ bottles in both the control and Fe-added treatments, although 341

the values were not statistically significant between CO₂ treatments (*t*-test, p > 0.05) 342

343 (Table 2).

344 All sequences obtained from the cDNA libraries were more than 95% similar to 345 sequences deposited in the GenBank. These sequences could be classified into the 346 following 11 phylogenetic groups: Chaetocerotaceae, Coscinodiscaceae, 347 Cymatosiraceae. Stephanodiscaceae. Thalassiosiraceae. unidentified centrics. 348 Bacillariaceae, Naviculaceae, Fragillariaceae, unidentified pennates, and other 349 eukaryotes by comparison with known rbcL sequences from GenBank. Sequences that 350 could not classified into a specific diatom family (e.g., closely related to two or more 351diatom families with same similarity score) were assigned as unidentified centrics or 352 unidentified pennates. Other eukaryotes consisted of haptophytes, pelagophytes, 353 dictyochophytes, dinoflagellates, and diatoms which could not be assigned to centrics 354and pennates. For all of the cDNA libraries, more than 88% of *rbcL* sequences were 355 most closely affiliated with those of cultured diatoms. In the initial cDNA library, the 356 most abundant sequences were closely affiliated with the diatom family Bacillariaceae 357 (46%), followed by other eukaryotes and Cymatosiraceae (17% and 14%, respectively) 358 (Fig. 6). The contributions of other diatom groups were less than 6% in the initial clone 359 library. In the control bottles, the contributions of Coscinodiscaseae increased to 12-360 14%, whereas those of Cymatosiraceae decreased to 4%. In the Fe-added bottles, the 361 contributions of Chaetocerotaceae and unidentified centrics to the total increased to 362 more than 8% and 20%, respectively. In contrast, the contributions of Bacillariaceae 363 decreased below 24% in both the Fe-380 and Fe-600 treatments. 364 Statistic analysis using LIBSHUFF revealed that the cDNA libraries in the control 365 treatments were not significantly different from the initial sample regardless of the CO₂ 366 level, whereas those in the Fe-added bottles differed significantly from the initial 367 assemblage (LIBSHUFF, p < 0.05) (Table 3). No significant difference in the cDNA 368 library was found between C-380 and C-600 treatments in the control bottles 369 (LIBSHUFF, p > 0.05). However, a significant difference between the Fe-380 and 370Fe-600 treatments was detected in the Fe-added bottles (LIBSHUFF, p < 0.05). In 371 addition, cDNA libraries in the Fe-added bottles differed significantly from those of the

372 control bottles in both the Fe-380 and Fe-600 treatments (LIBSHUFF, p < 0.05).

4 Discussion

375 4.1 Changes in phytoplankton community structure during incubation

376 Our CHEMTAX analysis suggested that the diatoms were the principal contributors 377 to the Chl *a* biomass in the initial phytoplankton community, followed by haptophytes 378 (Fig. 2). The results were consistent with those reported by Suzuki et al. (2002), who 379 examined the community structure in the Bering Sea during early summer of 1999. 380 These results suggest that diatoms and haptophytes are ecologically important 381 phytoplankton groups in the study area during the summer. Compared with previous 382 reports in the area (Suzuki et al., 2002; Yoshimura et al., 2013), a relatively high initial 383 Chl a concentration was observed in our experiment, possibly due to an intrusion of the 384 coastal seawater mass from the Aleutian trenches (Sugie et al., 2013). However, the Fe 385 infusion induced significant increases in Chl a biomass and concomitant rapid 386 drawdowns of macronutrients in our incubation bottles (Fig. S2). This indicates that the 387 seawater used for the incubation was Fe-limited for phytoplankton assemblages. Our 388 HPLC and CHEMTAX results suggested that the increase in phytoplankton biomass 389 was mainly due to an increase in diatoms (Figs. 1b and 2b).

390 We found that the growth of Fuco was less in the high CO₂ bottles in the control 391 treatments (Fig. 1a), suggesting that the elevated CO₂ levels could have a negative 392 impact on the diatom biomass in the study area. Negative effects on diatoms induced by 393 an increase in CO₂ availability were also reported in field incubation experiments 394 conducted in the Bering Sea and the Okhotsk Sea (Hare et al., 2007 and Yoshimura et 395 al., 2010, respectively). However, such trends have rarely been observed in other 396 regions of the world's oceans (e.g., Tortell et al., 2002; Kim et al., 2006; Feng et al., 397 2009; Hoppe et al., 2013; Endo et al., 2013). Therefore, the responses of phytoplankton 398 assemblages to ocean acidification can differ among geographic locations due to the 399 differences in the biogeography of phytoplankton and/or environmental conditions. 400 One possible cause of the geographic specificity in the open Bering Sea is the 401 differences in the species composition of diatoms. Our microscope data showed that 402 centric diatoms such as Chaetocerataceae and Rhizosoreniaceae were predominant at 403 the beginning of the incubation in terms of carbon biomass, and the coastal diatom 404 species Chaetoceros spp. became predominant in all incubation bottles after day 2 405(Sugie et al., 2013). Therefore, the relative decrease in Fuco biomass with increased

406 CO₂ levels might be partially explained by the decrease in *Chaetoceros* spp. A previous 407 field incubation experiment conducted in the Bering Sea also showed that the carbon 408 biomass of the *Chaetoceros* spp. decreased at higher CO₂ levels (600–960 µatm CO₂), 409 although it increased at 1190 µatm CO₂ (Yoshimura et al., 2013). However, Tortell et al. 410 (2008) demonstrated that relative abundance of Chaetoceros spp. increased under 411 elevated CO₂ levels in the Ross Sea. In the previous laboratory culture experiments, the 412effects of increased CO₂ on the growth and/or photosynthesis of *Chaetoceros* spp. were 413 also inconsistent. For example, Ihnken et al. (2011) demonstrated that the growth of 414 diatom Chaetoceros muelleri decreased with elevated CO₂ (decreased pH) levels 415 although their photosynthetic capacity increased. In contrast, Trimborn et al. (2013) 416 showed a significant increase in the growth rate of *Chaetoceros debilis* under high CO₂ 417condition. In addition, no CO₂-related change in the growth and photosynthetic 418 physiology of Chaetoceros brevis was found (Boelen et al., 2011). These results suggest 419 that the responses to elevated CO₂ differ among *Chaetoceros* species. 420The concentrations of 19'-Hex were significantly lower in the C-600 treatment than 421those in the C-380 treatment (Fig. 1c), suggesting that the ocean acidification could 422induce negative effects not only on the biomass of diatoms, but also on that of 423 haptophytes in the study area. Similar results were obtained from the previous field 424studies in other regions (e.g., Feng et al., 2010; Endo et al., 2013). One possible factor 425underlying these decreases is that the reduced carbonate-saturation states under high 426 CO₂ conditions. The energetic cost of calcification in coccolithophores will increase 427with a decrease in pH (Mackinder et al., 2010). Therefore, additional energy might be 428needed for cell growth in seawater with high CO₂ levels. In addition, non-calcifying 429haptophytes such as *Phaeocystis* spp. often dominate among haptophytes in the natural 430 phytoplankton community (Schoemann et al., 2005), although the effects of ocean 431acidification on them are still not well understood. Therefore, additional study using a 432wide range of haptophyte species would be required for a detailed understanding of the 433 responses of the haptophyte community to CO₂-induced ocean acidification. 434Our CHEMTAX outputs showed that the relative contributions of diatoms decreased 435with increased CO₂ levels, whereas the contributions of haptophytes increased in both 436 the control and Fe-added bottles (Fig. 2). This indicates that the negative impacts of

437 increased CO_2 on diatoms were greater than those on haptophytes and other

438 phytoplankton groups. Another possibility is that the competitions between diatoms and 439 other phytoplankton taxa could occur. For example, diatoms could become less 440 competitive when silicic acid is exhausted, because Si-depletion significantly depressed 441 the growth and could induce their cell death (Harrison et al., 1977; Jiang et al. 2014). 442However, concentrations of silicic acid were not significantly different among CO₂ 443 levels in the Fe-added treatments (Fig. S2f). Moreover, in the control treatments, silicic 444 acid was almost depleted in the low CO_2 treatment after day 5 but not in the high CO_2 445treatment (Fig. S2e). These results suggest that availability of silicic acid little affected 446 the decreases in relative diatom contribution to Chl *a* biomass. Larger diatoms can 447 contribute to efficient transfer of energy and organic compounds to higher trophic levels 448 because they would create a shorter food chain compared with nano- and pico-sized 449 phytoplankton (Michaels and Silver, 1988). Because diatoms form a large part of 450phytoplankton biomass in the Bering Sea basin (Suzuki et al., 2002; Takahashi et al., 4512002), the decrease in the relative contribution of diatoms with increasing CO₂ could reduce the energy transferred from the primary producers to the higher trophic levels. 452453The decreases in Fuco growth and relative contribution of diatoms were larger in the 454control bottles than those in the Fe-added treatments (Figs. 1 and 2), suggesting that the 455negative effect of CO₂ enrichment was greater in the Fe-limited conditions. These 456results are consistent with Sugie et al. (2013) and Yoshimura et al. (2014), who 457observed significant decreases in diatom carbon biomass and particulate organic carbon 458(POC) production under high CO₂ levels in the control treatments, whereas those were 459insignificantly changed in the Fe-added treatments. Sugie et al. (2013) indicated that the 460 Fe limitations for phytoplankton in the control bottles were enhanced at high CO₂ levels, likely due to the reduction of Fe bioavailability as reported in Shi et al. (2010). The 461 462combined effects of CO₂ and Fe availability were also tested in a diatom-dominated 463 phytoplankton community in the Southern Ocean (Hoppe et al., 2013). In their study, 464 net primary productivity in seawater decreased with increased pCO_2 levels in the 465 Fe-depleted treatments but not in the Fe-enriched treatments. These studies indicate that 466 an interactive effect of CO₂ enrichment and Fe limitation could occur in the 467 diatom-dominated natural phytoplankton assemblages in the HNLC region. 468

469 **4.2** *rbcL* expression in diatoms

470 A significant correlation between diatom *rbcL* copies per liter and Fuco 471concentration was found in this study (Fig. 3), suggesting the usefulness of the *rbcL* 472gene fragment as a proxy for diatoms. In addition, the cDNA sequences obtained from 473cloning were dominated by the diatom-derived rbcL gene (Fig. 6). These results 474indicate that the *rbcL* primers used successfully and selectively amplified the *rbcL* gene 475of diatoms. Suzuki et al. (2011) showed that Fuco concentration significantly correlated 476with diatom carbon biomass in the subarctic Pacific. Furthermore, Matsuda et al. (2011) 477showed that the number of *rbcL* gene per cell varies among diatom species, and it was 478positively correlated with cell size. Therefore, we concluded that the *rbcL* gene could 479serve as a potential molecular marker for diatom biomass.

480 The transcript abundance of the diatom-specific *rbcL* gene decreased with elevated 481 CO₂ levels in both the control and Fe-added treatments (Fig. 4). Because RubisCO 482expression is primarily controlled at the transcriptional level in the natural 483 phytoplankton community (Xu and Tabita, 1996; Wawrik et al., 2002), our results 484 suggest that increased CO₂ levels could reduce the amount of RubisCO in diatoms. It 485should be noted that significant decreases in *rbcL* expression with increased CO₂ levels 486were observed on days 2 or 3, when macronutrients still remained (Fig. S2). This 487indicates that the downregulation of *rbcL* expression in diatoms was probably caused by 488the increase in CO₂ availability. It has been shown that some land plants can increase 489their nitrogen utilization efficiency under elevated CO₂ levels by reducing the 490 investment of nitrogen in RubisCO (Curtis et al., 1989; Makino et al., 2003). Losh et al. 491 (2012; 2013) also demonstrated a decreased RubisCO contribution to the total protein in 492the California Current phytoplankton community with an increase in CO₂ level. 493 Because a decrease in the expression of RubisCO can result in a reduction of the 494potential capacity for carbon fixation in the natural environment (John et al., 2007b), 495our results indicate that an increase in CO₂ levels could have a negative impact on 496photosynthetic carbon fixation for diatoms in the study area. Recently, Gontero and 497Salvucci (2014) pointed out that RubisCO activase plays a key role in the modification 498of RubisCO activity, and consequently in the capacity of carbon fixation, although the 499occurrence of RubisCO activase in diatoms is not well understood. Further studies must 500be needed for better understanding of the impacts of elevated CO₂ on photosynthetic

501 physiology in diatoms.

502The negative effects of increasing CO_2 on diatom biomass were not severe in the 503Fe-added bottles relative to Fe-limited control bottles (Figs. 1a and b), whereas *rbcL* 504transcripts decreased with increased CO₂ regardless of Fe availability (Fig. 4). This 505suggests that the diatoms could overcome the decrease in RubisCO activity in the 506 Fe-added treatments. According to our cloning data (Fig. 6), a shift in phylogenetic 507composition of the diatoms actively transcribed *rbcL* was observed in the Fe-added 508bottles. In addition, $F_{\rm v}/F_{\rm m}$ values increased significantly with Fe enrichment in our incubation experiments (Sugie et al., 2013), indicating an increase in the photochemical 509510quantum efficiency of photosystem II for the diatoms. Therefore, the photosystem II 511activity might compensate for the decrease in RubisCO expression under Fe-replete 512conditions.

513It is generally recognized that phytoplankton autonomously regulate the transcription 514of the *rbcL* gene in response to environmental conditions such as light and nutrient 515availability (Pichard et al., 1996; Granum et al., 2009; John et al., 2010). However, the 516mechanisms controlling the transcription of RubisCO operon in diatoms are largely 517unknown. Recently, Minoda et al. (2010) showed that the red alga Cvanidioschyzon 518merolae increased rbcL transcription at high levels of NADPH, 3-phoshoglyceric acid 519(3-PGA), or ribulose-1,5-bisphosphate (RuBP) under the influence of the transcription 520factor Ycf30. In addition, it has been reported that regeneration of RuBP could be a 521limiting factor for the CBB cycle in high CO₂ conditions (von Caemmerer and Farquhar, 5221981; Stitt, 1991; Onoda et al., 2005). Thus, one possible mechanism underlying the 523reduction of diatom *rbcL* transcripts observed in our study is related to a decrease in 524RuBP concentration in the chloroplasts due to the increase in CO₂ availability for 525diatoms. Because diatoms possess the same type of RubisCO (Form ID) and gene 526homologs encoding the Ycf30 protein (i.e., ycf30) (Kowallik et al., 1995), they could 527control *rbcL* gene expression using the same mechanisms as *C. merolae*. Further studies 528using marine diatom cultures are required to obtain a better understanding of the 529physiological mechanisms controlling the expression of RubisCO. 530In our experiment, the rarefaction curves plateaued to some extent in all treatments

531 (Fig. 5), indicating that the clone numbers screened from each library were statistically

532sufficient for further diversity analysis. Taxonomic compositions in the cDNA library 533 were considerably different from those in the diatom carbon biomass revealed by 534microscopic analysis by Sugie et al. (2013), which were composed primarily by 535Chaetocerataceae. This implies that the predominant diatoms did not necessarily 536 become transcriptionally active rbcL phylotypes in our experiment. In addition, because 537 16-42% of the sequences were classified as unidentified diatoms or other eukaryotes, 538the primer set used in this study might be insufficient to estimate diatom composition at 539 the family level.

540The *rbcL* cDNA libraries in the Fe-added treatments differed significantly from the 541initial library, whereas those in the control treatments were not significantly different 542(Table 3), suggesting that the diatom blooms induced by Fe infusion were associated 543with the change in the relative contribution of *rbcL* expression in diatoms. For example, 544compared to the initial seawater, the relative contributions of Chaetocerataceae and 545unidentified centrics to the *rbcL* cDNA library increased markedly in the Fe-added 546bottles whereas they remained minor components in the control bottles (Fig. 6). This 547indicates that the relative significance of the RubisCO activity of these phylotypes could 548be increased by Fe enrichment. In addition, cDNA libraries were significantly different 549from each other at different CO₂ levels in the Fe-added bottles (Table 3). This indicates 550that the transcriptionally active phylotypes in diatoms could shift in response to an 551increase in the CO_2 level. On the other hand, the diversity indices for the 552diatom-specific *rbcL* cDNA sequences were not affected by CO₂ availability (Table 2). 553In addition, the highest chao1 (richness) value was observed in C-600 treatment. These 554results suggest that the richness and/or diversity of diatom phylotypes actively 555transcribing *rbcL* gene could remain under elevated CO₂ levels.

556

557 5 Conclusions

The present study showed that an increase in CO_2 levels could have negative impacts on diatom biomass in the Bering Sea, especially under Fe-limited conditions. Because diatoms play pivotal roles in carbon sequestration and food webs in the Bering Sea

- 561 (Springer et al., 1996; Takahashi et al., 2002), our results indicate that ocean
- acidification might alter the biogeochemical processes and ecological dynamics in the

- 563 study area. Although the present results cannot be extrapolated to other HNLC
- 564 ecosystems due to differences in other environmental conditions, our findings suggest
- that the combined effects of CO_2 and other environmental factors such as Fe availability
- need to be examined for a better understanding of the potential impacts of ocean
- acidification on marine ecosystems.
- 568 We examined, for the first time, the relationships between CO₂ levels or Fe 569availability and RubisCO expression of diatoms in the Bering Sea. Significant decreases 570in the *rbcL* expression of diatoms were observed at elevated CO₂ levels in both the 571Fe-limited and Fe-enriched treatments, suggesting that ocean acidification could reduce 572the primary productivity in the study area. Our results indicate that the amount of *rbcL* 573 transcripts could be an important indicator to assess the physiological responses of 574RubisCO activity in diatoms to environmental drivers. However, photosynthetic carbon 575fixation in diatoms can be controlled not only by RubisCO activity, but also other 576 processes such as carbon concentrating mechanisms (CCMs) and/or RuBP regeneration 577(Rost et al., 2003; Onoda et al., 2005). More detailed studies on molecular mechanisms 578are required to clarify the physiological responses of the diatom community to CO₂ and 579Fe availability.
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production: Microcosm experiments using iron deficient plankton communities in
open subarctic waters, J. Oceanogr., 69, 601–618, 2013.

829 **Table 1.** Carbonate chemistry, nutrients, and Fe parameters (value ± 1 standard 830 deviation, n = 3) during the incubation experiment. Carbonate parameters are the initial and mean values throughout the incubation. Macronutrients and Fe parameters are the 831 832 values at the initial or final sampling days (i.e., day 5 for the control and day 6 for the Fe-added treatments). Standard deviation was not assessed for initial TD-Fe 833 concentration because samples were collected from single source. See figures S1 and S2 834 for the complete data set. 835 836

	DIC	ТА	pCO ₂	CO ₂	pН
	$(\mu mol kg^{-1})$	$(\mu mol kg^{-1})$	(µatm)	$(\mu mol kg^{-1})$	(Total scale)
C-Initial	2086.4 ± 2.8	2249.1 ± 5.0	388.4 ± 18.1	18.4 ± 0.9	8.05 ± 0.02
C-380	2075.5 ± 8.1	2252.9 ± 10.8	355.7 ± 34.7	16.8 ± 1.6	8.09 ± 0.04
C-600	2151.6 ± 7.8	2250.9 ± 4.7	604.1 ± 36.2	28.5 ± 1.7	7.88 ± 0.02
Fe-Initial	2085.3 ± 0.8	2250.0 ± 4.9	383.4 ± 12.6	18.1 ± 0.6	8.06 ± 0.01
Fe-180	1959.9 ± 62.0	2244.1 ± 16.0	202.0 ± 50.9	9.5 ± 2.4	8.21 ± 0.10
Fe-380	2068.5 ± 27.7	2235.7 ± 14.9	375.9 ± 47.9	17.8 ± 2.3	8.01 ± 0.05
Fe-600	2120.6 ± 33.5	2248.5 ± 12.0	512.6 ± 135.5	24.2 ± 6.4	7.96 ± 0.11
Fe-1000	2200.2 ± 12.6	2248.4 ± 9.8	913.8 ± 159.8	43.2 ± 7.6	7.72 ± 0.07

837 838

Table 1. (Continued)

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	Nitrate	Phosphate	Silicic acid	TD-Fe
	$(\mu mol L^{-1})$	$(\mu mol L^{-1})$	$(\mu mol L^{-1})$	$(nmol L^{-1})$
C-Initial	18.06 ± 0.10	1.47 ± 0.01	16.95 ± 0.12	1.35
C-380	7.09 ± 0.27	0.65 ± 0.02	0.28 ± 0.05	0.27 ± 0.03
C-600	12.01 ± 0.27	0.98 ± 0.02	3.04 ± 0.32	0.29 ± 0.04
Fe-Initial	18.09 ± 0.11	1.47 ± 0.01	16.90 ± 0.12	5.50 ± 0.10
Fe-180	0.13 ± 0.04	0.10 ± 0.01	0.66 ± 0.09	4.60 ± 0.19
Fe-380	0.09 ± 0.00	0.12 ± 0.04	0.50 ± 0.01	4.48 ± 0.12
Fe-600	0.08 ± 0.00	0.10 ± 0.00	0.50 ± 0.01	4.34 ± 0.08
Fe-1000	0.08 ± 0.00	0.08 ± 0.02	0.47 ± 0.02	4.18 ± 0.24

Table 2. Number of OTUs, richness index, and diversity indices (value \pm 95% confidence interval) for *rbcL* cDNA libraries obtained from the initial seawater and the incubation bottles on day 2 (Fe-380 and Fe-600) and day 3 (C-380 and C-600).

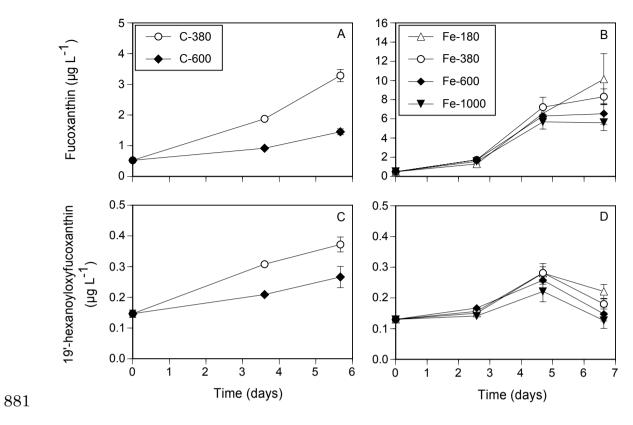
Library	No. of	No. of	Chao1	H	D
Library	sequences	OTUs	Chaor	11	
Initial	35	10	25.0	1.81 ± 0.32	0.197 ± 0.086
C-380	50	15	20.0	1.98 ± 0.36	0.232 ± 0.110
C-600	50	14	29.0	1.60 ± 0.41	0.369 ± 0.148
Fe-380	50	13	23.0	2.24 ± 0.23	0.116 ± 0.042
Fe-600	50	12	19.5	2.01 ± 0.26	0.158 ± 0.053

	Library (Y)				
	Initial	C-380	C-600	Fe-380	Fe-600
Library (X)					
Initial	—	0.434	0.573	0.383	0.587
C-380	0.153	—	0.086	0.101	0.898
C-600	0.523	0.500	—	0.004	0.033
Fe-380	<0.001	<0.001	<0.001	—	0.002
Fe-600	0.009	0.004	<0.001	0.030	_

Table 3. Significance levels for differences among *rbcL* cDNA libraries as calculated
847 with LIBSHUFF. *p* values < 0.05 are bolded.

850 Figure captions

851		
$852 \\ 853 \\ 854$	Figure 1.	(c and d) concentrations. Left (a and c) and right graphs indicate data from the control and Fe-added treatments, respectively. Error bars denote ± 1
$\begin{array}{c} 855\\ 856 \end{array}$		standard deviation (SD, $n = 3$). Standard deviations were not assessed on days 2 (Fe-added treatmets) and 3 (control treatments) because samples were
857		collected from each single bottle.
858		
859	Figure 2.	Mean contributions of each phytoplankton group to total Chl a biomass
860		estimated by CHEMTAX in the (A) control bottles at 380 and 600 ppm CO_2 ,
$\frac{861}{862}$		and (B) Fe-added bottles at 180, 380, 600 and 1000 ppm CO_2 (n = 1 or 3).
863	Figure 2	Polationship between fugewenthin concentration and distant aposition what
$\frac{863}{864}$	Figure 5.	Relationship between fucoxanthin concentration and diatom-specific <i>rbcL</i> copy number ($y = 7.62 \times 10^8 x + 1.90 \times 10^8$, $r^2 = 0.677$, $p < 0.001$, $n = 28$).
$\frac{864}{865}$		copy number $(y - 7.62 \times 10 x + 1.90 \times 10, 1 - 0.677, p < 0.001, n - 28).$
866	Figure 4	Abundances of the I mDNA (aDNA) normalized to the I gone construmber
$\frac{860}{867}$	rigure 4.	Abundances of <i>rbcL</i> mRNA (cDNA) normalized to <i>rbcL</i> gene copy number (<i>rbcL</i> cDNA/DNA) in the control bottles on day 3 and the Fe-added bottles
868		on day 2. Open bars and closed bars denote control and Fe-added bottles
869		respectively. Error bars indicate ± 1 SD (n = 3).
870		Tespectively. Entor bars indicate $\pm 1.5D$ (if -5).
871	Figure 5	Rarefaction analysis of the diatom-specific <i>rbcL</i> clone libraries. The
872	I Igui e o	rarefaction curves, plotting the number of operational taxonomic units
873		(OTUs) as a function of the number of sequences, were computed by the
874		software mothur. C and Fe indicate the control and Fe-added treatments,
875		respectively.
876		
877	Figure 6.	Relative phylotype contributions in the <i>rbcL</i> cDNA libraries obtained from
878	0	the initial seawater and the incubation bottles at day 2 (Fe-380 and Fe-600)
879		and day 3 (C-380 and C-600).
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000		



882 Figure 1

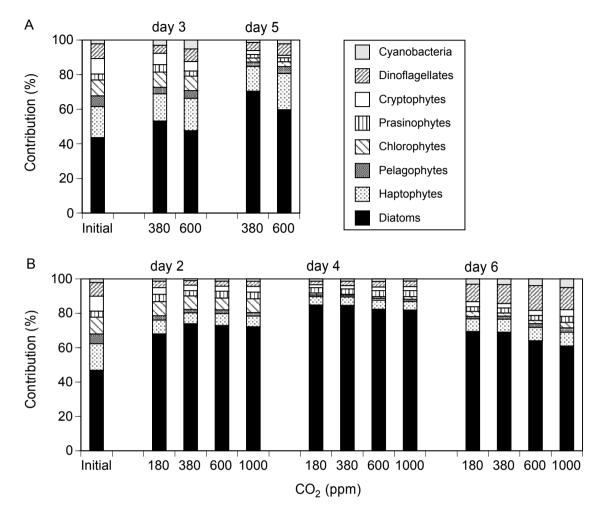
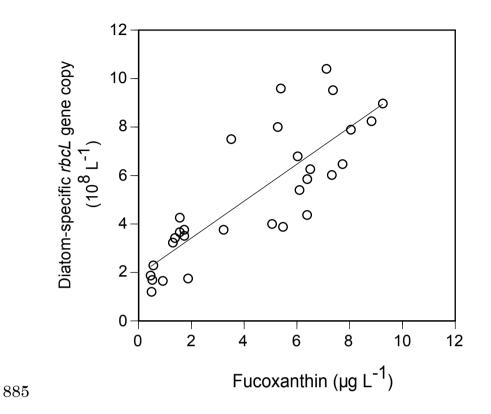
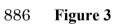
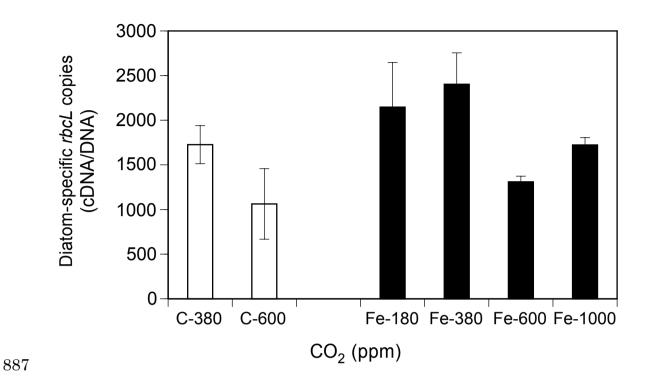


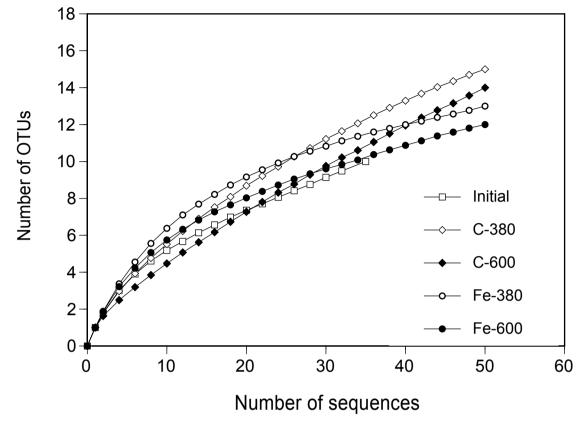
Figure 2







888 Figure 4



890 Figure 5

