

1 **Short-term effects of biogas digestate and cattle slurry**  
2 **application on greenhouse gas emissions and N availability**  
3 **from high organic carbon grasslands**

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16 nitrogen, emission factor, fertilization, N-use efficiency, N-balance.

17 **Abstract**

18 The change in the German energy policy resulted in a strong increase in the number  
19 of biogas plants in Germany. As a consequence, huge amounts of nutrient rich  
20 residues remain from the fermentative process, which are used as organic fertilizers.  
21 Drained peatlands are increasingly used to satisfy the huge demand for fermentative  
22 substrates (e.g. energy-crops, grass silage) and the digestate is returned to the  
23 peatlands. However, drained organic soils are considered as hot spots for nitrous  
24 oxide (N<sub>2</sub>O) emissions and organic fertilization is additionally known to increase N<sub>2</sub>O  
25 emissions from managed grasslands. Our study addressed the questions a) to what  
26 extent biogas digestate and cattle slurry application increase N<sub>2</sub>O, methane (CH<sub>4</sub>)  
27 and ammonia (NH<sub>3</sub>) fluxes as well as the mineral nitrogen use efficiency (NUE<sub>min</sub>)  
28 and grass yield, and b) how different soil organic matter contents (SOM) promote the  
29 production of N<sub>2</sub>O. The study was conducted at two areas within a grassland parcel,  
30 which differed in their soil organic carbon (SOC) contents. At each area (named C<sub>org</sub>-  
31 medium and C<sub>org</sub>-high) three sites were established, one was fertilized five times with  
32 biogas digestate, one with cattle slurry and the third served as control site. For each

33 site, fluxes of N<sub>2</sub>O and CH<sub>4</sub> were measured over two years using the closed chamber  
34 method. For NH<sub>3</sub> measurements we used the calibrated dynamic chamber method.  
35 On an annual basis, the application of biogas digestate significantly enhanced the  
36 N<sub>2</sub>O fluxes compared to the application of cattle slurry and additionally increased the  
37 NUE<sub>min</sub>. Furthermore, N<sub>2</sub>O fluxes from the C<sub>org</sub>-high site significantly exceeded N<sub>2</sub>O  
38 fluxes from the C<sub>org</sub>-medium sites. Annual cumulative emissions ranged from 0.91 ±  
39 0.49 kg N ha<sup>-1</sup> yr<sup>-1</sup> to 3.14 ± 0.91 kg N ha<sup>-1</sup> yr<sup>-1</sup>. Significantly different CH<sub>4</sub> fluxes  
40 between the investigated treatments or the different soil types were not observed.  
41 Cumulative annual CH<sub>4</sub> exchange rates varied between -0.21 ± 0.19 kg C ha<sup>-1</sup> yr<sup>-1</sup>  
42 and -1.06 ± 0.46 kg C ha<sup>-1</sup> yr<sup>-1</sup>. Significantly higher NH<sub>3</sub> losses, NUE<sub>min</sub> and grass  
43 yields from treatments fertilized with biogas digestate compared to those fertilized  
44 with cattle slurry were observed. The total NH<sub>3</sub> losses following splash plate  
45 application were 18.17 kg N ha<sup>-1</sup> for the digestate treatments and 3.48 kg N ha<sup>-1</sup> for  
46 the slurry treatments (36% and 15% of applied NH<sub>4</sub><sup>+</sup>-N). The observed linear  
47 increase of 16 days cumulative N<sub>2</sub>O-N exchange or rather annual N<sub>2</sub>O emissions,  
48 due to a higher mean groundwater level and a higher application rate of NH<sub>4</sub><sup>+</sup>-N,  
49 reveal the importance of site adapted N fertilization and the avoidance of N surpluses  
50 in C<sub>org</sub> rich grasslands.

## 51 **1. Introduction**

52 Germany has become the largest biogas producing country in the world, since the  
53 change in the German energy policy and the enactment of the German Renewable  
54 Energy Act (Weiland, 2010). At the end of 2012, more than 7,500 agricultural biogas  
55 plants operated in Germany (Fachverband Biogas, 2014). Heat and power from  
56 biogas substitute fossil fuels and therefore reduce greenhouse gas (GHG) emissions  
57 (Weiland, 2010; Don et al., 2011). The strong increase in the number of biogas plants  
58 caused a land-use change towards agro-biomass production and additionally raised  
59 the land-use intensity to satisfy the huge demand for fermentative substrates (Don et  
60 al., 2011). In 2011, the proportion of grass silage accounted for 9% of the total  
61 renewable resources for biogas production (DBFZ, 2012) and thus, grass silage  
62 represented the second most important fermentation substrate after maize silage.  
63 During the fermentative process high amounts of nutrient rich residues are left over.  
64 Today, this new form of organic fertilizer is used instead of mineral fertilizers or  
65 animal slurries to maintain soil fertility and productivity. Several studies reported a

66 significant increase in nitrous oxide (N<sub>2</sub>O) emissions due to the application of  
67 nitrogen fertilizers (e.g. Bouwman, 1996; Chadwick et al., 2000; Rodhe et al., 2006;  
68 Ruser, 2010). Additionally liquid organic fertilizers such as animal slurry add easily  
69 degradable organic carbon (Christensen 1983) and moisture, both favoring N<sub>2</sub>O  
70 losses through denitrification (Clayton et al., 1997). Enhanced N<sub>2</sub>O emissions are of  
71 great interest due to the fact that N<sub>2</sub>O acts as a radiative forcing greenhouse gas  
72 (IPCC, 2007) and contributes to the chemical destruction of stratospheric ozone  
73 (Crutzen, 1979). In Germany, about 67.4% of N<sub>2</sub>O emissions originate from the  
74 agricultural sector (Möller and Stinner, 2009). Particularly organic soils (e.g. drained  
75 peat soils and soils developed in wet conditions) are considered as hotspots of GHG  
76 emissions including N<sub>2</sub>O, which is due to the very high mineralization rates of  
77 degrading peat (Kasimir-Klemedtsson et al., 1997; Freibauer et al., 2004;  
78 Klemedtsson et al., 2005; Goldberg et al., 2010) and to soil moisture conditions  
79 which favor anaerobic micro-sites. According to Maljanen et al. (2010), N<sub>2</sub>O  
80 emissions from drained organic soils under agricultural use were on average four  
81 times higher than those from mineral soils. The few field studies of organic  
82 fertilization effects on annual N<sub>2</sub>O emissions from drained organic grassland soils  
83 revealed very high N<sub>2</sub>O emissions of up to 41.0 kg N ha<sup>-1</sup> yr<sup>-1</sup> (Velthof et al., 1996).  
84 In Germany, 40% of the drained peatlands are used as grasslands (Drösler et al.,  
85 2008), particularly in the smallholder structure of south Germany. Grassland soils in  
86 Europe and Germany produce more N<sub>2</sub>O per unit of fertilizer-N than croplands and  
87 emission factors further increase with soil organic carbon and nitrogen content  
88 (Freibauer and Kaltschmitt 2003; Dechow and Freibauer 2011). Moreover agricultural  
89 soils in the southern part of Germany emit, about three times more of the applied N  
90 as N<sub>2</sub>O than soils in the rest of Germany, which is attributed to the more frequent  
91 frost-thaw cycles and enhanced precipitation rates (Jungkunst et al. 2006, Dechow  
92 and Freibauer 2011). Thus, grasslands on organic soils in South Germany represent  
93 a wide-spread high-risk situation for high N<sub>2</sub>O emissions after cattle slurry or biogas  
94 digestate application, which has to our knowledge not yet been studied before.  
95 Biogas digestate is depleted in easily degradable C compounds and in organic dry  
96 matter content compared to fresh slurry due to anaerobic digestion (Möller and  
97 Stinner, 2009). In return, the pH value and the ammonium (NH<sub>4</sub><sup>+</sup>) content as well the  
98 NH<sub>4</sub><sup>+</sup>/N<sub>org</sub> ratio are higher than in fresh slurry (Wulf et al., 2002; Möller and Stinner,  
99 2009). Since digested products are more recalcitrant than fresh slurry it could be

100 assumed that microbial degradation is slow, resulting in less anoxic microsites and  
101 reduced N<sub>2</sub>O emissions than after fresh slurry application (Clemens and Huschka,  
102 2001; Oenema et al., 2005; Möller and Stinner, 2009). However, the few available  
103 field and laboratory experiments are contradictory regarding the effect of biogas  
104 digestate application on N<sub>2</sub>O emissions (e.g. Clemens and Huschka, 2001; Wulf et al.,  
105 2002; Clemens et al., 2006; Senbayram et al., 2009; Sanger et al., 2010), and very  
106 few studies exist for grasslands.

107 Slurry application also releases short-term methane (CH<sub>4</sub>) and ammonia (NH<sub>3</sub>)  
108 emissions. Methane acts as strong greenhouse gas, whereas NH<sub>3</sub> is considered as  
109 indirect greenhouse gas through ammonia deposition which could promote the  
110 formation of N<sub>2</sub>O (Moiser, 2001). Moreover, NH<sub>3</sub> deposition causes soil acidification  
111 and eutrophication of ecosystems (Dragosits et al., 2002; Sanderson et al., 2006; Ni  
112 et al., 2011). In Germany, agriculture is responsible for 95.3% of the anthropogenic  
113 NH<sub>3</sub> emissions (Haenel et al., 2010). Particularly high NH<sub>4</sub><sup>+</sup> contents and high pH  
114 values, which are typically for the biogas digestate, promote accelerated NH<sub>3</sub>  
115 volatilisation (Quakernack et al., 2011). High NH<sub>3</sub> emissions particularly occur after  
116 splash plate application on grassland (Rubæk et al., 1996; Quakernack et al., 2011),  
117 which is still common practice in the smallholder farms of South Germany.

118 The anaerobic fermentation leads to distinct differences in the composition of the  
119 remaining residues compared to untreated slurries, as mentioned before. The  
120 different properties of these fertilizers (e.g. higher NH<sub>4</sub><sup>+</sup> concentrations, narrower C/N  
121 ratio, higher pH values) directly effect N transformation processes, plant N availability  
122 and thus crop yield. Currently, the effect of anaerobic digestates on crop growth after  
123 surface application under field conditions is contradictory, since some authors  
124 reported higher crop yields compared to undigested slurries (e.g. Odlare, 2005 cited  
125 in Moller an Muller, 2012) whereas others found no effects (e.g. Moller et al., 2008).  
126 However, only a few studies exist for grassland but it seems that fertilization with  
127 biogas digestates positively affects grass yields, but only in single years (Elsasser et  
128 al., 1995; Rubæk et al., 1996; Moller et al., 2008; Moller and Muller, 2012). In general  
129 the application technique seems to have the greatest influence as this directly affects  
130 ammonia losses and thus immediately available N for plant growth (Moller and Muller,  
131 2012).

132 The objective of this study was to quantify short-term N<sub>2</sub>O, CH<sub>4</sub> and NH<sub>3</sub> emissions  
133 after application of biogas digestate and cattle slurry on grassland on two types of

134 high organic carbon soils in South Germany. Additionally it should be tested to what  
135 extent biogas digestate and cattle slurry application affect N availability and grass  
136 yield. We hypothesize: a) More N<sub>2</sub>O is emitted after biogas digestate than after slurry  
137 application because of higher NH<sub>4</sub><sup>+</sup>-N concentrations in the substrate. The more  
138 recalcitrant nature of the carbon in the biogas digestate does not matter for GHG  
139 formation in high organic carbon soils. b) N<sub>2</sub>O emissions increase with increasing soil  
140 C<sub>org</sub> content due to more favorable conditions for denitrification after organic fertilizer  
141 application. c) Biogas digestate leads to a significantly higher grass yield and N-use  
142 efficiency compared to cattle slurry due to the higher N availability of the digestate.

## 143 **2. Materials and methods**

### 144 **2.1. Study area**

145 The study was conducted on a permanent grassland at a drained fen peatland 30 km  
146 north-east of Munich (Freisinger Moos, 48°21'N, 11°41'E; 450 m a.s.l.). The  
147 dominant species were *Poa trivialis*, *Poa pratensis*, *Festuca pratensis*, *Dactylis*  
148 *glomerata* and *Alopecurus pratensis*. The grassland was mown two and three times  
149 in 2010 and 2011 respectively, as is the usual practise in this region. The grass was  
150 used as silage or hay for cattle or as substrate for biogas plants. According to the  
151 climate station in Weißenstephan, located 10 km northeast of the site, the 30-years  
152 mean annual temperature was 7.5 °C and the mean annual precipitation was 787  
153 mm (1961–1990). Annual atmospheric N deposition amounted to 6.22 and 7.20 kg N  
154 ha<sup>-1</sup> yr<sup>-1</sup>, with a NH<sub>4</sub><sup>+</sup>-N:NO<sub>3</sub><sup>-</sup>-N ratio of 46:54 and 49:51 in 2010 and 2011. Data of N  
155 deposition was collected by the Bavarian State Institute of Forestry at a German  
156 Level II monitoring area (Forest Intensive Monitoring Programme of the UNECE),  
157 located in 7 km distance to the investigated grassland. In October 2009, we selected  
158 two areas within the grassland parcel, which differed in their soil organic carbon  
159 (SOC) contents in the top soil (Table 1). According to the WRB (2006) soil types  
160 were classified as mollic Gleysol (named C<sub>org</sub>-medium) and as sapric Histosol  
161 (named C<sub>org</sub>-high) (Roßkopf personal communication).

### 162 **2.2. Experimental design**

163 At each area of the grassland parcel, three adjacent sites (site dimension 12 x 12 m)  
164 were selected. At one site biogas digestate and at another site cattle slurry was  
165 applied, whereas the third site served as control (without fertilization). Centrally at

166 each site, three PVC-collars for GHG measurements (inside dimension 75 x 75 cm)  
167 were permanently inserted 10 cm into the soil with a distance of 1.5 m to each other.  
168 To prevent oscillations of the peat through movements during the measurements,  
169 boardwalks were installed. At each area a climate station was set up in March 2010  
170 for the continuous recording (every 0.5 hour; CR200X Datalogger, Campbell Scientific)  
171 of air temperature and humidity at 20 cm above soil surface (CS215-L, Campbell  
172 Scientific), soil temperatures at the depth of -2, -5 and -10 cm (109-L, Campbell  
173 Scientific) and soil moisture content at -5 cm depth (SM200, Delta-T Devices). For  
174 NH<sub>3</sub> measurements, sensors for wind speed and wind direction (Kleinwindsensor,  
175 Thies Clima) in 2 m height were additionally integrated from May to July 2011, with a  
176 logging frequency of 5 seconds (GP1, Delta-T Devices). For measuring the ground  
177 water table, plastic perforated tubes (JK-casings DN 50, 60 mm diameter, 1 m length)  
178 were inserted close to each collar for plot-specific measurements of groundwater  
179 tables during gas flux measurements. In April 2010, we equipped one tube per site  
180 with a water level logger (Type MiniDiver, Schlumberger water services), which  
181 logged the water tables every 15 minutes. Additionally to the recorded data, site-  
182 specific soil temperatures in three soil depths (-2, -5 and -10 cm) were determined  
183 with penetration thermometers at the beginning and end of each gas flux  
184 measurement.

185 In 2010 and 2011, organic fertilizers were applied via splash plate (swivelling slurry  
186 spreader for biogas digestate; gooseneck scatterer for cattle slurry) on 14<sup>th</sup> June  
187 2010, 25<sup>th</sup> August 2010, 27<sup>th</sup> Mai 2011, 22<sup>th</sup> September 2011 and 04<sup>th</sup> November  
188 2011 by the landowners. The surface application technique via splash plate is the  
189 most common application technique in the smallholder structure of the region. The  
190 organic fertiliser was applied on the basis of equal volumetric rates per application  
191 event (between 20–25 m<sup>3</sup> ha<sup>-1</sup>). This method is typical for farming practices, but  
192 produces diverging N application rates per event between slurry and digestate based  
193 on NH<sub>4</sub><sup>+</sup> or N<sub>tot</sub> applications. It is known that the splash plate application technique  
194 can result in very uneven spreading regarding the application rate and/or the  
195 evenness. Both chosen spreading devices are known for the higher accuracy in their  
196 application evenness compared to conventional splash plates (Frick, 1999). In the  
197 present study, the application of an equal volumetric slurry rate was controlled via the  
198 barrel content and the tractor speed. At all sites, the tractor lane was 1 m in front of  
199 the collars which were placed in a row with a distance of 1.5 m to each other. Both

200 spreading systems had a spreading width of 12 m and no overlapping zones  
201 occurred. Nevertheless we can not give any estimation about the actually achieved  
202 accuracy of the application evenness.

203 The physical and chemical composition of the slurries and digestates varied between  
204 the four different application events (Table 2). Composition of organic fertilizers was  
205 analysed from 1 L samples which were taken from the slurry tank in the field. Slurries  
206 were immediately frozen at  $-20\text{ }^{\circ}\text{C}$  until analysis which was conducted by the  
207 AGROLAB Labor GmbH (Bruckberg, Germany). Due to technical problems at the  
208 first application event, cattle slurry was applied by watering cans on the plots and on  
209 a  $120\text{ m}^2$  adjacent area. To ensure an equal volumetric amount of organic fertilizer a  
210  $1\times 1\text{ m}$  grid, built by cords, was previously installed. The same method was used at  
211 the fourth application event for the digestate.

### 212 **2.3. $\text{N}_2\text{O}$ and $\text{CH}_4$ flux measurements**

213 As a background, we measured fluxes of  $\text{N}_2\text{O}$  and  $\text{CH}_4$  every second week from  
214 January 2010 to January 2012 using the static manual chamber method (volume 309  
215 L) (Livingston & Hutchinson; 1995). We removed, however, the gas fluxes measured  
216 in 2010 from the data set due to errors in the gas chromatography analysis and due  
217 to long vial storage. Intensive measurement campaigns were performed after the four  
218 fertilisation events on 14<sup>th</sup> June 2010, 25<sup>th</sup> August 2010, 27<sup>th</sup> Mai 2011, and 22<sup>th</sup>  
219 September 2011. Immediately after fertilization flux measurements were carried out  
220 daily for a week and on every second day for another eight to nine days. To minimize  
221 diurnal variation in the flux pattern, sampling was always carried out between 9.00  
222 a.m. and 11.30 a.m. A detailed description of chamber dimensions and configuration  
223 is given in Drösler (2005). Four gas samples were taken at four regular time intervals  
224 after chamber closure (enclosure time 60 min). The samples were collected in 20 ml  
225 glass vials, each sealed with a butyl rubber septum. The vials were flushed with  
226 chamber air for 30 seconds using a portable micro pump (KNF Neuberger GmbH,  
227 NMP015B), so that the air in the vials was exchanged 32 times. In addition the pump  
228 was used to build up an overpressure of approximately 550 mbar to protect the  
229 sample against fluctuations in atmospheric pressure during storage. Gas analyses  
230 were carried out with a gas chromatograph (Perkin & Elmer, Clarus 400 GC  
231 respectively Clarus 480 GC) equipped with a headspace auto sampler (Perkin &  
232 Elmer, TurboMatrix 110), a PoraPack 80/100 mesh column, an electron capture  
233 detector (ECD) for  $\text{N}_2\text{O}$  (ECD temperature  $380^{\circ}\text{C}$ ) and a flame ionization detector

234 (FID) for CH<sub>4</sub> analyses (FID temperature 310°C). Gas samples from the first  
235 fertilization event (14<sup>th</sup> June to 30<sup>th</sup> June of 2010) were immediately analysed at the  
236 Max Planck Institute for Biogeochemistry in Jena, whereas samples from the second  
237 fertilization event (25<sup>th</sup> August to 10<sup>th</sup> September of 2010) were analysed at the  
238 Thünen Institute in Braunschweig with a Varian CP-3800 GC-FID/-ECD using a  
239 headspace autosampler (QUMA Elektronik & Analytik GmbH, Germany) and similar  
240 conditions. Gas flux rates were calculated from the linear change in gas  
241 concentration over time considering chamber air temperature and atmospheric  
242 pressure. Gas fluxes were accepted when the linear regression was significant ( $P \leq$   
243 0.05). In case of small N<sub>2</sub>O or CH<sub>4</sub> fluxes, fluxes were also accepted if the coefficient  
244 of determination was  $\geq 0.90$  and the regression slope was between  $-1$  and  $1$  ppb  
245 min<sup>-1</sup>. The cumulative annual mean exchange rate was calculated by linear  
246 interpolation between the measurement dates.

#### 247 **2.4. NH<sub>3</sub> flux measurements**

248 Ammonia volatilization was measured at the third organic fertilizer application event  
249 on 27<sup>th</sup> of May 2011. Measurements were performed immediately after fertilizer  
250 application and thereafter in irregular time intervals of few hours (in total 96  
251 measurements). For NH<sub>3</sub> measurements we used the calibrated dynamic chamber  
252 method ('Dräger-Tube Method'; DTM) which was described in detail bei Pacholski et  
253 al. (2006). One day before application, eight stainless steel rings (104 cm<sup>2</sup>) were  
254 inserted into the upper soil (3 cm) at each treatment, from which four were grouped  
255 close together. Ambient air was sucked with a defined flow rate (1 L min<sup>-1</sup>) through  
256 four (via teflon tubes) connected conical stainless steel chambers to an ammonia  
257 indicator tube (Drägerwerk AG, Lübeck, Germany). The NH<sub>3</sub> volume concentration  
258 was corrected for air temperature and air pressure (Pacholski et al., 2006). To  
259 prevent overestimation of NH<sub>3</sub> volatilization through NH<sub>3</sub> enriched ambient air from  
260 surrounding area, ammonia concentration from the control treatments were  
261 subtracted from the fertilized treatments prior to NH<sub>3</sub> flux calculation. Different studies  
262 report a distinct underestimation of up to one order of magnitude of NH<sub>3</sub> fluxes  
263 determined by the DTM, mainly due to the low air exchange rate in the chambers  
264 (Roelcke, 2002; Pacholski et al., 2006). To avoid underestimation of cumulative NH<sub>3</sub>-  
265 N losses determined by the DTM, Pacholski et al. (2006) developed the following  
266 calibration formula to correct the NH<sub>3</sub> fluxes:

267



$$\ln(NH_3 flux_{IHF}) = 0.444 * \ln(NH_3 flux_{DTM}) + 0.590 * \ln(v_{2m}) \quad (1)$$

269

270 where  $NH_3 flux_{IHF}$  is  $NH_3$  flux measured by the integrated horizontal flux method (kg N  
 271  $ha^{-1} h^{-1}$ );  $NH_3 flux_{DTM}$  is  $NH_3$  flux measured by the DTM (kg N  $ha^{-1} h^{-1}$ );  $v_{2m}$  wind  
 272 speed at 2 m height ( $m s^{-1}$ ). Quakernack et al. (2011) compared the DTM method  
 273 with the frequently used micrometeorological method, concluding that the corrected  
 274 DTM method also allows quantitative  $NH_3$ -loss measurements. The total cumulative  
 275  $NH_3$  volatilization was estimated by curve fitting and integration of the area obtained  
 276 by the fitted curve between time zero and the time point where the  $NH_3$  flux was zero.

## 277 2.5. Grass yield, apparent N use efficiency and N-balances

278 The annual yield was determined by harvesting the grass inside the PVC-collars with  
 279 a scissor at each mowing event (same cutting height as the farmer, at about 5 cm).  
 280 Mowing events took place on 24<sup>th</sup> Mai 2010, 20<sup>th</sup> August 2010, 23<sup>th</sup> Mai 2011, 01<sup>st</sup>  
 281 August 2011 and 13<sup>th</sup> September 2011. To determine the dry mass (DM), grass  
 282 samples were oven dried at 60°C for 48 hours. To determine the total carbon ( $C_{tot}$ )  
 283 and total nitrogen ( $N_{tot}$ ) concentrations of plant biomass, dried grass samples were  
 284 milled (0.5 mm) and mixed sub samples were analysed according to DIN ISO 10694  
 285 and DIN ISO 13878 by the AGROLAB Labor GmbH (Bruckberg, Germany). The  
 286 apparent  $N_{tot}$  or rather  $N_{min}$  use efficiency (NUE,  $NUE_{min}$ ) was calculated as:

287

$$NUE \text{ or } NUE_{min} = \left( \frac{N \text{ uptake}_{treatment} - N \text{ uptake}_{control}}{total N \text{ applied}} \right) * 100\% \quad (2)$$

289

290 where  $N \text{ uptake}_{treatment}$  is the amount of N taken up by the plants in the fertilized  
 291 treatments,  $N \text{ uptake}_{control}$  is the amount of N taken up by the plants in the unfertilized  
 292 control, and  $total N \text{ applied}$  is the amount of  $N_{tot}$  or  $N_{min}$  applied, corrected by  $NH_3$ -N  
 293 losses (23% and 5% of  $N_{tot}$ , or 36% and 15% of  $N_{min}$  for biogas digestate and cattle  
 294 slurry, respectively).

295 Based on the measured gaseous N fluxes, the N uptake by plants and soil  $N_{min}$   
 296 contents a simple N balance was calculated as followed:

297

$$N \text{ balance} = (N \text{ applied} + (N_{min_{t2}} - N_{min_{t1}}) + N_{dep}) - (N \text{ uptake} + N_2 O_{cum} + NH_3_{cum}) \quad (3)$$

299

300 where  $N_{applied}$  is the amount of  $N_{tot}$  applied,  $N_{min,t1}$  and  $N_{min,t2}$  are the amounts of  
301  $N_{min}$  at time 1 (06<sup>th</sup> April 2011; early April represents the beginning of the vegetation  
302 period in 2011) and time 2 (18<sup>th</sup> October 2011; end of October represents the end of  
303 the vegetation period in 2011) for the soil depth 0–20 cm,  $N_{dep}$  is the annual  
304 atmospheric N deposition,  $N_{uptake}$  is the amount of N taken up by the plants  
305 (quantified in harvested biomass),  $N_2O_{cum}$  is the amount of the annual cumulative  
306  $N_2O$ -N losses, and  $NH_3_{cum}$  is the amount of the annual cumulative  $NH_3$ -N losses.

## 307 **2.6. Soil sampling and laboratory analyses**

308 For the determination of mineral N ( $N_{min} = NH_4^+-N + NO_3^--N$ ) contents, one mixed  
309 soil sample consisting of nine individual samples was collected at two soil depths (0–  
310 10, 10–20 cm) at each treatment during every gas flux measurement. Samples were  
311 immediately cooled and stored in an ice box before analyses. Mineral N was  
312 extracted after shaking 40 g of fresh soil with 160 ml  $CaCl_2$  (0.0125 M) for one hour.  
313 The extracts were filtered through a 4–7  $\mu m$  filter paper (Whatman 595  $\frac{1}{2}$ ) and the  
314 first 20 ml of the extract were discarded. The solution was frozen at  $-20\text{ }^\circ C$  until  
315 analysis, which was conducted by the AGROLAB Labor GmbH (Bruckberg,  
316 Germany). A subsample of 20–30 g was used to determine the gravimetric water  
317 content, which was taken into account for the calculation of mineral N concentrations.  
318 For determination of  $C_{tot}$  and organic carbon ( $C_{org}$ ) a mixed soil sample of nine  
319 individual samples was collected close to each collar at two soil depths (0–10, 10–20  
320 cm) using a 3 cm diameter auger. After drying for 72 hours at  $40\text{ }^\circ C$ , soil samples  
321 were sieved to 2 mm to remove stones and living roots. Analyses were conducted at  
322 the Division of Soil Science and Site Science (Humboldt Universität zu Berlin,  
323 Germany). For the determination of bulk density and porosity, three undisturbed core  
324 cutter samples ( $100\text{ cm}^3$ ) were randomly taken at four depths (0–5, 5–10, 10–15, 15–  
325 20 cm) for each treatment.

## 326 **2.7. Statistical analysis**

327 Statistical analyses were conducted using R 2.12.1 (R Development Core Team,  
328 2010). We used analysis of variance (ANOVA) (for grass yield, 16 days cumulative  
329  $N_2O$  emissions and treatment  $NO_3^-$  comparison) or the nonparametric Kruskal-Wallis  
330 Rank Sum test (for GW level) to compare means of samples. In case of significant  
331 differences among the means, we used Tukey's honest significant differences  
332 (TukeyHSD) or the non-parametric Pairwise Wilcoxon Rank Sum test with Bonferroni

333 correction for multiple comparisons. For testing two independent sample means, we  
334 use the Welch two sample t-test (for soil type  $\text{NO}_3^-$  comparison in 2010) or the non  
335 parametric Mann-Whitney U-test (for soil type  $\text{NO}_3^-$  comparison in 2011). For time  
336 series data ( $\text{N}_2\text{O}$ ,  $\text{CH}_4$  field measurements) we applied linear mixed effects models  
337 (Crawley 2007; Eickenscheidt et al., 2011; Hahn-Schöfl et al., 2011). We set up a  
338 basic model with soil type and fertilizer treatment as fixed effects and the spatial  
339 replication (individual plot) nested in time as random effect. Non-significant terms  
340 were removed from the fixed structure. We extended the basic model by a variance  
341 function when heteroscedasticity was observed. In case of significant serial  
342 correlation in data, a moving average or a first-order temporal autoregressive  
343 function was included in the model. Autocorrelation was tested using the Durbin-  
344 Watson test and by plotting the empirical autocorrelation structure (Eickenscheidt et  
345 al. 2011). The model extension was proved by the Akaike Information Criterion (AIC).  
346 For multiple comparisons we conducted Tukey contrasts using the General Linear  
347 Hypotheses function from the “multcomp” package (Hothorn et al., 2013).  
348 The assumption of normality of residuals was tested using the Lilliefors or Shapiro-  
349 Wilk test and by plotting the Quantile-Quantile plots. Homogeneity of variances of  
350 residuals was checked using the Levene or Breusch-Pagan test and by plotting the  
351 residuals against the fitted values. Where necessary, data were box-cox transformed  
352 prior to analyses. We used simple and multiple linear or non-linear regressions  
353 models to explain  $\text{N}_2\text{O}$ ,  $\text{CH}_4$  and  $\text{NH}_3$  fluxes. We accepted significant differences if  $P$   
354  $\leq 0.05$ . Results in the text are given as means  $\pm 1$  standard deviation.

### 355 **3. Results**

#### 356 **3.1. Environmental drivers**

357 Temperatures between the two investigated soil types did not differ. In 2010 and  
358 2011, air temperature in 20 cm height ranged from  $-17.5$  to  $39.5^\circ\text{C}$  with an annual  
359 mean of  $8.6^\circ\text{C}$  in 2011 at both investigated areas. Soil temperature in  $-2$  cm soil  
360 depth averaged  $10.3^\circ\text{C}$  at the  $\text{C}_{\text{org}}$ -medium sites and was slightly higher with  $10.5^\circ\text{C}$   
361 at the  $\text{C}_{\text{org}}$ -high sites in 2011. Air temperature in 20 cm height following 15 or 16 days  
362 after fertilization averaged  $16.0$ ,  $13.1$ ,  $15.4$  and  $11.5^\circ\text{C}$  for application events one to  
363 four at both investigated soil types. Soil temperature in  $-2$  cm soil depth was  
364 approximately  $2^\circ\text{C}$  above the mean air temperature in the same periods at both soil  
365 types. In 2010 and 2011 annual precipitation was 850 and 841 mm, which was

366 slightly above the 30-years mean of the period 1961–1990. Figure 1 shows the  
367 precipitation following the fertilizer application. With the exception of the third  
368 application event, no rainfall occurred during the application of the organic fertilizers.  
369 However, precipitation during and after the third application event was only weak and  
370 amounted to 3 mm in the time span between 16:00 and 00:00 hours.  
371 All treatments showed similar dynamics in their annual hydrographs (Fig. 2a) but  
372 mean annual groundwater levels of the C<sub>org</sub>-high treatments were significantly higher  
373 (all  $P < 0.001$ ) compared to the C<sub>org</sub>-medium treatments in 2010 and 2011 (Table 3).  
374 Mean groundwater levels following the fertilizer applications are shown in Table 3.

### 375 **3.2. N input and N availability**

376 The amount of N applied was 111 and 252 kg N ha<sup>-1</sup> for slurry treatments or rather  
377 101 and 174 kg N ha<sup>-1</sup> for digestate treatments in 2010 and 2011, respectively.  
378 However, due to the distinctly higher NH<sub>4</sub><sup>+</sup>-N/N<sub>tot</sub> ratio of the biogas digestate, total  
379 NH<sub>4</sub><sup>+</sup>-N input was comparable or slightly higher in 2010 and 2011 than at the slurry  
380 treatments (Table 2). Additional physical and chemical properties of the slurry and  
381 digestate are shown in Table 2.

382 The extractable N<sub>min</sub> contents of the soils were dominated by NO<sub>3</sub><sup>-</sup> whereas NH<sub>4</sub><sup>+</sup>  
383 was only of minor importance especially at the C<sub>org</sub>-medium sites (Fig. 2b and 2c).  
384 The NO<sub>3</sub><sup>-</sup> content was significantly higher ( $P < 0.001$ ) at the C<sub>org</sub>-high sites than at  
385 the C<sub>org</sub>-medium sites in 0–10 cm soil depth in both years and in 10–20 cm soil depth  
386 in 2010 ( $P < 0.01$ ) (Table 3). With exception of the first application event, all  
387 fertilization events increased the NO<sub>3</sub><sup>-</sup> contents of the soil for a short period (Fig. 2c,  
388 Table 3). However, only in 2011 the fertilized sites showed significantly ( $P < 0.01$ )  
389 higher NO<sub>3</sub><sup>-</sup> contents compared to the control treatments, but differences between  
390 digestate and slurry were generally not significant (except of 0–10 cm soil depth at  
391 the C<sub>org</sub>-medium site) (Table 3).

### 392 **3.3. N<sub>2</sub>O emissions**

393 Nitrous oxide fluxes were generally low at all treatments (Fig. 2d). Background  
394 emissions rarely exceeded 50 µg N m<sup>-2</sup> h<sup>-1</sup>. Highest N<sub>2</sub>O fluxes were found  
395 immediately after fertilizer application (Fig. 2d and 3), sometimes followed by a  
396 second phase of higher emissions after 6 to 12 days. In case of the C<sub>org</sub>-medium  
397 sites N<sub>2</sub>O fluxes returned to background emission level within 3 to 7 days, whereas

398 the C<sub>org</sub>-high sites had longer lasting increased N<sub>2</sub>O emissions, particularly at the  
399 digestate treatment.

400 Short term (16 days) N<sub>2</sub>O fluxes of fertilized treatments significantly ( $P < 0.01$ )  
401 exceeded N<sub>2</sub>O fluxes of control treatments at all fertilization events. However, only in  
402 one out of four fertilization events short term N<sub>2</sub>O fluxes were significantly ( $P < 0.001$ )  
403 higher at the digestate treatments compared to the slurry treatments. Additionally  
404 significantly ( $P < 0.001$ ) higher short term N<sub>2</sub>O fluxes were observed at the C<sub>org</sub>-high  
405 sites compared to the C<sub>org</sub>-medium sites in 2011, but the opposite was observed at  
406 the second fertilization event in 2010.

407 However, due to the high variability and the partially fast return to the background  
408 emission level, short term (16 days) cumulative N<sub>2</sub>O emissions were not significantly  
409 different from the control treatments in 2010 (Fig 4), but for 2011 short term  
410 cumulative N<sub>2</sub>O emissions had a clear trend in the order digestate > slurry > control  
411 (although not significant in one case).

412 On an annual basis organic fertilization led to significantly ( $P < 0.001$ ) higher N<sub>2</sub>O  
413 fluxes compared to unfertilized treatments. Additionally, the application of biogas  
414 digestate significantly ( $P < 0.01$ ) enhanced the N<sub>2</sub>O fluxes compared to the  
415 application of cattle slurry. Furthermore, N<sub>2</sub>O fluxes from the C<sub>org</sub>-high site  
416 significantly ( $P < 0.001$ ) exceeded N<sub>2</sub>O fluxes from the C<sub>org</sub>-medium sites. Annual  
417 cumulative emissions ranged from  $0.91 \pm 0.49$  kg N ha<sup>-1</sup> yr<sup>-1</sup> (control treatment, C<sub>org</sub>-  
418 medium site) to  $3.14 \pm 0.91$  kg N ha<sup>-1</sup> yr<sup>-1</sup> (digestate treatment, C<sub>org</sub>-high site) (Table  
419 4). Calculated emission factors (EF) based on the amount of N<sub>tot</sub> ranged from 0.12 to  
420 0.23 for the slurry treatments and from 0.55 to 1.13 for the digestate treatments  
421 (Table 4).

422 Observed N<sub>2</sub>O fluxes could not be explained by any of the measured environmental  
423 drivers. However, 53% of the temporal and spatial variation in the 16 days cumulative  
424 N<sub>2</sub>O-N exchange rates was explained by the amounts of applied NH<sub>4</sub><sup>+</sup>-N and the  
425 mean groundwater levels below surface during the same time (Fig. 5). A similar trend  
426 was observed for the annual cumulative N<sub>2</sub>O emissions but regression analysis was  
427 not possible due to the small sample size ( $n = 6$ ).

### 428 **3.4. CH<sub>4</sub> emissions**

429 Most of the time, CH<sub>4</sub> emissions could not be detected (Fig. 2e). Occasionally CH<sub>4</sub>  
430 peaks were only found immediately after cattle slurry application. However, with  
431 exception of the slurry treatment of the C<sub>org</sub>-high site at the first application event, the

432 organic fertilization did not result in significantly different short term (15 or 16 days)  
433 CH<sub>4</sub> fluxes between the treatments or the investigated soil types. The observed weak  
434 CH<sub>4</sub> emissions or uptakes amounted to cumulative annual CH<sub>4</sub> exchange rates of  
435  $-0.21 \pm 0.19 \text{ kg C ha}^{-1} \text{ yr}^{-1}$  to  $-1.06 \pm 0.46 \text{ kg C ha}^{-1} \text{ yr}^{-1}$ . Significantly different CH<sub>4</sub>  
436 fluxes between the investigated treatments or the different soil types could not be  
437 observed regarding the annual fluxes in 2011.

### 438 **3.5. NH<sub>3</sub> volatilisation**

439 Highest NH<sub>3</sub> losses were observed immediately after fertilization (Fig. 6). During the  
440 first 24 hours, 64% and 100% of total NH<sub>3</sub> losses occurred at the digestate and slurry  
441 treatments, respectively. Since differences in the response of NH<sub>3</sub> volatilization were  
442 not significant, treatment data were pooled by soil type prior to regression analysis.  
443 The total NH<sub>3</sub> loss following application was  $18.17 \text{ kg N ha}^{-1}$  for the digestate  
444 treatments and  $3.48 \text{ kg N ha}^{-1}$  for the slurry treatments. The relative N loss was 36%  
445 and 15% of applied NH<sub>4</sub><sup>+</sup>-N, or 23% and 5% of total applied N for the digestate and  
446 slurry treatments, respectively.

### 447 **3.6. Grass yield, apparent N use efficiency and estimated N balances**

448 In 2010 and 2011, the mean annual grass yield ranged from 4.5 (control C<sub>org</sub>-medium)  
449 to  $13.1 \text{ t DM ha}^{-1} \text{ yr}^{-1}$  (digestate C<sub>org</sub>-high) (Table 5). In both years the mean annual  
450 grass yield from the digestate treatments were significantly ( $P < 0.05$ ) higher  
451 compared to the slurry treatments. Additionally, the mean annual grass yield from the  
452 C<sub>org</sub>-high sites exceeded those from the C<sub>org</sub>-medium sites of both years, but  
453 differences were not significant.

454 The application of biogas digestate distinctively increased apparent NUE and NUE<sub>min</sub>  
455 compared to cattle slurry treatments (Table 5). NUE values were on average  $111 \pm$   
456  $133\%$  for biogas digestate treatments and  $21 \pm 18\%$  for cattle slurry. NUE<sub>min</sub> values  
457 were always  $>100\%$  for biogas digestate treatments, whereas for cattle slurry NUE<sub>min</sub>  
458 values averaged  $54 \pm 53\%$ . Beside fertilizer type effects, higher NUE and NUE<sub>min</sub>  
459 were observed at the C<sub>org</sub>-medium site compared to the C<sub>org</sub>-high site.

460 The estimated N balances revealed N surpluses of up to  $79 \text{ kg N ha}^{-1} \text{ yr}^{-1}$  for cattle  
461 slurry treatments but deficits of up to  $95 \text{ kg N ha}^{-1} \text{ yr}^{-1}$  for biogas digestate  
462 treatments, for the year 2011 (Table 6).

## 463 4. Discussion

### 464 4.1. Drainage and fertilizer effects on N-availability and N- 465 transformation

466 Mineral nitrogen contents were consistently higher at the C<sub>org</sub>-high treatments than at  
467 the C<sub>org</sub>-medium treatments, in line with the considerably higher amount of soil  
468 organic matter (SOM) at this site. It is well known that drainage enhances the  
469 degradation of SOM and thus stimulates net nitrogen mineralization and N  
470 transformation processes (Kasimir Klemetsson et al., 1997; Freibauer et al., 2004;  
471 Klemetsson et al., 2005; Goldberg et al., 2010). Various studies reported an annual  
472 N supply through peat mineralization of 70 to 292 kg N ha<sup>-1</sup> yr<sup>-1</sup> (Schothorst, 1977;  
473 Flessa et al., 1998; Sonneveld and Lantinga, 2011). It can be assumed that at a  
474 comparable aeration status and temperature, mineralization processes are more  
475 intensive at peatlands which were recently drained (Hacin et al., 2001; Renger et al.,  
476 2002; Sonneveld and Lantinga, 2011) or contain higher amounts of SOM.

477 As expected from literature the biogas digestates differed in their physical and  
478 chemical properties from the cattle slurries. The biogas digestates had narrower C/N  
479 ratios (e.g. Tambone et al., 2009), higher pH values (Wulf et al., 2002; Quakernack et  
480 al., 2011), wider NH<sub>4</sub><sup>+</sup>/N<sub>tot</sub> ratios and thus relatively higher NH<sub>4</sub><sup>+</sup> contents than the  
481 cattle slurries (Möller and Stinner 2009). However, the amounts of NH<sub>4</sub><sup>+</sup> were not  
482 distinctly different between the applied organic fertilizers (with one exception).

483 We observed an unexpected small change in the NH<sub>4</sub><sup>+</sup> content of the soil  
484 immediately after fertilizer application which could be attributed to different reasons.  
485 Firstly, the fertilizers partly remained on the plant canopy after splash plate  
486 application and therefore soil contact and infiltration was limited (Quakernack et al.,  
487 2011). Secondly, a significant fraction of NH<sub>4</sub><sup>+</sup> from the organic fertilizer was lost in a  
488 few hours after splash plate application via NH<sub>3</sub> volatilization. But most importantly, in  
489 well aerated soils applied NH<sub>4</sub><sup>+</sup> undergoes rapid nitrification, as indicated by the  
490 increasing soil NO<sub>3</sub><sup>-</sup> contents after fertilizer application in the upper soil layer. In  
491 general, the continuously observed absent or low NH<sub>4</sub><sup>+</sup> contents with simultaneously  
492 high extractable NO<sub>3</sub><sup>-</sup> in the soil indicate that net nitrification entirely controls net  
493 nitrogen mineralization at all treatments of the investigated study sites. Nitrification  
494 requires sufficient oxygen (O<sub>2</sub>) availability in the soil (Davidson et al., 1986) hence we

495 can assume well aerated soil conditions, at least in the upper soil layer, for most of  
496 the time at the study sites.

497 Several studies (e.g. Gutser et al., 2005; Jones et al., 2007) reported that the  
498 infiltration of organic fertilizer may enhance the soil N pool and further stimulates the  
499 SOM mineralization, leading to additional  $N_{\min}$ . This becomes also evident in the  
500 observed significantly higher  $\text{NO}_3^-$  contents of the fertilized treatments compared to  
501 the unfertilized control treatments, especially in the 0–10 cm soil layer. However,  
502 significant differences in the  $N_{\min}$  contents between the two investigated organic  
503 fertilizers were not found in 2010 and 2011. This may be due to the fact that the N  
504 uptake by plants at the digestate treatments was on average 27% higher and that  
505 distinct differences in the amount of  $N_{\text{tot}}$  and  $\text{NH}_4^+$  of the applied organic fertilizers  
506 were only observed in the second study year.

507 To maintain soil fertility and yield and to reduce harmful side effects (e.g.  $\text{N}_2\text{O}$  losses,  
508  $\text{NO}_3^-$  leaching) site adapted fertilization is necessary. The estimated negative N  
509 balances for biogas treatments are in line with Andres et al. (2013) who reported that  
510 positive N balances could only be achieved when the amount of applied digestate  
511 contains more than  $200 \text{ kg N ha}^{-1} \text{ yr}^{-1}$ . However, the strong negative N balances of  
512 the control treatments reveal that large amounts of up to  $148 \text{ kg N ha}^{-1} \text{ yr}^{-1}$  originate  
513 from peat mineralization, demonstrating the unsustainable agricultural use of drained  
514 peatlands. Assuming that the fertilized treatments received equal amounts of N from  
515 peat mineralization, all N balances of these treatments were strongly positive. N  
516 surpluses as estimated for the cattle slurry treatments enhance the soil N pool, but  
517 the gradual release of N at a non predictable stage from the soil N pool carries the  
518 risk of leaching or gaseous losses (Amon et al., 2006). Particularly in wintertime, high  
519 amounts of available  $\text{NO}_3^-$  in the soil, as observed especially at the fertilized  
520 treatments of the  $C_{\text{org}}$ -high sites, carry the risk of N leaching due to the reduced N  
521 demand by plant uptake and by the microbial community during this time (Merino et  
522 al., 2002; Sanger et al., 2010).

#### 523 **4.2. Fertilizer effect on N-use efficiency and grass yield**

524 In line with investigations from Schils et al. (2008) most of the applied and produced  
525  $N_{\min}$  was probably rapidly absorbed by the grassland as the soil  $N_{\min}$  content usually  
526 decreased within a few days after fertilizer application (Figure 2b, 2c). This becomes  
527 also evident in the apparent  $\text{NUE}_{\min}$ , especially from biogas digestate treatments. A  
528 significant effect of biogas digestate on crop yields and apparent  $\text{NUE}_{\min}$  as observed



529 in the present study were also reported from pot experiments (e.g. de Boer, 2008;  
530 Möller and Müller, 2012), but not for field applications without incorporation of the  
531 digestate into the soil (Möller and Müller, 2012). According to de Boer (2008) the  
532 higher  $NUE_{min}$  at digestate treatments can be attributed to the wider  $NH_4^+/N_{tot}$  ratio as  
533 well as to the narrower C/N ratio of the applied digestate. Thus more N was  
534 immediately available for plant growth after fertilization (Amon et al., 2006; Sanger et  
535 al., 2010), whereas the lower C/N ratio reduced the potential for immobilization of  
536 applied N (Velthof et al., 2003, de Boer, 2008). We hypothesized that the application  
537 of biogas digestate leads to a significantly higher grass yield and N-use efficiency  
538 compared to the application of cattle slurry due to the higher N availability of the  
539 digestate. This could partly be confirmed, but the much higher grass yields from  
540 biogas digestate treatments cannot solely be explained by differences in applied  
541  $NH_4^+$ , since differences were only small, in particular when accounting for  $NH_3$  losses.  
542 Many studies have shown that the utilization of N derived from organic fertilizer is  
543 relatively small in the year of application, due to the slow release of organically bound  
544 N (Jensen et al., 2000; Sørensen and Amato, 2002; Gutser et al., 2005). The  
545 consistently higher  $NUE_{min}$  of > 100% at the digestate treatments indicates that some  
546 organic N derived from the fertilizer or from the SOM pool has been mineralized  
547 (Gunnarsson et al., 2010). Since the digestate is considered as more recalcitrant  
548 (Clemens and Huschka, 2001; Oenema et al., 2005; Möller and Stinner, 2009), it can  
549 be assumed that the digestate enhanced SOM mineralization more than cattle slurry,  
550 or that N mineralized from SOM had a larger share in the uptake by the plants due to  
551 lower competition of microbial immobilization. The lower  $NUE$  at the  $C_{org}$ -high sites  
552 compared to  $C_{org}$ -medium sites reveals that plants are more independent of N input  
553 by fertilizer with increasing SOM at drained fen peatlands due to the extra  $N_{min}$   
554 derived from enhanced mineralization processes, as mentioned before. However, in  
555 the present study the observed yield differences between the treatments fertilized  
556 with biogas digestate and cattle slurry cannot fully be explained on the basis of  
557 available N in the applied fertilizers and further investigations are necessary.

#### 558 **4.3. Fertilizer and site induced $N_2O$ emissions**

559 The observed annual  $N_2O$  emissions were distinctly lower than the actual default  
560 emission factor from the Tier 1 approach for temperate, deep drained, nutrient rich  
561 grassland of  $8.2 \text{ kg } N_2O\text{-N ha}^{-1} \text{ yr}^{-1}$  (IPCC, 2014) and at the lower end of literature  
562 values from other organic soils. Studies from Germany reported much higher  $N_2O$

563 emissions, ranging from 1.15 to 19.8 kg N ha<sup>-1</sup> yr<sup>-1</sup> (Augustin et al., 1998; Flessa et  
564 al., 1997; Flessa et al., 1998; Beetz et al., 2013). Also investigations from other  
565 European countries showed that much higher N<sub>2</sub>O emissions can be released from  
566 grasslands on drained peatlands. For example, Velthof et al. (1996) and van Beek et  
567 al. (2010; 2011) reported N<sub>2</sub>O emissions, ranging from 4.2 to 41.0 kg N ha<sup>-1</sup> yr<sup>-1</sup> for  
568 the Netherlands, whereas at boreal regions N<sub>2</sub>O emissions of up to 9 kg N ha<sup>-1</sup> yr<sup>-1</sup>  
569 were measured (Nykänen et al., 1995; Maljanen et al., 2004; Regina et al., 2004).  
570 The observed N<sub>2</sub>O emissions were also in the range of those reported from  
571 grasslands on mineral soils in Germany, summarized by Jungkunst et al. (2006). In  
572 line with our results, Flessa et al. (1998) also found that N<sub>2</sub>O losses from peat soils  
573 are not always larger than from nearby mineral soils, but in contrast, Maljanen et al.  
574 (2010) found on average four times higher N<sub>2</sub>O emissions from cultivated organic  
575 soils than from mineral soils. The N<sub>2</sub>O emissions from the C<sub>org</sub>-high sites significantly  
576 exceeded those from the C<sub>org</sub>-medium sites in all treatments, which was in line with  
577 higher N<sub>min</sub> contents and higher groundwater levels. This probably could be attributed  
578 to the more favorable soil conditions for denitrification, due to higher C and N  
579 mineralization rates and alternating groundwater levels, promoting anaerobicity  
580 (Koops et al., 1996). Moreover, as mentioned before, net nitrification entirely controls  
581 net nitrogen mineralization, promoting also N<sub>2</sub>O losses, but probably to a lesser  
582 extent. However, the source of N<sub>2</sub>O production in soils is often uncertain because  
583 aerobic and anaerobic micro sites can occur within close proximity and thus  
584 nitrification and denitrification as well other abiotic processes producing N<sub>2</sub>O (e.g.  
585 nitrifier-denitrification, coupled nitrification-denitrification) can run simultaneously  
586 (Davidson et al., 1986; Butterbach-Bahl et al., 2013). Despite surprisingly low N<sub>2</sub>O  
587 emission levels, we confirmed our hypothesis that N<sub>2</sub>O emissions increase with  
588 increasing soil C<sub>org</sub> content probably due to more favorable conditions for  
589 denitrification.

590 The observed background emissions on the two organic soils correspond well to  
591 those on mineral agricultural soils (Bouwman, 1996). However, calculated emission  
592 factors as percentage of applied N without consideration of the NH<sub>3</sub> losses were  
593 lower for all treatments than the IPCC default value. Several other studies reported  
594 also emission factors < 1% of applied N (Chadwick et al., 2000; Velthof et al., 2003;  
595 Clemens et al., 2006; Jones et al., 2007; Möller and Stinner, 2009), but never for  
596 organic soils. Indeed, N<sub>2</sub>O studies on organic soils rarely differentiate between

597 fertilizer and soil derived N sources by unfertilized control plots as we do in this study.  
598 In line with Möller and Stinner (2009) the application of biogas digestate resulted in a  
599 distinctly higher percentage of produced N<sub>2</sub>O from applied N, compared to cattle  
600 slurry, yet at a low level.

601 One reason of generally low N<sub>2</sub>O emissions observed in the present study could be  
602 the small number of frost–thaw cycles in 2011. In general frost–thaw cycles are  
603 considered to favor high N<sub>2</sub>O emissions (Flessa et al., 1998, Jungkunst et al., 2006)  
604 but these observations seem to be more pronounced for croplands than for  
605 grasslands in Germany (Dechow and Freibauer, 2011). Denitrification activity is  
606 strongly related to the NO<sub>3</sub><sup>-</sup> content close to the groundwater level (van Beek et al.,  
607 2004). Given the high NO<sub>3</sub><sup>-</sup> contents, in particular in the C<sub>org</sub>-high soil, the evidence  
608 for fast nitrification and high net nitrogen mineralization, we argue that frequent but  
609 low dosage application of fertilizer and quick N uptake by plants avoid conditions  
610 favorable for high N<sub>2</sub>O emissions. Moreover through the splash plate application  
611 technique high amounts of NH<sub>4</sub><sup>+</sup> were rapidly lost as NH<sub>3</sub>, and therefore reduced the  
612 proportion of immediately available N for nitrification and denitrification.

613 As expected from the literature, highest N<sub>2</sub>O fluxes were found immediately after  
614 fertilizer application. The initial N<sub>2</sub>O peak could mainly be attributed to the  
615 denitrification of available soil NO<sub>3</sub><sup>-</sup>, presumably due to the more favorable conditions  
616 for denitrification through the addition of easily degradable organic C and water  
617 (Comfort et al., 1990; Chadwick et al., 2000; Velthof et al., 2003). Additionally, a  
618 probably smaller part of initial N<sub>2</sub>O could be ascribed to the rapid nitrification  
619 (Chadwick et al., 2000) or to nitrifier denitrification of slurry NH<sub>4</sub><sup>+</sup>. In contrast, the  
620 partially observed second N<sub>2</sub>O peak, mostly found a week after fertilizer application,  
621 can be attributed to the denitrification of mineralized and nitrified organic components  
622 of fertilizer N (Velthof et al., 2003).

623 Several authors proposed that the more recalcitrant digestate might reduce the rate  
624 of microbial degradation and oxygen consumption in the soil, thus resulting in  
625 reduced N<sub>2</sub>O emissions through less anaerobic soil conditions (Clemens and  
626 Huschka, 2001; Oenema et al., 2005; Möller and Stinner, 2009). In contrast, our  
627 study on organic soils found significantly higher N<sub>2</sub>O emissions from the digestate  
628 treatments compared to the slurry treatments. Higher N<sub>2</sub>O emissions derived from  
629 biogas digestates were also reported from a few other authors (e.g. Senbayram et al.,

630 2009; Sanger et al., 2010), whereas Clemens et al. (2006) found no differences  
631 between untreated and digested slurry.

632 It can be assumed that at drained organic soils, like in the present study, sufficient  
633 metabolizable C is generally widely available in the upper soil profile (e.g. van Beek  
634 al., 2004). Thus, as hypothesized, labile carbon is not limiting on organic soils. This  
635 was in line with Velthof et al. (2003) who supposed that the application of available C  
636 with the organic fertilizer has a larger effect on denitrification activity at soils with a  
637 lower  $C_{org}$  content compared to  $C_{org}$  rich soils. However, contrary to our hypothesis  
638 the significantly higher  $N_2O$  emissions from the digestate treatments can not solely  
639 be explained by the higher content of available N in the biogas digestate, since the  
640 amount of applied  $NH_4^+$ -N in the substrate was not distinctively different in particular  
641 when accounting for  $NH_3$  losses. As mentioned before, the high pH and the lower  
642 C/N ratio of the biogas digestate, obviously slightly enhanced SOM mineralization  
643 compared to cattle slurry fertilizer, leading to extra N for nitrification and denitrification.  
644 Thus the significantly higher  $N_2O$  emissions from the digestate treatments compared  
645 to the cattle slurry treatments could probably be attributed to a priming effect caused  
646 by increased SOM mineralization. However, further investigations are required to  
647 prove whether digestates enhanced SOM mineralization or if the additional released  
648  $N_{min}$  is derived from the organically bounded N in the fertilizer.

649 Nevertheless, the observed linear increase in the cumulative  $N_2O$ -N emissions during  
650 the first 16 days or annual  $N_2O$  emissions, due to a higher mean groundwater level  
651 and a higher application rate of  $NH_4^+$ -N reveal the importance of site adapted N  
652 fertilization and the avoidance of N surpluses during agricultural use of  $C_{org}$  rich  
653 grasslands. The same was also postulated for mineral soils by Ruser (2010).

#### 654 **4.4. Fertilizer and site induced $CH_4$ emissions**

655 The observed consumption rates of  $CH_4$  were in the range of  $CH_4$  uptakes reported  
656 by Flessa et al. (1998) for two different meadows in a southern German fen peatland.  
657 Slightly higher  $CH_4$  emissions of up to  $1.46 \text{ kg } CH_4\text{-C ha}^{-1} \text{ yr}^{-1}$  were reported from  
658 Beetz et al. (2013) for a drained intensive grassland in northern German and from  
659 Nykanen et al. (1995) for a drained grassland in Finland. It is known that drainage  
660 turns peatlands from a significant source back to a sink of  $CH_4$  (Crill et al., 1994). In  
661 peatlands the position of the groundwater table is considered as the key factor  
662 regulating methanogenic and methanotrophic processes (Whalen, 2005). In line with  
663 this, Flessa et al. (1998) showed that the consumption rate of  $CH_4$  increased with

664 lowering of the groundwater level. Nevertheless, significant differences in the amount  
665 of the annual CH<sub>4</sub> uptake capacity between the two study sites C<sub>org</sub>-medium and C<sub>org</sub>-  
666 high could not be seen, although distinct differences in the groundwater table were  
667 observed.

668 The occasionally observed CH<sub>4</sub> peak emissions were only found immediately after  
669 cattle slurry application. This was in line with several other studies which reported  
670 short-term CH<sub>4</sub> emissions immediately after organic fertilizer application due probably  
671 to volatilization of dissolved CH<sub>4</sub> from the applied substrate (Sommer et al., 1996;  
672 Chadwick et al., 2000; Wulf et al., 2002; Jones et al., 2005; Amon et al., 2006). The  
673 longer lasting CH<sub>4</sub> emissions observed after the first application event at the slurry  
674 treatment of the C<sub>org</sub>-high site might result from the degradation of volatile fatty acids  
675 by methanogenic bacteria (Chadwick et al., 2000; Wulf et al., 2002). Furthermore, the  
676 high groundwater level promotes the formation of CH<sub>4</sub> during this time period.  
677 However, we could not find any significant differences in the short term or annual  
678 CH<sub>4</sub> emissions between the two investigated fertilizers. According to Chadwick et al.  
679 (2000) more than 90% of total CH<sub>4</sub> emissions occur during the first 24h following  
680 fertilizer application. Therefore, we must assume that we have missed most of  
681 fertilizer induced CH<sub>4</sub> emissions. However, all studies from literature confirm the only  
682 minor importance of CH<sub>4</sub> emissions from applied organic fertilizers in the GHG  
683 balance of agricultural grasslands (Wulf et al., 2002; Amon et al., 2006; Dietrich et al.,  
684 2012).

#### 685 **4.5. N-losses by NH<sub>3</sub> volatilization**

686 The NH<sub>3</sub> losses measured after splash plate application at the third application event  
687 followed the typical pattern of lost ammonia (Clemens et al., 2006), particularly at the  
688 digestate treatments. Significantly higher NH<sub>3</sub> losses from treatments fertilized with  
689 biogas digestate were observed compared to those fertilized with cattle slurry. This is  
690 in line with several other studies (Amon et al., 2006; Möller and Stinner, 2009; Ni et  
691 al., 2011). However, it has to be taken into account that the present results are based  
692 only on measurements from a single application event, on which the largest  
693 differences in the fertilizer compositions occurred (see Table 2).

694 The higher NH<sub>3</sub> losses from treatments fertilized with biogas digestate could be  
695 attributed to the higher amount of NH<sub>4</sub><sup>+</sup> and the distinctly higher pH value of the  
696 applied digestate compared to the cattle slurry at the third fertilization event. Several  
697 authors propose that a lower dry matter content of slurries favors the infiltration into

698 the soil with a subsequent faster decrease of  $\text{NH}_3$  losses (Sommer et al. 1996; Ni et  
699 al. 2011). However, although the observed dry matter content of the biogas  
700 digestates was very low and at the lower end of values reported in literature (e.g.  
701 Gutser et al., 2005; Möller et al., 2008; Quarkernack et al., 2011) no corresponding  
702 effect was found in the present study. According to Döhler and Horlacher (2010),  
703 water saturated and dry soils lead to higher  $\text{NH}_3$ -losses due to the reduced infiltration  
704 of slurries. Thus it could be assumed that the infiltration of the slurries was possibly  
705 hampered due to the strong rain event which took place before the fertilizer  
706 application. Additionally, at low dosage applications a large part of the organic  
707 fertilizer remained on the plant canopy and thus soil contact and infiltration was  
708 limited after spreading. We conclude that this was also the main reason why no  
709 significant differences in the pattern of  $\text{NH}_3$  volatilization between the soil types were  
710 found in the present study.

711 The observed relative N losses of 15–36% of applied  $\text{NH}_4^+$ -N, were in the range  
712 reported in the literature (Sommer et al., 1996; Clemens et al., 2006; Quakernack et  
713 al., 2011). This demonstrates that  $\text{NH}_3$  volatilization is quantitatively the most  
714 important N-loss from slurry application, as was also proposed by Flessa and Beese  
715 (2000). Beside the negative effects of eutrophication and acidification of ecosystems  
716 (Dragosits et al., 2002; Sanderson et al., 2006; Ni et al., 2011), distinct  $\text{NH}_3$   
717 volatilization decreases the N fertilizer use efficiency. One of the most effective  
718 measures to reduce  $\text{NH}_3$  emissions from grassland is the incorporation of slurry  
719 (Rodhe et al., 2006). However, several studies reported a considerable increase of  
720 greenhouse gases (GHG), mainly  $\text{N}_2\text{O}$ , after injection of slurries and biogas  
721 digestates (Dosch and Gutser, 1996; Flessa and Beese, 2000; Wulf et al., 2002).  
722 However, up to date no study has examined the effect of the injection technique on  
723 organic soils.

## 724 **5. Conclusion**

725 We studied  $\text{N}_2\text{O}$ ,  $\text{CH}_4$  and  $\text{NH}_3$  fluxes after splash plate application of biogas  
726 digestate and cattle slurry in a region known for its risk of high  $\text{N}_2\text{O}$  and  $\text{NH}_3$   
727 emissions and we were the first to study digestate application on high organic carbon  
728 soils with 10 to 17%  $\text{C}_{\text{org}}$  content in the topsoil. To our surprise,  $\text{N}_2\text{O}$  emissions  
729 remained lower than typical rates and EFs observed on mineral soils in the vicinity of  
730 the sites. We attributed the low  $\text{N}_2\text{O}$  emissions to a mild winter without clear freeze-

731 thaw cycles, but maybe also to frequent application with low dosage of N, which was  
732 quickly taken up by the grass vegetation, as could be seen in the apparent  $NUE_{min}$ .  
733  $N_2O$  emissions increased with  $C_{org}$  content and fertilization. As hypothesized,  $N_2O$   
734 and  $NH_3$  emissions were distinctly higher after digestate than after slurry fertilization,  
735 which probably could be attributed to a priming effect caused by increased SOM  
736 mineralization for  $N_2O$ . Due to the deep drainage,  $CH_4$  emissions were of only minor  
737 importance independent of fertilizer type. Estimated N balances were negative for the  
738 control and the digestate treatments, but strongly positive in all cases when the net N  
739 supply from soil organic matter mineralization was considered. The observed linear  
740 increase in cumulative  $N_2O$  emissions with increasing  $NH_4^+$  fertilization and  
741 increasing groundwater table reveals the importance of site adapted N fertilization  
742 and the avoidance of N surpluses during agricultural use of  $C_{org}$  rich grasslands.

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1127 Spreading: II Greenhouse Gas Emissions, J. Environ. Qual., 31, 1795–1801, 2002.  
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1129 **Table 1** Soil properties of the study site

		C <sub>org</sub> -medium	C <sub>org</sub> -high	<i>n</i>
	Sampling depth			
Soil type (WRB 2006) <sup>1</sup>		mollic Gleysol	sapric Histosol	
Soil type (German classification KA5)		GMq	KV-KM	
Peat depth [cm] <sup>1</sup>		80	70	1
pH value*		4.1	4.2	
Total nitrogen [%]*		1.0	1.5	
Organic carbon [%]	0–10 cm	10.3 ± 0.2	17.0 ± 0.1	9
	10–20 cm	9.3 ± 0.2	16.3 ± 0.2	9
Bulk density [g cm <sup>-3</sup> ]	0–10 cm	0.79 ± 0.02	0.54 ± 0.02	18
	10–20 cm	0.90 ± 0.01	0.64 ± 0.01	18
Porosity [%]	0–10 cm	71 ± 1	78 ± 1	18
	10–20 cm	67 ± 1	72 ± 0	18

1130 Values present means ± standard error

1131 <sup>1</sup> World Reference Base for Soil Resources

1132 \* Relative to the upper horizon (C<sub>org</sub>-medium 0–20 cm; C<sub>org</sub>-high 0–15 cm); Roßkopf personal communication



**Table 2** Physical and chemical properties from the applied digestates and slurries.

	Cattle slurry					Biogas digestate				
	1. Application (14.06.2010)	2. Application (25.08.2010)	3. Application (27.05.2011)	4. Application (22.09.2011)	5. Application (04.11.2011)	1. Application (14.06.2010)	2. Application (25.08.2010)	3. Application (27.05.2011)	4. Application (22.09.2011)	5. Application (04.11.2011)
Fertilizer quantity [m <sup>3</sup> ha <sup>-1</sup> ]	20	20	25	20	20	20	20	25	20	20
Total carbon [kg ha <sup>-1</sup> ]	579	676	798	797	1073	384	373	167	184	178
Organic carbon [kg ha <sup>-1</sup> ]	410	573	655	706	960	306	337	148	161	178
Total nitrogen [kg ha <sup>-1</sup> ]	47	64	70	85	97	49	52	78	35	61
NO <sub>3</sub> <sup>-</sup> [kg N ha <sup>-1</sup> ]	0	0	0	0	0	0	0	0	0	0
NH <sub>4</sub> <sup>+</sup> [kg N ha <sup>-1</sup> ]	20	28	23	33	38	22	28	51	17	40
C/N ratio	12	11	11	9	11	8	7	2	5	3
pH (CaCl <sub>2</sub> )	–	–	6.8	7.0	7.0	–	–	7.7	7.4	7.7
Dry matter content [%]	5	7	7	9	10	4	4	2	2	3

**Table 3** Mean (minimum/maximum) groundwater level (GW), NO<sub>3</sub><sup>-</sup> and NH<sub>4</sub><sup>+</sup> content in the soil following organic fertilizer application and for the investigated years 2010 and 2011.

Sampling depth [cm]	C <sub>org</sub> -medium			C <sub>org</sub> -high			<i>n</i>	
	Control	Cattle slurry	Biogas digestate	Control	Cattle slurry	Biogas digestate		
1 Application (14.06. – 30.06.2010)								
GW level [cm]	-32 (-62/-2)	-39 (-60/-5)	-31 (-58/-2)	-21 (-46/-1)	-26 (-45/-7)	-33 (-45/-19)		
NO <sub>3</sub> <sup>-</sup> [mg N kg <sup>-1</sup> ]	0–10	5 (1/9)	5 (1/7)	7 (3/10)	6 (1/12)	8 (3/11)	8 (5/10)	12
	10–20	9 (6/12)	9 (5/13)	11 (8/15)	11 (8/15)	12 (7/15)	14 (6/19)	12
NH <sub>4</sub> <sup>+</sup> [mg N kg <sup>-1</sup> ]	0–10	0 (0/1)	0 (0/1)	0 (0/1)	1 (0/2)	1 (0/6)	1 (0/3)	12
	10–20	0 (0/1)	0 (0/1)	0 (0/1)	0 (0/2)	1 (0/11)	1 (0/2)	12

## 2 Application (25.08. – 10.09.2010)

GW level [cm]		-64 (-70/-49)	-58 (-63/-42)	-57 (-63/-40)	-36 (-40/-37)	-40 (-46/-22)	-37 (-43/-15)	
NO <sub>3</sub> <sup>-</sup> [mg N kg <sup>-1</sup> ]	0-10	13 (6/23)	27 (7/49)	21 (14/30)	25 (17/37)	50 (17/95)	25 (9/43)	12
	10-20	22 (17/28)	28 (19/37)	27 (17/38)	31 (26/35)	34 (11/45)	31 (12/48)	12
NH <sub>4</sub> <sup>+</sup> [mg N kg <sup>-1</sup> ]	0-10	0 (0/0)	2 (0/17)	0 (0/0)	3 (0/32)	0 (0/1)	1 (0/5)	12
	10-20	1 (0/10)	0 (0/0)	0 (0/1)		0 (0/1)	1 (0/3)	12

## 3 Application (27.05. – 11.06.2011)

GW level [cm]		-82 (-94/-57)	-76 (-89/-52)	-80 (-97/-46)	-41 (-60/-11)	-47 (-62/-16)	-49 (-62/-16)	
NO <sub>3</sub> <sup>-</sup> [mg N kg <sup>-1</sup> ]	0-10	9 (4/17)	17 (5/30)	40 (10/75)	17 (11/26)	29 (12/63)	29 (11/50)	12
	10-20	17 (11/22)	31 (18/44)	45 (18/75)	24 (18/30)	28 (18/40)	45 (21/148)	12
NH <sub>4</sub> <sup>+</sup> [mg N kg <sup>-1</sup> ]	0-10	0 (0/2)	2 (0/10)	21 (0/104)	0 (0/1)	1 (0/5)	10 (0/47)	12
	10-20	0 (0/1)	1 (0/2)	6 (0/26)	1 (0/2)	1 (0/3)	4 (0/12)	12

## 4 Application (22.09. – 07.10.2011)

GW level [cm]		-83 (-87/-72)	-77 (-81/-70)	-76 (-83/-58)	-54 (-60/-33)	-55 (-58/-46)	-53 (-57/-41)	
NO <sub>3</sub> <sup>-</sup> [mg N kg <sup>-1</sup> ]	0-10	18 (12/28)	48 (18/83)	62 (49/87)	23 (20/34)	43 (28/73)	45 (18/86)	12
	10-20	32 (18/46)	50 (21/79)	53 (35/66)	24 (20/30)	30 (22/39)	38 (23/86)	12
NH <sub>4</sub> <sup>+</sup> [mg N kg <sup>-1</sup> ]	0-10	0 (0/0)	1 (0/8)	1 (0/10)	0 (0/0)	3 (0/21)	0 (0/0)	12
	10-20	0 (0/0)	0 (0/0)	0 (0/1)	0 (0/1)	1 (0/3)	0 (0/0)	12

## 2010

GW level [cm]*		-67 (-94/-2)	-65 (-91/-2)	-63 (-92/0)	-41 (-68/2)	-45 (-64/-1)	-45 (-67/-1)	
NO <sub>3</sub> <sup>-</sup> [mg N kg <sup>-1</sup> ]	0-10	9 (1/26)	14 (1/49)	12 (3/30)	15 (1/37)	24 (3/95)	17 (4/43)	45
	10-20	14 (5/34)	17 (5/38)	17 (4/38)	19 (7/47)	23 (6/64)	21 (6/49)	45
NH <sub>4</sub> <sup>+</sup> [mg N kg <sup>-1</sup> ]	0-10	0 (0/4)	1 (0/17)	0 (0/9)	2 (0/32)	2 (0/19)	1 (0/14)	45
	10-20	0 (0/10)	0 (0/1)	0 (0/1)	0 (0/8)	1 (0/11)	1 (0/5)	45

## 2011

GW level [cm]		-76 (-98/-3)	-72 (-92/0)	-72 (-97/0)	-47 (-67/1)	-52 (-66/-3)	-50 (-65/-3)	
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NO <sub>3</sub> <sup>-</sup> [mg N kg <sup>-1</sup> ]	0–10	13 (4/31)	25 (5/83)	36 (8/111)	21 (11/41)	36 (12/98)	34 (11/91)	47
	10–20	24 (9/46)	34 (15/79)	40 (18/120)	27(14/52)	33 (18/78)	37 (10/148)	47
NH <sub>4</sub> <sup>+</sup> [mg N kg <sup>-1</sup> ]	0–10	0 (0/2)	1 (0/10)	6 (0/104)	1 (0/12)	2 (0/21)	4 (0/60)	47
	10–20	0 (0/3)	0 (0/4)	2 (0/26)	1 (0/7)	1 (0/27)	2 (0/12)	47

**Table 4** Calculated emission factors (EF) for the year 2011 and for single application events (16 days) (Apl. 1 – Apl. 4). EF based on the amount of total nitrogen (N<sub>tot</sub>) without consideration of NH<sub>3</sub>-N losses.

	C <sub>org</sub> -medium			C <sub>org</sub> -high		
	Control	Cattle slurry	Biogas digestate	Control	Cattle slurry	Biogas digestate
N <sub>2</sub> O exchange [kg N ha <sup>-1</sup> yr <sup>-1</sup> ]	0.91 ± 0.49	1.21 ± 0.05	1.86 ± 0.23	1.18 ± 0.07	1.77 ± 0.15	3.14 ± 0.91
Annual EF		0.12	0.55		0.23	1.13
EF Apl. 1		0.18	0.17		0.20	0.35
EF Apl. 2		0.11	0.05		0.11	0.21
EF Apl. 3		0.08	0.21		0.23	0.68
EF Apl. 4		0.09	0.33		0.15	0.56

**Table 5** N uptake and N use efficiency for the years 2010 and 2011.

Treatment	Cutting date	Fertilization date	N content plant [%]	DM [t ha <sup>-1</sup> yr <sup>-1</sup> ]	N uptake [kg N ha <sup>-1</sup> ]	N applied [kg N ha <sup>-1</sup> ]†	N <sub>min</sub> applied [kg N ha <sup>-1</sup> ]†	N use efficiency [%]	N <sub>min</sub> use efficiency [%]
Control C <sub>org</sub> -medium	24-May-10	–	2.04*	2.52	51	–	–	–	–

Control C <sub>org</sub> -high	24-May-10	–	2.14	2.93	63	–	–	–	–
Cattle slurry C <sub>org</sub> -medium	24-May-10	N.A.	2.37	3.19	76	–	–	–	–
Cattle slurry C <sub>org</sub> -high	24-May-10	N.A.	2.14	3.58	77	–	–	–	–
Biogas digestate C <sub>org</sub> -medium	24-May-10	N.A.	2.04	4.17	85	–	–	–	–
Biogas digestate C <sub>org</sub> -high	24-May-10	N.A.	2.27	4.39	100	–	–	–	–
Control C <sub>org</sub> -medium	20-Aug-10	–	2.03	2.02	41	–	–	–	–
Control C <sub>org</sub> -high	20-Aug-10	–	2.00	2.63	53	–	–	–	–
Cattle slurry C <sub>org</sub> -medium	20-Aug-10	14-Jun-10	2.19	3.06	67	45	17	58	153
Cattle slurry C <sub>org</sub> -high	20-Aug-10	14-Jun-10	1.93	3.23	62	45	17	22	57
Biogas digestate C <sub>org</sub> -medium	20-Aug-10	14-Jun-10	2.03	2.99	61	38	14	52	140
Biogas digestate C <sub>org</sub> -high	20-Aug-10	14-Jun-10	2.00	3.51	70	38	14	47	125
Control C <sub>org</sub> -medium	23-May-11	–	1.96	2.66	52	–	–	–	–
Control C <sub>org</sub> -high	23-May-11	–	1.70	3.82	65	–	–	–	–
Cattle slurry C <sub>org</sub> -medium	23-May-11	25-Aug-10	2.01	2.58	52	61	24	0	0
Cattle slurry C <sub>org</sub> -high	23-May-11	25-Aug-10	1.70	4.20	71	61	24	11	27
Biogas digestate C <sub>org</sub> -medium	23-May-11	25-Aug-10	1.96	3.97	78	40	18	64	144
Biogas digestate C <sub>org</sub> -high	23-May-11	25-Aug-10	1.83	4.54	83	40	18	45	101
Control C <sub>org</sub> -medium	1-Aug-11	–	1.71	2.06	35	–	–	–	–
Control C <sub>org</sub> -high	1-Aug-11	–	1.48	2.88	43	–	–	–	–
Cattle slurry C <sub>org</sub> -medium	1-Aug-11	27-May-11	1.71	2.73	47	67	20	17	58
Cattle slurry C <sub>org</sub> -high	1-Aug-11	27-May-11	1.51	3.19	48	67	20	8	28
Biogas digestate C <sub>org</sub> -medium	1-Aug-11	27-May-11	1.78	4.88	87	60	33	86	158
Biogas digestate C <sub>org</sub> -high	1-Aug-11	27-May-11	1.48	5.34	79	60	33	61	112
Control C <sub>org</sub> -medium	13-Sep-11	–	2.53	1.71	43	–	–	–	–
Control C <sub>org</sub> -high	13-Sep-11	–	2.26	2.27	51	–	–	–	–
Cattle slurry C <sub>org</sub> -medium	13-Sep-11	27-May-11	2.57	2.28	59	(55)‡	(8)‡	28	189
Cattle slurry C <sub>org</sub> -high	13-Sep-11	27-May-11	2.53	2.64	67	(61)‡	(14)‡	25	110
Biogas digestate C <sub>org</sub> -medium	13-Sep-11	27-May-11	2.53	3.15	80	(8)‡	(0)‡	436	–

Biogas digestate C <sub>org</sub> -high	13-Sep-11	27-May-11	2.26	3.25	74	(24)‡	(0)‡	94	–
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\* N contents from control treatments were estimated from fertilized treatments.

† Applied N<sub>tot</sub> and N<sub>min</sub> were corrected by NH<sub>3</sub>-N losses (23% and 5% from N<sub>tot</sub>, or rather 36% and 15% from N<sub>min</sub> for biogas digestate and cattle slurry, respectively).

‡ Hypothetically remaining N<sub>tot</sub> and N<sub>min</sub> from the application event 3 (27<sup>th</sup> May 2011).

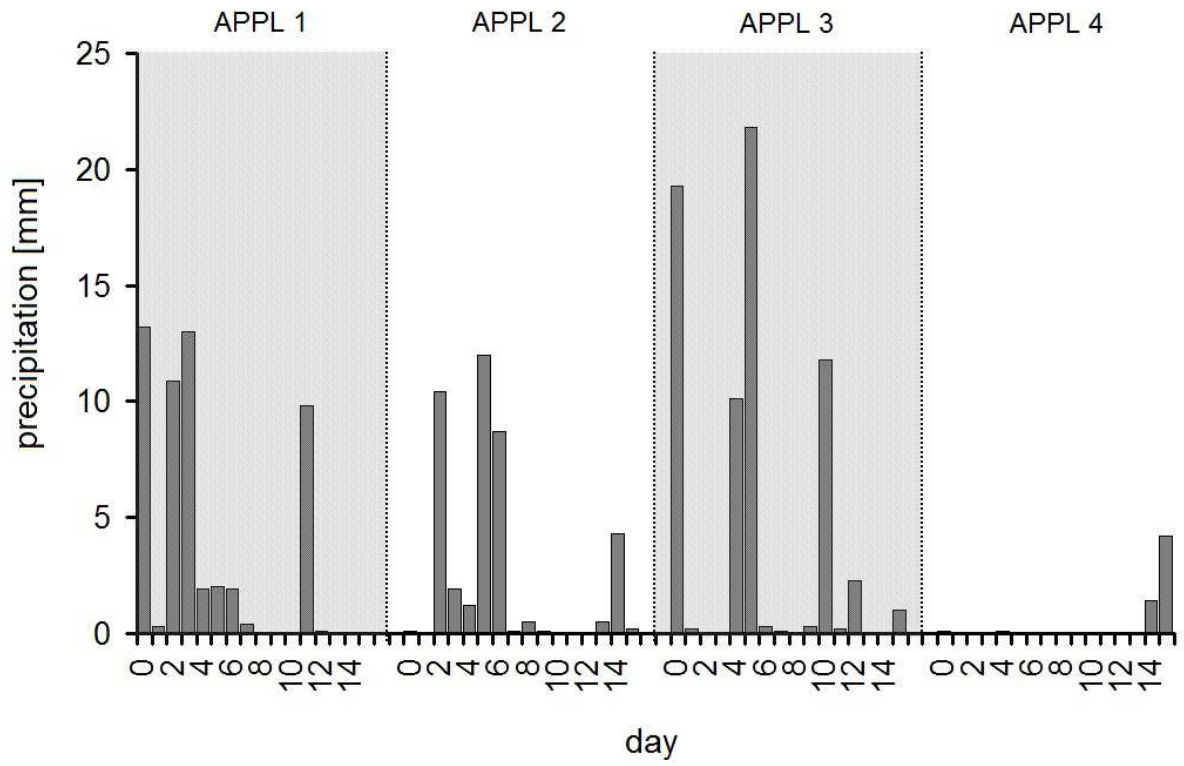
N.A. = not available.

**Table 6** Estimated nitrogen balance for the year 2011.

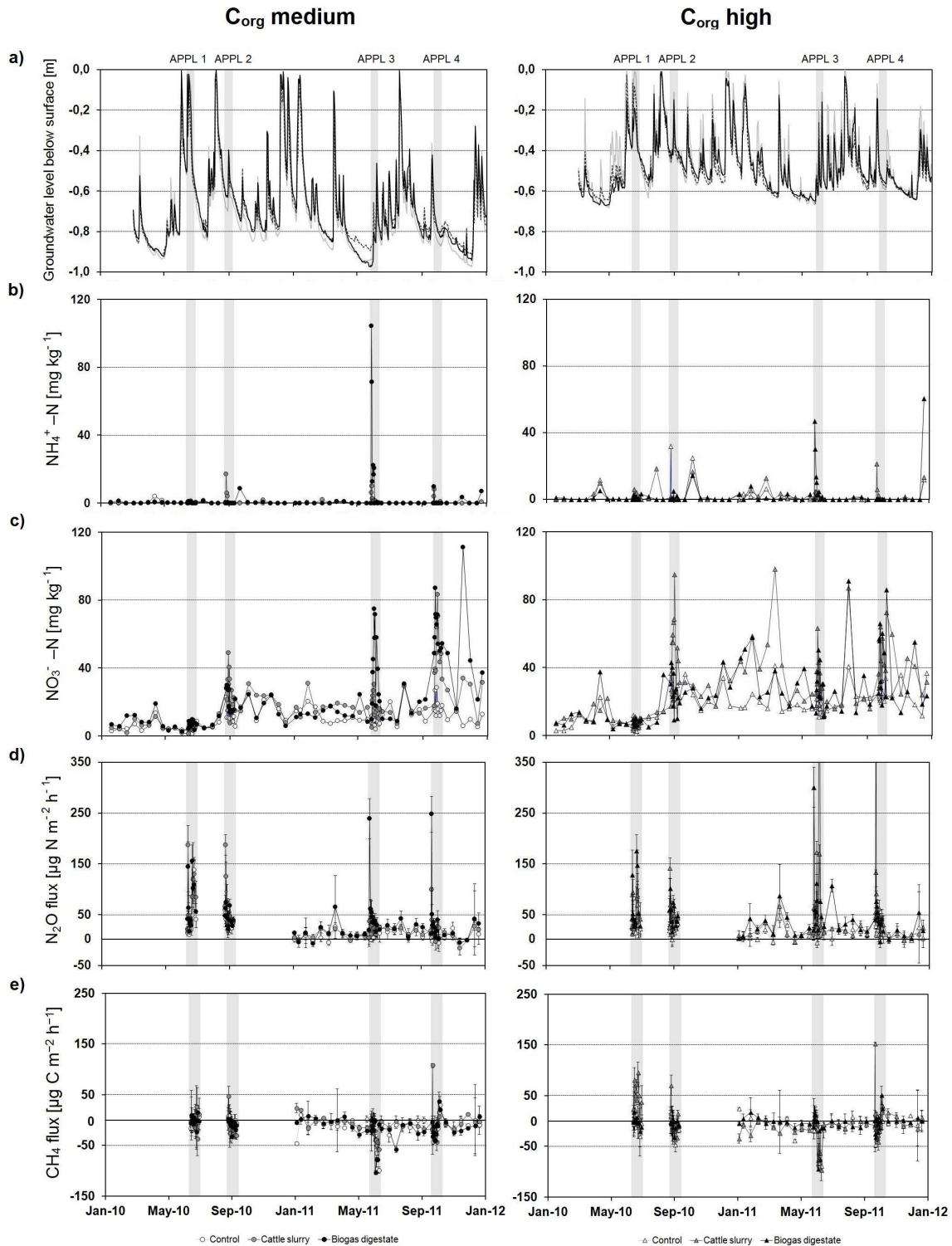
Treatment	N applied [kg N ha <sup>-1</sup> yr <sup>-1</sup> ]	N <sub>min</sub> t1* [kg N ha <sup>-1</sup> ]	N <sub>min</sub> t2* [kg N ha <sup>-1</sup> ]	N deposition [kg N ha <sup>-1</sup> yr <sup>-1</sup> ]	N uptake [kg N ha <sup>-1</sup> yr <sup>-1</sup> ]	N <sub>2</sub> O [kg N ha <sup>-1</sup> yr <sup>-1</sup> ]	NH <sub>3</sub> † [kg N ha <sup>-1</sup> yr <sup>-1</sup> ]	N balance [kg N ha <sup>-1</sup> yr <sup>-1</sup> ]
Control C <sub>org</sub> -medium	0	27.5	29.4	7.2	130	0.9	0.0	-122.4
Control C <sub>org</sub> -high	0	22.8	27.7	7.2	159	1.2	0.0	-148.0
Cattle slurry C <sub>org</sub> -medium	252	35.7	51.2	7.2	157	1.2	37.8	78.6
Cattle slurry C <sub>org</sub> -high	252	27.3	68.1	7.2	186	1.8	37.8	74.0
Biogas digestate C <sub>org</sub> -medium	174	29.8	83.3	7.2	244	1.9	40.0	-51.6
Biogas digestate C <sub>org</sub> -high	174	26.2	28.4	7.2	236	3.1	40.0	-95.3

\* Reference date for t1 is the 06<sup>th</sup> April 2011 and for t2 the 18<sup>th</sup> October 2011.

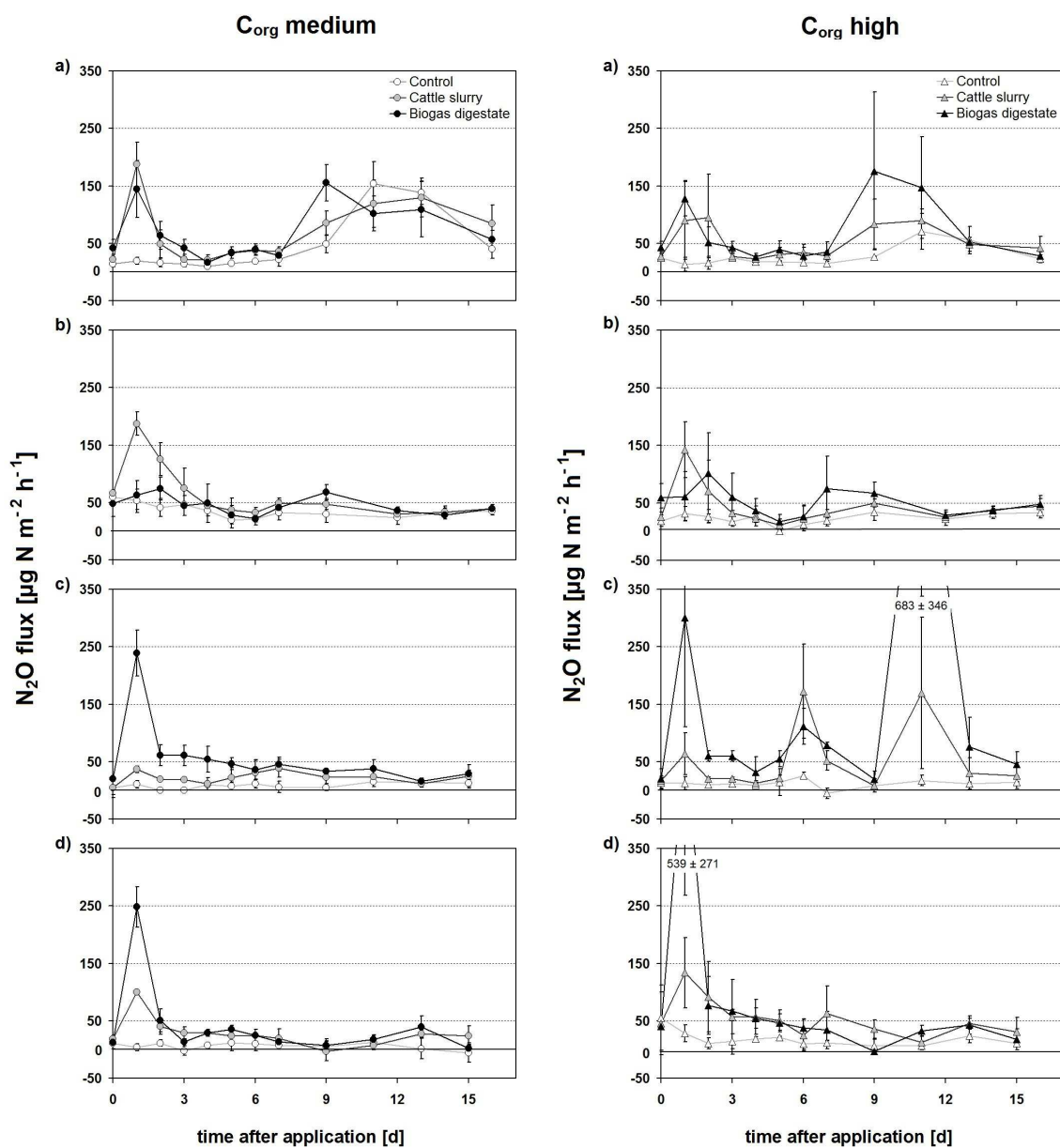
† NH<sub>3</sub>-N losses at the fourth and fifth application event were estimated based on NH<sub>3</sub> measurements at the third application event (23% and 5% from N<sub>tot</sub> for biogas digestate and cattle slurry, respectively).



**Fig. 1** Daily sums of precipitation following the organic fertilizer application events (APPL).

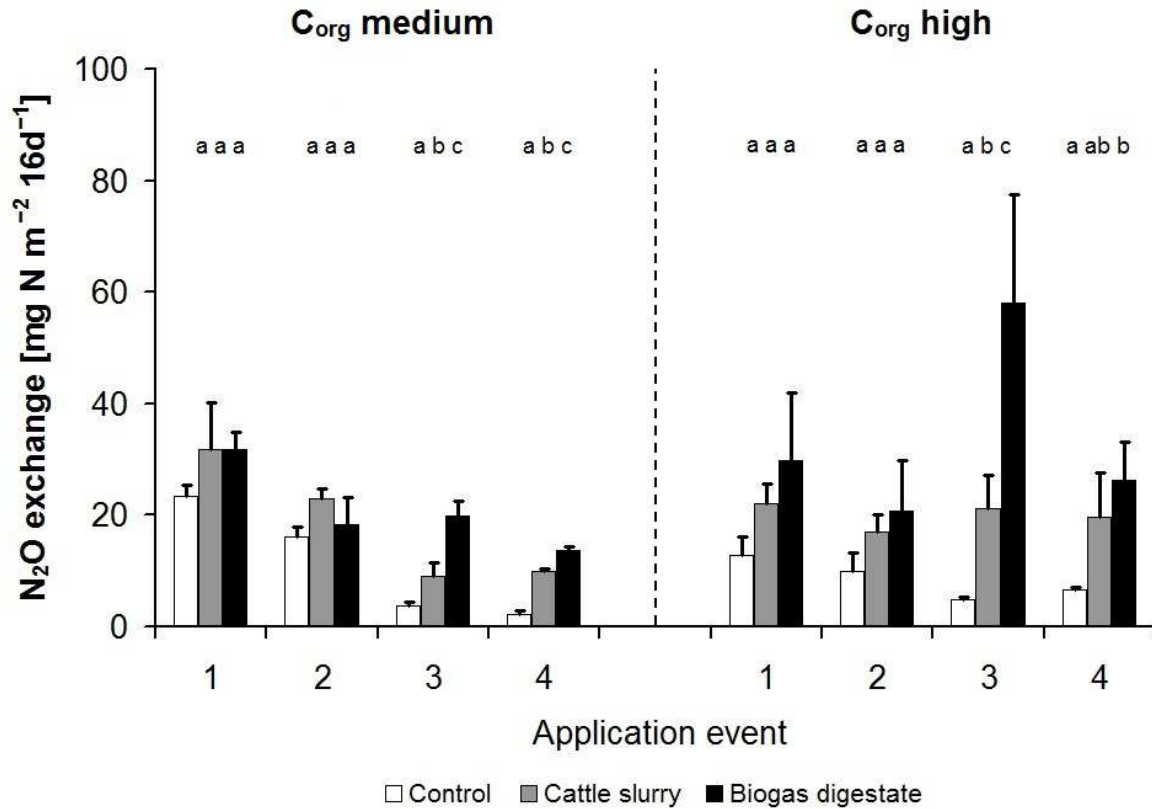


**Fig. 2** Variation in groundwater level (a), extractable  $\text{NH}_4^+$  (b) and  $\text{NO}_3^-$  (c) contents for the soil depth 0–10 cm,  $\text{N}_2\text{O}$  (d) and  $\text{CH}_4$  fluxes (e) (Mean  $\pm$  SD,  $n = 3$ ) of the  $\text{C}_{\text{org}}$ -medium and  $\text{C}_{\text{org}}$ -high sites from January 2010 to January 2012. Gray bars mark the 15 or 16 days period of intensiv gas flux measurements following organic fertilizer application.

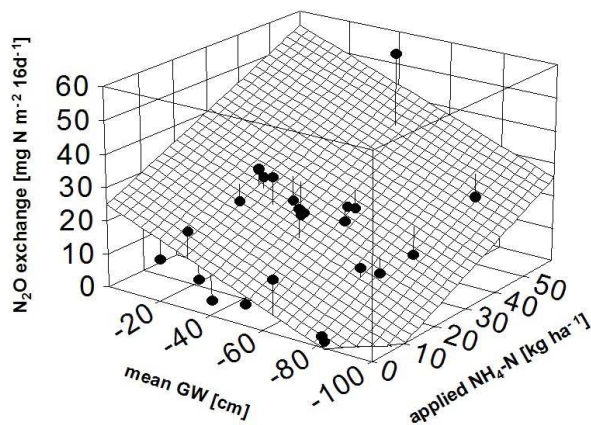


**Fig. 3** Mean ( $\pm$  SD,  $n = 3$ ) N<sub>2</sub>O fluxes following organic fertilizer application events (a) 14.06–30.06.2010; (b) 25.08.–10.09.2010; (c) 27.05.–11.06.2011 and (d) 22.09.–07.10.2011.

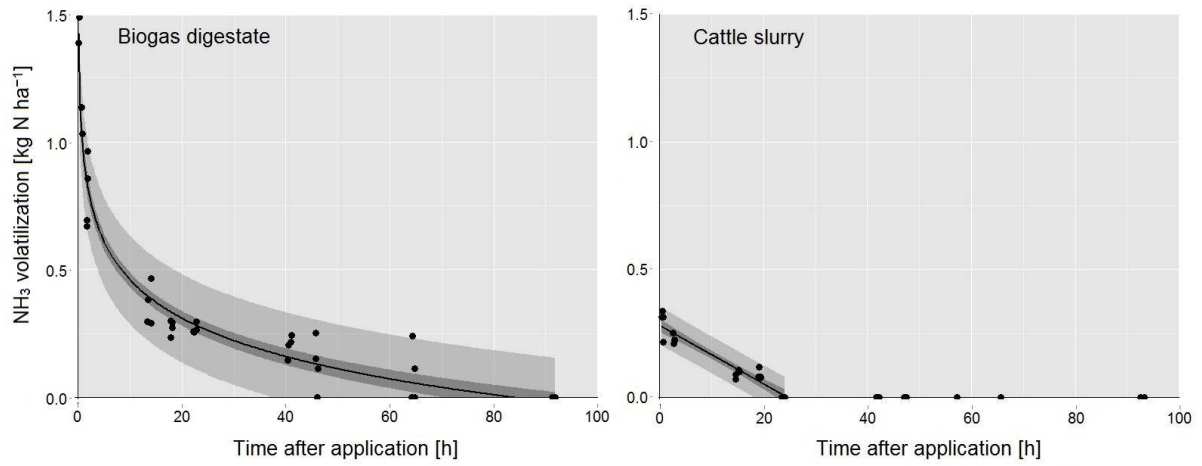




**Fig. 4** Cumulative N<sub>2</sub>O exchange during 16 days following organic fertilizer application. Bars indicate mean values + SD ( $n = 3$ ). Means followed by the same letter indicated no significant differences between treatments at a single application event for sites C<sub>org</sub>-medium and C<sub>org</sub>-high respectively (ANOVA, Tukey HSD-test at  $P \leq 0.05$ ).



**Fig. 5** Relationship of 16 days cumulative N<sub>2</sub>O–N emissions ( $y$ ) to mean groundwater level ( $x_1$ ) and the amount of applied NH<sub>4</sub>–N ( $x_2$ ). The regression equation is  $y = 24.98 (\pm 4.98) + x_1 \cdot 0.30 (\pm 0.09) + x_2 \cdot 0.51 (\pm 0.11)$ ;  $R^2 \text{ adj.} = 0.53$ ,  $P < 0.001$ ,  $df = 21$ . Solid lines indicate the deviation of measured data from the model surface.



**Fig. 6** Ammonia ( $\text{NH}_3$ ) volatilization following organic fertilizer application at event 3 (27.05.2011). Dots present single  $\text{NH}_3$  measurements for a time period of 94 hours. Black lines show the estimated  $\text{NH}_3$  volatilization with 95% confidence band (dark grey) and 95% prediction band (light grey). Model function for biogas digestate is:  $y = -0.2619 (\pm 0.025) \cdot \ln(x) + 0.9605 (\pm 0.008)$ ;  $r^2 = 0.96$ ;  $P < 0.0001$ ; Model function for cattle slurry is:  $y = 0.2818 (\pm 0.012) - 0.0114(x) (\pm 0.001)$ ;  $r^2 = 0.92$ ;  $P < 0.0001$ .