

1 **Short-term effects of biogas digestate and cattle slurry**
2 **application on greenhouse gas emissions affected by N**
3 **availability from grasslands on drained fen peatlands and**
4 **associated organic soils.**

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18 **Abstract**

19 The change in the German energy policy has resulted in a strong increase in the
20 number of biogas plants in Germany. As a consequence, huge amounts of nutrient
21 rich residues, the by-products of the fermentative process, are used as organic
22 fertilizers. Drained peatlands are increasingly used to satisfy the huge demand for
23 fermentative substrates (e.g. energy-crops, grass silage) and the digestate is
24 returned to the peatlands. However, drained organic soils are considered as hot
25 spots for nitrous oxide (N₂O) emissions and organic fertilization is additionally known
26 to increase N₂O emissions from managed grasslands. Our study addressed the
27 questions a) to what extent biogas digestate and cattle slurry application increase
28 N₂O and methane (CH₄) fluxes as well as the mineral nitrogen use efficiency (NUE_{min})
29 and grass yield, and b) how different soil organic matter contents (SOM) and nitrogen
30 (N) contents promote the production of N₂O. In addition NH₃ volatilization was
31 determined at one application event to obtain first clues with respect to the effects of

32 soil and fertilizer types. The study was conducted at two sites within a grassland
33 parcel, which differed in their soil organic carbon (SOC) and N contents. At each site
34 (named C_{org}-medium and C_{org}-high) three plots were established, one was fertilized
35 five times with biogas digestate, one with cattle slurry and the third served as control
36 plot. On each plot, fluxes of N₂O and CH₄ were measured on three replicates over
37 two years using the closed chamber method. For NH₃ measurements we used the
38 calibrated dynamic chamber method. On an annual basis, the application of biogas
39 digestate significantly enhanced the N₂O fluxes compared to the application of cattle
40 slurry and additionally increased the plant N-uptake and NUE_{min}. Furthermore, N₂O
41 fluxes from the C_{org}-high site significantly exceeded N₂O fluxes from the C_{org}-medium
42 sites. Annual cumulative emissions ranged from 0.91 ± 0.49 kg N ha⁻¹ yr⁻¹ to 3.14 ±
43 0.91 kg N ha⁻¹ yr⁻¹. Significantly different CH₄ fluxes between the investigated
44 treatments or the different soil types were not observed. Cumulative annual CH₄
45 exchange rates varied between -0.21 ± 0.19 kg C ha⁻¹ yr⁻¹ and -1.06 ± 0.46 kg C
46 ha⁻¹ yr⁻¹. Significantly higher NH₃ losses, NUE_{min} and grass yields from treatments
47 fertilized with biogas digestate compared to those fertilized with cattle slurry were
48 observed. The total NH₃ losses following the splash plate application were 18.17 kg
49 N ha⁻¹ for the digestate treatments and 3.48 kg N ha⁻¹ for the slurry treatments (36%
50 and 15% of applied NH₄⁺-N). The observed linear increase of 16 days cumulative
51 N₂O-N exchange or rather annual N₂O emissions, with mean groundwater level and
52 ammonium application rate, reveal the importance of site adapted N fertilization and
53 the avoidance of N surpluses in C_{org} rich grasslands.

54 **1. Introduction**

55 Germany has become the largest biogas producing country in the world, since the
56 change in the German energy policy and the enactment of the German Renewable
57 Energy Act (Weiland, 2010). At the end of 2012, more than 7,500 agricultural biogas
58 plants have been operated in Germany (Fachverband Biogas, 2014). Heat and
59 power from biogas substitute fossil fuels and therefore can reduce greenhouse gas
60 (GHG) emissions (Weiland, 2010; Don et al., 2011). The strong increase in the
61 number of biogas plants caused a land-use change towards agro-biomass production
62 and additionally raised the land-use intensity to satisfy the huge demand for
63 fermentative substrates (Don et al., 2011). In 2011, the proportion of grass silage
64 accounted for 9% of the total renewable resources for biogas production (DBFZ,

65 2012) and represented the second most important fermentation substrate after maize
66 silage.

67 During the fermentative process high amounts of nutrient rich residues are produced
68 as a by-product. Today, this new form of organic fertilizer is used instead of mineral
69 fertilizers or animal slurries to maintain soil fertility and productivity. Several studies
70 reported a significant increase in nitrous oxide (N₂O) emissions due to the application
71 of nitrogen fertilizers (e.g. Bouwman, 1996; Chadwick et al., 2000; Rodhe et al., 2006;
72 Ruser, 2010). Liquid organic fertilizers such as animal slurry add easily degradable
73 organic carbon (Christensen 1983) and moisture, both further favoring N₂O losses
74 through denitrification (Clayton et al., 1997). Enhanced N₂O emissions are of major
75 concern due to the fact that N₂O acts as a radiative forcing greenhouse gas (IPCC,
76 2007) and contributes to the chemical destruction of stratospheric ozone (Crutzen,
77 1979). In Germany, about 78% of N₂O emissions originate from the agricultural
78 sector (Umweltbundesamt, 2014). Particularly organic soils (e.g. drained peat soils
79 and soils developed in wet conditions) are considered as hotspots of GHG emissions
80 including N₂O, which is due to the very high mineralization rates of degrading peat
81 (Kasimir-Klmedtsson et al., 1997; Freibauer et al., 2004; Klmedtsson et al., 2005;
82 Goldberg et al., 2010) and to soil moisture conditions which favor anaerobic micro-
83 sites. According to Maljanen et al. (2010), N₂O emissions from drained organic soils
84 under agricultural use were on average four times higher than those from mineral
85 soils. The few field studies of organic fertilization effects on annual N₂O emissions
86 from drained organic grassland soils revealed very high N₂O emissions of up to 41.0
87 kg N ha⁻¹ yr⁻¹ (Velthof et al., 1996).

88 In Germany, 50% of the drained peatlands are used as grasslands (Drösler et al.,
89 2011), particularly in the smallholder structure of south Germany. Grassland soils in
90 Europe and Germany produce more N₂O per unit of fertilizer-N than croplands and
91 emission factors further increase with soil organic carbon and nitrogen content
92 (Freibauer and Kaltschmitt 2003; Dechow and Freibauer 2011). Moreover agricultural
93 soils in the southern part of Germany emit, about three times more of the applied N
94 as N₂O than soils in the rest of Germany, which is attributed to the more frequent
95 frost-thaw cycles and enhanced precipitation rates (Jungkunst et al. 2006, Dechow
96 and Freibauer 2011). Thus, grasslands on organic soils in South Germany represent
97 a wide-spread high-risk situation for high N₂O emissions after cattle slurry or biogas
98 digestate application, which has to our knowledge not yet been studied before.

99 Biogas digestate is depleted in easily degradable C compounds and in organic dry
100 matter content compared to fresh slurry due to anaerobic digestion (Möller and
101 Stinner, 2009). In return, the pH value and the ammonium (NH_4^+) content as well the
102 $\text{NH}_4^+/\text{N}_{\text{org}}$ ratio are higher than in fresh slurry (Wulf et al., 2002a; Möller and Stinner,
103 2009). Since digested products are more recalcitrant than fresh slurry it could be
104 assumed that microbial degradation is slow, resulting in less anoxic microsites and
105 reduced N_2O emissions than after fresh slurry application (Clemens and Huschka,
106 2001; Oenema et al., 2005; Möller and Stinner, 2009). However, the few available
107 field and laboratory experiments are contradictory regarding the effect of biogas
108 digestate application on N_2O emissions (e.g. Clemens and Huschka, 2001; Wulf et al.,
109 2002a; Clemens et al., 2006; Senbayram et al., 2009; Sängler et al., 2010), and very
110 few studies exist for grasslands (e.g. Wulf et al., 2002a, Clemens et al., 2006).

111 The different properties of biogas digestate and cattle slurry (e.g. higher NH_4^+ -N
112 concentrations, narrower C/N ratio, higher pH values) directly affect N transformation
113 processes, plant N availability and thus grass yield. Furthermore it can be assumed
114 that the plant N-uptake and the N_2O emissions are closely interconnected since N-
115 uptake can be considered as a proxy for N availability, affecting N gaseous losses as
116 well. Currently, the effect of anaerobic digestates on crop growth after surface
117 application under field conditions is contradictory, since some authors reported
118 higher crop yields compared to undigested slurries (e.g. Odlare, 2005 cited in Möller
119 and Müller, 2012) whereas others found no effects (e.g. Möller et al., 2008). However,
120 only a few studies exist for grassland but it seems that fertilization with biogas
121 digestates positively affects grass yields, but only in single years (Elsässer et al.,
122 1995; Rubæk et al., 1996; Möller et al., 2008; Möller and Müller, 2012).

123 Beside N_2O , slurry application also releases short-term methane (CH_4) and ammonia
124 (NH_3) emissions. Methane acts as strong greenhouse gas, whereas NH_3 is
125 considered as indirect greenhouse gas through ammonia deposition which could
126 promote the formation of N_2O (Moiser, 2001). Moreover, NH_3 deposition causes soil
127 acidification and eutrophication of ecosystems (e.g. Fangmeier et al., 1994; Galloway,
128 1995; Smith et al., 1999; Galloway, 2001). In Germany, agriculture is responsible for
129 95.3% of the anthropogenic NH_3 emissions (Haenel et al., 2010). Particularly high
130 NH_4^+ contents and high pH values, which are typically for the biogas digestate,
131 promote accelerated NH_3 volatilisation (Quakernack et al., 2011). High NH_3
132 emissions particularly occur after splash plate application on grassland (Rubæk et al.,

133 1996; Wulf et al., 2002b), which is still common practice in the smallholder farms of
134 South Germany.

135 The objective of this study was to quantify short-term N₂O and CH₄ emissions after
136 application of biogas digestate and cattle slurry on grassland on two types of organic
137 soils in South Germany, which differed in their soil organic carbon (SOC) and N
138 contents. Additionally it should be tested to what extent biogas digestate and cattle
139 slurry application affect N availability and grass yield. Furthermore NH₃ volatilization
140 was determined at one application event to obtain first clues with respect to the
141 effects of soil and fertilizer types. We hypothesize: a) More N₂O is emitted after
142 biogas digestate than after slurry application because of higher NH₄⁺-N
143 concentrations in the substrate and thus, higher NH₄⁺-N amounts when using equal
144 volumetric application rates. The more recalcitrant nature of the carbon in the biogas
145 digestate does not matter for GHG formation in high organic carbon soils. b) N₂O
146 emissions increase with increasing soil C_{org} and N content due to more favorable
147 conditions for denitrification after organic fertilizer application. c) Biogas digestate
148 leads to a significantly higher grass yield and N-use efficiency compared to cattle
149 slurry due to the higher N availability of the digestate.

150 **2. Materials and methods**

151 **2.1. Study site**

152 The study was conducted on a permanent grassland at a drained fen peatland 30 km
153 north-east of Munich (Freisinger Moos, 48°21'N, 11° 41'E; 450 m a.s.l.). The
154 dominant species were *Poa trivialis*, *Poa pratensis*, *Festuca pratensis*, *Dactylis*
155 *glomerata* and *Alopecurus pratensis*. The grassland was mown two and three times
156 in 2010 and 2011 respectively, as is the usual practice in this region. A summary of
157 the complete grassland management over both vegetation periods can be found in
158 Table 1. The grass was used as silage or hay for cattle or as substrate for biogas
159 plants. According to the climate station in Weißenstephan, located 10 km northeast of
160 the site, the 30-years mean annual temperature was 7.5 °C and the mean annual
161 precipitation was 787 mm (1961–1990). Annual atmospheric N deposition amounted
162 to 6.22 and 7.20 kg N ha⁻¹ yr⁻¹, with a NH₄⁺-N:NO₃⁻-N ratio of 46:54 and 49:51 in
163 2010 and 2011. Data of N deposition was collected by the Bavarian State Institute of
164 Forestry at a German Level II monitoring area (Forest Intensive Monitoring
165 Programme of the UNECE), located in 7 km distance to the investigated grassland.

166 In October 2009, we selected two sites within the grassland parcel, which differed in
167 their soil organic carbon (SOC) contents in the top soil (Table 2). According to the
168 WRB (2006) soil types were classified as mollic Gleysol (named C_{org}-medium) and as
169 sapric Histosol (named C_{org}-high) (Roßkopf personal communication).

170 **2.2. Experimental design**

171 At each site of the grassland parcel, three adjacent plots (plot dimension 12 x 12 m)
172 were selected. At one plot biogas digestate and at another plot cattle slurry was
173 applied, whereas the third plot served as control (without fertilization). Centrally at
174 each plot, three PVC-collars for GHG measurements (inside dimension 75 x 75 cm;
175 0.5625 m²) were permanently inserted 10 cm into the soil with a distance of 1.5 m to
176 each other. To prevent oscillations of the peat through movements during the
177 measurements, boardwalks were installed. At each site a climate station was set up
178 in March 2010 for the continuous recording (every 0.5 hour; CR200X Datalogger,
179 Campbell Scientific) of air temperature and humidity at 20 cm above soil surface
180 (CS215-L, Campbell Scientific) and soil temperatures at the depth of -2, -5 and -10
181 cm (109-L, Campbell Scientific). For NH₃ measurements, sensors for wind speed and
182 wind direction (Kleinwindsensor, Thies Clima) in 2 m height were additionally
183 integrated from May to July 2011, with a logging frequency of 5 seconds (GP1, Delta-
184 T Devices). For measuring the ground water table, plastic perforated tubes (JK-
185 casings DN 50, 60 mm diameter, 1 m length) were inserted close to each collar to
186 obtain individual groundwater tables for all repetitions during each gas flux
187 measurement. In April 2010, we equipped one tube per plot with a water level logger
188 (Type MiniDiver, Schlumberger water services), which logged the water tables every
189 15 minutes. Additionally to the recorded data, plot-specific soil temperatures in three
190 soil depths (-2, -5 and -10 cm) were determined with penetration thermometers at
191 the beginning and end of each gas flux measurement.

192 In 2010 and 2011, organic fertilizers were applied via splash plate (swivelling slurry
193 spreader for biogas digestate; gooseneck scatterer for cattle slurry) on 14th June
194 2010, 25th August 2010, 27th Mai 2011, 22th September 2011 and 04th November
195 2011 by the landowners (see Table 1). The surface application technique via splash
196 plate is the most common application technique in the smallholder structure of the
197 region. The organic fertiliser was applied on the basis of equal volumetric rates per
198 application event (20 or 25 m³ ha⁻¹). This method is typical for farming practices, but
199 produces diverging N application rates per event between slurry and digestate based

200 on NH_4^+ or N_{tot} applications. It is known that the splash plate application technique
201 can result in very uneven spreading regarding the application rate and/or the
202 evenness. Both chosen spreading devices are known for the higher precision in their
203 application evenness compared to conventional splash plates (approximately 15%
204 and 18–27% application variability for swivelling slurry spreader and gooseneck
205 scatterer and up to 47% for conventional splash plate; Frick, 1999). In the present
206 study, the application of an equal volumetric slurry rate was controlled via the barrel
207 content and the tractor speed. At all plots, the tractor lane were 1 m in front of the
208 collars which were placed in a row with a distance of 1.5 m to each other. Both
209 spreading systems had a spreading width of 12 m and no overlapping zones
210 occurred. Nevertheless we can not give any estimation about the actually achieved
211 precision of the application evenness.

212 The physical and chemical composition of the slurries and digestates varied between
213 the four different application events (Table 3). Composition of organic fertilizers was
214 analysed from 1 L samples which were taken from the slurry tank in the field. Slurries
215 were immediately frozen at $-20\text{ }^\circ\text{C}$ until analysis which was conducted by the
216 AGROLAB Labor GmbH (Bruckberg, Germany). Due to technical problems at the
217 first application event, cattle slurry was applied by watering cans on the ground of the
218 collars and on a 120 m^2 adjacent area. To ensure an equal volumetric amount of
219 organic fertilizer a $1\times 1\text{ m}$ grid, built by cords, was previously installed. The same
220 method was used at the fourth application event for the digestate.

221 **2.3. N_2O and CH_4 flux measurements**

222 As a background, we measured fluxes of N_2O and CH_4 every second week from
223 January 2010 to January 2012 using the static manual chamber method (volume 309
224 L) (Livingston & Hutchinson; 1995). We removed, however, the gas fluxes measured
225 in 2010 from the data set due to errors in the gas chromatography analysis and due
226 to long vial storage. Intensive measurement campaigns were performed after the four
227 fertilisation events on 14th June 2010, 25th August 2010, 27th Mai 2011, and 22th
228 September 2011. Immediately after fertilization flux measurements were carried out
229 daily for a week and on every second day for another eight to nine days. To minimize
230 diurnal variation in the flux pattern, sampling was always carried out between 9.00
231 a.m. and 11.30 a.m. A detailed description of chamber dimensions and configuration
232 is given in Drösler (2005). Four gas samples were taken at four regular time intervals
233 after chamber closure (enclosure time 60 min). The samples were collected in 20 ml

234 glass vials, each sealed with a butyl rubber septum. The vials were flushed with
235 chamber air for 30 seconds using a portable micro pump (KNF Neuberger GmbH,
236 NMP015B), so that the air in the vials was exchanged 32 times. In addition the pump
237 was used to build up an overpressure of approximately 550 mbar to protect the
238 sample against fluctuations in atmospheric pressure during storage. Gas analyses
239 were carried out with a gas chromatograph (Perkin & Elmer, Clarus 400 GC
240 respectively Clarus 480 GC) equipped with a headspace auto sampler (Perkin &
241 Elmer, TurboMatrix 110), a PoraPack 80/100 mesh column, an electron capture
242 detector (ECD) for N₂O (ECD temperature 380°C) and a flame ionization detector
243 (FID) for CH₄ analyses (FID temperature 310°C). Gas samples from the first
244 fertilization event (14th June to 30th June of 2010) were immediately analysed at the
245 Max Planck Institute for Biogeochemistry in Jena, whereas samples from the second
246 fertilization event (25th August to 10th September of 2010) were analysed at the
247 Thünen Institute in Braunschweig with a Varian CP-3800 GC-FID/-ECD using a
248 headspace autosampler (QUMA Elektronik & Analytik GmbH, Germany) and similar
249 conditions. Gas flux rates were calculated from the linear change in gas
250 concentration over time considering chamber air temperature and atmospheric
251 pressure. Gas fluxes were accepted when the linear regression was significant ($P \leq$
252 0.05). In case of small N₂O or CH₄ fluxes, fluxes were also accepted if the coefficient
253 of determination was ≥ 0.90 and the regression slope was between -1 and 1 ppb
254 min⁻¹. The cumulative annual mean exchange rate was calculated by linear
255 interpolation between the measurement dates.

256 **2.4. NH₃ flux measurements**

257 Ammonia volatilization was measured at the third organic fertilizer application event
258 on 27th of May 2011. Measurements were performed immediately after fertilizer
259 application and thereafter in irregular time intervals of few hours (in total 96
260 measurements). For NH₃ measurements we used the calibrated dynamic chamber
261 method ('Dräger-Tube Method'; DTM) which was described in detail bei Pacholski et
262 al. (2006). One day before application, eight stainless steel rings (104 cm²) were
263 inserted into the upper soil (3 cm) at each plot, from which four were grouped close
264 together. Ambient air was sucked with a defined flow rate (1 L min⁻¹) through four
265 (via teflon tubes) connected conical stainless steel chambers to an ammonia
266 indicator tube (Drägerwerk AG, Lübeck, Germany). The NH₃ volume concentration
267 was corrected for air temperature and air pressure (Pacholski et al., 2006). To

268 prevent overestimation of NH₃ volatilization through NH₃ enriched ambient air from
269 surrounding area, ammonia concentration from the control treatments were
270 subtracted from the fertilized treatments prior to NH₃ flux calculation. Different studies
271 report a distinct underestimation of up to one order of magnitude of NH₃ fluxes
272 determined by the DTM, mainly due to the low air exchange rate in the chambers
273 (Roelcke, 2002; Pacholski et al., 2006). To avoid underestimation of cumulative NH₃-
274 N losses determined by the DTM, Pacholski et al. (2006) developed the following
275 calibration formula to correct the NH₃ fluxes:

276

$$277 \ln(NH_3 flux_{IHF}) = 0.444 * \ln(NH_3 flux_{DTM}) + 0.590 * \ln(v_{2m}) \quad (1)$$

278

279 where $NH_3 flux_{IHF}$ is NH₃ flux measured by the integrated horizontal flux method (kg N
280 ha⁻¹ h⁻¹); $NH_3 flux_{DTM}$ is NH₃ flux measured by the DTM (kg N ha⁻¹ h⁻¹); v_{2m} wind
281 speed at 2 m height (m s⁻¹). Quakernack et al. (2011) compared the DTM method
282 with the frequently used micrometeorological method, concluding that the corrected
283 DTM method also allows quantitative NH₃-loss measurements. The total cumulative
284 NH₃ volatilization was estimated by curve fitting and integration of the area obtained
285 by the fitted curve between time zero and the time point where the NH₃ flux was zero.

286 **2.5. Grass yield, apparent N use efficiency and N-balances**

287 The annual yield was determined by harvesting the grass inside the PVC-collars with
288 a scissor at each mowing event (same cutting height as the farmer, at about 5 cm;
289 sample area = 0.5625 m²). There was no visible disturbance from trace gas
290 measurements in the collars. The grassland parcel showed a strong spatial
291 heterogeneity in grass yield so that only sampling inside the collars allowed to relate
292 grass yield and N uptake with N₂O emissions. Mowing events took place on 24th Mai
293 2010, 20th August 2010, 23th Mai 2011, 01st August 2011 and 13th September 2011
294 (see Table 1). To determine the dry mass (DM), grass samples were oven dried at
295 60°C for 48 hours. To determine the total carbon (C_{tot}) and total nitrogen (N_{tot})
296 concentrations of plant biomass, dried grass samples were milled (0.5 mm) and
297 mixed sub samples were analysed according to DIN ISO 10694 and DIN ISO 13878
298 by the AGROLAB Labor GmbH (Bruckberg, Germany). The apparent N_{tot} or rather
299 N_{min} use efficiency (NUE, NUE_{min}) was calculated as:

300

301
$$NUE \text{ or } NUE_{\min} = \left(\frac{N \text{ uptake}_{\text{treatment}} - N \text{ uptake}_{\text{control}}}{\text{total } N \text{ applied}} \right) * 100\% \quad 2)$$

302

303 where $N \text{ uptake}_{\text{treatment}}$ is the amount of N taken up by the plants in the fertilized
 304 treatments, $N \text{ uptake}_{\text{control}}$ is the amount of N taken up by the plants in the unfertilized
 305 control, and $\text{total } N \text{ applied}$ is the amount of N_{tot} or N_{min} applied, corrected by $\text{NH}_3\text{-N}$
 306 losses. Emission factors for NH_3 were taken from the German national greenhouse
 307 gas inventory (Haenel et al., 2014; Table 4.6 – broadcast, grassland, EF of 0.60 kg
 308 kg^{-1} related to applied $\text{NH}_4^+\text{-N}$), whereas for the third application event own
 309 estimated values based on our measurements were used.

310 Based on the measured gaseous N fluxes, the N uptake by plants and soil N_{min}
 311 contents a simple N balance was calculated as followed:

312

313
$$N \text{ balance} = (N \text{ applied} + (N_{\text{min}_{t_2}} - N_{\text{min}_{t_1}}) + N_{\text{dep}}) - (N \text{ uptake} + N_2O_{\text{cum}} + NH_{3\text{cum}}) \quad (3)$$

314

315 where $N \text{ applied}$ is the amount of N_{tot} applied, $N_{\text{min}_{t_1}}$ and $N_{\text{min}_{t_2}}$ are the amounts of
 316 N_{min} at time 1 (06th April 2011; early April represents the beginning of the vegetation
 317 period in 2011) and time 2 (18th October 2011; end of October represents the end of
 318 the vegetation period in 2011) for the soil depth 0–20 cm, N_{dep} is the annual
 319 atmospheric N deposition, $N \text{ uptake}$ is the amount of N taken up by the plants
 320 (quantified in harvested biomass), N_2O_{cum} is the amount of the annual cumulative
 321 $\text{N}_2\text{O-N}$ losses, and $\text{NH}_{3\text{cum}}$ is the amount of the annual cumulative $\text{NH}_3\text{-N}$ losses.

322 **2.6. Soil sampling and laboratory analyses**

323 For the determination of mineral N ($N_{\text{min}} = \text{NH}_4^+\text{-N} + \text{NO}_3^-\text{-N}$) contents, one mixed
 324 soil sample consisting of nine individual samples was collected at two soil depths (0–
 325 10, 10–20 cm) at each plot during every gas flux measurement. Samples were
 326 immediately cooled and stored in an ice box before analyses. Mineral N was
 327 extracted after shaking 40 g of fresh soil with 160 ml CaCl_2 (0.0125 M) for one hour.
 328 The extracts were filtered through a 4–7 μm filter paper (Whatman 595 ½) and the
 329 first 20 ml of the extract were discarded. The solution was frozen at $-20 \text{ }^\circ\text{C}$ until
 330 analysis, which was conducted by the AGROLAB Labor GmbH (Bruckberg,
 331 Germany). A subsample of 20–30 g was used to determine the gravimetric water
 332 content, which was taken into account for the calculation of mineral N concentrations.

333 For determination of C_{tot} and organic carbon (C_{org}) a mixed soil sample of nine
334 individual samples was collected close to each collar at two soil depths (0–10, 10–20
335 cm) using a 3 cm diameter auger. After drying for 72 hours at 40 °C, soil samples
336 were sieved to 2 mm to remove stones and living roots. Analyses were conducted at
337 the Division of Soil Science and Site Science (Humboldt Universität zu Berlin,
338 Germany). For the determination of bulk density and porosity, three undisturbed core
339 cutter samples (100 cm³) were randomly taken at four depths (0–5, 5–10, 10–15, 15–
340 20 cm) for each plot.

341 **2.7. Statistical analysis**

342 Statistical analyses were conducted using R 2.12.1 (R Development Core Team,
343 2010). We used analysis of variance (ANOVA) (for grass yield, 16 days cumulative
344 N₂O emissions and treatment NO₃⁻ comparison) or the nonparametric Kruskal-Wallis
345 Rank Sum test (for GW level) to compare means of samples. In case of significant
346 differences among the means, we used Tukey's honest significant differences
347 (TukeyHSD) or the non-parametric Pairwise Wilcoxon Rank Sum test with Bonferroni
348 correction for multiple comparisons. For testing two independent sample means, we
349 use the Welch two sample t-test (for soil type NO₃⁻ comparison in 2010) or the non
350 parametric Mann-Whitney U-test (for soil type NO₃⁻ comparison in 2011). For time
351 series data (N₂O, CH₄ field measurements) we applied linear mixed effects models
352 (Crawley 2007; Eickenscheidt et al., 2011; Hahn-Schöfl et al., 2011). We set up a
353 basic model with soil type and fertilizer treatment as fixed effects and the spatial
354 replication (individual collar) nested in time as random effect. Non-significant terms
355 were removed from the fixed structure. We extended the basic model by a variance
356 function when heteroscedasticity was observed. In case of significant serial
357 correlation in data, a moving average or a first-order temporal autoregressive
358 function was included in the model. Autocorrelation was tested using the Durbin-
359 Watson test and by plotting the empirical autocorrelation structure. The model
360 extension was proved by the Akaike Information Criterion (AIC). For multiple
361 comparisons we conducted Tukey contrasts using the General Linear Hypotheses
362 function from the "multcomp" package (Hothorn et al., 2013).

363 The assumption of normality of residuals was tested using the Lilliefors or Shapiro-
364 Wilk test and by plotting the Quantile-Quantile plots. Homogeneity of variances of
365 residuals was checked using the Levene or Breusch-Pagan test and by plotting the
366 residuals against the fitted values. Where necessary, data were box-cox transformed

367 prior to analyses. We used simple and multiple linear or non-linear regressions
368 models to explain N₂O, CH₄ and NH₃ fluxes. We accepted significant differences if P
369 ≤ 0.05. Results in the text are given as means ± 1 standard deviation.

370 **3. Results**

371 **3.1. Environmental drivers**

372 Temperatures between the two investigated soil types did not differ. In 2010 and
373 2011, air temperature in 20 cm height ranged from -17.5 to 39.5°C with an annual
374 mean of 8.6°C in 2011 at both investigated areas. Soil temperature in -2 cm soil
375 depth averaged 10.3°C at the C_{org}-medium sites and was slightly higher with 10.5°C
376 at the C_{org}-high sites in 2011. Air temperature in 20 cm height following 15 or 16 days
377 after fertilization averaged 16.0, 13.1, 15.4 and 11.5°C for application events one to
378 four at both investigated soil types. Soil temperature in -2 cm soil depth was
379 approximately 2°C above the mean air temperature in the same periods at both soil
380 types. In 2010 and 2011 annual precipitation was 850 and 841 mm, which was
381 slightly above the 30-years mean of the period 1961–1990. Figure 1 shows the
382 precipitation following the fertilizer application. With the exception of the third
383 application event, no rainfall occurred during the application of the organic fertilizers.
384 However, precipitation during and after the third application event was only weak and
385 amounted to 3 mm in the time span between 16:00 and 00:00 hours.

386 All treatments showed similar dynamics in their annual hydrographs (Fig. 2a) but
387 mean annual groundwater levels of the C_{org}-high treatments were significantly higher
388 (all *P* < 0.001) compared to the C_{org}-medium treatments in 2010 and 2011 (Table 4).
389 Mean groundwater levels following the fertilizer applications are shown in Table 4.

390 **3.2. N input and N availability**

391 The amount of N applied was 111 and 252 kg N ha⁻¹ for slurry treatments or rather
392 101 and 174 kg N ha⁻¹ for digestate treatments in 2010 and 2011, respectively.
393 However, due to the distinctly higher NH₄⁺-N/N_{tot} ratio of the biogas digestate, total
394 NH₄⁺-N input was comparable (2010) or slightly higher (15% more NH₄⁺-N; 2011)
395 than for the slurry treatments (Table 3). Additional physical and chemical properties
396 of the slurry and digestate are shown in Table 3.

397 The extractable N_{min} contents of the soils were dominated by NO₃⁻ whereas NH₄⁺
398 was only of minor importance especially at the C_{org}-medium sites (Fig. 2b and 2c).

399 The NO_3^- content was significantly higher ($P < 0.001$) at the C_{org} -high sites than at
400 the C_{org} -medium sites in 0–10 cm soil depth in both years and in 10–20 cm soil depth
401 in 2010 ($P < 0.01$) (Table 4). With exception of the first application event, all
402 fertilization events increased the NO_3^- contents of the soil for a short period (Fig. 2c,
403 Table 4). However, only in 2011 the fertilized sites showed significantly ($P < 0.01$)
404 higher NO_3^- contents compared to the control treatments, but differences between
405 digestate and slurry were generally not significant (except for 0–10 cm soil depth at
406 the C_{org} -medium site) (Table 4).

407 **3.3. N_2O emissions**

408 Nitrous oxide fluxes were generally low at all treatments (Fig. 2d). Background
409 emissions rarely exceeded $50 \mu\text{g N m}^{-2} \text{h}^{-1}$. Highest N_2O fluxes were found
410 immediately after fertilizer application (Fig. 2d and 3), sometimes followed by a
411 second phase of higher emissions after 6 to 12 days. In case of the C_{org} -medium
412 sites N_2O fluxes returned to background emission level within 3 to 7 days, whereas
413 the C_{org} -high sites had longer lasting increased N_2O emissions, particularly at the
414 digestate treatment.

415 Short term (16 days) N_2O fluxes of fertilized treatments significantly ($P < 0.01$)
416 exceeded N_2O fluxes of control treatments at all fertilization events. However, only in
417 one out of four fertilization events short term N_2O fluxes were significantly ($P < 0.001$)
418 higher at the digestate treatments compared to the slurry treatments. Additionally
419 significantly ($P < 0.001$) higher short term N_2O fluxes were observed at the C_{org} -high
420 sites compared to the C_{org} -medium sites in 2011, but the opposite was observed at
421 the second fertilization event in 2010.

422 However, due to the high variability and the partially fast return to the background
423 emission level, short term (16 days) cumulative N_2O emissions were not significantly
424 different from the control treatments in 2010 (Fig 4), but for 2011 short term
425 cumulative N_2O emissions had a clear trend in the order digestate > slurry > control
426 (although not significant in one case).

427 On an annual basis organic fertilization led to significantly ($P < 0.001$) higher N_2O
428 fluxes compared to unfertilized treatments. Additionally, the application of biogas
429 digestate significantly ($P < 0.01$) enhanced the N_2O fluxes compared to the
430 application of cattle slurry. Furthermore, N_2O fluxes from the C_{org} -high site
431 significantly ($P < 0.001$) exceeded N_2O fluxes from the C_{org} -medium sites. Annual
432 cumulative emissions ranged from $0.91 \pm 0.49 \text{ kg N ha}^{-1} \text{ yr}^{-1}$ (control treatment, C_{org} -

433 medium site) to $3.14 \pm 0.91 \text{ kg N ha}^{-1} \text{ yr}^{-1}$ (digestate treatment, C_{org} -high site) (Table
434 5). Calculated emission factors (EF) based on the amount of N_{tot} ranged from 0.12 to
435 0.23 for the slurry treatments and from 0.55 to 1.13 for the digestate treatments
436 (Table 5).

437 Observed collar specific cumulative annual N_2O fluxes were strongly related to collar
438 specific annual plant N-uptake (Fig. 5a and 5b). 53% of the temporal and spatial
439 variation in the 16 days cumulative N_2O -N exchange rates was explained by the
440 amounts of applied NH_4^+ -N and the mean groundwater levels below surface during
441 the same time (Fig. 6). A similar trend was additionally observed for the annual
442 cumulative N_2O emissions but regression analysis was not possible due to the small
443 sample size ($n = 6$).

444 **3.4. CH_4 emissions**

445 Most of the time, CH_4 emissions could not be detected (Fig. 2e). Occasionally CH_4
446 peaks were only found immediately after cattle slurry application. However, with
447 exception of the slurry treatment of the C_{org} -high site at the first application event, the
448 organic fertilization did not result in significantly different short term (15 or 16 days)
449 CH_4 fluxes between the treatments or the investigated soil types. The observed weak
450 CH_4 emissions or uptakes amounted to cumulative annual CH_4 exchange rates of
451 $-0.21 \pm 0.19 \text{ kg C ha}^{-1} \text{ yr}^{-1}$ to $-1.06 \pm 0.46 \text{ kg C ha}^{-1} \text{ yr}^{-1}$. Significantly different CH_4
452 fluxes between the investigated treatments or the different soil types could not be
453 observed regarding the annual fluxes in 2011.

454 **3.5. NH_3 volatilisation**

455 Highest NH_3 losses were observed immediately after fertilization (Fig. 7). During the
456 first 24 hours, 64% and 100% of total NH_3 losses occurred at the digestate and slurry
457 treatments, respectively. Since differences in the response of NH_3 volatilization were
458 not significant, treatment data were pooled by soil type prior to regression analysis.
459 The total NH_3 loss following application was $18.17 \text{ kg N ha}^{-1}$ for the digestate
460 treatments and $3.48 \text{ kg N ha}^{-1}$ for the slurry treatments. The relative N loss was 36%
461 and 15% of applied NH_4^+ -N, or 23% and 5% of total applied N for the digestate and
462 slurry treatments, respectively.

463 **3.6. Grass yield, apparent N use efficiency and estimated N balances**

464 In 2010 and 2011, the mean annual grass yield ranged from 4.5 (control C_{org}-medium)
465 to 13.1 t DM ha⁻¹ yr⁻¹ (digestate C_{org}-high) (Table 6). In both years the mean annual
466 grass yield from the digestate treatments were significantly (P < 0.05) higher
467 compared to the slurry treatments. This pattern was also found in the annual plant N-
468 uptake which showed a clear partitioning between the treatments investigated (Fig.
469 5a). Additionally, the mean annual grass yield from the C_{org}-high sites exceeded
470 those from the C_{org}-medium sites of both years, but differences were not significant.

471 The application of biogas digestate distinctively increased apparent NUE and NUE_{min}
472 compared to cattle slurry treatments (Table 6). NUE values were on average 115 ±
473 136% for biogas digestate treatments and 24 ± 23% for cattle slurry. NUE_{min} values
474 were always >100% for biogas digestate treatments, whereas for cattle slurry NUE_{min}
475 values averaged 74 ± 68% in 2011 but were >100% in 2010. Beside fertilizer type
476 effects, higher NUE and NUE_{min} were observed at the C_{org}-medium site compared to
477 the C_{org}-high site.

478 The estimated N balances revealed N surpluses of up to 70 kg N ha⁻¹ yr⁻¹ for cattle
479 slurry treatments but deficits of up to 108 kg N ha⁻¹ yr⁻¹ for biogas digestate
480 treatments, for the year 2011 (Table 7).

481 **4. Discussion**

482 **4.1. Drainage and fertilizer effects on N-availability and N-** 483 **transformation**

484 Mineral nitrogen contents were consistently higher at the C_{org}-high treatments than at
485 the C_{org}-medium treatments, in line with the considerably higher amount of SOC and
486 N at this site. It is well known that drainage enhances the degradation of SOM and
487 thus stimulates net nitrogen mineralization and N transformation processes (Kasimir
488 Klemedtsson et al., 1997; Freibauer et al., 2004; Klemedtsson et al., 2005; Goldberg
489 et al., 2010). Various studies reported an annual N supply through peat
490 mineralization of 70 to 292 kg N ha⁻¹ yr⁻¹ (Schothorst, 1977; Flessa et al., 1998;
491 Sonneveld and Lantinga, 2011). It can be assumed that at a comparable aeration
492 status and temperature, mineralization processes are more intensive at peatlands
493 which were recently drained (Hacin et al., 2001; Renger et al., 2002; Sonneveld and
494 Lantinga, 2011) or contain higher amounts of SOM.

495 As expected from literature the biogas digestates differed in their physical and
496 chemical properties from the cattle slurries. In the present study, the biogas
497 digestates had narrower C/N ratios (see also e.g. Tambone et al., 2009), higher pH
498 values (see also Wulf et al., 2002a and Quakernack et al., 2011), wider $\text{NH}_4^+/\text{N}_{\text{tot}}$
499 ratios and thus relatively higher NH_4^+ contents than the cattle slurries (see also Möller
500 and Stinner 2009). The amounts of NH_4^+ -N were not distinctly different between the
501 applied organic fertilizers but in 2011 biogas treatments received 15% more NH_4^+ -N
502 compared to cattle slurry treatments.

503 We observed an unexpected small change in the NH_4^+ content of the soil
504 immediately after fertilizer application which can be attributed to different reasons.
505 Firstly, the fertilizers partly remained on the plant canopy after splash plate
506 application and therefore soil contact and infiltration was limited (Quakernack et al.,
507 2011). Secondly, a significant fraction of NH_4^+ from the organic fertilizer was lost in a
508 few hours after splash plate application via NH_3 volatilization. Thirdly and probably
509 most important, in well aerated soils applied NH_4^+ underwent rapid nitrification, as
510 was indicated by the increasing soil NO_3^- contents in the upper soil layer after
511 fertilizer application. In general, the continuously observed absent or low NH_4^+
512 contents with simultaneously high extractable NO_3^- in the soil indicate that net
513 nitrification entirely controlled net nitrogen mineralization at all treatments of the
514 investigated study sites. Nitrification requires sufficient oxygen (O_2) availability in the
515 soil (Davidson et al., 1986) hence we can assume well aerated soil conditions, at
516 least in the upper soil layer, for most of the time at the study sites.

517 Several studies (e.g. Gutser et al., 2005; Jones et al., 2007) reported that the
518 infiltration of organic fertilizer may enhance the soil N pool and further stimulates the
519 SOM mineralization, leading to additional N_{min} . This becomes also evident in the
520 observed significantly higher NO_3^- contents of the fertilized treatments compared to
521 the unfertilized control treatments, especially in the 0–10 cm soil layer. However,
522 significant differences in the N_{min} contents between the two investigated organic
523 fertilizers were not found in 2010 and 2011. This may be due to the fact that the N
524 uptake by plants at the digestate treatments was on average 27% higher and that
525 distinct differences in the amount of N_{tot} and NH_4^+ of the applied organic fertilizers
526 were only observed in the second study year.

527 To maintain soil fertility and yield and to reduce harmful side effects (e.g. N_2O losses,
528 NO_3^- leaching) site adapted fertilization is necessary. The estimated negative N

529 balances for biogas treatments are in line with Andres et al. (2013) who reported that
530 positive N balances could only be achieved when the amount of applied digestate
531 contains more than 200 kg N ha⁻¹ yr⁻¹. However, the strong negative N balances of
532 the control treatments reveal that large amounts of up to 148 kg N ha⁻¹ yr⁻¹ originate
533 from peat mineralization, demonstrating the unsustainable agricultural use of drained
534 peatlands. Assuming that the fertilized treatments received equal amounts of N from
535 peat mineralization, all N balances of these treatments were strongly positive. N
536 surpluses as estimated for the cattle slurry treatments enhance the soil N pool, but
537 the gradual release of N at a non predictable stage from the soil N pool carries the
538 risk of leaching or gaseous losses (Amon et al., 2006). Particularly in wintertime, high
539 amounts of available NO₃⁻ in the soil, as observed especially at the fertilized
540 treatments of the C_{org}-high sites, carry the risk of N leaching due to the reduced N
541 demand by plant uptake and by the microbial community during this period (Merino et
542 al., 2002; Sanger et al., 2010).

543 **4.2. Fertilizer effect on N-use efficiency and grass yield**

544 In line with investigations from Schils et al. (2008) most of the applied and produced
545 N_{min} was probably rapidly absorbed by the grassland as the soil N_{min} content usually
546 decreased within a few days after fertilizer application (Figure 2b, 2c). This becomes
547 also evident in the apparent NUE_{min}, especially from biogas digestate treatments. A
548 significant effect of biogas digestate on crop yields and apparent NUE_{min} as observed
549 in the present study was also reported from pot experiments (e.g. de Boer, 2008;
550 Moller and Muller, 2012), but not for field applications without incorporation of the
551 digestate into the soil (Moller and Muller, 2012). According to de Boer (2008) the
552 higher NUE_{min} at digestate treatments can be attributed to the wider NH₄⁺/N_{tot} ratio as
553 well as to the narrower C/N ratio of the applied digestate. Thus more N was
554 immediately available for plant growth after fertilization (Amon et al., 2006; Sanger et
555 al., 2010), whereas the lower C/N ratio reduced the potential for microbial
556 immobilization of applied N (Velthof et al., 2003, de Boer, 2008). We hypothesized
557 that the application of biogas digestate leads to a significantly higher grass yield and
558 N-use efficiency compared to the application of cattle slurry due to the higher N
559 availability of the digestate. This could partly be confirmed, but the much higher grass
560 yields from biogas digestate treatments cannot solely be explained by differences in
561 applied NH₄⁺, since differences were only small. However, the much higher N-uptake
562 at the digestate treatments (Fig. 5a) indicated that much more N must have been

563 available at these treatments. Many studies have shown that the utilization of N
564 derived from organic fertilizer is relatively small in the year of application, due to the
565 slow release of organically bound N (Jensen et al., 2000; Sørensen and Amato, 2002;
566 Gutser et al., 2005). The consistently higher NUE_{min} of $> 100\%$ at the digestate
567 treatments indicates that some organic N derived from the fertilizer or from the SOM
568 pool has been mineralized (Gunnarsson et al., 2010). Since the digestate is
569 considered as more recalcitrant (Clemens and Huschka, 2001; Oenema et al., 2005;
570 Möller and Stinner, 2009), it can be assumed that the digestate enhanced SOM
571 mineralization more than cattle slurry, or that N mineralized from SOM or fertilizer
572 had a larger share in the uptake by the plants due to lower competition of microbial
573 immobilization as was reported by Gutser et al. (2010). Probably the assumed higher
574 SOM mineralization at the digestate treatments could partly be related to a priming
575 effect since the higher biomass production probably caused a higher release of root
576 exudates, containing easily available C and N which enhanced microbial activity
577 (Mounier et al. 2004; Henry et al. 2008) promoting SOM mineralisation. However,
578 further investigations are needed to prove this explanation. The lower NUE at the
579 C_{org} -high sites compared to C_{org} -medium sites reveals that plants are more
580 independent of N input by fertilizer with increasing SOM at drained fen peatlands due
581 to the extra N_{min} derived from enhanced mineralization processes, as mentioned
582 before.

583 **4.3. Fertilizer and site induced N_2O emissions**

584 The observed annual N_2O emissions were distinctly lower than the actual default
585 emission factor from the Tier 1 approach for temperate, deep drained, nutrient rich
586 grassland of $8.2 \text{ kg N}_2\text{O-N ha}^{-1} \text{ yr}^{-1}$ (IPCC, 2014) and at the lower end of literature
587 values from other organic soils. Studies from Germany reported much higher N_2O
588 emissions, ranging from 1.15 to $19.8 \text{ kg N ha}^{-1} \text{ yr}^{-1}$ (; Flessa et al., 1997; Augustin et
589 al., 1998; Flessa et al., 1998; Beetz et al., 2013). Also investigations from other
590 European countries showed that much higher N_2O emissions can be released from
591 grasslands on drained peatlands. For example, Velthof et al. (1996) and van Beek et
592 al. (2010; 2011) reported N_2O emissions, ranging from 4.2 to $41.0 \text{ kg N ha}^{-1} \text{ yr}^{-1}$
593 for the Netherlands, whereas at boreal regions N_2O emissions of up to $9 \text{ kg N ha}^{-1} \text{ yr}^{-1}$
594 were measured (Nykänen et al., 1995; Maljanen et al., 2004; Regina et al., 2004).
595 The observed N_2O emissions were also in the range of those reported from
596 grasslands on mineral soils in Germany, summarized by Jungkunst et al. (2006). In

597 line with our results, Flessa et al. (1998) also found that N₂O losses from peat soils
598 are not always larger than from nearby mineral soils, but in contrast, Maljanen et al.
599 (2010) found on average four times higher N₂O emissions from cultivated organic
600 soils than from mineral soils. The N₂O emissions from the C_{org}-high sites significantly
601 exceeded those from the C_{org}-medium sites in all treatments, which was in line with
602 higher N_{min} contents and higher groundwater levels. Additionally, the observed
603 distinctly stronger increase of cumulative annual N₂O emissions with increasing N-
604 uptake by plants (Fig. 5b) reveals that with increasing N availability a higher
605 proportion is lost as N₂O at the C_{org}-high sites compared to the C_{org}-medium sites.
606 This probably could be attributed to the more favorable soil conditions for
607 denitrification, due to higher C and N mineralization rates and alternating
608 groundwater levels, promoting anaerobicity (Koops et al., 1996). Moreover, as
609 mentioned before, net nitrification entirely controls net nitrogen mineralization,
610 promoting also N₂O losses, but probably to a lesser extent. However, the source of
611 N₂O production in soils is often uncertain because aerobic and anaerobic micro-sites
612 can occur within close proximity and thus nitrification and denitrification as well other
613 abiotic processes producing N₂O (e.g. nitrifier-denitrification, coupled nitrification-
614 denitrification) can run simultaneously (Davidson et al., 1986; Butterbach-Bahl et al.,
615 2013). Despite surprisingly low N₂O emission levels, we confirmed our hypothesis
616 that N₂O emissions increase with increasing soil C_{org} content probably due to more
617 favorable conditions for denitrification.

618 The observed background emissions on the two organic soils correspond well to
619 those on mineral agricultural soils (Bouwman, 1996). However, calculated emission
620 factors as percentage of applied N without consideration of the NH₃ losses were
621 lower for all treatments than the IPCC default value. Several other studies also
622 reported emission factors < 1% of applied N (Chadwick et al., 2000; Velthof et al.,
623 2003; Clemens et al., 2006; Jones et al., 2007; Möller and Stinner, 2009), but so far
624 not for organic soils. Indeed, N₂O studies on organic soils rarely differentiate between
625 fertilizer and soil derived N sources by unfertilized control plots as we do in this study.
626 In line with Möller and Stinner (2009) the application of biogas digestate resulted in a
627 distinctly higher percentage of N₂O produced from applied N, compared to cattle
628 slurry, yet at a low level.

629 One reason of generally low N₂O emissions observed in the present study could be
630 the small number of frost–thaw cycles in 2011. In general frost–thaw cycles are

631 considered to favor high N₂O emissions (Flessa et al., 1998, Jungkunst et al., 2006)
632 but these observations seem to be more pronounced for croplands than for
633 grasslands in Germany (Dechow and Freibauer, 2011). Denitrification activity is
634 strongly related to the NO₃⁻ content close to the groundwater level (van Beek et al.,
635 2004). Given the high NO₃⁻ contents, in particular in the C_{org}-high soil, the evidence
636 for fast nitrification and high net nitrogen mineralization, we argue that frequent but
637 low dosage application of fertilizer and quick N uptake by plants avoid conditions
638 favorable for high N₂O emissions. Moreover through the splash plate application
639 technique high amounts of NH₄⁺ were rapidly lost as NH₃, and therefore reduced the
640 proportion of immediately available N for nitrification and denitrification.

641 As expected from the literature, highest N₂O fluxes were found immediately after
642 fertilizer application. The initial N₂O peak could mainly be attributed to the
643 denitrification of available soil NO₃⁻, presumably due to the more favorable conditions
644 for denitrification through the addition of easily degradable organic C and water
645 (Comfort et al., 1990; Chadwick et al., 2000; Velthof et al., 2003). Additionally, a
646 probably smaller part of initial N₂O could be ascribed to the rapid nitrification
647 (Chadwick et al., 2000) or to nitrifier denitrification of slurry NH₄⁺. In contrast, the
648 partially observed second N₂O peak, mostly found a week after fertilizer application,
649 can be attributed to the denitrification of mineralized and nitrified organic components
650 of fertilizer N (Velthof et al., 2003).

651 Several authors proposed that the more recalcitrant digestate might reduce the rate
652 of microbial degradation and oxygen consumption in the soil, thus resulting in
653 reduced N₂O emissions through less anaerobic soil conditions (Clemens and
654 Huschka, 2001; Oenema et al., 2005; Möller and Stinner, 2009). In contrast, our
655 study on organic soils found significantly higher N₂O emissions from the digestate
656 treatments compared to the slurry treatments. Higher N₂O emissions derived from
657 biogas digestates were also reported from a few other authors (e.g. Senbayram et al.,
658 2009; Sanger et al., 2010), whereas Clemens et al. (2006) found no differences
659 between untreated and digested slurry.

660 It can be assumed that at drained organic soils, like in the present study, sufficient
661 metabolizable C is generally widely available in the upper soil profile (e.g. van Beek
662 al., 2004). Thus, as hypothesized, labile carbon is not limiting on organic soils. This
663 was in line with Velthof et al. (2003) who supposed that the application of available C
664 with the organic fertilizer has a larger effect on denitrification activity at soils with a

665 lower C_{org} content compared to C_{org} rich soils. However, contrary to our hypothesis
666 the significantly higher N_2O emissions from the digestate treatments can not solely
667 be explained by the higher content of available N in the biogas digestate, since the
668 amount of applied NH_4^+ -N in the substrate was not distinctively different.
669 Nevertheless, as mentioned before, the much higher N-uptake at the biogas
670 treatments (Fig. 5a) indicates that the application of this fertilizer resulted in a
671 distinctively higher N availability, promoting N_2O production. It could be assumed that
672 the high pH and the lower C/N ratio of the biogas digestate, obviously slightly
673 enhanced SOM mineralization probably due to increased microbial activity compared
674 to cattle slurry fertilizer, leading to extra N for nitrification and denitrification.
675 Moreover, as proposed before it is also conceivable that the higher biomass
676 production at these treatments itself is related to the increased N_2O emissions, due to
677 the stimulating effect of plant roots on denitrification activity (Klemedtsson et al. 1987;
678 Bakken, 1988). Considering that increasing biomass production means increasing
679 root growth and activity it could be assumed that exudation of easily available organic
680 C and N (Hailer and Stolp, 1985), as well as the O_2 demand due to root respiration is
681 higher at the digestate treatments, promoting more anaerobic microsites and thus
682 denitrification compared to cattle slurry (Erich et al. 1984; Klemedtsson et al. 1987).
683 Furthermore several authors have suggested that root exudates may increase
684 bacterial metabolism (Klemedtsson, 1987; Mounier et al. 2004; Henry et al. 2008),
685 further lowering the oxygen concentration and thus increasing denitrification
686 (Klemedtsson et al., 1987). However, the enhanced biomass production
687 simultaneously should have depleted mineral N in the soil and thus reduced available
688 N for nitrification and denitrification processes. Obviously, despite the negative
689 apparent N balance of the biogas digestate treatment, there was no real nitrogen
690 competition between plants and microbes. However, further investigations are
691 required to prove whether digestates enhanced SOM mineralization or to what extent
692 increased biomass production favors N_2O emissions and which N pathways and
693 processes are involved.

694 The observed linear increase in the cumulative N_2O -N emissions during the first 16
695 days or annual N_2O emissions, due to a higher mean groundwater level and a higher
696 application rate of NH_4^+ -N reveal the importance of site adapted N fertilization and
697 the avoidance of N surpluses during agricultural use of C_{org} rich grasslands. The
698 same was also postulated for mineral soils by Ruser (2010).

699 **4.4. Fertilizer and site induced CH₄ emissions**

700 The observed consumption rates of CH₄ were in the range of CH₄ uptakes reported
701 by Flessa et al. (1998) for two different meadows in a southern German fen peatland.
702 Slightly higher CH₄ emissions of up to 1.46 kg CH₄-C ha⁻¹ yr⁻¹ were reported from
703 Beetz et al. (2013) for a drained intensive grassland in northern German and from
704 Nykänen et al. (1995) for a drained grassland in Finland. It is known that drainage
705 turns peatlands from a significant source back to a sink of CH₄ (Crill et al., 1994). In
706 peatlands the position of the groundwater table is considered as the key factor
707 regulating methanogenic and methanotrophic processes (Whalen, 2005). In line with
708 this, Flessa et al. (1998) showed that the consumption rate of CH₄ increased with
709 lowering of the groundwater level. Nevertheless, significant differences in the amount
710 of the annual CH₄ uptake capacity between the two study sites C_{org}-medium and C_{org}-
711 high could not be seen, although distinct differences in the groundwater table were
712 observed.

713 The occasionally observed CH₄ peak emissions were only found immediately after
714 cattle slurry application. This was in line with several other studies which reported
715 short-term CH₄ emissions immediately after organic fertilizer application due probably
716 to volatilization of dissolved CH₄ from the applied substrate (Sommer et al., 1996;
717 Chadwick et al., 2000; Wulf et al., 2002a; Jones et al., 2005; Amon et al., 2006). The
718 longer lasting CH₄ emissions observed after the first application event at the slurry
719 treatment of the C_{org}-high site might result from the degradation of volatile fatty acids
720 by methanogenic bacteria (Chadwick et al., 2000; Wulf et al., 2002a). Furthermore,
721 the high groundwater level promotes the formation of CH₄ during this time period.
722 However, we could not find any significant differences in the short term or annual
723 CH₄ emissions between the two investigated fertilizers. According to Chadwick et al.
724 (2000) more than 90% of total CH₄ emissions occur during the first 24h following
725 fertilizer application. Therefore, we must assume that we have missed most of
726 fertilizer induced CH₄ emissions. However, all studies from literature confirm the only
727 minor importance of CH₄ emissions from applied organic fertilizers in the GHG
728 balance of agricultural grasslands (Wulf et al., 2002a; Amon et al., 2006; Dietrich et
729 al., 2012).

730 **4.5. N-losses by NH₃ volatilization**

731 The NH₃ losses measured after splash plate application at the third application event
732 followed the typical pattern of lost ammonia (Clemens et al., 2006), particularly at the
733 digestate treatments. Significantly higher NH₃ losses from treatments fertilized with
734 biogas digestate were observed compared to those fertilized with cattle slurry. This is
735 in line with several other studies (Messner, 1988, Döhler and Haring, 1989 (cited in
736 Döhler and Horlacher, 2010); Amon et al., 2006; Möller and Stinner, 2009; Pacholski
737 et al., 2010; Ni et al., 2011), whereas Pain et al. (1990), Rubæk et al. (1996), Wulf et
738 al. (2002b) and Clemens et al. (2006) found no differences between anaerobic
739 digested slurries compared to other animal slurries. However, it has to be taken into
740 account that the present results are based only on measurements from a single
741 application event. The observed relative N losses of 36% of applied NH₄⁺-N at the
742 biogas digestate treatments were in the range reported for liquid slurries and
743 digestates applied via surface application, whereas the significantly lower relative N
744 losses (15%) at the cattle slurry treatments stands in strong contrast to those
745 reported in literature (e.g. Döhler and Haring, 1989 (cited in Döhler and Horlacher,
746 2010); Smith et al., 2000; Wulf et al., 2002b; Chantigny et al., 2004). However,
747 compared to the EF of 60% used in the German national greenhouse gas inventory
748 both estimated NH₃ loss rates were rather low (Haenel et al., 2014). It can be
749 assumed that the higher concentration of NH₄⁺ (NH₄⁺/N_{tot} ratio 0.65 vs. 0.33) and the
750 distinctly higher pH value (7.7 vs. 6.8) of the applied digestate compared to the cattle
751 slurry caused the observed differences in the current study, since temperature and
752 wind speed were equal. According to Sommer and Hutchings (2001) a change in the
753 pH value from 7.7 to 8.0 will double the emission. However, the factors controlling the
754 rise in pH are complex (Sommer and Husted, 1995b cited in Sommer and Hutchings,
755 2001) and the pH value was not determined after fertilization in the present study.
756 Several authors propose that a lower dry matter content of slurries favors the
757 infiltration into the soil with a subsequent faster decrease of NH₃ losses (Sommer et
758 al. 1996; Ni et al. 2011). There over a limited range (slurry DM of 2–5%), NH₃ losses
759 increase by approximately 6% for every 1% DM content (Smith et al., 2000).
760 Although the observed dry matter content of the biogas digestates was very low and
761 at the lower end of values reported in literature (e.g. Gutser et al., 2005; Möller et al.,
762 2008; Quarkernack et al., 2011) no corresponding effect was found in the present
763 study as was also reported by Möller and Stinner (2009). According to Döhler and

764 Horlacher (2010) and Smith et al. (2000), water saturated grassland soils as well as
765 very dry grassland soils high in organic matter lead to higher NH_3 -losses due to the
766 reduced infiltration of slurries. Thus it could be assumed that the infiltration of the
767 slurries was possibly hampered in the current study, removing the effect of the
768 different DM contents, due to the strong rain event which took place before the
769 fertilizer application. The cattle slurry in our experiment had very favorable
770 characteristics for crust formation (high DM content, grass silage diet; Smith et al.
771 2007). Warm weather also supported crust formation after application of cattle slurry,
772 which can effectively inhibit NH_3 exchange with the atmosphere and has been
773 proposed as NH_3 mitigation measure for slurry storage (Smith et al. 2007). The
774 emission pattern observed in our study on soil with limited infiltration capacity
775 supports the effectiveness of crusts for low NH_3 losses. Additionally, at low dosage
776 applications a large part of the organic fertilizer remained on the plant canopy and
777 thus soil contact and infiltration was limited after spreading. We conclude that this
778 was also the main reason why no significant differences in the pattern of NH_3
779 volatilization between the soil types were found in the present study. Nevertheless,
780 the distinct lower relative N losses from cattle slurry compared to literature values
781 could not be explained in this way, but NH_3 volatilization reported in literature showed
782 a high variability in respect to climatic and soil conditions, slurry composition, and
783 application technique.

784 The observed relative N losses of 15–36% of applied NH_4^+ -N demonstrates that NH_3
785 volatilization is quantitatively the most important N-loss from slurry application, as
786 was also proposed by Flessa and Beese (2000). Beside the negative effects of
787 eutrophication and acidification of ecosystems (e.g. Fangmeier et al., 1994; Galloway,
788 1995; Smith et al., 1999; Galloway, 2001), distinct NH_3 volatilization decreases the N
789 fertilizer use efficiency. One of the most effective measures to reduce NH_3 emissions
790 from grassland is the incorporation of slurry (Rodhe et al., 2006). However, several
791 studies reported a considerable increase of greenhouse gases (GHG), mainly N_2O ,
792 after injection of slurries and biogas digestates (Dosch and Gutser, 1996; Flessa and
793 Beese, 2000; Wulf et al., 2002a). Up to date no study has examined the effect of the
794 injection technique on organic soils.

795 **5. Conclusion**

796 We studied N_2O , CH_4 and NH_3 fluxes after splash plate application of biogas
797 digestate and cattle slurry in a region known for its risk of high N_2O and NH_3
798 emissions and we were the first to study digestate application on high organic carbon
799 soils with 10 to 17% C_{org} content in the topsoil. To our surprise, N_2O emissions and
800 EF were lower than generally observed on mineral soils in the vicinity of the sites. We
801 attributed the low N_2O emissions to a mild winter without clear freeze-thaw cycles,
802 but maybe also to frequent application with low dosage of N, which was quickly taken
803 up by the grass vegetation, as was indicated by the apparent NUE_{min} . N_2O emissions
804 increased with C_{org} content and fertilization. As hypothesized, N_2O were distinctly
805 higher after digestate than after slurry fertilization, which probably could be attributed
806 to a priming effect caused by increased SOM mineralization or other sources of labile
807 carbon triggered by digestate application. Due to the deep drainage, CH_4 emissions
808 were of only minor importance and were independent of fertilizer type. Estimated N
809 balances were negative for the control and the digestate treatments, but strongly
810 positive in all cases when the net N supply from SOM mineralization was considered.
811 The observed linear increase in cumulative N_2O emissions with increasing NH_4^+
812 fertilization and increasing groundwater table reveals the importance of site adapted
813 N fertilization and the avoidance of N surpluses during agricultural use of C_{org} rich
814 grasslands.

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1284

1285 **Table 1** Summary of the grassland management for both vegetation periods.

Date	Management event
2010-03-26	levelling
2010-04-07	rolling
2010-05-24	mowing
2010-06-14	manuring [20 m ³ ha ⁻¹]
2010-08-20	mowing
2010-08-25	manuring [20 m ³ ha ⁻¹]
2010-09-23	herbicide against common sorrel (<i>Rumex acetosa</i>)
2011-03-16	levelling
2011-05-23	mowing
2011-05-27	manuring [25 m ³ ha ⁻¹]
2011-08-01	mowing
2011-09-13	mowing
2011-09-22	manuring [20 m ³ ha ⁻¹]
2011-11-04	manuring [25 m ³ ha ⁻¹]

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1288 **Table 2** Soil properties of the study sites.

		C _{org} -medium	C _{org} -high	<i>n</i>
Sampling depth				
Soil type (WRB 2006) ¹		mollic Gleysol	sapric Histosol	
Soil type (German classification KA5)		GMq	KV-KM	
Peat depth [cm] ¹		80	70	1
pH value*		4.1	4.2	
Total nitrogen [%]*		1.0	1.5	
Organic carbon [%]	0–10 cm	10.3 ± 0.2	17.0 ± 0.1	9
	10–20 cm	9.3 ± 0.2	16.3 ± 0.2	9
Bulk density [g cm ⁻³]	0–10 cm	0.79 ± 0.02	0.54 ± 0.02	18
	10–20 cm	0.90 ± 0.01	0.64 ± 0.01	18
Porosity [%]	0–10 cm	71 ± 1	78 ± 1	18
	10–20 cm	67 ± 1	72 ± 0	18

1289 Values present means ± standard error

1290 ¹ World Reference Base for Soil Resources

1291 * Relative to the upper horizon (C_{org}-medium 0–20 cm; C_{org}-high 0–15 cm); Roßkopf personal communication

Table 3 Physical and chemical properties from the applied digestates and slurries.

	Cattle slurry					Biogas digestate				
	1. Application (14.06.2010)	2. Application (25.08.2010)	3. Application (27.05.2011)	4. Application (22.09.2011)	5. Application (04.11.2011)	1. Application (14.06.2010)	2. Application (25.08.2010)	3. Application (27.05.2011)	4. Application (22.09.2011)	5. Application (04.11.2011)
Fertilizer quantity [m ³ ha ⁻¹]	20	20	25	20	20	20	20	25	20	20
Total carbon [kg ha ⁻¹]	579	676	798	797	1073	384	373	167	184	178
Organic carbon [kg ha ⁻¹]	410	573	655	706	960	306	337	148	161	178
Total nitrogen [kg ha ⁻¹]	47	64	70	85	97	49	52	78	35	61
NO ₃ ⁻ [kg N ha ⁻¹]	0	0	0	0	0	0	0	0	0	0
NH ₄ ⁺ [kg N ha ⁻¹]	20	28	23	33	38	22	28	51	17	40
NH ₄ ⁺ /N _{tot} ratio	0.42	0.44	0.33	0.38	0.39	0.45	0.53	0.65	0.49	0.66
C/N ratio	12	11	11	9	11	8	7	2	5	3
pH (CaCl ₂)	–	–	6.8	7.0	7.0	–	–	7.7	7.4	7.7
Dry matter content [%]	5	7	7	9	10	4	4	2	2	3

Table 4 Mean (minimum/maximum) groundwater levels (GW), NO₃⁻ and NH₄⁺ contents of the soils following organic fertilizer application and for the years 2010 and 2011.

Sampling depth [cm]	C _{org} -medium			C _{org} -high			<i>n</i>
	Control	Cattle slurry	Biogas digestate	Control	Cattle slurry	Biogas digestate	
1 Application (14.06. – 30.06.2010)							
GW level [cm]	-32 (-62/-2)	-39 (-60/-5)	-31 (-58/-2)	-21 (-46/-1)	-26 (-45/-7)	-33 (-45/-19)	
NO ₃ ⁻ [mg N kg ⁻¹]	0–10 5 (1/9)	5 (1/7)	7 (3/10)	6 (1/12)	8 (3/11)	8 (5/10)	12
	10–20 9 (6/12)	9 (5/13)	11 (8/15)	11 (8/15)	12 (7/15)	14 (6/19)	12
NH ₄ ⁺ [mg N kg ⁻¹]	0–10 0 (0/1)	0 (0/1)	0 (0/1)	1 (0/2)	1 (0/6)	1 (0/3)	12

	10–20	0 (0/1)	0 (0/1)	0 (0/1)	0 (0/2)	1 (0/11)	1 (0/2)	12
2 Application (25.08. – 10.09.2010)								
GW level [cm]		-64 (-70/-49)	-58 (-63/-42)	-57 (-63/-40)	-36 (-40/-37)	-40 (-46/-22)	-37 (-43/-15)	
NO ₃ ⁻ [mg N kg ⁻¹]	0–10	13 (6/23)	27 (7/49)	21 (14/30)	25 (17/37)	50 (17/95)	25 (9/43)	12
	10–20	22 (17/28)	28 (19/37)	27 (17/38)	31 (26/35)	34 (11/45)	31 (12/48)	12
NH ₄ ⁺ [mg N kg ⁻¹]	0–10	0 (0/0)	2 (0/17)	0 (0/0)	3 (0/32)	0 (0/1)	1 (0/5)	12
	10–20	1 (0/10)	0 (0/0)	0 (0/1)		0 (0/1)	1 (0/3)	12
3 Application (27.05. – 11.06.2011)								
GW level [cm]		-82 (-94/-57)	-76 (-89/-52)	-80 (-97/-46)	-41 (-60/-11)	-47 (-62/-16)	-49 (-62/-16)	
NO ₃ ⁻ [mg N kg ⁻¹]	0–10	9 (4/17)	17 (5/30)	40 (10/75)	17 (11/26)	29 (12/63)	29 (11/50)	12
	10–20	17 (11/22)	31 (18/44)	45 (18/75)	24 (18/30)	28 (18/40)	45 (21/148)	12
NH ₄ ⁺ [mg N kg ⁻¹]	0–10	0 (0/2)	2 (0/10)	21 (0/104)	0 (0/1)	1 (0/5)	10 (0/47)	12
	10–20	0 (0/1)	1 (0/2)	6 (0/26)	1 (0/2)	1 (0/3)	4 (0/12)	12
4 Application (22.09. – 07.10.2011)								
GW level [cm]		-83 (-87/-72)	-77 (-81/-70)	-76 (-83/-58)	-54 (-60/-33)	-55 (-58/-46)	-53 (-57/-41)	
NO ₃ ⁻ [mg N kg ⁻¹]	0–10	18 (12/28)	48 (18/83)	62 (49/87)	23 (20/34)	43 (28/73)	45 (18/86)	12
	10–20	32 (18/46)	50 (21/79)	53 (35/66)	24 (20/30)	30 (22/39)	38 (23/86)	12
NH ₄ ⁺ [mg N kg ⁻¹]	0–10	0 (0/0)	1 (0/8)	1 (0/10)	0 (0/0)	3 (0/21)	0 (0/0)	12
	10–20	0 (0/0)	0 (0/0)	0 (0/1)	0 (0/1)	1 (0/3)	0 (0/0)	12
<hr/>								
2010								
GW level [cm]*		-67 (-94/-2)	-65 (-91/-2)	-63 (-92/0)	-41 (-68/2)	-45 (-64/-1)	-45 (-67/-1)	
NO ₃ ⁻ [mg N kg ⁻¹]	0–10	9 (1/26)	14 (1/49)	12 (3/30)	15 (1/37)	24 (3/95)	17 (4/43)	45
	10–20	14 (5/34)	17 (5/38)	17 (4/38)	19 (7/47)	23 (6/64)	21 (6/49)	45
NH ₄ ⁺ [mg N kg ⁻¹]	0–10	0 (0/4)	1 (0/17)	0 (0/9)	2 (0/32)	2 (0/19)	1 (0/14)	45
	10–20	0 (0/10)	0 (0/1)	0 (0/1)	0 (0/8)	1 (0/11)	1 (0/5)	45
2011								

GW level [cm]		-76 (-98/-3)	-72 (-92/0)	-72 (-97/0)	-47 (-67/1)	-52 (-66/-3)	-50 (-65/-3)	
NO ₃ ⁻ [mg N kg ⁻¹]	0-10	13 (4/31)	25 (5/83)	36 (8/111)	21 (11/41)	36 (12/98)	34 (11/91)	47
	10-20	24 (9/46)	34 (15/79)	40 (18/120)	27(14/52)	33 (18/78)	37 (10/148)	47
NH ₄ ⁺ [mg N kg ⁻¹]	0-10	0 (0/2)	1 (0/10)	6 (0/104)	1 (0/12)	2 (0/21)	4 (0/60)	47
	10-20	0 (0/3)	0 (0/4)	2 (0/26)	1 (0/7)	1 (0/27)	2 (0/12)	47

Table 5 Calculated emission factors (EF) for the year 2011 and for single application events (16 days) (Apl. 1 – Apl. 4). The EF is based on the amount of total nitrogen (N_{tot}) without consideration of NH₃-N losses.

	C _{org} -medium			C _{org} -high		
	Control	Cattle slurry	Biogas digestate	Control	Cattle slurry	Biogas digestate
N ₂ O exchange [kg N ha ⁻¹ yr ⁻¹]	0.91 ± 0.49	1.21 ± 0.05	1.86 ± 0.23	1.18 ± 0.07	1.77 ± 0.15	3.14 ± 0.91
Annual EF		0.12	0.55		0.23	1.13
EF Apl. 1		0.18	0.17		0.20	0.35
EF Apl. 2		0.11	0.05		0.11	0.21
EF Apl. 3		0.08	0.21		0.23	0.68
EF Apl. 4		0.09	0.33		0.15	0.56

Table 6 Grass yields, N uptake and N use efficiency for the years 2010 and 2011.

Treatment	Cutting date	Fertilization date	N content plant [%]	DM [t ha ⁻¹ yr ⁻¹]	N uptake [kg N ha ⁻¹]	N applied [kg N ha ⁻¹]†	N _{min} applied [kg N ha ⁻¹]†	N use efficiency [%]	N _{min} use efficiency [%]
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Control C _{org} -medium	24-May-10	–	2.04*	2.52	51	–	–	–	–
Control C _{org} -high	24-May-10	–	2.14	2.93	63	–	–	–	–
Cattle slurry C _{org} -medium	24-May-10	N.A.	2.37	3.19	76	–	–	–	–
Cattle slurry C _{org} -high	24-May-10	N.A.	2.14	3.58	77	–	–	–	–
Biogas digestate C _{org} -medium	24-May-10	N.A.	2.04	4.17	85	–	–	–	–
Biogas digestate C _{org} -high	24-May-10	N.A.	2.27	4.39	100	–	–	–	–
Control C _{org} -medium	20-Aug-10	–	2.03	2.02	41	–	–	–	–
Control C _{org} -high	20-Aug-10	–	2.00	2.63	53	–	–	–	–
Cattle slurry C _{org} -medium	20-Aug-10	14-Jun-10	2.19	3.06	67	45	17	58	153
Cattle slurry C _{org} -high	20-Aug-10	14-Jun-10	1.93	3.23	62	45	17	22	57
Biogas digestate C _{org} -medium	20-Aug-10	14-Jun-10	2.03	2.99	61	38	14	52	140
Biogas digestate C _{org} -high	20-Aug-10	14-Jun-10	2.00	3.51	70	38	14	47	125
Control C _{org} -medium	23-May-11	–	1.96	2.66	52	–	–	–	–
Control C _{org} -high	23-May-11	–	1.70	3.82	65	–	–	–	–
Cattle slurry C _{org} -medium	23-May-11	25-Aug-10	2.01	2.58	52	61	24	0	0
Cattle slurry C _{org} -high	23-May-11	25-Aug-10	1.70	4.20	71	61	24	11	27
Biogas digestate C _{org} -medium	23-May-11	25-Aug-10	1.96	3.97	78	40	18	64	144
Biogas digestate C _{org} -high	23-May-11	25-Aug-10	1.83	4.54	83	40	18	45	101
Control C _{org} -medium	1-Aug-11	–	1.71	2.06	35	–	–	–	–
Control C _{org} -high	1-Aug-11	–	1.48	2.88	43	–	–	–	–
Cattle slurry C _{org} -medium	1-Aug-11	27-May-11	1.71	2.73	47	67	20	17	58
Cattle slurry C _{org} -high	1-Aug-11	27-May-11	1.51	3.19	48	67	20	8	28
Biogas digestate C _{org} -medium	1-Aug-11	27-May-11	1.78	4.88	87	60	33	86	158
Biogas digestate C _{org} -high	1-Aug-11	27-May-11	1.48	5.34	79	60	33	61	112
Control C _{org} -medium	13-Sep-11	–	2.53	1.71	43	–	–	–	–
Control C _{org} -high	13-Sep-11	–	2.26	2.27	51	–	–	–	–
Cattle slurry C _{org} -medium	13-Sep-11	27-May-11	2.57	2.28	59	(55)‡	(8)‡	28	189
Cattle slurry C _{org} -high	13-Sep-11	27-May-11	2.53	2.64	67	(61)‡	(14)‡	25	110

Biogas digestate C _{org} -medium	13-Sep-11	27-May-11	2.53	3.15	80	(8)‡	(0)‡	436	–
Biogas digestate C _{org} -high	13-Sep-11	27-May-11	2.26	3.25	74	(24)‡	(0)‡	94	–

* N contents from control treatments were estimated from fertilized treatments.

† Applied N_{tot} and N_{min} were corrected by NH₃-N losses (23% and 5% from N_{tot}, or rather 36% and 15% from N_{min} for biogas digestate and cattle slurry, respectively).

‡ Hypothetically remaining N_{tot} and N_{min} from the application event 3 (27th May 2011).

N.A. = not available.

Table 7 Estimated nitrogen balance for the year 2011.

Treatment	N applied [kg N ha ⁻¹ yr ⁻¹]	N _{min} t1* [kg N ha ⁻¹]	N _{min} t2* [kg N ha ⁻¹]	N deposition [kg N ha ⁻¹ yr ⁻¹]	N uptake [kg N ha ⁻¹ yr ⁻¹]	N ₂ O [kg N ha ⁻¹ yr ⁻¹]	NH ₃ † [kg N ha ⁻¹ yr ⁻¹]	N balance [kg N ha ⁻¹ yr ⁻¹]
Control C _{org} -medium	0	27.5	29.4	7.2	130	0.9	0.0	-122.4
Control C _{org} -high	0	22.8	27.7	7.2	159	1.2	0.0	-148.0
Cattle slurry C _{org} -medium	252	35.7	51.2	7.2	157	1.2	46.1	70.4
Cattle slurry C _{org} -high	252	27.3	68.1	7.2	186	1.8	46.1	66.1
Biogas digestate C _{org} -medium	174	29.8	83.3	7.2	244	1.9	52.4	-64.1
Biogas digestate C _{org} -high	174	26.2	28.4	7.2	236	3.1	52.4	-108.1

* Reference date for t1 is the 06th April 2011 and for t2 the 18th October 2011.

† NH₃-N losses at the fourth and fifth application event were estimated based on EF taken from the German national greenhouse gas inventory (EF =0.6 related to applied NH₄⁺; Haenel et al., 2014).

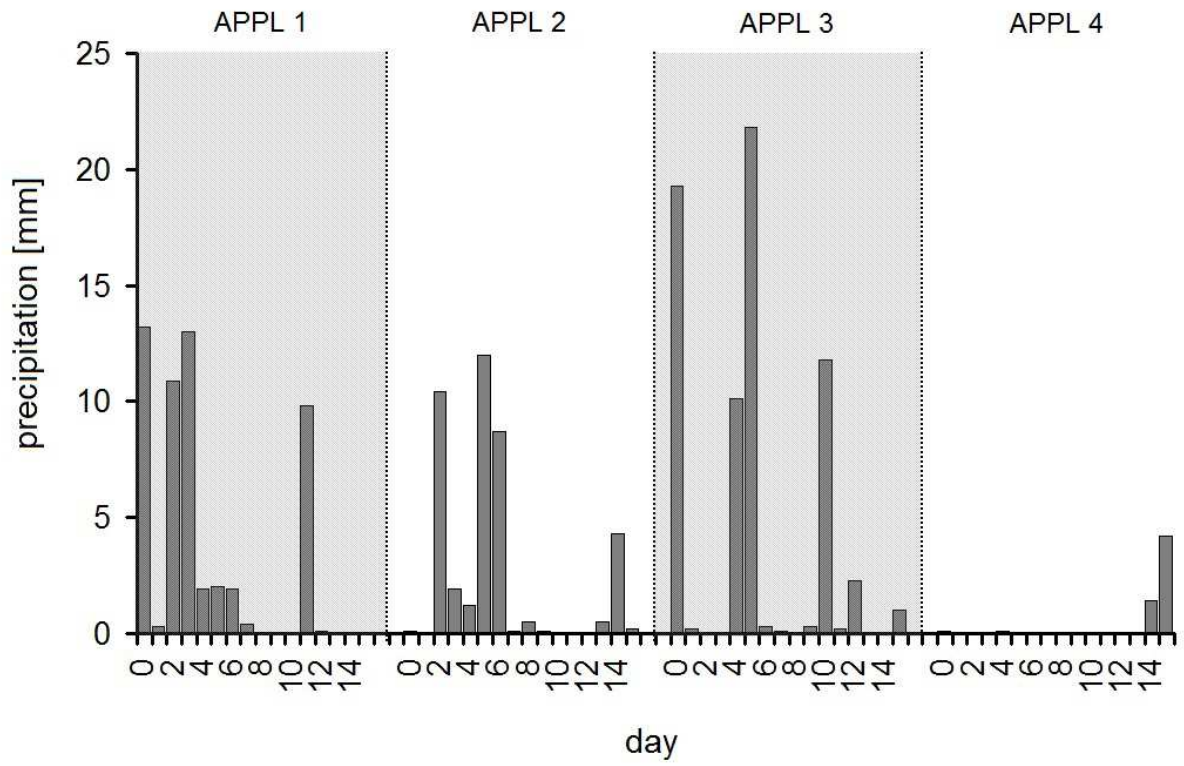


Fig. 1 Daily sums of precipitation following the organic fertilizer application events (APPL).

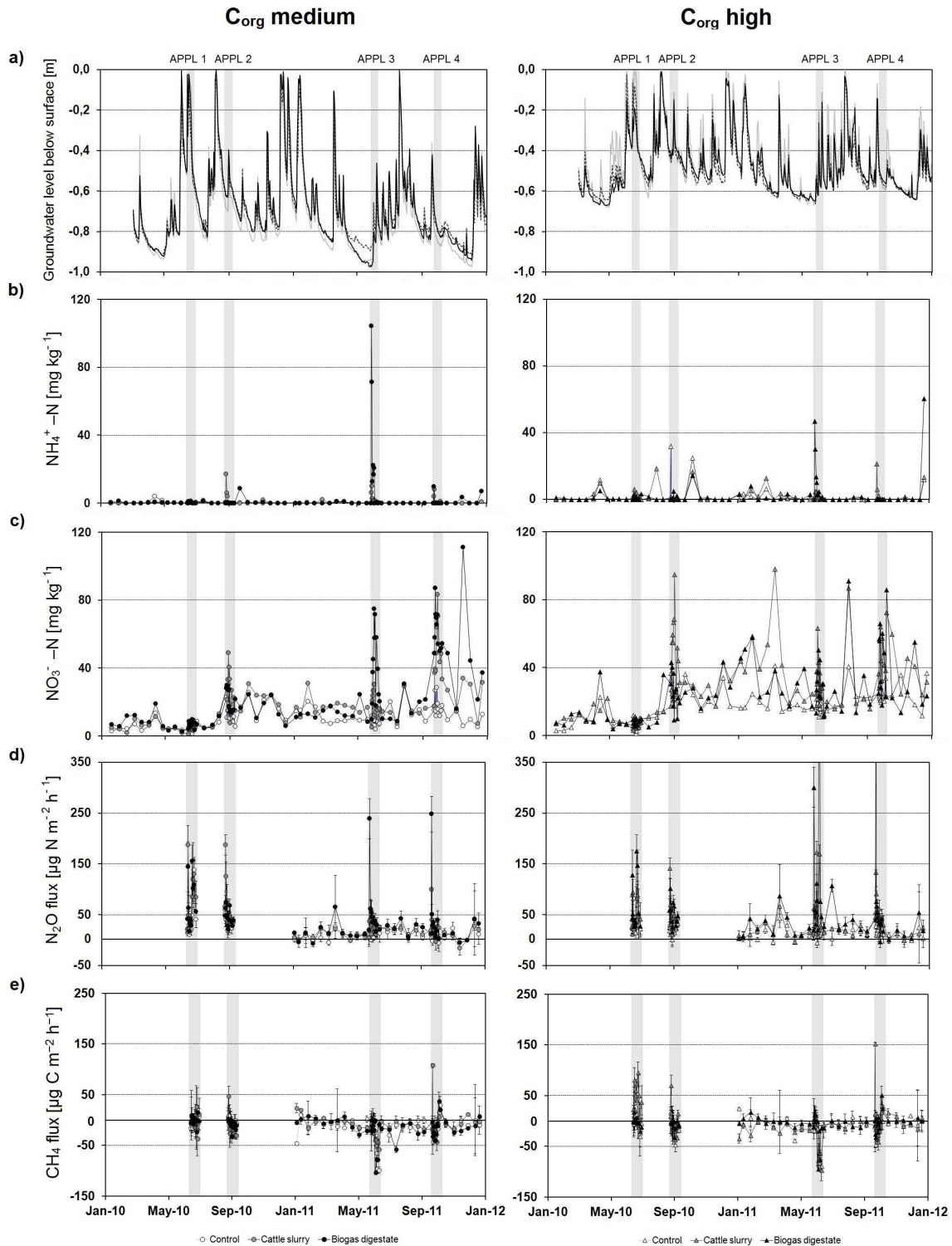


Fig. 2 Variation in groundwater level (a), extractable NH_4^+ (b) and NO_3^- (c) contents for the soil depth 0–10 cm, N_2O (d) and CH_4 fluxes (e) (Mean \pm SD, $n = 3$) of the C_{org} -medium and C_{org} -high sites from January 2010 to January 2012. Gray bars mark the 15 or 16 days period of intensiv gas flux measurements following organic fertilizer application.

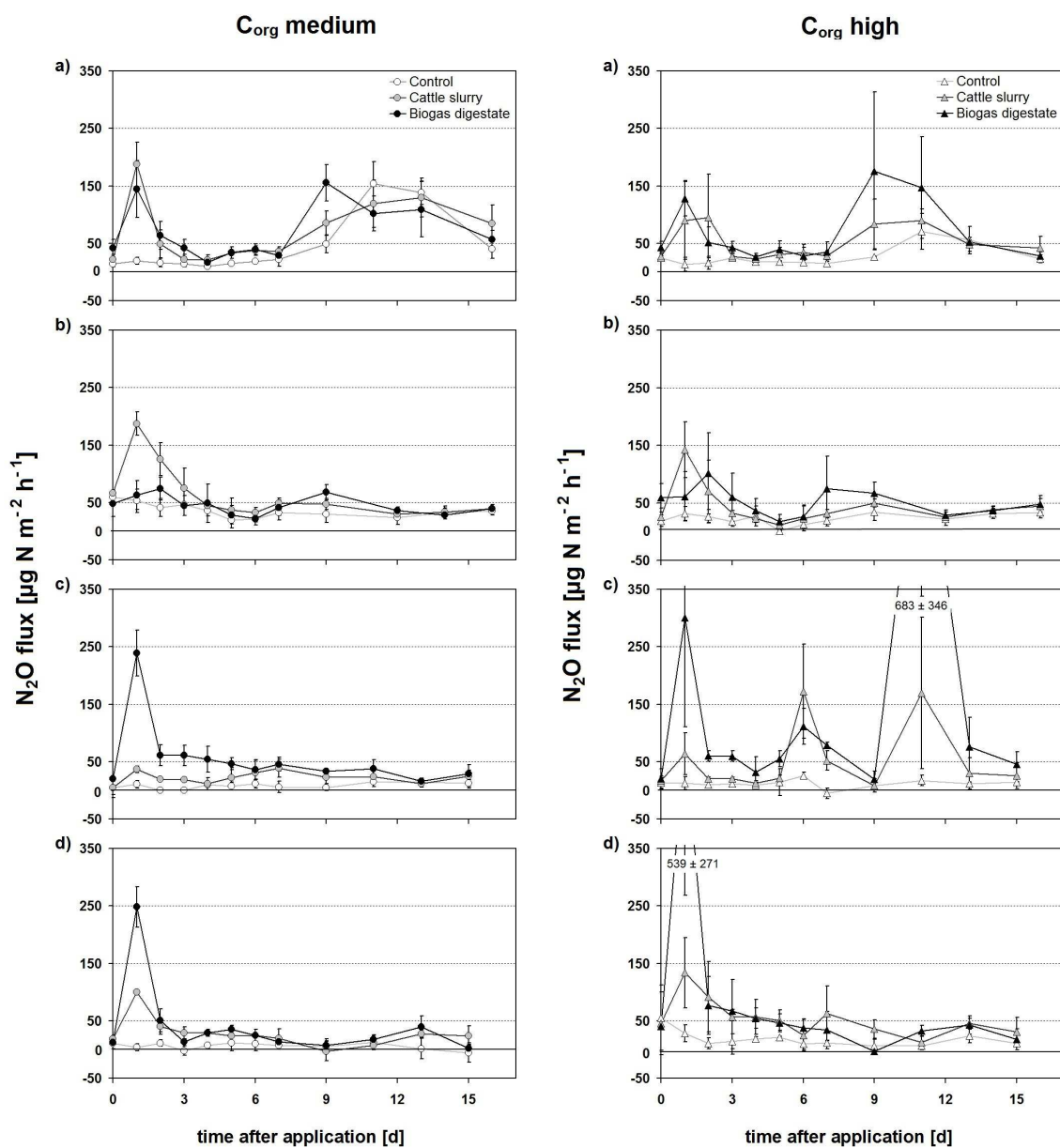


Fig. 3 Mean (\pm SD, $n = 3$) N₂O fluxes following organic fertilizer application events (a) 14.06–30.06.2010; (b) 25.08.–10.09.2010; (c) 27.05.–11.06.2011 and (d) 22.09.–07.10.2011.

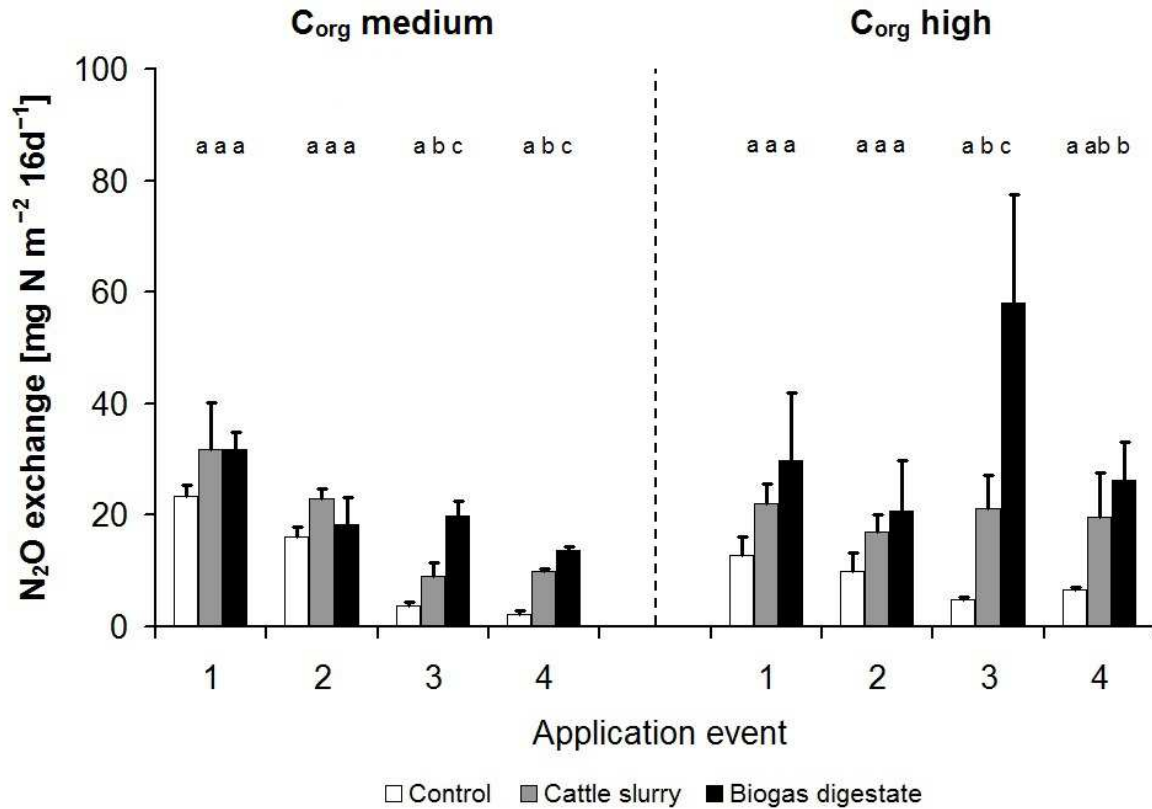


Fig. 4 Cumulative N₂O exchange during 16 days following organic fertilizer application. Bars indicate mean values + SD ($n = 3$). Means followed by the same letter indicated no significant differences between treatments at a single application event for sites C_{org}-medium and C_{org}-high respectively (ANOVA, Tukey HSD-test at $P \leq 0.05$).

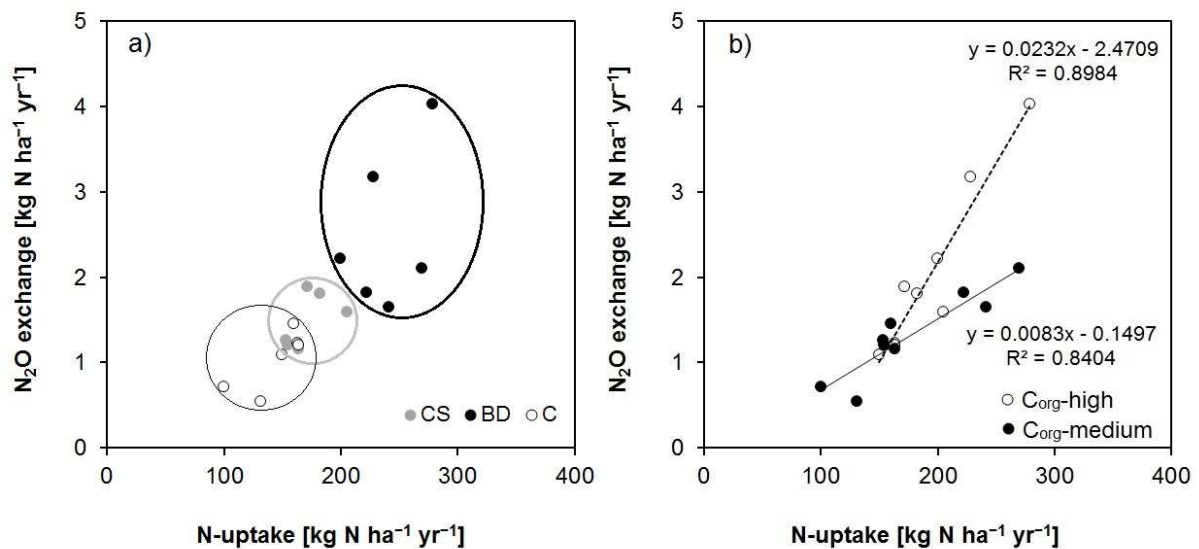


Fig. 5 Relationship between cumulative annual N₂O emissions and annual plant N uptake regarded for the treatments a) and for the investigated soil types b). Dots represent mean annual values of each PVC-collars. CS = Cattle slurry, BD = Biogas digestate, C = Control.

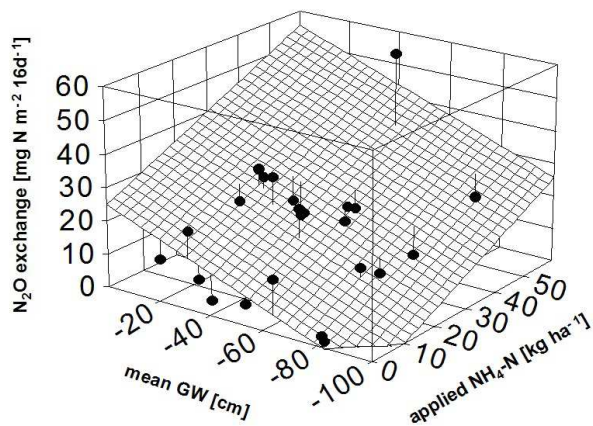


Fig. 6 Relationship of plot-wise mean 16 days cumulative N_2O -N emissions of the four application events (y) to mean groundwater level (x1) and the amount of applied $\text{NH}_4\text{-N}$ (x2). The regression equation is $y = 24.98 (\pm 4.98) + x_1 \cdot 0.30 (\pm 0.09) + x_2 \cdot 0.51 (\pm 0.11)$; $R^2 \text{ adj.} = 0.53$, $P < 0.001$, $df = 21$. Solid lines indicate the deviation of measured data from the model surface.

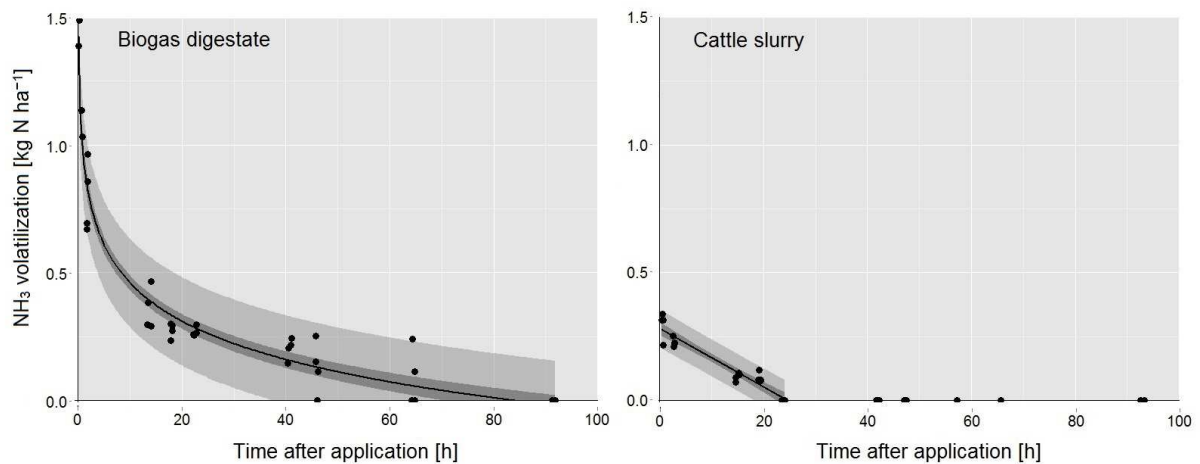


Fig. 7 Ammonia (NH_3) volatilization following organic fertilizer application at event 3 (27.05.2011). Dots present single NH_3 measurements for a time period of 94 hours. Black lines show the estimated NH_3 volatilization with 95% confidence band (dark grey) and 95% prediction band (light grey). The model function for biogas digestate is: $y = -0.2619 (\pm 0.025) \cdot \ln(x) + 0.9605 (\pm 0.008)$; $r^2 = 0.96$; $P < 0.0001$; the model function for cattle slurry is: $y = 0.2818 (\pm 0.012) - 0.0114(x) (\pm 0.001)$; $r^2 = 0.92$; $P < 0.0001$.