

1 **Stable carbon isotope biogeochemistry of lakes along a trophic gradient**

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16 **Abstract**

17 The stable carbon (C) isotope variability of dissolved inorganic and organic C (DIC and
18 DOC), particulate organic carbon (POC), glucose and polar-lipid derived fatty acids (PLFA)
19 were studied in a survey of 22 North American oligotrophic to eutrophic lakes. The $\delta^{13}\text{C}$ of
20 different PLFA were used as proxy for phytoplankton producers and bacterial consumers. Lake
21 pCO_2 was primarily determined by autochthonous production (phytoplankton biomass),
22 especially in eutrophic lakes, and governed the $\delta^{13}\text{C}$ of DIC. All organic-carbon pools showed
23 larger isotopic variability in eutrophic lakes compared to oligo-mesotrophic lakes because of the
24 high variability in $\delta^{13}\text{C}$ at the base of the food web (both autochthonous and allochthonous
25 carbon). Phytoplankton $\delta^{13}\text{C}$ was negatively related to lake pCO_2 over all lakes and positively
26 related to phytoplankton biomass in eutrophic lakes, which was also reflected in a large range in
27 photosynthetic isotope fractionation ($\epsilon_{\text{CO}_2\text{-phyto}}$, 8-25 ‰). The carbon isotope ratio of
28 allochthonous carbon in oligo-mesotrophic lakes was rather constant, while it varied in eutrophic
29 lakes because of maize cultivation in the watershed.

30

31 **1. Introduction**

32 Studies suggest that lakes contribute significantly to the global carbon budget via organic
33 matter burial and emission of CO₂ to the atmosphere (Cole et al., 2007). The balance between
34 primary production and external organic carbon input on the one hand and respiration and burial
35 of organic carbon on the other governs whether individual lakes are sources or sinks of CO₂.
36 This metabolic balance can be disturbed by changes in nutrient or organic matter inputs to the
37 lake. Primary (autochthonous) production increases with increasing nutrient concentrations and
38 lakes with high autochthonous carbon production, i.e. eutrophic lakes, may be sinks for CO₂
39 (Schindler et al., 1997). The loading of allochthonous (terrestrial) carbon is a key factor
40 controlling community respiration of lakes. The metabolic balance of lakes is directly influenced
41 by allochthonous organic carbon loading and trophic state (Del Giorgio and Peters, 1994;
42 Hanson et al., 2003).

43 Stable carbon isotope analysis is a powerful tool for studying carbon cycling in lakes
44 since it allows studying inorganic and organic carbon pools and changes therein. It can provide
45 information on the metabolic balance and the sources of organic matter fueling respiration.
46 Respiration yields ¹³C-depleted carbon dioxide from organic matter with the result that δ¹³C of
47 dissolved inorganic carbon of the lakes becomes lower (Parker et al., 2010). Primary producers
48 preferentially incorporate ¹²C in their organic matter with the consequence that the remaining
49 pool of dissolved inorganic carbon will be enriched in ¹³C (Herczeg, 1987; Parker et al., 2010).
50 The δ¹³C of the dissolved inorganic carbon pool thus integrates the relative importance of
51 respiration and primary production (Parker et al., 2010). The δ¹³C of organic carbon pools is
52 primarily governed by the δ¹³C of the dissolved inorganic carbon used by primary producers and
53 the isotope fractionation during carbon fixation. Terrestrial plants use atmospheric carbon
54 dioxide while aquatic primary producers utilize dissolved carbon dioxide or bicarbonate (Fry,
55 2006). The δ¹³C of terrestrial-derived organic carbon is therefore often distinct from that of
56 organic matter produced within the lakes and this difference can be used to trace carbon flows
57 and origins in plankton food webs.

58 A major challenge in stable isotope studies is to elucidate the isotopic composition of
59 microbial organisms (Middelburg, 2014), such as phytoplankton and bacteria, since it is difficult
60 to separate these potential carbon sources from bulk particulate organic carbon (POC).

61 Therefore, most studies use indirect methods to determine $\delta^{13}\text{C}$ of phytoplankton ($\delta^{13}\text{C}_{\text{phyto}}$),
62 bacteria ($\delta^{13}\text{C}_{\text{bac}}$) and allochthonous carbon ($\delta^{13}\text{C}_{\text{allo}}$). Common methods for determining $\delta^{13}\text{C}_{\text{phyto}}$
63 are the use of the $\delta^{13}\text{C}$ of particulate organic carbon (POC) with correction for non-
64 phytoplankton carbon and estimates based on $\delta^{13}\text{C}$ of dissolved inorganic carbon (DIC) with an
65 isotope fractionation factor (ϵ), obtained from experimental studies. Other methods are the use of
66 zooplankton consumers as a proxy for $\delta^{13}\text{C}_{\text{phyto}}$ or size fractionation of organic matter and
67 subsequent determination of $\delta^{13}\text{C}$ of different size classes (Marty and Planas, 2008).

68 Isotopic ratios of bacteria in field studies have been derived from re-growing bacteria in
69 bioassays (Coffin et al., 1989) or dialysis cultures (Kritzberg et al., 2004), with measurement of
70 ^{13}C in POC or respired CO_2 (McCallister et al., 2008) and from biomarkers like nucleic acids
71 (Coffin et al., 1990) and lipids (Bontes et al., 2006; Pace et al., 2007). Some studies used $\delta^{13}\text{C}$ of
72 DOC as proxy for $\delta^{13}\text{C}$ of bacteria, assuming that DOC was the primary carbon source for
73 bacteria (Taipale et al., 2008; Zigah et al., 2012).

74 A commonly used proxy for allochthonous $\delta^{13}\text{C}$ is the $\delta^{13}\text{C}$ of terrestrial C_3 plants, which
75 dominates most terrestrial vegetation and has a $\delta^{13}\text{C}$ of ~ -28 ‰ (Fry, 2006; Kohn, 2010). When
76 vegetation is dominated by C_4 plants, however, common in tropical areas and agricultural areas
77 with maize production ($\delta^{13}\text{C}$ of ~ -14 ‰; Fry, 2006), the isotopic composition of allochthonous
78 carbon can be significantly enriched. In lakes with large terrestrial input, $\delta^{13}\text{C}$ of DOC can be
79 used as a proxy for $\delta^{13}\text{C}_{\text{allo}}$, since terrestrial carbon forms the largest fraction of DOC (Kritzberg
80 et al., 2004; Wilkinson et al., 2013).

81 Compound specific isotope analysis (CSIA) of polar lipid-derived fatty acids (PLFA)
82 biomarkers has shown to be a valuable tool to determine the isotopic composition of plankton
83 producers and consumers (Boschker and Middelburg, 2002). Groups of phytoplankton and
84 bacteria have different fatty acid (FA) compositions, so by analyzing the $\delta^{13}\text{C}$ of specific FA, the
85 $\delta^{13}\text{C}$ of phytoplankton and bacteria can be inferred. The combined use of stable isotopes and FA
86 biomarkers has been successfully applied to study autochthonous and allochthonous carbon
87 contributions to zooplankton in a tidal river (Van den Meersche et al., 2009). Few studies have
88 applied CSIA to study carbon flows in plankton food webs in lakes. Examples are a
89 phytoplankton-zooplankton interaction study in a eutrophic lake (Pel et al., 2003), a

90 biomanipulation effect study (Bontes et al., 2006), a ^{13}C lake enrichment study (Pace et al.,
91 2007), and a cyanobacteria-zooplankton interaction study (de Kluijver et al., 2012).

92 In this study, we used compound-specific isotope analyses to examine carbon flows in
93 plankton food webs in temperate (North American) lakes. The lake survey encompassed a range
94 in trophic states from oligotrophic lakes, with an expected dominance of allochthonous input, to
95 eutrophic lakes, with an expected lower allochthonous input. In this trophic range, we explored
96 patterns of isotopic variability in dissolved inorganic and organic carbon, particulate organic
97 carbon and carbohydrates, phytoplankton, allochthonous carbon, heterotrophic bacteria and their
98 relationships.

99

100 **2. Material and Methods**

101 **2.1 Site description**

102 The 22 lakes sampled in this study are located in Iowa and Itasca County in Minnesota,
103 USA. Iowa lakes are mostly man-made and situated in a highly agricultural region, with maize
104 and soya beans as main products. This type of row-crop agriculture has a large impact on the
105 nutrient load of the lake watershed (Arbuckle and Downing, 2001). However, Itasca lakes are
106 natural and situated in a highly forested area. The catchment areas have developed since the last
107 glaciation episode ca. 12,000 years ago and consist of carbonate-poor glacial deposits (till)
108 (Grimley, 2000).

109 **2.2 Field sampling**

110 The lakes were sampled in July -August 2009 as part of the ongoing lake monitoring
111 program of the limnology laboratories of Iowa State University and Itasca Community College.
112 Key parameters, such as temperature, pH, Secchi transparency, oxygen, inorganic nutrients and
113 carbon concentrations were measured as part of and according to the lake monitoring program.
114 All samples were taken from up to 2 m of the upper mixed zone at the deepest point of each lake.
115 Water samples were taken between 10h-16h, a period of the day that yields relatively stable
116 water chemistry readings in these lakes. More information on data collection, lake
117 characteristics, and methods can be found on <http://limnoweb.eeob.iastate.edu/itascalakes> and

118 <http://limnology.eeob.iastate.edu/lakereport>. All nutrients were analyzed using certified methods
119 and strict quality assurance procedures.

120 Triplicate water samples were taken for stable isotope analyses and concentrations of the
121 major carbon pools. Headspace vials (20 ml and 2 ml) were filled on board with sampled water
122 using the overflow method and sealed with gas-tight caps for DIC isotope analyses and
123 concentrations, respectively. Mercury chloride was added for preservation and the samples were
124 stored upside-down at room temperature. For DOC analyses, 20 ml of sampled water was filtered
125 over GF/F (0.7 μm pore size, 25 mm diameter) and stored frozen in clean (acid and milli-Q
126 rinsed) vials until further analysis.

127 Seston samples for particulate organic carbon (POC) and carbohydrates were collected by
128 filtering 0.4 to 1 L of sampled water on pre-weighed and pre-combusted GF/F filters (0.7 μm
129 pore size, 47 mm diameter), which were subsequently dried at 60° for POC analysis or freeze-
130 dried for carbohydrates; PLFA samples were collected by filtering ~2 L sampled water on pre-
131 combusted GF/F filters (0.7 μm , 47 mm) and filters were stored frozen. Pigment samples were
132 taken for concentrations only and collected by filtering ~600 ml sampled water on GF/F filters
133 (0.7 μm , 47 mm) in the dark and filters were stored frozen.

134 **2.3 Laboratory analyses**

135 POC samples were analyzed for carbon content and isotope ratios on a Thermo Electron
136 Flash EA 1112 analyzer (EA) coupled to a Delta V isotope ratio mass spectrometer (IRMS) (c.f.
137 Nieuwenhuize et al., 1994). For DIC isotope analyses, a helium headspace was created in the
138 headspace vials and samples were acidified with H_3PO_4 solution. After equilibration, the CO_2
139 concentration and isotope ratio in the headspace was measured using EA-IRMS (Gilikin and
140 Bouillon, 2007). DIC concentrations were measured using spectrophotometry according to Stoll
141 et al., (2001). For DOC analyses, the samples were acidified and flushed with helium to remove
142 DIC and subsequently oxidized with sodium persulfate ($\text{Na}_2\text{S}_2\text{O}_8$); the isotope ratio and
143 concentration of CO_2 resulting from this treatment was measured using high performance liquid
144 chromatography - isotope ratio mass spectrometry (HPLC-IRMS) (Boschker et al. 2008). PLFA
145 samples were extracted according to a modified Bligh and Dyer method (Bligh and Dyer, 1959,
146 Middelburg et al., 2000). The lipids were fractionated in different polarity classes by column

147 separation on a heat-activated silic acid column and subsequent elution with chloroform, acetone,
148 and methanol. The methanol fractions, containing most of the PLFA, were collected and
149 transformed to fatty acid methyl esters (FAME) using methanolic NaOH. The 12:0 and 19:0
150 FAME were added as internal standards. Concentrations and $\delta^{13}\text{C}$ of individual PLFA were
151 measured using gas chromatography-combustion isotope ratio mass spectrometry (GC-C-IRMS)
152 (Middelburg et al., 2000). The isotopic compositions were corrected for the carbon added during
153 derivatization. Pigment samples were extracted with 90% acetone in purified (miliQ) water with
154 intense shaking. Concentrations of individual pigments were measured on HPLC (Wright et al.
155 1991). Carbohydrate samples were hydrolyzed in H_2SO_4 , neutralized with SrCO_3 , and
156 precipitated with BaSO_4 . The supernatant was collected and measured using HPLC-IRMS
157 according to Boschker et al., (2008).

158 **2.4 Data analyses**

159 The lakes were divided into eutrophic and oligo-mesotrophic lakes based on average
160 summer total phosphorus (TP) concentrations. Lakes with TP values $>24 \mu\text{g L}^{-1}$ and a
161 corresponding trophic state index >50 were classified as eutrophic, and lakes with TP values $<$
162 $24 \mu\text{g L}^{-1}$ as oligo-mesotrophic (Carlson 1977). All lakes in Iowa and one lake in Minnesota
163 were classified as eutrophic, while all oligo-mesotrophic lakes were located in Minnesota.

164 **2.4.1 CO_2 system and isotopic composition of CO_2**

165 The different components of the CO_2 system were calculated from temperature,
166 laboratory pH, and DIC concentrations using a salinity of 0 using the R package AquaEnv
167 (Hofmann et al., 2010). Stable isotope ratios were expressed in the delta notation ($\delta^{13}\text{C}$), which is
168 the $^{13}\text{C}/^{12}\text{C}$ ratio relative to VPDB standard, in part per thousand (‰). The isotope ratio of CO_2
169 (aq) ($\delta^{13}\text{C}_{\text{CO}_2}$) was calculated from $\delta^{13}\text{C}_{\text{DIC}}$ according to Zhang et al., (1995):

$$170 \quad \delta^{13}\text{C}_{\text{CO}_2} = \delta^{13}\text{C}_{\text{DIC}} - 0.0144 \times T(^{\circ}\text{C}) \times f\text{CO}_3^{2-} + 0.107 \times T(^{\circ}\text{C}) - 10.53 \quad (1)$$

171 where $f\text{CO}_3^{2-}$ is the fraction of CO_3^{2-} in total DIC, calculated from pH and DIC concentrations.

172 **2.4.2 $\delta^{13}\text{C}$ of phytoplankton and bacteria**

173 Poly-unsaturated fatty acids (PUFA) are abundant in most phytoplankton, and can
 174 generally be used as chemotaxonomic markers for this group (Dijkman and Kromkamp, 2006).
 175 The most abundant PUFA in all lakes were 18:3 ω 3 (α -linolenic acid), 18:4 ω 3 (stearidonic acid,
 176 SDA), 20:5 ω 3 (icosapentaenoic acid, EPA), 22:6 ω 3 (docosahexaenoic acid, DHA) and 20:4 ω 6
 177 (arachidonic acid, ARA), common PUFA's in freshwater phytoplankton (Taipale et al., 2013),
 178 and their concentration-weighted $\delta^{13}\text{C}$ were used to determine phytoplankton isotope ratios
 179 ($\delta^{13}\text{C}_{\text{phyto}}$). Note that phytoplankton is considered a mixture of eukaryotic algae and
 180 cyanobacteria. Branched fatty acids (BFA) are abundant in heterotrophic bacteria (Kaneda,
 181 1991) in contrast to phytoplankton. The most abundant BFA were i15:0, ai15:0 and i17:0 and
 182 their weighted $\delta^{13}\text{C}$ were used as a proxy for heterotrophic bacteria isotope ratios ($\delta^{13}\text{C}_{\text{bac}}$),
 183 which we further consider bacteria. Isotope fractionation (ϵ) between CO_2 and phytoplankton
 184 was calculated as

$$185 \quad \epsilon_{\text{CO}_2\text{-phyto}} (\text{‰}) = \frac{\delta^{13}\text{C}_{\text{CO}_2} - \delta^{13}\text{C}_{\text{phyto_cor}}}{1 + \delta^{13}\text{C}_{\text{phyto_cor}} / 1000} \quad (2)$$

186 $\delta^{13}\text{C}_{\text{phyto_cor}}$ was derived from $\delta^{13}\text{C}_{\text{phyto}}$ with a correction of +3‰ for the isotopic offset between
 187 PUFA and total cells ($\Delta\delta^{13}\text{C}_{\text{FA-cell}}$) (Schouten et al., 1998; Hayes, 2001), although this isotopic
 188 offset can be highly variable (Schouten et al. 1998).

189 2.4.3 $\delta^{13}\text{C}$ of allochthonous carbon

190 Allochthonous organic carbon ($\delta^{13}\text{C}_{\text{allo}}$), i.e. organic matter delivered to lakes as DOC or
 191 POC, cannot be measured directly as such and we therefore used two proxies: the measured
 192 isotopic ratios of DOC ($\delta^{13}\text{C}_{\text{DOC}}$) and calculated isotopic composition of particulate detritus
 193 ($\delta^{13}\text{C}_{\text{det}}$). The latter was calculated from a mass balance and mixing model, similar to Marty and
 194 Planas, (2008), amended with zooplankton and bacteria. We assumed that POC consists of
 195 phytoplankton, detritus, bacteria, and zooplankton, and that the $\delta^{13}\text{C}$ of POC represents a mixture
 196 of the weighted $\delta^{13}\text{C}$ of the individual groups. Subsequently, $\delta^{13}\text{C}_{\text{det}}$ in each lake was derived
 197 from $\delta^{13}\text{C}_{\text{POC}}$:

$$198 \quad \delta^{13}\text{C}_{\text{det}} (\text{‰}) = (\text{POC} \times \delta^{13}\text{C}_{\text{POC}} - C_{\text{phyto}} \times \delta^{13}\text{C}_{\text{phyto_cor}} - C_{\text{bac}} \times \delta^{13}\text{C}_{\text{bac}} - C_{\text{zoo}} \times \delta^{13}\text{C}_{\text{zoo}}) / C_{\text{det}} \quad (3)$$

$$199 \quad C_{\text{det}} (\text{mg C L}^{-1}) = \text{POC} - C_{\text{phyto}} - C_{\text{bac}} - C_{\text{zoo}} \quad (4)$$

200 The latter equation simply states that detrital organic matter is the non-living part of total
201 particulate organic matter pool.

202 Phytoplankton carbon (C_{phyto}) (mg C L^{-1}) was calculated as the average of biomass
203 estimates based on chl *a* concentration ($C: \text{Chl } a = 50$) as well as phytoplankton FA derived
204 biomass, to minimize the error associated with each method. Phytoplankton FA biomass was
205 calculated from the sum of phytoplankton PLFA ($\sum 18:3\omega3, 18:4\omega3, 20:5\omega3, 22:6\omega3,$ and
206 $20:4\omega6$) and a C: specific FA ratio of 60 based on culture studies, summarized in Dijkman and
207 Kromkamp (2006). The two approaches yielded similar results. Bacterial carbon (C_{bac}) (mg C L^{-1})
208 was calculated from the summed concentrations of bacteria specific FA (i15:0, ai15:0, and
209 i17:0) and a C_{bac} : FA ratio of 50 (Middelburg et al., 2000). Zooplankton carbon (C_{zoo}) used in
210 equation 3 was estimated to be ~10 % of C_{phyto} (Del Giorgio and Gasol, 1995) and zooplankton
211 $\delta^{13}\text{C}$ are based on de Kluijver (2012). Uncertainties in $\delta^{13}\text{C}$ and biomass of phytoplankton,
212 bacteria and zooplankton were not considered in calculating $\delta^{13}\text{C}_{\text{det}}$.

213 **2.4.4 Statistical analyses**

214 Data that were part of the lake monitoring program and pCO_2 values represent single
215 samples of each lake. Data on carbon concentrations and isotopic compositions in each lake
216 convey averages of triplicate samples. Statistical analyses were done with software package “R”
217 (R development core team, 2013). Prior to correlation analyses, data were checked for normal
218 distribution (Shapiro test) and log-transformed when necessary to achieve normal distribution.
219 Correlation coefficients were calculated using Pearson product-moment correlation coefficient
220 (normal distribution) or Spearman’s rank correlation coefficient (non-normal distribution). For
221 reasons of instructiveness we present the average \pm sd values for eutrophic lakes ($n=11$) and
222 oligo-mesotrophic lakes ($n=11$), but we do realize that any division based on a concentration is
223 somewhat arbitrary. The correlations were tested for total lakes ($n=22$) and for eutrophic lakes
224 ($n=11$) and oligo-mesotrophic lakes ($n=11$). Differences between eutrophic and oligo-
225 mesotrophic lakes were statistically tested using Student’s t-tests.

226

227

228 **3. Results**

229 **3.1 Lake chemistry**

230 The sampled lakes covered a large range of nutrients and CO₂ system characteristics
231 (Table 1). DIC values ranged from 0.052 to 4.6 mmol L⁻¹, alkalinity values ranged from 0.07 to
232 2.4 mmol L⁻¹ and pH ranged from 6.1 to 9.8 (Table 1). The calculated pCO₂ values were in the
233 range from 10-4500 µatm, covering a broad range from under- to super-saturation. The CO₂
234 system in eutrophic and oligo-mesotrophic lakes was clearly different. On average, the eutrophic
235 lakes had higher DIC, alkalinity, and pH than the oligo-mesotrophic lakes (Fig. 1, Table 1). In
236 the eutrophic lakes, there were positive correlations between alkalinity and DIC and pCO₂ values
237 (Fig. 1A, Table 2) and a negative correlation between pH and pCO₂ (Fig. 1B, Table 2). The
238 pCO₂ values were not related to pH, alkalinity, or DIC in the oligo-mesotrophic lakes. Both lake
239 systems showed super-saturation (average pCO₂ 838 µatm in both), but the pCO₂ range was
240 much larger in eutrophic lakes (10-4500 µatm) compared to oligo-mesotrophic lakes (310-3200
241 µatm) (Fig. 1, Table 1).

242 **3.2 Organic carbon and fatty acid concentrations**

243 POC, C_{phyto} and C_{bac} concentrations were higher and DOC concentrations were lower in
244 the eutrophic lakes compared to the oligo-mesotrophic lakes (Table 1). C_{phyto} was on average 1.3
245 ± 1.1 mg C L⁻¹ and 0.11 ± 0.03 mg C L⁻¹, corresponding to 44 % ± 17 % and 10 % ± 5 % of POC
246 in eutrophic and oligo-mesotrophic lakes, respectively. C_{phyto} and C_{bac} were significantly related
247 to TP (Table 2), but not to total nitrogen (TN) concentration. Average C_{bac} was 0.11 ± 0.08 mg C
248 L⁻¹ and 0.021 ± 0.017 mg C L⁻¹ in eutrophic and oligo-mesotrophic lakes, respectively.

249 Overall, lake pCO₂ decreased with increasing C_{phyto}, but the effect was strongest in
250 eutrophic lakes (Fig. 1C, Table 2). In the oligo-mesotrophic lakes, pCO₂ increased with
251 increasing DOC (Fig. 1D, Table 2), but this effect was caused by one point: the high pCO₂ at
252 high DOC in lake Sturgeon. In the eutrophic lakes, DOC concentrations were lower compared to
253 the oligo-mesotrophic lakes and did not act on lake pCO₂ (Fig. 1D, Table 1).

254 **3.3 δ¹³C of DIC and CO₂**

255 The isotope ratios of the major carbon pools in each lake are presented in Table 3 and in
 256 boxplots (median and percentiles) in Figure 2. $\delta^{13}\text{C}_{\text{DIC}}$ ranged from -9.3 to +1.5 ‰ and $\delta^{13}\text{C}_{\text{CO}_2}$
 257 (derived from $\delta^{13}\text{C}_{\text{DIC}}$) was on average 10.9 ± 0.3 ‰ depleted relative to DIC, with a range of -
 258 20.8 to -8.9 ‰. $\delta^{13}\text{C}_{\text{DIC}}$ and $\delta^{13}\text{C}_{\text{CO}_2}$ showed no correlation with alkalinity, DIC, pH, temperature
 259 and lake area. A weak negative relation between pCO_2 and $\delta^{13}\text{C}_{\text{DIC}}$ was observed, which was
 260 stronger in oligo-mesotrophic lakes than in eutrophic lakes (Fig. 3A, Table 2). The highest pCO_2
 261 lakes had the most depleted $\delta^{13}\text{C}_{\text{DIC}}$, suggesting that respiration of organic matter influenced
 262 $\delta^{13}\text{C}_{\text{DIC}}$. Low CO_2 lakes had enriched $\delta^{13}\text{C}_{\text{DIC}}$, indicating influence of primary production. Weak,
 263 but significant relations were observed for POC and DOC with $\delta^{13}\text{C}_{\text{DIC}}$ (Table 2). In eutrophic
 264 lakes, $\delta^{13}\text{C}_{\text{DIC}}$ increased with increasing POC (Fig. 3B, Table 2), while in oligo-mesotrophic
 265 lakes, $\delta^{13}\text{C}_{\text{DIC}}$ decreased with increasing DOC (Fig. 3C, Table 2).

266 3.4 $\delta^{13}\text{C}$ of organic carbon pools

267 The isotopic composition of DOC ($\delta^{13}\text{C}_{\text{DOC}}$) had the narrowest range of all carbon pools,
 268 only -28.8 to -27.0 ‰ (mean -28.0 ‰) in the oligo-mesotrophic lakes and a slightly larger range
 269 of -27.6 to -23.7 ‰ (mean -25.4 ‰) in the eutrophic lakes (Fig. 2, Table 3). The $\delta^{13}\text{C}$ isotopic
 270 range of POC ($\delta^{13}\text{C}_{\text{POC}}$) was larger than of DOC in both lake types and on average 2.0 ‰
 271 depleted compared to $\delta^{13}\text{C}_{\text{DOC}}$, with mean values of -27.8 ± 3.6 ‰ in eutrophic and -29.7 ± 2.8
 272 ‰ in oligo-mesotrophic lakes (Fig. 2, Table 3). $\delta^{13}\text{C}$ of particulate glucose ($\delta^{13}\text{C}_{\text{gluc}}$), the most
 273 abundant carbohydrate, was always enriched compared to $\delta^{13}\text{C}_{\text{POC}}$ and the enrichment was
 274 similar in eutrophic (3.1 ± 1.7 ‰) and oligo-mesotrophic lakes (2.8 ± 1.5 ‰) (Fig. 2). On the
 275 contrary, the concentration-weighted average $\delta^{13}\text{C}$ of all fatty acids ($\delta^{13}\text{C}_{\text{FA}_{\text{tot}}}$) was always
 276 depleted compared to $\delta^{13}\text{C}_{\text{POC}}$ (Fig. 2). The depletion of $\delta^{13}\text{C}_{\text{FA}_{\text{tot}}}$ relative to POC was higher in
 277 eutrophic lakes (5.2 ± 1.8 ‰) than in oligo-mesotrophic lakes (3.1 ± 1.1 ‰). The isotopic
 278 difference between glucose and $\delta^{13}\text{C}_{\text{FA}_{\text{tot}}}$: $\Delta\delta^{13}\text{C}_{\text{gluc-FA}_{\text{tot}}}$ was highly variable with a range of 1.6 to
 279 14.6 ‰. The isotopic differences between glucose and $\delta^{13}\text{C}_{\text{FA}_{\text{tot}}}$ did not correlate with CO_2 , but
 280 weakly with nutrient levels, i.e., $\Delta\delta^{13}\text{C}_{\text{gluc-FA}_{\text{tot}}}$ increased with increasing TP (Table 2).

281 There was a large variability among $\delta^{13}\text{C}$ of different FA with some consistent
 282 differences over all lakes. Compared to the $\delta^{13}\text{C}$ of 16:0 (the most abundant FA), the bacterial

283 FA markers were always enriched by 1.4 - 5.0 ‰ (e.g., the *iso*-15:0 FA in Fig. 4), therefore the
284 overall $\delta^{13}\text{C}$ of bacterial FA ($\delta^{13}\text{C}_{\text{bac}}$) was more enriched than $\delta^{13}\text{C}_{\text{FA}_{\text{tot}}}$ in both lake systems (Fig.
285 2). The poly-unsaturated fatty acids (PUFA) used as markers for phytoplankton showed
286 consistent differences throughout the lakes. DHA (22:6 ω 3), common in dinoflagellates
287 (Dalsgaard et al. 2003), was found to be enriched with 4.6 ‰ compared to 16:0 while PLFA
288 18:3 ω 3 (α -linolenic acid), common in cyanobacteria (de Kluijver et al., 2012), was 4.7 ‰
289 depleted compared to 16:0 (Fig. 4). The other phytoplankton markers were not statistically
290 different from 16:0. The weighted $\delta^{13}\text{C}$ of phytoplankton FA ($\delta^{13}\text{C}_{\text{phyto}}$) was the most depleted of
291 all carbon pools (Fig. 2, Table 3) with an average of -33.8 ± 5.3 ‰ in eutrophic lakes and $-33.4 \pm$
292 3.5 ‰ in oligo-mesotrophic lakes. $\delta^{13}\text{C}_{\text{bac}}$ were on average 4.7 ‰ enriched compared to $\delta^{13}\text{C}_{\text{phyto}}$
293 (Fig. 2).

294 3.5 Carbon isotopic composition of phytoplankton

295 $\delta^{13}\text{C}_{\text{phyto}}$ depends on the isotopic composition of substrate ($\delta^{13}\text{C}_{\text{CO}_2}$), the isotope
296 fractionation ($\epsilon_{\text{CO}_2\text{-phyto}}$) associated with primary production and the isotopic difference between
297 PUFA's and biomass. $\delta^{13}\text{C}_{\text{phyto}}$ in the eutrophic lakes became more enriched with increasing
298 C_{phyto} (Fig. 5A, Table 2) and decreasing pCO_2 (Fig. 5B, Table 2). No relation between $\delta^{13}\text{C}_{\text{phyto}}$
299 and C_{phyto} was observed in the oligo-mesotrophic lakes (Fig. 5A), but there was a strong negative
300 relation with pCO_2 (Fig. 5B, Table 2). The influence of C_{phyto} on $\delta^{13}\text{C}_{\text{phyto}}$ in the eutrophic lakes
301 was also reflected in fractionation; $\epsilon_{\text{CO}_2\text{-phyto}}$ was highly variable in eutrophic lakes, while it was
302 less variable in oligotrophic lakes (Fig. 5C). The range of $\epsilon_{\text{CO}_2\text{-phyto}}$ was 7.8 to 24.7 ‰ (mean 16.9
303 ‰) in eutrophic and 11.7 to 18.8 ‰ (mean 17.1 ‰) in oligo-mesotrophic lakes, when
304 $\delta^{13}\text{C}_{\text{phyto_cor}}$ was used (Table 3). The less variable ϵ in oligo-mesotrophic lakes resulted in a
305 strong correlation between $\delta^{13}\text{C}_{\text{CO}_2}$ and $\delta^{13}\text{C}_{\text{phyto}}$ (Table 2), which was absent in the eutrophic
306 lakes. $\epsilon_{\text{CO}_2\text{-phyto}}$ correlated negatively with C_{phyto} in eutrophic lakes, however (Table 2). The
307 variability in $\delta^{13}\text{C}_{\text{phyto}}$ in eutrophic lakes can be mainly attributed to the presence of two clusters:
308 a ^{13}C -enriched cluster at the highest C_{phyto} and a depleted cluster at lower C_{phyto} (Fig. 5A).
309 Interestingly, the eutrophic lakes within the enriched cluster also had high concentrations of
310 zeaxanthin, a marker pigment for cyanobacteria (data not shown here).

311 4. Discussion

312 4.1 Lake metabolism, pCO₂ and δ¹³C_{DIC}

313 In our study, about 3/4 of the lakes were supersaturated with pCO₂, consistent with the
314 literature that lakes generally emit carbon dioxide to the atmosphere (Cole et al., 1994, Cole et
315 al., 2007). This CO₂ excess can be due to in-lake respiration of terrestrially derived organic
316 carbon outbalancing carbon dioxide fixation by primary producers (negative metabolic balance)
317 or due to river and groundwater input of CO₂-rich waters (McDonald et al., 2013). Lake
318 metabolism also impacts δ¹³C_{DIC} dynamics. Previous studies have shown that δ¹³C of DIC in
319 lakes is driven by carbonate chemistry, hydrology (i.e., groundwater inflow), and metabolic
320 activity (Bade et al., 2004). Primary production enriches δ¹³C_{DIC} because of the preferential
321 uptake of ¹²C (isotope fractionation), while organic matter respiration depletes δ¹³C_{DIC} (Fry,
322 2006).

323 If pCO₂ and δ¹³C_{DIC} would have been only or primarily controlled by the balance between
324 respiration and production of organic matter, one would expect a tight correlation between
325 δ¹³C_{DIC} and pCO₂, which was not observed overall (Fig. 3A, Table 2), indicating that other
326 factors are involved. High inorganic carbon loadings of inflowing rivers and groundwater inputs
327 may sustain the CO₂ excess (McDonald et al., 2013) and govern the δ¹³C_{DIC} (Bade et al., 2004).
328 Moreover, carbon dioxide water-air exchange reactions may have modified δ¹³C_{DIC} values
329 because of isotope fractionation during water-air exchange, in particular at high pH/low CO₂
330 values (Herczeg and Fairbanks, 1987; Bade et al., 2004; Bontes et al., 2006).

331 The correlation between δ¹³C_{DIC} and pCO₂ in oligo-mesotrophic lakes was stronger (Table 2) and
332 pCO₂ was highest and δ¹³C_{DIC} was most depleted at the highest DOC in oligo-mesotrophic lakes
333 (Fig. 1C, 3C). Such a depletion of δ¹³C_{DIC} with increasing DOC, as an indicator of the
334 importance of respiration in oligo-mesotrophic lakes, has been shown previously in North-
335 American lakes (Lennon et al., 2006). In addition to community respiration, methanotrophic
336 bacteria in high DOC lakes could decrease δ¹³C_{DIC} (Jones et al. 1999). However, anoxic
337 hypolimnia are rare in these lakes, either due to low nutrients or polymixis, indicating that
338 methanogenesis was not of major importance in the lakes investigated. Furthermore, we

339 examined the $\delta^{13}\text{C}$ of fatty acids abundant in or specific to methanotrophs and these were not
340 more depleted in ^{13}C than other fatty acids.

341 Note that the lakes were only sampled at one point location and depth, representing
342 average conditions, so spatial variability per lake is not taken into account. Also diurnal variation
343 and variation over the year in each lake are not considered in this study. However, the aim of our
344 study is comparing snapshots of different lakes representing different metabolic states and not to
345 describe the biogeochemistry of each individual lake. So, although we miss some variation, this
346 shouldn't affect our main findings on carbon isotope biogeochemistry of these lakes.

347 **4.2 Allochthonous $\delta^{13}\text{C}$**

348 Dissolved and particulate organic carbon pools are mixtures of organic matter from
349 various sources with potentially different stable carbon isotopic compositions. The particulate
350 organic carbon pool comprises biomass from living organisms and remains from organisms
351 within the lake as well as allochthonous detritus. The relative importance of living biomass to
352 total POC pool, calculated based on equation 4, varies from 5.7 to 93 % (Table 1), with on
353 average about 53 ± 20 % in eutrophic lakes and only 13 ± 5 % in oligo-mesotrophic lakes. In
354 our study we have explicit carbon isotope data for the most important living compartments
355 (algae, bacteria and zooplankton; De Kluijver, 2012), but we have no direct measurement of the
356 $\delta^{13}\text{C}$ of organic carbon delivered from the watershed via atmospheric, riverine and groundwater
357 inputs. We have therefore used two proxies for $\delta^{13}\text{C}_{\text{allo}}$; the carbon isotope ratio of dissolved
358 organic carbon and that of detrital particulate organic carbon calculated by difference (equations
359 3, 4). Both proxies for $\delta^{13}\text{C}_{\text{allo}}$ provide an estimate for the total detrital pool, i.e. the sum of
360 aquatic and terrestrial detritus.

361 The oligo-mesotrophic lakes are surrounded by forest (C_3 vegetation) and $\delta^{13}\text{C}_{\text{DOC}}$ was -
362 28.0 ± 0.5 ‰, corresponding to a C_3 vegetation signal, suggesting that the dissolved organic
363 carbon pool is dominantly terrestrially derived, consistent with a combined carbon and hydrogen
364 isotope study of Wisconsin and Michigan lakes by Wilkinson et al., (2013). The other proxy for
365 allochthonous carbon, $\delta^{13}\text{C}_{\text{det}}$, was slightly more negative (-29.6 ± 2.1 ‰), but the two proxies
366 for allochthonous carbon were well correlated (Table 2). The 1.6 ‰ lighter isotopic composition
367 might reflect a relatively larger contribution of autochthonous detritus to the total detrital

368 particulate organic matter pool than to the dissolved pool. Consistently, Wilkinson et al., (2013)
369 reported that a lower contribution of terrestrial organic matter to the particulate pool than to the
370 dissolved organic matter pools in North American lakes.

371 Allochthonous carbon proxies in eutrophic lakes were more enriched and variable: -25.4
372 ± 1.1 ‰ for $\delta^{13}\text{C}_{\text{DOC}}$ and -26.6 ± 4.2 ‰ for $\delta^{13}\text{C}_{\text{det}}$. Moreover, $\delta^{13}\text{C}_{\text{DOC}}$ and $\delta^{13}\text{C}_{\text{det}}$ were not
373 significantly correlated. The enrichment of $\delta^{13}\text{C}_{\text{allo}}$ in eutrophic lakes can be partly explained by
374 land use in the water shed; almost all eutrophic lakes were located in the state of Iowa, where an
375 average of 92% of the land is under periodic cultivation for maize (C_4 plants, -14 ‰). There was
376 more uncertainty in $\delta^{13}\text{C}_{\text{allo}}$ in the eutrophic lakes for two main reasons. First, we expect a
377 substantial autochthonous contribution to DOC and detritus in productive lakes (Bade et al.,
378 2007), which contributes to the larger range in $\delta^{13}\text{C}_{\text{DOC}}$ and $\delta^{13}\text{C}_{\text{det}}$ (Fig. 2). Second, the presence
379 of C_3 and C_4 vegetation with their distinct isotopic compositions can create a variable $\delta^{13}\text{C}_{\text{allo}}$.
380 Variability in $\delta^{13}\text{C}_{\text{allo}}$ has received far less attention than that of aquatic primary producers. Our
381 results show distinct differences in the isotopic composition of external subsidies and argue
382 against a fixed value for allochthonous carbon, especially in areas with abundant C_4 vegetation,
383 such as maize.

384 **4.3 Phytoplankton $\delta^{13}\text{C}$**

385 The determination of $\delta^{13}\text{C}_{\text{phyto}}$ is one of the major challenges in aquatic ecology. Fatty
386 acid biomarkers as proxies for $\delta^{13}\text{C}_{\text{phyto}}$ have the advantage that there is a larger certainty that
387 measured $\delta^{13}\text{C}$ values represent parts of phytoplankton carbon. The main uncertainty using
388 $\delta^{13}\text{C}_{\text{FA}}$ as marker for $\delta^{13}\text{C}_{\text{phyto}}$ comes from the isotopic offset between lipids and total cells
389 ($\Delta\delta^{13}\text{C}_{\text{FA-cell}}$) which depends on species composition (summarized in, e.g., Hayes, 2001), growth
390 conditions (e.g., Riebesell et al., 2000) and the FA considered (Fig. 5).

391 Isotope fractionation between CO_2 and phytoplankton was variable ($8 - 25$ ‰) in our
392 study (Table 3). This implies that calculations of $\delta^{13}\text{C}_{\text{phyto}}$ from $\delta^{13}\text{C}_{\text{CO}_2}$ with a constant
393 fractionation factor provides inaccurate results, consistent with methodological comparisons by
394 Marty and Planas, (2008) and McCallister et al., (2008). The usual value for photosynthetic
395 fractionation in phytoplankton is ~ 20 ‰, based on C_3 photosynthesis (Fry, 2006), but several
396 studies that determined ϵ in lakes showed that actual fractionation is usually lower than this value

397 (Cole et al., 2002; Bade et al., 2006). Also, in our study, fractionation was lower (~17 ‰) on
398 average, and highly variable, especially in eutrophic lakes. There are several explanations for this
399 variability. 1) Actual fractionation has been shown to be dependent on several variables,
400 including growth rate (Bidigare et al., 1997) and CO₂ availability (Laws et al., 1995).
401 Fractionation is highest under high CO₂ availability and low growth rates. In the less productive
402 oligo-mesotrophic lakes, the conditions favor optimal fractionation, and therefore, fractionation
403 was rather constant (Fig. 5C). In the productive, eutrophic lakes, actual fractionation was
404 influenced by pCO₂ and C_{phyto}, with lowest fractionation and therefore most enriched ¹³C
405 phytoplankton in the most productive (low CO₂ and high C_{phyto}) lakes (Fig. 5A, 5B).

406 Two clusters in δ¹³C_{phyto} were present in the eutrophic lakes (Fig. 5A, 5B) and the shift
407 occurred when lakes were below 20 μmol L⁻¹ CO₂ in the eutrophic lakes. When CO₂ becomes
408 limiting, some phytoplankton can also shift to bicarbonate utilisation, which is isotopically
409 enriched by ~8 ‰ compared to CO₂. Direct uptake of carbonate and conversion in the
410 carboxysomes is very common in the Cyanobacteria that dominate eutrophic lakes (Bontes et al.,
411 2006). The lakes with ¹³C-enriched phytoplankton had high concentrations of zeaxanthin, a
412 biomarker for cyanobacteria. However, the enrichment in δ¹³C_{phyto} in high zeaxanthin lakes was
413 not a direct consequence of enrichment in Cyanobacteria. FA that are abundant in Cyanobacteria
414 (18:non) were not more enriched than FA that are absent in Cyanobacteria; in fact, they were the
415 most depleted of all FA (Fig 4). Cyanobacteria grown in laboratory cultures also showed higher
416 fractionation (up to 9 ‰) in lipids relative to total biomass than eukaryotic phytoplankton
417 (summarized in Hayes, 2001).

418 The most enriched phytoplankton FA was 22:6ω3, which is abundant in dinoflagellates
419 (Fig. 4) (Dalsgaard et al., 2003). Dinoflagellates were also more enriched in ¹³C compared to
420 other phytoplankton in a subtropical lake (Zohary et al., 1994). A possible explanation for ¹³C-
421 enriched dinoflagellates in field studies, can be their mixotrophic character, so that part of their
422 isotopic composition reflects consumer δ¹³C. However, PUFAs of autotrophic dinoflagellates
423 grown in continuous cultures were also more enriched to C16:0 than PUFA of other
424 phytoplankton (Schouten et al., 1998).

425 Finally, variability in Δδ¹³C_{FA-cell} can contribute to the observed variability. In laboratory
426 studies, the offset between lipids and bulk material has shown to be variable (van Dongen et al.,

427 2002, Fiorini et al., 2010). One can expect that in field studies, with multiple species, however,
428 these cellular variations would probably disappear within broader trends. If we assume an overall
429 mean $\Delta\delta^{13}\text{C}_{\text{FA-cell}}$, then the uncertainty in the actual value would affect the absolute fractionation
430 values, but not the observed variability in fractionation.

431 **4.4 Carbohydrates and lipid $\delta^{13}\text{C}$**

432 The enrichment of carbohydrates and depletion of lipids relative to total cells (mainly
433 amino acids) has been shown in culture studies of phytoplankton (Van Dongen et al., 2002) and
434 in culture studies of several primary producers and consumers (Teece and Fogel, 2007). Results
435 of this study show that the enrichment in glucose as well as the depletion in fatty acids relative to
436 bulk material can also be detected in field samples (Fig. 2, Table 3). We observed that
437 $\Delta\delta^{13}\text{C}_{\text{gluc-FA}_{\text{tot}}}$ increased with TP (Table 2), but whether this represents a general phenomenon for
438 lakes needs further exploration.

439 Bacterial FA were more enriched than phytoplankton FA in all lakes (Fig. 2, Fig. 4). This
440 observation can be explained by differences in carbon source or differences in $\Delta\delta^{13}\text{C}_{\text{FA-cell}}$
441 between phytoplankton and bacteria. Carbohydrates, present in high concentrations in DOC,
442 form an important carbon source for bacteria. Since carbohydrates were the most enriched
443 carbon source, a preferential use of carbohydrates, would result in ^{13}C enriched bacteria (Fig. 2).
444 Another explanation is that isotope fractionation during FA synthesis was smaller in bacteria
445 compared to phytoplankton. There are no field studies on $\Delta\delta^{13}\text{C}_{\text{FA-cell}}$ in freshwater bacteria, but
446 field studies on sediment and marine bacteria report a range of 0 – 5 ‰ in $\Delta\delta^{13}\text{C}_{\text{FA-cell}}$ (Hayes,
447 2001, Burke et al., 2003; Bouillon and Boschker, 2006). Burke et al., (2003) suggested that in
448 field samples with complex communities and substrates, $\Delta\delta^{13}\text{C}_{\text{FA-cell}}$ would be ~0 ‰. The results
449 of our study support this idea, since bacterial FA had a similar $\delta^{13}\text{C}$ as POC. If a similar $\Delta\delta^{13}\text{C}_{\text{FA-}}$
450 $_{\text{cell}}$ for phytoplankton and bacteria would be used, bacteria would be more enriched than its
451 potential carbon sources in half of the studied lakes, which is rather unlikely.

452

453

454

455 **5. Conclusions**

456 Our results show that trophic state has a large influence on lake metabolism and carbon
457 cycling in plankton food webs. Overall, eutrophic lakes had larger variability in $\delta^{13}\text{C}$ in all
458 organic carbon pools than oligo-mesotrophic lakes, caused by larger isotopic variability in the
459 base of the food web in eutrophic lakes (both allochthonous and autochthonous carbon). In
460 eutrophic lakes, $\delta^{13}\text{C}_{\text{phyto}}$ showed that two clusters of phytoplankton were present, with the most
461 enriched phytoplankton at high CO_2 and high chl *a*. Dominance of cyanobacteria played a role,
462 but enrichment was present in all phytoplankton, as seen in specific PLFA.

463

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474

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Table 1: Limnological characteristics of the sampled lakes. pCO₂ was determined from temperature, DIC, and pH. C_{phyto} presents the average of chl *a* and fatty acid based phytoplankton biomass. C_{bac} presents fatty acid derived bacteria carbon biomass.

name	U.S. state	trophic state	temperature (°)	pH	Alkalinity (mmol L ⁻¹)	DIC (mmol L ⁻¹)	pCO ₂ (µatm)	TN (mg L ⁻¹)	TP (µg L ⁻¹)	Chl <i>a</i> (µg L ⁻¹)	DOC (mg L ⁻¹)	POC (mg L ⁻¹)	C _{phyto} (mg L ⁻¹)	C _{bac} (mg L ⁻¹)
Beaver	I	Eu	24.1	9.5	1.03	1.70	30	1.69	152.8	72.6	2.10	5.07	2.26	0.251
Beeds	I	Eu	23.8	8.5	2.28	4.15	835	9.32	36.1	9.1	0.97	0.93	0.38	0.023
Big Creek	I	Eu	25.0	8.5	1.97	3.89	795	6.12	21.3	8.0	1.56	1.30	0.32	0.047
Coralville	I	Eu	24.2	7.8	2.37	4.55	4545	6.50	207.1	10.9	1.16	1.91	0.49	0.074
Little Splithand	M	Eu	19.2	8.2	1.04	1.69	635	0.01	29.5	19.5	3.70	2.07	0.58	0.026
Lower Pine	I	Eu	24.1	8.6	1.44	2.51	400	4.15	128.3	60.2	1.46	4.00	2.32	0.213
McBride	I	Eu	22.0	8.8	1.18	1.98	195	1.27	67.9	42.6	1.71	2.84	1.80	0.103
Meyers	I	Eu	27.1	9.8	1.08	1.49	10	2.14	208.7	86.2	2.70	9.91	3.65	0.190
Rodgers park	I	Eu	24.6	8.4	1.82	3.34	855	6.81	50.6	5.4	1.20	0.68	0.23	0.038
Saylorville	I	Eu	25.3	8.5	2.05	4.03	830	4.90	116.3	23.0	1.80	1.30	0.96	0.165
Three Mile	I	Eu	21.3	9.1	0.98	1.69	80	1.00	44.9	37.8	1.90	2.66	1.48	0.122
Beaver	M	O-M	19.7	6.9	0.09	0.09	520	0.12	16.2	3.6	4.34	1.50	0.13	0.017
Brush Shanty	M	O-M	20.3	7.4	0.28	0.29	655	0.34	10	1.4	4.75	0.56	0.06	0.014
Hatch	M	O-M	20.3	8.4	1.81	3.04	735	0.55	1.9	0.8	2.08	0.30	0.04	0.008
Horsehead	M	O-M	19.9	6.7	0.07	0.06	460	0.09	6.9	1.8	2.91	2.11	0.09	0.033
Kelly	M	O-M	20.2	6.9	0.08	0.05	310	0.01	8.9	2.6	2.50	1.22	0.09	0.017
Leighton	M	O-M	19.4	8.4	1.84	2.99	715	0.49	8.1	2.4	2.62	0.67	0.07	0.007
Little Sand	M	O-M	22.8	8.2	0.75	1.88	745	0.19	11.9	4.9	3.68	1.24	0.17	0.023
O'Leary	M	O-M	19.2	6.7	0.09	0.08	655	0.44	13.8	2.6	3.18	2.28	0.10	0.029
Sand Lake	M	O-M	19.9	7.7	1.18	0.61	710	0.66	20.4	3.1	3.81	0.85	0.12	0.016
South Sturgeon	M	O-M	18.6	6.1	0.22	0.20	3200	0.03	14.4	3.6	6.71	0.55	0.11	0.005
Thirty	M	O-M	22.0	7.1	0.13	0.12	525	0.10	15.7	6.3	3.65	3.02	0.23	0.066

Table 2: Significant correlation coefficients (r) between tested variables in all lakes and in eutrophic and oligo-mesotrophic lakes separately. Significance levels: * $p < 0.05$, ** $p < 0.01$, *** $p < 0.001$.

Variables	Eutrophic lakes (n=11) r	Oligo- mesotrop hic lakes (n=11) r	overall (n=22) r
log alkalinity and log pCO ₂	0.79 **		
DIC and log pCO ₂	0.82 **		
pH and log pCO ₂	-0.98 ***		
log C _{phyto} and log pCO ₂	-0.80 **		-0.59 ***
DOC and log pCO ₂		0.75 **	
log TP and log C _{phyto}	0.61 *	0.77 **	0.89 ***
log TP and log C _{bac}	0.74 **		0.82 ***
log pCO ₂ and $\delta^{13}\text{C}_{\text{DIC}}$	-0.63 *	-0.81 **	-0.48 *
POC and $\delta^{13}\text{C}_{\text{DIC}}$	0.61 *		
DOC and $\delta^{13}\text{C}_{\text{DIC}}$		-0.62 *	
log TP and $\Delta(\delta_{\text{gluc}} - \delta_{\text{FA}})$			0.52 *
log C _{phyto} and $\delta^{13}\text{C}_{\text{phyto}}$	0.90 ***		
log pCO ₂ and $\delta^{13}\text{C}_{\text{phyto}}$	-0.79 **	-0.90 ***	-0.73 ***
$\delta^{13}\text{C}_{\text{CO}_2}$ and $\delta^{13}\text{C}_{\text{phyto}}$		0.82 **	0.54 **
log C _{phyto} and ε	-0.70 *		
$\delta^{13}\text{C}_{\text{DOC}}$ and $\delta^{13}\text{C}_{\text{detritus}}$		0.79 **	0.54 *

Table 3: Carbon isotope values of sampled lakes. Isotope data are presented as average \pm sd (n=3). $\delta^{13}\text{C}_{\text{CO}_2}$ was calculated from $\delta^{13}\text{C}_{\text{DIC}}$ (equation 1). $\delta^{13}\text{C}_{\text{phyto}}$ and $\delta^{13}\text{C}_{\text{bac}}$ are not corrected for the offset between fatty acids and total cells, but $\epsilon_{\text{CO}_2\text{-algae}}$ (equation 2) used the corrected $\delta^{13}\text{C}$ of phytoplankton.

lake name	trophic state	$\delta^{13}\text{C}_{\text{DIC}}$ (‰)	$\delta^{13}\text{C}_{\text{CO}_2}$ (‰)	$\delta^{13}\text{C}_{\text{POC}}$ (‰)	$\delta^{13}\text{C}_{\text{DOC}}$ (‰)	$\delta^{13}\text{C}_{\text{Phyto}}$ (‰)	$\epsilon_{\text{CO}_2\text{-algae}}$	$\delta^{13}\text{C}_{\text{bac}}$ (‰)	$\delta^{13}\text{C}_{\text{gluc}}$ (‰)
Beaver (I)	Eu	-4.2 \pm 0.6	-15.2	-21.9 \pm 0.2	-24.2 \pm 0.0	-25.9 \pm 1.9	7.8	-22.4 \pm 0.3	-19.2
Beeds (I)	Eu	-2.4 \pm 0.1	-13.1	-32.2	-23.7 \pm 0.1	-38.0 \pm 2.5	22.7	-29.1 \pm 0.0	-31.1
Big Creek (I)	Eu	-3.2 \pm 0.1	-14.1	-32.2 \pm 0.1	-26.4 \pm 0.9	-40.8 \pm 2.2	24.7	-31.5 \pm 0.7	-30.4
Coralville (I)	Eu	-6.4 \pm 0.0	-17.6	-27.9 \pm 0.5	-24.6 \pm 0.3	-37.4 \pm 1.5	17.4	-29.4 \pm 0.9	-21.4
Little Split-hand (M)	Eu	-3.8 \pm 0.1	-14.7	-32.3 \pm 0.4	-27.6 \pm 0.4	-37.0 \pm 0.5	20.0	-31.7 \pm 0.7	-27.3
Lower Pine (I)	Eu	-2.4 \pm 0.1	-13.2	-26.2 \pm 0.2	-25.3 \pm 0.5	-30.6 \pm 1.1	14.8	-27.2 \pm 0.8	-24.0
McBride (I)	Eu	-2.8 \pm 0.1	-13.6	-25.6 \pm 0.4	-25.6 \pm 0.3	-27.3 \pm 0.6	11.0	-27.2 \pm 1.0	-21.5
Meyers (I)	Eu	1.0 \pm 0.3	-9.5	-22.1 \pm 0.1	-26.2 \pm 0.6	-29.3 \pm 1.7	17.3	-24.5 \pm 0.6	-19.7
Rodgers park (I)	Eu	-6.1 \pm 0.0	-17.3	-29.0	-24.8 \pm 0.3	-38.6	19.0	-30.3	-26.3
Saylorville (I)	Eu	-4.1 \pm 0.0	-15.1	-29.2 \pm 0.1	-25.8 \pm 1.2	-37.3 \pm 1.4	19.8	-29.5 \pm 0.4	-28.1
Three Mile (I)	Eu	-4.3 \pm 0.3	-15.3	-27.8 \pm 0.4	-25.5 \pm 1.0	-29.8 \pm 2.3	11.9	-28.5 \pm 0.6	-23.7
Beaver (M)	O-M	-4.1 \pm 0.5	-15.1	-30.5 \pm 0.2	-28.7 \pm 0.2	-32.4 \pm 0.2	14.8	-30.4 \pm 2.0	-28.3
Brush Shanty (M)	O-M	-2.5 \pm 0.2	-13.3	-29.6 \pm 0.2	-28.1 \pm 0.4	-33.6 \pm 2.1	17.9	-30.0 \pm 3.0	-25.0
Hatch (M)	O-M	-3.7 \pm 0.1	-14.6	-31.4 \pm 0.4	-28.8 \pm 0.2	-34.5 \pm 1.3	17.4	-30.6 \pm 2.4	-28.6
Horsehead (M)	O-M	-3.7 \pm 0.9	-14.7	-25.0 \pm 0.1	-27.0 \pm 0.1	-29.1 \pm 0.7	11.7	-27.8 \pm 2.0	
Kelly (M)	O-M	1.5 \pm 0.5	-8.9	-27.8 \pm 0.7	-27.9 \pm 0.4	-30.1 \pm 1.0	18.7	-26.6 \pm 2.6	-22.1
Leighton (M)	O-M	-3.3 \pm 0.0	-14.2	-32.3 \pm 0.3	-28.2 \pm 0.7	-34.4 \pm 1.3	17.7	-30.9 \pm 1.3	-31.8
Little Sand (M)	O-M	-2.2 \pm 0.1	-12.9	-30.4 \pm 0.2	-28.1 \pm 0.0	-33.7 \pm 1.7	18.4	-29.4 \pm 1.8	
O'Leary (M)	O-M	-1.3 \pm 0.7	-12.0	-28.1 \pm 0.3	-27.4 \pm 0.7	-31.3 \pm 0.4	16.7	-26.8 \pm 0.8	-26.3
Sand Lake (M)	O-M	-4.7 \pm 0.1	-15.7	-31.5 \pm 0.4	-28.0 \pm 0.1	-36.4 \pm 0.1	18.3	-29.6 \pm 1.8	-28.7
South Sturgeon (M)	O-M	-9.3 \pm 0.1	-20.8	-31.8 \pm 0.9	-28.0 \pm 0.0	-41.4 \pm 1.4	18.2	-30.5 \pm 1.2	-29.4
Thirty (M)	O-M	1.5 \pm 0.6	-8.9	-28.7 \pm 0.1	-27.8 \pm 0.1	-30.2 \pm 1.4	18.8	-32.0 \pm 1.2	-26.4

Figure legends

Figure 1: The relation of $p\text{CO}_2$ in eutrophic lakes (filled circles, $n=11$) and oligo-mesotrophic lakes (open circles, $n=11$) to A) DIC; B) pH; C) C_{phyto} ; D) DOC. The dashed line indicates atmospheric $p\text{CO}_2$ ($385 \mu\text{atm}$).

Figure 2: Box and whisker plot of the $\delta^{13}\text{C}$ of inorganic and organic carbon pools in eutrophic lakes (grey boxes, $n=11$) and oligo-mesotrophic lakes (white boxes, $n=11$). The dashed lines present typical values for C_3 and C_4 vegetation.

Figure 3: The relation of $\delta^{13}\text{C}_{\text{DIC}}$ in eutrophic lakes (filled circles, $n=11$) and oligo-mesotrophic lakes (open circles, $n=11$) to A) $p\text{CO}_2$; B) C_{phyto} ; C) DOC.

Figure 4: Box and whiskerplots of the $\Delta\delta^{13}\text{C}$ of individual biomarker fatty acids relative to 16:0 FA ($\delta^{13}\text{C}_{\text{FA}} - \delta^{13}\text{C}_{16:0}$) in all lakes.

Figure 5: The relation of $\delta^{13}\text{C}_{\text{phyto}}$ in eutrophic lakes (filled circles, $n=11$) and oligo-mesotrophic lakes (open circles, $n=11$) to A) C_{phyto} and B) $p\text{CO}_2$. Panel C presents a box whisker plot of calculated $\varepsilon_{\text{CO}_2-\text{algae}}$ in eutrophic (Eu) and oligo-mesotrophic (Oli-Meso) lakes.

Fig. 1

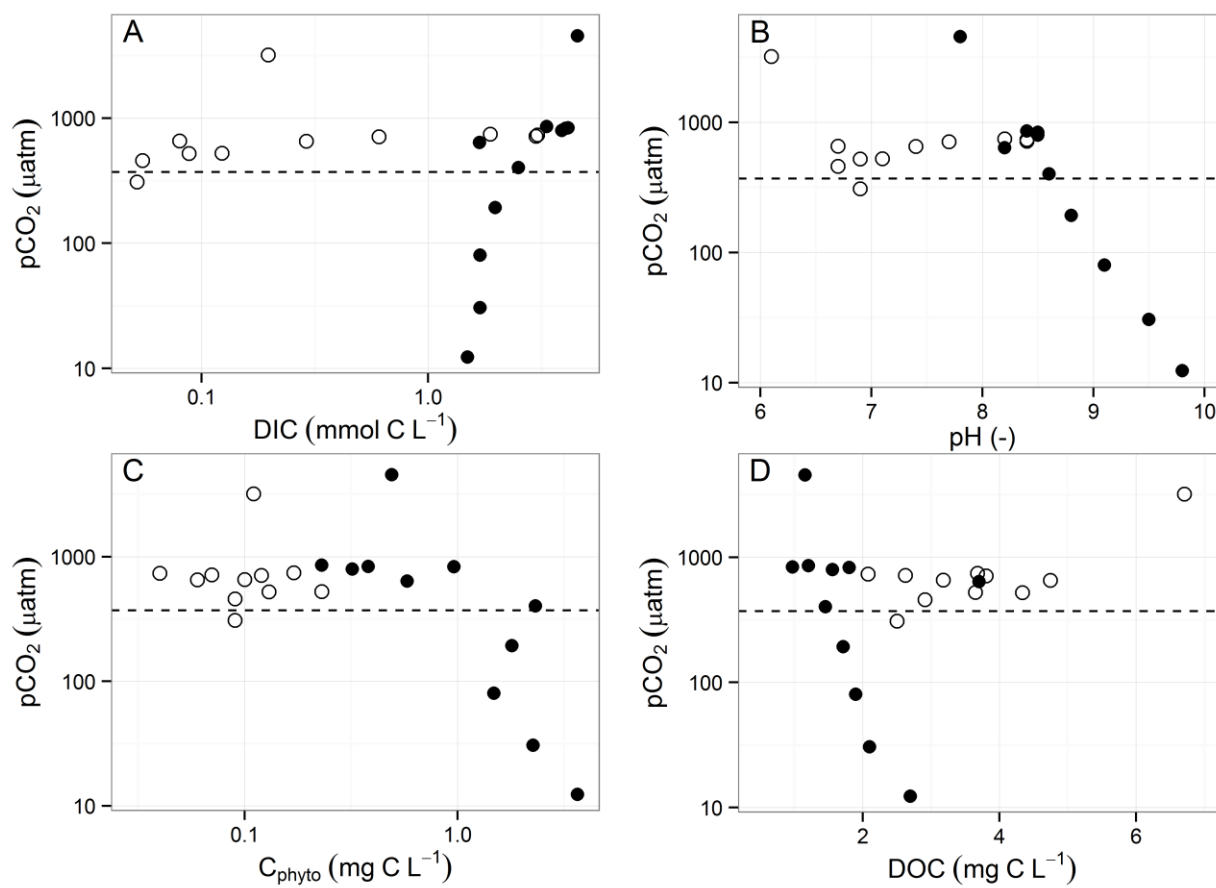


Fig. 2

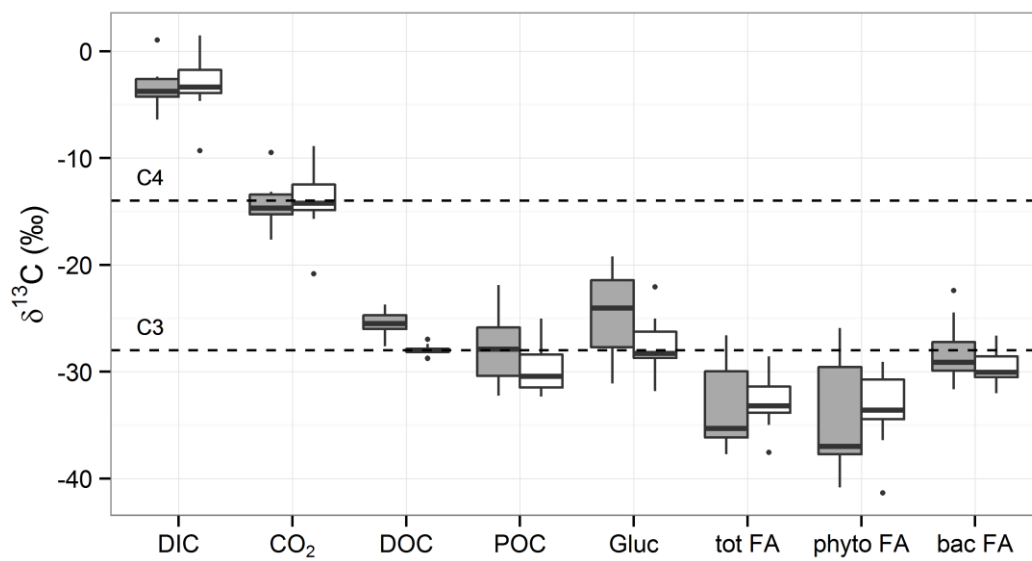


Fig. 3

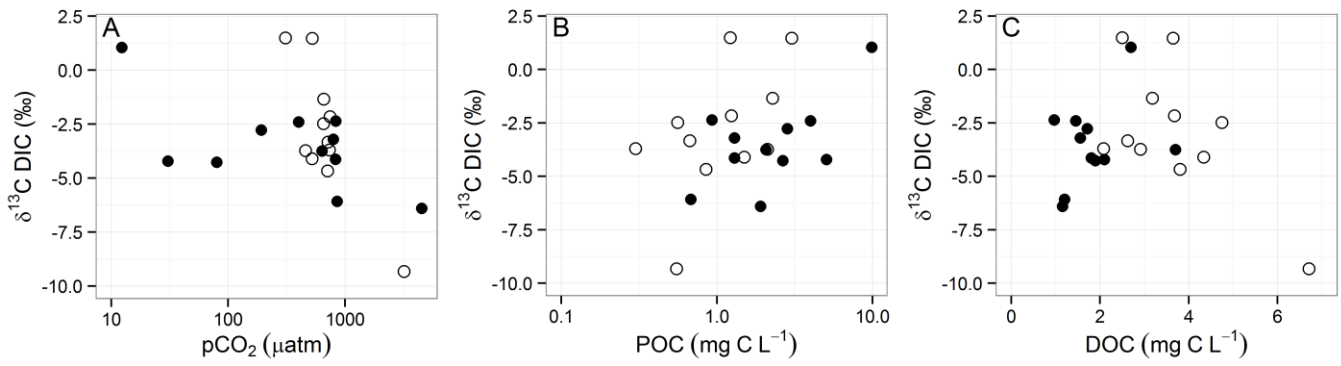


Fig. 4

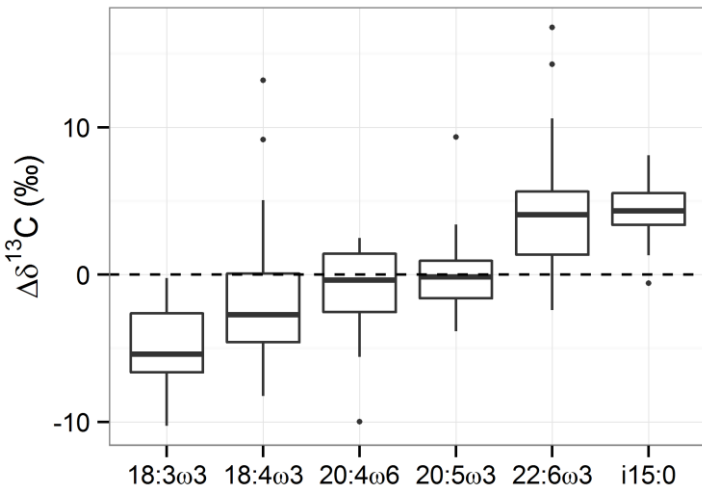


Fig. 5

