

# 1 Large methyl halide emissions from south Texas salt marshes

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## 10 11 Abstract

12 Coastal salt marshes are natural sources of methyl chloride (CH<sub>3</sub>Cl) and methyl bromide  
13 (CH<sub>3</sub>Br) to the atmosphere, but measured emission rates vary widely by geography. Here we  
14 report large methyl halide fluxes from subtropical salt marshes of south Texas. Sites with the  
15 halophytic plant, *Batis maritima*, emitted methyl halides at rates that are orders of magnitude  
16 greater than sites containing other vascular plants or macroalgae. *B. maritima* emissions were  
17 generally highest at midday; however, diurnal variability was more pronounced for CH<sub>3</sub>Br  
18 than CH<sub>3</sub>Cl, and surprisingly high nighttime CH<sub>3</sub>Cl fluxes were observed in July. Seasonal  
19 and intra-site variability were large, even taking into account biomass differences. Overall,  
20 these subtropical salt marsh sites show much higher emission rates than temperate salt  
21 marshes at similar times of the year, supporting the contention that low-latitude salt marshes  
22 are significant sources of CH<sub>3</sub>Cl and CH<sub>3</sub>Br.

## 23 24 1 Introduction

25 As atmospheric burdens of anthropogenic halocarbons decrease because of the Montreal  
26 Protocol, the relative importance of methyl halides for stratospheric ozone destruction  
27 increases. Methyl chloride (CH<sub>3</sub>Cl) and methyl bromide (CH<sub>3</sub>Br) are now the most abundant  
28 long-lived organochlorine and organobromine compounds, respectively (Montzka and  
29 Reimann, 2011). The atmospheric budgets of CH<sub>3</sub>Cl and CH<sub>3</sub>Br have large uncertainties

30 arising from the fact that they have a multitude of major anthropogenic (e.g., biomass burning,  
31 fumigation use of CH<sub>3</sub>Br, chemical feedstock use of CH<sub>3</sub>Cl) and natural sources (e.g., oceans,  
32 terrestrial ecosystems), some of which are poorly characterized. In our current understanding  
33 of the CH<sub>3</sub>Br budget, sinks outweigh the sources by about 30-35 Gg yr<sup>-1</sup>, or roughly 20-25%  
34 of the total annual flux (Montzka and Reimann, 2011). This large “missing source” for CH<sub>3</sub>Br  
35 is present in both pre-phaseout (1996-98) and current (2008) budgets and appears to be both  
36 natural and terrestrial in origin (Yvon-Lewis et al., 2009). The CH<sub>3</sub>Cl budget may be  
37 balanced with a very large low-latitude terrestrial source (Xiao et al., 2010), and a few *in situ*  
38 studies of subtropical (Yokouchi et al., 2002; Yokouchi et al., 2007) and tropical forests (Blei  
39 et al., 2010a; Saito et al., 2008) tentatively support this.

40 Coastal salt marshes have also been identified as globally significant sources of CH<sub>3</sub>Cl  
41 and CH<sub>3</sub>Br, with emissions associated with halophytic vascular plants. However, measured  
42 emissions show dramatic geographic variability, with large emissions from southern  
43 California (Manley et al., 2006; Rhew et al., 2002; Rhew et al., 2000) and much smaller  
44 emissions from higher latitude sites in Tasmania, Australia (Cox et al., 2004), Scotland (Blei  
45 et al., 2010b; Drewer et al., 2006) and northern California (Rhew and Mazéas, 2010).  
46 Measurements from lower latitude salt marshes have not yet been reported. In this study, we  
47 characterize the magnitude and seasonality of CH<sub>3</sub>Cl and CH<sub>3</sub>Br emissions from subtropical  
48 salt marshes in southern Texas. Obtaining a wider latitudinal range of measurements from  
49 coastal salt marshes is essential to constrain their role in the global budget of methyl halides.

50

## 51 **2 Site Description**

52 Five field outings were conducted between April 2006 and November 2009 at several salt  
53 marsh and coastal habitats on barrier islands in south Texas, USA off the Gulf of Mexico  
54 (**Table 1**). Sites were all located between 27 to 28° N and 97° to 98° W. All sites had sandy  
55 soils with elevations estimated at less than a meter above mean sea level. A total of 62 flux  
56 measurements were made at 24 individual sites from three different tidally influenced  
57 ecosystems.

58 The first field outing (TX1: April 28, 2006) took place on the southwestern shore of San  
59 Jose Island (27° 52'N, 97° 03' W), a sandy barrier island north of the city of Port Aransas.  
60 The goal of this initial outing was to survey emissions from predominant salt marsh plant

61 species of the region: *Borrchia frutescens* (sea ox-eye daisy), *Avicennia germinans* (black  
62 mangrove), *Monanthochloe littoralis* (shoregrass) and *Batis maritima* (maritime saltwort).

63 The second field outing (TX2: May 16-18, 2008) occurred at three different locations:  
64 San Jose Island (see TX1, two *B. maritima* sites, both slightly inundated during sampling),  
65 Mustang Island beach (27° 46'N, 97° 6'W, six beached seaweed sites), and the Mollie Beattie  
66 Habitat Community on the back bay of Mustang Island (27° 38'N, 97° 12'W, one *B. maritima*  
67 site). The goal of this second outing was to determine the daytime range of fluxes from the  
68 three *B. maritima* sites; to measure emissions from pelagic seaweed (*Sargassum* spp.)  
69 deposited on the Gulf-side beach at different stages of decomposition; and to determine the  
70 simultaneous gross consumption and production rates of methyl halides at all of these sites  
71 using a stable isotope tracer technique.

72 The third, fourth and fifth outings (TX3, TX4 and TX5) were all at the Mollie Beattie  
73 habitat (see TX2 above), on the fringe of a small saltwater pond, which was tidally connected  
74 with saline groundwater (**Text S1**). The purpose of these outings was to capture the full  
75 diurnal (24 hour) range of fluxes from a pair of *B. maritima* sites located within 20 meters of  
76 each other. These diurnal studies were conducted at three different times of the year: the early  
77 growing season (TX3: March 7-8, 2009), the peak growing season (TX4: July 19-20, 2009),  
78 and the end of the growing system (TX5: November 6-7, 2009). TX5 occurred after a period  
79 of heavy rain, and many of the *B. maritima* leaves were shed on the ground. Also, between 1  
80 a.m. and 11 a.m. during TX5, both sites were tidally inundated, with the shorter vegetation  
81 site mostly underwater during the 7:30 and 10:30 a.m. samplings.

82 Four vegetation-free control experiments were conducted: two beach sites after the  
83 removal of *Sargassum* (TX2), one salt marsh site with bare soil (TX4) and one salt marsh site  
84 inundated with 30 cm of tidal water (TX5).

85

### 86 **3 Methods**

87 Gas fluxes were measured with static flux chambers consisting of two components: a  
88 collar (61L, 0.264 m<sup>2</sup> footprint) placed in the wet sand > 2 cm depth and an insulated chamber  
89 lid (127 L) with a ¼" stainless steel sample line used to withdraw air samples and two internal  
90 fans to mix the chamber air. All-aluminum chambers were used to limit reactivity with  
91 methyl halides, and dark chambers have been shown to yield similar methyl halide fluxes as

92 light chambers in other salt marshes (Rhew and Mazéas, 2010). To initiate the enclosure  
93 period, the lid was placed into the water-filled channel on the rim of the base. Enclosure  
94 times were 30 minutes or less (30, 22-28, 20, 16 and 16 minutes for TX1-5, respectively), and  
95 three air samples were withdrawn from the chamber at equal time intervals. Samples were  
96 collected into previously evacuated 1 L electropolished stainless steel canisters  
97 (LabCommerce, San Jose, CA, USA) or 3 L fused silica lined canisters (Restek, Bellefonte,  
98 PA, USA). While sampling, a vent line was opened to equilibrate air pressure between inside  
99 and outside the chamber. In addition, ambient air samples were collected several times  
100 throughout each field campaign.

101 Air temperature (inside chamber and ambient air) and soil temperature (5 cm and 10 cm  
102 depth) were monitored with thermocouples (Omega Engineering Inc., Stamford, CT) during  
103 the first three outings and with stainless steel thermocouple data loggers (iButtons, Maxim  
104 Inc., Sunnyvale, CA, USA) for the last two outings. Soil moisture at 0-5 cm depth  
105 (ThetaProbe soil moisture sensor, Delta-T Devices, Cambridge, UK) and air pressure were  
106 monitored for each chamber experiment. For TX2-5, above-ground plant biomass was  
107 harvested, rinsed and drained before fresh weight was determined. Plants were then dried  
108 overnight at 65° C to determine the dry weight. Meteorological data including PAR, air  
109 pressure and air temperature were also measured at the Mission-Aransas National Estuarine  
110 Research Reserve monitoring station at the East Copano Bay, TX, USA  
111 (<http://lighthouse.tamucc.edu/MissionAransas/HomePage>).

112 Air samples were measured for halocarbons (CH<sub>3</sub>Br, CH<sub>3</sub>Cl and CHCl<sub>3</sub>) by gas  
113 chromatography-mass spectrometry (GC/MS, Agilent 6890N/5973). Details regarding the  
114 inlet system, chromatography, gas standards and calibration procedures are described  
115 elsewhere (Rhew, 2011). Concentration trends were calculated using a linear regression of  
116 the chamber air concentration versus time, with goodness of fit assessed both by R<sup>2</sup> and the  
117 standard error on the slope. For the *B. maritima* sites, for example, R<sup>2</sup> values averaged 0.997  
118 for CH<sub>3</sub>Cl and 0.995 for CH<sub>3</sub>Br. Net fluxes were calculated by multiplying this slope with the  
119 moles of air in the chamber, divided by the enclosed surface area; net flux errors were  
120 calculated by propagating the errors of each of these components. For consistency, all fluxes  
121 are reported in units of μmol m<sup>-2</sup> d<sup>-1</sup> unless otherwise indicated, with negative values  
122 representing consumption rates and positive values representing production rates. Also, a  
123 stable isotope tracer technique was applied in the TX2 outing to separate the net flux into the

124 gross production and gross consumption components (**Text S2**). All times are reported as  
125 U.S. Central Standard Time (CST= GMT – 6 hours).

126

## 127 **4 Results**

### 128 **4.1 TX1: April 2006**

129 Of the various vegetation sites sampled during TX1, the largest emissions by far were  
130 from the *B. maritima* site (triangles in **Fig. 1**), which emitted  $580 \pm 30 \mu\text{mol m}^{-2} \text{d}^{-1}$   $\text{CH}_3\text{Cl}$   
131 and  $30 \pm 12 \mu\text{mol m}^{-2} \text{d}^{-1}$   $\text{CH}_3\text{Br}$ . As a comparison, the largest reported emissions observed  
132 from a salt marsh previously were 570 and  $42 \mu\text{mol m}^{-2} \text{d}^{-1}$ , respectively (Rhew et al., 2002;  
133 Manley et al., 2006). The *A. germinans* site showed emissions <0.5% of the *B. maritima* site,  
134 while the two *B. frutescens* sites and the *M. littoralis* site showed small to insignificant net  
135 emissions of  $\text{CH}_3\text{Cl}$  ( $<0.3 \mu\text{mol m}^{-2} \text{d}^{-1}$ ) and  $\text{CH}_3\text{Br}$  ( $<0.012 \mu\text{mol m}^{-2} \text{d}^{-1}$ ).

136

### 137 **4.2 TX2: May 2008**

138 In TX2, the three *B. maritima* sites showed large net emissions of methyl halides,  
139 comparable to TX1. Emission rates increased throughout the day (8:50 a.m. to 2:20 p.m.  
140 CST), although the sampling period was too short to fully assess diurnal trends. One site had  
141 a maximum flux of  $620 \pm 20 \mu\text{mol m}^{-2} \text{d}^{-1}$   $\text{CH}_3\text{Cl}$  and  $39 \pm 2 \mu\text{mol m}^{-2} \text{d}^{-1}$   $\text{CH}_3\text{Br}$ , which at  
142 that point represented the largest  $\text{CH}_3\text{Cl}$  and second largest  $\text{CH}_3\text{Br}$  emission rate per unit area  
143 from a natural source yet observed.

144 The three freshly deposited *Sargassum* sites at the Gulf coast beach showed net  
145 emissions that were 3 orders of magnitude smaller than the *B. maritima* sites (**Fig. 1** and **Table**  
146 **1**). Three other sites of *Sargassum* that were visibly at a more advanced stage of  
147 decomposition and desiccation showed similar net emissions. When two of the *Sargassum*  
148 sites were cleared of seaweed and measured as control experiments on a bare sand surface, net  
149 emissions were an order of magnitude smaller still. Gross consumption rates measured with  
150 stable isotope tracers were negligibly small (**Text S2**).

151

### 152 **4.3 TX3, TX4 and TX5: March, July and November 2009**

153 The next three outings each captured the diurnal variability of  $\text{CH}_3\text{Cl}$  and  $\text{CH}_3\text{Br}$  fluxes  
154 over a 24 hour period from a pair of *B. maritima* dominated sites (**Fig. 2**). The day/night  
155 differences in emissions were much more pronounced for  $\text{CH}_3\text{Br}$  than for  $\text{CH}_3\text{Cl}$ . For  $\text{CH}_3\text{Br}$ ,  
156 the maximum daytime averages were 2.3 times greater than the nighttime averages (n=6

157 sites), whereas the difference for CH<sub>3</sub>Cl was 1.3 times. The molar ratio of CH<sub>3</sub>Cl to CH<sub>3</sub>Br  
158 fluxes also showed a day to night difference (**Fig. 2**), shifting from roughly 40:1 at night to  
159 20:1 during the daytime.

160 Surprisingly, the maximum observed CH<sub>3</sub>Cl emission flux in July ( $630 \pm 10 \mu\text{mol m}^{-2} \text{d}^{-1}$ )  
161 occurred in the middle of the night (1 a.m. CST). In fact, this represented the highest  
162 observed emission rate from all the outings, comparable to the highest flux from TX2. The  
163 lowest emissions were observed during November (TX5) during the morning at one site when  
164 the vegetation was almost entirely submerged by high tide. The other site also was  
165 submerged at the time, but had slightly more vegetation above the surface of the water.

166 To derive an integrated daily flux, the fluxes at each site were modeled by a cosine  
167 function during daylight hours, with steady emissions assumed at night (**Fig. 2, Text S3**). Of  
168 these three outings, the largest average diel emissions were in July (TX4) at  $455 \pm 130 \mu\text{mol}$   
169  $\text{m}^{-2} \text{d}^{-1}$  for CH<sub>3</sub>Cl and  $22 \pm 5 \mu\text{mol m}^{-2} \text{d}^{-1}$  for CH<sub>3</sub>Br. March emissions were roughly half of  
170 those, and November emissions were slightly lower than March (**Table 1, Fig. 3**). Even  
171 though nighttime measurements were not used in the model, the difference between the  
172 modeled to measured nighttime values was only  $-4 \pm 11\%$  (or  $-20 \pm 50 \mu\text{mol m}^{-2} \text{d}^{-1}$ ) for  
173 CH<sub>3</sub>Cl and  $5 \pm 22\%$  (or  $1 \pm 2 \mu\text{mol m}^{-2} \text{d}^{-1}$ ) for CH<sub>3</sub>Br. Thus the model was applied to the  
174 May (TX2) sites as well (**Table 1**).

175

## 176 **5 Discussion**

177 The predominance of *B. maritima* emissions over emissions from other measured plant  
178 and macroalgal species is similar to observations from southern California salt marshes,  
179 where *B. maritima* was one of the two largest emitters of methyl halides (Manley et al., 2006;  
180 Rhew et al., 2002). However, *B. maritima* sites from Texas generally showed much larger  
181 diel averaged emissions of CH<sub>3</sub>Cl and CH<sub>3</sub>Br than those from southern California, especially  
182 outside the peak summer growing season (**Fig. 3**). Even normalized by biomass, emission  
183 rates from Texas sites were roughly ten times larger than Newport Bay California sites  
184 (monthly averages) (Manley et al., 2006).

185 The production of CH<sub>3</sub>Cl and CH<sub>3</sub>Br at *B. maritima* sites are related, as illustrated by a  
186 strong linear correlation ( $R^2=0.78$ ). These fluxes also showed moderate correlations with  
187 chamber air temperature, surface soil temperature and biomass ( $R^2 = 0.40$  to  $0.53$ , **Figs. S1**

188 **and S2).** Within individual outings, however, these environmental factors were poor  
189 predictors. For example, large flux differences were observed between two adjacent sites  
190 with similar biomass (e.g., TX4 and TX5) and could even show a slightly negative  
191 relationship (e.g., TX3). CH<sub>3</sub>Cl and CH<sub>3</sub>Br showed no correlation with net fluxes of  
192 chloroform and carbonyl sulfide (Whelan et al., 2013) that were measured simultaneously  
193 **(Figs. S1 and S2).**

194 The very large nighttime emissions in July when temperatures were also high suggest that  
195 temperature is a more proximate control on emission rates than insolation. This is consistent  
196 with studies in southern California (Rhew et al., 2002) and Scotland (Blei et al., 2010b), but  
197 contrasts with earlier studies in Scotland (Drewer et al., 2006) and Ireland (Dimmer et al.,  
198 2001). At another salt marsh site in southern California, Manley et al. (2002) found that *B.*  
199 *maritima* emissions were less correlated with either temperature or insolation compared to  
200 other plants. We suggest that for studies that use transparent chambers, the effect of  
201 insolation and temperature may be difficult to separate without monitoring leaf temperatures  
202 directly or actively modulating the temperature in the chamber. This does not discount the  
203 importance of insolation, which regulates seasonal changes in temperature and biomass.

204 The average CH<sub>3</sub>Cl:CH<sub>3</sub>Br molar flux ratio of  $22 \pm 9$  is slightly greater than southern and  
205 northern California salt marsh averages (7-17) (Manley et al., 2006; Rhew et al., 2002; Rhew  
206 and Mazéas, 2010) and is much higher than the ratios of 2 to 4 reported from higher latitude  
207 salt marshes (Blei et al., 2010b; Cox et al., 2004; Dimmer et al., 2001). This is consistent  
208 with the observation of Blei et al. (2010b) that the salt marshes from more temperate climates  
209 generally have lower emission ratios. However, this is not a consequence of higher  
210 temperatures leading to higher ratios. At the Texas *B. maritima* sites, molar ratios did not  
211 dramatically shift with the seasons, and the molar ratios of emissions were higher at night  
212 (~40) and lower during the day (~20), opposite of the temperature trends (**Fig. 2**).

213 This diurnal trend in ratios is clearly related to the much larger diurnal variation in CH<sub>3</sub>Br  
214 flux compared to CH<sub>3</sub>Cl flux, as illustrated by the pronounced midday CH<sub>3</sub>Br peak in this  
215 study (**Fig. 2**). Interestingly, this same trend in molar ratios was also observed in a San Diego  
216 salt marsh (Rhew et al., 2002), where it mirrored a diurnal shift in the carbon isotopic ratio  
217 ( $\delta^{13}\text{C}$ ) of CH<sub>3</sub>Cl and CH<sub>3</sub>Br. In that study, carbon isotopic signatures were heavier at night (-  
218 50‰ CH<sub>3</sub>Cl and -10 ‰ CH<sub>3</sub>Br) compared to daytime (-70‰ CH<sub>3</sub>Cl and -60‰ CH<sub>3</sub>Br), with  
219 the isotopic shift much more pronounced for CH<sub>3</sub>Br than CH<sub>3</sub>Cl (Bill et al., 2002).

220 Two hypotheses were proposed to explain these diurnal trends of ratios and isotopic  
221 signatures (Bill et al., 2002; Rhew et al., 2002): 1) biogenic production dominates during the  
222 day, while soil consumption becomes more significant at night; and 2) two different  
223 production mechanisms with different isotopic signatures and ratios of production occur  
224 simultaneously.

225 The first hypothesis could explain the lower overall net emission rates and heavier  
226 isotopic signatures at night (since consumption favors lighter isotopes (Miller et al., 2001)),  
227 but this study and others (Rhew and Mazéas, 2010) suggest that gross consumption is trivial  
228 in salt marshes compared to *B. maritima* production rates, even at night (**Text S2**). Also,  
229 gross consumption generally favors CH<sub>3</sub>Cl uptake over CH<sub>3</sub>Br by a molar factor of 30 to 40  
230 (Rhew, 2011; Rhew and Mazéas, 2010), such that if nighttime consumption is important, the  
231 net emission ratio of CH<sub>3</sub>Cl to CH<sub>3</sub>Br should decrease at night, not increase.

232 The second hypothesis is supported by having two known production mechanisms of  
233 methyl halides from *B. maritima*: the enzymatically mediated methylation of halides (Ni and  
234 Hager, 1999; Wuosmaa and Hager, 1990) and an abiotic reaction between plant pectin and  
235 halides (Hamilton et al., 2003; Wishkerman et al., 2008). Because the abiotic mechanism  
236 yields a very light isotopic signature ( $\delta^{13}\text{C}$  of -78‰ for *B. maritima*) (Keppler et al., 2004), a  
237 large abiotic increase during the daytime relative to enzymatic production could explain the  
238 isotopic shift. However, the CH<sub>3</sub>Cl:CH<sub>3</sub>Br molar ratio of production for the abiotic  
239 mechanism is also larger (45 to 58) (Wishkerman et al., 2008) than the predicted enzymatic  
240 production (20 to 1) (Rhew et al., 2002), which would yield a larger diurnal shift for CH<sub>3</sub>Cl  
241 than CH<sub>3</sub>Br, which is not observed.

242 An alternative hypothesis involves both a diurnal shift in the isotope signature of the  
243 carbon substrate used to produce methyl halides combined with a shift in methylation ratios of  
244 the halides. If production is predominantly biological, a diurnal shift in the  $\delta^{13}\text{C}$  signature of  
245 the methyl donor (*S*-adenosyl-L-methionine) (Ni and Hager, 1999) and/or higher isotopic  
246 fractionation rates during the daytime could yield the observed isotopic signal. It is also  
247 possible that the abiotic production mechanism produces lighter  $\delta^{13}\text{C}$  methyl halides at higher  
248 temperatures, but the carbon source for abiotic production comes from structural components  
249 of a plant that are not necessarily expected to have diurnal variation in  $\delta^{13}\text{C}$  (Keppler et al.,  
250 2004).

251 The other half of this hypothesis involves the Cl/Br ratios in the plant changing during the  
252 day. Bromide is preferentially halogenated by both biotic and abiotic mechanisms relative to  
253 their availability (Ni and Hager, 1998; Wishkerman et al., 2008) and may be replenished  
254 during the day and depleted at night, perhaps in conjunction with transpiration rates.  
255 However, the amount of chloride and bromide that is volatilized daily via methyl halide  
256 emission is not large enough to substantially change the overall Cl<sup>-</sup> and Br<sup>-</sup> content in plant  
257 tissue. In this study, the biomass normalized diel emission rates of CH<sub>3</sub>Cl and CH<sub>3</sub>Br were  
258  $1.25 \pm 0.40 \mu\text{mol gdw}^{-1} \text{d}^{-1}$  and  $0.062 \pm 0.014 \mu\text{mol gdw}^{-1} \text{d}^{-1}$ , respectively (n=9 sites). If we  
259 assume that the *B. maritima* tissue halide contents are similar to those measured in southern  
260 California *B. maritima* plants (210 mg g<sup>-1</sup> for Cl<sup>-</sup> and 2900 μg g<sup>-1</sup> for Br<sup>-</sup> (Manley et al.,  
261 2006)), then we estimate that roughly 0.02% of Cl and 2% Br in the leaf tissue is removed  
262 daily via methyl halide emissions. Thus, to impact halide availability, there would need to be  
263 a small segregated subset of ‘active’ halides at the enzyme site. If this ‘active’ halide pool  
264 was 0.5% of the overall tissue content, then the methyl halide emissions could reduce that  
265 pool by 4% for Cl and 34% for Br daily. This would lead to an increased CH<sub>3</sub>Cl to CH<sub>3</sub>Br  
266 emission ratio, until the halide levels were replenished. A subset of ‘active’ halides in the  
267 cytoplasm of plant cells is implied by Ni and Hager (1998, 1999), who proposed that the  
268 function of halide methyltransferase is to dispose of excess chloride to regulate internal  
269 concentrations.

270 The Texas salt marsh fluxes measured over several months strongly suggest a seasonality  
271 of fluxes. Assuming that the seasonality can be characterized with a sinusoidal fit to the diel  
272 averaged data (**Fig. 3**) and that these measurements are temporally and spatially  
273 representative, we derive an estimated annual flux of 92 mmol m<sup>-2</sup> yr<sup>-1</sup> for CH<sub>3</sub>Cl and 4.7  
274 mmol m<sup>-2</sup> yr<sup>-1</sup> for CH<sub>3</sub>Br. These annual values are 2 to 3 times larger than those estimated for  
275 the *B. maritima* sites in Upper Newport Bay (28 mmol m<sup>-2</sup> yr<sup>-1</sup> for CH<sub>3</sub>Cl and 2.4 mmol m<sup>-2</sup>  
276 yr<sup>-1</sup> for CH<sub>3</sub>Br) (Manley et al., 2006). Sampling over the full range of environmental  
277 conditions would help refine these estimates of the annual flux.

278 The surface area coverage of *B. maritima* in Texas salt marshes was not quantified for this  
279 study. In Newport Bay, California, *B. maritima* covered 10% of the entire salt marsh area  
280 (including barren areas) and 18% of the vegetated area (Manley et al., 2006). For the purpose  
281 of comparison, we will assume that these Texas salt marshes have the same *B. maritima*  
282 coverage and that the remaining 82-90% of salt marsh has negligible emission rates. Spatially

283 averaged emissions for the entire salt marsh are then estimated at 9-17 and 0.47-0.84 mmol m<sup>-2</sup>  
284 yr<sup>-1</sup> for CH<sub>3</sub>Cl and CH<sub>3</sub>Br, respectively. These fluxes are slightly greater than those reported  
285 from Newport Bay salt marshes (3-8 and 0.2-0.7 mmol m<sup>-2</sup> yr<sup>-1</sup> for CH<sub>3</sub>Cl and CH<sub>3</sub>Br,  
286 respectively, with the range representing total area and only vegetated areas), which have  
287 other large emissions associated with *Frankenia grandifolia* (Manley et al., 2006). However,  
288 these rates are one to two orders of magnitude greater than annually averaged salt marsh  
289 fluxes in Scotland (0.11 mmol m<sup>-2</sup> yr<sup>-1</sup> for CH<sub>3</sub>Cl and 0.03 mmol m<sup>-2</sup> yr<sup>-1</sup> for CH<sub>3</sub>Br) (Blei et  
290 al., 2010b; Drewer et al., 2006). The Scotland salt marsh fluxes are similar in magnitude to  
291 other high latitude salt marshes, including Tasmania, Australia (Cox et al., 2004) and northern  
292 California (Rhew and Mazéas, 2010).

293 Collectively, these studies show that methyl halide emissions from coastal salt marshes  
294 have a strong climatic dependence, with small emissions at higher latitudes and large  
295 emissions at lower latitudes. This climatic dependence may be related to both temperature  
296 (higher temperatures yield faster enzymatic and abiotic production rates of methyl halides)  
297 and insolation (greater photosynthesis rates lead to greater biomass, with associated increases  
298 in relevant secondary metabolites and enzymes). *B. maritima* alone may be responsible for  
299 globally significant amounts of methyl halides, as it is an evergreen succulent shrub found  
300 widely in tropical and subtropical salt marshes, brackish marshes and mangrove swamps  
301 ranging from northern Brazil (3°S) to South Carolina (33°N) (Lonard et al., 2011). A major  
302 uncertainty involves the spatial distribution and global coverage of coastal wetlands, with 2.2-  
303 40 Mha of tidal marsh and 13.8-15.2 Mha of mangroves (Pendleton et al., 2012). As an  
304 illustrative exercise, if *B. maritima* or similarly emitting plants cover 10% of the surface area  
305 of tidal marshes and mangroves, and if averaged emissions are as calculated here, then this  
306 subset of salt marsh vegetation would contribute 30-90 Gg CH<sub>3</sub>Cl and 3-9 Gg CH<sub>3</sub>Br per  
307 year. Deriving a more accurate global source strength will require a much broader geographic  
308 distribution of measurements, along with better estimates of ecosystem surface areas and plant  
309 distributions. Clarifying the importance of coastal salt marsh vegetation in the global budgets  
310 of CH<sub>3</sub>Cl and CH<sub>3</sub>Br will require further measurements at low latitude salt marsh sites.

311

## 312 **6 Conclusion**

313 Large emissions of CH<sub>3</sub>Cl and CH<sub>3</sub>Br were observed from subtropical salt marshes  
314 located on the Gulf coast of Texas. These large emissions were associated with *B. maritima*,  
315 a widespread succulent salt marsh plant that was also observed to be a large emitter in

316 southern California salt marshes. However, *B. maritima* emission rates in this study were 2 to  
317 3 times larger than those reported from California, and spatially averaged emission rates from  
318 Texas salt marshes were much larger overall than those reported from higher latitude salt  
319 marsh sites. Diurnal trends in CH<sub>3</sub>Cl and CH<sub>3</sub>Br emission rates, along with their ratio of  
320 emissions, were similar to those observed in southern California salt marshes. To derive a  
321 better estimate of the global salt marsh contribution to the atmospheric budgets of the methyl  
322 halides, more information is needed about the spatial extent, vegetation cover and methyl  
323 halide emission rates from low latitude salt marsh sites.

324

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331

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415

416 **Table 1.** Field conditions and predominant vegetation at Texas coastal field sites

417	Outing/date: (dd/mm/yy)	Location	Net flux ( $\mu\text{mol m}^{-2} \text{d}^{-1}$ )		Modeled diel flux*		<i>Batis</i> biomass	
418	Enclosed species	n	CH <sub>3</sub> Cl	CH <sub>3</sub> Br	CH <sub>3</sub> Cl	CH <sub>3</sub> Br	kg/m <sup>2</sup>	kg/m <sup>2</sup>
419	TX1: 28/04/06	San Jose Island					fresh	dry
420	<i>Batis maritima</i>	1	584 ± 27	30 ± 12	-	-	n.d.	n.d.
421	<i>Avicennia germinans</i>	1	1.8 ± 0.4	0.023 ± 0.010	-	-	-	-
422	Various†	3	≤ 0.2	≤ 0.02	-	-	-	-
423	TX2: 16/05/08-18/05/08	San Jose Island, Mustang Island Beach and Mollie Beattie						
424	<i>B. maritima</i> (site A) ‡	3	409 ± 115	23 ± 5	296	16	1.44	0.25
425	<i>B. maritima</i> (site B) ‡	3	371 ± 80	22 ± 4	230	15	1.38	0.23
426	<i>Sargassum</i> (fresh)	3	0.56 ± 0.55	0.029 ± 0.027	-	-	-	-
427	<i>Sargassum</i> (decaying)	3	0.84 ± 0.72	0.030 ± 0.014	-	-	-	-
428	Sand (beach)	2	0.04 ± 0.02	0.004 ± 0.003	-	-	-	-
429	<i>B. maritima</i> (site C)	3	494 ± 115	29 ± 8	287	16	1.58	0.30
430	TX3: 07/03/09-08/03/09	Mollie Beattie						
431	<i>B. maritima</i> (site A)	5	220 ± 30	13 ± 5	222	13	1.52	0.28§
432	<i>B. maritima</i> (site B)	5	270 ± 40	13 ± 4	266	13	1.34	0.24§
433	TX4: 20/07/09-21/07/09	Mollie Beattie						
434	<i>B. maritima</i> (site A)	7	571 ± 43	27 ± 7	547	25	1.66	0.31
435	<i>B. maritima</i> (site B)	7	374 ± 28	20 ± 6	362	18	1.68	0.29
436	Sand (marsh)	1	-0.073 ± 0.061	0.004 ± 0.002	-	-	-	-
437	TX5: 06/11/09-07/11/09	Mollie Beattie						
438	<i>B. maritima</i> (site A) ‡	7	165 ± 30	7.0 ± 2.9	156	6	0.66	0.13
439	<i>B. maritima</i> (site B) ‡	7	207 ± 117	10.7 ± 9.4	265	11	0.75	0.13
440	Saltwater (marsh) ‡	1	2.40 ± 0.07	0.037 ± 0.001	-	-	-	-

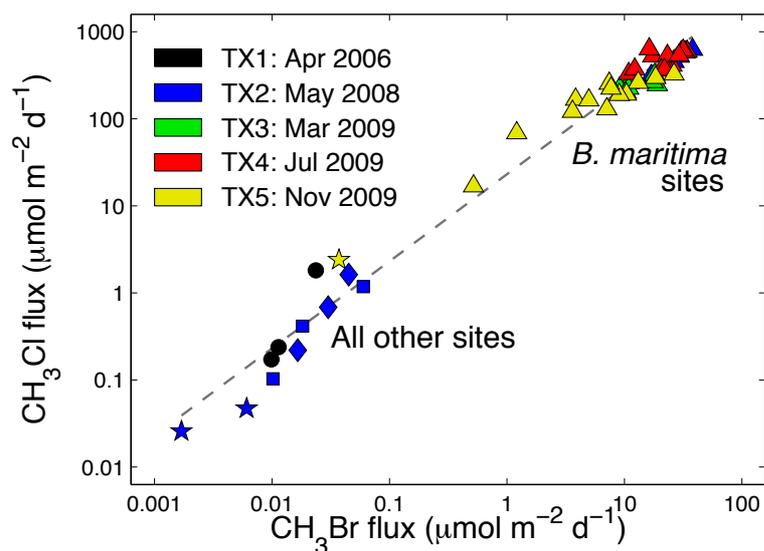
441  
442 \* modeled diel flux ( $\mu\text{mol m}^{-2} \text{d}^{-1}$ ) based on daytime measurements

443 † *Borrhichia frutescens* and *Monanthochloe littoralis*

444 ‡ soil surface covered with water for some or all measurements

445 § vegetation H<sub>2</sub>O estimated as 81.9% based on average of other outings

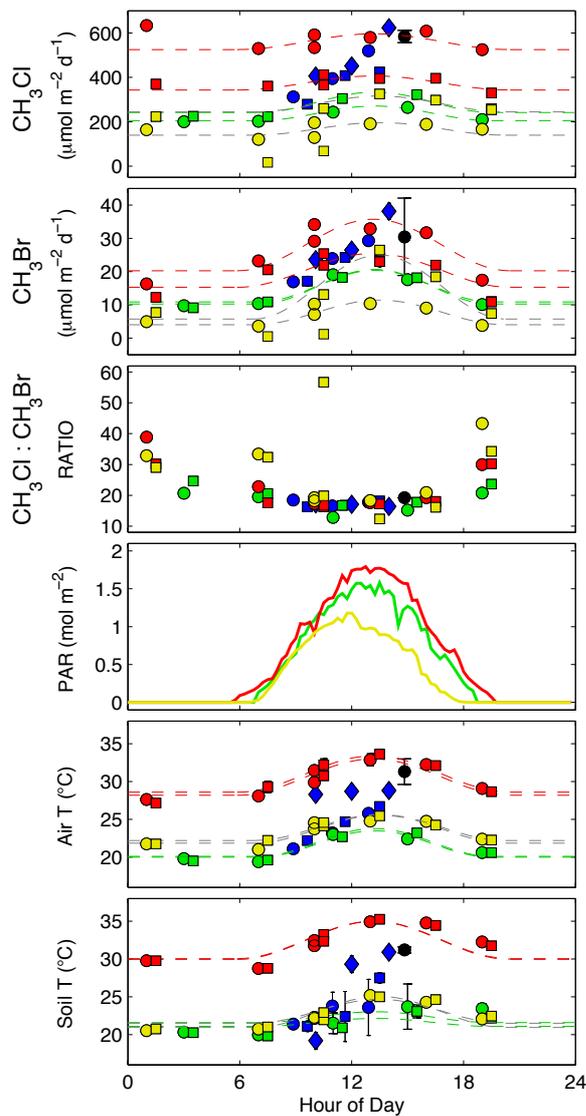
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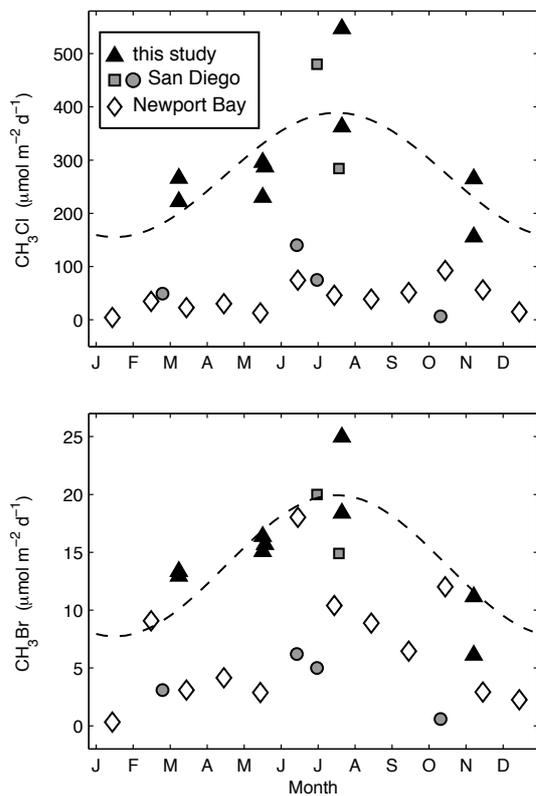
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448

449 **Figure 1.** Net fluxes of CH<sub>3</sub>Cl and CH<sub>3</sub>Br for all field sites in south Texas. Triangles = *B.*  
 450 *maritima*; squares = fresh *Sargassum*; diamonds = decaying *Sargassum*; circles = other  
 451 vegetation (*M. littoralis*, *A. germinans*, *B. frutescens*), stars= vegetation-free control. Note the  
 452 log-log scale. The gray dashed line shows the overall average 23:1 molar ratio. Two  
 453 chambers that had small negative fluxes are not included.



454  
 455 **Figure 2.** Net fluxes of  $\text{CH}_3\text{Cl}$ , net fluxes of  $\text{CH}_3\text{Br}$ , the  $\text{CH}_3\text{Cl}$  to  $\text{CH}_3\text{Br}$  flux ratio,  
 456 photosynthetically active radiation (PAR), chamber air temperature and surface soil  
 457 temperature at *B. maritima* sites versus time of day (Central Standard Time). Colors represent  
 458 different outings, as in Fig. 1; different symbols (circles, squares, diamonds) represent  
 459 different sites at the same outing. Error bars that are smaller than the symbols are not shown  
 460 The dashed lines represents the model fit to TX3-5 (March, July, and November) results.  
 461 PAR is a 15 minute interval measurement averaged over the two days of each field outing.



462  
 463 **Figure 3.** Comparison of *B. maritima* methyl halide emissions from three coastal salt  
 464 marshes. This study (black triangles) shows diel averages. The San Diego, California sites  
 465 show diel averages (gray squares) at two *B. maritima* sites and daytime fluxes (gray circles) at  
 466 mixed *B. maritima* /*Salicornia bigelovii* sites [Rhew et al., 2000; Rhew et al., 2002]). The  
 467 Upper Newport Bay, California sites (white diamonds) show daytime fluxes of monospecific  
 468 *B. maritima* sites [Manley et al., 2006]). The dashed line is a sinusoidal curve fit to the Texas  
 469 data.