

Interactive comment on “Organomineral nanocomposite carbon burial during Oceanic Anoxic Event 2” by S. C. Löhrr and M. J. Kennedy

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Received and published: 13 August 2014

We thank the reviewers for their positive feedback, constructive comments and helpful suggestions. Where appropriate, referees comments are reproduced in *italics* while our responses to each point are given in plain text.

Anonymous Referee 1

This paper was a great pleasure to review and an eye opener in several respects.

The methods and very broad range of sophisticated experiments are at the cutting edge and appear to be thoroughly performed and are well presented. The correlation between TOC and MSA is very strong and convincing, amazingly good, in fact, for such a complex system. Thus, I was initially worried that the MSA/TOC correlations

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might partially be an artifact of the methodology, i.e. that the EGME method for MSA determination essentially measures organic matter surfaces, not mineral surfaces (thus creating an autocorrelation). However, this concern is dispelled by the correlation in Fig 3 between MSA before and after organic matter removal. **For this correlation, it would be useful to add evidence about the efficiency of OM removal using hydrogen peroxide (i.e., was OM entirely eliminated?).**

We note in the caption to Figure 3 that efficiency of OM removal ranged between 61 and 92 percent. In addition, we are making available all the data presented in the manuscript as supplementary information, including the percentage of OM removed after treatment with hydrogen peroxide.

Anonymous Referee 2

The reviewer requested that a more clear tie between data and interpretation be made, specifically they wanted these issues addressed:

- In the manuscript they point out that 'OC is preserved mainly as sub- μm organic matter in close association with clays, rather than as discrete, μm -scale organic particles' and 'the great majority of OC is quantitatively associated with and stabilized by clay mineral surfaces' (p. 6822). However, I cannot find any discussion about quantitative estimation of their relative abundances. Thus, it is difficult to agree with the authors claim. Isn't it possible to quantitatively estimate their ratios? In Fig. 6D, some areas (few to 10 μm in size) with high carbon content are seen. They look like discrete OM. I feel that the manuscript needs careful discussion on relative abundances of μm -size discrete OM and sub- μm size OM associated with clay minerals.

Our assessment of the relative abundance of clay-associated vs discrete OM is based on the strong bulk sample relationship between MSA and TOC, which shows an R2 value of >0.9 in some sections. As we note in the manuscript, the strength of the relationship implies that the great majority of OM is associated with clay minerals, and that the OM is sufficiently small to meaningfully interact and associate with clays. We

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assume that this type of OC preservation maintains a proportionality between TOC and MSA since a mechanistic control of OC preservation by mineral surfaces should result in a reproducible relationship between these two parameters (Mayer, 1994; Keil and Mayer, 2014; Kennedy et al., 2014). Not only is discrete OM detritus too large to meaningfully interact with the clay mineral matrix, if particulate OC from pelagic sources (and unrelated to mineral surface associated OC) where an important contributor to some of the samples these would plot above the regression evident in Figure 2 (high TOC with limited surface area) indicating the presence of OC that was independent and unrelated to the MSA scaling relationship controlled by surface area. Yet samples with high TOC relative to MSA are not apparent in our dataset.

Alternatively the relationship between MSA and TOC could be simply a coincidence. However, we found this relationship repeated in each core we sampled, and it is reported from other unrelated deposits (Mayer, 1994; Kennedy et al., 2002; Blair and Aller, 2012; Kennedy et al., 2014) of varying age and depositional conditions making a coincidental scaling unlikely. Still, the correlation does not require causation, and to provide further evidence that nanocomposites were forming we directly imaged the relationship using high-resolution synchrotron and SEM imaging. This shows that rather than being concentrated in discrete particles, OM is dispersed throughout the sample, where it is associated with clays forming organo-clay aggregates. SEM images and STXM maps are consistent with our interpretation of the MSA-TOC relationship that the majority of OM is aggregated and associated with clays.

We recognize that because of the μm size of the image, establishing that any given sample is representative of the primary form of OC is problematic, and so we offer this type of data only as a proof of concept while relying on the bulk rock relations (MSA-TOC) to identify the extent of this relationship. Additionally, however, the clay-organic association is also evident in our XRD data, which shows the presence of molecular-scale OM intercalated within the expandable smectite interlayer (Figure 7), and this relationship is evident from bulk sample material.

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We have revised the manuscript to further clarify these points and the rationale behind them.

- I am not fully convinced by the FTIR spectroscopic maps of aliphatic carbon and clay (Figs 6B and 6C) that were used to exemplify the close association between organic matter and clay minerals. In my view, the distribution of aliphatic carbon significantly differs from that of clay.

We have revised the manuscript to clarify that the FTIR spectroscopic maps show that both OM and clays are broadly distributed throughout the sample, with overlapping distributions, suggesting a close association between clay minerals and OM. The resolution of the technique is not sufficient to do better than this. What it can do, however, is rule out the presence of $> \mu\text{m}$ scale organic particles that are independent of clay minerals. The intimate ($< 1 \mu\text{m}$) relationship between clay and organic material is further indicated in backscatter electron images, which show that OM is present throughout the sediment and is primarily present within organo-clay aggregates.

- The analytical position of electron microprobe analysis with EDAX (Fig. 5D) should be given in Fig. 5C. In general, electron beam expands about $3 \mu\text{m}$ in the sample surface. Because of the limit of spatial resolution of electron beam, I think that the EDAX data is not good enough to demonstrate the co-existence of organic matter and clay minerals on the sub-micrometer scale.

As reviewer 2 points out electron beams have sample interaction volumes of up to about $3 \mu\text{m}$ diameter, depending on factors such as accelerating voltage and sample composition, so that EDS analyses with sub- μm resolution are not feasible. Fortunately, sub- μm elemental analyses are not required for the point we are making here. We simply seek to demonstrate that the relatively homogeneous $\approx 5 \mu\text{m}$ size aggregates shown in Figure 5B and C, which we interpreted as organo-mineral aggregates based on petrographic and back-scatter electron contrast criteria (i.e. presence of bright, sub- μm , elongate particles morphologically consistent with clays, coated and

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aggregated by an amorphous, darker material interpreted to be OM), are indeed composed mainly of Si, Al, O and C. This is straightforward with the EDS since the electron beam interaction volume is confined to the relatively homogeneous aggregate. Thus it is the SEM BSE image that demonstrates the co-existence of organic matter and clay minerals at the sub-micrometer scale, the EDS analyses simply serve as an independent confirmation that the average composition of the sub- μm materials visible in the SEM images is consistent with clays and organic matter.

We have revised Figure 5C to show more clearly the area analysed by EDS.

- The regression lines shown in Fig. 2 do not run near the zero point, but they have intercepts around 200 m²/g. What does the intercept mean? Attention must be paid to compare their results with their previous data (Kennedy et al., 2002, 2014; Kennedy and Wagner, 2011). For example, the 'laminated anoxic' sediments of late Cretaceous interval from the deep Ivorian Basin, Site 959 (Fig. 2B in Kennedy and Wagner, 2011), are plotted in a different field with a different slope (0.7 mg-OC/m²) compared with the data presented in this paper (Fig. 2). What does this mean? Does this indicate that mechanism governing the OM preservation is not as simple as the authors consider, and that we should think other mechanisms to explain the preservation of OM?

Reviewer 2 notes that the x-axis intercept of the MSA to TOC regression shown in Figure 2 is significantly offset from zero (≈ 200 m²/g). Similar offsets have been observed in other studies of ancient OC rich sediments, with offsets of ≈ 260 m²/g and ≈ 290 m²/g for the anoxic/suboxic and the oxic facies, respectively, of OAE3 at ODP Site 959 (Kennedy and Wagner, 2011) and an offset of ≈ 60 m²/g for the Cretaceous Pierre Shale (Kennedy et al., 2002). This x-intercept offset implies MSA that did not acquire OM or subsequently lost OM, and has previously been interpreted to reflect loss of a more readily oxidised fraction of OM that was associated with external clay mineral surfaces and edges, which can exceed 200 m²/g in smectitic sediments, whereas OM preserved within the smectite interlayer is considered to be more refractory (Kennedy and Wagner, 2011; Kennedy et al., 2014). However, this interpretation is inconsistent

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with our SEM observation of organoclay aggregates, which suggests that a substantial component of OM is retained in association with external clay surfaces, encapsulated within organoclay aggregates. Alternatively, the observed offset may be the product of lower OM concentrations in the outermost zone of organoclay aggregates, as OM encapsulated by and sorbed to clays in aggregate interiors is better protected relative to OM initially present closer to the aggregate exterior (Keil and Mayer, 2014), as also observed in soil organomineral aggregates (Kinyangi et al., 2006). If correct, this would result in a systematic x-axis intercept offset that is dependent primarily on aggregate size (surface area/volume ratio).

The slope of the MSA to TOC regression has been shown to vary as a function of the depositional environments, sediment diagenesis, and oxic versus anoxic sediments (Blair and Aller, 2012; Kennedy et al., 2014). While TOC generally correlates with MSA in modern continental margin sediments, the highest OM loadings per unit MSA (slope) are found in sediments from high productivity and/or oxygen depleted settings whereas the lowest loadings are typical of sediments from higher energy settings or deep-sea deposits that are subject to long oxygen exposure times (Blair and Aller, 2012). Similar differences are apparent in ancient sediments, with higher OM loadings recorded in samples from anoxic facies (0.7 mg OC/m²) compared to oxic and suboxic facies (0.4 mg OC/m²) at ODP Site 959 (Kennedy and Wagner, 2011). While the ODP 959 and Demerara Rise datasets largely overlap, the slope of the MSA to TOC regression at Demerara Rise sites ranges between 0.44 mg OC/m² (Site 1258) and 0.52 mg OC/m² (Site 1260), somewhat lower than the laminated, anoxic facies at Site 959 but greater than the oxic to suboxic facies at Site 959. Although this might indicate a greater oxygen effect at Demerara Rise relative to Site 959, comparison to modern sediments show that the OM loading ratio is equally reduced in relatively higher energy or lower productivity settings.

We have revised the manuscript to expand on these points.

- *How are the slopes and intercepts of the MSA-OM crossplots (Fig. 2) sensitive to*

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mineral compositions? Don't those slopes change in response to relative abundance of smectite (e.g., smectite / total clay minerals) during OAE2, when relative abundance of smectite increased?

Apart from higher temperature clay diagenesis during burial (Kennedy et al., 2014), which can be ruled out in the thermally immature sediments studied here, mineralogical changes are unlikely to influence the slope. This is because the > 100 m²/g surface areas measured here are diagnostic of smectite clay, which dominated sediment surface areas even where representing <5% in any case, and as we note in the manuscript, our data does not show an increased OC richness or smectite abundance in the non-carbonate fraction during OAE2 at Demerara Rise, although we suggest that an increase in volcanic-derived smectite at other sites may account for increased global OC burial during this period and the C isotope anomaly associated with OAE2.

References: Blair, N. E., and Aller, R. C. (2012). The Fate of Terrestrial Organic Carbon in the Marine Environment. *Annual Review of Marine Science*, 4(1), 401–423.

Keil, R. G. and Mayer, L. M.: Mineral matrices and organic matter, in: *Treatise on Geochemistry*, Elsevier, 337–359, 2014 Kennedy, M. J., Pevear, D. R., and Hill, R. J. (2002). Mineral surface control of organic carbon in black shale. *Science*, 295(5555), 657–660.

Kennedy, M. J., and Wagner, T. (2011). Clay mineral continental amplifier for marine carbon sequestration in a greenhouse ocean. *PNAS*, 108(24), 9776–9781.

Kennedy, M. J., Löhr, S. C., Fraser, S. A., and Baruch, E. T. (2014). Direct evidence for organic carbon preservation as clay-organic nanocomposites in a Devonian black shale; from deposition to diagenesis. *Earth and Planetary Science Letters*, 388, 59–70.

Kinyangi, J., Solomon, D., Liang, B., Lerotic, M., Wirick, S., and Lehmann, J. (2006). Nanoscale Biogeocomplexity of the Organomineral Assemblage in Soil. *Soil Science Society of America Journal*, 70(5), 1708–1718.

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Mayer, L. M. (1994). Surface area control of organic carbon accumulation in continental shelf sediments. *Geochimica Et Cosmochimica Acta*, 58(4), 1271–1284.

Ransom, B., Kim, D., Kastner, M., and Wainwright, S. (1998). Organic matter preservation on continental slopes: importance of mineralogy and surface area. *Geochimica Et Cosmochimica Acta*, 62(8), 1329–1345.

Interactive comment on Biogeosciences Discuss., 11, 6815, 2014.

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11, C4346–C4353, 2014

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