SE: Example M and I calculations for organic matter cycling (i = 2) at depth

$$\boldsymbol{M}_{2} = \boldsymbol{\sigma}_{R_{2}} \left| \sum_{j=1}^{7} \frac{\partial \Omega_{C}}{\partial X_{j}} \frac{\partial X_{j,2}}{\partial R_{i}} \right| = \boldsymbol{\sigma}_{R_{2}} \left| S_{R_{2}} \right|$$

 R_2 is phosphate (P), and the values of the various terms are from Table 1 and A1. If we remineralize σ_{R_2} µmols/kg of phosphate worth of organic matter, this...

$$\begin{split} M_i &= 0.60 \ \mu\text{mol P/kg} \left[\left(0 \, \frac{1}{\text{db}} \times -0.00028 \, \frac{\text{db}}{\mu\text{mol P/kg}} \right) + \right. \\ &\quad \text{... has no effect on pressure} \\ &\quad + \left(0 \, \frac{1}{\text{°C}} \times 0.014 \, \frac{\text{°C}}{\mu\text{mol P/kg}} \right) + \\ &\quad + \left(0 \times -0.011 \, \frac{1}{\mu\text{mol P/kg}} \right) + \\ &\quad + \left(1 \, \frac{1}{\mu\text{mol P/kg}} \times -0.0085 \, \frac{\mu\text{mol P/kg}}{\mu\text{mol P/kg}} \right) + \\ &\quad + \left(0 \, \frac{1}{\mu\text{mol Si/kg}} \times -0.00012 \, \frac{\mu\text{mol Si/kg}}{\mu\text{mol P/kg}} \right) + \\ &\quad + \left(0 \, \frac{1}{\mu\text{mol Si/kg}} \times -0.00012 \, \frac{\mu\text{mol Si/kg}}{\mu\text{mol P/kg}} \right) + \\ &\quad + \left(-20.16 \, \frac{1}{\mu\text{mol } A_{\text{T}}/\text{kg}} \times 0.0082 \, \frac{\mu\text{mol } A_{\text{T}}/\text{kg}}{\mu\text{mol P/kg}} \right) + \\ &\quad + \left(117 \, \frac{1}{\mu\text{mol } C_{\text{T}}/\text{kg}} \times -0.0079 \, \frac{\mu\text{mol } C_{\text{T}}/\text{kg}}{\mu\text{mol P/kg}} \right) + \\ &\quad + \left(0 \, \frac{1}{\mu\text{mol } A_{\text{T}}/\text{kg}} \times -0.0079 \, \frac{\mu\text{mol } C_{\text{T}}/\text{kg}}{\mu\text{mol P/kg}} \right) + \\ &\quad + \left(0 \, \frac{1}{\mu\text{mol } C_{\text{T}}/\text{kg}} \times -0.0079 \, \frac{\mu\text{mol } C_{\text{T}}/\text{kg}}{\mu\text{mol P/kg}} \right) + \\ &\quad + \left(0 \, \frac{1}{\mu\text{mol } C_{\text{T}}/\text{kg}} \times -0.0079 \, \frac{\mu\text{mol } C_{\text{T}}/\text{kg}}{\mu\text{mol P/kg}} \right) + \\ &\quad + \left(0 \, \frac{1}{\mu\text{mol } C_{\text{T}}/\text{kg}} \times -0.0079 \, \frac{\mu\text{mol } C_{\text{T}}/\text{kg}}{\mu\text{mol P/kg}} \right) + \\ &\quad + \left(0 \, \frac{1}{\mu\text{mol } C_{\text{T}}/\text{kg}} \times -0.0079 \, \frac{\mu\text{mol } C_{\text{T}}/\text{kg}}{\mu\text{mol P/kg}} \right) + \\ &\quad + \left(0 \, \frac{1}{\mu\text{mol } C_{\text{T}}/\text{kg}} \times -0.0079 \, \frac{\mu\text{mol } C_{\text{T}}/\text{kg}}{\mu\text{mol P/kg}} \right) + \\ &\quad + \left(0 \, \frac{1}{\mu\text{mol } C_{\text{T}}/\text{kg}} \times -0.0079 \, \frac{\mu\text{mol } C_{\text{T}}/\text{kg}}{\mu\text{mol P/kg}} \right) + \\ &\quad + \left(0 \, \frac{1}{\mu\text{mol } C_{\text{T}}/\text{kg}} \times -0.0079 \, \frac{\mu\text{mol } C_{\text{T}}/\text{kg}}{\mu\text{mol P/kg}} \right) + \\ &\quad + \left(0 \, \frac{1}{\mu\text{mol } C_{\text{T}}/\text{kg}} \times -0.0079 \, \frac{\mu\text{mol } C_{\text{T}}/\text{kg}}{\mu\text{mol P/kg}} \right) + \\ &\quad + \left(0 \, \frac{1}{\mu\text{mol } C_{\text{T}}/\text{kg}} \times -0.0079 \, \frac{\mu\text{mol } C_{\text{T}}/\text{kg}}{\mu\text{mol P/kg}} \right) + \\ &\quad + \left(0 \, \frac{1}{\mu\text{mol } C_{\text{T}}/\text{kg}} \times -0.0079 \, \frac{\mu\text{mol } C_{\text{T}}/\text{kg}}{\mu\text{mol } C_{\text{T}}/\text{kg}} \right) + \\ &\quad + \left(0 \, \frac{1}{\mu\text{mol } C_{\text{T}}/\text{kg}} \times -0.0079 \, \frac{\mu\text{mol } C_{\text{T}}/\text{kg}}{\mu\text{mol } C_{\text{T}}/\text{kg}} \right) + \\ &\quad + \left(0 \, \frac{1}{\mu\text{mol } C_{\text{T}}/\text{kg}} \times -0.0079 \, \frac{\mu\text{mol } C_{\text{T}}/\text{kg}}{\mu\text{mol } C_{\text{T$$

This M value corresponds to an I value of...

$$I_i = 100\% \times \frac{M_i}{\sum_{i=1}^{6} M_i}$$

$$I_2 = 100\% \times \frac{M_2}{M_1 + M_2 + M_3 + M_4 + M_5 + M_6}$$

$$I_2 = \frac{100\% \times 0.66}{0.23 + 0.66 + 0.011 + 0.4 + 0.06 + 0.017} = 48\%$$