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# Properties of dissolved and total organic matter in throughfall, stemflow and forest floor leachate of Central European forests

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Abstract. For the first time, we investigated the composition of dissolved organic matter (DOM) compared to total OM (TOM, consisting of DOM,  $< 0.45 \,\mu\text{m}$  and particulate OM,  $0.45 \,\mu\text{m} < \text{POM}$   $< 500 \,\mu\text{m}$ ) in throughfall, stemflow and forest floor leachate of common beech (*Fagus sylvatica* L.) and Norway spruce (*Picea abies* (L.) H. Karst.) forests using solid state <sup>13</sup>C nuclear magnetic

5 resonance (NMR) spectroscopy. We hypothesized that the composition and properties of OM in forest ecosystem water samples differ between DOM and TOM and between the two tree species.

The <sup>13</sup>C NMR results, derived from 21 samples, point to pronounced differences in the composition of DOM and TOM in throughfall solution at the beech sites with TOM exhibiting higher relative intensities for the alkyl C region, representing aliphatic C from less decomposed organic material

10 compared to DOM. Furthermore, TOM shows lower intensities for lignin-derived and aromatic C of the aryl C region resulting in lower aromaticity indices and a diminished degree of humification. Along the ecosystem compartments, differences in the structural composition of DOM and TOM under beech lessened in the order throughfall > stemflow ≈ forest floor leachate.

In contrast to the broadleaved sites, differences between DOM vs. TOM in throughfall solution 15 under spruce were only less pronounced and spectra were overall dominated by the alkyl C region, representing aliphatic C. Explanations to the reported results might be substantiated in differences in tree species-specific structural effects, leaching characteristics or differences in the microbial community of the tree species' phyllosphere and cortisphere. However, the fact that throughfall DOM under beech showed the highest intensities of recalcitrant aromatic and phenolic C among all sam-

20 ples analyzed, likely points to a high alleleopathic potential of beech trees negatively affecting other organisms and hence ecosystem processes and functions.

**Keywords.** Dissolved organic matter (DOM), particulate organic matter (POM), total organic matter (TOM), European beech (*Fagus sylvatica* L.), solid state <sup>13</sup>C CPMAS NMR spectroscopy, Biodiversity Exploratories

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# Abbreviations.

<sup>13</sup> C CPMAS NMR	Carbon-13 cross-polarization and magic angle spinning nuclear magnetic
	resonance
ALB	Schwäbische Alb exploratory
A/O-A	Alkyl C to O-alkyl C ratio
BP	Bulk precipitation
DOM	Dissolved organic matter
FF	Forest floor
HAI	Hainich–Dün exploratory
OM	Organic matter
PE	Polyethylene
POM	Particulate organic matter
SCH	Schorfheide-Chorin exploratory
SF	Stemflow
TF	Throughfall
ТОМ	Total organic matter

# 1 Introduction

Dissolved organic matter (DOM), often described as solutes passing filter < 0.45 μm in pore size,</li>
contributes substantially to the biogeochemistry of carbon (C) and nitrogen (N) in forest ecosystems (Qualls et al., 1991; Michalzik and Matzner, 1999; Fröberg et al., 2006; Kindler et al., 2011). While many investigations addressed the nature and dynamics of DOM in terrestrial ecosystems (e.g., Qualls and Haines, 1991; Currie et al., 1996), only a few investigated the dynamics (Carlisle et al., 1966; Sollins et al., 1980; le Mellec et al., 2009, 2010) and none the character of water-bound

- total OM (TOM) including the particulate OM fraction (POM;  $0.45 \,\mu\text{m} < \text{POM} < 500 \,\mu\text{m}$ ). Since water-bound element and nutrient concentrations are conventionally measured after filtration (frequently to  $< 0.45 \,\mu\text{m}$  pore size), the exclusion of the POM fraction potentially results in misleading inferences and budgeting gaps in studies of nutrient and energy fluxes in ecosystems (Michalzik and Stadler, 2005; Stadler et al., 2006). We are aware of only a handful of studies over the last 35
- 40 years which address POM or size equivalent in temperate broadleaved and coniferous ecosystems (e.g., Carlisle et al., 1966; Sollins et al., 1980; Likens et al., 1983; Lamersdorf and Blank, 1995; Levia et al., 2013). The majority of these studies highlighted the importance of  $OM > 0.45 \,\mu\text{m}$  to the overall budgets of C, N and other nutrients. (Lamersdorf and Blank, 1995) revealed a significant

additional input of organic carbon and nitrogen to the soil with throughfall (TF) via the particulate

- 45 fraction (> 0.45 μm). These results are corroborated by recent findings for POM and DOM in TF of beech, spruce and pine forest stands (le Mellec and Michalzik, 2008; le Mellec et al., 2009, 2010), where the additional input of C and N by particulate organic matter represented between 20–30 % of the total organic C and 10–20 % of the total N fluxes with throughfall. The low C-to-N ratios of 12–14 for POM as reported by Lamersdorf and Blank (1995) furthermore suggest that this material
- 50 might be particularly relevant to promote microbial decomposition processes (Fontaine et al., 2004), while wider C-to-N ratios of > 30 tend to decelerate decomposition rates (Pérez-Harguindeguy et al., 2000).

Although some knowledge of the amount and temporal dynamics of DOM and POM is available, the composition especially of TOM in solutions and its alteration traversing forest canopies, stem surfaces and subsequently the forest floor (FF) is largely unknown. The chemical nature of mobile

- 55 surfaces and subsequently the forest floor (FF) is largely unknown. The chemical nature of mobile OM is a prerequisite for advancing understanding of the C and nutrient cycling as well as other ecosystem processes in forests (Schnitzer and Khan, 1972; Guggenberger and Zech, 1994; Wardle et al., 1998). Alteration in the composition of OM in throughfall, stemflow (SF) and FF solutions are assumed to be driven by various abiotic and biotic processes. In this context, precipitation patterns
- 60 (Mercier and Lindow, 2000) and canopy architecture (Levia and Frost, 2006; Levia et al., 2012) basically alter the release of OM with throughfall. Atmospheric deposition (e.g. pollen, inorganic N deposition, Lamersdorf and Blank (1995)) and increased nutrient leaching from damaged leaves and immobilization processes within the canopy mediated by phyllosphere microorganisms (Guggenberger and Zech, 1994; Stadler and Müller, 2000) appear to influence the chemical qualities of OM
- 65 from the forest canopy. Interspecific differences in bark traits and cortisphere properties, e.g. bark texture and colour, influence the microclimate and arthropod fauna (Nicolai, 1986), the plant epiphyte (Wyse and Burns, 2011) and microbial community (Andrews and Harris, 2000) consequently affecting the amount and composition of OM as well. For example, enhanced bark microrelief increases the contact time of stemflow solutions with bark surfaces leading to an enrichment of aro-
- 70 matic hydrocarbons (Levia et al., 2012) likely due to the enhanced degradation and release of soluble lignin compounds (Guggenberger et al., 1994).

Previous work has demonstrated that tree species and spatial and temporal patterns of environmental conditions such as chemical soil properties influence the input and cycling of cations (Gersper and Holowaychuk, 1971; Finzi et al., 1998; Levia et al., 2011; Van Stan et al., 2012) and DOM in

- 75 temperate forests (Levia et al., 2012). However, it is unclear how tree species influence chemical properties of DOM and especially of POM in TF, SF and FF leachates and hence subsequently affect soil processes by varying chemical composition and associated functionality. In this context, beech trees potentially exhibit the highest amounts of phenolics in leaves (Bussotti et al., 1998), bark (Dübeler et al., 1997), wood (Mämmelä, 2001) and roots (Weissen and Van Praag, 1991) compared to
- 80 co-occurring species. Phenolic compounds released from living or dead plant material have been

shown to function as allelochemicals, affecting associated species (Rice, 1984; Wardle et al., 1998). An interspecific allelopathic potential on other organisms may result in net changes in ecosystem processes and functions (e.g. herbivory, decomposition and nutrient mineralization) (Wardle et al., 1998). However, the release of phenolic or other allelopathic compounds from living plant material

via throughfall and stemflow and the effects on ecosystem processes and function is still insuffi-85 ciently understood.

To elucidate the composition of DOM and TOM in terrestrial ecosystem water samples we applied solid state <sup>13</sup>C nuclear magnetic resonance spectroscopy with cross-polarization and magic angle spinning (CPMAS-NMR), which is a nondestructive method that provides information on the

- chemical surroundings of C atoms forming the major component of OM. Consequently, solid state 90  $^{13}$ C NMR spectroscopy is widely used for the characterization of a range of organic materials from various natural environments (Baldock et al., 2004). It has been applied to soil OM (Helfrich et al., 2006; Keeler et al., 2006), hot water extracts (Leinweber et al., 1995; Balaria et al., 2009) and DOM in aqueous solutions from aquatic (Schumacher et al., 2006) and terrestrial ecosystems (Guggen-
- berger et al., 1994; Fröberg et al., 2003; Sanderman et al., 2008). Although the technique has been 95 applied to POM from aquatic systems (McKnight et al., 1997; Lankes et al., 2008), to the best of our knowledge, it has never been applied to POM or TOM in aqueous samples from terrestrial ecosystems like TF, SF or FF leachates. Therefore, we present the first study investigating the composition of DOM compared to TOM by solid state <sup>13</sup>C NMR spectroscopy.
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We tested the hypotheses that (i) the structural composition of DOM in filtered and TOM in unfiltered forest ecosystem water samples differ along ecosystem compartments and (ii) between tree species, exemplarily tested between beech and spruce at one of our study regions.

### 2 Methods

#### 2.1 Study sites

- The sites are in three landscapes forming part of the German "Biodiversity Exploratories" (www. 105 biodiversity-exploratories.de), which were established in 2006 as a scientific platform for biodiversity research funded by the German Research Foundation (DFG SPP 1374 "Exploratories for Largescale and Long-term Functional Biodiversity Research"; see Fischer et al. (2010) for a detailed description). The three regions differ in environmental conditions (Table 1) and are representative
- 110 of large areas of Central Europe. We selected three extensively managed "close to native" European beech forests (Fagus sylvatica L.; mature forest stands, protected for at least 60 years; see Fischer et al., 2010) in each of the three exploratories. These forests are comparable with regard to tree age and tree species composition, and are similar to stand density, mean tree diameter at breast hight, basal area and forest management (Table 2). In the Hainich-Dün exploratory (HAI) we additionally
- 115 collected samples from three Norway spruce (Picea abies (L.) H. Karst.) plantations (Table 2). We

only choose the HAI spruce sites, because the Schorfheide-Chorin (SCH) exploratory only provides pine (Pinus sylvestris L.) stands and the spruce stands of the Schwäbische Alb exploratory (ALB) differ in tree species composition (Schwarz et al., 2014).

# 2.2 Experimental setup

- The experimental setup was identical for all the sites. At each exploratory, bulk precipitation (BP) 120 was collected in five replicates from three open areas using 2 L polyethylene (PE) open funnel samplers (diameter 0.12 m). TF was sampled using rain collectors of the same type arranged in a crossshaped grid with 20 replicates per forest site. The sampling bottles were wrapped with aluminum foil to minimize the impact of irradiation and algal growth and were equipped with a polyester net
- 125 (mesh width 1.6 mm) to prevent sample contamination with coarse matter. Stemflow was collected in three replicates per site using sliced polyurethane hoses (diameter 4 cm) as a collar installed at 1.60 m height, sealed with polyurethane glue and connected with a polypropylene or PE barrel via a PE tube. To address the element input from the organic layer to the mineral soil, FF leachates were collected with three zero-tension lysimeters per site  $(280 \,\mathrm{cm}^2 \,\mathrm{sampling area})$ , installed in the
- vicinity of the TF samplers. They consisted of polyvinyl chloride plates covered with a PE net (mesh 130 width 0.5 mm) connected with PE hoses to 2 L PE bottles stored in a box below ground.

#### Sampling procedure and analysis 2.3

Samples of BP, TF, SF and FF leachate were collected in September 2010 for ALB and the HAI beech sites and in August/ September 2011 for SCH and the HAI spruce sites after a bi-weekly routine sam-

- 135 ple collection interval. The reasons to choose these sampling dates were threefold: beech canopies were fully developed, external inferring events such as the release of pollen by coniferous trees was over (mid May to mid June) and generally sufficient precipitation water occurs during this time of the year to perform the freeze-drying of large amounts of sample solutions for solid state  ${}^{13}$ C NMR analysis. Individual samples of BP (3 sites  $\times$  5 replicates = 15), TF (3 sites  $\times$  20 replicates = 60), SF
- $(3 \text{ sites} \times 3 \text{ replicates} = 9)$  and FF leachates  $(3 \text{ sites} \times 3 \text{ replicates} = 9)$  were pooled to one volume-140 weighted sample per sample type, sampling date and exploratory, respectively. The samples were stored on ice in cool boxes until further processing in the laboratory. At latest 24 h after collection, an aliquot of the BP, TF, SF and FF leachate samples was bottled unfiltered, whereas another was filtered with  $< 0.45 \,\mu\text{m}$  cellulose acetate filters (Sartorius, Göttingen, Germany). Both the unfiltered and filtered aliquots were immediately frozen to prevent microbial alteration.
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For solid state <sup>13</sup>C NMR analysis, between 400 and 800 mL of the filtered and unfiltered sample aliquots were freeze-dried (Christ Alpha 1-4, Osterode/ Harz, Germany). Samples of BP contained insufficient amount (< 30 mg) of dry matter and could therefore not be analyzed with NMR spectroscopy. The same held true for three other ALB samples, encompassing the filtered TF sample and

150 the filtered and unfiltered SF samples. In total, 21 samples were analyzed by solid state <sup>13</sup>C NMR spectroscopy. Table 3 provides an overview over the sample consortium.

Solid state <sup>13</sup>C NMR measurements were obtained with a Bruker DSX 200 NMR spectrometer (Bruker, Karlsruhe, Germany). CPMAS was applied with a <sup>13</sup>C resonance frequency of 50.32 MHz and a spinning speed of 6.8 kHz. A ramped <sup>1</sup>H pulse starting at 100 % and decreasing to 50 % of

- 155 the initial power was used during a contact time of 1 ms in order to circumvent spin modulation during the Hartmann–Hahn contact. Pulse delays of 2 s were used for all spectra. Pre-experiments confirmed that the pulse delays were long enough to avoid saturation. Depending on the C content of the samples, 2070 to 130 939 scans were accumulated and a line broadening between 50 and 100 Hz was applied. The <sup>13</sup>C chemical shifts were calibrated relative to tetramethylsilane (0 ppm). Relative
- 160 contributions from the various C groups were determined by integration of the signal intensity in the respective chemical shift regions according to Knicker et al. (2005). The region from 220 to 160 ppm was assigned to carbonyl (aldehyde and ketone) and carboxyl/ amide C. Olefinic and aromatic C were detected between 160 and 110 ppm. O-alkyl and N-alkyl C signals were found from 110 to 60 ppm and from 60 to 45 ppm, respectively. Resonances of alkyl C were assigned to the region
- 45 to -10 ppm. We calculated the alkyl C/O-alkyl C (A/O-A) ratio for each fraction according to Baldock et al. (1997) as a sensitive index for the extent of decomposition. The relative aromaticity (in %) was estimated from the ratio aryl C/(aryl C + O-alkyl C + alkyl C) as proposed by Hatcher et al. (1981).

# 3 Results and discussion

- All spectra shown in Fig. 1 were dominated by the O-alkyl C region (110–45 ppm) and exhibited pronounced signals at ca. 72 ppm, corresponding to secondary carbinol C (i.e. C-2, C-3 and C-5 of pyranoside rings) in cellulose and hemicellulose (Almendros et al., 2000; Keeler et al., 2006; Lemma et al., 2007; Conte et al., 2010). Resonances at ca. 103 ppm were assigned to anomeric C in carbohydrates (Rodríguez-Murillo et al., 2011), whereas N-substituted C of amino
  sugars contributed to the region between 58 and 51 ppm (Knicker and Kögel-Knabner, 1998).
- In all samples, a maximum peak in the carboxyl C region at ca. 172 ppm indicated the presence of carboxylic acids, amides and esters originating from compounds such as cutin, proteins and hydrolysable tannins (Lemma et al., 2007).

### 180 Throughfall

Unfiltered and filtered TF samples from the beech sites differed remarkably with regard to C composition. Unfiltered samples revealed clearly decreased signal intensities for the aryl C region (160–110 ppm) that mostly refers to lignin systems [ca. 135 ppm, unsubstituted; ca. 145 ppm, heterosubstituted; ca. 153 ppm, ether-linked; Conte et al., 2010; Rodríguez-Murillo et al., 2011] and

increased intensities for the alkyl C region (45 to -10 ppm), corroborating with higher A/O-A ratio 185 values and lower aromaticity indices for TOM vs. DOM (Tables 3 and 4).

This might likely point to aliphatic C from less decomposed organic material accompanied by reduced humification intensity compared with filtered TF samples, where the intensity of the aromatic C region (140–110 ppm) was amplified by 60 % on average. For the O-alkyl C (110–45 ppm)

190 and carboxyl C (220-160 ppm) regions, the trends appeared inconsistent (Table 3, Fig. 1). In essence, DOM in TF under beech was dominated by aryl C at the expense of alkyl C, resulting in remarkably higher aromaticity index values. A high proportion in aromatic C in DOM points to increased recalcitrance of DOM towards microbial degradability due to the inhibitory effects of these compounds on enzyme activity (e.g.  $\alpha$ -amylase, trypsin, lysozyme) (Rohn et al., 2002; Marschner 195

and Kalbitz, 2003).

About 30% of the aryl C was composed of phenolic C (Table 3), partly forming secondary metabolites, which are known to inhibit both germination and growth of plants, thereby affecting the distribution patterns and composition of understory vegetation of forests (Rice, 1984; Muscolo et al., 2001; Hane et al., 2003). Furthermore, they affect the growth of mycorrhizal fungi (Lindeberg,

- 200 1985) and inhibit feeding by various fungi (Alfredsen et al., 2008) and potential herbivores (Dübeler et al., 1997; Petrakis et al., 2011). In their review, Wardle et al. (1998) concluded that the effect of plants with allelopathic potential on other organisms may ultimately result not only in net change at the population level (e.g. plant-plant interference), but in net change in ecosystem processes and functions (e.g. herbivory, decomposition and nutrient mineralization), which may be important in shaping plant community structure in the long term.
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In comparison with beech, the analysis of <sup>13</sup>C chemical shift regions of the unfiltered and filtered TF samples from the HAI spruce forests indicated only minor differences in C fraction composition (Table 3, Fig. 1). In this context, the C of the unfiltered TF samples exhibited slightly higher relative intensities for aliphatic C (45 to -10 ppm) and carbohydrate C (90–60 ppm) and lower relative in-

210 tensities for aryl C (160–110 ppm) and carboxyl C (220–160 ppm) compared with filtered samples. This aspect was mirrored by the lower aromaticity index of the unfiltered samples (17.0) compared with the filtered samples (19.6) (Table 4).

The less pronounced differences in DOM and TOM composition under spruce compared to beech appear to be substantiated in differing interspecific canopy metrics. Spruce trees in Germany exhibit a denser foliage and higher leaf area index of  $20.5 \text{ m}^2 \text{ m}^{-2}$  compared with  $6.5 \text{ m}^2 \text{ m}^{-2}$  for beech, as 215 well as higher needle surface roughness and higher leaf longevity (Rothe et al., 2002). Consequently, spruce canopies provide a higher filtering capacity for airborne compounds and higher accumulation rate than beech canopies. Additionally, spruce canopies exhibit higher rainfall interception rate values of 30-40% compared with 17-20% for beech canopies (Rothe et al., 2002). This might lead to

longer periods of water availability, longer contact time for rainwater and needle surfaces and sub-220 strate availability and hence more stable environmental conditions within the phyllosphere, likely promoting the microbial decomposition of TOM to DOM and hence diminishing structural differences.

Compared to the HAI spruce sites, Sanderman et al. (2008) found similar distributions in the relative intensities of chemical shift regions for DOM in TF of a nearly 100 years old mixed Redwood and Douglas-fir stand in California. In contrast, Fröberg et al. (2003) reported differing intensities for DOM in TF from a a 35 yr old Norway spruce stand in southern Sweden, with lower carbonyl C and aryl C, but higher O-alkyl C contents, even resulting in lower A/O-A ratio (0.42) and aromaticity index (12.24 %). To our knowledge, there are no comparable studies of

230 beech-derived TF samples as of SF samples in general. The diverting results from Sweden are likely due to the younger stand age or differences in stand structure and climate.

# Stemflow

As illustrated by difference spectra, unfiltered and filtered SF samples under beech as well as under spruce showed similar distributions and signal intensities (Fig.1). However, the difference spectra from SCH samples exhibited pronounced peaks at ca. 30 ppm within the alkyl C region, which can be assigned to methylenes in fatty acyl chains (Lundberg et al., 2001). These peaks point to a local event at the SCH beech sites which might be derived from excretions of canopy insects such as the beech woolly aphid (*Phyllaphis fagi* L.) or other leaf-sucking insects. Herbivorous

- 240 insect species are known to temporarily occur in mass infestations and then alter the input fluxes, especially of POM, by e.g. green leaf litter, wax wool excretion and cell tissue (Stadler et al., 2001; le Mellec and Michalzik, 2008). In this context, Greenway et al. (1974) reported on the composition of triglycerides found in cornicle secretions from a range of aphid families, including *P. fagi.* Additionally, Stanley-Samuelson et al. (1988) reviewed the formation of fatty acids in cuticular
- 245 secretion components of other insect orders.

# Forest floor leachates

Regarding FF leachates under beech, unfiltered samples of two (HAI and SCH) out of the three exploratories exhibited higher intensities within the O-alkyl C region (110–45 ppm) and lower

- 250 signals in the carboxyl C region (220–160 ppm) compared with the filtered samples (Fig. 1). While the aromaticity index among the FF leachate fractions were similar, with values around 22 and 24 %, the A/O-A ratio values were lower for the TOM than the DOM fractions (Table 4). This trend was observed for FF leachates from all the sites and for both tree species, except for the ALB beech sites. Here, the two fractions of FF leachate revealed only minor differences in
- 255 C composition and A/O-A ratio. Again, no corresponding data for beech stands were available in the literature, but for a spruce forest site in southern Sweden, Fröberg et al. (2003) reported similar signal intensities from solid state <sup>13</sup>C NMR spectra of Oa layer solutions. In contrast, <sup>13</sup>C NMR data for FF leachates of a Californian mixed Redwood stand reported by Sanderman et al.

(2008) exhibited lower intensities for alkyl C and remarkably higher intensities for the aromatic

- 260 C region, resulting in an approximately halved A/O-A ratio compared with our spruce sites. We assume that the diverting sampling period and season along with different climatic and hydrological conditions (Mediterranean vs. Temperate) are responsible for the differences in DOM composition. In this context, enhanced hydrological flux generation (e.g. throughout rainy seasons) was shown to increase the aromaticity of DOM by the intensified degradation of soluble lignin compounds as
- observed by Guggenberger et al. (1994) and Levia et al. (2012).

In summary, the compositional differences between DOM and TOM at the beech sites diminished from  $TF > SF \approx FF$  leachates. This might be related to longer stemflow–bark contact time due to the bark micro-relief, less changing environmental conditions and better substrate availability

270 on the tree stem (Levia et al., 2012) and stagnant water on the litter surfaces of the forest floor favoring degradation of canopy-derived OM by the soil biota and epiphytic microorganisms that level out differences in the composition and quality between the two size fractions.

Differing composition of DOM under beech vs. spruce might be substantiated in differences in tree species-specific structural effects, leaching characteristics or differences in the microbial community

of the tree species' phyllosphere and cortisphere (Guggenberger and Zech, 1994; Stadler and Müller, 2000).

The discussion generally lacks of comparable data on TOM composition as, to the best of our knowledge, there are no studies available reporting on solid state <sup>13</sup>C NMR spectroscopy-derived TOM characteristics in TF, SF and FF leachate samples under beech or other tree species.

# 280 4 Conclusions

Structural data derived from solid state <sup>13</sup>C NMR analysis confirmed pronounced differences between filtered and unfiltered beech TF samples. The compositional differences between DOM and TOM diminished from TF to SF and FF leachates. This might be due to increased contact time and intensified microbial activity levelling out the compositional differences between the two OM frac-

tions. Since differences in the structural composition of DOM and TOM in TF and SF were less pronounced under spruce than under beech, we suspect a tree species-related effect on the origin of OM composition and resulting properties such as recalcitrance and allelopathic potential. To consolidate our findings on tree species-specific differences in the amount and functionality of OM in TF samples and ecological implications, research has to be extended spatially and to further tree 290 species.

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	Schorfheide-Chorin (SCH)	Hainich-Dün (HAI)	Schwäbische Alb (ALB)
Location	Northeast Germany (53° 2′ N, 13° 51′ E)	Central Germany (51° 10' N, 10° 23' E)	Southwest Germany (38° 24' N, 9° 24' E)
Altidute above see level	3–140 m	285–550 m	460–860 m
Mean annual temperature	8–8.5 °C	6.5–8 °C	6–7 °C
Mean annual precipitation	500–600 mm	500-800 mm	700–1000 mm
Geological substrate	Quarzitic glacial till	Triassic limestone with loess cover	Jurassic limestone
Main soil types	Cambisols	Luvisols	Leptosols
	Albeluvisols	Stagnosols	Cambisols
Main tree species	Fagus sylvatica L.	Fagus sylvatica L.	Fagus sylvatica L.
	Pinus sylvestris L.	Picea abies (L.) H. Karst.	Picea abies (L.) H. Karst
	Quercus spp.	Fraxinus excelsior L.	

**Table 1.** Environmental properties of the study regions (mainly given by Fischer et al., 2010; soil classificationaccording to IUSS Working group WRB, 2006).

**Table 2.** Stand structural properties and management type of the study plots (dbh = tree diameter at breast hight; data from Schwarz et al., 2014).

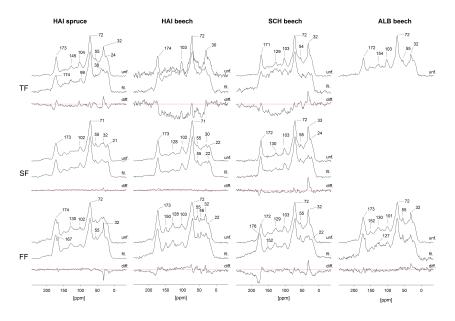
Region	Plot ID	Main tree species	Management type	Stand density (n ha <sup>-</sup> 1)	Mean dbh (cm)	Basal area $(m^2 ha^- 1)$
Schwäbische Alb	AEW7	Beech	extensively managed	200	85.2	33.3
	AEW8	Beech	extensively managed	276	56.9	40.0
	AEW9	Beech	extensively managed	376	42.5	31.7
Hainich-Dün	HEW1	Spruce	age-class forest	276	49.4	43.0
	HEW2	Spruce	age-class forest	660	35.9	41.3
	HEW3	Spruce	age-class forest	651	32.2	42.7
	HEW10	Beech	extensively managed	377	50.3	34.3
	HEW11	Beech	extensively managed	568	51.1	36.7
	HEW12	Beech	extensively managed	326	54.9	36.5
Schorfheide-Chorin	SEW7	Beech	extensively managed	155	61.2	36.9
	SEW8	Beech	extensively managed	154	68.0	38.3
	SEW9	Beech	extensively managed	259	52.5	43.6

max.         min.           220         185         Carboxyl C         1.4         2.8         2.5         2.4         3.1         2.6           185         160         Carboxyl C         9.2         9.7         10.2         10.2         12.8         12.7           160         140         Aryl C         Phenolic C         4.9         5.7         5.6         5.5         6.0         5.8           140         110         Aromatic C         10.3         11.5         11.5         11.1         12.8         12.2           90         60         Carbohydrate C         29.3         27.4         29.9         30.4         25.3         24.8           60         45         Methoxyl C         10.0         9.7         9.7         9.7         10.1         9.8           45         -10         Alkyl C         Aliphatic C         25.8         24.7         20.9         21.1         21.3         23.9           101         90         O-alkyl C         Phenolic C         5.1         9.0         5.3         4.9         7.1         6.9           110         90         O-alkyl C         Di-o-akyl C         7.4         11.7         9		Integrat	ed region	Chemica	ll shift region	TF unf.	TF fil.	SF unf.	SF fil.	FF unf.	FF fil.
PUP         185         160         Carboxyl C         9.2         9.7         10.2         10.2         10.2         12.8         12.7           160         140         Aryl C         Phenolic C         4.9         5.7         5.6         5.5         6.0         5.8           140         110         Aromatic C         10.3         11.5         11.5         11.1         12.8         12.2           90         60         Carbohydrate C         29.3         27.4         29.9         30.4         25.3         24.8           60         45         Methoxyl C         10.0         9.7         9.7         9.7         10.1         9.8           45         -10         Alkyl C         Aliphatic C         25.8         24.7         20.9         21.1         21.3         23.9           220         185         Carboxyl C         Carboxyl C         3.9         1.5         2.4         2.1         2.4         4.4           185         160         Carboxyl C         7.4         11.7         9.7         9.3         13.7         12.9           110         90         0-alkyl C         7.4         1.7         9.7         9.3         8.2<		max.	min.								
operation         160         140         Aryl C         Phenolic C         4.9         5.7         5.6         5.5         6.0         5.8           140         110         Aromatic C         10.3         11.5         11.5         11.1         12.8         12.2           110         90         O-alkyl C         Di-O-alkyl C         90         8.6         9.7         9.6         8.7         8.2           90         60         Carbohydrate C         29.3         27.4         29.9         30.4         25.3         24.8           60         45         Methoxyl C         10.0         9.7         9.7         9.7         10.1         9.8           45         -10         Alkyl C         Aliphatic C         25.8         24.7         20.9         21.1         21.3         23.9           220         185         Carboxyl C         Carboxyl C         10.2         8.8         11.3         11.0         13.3         13.7           160         140         Aryl C         Phenolic C         5.1         9.0         5.3         4.9         7.1         6.9           110         90         0-alkyl C         Di-o-alkyl C         7.4         11.7		220	185	Carboxyl C	Carbonyl C	1.4	2.8	2.5	2.4	3.1	2.6
OPE         140         110         Aromatic C         10.3         11.5         11.1         12.8         12.2           110         90         O-alkyl C         Di-O-alkyl C         90         8.6         9.7         9.6         8.7         8.2           90         60         Carbohydrate C         29.3         27.4         29.9         30.4         25.3         24.8           60         45         Methoxyl C         10.0         9.7         9.7         9.7         10.1         9.8           45         -10         Alkyl C         Aliphatic C         25.8         24.7         20.9         21.1         21.3         23.9           220         185         Carboxyl C         Carboxyl C         10.2         8.8         11.3         11.0         13.3         13.7           160         140         Aryl C         Phenolic C         5.1         9.0         5.3         4.9         7.1         6.9           110         90         O-alkyl C         Di-O-alkyl C         7.4         11.7         9.7         9.3         8.2           90         60         Carbohyl C         28.3         24.5         30.8         31.3         24.2		185	160		Carboxyl C	9.2	9.7	10.2	10.2	12.8	12.7
90         60         Carbohydrate C         29.3         27.4         29.9         30.4         25.3         24.8           60         45         Methoxyl C         10.0         9.7         9.7         9.7         10.1         9.8           45         -10         Alkyl C         Aliphatic C         25.8         24.7         20.9         21.1         21.3         23.9           220         185         Carboxyl C         Carboxyl C         10.2         8.8         11.3         11.0         13.3         13.7           160         140         Aryl C         Phenolic C         5.1         9.0         5.3         4.9         7.1         6.9           110         90         O-alkyl C         Di-O-alkyl C         7.4         11.7         9.7         9.3         9.3         8.2           90         60         Carbonyl C         16.4         11.1         9.3         9.3         8.2           90         60         Carbonyl C         28.3         24.5         30.8         31.3         24.2         22.6           60         45         Methoxyl C         28.5         8.6         9.5         9.6         10.4         9.9	ece	160	140	Aryl C	Phenolic C	4.9	5.7	5.6	5.5	6.0	5.8
90         60         Carbohydrate C         29.3         27.4         29.9         30.4         25.3         24.8           60         45         Methoxyl C         10.0         9.7         9.7         9.7         10.1         9.8           45         -10         Alkyl C         Aliphatic C         25.8         24.7         20.9         21.1         21.3         23.9           220         185         Carboxyl C         Carboxyl C         10.2         8.8         11.3         11.0         13.3         13.7           160         140         Aryl C         Phenolic C         5.1         9.0         5.3         4.9         7.1         6.9           110         90         O-alkyl C         Di-O-alkyl C         7.4         11.7         9.7         9.3         9.3         8.2           90         60         Carbonyl C         16.4         11.1         9.3         9.3         8.2           90         60         Carbonyl C         28.3         24.5         30.8         31.3         24.2         22.6           60         45         Methoxyl C         28.5         8.6         9.5         9.6         10.4         9.9	spru	140	110		Aromatic C	10.3	11.5	11.5	11.1	12.8	12.2
90         60         Carbohydrate C         29.3         27.4         29.9         30.4         25.3         24.8           60         45         Methoxyl C         10.0         9.7         9.7         9.7         10.1         9.8           45         -10         Alkyl C         Aliphatic C         25.8         24.7         20.9         21.1         21.3         23.9           220         185         Carboxyl C         Carboxyl C         10.2         8.8         11.3         11.0         13.3         13.7           160         140         Aryl C         Phenolic C         5.1         9.0         5.3         4.9         7.1         6.9           110         90         O-alkyl C         Di-O-alkyl C         7.4         11.7         9.7         9.3         9.3         8.2           90         60         Carbonyl C         16.4         11.1         9.3         9.3         8.2           90         60         Carbonyl C         28.3         24.5         30.8         31.3         24.2         22.6           60         45         Methoxyl C         28.5         8.6         9.5         9.6         10.4         9.9	IAI	110	90	O-alkyl C	Di-O-alkyl C	9.0	8.6	9.7	9.6	8.7	8.2
45         -10         Alkyl C         Aliphatic C         25.8         24.7         20.9         21.1         21.3         23.9           220         185         Carboxyl C         Carboxyl C         3.9         1.5         2.4         2.1         2.4         4.4           185         160         Carboxyl C         10.2         8.8         11.3         11.0         13.3         13.7           160         140         Aryl C         Phenolic C         5.1         9.0         5.3         4.9         7.1         6.9           110         90         O-alkyl C         Di-O-alkyl C         7.4         11.7         9.7         9.3         8.2           90         60         Carbohydrate C         28.3         24.5         30.8         31.3         24.2         22.6           60         45         Methoxyl C         8.5         8.6         9.5         9.8         10.4         9.9           45         -10         Alkyl C         Aliphatic C         26.8         18.8         21.1         22.5         19.5         21.4           185         160         Carboxyl C         9.3         10.2         9.5         10.7         9.6	Ц	90	60		Carbohydrate C	29.3	27.4	29.9	30.4	25.3	24.8
Provide         Provide         Structure         St		60	45		Methoxyl C	10.0	9.7	9.7	9.7	10.1	9.8
HS         160         Carboxyl C         10.2         8.8         11.3         11.0         13.3         13.7           HO         140         Aryl C         Phenolic C         5.1         9.0         5.3         4.9         7.1         6.9           140         110         Aromatic C         9.7         17.1         9.9         9.1         13.7         12.9           110         90         O-alkyl C         Di-O-alkyl C         7.4         11.7         9.7         9.3         9.3         8.2           90         60         Carbohydrate C         28.3         24.5         30.8         31.3         24.2         22.6           60         45         Methoxyl C         8.5         8.6         9.5         9.8         10.4         9.9           45         -10         Alkyl C         Aliphatic C         26.8         18.8         21.1         22.5         19.5         21.4           20         185         Carboxyl C         9.3         10.2         9.5         10.7         9.6         11.9           160         140         Aryl C         Phenolic C         7.0         10.2         3.9         5.3         6.5         6.2<		45	-10	Alkyl C	Aliphatic C	25.8	24.7	20.9	21.1	21.3	23.9
Homology         160         140         Aryl C         Phenolic C         5.1         9.0         5.3         4.9         7.1         6.9           140         110         Aromatic C         9.7         17.1         9.9         9.1         13.7         12.9           110         90         O-alkyl C         Di-O-alkyl C         7.4         11.7         9.7         9.3         9.3         8.2           90         60         Carbohydrate C         28.3         24.5         30.8         31.3         24.2         22.6           60         45         Methoxyl C         8.5         8.6         9.5         9.8         10.4         9.9           45         -10         Alkyl C         Aliphatic C         26.8         18.8         21.1         22.5         19.5         21.4           220         185         Carboxyl C         9.3         10.2         9.5         10.7         9.6         11.9           160         140         Aryl C         Phenolic C         7.0         10.2         3.9         5.3         6.5         6.2           110         90         O-alkyl C         Di-O-alkyl C         10.2         11.9         8.2		220	185	Carboxyl C	Carbonyl C	3.9	1.5	2.4	2.1	2.4	4.4
Page H40         110         Aromatic C         9.7         17.1         9.9         9.1         13.7         12.9           110         90         O-alkyl C         Di-O-alkyl C         7.4         11.7         9.7         9.3         9.3         8.2           90         60         Carbohydrate C         28.3         24.5         30.8         31.3         24.2         22.6           60         45         Methoxyl C         8.5         8.6         9.5         9.8         10.4         9.9           45         -10         Alkyl C         Aliphatic C         26.8         18.8         21.1         22.5         19.5         21.4           220         185         Carboxyl C         Garboxyl C         9.3         10.2         9.5         10.7         9.6         11.9           160         140         Aryl C         Phenolic C         7.0         10.2         3.9         5.3         6.5         6.2           110         90         O-alkyl C         Di-O-alkyl C         10.2         11.9         8.2         8.8         9.9         8.1           90         60         Carboxyl C         13.3         19.0         8.2         9.7		185	160		Carboxyl C	10.2	8.8	11.3	11.0	13.3	13.7
90         60         Carbohydrate C         28.3         24.5         30.8         31.3         24.2         22.6           60         45         Methoxyl C         8.5         8.6         9.5         9.8         10.4         9.9           45         -10         Alkyl C         Aliphatic C         26.8         18.8         21.1         22.5         19.5         21.4           220         185         Carboxyl C         Carbonyl C         1.4         3.6         0.6         0.8         1.6         6.7           185         160         Carboxyl C         9.3         10.2         9.5         10.7         9.6         11.9           160         140         Aryl C         Phenolic C         7.0         10.2         3.9         5.3         6.5         6.2           110         90         O-alkyl C         Di-O-alkyl C         10.2         11.9         8.2         9.7         13.4         12.0           90         60         Carbohydrate C         27.0         23.5         30.6         29.4         27.5         24.0           45         -10         Alkyl C         Aliphatic C         22.5         14.3         28.8         24.9 <td>ų</td> <td>160</td> <td>140</td> <td>Aryl C</td> <td>Phenolic C</td> <td>5.1</td> <td>9.0</td> <td>5.3</td> <td>4.9</td> <td>7.1</td> <td>6.9</td>	ų	160	140	Aryl C	Phenolic C	5.1	9.0	5.3	4.9	7.1	6.9
90         60         Carbohydrate C         28.3         24.5         30.8         31.3         24.2         22.6           60         45         Methoxyl C         8.5         8.6         9.5         9.8         10.4         9.9           45         -10         Alkyl C         Aliphatic C         26.8         18.8         21.1         22.5         19.5         21.4           220         185         Carboxyl C         Carbonyl C         1.4         3.6         0.6         0.8         1.6         6.7           185         160         Carboxyl C         9.3         10.2         9.5         10.7         9.6         11.9           160         140         Aryl C         Phenolic C         7.0         10.2         3.9         5.3         6.5         6.2           110         90         O-alkyl C         Di-O-alkyl C         10.2         11.9         8.2         9.7         13.4         12.0           90         60         Carbohydrate C         27.0         23.5         30.6         29.4         27.5         24.0           45         -10         Alkyl C         Aliphatic C         22.5         14.3         28.8         24.9 <td>bee</td> <td>140</td> <td>110</td> <td></td> <td>Aromatic C</td> <td>9.7</td> <td>17.1</td> <td>9.9</td> <td>9.1</td> <td>13.7</td> <td>12.9</td>	bee	140	110		Aromatic C	9.7	17.1	9.9	9.1	13.7	12.9
90         60         Carbohydrate C         28.3         24.5         30.8         31.3         24.2         22.6           60         45         Methoxyl C         8.5         8.6         9.5         9.8         10.4         9.9           45         -10         Alkyl C         Aliphatic C         26.8         18.8         21.1         22.5         19.5         21.4           220         185         Carboxyl C         Carbonyl C         1.4         3.6         0.6         0.8         1.6         6.7           185         160         Carboxyl C         9.3         10.2         9.5         10.7         9.6         11.9           160         140         Aryl C         Phenolic C         7.0         10.2         3.9         5.3         6.5         6.2           110         90         O-alkyl C         Di-O-alkyl C         10.2         11.9         8.2         9.7         13.4         12.0           90         60         Carbohydrate C         27.0         23.5         30.6         29.4         27.5         24.0           45         -10         Alkyl C         Aliphatic C         22.5         14.3         28.8         24.9 <td>IAI</td> <td>110</td> <td>90</td> <td>O-alkyl C</td> <td>Di-O-alkyl C</td> <td>7.4</td> <td>11.7</td> <td>9.7</td> <td>9.3</td> <td>9.3</td> <td>8.2</td>	IAI	110	90	O-alkyl C	Di-O-alkyl C	7.4	11.7	9.7	9.3	9.3	8.2
45         -10         Alkyl C         Aliphatic C         26.8         18.8         21.1         22.5         19.5         21.4           220         185         Carboxyl C         Carbonyl C         1.4         3.6         0.6         0.8         1.6         6.7           185         160         Carboxyl C         9.3         10.2         9.5         10.7         9.6         11.9           160         140         Aryl C         Phenolic C         7.0         10.2         3.9         5.3         6.5         6.2           110         90         O-alkyl C         Di-O-alkyl C         10.2         11.9         8.2         9.7         13.4         12.0           90         60         Carbohydrate C         27.0         23.5         30.6         29.4         27.5         24.0           60         45         Methoxyl C         9.3         7.3         10.2         10.4         9.7         9.1           45         -10         Alkyl C         Aliphatic C         22.5         14.3         28.8         24.9         21.7         22.0           185         160         Carboxyl C         10.1         n.a.         n.a.         n.a.	-	90	60		Carbohydrate C	28.3	24.5	30.8	31.3	24.2	22.6
220         185         Carboxyl C         Carbonyl C         1.4         3.6         0.6         0.8         1.6         6.7           185         160         Carboxyl C         9.3         10.2         9.5         10.7         9.6         11.9           160         140         Aryl C         Phenolic C         7.0         10.2         3.9         5.3         6.5         6.2           140         110         Aromatic C         13.3         19.0         8.2         9.7         13.4         12.0           90         60         Carbohydrate C         27.0         23.5         30.6         29.4         27.5         24.0           60         45         Methoxyl C         9.3         7.3         10.2         10.4         9.7         9.1           45         -10         Alkyl C         Aliphatic C         22.5         14.3         28.8         24.9         21.7         22.0           185         160         Carboxyl C         10.1         n.a.         n.a.         n.a.         1.2         13.2           160         140         Aryl C         Phenolic C         6.9         n.a.         n.a.         n.a.         1.2.2		60	45		Methoxyl C	8.5	8.6	9.5	9.8	10.4	9.9
185         160         Carboxyl C         9.3         10.2         9.5         10.7         9.6         11.9           160         140         Aryl C         Phenolic C         7.0         10.2         3.9         5.3         6.5         6.2           140         110         Aromatic C         13.3         19.0         8.2         9.7         13.4         12.0           90         60         Carbohydrate C         27.0         23.5         30.6         29.4         27.5         24.0           60         45         Methoxyl C         9.3         7.3         10.2         10.4         9.7         9.1           45         -10         Alkyl C         Aliphatic C         22.5         14.3         28.8         24.9         21.7         22.0           185         160         Carboxyl C         1.9         n.a.         n.a.         n.a.         1.2         13.2           160         140         Aryl C         Phenolic C         6.9         n.a.         n.a.         n.a.         1.2         13.2           160         140         Aryl C         Phenolic C         6.9         n.a.         n.a.         n.a.         1.6.1		45	-10	Alkyl C	Aliphatic C	26.8	18.8	21.1	22.5	19.5	21.4
Homology         160         140         Aryl C         Phenolic C         7.0         10.2         3.9         5.3         6.5         6.2           140         110         Aromatic C         13.3         19.0         8.2         9.7         13.4         12.0           110         90         O-alkyl C         Di-O-alkyl C         10.2         11.9         8.2         9.7         13.4         12.0           90         60         Carbohydrate C         27.0         23.5         30.6         29.4         27.5         24.0           60         45         Methoxyl C         9.3         7.3         10.2         10.4         9.7         9.1           45         -10         Alkyl C         Aliphatic C         22.5         14.3         28.8         24.9         21.7         22.0           220         185         Carboxyl C         1.9         n.a.         n.a.         n.a.         1.2         13.2           160         140         Aryl C         Phenolic C         6.9         n.a.         n.a.         1.2         13.2           160         140         Aryl C         Phenolic C         6.9         n.a.         n.a.         n.a. <td></td> <td>220</td> <td>185</td> <td>Carboxyl C</td> <td>Carbonyl C</td> <td>1.4</td> <td>3.6</td> <td>0.6</td> <td>0.8</td> <td>1.6</td> <td>6.7</td>		220	185	Carboxyl C	Carbonyl C	1.4	3.6	0.6	0.8	1.6	6.7
Provide         140         110         Aromatic C         13.3         19.0         8.2         9.7         13.4         12.0           110         90         O-alkyl C         Di-O-alkyl C         10.2         11.9         8.2         9.7         13.4         12.0           90         60         Carbohydrate C         27.0         23.5         30.6         29.4         27.5         24.0           60         45         Methoxyl C         9.3         7.3         10.2         10.4         9.7         9.1           45         -10         Alkyl C         Aliphatic C         22.5         14.3         28.8         24.9         21.7         22.0           220         185         Carboxyl C         Carboxyl C         1.9         n.a.         n.a.         n.a.         1.2         13.2           160         140         Aryl C         Phenolic C         6.9         n.a.         n.a.         n.a.         n.a.         1.2         13.2           160         140         Aryl C         Phenolic C         6.9         n.a.         n.a.         n.a.         1.2         13.2           110         90         O-alkyl C         Di-O-alkyl C		185	160		Carboxyl C	9.3	10.2	9.5	10.7	9.6	11.9
90         60         Carbohydrate C         27.0         23.5         30.6         29.4         27.5         24.0           60         45         Methoxyl C         9.3         7.3         10.2         10.4         9.7         9.1           45         -10         Alkyl C         Aliphatic C         22.5         14.3         28.8         24.9         21.7         22.0           220         185         Carboxyl C         Carboxyl C         1.9         n.a.         n.a.         n.a.         1.2         1.7         22.0           185         160         Carboxyl C         10.1         n.a.         n.a.         n.a.         n.a.         1.2         13.2           160         140         Aryl C         Phenolic C         6.9         n.a.         n.a.         n.a.         1.2         13.2           140         110         Aromatic C         12.7         n.a.         n.a.         n.a.         n.a.         12.6         13.6           110         90         O-alkyl C         Di-O-alkyl C         10.3         n.a.         n.a.         n.a.         8.9         9.4           90         60         Carbohydrate C         28.6         n.	-H	160	140	Aryl C	Phenolic C	7.0	10.2	3.9	5.3	6.5	6.2
90         60         Carbohydrate C         27.0         23.5         30.6         29.4         27.5         24.0           60         45         Methoxyl C         9.3         7.3         10.2         10.4         9.7         9.1           45         -10         Alkyl C         Aliphatic C         22.5         14.3         28.8         24.9         21.7         22.0           220         185         Carboxyl C         Carboxyl C         1.9         n.a.         n.a.         n.a.         1.2         1.7         22.0           185         160         Carboxyl C         10.1         n.a.         n.a.         n.a.         n.a.         1.2         13.2           160         140         Aryl C         Phenolic C         6.9         n.a.         n.a.         n.a.         1.2         13.2           140         110         Aromatic C         12.7         n.a.         n.a.         n.a.         n.a.         12.6         13.6           110         90         O-alkyl C         Di-O-alkyl C         10.3         n.a.         n.a.         n.a.         8.9         9.4           90         60         Carbohydrate C         28.6         n.	bee	140	110		Aromatic C	13.3	19.0	8.2	9.7	13.4	12.0
90         60         Carbohydrate C         27.0         23.5         30.6         29.4         27.5         24.0           60         45         Methoxyl C         9.3         7.3         10.2         10.4         9.7         9.1           45         -10         Alkyl C         Aliphatic C         22.5         14.3         28.8         24.9         21.7         22.0           220         185         Carboxyl C         Carboxyl C         1.9         n.a.         n.a.         n.a.         1.2         1.7         22.0           185         160         Carboxyl C         10.1         n.a.         n.a.         n.a.         n.a.         1.2         13.2           160         140         Aryl C         Phenolic C         6.9         n.a.         n.a.         n.a.         1.2         13.2           140         110         Aromatic C         12.7         n.a.         n.a.         n.a.         n.a.         12.6         13.6           110         90         O-alkyl C         Di-O-alkyl C         10.3         n.a.         n.a.         n.a.         8.9         9.4           90         60         Carbohydrate C         28.6         n.	SCH	110	90	O-alkyl C	Di-O-alkyl C	10.2	11.9	8.2	8.8	9.9	8.1
45         -10         Alkyl C         Aliphatic C         22.5         14.3         28.8         24.9         21.7         22.0           220         185         Carboxyl C         Carbonyl C         1.9         n.a.         n.a.         n.a.         n.a.         2.2         2.7           185         160         Carboxyl C         10.1         n.a.         n.a.         n.a.         n.a.         1.2         13.2           160         140         Aryl C         Phenolic C         6.9         n.a.         n.a.         n.a.         n.a.         6.1         6.3           140         110         Aromatic C         12.7         n.a.         n.a.         n.a.         n.a.         6.1         6.3           110         90         O-alkyl C         Di-O-alkyl C         10.3         n.a.         n.a.         n.a.         8.9         9.4           90         60         Carbohydrate C         28.6         n.a.         n.a.         n.a.         25.7         25.8           60         45         Methoxyl C         8.7         n.a.         n.a.         n.a.         10.4         9.7	•	90	60		Carbohydrate C	27.0	23.5	30.6	29.4	27.5	24.0
220         185         Carboxyl C         Carbonyl C         1.9         n.a.         n.a.         n.a.         n.a.         2.2         2.7           185         160         Carboxyl C         10.1         n.a.         n.a.         n.a.         n.a.         12.2         13.2           185         160         Carboxyl C         10.1         n.a.         n.a.         n.a.         n.a.         12.2         13.2           185         160         140         Aryl C         Phenolic C         6.9         n.a.         n.a.         n.a.         n.a.         6.1         6.3           140         110         Aromatic C         12.7         n.a.         n.a.         n.a.         n.a.         1.6.         13.6           110         90         O-alkyl C         Di-O-alkyl C         10.3         n.a.         n.a.         n.a.         8.9         9.4           90         60         Carbohydrate C         28.6         n.a.         n.a.         n.a.         10.4         9.7           60         45         Methoxyl C         8.7         n.a.         n.a.         n.a.         10.4         9.7		60	45		Methoxyl C	9.3	7.3	10.2	10.4	9.7	9.1
185         160         Carboxyl C         10.1         n.a.         n.a.         n.a.         12.2         13.2           160         140         Aryl C         Phenolic C         6.9         n.a.         n.a.         n.a.         6.1         6.3           140         110         Aromatic C         12.7         n.a.         n.a.         n.a.         12.6         13.6           110         90         O-alkyl C         Di-O-alkyl C         10.3         n.a.         n.a.         n.a.         12.6         13.6           90         60         Carbohydrate C         28.6         n.a.         n.a.         n.a.         1.0.         8.9         9.4           60         45         Methoxyl C         8.7         n.a.         n.a.         n.a.         10.4         9.7		45	-10	Alkyl C	Aliphatic C	22.5	14.3	28.8	24.9	21.7	22.0
I60         140         Aryl C         Phenolic C         6.9         n.a.         n.a.         n.a.         n.a.         6.1         6.3           140         110         Aromatic C         12.7         n.a.         n.a.         n.a.         n.a.         1.0         13.6           110         90         O-alkyl C         Di-O-alkyl C         10.3         n.a.         n.a.         n.a.         8.9         9.4           90         60         Carbohydrate C         28.6         n.a.         n.a.         n.a.         10.4         9.7           60         45         Methoxyl C         8.7         n.a.         n.a.         n.a.         10.4         9.7		220	185	Carboxyl C	Carbonyl C	1.9	n.a.	n.a.	n.a.	2.2	2.7
Product         Product <t< td=""><td></td><td>185</td><td>160</td><td></td><td>Carboxyl C</td><td>10.1</td><td>n.a.</td><td>n.a.</td><td>n.a.</td><td>12.2</td><td>13.2</td></t<>		185	160		Carboxyl C	10.1	n.a.	n.a.	n.a.	12.2	13.2
90         60         Carbohydrate C         28.6         n.a.         n.a.         n.a.         25.7         25.8           60         45         Methoxyl C         8.7         n.a.         n.a.         10.4         9.7	÷	160	140	Aryl C	Phenolic C	6.9	n.a.	n.a.	n.a.	6.1	6.3
90         60         Carbohydrate C         28.6         n.a.         n.a.         n.a.         25.7         25.8           60         45         Methoxyl C         8.7         n.a.         n.a.         10.4         9.7	bee	140	110		Aromatic C	12.7	n.a.	n.a.	n.a.	12.6	13.6
90         60         Carbohydrate C         28.6         n.a.         n.a.         n.a.         25.7         25.8           60         45         Methoxyl C         8.7         n.a.         n.a.         10.4         9.7	ALB	110	90	O-alkyl C	Di-O-alkyl C	10.3	n.a.	n.a.	n.a.	8.9	9.4
	4	90	60		Carbohydrate C	28.6	n.a.	n.a.	n.a.	25.7	25.8
45 –10 Alkyl C Aliphatic C 20.9 n.a. n.a. n.a. 21.9 19.3		60	45		Methoxyl C	8.7	n.a.	n.a.	n.a.	10.4	9.7
		45	-10	Alkyl C	Aliphatic C	20.9	n.a.	n.a.	n.a.	21.9	19.3

**Table 3.** Relative signal intensity (in %) of chemical shift regions revealed by solid-state <sup>13</sup>C NMR spectroscopy (filtered (fil.) and unfiltered (unf.) samples of throughfall (TF), stemflow (SF) and forest floor (FF) leachates; n.a., not analyzed; ALB, Schwäbische Alb; HAI, Hainich-Dün; SCH, Schorfheide-Chorin).

**Table 4.** Alkyl C/O-alkyl C (A/O-A) ratios and aromaticity (in %) calculated from chemical shift regions revealed by solid-state <sup>13</sup>C NMR spectroscopy of filtered (fil.) and unfiltered (unf.) samples of throughfall (TF), stemflow (SF) and forest floor (FF) leachates; n.a., not analyzed; ALB, Schwäbische Alb; HAI, Hainich-Dün; SCH, Schorfheide-Chorin.

	Sample	A/O-A ratio	Aromaticity [%]
HAI spruce	TF unf.	0.53	17.0
	TF fil.	0.54	19.6
	SF unf.	0.42	19.7
IAI	SF fil.	0.42	19.0
j.L.	FF unf.	0.48	22.4
	FF fil.	0.56	21.2
	TF unf.	0.61	17.3
ų	TF fil.	0.42	29.1
beec	SF unf.	0.42	17.7
HAI beech	SF fil.	0.45	16.1
щ	FF unf.	0.44	24.7
	FF fil.	0.53	24.2
SCH beech	TF unf.	0.49	22.7
	TF fil.	0.33	33.9
	SF unf.	0.59	13.5
	SF fil.	0.51	16.9
	FF unf.	0.46	22.5
	FF fil.	0.53	22.4
ALB beech	TF unf.	0.44	22.3
	TF fil.	n.a.	n.a.
	SF unf.	n.a.	n.a.
ALB	SF fil.	n.a.	n.a.
A	FF unf.	0.49	21.8
	FF fil.	0.43	23.7



**Figure 1.** Solid-state CPMAS <sup>13</sup>C NMR spectra obtained from filtered (fil.) and unfiltered (unf.) forest ecosystem water samples as well as calculated difference spectra (diff.) of both fractions; TF = throughfall, SF = stemflow, FF = forest floor leachate, ALB = Schwäbische Alb, HAI = Hainich-Dün, SCH = Schorfheide-Chorin; red lines indicate zero level within difference spectra.