

Effects of CO₂-driven ocean acidification (OA) on early life stages of Marine Medaka (*Oryziasmelastigma*)

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1 **Abstract**

2 The potential effects of high CO₂ and associated ocean acidification (OA) in marine
3 fishes and other non-calcified organisms are less well understood. In this study, we
4 investigated the responses of early life stages (ELS) of marine medaka (*Oryzias*
5 *melastigma*) exposed to a series of experimental manipulation of CO₂ levels. Results
6 showed that CO₂-driven seawater acidification (pH 7.6 and pH 7.2) had no detectable
7 effect on hatching time, hatching rate, or heart rate of embryos. However, the
8 deformity rate of larvae in the pH 7.2 treatment was significantly higher than that in
9 the control treatment. There is no significant difference between the left and right
10 otolith areas in each treatment. However, the average otolith area of larvae in the pH
11 7.6 treatment was significantly smaller than that in the control. Such alterations in the
12 developmental abnormalities and otolith size of marine medaka larvae due to
13 elevated-CO₂ levels suggests that this species will be increasingly challenged by
14 future OA. Further studies of the impacts of OA on marine fish to assess whether or
15 not the environmental influence in one generation can affect the later life history and
16 the phenotype of subsequent generations are needed.

17 **Keywords:** climate change, non-calcified organisms, marine fish, early life stage,
18 otolith, calcification

19 1. Introduction

20 In the 20th century, the atmospheric carbon dioxide (CO₂) concentration
21 continued to increase as a result of fossil fuel combustion and other human activities.
22 It was in turn taken up by the ocean gradually through air-sea exchange. Oceanic CO₂
23 can hydrolyze to increase the concentration of hydrogen ions (H⁺), which leads to the
24 reduction of pH in the ocean by 0.1 units (Orr et al., 2005). Based on the amount of
25 global CO₂ emission at present, the pH of ocean is likely to drop by 0.3 ~ 0.4 units by
26 the end of the 21st century and by 0.7 units after 300 years. The current and predicted
27 levels of CO₂ and acidity of seawater of nearshore, estuarine, and higher-latitude
28 habitats are expected to be even greater and substantially more variable than those of
29 the open ocean (Gruber et al., 2012; Zhai et al., 2014). One alarming consequence is
30 the rapid change in seawater chemistry and decrease of ocean pH, which could have
31 great impacts on marine ecosystems, and pose a threat to marine life (Frommel et al.,
32 2013a; Kerr, 2010). Elevated CO₂ concentrations can disturb the acid-base regulation,
33 blood circulation, respiration, and the nervous system of marine organisms, leading to
34 long-term effects such as reduced growth rates and reproduction (Frommel et al.,
35 2013a). Other direct response to ocean acidification (OA) have been found in the
36 alteration of behavior (Dixson et al., 2010; Munday et al., 2009a), development
37 (Frommel et al., 2012b), RNA/DNA ratio (Franke and Clemmesen, 2011), and otoliths
38 (Checkley et al., 2009; Maneja et al., 2013; Munday et al., 2011b) of marine fish
39 larvae. However, the emerging picture remains intriguingly complex. While the
40 majority of responses to high CO₂ appear to be negative (Branch et al., 2013) with
41 highest sensitivities observed during the early life stages (ELS) and in the calcifying
42 invertebrates such as corals, bivalves, pteropods, and echinoderms, there are
43 substantial evidences for non-linear, neutral, or even positive reaction to increasing
44 CO₂ conditions (Hurst et al., 2013; Munday et al., 2011b; Murray et al., 2014).
45 Moreover, marine fish exemplifies this complexity. Decades of empirical data suggest
46 that juvenile and adult fish possess sufficient acid-base and osmoregulatory
47 capabilities for the toleration of very high metabolic and ambient CO₂ levels (> 2000

48 μatm) (Murray et al., 2014). Although fish possesses the ability of acid-base balance
49 regulation, its physiological function will certainly decline under such regulation for a
50 long time from the perspective of energetics, especially in its most fragile and
51 sensitive early life stage during its life history. In addition, in ELS of multiple taxa
52 including fish, elevated CO_2 was shown to affect calcification of shells and skeletons
53 due to a drop in the carbonate availability (Riebesell, et al., 2010). Munday et al.
54 (2011a) observed no effect on spiny damselfish otolith calcification at 850 μatm ,
55 while Munday et al. (2011b) and Checkley et al. (2009) highlighted an otolith
56 hypercalcification in white seabass (*Atractoscion nobilis*) larvae exposed at 993 μatm
57 and 2558 $\mu\text{atm } p\text{CO}_2$ and in clownfish (*Pomacentridae*) larvae at 1721 $\mu\text{atm } p\text{CO}_2$,
58 respectively. In case of calcification modulation, otolith morphology can be affected,
59 which may have negative repercussions on the behavior and acoustic function of fish
60 and decrease their survival probabilities (Bignami et al., 2013; Réveillac et al., 2015).

61 Marine medaka, *Oryzias melastigma* or *Oryzias javanicus*, is one of the 14 species
62 belonging to the genus *Oryzias*, which distribute in estuarine waters from East to
63 Southeast Asia (Koyama et al., 2008). It has been proposed as a model species in
64 marine environmental risk assessments (Mu et al., 2014). However, few studies have
65 addressed OA effects on the ELS of marine medaka so far. The objective of this study
66 was to examine how CO_2 -driven OA affected the embryos and newly hatched larvae
67 of marine medaka after 21 d exposure through investigating the embryonic, larval,
68 and otolith development.

69 **2. Materials and methods**

70 **2.1 Fish rearing**

71 Marine medaka, *O. melastigma*, were provided by the Key Laboratory of Coastal
72 Ecological Environment of State Oceanic Administration. Fish were maintained in
73 aquatic habitats system (Aquatic Habitats, USA) with a salinity of 30 ± 2 , temperature
74 of 26 ± 1 °C, and a photoperiod of 14 h:10 h (light:dark). All fishes were fed with
75 nauplii of *Artemia* three times a day and synthetic food (New life spectrum thera-A
76 formula, Made in the Newlife International, Inc, USA.) twice a day. One-tenth of the

77 total amount of water in the system was automatically renewed daily. To ensure
78 developmental synchronization of embryos during experiment, all eggs were collected
79 within 3 ~ 5 h after initiation of spawning, and fertilized and viable ones were
80 selected under dissecting microscope.

81 The experimental seawater (salinity of 30.7 ± 0.1) was prepared by diluting sea
82 salts (Instant Ocean, Aquarium Systems, USA) with deionized water. The standard
83 NBS pH was 8.2 ± 0.004 .

84 **2.2 Seawater manipulation and experimental design**

85 The design of seawater pH control system was based on Riebesell et al.(Riebesell,
86 et al., 2010) with some modifications. Briefly, partial pressure of CO_2 ($p\text{CO}_2$) was
87 adjusted by pH modulator (aquastar pH Modul II, IKS) with standard deviation of
88 ± 0.01 . Three pH gradients, 8.2, 7.6 and 7.2, were set according to the predicted levels
89 upon CO_2 emission at present, after 100 and 300 years (Orr et al., 2005), respectively.
90 The pH control system **consists of** three parts, namely monitor, controller and aeration
91 (Fig. 1). The pH meter in water **monitored** **the** real-time pH changes during
92 experiment. The controller associated with pH meter **was also connected with**
93 electromagnetic valve, which **opened or closed** the electromagnetic valve based on the
94 feedback of pH meter. The intake of electromagnetic valve **connected** to a cylinder
95 equipped with a high concentration of CO_2 (0.1 % CO_2 : 99.9 % air, $p\text{CO}_2$ of 1000
96 ppm), and its outtake **connected** to silicone tube, drying tube, check valve and refiner
97 which insert into seawater. The refiner was placed in the middle of the aquarium (10 L)
98 bottom to make the gas bubbled into water quickly and homogeneously. When the pH
99 in seawater was higher than the set value, the electromagnetic valve **opened**
100 automatically to pipe **concentrated** CO_2 into the water until the pH **dropped** to the set
101 value, and then the valve closed. During the exposure experiment, parameters
102 including pH, inorganic carbon (DIC), temperature, salinity, total alkalinity (TA) and
103 dissolved oxygen (DO) were continuously monitored and analyzed to ensure the
104 stability of pH control system.

105 **For each pH treatment, 90 fertilized eggs were randomly assigned to three tanks**
106 **(three replicates) with 30 eggs per replicate. These tanks were monitored daily for**

107 dead embryos, hatched larvae, and hatching time. Subsets of hatched larvae per tank
108 were then transferred to the alternative aquariums with the same exposure conditions
109 to start the larval exposure sub-experiment (3 CO₂ levels × 3 replicates). Larvae were
110 monitored daily and dead ones were removed until the termination of experiment at
111 21 days (approximately one-week post-hatch). By the end of experiments, the survival
112 larvae were anesthetized and photographed under microscope (Leica DMI4000B) for
113 deformity analyses, and the otoliths were then removed and dry-stored in well-plates.

114 **[Figure.1]**

115 **2.3 Determination of Water Quality Parameters**

116 The determination of pH, TA and DIC referred to the methods of Dickson et al.
117 (2007). In brief, samples were collected into vials without obvious bubbles by an
118 overflow manner, and then fixed with 0.1 % saturated HgCl₂ solution. The pH was
119 detected using combined electrode (Orion 8102 BN Ross) and high-precision pH
120 meter (Thermo Orion 3-Star, USA) in 25°C water bath within 2 h after sampling. The
121 deviation was less than 0.01. TA and DIC were detected by TA analyzer (Apollo
122 AS-ALK2, USA) and DIC analyzer (Apollo AS-C3, USA) with an accuracy of more
123 than ± 2 μmol/kg, respectively. Salinity, temperature and DO of seawater were
124 detected by YSI-85 water quality monitor (YSI Inc, USA), and the accuracy of each
125 parameter was more than ± 0.1, ± 0.1 °C and ± 2% air saturation, respectively.
126 Aragonite saturation (Ω_{Ar}) was calculated based on temperature, salinity and measured
127 TA and DIC through CO₂-SYS carbonate system software (Pelletier et al., 2011).
128 Other parameters adoption including dissociation constants of carbonic acid and
129 sulfuric acid, saturated solubility product of CaCO₃ were consistent with those
130 internationally applied (Millero et al., 2006).

131 **2.4 Developmental toxicity**

132 The numbers of embryos surviving to hatching were counted based on daily
133 inspection of the embryos in each treatment. Hatching rate data were summed and
134 converted to proportions of survival numbers out of 30 eggs in per replicate. After 8

135 days post fertilization, and 3 days before expected hatching, embryos were inspected
136 at least twice a day and hatching numbers were recorded. Heart rates were estimated
137 by counting the number of heart beats over a 30 s period ($n = 10$) at day 8. The time
138 when ≥ 50 % of the embryos had hatched was recorded as the "hatching time"
139 (Forsgren et al., 2013). As observations of spawning and hatching were made at
140 somewhat irregular intervals over the course of the study, spawning and hatching
141 times were analyzed. The embryonic hatching time was calculated as the time elapsed
142 between spawning and hatching.

143 On day 21, thirty larvae (10 larvae per replicate) from each CO₂ treatment were
144 randomly selected and photographed for deformity analyses. The deformity rate was
145 calculated based on the proportions of abnormal larvae numbers out of 10 eggs in per
146 replicate. Survival rate of larvae was the obtained proportions through dividing the
147 larvae numbers remaining at termination of the experiment by the larvae numbers
148 initially newly hatched in per replicate.

149 **2.5 Otolith measurement**

150 The measurement of marine medaka otolith was based on the method of Franke
151 and Clemmesen (2011). Briefly, the left and right otoliths were removed from 16 fish
152 larvae randomly selected from each CO₂ treatment. Each otolith was observed and
153 photographed under microscope (Leica DMI4000B). Digital pictures of otolith were
154 taken at 1000 \times magnification using the microscope equipped with Leica DFC420C
155 Digital Camera. Otolith area (μm^2) was calculated through Image-Pro Plus 5.0
156 software after calibration and gray-scale processing of photos.

157 **2.6 Statistical Analyses**

158 Data analyses were performed using SPSS ver.16.0 (Chicago, IL) software. All
159 data were tested for normal distribution using the Kolmogorov-Smirnov test.
160 Non-normally distributed data were log-transformed. The difference between
161 measured and nominal pH was analyzed by T-test. For heart rate, hatching rate,
162 hatching time, and deformity rate, one-way analysis of variance (ANOVA) followed
163 by Bonferroni post hoc tests were applied to test the differences between and among
164 groups. An independent sample test was used to compare the difference of otolith

165 areas between left and right sides in each treatment. If there was a significant
166 difference, one-way ANOVA was used to further compare the difference between
167 treatments for left and right sides, respectively. If not, one-way ANOVA was
168 performed after data combining of left and right sides. Results were expressed as
169 means \pm standard deviation (SD).

170 **3. Results**

171 **3.1 Seawater chemical parameters**

172 Measured pH in three treatments and different chemical parameters in seawater
173 were shown in Figure 2 and Table 1, respectively. During the 21 d of exposure,
174 measured pH in pH 8.2, 7.6 and 7.2 groups were 8.22 ± 0.004 , 7.63 ± 0.007 and 7.22
175 ± 0.002 , respectively. The fluctuation was less than 0.05 (Fig.2), indicating the
176 stability of pH control system.

177 **[Figure 2.]**

178 **[Table 1.]**

179 **3.2 Embryonic development**

180 Three replicates produced a total of 90 eggs in each CO₂ treatment. The hatching
181 times were extended with decreasing pH level, but there was no significant difference
182 among the three pH treatments ($F_{2,6} = 5.8$, $p = 0.066$) (Fig. 3A). On average, 83
183 percent of eggs in three pH treatments survived to hatch, and the hatching rate of eggs
184 was not significantly different among the three pH treatments ($F_{2,6} = 1.1$, $p = 0.4$) (Fig.
185 3B). For the heart rates of embryos, pH 7.6 and 7.2 groups were not significantly
186 different from those in the control group ($F_{2,28} = 1.7$, $p = 0.7$) (Fig. 3C).

187 **[Figure 3.]**

188 **3.3 Larval development**

189 Three replicates produced a total of 66 ~75 newly hatched larvae in each CO₂
190 treatment level. By the end of experiment, larvae survival rate was highly variable but
191 did not differ significantly between the control and acidified water groups ($F_{2,6} = 0.3$,
192 $p = 0.7$) (Fig. 4B).

193 However, the two lower pH treatments (pH 7.6 and pH 7.2) can both cause spinal
194 deformities, craniofacial deformities, stretched heart and pericardial edema of marine
195 medaka larvae (Fig.5). Furthermore, in pH 7.2 treatment, the deformity rate was
196 significantly higher than that of control group ($F_{1,4} = 32, p = 0.005$) (Fig. 4A).

197 [Figure 4.]

198 [Figure 5.]

199 3.4 Otolith development of larvae

200 The effects of different pH treatments on otolith size of marine medaka larvae
201 were shown in Figure 6. There was no statistically significant difference between the
202 areas of left and right sides in each pH treatment (pH 8.2: $F_{1,59} = 0.092, p = 0.76$; pH
203 7.6: $F_{1,67} = 0.045, p = 0.83$; pH 7.2: $F_{1,68} = 0.005, p = 0.95$, respectively) (Fig. 6A). In
204 pH 7.6 treatment, the average areas of left and right sides were significantly smaller
205 than those of the control treatment ($F_{1,128} = 8.8, p = 0.013$) (Fig. 6B).

206 [Figure 6.]

207 4. Discussions

208 Assessment of species sensitivity or tolerance to CO₂-driven acidification in
209 marine environment is critical to evaluate the impacts of OA on marine biodiversity
210 and ecosystem function (Fabry et al., 2008; Melzner et al., 2009). A number of studies
211 found that CO₂-driven acidification had obvious influences on ELS of many marine
212 invertebrates, especially calcified organisms including coral (Doropoulos et al., 2012;
213 Fabricius et al., 2011), *coccolithophores* (Berry et al., 2002), and mollusk (Kroeker et
214 al., 2013; Thomsen et al., 2013; Waldbusser et al., 2011). OA was predicted to
215 potentially affect individual behavior such as development, growth, survival and
216 swimming particularly during the early life stage of marine organisms (Munday et al.,
217 2008). In our experiments, the duration of embryonic stage, egg survival and
218 embryonic heart rate of marine medaka were unaffected by acidification water with
219 pH 7.6 and pH 7.2. There was a slight increase in the embryonic duration of the eggs,

220 but the size effect was not different among the three pH treatments. Overall, these
221 results suggest that the egg stage of marine medaka is relatively tolerant to elevated
222 CO₂ and low pH level, which were consistent with the results reported by other
223 studies on a diverse set of marine fishes. For instance, Munday et al., (2009) found the
224 survival to hatch of orange clownfish (*Amphiprion percula*) from the Great Barrier
225 Reef, Australia, to be nonresponsive to pCO₂ levels to 1020 ppm (pH 7.8). Similarly,
226 Franke and Clemmesen (2012) found no significant effect of elevated pCO₂ levels
227 from 460 to 4635 ppm (corresponding to pH 8.08 ~ pH 7.05) on survival to hatch of
228 Atlantic herring from the western Baltic Sea. In the study of Frommel et al. (2013),
229 the survival of embryos of Atlantic cod from the Bornholm Basin of the western
230 Baltic Sea was not altered at pCO₂ levels up to 4000 ppm (pH 7.2). Hurst et al. (2013)
231 also reported no effect on embryo survival of walleye pollock (*Theragra*
232 *chalcogramma*), common in the temperate eastern North Pacific, at pCO₂ levels up to
233 1933 ppm (pH 7.4). In other cases, however, a strong effect of CO₂ was observed
234 evident on the embryo survival of summer flounder (*Paralichthys dentatus*), an
235 ecologically and economically important flatfish of the inshore and nearshore waters
236 of the Mid-Atlantic Bight (Chambers et al., 2013). The relative survival of summer
237 flounder embryos was reduced to 48% when maintained at 1808 ppm pCO₂ (pH 7.5)
238 and to 16% when maintained at 4714 ppm pCO₂ (pH 7.1). Baumann et al. (2012) also
239 reported a 74% reduction in survival of embryos and young larvae of inland silverside,
240 *Menidia beryllina*, native to estuaries of the US Atlantic coast, when maintained at
241 1100 ppm pCO₂ compared to those held at 410 ppm pCO₂. All of these studies varied
242 in the number of parents used, the time lapse between egg fertilization and initiation
243 of CO₂ treatment, and how and when survival was scored. For example, the CO₂
244 treatments of inland silverside by Baumann et al. (2012) began at approximately 24 h
245 post-fertilization, and the survival was scored at approximately 1 week post-hatching.
246 The different approaches used in previous studies may preclude a fair cross-study
247 comparison (Chambers et al., 2014); however, the overall present of effect of elevated
248 CO₂ environments on embryo survival is in contrast to the findings here. Habitats
249 occupied of species, particularly in the their ELS, may play a role in their sensitivities.

250 It is counter to expectations and requires further attention that species in their ELS are
251 found in estuarine (marine medaka) and inner shelf (summer flounder) habitats, both
252 with relatively high ambient CO₂ levels, but exhibit different sensitivities to
253 experimentally elevated-CO₂ levels.

254 An unexpected result of our study was that elevated levels of CO₂ affected larval
255 development abnormalities, and the average deformity rate of marine medaka larvae
256 (approximately one-week post-hatch) increased significantly by 16 % as CO₂
257 increased from control level (pH 8.2) to high-CO₂ level (pH 7.2). Although
258 CO₂-induced acidification up to the high-CO₂ level (pH 7.2) had no noticeable effect
259 on larval survival by the end of experiments (21 d), the larval development
260 abnormalities may ultimately influence the later life consequences and therefore
261 further reduce the productivity of fish stock in future acidified oceans. Chambers et al.
262 (2014) found no reduction in survival with CO₂ for larvae during the first four weeks
263 of larval life (experiment ended at 28 d post-hatching (dph)), however, the sizes,
264 shapes, and developmental status of larvae showed initially longer and faster growing
265 when reared at pH 7.5 and pH 7.1 levels, and the tissue damage was evident in larvae
266 as early as 7 dph from both elevated-CO₂ levels. At present, it is unknown that how
267 increasing CO₂ levels affect development and survival in fish ELS. Even if fish
268 embryos and early larvae are capable of physiological adaptation to increased CO₂
269 somehow, this would incur further metabolic costs and thus reduce energy available
270 for tissue synthesis or post-hatch survival on diminished yolk reserves. As some fish
271 eggs, including those of *O. melastigma*, seem to be tolerant to low-pH conditions, the
272 high levels of CO₂ or associated changes in carbonate chemistry may be more
273 important to larval-fish development than hydrogen ion concentrations. (Baumann et
274 al., 2011; Ishimatsu et al., 2008).

275 The pH drop driven by CO₂ can change concentrations of bicarbonate and
276 non-bicarbonate ions during which elevated CO₂ affects saturation states of calcium
277 ions carbonate polymorphs (Munday et al., 2008). Otoliths are bony structures of fish
278 to sense orientation and acceleration and consist of aragonite-protein bilayers, which
279 document fish age and growth (Checkley et al., 2009). Its formation starts during

280 embryonic development, and any alteration of otolith size or shape is important for
281 physical performance and individual adaptability of fish. Therefore, any substantial
282 change in the size, shape, or symmetry of otoliths could have serious implications for
283 individual performance and survival (Munday et al., 2011a; Munday et al., 2008). In
284 this study, we found no significant difference existing between the left and right sides
285 of marine medaka larval otolith under the same pH level. However, otolith area of
286 larval fish exposed to the intermediate-CO₂ level (pH 7.6) was smaller than that of
287 control. Results suggested that there was no significant pCO₂ effect on otolith
288 symmetry of marine medaka, defined as the difference between the right and left sides.
289 However, the otolith area was significantly affected. The trend of reduction in otolith
290 area of marine medaka larvae exposed to elevated CO₂ environments found here has
291 not been reported in most previous studies focusing on other marine fishes. For
292 instance, Checkley et al. (2009) found that otolith area of white seabass (*Atractoscion*
293 *nobilis*) larvae increased by 7% ~ 9% and 10% ~ 14% after exposure to 993 ppm and
294 2558 ppm CO₂, respectively. Munday et al. (2011b) found that the size, shape, and
295 symmetry of otoliths in larval clownfish was unaffected by exposure to simulated
296 levels of OA (pH 7.8 and 1050 µatm CO₂); however, in a more extreme treatment (pH
297 7.6 and 1721 µatm CO₂) otolith area and maximum length were larger than those of
298 control otoliths. Maneja et al.(2013) found that elevated CO₂ had no significant effect
299 on the shape of the otoliths nor was there any trend in the fluctuating asymmetry,
300 while increased otolith growth was observed in 7 to 46 d post hatch cod larvae in two
301 pCO₂ treatments of 1800 µatm and 4200 µatm. In contrast, Munday et al. (2011a) did
302 not detect any effect of elevated CO₂ on otolith size of juvenile spiny damselfish,
303 *Acanthochromis polyacanthus*, which were reared for 3 weeks in treatments up to 841
304 µatm CO₂. Our results seemed to support the hypothesis that otoliths of larvae reared
305 in seawater with elevated CO₂ would grow more slowly than they do in seawater with
306 normal CO₂. The reduction of otolith area was likely associated with reduced CaCO₃
307 saturation which slowed down its formation. We do not know whether smaller otoliths
308 have a deleterious effect, although we do know that asymmetry between otoliths can
309 be harmful (Checkley et al., 2009).The difference between our results and other

310 studies may be related to: (1) different $p\text{CO}_2$ levels; (2) different life histories; or (3)
311 different exposure duration (Munday et al., 2011a). However, another interesting
312 result from the present study was that the otolith area of marine medaka larvae under
313 the extreme CO_2 level (pH 7.2) tended to increase instead of continuously reducing.
314 We should not ignore its own acid-base regulation ability that increased the available
315 amount of carbonate by compensation mechanism for otolith to intensify the
316 calcification process under such acidic condition (Checkley et al., 2009). Calcium
317 incorporation into the otolith was modulated by the seawater pH. This question the
318 stability of the elemental: Ca ratio under environmental hypercapnia. During the
319 biomineralization of the otolith, chemical elements such as metals and metalloids are
320 supposed to substitute for calcium (Réveillac et al., 2015). The changes of pH and
321 seawater chemistry caused by increased CO_2 can modify the speciation of metals and
322 their subsequent bioavailability to organisms (Millero et al., 2006). The physiological
323 response of fish to hypercapnia might in turn stimulate processes to compensate for
324 acidosis based on the key role of ion transporters. In present study, OA may interfere
325 with trace element uptake and body concentrations and therefore could affect otolith
326 growth and microchemical constituent. Further studies are thus needed to investigate
327 the possibility that OA impacts on the trace metals properties, molecular-binding
328 affinities and incorporation pathway into the otolith.

329 In conclusion, this study demonstrated that, under projected near-future $p\text{CO}_2$
330 levels, the ELS of marine medaka exhibited a dramatic increase of larval
331 developmental deformity and otolith calcification while their survival was not
332 affected. Importantly, the observed CO_2 -induced abnormal development of larvae
333 might have predictably negative consequences on the recruitment of fish population,
334 the effects of which on later life history and phenotype of subsequent generations
335 should be concerned. As the otolith is an essential tool used in reconstructing fish life
336 history in terms of age, somatic growth and attended habitats, further studies should
337 investigate the process of otolith biomineralization. Finally, we emphasize that there is
338 considerable variation among species in their sensitivities to elevated CO_2 and
339 reduced pH. Determining the traits that render some species more susceptible than

340 others will be helpful and valuable to predict the long-term and ecological effects of
341 OA.

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468 **Table list**

469 **Table 1. Summary of chemical parameters in control and acidic seawater ($n = 3$)**

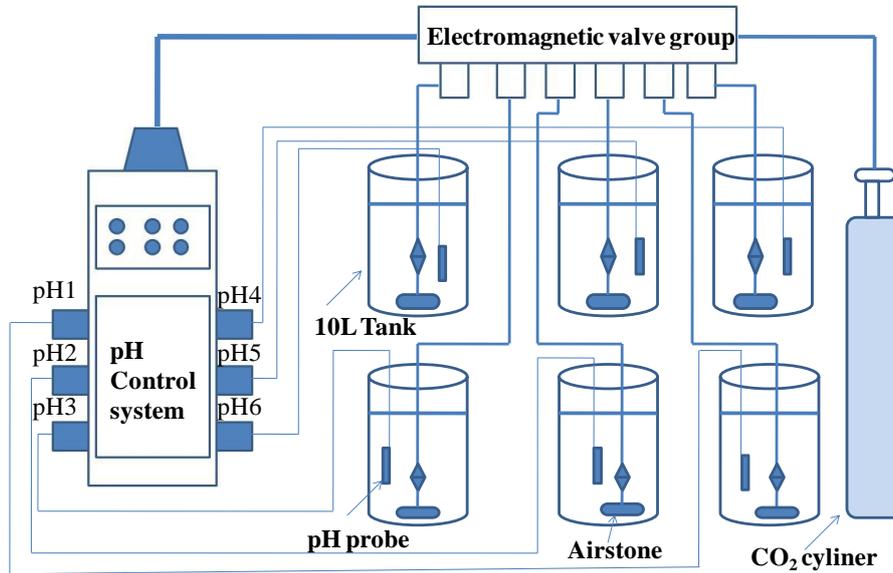
pH_{NBS}^*	DIC ($\mu\text{mol/kg}$)	$p\text{CO}_2$ (μatm)	CO_2 ($\mu\text{mol/kg}$)	HCO_3^- ($\mu\text{mol/kg}$)	CO_3^{2-} ($\mu\text{mol/kg}$)	Ω_{Ar}
8.22 ± 0.004	2645.1 ± 28.5	495.9 ± 2.2	14.4 ± 0.1	2380.3 ± 10.1	280.4 ± 3.9	4.5 ± 0.06
7.63 ± 0.007	3014.2 ± 74.3	2372.6 ± 52.3	68.7 ± 1.4	2861.0 ± 20.7	84.5 ± 0.3	1.4 ± 0.006
7.22 ± 0.002	3202.7 ± 18.5	6165.7 ± 56.4	178.4 ± 1.8	2988.8 ± 9.3	35.5 ± 0.6	0.6 ± 0.01

470 * pH_{NBS} : The fundamental definition of pH in terms of the hydrogen ion activity; NBS: National Bureau
 471 of Standard.

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473 **Figures and captions**

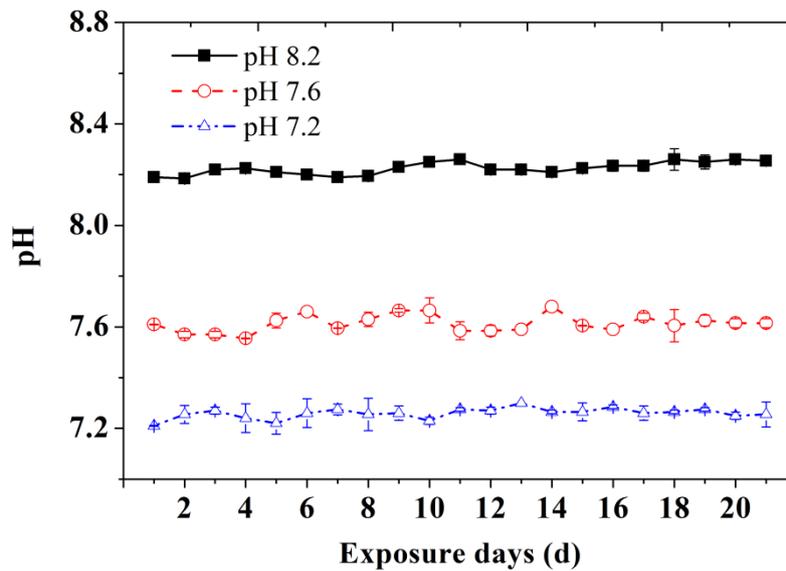
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476 **Figure 1. Schematic illustration of the pH control system applied in exposure experiment (For details refer**
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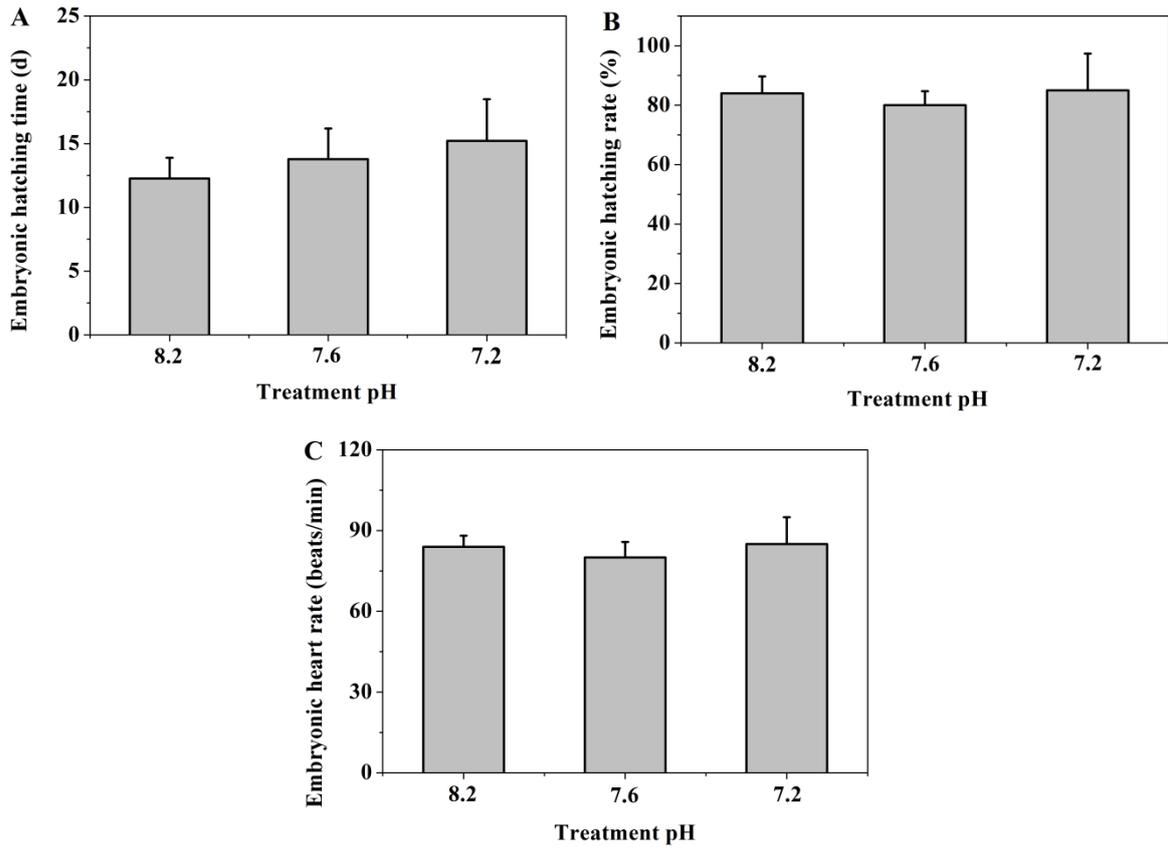


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480 **Figure 2. Measured mean pH_{NBS} of seawater in three pH treatments during 21 d of exposure ($n = 3$).**

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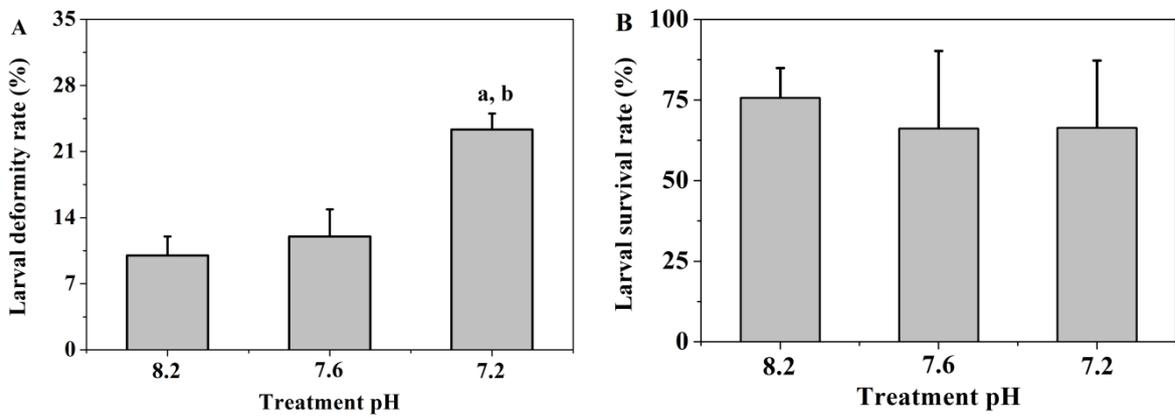
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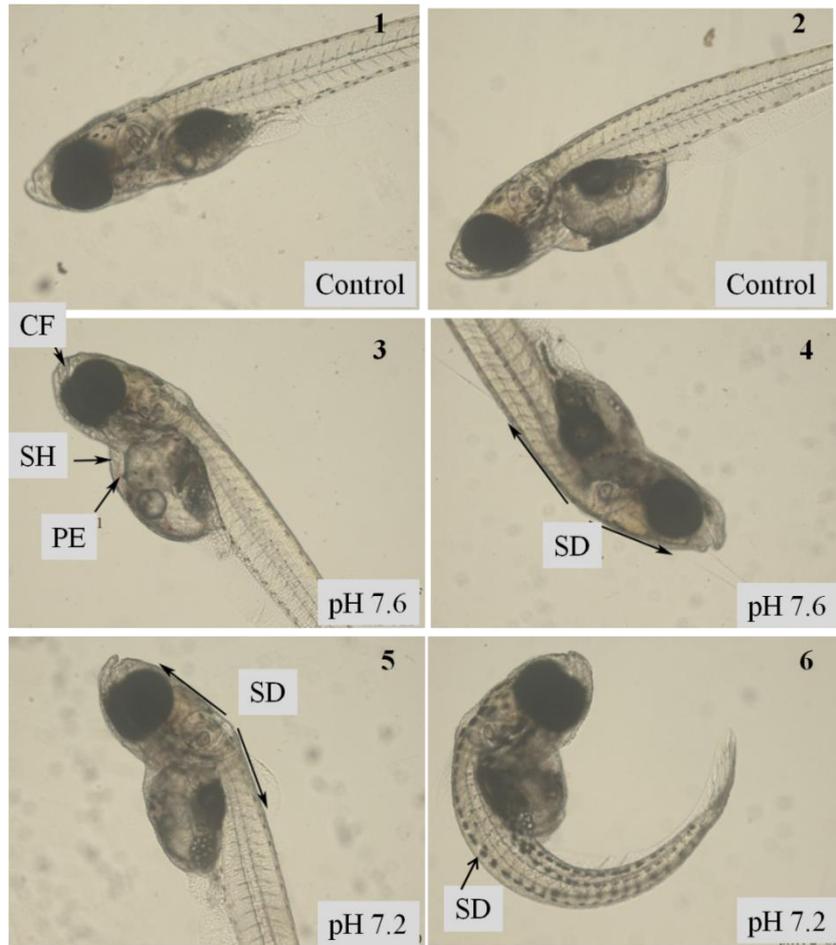
485 **Figure 3. The hatching time, hatching rate, and heart rate of marine medaka embryos exposed to three pH**
 486 **levels. (A) Hatching time; (B) Hatching rate; (C) Heart rate.**



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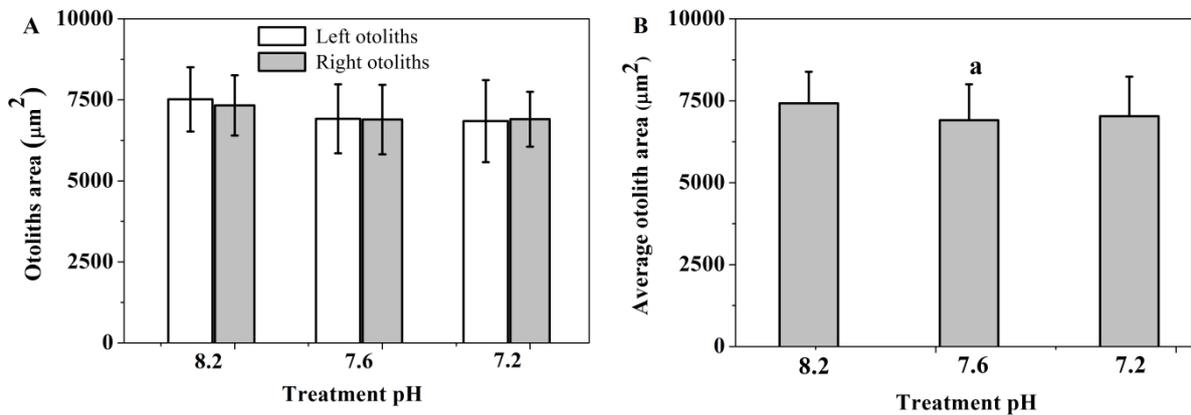
488 **Figure 4. The deformity and survival rates of larvae exposed to three pH levels. (A)Deformity rate; (B)**
 489 **Survival rate. The a indicates that the value in pH 7.2 or in pH 7.6 differs significantly from that in the**
 490 **control (pH 8.2), and b indicates that the value in pH 7.2 differs significantly from that in pH 7.6.**

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Figure 5. Morphological changes of medaka larvae exposed to three pH levels. 1~2: Control: Normal (pH 8.2); 3~4: pH 7.6 treatment; 5~6: pH 7.2 treatment. SD: Spinal deformities; CF: Craniofacial deformities; PE: Pericardial edema; SH: Stretched heart.



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Figure 6. The effects of different pH levels on the otolith area of marine medaka larvae after 21 d of exposure. The a indicates that the value in pH 7.2 or in pH 7.6 differs significantly from that in the control (pH 8.2).