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# **Reviews and Syntheses: Effects of** permafrost thaw on arctic aquatic ecosystems

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Discussion Paper

Discussion Paper

Close Full Screen / Esc

Printer-friendly Version

Interactive Discussion



**BGD** 

12, 10719-10815, 2015

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Title Page

**Abstract** 

Introduction

Conclusions

References

**Tables** 

**Figures** 











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**BGD** 

12, 10719-10815, 2015

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Title Page Abstract Introduction Conclusions References **Tables Figures** 

> Back Close

Full Screen / Esc

Printer-friendly Version



)

ussion Paper

Discussion Paper

Discussion Paper

Discussion Paper

**BGD** 

12, 10719–10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Abstract Introduction

Title Page

Conclusions

Tables Figures

I4 FI

Back Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



release of organic matter as greenhouse gases (CO<sub>2</sub> and CH<sub>4</sub>), its burial in sediments,

Interactive Discussion



and its loss downstream. The magnitude of thaw impacts on northern aquatic ecosystems is increasing, as is the prevalence of thaw-impacted lakes and streams. There is therefore an urgent need to address the key gaps in understanding in order to predict the full effects of permafrost thaw on aquatic ecosystems throughout the Arctic, and their consequential feedbacks to climate.

#### Introduction

Permafrost is perennially frozen ground that underlies about a quarter of the landmass of the Northern Hemisphere (Brown et al., 1998). It consists of various soil types ranging from frozen peat and frozen mineral soil to frozen Pleistocene deposits rich in massive ground ice. The distribution of permafrost is generally divided into four zones based on the percentage of the land that is underlain by permafrost: continuous (90-100%), discontinuous (50-90%), sporadic (10-50%) and isolated (< 10%) (Fig. 1). Terrestrial permafrost (we do not consider subsea permafrost in this review) hosts about  $1330-1580 \,\mathrm{Pg}$  (1  $\mathrm{Pg} = 10^{15} \,\mathrm{g}$ ) organic carbon (OC) (Schuur et al., 2015). Over the last few decades, permafrost ground temperatures have been slowly increasing (Romanovsky et al., 2010) as a result of increased surface warming in arctic regions (IPCC, 2013).

The Arctic is extremely rich in water. Lakes, reservoirs, rivers, and various types of wetlands, floodplains, bogs, fens and mires, on average occupy 16 % of the landscape underlain by permafrost (Fig. 1; Global Lakes and Wetlands Database; Lehner and Döll, 2004). Discontinuous, sporadic and isolated permafrost regions are relatively rich in surface water with 20, 23 and 18 % landscape coverage, respectively, compared to continuous permafrost with only 11 %. Water, in all its forms, connects all components of the landscape and plays a key role in the storage and transport pathways of sediments, organic matter, nutrients, and other constituents (Battin et al., 2009; Vonk and Gustafsson, 2013). The role of hydrology is therefore key in both the response and the effects of permafrost thaw, and it strongly influences the balance of carbon dioxide

#### **BGD**

12, 10719–10815, 2015

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Title Page **Abstract** Introduction

Conclusions References

**Tables Figures** 

Back Close

Full Screen / Esc

(CO<sub>2</sub>) and methane (CH<sub>4</sub>) emissions. At the same time, permafrost thaw will also create new aquatic ecosystems and modify existing aquatic ecosystems. In this review we provide an overview of the effects of permafrost thaw on aquatic ecosystems and their potential feedbacks to climate, with a consideration of all aquatic ecosystems located within the permafrost zones defined by Brown et al. (1998; Fig. 1).

When permafrost thaws, the soil organic matter and minerals within it become available for remobilization and introduction into aquatic systems. The type of thaw will largely determine the rate and effects of this remobilization. Here, we will distinguish between two types of thaw: (i) thaw of ice-rich permafrost, also called thermokarst, with a more abrupt or episodic character that tends to manifest as a *pulse disturbance*, and (ii) thaw of permafrost with (relatively) low ground-ice content, with a gradual but persistent and longer-term character that tends to manifest as a *press disturbance* (Grosse et al., 2011). We consider the effects of both thaw types on aquatic ecosystems, although in the different sections below, some thaw regimes have been given particular emphasis (e.g., thermokarst lakes) depending on the topic discussed.

Thaw of ice-rich permafrost leads to a range of landscape features (Fig. 2) that are collectively referred to as thermokarst, and that are typically divided into three primary groups based on where and how they form (Kokelj and Jorgenson, 2013): (i) hillslope processes, (ii) wetland processes, and (iii) thermokarst lake processes. *Hillslope processes* result in dramatic features such as retrogressive thaw slumps (Fig. 2b), active layer detachment slides and thermal erosional gullies (Fig. 2c), which together we refer to as thermo-erosional features (TEFs). The scale of these features is local but the transport of sediment, nutrients and organic matter into aquatic ecosystems can be large, and these features may form in a matter of hours and slowly grow for several years. *Wetland processes*, not specifically discussed in this review, include peatland collapse and the development of bogs and fens. *Thermokarst lake processes*, which include lake formation, expansion and drainage, are the most abundant (10–50 % of permafrost-impacted landscapes; Jorgenson et al., 2006; Kokelj and Jorgenson, 2013)

**BGD** 

12, 10719–10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

I∢













Printer-friendly Version



**BGD** 

12, 10719–10815, 2015

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Title Page

**Abstract** Introduction Conclusions References **Tables Figures** Back Close Full Screen / Esc

Printer-friendly Version

Interactive Discussion



and most easily recognizable form of thermokarst, and are particularly emphasized in this review.

Thaw of permafrost with lower ice content results in a more gradual top-down thawing process, and occurs through active layer deepening and talik formation (Schuur 5 et al., 2008). This type of thaw is gradual but can occur over entire landscapes (Shiklomanov et al., 2013; Åkerman and Johansson, 2008). Although its impact on aquatic ecosystems is harder to detect and requires long-term monitoring, such gradual thaw can cause striking changes to regional hydrology and chemistry (e.g., Walvoord et al., 2012; Striegl et al., 2005; Keller et al., 2010). Talik formation occurs under water bodies, but also when the active layer has deepened to such an extent that the soil does not refreeze completely in winter. Taliks create new hydrological hotspots in the landscape, and may increase groundwater flow and deeper flow paths, particularly in the case of open taliks that fully penetrate the permafrost profile (Walvoord et al., 2012), and alter fluxes of constituents (Walvoord and Striegl, 2007).

While ground ice content and topography are important modifying variables that determine how permafrost thaw manifests itself within landscapes, factors such as the local composition of soils and regional extent of permafrost will also play an important role in determining the effect of permafrost thaw on aquatic ecosystems (Fig. 3). For example, continued permafrost degradation, and in particular the transition from continuous to discontinuous permafrost, will considerably lengthen flow paths, often enable greater inputs of groundwater to freshwater systems, and affect the processing of water en route to aquatic systems (e.g., Striegl et al., 2005; Walvoord et al., 2012). Similarly, local soil conditions will affect how permafrost thaw changes soil-water interactions as water moves across landscapes, with the mobilization or exposure of organic vs. mineral soils being an important potential regulator of how thaw affects impacted aquatic systems.

In this review, we first describe the general impacts of thaw on aquatic ecosystems such as changes in physical, optical and chemical limnology and the release of materials from land to water (Sect. 2). We identify the various pathways of organic carbon and

contaminant degradation (bio- and photodegradation), and the resulting changes in microbial community structure and gas fluxes from thaw-impacted systems (Sect. 3). We then evaluate the broader consequences of permafrost thaw for aquatic ecosystems considering the release of old carbon into waters and the atmosphere, carbon burial, effects for ecosystem structure and functioning, and exports to oceans (Sect. 4). We end this review with a summary of our findings, an overview of potential climate feedbacks and recommendations for future research (Sect. 5).

#### 2 Impacts of thaw on aquatic ecosystems

### 2.1 Physical and optical limnology

The land area above latitude 45.5° N totals 41.3 million km² and contains around 200 000 lakes, with 73 % of them occurring in permafrost landscapes (Smith et al., 2007; Grosse et al., 2011, 2013). Thermokarst lakes are among the most abundant of these waterbodies, and can be found throughout the circumpolar north, from North America to Europe and Russia. They encompass a wide range of physical characteristics, which in turn contribute to the large variations in their biogeochemical properties. In this section, we focus on the effects of thermokarst on the physical and optical limnology of ponds and lakes (Fig. 4). We acknowledge, however, that thermokarst processes will also have significant effects on the physical properties of stream and river systems, for example via the delivery of colored or chromophoric dissolved organic matter (CDOM) and sediments to these environments (see Sect. 2.2).

Thermokarst lakes occur in cold, high-latitude environments that experience prolonged winters with sub-zero temperatures for 8 months or more of the year, and short summers with air temperatures that may rise well above 10 °C depending on location. They tend to have minimal hydrological connectivity given their frozen surroundings, which inhibit the infiltration and exchange of water. As a result, the hydrological balance of these closed basins is strongly influenced by snowmelt in spring, evaporation

**BGD** 

12, 10719–10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Abstract Introduction

Conclusions

Tables Figures

I**∢** ►I

Back Close

Full Screen / Esc

Printer-friendly Version



Discussion

Pape

Discussion Paper

**Tables** 

**Figures** 

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



and precipitation in summer, and water inflow from local permafrost soils as a result of thermokarst processes (Fig. 4; Dingman et al., 1980; Bowling et al., 2003; Smith et al., 2007). Notable exceptions are the thermokarst lakes located in floodplain deltas, where river waters may flood and connect the lakes each year, depending on their height above the main stem of the river (McKnight et al., 2008). Thermokarst lake disappearance by drainage into rivers is a widespread scenario in Siberia (e.g., Kravtsova and Bystrova, 2009; Kravtsova and Tarasenko, 2010).

Thermokarst lakes vary greatly in surface area, from ponds that are only a few meters across (Bouchard et al., 2011; Breton et al., 2009) to lakes that are several kilometers in their maximum dimension (Arp et al., 2011; Pelletier, 2005; Pokrovsky et al., 2011). In some thermokarst-impacted landscapes, ponds and lakes cover up to 30% of land surface area (Hinkel et al., 2005; Côté and Burn, 2002). Their depth is limited by local geomorphology and thaw depth of the permafrost, resulting in shallow ponds in many areas. For thermokarst lakes in northern Québec, in a region extending from continuous to discontinuous and sporadic permafrost, maximum lake depths range from 1 to 3.5 m (Breton et al., 2009; Bouchard et al., 2011; Crevecoeur et al., 2015; Laurion et al., 2010). Similarly, lake depths range from 0.5 to 1.5 m on a discontinuous permafrost tundra in western Siberia (Pokrovsky et al., 2013), from 0.4 to 2.6 m in an area of continuous permafrost in northern Alaska (Arp et al., 2011), and from 1 to 3.5 m in continuous permafrost of the Arctic Coastal Plain (Hinkel et al., 2012). Thermokarst lakes approaching 10 m depth can be found in interior Alaska (Sepulveda-Jáuregui et al., 2015) and lakes deeper than 10 m can be found on the Seward Peninsula, Alaska (Hopkins, 1949) and northeast Siberia (Walter Anthony and Anthony, 2013); thermokarst lakes as deep as 22 m exist on the Yukon Coastal Plain of north-western Canada (West and Plug, 2008). The size and abundance of thermokarst lakes are changing, but with pronounced differences between regions. In Russia, thermokarst lake area has been increasing in mid-latitudes but decreasing in southern Siberia (Kirpotin et al., 2008; Sharonov et al., 2012). However, high-arctic thermokarst lakes have also been expanding (Sannikov et al., 2012). Lake area changes may vary substantially

# **BGD**

12, 10719–10815, 2015

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Title Page Introduction **Abstract** 

Conclusions References

Back

Paper

Conclusions

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



(i.e., both increasing and decreasing) even within smaller regions, e.g. in Altai (Polishuk et al., 2013) or northwest Siberia (Bryksina et al., 2011). In summary, thermokarst lakes and ponds occur in wide varieties (Fig. 2) depending on, for example, their formation process, surface area, and depth; from here on we collectively refer to all of these 5 waterbodies as thermokarst lakes.

Ongoing thermokarst can result in variable inputs of dissolved organic carbon (DOC), inorganic solutes (including nutrients and major ions), and particulate organic and mineral materials into these lakes. Consequently, their limnological properties are highly influenced by their regional surroundings (Fig. 4; Prowse et al., 2006; Bowden et al., 2008; Watanabe et al., 2011; see Sect. 2.2 below). One of the expressions of that variability is the color of the water, which although typically brown, may sometimes be blue, green, black or even white. A study on water color and light attenuation in the thaw waters of Nunavik, Québec (Fig. 2h), showed that the lake surface color (water-leaving spectral radiance) was dependent upon the combined concentrations of CDOM and suspended non-algal particulate material, allowing certain biogeochemical properties such as DOC to be estimated from satellite remote sensing (Watanabe et al., 2011). Waters rich in CDOM (or in clays and silts) strongly attenuate solar radiation, and suppress primary production (see Sect. 4.3.1). In shallower and clearer waters, sufficient photosynthetically-active radiation may penetrate to the bottom of the lakes to allow the development of aquatic macrophytes; for example in lakes of the Mackenzie River Delta that are less affected by flooding with turbid river water (Squires and Lesack, 2003).

Depending on their CDOM and particle content, the surface waters of thermokarst lakes may strongly absorb solar radiation, and this gives rise to pronounced surface warming of the more colored lakes. In combination with the cooling of their bottom waters by the permafrost beneath, this means that thermokarst lakes can have pronounced vertical thermal and density gradients and may be strongly stratified in the summer despite their shallow depths (Fig. 4; Sepulveda-Jáuregui et al., 2015). In winter, ice-cover reduces transfer of energy with the atmosphere and eliminates potential

#### **BGD**

12, 10719–10815, 2015

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Title Page Introduction

**Abstract** 

References

**Tables** 

**Figures** 



Back

Back

Printer-friendly Version

Interactive Discussion



wind mixing, although some transfer of heat may result from the oxidation of organic matter in bacterial sediment processes (Mortimer and Mackereth, 1958). Strong thermal stratification during summer has been reported in many thermokarst lakes, for example even in waters less than 2 m deep in northern Québec (Laurion et al., 2010; Deshpande et al., 2015). Similarly, in a north-south transect in Alaska, the lakes on yedoma-like permafrost were typically stratified in summer despite their shallow depths, with less than 0.1 mg O<sub>2</sub> L<sup>-1</sup> at the bottom of the water column. On the other hand, lakes in non-yedoma permafrost or non-permafrost catchments tended to be less stratified and had well-oxygenated bottom waters (Sepulveda-Jáurequi et al., 2015). In other thermokarst lakes, little or no stratification has been observed during summer (Burn, 2003; Hinkel et al., 2012; Pokrovsky et al., 2013), possibly as a result of greater wind exposure, increased convective mixing or less near-surface heating.

Deep thermokarst lakes (> 2 m) show strong inverse stratification beneath their icecover during winter (e.g., Laurion et al., 2010; Deshpande et al., 2015; Sepulveda-Jáurequi et al., 2015), including those lakes that show little or no stratification in summer (e.g., Burn, 2003). Shallow thermokarst lakes may freeze all the way to the bottom in winter; for example on the Arctic Coastal Plain of Alaska, where the ice thickness reaches 1.5–2 m depths and lakes with a maximum depth of < 2 m tend to freeze completely (Arp et al., 2011).

Many thermokarst lakes are likely to be cold polymictic, undergoing stratification events that become established and then break down over diurnal, or several day, cycles. As a result, the increased use of high resolution, automated temperature loggers is likely to yield new insights into these short term stratification and mixing dynamics, even in those lakes currently considered to be well-mixed in summer. For example, in western Siberia, an air-temperature increase of 15 °C from an anomalous heat wave resulted in increased surface-water temperatures of 10°C, and the formation of a strong temperature gradient in the water column (Pokrovsky et al., 2013). Additionally, there may be strong diurnal variations in stratification and mixing. On the other hand, very shallow, small thermokarst lakes (e.g., collapsed ice-wedge trough ponds; Fig. 2) can

#### **BGD**

12, 10719–10815, 2015

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Title Page Introduction **Abstract** 

Conclusions References

> **Tables Figures**

Close

Full Screen / Esc

remain stratified for most of the summer, with only surface waters showing these diurnal dynamics (Negandhi et al., 2014; Bouchard et al., 2015).

Stratification within lakes is also influenced by gradients in salinity and gas concentrations that contribute to density differences down the water column (Kirillin et al., 2012). Such effects are also likely in thermokarst lakes, but have received little attention to date. Thermokarst processes result in inputs of eroded permafrost material into these water bodies, including POC and DOC, nutrients, and ions (Kokelj et al., 2005; Prowse et al., 2006; Bowden et al., 2008), which are further concentrated with the lake freeze-up during winter (Grasby et al., 2013; Deshpande et al., 2015), in turn affecting water column stability. The hydrodynamic effects of high dissolved CH<sub>4</sub> and CO<sub>2</sub> gradients under ice in stratified thermokarst lakes and of the gas bubble trains associated with ebullition from sediments (Walter et al., 2006) have received little attention.

The combination of high rates of bacterial metabolism, small lake volumes and prolonged ice cover means that permafrost thaw lakes can experience full water column anoxia for much of the year, in striking contrast to deeper, less productive lakes in the Arctic such as Toolik Lake, Alaska, and Char Lake, Canada (Deshpande et al., 2015). In northern Québec thaw lakes, mixing in spring occurs during an extremely short period of time (< 5 days) before the lakes restratify (Laurion et al., 2010). Mixing at that time may be insufficient to completely re-oxygenate the water column, and the more important period is fall, when prolonged mixing, likely aided by convective processes, results in the transfer of oxygen to the bottom of the lake (Deshpande et al., 2015). In a study near Mayo, Yukon Territory (Canada), a similar pattern was observed, of prolonged, substantial mixing in fall but only a short period of mixing in spring (Burn, 2003), which may favor the continuation of bottom water anoxia throughout summer. The fall mixing period is likely to be especially important for gas exchange with the atmosphere, and for stimulating aerobic processes such as bacterial respiration and methanotrophy throughout the water column.

Presently, there are major gaps in our understanding of the physical and hydrological dynamics of thermokarst lakes, including measurements of heat transfer from the

BGD

12, 10719–10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Abstract Introduction

Conclusions

Tables Figures

.

•

Back

Close

Full Screen / Esc

Printer-friendly Version



sediments, the penetration of solar radiation through winter ice cover, the potential for internal seiches in winter influenced by floating ice, and the nature of groundwater flows (Kirillin et al., 2012). Advective transfer of liquid water from shallower littoral zones to the pelagic bottom waters due to differential cooling may also play a role in material transfer within these waterbodies, as has been observed elsewhere (MacIntyre and Melack, 1995); no studies to date have addressed the three dimensional hydrodynamics of thaw waters.

### 2.2 Chemical limnology and the transfer of materials from land to water

Permafrost thaw and thermokarst processes can have a major effect on the chemistry of impacted aquatic ecosystems. Particularly in thermokarst lakes, these changes may occur as a result of the optical and physical processes described above. Where thermokarst affects stratification in thermokarst lakes, changes in redox and phosphorus remobilization from sediments could occur (e.g., Sondergaard et al., 2003). However these effects have been little studied in thaw-impacted systems. Changes in water column and sediment oxygenation will in turn change the prevalence of redox reactions, including the bacterially mediated processes described in Sects. 3.4 and 3.5, below. In addition to these classic "limnological" changes, permafrost thaw can be expected to have a fundamental effect on the transfer of sediments and chemical constituents from land to water. A recent review by Frey and McClelland (2009) provides an indepth discussion of how this change may play out for nutrients, DOC and major ions in stream and river systems. In this section, we update this previous work, and add a consideration of the specific effects of thermokarst and permafrost thaw via active layer deepening on lake water chemistry.

# 2.2.1 Press vs. pulse disturbances

The degree to which nutrients, organic matter, and sediments are released to aquatic systems is likely to depend on the type of permafrost thaw. For example, the *slow press* 

BGD

12, 10719–10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Abstract Introduction

Conclusions References

Ticicionicos

Tables Figures

Back Close

Full Screen / Esc

Printer-friendly Version



Back



of permafrost thaw through active layer thickening will likely favor the delivery of soluble materials (nutrients, base cations, DOC), although the mechanisms that deliver soluble materials to aquatic systems are complex, with research results often providing contradictory evidence for the effect of slow permafrost thaw on land-to-water constituent flux (Frey and Smith, 2005; Striegl et al., 2005; Frey and McClelland, 2009). In contrast, the fast pulse of permafrost thaw through thermo-erosional processes is likely to favor the delivery of particulate over soluble materials.

The spatial distribution and life cycle of these contrasting thaw processes will also govern the impact that they have on the delivery of biogeochemical constituents to aquatic ecosystems. For example, TEFs are discretely distributed across the landscape, following variations in topography and ground ice content. While these features can be numerous in impacted areas (Lacelle et al., 2015) they take up a relatively small percentage of the total landscape area (1.5% for work in Alaska; Krieger, 2012). Individual TEFs have lifecycles on the order of decades (Kokelj et al., 2013; Pearce et al., 2014), and – while they are active – may have intense local impacts on sediment and ionic fluxes to freshwater systems (see below). These local impacts, however, may become more muted when averaged over wider landscapes. In contrast, the slow press of active layer thickening occurs more universally across the landscape, but results in changes that are much more subtle, and may require long-term monitoring programs to detect. Slow press changes can operate over decades, altering aquatic ecosystems (e.g., Keller et al., 2010; Walvoord et al., 2012), and causing entire landscapes to slowly subside (Shiklomanov et al., 2013).

### Sediment delivery to aquatic ecosystems

Thermo-erosional features either directly adjacent to, or within the catchment of, aquatic systems can significantly increase suspended sediment concentrations, particularly in streams and rivers where turbulence causes materials to remain entrained. Thaw slumps and gullies directly adjacent to streams have been shown to cause orderof-magnitude increases in suspended sediment concentrations (Bowden et al., 2008; **BGD** 

12, 10719–10815, 2015

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Title Page Introduction **Abstract** 

Conclusions References

> **Tables Figures**

Close

Full Screen / Esc

Printer-friendly Version

Calhoun, 2012) that can continue to be seen for considerable distance downstream (Kokelj et al., 2013). For example, one small thermokarst gulley that formed in 2003 and intersected a small, headwater beaded-stream (the Toolik River) in a 0.9 km² Alaskan catchment delivered more sediment downslope to the river than is normally delivered in 18 years from a 132 km² adjacent reference catchment of the upper Kuparuk River (Bowden et al., 2008). Similarly, streams impacted by within-catchment active layer detachments have been shown to exhibit elevated sediment levels at their outflow when compared to non-impacted sites (Lamoureux and Lafreniere, 2014). Where slumping occurs directly adjacent to lakes, however, slump-associated sediments can settle out of suspension rapidly. Thus, while lakes impacted by permafrost slumping experience altered sedimentation rates (Deison et al., 2012), water column sediment loads are generally not impacted (Dugan et al., 2012; Kokelj et al., 2005). Similarly, increases in sediment loads are atypical in systems where thermokarst causes landscape collapse without significant exposure of soils, such as in the creation of thermokarst lakes in lowland regions.

#### 2.2.3 Organic matter delivery to aquatic ecosystems

Changes in sediment delivery to aquatic ecosystems will affect the flux of particulate organic carbon (POC) to these systems. In the active-layer detachment system described above, increases in suspended sediments were accompanied by measured increases in POC (Lamoureux and Lafreniere, 2014). In lakes, permafrost thaw can change the rate of accumulation of organic matter in sediments, and depending on the composition of eroding materials, either increase (Vonk et al., 2012a) or decrease (Deison et al., 2012) the concentration of sediment organic matter. Notably, the POC that travels to aquatic systems as a result of permafrost thaw may be only partially derived from permafrost carbon, because the action of thaw and landscape collapse will also expose and mobilize soils from the seasonally unfrozen active layer (e.g., Kokelj and Jorgenson, 2013). In Sect. 4.1.2 we review the effect of permafrost thaw on the mobilization of old organic carbon.

12, 10719-10815, 2015

**BGD** 

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Abstract Introduction

Conclusions References

Tables

Figures

I₫

►I

■ ...



Back

Close

Full Screen / Esc

Printer-friendly Version



Discussion

Back

Printer-friendly Version

Interactive Discussion



Where permafrost thaw enhances contact between water and organic soil horizons, increases in DOC concentrations are likely to occur. Direct slumping of old, yedoma carbon into streams causes striking increases in DOC in adjacent receiving waters in the Kolyma River watershed (Vonk et al., 2013), while slumping adjacent to streams in the Alaskan Arctic is also associated with significantly increased streamwater DOC at the site of impact (Abbott et al., 2014). During thaw of ice-rich permafrost, DOC stored in ice wedges and other ground ice (Fritz et al., 2015) is also released. Thermokarst lakes that form in organic-rich terrains can have significantly elevated concentrations of DOC as a result of direct contact between overlying water and recently submerged soils, and continued thermokarst expansion into new soils at the lake margin (e.g., Breton et al., 2009). In regions where slumping increases delivery of inorganic particles from land to water, however, aquatic DOC concentrations can decrease, as a result of the adsorption of organics onto sediment surfaces that settle after suspension (Kokeli et al., 2005).

In addition, thaw-enabled changes in flowpaths can also be expected to affect the transport of DOC to aquatic ecosystems. Although there is little direct evidence for the effect of water interactions with deeper soil layers as active layers deepen in organicrich regions, there are parallels to be drawn with more transitional (sub-arctic) systems, where permafrost peatland plateaus are associated with low annual export (2- $3 \,\mathrm{g\, C\, m^{-2}\, yr^{-1}}$ ) dominated by the snow melt period ( $\sim 70 \,\%$ ) and non-permafrost fens are characterized by much higher DOC export (7 g C m<sup>-2</sup> yr<sup>-1</sup>) due to more sustained annual hydrological connectivity (Olefeldt and Roulet, 2014). Conversely, where soils are characterized by shallow organic layers, growing season export of flow-weighted DOC has been shown to decrease significantly from 1978–1980 to 2001–2003 (Striegl et al., 2005), likely as a result of the combined effect of increased flow path (deeper active layer), residence time (Koch et al., 2013), and microbial mineralization of DOC in the unfrozen soil and groundwater zone. Permafrost thaw as a result of wildfire has been shown to increase hydrologic connectivity between burned hillslopes and catchment surface waters, such that burned soils can become a dominant source of water

#### **BGD**

12, 10719–10815, 2015

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Title Page Introduction Abstract Conclusions References

> **Tables Figures**

Full Screen / Esc

References

**Tables** 

**Figures** 

Back

Close

Printer-friendly Version

Interactive Discussion



and solutes to streams during summer, whereas unburned hillslopes provide longer term storage of water and solutes (Koch et al., 2014). In these regions, it has been suggested that organic matter sorption onto newly thawed mineral soils may also be important (Petrone et al., 2007). Over geographic gradients, changes in permafrost extent have regionally-variable effects on DOC flux from land to water, with decreasing permafrost extent (and presumably increasing contact with deeper soils and groundwater inflows) causing increasing DOC fluxes in organic rich regions, but decreases in DOC fluxes in regions with poorly developed organic horizons (Tank et al., 2012; Frey and Smith, 2005). Controlled leaching experiments of soils from the Alaskan and western Canadian Arctic have also found that regardless of temperature and leaching time, only small amounts of DOC could be released from permafrost-impacted soils and that mobilization of OC occurred largely in the POC phase (Guo et al., 2007).

### 2.2.4 Nutrient delivery to aquatic ecosystems

Similar to DOC, the effect of permafrost thaw on nutrient concentrations may also be regionally, or landscape specific. For thermo-erosional processes, direct slumping into Alaskan streams has been shown to increase dissolved inorganic and organic nutrient concentrations (Abbott et al., 2014; Bowden et al., 2008), while in the western Canadian Arctic, total dissolved N and P can be lower in shallow lakes directly impacted by permafrost slumping. This latter effect has been hypothesized to occur as a result of the adsorption of organic nitrogen and phosphorus onto settling mineral particles, or following rapid uptake and senescence, and then burial, of photosynthesizing cells (Thompson et al., 2012). In lakes that remain turbid following thermokarst disturbances, total phosphorus concentrations can be high, following phosphorus adsorption onto clays that are transported into aquatic systems (Breton et al., 2009). Similarly, slumping can also increase sediment nutrient concentrations (Mesquita et al., 2010), while shoreline expansion of thermokarst lakes in yedoma regions can enable nutrient-rich yedoma soils, and the nutrient-rich plants that these soils support, to enter lakes (Walter Anthony et al., 2014).

12, 10719–10815, 2015

**BGD** 

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Title Page

**Abstract** Introduction

Conclusions

Full Screen / Esc

Warming, coupled with landscape changes that decrease water contact with organic soils and increase water contact with inorganic soils has been shown in several arctic regions to lead to higher nitrate concentrations in adjacent streams, as a result of decreased NO<sub>3</sub> uptake and increased denitrification (Harms and Jones, 2012; Jones et al., 2005; Louiseize et al., 2014). On the Alaskan North Slope, nitrate export from the upper Kuparuk River increased over a period of several decades, via mechanisms that may be linked to warming and permafrost thaw (McClelland et al., 2007). Conversely, deeper flowpaths that decrease water contact with organic soils but increase contact with mineral soils are expected to decrease dissolved organic nitrogen exports (Harms and Jones, 2012; Walvoord and Striegl, 2007; Koch et al., 2013). This may also lead to increased phosphorus concentrations, because mineral weathering is the primary source of phosphorus in soil waters (e.g., Frey and McClelland, 2009).

#### 2.2.5 Delivery of major ions

Permafrost thaw is also expected to increase the concentration of weathering-derived ions in receiving waters, as slumping or deepening flowpaths increases the contact between water and deeper mineral soil layers (see review in Frey and McClelland, 2009). For example, in the Alaskan Arctic, increasing thaw depths have been associated with increasing surface water concentrations of calcium and bicarbonate (Keller et al., 2010), while near-surface permafrost has been found to be ion-rich in the western Canadian Arctic. In Siberia, and at multiple locations throughout the pan-Arctic, decreasing permafrost extent has also been associated with increasing streamwater concentrations of major weathering ions (Frey et al., 2007; Tank et al., 2012). This effect can be pronounced when thermokarst slumping occurs directly adjacent to aquatic systems: both streams (Malone et al., 2013) and lakes (Kokelj et al., 2005) can exhibit strikingly elevated ionic concentrations when directly impacted by permafrost slumping.

BGD

12, 10719–10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Abstract Introduction

Conclusions References

Tables Figures

Title Page

■ Back Close

Full Screen / Esc

Printer-friendly Version



Overall, the manner in which permafrost thaw affects surface water chemistry as a result of changing land-to-water fluxes will be dependent on the constituent and the landscape. While evidence suggests that constituents such as suspended sediments and weathering ions will experience neutral to increasing effects in response to permafrost thaw, the effect on constituents such as DOC and nutrients is likely to be more variable. As a result, we must consider local conditions, including soil composition (mineral vs. organic-rich), the nature of thaw (e.g., thermokarst processes on hillslopes vs. within lowlands, vs. active layer deepening), and the current extent of permafrost (continuous vs. discontinuous) to best understand the effects of thaw on land-water fluxes of chemical constituents. It seems clear that different types of thaw will differ in their effect on land-to-water constituent flux. TEFs seem likely to act as a population of features. only a small portion of which will be active at any one time within the landscape. Over the landscape as a whole, this will decrease their area-averaged rate of loading (mass per unit area) when compared to their effect at the site of impact. TEFs will also have a strong effect on particulate flux to aquatic systems. In contrast, long-term active layer thickening will likely support the delivery of soluble materials at low area-specific rates

Particularly for streams, hydrological connectivity between sites of thaw and the stream system is also an important consideration, because lateral inputs are typically more influenced by conditions in the riparian and deeper subsurface zones than by conditions at ridge tops and the near surface (Rastetter et al., 2004; Stieglitz et al., 2003). Thus, in areas where permafrost is thawing rapidly, the proximity of thaw features to aquatic systems, and the hydrological conditions of local stream and subsurface zones are likely to have a strong influence on the biogeochemical imprint of lateral inputs to streams. While TEFs have the capacity to move large quantities of soils and nutrients downslope, hydrological connectivity must be present to enable these constituents to reach the stream for biogeochemical impact to occur.

over a long time period, from a large area of the landscape.

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Paper

Discussion Paper

Discussion Paper

**Discussion Paper** 

12, 10719-10815, 2015

**BGD** 

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Conclusions References

Tables Figures

• •

Abstract

P1

Back

Close

Full Screen / Esc

Printer-friendly Version



Contaminants reach remote polar regions following long-range atmospheric transport and deposition, as well as through increased local waste production from marine transport and industrial or mining activities. The frozen soils of permafrost have historically been considered a barrier to the movement of contaminants and many waste and dump sites use containment strategies, which rely on the low mobility of contaminants in permafrost soils (Grannas et al., 2013). However, the warming arctic climate may lead to an increased mobility of contaminants, either stored in soils at waste sites or historically-accumulated in permafrost, into arctic surface waters (Armitage and Wania, 2013; Chételat et al., 2014).

Thawing permafrost may have major consequences for contaminant transport and transformations in the Arctic due to: (a) physical changes in the hydrological cycle leading to the remobilization of contaminants from contaminated soils or sediments, (b) chemical changes due to the release of nutrients and organic carbon from previously frozen soils and (c) biological changes via the microbial transformation of contaminants. Most studies on the interactions between contaminants and permafrost soils have concentrated on fuel products, persistent organic pollutants (POPs) and metals such as mercury (Hg) and lead (Pb).

Permafrost thaw may cause increased mobility of contaminants from catchment soils to surface waters due to accelerated soil/peat erosion, altered hydrological flow (increasing hydrological connectivity) and increased runoff leading to exposure of soluble contaminants (Fortier et al., 2007; Klaminder et al., 2008). Increased lateral hydraulic conductivity may accelerate the downhill movement of contaminants through the large pores, lenses and veins created in the active layer by the thawing of ice-rich permafrost (Dyke, 2001). After thaw, permafrost is no longer an impermeable barrier to contaminants, allowing for infiltration into soils and aquatic systems (Grannas et al., 2013). The reduced surface area of thermokarst lakes in some areas due to the creation of

BGD

Paper

Discussion Paper

Discussion Paper

Discussion Paper

12, 10719-10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Abstract Introduction

Conclusions References

Tables Figures

**♦** Back Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



10737

Discussion Paper

Conclusions **Tables** 

**Figures** 



Back



Full Screen / Esc

Printer-friendly Version

Interactive Discussion



drainage channels may lead to increased contaminant concentrations in the remaining surface waters (Macdonald et al., 2005).

Traditionally, the distinction is made between (i) organic contaminants, and (ii) inorganic contaminants. Studies of organic contaminants (hydrocarbons or non-aqueous phase liquids) predict lateral movement in the active layer with limited vertical transport in areas of continuous permafrost (Carlson and Barnes, 2011; Dyke, 2001). Vertical migration is possible in some regions (McCarthy et al., 2004) yet permafrost acts as a low-permeability barrier in others (Curtosi et al., 2007). Organic contaminants may migrate downwards into frozen soils in areas of discontinuous permafrost due to more variable distribution (Carlson and Barnes, 2011). Few studies have yet directly focused on physical changes of permafrost on inorganic contaminant mobility. Studies using mass-balance calculations and paleoecological techniques have linked thermokarst erosion in peatlands and the release of Hg into lake surface waters (Klaminder et al., 2008; Rydberg et al., 2010). Stable isotope analysis also suggests that the weak recovery of Pb contamination in two sub-arctic lakes (despite dramatically reduced atmosphere inputs) may be linked to the subsidence of thawing permafrost soils (Klaminder et al., 2010). Permafrost degradation may affect the mobility of inorganic contaminants differently across different Arctic regions. One study found that lakes impacted by the development of retrogressive thaw slumps in siliciclastic soils had lower levels of Hg in surface sediments when compared to reference lakes (Deison et al., 2012). In this instance, thaw slumping may have led to a dilution of organic material and associated mercury (Hg) due to high inorganic sedimentation rates. On the other hand, a recent study on small thermokarst lakes located in areas dominated by slumping organic soils (i.e. peat) showed elevated concentrations of Hg and toxic methylmercury (MeHg) in these systems, strongly related to inputs of organic matter and nutrients into surface waters (MacMillan et al., 2015).

Permafrost has historically been considered an impermeable barrier to the movement of contaminants due to their low infiltration and mobility in frozen soils (Grannas et al., 2013). However, ongoing and future climate warming will likely disrupt the se-

12, 10719–10815, 2015

**BGD** 

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Title Page Introduction **Abstract** 

References



questration of contaminants in permafrost soils and lead to enhanced leaching and therefore the mobility of organic and inorganic contaminants to nearby aquatic systems across all permafrost zones. This, however, will not necessarily result in increased contaminant concentrations due to a dilution effect by other materials transported with the contaminant.

# Pathways of degradation

The carbon, nutrients, and contaminants that are delivered to aquatic ecosystems as a result of permafrost thaw will have a critical effect on the functioning of these systems. Changes in biological function (see Sect. 4) will determine the relative balance between processing within freshwater systems, vs. loss via potential outflow pathways and the eventual delivery of these constituents to sediments (see Sect. 4.2) or "downstream" to the Arctic Ocean (see Sect. 4.5). In the case of carbon, differences in quality will affect whether processing and uptake results in the release of greenhouse gases (GHG), or C incorporation into microbially-based food webs. At the same time, changes in stratification, redox, solubility and oxygen availability (Sects. 2.1, 2.2, and 3.5.1) will affect the balance between CO<sub>2</sub> and CH<sub>4</sub> release. In this section, we review the dominant pathways of carbon and contaminant degradation in permafrost-thaw impacted systems. We focus specifically on bio- and photo-degradation pathways, and the effect of permafrost thaw on GHG emissions from thaw-impacted systems. The manner in which permafrost thaw affects nutrient uptake within aquatic systems is addressed in Sect. 4.3.

# Biodegradation of organic carbon

In the Arctic, where transfers of organic C from soils to aquatic ecosystems can be especially strong (Kling et al., 1991), C fluxes from surface waters to the atmosphere, and from land to ocean, may represent up to 40% of the net land-atmosphere C

**BGD** 

12, 10719–10815, 2015

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Title Page

**Abstract** Introduction Conclusions

References

**Tables** 

**Figures** 

Back

Close

Full Screen / Esc

Printer-friendly Version



**Figures** 

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



exchange (maximum flux of ~ 0.16 Pg Cyr<sup>-1</sup> compared to a net terrestrial sink of  $0.4 \pm 0.4 \,\mathrm{Pg}\,\mathrm{C}\,\mathrm{yr}^{-1}$ ; McGuire et al., 2009). In many cases, these C fluxes from freshwaters to the atmosphere or to the coastal oceans are supported by the degradation of terrestrially-derived DOC (Spencer et al., 2008), although DIC derived from weath-<sub>5</sub> ering can also be an important CO<sub>2</sub> source (Striegl et al., 2012). Biological processing of DOC occurs prior to and upon hydrologic delivery to surface waters (Michaelson et al., 1998; Spencer et al., 2015). The biodegradability of DOC in arctic systems is dependent on several factors including DOC source and chemical character (Michaelson et al., 1998; Wickland et al., 2007, 2012; Balcarcyzk et al., 2009; Mann et al., 2012; Olefeldt et al., 2013; Abbott et al., 2014), nutrient availability (Holmes et al., 2008; Wickland et al., 2012; Mann et al., 2012; Abbott et al., 2014), water temperature (Wickland et al., 2012), and prior processing (Michaelson et al., 1998; Wickland et al., 2007; Spencer et al., 2015). There are strong seasonal patterns in DOC biodegradability in large arctic rivers, where the relative amount of biodegradable DOC (BDOC) is greatest in winter and spring, and generally declines through the summer and fall (Holmes et al., 2008; Wickland et al., 2012; Mann et al., 2012; Vonk et al., 2015), reflecting the influences of seasonal thaw depth on DOC sources and hydrologic connectivity. Soil BDOC does not show a strong seasonality (Wickland et al., 2007; Vonk et al., 2015), supporting the notion that changes in DOC residence time and processing in soils prior to delivery to surface waters is a primary control on aquatic BDOC (Striegl et al., 2005; Vonk et al., 2015).

Permafrost presence and extent has direct and indirect influences on biodegradable DOC in aquatic ecosystems through its controls on potential sources and on hydrologic pathways and rate of delivery. A synthesis study of BDOC in arctic soils and surface waters finds higher BDOC in soils and aquatic systems with increasing permafrost extent (Vonk et al., 2015). In the absence of direct slumping, sources of DOC in permafrost-impacted areas are restricted to surface litter and modern active layer soils, which have relatively high C content and are presumably less degraded, and thus are strong potential sources of BDOC. Deeper soils having generally lower C con12, 10719-10815, 2015

**BGD** 

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Title Page

Introduction **Abstract** 

Conclusions References

**Tables** 



Back

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



tent are more accessible in discontinuous permafrost areas, and hydrologic flowpaths with longer residence times allow for greater opportunity of DOC processing during transport (Walvoord and Striegl, 2007) generally resulting in lower potential BDOC delivery to aquatic systems (Vonk et al., 2015). However, there are observations of increasing delivery of biodegradable DOC to aquatic ecosystems in areas of decreasing permafrost extent (Kawahigashi et al., 2004; Balcarcyzk et al., 2009), with one explanation being the preferential sorption of more recalcitrant hydrophobic DOC constituents to increasingly-exposed/accessible mineral soils (Kawahigashi et al., 2004). Therefore broad generalizations of permafrost control on BDOC are still difficult to make with certainty.

Permafrost thaw and thermokarst formation can potentially impact biodegradable DOC in aquatic ecosystems by altering sources, rates and pathways of hydrologic delivery. Newly thawed soils are potential DOC sources that can exceed DOC released from seasonally thawed active layer soils (Roehm et al., 2009; Waldrop et al., 2010). Studies of distinct DOC sources suggest that non-permafrost-derived soil pore water DOC is moderately biodegradable (Wickland et al., 2007; Roehm et al., 2009; Vonk et al., 2015), whereas certain permafrost soil-derived DOC can be highly biodegradable, particularly Pleistocene yedoma DOC (Vonk et al., 2013; Abbott et al., 2014; Spencer et al., 2015). These studies point to a high susceptibility of permafrost DOC to degradation during transport from soils, and within surface waters, compared to nonpermafrost DOC, with aliphatic DOC originating in permafrost being preferentially degraded (Spencer et al., 2015). Surface waters contain a mixture of DOC from different sources, and therefore it is difficult to isolate the relative biodegradability of permafrost soil vs. non-permafrost soil (active layer) derived DOC within surface waters (Holmes et al., 2008; Balcarczyk et al., 2009; Mann et al., 2012; Wickland et al., 2012). Actively thawing permafrost features release elevated amounts of BDOC to low order water tracks and outflows (Abbott et al., 2014), but significant biodegradation during transport to higher order streams and rivers remove substantial amounts of permafrost DOC before reaching major arctic rivers and the ocean (Spencer et al., 2015). Increas-

## **BGD**

12, 10719–10815, 2015

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Title Page

**Abstract** Introduction

Conclusions References

> **Tables Figures**

ing hydrologic flowpath lengths and residence time in soils in some areas as a result of permafrost thaw likely promote DOC processing and sorption within watersheds prior to discharge to surface waters (Striegl et al., 2005), further reducing the likelihood of biodegradable permafrost DOC reaching aquatic systems.

Thermokarst can also release significant quantities of POC to aquatic ecosystems (Sect. 2.2), often outweighing the amount of released DOC by factors of 50 or more (Vonk et al., 2013; C. Bulger and S. Tank, unpublished data; S. Weege, unpublished data). In thermokarst lakes, contact between slumping shorelines and overlying water can significantly increases CO<sub>2</sub> and CH<sub>4</sub> production (Sect. 3.5.1, below). In streams impacted by thermokarst, POC will remain entrained for long distance downstream (Kokelj et al., 2013), and may reasonably be subject to significant biological decomposition as a result of this entrainment, as has been shown for other systems (Richardson et al., 2013). Degradation of the POC delivered to aquatic systems as a result of permafrost thaw, however, has received little attention to date.

# 3.2 Photodegradation of organic carbon

Recent studies indicate that sunlight can play an important role in dissolved organic matter (DOM) degradation in thermokarst-impacted lakes (Cory et al., 2013, 2014; Laurion and Mladenov, 2013; Koehler et al., 2014). To understand why sunlight can be an important control on DOM degradation in thermokarst-impacted systems, we review controls on DOM photo-degradation and the specific characteristics of thermokarst lakes that can maximize opportunities for photo-degradation.

Sunlight breaks down DOM into three broad classes of products: (1)  $\rm CO_2$  (and CO that can be subsequently oxidized to  $\rm CO_2$ ; photo-mineralization), (2) partially oxidized or degraded DOM that bacteria can then readily respire to  $\rm CO_2$  (photo-stimulated bacterial respiration), and (3) partially oxidized or degraded DOM that is more recalcitrant to bacteria (Cory et al., 2010, 2013, 2014; Hong et al., 2014; Laurion and Mladenov, 2013). Where the presence of thermokarst increases the transfer of old permafrost carbon from land to water, photochemical conversion of DOC to  $\rm CO_2$  and photo-stimulated

**BGD** 

12, 10719-10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Abstract Introduction

Conclusions References

Tables Figures

•

Back

Close

Full Screen / Esc

Printer-friendly Version



**Figures** 

Introduction

References



Abstract

Conclusions

**Tables** 







Full Screen / Esc

**BGD** 

12, 10719–10815, 2015

**Reviews and** 

**Syntheses: Effects of** 

permafrost thaw

J. E. Vonk et al.

Title Page

Printer-friendly Version

Interactive Discussion



bacterial respiration may therefore increase C transfer to the atmosphere on relatively short time scales, thus providing a positive feedback to global warming (e.g., Cory et al., 2013, 2014; Laurion and Mladenov, 2013). The water column rate of DOM photodegradation to CO<sub>2</sub> or to partially oxidized DOM increases with increasing UV radiation from sunlight, and also depends on the rate of light absorption by DOM in the water column, and the lability of DOM to be converted to CO<sub>2</sub> or to partially oxidized DOM during light exposure.

UV irradiance entering the water column depends on the presence of snow and ice cover (Vincent and Belzile, 2003), the solar zenith angle (i.e., latitude, date and time of day) and the composition of the atmosphere (e.g., ozone, and the amount and type of clouds and aerosols) (Bernhard et al., 2013; Vavrus et al., 2010). In surface waters across the Alaskan Arctic (from Toolik Lake to Barrow, AK), the sun is above the horizon from approximately mid-May through mid-July, but ~ 90 % of the daily UV flux involved in DOM degradation reaches surface waters during the day due to the low solar zenith angle overnight (Cory et al., 2014). Clouds generally decrease surface UV, but the effect of clouds can be offset by ozone levels, making it difficult to predict surface UV based only on latitude and date across the Arctic (Bernhard et al., 2013; Vavrus et al., 2010). Although less UV generally reaches surface waters in the Arctic compared to lower latitudes due to lower solar zenith angles, thermokarst lakes can often have high concentrations of light-absorbing DOM that is susceptible to photo-degradation, thus potentially counter-balancing the lower UV (Koelher et al., 2014; Cory et al., 2014).

The many shallow lakes across the Arctic often contain high concentrations of lightabsorbing CDOM (e.g., Hobbie, 1980). For example, Naperian absorption coefficients for CDOM were reported to range from 9-17 m<sup>-1</sup> at 330 nm in lakes of the Mackenzie Delta (Gareis et al., 2010), while a survey of 380 lakes in the Alaskan Arctic reported a mean of 11 m<sup>-1</sup> vs. 31 m<sup>-1</sup> at 320 nm for lakes near the foothills of the Brooks Range vs. lakes on the coastal plain, respectively (Cory et al., 2014). For small thermokarst lakes of eastern Canada, Watanabe et al. (2011) present a large range of absorption coefficients at 320 nm (9.9 to 56 m<sup>-1</sup>) at the southern limit of permafrost along eastern

**BGD** 

12, 10719–10815, 2015

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Title Page

Introduction **Abstract** Conclusions References **Tables Figures** Back Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



Hudson Bay, while even a wider range was obtained by Breton et al. (2009) on a series of sub-arctic and arctic lakes (reaching up to 171 m<sup>-1</sup>), although these latter systems are typically also affected by high levels of non-CDOM UV absorbance (see below).

Near-surface rates of DOM photo-degradation increase linearly with increasing 5 CDOM, while the depth-integrated rates of photo-degradation in the water column depend non-linearly on CDOM concentrations due to the attenuation of light (Miller, 1998). In thermokarst lakes, CDOM is often the main UV-absorbing constituent (e.g., Cory et al., 2014; Gareis et al., 2010; Hobbie, 1980), and thus controls the depth of UV light penetration. However, non-algal particles and especially fine inorganic particles can contribute a significant portion of UV attenuation in thermokarst lakes influenced by marine clays and silts (Watanabe et al., 2011). In such cases, UV is attenuated to a much larger extent, limiting photochemical reactions to the very surface under the strongly stratified conditions observed for these systems (Laurion et al., 2010). Although CDOM concentrations and thus light attenuation are often high in lakes across the Arctic, the whole water column can still be exposed to UV because many of these systems are shallow (Cory et al., 2014; Gareis et al., 2010; see also Sect. 2.1). For example, a survey of CDOM and UV light in thermokarst lakes of the Mackenzie River delta concluded that 19 and 31 % of the water column was exposed to UVB and UVA radiation, respectively (Gareis et al., 2010). For a series of thermokarst lakes of the coastal plain in the Alaskan Arctic, up to 20% of the water column was exposed to UVB while 30-100 % was exposed to UVA (Cory et al., 2014). Exceptions include turbid streams, and lakes impacted by thermokarst slumping (Bowden et al., 2008; Cory et al., 2013; Gareis et al., 2010; Watanabe et al., 2011), or lakes with abundant macrophyte production (Gareis et al., 2010), where UV penetration is low.

The degree to which DOM drained from catchment soils underlain by permafrost is susceptible to photo-degradation, quantified as the apparent quantum yield for each major class of DOM photo-products, has been measured to be on the high end of the range reported for aquatic DOM (Cory et al., 2013, 2014; Hong et al., 2014; Koehler et al., 2014). These findings suggest that DOM originating from soils underlain by per-

References

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



mafrost may be more labile to photo-degradation relative to DOM in freshwaters outside the Arctic, consistent with prior studies suggesting high photo-reactivity of DOM in arctic surface waters (Laurion and Mladenov, 2013; Mann et al., 2012). However, DOM leached specifically from the permafrost soil layer sampled across the Arctic has a consistently lower concentration of aromatic, light absorbing carbon (i.e., lower CDOM per DOC concentration), often quantified as lower SUVA<sub>254</sub> values (Abbott et al., 2014; Cory et al., 2013; Mann et al., 2012, 2014; Ward and Cory, 2015), compared to DOM draining from the active, organic surface layer. Despite lower concentrations of lightabsorbing C, permafrost DOM has been measured to be equally or more sensitive to photo-degradation in a series of sites in the Alaskan Arctic (i.e., when comparing apparent quantum yields for photochemical CO<sub>2</sub> production for example, corrected for differences in rates of light-absorption; Cory et al., 2013). These results may suggest that the chemical composition of permafrost DOM makes it more reactive to sunlight than expected based on aromatic C content alone.

Overall, the typical range of CDOM concentrations and the shallow water depths in thermokarst lakes can mean that a greater fraction of DOM is exposed to UV in the water column compared to deeper, non-thermokarst lakes. Especially for lakes with no outlet, the residence time of DOM and its exposure to UV light is high, thus confining DOM to a thin boundary layer where opportunities for photo-degradation are maximized (e.g., Lougheed et al., 2011; Olefeldt and Roulet, 2012). The alternation of stratification periods (intensive UV exposure at the very surface) with nighttime cooling and mixing (renewal of surface water DOM) observed in many shallow thermokarst lakes and lakes may also offer a greater opportunity for efficient DOM photo-degradation. With forthcoming climate change and deeper permafrost thaw, C flux from peaty soils to thermokarst lakes may also be enhanced in some regions, and the released DOM will be subject to UV-induced mineralization, especially as the summer season lengthens (Erickson et al., 2015). On the other hand, a study of 73 lakes in the Mackenzie Delta uplands found that slump-impacted lakes had significantly lower CDOM than unimpacted lakes (Kokelj et al., 2009; Thompson et al., 2012; see Sect. 2.2), indicating

# **BGD**

12, 10719–10815, 2015

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Title Page **Abstract** 

Introduction

Conclusions

**Tables** 

**Figures** 



that photodegradation may decline in some slump-impacted systems due to adsorption of CDOM to basic cations and clay particles. To understand the role of sunlight in DOM processing in thermokarst waters, future work must quantify UV irradiance in the water column, residence time of DOM in the UV-exposed portion of the water, and identify the factors that control vertical losses of DOM and the lability of permafrost DOM to photo-degradation.

#### 3.3 Photochemical and microbial transformation of contaminants

There are likely to be multiple effects of permafrost thaw on the mobility and transformation of contaminants in arctic environments. The climate-triggered release of contaminants may increase contaminant transport into aquatic systems through increased leaching and hydrological connectivity. Thermokarst lakes with anoxic sediments may act as sinks for some trace metals, but as a source for others, and may also allow enhanced microbial activity and hypolimnia and bacterial metal alkylation in the hypolimnion (e.g., production of the neurotoxin methylHg). Here, we examine the potential mechanisms that may allow arctic warming to lead to increases in the degradation and transformation of contaminants.

Photochemical transformations may affect both the mobility and availability of photoreactive contaminants such as the DOC-driven photoredox transformations of As (Buschmann et al., 2005) and Hg. Studies have shown that Hg can be photoreduced and volatilized in surface waters (Tseng et al., 2004), whereas the neurotoxin methylHg (MeHg) can be photodegraded (Hammerschmidt and Fitzgerald, 2010). Photochemical transformation of contaminants may become more important in regions where total thermokarst lake area is increasing, as these lakes are shallow ecosystems and are irradiated constantly during the summer months.

Changes in the release of DOC and POC into aquatic ecosystems may significantly affect the phase partitioning and solubility of contaminants. In discontinuous permafrost regions, an increase in labile organic carbon (i.e. DOC) with thawing may lead to enhanced microbial activity and hence alter the microbial transformation of some contam-

BGD

12, 10719–10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Abstract Introduction

Conclusions

Tables Figures

•

Back

Close

Full Screen / Esc

Printer-friendly Version



Discussion Paper

Printer-friendly Version

inants (Roehm et al., 2009). Microbes can influence contaminant cycling by degrading organic contaminants, alkylating metals, and creating redox gradients that may modify the mobility and toxicity of toxic metals. Permafrost thaw can affect microbial diversity and microbial activity. Microbial diversity typically is highest in the surface active layer 5 and decreases towards the permafrost table (Frank-Fahle et al., 2014). Deepening of the active layer will increase microbial diversity and allow microbes to speed up the degradation and transformation of some contaminants. Changing microbial diversity in combination with nutrient and temperature effects can affect the microbial degradation of organic pollutants in thawing permafrost (Bell et al., 2013). Microbial activity can be enhanced when permafrost thaw creates new environments such as warm, stratified thermokarst lakes, which may be potential sites for bacterial metal alkylation (Stern et al., 2012). Thermokarst lakes with hypoxic or anoxic bottom waters may be sites that are highly conducive to microbial Hg(II) methylation, and increasing inputs of organic matter and nutrients from thawing permafrost into these systems may have potentially important consequences for the transport or in situ production of methylmercury (MacMillan et al., 2015).

Studies in thermokarst lakes in Siberia (Manasypov et al., 2014) have shown close relationships between diagenetic processes and the remobilization of contaminants from the sediments. This remobilization is tied to diagenetic reactions occurring in these lakes due to the microbial mineralization of natural organic matter. The anoxic conditions in lake sediments during the early stages of thermokarst lake development result in microbe-mediated reactions causing authigenic sulphide precipitation (i.e., the reduction of sulphate) that can create a sink for metals in the sediments (Audry et al., 2011). Early diagenetic reactions in thermokarst lakes and the resulting shift in redox are responsible for the partitioning of trace elements, including several major contaminants (As, Cu, Zn, Cd, Pb, Ni). During all stages of lake development, the sediments may be a source of dissolved Ni and As to the water column (Audry et al., 2011). Overall, permafrost thaw may initially lead to higher levels of dissolved (bioavailable)

#### **BGD**

12, 10719–10815, 2015

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Title Page Introduction **Abstract** Conclusions References

> **Tables Figures**

Close

Full Screen / Esc

Back

trace metal concentrations in thermokarst lakes; as lake development continues, there is likely to be a decrease in dissolved metal concentrations.

#### 3.4 Microbiology of thaw waters

Biogeochemical data collected to date from thermokarst lakes, rivers, lakes and wetlands point to the importance of these environments as sites of intense microbial activity in the northern landscape. As a result there is an increasing effort to apply molecular microbiological methods, particularly next generation nucleic acid sequencing techniques, to understand the biodiversity, network relationships and biogeochemical capabilities of these microbial communities. This research theme is still at an early stage of development, but the picture that is emerging is one of complex microbial consortia, with all domains of life well represented, and dominance by certain groups that play key biogeochemical roles.

Methanogenic archaea occur in high abundance in the anoxic,  $CH_4$ -rich waters of permafrost thaw waters and wetlands, and molecular techniques have revealed a variety of taxa. In the High Arctic, gene signatures from acetoclastic and hydrogenotrophic methanogenic Archaea were detected in ponds associated with ice-wedge polygons (Negandhi et al., 2013), while a study across a permafrost gradient in northern Sweden showed that partially thawed sites were often dominated by a single taxon ("Methanoflorens stordalenmirensis") that belongs to the uncultivated archaeal lineage "Rice Cluster II" (Mondav et al., 2014). Metagenomic analysis showed that this microorganism has the genes for hydrogenotrophic methanogenesis. A subsequent molecular study combined with isotopic analyses showed that the abundance of this taxon is a predictor of the relative proportions of carbon released from the thawing permafrost as  $CH_4$  vs. carbon  $CO_2$  (McCalley et al., 2014).

There is now a rapidly growing DNA data base for bacteria in permafrost soils, which often contain anaerobic groups such as sulphate reducers, Fe(III) reducers and denitrifiers, and many aerobic groups including actinobacteria and methanotrophs (Jansson and Tas, 2014). By comparison, much less is known about the microbial constituents

**BGD** 

12, 10719–10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Abstract Introduction

Conclusions References

Tables Figures

I₫

•

Back

Full Screen / Esc

Close

Printer-friendly Version



Interactive Discussion



of thaw waters. Soil crusts in the High Arctic polar desert have been shown to contain remarkably diverse communities of bacteria, with evidence that their populations of cyanobacteria and acidobacteria are stimulated by water track flows over the permafrost (Steven et al., 2013). High throughput analysis of bacterial samples from High 5 Arctic polygonal ponds showed that the planktonic sequences in these waters were dominated by carbon degrading taxa in the Bacteroidetes, Betaproteobacteria and Actinobacteria (Negandhi et al., 2014). In contrast, the sediment community had a higher alpha-diversity and the sequences included carbon degraders (29-46%), cyanobacteria (20-27%), purple non-sulfur bacteria (6-13%), methanotrophs (11-20%) and methanogen symbionts (1–2%).

DNA clone library analysis of thermokarst lakes in a sporadic permafrost region revealed large differences in the assemblages inhabiting the different water layers, and the presence of methanotrophic bacteria (Rossi et al., 2013). Subsequent analysis by high throughput RNA sequencing showed that the dominant bacterial taxa were betaproteobacteria, especially the genera Variovorax and Polynucleobacter (both known to degrade a wide variety of organic compounds), and that methanotrophs (notably Methylobacter) were also well represented (Crevecoeur et al., 2015). Methanotrophic taxa accounted for up to 27% of the total bacterial sequences, indicating the importance of CH<sub>4</sub> as an energy source in these ecosystems. A puzzling observation was that the anoxic bottom waters in most of these thermokarst lakes had abundant methanotrophs, accounting for up to 23% of the sequences. This could be the result of intermittent injection of oxygen into these bottom waters by mixing, or sustained viability of the methanotrophs mixed down from the aerobic surface zone. Such mixing occurs mostly during fall (Deshpande et al., 2015), which would imply prolonged survival under anoxic conditions, and the availability of an inoculum for rapid response to oxygen resupply during mixing. However, the composition of winter microbial communities in ice-covered thermokarst lakes is at present unknown, and will require close attention in the future.

**BGD** 

12, 10719–10815, 2015

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Title Page Introduction **Abstract** 

Conclusions References

> **Tables Figures**

Back Close

Full Screen / Esc

Full Screen / Esc

Interactive Discussion



High-throughput DNA sequencing has also been used to examine biogeographical patterns. The bacterial composition of thermokarst lakes was examined over a North-South gradient of permafrost degradation in the eastern Canadian sub-Arctic and showed that greater differences occurred among valleys across this gradient than among lakes within a valley, despite marked differences in limnological properties among neighboring lakes (Comte et al., 2015). This implies that the taxonomic composition and perhaps also the biogeochemical functioning of thermokast lake bacterial assemblages are regulated by local landscape features, such as the extent of permafrost thaw.

Phototrophic organisms in thaw waters include photosynthetic sulphur bacteria (Rossi et al., 2013; Crevecoeur et al., 2015), benthic cyanobacteria and purple nonsulphur bacteria (Negandhi et al., 2014), picocyanobacteria, as well as eukaryotic algae of diverse phylogenetic groups (Przytulska et al., unpublished), however the network associations among these organisms and other microbial taxa has yet to be explored. Such analyses have been applied to temperate lake communities, combining bacterial DNA sequence data with phytoplankton and zooplankton counts by microscopy, and these reveal highly connected, potential keystone taxa in the mixed communities (Peura et al., 2015).

In large arctic river systems, bacterial communities showed a large spatial synchrony, along with clear seasonal community differences driven by shifts in hydrology and biogeochemistry that reassembled annually (Crump et al., 2009). Furthermore, Crump et al. (2012) observed a decreasing species diversity downslope in a soil-stream-lake sequence in Alaska. Soil waters and headwater streams showed highest species richness, whereas lake waters show a lower diversity. They suggest that bacterial and archaeal diversity in freshwaters is initially structured by inoculation of soil microbes, and then subject to a species-sorting process during downslope dispersal. Permafrost thaw could lead to a greater transfer of soil microbes into aquatic communities.

A conspicuous gap in information available to date is the diversity and role of viruses in these microbe-rich habitats. Viruses are likely to be the biologically most abundant **BGD** 

12, 10719–10815, 2015

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Title Page Introduction **Abstract** 

Conclusions References

> **Tables Figures**

Back

Close

Printer-friendly Version

particles in these waters, as elsewhere, and may influence the species succession of microbes in all domains of life, affect carbon cycling by their lytic activities, and have a controlling influence on evolutionary processes via horizontal gene transfer (Suttle et al., 2007). New viral lineages are being discovered in environments elsewhere, including high latitude freshwaters (Chénard et al., 2015), and thermokarst lakes will likely yield additional new groups given the diversity of potential host taxa.

#### 3.5 Aquatic gas fluxes

#### 3.5.1 Emission of CO<sub>2</sub> and CH<sub>4</sub> from permafrost-thaw impacted systems

In well-drained terrestrial environments, permafrost thaw leads to microbial decomposition resulting in variable production and emission of  $CO_2$  (e.g., Schuur et al., 2009; Schädel et al., 2014). Thaw of ice-rich permafrost, particularly in poorly drained low-land areas, results in ground subsidence and saturated soils that take the form of thermokarst lakes, wetlands and slumping into streams. Since waterlogging slows the diffusion of oxygen from the atmosphere into soils, this results in anoxic conditions in sediments as well as in portions of the overlying water columns of many thermokarst water bodies. Under anaerobic conditions, decomposition of organic matter also produces  $CH_4$ , a potent greenhouse gas (Myhre et al., 2013; Wickland et al., 2006). Where soils surrounding thermokarst lakes are anoxic, lateral inputs of  $CH_4$  produced within the active layer can also occur (Paytan et al., 2015).

In a study of 40 Alaskan thermokarst lakes (Fig. 5) that span large gradients of climate, vegetation, geology and permafrost regimes, Sepulveda-Jáuregui et al. (2015) found that all lakes were net sources of atmospheric  $CH_4$  and  $CO_2$  (when integrated over a year) as also noted earlier by Kling et al. (1991, 1992). On a C mass basis,  $CO_2$  emissions from Alaskan lakes were  $\sim$  6-fold higher than  $CH_4$  emissions. However, considering the  $\sim$  30-fold stronger global warming potential of  $CH_4$  vs.  $CO_2$  over 100 years (GWP<sub>100</sub>; Myhre et al., 2013),  $CH_4$  emissions had nearly twice the impact on climate as  $CO_2$  emissions in this region.

**BGD** 

12, 10719–10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Abstract Introduction

Conclusions References

Tables Figures

Back Close

Full Screen / Esc

Printer-friendly Version



**Figures** 





Full Screen / Esc

Printer-friendly Version

Interactive Discussion



In the Eastern Canadian Arctic, a thermokarst lake (deep enough to have unfrozen water in winter and likely a talik underneath) was shown to be a relatively small GHG emitter in July (Bouchard et al., 2015), although its thermal structure suggests that GHG stored in the hypolimnion is transferred to the atmosphere at the autumnal overturn. Also in the Eastern Canadian Arctic, thermokarst ponds (smaller in size and freezing to bottom in winter) at the southern limit of permafrost in Nunavik and in the continuous permafrost zone on Bylot Island, Nunavut, show large variations in summertime CO<sub>2</sub> and CH<sub>4</sub> fluxes. Turbid, sub-arctic thermokarst ponds were all GHG emitters, but showed on average a 530-fold higher CO2 than CH4 diffusive flux in summer, with strong GHG gradients in the hypolimnion (summer storage). In the high-Arctic, polygonal ponds over low-centered ice wedge polygons were CO2 sinks because of colonization by active cyanobacterial mats (Laurion et al., 2010), while shallower ice-wedge trough ponds were identified as the main GHG emitters (Negandhi et al., 2013), with summer CO₂ fluxes ~ 25-fold higher than CH₄ diffusive flux. At this site, the CH₄ ebullition flux (likely background ebullition) was in the same range as the diffusive flux.

In streams and rivers, emission of CO<sub>2</sub> is typically much greater than emission as  $CH_4$  (Striegl et al., 2012). On a catchment scale, sub-arctic and arctic streams within permafrost zones can emit relatively high amounts of GHG relative to their areal extent (Koprivnjak et al., 2010; Teodoru et al., 2009; Denfeld et al., 2013; Lundin et al., 2013; Striegl et al., 2012; Crawford et al., 2013), and gaseous emissions can account for up to 50 % of total C exports (Striegl et al., 2012). For example, in northern Sweden streams accounted for 4% of the aquatic surface area yet accounted for 95% of the total aquatic emission (Lundin et al., 2013), whereas in northern Québec stream accounted for 1 % of the aquatic surface and accounted for 25 % of the aquatic emissions (Teodoru et al., 2009). Stream  $CH_4$  emissions can also be significant to total catchment emissions; for example in interior Alaska stream CH<sub>4</sub> emission was estimated to be up to 10% of catchment terrestrial emissions despite the very low surface area (< 0.2% of catchment area; Crawford et al., 2013). The relatively high emissions can be attributed both to supersaturation relative to the atmosphere as well as high gas transfer

#### **BGD**

12, 10719–10815, 2015

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Title Page Introduction **Abstract** 

Conclusions References

**Tables** 

Back

Close

velocities associated with these more turbulent waters (Kling et al., 1992; Striegl et al., 2012; Lundin et al., 2013; Denfeld et al., 2013). To date, however, there are no published studies to show how gas fluxes are affected by the direct action of thermokarst slumping into streams.

#### 5 3.5.2 Scale and distribution of GHG measurements

Since the solubility of CO<sub>2</sub> exceeds that of CH<sub>4</sub>, CO<sub>2</sub> evades aquatic ecosystems primarily by diffusion, while CH<sub>4</sub> more readily comes out of solution, forming bubbles in sediments that escape to the atmosphere by ebullition. Emission of CH<sub>4</sub> through diffusion from aquatic systems can, however, also be high, particularly in wetlands, lakes and other standing open water (Reeburgh et al., 1998; Lundin et al., 2013). Due to large heterogeneity in the spatial and temporal dynamics of ebullition, this mode of CH₄ emission is less commonly studied than diffusion (Bastviken et al., 2011; Wik et al., 2015), although ebullition has been found to be the dominant form of CH<sub>4</sub> emission in many thermokarst lakes (Bartlett et al., 1992; Walter et al., 2006; Sepulveda-Jáuregui et al., 2015; but see discussion on the Eastern Canadian Arctic, above). Recent studies focusing on ebullition dynamics in thermokarst lakes distinguished multiple sub-modes of ebullition emission including seep ebullition, background ebullition, and ice-bubble storage (Walter et al., 2006; Greene et al., 2014; Sepulveda-Jáuregui et al., 2015; Langer et al., 2015; Fig. 5). Background ebullition is most commonly reported in the literature and consists predominately of distributed bubbling from seasonally warm surface sediments. In contrast, seep ebullition involves bubbling of CH<sub>4</sub> formed at depth in dense sediments (Fig. 4), which are typically found in thaw bulbs beneath thermokarst lakes and streams (Walter Anthony et al., 2014). Seep ebullition occurs repeatedly from the same point-source locations and occurs year-round due to the thermal lag that results in warmer temperatures in deep sediments through the fall to winter. Bubbling rates in hotspot seeps are high enough to maintain open holes in thermokarst lake ice, resulting in the emission of CH<sub>4</sub>-rich bubbles to the atmosphere throughout winter (Zimov et al., 2001; Greene et al., 2014; Fig. 5). In thermokarst lakes where both seep

BGD

12, 10719-10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

.....

Figures

I₫



■ ...



Back



Full Screen / Esc

Printer-friendly Version



Discussion Paper

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



and background ebullition were measured, seep ebullition was found to dominate CH<sub>4</sub> emissions despite occupying only a small fraction of the lake surface area (Walter et al., 2006). More recently, ice-bubble storage, the release of ebullition bubbles seasonally trapped by winter lake ice upon spring melt, was also recognized as an important, additional mode of ebullition. It contributed 9-13% of total annual CH<sub>4</sub> emissions from thermokarst (and non-thermokarst) lakes in Alaska (Greene et al., 2014; Sepulveda-Jáurequi et al., 2015) and was also recognized as an important springtime emission mode in West Siberian lakes (Golubyatnikov and Kazanstev, 2013). Ice-bubble storage is likely an important mode of emission in many Arctic systems since CH<sub>4</sub>-rich ice-bubbles have been observed in aquatic systems in Northeast Siberia (Walter et al., 2006; Langer et al., 2015), Sweden (Boereboom et al., 2012; Wik et al., 2011), Finland (Walter Anthony et al., unpublished data); Greenland (Walter Anthony et al., 2012), Alaska (Walter et al., 2007; Brosius et al., 2012; Sepulveda-Jáurequi et al., 2015) and Canada (Duguay et al., 2002; Brosius et al., 2012).

One promising technique for measuring aquatic gas fluxes in permafrost-impacted systems is eddy covariance (EC). Although EC data on inland freshwater ecosystems are still rare, this approach shows great promise for understanding aquatic carbon fluxes. Currently, on-going EC measurements focus on CO<sub>2</sub> and CH<sub>4</sub> fluxes over thermokarst lakes in Siberia (Russia) (T. Sachs, personal communication, 2015; L. Belelli-Marchesini, personal communication, 2015) or sub-arctic lakes within thawing permafrost environments (M. Jammet, personal communication, 2015). Using EC, Eugster et al. (2003) found efflux rates of 114 mg C m<sup>-2</sup> day<sup>-1</sup> over an arctic Alaskan lake in late July, which agreed well with two other continuous flux estimation techniques (boundary layer and surface renewal models). In a boreal lake, CO<sub>2</sub> effluxes determined by episodic floating chamber measurements were about 100 % larger than fluxes measured with EC, indicating potential biases related to inadequate spatial and/or temporal sampling intervals of the chamber method (Podgrajsek et al., 2014a).

While proving the feasibility of eddy covariance measurements in freshwater ecosystems, aquatic EC work also highlights challenges related to the application of this **BGD** 

12, 10719–10815, 2015

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Title Page **Abstract** Introduction

Conclusions References

> **Tables Figures**

Back Close terrestrially-optimized approach to aquatic systems (Eugster et al., 2011; Vesala et al., 2006). Overall, EC shows great promise with respect to: (1) integration of all gas flux pathways from the lake sediments to the atmosphere, namely diffusion, ebullition and plant-mediated transport in the case of CH<sub>4</sub>, (2) continuous flux monitoring over time, enabling the capture of episodic ebullition events of CH<sub>4</sub> in lakes (Eugster et al., 2011), and (3) the analysis of dynamic responses of lake-atmosphere carbon fluxes to temporal (including diurnal) changes in environmental variables (Eugster, 2003; Podgrajsek et al., 2014b; Vesala et al., 2006). A significant portion of gaseous carbon emissions from seasonally ice-covered lakes appears to occur during spring ice-thaw (e.g. Karlsson et al., 2013), stressing the importance of year-round carbon flux monitoring on thermokarst lakes.

## 3.5.3 Lake morphology and evolution

The morphological diversity of thermokarst lakes will have important consequences for hydrology, physicochemistry, and thus ultimately the microbial processes responsible for GHG production and evasion dynamics at the air-water interface. Within the context of permafrost soil organic carbon content, thermokarst lakes have been classified depending on whether they are surrounded by yedoma-type permafrost or non-yedoma substrates (Walter Anthony et al., 2012; Sepulveda-Jáuregui et al., 2015). Yedoma is typically thick (tens of meters), Pleistocene-aged loess-dominated permafrost sediment with high organic carbon (~ 2 % by mass) and ice (50-90 % by volume) contents (Zimov et al., 2006). When yedoma thaws and ground ice melts, deep thermokarst lakes with high CH<sub>4</sub> production potentials form (Zimov et al., 1997; Kanevskiy et al., 2011; Walter Anthony and Anthony, 2013). Because these deep (> 2 m) lakes are often humic and underlain by a talik, they are stratified for most of the year and are likely to have an anoxic hypolimnion controlling GHG producers and consumers. These systems present large GHG seepage ebullition throughout the year, with a characteristic seasonal pattern in GHG evasion (Walter et al., 2006; Sepulveda-Jáuregui et al., 2015). Smaller but very turbid thermokarst ponds studied in the Eastern coast of Hudson Bay

BGD

12, 10719–10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Abstract Introduction

Conclusions References

Tables Figures

14

•

Back

Close

Full Screen / Esc

Printer-friendly Version



**BGD** 

12, 10719–10815, 2015

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Title Page

Introduction **Abstract** Conclusions References **Tables Figures** Back Close

Printer-friendly Version

Full Screen / Esc

Interactive Discussion



also do not freeze to the bottom and can similarly be highly stratified and anoxic (Deshpande et al., 2015; Laurion et al., 2010). Some non-yedoma permafrost soils can also have high organic carbon and excess

ice concentrations within several meters of the ground surface; however, these organic-5 and ice-rich permafrost horizons are typically thinner than yedoma deposits (Ping et al., 2008; Tarnocai et al., 2009; Bouchard et al., 2015). As a result, thermokarst lakes formed in non-yedoma permafrost soils are commonly shallower than yedoma lakes and have been shown to emit less CH<sub>4</sub> (West and Plug, 2008; Grosse et al., 2013; Walter Anthony and Anthony, 2013). For instance, CH<sub>4</sub> emissions from thermokarst lakes formed in carbon-rich yedoma permafrost were 6-fold higher than emissions from other lake types across Alaska (Sepulveda-Jáuregui et al., 2015). Other larger nonyedoma lakes susceptible to permafrost thaw have also been reported to cover vast areas in the Hudson Bay Lowlands and within the Yukon (Bouchard et al., 2013; Turner et al., 2014). To date, GHG gas fluxes in these systems has not been studied.

Shallow thermokarst lakes (e.g., Turner et al., 2014) may allow colonization by plants and benthic photosynthesizing mats, creating CO<sub>2</sub> sink periods while they remain CH<sub>4</sub> emitters (Laurion et al., 2010; Negandhi et al., 2014; Tank et al., 2009). These lakes can freeze to the bottom (no talik, depending on latitude), which limits active GHG production to the unfrozen period of the year. In these shallow lakes there is less opportunity for the dissolution of ebullitive CH<sub>4</sub> before it escapes to the atmosphere, and thus for its consumption by methanotrophic bacteria. Furthermore, large and shallow lakes are generally polymictic with GHG evasion largely influenced by winds, generating oxic conditions. For very small water bodies (a few m<sup>2</sup>) such as ice-wedge trough ponds, microtopography will be the main regulator of thermal structure, and gas exchange will be most affected by heat flux. Even though trough ponds are very shallow systems (< 1 m), they can be highly stratified with only occasional mixing events during the summer (Bouchard et al., 2015), resulting in large periods of hypoxic to anoxic bottom waters, and evasion of GHG stored in bottom waters following changes in meteorological conditions. Depending on their erosional features these ponds can also

Back



be colonized by aquatic plants associated with efficient methanotrophic communities (e.g., Liebner et al., 2011; Siberian ponds).

Finally, the evolution of thermokarst lake landscapes is a critical determinant of the overall carbon balance of these systems (van Huissteden et al., 2011; Walter Anthony 5 et al., 2014). This landscape evolution can be observable on timescales as short as 30-40 years (Polishuk et al., 2012; Bryksina et al., 2011; Smith et al., 2005) and is characterized by cyclical flooding of vegetated soils and recolonization of drained lake bottoms, and an evolution from strong CH<sub>4</sub> emission during the initial phase of lake formation, through a phase of carbon accumulation associated with higher within-lake primary production and the creation of terrestrial wetlands as lakes drain (Ovenden, 1986; van Huissteden et al., 2011; Walter Anthony et al., 2014).

## Consequences

# Release of old permafrost OC into aquatic systems and the atmosphere

## 4.1.1 Release of old permafrost OC into aquatic systems

A recent study of Eurasian Arctic river basins concluded that climate change-induced mobilization of old permafrost OC is well underway in the Arctic (Feng et al., 2013). In this section we review the evidence that currently exists for the release of old permafrost OC into aquatic systems.

Mobilization of old permafrost OC to surface waters could occur in the form of DOC, POC or gaseous (CO<sub>2</sub> or CH<sub>4</sub>). Release of old C can be measured with radiocarbon isotopes, either on bulk OC or on compound specific biomarkers, or a combination of both. Since permafrost and Yedoma deposits contain organic C with ages of > 30 000 yr BP (e.g. Zimov et al., 2006), radiocarbon could be an excellent marker to detect change in arctic aquatic environments.

#### **BGD**

12, 10719–10815, 2015

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Title Page **Abstract** Introduction Conclusions References

> **Tables Figures**

Close

Full Screen / Esc

Conclusions

**Figures** 



Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



Research that includes bulk radiocarbon measurements in rivers has largely focused on DOC and POC. Generally, DOC in larger river systems tends to be young. In large arctic rivers, Amon et al. (2012) and Guo et al. (2007) measured <sup>14</sup>C-DOC values ranging from 83 to 113% modern (1440 to modern yrBP). Within large arctic river basins there is significant spatial variability in riverine <sup>14</sup>C-DOC values, reflecting dominant water and carbon source materials (Aiken et al., 2014; O'Donnell et al., 2014). Export of contemporary DOC in rivers dominates the spring freshet event, a time of year when the majority of water and DOC export occurs. In the Ob', Yenisey, Lena, Mackenzie, and Yukon rivers, Raymond et al. (2007) estimated that ~ 90 % of DOC exported at this time was less than 20 years old. The presence of young DOC during the spring freshet and the slight aging of DOC (675 yr BP) during the late summer low flow season in the Kolyma watershed (Siberia) demonstrated the significance of seasonal change in the release of DOC of variable age during the hydrological year (Neff et al., 2006). Winter flow is most <sup>14</sup>C-depleted, although there can be significant variation within a large river basin even in winter. O'Donnell et al. (2014) measured winter radiocarbon ages ranging from 35 to 445 yr BP, likely related to regional groundwater travel times.

Whilst the overall picture for major arctic rivers is one of export of large amounts of young semi-labile DOC to the Arctic Ocean, there are also examples of mobilization of old DOC, particularly within smaller systems. In the Sag River, draining north Alaskan tundra, DOC age was 2170-4950 yr BP (Guo et al., 2007). This age was similar to the age of soil organic matter in the river basin and the presence of old DOC in surface waters was linked to active layer and soil cryoturbation. The most striking and the oldest DOC ever dated in surface waters is from small (first-order) sediment-rich thaw streams draining directly into the Kolyma River, Siberia (Vonk et al., 2013; Spencer et al., 2015). This Pleistocene-aged (> 21 000 yr BP) DOC is being mobilized from old Yedoma deposits (aged up to 45 000 yr BP) into small DOC-rich streams. This old DOC also shows very high biodegradation potential (Vonk et al., 2013; Spencer et al., 2015; Sect. 3.1), indicating that it may be degraded to CO<sub>2</sub> well before reaching the mouth of larger river systems. Additional sources of ancient DOC in large river systems include

# **BGD**

12, 10719–10815, 2015

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Title Page Introduction

Abstract

References

**Tables** 

Back

deep groundwater and, in some basins, glacial meltwater (Aiken et al., 2014). The DOC concentrations in these source waters, however, are typically low (Aiken et al., 2014).

There is abundant evidence of mobilization of old POC into arctic lakes (Abbott and Stafford, 1996), rivers and estuarine sediments associated with permafrost thaw, bank erosion and transport of organic C. For example, Guo et al. (2007) reported <sup>14</sup>C ages of sediment and suspended POC in large arctic rivers of 4430–7970 yr BP and concluded that POC release and age would increase in arctic river systems subject to global warming. Additionally, there has been some work in smaller coastal watersheds by Lamoureux and Lafreniere (2014) who concluded that recent permafrost disturbance delivered old (up to 6740 yr BP) POC to the aquatic system. Yedoma-derived Pleistocene aged POC has also been identified in sediments from the Colville River Delta, which drains into the west Beaufort Sea (Schreiner et al., 2014; 10 000–16 000 yr BP), and in thaw streams draining Yedoma deposits, Siberia (Vonk et al., 2013; 19 000–38 000 yr BP). One caveat to studies in coastal or estuarine settings is that marine sediments and microfossils could potentially influence the <sup>14</sup>C age of particulate material.

The main problem with <sup>14</sup>C dating on bulk OC is the potential complicating factor of multiple sources, including the influence of <sup>14</sup>C dead organic C, such as geological (sedimentary) C. This has lead to an increase in the use of compound-specific biomarkers. Spencer et al. (2008) found elevated lignin C-normalized yields during the spring freshet across the Yukon River basin, identifying surface vegetation as strong DOC sources. Amon et al. (2012) similarly found that biomarker abundance changed in six of the largest arctic rivers according to season, with high concentrations of lignin phenols in the spring freshet (indicative of fresh vegetation) and elevated levels of p-hydroxybenzenes during the low flow season (indicative of moss and peat-derived OM). They also found a strong relationship between <sup>14</sup>C-DOC age and lignin phenol concentration in five out of six rivers, with the youngest DOC associated with the spring freshet. Concentration differences in source—tracing organic molecules, namely <sup>14</sup>C-young, vascular plant-derived lignin phenols and <sup>14</sup>C-old permafrost-derived waxy

**BGD** 

12, 10719–10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Abstract Introduction

Conclusions

Tables Figures

l∢ ≯l

■ Back Close

Full Screen / Esc

Printer-friendly Version



**Figures** 

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



lipids, were found to show a relationship between <sup>14</sup>C age and permafrost coverage (Feng et al., 2013). Drainage basins associated with increasing amounts of discontinuous permafrost were characterized by older OC. Likewise Gustafsson et al. (2011) found that the average age of n-alkanes in estuarine sediments increased (1140 to 6400 yr BP) from east to west across the Siberian Arctic, consistent with warmer climatic conditions and more discontinuous permafrost towards the west. Additional biomarkers such as membrane lipids (ex. glycerol dialkyl glycerol tetraethers, GDGTs; bacteriohopanepolyols, BHPs; and intact polar membrane lipids, IPLs) may have the potential to trace terrigenous OC stored in permafrost and remobilized along arctic land-river-ocean transects (Doğrul Selver et al., 2012, 2015; Rethemeyer et al., 2010).

#### 4.1.2 Release of old permafrost OC as greenhouse gases

The consequence of permafrost thaw beneath and adjacent to thermokarst lakes, wetlands and streams is the potential mobilization and return of old carbon to the atmosphere. Schaefer et al. (2014) defined the permafrost carbon feedback as amplification of anthropogenic warming due to carbon emissions from thawing permafrost. Direct evidence for a positive permafrost carbon feedback to climate in thermokarst lakes is found in the radiocarbon ages and deuterium values of CH4 in bubbles and in spatial patterns of CH<sub>4</sub> emissions in thermokarst lakes. Zimov et al. (1997) first revealed that methanogenesis in deep, cold thaw bulbs where Pleistocene-aged yedoma is thawing beneath lakes in Siberia leads to the release of Pleistocene-aged CH<sub>4</sub>. The release of permafrost-derived carbon to the atmosphere in <sup>14</sup>C-depleted CH<sub>4</sub>-rich bubbles contributes to climate warming, which in turn causes permafrost to thaw and more CH<sub>4</sub> to be produced in a positive feedback cycle (Walter et al., 2006). Field observations and modeling showed that permafrost-derived CH<sub>4</sub> emissions were highest along thermokarst margins in Siberian and Alaskan lakes and in younger stages of lake development, where permafrost thaw is most active (Walter et al., 2006, 2007; Desyatkin et al., 2009; Kessler et al., 2012). This permafrost carbon feedback was

12, 10719-10815, 2015

**BGD** 

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

**Abstract** Introduction

Conclusions References

**Tables** 







Back Full Screen / Esc

Printer-friendly Version

Interactive Discussion



affirmed in Alaskan thermokarst lakes by independent evidence from deuterium. Walter et al. (2008) and Brosius et al. (2012) found that  $\delta D$  values of ebullition CH<sub>4</sub> in yedoma-type lakes in Alaska and Siberia reflected CH<sub>4</sub> formation from Pleistoceneorigin melt water, which has a highly negative isotopic signature. In contrast, bubbles 5 emitted from the centers of older yedoma lakes where permafrost is no longer thawing (Kessler et al., 2012), and from non-yedoma lakes, contained higher δD-CH<sub>4</sub> values and younger <sup>14</sup>C-CH<sub>4</sub> ages, pointing to Holocene-aged meteoric water and carbon as the substrates for methanogenesis (Brosius et al., 2012). On the other hand, recent work on an eastern Canadian thermokarst lake (Holocene deposits) shows a different trend, where ebullition  $CH_4$  emitted from the lakeshore was younger ( $\sim 1550 \, \text{yr BP}$ ) and had a more negative  $\delta D$ -CH<sub>4</sub> than from the lake center ( $\sim 3250 \,\mathrm{yr}$  BP; Bouchard et al., 2015). Most interestingly, thermokarst lakes emitted modern CH₄ even though they are exposed to peat slumping and erosion down at least to the active layer (base of active layer  $\sim 2200$  to 2500 yr BP) with  $\delta$ D-CH<sub>4</sub> reaching down to -448% (Bouchard et al., 15 2015).

It is important to note that the present-day permafrost carbon feedback from thermokarst lakes to climate warming is likely smaller than it was in the early Holocene when thermokarst lakes first formed on the permafrost landscape (Walter et al., 2007; Brosius et al., 2012). Walter Anthony et al. (2014) estimated rates of carbon loss from yedoma-type lakes (in North Siberia, Alaska and northwest Canada) to the atmosphere from 20 kyr ago to the present. Their results indicate widespread lake formation between 14-9 kyr ago, generating a major northern source of <sup>14</sup>C-depleted atmospheric CH<sub>4</sub> during deglaciation. The subsequent slow-down of first-generation thermokarstlake formation throughout the Holocene combined with the acceleration of other negative feedback processes (e.g. carbon sequestration by lakes, see next section) results in lower present-day CH<sub>4</sub> emissions, a smaller permafrost carbon feedback, and a net negative radiative forcing of carbon exchange between lakes and the atmosphere on climate.

#### **BGD**

12, 10719–10815, 2015

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Title Page **Abstract** Introduction

Conclusions References

> **Tables Figures**

Close

#### 4.2.1 Carbon burial in arctic aquatic ecosystems

Inland waters receive large quantities of organic matter from their watersheds, but, globally, less than half of this carbon reaches the ocean (Battin et al., 2009). The loss en route is attributed to (i) mineralization to CO<sub>2</sub> and CH<sub>4</sub> (see Sect. 3), and (ii) and sequestration into sediments of lakes and reservoirs (Cole et al., 2007). Sediment sequestration of carbon can be substantial in relatively lake-rich boreal and arctic land-scapes (Lehner and Döll, 2004; Fig. 1), but still receives little attention.

Generally, total carbon mineralization rates exceed carbon burial (Battin et al., 2009; Tranvik et al., 2009), but there are some exceptions, for example in the case of deep thermokarst lakes (see Sect. 4.2.2). Lake shape is a key regulator in carbon burial; small and deep lakes (Ferland et al., 2012; Kortelainen et al., 2004) bury carbon more efficiently than large and shallow lakes. This is explained by a higher benthic metabolic capacity to process incoming carbon and greater particle resuspension in large, shallow, and thus well-mixed lakes. Prior to burial, degradation occurs in the water column and the uppermost sediment layers. This can be substantial with, for example, averages up to 75 % of the OC mineralized over the first few decades following sediment deposition in boreal lakes in Québec (Ferland et al., 2014). Long-term accumulation rates in sediments of arctic and boreal non-thermokarst lakes ranged between 0.2 and  $13\,\mathrm{g\,C\,m^{-2}\,yr^{-1}}$  across sites in Greenland, boreal Québec, and boreal Finland (N. J. Anderson et al., 2009; Ferland et al., 2014; Kortelainen et al., 2004; Sobek et al., 2014). Thermokarst lakes in yedoma regions, however, show much larger long-term sediment accumulation rates (47 ± 10 g C m $^{-2}\,\mathrm{yr^{-1}}$ ; Walter Anthony et al., 2014; see Sect. 4.2.2).

Similarly, coastal shelf regions bordering yedoma-rich Eastern Siberia receive rather large amounts of carbon with accumulation rates of  $36\pm17\,\mathrm{g\,C\,m^{-2}}$  annually (Vonk et al., 2012b). Long-term (Holocene) carbon accumulation rates in this region, however, vary between 0.1 and 2.7 g C m<sup>-2</sup> yr<sup>-1</sup> (Bauch et al., 2001; Stein and Fahl, 2000) suggesting significant decomposition in the sediments after deposition and/or increases

Discussion

- D:

Discussion Paper

Discussion Paper

Discussion Paper

**BGD** 

12, 10719-10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Abstract

Conclusions References

Tables Figures

I4 ►FI

Back Close

Full Screen / Esc

Printer-friendly Version



in recent accumulation rates. Furthermore, recent studies in this region suggest that permafrost-derived carbon is preferentially buried, when compared with marine or modern terrestrial carbon (Vonk et al., 2014). This appears to be in contrast with high initial biodegradability of (yedoma) permafrost DOC upon aquatic release (see Sect. 3.1). We hypothesize that this apparent contrast can be explained by the parallel thaw-release of different pools of organic matter in permafrost (Vonk et al., 2010; Karlsson et al., 2011). On the one hand, DOC and buoyant, non-mineral bound POC are released that are highly sensitive to biodegradation (e.g. Abbott et al., 2014; Vonk et al., 2013) leading to rapid removal in aquatic systems (Spencer et al., 2015), whereas mineral-bound, ballasted POC is resistant to degradation and preferentially transported to (and buried in) coastal shelf sediments (Vonk et al., 2011; Karlsson et al., 2011).

#### 4.2.2 Carbon burial in yedoma thermokarst lakes

Since the last deglaciation (the past 14.7 kyr), about 70 % of all yedoma deposits has thawed through the formation of thermokarst lakes and streams (Strauss et al., 2013). This has released GHG to the atmosphere and OC to lake basin sediments and downstream export. Formation of thermokarst systems, however, also caused atmospheric  $CO_2$  to be absorbed through contemporary plant photosynthesis, senescence and burial. While initial thermokarst basin formation caused significant efflux of  $CO_2$  and  $CH_4$ , as these basins evolved, nutrient-rich sediments facilitated terrestrial and aquatic plant proliferation, leading to sequestration of OC in sediments of drained lake basins during the Holocene. The long-term organic carbon accumulation rate in deep, yedoma thermokarst lakes was found to be on average five times higher than in other northern lakes throughout the Holocene (Walter Anthony et al., 2014). The anomalously high carbon sequestration in yedoma thermokarst lakes was attributed to (a) thermokarst-related shore erosion and deposition of terrestrial organic matter in lake bottoms, (b) high aquatic productivity enhanced by nutrient supply from thawing yedoma, and (c) unique preservation conditions in deep thermokarst lakes.

BGD

12, 10719–10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

I₫

4



Back



Full Screen / Esc

Printer-friendly Version



Since GHG emissions and carbon sequestration have counteractive effects on climate (warming vs. cooling, respectively), the radiative impacts of both processes must be upscaled to understand their overall impact over long time scales. Walter Anthony et al. (2014) developed trajectories of thermokarst-basin carbon flux (for yedoma land-scapes) from the last major glaciation to present, based on estimates of contemporary CH<sub>4</sub> flux, total yedoma carbon lost as CO<sub>2</sub> and CH<sub>4</sub>, total accumulated carbon, and thermokarst-lake initiation dates. Model results indicated that yedoma thermokarst lakes caused a net climate warming at the peak of their formation during deglaciation, driven primarily by CH<sub>4</sub> release from thawed, decaying yedoma. However, high carbon accumulation in existing basins and a slowdown of lake formation caused thermokarst-lake impact on climate to switch from net warming to net cooling around 5 kyr ago, such that these basins are now net GHG sinks. Notably, long-term trends in the climate feedback potential of non-yedoma thermokarst lakes have not yet been extensively investigated, despite the fact that non-yedoma permafrost stores 75 % of the global carbon permafrost pool (Schuur et al., 2015; Zimov et al., 2006).

## 4.3 Ecosystem structure and function

While there is a growing body of literature quantifying the nature, timing, and extent of permafrost thaw in the Arctic, there is considerably less literature on the direct effects of permafrost thaw on the structure and function of aquatic ecosystems, especially streams. The likelihood that permafrost thaw will substantially affect major ecological functions (e.g., photosynthesis, respiration, nutrient uptake) or food web characteristics (e.g., benthic algal biomass, macroinvertebrate community structure) is dependent on several factors, most notably the intensity, spatial extent, temporal duration, and hydrological connectivity of the disturbance associated with permafrost thaw. Here, we provide an overarching review of the potential effects of permafrost thaw on aquatic ecosystem structure and function.

BGD

12, 10719–10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

I₫











Full Screen / Esc

Printer-friendly Version



Arctic regions contain numerous lakes with large differences in abiotic and biotic conditions (Hamilton et al., 2001; Rautio et al., 2011), suggesting that the consequences of permafrost thaw on ecosystem function are likely to vary across lakes. Thawing permafrost and associated changes in export of nutrients and DOM is expected to have pronounced effects on recipient lake ecosystems. Input of nutrients per se will stimulate primary production (mainly via effects on pelagic algae) (Levine and Whalen, 2001; O'Brien et al., 2005). Input of sediments and DOM, on the other hand, will decrease primary production if it leads to suboptimal conditions for photosynthesis, mainly affecting benthic algae but also planktonic algae when lakes are very turbid (Ask et al., 2009; Roiha et al., 2015). In regions where thaw increases both nutrients and DOM we could expect stimulation of total primary production in clear and shallow lakes but suppression of primary production in more colored or deeper lakes (Seekell et al., 2015). In regions where retrogressive thaw slumping delivers mineral-rich sediments to lakes (Mesquita et al., 2010; Thompson et al., 2008), permafrost degradation has led to significantly greater dissolved ion content, lower DOC concentrations and increased water transparency. This has led to enhanced macrophyte development and higher abundance of benthic macroinvertebrates (Mesquita et al., 2010; Moquin et al., 2014) and higher abundance and diversity of periphytic diatoms (Thienpont et al., 2013). Further, DOM released following thaw is relatively labile and could support bacterial metabolism (Roehm et al., 2009; Vonk et al., 2013), resulting in increasing rates of bacterial respiration and production relative to primary production (Breton et al., 2009; Karlsson et al., 2010; Roiha et al., 2015).

These changes at the base of the food web are expected to result in a shift in the relative importance of different OC resources supporting higher consumers, by decreasing the importance of benthic algae and increasing the reliance on pelagic and terrestrial resources with increasing DOM. Heterotrophic bacteria transfer DOM to mixotrophic algae and heterotrophic protozoans, to zooplankton and zoobenthos feeding on bacteria,

Paper

Discussion Paper

Discussion Paper

Discussion Paper

**BGD** 

12, 10719-10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Abstract Introduction

Conclusions References

Tables

Figures

I₹







Full Screen / Esc

Printer-friendly Version

Interactive Discussion



10765

and via predation to higher trophic levels (Jansson et al., 2007). Another consequence of thaw and increased DOM export is an increasing degree of net heterotrophy, i.e., a decrease in the ratio between gross primary production and community respiration (Ask et al., 2012). Heterotrophic bacteria benefit from fresh and high carbon inputs from the catchment and the high nutrient concentrations below the thermocline (Breton et al., 2009; Roiha et al., 2015). Respiration rates can be very high in thermokarst lakes, favoring rapid oxygen depletion and prolonged anoxia (Deshpande et al., 2015). This has implications for GHG production and exchange with the atmosphere (see Sect. 3.5).

The special conditions in many thermokarst lakes with long periods of anoxia under the ice (8 months in winter) and under the thermocline (up to 4 months in summer in some lakes, Laurion et al., 2010; Deshpande et al., 2015) exclude fish from the majority of these lakes. Increased rates of carbon import and decreased  $O_2$  concentrations with further permafrost degradation may particularly affect zooplankton and macroinvertebrate communities in the future, which are currently very abundant in oxygen-rich surface waters of thermokarst lakes (Rautio et al., 2011). Further,  $CH_4$ -oxidizing bacteria, found as relatively abundant in many stratified thermokarst lakes (Rossi et al., 2013; Crevecoeur et al., 2015), may play an important role in the carbon transfer through the food web. These bacteria are known to occur in environments where both oxygen and  $CH_4$  are available, and they have been suggested to contribute to the zooplankton diet (Jones, 2000; Kankaala et al., 2006). Permafrost thaw may stimulate this C pathway.

#### 4.3.2 Streams

In Sect. 2.2 we describe how permafrost thaw is likely to affect the delivery of sediment and nutrients to stream ecosystems. Sediment and nutrient concentrations are two of the fundamental factors that influence the structure and function of streams. However, as in lakes, there is a trade-off between the negative effects (smothering, shading, and scouring) caused by elevated levels of sediment loading vs. the positive effects (fertilization) caused by elevated concentrations of soluble nutrients. In the short-term

**BGD** 

12, 10719–10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Abstract Introduction

Conclusions References

Title Page

Tables Figures

4

Back Close

Full Screen / Esc

Printer-friendly Version



and locally, if TEFs intersect a stream they are likely to have significant impacts caused by massive inputs of sediment. While in some systems, TEFs have impacts that can be seen over broad catchment scales (e.g., Kokelj et al., 2012), in others, the long-term and regionally-averaged effect of TEFs on suspended sediments may be relatively small (see Sect. 2.2). At the same time, ongoing active layer deepening (and long-term, deep impacts from older TEFs) may add low levels of nutrients to arctic streams over longer time scales. Currently there is very little literature to support these potential impacts.

Even subtle increases in the loading of limiting nutrients can have profound impacts on highly-oligotrophic, arctic stream ecosystems. A whole-ecosystem nutrient fertilization experiment on the Kuparuk River has shown that long-term (30 year), low-level increases in soluble reactive phosphorus alone can have important influences on benthic autotrophic and macroinvertebrate community structure and can significantly increase primary and secondary production (Bowden et al., 1994; Cappelletti, 2006; Peterson et al., 1985). On the other hand, sediment loading may offset the stimulatory effects of introduced nutrients and interfere with benthic stream structure and function. Adverse effects of sediment influx to streams include damage to primary producers, especially from scour during storms, which can reduce primary production and ecosystem respiration. Increased sediment loading may clog the interstices among streambed particles, which could reduce the connectivity between the hyporheic zone and surface waters, interfering with exchange of nutrients and dissolved oxygen (Kasahara and Hill, 2006). Sediment loading may also lead to instability on the stream bottom, affecting the ability of benthic macroinvertebrates to establish and feed (Uehlinger and Naegeli, 1998).

Recent studies have evaluated the higher order effects of sediment and nutrient loading from thermokarst and detected significant impacts on some aspects of the biological function of receiving waters. Daily rates of riverine production and respiration decreased by 63 and 68 %, respectively, in the Selawik River in northwest Alaska in response to elevated turbidity levels that increased by several orders of magnitude below

BGD

12, 10719–10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Abstract Introduction

Conclusions References

Tables Figures

I4 PI

Back Close

Full Screen / Esc

Printer-friendly Version



Abstract Conclusions

References

Introduction

**BGD** 

12, 10719–10815, 2015

**Reviews and** 

**Syntheses: Effects of** 

permafrost thaw

J. E. Vonk et al.

Title Page

**Tables** 

**Figures** 

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



a massive thaw slump (Calhoun, 2012). Larouche et al. (2015) studied biogeochemical characteristics of a tundra stream on the North Slope, Alaska over a period of three summer seasons (2009–2011) after a gully thermokarst feature impacted this stream several years earlier (2005). They found that 4-6 years after the initial disturbance the TEF still caused modest increases in the loading of sediment and dissolved solutes. Furthermore, rates of ecosystem production and respiration and benthic chlorophyll a in the impacted reach of this stream, were significantly lower during the driest of the three summers. Rates of ammonium and soluble reactive phosphorus uptake were consistently lower in the impacted reach.

Benthic macroinvertebrates are typically the dominant vector of energy flow in lotic systems, connecting primary production to higher trophic levels (e.g., Hynes, 1970; Merritt et al., 2008). These communities are generally patchy and are sensitive to minor disturbance regimes (Lake, 2000). Allochthonous sediment has been shown to significantly impact habitat composition, leading to profound effects on the distribution of individual organisms (Lenat et al., 1981; Parker and Huryn, 2006). Shifts in community structure can indicate event severity, given that benthic macroinvertebrate diversity and overall community composition are strongly related to stream ecosystem structure and function (Carter et al., 2006; Vannote et al., 1980). In the study conducted by Larouche et al. (2015) initial macroinvertebrate richness and diversity was low but increased late in the season (August). Overall, the shifts in community structure were subtle.

Overall, it appears that arctic headwater streams may be resilient and regain considerable functionality as local disturbance features begin to repair, particularly in the case of smaller features such as gulley thermokarst disturbance that experience stabilization by re-vegetation (Jorgenson et al., 2006). While the acute impacts of slumping are obvious and notable, it is the chronic impacts of long-term, elevated nutrient and sediment loading that are of greater interest. There is growing evidence that subtle differences in sediment and nutrient delivery to arctic headwater streams can still be apparent many years after disturbance, and that thermokarst slumping may significantly affect primary producer biomass, benthic organic nutrients, benthic invertebrate community structure

and key ecosystem functions such as whole-stream metabolism and nutrient update. Averaged over thaw-impacted landscapes as a whole, these effects may often be subtle (Larouche et al., 2015). However, it is less clear how long-term nutrient and sediment loading will change in thaw-impacted stream systems. Certainly, there is the possibility of large and important shifts in community structure and function if measured changes in sediment and nutrient flux do occur.

## 4.4 Export to ocean

Permafrost degradation can lead to clear changes in the biogeochemical flux of constituents from land to water (see Sect. 2.2). Beyond the immediate site of impact, however, this constituent flux can be expected to have effects that range well into downstream environments. In many cases, changes on land can have clear impacts on coastal ocean processes, as has been shown within the estuarine zones of many large, southern rivers (Bianchi and Allison, 2009). In permafrost-impacted systems, understanding how changes in constituent flux at the terrestrial—aquatic interface will translate to changing export to the Arctic Ocean is still a challenging task. In this section, we describe the potential effects of permafrost thaw on land to ocean constituent flux, highlighting some major knowledge gaps in our current understanding of this process.

In the absence of direct, long-term measurements at river mouths, scaling changes that are being observed at the small catchment scale to changes in ocean-ward transport requires an understanding of the rates of deposition, uptake, and decomposition of various thaw-released constituents, both in absolute terms and relative to their non-permafrost derived counterparts. These rates will vary considerably among constituent types. For example, nutrients released as a result of permafrost thaw may be taken up rapidly following release to aquatic environments (Bowden et al., 2008), while major ions released following the exposure of mineral soils are largely conservative, and can be detected far downstream (Kokelj et al., 2013; Malone et al., 2013). Similarly, in some regions, sediment pulses associated with thermokarst disturbances decline markedly

**BGD** 

12, 10719–10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

. . .



4



Back



Full Screen / Esc

Printer-friendly Version













Printer-friendly Version

Interactive Discussion



with movement downstream (Bowden et al., 2008), while in others, thaw-associated sediment signatures can be across broad, catchment-wide scales (Kokelj et al., 2013), and the signature of permafrost thaw-origin sediments and particulate organics has been detected at the mouths of several large, arctic rivers (Guo et al., 2007; Gustafs-<sub>5</sub> son et al., 2011), and in increasing concentrations in some arctic coastal sediments (Feng et al., 2013).

Where direct observational evidence does exist to indicate changing biogeochemical flux at the mouths of arctic rivers, it appears that some constituents are changing relatively synchronously across arctic regions, while others may show significant regional differences in their trends. For example, flow-weighted bicarbonate fluxes appear to be increasing modestly throughout several arctic regions, including at the mouths of the Yukon (Striegl et al., 2005), and Mackenzie (S. Tank, unpublished data) Rivers. These trends are further corroborated by studies examining variation in riverine bicarbonate flux across arctic watersheds with differing permafrost coverage (Tank et al., 2012a, c). In contrast, the downstream transport of DOC to coastal areas appears to be increasing in some, but decreasing in other, regions, based both on direct river-mouth measurements over time and sub-watershed comparisons across permafrost gradients (Kawahigashi et al., 2004; Striegl et al., 2005; Tank et al., 2012a; S. Tank unpubl. data for the Mackenzie catchment). Furthermore, DOC originating in old permafrost may be preferentially degraded within stream networks, and thus may not be detectable at the river mouth (Spencer et al., 2015). It therefore remains difficult to attribute the few documented changes at the mouths of arctic rivers to either up-catchment permafrost degradation or, for example, to the more widespread effects of changing temperature and precipitation patterns. For example, fluxes of DOC will be affected by changes in the composition and overall production and decomposition of vegetation (e.g., Laudon et al., 2012), in addition to the exposure of organic soils via permafrost thaw. Similarly, fluxes of bicarbonate will be affected not only by the thaw-mediated exposure of deeper mineral soils, but also by increases in root respiration that affect weathering rates (Beaulieu et al., 2012). For many constituents, changes in the seasonality of pre-

12, 10719–10815, 2015

**BGD** 

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Title Page Introduction

**Abstract** 

References

Conclusions

**Tables** 

**Figures** 







cipitation may have a resultant effect on constituent flux and concentration (Spence et al., 2011). Research to explore how permafrost thaw affects aquatic biogeochemistry across nested spatial scales, and to further elucidate the mechanisms of changing chemistry at scales where direct, mechanistic, observations are possible, will greatly aid our ability to understand how up-catchment permafrost degradation affects biogeochemical flux to the coastal ocean.

The effect of changing riverine fluxes on coastal ocean processes cannot be considered without also considering thaw-induced changes in coastal erosion. Rates of coastal erosion vary by region, in part because of regional differences in ground ice content and bluff height (Lantuit et al., 2013). Overall, however, the impact of coastal erosion on biogeochemical flux to the ocean appears to be significant. Estimates of carbon release by coastal erosion vary significantly, and range between 5–14 Tg OC yr<sup>-1</sup> for the entire Arctic combined (Rachold et al., 2004; Jorgenson and Brown, 2005; Vasiliev et al., 2005; Couture, 2010; Vonk et al., 2012b) with highest delivery rates in the Laptev and East Siberian Seas (Vonk et al., 2012b). This value is nearly a third of the combined, circum-arctic, estimated delivery of DOC (34 Tg; Holmes et al., 2012) and POC (6 Tg; McGuire et al., 2009) via rivers each year. Rates of coastal erosion, however, appear to be increasing both in the Russian and the Alaskan Arctic (Günther et al., 2013; Jones et al., 2009) which is due to decreasing sea ice content, allowing for higher storm intensity and wave fetch, along with increasing summertime sea surface temperature and a rising sea (IPCC, 2013).

Documented changes in riverine biogeochemistry and coastal erosion rates may in turn have a significant effect on carbon and nutrient cycles in the near shore ocean. Where the delivery of DOC and POC increase, the attenuation of light will reduce the photosynthetic uptake of  $CO_2$  (Retamal et al., 2008), while increases in both bacterial production, and the decomposition of this organic matter to carbon dioxide  $CO_2$  may result. On the East Siberian Shelf, for example, large zones of  $CO_2$  outgassing have been shown to occur alongside plumes of DOC that have a clear terrestrial isotopic signature (L. G. Anderson et al., 2009a). In addition, increases in light caused by sea ice

**BGD** 

12, 10719–10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Abstract Introduction

Conclusions References

Tables Figures

4 >

Back Close

Full Screen / Esc

Printer-friendly Version



retreat could also increase photochemical degradation of riverine DOM to CO<sub>2</sub> (e.g., Tank et al., 2012b). If changing delivery of organics does affect coastal CO<sub>2</sub> saturation, this could combine with changes in bicarbonate flux to have a significant impact on nearshore aragonite saturation, compounding the effects of temperature and sea 5 ice melt on ocean acidification in the Arctic (Steinacher et al., 2009; Yamamoto-Kawai et al., 2009). In much of the Russian Arctic, organic carbon transport from land to ocean is already high (Holmes et al., 2012; Vonk et al., 2012b) and rivers are relatively bicarbonate-poor (Tank et al., 2012c). Thus, increasing delivery of organics to coastal zones, and the resultant CO<sub>2</sub> production, could further reduce aragonite saturation in a region where near shore regions are already poorly buffered (Anderson et al., 2011; Tank et al., 2012c). This effect may vary regionally, however. Although aragonite undersaturation has been detected in river water plumes of the North American Arctic (Mathis et al., 2011), many rivers in the North American Arctic are better buffered than their Siberian counterparts (Tank et al., 2012c), and in some cases are experiencing declining discharge-normalized fluxes of DOC (Striegl et al., 2005). For these, and other constituents such as nutrients, we still have much to learn about how fluxes to the coastal ocean are changing, and how this change may affect near shore biogeochemical function (Le Fouest et al., 2013; Letscher et al., 2013; Tank et al., 2012b). Understanding the effect of these changing fluxes in general, and the specific importance of permafrost thaw for changing biogeochemistry in the coastal Arctic, remains

#### 5 Summary, feedbacks and future research needs

a clear priority for future research.

Permafrost thaw has a broad range of effects on the functioning of aquatic ecosystems. These include changes in optical and thermal properties, altered chemistry of the water column and of sediments, changes in contaminant loads, and altered potential for bio-and photo-degradation, which in turn affect gas fluxes, carbon burial and export of thaw-released constituents downstream and to the coastal ocean. In addition, these

BGD

12, 10719–10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Abstract Introduction

Conclusions

Tables Figures

I₫

•

Close

Back

Full Screen / Esc

Printer-friendly Version



changes affect microbial communities and processes, primary production, and trophic structure in thaw-impacted systems.

In addition to these effects, thermokarst can be expected to initiate important feed-backs to climate change, which will often result from processes that range beyond those directly impacting aquatic systems. Below, we present some summary thoughts, discuss broad scale feedbacks to climate, and provide an assessment of future research needs based on the summary presented in this review.

## 5.1 Summary

Ground-ice content, topography and soil type are the main drivers for both types of permafrost thaw (press vs. pulse) and associated release of constituents into aquatic systems: (i) When thaw is manifested as a pulse disturbance, this leads to thermokarst lakes (lowland terrain) or slumping (hillslope terrain). For both features, the soil type of the pulsed material generally determines the release and effect of constituents; a pulse of OC-rich soils will color thermokarst lakes (affecting stratification etc.), will cause low transparency (high turbidity) in aquatic ecosystems after thaw, and will lead to increasing OC input. A pulse of mineral-rich soils, however, might lead to clearer thermokarst lakes and decreasing OC in the water column, due to sorption of matter to mineral surfaces. (ii) When thaw is manifested as a press disturbance, this generally leads to longer flow paths and increasing residence time in soils. Here, the soil type of the thawed material generally determines the release and effect of constituents; thaw of OC-rich soils will lead to higher OC input whereas thaw of mineral-rich soils will lead to lower OC inputs.

The fate of released constituents and their feedbacks to climate depends on the propensity of these constituents for degradation vs. burial, which is determined by environmental parameters as well as intrinsic properties. Also, it is important to distinguish between thaw-mobilization of old permafrost OC vs. contemporary OC, when considering the feedback potential. The degradability of released OC, representing the most direct carbon-climate "linkage", can be divided into a dissolved and a particu-

**BGD** 

12, 10719–10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Abstract

Conclusions

References

Introduction

Tables

Figures

1.9....











Printer-friendly Version



Back

Printer-friendly Version

Interactive Discussion



late fraction; the biodegradability of DOC is determined by source, chemical character, nutrient availability, temperature and prior microbial and photochemical processing. Furthermore, DOC is generally more degradable when flushed from continuous permafrost regions, surface litter, active layer soils, and yedoma, but less degradable when flushed from deeper mineral soil layers. Photodegradation of DOC is relatively high in thaw-impacted systems that are shallow, rich in light-absorbing DOM, or that undergo short-term (days to weeks) stratification events. Photodegradation can however be hampered by slumping of OC-rich soils (decreasing transparency) or slumping of mineral-rich soils (adsorbing CDOM). Our understanding of degradation of POC and the factors influencing it is still remarkably poor. Burial of OC is generally lower than total OC remineralization to CO<sub>2</sub> and CH<sub>4</sub>, particularly in large and shallow lakes, but can be higher in small and deep lakes, thermokarst-yedoma lakes, and on the coastal shelf.

There are few studies on the effects of permafrost degradation on the ecology and food web structure of aquatic ecosystems. Thus, we can only speculate about future ecological conditions based on our current understanding of the function of high latitude systems and the effects of permafrost degradation on the physics and chemistry of impacted systems. The impact of permafrost thaw on foodwebs and ecosystem functioning appear likely to be driven by changes in the inputs of nutrients, DOM and sediments. Primary production is stimulated by nutrient input but can be hindered by light suppression following increasing input of CDOM or OC-rich sediments. particularly when DOM concentrations are great enough that the positive effect of increasing organic nutrients is overwhelmed by the negative effect of decreasing light penetration (Seekell et al., 2015). Benthic communities can be destabilized by high sediment loading but may thrive when slumping of mineral-rich sediment leads to increasing water clarity. Overall, food web changes may lead to shifts in (i) OC resources supporting higher consumers, and (ii) the net heterotrophy of systems. Increasing terrestrial DOM input may be transferred to zooplankton and zoobenthos via increased heterotrophic

#### **BGD**

12, 10719–10815, 2015

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Title Page Introduction **Abstract** Conclusions References

> **Tables Figures**

Full Screen / Esc

Close

bacterial production. On the other hand, in response to elevated turbidity levels, production and respiration may decrease.

#### 5.2 Climate feedbacks

Permafrost thaw enables important feedbacks to climate through activating and remobilizing previously frozen carbon pools (Schuur et al., 2015) that generate fluxes of CO<sub>2</sub> and CH<sub>4</sub> (a net climate warming) or generate increasing accumulation in sedimentary basins (a net climate cooling). There are, however, more climate feedbacks that, while not directly related to aquatic processes, do have an important effect on the functioning of these systems. For example, thermokarst-enabled increases in inundated landscape area can be followed by loss of tree cover in the area surrounding thaw depressions, and a shift towards sedge-dominated fen vegetation (e.g. Jorgenson et al., 2001). The resulting change in albedo could subsequently affect regional radiative forcing, in a direction which will depend upon the manner in which vegetation changes affect snow cover (Notaro and Liu, 2007), and the relative change in cover of peat, forest, and water, because of the differences in albedo between these land cover types (e.g. Lohila et al., 2010). Additionally, energy partitioning into latent and sensible heat fluxes may be altered significantly upon thaw, as lake surfaces would be increasingly more important in certain regions, creating a distinct microclimate with high evapotranspiration and low sensible heat flux (e.g. Rouse et al., 2005). Lake energy balance varies widely with depth (e.g. Eaton et al., 2001), adding importance to the temporal changes in thermokarst lake sizes. Finally, thermokarst-enabled changes in the emission of biogenic volatile organic compounds (BVOCs) through landscape shifts could also affect regional climate, because secondary aerosols originating from BVOCs facilitate cloud formation (e.g. Ehn et al., 2014). Taken as a whole, permafrost thaw and the formation of thermokarst landscapes will produce diverse and contradirectional climatic effects at regional and global scales. An approach that considers these multiple effects is therefore needed to understand how thermokarst feeds back to regional and global climates.

BGD

12, 10719–10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

Figures

I₫











Full Screen / Esc

Printer-friendly Version



Aguatic ecosystems are widely recognized as locations of active processing and burial of the organic matter they receive (Cole et al., 2007; Battin et al., 2009), and, lately, also receive more attention in climate-carbon interactions in the Arctic (e.g. Sobek <sub>5</sub> et al., 2003; Feng et al., 2013; Vonk and Gustafsson, 2013; Olefeldt and Roulet, 2014). However, in this review we have also identified numerous gaps in our knowledge of the diverse effects of permafrost thaw on aquatic ecosystems and the consequential feedbacks on climate. We therefore make the following recommendations for future research directions, where we make the division into general directions, directions specific to streams and rivers, directions specific to thermokarst lakes, and the use of specific techniques:

#### General future research directions 5.3.1

Fluxes and degradation of particulate OC from thawing permafrost. Permafrost thaw. particularly when manifested as a pulse disturbance, can deliver substantial POC inputs to aquatic systems by exposing and rapidly thawing deep permafrost deposits. However, studies focusing on OC fluxes from thawing permafrost have to date mostly focused on DOC. Whereas DOC from collapsing permafrost is among the most biodegradable reported in natural systems (Vonk et al., 2013a; Abbott et al., 2014; Spencer et al., 2015), the biodegradability of POC, released in far larger quantities, has never been properly assessed.

The relative mobilization and degradability of old vs. contemporary carbon. As thawing permafrost increases shoreline contact between lakewater and soils, and increases direct slumping into lake and stream systems, the carbon that is introduced to aquatic systems will be from both shallow, contemporary, soil layers and from older, permafrost soils. Although permafrost DOC derived from yedoma appears to be highly degradable (see Sect. 3.1), there is also evidence that some thermokarst lakes emit modern carbon (see Sect. 4.1.2). Understanding the relative susceptibility of OC pools with

Paper

Discussion Paper

**BGD** 

12, 10719–10815, 2015

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Discussion Paper

Discussion Paper Printer-friendly Version Interactive Discussion



Title Page **Abstract** Introduction Conclusions References **Tables Figures** Back Close Full Screen / Esc

10776

BGD

12, 10719-10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Abstract Introduction

Conclusions References

Tables Figures

I ◆ ▶I

◆ ▶ Back Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



permafrost-origin vs. contemporary-origin to bio- and photodegradation, and the relative mobilization of these two pools as a result of permafrost thaw across various regions and aquatic ecosystems, will help our ability to quantify feedbacks to climate in thaw-impacted systems. The priming effects generated by, for example, light and photosynthetic exudates on the consumption of old OC also needs to be further explored.

Influence of permafrost thaw on fluxes to coastal ocean. Our understanding of the effect of changing constituent fluxes following permafrost thaw on the optical characteristics, primary production and biogeochemistry of coastal arctic systems is still limited and remains a clear priority for future research.

Resiliency of stream and lake ecosystems to direct thermokarst impacts. Very few studies have reported on the intensity or the duration of thermokarst impacts on the structure of biological communities or the function of key ecological processes (e.g., photosynthesis, respiration, and nutrient uptake) in arctic streams (Calhoun, 2012; Larouche et al., 2015). It is impossible, therefore, to do more than speculate about how the ecosystem services provided by arctic streams and rivers are changing in response to this regional impact of climate change.

Ecosystem structure and functioning. The structure and functioning of arctic aquatic ecosystems remains poorly studied, particularly related to impacts of permafrost thaw on ecosystem scale processes. For example, resource use and growth by consumers in thermokarst lakes has not been quantified to date. Long-term effects of nutrient and sediment loading in thaw-impacted stream systems are still understudied, but are vital for effects on and shifts in receiving foodwebs.

*Microbial diversity and processes.* Microbial diversity studies have only recently begun, and there are many gaps in understanding. For example, the diversity and roles of viruses, likely the biologically most abundant particles in thaw waters, have not received attention to date. The composition of winter microbial communities in ice-covered thaw lakes is at present unknown, and only minimally studied in rivers (Crump et al., 2009) and the microbial processes operating under the ice have been little explored. These deserve special attention, given the long duration of ice-cover in northern lakes, and

the evidence of prolonged anoxia in these waters that favor anaerobic processes such as methanogenesis. The spring period of ice melt and partial mixing, and the prolonged period of mixing in fall, may be important for gas exchange as well as key aerobic microbial processes such as methanotrophy, and these transition periods also require closer study.

Improved assessment of underwater UV irradiance. To specifically understand photodegradation of both old and contemporary DOM in thaw waters (see also bullet above), we further recommend work to (i) quantify UV spectral irradiance in thawimpacted water columns, (ii) understand the residence time of DOM in the UV-exposed portion of the water column and its variability with changing mixing regimes, and (iii) identify the factors that control vertical losses of DOM and the lability of permafrost DOM to photo-degradation.

#### 5.3.2 Research direction specific to streams and rivers

Watersheds of small and intermediate size. Research to date has been somewhat skewed towards large rivers basins and estuaries; small watersheds and headwater streams where processes and change are easier to elucidate remain under-studied, meaning that it is often difficult to link measurements to clear source areas or processes.

Sediment erosion vs. delivery to streams. Streambank erosion effectively delivers 100 % of eroded sediments to the streams. But TEFs that form at some distance from streams may deliver far less (down to zero) sediment mass, C, N, and P to streams. To scale the effects of hillslope thermokarst to aquatic systems at broad spatial scales, we must better quantify how the position of various TEF features in the landscape moderates their effect on aquatic ecosystems.

Influences of hyporheic processes. It is well known that the hyporheic zone (region below and alongside stream) contributes substantially to nutrient and carbon processing in temperate and tropical streams (e.g., Boulton et al., 1998). Recent research has shown that despite the presence of permafrost, the hyporheic zone is equally important

BGD

12, 10719–10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Introduction

Conclusions References

Tables Figures

•

Back

**Abstract** 

Full Screen / Esc

Close

Printer-friendly Version



to the ecological functions of arctic streams (e.g., Zarnetske et al., 2008). However, we do not know how thermokarst impacts will affect hyporheic processes or vice versa.

#### 5.3.3 Research directions specific to thermokarst lakes

Thermokarst lake processes in non-yedoma systems. The emission and burial of thaw-released OC in yedoma thermokarst lakes has been a relatively large focus of research attention (e.g. Walter Anthony et al., 2007, 2014; Sepulveda-Jáuregui et al., 2015), but the fate of thaw-released OC in thermokarst lakes in non-yedoma regions is still understudied. Considering that the yedoma region holds 210–456 Pg C and the total permafrost region holds 1330–1580 Pg C (Schuur et al., 2015), this makes non-yedoma regions holding 66–87 % of the total permafrost important to consider, particularly where non-yedoma regions are also lake-rich.

Physical and hydrological dynamics of thermokarst lakes. Current gaps in our understanding of the physical and hydrological dynamics of thermokarst lakes include the quantification of sediment heat transfer, penetration of solar radiation through ice cover, modeling of diffusive GHG exchange in small lakes, wave and energy dynamics associated with floating ice, and the extent of groundwater flow (Kirillin et al., 2012). Furthermore, material transport caused by advective water transfer between shallow zones to bottom waters such as found in MacIntyre and Melack (1995) in other systems has not been addressed in thaw waters.

Hydrodynamic effects of high CH<sub>4</sub> and CO<sub>2</sub> concentrations. Gradients in gas concentrations (particularly during ice-covered periods) can cause density differences in the water column that can modify stratification such as suggested by Deshpande et al. (2015) but these effects have been little studied to date. Also, the effects on stratification by gas bubble trains associated with ebullition from sediments (Walter et al., 2006) have received little attention.

Role of  $CH_4$  oxidation in thermokarst systems. The emission of  $CH_4$  from aquatic ecosystems is significantly offset by microbial oxidation of  $CH_4$  (Trotsenko and Murrell, 2008). For example, in northern lakes, up to 88% of  $CH_4$  produced in sediments is

**BGD** 

12, 10719–10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Abstract Introduction

Conclusions References

Tables Figures

Back Close

Full Screen / Esc

Printer-friendly Version



oxidized by microbes (e.g., Bastviken et al., 2008) and abundant methanotrophs have been observed in thermokarst lakes. Oxidation of CH<sub>4</sub> has recently been detected through laboratory incubation studies of thermokarst lakes in the boreal and tundra zones of Alaska (Martinez-Cruz et al., 2015), however, numerous questions remain to 5 be answered with respect to (i) the extent to which CH<sub>4</sub> oxidation offsets whole-lake emissions in thermokarst-lake systems, (ii) which CH<sub>4</sub>-carbon pools are subject to oxidation (contemporary vs. old carbon), (iii) microbial community dynamics, and (iv) biogeochemical and ecological controls over CH<sub>4</sub> oxidation among different thermokarst lake types.

Lake carbon burial. Our knowledge of the relative role of burial vs. processing in northern lakes remains poor. Tranvik et al. (2009) project that carbon burial in polar lakes will decrease whereas carbon burial in boreal lakes will increase. This review, however, points out that other factors such as permafrost type (yedoma vs. nonvedoma; Walter Anthony et al., 2014) or lake shape (small and deep vs. large and shallow; Ferland et al., 2012) strongly affect burial efficiencies and may overrule the distinction between boreal and polar regions. More research is needed to shed light on these processes.

#### 5.3.4 The use of specific techniques in future research

Usage of high-resolution automated loggers in thermokarst lakes. Many thermokarst lakes undergo rapidly cycling stratification events (i.e. diurnal or several day) that are hard to capture with sparse measurements. For example, strong anomalies in air temperature can result in the formation of strong temperature gradients (Pokrovsky et al., 2013). The increased use of high resolution, automated temperature and O<sub>2</sub> loggers is likely to yield new insights into short term stratification and mixing dynamics, even in those lakes currently considered to be well mixed in summer.

Remote sensing. We recommend increasing usage of high-resolution satellite remote sensing to assess (i) the changing areal coverage of thermokarst lakes in both discontinuous and continuous permafrost regions as well as (ii) changing DOC lake

**BGD** 

12, 10719–10815, 2015

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Title Page **Abstract** Introduction

Conclusions References

**Tables Figures** 

Back Close Full Screen / Esc

Printer-friendly Version



**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Title Page **Abstract** Introduction Conclusions References **Tables Figures** Back Close Full Screen / Esc

Printer-friendly Version

Interactive Discussion

concentrations (e.g. Watanabe et al., 2011) derived from changing lake surface color as a result of permafrost thaw.

Radiocarbon dating. The arctic aquatic system should provide an early and sensitive signal of change in the cycling of OC in the terrestrial environment. The develop-5 ment of new direct methods to date aquatic dissolved CO<sub>2</sub> (Billett et al., 2012; Garnett et al., 2012) has significantly increased our capacity to measure the source and age of CO<sub>2</sub> released from arctic landscapes; these along with existing dating tools for POC, POC and CH<sub>4</sub>, provide researchers with a strong methodological basis to quantify and detect the release of aged C into the aquatic environment. This will allow us to detect change or rates of change in areas of the arctic undergoing differential rates of climate warming and address the key issue of whether "old" carbon (fixed 100s or 1000s of year BP) is being released directly or indirectly into the atmosphere.

Eddy correlation flux measurements on thermokarst lakes. Given its general applicability for studying lake-atmosphere exchanges of carbon (e.g. Vesala et al., 2012), we recommend increasing the application of eddy covariance on permafrost thermokarst lakes. We suggest that particular attention be paid to the following challenges:

1. Eddy flux footprint analysis. Because the flux footprint will often consist of a mixture of terrestrial and aquatic fluxes (Wille et al., 2008), it is important to use an appropriate footprint model (Vesala et al., 2008), preferably supplemented with localized flux measurements situated within the larger footprint to successfully interpret eddy flux dynamics (Pelletier et al., 2014; Sachs et al., 2010).

20

2. Usage of low-maintenance instrumentation. Robust, low power, open-path and enclosed-path gas analyzers for CO<sub>2</sub> and CH<sub>4</sub> (Burba et al., 2012) that require minimal maintenance have recently been developed, offering new opportunities for quasi-continuous gas flux measurements in remote locations. Given a sufficient power supply on-site, robust, field deployable closed-path gas analyzers are also suitable for continuous eddy flux measurements (e.g. Vesala et al., 2006).

12, 10719–10815, 2015

3. Harmonized data processing protocols. Past efforts to compile datasets from large, terrestrial eddy covariance networks, such as FLUXNET (Baldocchi et al., 2001) or CarboEurope (Papale et al., 2006), have shown the importance of consistent data processing protocols to ensure comparability between sites. Processing protocols should be revised for application over lakes due to significant differences in surface processes of aquatic and terrestrial ecosystems (e.g. Vesala et al., 2012). Given the wide range of thermokarst lake sizes and types, a network of several flux towers has a great potential to better understand lake-atmosphere interactions of these ecosystems.

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BGD

12, 10719–10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

Figures

I◀



•



Back



Full Screen / Esc

Printer-friendly Version



- BGD
- 12, 10719-10815, 2015
- Reviews and Syntheses: Effects of permafrost thaw
  - J. E. Vonk et al.
- Printer-friendly Version
  - Interactive Discussion

Full Screen / Esc

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J. E. Vonk et al.

- Title Page

  Abstract Introduction

  Conclusions References

  Tables Figures

  I◀ ▶I
  - **◄** ► Back Close
  - Full Screen / Esc

Printer-friendly Version



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Close

Printer-friendly Version

Back

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Full Screen / Esc

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J. E. Vonk et al.

Title Page

Abstract Introduction

Conclusions References

Tables Figures

•

Close

Full Screen / Esc

Back

Printer-friendly Version

- 26,
- BGD
- 12, 10719–10815, 2015

# Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Printer-friendly Version

Full Screen / Esc

Close

Back

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- Title Page **Abstract** Introduction Conclusions References **Tables Figures** Back Close Full Screen / Esc
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Title Page Abstract Introduction

Conclusions References

> **Tables Figures**

Close

Back

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J. E. Vonk et al.

Title Page

Abstract Introduction

Conclusions References

Tables Figures

4

•

Back

Close

Printer-friendly Version

Full Screen / Esc



- 12, 10719–10815, 2015

**BGD** 

- **Reviews and Syntheses: Effects of** permafrost thaw
  - J. E. Vonk et al.
- Title Page **Abstract** Introduction Conclusions References **Tables Figures** Back Close
- Printer-friendly Version
  - Interactive Discussion

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- nic ob.
- 12, 10719–10815, 2015

**BGD** 

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

- Title Page

  Abstract Introduction

  Conclusions References

  Tables Figures

  I◀ ▶I
  - Close
  - Full Screen / Esc

Back

Printer-friendly Version



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**BGD** 

12, 10719–10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Abstract Introduction

Conclusions References

Tables Figures

•

Close

Back

Full Screen / Esc

Printer-friendly Version



- **BGD**
- 12, 10719–10815, 2015
- Reviews and Syntheses: Effects of permafrost thaw
  - J. E. Vonk et al.
- - Full Screen / Esc

    Printer-friendly Version

Close

Back

- Interactive Discussion
  - © **()**

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## Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

- Title Page

  Abstract Introduction

  Conclusions References

  Tables Figures

  I ← ►I

  Back Close
- Printer-friendly Version
  - Interactive Discussion

Full Screen / Esc

© O

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Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

- Title Page

  Abstract Introduction

  Conclusions References

  Tables Figures
  - **→**
- Back
- Close

Full Screen / Esc

Printer-friendly Version



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  - 12, 10719-10815, 2015
  - Reviews and Syntheses: Effects of permafrost thaw
    - J. E. Vonk et al.
  - Title Page

    Abstract Introduction

    Conclusions References

    Tables Figures

    I 

    I 

    I 

    Back Close
    - Printer-friendly Version

Full Screen / Esc

- Interactive Discussion
  - © BY

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- BGD
  - 12, 10719-10815, 2015
  - Reviews and Syntheses: Effects of permafrost thaw
    - J. E. Vonk et al.
  - Title Page

    Abstract Introduction

    Conclusions References

    Tables Figures

    I 

    I 

    Back Close

    Full Screen / Esc
    - Printer-friendly Version
    - Interactive Discussion
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**BGD** 

12, 10719–10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Abstract Introduction

Conclusions References

Tables Figures

4

Back Close

Full Screen / Esc

Printer-friendly Version



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12, 10719–10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Abstract Introduction

Conclusions References

Tables Figures

4

•

Close

Back

Full Screen / Esc

Printer-friendly Version

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12, 10719–10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Abstract Introduction

Conclusions References

Tables Figures

4

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



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12, 10719–10815, 2015

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Title Page

Conclusions References

> **Tables Figures**

**Abstract** 

Close

Introduction

Back

Full Screen / Esc

- **BGD** 12, 10719–10815, 2015
- **Reviews and Syntheses: Effects of** permafrost thaw
  - J. E. Vonk et al.
- Title Page Introduction **Abstract** Conclusions References **Tables Figures**
- Printer-friendly Version

Full Screen / Esc

Close

Back

- Interactive Discussion

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12, 10719-10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

- Title Page

  Abstract Introduction

  Conclusions References
  - onclusions
  - Tables Figures
  - 4
- Close
- Back
  - Full Screen / Esc

Printer-friendly Version



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**BGD** 

12, 10719–10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Abstract Introduction

Conclusions References

Tables Figures

l∢ ≻l

•

Close

Back

Full Screen / Esc

Printer-friendly Version

**Discussion Paper** 

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**BGD** 

12, 10719–10815, 2015

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Title Page

Introduction **Abstract** 

Conclusions References

> **Tables Figures**



Close

Back



Printer-friendly Version

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12, 10719–10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Abstract Introduction

Conclusions References

Tables Figures

4

Back

Close

Full Screen / Esc

Printer-friendly Version



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15

**BGD** 

12, 10719-10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Title Page

Abstract Introduction

Conclusions References

Tables Figures

I 

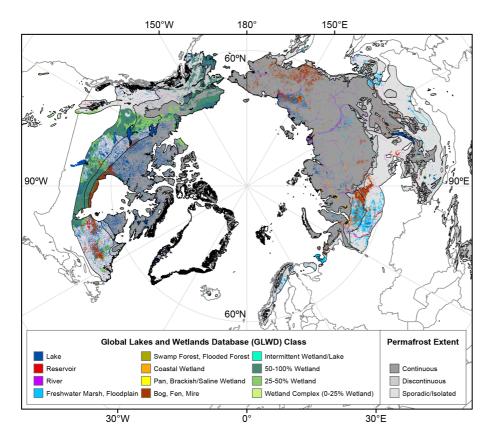
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Back Close

Full Screen / Esc

Printer-friendly Version





**Figure 1.** Map of the permafrost zones in the Northern Hemisphere (grey scale; Brown et al., 1998) superimposed on waterbodies from the Global Lakes and Wetlands Database (Lehner and Döll, 2004).

12, 10719-10815, 2015

Reviews and Syntheses: Effects of permafrost thaw

J. E. Vonk et al.

Printer-friendly Version



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Interactive Discussion



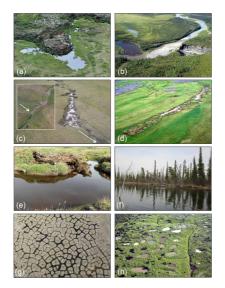


Figure 2. Photos of typical thermokarst processes: (a) Thermokarst lake SAS1, located in a subarctic peat bog near Kuujjuarapik-Whapmagoostui, Québec, Canada. The waterbody lies next to a thawing, organic-rich palsa (permafrost mound), and is 25 m in its maximum dimension. (b) Massive thaw slump on the Selawik River near Selawik, Alaska, US. Sediment discharge from the feature has entirely blocked the river. Note the turbidity downstream. (c) Gully thermokarst on the Toolik River, Alaska, US, and impact on receiving stream (inset), (d) active layer detachment slides near the Anaktuvuk River burn area on the North Slope of Alaska, (e) trough pond on Bylot Island, Nunavut, Canada featuring active erosion, (f) thermokarst lakes in Mackenzie Delta, Northwest Territories, Canada, showing active shoreline slumping, (q) polygonal landscape on Bylot Island, showing ice-wedge trough ponds, and (h) thermokarst lakes and ponds with a wide range in color near Kuujjuarapik-Whapmagoostui, Québec, Canada. Photo credits: (a), Bethany Deshpande; (b), Benjamin Crosby; (c, d), William Breck Bowden; (e), (g), and (h), Isabelle Laurion; (f), Suzanne Tank.

**BGD** 

12, 10719–10815, 2015

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Title Page

**Abstract** 

Introduction

Conclusions

References

**Tables** 

**Figures** 













Back

Full Screen / Esc

Printer-friendly Version



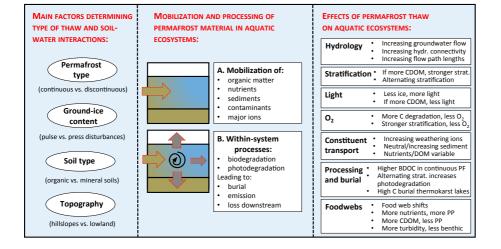


Figure 3. Conceptual diagram of (left) factors determining thaw type and soil-water interactions, (middle) mobilization and processing of permafrost material into aquatic ecosystems, and (right) effects of permafrost thaw on aquatic ecosystems.

## **BGD**

12, 10719-10815, 2015

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

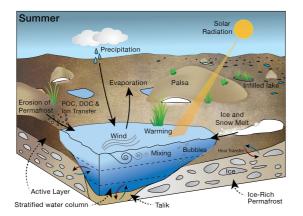
Title Page Introduction **Abstract** Conclusions References Tables **Figures** 



Close

Interactive Discussion





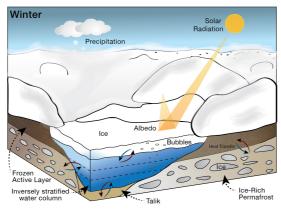


Figure 4. Physical limnological chracteristics of permafrost thaw lakes in summer and winter. A talik is a region of ground that remains unfrozen throughout the year as a result of heat transfer through the water column of lakes (or other freshwater bodies). A palsa is a raised mound of permafrost found in many subarctic and arctic wetland areas.

12, 10719-10815, 2015

**BGD** 

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Title Page

**Abstract** 

References

Introduction

**Tables** 

Conclusions

**Figures** 

Close

Back

Printer-friendly Version

Interactive Discussion



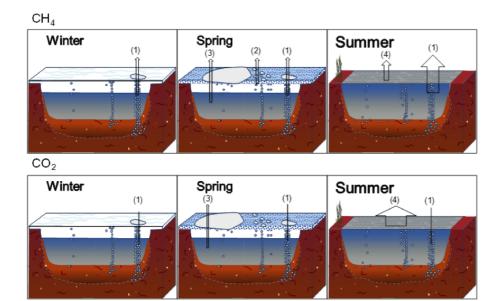


Figure 5. Schematic of CH<sub>4</sub> and CO<sub>2</sub> emission pathways during different seasons in thermokarst lakes. The thickness of arrows indicates the relative magnitude of contribution from each pathway according to a study of 40 Alaskan lakes (Sepulveda-Jáuregui et al., 2015): (1) Direct ebullition through ice-free hotspot seeps in winter and from all seep classes during the last month of ice cover in spring and in summer, (2) ice-bubble storage emission during spring ice melt, (3) Storage emission of dissolved gases accumulated under lake ice when ice melts in spring; (4) Diffusion emission from open water in summer. The background ebullition mode, discussed in the text, is not shown. The dashed line indicates the boundary between the thaw bulb under lakes and the surrounding permafrost. Figure modified from Sepulveda-Jáurequi et al., 2015.

## **BGD**

12, 10719–10815, 2015

**Reviews and Syntheses: Effects of** permafrost thaw

J. E. Vonk et al.

Title Page **Abstract** Introduction

Conclusions References

> **Tables Figures**



Full Screen / Esc