

This discussion paper is/has been under review for the journal Biogeosciences (BG).  
Please refer to the corresponding final paper in BG if available.

# Concentration maxima of volatile organic iodine compounds in the bottom layer water and the cold, dense water over the Chukchi Sea in the western Arctic Ocean: a possibility of production related to degradation of organic matter

A. Ooki<sup>1</sup>, S. Kawasaki<sup>1</sup>, K. Kuma<sup>1</sup>, S. Nishino<sup>2</sup>, and T. Kikuchi<sup>2</sup>

<sup>1</sup>Faculty of Fisheries Sciences, Hokkaido University, Hakodate, Japan

<sup>2</sup>Institute of Arctic Climate and Environment Research, Japan Agency for Marine-Earth Science and Technology, Yokosuka, Japan

Received: 28 April 2015 – Accepted: 16 June 2015 – Published: 17 July 2015

Correspondence to: A. Ooki (ooki@fish.hokudai.ac.jp)

Published by Copernicus Publications on behalf of the European Geosciences Union.

**BGD**

12, 11245–11278, 2015

**Volatile organic iodine compounds in seawater related to degradation of organic matter**

A. Ooki et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

## Abstract

We conducted a shipboard observation over the Chukchi Sea and the Canada Basin in the western Arctic Ocean in September and October 2012 to obtain vertical distributions of four volatile organic iodine compounds (VOIs) in seawater. ~~VOIs are believed to play a role in ozone destruction in the troposphere and lower stratosphere.~~ The VOIs observed in this study were iodomethane ( $\text{CH}_3\text{I}$ ), iodoethane ( $\text{C}_2\text{H}_5\text{I}$ ), diiodomethane ( $\text{CH}_2\text{I}_2$ ) and chloriodomethane ( $\text{CH}_2\text{ClI}$ ). Maximum concentrations of the four VOIs were found in the bottom layer water over the Chukchi Sea shelf, in which layer the maximum concentration of ammonium ( $\text{NH}_4^+$ ) also occurred. A significant correlation was observed between  $\text{C}_2\text{H}_5\text{I}$  and  $\text{NH}_4^+$  (correlation coefficient  $R = 0.93$ ) and between  $\text{CH}_3\text{I}$  and  $\text{NH}_4^+$  ( $R = 0.77$ ), suggesting production of these VOIs increased with degradation of organic matter. Over the ~~northern Chukchi Sea shelf-slope area~~, concentration maxima of  $\text{CH}_2\text{I}_2$ ,  $\text{CH}_2\text{ClI}$ , and  $\text{CH}_3\text{I}$  were found in the subsurface cold, dense water (CDW). A large nitrogen deficit ( $\text{N-deficit} = \text{NH}_4^+ + \text{NO}_3^- + \text{NO}_2^- - 16\text{PO}_4^{3-}$ ) simultaneously occurred in this water, suggesting production of the three VOIs in the sediment or the bottom layer water over the shelf, probably in association with degradation of organic matter. We conclude that VOI production over the Chukchi Sea shelf can be largely attributed to the degradation of organic matter that is produced in the highly productive shelf water. High concentrations of  $\text{CH}_2\text{ClI}$  were also found in the Alaskan Coastal Water (ACW) from the Bering Strait to the surface of the northern Chukchi slope. The VOIs that originated at the Chukchi Sea shelf are expected to be laterally transported to the Arctic Ocean Basin ~~through the subsurface CDW and the surface ACW.~~

## 1 Introduction

Volatile organic iodine compounds (VOIs) such as iodomethane ( $\text{CH}_3\text{I}$ ), iodoethane ( $\text{C}_2\text{H}_5\text{I}$ ), diiodomethane ( $\text{CH}_2\text{I}_2$ ) and chloriodomethane ( $\text{CH}_2\text{ClI}$ ) are known to pro-

BGD

12, 11245–11278, 2015

**Volatile organic iodine compounds in seawater related to degradation of organic matter**

,  $P < 0.01$ ,  $n = 64$

,  $P < 0.01$ ,  $n = 64$

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



vide iodine atoms to the atmosphere, resulting in catalytic ozone destruction in the troposphere and lower stratosphere (World Meteorological Organization (WMO), 2010). These VOIs have the potential to deplete ozone in boundary layer air on a wide scale (e.g., Carpenter et al., 1999). Arctic ozone depletion during polar sunrise has been indicated, by both observational and theoretical studies (e.g., Spicer et al., 2002), to result from the release of gaseous Br<sub>2</sub> and BrCl from the surface of sea ice and/or snow. More recent theoretical studies have indicated that reactive VOIs such as CH<sub>2</sub>I<sub>2</sub> would be extremely effective agents for tropospheric Arctic ozone depletion in polar sunrise, and that iodine compounds added to a Br<sub>2</sub>/BrCl mixture have a significantly greater ozone depletion effect than additional Br<sub>2</sub> and BrCl molecules only (Carpenter et al., 2005). To understand the flux patterns of VOIs emitted from the ocean to the Arctic air, it is necessary to investigate VOI distribution in seawater and VOI production and loss processes.

The concentrations of VOIs in seawater of the Arctic Ocean have been measured in the following studies. CH<sub>3</sub>I, CH<sub>2</sub>I<sub>2</sub>, CH<sub>2</sub>ClI, and C<sub>3</sub>H<sub>7</sub>I were measured around Ny Ålesund (80° N) in Spitzbergen, Norway in September (Schall and Heumann, 1993). CH<sub>3</sub>I was measured in the Greenland and Norwegian Seas (63–75° N) in November (Happell and Wallace, 1996). CH<sub>2</sub>ClI was measured in the Arctic Ocean Basin from Barrow, Alaska to Svalbard (71–90° N) (Karlsson et al., 2013). CH<sub>3</sub>I, C<sub>2</sub>H<sub>5</sub>I, CH<sub>2</sub>ClI, and C<sub>3</sub>H<sub>7</sub>I were measured in the Greenland and Norwegian Seas (78–82° N) in June and July (Atkinson et al., 2014). ~~As can be seen, each of these studies considered VOIs in the Atlantic sector of the Arctic Ocean. In contrast, there have been very few measurements of VOIs in the western (Pacific sector) Arctic Ocean, which has vast continental shelves.~~

In recent global mapping of VOIs in the surface mixed layer (Ooki et al., 2015), the Chukchi Sea shelf in the western Arctic Ocean was found to have the second-highest concentration of CH<sub>2</sub>ClI (ave. 5.7 pmol L<sup>-1</sup>) among all water types (tropical, subtropical, transitional, subpolar and polar) and ocean areas (shelf, slope, and basin). In the shelf–slope areas, CH<sub>2</sub>ClI had the largest flux to the atmosphere among three VOIs

BGD

12, 11245–11278, 2015

**Volatile organic  
iodine compounds in  
seawater related to  
degradation of  
organic matter**

A. Ooki et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

(Karlsson et al.,  
2013)

Printer-friendly Version

Interactive Discussion

(CH<sub>3</sub>I, C<sub>2</sub>H<sub>5</sub>I, and CH<sub>2</sub>ClI). The vast Chukchi Sea shelf and the adjacent sea areas might have an impact on organic iodine flux to the Arctic atmosphere. Although these various studies have measured distribution of VOIs, there is no established approach that will enable us to determine how VOI distributions and processes that produce them are impacted by the environmental conditions – including water mass structure and biochemical parameters, such as nutrients and chlorophyll *a*.

The vast Chukchi Sea shelf and the adjacent Canada Basin characterize the geography of the western Arctic Ocean. ~~The Bering and Chukchi Sea shelves in the western Arctic Ocean, which are among the largest continental shelves in the world, are known to have high biological productivity (e.g., Cota et al., 1996).~~ Three Pacific-origin water masses – the Anadyr Water (AnW), Bering Shelf Water (BSW) and Alaskan Coastal Water (ACW) – flow through the Bering Strait into the Chukchi Sea (Grebmeier et al., 1988, 1989). The AnW and BSW form a modified Bering Shelf–Anadyr Water (BSAW). BSAW flowing over the Chukchi shelf is characterized by low temperature (–1 to +2.0 °C) and relatively high salinity (> 31.8). It is originally enriched in nutrients, and occupies the bulk of the central Chukchi Sea. ACW flowing over the eastern Chukchi shelf along the coastline of the Alaskan Peninsula is characterized by higher temperatures (> +2 °C) and lower salinity (< 31.8). The nutrient level of ACW is originally low compared with that of BSAW. ACW and BSAW are described as the Pacific-origin Summer Water (PSW) in the western Arctic Ocean. ~~The surface mixed layer in the Arctic Ocean has a seasonally low salinity signature in the summer as a result of sea ice melt and river runoff.~~ In the Canada basin, the surface mixed layer has a seasonally low salinity signature (< 30) with relatively low temperature (< +2 °C) in the summer as a result of sea ice melting and river runoff, is called Polar Mixed Layer Water (PMLW) (Codispoti et al., 2005). ~~The PMLW covering the surface layer contributes to sea ice formation in the fall and winter.~~ Cold, dense water (CDW), which is characterized by its low temperature (–1.7 to –1.0 °C) and relatively high salinity (> 31.8), ~~is created as a result of sea ice formation and brine production, forming upper and lower halocline layers (Anderson et al., 2013).~~ The CDW that is transported

BGD

12, 11245–11278, 2015

**Volatile organic  
iodine compounds in  
seawater related to  
degradation of  
organic matter**

A. Gokti et al.

The Chukchi Sea shelf is one of the largest continental shelf in the world and has high biological productivity

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

32 - 33

Interactive Discussion

northward on the bottom of the Chukchi Sea shelf is described as the Pacific-origin Winter Water (PWW). This CDW (or PWW) flows in the subsurface layer of the Canada Basin. Atlantic Water (AW) with high salinity ( $> 34.2$ ) extends from below the halocline layer to the bottom of the Canada Basin.

In the present study, we conducted shipboard observation on the R/V *Mirai* (MR12-E03) over the Chukchi Sea and the Canada Basin in the western Arctic Ocean in September and October 2012. We identified the vertical distributions of four VOIs ( $\text{CH}_3\text{I}$ ,  $\text{C}_2\text{H}_5\text{I}$ ,  $\text{CH}_2\text{ClI}$ , and  $\text{CH}_2\text{I}_2$ ) in relation to the water mass structure. We also attempted to find a clue that will help explain the production processes of VOIs in seawater in association with biochemical parameters, such as nutrients and  $\text{Chl } a$ .

## 2 Method

### 2.1 Shipboard observation and seawater sampling

Shipboard observation was conducted on the R/V *Mirai* (MR12-E03), which is owned and operated by the Japan Agency for Marine-Earth Science and Technology (JAMSTEC), between 15 September and 04 October 2012 in the western Arctic Ocean as a part of GRENE Arctic Climate Change Research Project initiated by the Ministry of Education, Culture, Sports, Science and Technology (MEXT) Japan. A general description of the R/V *Mirai* is provided in the cruise report, which is available to the public on the JAMSTEC data website (<http://www.godac.jamstec.go.jp/darwin/e>). The locations of the 19 sampling stations for the present study are shown in Fig. 1.

Seawater samples for VOI measurement were collected in Teflon-coated, 10-L Niskin-X sampling bottles attached to CTD-RMSs (conductivity-temperature-depth probe-rosette multi-samplers). The sampling depths were 0 (bucket sampling), 5, 10, 20, 30, 40, 50, 100, 200, 500, 1000, 2000, and 3000 m (or bottom depth  $-5$  m).

Seawater aliquots (125 mL) were collected in dark glass bottles, overflowing approximately 250 mL of seawater. To arrest microbial activity, 50  $\mu\text{L}$  of saturated mercuric

BGD

12 11245–11278, 2015

The lower halocline layer water is formed by the mixture of CDW and AW.

We also compare VOIs with nutrients and  $\text{Chl } a$  and discuss possible processes to produce VOIs in seawater.

A. Ooki et al.

Methods

A general information of MR12-E03

Abstract Introduction

Conclusions References

Tables Figures

◀ ▶

◀ ▶

Back Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

chloride ( $\text{HgCl}_2$ ) solution, which had been previously bubbled with high-purity nitrogen for 30 min at 90–100 °C to remove any contaminating gases, was added to each bottle. The bottles were crimp-sealed with 0.5 mL headspace using a butyl rubber-lined septum and an aluminum cap. The sample bottles were kept in the dark at 0–4 °C until pre-treatment, typically within one month.

Dissolved halocarbons were collected from the seawater samples by the purge-and-trap method. The total volume of seawater (125 mL) in the sample bottle was introduced to a purge vessel by a high-purity nitrogen carrier at 65 mL  $\text{L}^{-1}$  and the dissolved gases were consecutively purged by bubbling with nitrogen for 35 min. The purged gas sample was collected in a cold-trap containing Tenax TA (10 mg) cooled to –90 °C. This cold-trap sample was sealed with 1/16 in. screw nuts (Swagelok®) and stored in a freezer (–30 °C) until gas chromatography–mass spectrometry (GC–MS) analysis, typically within 7 days. The purge efficiencies for  $\text{CH}_3\text{I}$ ,  $\text{C}_2\text{H}_5\text{I}$ ,  $\text{CH}_2\text{ClI}$  and  $\text{CH}_2\text{I}_2$  were 94, 95, 83, and 66 %, respectively.

Concentrated VOIs in the cold-trap were thermally (200 °C) desorbed and transferred to an automated pre-concentration GC–MS analytical system. Details of the GC–MS analysis are described elsewhere (Ooki and Yokouchi, 2011a, b). Briefly, the thermally desorbed sample was collected in a trap containing Carboxene 1000 and Carboxene B cooled to –50 °C in a small freezer. Concentrated VOIs in this first trap were thermally (200 °C) desorbed and transferred to a second trap, which contained Tenax TA and Carboxene 1000 cooled to –50 °C. Then, the second trap was heated to 200 °C, and the desorbed components were transferred to a capillary column (Porabond Q, 0.32 mm, 50 m) in preparation for GC–MS analysis (Agilent 5973, 6890).

To check the possibility of losses and contaminations of VOIs in the cold-trap during storage in the freezer (–30 °C), two types of standard gases were separately concentrated in each cold-trap. The first type is a purge-and-trap standard containing three VOIs in a water bottle (125 mL). The VOIs standard in a standard bottle (125 mL) was collected in a cold-trap using the purge-and-trap method mentioned above. The second type is mixed standard gases containing 13 halocarbons in a high-pressure cylin-

BGD

, for avoiding the damage of the glass bottle,

Volatile organic  
iodine compounds in  
seawater related to

65 ml min<sup>-1</sup>

organic matter

A. Ooki et al.

, respectively (Ooki  
and Yokouchi,  
2011a)

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

der. The mixed standard diluted with high-purity nitrogen was introduced to a cold-trap. Blank cold-traps (no VOI standard) and the two types of standard cold-traps were stored in a freezer ( $-30^{\circ}\text{C}$ ) for 7 to 30 days. The results of the GC–MS analysis of the standard and blank traps showed that no significant losses or contaminations occurred during storage.

## 2.2 Chlorophyll *a* and nutrients

Chl *a* concentrations were measured with a fluorescence sensor attached to the CTD observation system. The fluorescence data were calibrated with Chl *a* concentrations in discrete seawater samples measured using the fluorometric Welshmeyer method (Welshmeyer, 1994). Nutrient ( $\text{NO}_3$ ,  $\text{NO}_2$ ,  $\text{NH}_4$ ,  $\text{SiO}_2$ , and  $\text{PO}_4$ ) concentrations were measured by the colorimetric method using a QuAatro system and in accordance with “The GO-SHIP Repeat Hydrography Manual” (Hydes et al., 2010). Analytical precision was 0.12 % for  $\text{NO}_3^-$ , 0.21 % for  $\text{NO}_2^-$ , 0.19 % for  $\text{PO}_4^{3-}$ , 0.11 % for  $\text{SiO}_2$ , and 0.34 % for  $\text{NH}_4^+$ .

We calculated the value of nitrogen deficit relative to phosphate, N-deficit ( $= [\text{NH}_4^+] + [\text{NO}_3^-] + [\text{NO}_2^-] - r \times [\text{PO}_4^{3-}]$ ), where  $r$  is a Redfield ratio of 16. We regard the large N-deficit as an indicator of the influence of organic matter decomposition in the sediment over the shelf.

denitrification in shelf sediment.

## 3 Results

### 3.1 Overview of hydrographic condition

The spatial distribution of surface seawater temperature (SST) is shown in Fig. 1. The water masses were classified according to Grebmeier et al. (1989). The warm ACW with temperature  $> +2^{\circ}\text{C}$ , as shown in light blue–red color in the figure, and salinity  $< 31.8$ , was found extensively in the east Chukchi Sea shelf–slope area. The main flow

BGD

12, 11245–11278, 2015

#### Volatile organic

Concentration ratio of nitrogenous nutrient and phosphate in seawater is usually fixed with  $\text{N/P}=16$ . In anoxic region, such as bottom sediment, in which nitrate is used in the destruction of organic matter, phosphate can increase with decrease in nitrate.

Conclusions

References

Tables

Figures

► I

►

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

of ACW (temp. > 4 °C), shown in both orange and red color in the figure, was along the coastline of the Alaskan Peninsula. This flow covered the sampling locations of St1, St14, St76, and St74. The ACW seemed to distribute westward around 70° N, covering the sampling locations of St78–80. Its eastward flow reached the Barrow Canyon (St53–60) and branched to the northwest along the Chukchi Sea slope from St48–45 (light blue in the figure). The PMLW (–1 < temp. < 2 °C, sal. < 30) occupied the surface over the northern Chukchi Sea and the Canada Basin, covering the sampling locations of St72, St24, St64, St39, St29, St33, and St32.

Geographical information is summarized in Table 1. This information includes sampling position (Lat., Long.), bottom depth, area type (shelf with bottom depth < 200 m, slope with bottom depth 200–2000 m or basin with bottom depth > 2000 m). The table also includes hydrographic conditions, which include SST, sea surface salinity (SSS), surface mixed layer depth (MLD), and water mass types (ACW, BSAW, PMLW, CDW, or AW). The sampling stations are listed in order of latitude from the southernmost St1 to the northernmost St45 under the influence of ACW and in order of bottom depth from the shallowest St72 to the deepest St64 under the influence of PMLW. The MLD was defined as the depth where the potential density increases by 0.125 kg m<sup>–3</sup> compared with the reference depth of 5 m. The subsurface layer was defined as the layer between the surface mixed layer and the CDW or ocean floor.

In the present study, the southern Chukchi Sea area (St1–74) was defined as the area between the Bering Strait and 71° N, where there is a shallow continental shelf. The central Chukchi Sea area (St53–24) was defined as the area between 71° N and 74° N, and the northern slope-basin area (St29–64) was defined as the northern area (74° N).

In the water structure analysis, the ACW occupied the water column under the main flow of ACW (St1, St14, St76, St74), and the BSAW was found in the subsurface layer at the other stations. The CDW (or PWW) was found below the subsurface layer at the central Chukchi Sea and the northern areas. The AW underlaid the CDW at the deeper stations (bottom depth > 124 m) of the central and northern areas.

BGD

12, 11245–11278, 2015

## Volatile organic iodine compounds in

In the present study, we divided our study area into three groups: Southern Chukchi shelf, Central Chukchi shelf-slope, and Northern slope-basin, defined as area between Bering Strait and 71N, 71N and 74N, and north of 74N, respectively (Table 2).

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

## 3.2 Distributions of Chlorophyll *a* (Chl *a*), Ammonium (NH<sub>4</sub><sup>+</sup>), N-deficit, and VOIs

The average concentrations of Chl *a*, NH<sub>4</sub><sup>+</sup>, N-deficit, and the four studied VOIs (CH<sub>3</sub>I, C<sub>2</sub>H<sub>5</sub>I, CH<sub>2</sub>I<sub>2</sub> and CH<sub>2</sub>ClI) in the surface mixed layer, subsurface layer, CDW and AW at each station are listed in Tables 2–7. The vertical distributions of potential density, NH<sub>4</sub><sup>+</sup>, N-deficit and the four VOIs in the southern Chukchi Sea shelf (St10, St80, St80 and St76), and the central Chukchi Sea and northern areas (St24, St29, St64, St32, and St68) are shown in Fig. 2a–g and Fig. 3a–g, respectively.

### 3.2.1 Chlorophy

The averages of Chl *a* in ACW in the surface mixed layer (0.75 ± 0.33 μg L<sup>-1</sup>) and the subsurface layer (0.67 ± 0.26 μg L<sup>-1</sup>) were higher than the total average in all Pacific-origin Water.

The total average concentration of Chl *a* in all Pacific-origin Water (ACW + BSAW + PMLW + CDW) was 0.42 μg L<sup>-1</sup>. We regard any Chl *a* concentration above 0.42 μg L<sup>-1</sup> as “high” in the western Arctic Ocean. The ACW had high averages of 0.75 (±0.33σ) μg L<sup>-1</sup> in the surface mixed layer and 0.67 (±0.26σ) μg L<sup>-1</sup> in the subsurface layer. The PMLW and BSAW had lower Chl *a* concentrations, with averages of 0.19 (±0.17σ) μg L<sup>-1</sup> in the surface mixed layer and 0.31 (±0.16σ) μg L<sup>-1</sup> in the subsurface layer. In the northern slope–basin stations (St64, St39, St29, St33, St32), where the PMLW occupied the surface layer, the Chl *a* concentrations were very low (0.08–0.16 μg L<sup>-1</sup>) in the surface layer. But, concentration maxima were found in the subsurface layer at station St29 (0.53 μg L<sup>-1</sup>) and the CDW at St33 (0.48 μg L<sup>-1</sup>) (Table 2).

### 3.2.2 Ammonium (NH<sub>4</sub><sup>+</sup>) and N-deficit

The total average concentration of NH<sub>4</sub><sup>+</sup> in all Pacific-origin Water (ACW + BSAW + PMLW + CDW) was 1.0 μmol L<sup>-1</sup>. We regard any NH<sub>4</sub><sup>+</sup> concentration above 1.0 μmol L<sup>-1</sup> as “high” in the western Arctic Ocean. Over the southern Chukchi Sea shelf (St1–74), the concentration maxima (1.0–5.8 μmol L<sup>-1</sup>) were found

BGD

12, 11245–11278, 2015

Volatile organic  
iodine compounds in  
seawater related to  
degradation of  
organic matter

A. Ooki et al.

Discussion Paper | Discussion Paper | Discussion Paper | Discussion Paper | Discussion Paper

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



in the bottom layer water as shown in Fig. 2b for St10, St80 and St76. The highest concentration ( $5.8 \mu\text{mol L}^{-1}$ ) was found near the bottom of the BSAW around the Hope Valley over the Chukchi Sea shelf (St10), where the “Hope Valley biological hotspot” is known to exist (Grebmeier et al., 2015). The dissolved oxygen minimum ( $243 \mu\text{mol Kg}^{-1}$ ) and large N-deficit ( $-12.3 \mu\text{mol L}^{-1}$ ) occurred simultaneously near the bottom in this area.

Over the central Chukchi Sea shelf-slope-area, concentration maxima ( $1.3$ – $3.8 \mu\text{mol L}^{-1}$ ) were found in the subsurface BSAW and CDW above AW at St53, St60, St68, St72 and St24 (Table 3). The vertical distributions at St68 and St24 are shown in Fig. 3b.

In the northern slope-basin-areas (St29–64),  $\text{NH}_4^+$  concentrations were very low, with an average of  $0.05 (\pm 0.07\sigma) \mu\text{mol L}^{-1}$ . Note that small concentration peaks were found in the CDW at St29 ( $0.27 \mu\text{mol L}^{-1}$ ) at 60 m depth, as shown in Fig. 3b. The same were found in the CDW at St33 ( $0.56 \mu\text{mol L}^{-1}$ ) at 51 m depth over the Chukchi Sea slope. The latter is the location where Chl *a* peaks ( $0.48$ – $0.53 \mu\text{g L}^{-1}$ ) also occurred.

The high  $\text{NH}_4^+$  concentration in the Chukchi Sea shelf-slope-areas can be explained as follows. Ammonium regenerated from the organic matter in sediment is added to the bottom layer water over the Chukchi Sea shelf in summer. The CDW, which has an added high amount of  $\text{NH}_4^+$  from the sediment of the northernmost end of the Chukchi Sea shelf, has been moved to the slope area (Nishino et al., 2005). The supply of  $\text{NH}_4^+$  regenerated from organic matter in the ocean sediment contributes to keeping the  $\text{NH}_4^+$  concentration at several  $\mu\text{mol L}^{-1}$  in the water column. In contrast, the low concentrations of  $\text{NH}_4^+$  in the northern slope-basin area can be explained by the biological consumption of  $\text{NH}_4^+$  ( $\text{NH}_4^+$  oxidation as a first step of nitrification) in the water column.

Large N-deficits ( $-11.1$ – $-20.4 \mu\text{mol L}^{-1}$ ) were found in the bottom layer water of the BSAW and CDW from the southern Chukchi Sea shelf to the northern slope-basin area. The largest N-deficit ( $-20.4 \mu\text{mol L}^{-1}$ ) occurred in the CDW at the northern tip of the Chukchi Sea shelf (St24), as shown in Fig. 3c. These N-deficits largely resulted

BGD

12, 11245–11278, 2015

**Volatile organic  
iodine compounds in  
seawater related to  
degradation of  
organic matter**

A. Ooki et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

from denitrification in the ocean sediment over the shelf. Within a narrow depth range in the subsurface waters of the BSAW and CDW over the northern Chukchi Sea shelf-slope area and the Canada Basin, the ~~N\*\* minima (or N-deficit minima)~~ were coincident with the remarkably high maxima of nutrients, humic-like F-DOM, and dissolved Fe – which are all attributed to emissions from the bottom sediment, which in turn are linked to organic matter decomposition (Hioki et al., 2014).

### 3.2.3 Iodomethane (CH<sub>3</sub>I)

The total average concentration of CH<sub>3</sub>I in all Pacific-origin water (ACW + BSAW + PMLW + CDW) was 2.2 pmol L<sup>-1</sup>, which corresponds to the global ocean surface average values in the range of 2.3–2.6 pmol L<sup>-1</sup> in shelf-slope areas (Ooki et al., 2015). Globally, the polar surface waters had the lowest concentrations of CH<sub>3</sub>I (1.0–1.5 pmol L<sup>-1</sup>) among all water types from tropical to polar (total average 2.3–3.7 pmol L<sup>-1</sup>). ~~We regard any CH<sub>3</sub>I concentration above 2.2 pmol L<sup>-1</sup> as “high” in the Arctic Ocean.~~

High concentration peaks of CH<sub>3</sub>I were occasionally found in the bottom layer water over the southern Chukchi Sea shelf (Fig. 2d), with average values in the range of 2.1–5.2 pmol L<sup>-1</sup>, and in the subsurface ACW/BSAW and CDW over the central Chukchi Sea, with average values in the range of 1.5–8.1 pmol L<sup>-1</sup> (Table 4). These concentrations are consistent with the surface mixed layer averages of 2.5 (1.2–4.3) pmol L<sup>-1</sup> in the Amundsen Sea and 3.0 (2.1–4.7) pmol L<sup>-1</sup> in the Ross Sea in Antarctica. In summer, considerable algal bloom occurs, and maximum Chl *a* concentrations are 8.4–10 μg L<sup>-1</sup> (Mattson et al., 2012).

The highest concentration of CH<sub>3</sub>I (8.1 pmol L<sup>-1</sup>) occurred in the CDW, near the bottom over the northern tip of the Chukchi Sea shelf (St24), where the largest N-deficit (–20.4 μmol L<sup>-1</sup>) also occurred. A significant peak (4.0 pmol L<sup>-1</sup>) occurred in the CDW above the AW over the northern Chukchi Sea slope (St68 in Table 4), coincident with the concentration peak of NH<sub>4</sub><sup>+</sup> (3.2 μmol L<sup>-1</sup>), as shown in Fig. 2b and d. In the north-

BGD

12, 11245–11278, 2015

**Volatile organic  
iodine compounds in  
seawater related to  
degradation of**

(> 2.2 pmol L<sup>-1</sup>) above  
the total average in all  
Pacific-origin water

A. Ooki et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

ern slope–basin ~~area (St29–64)~~, the  $\text{CH}_3\text{I}$  concentrations were low with average values in the range of  $0.37\text{--}1.9\text{ pmol L}^{-1}$ .

### 3.2.4 Iodoethane ( $\text{C}_2\text{H}_5\text{I}$ )

The total average concentration of  $\text{C}_2\text{H}_5\text{I}$  in all Pacific-origin water (ACW + BSAW + PMLW + CDW) was  $0.84\text{ pmol L}^{-1}$ . We regard any  $\text{C}_2\text{H}_5\text{I}$  concentration above  $0.84\text{ pmol L}^{-1}$  as “high” in the western Arctic Ocean. This threshold is 1.5–3 times the global ocean surface average values in the range of  $0.23\text{--}0.59\text{ pmol L}^{-1}$  in shelf–slope areas (Ooki et al., 2015).

High concentration peaks of  $\text{C}_2\text{H}_5\text{I}$  were occasionally found in the bottom layer water over the southern Chukchi Sea shelf (average values in the range of  $0.9\text{--}3.9\text{ pmol L}^{-1}$  in Table 5), as shown in Fig. 2e, and in subsurface BSAW and CDW over the central shelf–slope areas (average values in the range of  $1.2\text{--}4.4\text{ pmol L}^{-1}$ ), as shown in Fig. 3e.

The highest concentration of  $\text{C}_2\text{H}_5\text{I}$  ( $4.4\text{ pmol L}^{-1}$ ) occurred in the CDW over the central Chukchi Sea slope (St68), at which depth a large N-deficit ( $-13.2\text{ }\mu\text{mol L}^{-1}$ ) and high concentration of  $\text{NH}_4^+$  ( $3.2\text{ }\mu\text{mol L}^{-1}$ ) simultaneously occurred. In the northern slope–basin ~~area~~, the  $\text{C}_2\text{H}_5\text{I}$  concentrations were low ( $\text{DL}\text{--}0.70\text{ pmol L}^{-1}$ ). A small peak of  $\text{C}_2\text{H}_5\text{I}$  ( $1.5\text{ pmol L}^{-1}$ ) was found in the CDW over the northern tip of the Chukchi Sea shelf (St24), where the  $\text{CH}_3\text{I}$  maximum and a large N-deficit simultaneously occurred.

### 3.2.5 Diiodomethane ( $\text{CH}_2\text{I}_2$ )

Very high levels of  $\text{CH}_2\text{I}_2$  ( $30\text{--}47\text{ pmol L}^{-1}$ ) were found in the ACW at the Bering Strait (St1). Similar high levels of  $\text{CH}_2\text{I}_2$  (ave.  $23\text{ pmol L}^{-1}$ , min–max: DL ( $< 0.9$ )– $48\text{ pmol L}^{-1}$ ) were reported in the macro-algal field seawater in the fjord of the NyÅlesund ( $80^\circ\text{ N}$ ,  $12^\circ\text{ E}$ ) on Spitzbergen, Norway (Schall and Heumann, 1993). The total average concentration of  $\text{CH}_2\text{I}_2$  in all Pacific-origin water (ACW + BSAW + PMLW + CDW) except for the Bering Strait water (St1) was  $4.4\text{ pmol L}^{-1}$ . We regard any  $\text{CH}_2\text{I}_2$  concentration

BGD

12, 11245–11278, 2015

( $> 0.84\text{ pmol L}^{-1}$ ) above the total average in Pacific-origin water

seawater related to degradation of organic matter

A. Ooki et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

above  $4.4 \text{ pmol L}^{-1}$  as “high” in the western Arctic Ocean. This threshold is within the range of surface averages from  $2.5 \text{ pmol L}^{-1}$  in the coastal area (10 km offshore) of the English Channel ( $50^\circ \text{ N}$ ,  $4^\circ \text{ W}$ ) (Archer et al., 2007) to  $6.2 \text{ pmol L}^{-1}$  in the fjord of the Ny Ålesund (Schall and Heumann, 1993). High concentrations were found in the bottom layer water of the southern Chukchi Sea shelf and the subsurface layer of the ACW, BSAW, and CDW over the central to northern Chukchi shelf–slope areas, as shown in Figs. 2f and 3f. The low concentrations ( $0.80\text{--}1.50 \text{ pmol L}^{-1}$ ) occasionally found in the surface mixed layer water over the shelf (St76 and St45 in Table 6) are probably due to the photolysis of  $\text{CH}_2\text{I}_2$  in seawater (Jones and Carpenter, 2005).

A significant peak of  $\text{CH}_2\text{I}_2$  ( $20.9 \text{ pmol L}^{-1}$ ), which is the highest concentration other than the Bering Strait samples, was found in the CDW, near the bottom over the northern tip of the Chukchi Sea shelf (St24 in Fig. 3f). The largest N-deficit and the  $\text{CH}_3\text{I}$  maximum also occurred in this water. In the northern slope–basin area, the lowest levels of  $\text{CH}_2\text{I}_2$  ( $DL\text{--}0.05 \text{ pmol L}^{-1}$ ) were found in the surface mixed layer and the underlying deeper AW. In contrast, significant peaks of  $\text{CH}_2\text{I}_2$  ( $0.47\text{--}14 \text{ pmol L}^{-1}$ ) were found in the subsurface BSAW and CDW over the slope–basin area. The peak concentrations decreased with increasing bottom depth from the shelf (St24) to the far basin (St64).

### 3.2.6 Chloriodomethane ( $\text{CH}_2\text{ClI}$ )

The total average concentration of  $\text{CH}_2\text{ClI}$  in all Pacific-origin water (ACW + BSAW + PMLW + CDW) was  $5.0 \text{ pmol L}^{-1}$ . We regard any  $\text{CH}_2\text{ClI}$  concentration above  $5.0 \text{ pmol L}^{-1}$  as “high” in the western Arctic Ocean. This threshold is within the global ocean surface average values in the range of  $3.7\text{--}6.6 \text{ pmol L}^{-1}$  in shelf–slope areas (Ooki et al., 2015). High concentrations of  $\text{CH}_2\text{ClI}$  were found in all the ACW samples except for St76, with an average of  $8.7 \pm 3.6(1\sigma) \text{ pmol L}^{-1}$  from the Bering Strait to the central Chukchi Sea slope area (Table 7). Low concentrations of  $\text{CH}_2\text{I}_2$  in the surface ACW were occasionally found over the shelf (St76) and slope

BGD

( $> 4.4 \text{ pmol L}^{-1}$ )

12, 11245–11278, 2015

**Volatile organic  
iodine compounds in  
seawater related to  
degradation of  
organic matter**

A. Ooki et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

(St45). The occasional low concentrations of  $\text{CH}_2\text{I}_2$  in the surface layer are attributed to photochemical degradation. Since the photolysis rate of  $\text{CH}_2\text{CII}$  is much slower than that of  $\text{CH}_2\text{I}_2$  in seawater (Jones and Carpenter, 2005),  $\text{CH}_2\text{CII}$  in the surface layer would exist for a longer time than  $\text{CH}_2\text{I}_2$ . The lateral transport of  $\text{CH}_2\text{CII}$  and  $\text{CH}_2\text{I}_2$  will be discussed in Sect. 4.4.

A high concentration peak of  $\text{CH}_2\text{CII}$  ( $11.0 \text{ pmol L}^{-1}$ ) was found in the CDW, near the bottom over the northern tip of the Chukchi Sea shelf (St24 in Fig. 3g). Significant peaks of  $\text{CH}_2\text{I}_2$  and  $\text{CH}_3\text{I}$ , and the largest N-deficit also occurred in this water.

## 4 Discussion

### 4.1 Correlation analysis and VOI production

Correlation analysis was performed to identify potential correlation between VOIs, Chl *a*,  $\text{NH}_4^+$ , and N-deficit for all the samples and ACW and BSAW/PMLW/CDW samples. The correlate coefficients are summarized in Table 8. Scattering diagrams of VOIs as compared to  $\text{NH}_4^+$  are shown in Fig. 4.

High correlations were found in the BSAW/PMLW/CDW between  $\text{NH}_4^+$  and three VOIs:  $\text{CH}_3\text{I}$  ( $R = 0.77$ ),  $\text{C}_2\text{H}_5\text{I}$  ( $R = 0.93$ ) and  $\text{CH}_2\text{CII}$  ( $R = 0.66$ ). High correlations were also found between N-deficit and  $\text{CH}_2\text{I}_2$  ( $R = 0.67$ ). Other nutrient components ( $\text{NO}_3^-$ ,  $\text{NO}_2^-$ ,  $\text{PO}_4^{3-}$ ,  $\text{SiO}_2$ ) showed lower correlations with VOIs than  $\text{NH}_4^+$  and N-deficit.

We could not find any correlation between VOIs and Chl *a*, even though former incubation experiments have indicated that many marine phytoplankton species produce VOIs as they grow (Tokarczyk and Moore, 1994; Moore et al., 1996; Manley and de la Cuesta, 1997). However, high and low concentrations of VOIs in surface seawater have also been attributed to varying high and low biological productivity in different types of water (e.g., from tropical to polar) and different sea areas (shelf, slope, basin) (Ooki et al., 2015). It seems that Chl *a* is not a main factor in the vertical–horizontal

BGD

12, 11245–11278, 2015

**Volatile organic  
iodine compounds in  
seawater related to  
degradation of  
organic matter**

A. Ooki et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

distributions of VOIs on a regional scale, such as the Chukchi Sea and adjacent sea area.

#### 4.1.1 Production of Iodomethane ( $\text{CH}_3\text{I}$ ) and Iodoethane ( $\text{C}_2\text{H}_5\text{I}$ )

The significant correlation between  $\text{C}_2\text{H}_5\text{I}$  and  $\text{NH}_4^+$  implies that  $\text{C}_2\text{H}_5\text{I}$  production in seawater is occurring at the same time as regeneration of  $\text{NH}_4^+$  from organic matter. The  $\text{C}_2\text{H}_5\text{I}$  consumption, which is probably bacterial consumption, might occur at a rate similar to the biological  $\text{NH}_4^+$  consumption. The consumption rates for  $\text{NH}_4^+$  in seawater over the Chukchi Sea shelf are reported to vary widely from  $0.15\text{--}3.6\text{ nmol L}^{-1}\text{ d}^{-1}$  (Christman et al., 2011) to  $0.24\text{--}7.2\text{ }\mu\text{mol L}^{-1}\text{ d}^{-1}$  (Souza et al., 2014), with a variable turnover from within a day to over ten days. The high correlation between  $\text{CH}_3\text{I}$  and  $\text{NH}_4^+$  ( $R = 0.77$ ) would also be attributable to the  $\text{CH}_3\text{I}$  production linked to the degradation of organic matter.

Note that two plots below the linear regression line in Fig. 4 are the data collected from the bottom layer water at the Hope Valley biological hotspot (St10) with high nutrients and low dissolved oxygen. We do not currently propose a reason for this exceptional plot.

The production of  $\text{C}_2\text{H}_5\text{I}$  and  $\text{CH}_3\text{I}$  associated with the bacterial degradation of organic matter was first demonstrated by an incubation experiment for marine biogenic aggregates, which mainly consisted of diatom collected from the Arctic seawater. The concentrations of  $\text{C}_2\text{H}_5\text{I}$  and  $\text{CH}_3\text{I}$  in the incubation vessel increased sharply, simultaneously with the enhancement of bacterial heterotrophic production rate (Hughes et al., 2008).

Previously,  $\text{NH}_4^+$  concentrations in coastal subarctic seawater have been observed to increase in the bottom layer water ( $\sim 90\text{ m}$  depth) after the spring bloom of diatoms (Kudo et al., 2007). Therefore, we suggest that the production of  $\text{C}_2\text{H}_5\text{I}$  and  $\text{CH}_3\text{I}$  associated with the degradation of organic matter occurs in the bottom layer water, probably after the diatom bloom, over the Chukchi Sea shelf.

, implying that large fraction of VOIs abundances in coastal sea area were not directly emitted from living phytoplankton as discussed in later sections 4.1.1 and 4.1.2.

### iodine compounds in seawater related to degradation of organic matter

A. Ooki et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

### 4.1.2 Production of Diiodomethane ( $\text{CH}_2\text{I}_2$ ) and Chloriodomethane ( $\text{CH}_2\text{ClI}$ )

The notable feature of  $\text{CH}_2\text{I}_2$  and  $\text{CH}_2\text{ClI}$  distributions was that high concentration peaks were found (1) in the CDW and subsurface BSAW over the Chukchi Sea shelf (St24, St53, St60, St72) and slope (St68), where large N-deficit and high  $\text{NH}_4^+$  concentration occurred simultaneously, and (2) at the northern Chukchi Sea slope (St29), where large N-deficit and low  $\text{NH}_4^+$  concentration occurred simultaneously. Concentration peaks of  $\text{C}_2\text{H}_5\text{I}$  and  $\text{CH}_3\text{I}$  were not found over the northern slope area (St29) where  $\text{NH}_4^+$  concentration was low.

From the marine biogenic aggregates incubation (Hughes et al., 2008) mentioned in the last section, the concentrations of  $\text{CH}_2\text{I}_2$  and  $\text{CH}_2\text{ClI}$  did not increase in the incubation vessel during the incubation period of 70 h, whereas  $\text{C}_2\text{H}_5\text{I}$  showed significant increases in concentration with the enhancement of bacterial production rate after 30 h of incubation. It is possible that the production processes of  $\text{CH}_2\text{I}_2$  in bottom layer water are quite different from those of  $\text{C}_2\text{H}_5\text{I}$ , even if the production of both VOIs is associated with the degradation of organic matter in the sediment or the bottom layer water.

Recent studies have suggested that the production of  $\text{CH}_2\text{I}_2$  in seawater occurs after the reaction of organic matter with  $\text{I}_2$ , which is known as the “iodoform ( $\text{CHI}_3$ ) reaction.” The production of  $\text{CHI}_3$  and  $\text{CH}_2\text{I}_2$  has been clarified to occur after the reaction of dissolved organic matter with  $\text{I}_2$ , which is produced by  $\text{I}^-$  oxidizing bacteria in a culture vessel (Fuse et al., 2003; Amachi et al., 2005). It is noted that small amounts of  $\text{CH}_2\text{ClI}$  and  $\text{CH}_3\text{I}$  have been detected in the culture vessel with drastic increases of  $\text{CH}_2\text{I}_2$ , whereas  $\text{C}_2\text{H}_5\text{I}$  production has not yet clarified from the iodoform reaction.

The production of  $\text{I}_2$  in seawater is believed to be initiated with  $\text{I}^-$  oxidations. However, it remains unclear where the  $\text{I}^-$  oxidations occur in the ocean environment. Recently, iodide-oxidizing bacteria, which oxidize iodide ( $\text{I}^-$ ) to molecular iodine ( $\text{I}_2$ ), have been detected in natural seawater samples after incubation under the iodide-rich condition ( $\sim 1 \text{ mmol L}^{-1}$ ) (Amachi et al., 2005). The iodide-rich condition has been found in pore water of ocean sediment, with  $\text{I}^-$  concentration of  $\sim 7 \text{ } \mu\text{mol L}^{-1}$ , which is 70 to 700 times

BGD

12, 11245–11278, 2015

**Volatile organic  
iodine compounds in  
seawater related to  
degradation of  
organic matter**

A. Ooki et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

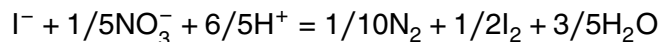
Full Screen / Esc

Printer-friendly Version

Interactive Discussion

the seawater concentration ( $10\text{--}100\text{ nmol L}^{-1}$ ) (Anschutz et al., 2000). Iodide-oxidizing bacteria might actively produce  $\text{I}_2$  in pore water of the ocean sediment. Therefore, it is supposed that  $\text{I}_2$  production by  $\text{I}^-$ -oxidizing bacteria in the sediment over the Chukchi Sea shelf has resulted in the production of  $\text{CH}_2\text{I}_2$ ,  $\text{CH}_2\text{CII}$ , and  $\text{CH}_3\text{I}$  in the sediment or the bottom layer water.

Moreover, iodide can be oxidized by nitrate in a reaction that yields  $\text{N}_2$  and  $\text{I}_2$  in marine sediments (Anschutz et al., 2000).



This reaction, which is thermodynamically favorable at all pH encountered in marine sediments ( $\text{pH} < 8$ ), would promote denitrification. Therefore, we suppose that the high concentrations of  $\text{CH}_2\text{I}_2$  and  $\text{CH}_2\text{CII}$  in the bottom layer water and CDW over the Chukchi Sea shelf–slope area, with the large N-deficit, were linked to denitrification in the sediment.

## 4.2 Lateral transport of VOIs

The significant concentration peaks of the four VOIs in the CDW over the northern shelf–slope area suggest the lateral transport of VOIs through the CDW layer from the shelf to the basin area, as well as the transport of humic-like DOM, nutrients, and dissolved Fe (Hioki et al., 2014) – which are all believed to be derived from organic matter decomposition in the sediment over the shelf.

High concentrations of  $\text{CH}_2\text{CII}$  in the surface mixed layer over the central Chukchi Sea slope would be due to the lateral transport of ACW from the Bering Strait and the southern Chukchi Sea. Lateral transport of  $\text{CH}_3\text{I}$ ,  $\text{C}_2\text{H}_5\text{I}$ , and  $\text{CH}_2\text{I}_2$  in ACW through the surface mixed layer to the central slope area was not observed.

The lateral transport of  $\text{CH}_2\text{CII}$  from the Chukchi Sea to the northern Canada Basin in ACW through the mixed layer, and to the Alpha Ridge ( $85^\circ\text{ N}$ ) in the Pacific-origin CDW through the subsurface upper halocline layer, has recently been reported by Karlsson

**BGD**

12, 11245–11278, 2015

**Volatile organic  
iodine compounds in  
seawater related to  
degradation of  
organic matter**

A. Ooki et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



et al. (2013). The VOIs produced over the Chukchi Sea shelf might have the potential to supply iodine to Arctic air far from the shelf.

## 5 Summary

Shipboard observation was conducted over the Chukchi Sea shelf-slope and the Canada Basin in the western Arctic Ocean in September and October 2012 by the R/V *Mirai* (MR12-E03) as part of GRENE Arctic Climate Change Research Project. The purpose of the study was to obtain vertical distributions of iodomethane ( $\text{CH}_3\text{I}$ ), iodoethane ( $\text{C}_2\text{H}_5\text{I}$ ), diiodomethane ( $\text{CH}_2\text{I}_2$ ) and chloriodomethane ( $\text{CH}_2\text{ClI}$ ) in seawater. The following is a summary of the characteristics of the vertical distributions of these VOIs, and hypotheses concerning lateral transport of VOIs in the area and mechanisms responsible for their production in relation to the degradation of organic matter.

High concentrations of four VOIs were found in the bottom layer water over the Chukchi Sea shelf, in which layer the concentration maximum of ammonium ( $\text{NH}_4^+$ ) simultaneously occurred. High correlations were found between  $\text{NH}_4^+$  and  $\text{C}_2\text{H}_5\text{I}$  with correlate coefficient  $R = 0.93$ , and between  $\text{NH}_4^+$  and  $\text{CH}_3\text{I}$  with  $R = 0.77$ . These results suggest that production of these VOIs is related to degradation of organic matter. The significant correlation between  $\text{C}_2\text{H}_5\text{I}$  and  $\text{NH}_4^+$  implies similar consumption rates for both compounds in seawater, probably owing to bacterial consumption. Over the northern Chukchi Sea shelf-slope, concentration maxima of  $\text{CH}_2\text{I}_2$ ,  $\text{CH}_2\text{ClI}$ , and  $\text{CH}_3\text{I}$  were found in the subsurface BSAW and CDW alongside large N-deficits and low concentrations of  $\text{NH}_4^+$ . These results suggest that the production of the three VOIs are quite different from the production process of  $\text{C}_2\text{H}_5\text{I}$ , while the productions of the four VOIs are all suspected to be relevant to the decompositions of organic matter in the sediment or the bottom layer water over the shelf.

Concentration peaks of four VOIs in the subsurface BSAW and CDW over the northern Chukchi Sea slope, and the peaks of  $\text{CH}_2\text{ClI}$  in the surface ACW over the central

BGD

12, 11245–11278, 2015

### Volatile organic iodine compounds in seawater related to degradation of organic matter

A. Ooki et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

CH<sub>2</sub>ClI

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



Chukchi Sea slope, suggest lateral transport from the Chukchi Sea shelf to slope–basin areas. The production of VOIs over the Chukchi Sea, in relation to decomposition of organic matter created in the high productive Chukchi Sea shelf, would have the potential to supply iodine to Arctic air far from the shelf.

5 *Author contributions.* A. Ooki and S. Kawasaki measured VOIs in seawater, and analyzed the data. K. Kuma, S. Nishino, and T. Kikuchi designed the ship observation. A. Ooki, S. Kawasaki, S. Nishino, and T. Kikuchi carried out the observation. A. Ooki prepared the manuscript with contributions from all co-authors.

10 *Acknowledgements.* We thank the captain, officers, and crew of the R/V *Mirai*, which was operated by Global Ocean Development, Inc. We also thank the staff of Marine Works Japan, Ltd., for their skillful work aboard the ship and for data processing, and A. Fujiwara for preparing the SST map from satellite data. This study was supported by GRENE Arctic Climate Change Research Project, and by JSPS KAKENHI grant number 24 681 001.

## References

- 15 Amachi, S., Muramatsu, Y., Akiyama, Y., Miyazaki, K., Yoshiki, S., Hanada, S., Kamagata, Y., Ban-nai, T., Shinoyama, H., and Fujii, T.: Isolation of iodide-oxidizing bacteria from iodide-rich natural gas brines and seawaters, *Microb. Ecol.*, 49, 547–557, doi:10.1007/s00248-004-0056-0, 2005.
- 20 Anderson, L. G., Andersson, P. S., Bjork, G., Jones, E. P., Jutterstrom, S., and Wahlstrom, I.: Source and formation of the upper halocline of the Arctic Ocean, *J. Geophys. Res.*, 118, 410–421, doi:10.1029/2012JC008291, 2013.
- Anschutz, P., Sundby, B., Lefrancois, L., Luther, G. W., and Mucci, A.: Interactions between metal oxides and species of nitrogen and iodine in bioturbated marine sediments, *Geochim. Cosmochim. Ac.*, 64, 2751–2763, doi:10.1016/S0016-7037(00)00400-2, 2000.
- 25 Archer, S. D., Goldson, L. E., Liddicoat, M. I., Cummings, D. G., and Nightingale, P. D.: Marked seasonality in the concentrations and sea-to-air flux of volatile iodocarbon compounds in the western English Channel, *J. Geophys. Res.*, 112, C08009, doi:10.1029/2006JC003963, 2007.

- Atkinson, H. M., Hughes, C., Shaw, M. J., Roscoe, H. K., Carpenter, L. J., and Liss, P. S.: Halocarbons associated with Arctic sea ice, *Deep-Sea Res.*, 92, 162–175, doi:10.1016/j.dsr.2014.05.012, 2014.
- Carpenter, L. J., Sturges, W. T., Penkett, S. A., Liss, P. S., Aliche, B., Hebestreit, K., and Platt, U.: Short-lived alkyl iodides and bromides at Mace Head, Ireland: links to biogenic sources and halogen oxide production, *J. Geophys. Res.*, 104, 1679–1689, doi:10.1029/98JD02746, 1999.
- Carpenter, L. J., Hopkins, J. R., Jones, C. E., Lewis, A. C., Parthipan, R., Wevill, D. J., Poissant, L., Pilote, M., and Constant, P.: Abiotic source of reactive organic halogens in the sub-arctic atmosphere, *Environ. Sci. Technol.*, 39, 8812–8816, doi:10.1021/es050918w, 2005.
- Christman, G. D., Cottrell, M. T., Popp, B. N., Gier, E., and Kirchman, D. L.: Abundance, diversity, and activity of ammonia-oxidizing prokaryotes in the coastal Arctic Ocean in summer and winter, *Appl. Environ. Microb.*, 77, 2026–2034, doi:10.1128/AEM.01907-10, 2011.
- Codispoti, L. A., Flagg, C., Kelly, V., and Swift, J. H.: Hydrographic conditions during the 2002 SBI process experiments, *Deep-Sea Res. II*, 52, 3199–3226, doi:10.1016/j.dsr.2005.10.007, 2005.
- Cota, G. F., Pomeroy, L. R., Harrison, W. G., Jones, E. P., Peters, F., Sheldon, W. M., and Weingartner, T. R.: Nutrients, primary production and microbial heterotrophy in the southeastern Chukchi Sea: Arctic summer nutrient depletion and heterotrophy, *Mar. Ecol.-Prog. Ser.*, 135, 247–258, doi:10.3354/meps135247, 1996.
- Devol, A. H., Codispoti, L. A., and Christensen, J. P.: Summer and winter denitrification rates in western Arctic shelf sediments, *Cont. Shelf Res.*, 17, 1029–1050, doi:10.1016/S0278-4343(97)00003-4, 1997.
- Fuse, H., Inoue, H., Murakami, K., Takimura, O., and Yamaoka, Y.: Production of free and organic iodine by *Roseovarius* spp., *FEMS Microbiol. Lett.*, 229, 189–194, doi:10.1016/S0378-1097(03)00839-5, 2003.
- Grebmeier, J. M., McRoy, C. P., and Feder, H. M.: Pelagic-benthic coupling on the shelf of the northern Bering and Chukchi Seas. I. Food supply source and benthic biomass, *Mar. Ecol.-Prog. Ser.*, 48, 57–67, 1988.
- Grebmeier, J. M., Feder, H. M., and McRoy, C. P.: Pelagic-benthic coupling on the shelf of the northern Bering and Chukchi Seas, II. Benthic community structure, *Mar. Ecol.-Prog. Ser.*, 51, 253–268, 1989.

# Volatile organic iodine compounds in seawater related to degradation of organic matter

A. Ooki et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



# Volatile organic iodine compounds in seawater related to degradation of organic matter

A. Ooki et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

- Grebmeier, J. M., Bluhm, B. A., Cooper, L. W., Danielson, S., Arrigo, K. R., Blanchard, A. L., Clarke, J. T., Day, R. H., Frey, K. E., Gradinger, R. R., Kedra, M., Konar, B., Kuletz, K. J., Lee, S. H., Lovvorn, J. R., Norcross, B. L., and Okkonen, S. R.: Ecosystem characteristics and processes facilitating persistent macrobenthic biomass hotspots and associated benthivory in the Pacific Arctic, *Prog. Oceanogr.*, 136, 92–114, doi:10.1016/j.pocean.2015.05.006, 2015.
- Happell, J. D. and Wallace, D. W. R.: Methyl iodide in the Greenland/Norwegian Seas and the tropical Atlantic Ocean: evidence for photochemical production, *Geophys. Res. Lett.*, 23, 2105–2108, doi:10.1029/2006JC003963, 1996.
- Hioki, N., Kuma, K., Morita, Y., Sasayama, R., Ooki, A., Kondo, Y., Obata, H., Nishioka, J., Yamashita, Y., Nishino, S., Kikuchi, T., and Aoyama, M.: Laterally spreading iron, humic-like dissolved organic matter and nutrients in cold, dense subsurface water of the Arctic Ocean, *Scientific Reports*, 4, 6775, doi:10.1038/srep06775, 2014.
- Hughes, C., Malin, G., Turley, C. M., Keely, B. J., and Nightingale, P. D.: The production of volatile iodocarbons by biogenic marine aggregates, *Limnol. Oceanogr.*, 53, 867–872, doi:10.4319/lo.2008.53.2.0867, 2008.
- Hydes, D. J., Aoyama, M., Aminot, A., Bakker, K., Becker, S., Coverly, S., Daniel, A., Dickson, A. G., Grosso, O., Kerouel, R., van Ooijen, J., Sato, K., Tanhua, T., Woodward, E. M. S., and Zhang, J. Z.: Determination of dissolved nutrients (N, P, Si) in seawater with high precision and inter-comparability using gas-segmented continuous flow analysers, in: *The GO-SHIP Repeat Hydrography Manual: a Collection of Expert Reports and Guidelines*, ICPO publ. Ser., No. 134, edited by: Hood, E. M., Sabine, C. L., and Sloyan, B. M., IOCCP Rep. 14, U. N. Educ.Sci. and Cult. Organ. Intergov. Oceanotr. Comm., Paris, 1–88, 2010.
- Jones, C. E. and Carpenter, L. J.: Solar photolysis of  $\text{CH}_2\text{I}_2$ ,  $\text{CH}_2\text{ICl}$ , and  $\text{CH}_2\text{IBr}$  in water, salt-water, and seawater, *Environ. Sci. Technol.*, 39, 6130–6137, doi:10.1021/es050563g, 2005.
- Karlsson, A., Theorin, M., and Abrahamsson, K.: Distribution, transport, and production of volatile halocarbons in the upper waters of the ice-covered high Arctic Ocean, *Global Biogeochem. Cy.*, 27, 1246–1261, doi:10.1002/2012GB004519, 2013.
- Kudo, I., Yoshimura, T., Lee, C. W., Yanada, M., and Maita, Y.: Nutrient regeneration at bottom after a massive spring bloom in a subarctic coastal environment, Funka Bay, Japan, *J. Oceanogr.*, 63, 791–801, doi:10.1007/s10872-007-0067-9, 2007.
- Manley, S. L. and de la Cuesta, J. L.: Methyl iodide production from marine phytoplankton cultures, *Limnol. Oceanogr.*, 42, 142–147, doi:10.1029/2006JC003963, 1997.

# Volatile organic iodine compounds in seawater related to degradation of organic matter

A. Ooki et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

- Mattson, E., Karlsson, A., Smith, W. O., and Abrahamsson, K.: The relationship between biophysical variables and halocarbon distributions in the waters of the Amundsen and Ross Seas, Antarctica, *Mar. Chem.*, 140, 1–9, doi:10.1016/j.marchem.2012.07.002, 2012.
- Moore, R. M., Webb, M., Tokarczyk, R., and Wever, R.: Bromoperoxidase and iodoperoxidase enzymes and production of halogenated methanes in marine diatom cultures. *J. Geophys. Res.*, 101, 20899–20908, doi:10.1029/96JC01248, 1996.
- Nishino, S., Shimada, K., and Itoh, M.: Use of ammonium and other nitrogen tracers to investigate the spreading of shelf waters in the western Arctic halocline, *J. Geophys. Res.*, 110, C10005, doi:10.1029/2003JC002118, 2005.
- Ooki, A. and Yokouchi, Y.: Dichloromethane in the Indian Ocean: evidence for in-situ production in seawater, *Mar. Chem.*, 124, 119–124, doi:10.1016/j.marchem.2011.01.001, 2011a.
- Ooki, A. and Yokouchi, Y.: Determination of Henry's law constant of halocarbons in seawater and analysis of sea-to-air flux of iodoethane ( $C_2H_5I$ ) in the Indian and Southern oceans based on partial pressure measurements, *Geochem. J.*, 45, E1–E7, 2011b.
- Ooki, A., Nomura, D., Nishino, S., Kikuchi, T., and Yokouchi, Y.: A global-scale map of isoprene and volatile organic iodine in surface seawater of the Arctic, northwest Pacific, Indian, and Southern Oceans, *J. Geophys. Res.*, doi:10.1002/2014JC010519, 2015.
- Schall, C. and Heumann, K. G.: GC determination of volatile organoiodine and organobromine compounds in Arctic seawater and air samples, *Fresenius J. Anal. Chem.*, 346, 717–722, doi:10.1016/S0967-0645(02)00204-7, 1993.
- Souza, A. C., Gardner, W. S., and Dunton, K. H.: Rates of nitrification and ammonium dynamics in northeastern Chukchi Sea shelf waters, *Deep-Sea Res.*, 102, 68–76, doi:10.1016/j.dsr.2013.12.017, 2014.
- Spicer, C. W., Plastringe, R. A., Foster, K. L., Finlayson-Pitts, B. J., Bottenheim, J. W., Grannas, A. M., and Shepson, P. B.: Molecular halogens before and during ozone depletion events in the Arctic at polar sunrise: concentrations and sources, *Atmos. Environ.*, 36, 2721–2731, doi:10.1016/S1352-2310(02)00125-5, 2002.
- Tokarczyk, R. and Moore, R. M.: Production of volatile organohalogens by phytoplankton cultures, *Geophys. Res. Lett.*, 21, 285–288, doi:10.1029/94GL00009, 1994.
- Welshmeyer, N. A.: Fluorometric analysis of chlorophyll *a* in the presence of chlorophyll *b* and pheopigments, *Limnol. Oceanogr.*, 39, 1985–1992, doi:10.4319/lo.1994.39.8.1985, 1994.
- World Meteorological Organization (WMO): Scientific Assessment of Ozone Depletion: 2010, Global Res. Monit. Proj. Rep. 52, WMO, Geneva, Switzerland, 2010.

# Volatile organic iodine compounds in seawater related to degradation of organic matter

A. Ooki et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

**Table 1.** Hydrographic conditions in the Chukchi Sea and the Arctic Ocean Basin.

St.	Location	Lat.	Long.	Bottom depth (m)	Area type	SST (°C)	SSS	MLD (m)	Surface mixed layer (m)	Subsurface layer (m)	CDW (m)	AW (m)
1	Southern	65.7	191.8	43	Shelf	8.1	24.1	5	0–5 (ACW)	6–36 (ACW)	–	–
10	Chukchi	68.0	192.0	55	Shelf	3.9	28.9	6	0–6 (ACW)	7–22 (ACW), 23–48 (BSAW)	–	–
14	shelf	68.3	193.0	38	Shelf	7.1	28.0	7	0–6 (ACW)	7–32 (ACW)	–	–
80		70.8	192.0	48	Shelf	2.7	31.9	20	0–20 (ACW)	21–22 (ACW), 23–39 (BSAW)	–	–
78		70.7	194.0	40	Shelf	3.1	31.2	11	0–11 (ACW)	12–28 (ACW), 29–33 (BSAW)	–	–
76		70.8	196.0	47	Shelf	4.5	29.6	15	0–15 (ACW)	16–39 (ACW)	–	–
74		70.8	198.0	43	Shelf	4.5	31.1	33	0–33 (ACW)	34–35 (ACW)	–	–
53	Central	71.4	202.5	124	Shelf	3.6	29.7	9	0–9 (ACW)	10–24 (ACW), 25–36 (BSAW)	37–99	> 100
60	Chukchi	71.7	204.9	257	Slope	3.2	30.3	6	0–6 (ACW)	7–25 (ACW), 26–54 (BSAW)	55–111	> 112
48	shelf –	72.5	204.0	1896	Slope	4.0	26.7	9	0–9 (ACW)	10–40 (ACW), 41–70 (BSAW)	71–207	> 208
68	slope	72.9	202.0	1573	Slope	2.6	28.4	14	0–14 (ACW)	15–20 (ACW), 21–55 (BSAW)	56–151	> 152
45		73.3	200.0	1390	Slope	2.7	26.9	10	0–10 (ACW)	11–13 (ACW), 14–81 (BSAW)	82–203	204
72		72.0	200.0	30	Shelf	0.2	28.2	8	0–8 (PMLW)	9–15 (BSAW)	16–20	–
24		73.5	191.3	118	Shelf	–0.2	26.9	9	0–8 (PMLW)	9–32 (BSAW)	32–105	–
29	Northern	75.2	187.4	453	Slope	–0.3	26.6	13	0–13 (PMLW)	14–51 (BSAW)	52–192	> 193
33	slope –	75.2	182.5	721	Slope	0.0	26.7	18	0–18 (PMLW)	19–45 (BSAW)	46–162	> 163
39	basin	75.0	198.1	1967	Slope	0.8	25.7	20	0–20 (PMLW)	21–91 (BSAW)	92–240	> 240
32		76.0	186.0	2137	Basin	–0.4	25.8	23	0–20 (PMLW)	21–83 (BSAW)	84–225	> 226
64		74.5	206.0	3851	Basin	0.1	25.1	17	0–18 (PMLW)	19–105 (BSAW)	106–263	> 264

Shelf area with bottom depth &lt; 200 m, slope area with bottom depth between 200 and 2000 m, basin area with bottom depth &gt; 2000 m.

Sampling stations in the shelf area are displayed in italics.

ACW: Alaskan Coastal Water (SST &gt; +2°C; S &lt; 31.8).

BSAW: Bering Shelf–Anadyr Water (T = –1 ~ –2.0°C).

PMLW: Polar Mixed Layer Water (T &lt; +2°C, S &lt; 30).

Subsurface layer: between ML and CDW.

CDW: Cold, Dense Water (Halocline Layer) (T = –1 ~ –1.7°C; S = 32 ~ 34).

AW: Atlantic Water (S &gt; 34.2).

**Table 2.** Concentrations of Chl *a*.

St.	ML ACW	ML PMLW	bML ACW	bML BSAW	CDW	AW
Southern Chukchi shelf						
1	<b>1.28</b>		<b>1.24</b>			
10	<b>1.14</b>		<b>0.66</b>	0.23		
14	<b>0.98</b>		<b>0.46</b>			
80	<b>0.71</b>			0.21		
78	<b>0.64</b>		<b>0.85</b>	0.22		
76	0.24		0.29			
74	<b>0.73</b>		<b>0.69</b>			
Central Chukchi shelf – slope						
53	<b>0.93</b>		<b>0.59</b>	0.18	0.20	0.21
60	<b>0.97</b>		<b>0.58</b>	0.14	0.12	0.12
48	<b>0.47</b>		<b>0.51</b>		0.08	0.06
68	<b>0.63</b>		<b>0.81</b>	0.20	0.10	0.06
45	0.22			0.28	0.03	0.02
72		<b>0.56</b>		<b>0.69</b>	0.27	
24		0.14		<b>0.46</b>	0.13	
Northern slope – basin						
29		0.14		<b>0.53</b>	0.11	0.07
33		0.13		0.31	<b>0.48</b>	0.03
39		0.16		0.38	0.10	0.06
32		0.08		0.29	0.04	0.01
64		0.12		0.16		0.05

High Chl *a* ( $> 0.42 \mu\text{g L}^{-1}$ ) concentrations are denoted in bold.

# Volatile organic iodine compounds in seawater related to degradation of organic matter

A. Ooki et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

**Table 3.** Concentrations of  $\text{NH}_4^+$  and N-deficit, ( $\text{NH}_4^+$ /N-deficit).

St.	ML ACW	ML PMLW	bML ACW	bML BSAW	CDW	AW
Southern Chukchi shelf						
1	0.91/–10.0		<b>1.03/–10.8</b>			
10	<b>1.38/–9.8</b>		<b>2.73/–10.9</b>	<b>5.77/–15.2</b>		
14	0.31/–9.1		<b>2.08/–9.9</b>			
80	<b>1.01/–7.9</b>			<b>4.58/–16.8</b>		
78	0.47/–7.6		0.52/–8.0	<b>3.35/–14.3</b>		
76	<b>1.01/–8.1</b>		<b>2.28/–11.4</b>			
74	<b>1.40/–11.5</b>		<b>1.61/–11.6</b>			
Central Chukchi shelf-slope						
53	0.74/–10.7		<b>1.49/–12.1</b>	<b>3.39/–15.5</b>	<b>3.83/–15.0</b>	0.67/–7.7
60	0.17/–9.3		<b>1.36/–11.6</b>	<b>1.67/–13.7</b>	<b>2.15/–11.9</b>	0.03/–2.9
48	0.04/–8.5		0.56/–10.2		0.11/–11.1	DL/–2.3
68	0.11/–8.0		0.50/–9.2	<b>1.44/–13.6</b>	<b>3.22/–16.1</b>	0.07/–2.1
45	DL/–8.8			DL/–13.2	DL/–10.9	DL/–2.1
72		DL/–7.5		DL/–8.0	<b>1.34/–15.6</b>	
24		0.05/–9.3		0.13/–11.1	<b>1.81/–20.4</b>	
Northern slope-basin						
29		0.02/–10.0		<b>0.12/–12.5</b>	DL/–15.5	DL/–4.7
33		0.05/–9.6		0.02/–10.2	<b>0.29/–15.5</b>	DL/–2.3
39		0.03/–9.2		0.05/–13.2	0.02/–13.6	0.01/–2.0
32		0.05/–9.1		0.02/–12.4	0.02/–13.6	DL/–2.1
64		0.05/–8.5		0.02/–12.4		DL/–2.2

High  $\text{NH}_4^+$  ( $> 1.0 \mu\text{mol L}^{-1}$ ) concentrations and large N-deficits ( $< -11 \mu\text{mol L}^{-1}$ ) are denoted in bold.

# Volatile organic iodine compounds in seawater related to degradation of organic matter

A. Ooki et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



**Table 4.** Concentrations of  $\text{CH}_3\text{I}$  in each water type.

St.	ML ACW	ML PMLW	bML ACW	bML BSAW	CDW	AW
Southern Chukchi shelf						
1	<b>3.4</b>		<b>5.20</b>			
10	1.7		<b>3.35</b>	<b>4.70</b>		
14	<b>2.5</b>		<b>4.05</b>			
80	1.7			<b>4.05</b>		
78	1.2		1.45	<b>2.75</b>		
76	0.9		2.13			
74	<b>2.2</b>		<b>2.40</b>			
Central Chukchi shelf-slope						
53	1.5		<b>2.30</b>	<b>4.30</b>	<b>4.45</b>	1.10
60	2.0		<b>2.35</b>	<b>2.65</b>	2.10	0.30
48	1.4		<b>3.13</b>		0.65	0.50
68	1.7		<b>2.60</b>	<b>2.65</b>	<b>4.00</b>	0.58
45	1.1			1.50	0.60	0.27
72		1.30		1.50	<b>3.70</b>	
24		1.10		<b>2.57</b>	<b>8.10</b>	
Northern slope-basin						
29		1.00		1.80	1.20	0.75
33		1.15		1.40	1.60	0.63
39		0.93		1.20	0.70	0.50
32		0.80		1.90	0.60	0.37
64		0.87		1.03		0.48

High  $\text{CH}_3\text{I}$  ( $> 2.2 \text{ pmol L}^{-1}$ ) concentrations are denoted in bold.

# Volatile organic iodine compounds in seawater related to degradation of organic matter

A. Ooki et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



**Table 5.** Concentrations of  $C_2H_5I$  in each water type.

St.	ML ACW	ML PMLW	bML ACW	bML BSAW	CDW	AW
Southern Chukchi shelf						
1	0.50		<b>1.58</b>			
10	0.45		<b>1.45</b>	<b>2.70</b>		
14	0.30		0.70			
80	0.53			<b>3.90</b>		
78	0.30		0.35	<b>2.55</b>		
76	0.17		<b>1.40</b>			
74	0.83		<b>0.90</b>			
Central Chukchi shelf-slope						
53	0.40		<b>0.85</b>	<b>2.70</b>	<b>2.95</b>	0.70
60	0.30		<b>0.85</b>	<b>1.30</b>	<b>1.90</b>	0.10
48	0.15		0.43		0.30	0.07
68	0.20		0.50	<b>1.15</b>	<b>4.40</b>	0.10
45	0.13			0.17	0.25	0.07
72		0.23		0.30	<b>2.10</b>	
24		0.10		0.17	<b>1.50</b>	
Northern slope-basin						
29		0.10		0.25	0.70	0.20
33		0.05		0.10	0.55	0.00
39		0.13		0.20	0.25	0.10
32		0.10		0.15	0.45	0.07
64		0.10		0.17		0.08

High  $C_2H_5I$  ( $> 0.84 \text{ pmol L}^{-1}$ ) concentrations are denoted in bold.

# Volatile organic iodine compounds in seawater related to degradation of organic matter

A. Ooki et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



**Table 6.** Concentrations of  $\text{CH}_2\text{I}_2$  in each water type.

St.	ML ACW	ML PMLW	bML ACW	bML BSAW	CDW	AW
Southern Chukchi shelf						
1	<b>47.40</b>		<b>30.00</b>			
10	0.65		2.45	4.00		
14	<b>7.20</b>		<b>9.95</b>			
80	2.63			<b>9.10</b>		
78	3.80		<b>6.10</b>	<b>6.30</b>		
76	0.80		<b>6.20</b>			
74	<b>9.38</b>		<b>13.90</b>			
Central Chukchi shelf-slope						
53	2.77		<b>5.40</b>	<b>10.10</b>	<b>9.80</b>	1.70
60	4.20		<b>5.55</b>	4.20	<b>7.60</b>	0.20
48	4.10		<b>11.30</b>		0.60	0.27
68	3.60		<b>6.90</b>	<b>5.15</b>	<b>13.40</b>	0.20
45	1.50			1.60	0.85	0.10
72		0.27		1.00	<b>5.90</b>	
24		0.10		2.37	<b>20.90</b>	
Northern slope-basin						
29		0.00		<b>7.40</b>	1.30	0.45
33		0.00		0.60	2.30	0.03
39		0.05		1.45	0.15	0.00
32		0.03		1.55	0.35	0.03
64		0.00		0.47		0.04

High  $\text{CH}_2\text{I}_2$  ( $> 4.4 \text{ pmol L}^{-1}$ ) concentrations are denoted in bold.

**Table 7.** Concentrations of CH<sub>2</sub>ClI in each water type.

St.	ML ACW	ML PMLW	bML ACW	bML BSAW	CDW	AW
Southern Chukchi shelf						
1	<b>10.95</b>		<b>7.70</b>			
10	<b>5.90</b>		<b>7.10</b>	3.95		
14	<b>17.80</b>		<b>13.35</b>			
80	<b>5.97</b>			4.80		
78	<b>8.37</b>		<b>9.40</b>	<b>5.65</b>		
76	3.50		4.83			
74	<b>12.68</b>		<b>14.60</b>			
Central Chukchi shelf-slope						
53	<b>6.93</b>		<b>6.05</b>	<b>5.10</b>	4.30	0.70
60	<b>9.60</b>		<b>6.55</b>	3.10	2.90	0.20
48	<b>10.60</b>		<b>7.50</b>		0.50	0.13
68	<b>10.45</b>		<b>8.00</b>	3.10	4.60	0.08
45	4.23			1.37	0.60	0.00
72		2.00		3.50	<b>5.30</b>	
24		0.50		3.70	<b>11.00</b>	
Northern slope-basin						
29		0.20		2.10	1.90	0.25
33		0.25		1.30	2.15	0.13
39		0.58		1.10	0.55	0.00
32		0.17		1.35	0.70	0.03
64		0.23		1.00		0.02

High CH<sub>2</sub>ClI (> 5.0 pmol L<sup>-1</sup>) concentrations are denoted in bold.

**BGD**

12, 11245–11278, 2015

**Volatile organic  
iodine compounds in  
seawater related to  
degradation of  
organic matter**

A. Ooki et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



# Volatile organic iodine compounds in seawater related to degradation of organic matter

A. Ooki et al.

**Table 8.** Correlate coefficients  $R$  (all water types ( $n = 147$ )/warm ACW ( $n = 54$ )/cold BSAW-PMLW-CDW ( $n = 64$ )).

	CH <sub>3</sub> I	C <sub>2</sub> H <sub>5</sub> I	CH <sub>2</sub> I <sub>2</sub>	CH <sub>2</sub> ClI	NH <sub>4</sub> <sup>+</sup>	Chl <i>a</i>	N-deficit
CH <sub>3</sub> I		<b>0.73/0.71/0.76</b>	0.55/0.47/ <b>0.84</b>	0.55/0.21*/ <b>0.91</b>	<b>0.70/0.43/ 0.77</b>	0.41/0.28*/0.09*	<b>0.61/0.49/0.63</b>
C <sub>2</sub> H <sub>5</sub> I	<b>0.73/0.71/0.76</b>		0.27/0.08*/ <b>0.73</b>	0.24/0.10*/ <b>0.70</b>	<b>0.90/0.79/0.93</b>	0.05*/0.07*/0.11*	0.59/ <b>0.76/0.61</b>
CH <sub>2</sub> I <sub>2</sub>	0.55/0.47/ <b>0.84</b>	0.27/0.08*/ <b>0.73</b>		0.50/0.27*/ <b>0.85</b>	0.23/0.05*/ <b>0.64</b>	<b>0.62/0.64/0.13*</b>	0.22/0.01*/ <b>0.67</b>
CH <sub>2</sub> ClI	0.55/0.21*/ <b>0.91</b>	0.24/0.10*/ <b>0.70</b>	0.50/0.27*/ <b>0.85</b>		0.31/0.10*/ <b>0.66</b>	<b>0.69/0.30*/0.18*</b>	0.29/0.03*/ <b>0.63</b>
NH <sub>4</sub> <sup>+</sup>	<b>0.70/0.43/0.77</b>	<b>0.90/0.79/0.93</b>	0.23/0.05*/0.64	0.31/0.10*/0.66		0.06/0.32*/0.08*	0.52/ <b>0.68/0.55</b>
Chl <i>a</i>	0.41/0.28*/0.09*	0.05*/0.07*/0.11*	0.62/0.64/0.13*	0.69/0.30*/0.18*	0.06/0.32*/0.08*		0.20*/0.18*/0.04*
N-deficit	0.61/0.49/0.63	0.59/ <b>0.76/0.61</b>	0.22/0.01*/0.67	0.29/0.03*/0.63	0.52/0.68/0.55	0.20*/0.18*/0.04*	

All water types: ACW + BSAW + PMLW + CDW + AW.  
 $\rho$  value < 0.01, \*  $\rho$  value > 0.01.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

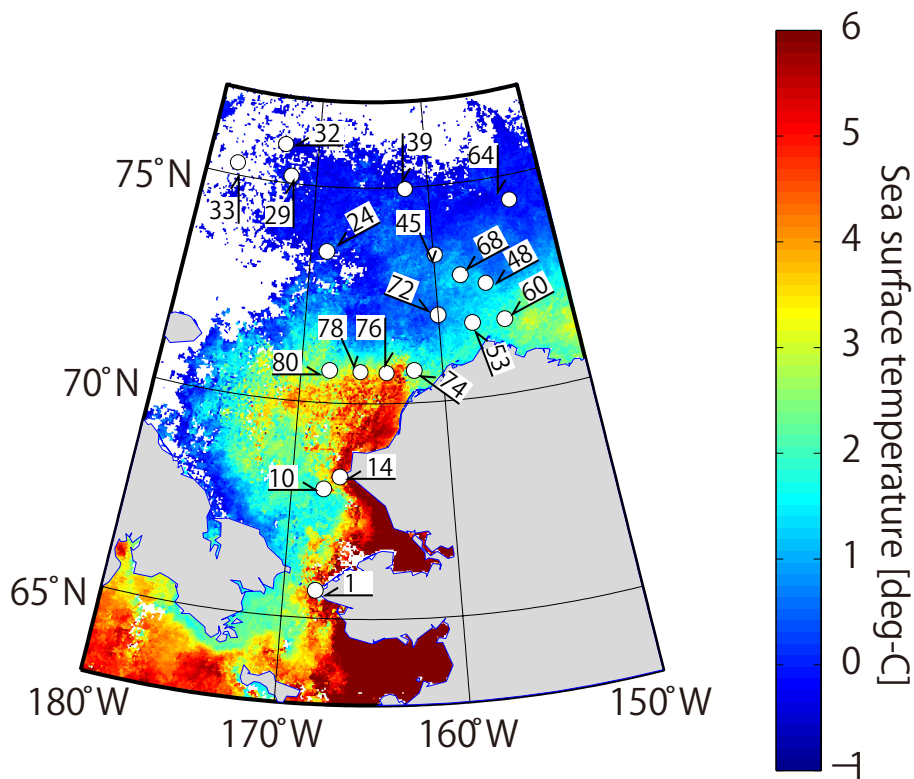
Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



**Figure 1.** Sampling stations in the present study and sea surface temperature (color) by satellite images on October 2012 (NASA Goddard Space Flight Center Ocean Color; <http://oceancolor.gsfc.nasa.gov/cms/>).

# Volatile organic iodine compounds in seawater related to degradation of organic matter

A. Ooki et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

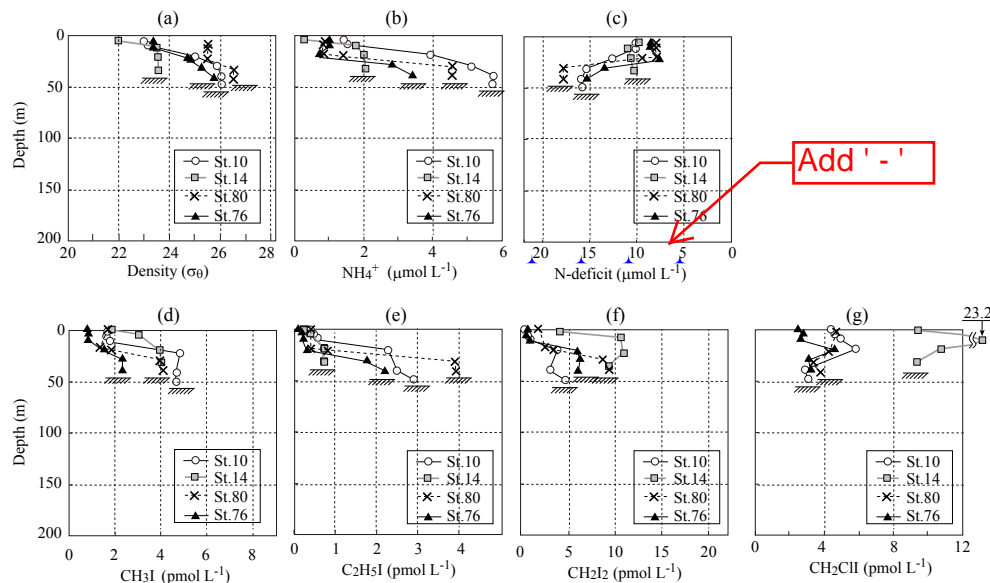
Full Screen / Esc

Printer-friendly Version

Interactive Discussion

# Volatile organic iodine compounds in seawater related to degradation of organic matter

A. Ooki et al.



**Figure 2.** Vertical distributions of potential density (a),  $\text{NH}_4^+$  (b), N-deficit (c),  $\text{CH}_3\text{I}$  (d),  $\text{C}_2\text{H}_5\text{I}$  (e),  $\text{CH}_2\text{I}_2$  (f),  $\text{CH}_2\text{ClI}$  (g) at St10, St14, St80, and St76 over the southern Chukchi Sea shelf.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

# Volatile organic iodine compounds in seawater related to degradation of organic matter

A. Ooki et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

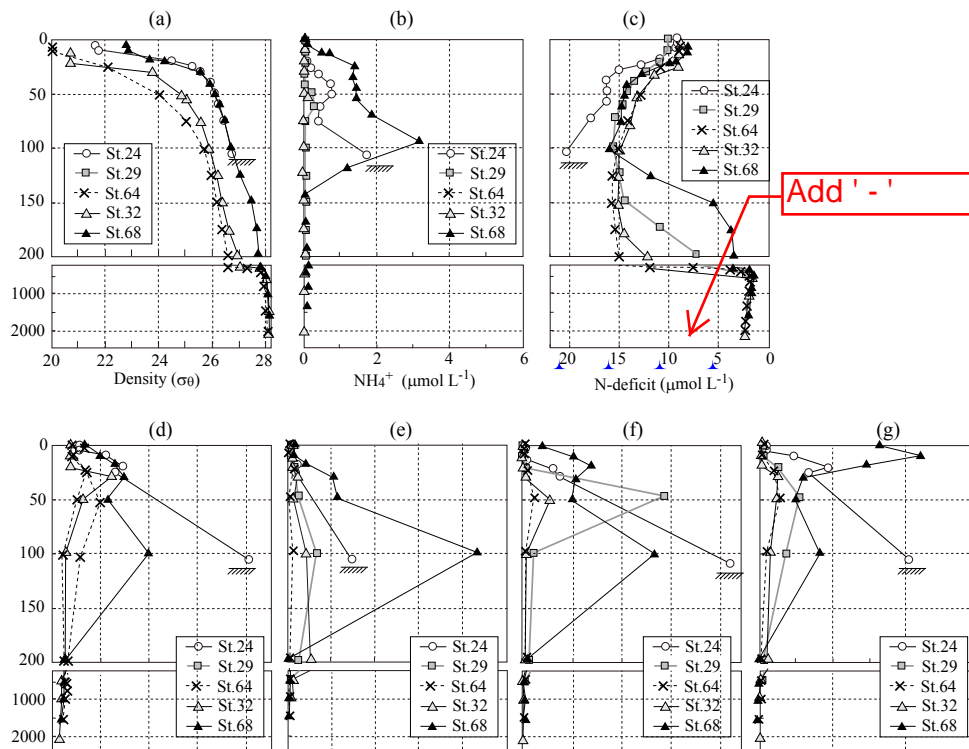
Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



**Figure 3.** Vertical distributions of potential density (a),  $\text{NH}_4^+$  (b), N-deficit (c),  $\text{CH}_3\text{I}$  (d),  $\text{C}_2\text{H}_5\text{I}$  (e),  $\text{CH}_2\text{I}_2$  (f),  $\text{CH}_2\text{ClI}$  (g) at St24, St29, St64, St32, and St68 over the central Chukchi Sea shelf – slope area and the northern slope – basin area.

# Volatile organic iodine compounds in seawater related to degradation of organic matter

A. Ooki et al.

Title Page

Abstract

Introduction

Conclusions

References

Tables

Figures

◀

▶

◀

▶

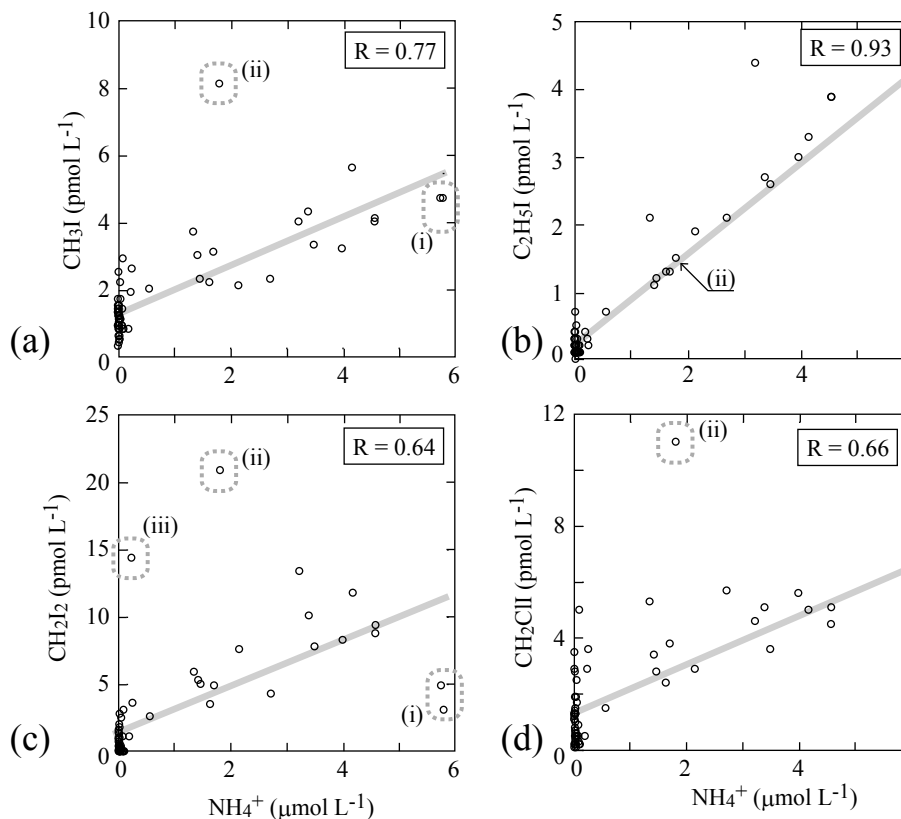
Back

Close

Full Screen / Esc

Printer-friendly Version

Interactive Discussion



**Figure 4.** Scatter plots of  $\text{CH}_3\text{I}$  and  $\text{NH}_4^+$  **(a)**,  $\text{C}_2\text{H}_5\text{I}$  and  $\text{NH}_4^+$  **(b)**,  $\text{CH}_2\text{I}_2$  and  $\text{NH}_4^+$  **(c)**, and  $\text{CH}_2\text{ClI}$  and  $\text{NH}_4^+$  **(d)**. The data for St10 in the subsurface bottom layer water of the Hope Valley biological hotspot are enclosed with a dashed line (i); data for St24 in the bottom layer water over the northern tip of the Chukchi Sea shelf is enclosed with a dashed line (ii); and data for St29 in the subsurface BSAW (48 m) over the northern slope area is enclosed with a dashed line (iii).