

1 **Carbon dynamics in highly heterotrophic subarctic thaw ponds**

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12

13 **Abstract**

14 Global warming has accelerated the formation of permafrost thaw ponds in several subarctic and  
15 arctic regions. These ponds are net heterotrophic as evidenced by their greenhouse gas (GHG)  
16 supersaturation levels (CO<sub>2</sub> and CH<sub>4</sub>), and generally receive large terrestrial carbon inputs from the  
17 thawing and eroding permafrost. We measured seasonal and vertical variations in the concentration  
18 and type of dissolved organic matter (DOM) in five subarctic thaw (thermokarst) ponds in northern  
19 Quebec, and explored how environmental gradients influenced heterotrophic and phototrophic  
20 biomass and productivity. Late winter DOM had low aromaticity indicating reduced inputs of  
21 terrestrial carbon, while the high concentration of dissolved organic carbon (DOC) suggests that  
22 some production of non-chromophoric dissolved compounds by the microbial food web took place  
23 under the ice cover. Summer DOM had a strong terrestrial signature, but was also characterized  
24 with significant inputs of algal-derived carbon, especially at the pond surface. During late winter,  
25 bacterial production was low (maximum of 0.8 mg C m<sup>-3</sup> d<sup>-1</sup>) and was largely based on free-living  
26 bacterioplankton (58%). Bacterial production in summer was high (up to 58 mg C m<sup>-3</sup> d<sup>-1</sup>),  
27 dominated by particle-attached bacteria (67%), and strongly correlated to the amount of terrestrial  
28 carbon. Primary production was restricted to summer surface waters due to strong light limitation  
29 deeper in the water column or in winter. The phototrophic biomass was equal to the heterotrophic  
30 biomass, but as the algae were mostly composed of mixotrophic species, most probably they used  
31 bacteria rather than solar energy in such shaded ponds. Our results point to a strong heterotrophic  
32 energy pathway in these thaw pond ecosystems, where bacterioplankton dominates the production  
33 of new carbon biomass in both summer and winter.

34

## 35 **1 Introduction**

36 Traditional view of inland waters as sinks of carbon has changed during the past decades, and at  
37 present, lakes and ponds are considered net sources of carbon to the atmosphere (Tranvik et al.,  
38 2009). Increasing attention has been given to thaw ponds and lakes after recognizing the cumulative  
39 effect of their high abundance and greenhouse gas (GHG) emissions on global warming (Kling et  
40 al., 1992; Walter et al., 2006; Laurion et al., 2010), especially when they emit old carbon thus  
41 having the potential to act as a positive feedback mechanism on climate (Walter Anthony et al.,  
42 2014). Thaw ponds and lakes are dominant in continuous and discontinuous permafrost areas, for  
43 example in permafrost regions of Siberia where they represent 90% of all lakes (Walter et al.,  
44 2006), but until recently, little has been known on the limnological properties and microbial  
45 communities of this important freshwater ecosystem (Vonk et al., 2015).

46 Subarctic thaw ponds can be highly turbid and have relatively high nutrient concentrations (Breton  
47 et al., 2009; Rautio et al., 2011a) compared to most high latitude freshwater ecosystems (Pienitz et  
48 al., 1997; Hamilton et al., 2001; Medeiros et al., 2012). Moreover, high perimeter/volume ratio, the  
49 presence of palsas and bogs and the thawing of organic-rich permafrost in their watershed favor  
50 high inputs of dissolved organic matter (DOM) to the ponds. Chromophoric DOM (CDOM) can  
51 influence water column temperature regimes through its absorption of sunlight (Caplanne and  
52 Laurion, 2008), and with suspended particles, they contribute to form a distinct and stable  
53 thermocline (Laurion et al., 2010). The rate of DOM input to aquatic systems has been documented  
54 to increase in boreal lakes over the last decades (Hudson et al., 2003), and while no such  
55 information is yet available for thaw ponds, the recent mobilization of terrestrial carbon stocks  
56 stored for thousands of years in the permafrost (Vonk et al., 2012) suggests a similar DOM increase  
57 is taking place also in the North.

58 High DOM concentration also strongly regulates the depth of the euphotic zone. Light is usually  
59 limited to the first meter in thaw ponds (Squire and Lesack, 2003; Watanabe et al., 2011), therefore  
60 restricting the water volume where photosynthesis can occur. This type of environment favors  
61 heterotrophic bacterial production (BP) that was shown to reach similar rates as in eutrophic lakes  
62 (Breton et al., 2009). The limnological characteristics of thaw ponds are likely also beneficial to  
63 mixotrophic algae and heterotrophic protozoans, as flagellated species are generally well adapted  
64 when lakes are rich in humic compounds (Jones, 2000). In low light conditions they can utilize

65 bacteria as an energy source, and as they are motile they also benefit from the nutrient-rich bottom  
66 waters (Arvola et al., 1991). They have been shown to be a key link in the carbon transfer from low  
67 to high trophic levels in small boreal forest lakes (Salonen and Rosenberg, 2000).

68 Circumpolar thaw ponds are ice covered for a large fraction of the year, which further affects  
69 carbon cycling, and generates large seasonal variations. The aim of this study was to demonstrate  
70 the difference in GHG accumulation, CDOM characteristics, and autotrophic and heterotrophic  
71 carbon pools between two seasons and depths. Comparisons were made between late winter and  
72 summer, and between surface and bottom waters in summer. We hypothesized that in the absence  
73 of light in winter or in the aphotic summer hypolimnion, autotrophic productivity stops while  
74 respiration continues, resulting in large summer and winter storage of GHG in the water column.  
75 We further hypothesized that the thermal structure in summer generates an accumulation of organic  
76 carbon in the hypolimnion where heterotrophy dominates. This study follows earlier ones looking at  
77 the limnological characteristics and bacterial communities of subarctic thaw ponds in summer  
78 (Breton et al., 2009; Negandhi et al., 2013; Rossi et al., 2013; Crevecoeur et al., 2015).

79

## 80 **2 Methods**

### 81 **2.1 Sample collection and limnological analyses**

82 The study was carried out in the discontinuous permafrost region near the village of  
83 Whapmagoostui-Kuujuarapik (55° 20' N 77° 30' W), Northern Quebec, Canada, along the River  
84 Kwakwatanikapistikw (local Cree name; KWK is used to name the ponds hereafter) where a large  
85 number of thaw ponds can be found. At this site, ponds are thermokarstic and form over an  
86 impermeable clay-silt bed, preventing most hydrological interactions among ponds, on center of  
87 thawed lithalsas (or inorganic palsas). They are surrounded by dense shrubs and sparse trees, and  
88 occasional areas of mosses and aquatic plants. More details on the study area are provided in  
89 Bouchard et al. (2014) and Bouchard et al. (2011).

90 Five ponds were studied during late winter (6-11 April) and summer (8-14 August) in 2009. They  
91 are all small (averaged diameter 17 m  $\pm$  1.3 m) and shallow (average depth 2.5 m  $\pm$  0.4 m).  
92 Sampling was done with a Kemmerer water sampler (Wildco® USA) through a borehole drilled  
93 into the ice in winter, and from an inflatable boat in summer at the middle of the pond. During

94 winter sampling, the water started pouring over the ice through the hole because of the pressure  
95 from the weight of snow and ice, and possibly resulting in a mixing of surface with deeper water.  
96 This was suggested as the factor explaining the similar GHG concentrations obtained in surface  
97 (just under the ice) and bottom waters (approximately 0.5 m above the sediments), despite the  
98 expected inverse thermal stratification at this period of the year (Laurion et al., 2010). All other  
99 samples were collected from about 1 m below the ice and considered as representative of an  
100 integrated water column. In summer, surface and bottom waters were sampled from just below the  
101 surface and approximately 0.5 m above the bottom sediments.

102 Measures of temperature, dissolved oxygen, pH and conductivity were done with a 600R YSI  
103 multiparametric probe (Yellow Springs Inc.). Further analyses were performed on water prefiltered  
104 through a 50- $\mu\text{m}$  sieve. Total suspended solids (TSS) were obtained by filtering 500-1000 mL of  
105 water onto precombusted and preweighted GF/F filters (Advantec MFS Inc.). The 50- $\mu\text{m}$   
106 prefiltration reduced the TSS by an average 5.8% in comparison to bulk water samples, both  
107 analyzed in summer (data not shown). Filters were stored at  $-20^{\circ}\text{C}$  until they were dried for 24 h at  
108  $60^{\circ}\text{C}$ , and weighted to get sestonic dry weight or TSS. They were subsequently combusted at  $450^{\circ}\text{C}$   
109 for 2 h to be able to calculate the organic fraction of TSS. An aliquot of water was filtered through a  
110 pre-rinsed cellulose acetate filter (0.2  $\mu\text{m}$  pore size; Advantec MFS Inc.) to analyse soluble reactive  
111 phosphorus (SRP; unpreserved samples) and iron (Fe; samples preserved with  $\text{HNO}_3$  at 0.15% final  
112 concentration). Total phosphorus (TP) and total nitrogen (TN) were analysed from unfiltered water  
113 preserved with  $\text{H}_2\text{SO}_4$  (final concentration of 0.15%). All nutrient samples were stored in acid-  
114 washed glass bottles in dark and cold ( $4^{\circ}\text{C}$ ) until further analysis as in Breton et al. (2009).

115

## 116 **2.2 Carbon characterization**

117 Dissolved  $\text{CO}_2$  and  $\text{CH}_4$  were determined as in Laurion et al. (2010). In brief, 2 L of pond water  
118 were equilibrated into 20 mL of ambient air for 3 min, with the headspace sampled in duplicated  
119 glass vials (Vacutainers<sup>®</sup>) previously flushed with helium and vacuumed. Gas samples were  
120 analyzed by gas chromatography (Varian 3800), and dissolved gas concentrations were calculated  
121 according to Henry's Law.

122 The water filtered through pre-rinsed cellulose acetate filters as above was also used to analyze  
123 dissolved organic carbon concentration (DOC), and for optical analyses of CDOM. Samples were

124 stored in amber glass bottles in dark and cold (4°C) until the analysis. DOC was quantified using a  
125 carbon analyzer (Shimadzu TOC-5000A) calibrated with potassium biphthalate. The absorbance  
126 was measured from 200 to 800 nm with a spectrophotometer (Varian Cary 300), using 1-cm quartz  
127 cuvettes on dual-beam mode, at 1 nm intervals for winter samples, and 4 nm intervals for summer  
128 samples. Null-point adjustment was performed using the mean value from 750-800 nm, and the  
129 absorption coefficients ( $a_\lambda$ ) were calculated from absorbance measurements ( $A_\lambda$ ) at 254 and 320 nm  
130 using  $a_\lambda = 2.303 A_\lambda/L$ , where L is the length of the cuvette in meters (Mitchell et al., 2002). The  
131 absorption coefficient at 320 nm ( $a_{320}$ ) was used as an index of CDOM concentration. Specific UV  
132 absorbance–index ( $SUVA_{254}$ ) was determined from DOC normalized  $A_{254}$ , and used as an index of  
133 aromaticity (Weishaar et al., 2003).

134 CDOM absorption spectra were further analyzed to calculate the spectral slopes  $S_\lambda$  (Loiselle et al.,  
135 2009) between 250 nm and 450 nm (Galgani et al., 2011). Slopes were calculated over 20 nm  
136 intervals with a 1 nm step (i.e. 250-269, 251-270,...) for winter samples, and 4 nm steps for  
137 summer samples (i.e. 250-269, 254-273,...). The resulting set of spectral slopes was plotted by  
138 center wavelengths. Calculations were performed in open source software package SciLab 4.15.  
139 The individual spectral slope  $S_{289}$  was used to evaluate the amount of fulvic and humic acids likely  
140 related to the autochthonous production (Loiselle et al., 2009). Algal-derived carbon has a  
141 maximum slope at 289 nm, while terrestrial carbon spectral slopes lack this peak and present a  
142 spectrally variable distribution with a steady increase from 260 to 390 nm.

143

### 144 **2.3 Microbial abundance and productivity**

145 For the determination of chlorophyll *a* (Chl-*a*) concentrations, 1-2 L of 50 µm-sieved pond water  
146 were filtered onto GF/F filters. Samples were collected in duplicates and stored at -80°C until  
147 pigment extraction and fluorometric analysis were done in hot ethanol according to Nush (1980).  
148 Water samples for the enumeration of nanoflagellates (PNF: phototrophic and mixotrophic cells;  
149 HNF: heterotrophic cells) and phototrophic picoplankton (PPA) were preserved with 0.2 µm-  
150 filtered glutaraldehyde solution at a final concentration of 1%. Three replicate aliquots (3-5 ml)  
151 were filtered through 0.6 µm black polycarbonate filters (Nuclepore). Samples were stained with 4-  
152 ,6-diamido-2-phenylindole (DAPI), mounted to slides and stored at -20°C until counting under UV  
153 excitation with an epifluorescence microscope (Zeiss Axiovert 200) at 1000× magnification.

154 Discrimination between pigmented and heterotrophic nanoflagellates was done with green  
155 excitation light. At least 50 cells or a maximum of 30 fields were counted from each sample.  
156 Nanoflagellate cells were further divided into three size categories:  $< 5 \mu\text{m}$ ,  $5\text{-}10 \mu\text{m}$  and  $> 10 \mu\text{m}$ .  
157 Volumes were converted to carbon biomass using the carbon content coefficient of  $0.19 \text{ pg C } \mu\text{m}^{-3}$   
158 for PNF and HNF (Putt and Stoecker, 1989). Phototrophic picoplankton carbon content was  
159 calculated using the carbon conversion equation ( $\text{pg C} = 0.433 \times V^{0.866}$ , where  $V$  is in  $\mu\text{m}^3$ ) for  
160 picoeukaryotic cells (Verity et al., 1992; Campbell et al., 1994).

161 Bacterial abundance (BA) was measured with a flow cytometer (FACSCalibur, Becton-Dickinson)  
162 as in Rossi et al. (2013) before and after sonication to obtain estimations of free-living and particle-  
163 attached bacterial abundance. Biomass calculation for bacterioplankton was based on biovolumes  
164 measured from digital images acquired from DAPI-stained microscope slides using the *Cell C*  
165 program (Selinummi et al., 2005). Average bacterial carbon content was first estimated with  
166 allometric conversion formula recommended for DAPI stained cells ( $\text{fgC} = 218 \times V^{0.86}$ , where  $V$  is  
167 in  $\mu\text{m}^3$ ) (Posch et al., 2001). The average cellular carbon content ( $18.1 \text{ fg C cell}^{-1}$ ) was then  
168 multiplied by the bacterial abundance.

169 Bacterial production was measured from unfiltered and filtered ( $3 \mu\text{m}$ ) water samples using tritiated  
170 leucine ( $^3\text{H}$ -leucine) incorporation with centrifugation (Smith and Azam, 1992). A solution of  $^3\text{H}$ -  
171 leucine (specific activity of  $164 \text{ Ci mmol}^{-1}$ ) was added to each vial to obtain a final leucine  
172 concentration of  $30 \text{ nM}$ , at which bacteria were saturated (experimentally tested). During the winter,  
173 the samples were incubated at  $\sim 0^\circ\text{C}$  in the dark for 2 hours. During the summer, samples were  
174 incubated at *in situ* temperatures ( $5\text{-}18^\circ\text{C}$ ), depending on ponds. BP was stopped by adding  
175 trichloroacetic acid ( $5\%$  final concentration), and samples were then stored at  $-20^\circ\text{C}$  before the  
176 centrifugation step. Winter samples were radioassayed with a Perkin Elmer Tri-Carb 2800, and  
177 summer samples with a Beckman LS 6500. Leucine incorporation was converted to carbon biomass  
178 using coefficients from Simon and Azam (1989). Bacterial respiration (BR) was estimated from BP  
179 according to Del Giorgio and Cole (1998) BR model II.

180 Primary production (PP) was measured from the mixed water column during winter and from the  
181 surface in summer. An incubation system generating a gradient of photosynthetically active  
182 radiation (PAR), introduced by Rae and Vincent (1998), was used to obtain photosynthesis versus  
183 irradiance curves from complete darkness to full sunlight ( $E_{\text{max}}$ ). The value for  $E_{\text{max}}$  was calculated

184 from *in-situ* PAR measurements available from a weather station located in Whapmagoostui-  
185 Kuujjuarapik, and assuming the air-water surface in these humic waters to cause a 50% decrease in  
186 radiation levels (Doxaran et al., 2004). A water volume of 20 mL was inoculated with a working  
187 solution of [<sup>14</sup>C]-HCO<sub>3</sub> (80 μCi ml<sup>-1</sup>) to a final concentration of 0.2 μCi ml<sup>-1</sup>. In winter, the  
188 incubations were done outside (~0°C) under natural light conditions in a water bath for 45-60 min,  
189 while in summer a circulating water bath was used to keep samples at *in situ* temperature (14 – 18  
190 °C) and incubations were run for 120 min. The incubations were terminated by filtrating samples  
191 onto GF/F filters, the method only capturing <sup>14</sup>C fixed in particulate organic matter and not the  
192 carbon allocated to exudates. The filters were subsequently frozen before they were radioassayed  
193 with a Perkin Elmer Tri-Carb 2800 liquid scintillation counter. Chl-a normalized carbon fixation  
194 rates were fitted to the equation of Platt et al. (1980) or Jassby and Platt (1976) depending on the  
195 presence or absence of photoinhibition, respectively. Data fit to the equations had on average a r<sup>2</sup>  
196 value of 0.92, and always higher than 0.81. The iterative non-linear regression of SigmaPlot 11.0  
197 was used to obtain the maximum photosynthetic rate (P<sub>max</sub>). PP at the pond bottom was estimated  
198 from the photosynthesis-irradiance curves and PAR estimated from *in situ* surface measurements,  
199 and diffuse attenuation coefficients calculated from a linear regression using TSS and DOC  
200 (Watanabe et al., 2011). Without light under the ice and snow cover, PP was established at zero.  
201 Light was measured under the ice with a Li-Cor Li-192 submersible PAR Quantum sensor.

202

## 203 **2.4 Data analysis**

204 Differences in environmental and microbial variables among waters were tested using a one-way  
205 ANOVA (fixed factor; JMP software, version 11.0.0, SAS Institute Inc. 2013). Conductivity, BP,  
206 PPA, SRP, and PNF biomass were log-transformed, while TP, TN, and BB were square-root  
207 transformed to achieve normality and homogeneity of variance in residuals (visually checked with  
208 predicted values). When differences were detected, a Tukey HSD multiple comparison test was  
209 performed. Environmental variables are illustrated with a principal component analysis (PCA)  
210 (PAST version 2.17c), where biological variables were entered as vectors which length and  
211 direction were determined by Spearman's correlation analysis between biological variables and  
212 PCA axis 1 and 2. Correlations between organisms (abundance, biomass and productivity  
213 measurements) and environmental variables were tested with non-parametric Spearman's rank



214 correlation coefficient analysis (IBM SPSS statistic 20.0). A significant level of  $\alpha = 0.05$  was used  
215 for all statistical tests.

216

## 217 **3 Results**

### 218 **3.1 Physicochemical properties of thaw ponds**

219 In late winter, the ponds were covered with 50 to 70 cm of snow and approximately 30 to 60 cm of  
220 ice, which absorbed all light. Water temperature was 2.0°C under the ice and the ponds were likely  
221 O<sub>2</sub> depleted at the bottom because of the smell of hydrogen sulfide, while O<sub>2</sub> concentration was 4.5  
222  $\pm 3.5$  mg L<sup>-1</sup> in the integrated water column sample. In summer, despite their shallow depth, ponds  
223 are strongly stratified with nearly 10°C difference in temperature and 10 mg L<sup>-1</sup> difference in O<sub>2</sub>  
224 between surface and bottom (Fig. 1). Some ponds (KWK 2, 6 and 20) had a thermally homogenous  
225 epilimnion until 1 to 1.5 m depth, while in KWK 12 and 23 the temperature declined nearly linearly  
226 towards the bottom with a rate of 1-2°C every 50 cm. The pH varied between 6.0 and 7.5, with  
227 lowest and highest values respectively measured in summer bottom and surface waters. There was  
228 no difference ( $p = 0.9034$ ) in conductivity between winter ( $50 \pm 7$   $\mu$ S/cm) and summer at the  
229 surface ( $46 \pm 10$   $\mu$ S/cm), but bottom waters had significantly ( $p < 0.009$ ) higher conductivity ( $194 \pm$   
230  $67$   $\mu$ S/cm) than winter and summer surface waters. Nutrients were relatively high in all samples.  
231 TN concentrations were elevated in winter ( $958 \pm 235$   $\mu$ g N L<sup>-1</sup>) compared to the summer values  
232 ( $264 \pm 118$   $\mu$ g N L<sup>-1</sup>) ( $p = 0.0001$ ). TP was relatively high in winter ( $97 \pm 43$   $\mu$ g P L<sup>-1</sup>), and in  
233 summer the concentrations were clearly lower at the surface ( $55 \pm 29$   $\mu$ g P L<sup>-1</sup>) than at the bottom  
234 ( $3110 \pm 1043$   $\mu$ g P L<sup>-1</sup>) ( $p = 0.0001$ ). SRP was available during the whole year, with higher  
235 concentrations in late winter ( $26 \pm 15$   $\mu$ g P L<sup>-1</sup>) and summer bottom waters ( $31 \pm 36$   $\mu$ g P L<sup>-1</sup>) than  
236 in summer surface where concentrations were relatively low ( $3.1 \pm 3.0$   $\mu$ g P L<sup>-1</sup>) but not statistically  
237 different from the other values ( $p = 0.064$ ). All ponds had higher TSS in bottom waters ( $85 \pm 43$  mg  
238 L<sup>-1</sup>), while in winter and summer surface waters the concentrations were generally an order of  
239 magnitude lower. In most ponds, organic fraction of TSS followed the same trend as TSS and made  
240 on average 24% of TSS. Fe was lower at the surface ( $0.3 \pm 0.1$  mg L<sup>-1</sup>) than at the bottom of ponds  
241 ( $3.1 \pm 1.6$  mg L<sup>-1</sup>) in summer. No Fe measurements were done during late winter. Limnological  
242 properties of the ponds are summarized in Table 1.

243

### 244 **3.2 Carbon characterization**

245 CO<sub>2</sub> and CH<sub>4</sub> concentrations were high in winter (up to 357 and 14.1 μM respectively), and  
246 particularly high at the pond bottom in summer (up to 815 and 312 respectively; Table 2). Surface  
247 waters, in contact with the atmosphere, had one order of magnitude lower concentrations, but values  
248 remained supersaturated. The inverse relationship ( $r = -0.9228$ ;  $p < 0.0001$ ) between O<sub>2</sub> and CO<sub>2</sub>  
249 suggests that the CO<sub>2</sub> originated from O<sub>2</sub> respiration and not from external sources such as  
250 weathering, runoff and groundwater inputs that are unlikely in the studied ponds. DOC ranged from  
251 4.1 to 10.5 mg L<sup>-1</sup>, with the highest concentrations obtained in winter. Despite the higher DOC  
252 observed in winter, CDOM was lowest at this period, as indicated by a<sub>320</sub> values ( $7.1 \pm 2.5 \text{ m}^{-1}$   
253 compared to  $27 \pm 10$  and  $70 \pm 37 \text{ m}^{-1}$  at the surface and bottom in summer, respectively). SUVA<sub>254</sub>  
254 followed the same trend as a<sub>320</sub>, with lower values in winter ( $0.8 \pm 0.2 \text{ L mg C}^{-1} \text{ m}^{-1}$ ) compared to  
255 summer surface ( $4.8 \pm 1.0 \text{ L mg C}^{-1} \text{ m}^{-1}$ ) and bottom waters ( $7.7 \pm 2.4 \text{ L mg C}^{-1} \text{ m}^{-1}$ ), indicating the  
256 presence of more aromatic compounds, and possibly DOM-Fe complexes in summer. The shape of  
257 absorption spectral slopes showed a clear maximum at 382 nm, indicating the presence of terrestrial  
258 fulvic and humic acids. Consistent with the SUVA<sub>254</sub> index, S<sub>382</sub> values were relatively low in  
259 winter ( $0.0137 \pm 0.0016 \text{ nm}^{-1}$ ), with similar values in summer surface waters ( $0.0158 \pm 0.0014 \text{ mg}$   
260  $\text{nm}^{-1}$ ), but show significant difference with bottom waters ( $0.0169 \pm 0.0007 \text{ nm}^{-1}$ ) ( $p = 0.0061$ ). The  
261 individual spectral slope S<sub>289</sub> indicates that higher amounts of algal-derived carbon were present in  
262 summer surface waters ( $0.0148 \pm 0.0009 \text{ nm}^{-1}$ ), especially in the most transparent pond KWK6  
263 ( $0.0162 \text{ nm}^{-1}$ ), while their role was limited in the dark habitats under the ice ( $0.0113 \pm 0.0011 \text{ nm}^{-1}$ )  
264 and in bottom waters ( $0.0112 \pm 0.0007 \text{ nm}^{-1}$ ). Carbon quantity and CDOM quality properties are  
265 summarized in Table 2, and the seasonal differences in absorption spectral slopes are shown in Fig.  
266 2.

267

### 268 **3.3 Microbial abundance and productivity**

269 The abundance of all organisms examined was highest in the summer bottom waters, but the  
270 magnitude and seasonal distribution differed among taxa and ponds (Table 3). Especially, the  
271 abundant motile PNF accumulated in the nutrient-rich bottom waters, where their abundance was in  
272 most ponds an order of magnitude higher than at the surface ( $110 \pm 89$  vs.  $27 \pm 8 \times 10^6 \text{ L}^{-1}$ ). The  
273 less abundant HNF also favored the bottom waters, with an order of magnitude difference in the

274 abundance between bottom and surface waters ( $9.2 \pm 11.6$  vs.  $0.8 \pm 0.7 \times 10^6 \text{ L}^{-1}$ ). The smaller  
275 phototrophic picoplankton (PPA) and bacteria were more uniformly distributed, although the  
276 abundances were significantly higher ( $p < 0.05$ ) at the bottom. Abundance was also quantified for  
277 free-living and particle-attached bacteria. During late winter,  $74 \pm 9\%$  of bacterioplankton was  
278 attached to particles, and  $46 \pm 21\%$  and  $70 \pm 17\%$  in summer surface and bottom waters (Table A1).

279 When converted to biomass, the seasonal and vertical distribution pattern was even more  
280 accentuated (Table 3, Fig. 3). Overall, lower biomasses were encountered in winter, and higher at  
281 the bottom of the water column in summer. Total phototrophic biomass as Chl-*a* concentration ( $5.6$   
282  $\pm 2.5 \mu\text{g L}^{-1}$ ) as well as the PNF biomass ( $152 \pm 68 \mu\text{g C L}^{-1}$ ) under the ice were relatively high.  
283 Summer surface phototrophic biomass was similar to winter waters ( $p = 0.1213$ ), and it was  
284 different ( $p < 0.05$ ) and highest in the dark bottom waters, likely reflecting cell sedimentation. The  
285 biomass of strictly phototrophic PPA was very low in winter ( $5.8 \pm 4.4 \mu\text{g C L}^{-1}$ ) and an order of  
286 magnitude higher in summer. The bacterial biomass was always high when compared to literature,  
287 with the lowest biomass occurring in winter ( $109 \pm 67 \mu\text{g C L}^{-1}$ ), two times higher biomass in  
288 summer at the surface ( $223 \pm 36 \mu\text{g C L}^{-1}$ ), and five times higher at the bottom ( $520 \pm 146 \mu\text{g C L}^{-1}$ ).  
289 The HNF contributed little to the total microbial biomass, especially in the illuminated summer  
290 surface waters where they averaged  $3.2 \pm 2.9 \mu\text{g C L}^{-1}$ . Their biomass was higher at the bottom  
291 ( $39.2 \pm 49.4 \mu\text{g C L}^{-1}$ ) and in winter samples ( $8.9 \pm 4.7 \mu\text{g C L}^{-1}$ ).

292 Snow and ice cover prevented light penetration under the ice, and therefore we assumed that no PP  
293 occurred *in situ* in late winter. However, the photosynthetic capacity of late winter phytoplankton  
294 community was not null, and represents the potential at the ice-out, which took place 2-3 weeks  
295 later. The maximum photosynthetic rate ( $P_{\text{max}}$ ) of late winter phytoplankton community was for  
296 most ponds an order of magnitude lower than  $P_{\text{max}}$  of the summer community ( $2.0 \pm 2.1$  vs.  $30.6 \pm$   
297  $26.1 \text{ mg C m}^{-3} \text{ d}^{-1}$ ). BP rates were very low in winter ( $0.4 \pm 0.2 \text{ mg C m}^{-3} \text{ d}^{-1}$ ) and different ( $p =$   
298  $0.0001$ ) compared to the summer when surface ( $31 \pm 14 \text{ mg C m}^{-3} \text{ d}^{-1}$ ) and bottom ( $42 \pm 19 \text{ mg C}$   
299  $\text{m}^{-3} \text{ d}^{-1}$ ) BP rates. BR rates calculated from BP were low in winter ( $2.0 \pm 0.6 \text{ mg C m}^{-3} \text{ d}^{-1}$ ) and  
300 equally high during summer in surface ( $28 \pm 6 \text{ mg C m}^{-3} \text{ d}^{-1}$ ) and in bottom ( $33 \pm 6 \text{ mg C m}^{-3} \text{ d}^{-1}$ ).  
301 Particle-attached bacterioplankton played an important role in pond production at both sampling  
302 seasons (late winter:  $42 \pm 20\%$ ; summer surface:  $66 \pm 20\%$ ; bottom:  $68 \pm 44\%$ ; Table A1).

303 Winter, surface and bottom environmental characteristics formed three distinct groups on the PCA  
304 (Fig. 4a), with the first two axes explaining 72.9% of the variation in the environmental variables  
305 (PC1 46.0% and PC2 26.9%). Axis PC1 has strong positive correlations with TP ( $r=0.91$ ),  $a_{320}$   
306 ( $r=0.80$ ), TSS ( $r=0.83$ ) and GHG ( $\text{CO}_2$   $r=0.92$  and  $\text{CH}_4$   $r=0.90$ ), and strong negative correlations  
307 with pH ( $r=-0.81$ ) and  $S_{289}$  ( $r=-0.65$ ), whereas PC2 has positive correlation with temperature  
308 ( $r=0.81$ ),  $S_{382}$  ( $r=0.69$ ) and  $\text{SUVA}_{254}$  ( $r=0.76$ ), and negative correlation with TN ( $r=-0.93$ ) and DOC  
309 ( $r=-0.67$ ). Heterotrophs (BB and HNF) were correlated with PC1 while phototrophs (PP, PNF and  
310 PPA) were more strongly correlated to PC2 (Fig. 4b). Further, BP and BB were best correlated with  
311 carbon quality indices  $a_{320}$ ,  $\text{SUVA}_{254}$ , and  $S_{289}$ , while HNF were best correlated with conductivity  
312 (Table 4). PP, PNF and PPA were best correlated with a combination of carbon quality indices,  
313 DOC and nutrients (Table 4).

314

## 315 **4 Discussion**

316 The large variations in physicochemical characteristics of thaw ponds between late winter and  
317 summer and between surface and bottom waters were reflected in their ecological properties. The  
318 cold and nutrient-rich winter waters had the lowest microbial abundance and productivity. The  
319 warm and illuminated summer surface waters favored primary production, while the colder and  
320 darker bottom waters accumulated nutrients, carbon compounds and microorganisms. The overall  
321 supersaturation in GHG and the clear dominance of heterotrophy over phototrophy, as expressed in  
322 the biomass and activity of the microbial food web components, indicate that the majority of the  
323 energy was flowing through the heterotrophic food web.

324

### 325 **4.1 Spatial and seasonal variations in GHG**

326 Greenhouse gas storage in pond waters occurred in winter under the ice cover and in summer under  
327 the steep thermocline, due to the isolation of these water masses from atmospheric venting. Partial  
328 pressures measured were especially large in summer hypolimnetic waters (in average 13 674  $\mu\text{atm}$   
329 of  $\text{CO}_2$  and 4783  $\mu\text{atm}$  of  $\text{CH}_4$ ) compared to values at the surface (in average 1206  $\mu\text{atm}$  of  $\text{CO}_2$  and  
330 9  $\mu\text{atm}$  of  $\text{CH}_4$ ). Such heterogeneity in the vertical distribution of GHG is typical for seasonally  
331 stratified lakes where bottom values are often many folds higher than at the surface (Eller et al.,

332 2005; Gu erin et al., 2006; Bastviken et al., 2008). Partial pressures of CO<sub>2</sub> at the surface of thaw  
333 ponds (598-1545  $\mu\text{atm}$ ) were comparable to values obtained in other regions, for example in  
334 European boreal lakes (990  $\mu\text{atm}$ ; Kortelainen et al. 2006), Canadian boreal streams and rivers  
335 (1850  $\mu\text{atm}$ ; Teodoru et al. 2009), Western Siberian lakes and ponds (1935  $\mu\text{atm}$ ; Shirokova et al.  
336 2012), or Canadian subarctic and arctic ponds (1896  $\mu\text{atm}$ ; Breton et al. 2009). Surface water CH<sub>4</sub>  
337 partial pressures (6-13  $\mu\text{atm}$ ) were also in the same range as values reported in lakes and ponds  
338 from North-America (11-59  $\mu\text{atm}$ ; Bastviken et al. 2008) Finland (23  $\mu\text{atm}$ ; Juutinen et al. 2009),  
339 Canadian Subarctic and Arctic (22  $\mu\text{atm}$ ; Breton et al. 2009), and Western Siberia (30  $\mu\text{atm}$ ;  
340 Shirokova et al. 2012).

341 The dominance of BR over PP, combined with the isolation of water masses during the long winter,  
342 followed by a very short spring mixing period (Laurion et al., 2010) and the steep summer  
343 stratification of these ecosystems controlled the accumulation of GHG in the hypolimnion. When  
344 normalized by the duration of the known isolation period (Laurion et al., 2010), GHG accumulated  
345 more rapidly in summer at the pond bottom ( $6.8 \pm 2.2 \mu\text{M CO}_2 \text{ d}^{-1}$  and  $1.9 \pm 0.9 \mu\text{M CH}_4 \text{ d}^{-1}$ ) than  
346 in winter under the ice cover ( $1.5 \pm 0.3 \mu\text{M CO}_2 \text{ d}^{-1}$  and  $0.030 \pm 0.027 \mu\text{M CH}_4 \text{ d}^{-1}$ ), assuming a  
347 constant accumulation rate during the two sampling periods and excluding freezing exclusion  
348 effects. This may be partly due to the more abundant microbial community in summer, although  
349 most GHG produced likely came from benthic respiration, especially in such shallow systems  
350 (Kortelainen et al., 2006). Summer also provides fresh organic matter that may stimulate GHG  
351 producers that indeed presented higher respiration rates. The smaller CH<sub>4</sub> accumulation rates in  
352 winter might also result from more optimal conditions for methanotrophy under the ice cover where  
353 bubbling CH<sub>4</sub> is trapped and dissolves in water. The amount of CH<sub>4</sub> in a lake was shown to depend  
354 on the size of the anoxic layer (Fendinger et al., 1992; Bastviken et al., 2004), and in the present  
355 study, this layer represents a particularly large proportion of the water volume in summer. During  
356 the autumnal turnover period (lasting about 2 months), the whole water column becomes  
357 oxygenated, and in the first part of the winter surface layer of the pond may contain sufficient  
358 oxygen (Deshpande et al., 2015) for efficient CH<sub>4</sub> consumption. Anoxic methanotrophy has also  
359 been reported (Sivan et al., 2011) and may occur at our study sites. Nevertheless, these GHG  
360 storage periods indicate that subarctic thaw ponds will release large amounts of CH<sub>4</sub> during the  
361 autumnal mixing period, and also during spring turnover if it lasts long enough (Walter et al., 2006;  
362 Laurion et al., 2010).

363

#### 364 **4.2 Seasonality in CDOM properties**

365 We used  $a_{320}$  as an indicator of the quantity of CDOM, and  $SUVA_{254}$ ,  $S_{289}$  and  $S_{382}$  as indicators of  
366 the aromatic content of CDOM and the relative proportions in allochthonous (terrestrial) versus  
367 autochthonous (algal) carbon sources. The high  $a_{320}$  and  $SUVA_{254}$  values indicate a large  
368 contribution by terrestrial humic substances in DOM at the surface of thaw ponds in summer, with  
369 values similar to a few years earlier in the same ponds (Breton et al., 2009; Watanabe et al., 2011).  
370 CDOM values were clearly smaller in winter (late winter  $a_{320}$ ; Table 2) and less aromatic  
371 ( $SUVA_{254}$ ) than in summer, suggesting that carbohydrates or protein-like compounds from bacteria-  
372 induced CDOM-degradation made up a larger fraction of DOM in winter. Laurion et al.  
373 (unpublished data) tested the role of bacterial CDOM degradation in arctic thaw ponds and found  
374 that its importance can almost match that of photodegradation in certain cases, without significant  
375 losses in DOC, potentially indicating a bacterial transformation into less colored compounds. The  
376 higher DOC values observed in winter (average of  $8.3 \text{ mg L}^{-1}$ ) compared to summer at the surface  
377 ( $5.8 \text{ mg L}^{-1}$  in) thus likely resulted from a combination of a generation of dissolved compounds  
378 from the uninterrupted but slower microbial activity, and out-freezing inputs associated to the  
379 formation of ice. Although BP in late winter was low, the long subarctic winters allow  
380 accumulation effects that could lead to a decrease in CDOM. The labile part of DOC is known to be  
381 consumed within days by bacterioplankton, leaving the recalcitrant fraction of DOC to dominate in  
382 environments where fresh inputs are limited (Del Giorgio and Davis, 2002; Roehm et al., 2009;  
383 Guillemette and del Giorgio, 2011). However this is not clearly observed in the present study thaw  
384 ponds, where less aromatic and non-chromophoric fractions increased along winter, likely as the  
385 result of DOM recycling by a rich microbial food web, together with a lack of new carbon inputs  
386 from the frozen catchment.

387 The high  $a_{320}$  and  $SUVA_{254}$  values found in bottom waters are very likely caused by the elevated  
388 dissolved iron concentrations (up to  $5.1 \text{ mg L}^{-1}$ ;  $r^2 = 0.947$  between  $SUVA_{254}$  and Fe in summer  
389 samples,  $n=10$ ). High concentrations of iron ( $>2 \text{ mg L}^{-1}$ ) were shown to form complexes with humic  
390 substances and increase DOM absorbance (Maloney et al., 2005; Xiao et al., 2013). During the  
391 autumnal turnover, CDOM-Fe complexes from bottom waters are exposed to photochemical  
392 reactions at the surface, potentially generating hydroxyl radicals that dissociate carboxyl groups

393 from humic substances. These reactions can lead to lower molecular weight carbon components at  
394 the same time as increasing the water clarity (Brinkmann et al., 2003). The spectral slope values  
395 indicate that algal-derived proteins ( $S_{289}$ ; Galgani et al. 2011) were more abundant in summer  
396 surface waters when primary production and Chl-*a* were highest, and at the same time fulvic and  
397 humic acids ( $S_{382}$ ) also seemed more abundant than in winter. Algal proteins are generally highly  
398 available to heterotrophic organisms and were likely supporting the higher summer BP (Kritzberg et  
399 al., 2004). Nevertheless, Xiao et al. (2013) showed how high Fe concentrations, such as found in  
400 bottom waters, can significantly lower DOM absorption slopes, thus interpretations of bottom water  
401  $S_{\lambda}$  should be done cautiously.

402

### 403 **4.3 Thaw pond microbial food web**

404 Primary production rates in summer at the surface of thaw ponds were in the same range as in other  
405 arctic and subarctic lakes (Lizotte, 2008), while in late winter snow and ice cover prevented *in situ*  
406 PP. However, when exposed to light, the production of carbon by the winter phytoplankton  
407 community was in the same range as the heterotrophic production, suggesting that the high  
408 phytoplankton biomass encountered during late winter were made of mixotrophic species that  
409 switched from hetero- to autotrophic production at ice-out. Similarly, the high Chl-*a* and PNF  
410 biomass in the light-limited bottom waters in summer were most likely composed of mixotrophs.  
411 Many species of Chrysophyceae, Cryptophyceae and Dinophyceae are known to be 99%  
412 mixotrophic (Ollrik 1998 and references therein). Chrysophyceae were dominating the  
413 phytoplankton community in KWK2 and KWK20 (Dupont 2008, L. Forsström, unpublished data).  
414 Many Chrysophyceae species have flagella, and thus can thrive in sheltered humic ponds that have  
415 been shown to favor diurnal migrations of phytoplankton (Jones, 1991). This is especially  
416 advantageous in steeply stratified and shallow ponds where phytoplankton can migrate within a few  
417 hours (Arvola et al., 1991), and are able to use nutrient-rich hypolimnetic waters (Salonen et al.,  
418 1984). The high nutrient concentrations in bottom waters likely explain the high biomass of PNF  
419 observed in these waters. Anoxic bottom waters were also devoid of *Daphnia* sp. (M. Wauthy et al.,  
420 unpublished data) providing to phytoplankton a refuge from grazing. Absence of light, hypoxia and  
421 low temperatures further prevent photo-oxidation of algal pigments, preserving them for extended  
422 periods (Vallentyne, 1960). Thus some of the Chl-*a* and PNF encountered at the bottom may have

423 been from dead yet still not degraded cells. Another factor that could have caused bias in very high  
424 bottom water Chl-*a* values is the occurrence of chlorophyll-*b* containing green-sulphur bacteria that  
425 are known to colonize the anoxic layer of the studied ponds (Rossi et al., 2013), and that may  
426 interfere with the spectrophotometric analyses of Chl-*a*. However, green-sulphur bacteria were not  
427 found in all ponds with high values of Chl-*a*. In winter, the lack of PP and allochthonous inputs of  
428 carbon, and a DOM pool composed by a large fraction of molecules with low aromaticity (low  
429 SUVA<sub>254</sub> and S<sub>289</sub>) suggest that intensively recycled DOM formed the basal organic carbon pool in  
430 thaw ponds.

431 BP and BB were strongly linked to CDOM originating from a terrestrial source (high SUVA<sub>254</sub> and  
432 a<sub>320</sub>), indicating a positive link between bacterioplankton and terrestrial carbon (Tranvik, 1988;  
433 Crump et al., 2003). These results are in accordance with the experiment from Breton et al. (2009)  
434 which showed that bacterioplankton in thaw ponds was more carbon than nutrient-limited (glucose  
435 was used as a labile carbon source). On the contrary to several previous studies (Granéli et al.,  
436 2004; Sävström et al., 2007; Roiha et al., 2012), the amount of DOC did not seem to control the BP  
437 in our study. The studies that reported a positive correlation between DOC and BP have been  
438 conducted in summer, when BP and DOC concentrations were highest, but without necessarily  
439 taking into account the variability in DOC type, which had a dominant role in our study.

440 The presence of a mixture of autochthonous and allochthonous carbon may have helped in making  
441 the terrestrial carbon available to bacteria. Phytoplankton and macrophytes apparently produced  
442 significant amounts of dissolved carbon in surface waters during summer as indicated by the larger  
443 S<sub>289</sub> peak. This carbon is considered more labile (lower molecular weight) and with a higher  
444 nutritional value to heterotrophs (Kritzberg et al., 2004; Brett et al., 2009). The availability of high  
445 quality phytoplankton carbon during summer possibly enhanced the bacterial use of more  
446 recalcitrant terrestrial carbon through the priming effect (Bianchi, 2011), that is a small supplement  
447 of nutritionally rich algal cells increasing bacterioplankton terrestrial carbon assimilation.

448 Correlation between TSS and TP ( $r=0.903$   $p<0.001$ ,  $n=15$ ) indicates that phosphorus was binding to  
449 particles, which may be more accessible to attached bacterioplankton than to phytoplankton in  
450 summer surface waters where SRP concentrations were low. The high proportion of particle-  
451 attached bacterial abundance ( $62 \pm 30$  %) and productivity ( $59 \pm 30$  %) in thaw ponds suggests that  
452 particle attachment is beneficial. Although studies have shown that BP is often nutrient-limited



453 (Kritzberg et al., 2004; Vrede, 2005), there were no correlation between BP and nutrients in thaw  
454 ponds.

455 One possible additional explanation for the difference between winter and summer BP is the  
456 difference in temperature that is known to have a large impact on production (Adams et al., 2010).  
457 On the other hand, bacterial communities are known to adapt at low temperatures, and some  
458 communities can have multiple temperature maxima (Adams et al., 2010). In our study, the highest  
459 BP was found in relatively cool bottom waters. Moreover, bacteria biomass (BB) was not correlated  
460 to temperature, although several studies found temperature as a main driver of bacterial biomass  
461 (Ochs et al., 1995; Rae and Vincent, 1998). Trophic interactions certainly contributed to the  
462 observed patterns in microbial abundance. Rotifers are the dominant zooplankton group in the  
463 studied thaw ponds, while the cladoceran community is composed of a few *Daphnia* sp. observed  
464 only at the surface (Bégin, 2014). These two zooplankton groups were only found in summer, when  
465 they could exert a grazing pressure on bacteria and phytoplankton, but experiments would be  
466 needed to confirm the extent of grazing. We tentatively estimated the top-down control of  
467 nanoflagellates on bacteria, using the average nanoflagellate ingestion rate of  $0.66 \text{ pg C cell}^{-1} \text{ day}^{-1}$   
468 measured for *Dinobryon* sp. in Svalbard (Laybourn-Parry and Marshall, 2003). From these  
469 calculations, it seems that nanoflagellate grazing on bacteria could have been efficient enough in  
470 late winter when the grazing potential exceeded 100% of the BP, while summer removal  
471 estimations corresponded to only 1.7% at the surface and 9.3% at the bottom of the BP, possibly  
472 contributing to explaining the accumulation of BB in summer

473

## 474 **5 Conclusions**

475 Low light intensity, strong stratification and high concentrations of CDOM in subarctic thaw ponds  
476 are the main drivers creating an environment highly beneficial to heterotrophs. We have also shown  
477 some important differences in their ecosystem dynamics between winter and summer. Although a  
478 high biomass of bacterioplankton is found under the ice, the ponds are producing relatively small  
479 amounts of carbon in late winter, likely caused by the lack of fresh carbon inputs and its associated  
480 priming effect. Nevertheless, the ponds are accumulating GHG during winter. The relatively high  
481 phototrophic biomass and primary production potential encountered under the ice are at the same  
482 level as the heterotrophic biomass and production, indicating the importance of mixotrophy in

483 winter. In summer, fresh primary production and terrestrial carbon inputs enhance the heterotrophic  
484 production. Although a higher phototrophic (mixotrophic) biomass is encountered in summer,  
485 primary production is restricted by light and space and is significant only at the pond surface, thus  
486 the heterotrophic carbon production remains significantly higher. Consequently, GHG continue to  
487 accumulate in hypolimnetic waters during this strongly stratified summer period, and are most  
488 likely released during the autumnal turnover period, in addition to the constant emissions of GHG  
489 throughout the summer from the highly supersaturated surface waters.

490

## 491 **6 Acknowledgements**

492 We thank Academy of Finland (grants 119205 and 140775), the Societas Biologica Fennica  
493 Vanamo, Maa- ja Vesitekniikan Tuki Ry and the Natural Sciences and Engineering Research  
494 Council of Canada for their financial support, and Denis Sarrazin, Paul-Georges Rossi, Frédéric  
495 Bouchard, Annabelle Warren, Benoit Ginoux, Catherine Girard and Jonna Kuha for their assistance  
496 in the field and laboratory.

497

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- 745

746 **Figure captions**

747  
748 Figure 1. Summer temperature and oxygen profiles from sampled ponds.

749  
750 Figure 2. Absorption spectral slope curve ( $S_\lambda$ ) of dissolved organic matter as an indicator of the  
751 presence of algal-derived carbon (peak at  $S_{289}$ ) and terrestrial carbon (steady increase in the slopes  
752 from 260-390 nm).

753  
754 Figure 3. Seasonal changes in the biomass of phototrophic picoplankton (PPA), pigmented  
755 nanoflagellates (PNF), bacteria (BB), and heterotrophic nanoflagellates (HNF). Error bars  
756 represent standard errors.

757  
758 Figure 4. a) Seasonality in pond environmental variables illustrated with principal component  
759 analysis during late winter (triangles), summer surface (diamonds) and summer bottom (crosses). b)  
760 Spearman correlation coefficients between biological variables and principal components 1 (x-axis)  
761 and 2 (y-axis). BP=bacterial production, BB=bacterial biomass, HNF=heterotrophic nanoflagellate  
762 biomass, PP=primary production, PNF=pigmented nanoflagellate biomass, and PPA=phototrophic  
763 picoplankton biomass.

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765

766 **Table 1.** Limnological properties of the subarctic ponds sampled in late winter (W) and during  
 767 summer at the surface (S) and bottom (B) of the water column, including pH, conductivity  
 768 (Cond.), total phosphorus (TP), total nitrogen (TN), soluble reactive phosphorus (SRP), total  
 769 suspended solids (TSS), particulate organic carbon (POC) and iron (Fe).

Site	pH	Cond. ( $\mu\text{S cm}^{-1}$ )	TP ( $\mu\text{g L}^{-1}$ )	TN ( $\mu\text{g L}^{-1}$ )	SRP ( $\mu\text{g L}^{-1}$ )	TSS ( $\text{mg L}^{-1}$ )	POC ( $\text{mg L}^{-1}$ )	Fe ( $\text{mg L}^{-1}$ )
Winter								
KWK 2	6.6	60	61.9	957.4	20.8	2.7	0.8	na
KWK 6	6.8	49	50.6	687.8	7.7	3.7	1.1	na
KWK 12	6.6	46	73.2	1038.4	18.4	4.1	1.2	na
KWK 20	6.7	52	180.7	1301.6	47.1	53.9	16.6	na
KWK 23	6.4	42	119.3	802.7	33.9	23.5	6.8	na
Summer surface								
KWK 2	7.4	38	34.9	289.3	1.2	3.4	0.9	0.3
KWK 6	6.9	61	43.9	227.9	0.9	7.2	2.3	0.2
KWK 12	7.3	36	23.9	311.9	0.7	2.6	0.9	0.3
KWK 20	7.5	47	92.6	263.4	7.1	14.2	1.8	0.5
KWK 23	7.1	48	78.0	227.9	5.7	16.1	2.1	0.4
Summer bottom								
KWK 2	6.2	200	341.5	496.1	1.5	27.7	4.4	2.5
KWK 6	6.2	265	197.7	389.4	na	13.6	2.3	1.1
KWK 12	6.1	247	207.1	447.6	1.0	37.0	8.8	2.9
KWK 20	6.0	155	377.0	289.3	73.8	85.8	5.7	5.1
KWK 23	6.2	102	431.8	266.7	45.7	126.8	7.8	4.1

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771

772 **Table 2.** Concentrations of carbon dioxide (CO<sub>2</sub>), methane (CH<sub>4</sub>), and dissolved organic carbon  
 773 (DOC), and dissolved organic matter (DOM) optical properties, including absorption coefficient  
 774 of DOM at 320 nm (a<sub>320</sub>), specific UV-absorbance index (SUVA<sub>254</sub>), and absorption spectral  
 775 slopes at 289 nm (S<sub>289</sub>) and 382 nm (S<sub>382</sub>), of subarctic pond water sampled in late winter and  
 776 during summer at the surface and bottom of the water column.

Site	CO <sub>2</sub> (μM)	CH <sub>4</sub> (μM)	DOC (mg L <sup>-1</sup> )	a <sub>320</sub> (m <sup>-1</sup> )	SUVA <sub>254</sub> (L mg C <sup>-1</sup> m <sup>-1</sup> )	S <sub>289</sub> (nm <sup>-1</sup> )	S <sub>382</sub> (nm <sup>-1</sup> )
Winter							
KWK 2	316	14.1	7.7	8.5	0.9	0.0099	0.0155
KWK 6	354	5.1	5.2	3.0	0.5	0.0112	0.0122
KWK 12	215	2.4	10.5	9.5	0.8	0.0105	0.0154
KWK 20	na	na	10.3	8.1	0.8	0.0125	0.0130
KWK 23	357	2.5	7.9	6.6	0.8	0.0124	0.0123
Summer surface							
KWK 2	72	0.6	5.6	26.4	4.7	0.0141	0.0169
KWK 6	27	0.5	4.1	11.8	3.4	0.0162	0.0135
KWK 12	55	0.3	6.0	26.5	4.5	0.0142	0.0170
KWK 20	71	0.3	7.1	38.9	6.1	0.0145	0.0156
KWK 23	53	0.3	6.3	30.7	5.4	0.0150	0.0161
Summer bottom							
KWK 2	na	na	6.1	51.5	6.7	0.0109	0.0156
KWK 6	422	145.2	4.2	21.4	4.3	0.0125	0.0173
KWK 12	761	259.0	7.4	67.9	7.4	0.0111	0.0173
KWK 20	815	311.9	9.3	115.6	9.8	0.0107	0.0172
KWK 23	570	131.6	7.5	94.5	10.2	0.0109	0.0171

777

778

779 **Table 3.** Phototrophic and heterotrophic properties of subarctic pond water sampled in late winter  
 780 and during summer at the surface and bottom of the water column, including chlorophyll-*a*  
 781 concentration (Chl-*a*), maximum photosynthesis parameter ( $P_{\max}$ ), *in situ* primary production (PP),  
 782 pigmented nanoflagellate abundance (PNF), phototrophic picoplankton abundance (PPA),  
 783 bacterial production (BP), bacterial respiration (BR), bacterial abundance (BA) and heterotrophic  
 784 nanoflagellate abundance (HNF).

Site	Chl- <i>a</i> ( $\mu\text{g L}^{-1}$ )	$P_{\max}$ ( $\text{mg C m}^{-3} \text{d}^{-1}$ )	PP ( $\text{mg C m}^{-3} \text{d}^{-1}$ )	PNF ( $\times 10^6 \text{L}^{-1}$ )	PPA ( $\times 10^5 \text{mL}^{-1}$ )	BP ( $\text{mg C m}^{-3} \text{d}^{-1}$ )	BR ( $\text{mg C m}^{-3} \text{d}^{-1}$ )	BA ( $\times 10^6 \text{mL}^{-1}$ )	HNF ( $\times 10^6 \text{L}^{-1}$ )
Winter									
KWK 2	2.8	1.2	0	33.0	0.2	0.2	1.3	4.2	2.0
KWK 6	9.9	0.8	0	51.0	0.9	0.5	2.2	11.2	3.3
KWK 12	11.5	5.8	0	29.1	0.7	0.3	1.6	2.5	2.4
KWK 20	2.8	0.8	0	6.3	0.2	0.8	3.0	8.7 <sup>a</sup>	2.4
KWK 23	0.8	1.5	0	10.0	0.1	0.4	2.0	3.7	0.3
Summer surface									
KWK 2	5.4	4.5	2.6	16.3	3.3	27.2	25.7	13.3	0.0
KWK 6	13.4	27.9	15.3	20.0	4.2	37.4	31.2	11.3	0.8
KWK 12	2.2	5.9	3.2	28.8	0.9	31.2	27.9	9.6	0.7
KWK 20	14.3	59.0	26.4	33.1	17.3	15.2	18.0	14.8	1.9
KWK 23	12.3	55.6	15.6	35.0	3.8	45.9	35.3	12.7	0.4
Summer bottom									
KWK 2	180.2	na	0.8	149.6	9.4	57.6	40.5	35.7	26.2
KWK 6	87.2	na	4.2	201.6	3.0	39.0	32.0	14.3	6.3
KWK 12	158.9	na	0.3	72.5	26.8	48.8	36.6	38.0	3.7
KWK 20	203.4	na	0.3	na	4.0	39.1	32.0	27.2	na
KWK 23	37.1	na	0.0	4.5	6.2	23.0	23.2	24.3	0.6

785 <sup>a</sup> BA was calculated from DAPI-stained microscope slide

786

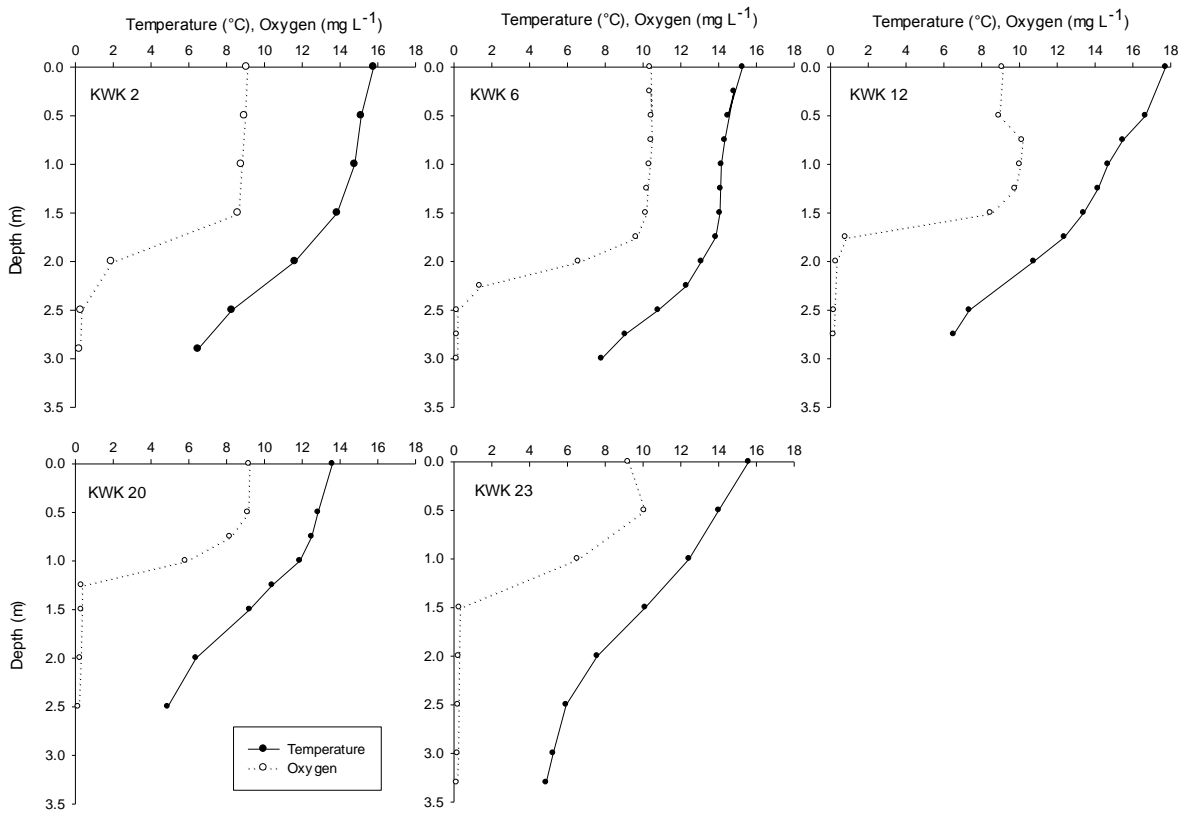
787 **Table 4.** Spearman correlations between biological and environmental variables. Only statistically  
 788 significant correlations are given.

<b>Biological variable</b>	<b>Environmental variable</b>	<b>n</b>	<b>r</b>	<b>p</b>
Bacteria production (BP)	Temperature	15	0.600	0.018
	TN	15	-0.524	0.045
	a <sub>320</sub>	15	0.704	0.003
	SUVA <sub>254</sub>	15	0.699	0.004
	S382	15	0.633	0.011
Bacteria biomass (BB)	O <sub>2</sub>	15	0.555	0.049
	Conductivity	15	0.668	0.007
	TP	15	0.600	0.018
	TSS	15	0.521	0.046
	Fe	10	0.879	0.001
	CH <sub>4</sub>	13	0.588	0.035
	a <sub>320</sub>	15	0.825	<0.001
	SUVA <sub>254</sub>	15	0.857	<0.001
HNF	Conductivity	15	0.657	0.011
	CH <sub>4</sub>	15	0.636	0.026
Primary production (PP)	Temperature	10	0.896	<0.001
	SRP	10	-0.662	0.007
	TN	10	-0.827	<0.001
	DOC	10	-0.610	0.016
	SUVA <sub>254</sub>	10	0.533	0.041
	S289	10	0.732	0.002
PNF	SRP	14	-0.732	0.003
	DOC	14	-0.647	0.012
PPA	TN	15	-0.665	0.007
	a <sub>320</sub>	15	0.825	<0.001
	SUVA <sub>254</sub>	15	0.717	0.003
	S382	15	0.564	0.029



789 **Figure 1.**

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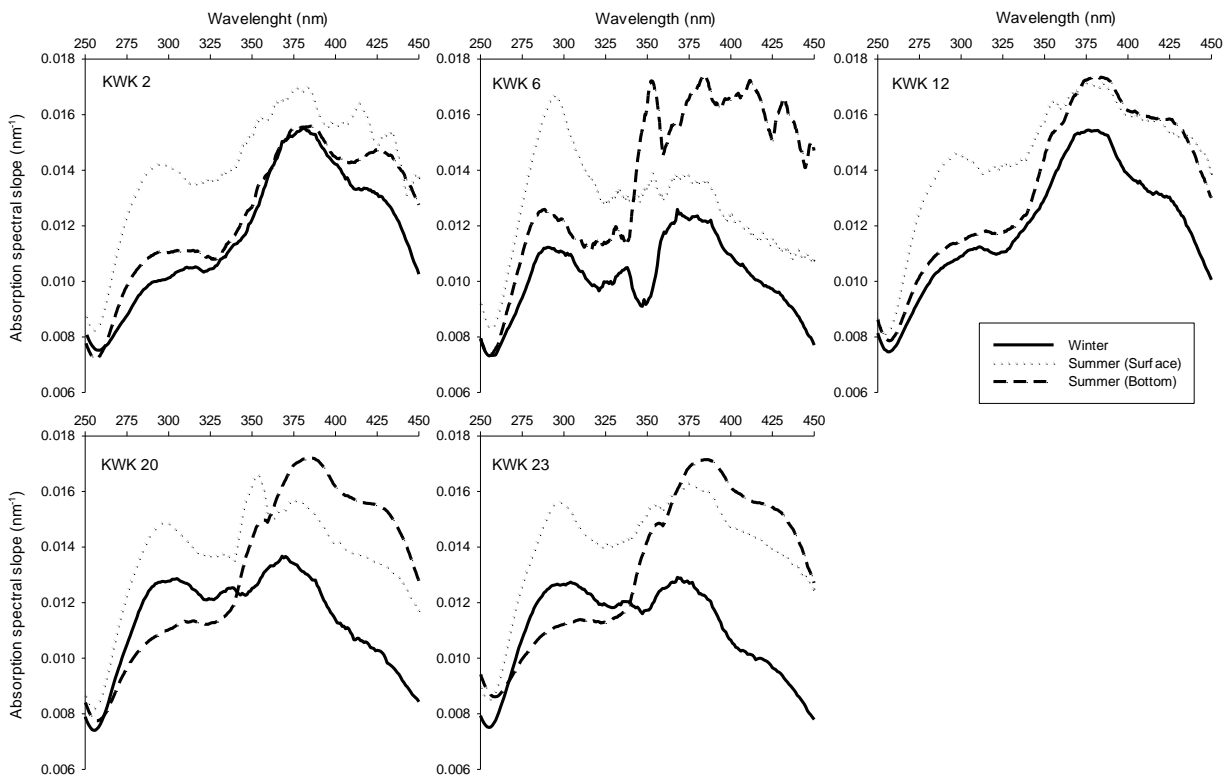
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795 **Figure 2.**

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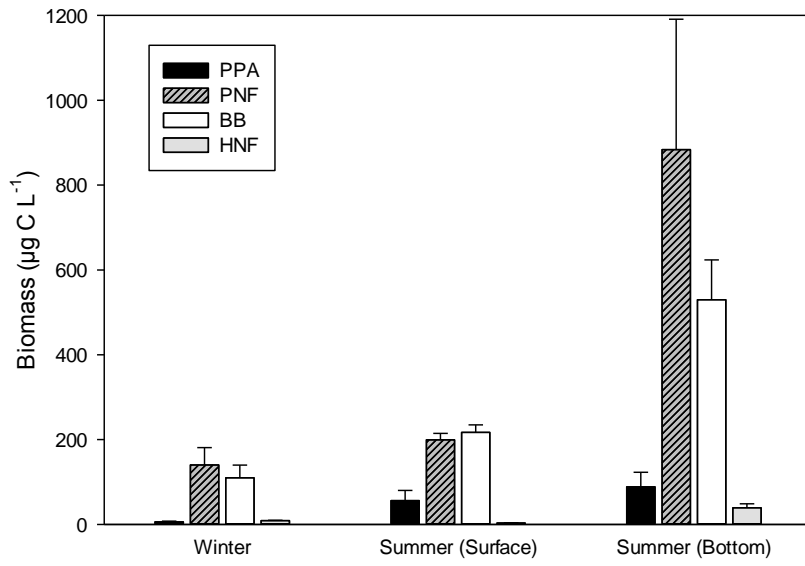
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800 **Figure 3.**

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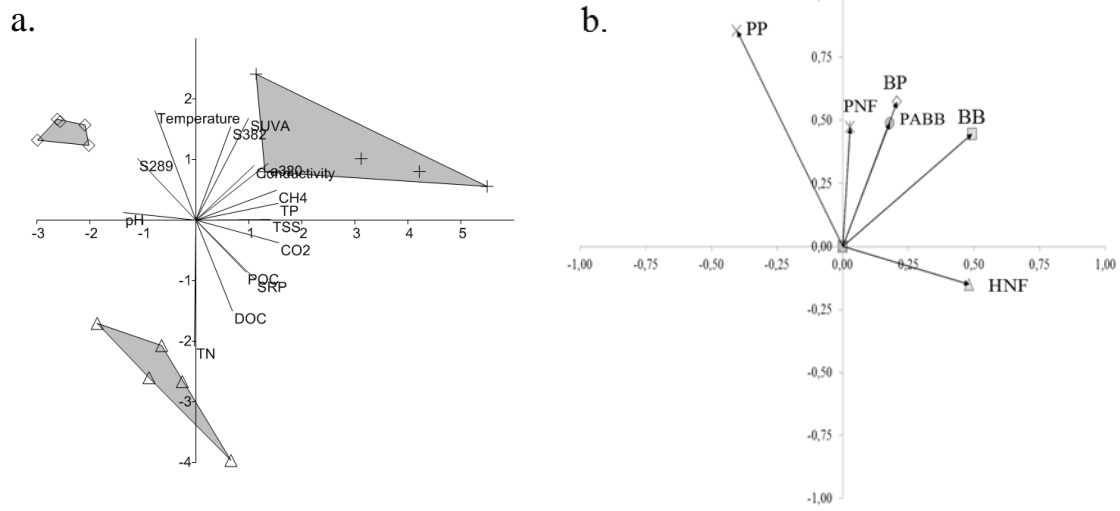
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805 **Figure 4.**

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812 **Appendix A:**

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814 **Table A1.** Particle attachment of bacterioplankton in winter, during summer at the  
 815 surface and bottom of ponds, including sonicated bacterial abundance ( $BA_T$ )  
 816 representing the total abundance, bacterial abundance before sonication ( $BA_{FL}$ )  
 817 representing free-living bacteria, proportion of particle-attached bacteria ( $BA_P$ ), total  
 818 bacterial production (BP; as in Table 3), filtered ( $3\mu m$ ) bacterial production ( $BP_{FL}$ ) and  
 819 proportion of particle attached bacterial production ( $BP_P$ ).

Site	$BA_T$ ( $\times 10^6 \text{ mL}^{-1}$ )	$BA_{FL}$ ( $\times 10^6 \text{ mL}^{-1}$ )	$BA_P$ (%)	BP ( $\text{mg C m}^{-3} \text{ d}^{-1}$ )	$BP_{FL}$ ( $\text{mg C m}^{-3} \text{ d}^{-1}$ )	$BP_P$ (%)
Winter						
KWK 2	4.2	0.8	81	0.2	0.1	54
KWK 6	11.2	3.2	72	0.5	0.2	54
KWK 12	2.5	0.5	79	0.3	0.2	48
KWK 20	na	na	na	0.8	0.4	47
KWK 23	3.7	1.4	62	0.4	0.4	7
Summer surface						
KWK 2	13.3	5.8	57	27.2	4.7	83
KWK 6	11.3	8.5	24	37.4	3.3	91
KWK 12	9.6	6.0	37	31.2	17.4	56
KWK 20	14.8	4.1	73	15.2	7.0	54
KWK 23	12.7	7.7	40	45.9	18.6	60
Summer bottom						
KWK 2	35.7	11.3	68	57.6	1.7	97
KWK 6	14.3	8.3	42	39.0	1.5	96
KWK 12	38.0	11.2	70	48.8	26.0	47
KWK 20	27.2	3.7	86	39.1	1.0	97
KWK 23	24.3	4.2	83	23.0	26.6	0

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