# **Anonymous Referee #1**

We would like to thank the reviewer for her/his positive evaluation and the very helpful comments and suggestions. Below we reply on each specific comment.

- 119 "the exponent was fixed to 0.5" could you please better explain why and for what this exponent stands, I am not sure every reader would know.

Reply: We added the following information to the ms:

"The Schmidt-number exponent *n* describes the dependence of the gas exchange velocity of a particular gas on the diffusion coefficient of this gas in water. We used n=0.5, which showed best agreement with measurements for wave-covered and turbulent water surfaces (Jähne and Haußecker, 1998).

- 337 I find a difference in flow of only 0.02 between left and middle very minor, is this correct or wrong spelled/written? Shouldn't the difference between those both speeds more?

Reply: The flow velocities refer to the mean flow velocities in the flume, without chamberinduced disturbances, and are correct. Due to technical limitations, we could only use a rather narrow range of flow velocities in the flume.

- 39 velocity can also be derived
- 45 presence
- 115 comma instead of point
- 225 delete one "however"
- 298 for up to 5 minutes every 30s: seems difficult for me or do not understand
- 322 reference after point Fig. 1 "The solid line shows...

Reply: These typos were corrected.

# Anonymous Referee #2

Reply: We would like to thank the reviewer for her/his positive evaluation and the very helpful comments and corrections. Below we reply on each specific comment.

- Abstract – in the list of four points made from this study, switch numbers 1 and 2 so that you start by stating that anchored chambers produce turbulence. It seems to make more sense to start by saying that you find something in one method and then that you didn't in the other, instead of the other way around.

Reply: We followed the reviewer's suggestion and changed the abstract accordingly.

- P14622, L19-22 – I am not sure if this sentence regarding microbubbles fits here, especially since the papers you cited refer to lakes. Perhaps you can be clearer with what you are trying to convey with this point and you could use Beaulieu 2012 as a better reference for elevated kCH4 in rivers.

Reply: The sentence was changed to: "Moreover, recent studies revealed that the gas exchange velocity of  $CH_4$  can be enhanced by microbubbles (Beaulieu et al., 2012) and can therefore differ from that of the volatile tracer."

- P14623, L8 - Vachon et al. 2010 also discusses turbulence bias of chambers

Reply: We relate our findings to the study of Vachon (which refers to lakes and reservoirs) in the discussion section.

P14624, L6-7 – Change 'produced' to 'produce' and add a question mark at the end of the question.

Reply: Corrected.

- P14624, L8 – More details should be give about the size or stream order of the 9 rivers, especially since Hotchkiss et al 2015 just found that CO2 emissions change with size of streams. This new article should be cited and discussed in your manuscript. (Hotchkiss, E. R., Hall Jr, R. O., Sponseller, R. A., Butman, D., Klaminder, J., Laudon, H., ... & Karlsson, J. (2015). Sources of and processes controlling CO2 emissions change with the size of streams and rivers. Nature Geoscience, 8(9), 696-699.)

Reply: Information on discharge are now provided in Table 2.

We now refer to the new article in our introduction section. Because the major purpose of the present manuscript is on methodological issues, we consider a more detailed discussion of the magnitude of the observed fluxes and their relation to controlling processes at the landscape scale is not appropriate.

- P14623, L12-17 – Does using these different methods influence your flux results? I presume not seeing as what is important is the rate of accumulation and not absolute concentrations; however, you should make this clear in the text as it may cause confusion.

Reply: We assume the reviewer is referring to p. 14624. The uncertainties related to the relative accuracy of the different instruments is small compared to differences observed between different deployment types.

- P14625, L9 – add '(F)' after 'fluxes' to define the variable in eq.3

Reply: The flux symbol F is defined before, above Eq. 1.

- P14626, L20-25 – this section is a bit hard to follow. Is it possible to include a supplemental figure that will help the reader understand this process?

Reply: We did not find an intuitive way to illustrate this procedure in a figure. The text was changed to:

"The longitudinal extent of the observed flow fields (433 mm for anchored and 395 mm for drifting chambers) covered the complete chamber diameter and velocities are reported as a function of distance from the leading chamber edge in both the anchored and the drifting deployment."

- P14627, L15 - delete 'measured'

- P14628, L6-7 – rewrite 'than those under drifting chambers, with individual measurements of k600\_CO2\_a being up to 20 times higher than k600\_CO2\_d. The average ratio'

Reply: We applied the suggested changes.

- Figures 2a and b can have fit lines that refer to those discussed on P14628, L20 & L23

A fit line was added to Fig. 2b only. We realized that the individual fit lines for the various data sets in in Fig. 2a would add too much complexity to the figure and decided to skip these in favor of clarity.

Technical Note: Drifting versus anchored flux chambers for measuring greenhouse gas emissions from running waters Andreas Lorke<sup>1\*</sup>, Pascal Bodmer<sup>2,3</sup>, Christian Noss<sup>1</sup>, Zeyad Alshboul<sup>1</sup>, Matthias Koschorreck<sup>4</sup>, Celia Somlai<u>-Haase<sup>1</sup></u>, David Bastviken<sup>5</sup>, Sabine Flury<sup>2</sup>, Daniel F. McGinnis<sup>2,6</sup>, Andreas Maeck<sup>7</sup>, Denise Müller<sup>8,9</sup>, Katrin Premke<sup>2,10</sup>

<sup>1</sup> Institute for Environmental Sciences, University of Koblenz-Landau, Fortstr. 7, 76829 Landau, Germany

<sup>2</sup> Leibniz-Institute of Freshwater Ecology and Inland Fisheries, Chemical Analytics and Biogeochemistry, Müggelseedamm 310, 12587 Berlin, Germany

<sup>3</sup> Institute of Biology, Freie Universität Berlin, Berlin, Germany

<sup>4</sup> Helmholtz Centre for Environmental Research – UFZ, Department Lake Research, Brückstr.3a, 39114 Magdeburg, Germany

<sup>5</sup> Linköping University, Department of Thematic Studies – Environmental Change, 58183 Linköping, Sweden,

<sup>6</sup> Institute F.-A. Forel, Section of Earth and Environmental Sciences, University of Geneva,

Geneva, Switzerland

<sup>7</sup> Senect GmbH & Co. KG, An 44 – No. 11, 76829 Landau, Germany

<sup>8</sup> Institute of Environmental Physics (IUP), Otto-Hahn-Allee 1, 28359 Bremen, Germany

<sup>9</sup> Center for Tropical Marine Ecology (ZMT), Fahrenheitsstr. 8, 28359 Bremen, Germany

<sup>10</sup> Leibniz Centre for Agricultural Landscape Research, Institute for Landscape Biogeochemistry, Eberswalder Straße 84, 15374 Müncheberg, Germany

\* Corresponding author: <u>lorke@uni-landau.de</u>

Feldfunktion geändert

# Abstract

1	Stream networks were recently discovered as major but poorly constrained natural greenhouse
2	gas (GHG) sources. A fundamental problem is that several measurement approaches have
3	been used without cross comparisons. Flux chambers represent a potentially powerful
4	methodological approach if robust and reliable ways to use chambers on running water can be
5	defined. Here we compare the use of anchored and freely drifting chambers on various
6	streams having different flow velocities. The study clearly shows that (1) anchored chambers
7	enhance turbulence under the chambers and thus elevate fluxes, (2) drifting chambers have a
8	very small impact on the water turbulence under the chamber and thus generate more reliable
9	fluxes, (3) the bias of the anchored chambers greatly depends on chamber design and
10	sampling conditions, and (4) there is a promising method to reduce the bias from anchored
11	chambers by using a flexible plastic foil seal to the water surface rather than having rigid
12	chamber walls penetrating into the water. Altogether, these results provide novel guidance on
13	how to apply flux chambers in running water, which will have important consequences for

14 measurements to constrain the global GHG balances.

**Gelöscht:** drifting chambers have a very small impact on the water turbulence under the chamber and thus generate more reliable fluxes **Gelöscht:** anchored chambers enhance turbulence under the chambers and thus elevate fluxes

# 19 1 Introduction

20	Rivers and streams have been identified as important links in the global carbon cycle. They	
21	receive and transport terrestrial carbon from the land to the ocean and are also shown to be a	
22	net source of greenhouse gases (GHG), i.e carbon dioxide (CO <sub>2</sub> ) and methane (CH <sub>4</sub> )	
23	(Aufdenkampe et al., 2011;Battin et al., 2008;Cole et al., 2007;Tranvik et al., 2009). In a	
24	recent study, the global $CO_2$ emissions from rivers and streams were estimated to be 1.8±0.25	
25	Gt C year-1 (Raymond et al., 2013), which corresponds to 70% of the global ocean carbon	
26	sink (Le Quéré et al., 2014). Due to the lack of knowledge of surface area and gas exchange	
27	velocity, the smallest streams are considered as a major unknown component of regional to	
28	global scale GHG emission estimates (Bastviken et al., 2011;Cole et al., 2007). Despite these	
29	knowledge gaps, there are strong indications that small streams have the highest gas exchange	
30	velocities (Aufdenkampe et al., 2011), highest CO <sub>2</sub> partial pressures (Koprivnjak et al., 2010)	
31	and cover the largest fractional surface area within fluvial networks (Butman and Raymond,	
32	2011). A continental-scale analysis of CO <sub>2</sub> efflux from streams and rivers revealed a	
33	continuous decline of the fluxes with increasing size and discharge of the aquatic systems	
33 34	continuous decline of the fluxes with increasing size and discharge of the aquatic systems (Hotchkiss et al., 2015).	
33 34 35	<u>continuous decline of the fluxes with increasing size and discharge of the aquatic systems</u> (Hotchkiss et al., 2015) <u>.</u> Ecosystem-scale fluxes of CO <sub>2</sub> and CH <sub>4</sub> from running waters are often derived indirectly	
<ul><li>33</li><li>34</li><li>35</li><li>36</li></ul>	<ul> <li><u>continuous decline of the fluxes with increasing size and discharge of the aquatic systems</u></li> <li>(Hotchkiss et al., 2015).</li> <li>Ecosystem-scale fluxes of CO<sub>2</sub> and CH<sub>4</sub> from running waters are often derived indirectly</li> <li>using measured gas partial pressure in the surface water in combination with estimates of a</li> </ul>	
<ul> <li>33</li> <li>34</li> <li>35</li> <li>36</li> <li>37</li> </ul>	<ul> <li>continuous decline of the fluxes with increasing size and discharge of the aquatic systems</li> <li>(Hotchkiss et al., 2015).</li> <li>Ecosystem-scale fluxes of CO<sub>2</sub> and CH<sub>4</sub> from running waters are often derived indirectly</li> <li>using measured gas partial pressure in the surface water in combination with estimates of a</li> <li>gas exchange velocity. For sparingly soluble gases, the exchange velocity is mainly controlled</li> </ul>	
<ol> <li>33</li> <li>34</li> <li>35</li> <li>36</li> <li>37</li> <li>38</li> </ol>	<ul> <li>continuous decline of the fluxes with increasing size and discharge of the aquatic systems</li> <li>(Hotchkiss et al., 2015).</li> <li>Ecosystem-scale fluxes of CO<sub>2</sub> and CH<sub>4</sub> from running waters are often derived indirectly</li> <li>using measured gas partial pressure in the surface water in combination with estimates of a</li> <li>gas exchange velocity. For sparingly soluble gases, the exchange velocity is mainly controlled</li> <li>by turbulence at the water-side of the air-water interface. In smaller rivers and streams,</li> </ul>	
<ul> <li>33</li> <li>34</li> <li>35</li> <li>36</li> <li>37</li> <li>38</li> <li>39</li> </ul>	<ul> <li>continuous decline of the fluxes with increasing size and discharge of the aquatic systems</li> <li>(Hotchkiss et al., 2015).</li> <li>Ecosystem-scale fluxes of CO<sub>2</sub> and CH<sub>4</sub> from running waters are often derived indirectly</li> <li>using measured gas partial pressure in the surface water in combination with estimates of a</li> <li>gas exchange velocity. For sparingly soluble gases, the exchange velocity is mainly controlled</li> <li>by turbulence at the water-side of the air-water interface. In smaller rivers and streams,</li> <li>turbulence is driven by stream velocity, depth and bottom roughness (Marion et al., 2014),</li> </ul>	
<ul> <li>33</li> <li>34</li> <li>35</li> <li>36</li> <li>37</li> <li>38</li> <li>39</li> <li>40</li> </ul>	<ul> <li>continuous decline of the fluxes with increasing size and discharge of the aquatic systems</li> <li>(Hotchkiss et al., 2015).</li> <li>Ecosystem-scale fluxes of CO<sub>2</sub> and CH<sub>4</sub> from running waters are often derived indirectly</li> <li>using measured gas partial pressure in the surface water in combination with estimates of a</li> <li>gas exchange velocity. For sparingly soluble gases, the exchange velocity is mainly controlled</li> <li>by turbulence at the water-side of the air-water interface. In smaller rivers and streams,</li> <li>turbulence is driven by stream velocity, depth and bottom roughness (Marion et al., 2014),</li> <li>and the resulting gas exchange velocities are often parameterized with one or more of the</li> </ul>	
<ul> <li>33</li> <li>34</li> <li>35</li> <li>36</li> <li>37</li> <li>38</li> <li>39</li> <li>40</li> <li>41</li> </ul>	<ul> <li>continuous decline of the fluxes with increasing size and discharge of the aquatic systems</li> <li>(Hotchkiss et al., 2015).</li> <li>Ecosystem-scale fluxes of CO<sub>2</sub> and CH<sub>4</sub> from running waters are often derived indirectly</li> <li>using measured gas partial pressure in the surface water in combination with estimates of a</li> <li>gas exchange velocity. For sparingly soluble gases, the exchange velocity is mainly controlled</li> <li>by turbulence at the water-side of the air-water interface. In smaller rivers and streams,</li> <li>turbulence is driven by stream velocity, depth and bottom roughness (Marion et al., 2014),</li> <li>and the resulting gas exchange velocities are often parameterized with one or more of the</li> <li>following terms: stream order, slope, flow velocity, discharge, width and depth (Alin et al.,</li> </ul>	
<ul> <li>33</li> <li>34</li> <li>35</li> <li>36</li> <li>37</li> <li>38</li> <li>39</li> <li>40</li> <li>41</li> <li>42</li> </ul>	<ul> <li>continuous decline of the fluxes with increasing size and discharge of the aquatic systems</li> <li>(Hotchkiss et al., 2015).</li> <li>Ecosystem-scale fluxes of CO<sub>2</sub> and CH<sub>4</sub> from running waters are often derived indirectly</li> <li>using measured gas partial pressure in the surface water in combination with estimates of a</li> <li>gas exchange velocity. For sparingly soluble gases, the exchange velocity is mainly controlled</li> <li>by turbulence at the water-side of the air-water interface. In smaller rivers and streams,</li> <li>turbulence is driven by stream velocity, depth and bottom roughness (Marion et al., 2014),</li> <li>and the resulting gas exchange velocities are often parameterized with one or more of the</li> <li>following terms: stream order, slope, flow velocity, discharge, width and depth (Alin et al.,</li> <li>2011;Raymond et al., 2012;Wallin et al., 2011). In small streams, reach-scale estimates of the</li> </ul>	

Formatiert: Tiefgestellt

- 44 tracer (e.g., propane or sulfur hexafluoride) is injected upstream and the longitudinal decrease
- 45 of its dissolved concentration is measured (Halbedel and Koschorreck, 2013;Raymond et al.,
- 46 2012). For practical reasons, tracer gas injections are limited to application in small streams
- 47 and alternative methods suitable for a greater range of stream sizes are needed. Moreover,
- 48 recent studies revealed that the gas exchange velocity of CH<sub>4</sub> can be <u>enhanced by</u>
- 49 microbubbles (Beaulieu et al., 2012) and can therefor differ from that of the volatile tracer. To
- 50 better constrain ecosystem-scale estimates of GHG emissions and to improve the
- 51 understanding of the flux drivers in small running waters, reliable methods are required that
- 52 allow direct measurements.
- 53 As eddy-covariance (Baldocchi, 2014) measurements are not suitable for small streams, gas
- 54 flux chambers that float on the water surface are a straightforward and inexpensive method
- 55 for direct measurements of gas fluxes, and can easily be replicated over time and space
- 56 (Bastviken et al., 2015). The gas flux is determined from the change of the gas concentration
- 57 in the chamber headspace over time. Floating chambers have been frequently applied for
- 58 measuring gas fluxes in large rivers, reservoirs and lakes (e.g., Beaulieu et al.,
- 59 2014;DelSontro et al., 2011;Eugster et al., 2011).
- 60 Chamber measurements have been criticized because submerged chamber edges are thought
- 61 to disrupt the aquatic boundary layer, thereby affecting the gas exchange (Kremer et al.,
- 62 2003). Comparisons of floating chambers with other flux measurement techniques were
- 63 performed in lakes, rivers and estuaries. While some studies have reported a tendency of
- 64 floating chambers to yield higher fluxes than other methods (Raymond and Cole,
- 2001;Teodoru et al., 2015), others found reasonable agreement (Gålfalk et al., 2013;Cole et
  al., 2010).
- 67 In streams and rivers, floating chambers have been deployed anchored at one spot (anchored
- 68 chambers) (Sand-Jensen and Staehr, 2012;Crawford et al., 2013), or freely drifting with the
- 69 water (drifting chambers) (Alin et al., 2011;Beaulieu et al., 2012). Although based on the

# Gelöscht: y

Gelöscht: significantly larger Gelöscht: than that of CO<sub>2</sub>, which has been attributed to the presence of microbubbles (McGinnis et al., 2014;Prairie and del Giorgio, 2013)

75	same principle, the two deployment modes have fundamental differences. Because of the	
76	higher velocity difference between the chamber and the surface water, anchored chambers in	
77	running waters may create additional turbulence around the chamber edges (Kremer et al.,	
78	2003). If the effect of this turbulence on fluxes is minor, anchored chambers would be	Gelöscht: However, i
79	advantageous as the area covered by the chamber can be controlled and because practical	
80	work with anchored chambers is relatively simple. Drifting chambers will likely induce less	
81	turbulence in the surface water, however it is difficult to control their coverage, potentially	
82	resulting in spatially biased measurements. Drifting chambers are also complicated for several	
83	reasons, e.g., the presence of obstacles in the streams or in terms of logistics, as the chambers	
84	may travel far during measurement periods.	
85	While establishing efficient methods for running water gas emissions are needed to improve	
86	the global GHG budgets, progress in chamber based methods is prevented by the lack of	
87	comparative assessments of anchored versus drifting chambers. In this study, we compared	
88	measurements of GHG fluxes and the gas exchange velocity using drifting and anchored	
89	chambers in various streams and rivers. Because chamber performance is expected to depend	
90	strongly on chamber design, the field experiments were conducted using three different	
91	chamber types. In laboratory experiments, we analyzed the flow field and the turbulence	
92	under both anchored and drifting chambers at different flow velocities. The primary objective	
93	of this study was to answer the question: Do anchored chambers produce, reliable	Gelöscht: d
94	measurements of localized GHG fluxes in running waters?	Gelöscht:
95	2 Methods	
96	2.1 Chamber measurements in the field	
97	Field measurements were conducted in nine different rivers and streams in Germany and	
98	Poland using three different chambers ( <u>Table 1</u> ). All three data sets included <i>anchored</i>	Feldfunktion geändert
99	measurements, where the chambers were tethered to stay at a fixed position as well as drifting	

103	measurements, where the chambers were freely moving with the current. In two of the data	
104	sets (A and B), the temporal change of $\mathrm{CO}_2$ and $\mathrm{CH}_4$ concentration in the chamber headspace	
105	was measured on a boat using infrared gas analyzers (A: OA-ICOS gas analyzer, UGGA, Los	
106	Gatos Research Inc. USA, B: FTIR analyzer, Gasmet 4010, Gasmet, Finland). In the third	
107	data set (C), the gas concentration was measured using a built-in and low-cost CO2 sensor	
108	(ELG, SenseAir, Sweden). The chamber used in (C) is described in detail elsewhere	
109	(Bastviken et al., 2015).	
110	The chamber flux measurements were supplemented by measurements of dissolved gas	
111	concentrations (CO <sub>2</sub> and in data set A and B also CH <sub>4</sub> ) in the stream water and in the	
112	atmosphere (Table 1). Additional measurements include water temperature and near-surface	Feldfunktion geändert
113	current velocity, which was measured at selected sites within the study reaches using acoustic	
114	or electromagnetic current meters. More details on sampling and instrumentation are provided	
115	in <u>Appendix A</u> .	Feldfunktion geändert
115 116	in <u>Appendix A</u> . The flux $F$ (mmol m <sup>-2</sup> d <sup>-1</sup> ) of CO <sub>2</sub> (all data sets) and CH <sub>4</sub> (parts of data set A and B), was	Feldfunktion geändert
115 116 117	in <u>Appendix A</u> . The flux $F$ (mmol m <sup>-2</sup> d <sup>-1</sup> ) of CO <sub>2</sub> (all data sets) and CH <sub>4</sub> (parts of data set A and B), was calculated from the observed rate of change of the mole fraction $S$ (ppm s <sup>-1</sup> ) of the respective	Feldfunktion geändert
115 116 117 118	in <u>Appendix A</u> . The flux $F$ (mmol m <sup>-2</sup> d <sup>-1</sup> ) of CO <sub>2</sub> (all data sets) and CH <sub>4</sub> (parts of data set A and B), was calculated from the observed rate of change of the mole fraction $S$ (ppm s <sup>-1</sup> ) of the respective gas in the chamber using (Campeau and Del Giorgio, 2014):	Feldfunktion geändert
115 116 117 118 119	in <u>Appendix A</u> . The flux <i>F</i> (mmol m <sup>-2</sup> d <sup>-1</sup> ) of CO <sub>2</sub> (all data sets) and CH <sub>4</sub> (parts of data set A and B), was calculated from the observed rate of change of the mole fraction <i>S</i> (ppm s <sup>-1</sup> ) of the respective gas in the chamber using (Campeau and Del Giorgio, 2014): $F = (S \cdot V/A) \cdot t_1 \cdot t_2$ (1)	Feldfunktion geändert
<ol> <li>115</li> <li>116</li> <li>117</li> <li>118</li> <li>119</li> <li>120</li> </ol>	in <u>Appendix A</u> . The flux <i>F</i> (mmol m <sup>-2</sup> d <sup>-1</sup> ) of CO <sub>2</sub> (all data sets) and CH <sub>4</sub> (parts of data set A and B), was calculated from the observed rate of change of the mole fraction <i>S</i> (ppm s <sup>-1</sup> ) of the respective gas in the chamber using (Campeau and Del Giorgio, 2014): $F = (S \cdot V / A) \cdot t_1 \cdot t_2$ (1) Where <i>V</i> is the chamber gas volume (m <sup>3</sup> ), <i>A</i> is the chamber area (m <sup>2</sup> ), $t_1$ =8.64 × 10 <sup>4</sup> s d <sup>-1</sup> is the	Feldfunktion geändert
<ol> <li>115</li> <li>116</li> <li>117</li> <li>118</li> <li>119</li> <li>120</li> <li>121</li> </ol>	in <u>Appendix A</u> . The flux <i>F</i> (mmol m <sup>-2</sup> d <sup>-1</sup> ) of CO <sub>2</sub> (all data sets) and CH <sub>4</sub> (parts of data set A and B), was calculated from the observed rate of change of the mole fraction <i>S</i> (ppm s <sup>-1</sup> ) of the respective gas in the chamber using (Campeau and Del Giorgio, 2014): $F = (S \cdot V / A) \cdot t_1 \cdot t_2$ (1) Where <i>V</i> is the chamber gas volume (m <sup>3</sup> ), <i>A</i> is the chamber area (m <sup>2</sup> ), $t_1$ =8.64 × 10 <sup>4</sup> s d <sup>-1</sup> is the conversion factor from seconds to days, and $t_2$ is a conversion factor from mole fraction	Feldfunktion geändert
<ol> <li>115</li> <li>116</li> <li>117</li> <li>118</li> <li>119</li> <li>120</li> <li>121</li> <li>122</li> </ol>	in <u>Appendix A</u> . The flux <i>F</i> (mmol m <sup>-2</sup> d <sup>-1</sup> ) of CO <sub>2</sub> (all data sets) and CH <sub>4</sub> (parts of data set A and B), was calculated from the observed rate of change of the mole fraction <i>S</i> (ppm s <sup>-1</sup> ) of the respective gas in the chamber using (Campeau and Del Giorgio, 2014): $F = (S \cdot V/A) \cdot t_1 \cdot t_2$ (1) Where <i>V</i> is the chamber gas volume (m <sup>3</sup> ), <i>A</i> is the chamber area (m <sup>2</sup> ), $t_1$ =8.64 × 10 <sup>4</sup> s d <sup>-1</sup> is the conversion factor from seconds to days, and $t_2$ is a conversion factor from mole fraction (ppm) to concentration (mmol m <sup>-3</sup> ) at <i>in-situ</i> temperature ( <i>T</i> in K) and atmospheric pressure ( <i>p</i>	Feldfunktion geändert
<ol> <li>115</li> <li>116</li> <li>117</li> <li>118</li> <li>119</li> <li>120</li> <li>121</li> <li>122</li> <li>123</li> </ol>	in <u>Appendix A</u> . The flux <i>F</i> (mmol m <sup>-2</sup> d <sup>-1</sup> ) of CO <sub>2</sub> (all data sets) and CH <sub>4</sub> (parts of data set A and B), was calculated from the observed rate of change of the mole fraction <i>S</i> (ppm s <sup>-1</sup> ) of the respective gas in the chamber using (Campeau and Del Giorgio, 2014): $F = (S \cdot V / A) \cdot t_1 \cdot t_2$ (1) Where <i>V</i> is the chamber gas volume (m <sup>3</sup> ), <i>A</i> is the chamber area (m <sup>2</sup> ), $t_1$ =8.64 × 10 <sup>4</sup> s d <sup>-1</sup> is the conversion factor from seconds to days, and $t_2$ is a conversion factor from mole fraction (ppm) to concentration (mmol m <sup>-3</sup> ) at <i>in-situ</i> temperature ( <i>T</i> in K) and atmospheric pressure ( <i>p</i> in Pa), according to the ideal gas law:	Feldfunktion geändert
<ol> <li>115</li> <li>116</li> <li>117</li> <li>118</li> <li>119</li> <li>120</li> <li>121</li> <li>122</li> <li>123</li> <li>124</li> </ol>	in <u>Appendix A</u> . The flux <i>F</i> (mmol m <sup>-2</sup> d <sup>-1</sup> ) of CO <sub>2</sub> (all data sets) and CH <sub>4</sub> (parts of data set A and B), was calculated from the observed rate of change of the mole fraction <i>S</i> (ppm s <sup>-1</sup> ) of the respective gas in the chamber using (Campeau and Del Giorgio, 2014): $F = (S \cdot V / A) \cdot t_1 \cdot t_2$ (1) Where <i>V</i> is the chamber gas volume (m <sup>3</sup> ), <i>A</i> is the chamber area (m <sup>2</sup> ), $t_1$ =8.64 × 10 <sup>4</sup> s d <sup>-1</sup> is the conversion factor from seconds to days, and $t_2$ is a conversion factor from mole fraction (ppm) to concentration (mmol m <sup>-3</sup> ) at <i>in-situ</i> temperature ( <i>T</i> in K) and atmospheric pressure ( <i>p</i> in Pa), according to the ideal gas law: $t_2 = p / (8.31 \text{ J K}^{-1} \text{ mole}^{-1} \cdot T) \cdot 1000$ (2)	Feldfunktion geändert
<ol> <li>115</li> <li>116</li> <li>117</li> <li>118</li> <li>119</li> <li>120</li> <li>121</li> <li>122</li> <li>123</li> <li>124</li> <li>125</li> </ol>	in <u>Appendix A</u> . The flux <i>F</i> (mmol m <sup>-2</sup> d <sup>-1</sup> ) of CO <sub>2</sub> (all data sets) and CH <sub>4</sub> (parts of data set A and B), was calculated from the observed rate of change of the mole fraction <i>S</i> (ppm s <sup>-1</sup> ) of the respective gas in the chamber using (Campeau and Del Giorgio, 2014): $F = (S \cdot V / A) \cdot t_1 \cdot t_2$ (1) Where <i>V</i> is the chamber gas volume (m <sup>3</sup> ), <i>A</i> is the chamber area (m <sup>2</sup> ), $t_1$ =8.64 × 10 <sup>4</sup> s d <sup>-1</sup> is the conversion factor from seconds to days, and $t_2$ is a conversion factor from mole fraction (ppm) to concentration (mmol m <sup>-3</sup> ) at <i>in-situ</i> temperature ( <i>T</i> in K) and atmospheric pressure ( <i>p</i> in Pa), according to the ideal gas law: $t_2 = p / (8.31 \text{ J K}^{-1} \text{ mole}^{-1} \cdot T) \cdot 1000$ (2) The gas exchange velocity of the respective gas at in-situ temperature <i>k</i> (m d <sup>-1</sup> ) was estimated	Feldfunktion geändert

$$k = F / (K_H \cdot (p^{water} - p^{air}))$$
(3)

128	using the partial pressure of $CO_2$ and $CH_4$ in the stream water ( $p^{water}$ ) and in the atmosphere	
129	$(p^{air})$ . The partial pressures were obtained by multiplication of the measured mole fraction	
130	with atmospheric pressure. $K_H$ is the temperature-dependent Henry constant (mmol m <sup>-3</sup> Pa <sup>-1</sup> )	
131	(Goldenfum, 2011). The <i>in-situ</i> gas exchange velocities were converted to a standardized	
132	(independent of temperature and gas diffusivity) exchange velocity $k_{600}$ using the Schmidt	
133	number dependence:	
134	$k_{600} = k \cdot (600 / Sc)^{-n} \tag{4}$	
135	where the temperature-dependent Schmidt numbers $(Sc)$ of both gases were estimated	
136	according to Goldenfum (2011). The Schmidt-number exponent <i>n</i> describes the dependence	
137	of the gas exchange velocity of a particular gas on the diffusion coefficient of this gas in	
138	water. We used $n=0.5$ , which showed best agreement with measurements for wave-covered	Gelöscht: was
120	and turbulant water surfaces (Jähne and Haufesher, 1008)	Gelöscht: fixed at
139	and turburent water surfaces (Jamie and Haubecker, 1998).	<b>Formatiert:</b> Schriftart: K
140		
141	2.2 Turbulence measurements in the lab	
142	The flow fields under freely drifting and anchored chambers were measured using particle	
143	image velocimetry (PIV) in a 3 m long laboratory flume. The chamber type and geometry was	
144	identical to the chamber in data set C (Table 1). The flow field under the drifting chamber was	Feldfunktion geändert
145	measured for 50 repeated chamber runs (58 s cumulative velocity observations under the	
146	chamber) at a mean flow velocity of 0.10 m s <sup>-1</sup> , the highest flow velocity that could be	
147	realized in the flume. Measurements under anchored chambers were performed for 90 s at a	
148	mean flow velocity of 0.10 m s <sup>-1</sup> . Additional measurements were performed at reduced mean	
149	flow velocities of 0.08 and 0.06 m s <sup>-1</sup> . As a reference, the undisturbed flow field without	
150	chambers was measured for 90 s. Due to the limited length of the laboratory flume it was not	
151	possible to measure gas fluxes or estimate the gas exchange velocities.	
152	The flow fields were analyzed by illuminating neutrally buoyant seeding particles (diameter	
153	of 20 $\mu$ m, polyethylene) within a thin light sheet produced by a double-pulse laser	
	7	

cht: was cht: fixed at tiert: Schriftart: Kursiv



<sup>172 2.3</sup> Statistics

174 sets were compared using paired *t*-tests, comparisons between the data sets were performed

175 using 2-sample *t*-tests. Spearman rank correlations coefficients (*r*<sub>S</sub>) were estimated when

176 testing for correlations between gas exchange velocities from anchored and drifting chambers

177 for each data set. All analyses were performed at a significance level p < 0.05, unless stated

178 otherwise.

Gelöscht: . Gelöscht: These flow fields were composed from individual observations by assembling the velocity vectors relative to the (moving) leading edge of the chamber.

Gelöscht:  $TKE = \frac{3}{4} \overline{(u'^2 + w'^2)}$ 

Gelöscht: Longitudinally extended

<sup>173</sup> The mean fluxes measured with anchored and drifting chambers in the respective field data

# 185 3 Results

186	3.1 Drifting vs. anchored chamber measurements in the field		
187	In all measurements, the CO2 and CH4 fluxes were positive, i.e. the streams were sources of	_	Gelöscht: measured
188	both gases to the atmosphere. While the mean CO <sub>2</sub> fluxes measured by drifting chambers did		
189	not differ significantly among the data sets B and C, they were about seven-fold higher in data		
190	set A ( <u>Table 2</u> ). In all data sets, anchored chamber fluxes were significantly higher than the		Feldfunktion geändert
191	corresponding drifting chamber fluxes.		
192	Gas exchange velocities $k_{600}$ estimated from CO <sub>2</sub> measurements in the drifting chamber		
193	deployments ( $k_{600\_CO2\_d}$ ) ranged between 0.2 and 8.1 m d <sup>-1</sup> . They varied widely within each		
194	data set ( <u>Table 2</u> ), but in contrast to the current velocities mean values of $k_{600\_CO2\_d}$ did not		Feldfunktion geändert
195	significantly differ among the data sets. In all data sets, however, $k_{600}$ from anchored		
196	chambers ( $k_{600\_CO2\_a}$ ) differed significantly from that of drifting chambers (Fig. 1A). Except		Feldfunktion geändert
197	for data set A, both were weakly correlated to each other ( $r_s = 0.49$ , $p=0.01$ and $r_s = 0.76$ ,		
198	p<0.001 for data set B and C, respectively) (Fig. 1B). With only a few exceptions, the gas		Feldfunktion geändert
199	exchange velocities under anchored chambers were higher than those under drifting chambers		Gelöscht: .
200	with individual measurements, $k_{600} c_{O2_a}$ being up to 20 times higher than $k_{600} c_{O2_a}$ . The		Gelöscht: While
201	average ratio of both velocities was 2.2, 6.2 and 4.0 for data set A, B and C, respectively	$\square$	Gelöscht: in Gelöscht: was
202	(Table 2).		Gelöscht: ,
202	When both access were measured, the acc exchange valuation estimated from CO. fluxes were		Feldfunktion geändert
205	when bour gases were measured, the gas exchange velocities estimated from CO <sub>2</sub> fluxes were		
204	strongly correlated to those estimated from CH <sub>4</sub> measurements for both deployment types.		
205	Small but significant differences were observed between $k_{600\_CO2\_d}$ and $k_{600\_CH4\_d}$ , whereas the		
206	$\mathrm{CO}_2$ based estimates were on average slightly higher in data set A and lower in data set B		
207	(Fig. 1A). In accordance with the CO <sub>2</sub> based estimates, $k_{600}$ estimated from CH <sub>4</sub> was higher		Feldfunktion geändert
208	under anchored than under drifting chambers ( <u>Table 2</u> ) and the ratio $k_{600_a} / k_{600_d}$ did not		Feldfunktion geändert
209	differ significantly between both gases.		

217	When combining all data sets, there was no correlation between gas exchange velocities and		
218	the measured current velocity for drifting chambers for either CO <sub>2</sub> or CH <sub>4</sub> (Fig. 2A).	(	Feldfunktion geändert
219	However, for anchored chamber deployments, $k_{600_a}$ was positively correlated to current speed		
220	in data set A ( $r_s$ =0.54, $p$ =0.02) and B ( $r_s$ =0.7, $p$ <0.001). The ratio of the gas exchange		
221	velocities estimated from both deployment types was positively correlated to current speed		
222	when all three data sets were combined ( $r_s=0.66$ , $p<0.001$ ), but no significant correlations		
223	were observed within the individual data sets (Fig. 2B).		Feldfunktion geändert
224			
225	3.2 Flow field and turbulence under chambers		
226	The laboratory measurements revealed pronounced differences in the flow fields and		
227	turbulence under the anchored and drifting chambers. The mean longitudinal flow velocity		
228	was strongly reduced within the submerged part of the anchored chamber and increased		
229	below the submerged chamber edge. Recirculating eddies were formed under the leading		
230	(upstream) edge of the chamber (vector graphs of the mean velocity distributions are provided		
231	in <u>Appendix B</u> ). These eddies detached and injected turbulence below the chamber (Fig. 3).		Feldfunktion geändert
232	The turbulent kinetic energy which was produced by the submerged edge of the anchored		Feldfunktion geändert
233	chambers increased with increasing current speed ( <u>Appendix B</u> ). Under the drifting chambers,	(	Feldfunktion geändert
234	the flow velocities were slightly enhanced below the submerged chamber edge, but no		
235	recirculating eddies were formed.		
236	The penetration depth of the chamber edges varied with time as the chamber moved vertically		
237	on the rough water surface (see Appendix B for snapshots of instantaneous velocity	(	Feldfunktion geändert
238	distributions and chamber penetration). However, at the same flow velocity the average		
239	penetration depth of the anchored chamber was higher than that of the drifting chamber (Fig.		Feldfunktion geändert
240	<u>3</u> ).		

# 241 4 Discussion

# 242 4.1 Chamber bias in anchored deployments

243	Our field observations showed consistently higher gas exchange velocities and gas fluxes
244	measured with anchored in comparison to freely drifting chambers in a variety of small
245	streams with flow velocities between 0.08 and 0.8 m s <sup>-1</sup> . Detailed observations of the flow
246	field and turbulence under both types of chambers in the laboratory revealed a reduction of
247	mean flow velocity and the generation of chamber-induced turbulence due to the shedding of
248	eddies at the upstream part of the submerged edge of the anchored chamber. Under identical
249	hydraulic conditions, anchored chambers penetrated deeper into the water, which we attribute
250	to a partial diversion of the strong horizontal drag force imposed by the flow into the vertical
251	direction. In combination, horizontal current shear and deeper penetration caused an increase
252	in magnitude of chamber-induced turbulence with increasing difference in velocity between
253	the water flow and the chamber (Fig. B1). This mechanism has been suggested in previous
254	studies of floating chamber performance in water bodies, although there are mixed results
255	regarding its importance (Cole et al., 2010;Gålfalk et al., 2013;Vachon et al., 2010).
256	The laboratory observation agrees with our field measurements, where the ratio of the fluxes
257	measured with anchored and with drifting chambers was comparably small at flow velocities
258	<0.2 m s <sup>-1</sup> . However, even at low flow velocities, the gas exchange velocity was enhanced by
259	more than a factor of two in the anchored deployment. At higher flow velocities (> $0.2 \text{ m s}^{-1}$ )
260	typical for rivers and streams, chamber-induced turbulence obviously dominated the gas flux
261	into the anchored chambers.
262	The large (several-fold) potential overestimation of fluxes measured with anchored chambers
263	calls into question its suitability for application in running waters, particularly at high flow

264 rates. This agrees with the observations of Teodoru et al. (2015) who reported a linear

- 265 dependency of the gas exchange velocity under anchored chambers on the water velocity
- 266 relative to the chamber in a large river.
- 267 4.2 Correction methods and chamber optimization
- 268 The correlation of the anchored chamber gas exchange velocity with flow velocity observed
- 269 in our study could provide a potential means for correcting the artificial chamber flux, if the
- 270 corresponding drifting chamber gas exchange velocity was also a function of flow velocity.
- 271 However, no such correlation was present in our field observations, indicating that near-
- 272 surface flow velocity is a poor predictor for the gas exchange velocities in streams. Therefore,
- it can be expected that river depth and bed roughness affect the near-surface turbulence more
- than flow velocity (Moog and Jirka, 1999;Raymond et al., 2012).
- 275 As the correction of the effects of chamber-induced turbulence on measured fluxes seems
- 276 unlikely, it would be more reasonable to optimize the chamber design to completely avoid or
- 277 to at least reduce this effect. The rectangular chamber B produced the largest error, although it
- 278 remained unclear from our measurements whether this was caused by the geometry of the
- 279 chamber or by the high flow velocity in data set B. On this basis, we recommend the use of
- 280 more streamlined circular chambers to minimize the error under drifting conditions. Crawford
- et al. (2013) and McMahon and Dennehy (1999) used streamlined (canoe-shaped) instead of
- 282 cylindrical or rectangular chambers to minimize the generation of chamber-induced
- 283 turbulence at the upstream chamber edge during anchored chamber deployments. However,
- they did not provide evidence that this goal was reached.
- 285 Another approach to minimize the bias of anchored chambers would be to design chambers
- 286 without submerged rigid walls. Submergence of the chamber edges can be avoided
- 287 completely by using a thin plastic foil which adheres to the water surface to seal the chamber
- headspace (Fig. 4A). Laboratory (PIV) measurements of the flow field were performed under
- a foil, mimicking a chamber deployed in anchored mode. The measurements revealed a strong

reduction of flow disturbances and chamber-induced turbulence (Fig. 4) in comparison to
both anchored and drifting chambers. Such "flying chambers" require a frame to keep the
chamber above the water surface, which can be supported by floats at a larger lateral distance
to the chamber or, in small streams, also by a fixation at the river bank.

294

#### 295 4.3 Implications for chamber-based flux measurements

296 Our study clearly shows that anchored chambers strongly overestimate the gas flux in running 297 water and are not suited to quantify greenhouse gas fluxes in streams and rivers. One possible 298 way forward to reduce this bias while still maintaining the practical advantages of the 299 anchored chambers could be flying (anchored) chambers with flexible foil sealing at the water 300 surface. Drifting chambers provide a practical and reliable solution, although they are not free 301 of potential spatial bias. Because their measurement locations are difficult to control, their 302 trajectories may not be representative for the areal mean flux from the study reach. Regions 303 with locally enhanced turbulence, e.g., stream-reaches with large emerging roughness of the 304 river bed, cannot be surveyed with drifting chambers, however the gas exchange velocity is 305 highest at these sites (Moog and Jirka, 1999). Similarly, mean-flow trajectories may bypass 306 backwaters and regions of reduced flow velocity along the stream banks. Observations in 307 reservoirs and river impoundments revealed that the enhanced sedimentation of particulate 308 organic matter can make these zones emission hot spots (Maeck et al., 2013;DelSontro et al., 309 2011). Anchored chamber deployments may provide a useful extension of drifting chamber 310 measurements at such sites, if the flow velocity is sufficiently small. To truly validate a 311 reliable chamber method for small streams, a multi-method comparison study, including 312 tracer additions, should be performed. 313 This study shows that flux chamber approaches to measure GHG fluxes from running waters

314 have a high potential, given sufficient knowledge about appropriate chamber design and

315 deployment approaches. Thus, flux chambers are emerging as an important method to

- 316 constrain greenhouse gas fluxes from stream networks.
- 317

### 318 Acknowledgments

- 319 Parts of this study were financially supported by the German Research Foundation (grant no.:
- 320 LO 1150/9-1) and conducted within the LandScales project ('Connecting processes and
- 321 structures driving the landscape carbon dynamics over scales') financed by the the Leibniz
- 322 Association within the Joint Initiative for Research and Innovation (BMBF) and (partially)
- 323 carried out within the SMART Joint Doctorate (Science for the MAnagement of Rivers and
- 324 their Tidal systems) funded with the support of the Erasmus Mundus program of the European
- 325 Union and the Swiss National Science Foundation (Grant Nr. PA00P2\_142041). The
- 326 development and production of the chambers with built in CO<sub>2</sub> loggers (data set C) was
- 327 supported by the Swedish Research Council VR. Funding for an initial workshop was carried
- 328 out by the IGB cross-cutting research domain 'Aquatic Boundaries and Linkages'. We
- 329 gratefully acknowledge the financial support of German Academic Exchange Service
- 330 (DAAD) (Sustainable water management Program (NAWAM), Grant number: A/12/91768).
- 331 We thank Simone Langhans for her fruitful input which shaped the core idea of the presented
- 332 <u>study. Finally, we thank the two anonymous reviewers for constructive inputs that improved</u>
- 333 <u>the manuscript.</u>
- 334

### 335 Appendices

#### 336 Appendix A: Additional information on the field data sets

#### 337 A1: Data set A

- 338 Field measurements of five streams in North Central European Plains in Germany and Poland
- 339 were conducted during October 2014. Gaseous CO<sub>2</sub> and CH<sub>4</sub> emissions were measured at the
- 340 water-air interface with a drifting chamber attached to an Ultraportable Greenhouse Gas
- 341 Analyzer (UGGA; Los Gatos Research, Inc., USA). The chamber was connected to the
- 342 UGGA placed in a boat via two gas tight tubes (Tygon 2375), creating a circulation of air
- 343 being sucked in and pumped out. For the anchored measurements, we tethered the chamber to
- 344 a rack in the middle of the respective stream, in which we placed the sensors for continuously
- 345 dissolved CO<sub>2</sub> and CH<sub>4</sub> measurements (HydroC<sup>™</sup>; CONTROS Systems & Solutions GmbH,
- 346 Germany). Subsequently, we floated down a predefined stream section with the same
- 347 chamber following freely the boat or vice versa at the speed of the current. During the
- 348 chamber measurements, the UGGA continuously measured the gaseous CO<sub>2</sub> and CH<sub>4</sub>
- 349 accumulation in the chamber (frequency 1 s). Flow velocity was measured with an Acoustic
- 350 Digital Current Meter (OTT, Germany).

- 352 A2: Data set B
- 353 Measurements were performed on the Bode River between Egeln-Nord and Staßfurt on 7
- April 2014 (summer base flow 7.7 m<sup>3</sup> s<sup>-1</sup>) and 12 March 2015 (winter high flow 12.8 m<sup>3</sup> s<sup>-1</sup>).
- 355 The flux of CO<sub>2</sub> and CH<sub>4</sub> between water and atmosphere was measured by a rectangular
- 356 floating chamber, which was connected to an FTIR analyzer (GASMET 4010, Finland).
- 357 Measurements were performed from a boat while drifting down the river. For a single
- measurement, the chamber was placed at the water surface <u>for up to five minutes</u> and CO<sub>2</sub> and

359	CH4 change inside the chamber was measured every 30 s. To compare drifting and fixed	Gelöscht: for up to 5 minu
360	chamber measurements, the boat was then stopped by an anchor and measurements continued	
361	for another 3-5 min. During this stationary measurement, current velocity was measured with	
362	an electromagnetic current meter (MF-Pro, Ott, Germany) and water temperature were	
363	measured by hand held probes (ProfiLine Multi,WTW, Germany).	
364	The concentration of $\text{CO}_2$ in the water was continuously measured by a submersible probe	
365	(HydroC <sup>TM</sup> ; CONTROS Systems & Solutions GmbH, Germany). Additionally samples for	
366	CH4 analysis were taken in plastic syringes and later analyzed by headspace GC.	
367	Water temperature was continuously measured by temperature loggers (Tidbit, Onset,	
368	U.S.A.). The barometric pressure was recorded by the FTIR analyzer.	
369	Under drifting conditions the CH4 flux was often below the detection limit while there was	
370	always a positive CH4 flux in anchored chamber deployments.	
371		
372	A3: Data set C	
373	Chambers with a cross-sectional area of 0.066 $m^2$ and volume of 6.8 L were covered by	
374	aluminum foil to reduce the internal heating and equipped with a Styrofoam material to keep	
375	the chamber body floating on water surface. The chambers were equipped with an internal	
376	CO2 logger system that is positioned inside the headspace of the chamber (Bastviken et al.,	
377	2015). The non-dispersive infrared (NDIR) CO2 logger (ELG, SenseAir, Sweden,	
378	www.senseair.se) measures CO2 in the range of 0-5000 ppm. The logger measures	Feldfunktion geändert
379	simultaneously CO2, temperature and relative humidity, and operates at temperature and	
380	humidity of 0-50 °C and 0-99% (non-condensing conditions) respectively. The loggers were	
381	calibrated by the manufacturer and operated with 9 VDC batteries. The measurement interval	
382	was adjusted to be 30 s, more information of technical specifications are provided elsewhere	
383	(Bastviken et al., 2015).	

or up to 5 minutes

385	Chambers were deployed fixed at a certain position (anchored) and freely drifting. Triplicate
386	measurements were conducted during each drifting run, and three runs were conducted at
387	each site. The anchored chambers were then used for measuring the flux of $CO_2$ at different
388	locations along the pathways of the drifting chambers. The chamber flux measurements were
389	supplemented by measurements of dissolved gas CO2 and CH4 concentrations in the stream
390	waters at each anchored stations for each run. Continuous measurements of CO <sub>2</sub> and methane
391	in the middle of the stream were conducted using a membrane equilibrator (Liqui-Cel
392	MiniModule, Membrana, USA) connected with an Ultraportable Greenhouse Gas Analyzer
393	(UGGA; Los Gatos Research, Inc., USA). The water samples were pumped through the

394 membrane contactor using a peristaltic pump at a constant flow rate.









# 407 Tables

### 408 Table 1

г

- 409 **Table 1:** Summary of the three data sets obtained in field measurements. Pictures show the
- 410 three different chambers used for the anchored and drifting approach. Additional information
- 411 about the sampling procedures are provided in the <u>Supplementary Information</u>.

Data set	Α	В	С
Site	5 different streams	Bode river,	3 different streams,
	North-Central	Harz mountains,	Upper Rhine Valley,
	European Plains in	Central Germany	South-West Germany
	Germany and Poland		
Chamber volume (m <sup>3</sup> )	0.0168	0.0147	0.0068
Chamber area (m <sup>2</sup> )	0.126	0.098	0.066
(shape)	(circular)	(rectangular)	(circular)
Chamber height	0.175	0.15	0.13
(m)			
Penetration depth	0.018	0.023	0.025
(m)			
Chamber gas	LosGatos, CO <sub>2</sub> , CH <sub>4</sub>	FTIR analyzer	Built-in low-cost CO <sub>2</sub>
measurement	on boat	(GASMET, Finland)	logger (ELG by
		on boat	SenseAir, Sweden)
Dissolved gas	Contros CO <sub>2</sub> and CH <sub>4</sub>	Contros CO <sub>2</sub> , CH <sub>4</sub>	UGGA with
measurement		with GC	membrane contactor
Drifting	following boat or vice	Freely drifting while	Freely drifting
measurements	versa	followed with boat	
Anchored	Tethered to a rack in	Tethered to anchored	Tethered with rope
measurements	the middle of the stream	boat	from above
Number of	At 5 sites: 2-5 pairs of	For two different	At 3 sites: 2-3
measurements	anchored chamber	discharge situations:	subsequent floating
	measurements	10-13 pairs of	chamber runs and 5
	(upstream) and	subsequent drifting	parallel anchored
	subsequent floating	and anchored	chambers distributed
	chamber runs	chamber	along the trajectory of
		measurements down	the floating chamber
		the river using a	
		single chamber	

# 412 Table 2

413	Table 2: Discharge rate, flow velocities, gas fluxes ( $F_{CO2}$ , $F_{CH4}$ ), and gas exchange velocities	Gelöscht: F
414	$(k_{600\_CO2}, k_{600\_CH4})$ estimated from drifting (subscript d) and from anchored (subscript a)	
415	chambers during the three field campaigns (A-C, cf. Table 1). Except for discharge, all values	Gelöscht: A
116	are given as mean + standard deviation	Feldfunktion geändert
#10	are given as mean $\pm$ standard deviation,	 Gelöscht: .
I	Data set A B C	

Data set	Α	В	С
No. of samples <i>n</i>	n <sub>CO2</sub> =18	<i>n</i> <sub>CO2</sub> =27	<i>n</i> <sub>CO2</sub> =24
	n <sub>CH4</sub> =18	<i>n</i> <sub>CH4</sub> =9	<i>n</i> <sub>CH4</sub> =0
Discharge (m <sup>3</sup> s <sup>-1</sup> )	<u>0.6 – 1.4</u>	<u>7.7 – 12.8</u>	<u>0.1 – 7.6</u>
Flow velocity (m s <sup>-1</sup> )	$0.21\pm0.07$	$0.60 \pm 0.12$	$0.30\pm0.07$
$F_{\text{CO2}_a} (\text{mmol } \text{m}^{-2} \text{ day}^{-1})$	$742 \pm 282$	$302\pm148$	$103 \pm 47$
$F_{\text{CO2}_d} (\text{mmol m}^{-2} \text{ day}^{-1})$	363 ± 139	55 ± 30	$49 \pm 36$
$k_{600}$ _CO2_a (m day <sup>-1</sup> )	$6.5 \pm 1.4$	$17 \pm 6.4$	$4.1 \pm 2.8$
$k_{600}$ _CO2_d (m day <sup>-1</sup> )	3.3 ± 1.1	$3.2 \pm 1.5$	$2.1 \pm 2.5$
k <sub>600_CO2_a</sub> / k <sub>600_CO2_d</sub>	$2.2\pm0.9$	$6.2 \pm 3.2$	$4.0 \pm 5.0$
$F_{CH4_a}$ (mmol m <sup>-2</sup> day <sup>-1</sup> )	$4.31 \pm 1.35$	$1.55\pm0.71$	-
$F_{CH4_d}$ (mmol m <sup>-2</sup> day <sup>-1</sup> )	$2.12 \pm 0.86$	$0.37\pm0.16$	-
$k_{600}$ _CH4_a (m day <sup>-1</sup> )	$6.0 \pm 1.4$	$23.0\pm10.8$	-
$k_{600}$ _CH4_d (m day <sup>-1</sup> )	$2.9\pm0.9$	$5.5 \pm 2.4$	-
k <sub>600_CH4_a</sub> / k <sub>600_CH4_d</sub>	$2.3 \pm 1.0$	$4.8 \pm 2.1$	-

ormatiert: Schriftart: Nicht Fett		
formatiert: Links		
Formatiert: Hochgestellt		
ormatiert: Schriftart: Nicht Fett		

### 421 Figures

### 422 Figure 1



**Fig. 1:** A) Box plots of the standardized gas exchange ( $k_{600}$ ) velocity measured using drifting (solid lines) and anchored (dashed lines) flux chambers in data set A (black), B (red) and C (blue). The diamond-shaped boxes encompass the 25-75 percentile range, whiskers show minimum and maximum, open squares and horizontal lines mark mean and median values, respectively. B)  $k_{600}$  estimated from anchored chamber deployments versus that from drifting chambers for the data sets A-C (color). Filled symbols show  $k_{600}$  estimated from CO<sub>2</sub> fluxes, open symbols are based on CH<sub>4</sub> fluxes. The solid line show<u>s</u> a 1:1 relationship.



426 **Fig. 2:** A) Gas exchange velocity  $k_{600}$  from anchored (triangles) and drifting (circles)

427 chambers versus current velocity for the three field data sets (A-C, colors). Filled symbols

428 show data obtained from CO<sub>2</sub>, open symbols are based on CH<sub>4</sub> fluxes. B) Ratio of the gas

429 exchange velocities from anchored and drifting chambers versus current speed (filled

430 symbols: CO<sub>2</sub>, open symbols: CH<sub>4</sub>, symbol color indicates data set). The dashed line indicates

431 a constant ratio of one <u>and the solid line shows a linear regression of the combined data sets</u>

432  $(r_{s}=0.66, p<0.001).$ 

### 435 Figure 3



437 Fig. 3: Laboratory measurements of the mean longitudinal flow velocities (U) A) below a

438 drifting and B) below an anchored chamber. Mean turbulent kinetic energy (TKE) of the flow

- 439 fields below C) the drifting chamber and D) the anchored chamber. *z* and *x* refer to depth and
- 440 longitudinal distance respectively. Chamber edges are masked out (white) and regions without
- 441 sufficient observations for temporal averaging are marked by dark blue color. The flow
- 442 direction is from left to right and the mean flow velocity was  $0.1 \text{ m s}^{-1}$ .

443

### 444 Figure 4



**Fig. 4:** A) Flying chamber design without penetration of the water surface by the chamber edges but using a plastic foil collar (marked by the red arrow) for sealing. The chamber is fixed above the water surface by a supporting frame. B) Distribution of mean longitudinal flow velocities (U) and B) turbulent kinetic energy (TKE) of the flow field below the front edge of a static foil (marked by black bar) at the water surface. The direction of flow was from left to right, x and y refer to longitudinal distance and depth, respectively. The mean flow velocity was 0.10 m s<sup>-1</sup>. Color scales are identical to that of Fig. 3.

#### References 446

- 447 Alin, S. R., Rasera, M., Salimon, C. I., Richey, J. E., Holtgrieve, G. W., Krusche, A. V., and 448 Snidvongs, A.: Physical controls on carbon dioxide transfer velocity and flux in low-449 gradient river systems and implications for regional carbon budgets, J. Geophys. Res.-450 Biogeo., 116, G01009, 10.1029/2010jg001398, 2011.
- 451 Aufdenkampe, A. K., Mayorga, E., Raymond, P. A., Melack, J. M., Doney, S. C., Alin, S. R., 452 Aalto, R. E., and Yoo, K.: Riverine coupling of biogeochemical cycles between land, 453 oceans, and atmosphere, Front. Ecol. Environ., 9, 53-60, 10.1890/100014, 2011.
- 454 Baldocchi, D.: Measuring fluxes of trace gases and energy between ecosystems and the 455 atmosphere - the state and future of the eddy covariance method, Global Change Biol., 456 20, 3600-3609, 10.1111/gcb.12649, 2014.
- 457 Bastviken, D., Tranvik, L. J., Downing, J. A., Crill, P. M., and Enrich-Prast, A.: Freshwater 458 Methane Emissions Offset the Continental Carbon Sink, Science, 331, 50-50, 459 10.1126/science.1196808, 2011.
- 460 Bastviken, D., Sundgren, I., Natchimuthu, S., Reyier, H., and Gålfalk, M.: Technical Note: 461 Cost-efficient approaches to measure carbon dioxide (CO2) fluxes and concentrations in 462 terrestrial and aquatic environments using mini loggers, Biogeosciences, 12, 3849-3859, 463 10.5194/bg-12-3849-2015, 2015.
- 464 Battin, T. J., Kaplan, L. A., Findlay, S., Hopkinson, C. S., Marti, E., Packman, A. I., 465 Newbold, J. D., and Sabater, F.: Biophysical controls on organic carbon fluxes in fluxial 466 networks, Nature Geosci., 1, 95-100, 10.1038/ngeo101, 2008.
- 467 Beaulieu, J. J., Shuster, W. D., and Rebholz, J. A.: Controls on gas transfer velocities in a 468 large river, J. Geophys. Res.-Biogeo., 117, G02007, 10.1029/2011jg001794, 2012.
- 469 Beaulieu, J. J., Smolenski, R. L., Nietch, C. T., Townsend-Small, A., and Elovitz, M. S.: High 470 Methane Emissions from a Midlatitude Reservoir Draining an Agricultural Watershed, 471 Environ. Sci. Technol., 48, 11100-11108, 10.1021/es501871g, 2014.
- 472 Butman, D., and Raymond, P. A .: Significant efflux of carbon dioxide from streams and 473 rivers in the United States, Nature Geosci., 4, 839-842, 2011.
- Campeau, A., and Del Giorgio, P. A.: Patterns in CH4 and CO2 concentrations across boreal 474 475 rivers: Major drivers and implications for fluvial greenhouse emissions under climate 476 change scenarios, Glob Chang Biol, 20, 1075-1088, 2014.
- 477 Cole, J. J., Prairie, Y. T., Caraco, N. F., McDowell, W. H., Tranvik, L. J., Striegl, R. G., 478 Duarte, C. M., Kortelainen, P., Downing, J. A., Middelburg, J. J., and Melack, J.: 479 Plumbing the global carbon cycle: Integrating inland waters into the terrestrial carbon 480 budget, Ecosystems, 10, 171-184, 10.1007/s10021-006-9013-8, 2007.
- 481 Cole, J. J., Bade, D. L., Bastviken, D., Pace, M. L., and Van de Bogert, M.: Multiple 482 approaches to estimating air-water gas exchange in small lakes, Limnol. Oceanogr. 483 Meth., 8, 285-293, 10.4319/lom.2010.8.285, 2010.
- 484 Crawford, J. T., Striegl, R. G., Wickland, K. P., Dornblaser, M. M., and Stanley, E. H.: 485 Emissions of carbon dioxide and methane from a headwater stream network of interior 486 Alaska, J. Geophys. Res.-Biogeo., 118, 482-494, 10.1002/jgrg.20034, 2013.
- 487 DelSontro, T., Kunz, M. J., Kempter, T., Wüest, A., Wehrli, B., and Senn, D. B.: Spatial 488 Heterogeneity of Methane Ebullition in a Large Tropical Reservoir, Environ. Sci. 489 Technol., 45, 9866-9873, 10.1021/es2005545, 2011.
- 490 Eugster, W., DelSontro, T., and Sobek, S.: Eddy covariance flux measurements confirm 491 extreme CH4 emissions from a Swiss hydropower reservoir and resolve their short-term 492
- variability, Biogeosciences, 8, 2815-2831, 10.5194/bg-8-2815-2011, 2011.

Formatiert: Englisch (USA)

- 493 Gålfalk, M., Bastviken, D., Fredriksson, S., and Arneborg, L.: Determination of the piston 494 velocity for water-air interfaces using flux chambers, acoustic Doppler velocimetry, and 495 IR imaging of the water surface, J. Geophys. Res.-Biogeo., 118, 770-782, 496 10.1002/jgrg.20064, 2013.
- 497 Goldenfum, J. A.: GHG Measurement Guidelines for Freshwater Reservoirs, UNESCO/IHA, 498 London, 139 pp., 2011.
- 499 Halbedel, S., and Koschorreck, M.: Regulation of CO2 emissions from temperate streams and 500 reservoirs, Biogeosciences, 10, 7539-7551, 10.5194/bg-10-7539-2013, 2013.
- 501 Hotchkiss, E. R., Hall Jr, R. O., Sponseller, R. A., Butman, D., Klaminder, J., Laudon, H., 502 Rosvall, M., and Karlsson, J.: Sources of and processes controlling CO2 emissions 503 change with the size of streams and rivers, Nature Geosci., 8, 696-699, 504 10.1038/ngeo2507, 2015.
- 505 Jähne, B., and Haußecker, H.: Air-water gas exchange, Ann. Rev. Fluid Mech., 30, 443-468, 506 1998.
- 507 Koprivnjak, J. F., Dillon, P. J., and Molot, L. A.: Importance of CO<sub>2</sub> evasion from small 508 boreal streams, Global Biogeochem. Cycles, 24, Gb4003, 10.1029/2009gb003723, 2010.
- 509 Kremer, J. N., Nixon, S. W., Buckley, B., and Roques, P.: Technical Note: Conditions for 510 Using the Floating Chamber Method to Estimate Air-Water Gas Exchange, Estuaries, 26, 511 985-990, 10.1007/BF02803357, 2003.
- 512 Le Quéré, C., Peters, G. P., Andres, R. J., Andrew, R. M., Boden, T. A., Ciais, P., 513 514 Friedlingstein, P., Houghton, R. A., Marland, G., Moriarty, R., Sitch, S., Tans, P., Arneth, A., Arvanitis, A., Bakker, D. C. E., Bopp, L., Canadell, J. G., Chini, L. P., Doney, S. C.,
- 515 Harper, A., Harris, I., House, J. I., Jain, A. K., Jones, S. D., Kato, E., Keeling, R. F.,
- 516 Klein Goldewijk, K., Körtzinger, A., Koven, C., Lefèvre, N., Maignan, F., Omar, A.,
- 517 Ono, T., Park, G. H., Pfeil, B., Poulter, B., Raupach, M. R., Regnier, P., Rödenbeck, C.,
- 518 Saito, S., Schwinger, J., Segschneider, J., Stocker, B. D., Takahashi, T., Tilbrook, B., van 519 Heuven, S., Viovy, N., Wanninkhof, R., Wiltshire, A., and Zaehle, S.: Global carbon
  - budget 2013, Earth Syst. Sci. Data, 6, 235-263, 10.5194/essd-6-235-2014, 2014.
- 520 521 Maeck, A., DelSontro, T., McGinnis, D. F., Fischer, H., Flury, S., Schmidt, M., Fietzek, P., 522 and Lorke, A.: Sediment trapping by dams creates methane emission hotspots, Environ. 523 Sci. Technol., 47, 8130-8137, 2013.
- 524 Marion, A., Nikora, V., Puijalon, S., Bouma, T., Koll, K., Ballio, F., Tait, S., Zaramella, M., 525 Sukhodolov, A., O'Hare, M., Wharton, G., Aberle, J., Tregnaghi, M., Davies, P., Nepf, 526 H., Parker, G., and Statzner, B.: Aquatic interfaces: a hydrodynamic and ecological 527 perspective, J. Hydraul. Res., 1-15, 10.1080/00221686.2014.968887, 2014.
- 528 McMahon, P. B., and Dennehy, K. F.: N<sub>2</sub>O emissions from a nitrogen-enriched river, 529 Environ. Sci. Technol., 33, 21-25, 10.1021/es980645n, 1999.
- 530 Moog, D., and Jirka, G.: Stream Reaeration in Nonuniform Flow: Macroroughness 531 Enhancement, J. Hydraul. Eng., 125, 11-16, doi:10.1061/(ASCE)0733-9429(1999)125:1(11), 1999. 532
- 533 Raymond, P. A., and Cole, J. J.: Gas exchange in rivers and estuaries: Choosing a gas transfer 534 velocity, Estuaries, 24, 312-317, 10.2307/1352954, 2001.
- 535 Raymond, P. A., Zappa, C. J., Butman, D., Bott, T. L., Potter, J., Mulholland, P., Laursen, A. 536 E., McDowell, W. H., and Newbold, D.: Scaling the gas transfer velocity and hydraulic 537 geometry in streams and small rivers, Limnology & Oceanography: Fluids & 538 Environments, 2, 41-53, 10.1215/21573689-1597669, 2012.
- 539 Raymond, P. A., Hartmann, J., Lauerwald, R., Sobek, S., McDonald, C., Hoover, M.,
- 540 Butman, D., Striegl, R., Mayorga, E., Humborg, C., Kortelainen, P., Durr, H., Meybeck,
- 541 M., Ciais, P., and Guth, P.: Global carbon dioxide emissions from inland waters, Nature, 542 503, 355-359, 10.1038/nature12760, 2013.

- Sand-Jensen, K., and Staehr, P.: CO<sub>2</sub> dynamics along Danish lowland streams: water–air
  gradients, piston velocities and evasion rates, Biogeochemistry, 111, 615-628,
  10.1007/s10533-011-9696-6, 2012.
- Teodoru, C. R., Nyoni, F. C., Borges, A. V., Darchambeau, F., Nyambe, I., and Bouillon, S.:
  Dynamics of greenhouse gases (CO<sub>2</sub>, CH<sub>4</sub>, N<sub>2</sub>O) along the Zambezi River and major
  tributaries, and their importance in the riverine carbon budget, Biogeosciences, 12, 24312453, 10.5194/bg-12-2431-2015, 2015.
- Tranvik, L. J., Downing, J. A., Cotner, J. B., Loiselle, S. A., Striegl, R. G., Ballatore, T. J.,
  Dillon, P., Finlay, K., Fortino, K., Knoll, L. B., Kortelainen, P. L., Kutser, T., Larsen, S.,
  Laurion, I., Leech, D. M., McCallister, S. L., McKnight, D. M., Melack, J. M., Overholt,
  E., Porter, J. A., Prairie, Y., Renwick, W. H., Roland, F., Sherman, B. S., Schindler, D.
  W., Sobek, S., Tremblay, A., Vanni, M. J., Verschoor, A. M., von Wachenfeldt, E., and
  Weyhenmeyer, G. A.: Lakes and reservoirs as regulators of carbon cycling and climate,
  Limnol. Oceanogr., 54, 2298-2314, 10.4319/lo.2009.54.6\_part\_2.2298, 2009.
- Limnol. Oceanogr., 54, 2298-2314, 10.4319/lo.2009.54.6\_part\_2.2298, 2009.
  Vachon, D., Prairie, Y. T., and Cole, J. J.: The relationship between near-surface turbulence and gas transfer velocity in freshwater systems and its implications for floating chamber measurements of gas exchange, Limnol. Oceanogr., 55, 1723-1732, 10.4319/lo.2010.55.4.1723, 2010.
- Wallin, M. B., Oquist, M. G., Buffam, I., Billett, M. F., Nisell, J., and Bishop, K. H.:
  Spatiotemporal variability of the gas transfer coefficient (*K*(CO<sub>2</sub>)) in boreal streams:
  Implications for large scale estimates of CO<sub>2</sub> evasion, Global Biogeochem. Cycles, 25,
  Gb3025, 10.1029/2010gb003975, 2011.
- 565