Dear Editor,

We are grateful for the constructive comments on our manuscript from the two referees. Please find below our point by point responses to each referee comment and suggestion, as well as a revised version of our manuscript 'No observed effect of ocean acidification on nitrogen biogeochemistry in a summer Baltic Sea plankton community'.

As suggested, the figures have been revised and condensed, and material and figures regarding the addition of the  $^{15}$ N-N<sub>2</sub> tracer has been shifted to the supplementary materials. These changes and others made to the manuscript are in blue and red text below. We are currently formatting the data files to be uploaded to the PANGAEA database. These files will then be linked to this manuscript and others in the Special Issue.

We thank you for the opportunity to submit a revised manuscript for consideration in Biogeosciences and look forward to hearing a response on the manuscript soon.

Yours Sincerely,

Allanah Paul Corresponding author, on behalf of all authors

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#### Response to Reviewer #1 (D. Campbell)

We thank D. Campbell for his useful and constructive comments on this manuscript which helped to clarify a number of points and tune the figures in the manuscript. Our responses to reviewer comments, including modifications to the manuscript, are detailed in the following:

**Comment 1 by D. Campbell:** The mesocosms were closed at the bottom. Would this alter their response by cutting off upwelling supplies of  $NH_4^+$ ?

**Author response:** As the reviewer correctly points out, once the mesocosm bags were pulled up above the water surface during closure on *t-5*, the mesocosms were then closed water masses with no exchange with the outside environment, only with the atmosphere. Therefore, there was no supply of nutrients through upwelling inside the mesocosms. Sporadic, wind-driven upwelling events are known to stimulate blooms of N<sub>2</sub>-fixing filamentous cyanobacteria in the Baltic Sea in summer (Nausch et al., 2009; Wasmund et al., 2012) by bringing up phosphate rich water (Nausch et al., 2007). We observed this phenomenon outside the mesocosms during Phase II (*t17* to *t30*). Hence, it is likely that the response of the plankton community would have been altered by the addition of nutrients (either ammonium/NH<sub>4</sub><sup>+</sup> or phosphate). However, here we were primarily interested in the response of the plankton assemblage and N cycle to  $CO_2$  in a low nutrient, closed system.

**Comment 2 by D. Campbell:** Abstract: "(average treatment  $fCO_2$ : 365–1231 µatm)" This statement needs to be clarified; I think: (average treatments  $fCO_2$ : 365, 1232 µatm). Line 12 in the Materials & Methods has a different range of  $fCO_2$ .

**Author response:** The two reported ranges of  $CO_2$  are different as the one in the Abstract refers to the average  $fCO_2$  over the study period, whereas in the Materials and Methods, the  $fCO_2$  range refers to the initial treatments present on t4 after  $fCO_2$  was manipulated (p. 17511, line 11-12: 'Initial  $fCO_2$  ranged from ~ 240 µatm in the two ambient control mesocosms to up to 1650 µatm (Fig. 1a).'). These are not the same range because  $fCO_2$  was not constant during the study period due to outgassing of  $CO_2$  from the treatment mesocosms (Fig. 1a). A statement, as included in the Fig. 1 caption, will also be added to the text in the revised manuscript to clarify this important distinction to read (p. 17517, line 10): 'These phases are also used to assist with data interpretation in this manuscript. Average  $fCO_2$  was calculated for each mesocosm between t1 and t43.'

#### Comment 3 by D. Campbell: line 22: nor, not or. Nor follows a negative.

**Author response:** We thank the reviewer for pointing this out and this will be changed for p. 17508, line 22 in the revised manuscript.

## **Comment 4 by D. Campbell:** *Materials & Methods: line 26, "KOSMOS,"??? Undefined acronym/abbreviation?*

**Author response:** KOSMOS stands for 'Kiel Off-Shore Mesocosms for future Ocean Simulations'. This definition will be added to p. 17510, line 26 in the revised manuscript.

## **Comment 5 by D. Campbell:** *Table 1: I do not understand why this table is organized into 10 columns. It looks to me like it should be 5 columns, twice as many rows.*

Author response: The format of Table 1 will be changed accordingly in the revised manuscript.

**Comment 6 by D. Campbell:** Figure 1. Would it be worth showing DIC? Is there any change? **Author response:** Changes in DIC are the primary driver in changes in  $fCO_2$ , hence mirror changes in  $fCO_2$  over the study period. Details on the carbonate chemistry are presented in an accompanying paper (Paul et al., 2015).

**Comment 7 by D. Campbell:** Figure 1: insert legends are very small; I cannot read them at printed page size. I am getting weird colour changes (artefacts) in the roman numerals for the experiment stage labelling I, II, III. This is probably a .pdf generation issue, but it is distracting.

**Author response:** The inserted legend will be removed and added to a separate panel in the figure to improve readability in the revised manuscript. Regarding the colour changes for the roman numerals, this artefact appears to disappear if the figures are viewed highly zoomed and in the printed version. The figures will be revised with changes to the legends and we will try to resolve this issue for computer screens.

**Comment 8 by D. Campbell:** Again, the listed  $fCO_2$  levels differ from the materials and methods, and from the abstract.

Author response: Please see Author response to Comment 2 by D. Campbell.

**Comment 9 by D. Campbell:** *Figure 3: It would be good to have the colour/symbol legend for each figure, to stand on its own without reference back to Figure 1.* 

Author response: The colour/symbol legend will be added to all figures in the revised manuscript.

**Comment 10 by D. Campbell:** Figure 6: Data points with uncontaminated gas are below detection, all detected rates are from the contaminated period. Should this data be presented? I am reading page 17519 but am not clear on the origin of the data in Figure 6.

**Author response:** The detection limit was determined to be 0.15 nmol N L<sup>-1</sup> day<sup>-1</sup> and we detected rates above the detection limit in some mesocosms from t3 and in all mesocosms from t11 until t21 (apart from M1 on t15). For clarity we will remove the data points below the detection limit from Figure 5 as done in Figure 6. The data in Figure 6 is to indicate the sudden and large increase in apparent rates which is an artefact and the result of contaminated gas used for labelling. This figure will be removed from the revised manuscript as suggested by Reviewer #2.

**Comment 11 by D. Campbell:** Results: P.17521 The extrapolations in the absence of actual  $N_2$  fixation rates seem reasonable, but are based upon multiple assumptions on  $N_2$  rates, N:P ratios and N exudation rates.

Author response: As correctly highlighted, there are a number of assumptions included in the calculation of N inputs through  $N_2$ -fixation which are acknowledged in the methods section (p. 17520).

**Comment 12 by D. Campbell:** *P.17521, the N contamination issue is serious given the patchy cyanobacterial data.* 

**Author response:** We agree with this statement by D. Campbell and have ensured that this is transparently presented in the manuscript.

**Comment 13 by D. Campbell:** Discussion: "In fact, nitrate concentrations continually increased throughout the experiment at an average 10 net rate of 1 nmol N  $L^{-1}$  day<sup>-1</sup> (Fig. 1c) .... "

Summary: "Thus N uptake rates were well balanced with supply or any net differences were too small to be detected in N pool sizes across the range of simulated ocean acidification scenarios" These statements appear contradictory. I think the summary needs to be clarified that  $fCO_2$  did not provoke changes in N pool sizes?

**Author response:** We thank D. Campbell for highlighting an apparent inconsistency between these two statements. **'CO<sub>2</sub>-related** ' will be added to the Summary in the revised manuscript to clarify this point and will then read: "Thus N uptake rates were well balanced with supply or any net **CO<sub>2</sub>-related** differences were too small to be detected in N pool sizes across the range of simulated ocean acidification scenarios."

#### **References:**

Nausch, G., Feistel, R., Lass, H. U., Nagel, K., and Siegel, H.: Hydrographisch-chemische Zustandseinschätzung der Ostsee 2006, Marine Science Reports, Warnemünde, 70, 1 – 91, 2007.

Nausch, M., Nausch, G., Lass, H. U., Mohrholz, V., Nagel, K., Siegel, H., and Wasmund, N.: Phosphorus input by upwelling in the eastern Gotland Basin (Baltic Sea) in summer and its effects on filamentous cyanobacteria, Estuarine, Coastal and Shelf Science, 83 (4), 434 – 442, doi:10.1016/j.ecss.2009.04.031, 2009.

Paul, A. J., Bach, L. T., Schulz, K.-G., Boxhammer, T., Czerny, J., Achterberg, E. P., Hellemann, D., Trense, Y., Nausch, M., Sswat, M., and Riebesell, U.: Effect of elevated  $CO_2$  on organic matter pools and fluxes in a summer Baltic Sea plankton community, Biogeosciences, 12, 6818 – 6203, doi:10.5194/bg-12-6181-2015, 2015.

Wasmund, N., Nausch, G., and Voss, M.: Upwelling events may cause cyanobacteria blooms in the Baltic Sea, Journal of Marine Systems, 90 (1), 67 – 76, doi:10.1016/j.jmarsys.2011.09.001, 2012.

#### Response to Reviewer #2 (Anonymous Reviewer)

We thank the reviewer for the constructive comments on this manuscript. We have taken them on board and our responses to reviewer comments, including potential modifications to the manuscript, are detailed in the following:

**Comment 1 by Reviewer #2:** Overall the results show little significant effect of  $CO_2$  except for phosphate availability, the implications of which should be considered more in the Discussion and Summary.

**Author response:** We acknowledge that  $CO_2$  seemed to have a minimal effect on the nitrogen cycling as indicated in the title of this manuscript, whereas phosphate availability seemed to be more affected. Please see an accompanying manuscript (Nausch et al., 2015) for coverage and in-depth discussion of the phosphorus pools and cycle, which is now available in Biogeosciences Discussion. We will update this reference.

**Comment 2 by Reviewer #2:** Despite an interesting and novel approach to measuring nitrogen fixation in the latter part of the experiment, this was unfortunately confounded by contaminated isotopically-labelled dinitrogen. Frustrating as it is, particularly after what must have been a lot of hard work, the methodology and post-t21 results and interpretation on nitrogen fixation should be removed, as this contamination renders them unusable and confusing.

**Author response:** In accordance with the suggestions by Reviewer #2, we will remove Figure 2 and Figure 6 from the manuscript, along with the reference and discussion of these in the text body, and will put them as supplementary materials in the revised manuscript. Interpretation of  $N_2$ -fixation rates is only made up until *t21* where the data is not contaminated and afterwards any indication is from either P\*, N pool sizes or abundance of *A. flos-aquae*. While these data is not conclusive on the effect of ocean acidification on  $N_2$ -fixation for this study, these are indicators which we feel are reasonable to present and discuss in this manuscript in relation to  $N_2$ -fixation for the period where the estimated rates from incubations were affected by the contamination issue.

**Comment 3 by Reviewer #2:** Also, the reasons for the low nitrogen fixation rate and biomass of Aphanizomenon flos-aquae in the mesocosms relative to the surrounding water remain unclear. This may reflect an artefact of the mesocosms, or the possibility that, as the mesocosm water was initially filtered at 50um this may have removed some of the Aphanizomenon flos-aquae, particularly the colonial forms. This could then explain the observed low densities and nitrogen fixation rates relative to water outside the mesocosms.

Author response: Please refer to Author response to Comment 4 by Reviewer #2 below.

#### METHODS

**Comment 4 by Reviewer #2:** Filtration at 50  $\mu$ m may have excluded the large, colonial nitrogenfixers. Aphanizomenon flos-aquae is a reasonably large filamentous cyanobacterium, particularly when in colonial form, and the low biomass in the mesocosms, below that of the surrounding water, may reflect removal of a proportion of the A. flos-aquae biomass during mesocosm filling.

Author response: Only the water used for the  $CO_2$  enrichment was filtered at 50  $\mu$ m, whereas a net of 3 mm mesh size covered both ends of the mesocosm bag as the bottom end was lowered and the upper end pulled above the water surface during mesocosm closure. These details are described in

depth in Paul et al. 2015. While we cannot exclude the possibility that this 3 mm net may have removed some of the *A. flos-aquae* biomass during the lowering of the mesocosm bags, this was done only at the beginning of the experiment when there were no filamentous cyanobacteria aggregates visible in the Tvärminne Störfjärden. It is more likely that the upwelling of phosphate-rich water outside the mesocosm around *t17* encouraged growth of *A. flos-aquae*, rather than the removal of *A. flos-aquae* during mesocosm closure. This stimulation could not happen inside the mesocosms as they were tightly sealed and no phosphate-rich water was introduced and no nutrients were added.

## **Comment 5 by Reviewer #2:** The text should clarify that the nitrogen fixation techniques were modified from that of Mohr et al (2010).

**Author response:** We will change p. 17513 (line 14-15) in the revised manuscript to read: 'Incubations for determination of  $N_2$ -fixation rates were carried out using an approach described by Mohr et al. (2010), with some modifications for the preparation of the <sup>15</sup>N-N<sub>2</sub> enriched seawater (see Section 2.3 for details).'

**Comment 6 by Reviewer #2:** The replacement of 70-90ml of water with degassed water and, to a lesser extent, the sampling & transfer of water samples, would have reduced the  $CO_2$  content and raised pH of the incubation samples. Was pH measured before or after the nitrogen fixation incubations?

**Author response:** This is an important point to consider when using this degassing method in ocean acidification studies. While pH was not measured in individual incubation bottles, the effect of degassing on the carbonate system was investigated during a different experiment also in the Baltic Sea (A. Paul, unpublished data). Dissolved organic carbon (DIC) concentrations were determined in samples before and after degassing. The reduction in DIC in water from the south-western Baltic Sea (S = 15.3, T = 22.5°C, TA = 1965.5  $\mu$ mol kg<sup>-1</sup>) by the degassing system was on the order of 100  $\mu$ mol kg<sup>-1</sup>, less than 10% of total DIC (~1800  $\mu$ mol kg<sup>-1</sup>), using the same method as in the study presented here. Note that a water sample can be stripped of many dissolved gases relatively quickly. For CO<sub>2</sub> however, it will take considerably longer as most DIC is present in the form of bicarbonate and carbonate.

In the incubations reported in this manuscript (northern, central Baltic Sea, S = 5.7,  $T \sim 8.0-15.9^{\circ}C$ , TA ~1520 µmol kg<sup>-1</sup>), about 70 mL of this degassed water with reduced DIC was added. Assuming a similar amount of DIC was removed (100 µmol kg<sup>-1</sup>), this corresponds to a calculated decrease in DIC of ~3 µmol kg<sup>-1</sup> in each incubation bottle. This is insignificant for  $fCO_2$  levels considering the range applied in our study. Due to the lower alkalinity present in this study, the amount of DIC removed through the degassing procedure may lead to decreased DIC in the incubation bottles of slightly more than 3 µmol kg<sup>-1</sup>. However this would still be on the same order of magnitude as for the other study from the Baltic Sea, hence we do not consider that this would have a substantial effect on pH or  $CO_2$  content these incubations.

**Comment 7 by Reviewer #2:** Figure 2 shows the  ${}^{15}N-N_2$  enriched seawater entering the overflow system & degassing. Should the  ${}^{15}N-N_2$  supply line connect to the airstone in the overflow system, rather than the  ${}^{15}N-N_2$  enriched seawater?

**Author response:** Figure 2 is correctly shown but will be moved in the revised manuscript to the supplementary materials in accordance with suggestions by Reviewer #2 (see also Author response to Comment 2 by Reviewer #2).

**Comment 9 by Reviewer #2:** What was the final atom%  $^{15}$ N-N<sub>2</sub> in the mesocosms following addition of isotopically labelled N<sub>2</sub> at t21?

**Author response:** Peak enrichment of <sup>15</sup>N in N<sub>2</sub> after  $2^{nd}$  addition of the isotopically labelled N<sub>2</sub> on *t27* ranged between 0.53 and 0.57 atom%.

#### RESULTS

**Comment 10 by Reviewer #2:** In Fig 1f both the key and the ammonium data are too small to read. As the key is important it should be larger, and ideally replicated on the other timeline figures. **Author response:** The inserted legend will be removed and added to a separate panel in the figure to improve readability in the revised manuscript. The colour/symbol legend will be added to all figures

in the revised manuscript.

**Comment 11 by Reviewer #2:** In Fig 3a the increase in P\* in Phase II occurs only in some of the treatments at the onset of Phase II, and otherwise Phase II is dominated by uniform concentrations, so the description is incorrect. The increase in P\* in Phase III is similar to Phase II, if not more significant.

**Author response:** The statement in the manuscript refers to inorganic phosphate concentrations shown in Fig. 1d. We agree that this point was not clearly explained and thank Reviewer #2 for bringing this up. This variation may also be partly masked by the choice of scale so that the Baltic Sea data could also be shown. To clarify this point, this statement will be rephrased in the revised manuscript and will read (p.17517, line 22-23): 'Inorganic phosphate concentrations decreased during Phase I, followed by an increase at the beginning of Phase II. Concentrations in the ambient/control treatments remained higher than in the higher CO<sub>2</sub> treatments in Phase III.'

**Comment 12 by Reviewer #2:** "Nitrate concentrations increased throughout the experiment with a possible small drawdown after t39 in all treatments" – this drawdown is not really evident in Fig 3c. **Author response:** This will be removed in a revised version of this manuscript to read (p. 17517, line 23-25): 'Nitrate concentrations increased slightly throughout the experiment, whereas  $NH_4^+$  concentrations were variable'.

#### **Comment 13 by Reviewer #2:** What is the source of the spikes in nitrate concentration?

**Author response:** This is likely related to the low concentrations observed. The nitrate concentrations were typically in the nanomolar range and therefore sampling and sample handling in the mesocosm environment is challenging. It is therefore likely that this variable is prone to unintended contamination during sampling and sample handling.

**Comment 14 by Reviewer #2:** Perhaps combine Figs 1 and 3 to allow comparison, & also to reflect the text in the Results section. The rainfall data is not required as there was no relationship with measured variables.

**Author response:** As suggested, we will remove the panel with the rainfall data and instead combine the two panels from Figure 3.

**Comment 15 by Reviewer #2:** "BSi in Phase II where a positive effect was detected (p = 0.034)" – why not include this in Table 1?

**Author response:** This data is presented in an accompanying paper (Paul et al., 2015) as cited on p. 17518 (line 18) and was thus not included in the table. Only new data and results of new statistical analyses presented here were included in Table 1. In the revised manuscript, these data (as well as PON, see also Comment 21 by Reviewer #2) will be included in Table 1. An asterisk (\*) will be added to these data points to indicate the statistical analyses and data originate from Paul et al. 2015.

**Comment 16 by Reviewer #2:** "A. flos-aquae abundances ....... were highest in Phases II/III and lowest in Phase I" – sporadic spikes in certain treatments were higher in Phase III than Phase I, but overall Fig 5d shows similar A. flos-aquae abundances in Phase I & III.

**Author response**: This statement was made based on the abundances of *A. flos-aquae* determined by microscopy counts (Fig. 5a). Nonetheless it is correct that the temporal variation in abundances of *A. flos-aquae* and the Aphanizophyll marker pigment concentrations do not fit exactly together. This may be in part influenced by missing data points in Aphanizophyll concentrations on *t35* and *t39* where the microscopy counts indicate peaks in *A. flos-aquae* biomass. In a revised figure, we will remove the connecting lines in the Aphanizophyll marker pigment panel (1c) so this issue becomes clearer.

**Comment 17 by Reviewer #2:** Although the  ${}^{15}N-N_2$  addition to the mesocosms (from t21) is interesting, the authors identify that these results are unusable due to gas contamination, and so the methodology and results (including Figure 6) should be omitted from the paper, as they do not assist the discussion and interpretation.

**Author response:** Figure 6 will be moved to the supplementary materials in the revised manuscript. Please also see Author response to Comment 2 by Reviewer #2.

**Comment 18 by Reviewer #2:** "This indicated potential input of atmospheric N with a low  $d^{15}N$  into particulate matter via  $N_2$ -fixation...". The authors should consider that this could alternatively reflect the uptake of ammonium depleted in <sup>15</sup>N produced during ammonification.

Author response: Ammonification is a process which likely occurred in this study, although this was not quantified. Thus, this may have influenced the  $\delta^{15}N$  signal in particulate matter reported here through production of ammonium depleted in  $^{15}N$  and consequent assimilation into the PON pool. We thank Reviewer #2 for commenting on this and will add this point to the revised manuscript to p. 17520, line 5-6 to read: 'This indicated either potential input of atmospheric N with a low  $\delta^{15}N$  into particulate matter via N<sub>2</sub>-fixation during this period or potential uptake of ammonium with a  $\delta^{15}N$  signature depleted through ammonification.

**Comment 19 by Reviewer #2:** "This was one day after the mesocosm walls were cleaned indicating that there were likely diazotrophic species and diatoms attached to the mesocosm walls". Was this the only time the mesocosm walls were cleaned (in Paul et al, 2015, it mentions "Mesocosm bags were cleaned occasionally inside and outside throughout the experiment"), If not, were other trap samples affected on other days? Should the Aphanizophyll spike on t15 be regarded as an artefact? Author response: The mesocosm bags were cleaned for the first time on t16 with cleaning occurring more regularly after this (t22/23, t29, t36 and t42) as detailed in Paul et al. 2015 (refer to Figure 3).

Hence the effect of wall-growth on the material collected in the sediment traps was probably more important on t17 (~2.5 weeks of growth) than on the other sampling days after cleaning.

**Comment 20 by Reviewer #2:** *"The assessment for between t23 and t43 is based on the premise of continued elevated ...... – why not just do this comparison up to t21 to remove any uncertainty?* **Author response:** As suggested, we calculated the N input up until *t21*. Using the same method as for between *t23* and *t43*, a mean N<sub>2</sub>-fixation rate of 0.57 nmol N L<sup>-1</sup> day<sup>-1</sup> over 24 days (*t-3 – t21*) and an assumed 50% exudation of DON/NH<sub>4</sub><sup>+</sup>, leads to a calculated N input of 27.55 nmol N. This is a little higher than the estimate from isotope uptake incubations of 20 nmol N. Nonetheless we believe that this is still in reasonable agreement with the original estimation, particularly considering the variability between mesocosms in both N<sub>2</sub>-fixation rates and *Aphanizomenon flos-aquae* abundances.

#### DISCUSSION

**Comment 21 by Reviewer #2:** "The only statistically significant, but very minor, correlation was a positive relationship between  $CO_2$  and PON concentrations" – why not include this in Table 1? **Author response:** This data will be included in Table 1 in the revised manuscript. Please also see Author response to Comment 15 by Reviewer #2 above.

**Comment 22 by Reviewer #2:** "This is due to the rather low A. flos-aquae biomass ....." – might this reflect the 50  $\mu$ m filtration when filling the mesocosms?

**Author response:** A net with a 3 mm mesh size was used to exclude larger particles and organisms during mesocosm closure, whereas the 50  $\mu$ m gauze was used only for the water for CO<sub>2</sub>- enrichment. Please also see Author response to Comment 4 by Reviewer #2 above.

**Comment 23 by Reviewer #2:** "Diazotrophic organisms typically have slower growth rates than other organisms. Hence any potential influence of ocean acidification on their physiology may take longer to become apparent in biogeochemical parameters sampled in larger-scale field studies." As growth rates will be the same in the field and the lab, the difference in the response of nitrogen fixation to  $CO_2$  from reported lab experiments results more likely reflects ecosystem interactions (grazing, competition, nutrient availability) in field studies.

**Author response:** It is difficult to say that the growth rates would be the same as in the lab as it is difficult to exactly replicate macro/micro-nutrient availability and light conditions. However we agree that these ecosystem interactions will likely modulate the physiological response to  $CO_2$  observed in laboratory studies. To reflect this, we will modify p. 17523 (lines 10-12) in the revised manuscript to read: 'Hence any potential influence of ocean acidification on their physiology may take longer to become apparent in biogeochemical parameters sampled in larger-scale field studies where most sampled parameters such as PON are a mixture of organic compounds of various origin and isotopic composition. In addition, the overall response to  $CO_2$  observed in laboratory experiments, with trophic interactions such as grazing and competition between species for nutrients and light.'

**Comment 24 by Reviewer #2:** "Hence natural exposure to highly variable carbonate chemistry conditions..." this is an interesting idea, but does not explain why most of the papers reporting  $CO_2$  enhancement of marine N fixation showed it in Trichodesmium (see Hutchins et al papers) which would experience similar highly variable conditions. This caveat should be mentioned.

Author response: We thank the reviewer for pointing out this interesting caveat. We would like to reiterate that the filamentous, heterocystous cyanobacteria species found in the Baltic Sea such as A. flos-aquae have generally shown the opposing physiological response to tropical/subtropical species such as *Trichodesmium* sp. under high  $CO_2$  (as summarised in the Introduction, p17510, lines 1 – 11), despite both being aggregate forming species. Currently, to the best of our knowledge, there is no clear explanation for this inconsistent response to elevated CO<sub>2</sub> between cyanobacteria morphologies or species in physiological single-strain culture studies although a variety of hypotheses exist. In addition, we are not aware of any published study on the microenvironments in Trichodesmium aggregates which could be used as a comparison. Hence, to avoid any potential for confusion here and refocus the discussion on the A. flos-aquae, for which there is data on aggregate microenvironments, we will remove the reference to *Trichodesmium* sp. in the revised manuscript on p.17524 (line 7-14) to read: 'In addition, filamentous cyanobacteria in the Baltic Sea form characteristic surface aggregations. Inside aggregations of A. flos-aquae, microenvironments can create substantially different conditions compared to the surrounding water with large diurnal fluctuations in pH (7.4 vs. 9.0) and O<sub>2</sub> concentrations (~ 150–450  $\mu$ mol O<sub>2</sub> L<sup>-1</sup>) and thus also inorganic carbon availability (Ploug, 2008). '

## **Comment 25 by Reviewer #2:** 4.2 What is the explanation for the coincident increases in PON and nitrate from Phase II to Phase III?

**Author response:** Perhaps the reviewer means DON here, rather than PON? We do not have a clear reason or mechanism as to why either variable (DON or nitrate) increased over time as no rate measurements were made. We can only speculate that some DON was released through 'sloppy feeding' by zooplankton who were abundant at the time (see Lischka et al., 2015) or nitrate was produced by nitrification in the water column (0 - 17m). We consider it unlikely that these increases are directly related to each other as the increase in nitrate (in the range of tens of nanomol per liter) is an order of magnitude smaller than the apparent increase in DON (in the range of a few hundred nanomol per liter).

#### SUMMARY

**Comment 26 by Reviewer #2:** Bearing in mind the only significant correlation with elevated  $CO_2$  was a reduction in phosphate, the Summary should consider the implications of this for future nutrient budgets and productivity in the Baltic Sea.

**Author response:** As mentioned in Author response to Comment 1 from Reviewer #2, please note that discussion of the phosphorus pools and cycle are discussed in an accompanying paper (Nausch et al., 2015). We also consider that any speculation here, with reference to nutrient budgets and diazotrophic cyanobacteria in the future, is not justified considering the lack of response observed.

#### **References:**

Lischka, S., Bach, L. T., Schulz, K.-G., and Riebesell, U.: Micro- and mesozooplankton community response to increasing  $CO_2$  levels in the Baltic Sea: insights from a large-scale mesocosm experiment, Biogeosciences Discussions, 12, 20025 – 20070, doi:10.5194/bgd-12-20025-2015, 2015.

Nausch, M., Bach, L., Czerny, J., Goldstein, J., Grossart, H. P., Hellemann, D., Hornick, T., Achterberg, E., Schulz, K., and Riebesell, U.: Effects of  $CO_2$  perturbation on phosphorus pool sizes and uptake in a mesocosm experiment during a low productive summer season in the northern Baltic Sea, Biogeosciences Discussions, 12, 17543 – 17593, doi:10.5194/bgd-12-17543-2015, 2015, 2015.

Paul, A. J., Bach, L. T., Schulz, K.-G., Boxhammer, T., Czerny, J., Achterberg, E. P., Hellemann, D., Trense, Y., Nausch, M., Sswat, M., and Riebesell, U.: Effect of elevated  $CO_2$  on organic matter pools and fluxes in a summer Baltic Sea plankton community, Biogeosciences, 12, 6818 – 6203, doi:10.5194/bg-12-6181-2015, 2015.

# No observed effect of ocean acidification on nitrogen biogeochemistry in a summer Baltic Sea plankton community

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13

#### 14 Abstract

15 Nitrogen fixation by filamentous cyanobacteria supplies significant amounts of new nitrogen (N) to the Baltic Sea. This balances N loss processes such as denitrification and anammox and 16 17 forms an important N source supporting primary and secondary production in N-limited postspring bloom plankton communities. Laboratory studies suggest that filamentous diazotrophic 18 19 cyanobacteria growth and N<sub>2</sub>-fixation rates are sensitive to ocean acidification with potential 20 implications for new N supply to the Baltic Sea. In this study, our aim was to assess the effect 21 of ocean acidification on diazotroph growth and activity as well as the contribution of 22 diazotrophically-fixed N to N supply in a natural plankton assemblage. We enclosed a natural 23 plankton community in a summer season in the Baltic Sea near the entrance to the Gulf of Finland in six large-scale mesocosms (volume  $\sim 55 \text{ m}^3$ ) and manipulated fCO<sub>2</sub> over a range 24 relevant for projected ocean acidification by the end of this century (average treatment  $fCO_2$ : 25  $365 - 1231 \mu$ atm). The direct response of diazotroph growth and activity was followed in the 26 27 mesocosms over a 47 day study period during N-limited growth in the summer plankton community. Diazotrophic filamentous cyanobacteria abundance throughout the study period 28 29 and N<sub>2</sub>-fixation rates (determined only until day 21 due to subsequent use of contaminated commercial <sup>15</sup>N-N<sub>2</sub> gas stocks) remained low. Thus estimated new N inputs from diazotrophy 30 were too low to relieve N limitation and stimulate a summer phytoplankton bloom. Instead 31

regeneration of organic N sources likely sustained growth in the plankton community. We
could not detect significant CO<sub>2</sub>-related differences in inorganic <u>n</u>or organic N pools sizes, or
particulate matter N:P stoichiometry. Additionally, no significant effect of elevated CO<sub>2</sub> on
diazotroph activity was observed. Therefore, ocean acidification had no observable impact on
N cycling or biogeochemistry in this N-limited, post-spring bloom plankton assemblage in the
Baltic Sea.

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#### 8 1 Introduction

9 Nitrogen (N) is an essential element for cell functioning in the biosphere due to its presence in 10 many important biomolecules such as nucleic acids and proteins. However, in many marine 11 ecosystems N is considered the limiting nutrient for important cellular processes in 12 phytoplankton (Vitousek and Howarth, 1991), as indicated through stimulation carbon 13 fixation and pigment synthesis through addition of inorganic N (e.g. Moore et al. (2008, 14 2013)). This low N availability also prevails in post-spring bloom plankton communities in the Baltic Sea, as the nitrate pool is exhausted during the spring-bloom leaving behind an 15 excess of dissolved inorganic phosphorus (Wasmund et al., 2001). Consequently, filamentous 16 diazotrophic (N<sub>2</sub>-fixing) cyanobacteria, in particular heterocystous Nodularia spumigena and 17 Aphanizomenon flos-aquae, capitalise on this excess phosphate and increasing water column 18 temperatures in summer months (Kononen et al., 1996; Pliński and Jóźwiak, 1999; Wasmund, 19 20 1997) and commonly form extensive blooms and surface aggregations (e.g. Kahru and Elmgren, 2014). The atmospheric nitrogen gas  $(N_2)$  fixed by these heterocystous 21 22 cyanobacteria during the summer months forms a key N source for the wider plankton community in the Baltic Sea, since a significant fraction of the fixed N can be released as 23 24 ammonium (Ohlendieck et al., 2000; Ploug et al., 2010; Stal et al., 2003; Wannicke et al., 25 2013) and dissolved organic N compounds (Ohlendieck et al., 2000, 2007; Wannicke et al., 26 2013). Thus in addition to N in diazotroph biomass, newly fixed N is also available for direct 27 assimilation by phytoplankton and bacteria and is estimated to support up to 20 - 45% of 28 annual primary production in the Baltic Sea (Gustafsson et al., 2013). This new N input partly 29 replenishes N loss processes such as anammox and denitrification in the deep anoxic basins (Vahtera et al., 2007). Furthermore, this fixed N can also be directly transferred to higher 30 31 trophic levels through grazing by zooplankton (Engström-Öst et al., 2011; Hogfors et al., 32 2014; Wannicke et al., 2013).

Changes in seawater carbonate chemistry due to increased atmospheric CO<sub>2</sub> concentrations 1 2 are expected to induce changes in phytoplankton physiology. The associated decrease in 3 seawater pH is called ocean acidification. Numerous single-strain culture studies have 4 investigated the physiological responses of a variety of diazotrophic organisms and generally 5 indicated increased N<sub>2</sub>-fixation and diazotroph growth rates under elevated CO<sub>2</sub> (Barcelos e Ramos et al., 2007; Fu et al., 2008; Hutchins et al., 2007; Kranz et al., 2010; Levitan et al., 6 7 2007), with contrasting evidence under iron limitation (Shi et al., 2012) and with freshwater strains of A. flos-aquae (Yamamoto and Nakahara, 2005). Three studies on the common 8 9 Baltic Sea species, N. spumigena, produced contrasting results with two studies under phosphate repletion suggesting a negative effect (Czerny et al., 2009; Eichner et al., 2014), 10 11 and one study, under low inorganic phosphate availability, indicating a positive effect (Wannicke et al., 2012) of increased CO<sub>2</sub> on growth and N<sub>2</sub>-fixation rates. This discrepancy 12 13 may, however, be due to differences in phosphate availability (Eichner et al., 2014). Considering the contribution of diazotrophs to the N budget and primary productivity in the 14 15 Baltic Sea, it is vital to understand the influence of future changes in  $pCO_2$  on new N inputs by diazotrophs. 16

17 In this mesocosm study, our aim was to assess diazotroph growth and rates of  $N_2$ -fixation 18 under a range of  $CO_2$  concentrations in a natural plankton community. N limitation of 19 phytoplankton growth was reported in the study area in the Finland Archipelago Sea 20 (Kirkkala et al., 1997; Tamminen and Andersen, 2007). By utilizing the naturally occurring 21 low N conditions in the Baltic Sea we wanted to examine the importance of new N inputs by 22 diazotrophic organisms to the wider plankton community N supply under projected future 23 ocean acidification scenarios.

24

#### 25 2 Materials and methods

#### 26 2.1 Experimental set-up and sampling

The study took place in the period between June and August 2012 in Tvärminne Storfjärden which is situated in the Archipelago Sea on the southwestern tip of Finland. Six pelagic mesocosms (total volume ~55m<sup>3</sup>, <u>Kiel Off-Shore Mesocosms for future Ocean Simulations -</u> KOSMOS, Riebesell et al. (2013)) were deployed on 12 June 2012 (day of experiment -10 = t-10, i.e. 10 days before CO<sub>2</sub> manipulation) and moored at 59° 51.5' N, 23° 15.5' E. The cylindrical mesocosm bags of 2 m in diameter extended from 1.5 m above to 19 m below the water surface and were closed at the bottom by a 2 m long sediment trap funnel on *t-5*. A 3
 mm net was used to exclude larger organisms or particles before mesocosm closure.

3 A gradient of CO<sub>2</sub> treatments across the mesocosms was established over a four day period by 4 additions of filtered (50 µm), CO<sub>2</sub>-saturated seawater evenly distributed in the water column, as described by Riebesell et al. (2013).  $CO_2$  additions were carried out in the afternoons of tO 5 6 -t4 not to interfere with the daily sampling. A CO<sub>2</sub> addition was also made in the upper 7 m 7 on t15 to counter strong outgassing in the upper water column. Initial  $fCO_2$  ranged from ~240 8 µatm in the two ambient control mesocosms to up to 1650 µatm (Fig. 1A). Unenriched 9 filtered (50 µm) seawater was added to the two control mesocosms (M1, M5). The seawater 10 used for the additions to the mesocosms was collected from the Tvärminne Storfjärden from a 11 depth of 10 m by a pipe connected to the laboratory at the research station.

12 Depth-integrating water samplers (IWS, HYDRO-BIOS, Kiel) were used to collect water 13 from 0 - 17 m depth in each mesocosm for analysis of particulate matter, dissolved inorganic 14 and organic matter, phytoplankton pigments, phytoplankton abundances, carbonate chemistry 15 variables. Samples for carbonate chemistry variables were taken directly from the IWS on 16 board the sampling boat whereas all other samples were pooled in 10 L plastic carboys and stored on board in the dark until sub-sampling on shore (Paul et al., 2015). Particulate matter 17 18 collected in the sediment trap was pumped to the surface and collected in sampling bottles 19 (Boxhammer et al., 2015).

20 Particulate matter (C, N, P) and phytoplankton pigment samples were collected onto GF/F 21 filters (nominal pore size of 0.7 µm, 25 mm diameter, Whatman) by gentle vacuum filtration 22 (pressure <200 mbar). Filters and glass petri dishes were combusted at  $450^{\circ}$ C for 6 hours 23 before use. Collected particulate sediment material was concentrated, freeze-dried and ground to a homogenous powder, while supernatant subsamples were filtered and subsequently 24 25 analysed as for water column material. Total particulate carbon and nitrogen (TPC and PON) content and isotopic composition were analysed according to Sharp et al. (1974) using an 26 elemental analyser (EuroEA) coupled by either a Conflo II to a Finnigan Delta<sup>Plus</sup> isotope 27 ratio mass spectrometer or by a Conflo III to a Thermo Finnigan Delta<sup>Plus</sup> XP isotope ratio 28 29 mass spectrometer. Stable N isotope composition of particulate N is reported in permil (‰) relative to the atmospheric N<sub>2</sub> standard (AIR). Total particulate phosphorus (TPP) 30 31 concentrations were determined spectrophotometrically following sample digestion as 32 described in Hansen and Koroleff (1999). Samples for biogenic silica (BSi) analyses were 33 collected on cellulose acetate filters (pore size of 0.65 µm, 25 mm diameter, Whatman) by

filtration as described above for particulate matter. Concentrations were determined 1 2 spectrophotometrically following sample digestion according to Hansen and Koroleff (1999). 3 Samples for determination of nanomolar concentrations of dissolved inorganic nutrients were filtered (GF/F, nominal pore size of 0.7 µm, Fisher Scientific). Nitrate and nitrite (hereafter 4 5 nitrate) and dissolved inorganic phosphate concentrations were then analysed colorimetrically 6 using a 2 m liquid waveguide capillary cell (Patey et al., 2008; Zhang and Chi, 2002) and a 7 miniaturised detector (Ocean Optics Ltd). Concentrations of ammonium (NH<sub>4</sub><sup>+</sup>) were determined fluorimetrically (Trilogy, Turner) according to Kérouel and Aminot (1997). Total 8 9 dissolved nitrogen (TDN) was analysed using a high-temperature catalytic combustion technique with a Shimadzu TOC-TN V analyser as described by Badr et al. (2003). Samples 10 11 were filtered (GF/F, nominal pore size of 0.7 µm, Fisher Scientific) to remove particulate 12 material and collected in clean glass vials, acidified with HCl to pH 1.9 and flame sealed. 13 Filters and vials were combusted for 6 hours at 450°C before use. Dissolved organic nitrogen (DON) concentrations were calculated by subtracting the inorganic N concentrations from 14 15 TDN. Phytoplankton pigments were extracted in acetone (90%) and after homogenisation and centrifugation, the supernatant was filtered (0.2 µm PTFE filters, VWR International) and 16 17 concentrations were determined by reverse phase high performance liquid chromatography 18 (HPLC; WATERS HPLC with a Varian Microsorb-MV 100-3 C8 column; Barlow et al. 19 (1997), Derenbach et al. (1969)). A library of pre-measured commercial standards was used to 20 calibrate peaks.

Phosphate excess (P\*, Deutsch et al. (2007)) was calculated from the dissolved inorganic
phosphate, nitrate and ammonium concentrations according to:

23 
$$P^* = [PO_4^{3-}] - \frac{[NO_3^-] + [NH_4^+]}{16}$$
 (1)

Dissolved silicate (DSi) drawdown was calculated as the difference in DSi concentration on agiven sampling day (*tx*) and *t1*:

26 DSi drawdown = 
$$[DSi]_{tl} - [DSi]_{tx}$$
 (2)

A comprehensive description of mesocosm deployment, set-up and sampling procedures including sample collection, handling and analyses for particulate matter, dissolved inorganic and organic matter, phytoplankton pigments, and sediment trap particulate matter is covered in Paul et al. (2015), also in this Special Issue. An overview table of sampled variables for the entire experiment, including sampling frequency, is also presented in this accompanying manuscript.

#### 1 2.2 N<sub>2</sub>-fixation rate incubations

2 Incubations for determination of N<sub>2</sub>-fixation rates were carried out using an approach described by Mohr et al. (2010), with some modifications for the preparation of the  ${}^{15}$ N-N<sub>2</sub> 3 enriched seawater. Seawater used for <sup>15</sup>N-N<sub>2</sub> enrichments was filtered (polycarbonate 4 Isopore<sup>TM</sup> filter, pore size of 0.22 µm, 47 mm diameter) before being pumped through a 5 6 degassing membrane (Membrana Mini Module G542) attached to a water-jet pump to remove 7 ambient N<sub>2</sub>. The degassing system was cleaned with 5% HCl before and after use, followed by cycling with deionised water (MilliQ, Millipore) to remove any traces of acid. Seawater 8 9 from the Tvärminne Storfjärden was collected from a depth of 10 m and cycled once through 10 the degassing system before collection in an air-tight, acid-cleaned bag with septum (SKC 11 Tedlar® Bag with single polypropylene fitting) without exposure to the atmosphere. 1 mL of <sup>15</sup>N-N<sub>2</sub> gas (98 atom % <sup>15</sup>N, Sigma Aldrich, Lot no.: CX0937 until *t21*, SZ1670V after *t21*) 12 was injected through the septum into the bag for every 100 mL of sample. The resulting 13 bubble was dissolved and the <sup>15</sup>N-N<sub>2</sub>-enriched seawater was stored at in situ temperature of 14 the mesocosms until addition to incubation bottles. Seawater for the blank incubations was 15 16 prepared in a separate bag using the same process however ambient air was added instead of isotopically labelled <sup>15</sup>N-N<sub>2</sub> gas. 17

18 Water samples for N<sub>2</sub>-fixation rate incubations were directly transferred in a gentle manner 19 from the integrating water sampler into 2.3 L polycarbonate bottles on board the sampling 20 boat using silicon tubing. The bottles were stored in a closed cool box to control temperature 21 and to block sunlight until return to the on shore laboratory. Each bottle was weighed and 22 homogenised by gentle rotation before 70 - 90 mL of water was removed to make space for the <sup>15</sup>N-enriched seawater. Enriched or 'blank' seawater was transferred from the Tedlar® 23 bags to the respective bottles through Tygon<sup>TM</sup> tubing, immersed in the sample bottle, using a 24 25 peristaltic pump to minimise tracer loss through exposure to atmosphere. Incubation bottles 26 were filled with no headspace. After addition, the caps were immediately screwed on to seal 27 the bottles air tight. During these procedures, the bottles were reweighed at each step in order to determine the exact amount of isotope label inside each bottle. The final <sup>15</sup>N-enrichment of 28 dissolved N<sub>2</sub> gas in each bottle was between 1.0 - 3.5 atom %. The bottles were then mixed 29 by gentle rotation and placed in a climate chamber at in situ temperature and under controlled 30 light conditions (~73  $\pm$  1 µmol photons m<sup>-2</sup> s<sup>-1</sup>, mean  $\pm$  S.D). Irradiance was measured using a 31 LI-COR LI-192 quantum sensor. Measured irradiance were within the range of average depth-32 33 integrated (0 - 17 m) irradiance in the mesocosms taken from daily CTD profiles at between 13:30 and 14:30 LT (20 to 300 µmol photons m<sup>-2</sup> s<sup>-1</sup>). The light-dark cycle followed the 34

natural sunrise-sunset variation which on the summer solstice (21 June 2012, *t-1*) was 19:5 hours (L:D). Climate chamber temperature was programmed to follow the daily integrated water column temperature as recorded by the afternoon CTD sampling and thus is reported as in situ temperature. Consistency between irradiance conditions at each bottle position were achieved by a rotation regime. Bottles were rotated gently to mix and the bottle position rotated systematically approximately every three hours during the light cycle. Time of rotation was recorded allowing the calculation of average irradiance between each individual bottle.

8 Incubations were terminated after 24 hours by filtration through a combusted (6 h at 450°C) 9 and acid rinsed (1% HCl) GF/F filter (0.7 µm pore size, 25 mm diameter, Whatman) under 10 reduced vacuum (<200 mbar). Filters were placed in glass petri dishes (combusted 6 h, 11 450°C), frozen immediately and stored at -20°C until analysis on a mass spectrometer as 12 described for particulate C and N analyses above and also in Paul et al. (2015). Rates were calculated according to Montoya et al. (1996). Estimated internal analytical uncertainty in 13 14 calculated N<sub>2</sub>-fixation rates was less than  $\pm 10\%$  when rates were above the detection limit. The detection limit was determined as a difference in  $\delta^{15}N$  between initial and final values of 15 larger than 1.0%. This corresponded to a calculated rate of more than 0.15 nmol N L<sup>-1</sup> d<sup>-1</sup>. 16

#### 17 2.3 Phytoplankton counts

Counts of phytoplankton cells >20 µm were made from 50 mL samples fixed with acidic 18 19 Lugol's iodine solution (1% final concentration). Samples were concentrated using gravitational settling and counted under an inverted microscope (ZEISS Axiovert 100) after 20 21 Utermöhl (1958) and following the guidelines for determination of phytoplankton species 22 composition, abundance and biomass for the COMBINE programme provided by HELCOM 23 (Annex C-6). The cells were counted either on half of the chamber at 100 fold or on 3 to 4 24 strips at 200 fold magnification. Filamentous cyanobacteria were counted in 50 µm length 25 units. Plankton were identified where possible to the species level according to Hoppenrath et al. (2009), Kraberg et al. (2010) and Tomas (1997). Biovolumes of counted plankton cells 26 27 were calculated according to Olenina et al. (2006) and converted to cellular organic carbon 28 quotas by the equations of Menden-Deuer and Lessard (2000).

#### 29 2.4 Statistical analyses

30 A linear regression analysis was applied to determine the relationship between mean  $fCO_2$  and 31 the mean response of each variable for the three experimental phases (Phase I, II and III), as 1 described in Paul et al. (2015). Linear regression analyses were undertaken using R (R Core
2 Team, 2015).

3

#### 4 3 Results

5 Three experimental phases after initial CO<sub>2</sub> manipulation on t0 were defined in Paul et al. 6 (2015) using temperature and chlorophyll *a* (Chl *a*) fluctuations: Phase I (t1 - t16), Phase II 7 (t17 - t30) and Phase III (t31 - t43). These phases are also used to assist with data 8 interpretation in this manuscript. Reported average  $fCO_2$  was calculated for each mesocosm 9 between t1 and t43.

#### 10 **3.1** Inorganic nutrient availability and nutrient limitation

There were low concentrations of inorganic N present throughout the study period with inorganic nitrate concentrations in the range of 3 - 107 nmol L<sup>-1</sup> (Fig. 1C). Ammonium was the dominant source of inorganic N with concentrations ranging between 20 and 289 nmol L<sup>-1</sup> <sup>1</sup>. Hence NH<sub>4</sub><sup>+</sup> was also included in the calculations of P\* (excess phosphate) and inorganic nutrient elemental stoichiometry according to the Redfield ratio (Fig. 1E, Eqn. 1).

There was an excess of inorganic phosphate to inorganic N in all mesocosms ( $P^*>0$  nmol L<sup>-1</sup>, 16 Fig. 1E) and the surrounding waters throughout the study period, with phosphate 17 concentrations ranging between 72 and 214 nmol  $L^{-1}$  in the mesocosms and up to 410 nmol  $L^{-1}$ 18 outside the mesocosms in the surrounding Archipelago Sea. Inorganic phosphate 19 20 concentrations decreased during Phase I, followed by an increase at the beginning of Phase II. Concentrations in the ambient/control treatments remained higher than in the higher CO<sub>2</sub> 21 treatments in Phase III. Nitrate concentrations increased slightly throughout the experiment, 22 whereas  $NH_4^+$  concentrations were variable. Samples for  $NH_4^+$  analyses were lost on t27 and 23 t29. There did not appear to be any remarkable relationship linking accumulated precipitation 24 25 (between sampling days), and the increase in nitrate, indicating that wet atmospheric 26 deposition of nitrate into the mesocosms was effectively prevented by the mesocosm roofs 27 and did not affect the nitrate pool. Precipitation data for the Hanko weather station (ID no.: 28 GHCND:FIE00142025, latitude: 59.8439, longitude: 23.2517) were obtained from the National Oceanographic Data Center (NOAA). 29

#### 30 **3.2** Diatom abundance, silicate dynamics and dissolved N utilisation

Diatoms were mostly abundant at the beginning of the experiment with the species 1 2 Chaetoceros sp. and Skeletonema marinoi present in the large size class (>20 µm, Fig. 2). 3 Fucoxanthin marker pigment concentrations in this size class and suspended BSi 4 concentrations (>0.65  $\mu$ m) declined markedly during the first few days in Phase I and the 5 dynamics fitted well to the microscopy counts of both Chaetoceros sp. and Skeletonema marinoi. Dissolved silicate (DSi) concentrations continued to decrease up until t13. No 6 7 statistically significant difference between CO<sub>2</sub> treatments was detected for diatom abundance 8 (microscopy counts), DSi drawdown or BSi concentrations (Table 1, Figs. <u>2 C, E</u>), apart from 9 BSi in Phase II where a positive effect was detected (p = 0.034, see Paul et al. (2015) for 10 statistical analyses).

Dissolved organic nitrogen (DON) concentrations ranged between 20 and 25 µmol L<sup>-1</sup> (Fig.
<u>2A</u>). DON concentrations appeared to decrease during Phase I, however considerable variability in the data meant this DON drawdown could not be accurately quantified.

#### 14 3.3 Diazotroph abundance and N<sub>2</sub>-fixation rates, $\delta^{15}$ N in particulate N

15 The abundance of filamentous diazotrophic cyanobacteria remained low throughout the experiment with no significant bloom development (<6  $\mu$ g C L<sup>-1</sup>, Fig. <u>3A</u>). The most 16 dominant species, A. *flos-aquae*, had a maximum biomass of 4.9 µg C L<sup>-1</sup> in the mesocosms 17 (M1, t27), whereas the next most abundant species, Anabaena sp., had a maximum biomass in 18 the water column of 0.18  $\mu$ g C L<sup>-1</sup> (M1, *t17*). Aphanizophyll, a pigment present in A. flos-19 20 aquae and Anabaena sp. (Schluter et al., 2004), was detected in both suspended material in 21 the water column (>20  $\mu$ m), and in the sinking material collected in the sediment trap. 22 Concentrations of this pigment increased at the end of Phase I concurrent with an increase in 23  $N_2$ -fixation rates (Fig. 3). Although numbers in the mesocosms remained generally low, A. flos-aquae abundances based on microscopy counts and phytoplankton pigment analyses, 24 were highest in Phases II/III and lowest in Phase I (Fig. 3). A. flos-aquae biomass outside the 25 mesocosms was up to 30  $\mu$ g C L<sup>-1</sup> on *t15* and is supported by high Aphanizophyll pigment 26 concentrations of 109 ng  $(mg TPC)^{-1}$  also on *t15* (data not shown). 27

Rates of N<sub>2</sub>-fixation until *t21* ranged from below the detection limit at the beginning of the experiment, up to 4.4 nmol N L<sup>-1</sup> d<sup>-1</sup> inside the mesocosms and up to 37.9 nmol L<sup>-1</sup> day<sup>-1</sup> in the waters outside. We observed a substantial increase in the N<sub>2</sub>-fixation rates from 2.6 to 4.4 nmol L<sup>-1</sup> day<sup>-1</sup> up to 50 to 60 nmol L<sup>-1</sup> day<sup>-1</sup> between *t21* and *t23* without any remarkable change in diazotroph abundance of the same magnitude (Fig. 3). This was also evident in *A*.

flos-aquae biomass-related N2-fixation rates (see Fig. B, Supplementary Materials). This 1 increase coincided with the use of a new <sup>15</sup>N-N<sub>2</sub> gas bottle with a lot number which was 2 reported two years later as contaminated with  $^{15}$ N-labelled NH<sub>4</sub><sup>+</sup> and NO<sub>3</sub><sup>-</sup> by Dabundo et al. 3 (2014) (Sigma Aldrich, Lot no. SZ1670V). The measured rates from t23 on are therefore not 4 exclusively N<sub>2</sub>-fixation and are not reliable thus they were excluded from analyses. In 5 6 addition to the bottle assays, the <sup>15</sup>N-N<sub>2</sub> isotope tracer was also added directly to all mesocosms except for M1 (control) and M7 (see Supplementary Materials). Therefore these 7 two mesocosms were not affected by this contamination issue. Hence, the natural abundance 8  $\delta^{15}$ N data from the suspended material in the water column and the sinking material from the 9 sediment trap is reported for the entire experiment (t-3 until t43) for M1 and M7 mesocosms 10 (Figs. 3E, F) but only until t21 for M3, M5, M6 and M8. Any NH<sub>4</sub><sup>+</sup> or nitrate added to the 11 four mesocosms with the isotope tracer was highly isotopically enriched in <sup>15</sup>N but was in 12 13 very low concentration and so was insignificant for the nutrient budget.

The natural abundance  $\delta^{15}$ N in suspended particulate N in the mesocosms decreased during 14 the period of higher Chl a in Phase I from  $6.0 \pm 0.5$  % on t1 to  $2.6 \pm 0.5$  % on t15 (mean  $\pm$ 15 S.D.). This indicated potential input of atmospheric N with a low  $\delta^{15}$ N into particulate matter 16 via N<sub>2</sub>-fixation during this period or potential uptake of ammonium with a  $\delta^{15}$ N signature 17 depleted through ammonification. A sharp decrease in  $\delta^{15}$ N in the sinking particulate material 18 occurred on t17, the same day that considerable amounts of Aphanizophyll and Fucoxanthin 19 were found in the sediment trap material (Figs. 3D, F, Fucoxanthin not shown). This was one 20 day after the mesocosm walls were cleaned indicating that there were likely diazotrophic 21 22 species and diatoms attached to the mesocosm walls. Identification from microscope photos 23 revealed the presence of filamentous cyanobacteria with heterocysts in the collected sediment trap material. Aside from this, there were no remarkable fluctuations in  $\delta^{15}N$  in either the 24 suspended or sinking particulate matter pools, including after *t21* in M1 and M7 (Figs. <u>3E, F</u>). 25

Assessment of in situ N<sub>2</sub>-fixation rates based on <sup>15</sup>N -uptake from the combined dissolved N 26 pool of  $NO_3^-$ ,  $NH_4^+$  and  $N_2$  was abandoned due to high uncertainty in initial <sup>15</sup>N enrichment 27 28 and concentrations of the combined dissolved N pool, and fast saturation of label uptake after 29 ca. four days (two successive sampling days). To assess the contribution of diazotrophy to N supply in the mesocosms, we calculated a theoretical cumulative diazotrophic N input using 30 measured N<sub>2</sub>-fixation rates from bioassays up until t21 (M1 = 20 nmol N L<sup>-1</sup>), and then 31 assumed a constant N<sub>2</sub>-fixation rate of 4 nmol N  $L^{-1} d^{-1}$  into particulate N between *t23* and *t43* 32 (total = 80 nmol N  $L^{-1}$ ). The assessment for between t23 and t43 is based on the premise of 33 34 continued elevated A. flos-aquae biomass and assuming 50% exudation of fixed N as DON or

 $NH_4^+$  ( $< t21 = 20 \text{ nmol N } L^{-1}$ ,  $> t21 = 80 \text{ nmol } L^{-1}$ , total = 100 nmol N  $L^{-1}$ ). This yielded a 1 theoretical new N input from A. flos-aquae of only 200 nmol N L<sup>-1</sup>, amounting to ~5% of 2 mean PON pool standing stock (~  $3 \mu mol L^{-1}$ ) and is clearly at the higher end of estimations. 3 We calculated corresponding N requirement of the plankton community of 27.2 nmol N  $L^{-1} d^{-1}$ 4 <sup>1</sup> from the average phosphorus uptake rate across all treatments of 1.7 nmol  $PO_4^{3-}L^{-1}d^{-1}$  from 5 t1 - t30 as reported by Nausch et al. (2015), by assuming Redfield nutrient uptake 6 7 stoichiometry (16N:1P). This is almost seven times larger than estimated daily diazotrophic N inputs of ~4 nmol N  $L^{-1} d^{-1}$ , corresponding to 14% of calculated community N requirement. 8

Low filamentous diazotrophic cyanobacteria abundances exacerbated the inherent sampling 9 10 error in both microscopy and pigment analyses due to patchy distribution and the tendency of 11 filaments to aggregate. Hence, unfortunately no reliable statistical analyses on the effect of 12 higher  $fCO_2$  on diazotroph abundance or marker pigment concentration could be undertaken, for any phase of the experiment. Any potential CO<sub>2</sub> effect on diazotroph abundance was also 13 14 not obvious on visual data inspection, and no effect could be detected on N2-fixation rates or  $\delta^{15}$ N natural abundance in suspended particulate matter from the water column or sediment 15 16 trap particulate matter up until t21 (Table 1), when rates were reliable and there was data from a sufficient number of CO<sub>2</sub> treatments. 17

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#### 19 4 Discussion

#### 20 4.1 Effects of elevated CO<sub>2</sub> on diazotrophic N inputs

21 Bioavailable N was present in low concentrations and was probably the limiting 22 macronutrient in the plankton community. Hence, higher phytoplankton biomass and lower 23 phosphate concentrations at higher CO<sub>2</sub> observed in this same mesocosm study (Paul et al., 2015), may have suggested relief of N limited growth by potentially increased N<sub>2</sub>-fixation. 24 25 However we have no strong evidence to support this hypothesis based on N pool standing 26 stocks and estimated diazotrophic N inputs. The only statistically significant, but very minor, 27 correlation was a positive relationship between  $CO_2$  and PON concentrations (Fig. 1G, Table 1, 0.08  $\mu$ mol L<sup>-1</sup>, 3% difference in PON, slope = 1.75 x 10<sup>-4</sup>  $\mu$ mol L<sup>-1</sup>  $\mu$ atm<sup>-1</sup>, data from Paul 28 et al. (2015)). No significant difference in N<sub>2</sub>-fixation rates (until t21) or A. flos-aquae 29 abundance at elevated  $CO_2$  compared to the ambient treatments was detected (Table 1, Fig. 3). 30 31 Phosphate turnover rates, a potential indicator of P demand for N<sub>2</sub>-fixation, were also unaffected by CO<sub>2</sub> in Phases I or II (Nausch et al., <u>2015</u>). These variables (N<sub>2</sub>-fixation and 32

1 phosphate uptake rates) provide a more sensitive measure of turnover rates of N and P than 2 assessing changes in N pool standing stocks in this tightly-coupled regenerative plankton community. Unfortunately, we only have reliable  $N_2$ -fixation rates from incubations until t213 due to contamination of <sup>15</sup>N-N<sub>2</sub> gas with bioavailable N compounds (Dabundo et al., 2014) 4 and not after  $\sim t25$  when significant CO<sub>2</sub>-related differences in C and P pools were apparent. 5 Hence, in the later stages of the experiment (Phase II and III), it is possible that there was a 6 7 divergence in N<sub>2</sub>-fixation rates between treatments that was missed, despite low abundances of A. flos-aquae, the dominant filamentous diazotrophic cyanobacterium present. Nonetheless 8 we estimate that the contribution of diazotrophy to N supply in the mesocosms over the study 9 duration of 43 days was small (~200 nmol  $L^{-1}$ ). Maximum measured N<sub>2</sub>-fixation rates of 4.4 10 nmol N L<sup>-1</sup> d<sup>-1</sup> were low compared to reported for the Baltic Sea in mid-summer which range 11 from 1.7 up to 550 nmol N  $L^{-1} d^{-1}$  (Farnelid et al., 2013; Ohlendieck et al., 2000, 2007; 12 Wasmund et al., 2001). This is due to the rather low A. flos-aquae biomass in the mesocosms 13 compared to literature values (this study: maximum biomass = 5  $\mu$ g C L<sup>-1</sup> integrated over 0 – 14 17 m; Gulf of Finland:  $22 - 26 \mu g C L^{-1}$  in the surface 5 m,  $6 - 7 \mu g C L^{-1}$  at 20 m deep in July 15 (Laamanen and Kuosa, 2005)). Thus even if all newly-fixed N by diazotrophs was transferred 16 17 to diazotroph and plankton biomass (i.e PON pool), this small accumulation would most likely remain below the detection limits in the suspended PON pool (~10% = 0.3  $\mu$ mol L<sup>-1</sup>). 18 19 On top of this, any CO<sub>2</sub>-related differences in N<sub>2</sub>-fixation would be near impossible to resolve 20 in this small contribution by diazotrophs.

21 The absence of any detectable effect may of course be influenced by the relatively low 22 abundances of filamentous diazotrophic cyanobacteria in this study, as temperatures were 23 mostly below temperatures thought to stimulate bloom development (16°C, Wasmund et al. (1997); this study 8 – 16°C, Paul et al. (2015)). Nevertheless our results from this  $CO_2$ 24 manipulation study are in agreement with studies from both the marine (Böttjer et al., 2014; 25 26 Law et al., 2012) and freshwater (Shapiro, 1997; Yamamoto, 2009) realms which detected no significant effect of decreased pH/increased CO<sub>2</sub> on diazotroph abundance and/or activity in 27 28 natural plankton communities. These four independent studies all contradict physiological 29 investigations in single-strain culture experiments where diazotroph growth and activity was modulated by CO<sub>2</sub> availability (e.g. Barcelos e Ramos et al., 2007; Czerny et al., 2009; 30 31 Eichner et al., 2014; Fu et al., 2008; Hutchins et al., 2013; Wannicke et al., 2012). 32 Diazotrophic organisms typically have slower growth rates than other organisms. Hence any 33 potential influence of ocean acidification on their physiology may take longer to become 34 apparent in biogeochemical parameters sampled in larger-scale field studies, where most

1 sampled variables such as PON are a mixture of organic compounds of various origin and 2 isotopic composition. In addition, the overall response to CO<sub>2</sub> observed in such field studies is a combination of the pure physiological response, which can be observed in laboratory 3 experiments, with trophic interactions such as grazing and competition between species for 4 5 nutrients and light. However to the best of our knowledge, there are no direct N<sub>2</sub>-fixation rate measurements from CO<sub>2</sub>-manipulation studies with A. flos-aquae in the field which could 6 7 shed light on any underlying physiological response of this diazotroph and confirm laboratory findings in the field. Furthermore, high grazing pressure, hence top-down control, particularly 8 after t17 (Lischka et al., 2015) may have overridden any potential CO<sub>2</sub> effect of bottom-up 9 control on diazotroph growth. 10

11 In addition to these highly visible filamentous N<sub>2</sub>-fixers, there is growing evidence to support 12 the role of heterotrophic and non-phototrophic N<sub>2</sub>-fixation by smaller unicellular organisms in diverse ecosystems (Halm et al., 2012; Loescher et al., 2014; Moisander et al., 2010; Zehr et 13 al., 2008) including in the Baltic Sea and Kattegat (Bentzon-Tilia et al., 2015; Farnelid et al., 14 15 2009), which cannot be quantified by common microscopic methods used in this experiment. Hence, while there appeared to be a good correlation between A. flos-aquae abundance and 16  $N_2$ -fixation rates until t21 in this study, we cannot rule out the contribution of heterotrophic 17 18 organisms to the measured rates. However, regardless of the diazotroph community present, N<sub>2</sub>-fixation rates were low and diazotrophy made only a small contribution (< 200 nmol  $L^{-1}$ ) 19 to the N cycle in this study. Thus we have no direct evidence from observations in this study 20 that  $N_2$ -fixation or diazotroph abundance (Fig. 3) were significantly influenced by  $CO_2$  nor 21 22 that this could explain the observed higher particulate matter concentrations or lower 23 phosphate concentrations in the higher CO<sub>2</sub> treatments (Paul et al., 2015) based on 24 hypothesised relief of N-limitation.

25 In this area of the Baltic Sea, plankton communities, containing filamentous diazotrophic 26 cyanobacteria, are exposed to large diurnal and seasonal changes in pH (Almén et al., 2014; 27 Brutemark et al., 2011). In addition, filamentous cyanobacteria form characteristic surface 28 aggregations. Inside these aggregations, microenvironments can create substantially different conditions compared to the surrounding water with large diurnal fluctuations in pH (7.4 vs 29 9.0) and O<sub>2</sub> concentrations (~150 - 450  $\mu$ mol O<sub>2</sub> L<sup>-1</sup>) and thus also inorganic carbon 30 availability (Ploug, 2008). Hence natural exposure to highly variable carbonate chemistry 31 32 conditions may have also played a role in dampening any potential influence of ocean 33 acidification in this plankton community.

## 4.2 Evidence from N pools of the importance of regenerative production and effects of CO<sub>2</sub>

3 Productivity in this plankton community appeared to be dominated by regenerative production 4 (sensu Dugdale and Goering (1967)) under low nitrate availability during Phase I, as has been observed in summer plankton communities in the Baltic Sea (Kuparinen, 1987; Sahlsten and 5 6 Sörensson, 1989; Tamminen, 1995). DON appeared to be a more important N source than N 7 derived from N<sub>2</sub>-fixation. Any relatively fresh and labile N-rich dissolved organic matter 8 (DOM) present after the decline of the spring bloom was likely remineralised by the bacterial 9 community. Here, simultaneous drawdown of DSi and DON between t-1 and t15 suggests that in particular diatoms, also persisting from the spring bloom, were beneficiaries of this organic 10 N turnover. Available  $NH_4^+$  (~100 nmol L<sup>-1</sup>) could not have supported the DSi uptake (~0.4 11  $\mu$ mol L<sup>-1</sup>) as the sole N source based on ~1:1 molar Si:N requirement by diatoms, thus 12 suggesting instead potential rapid resupply of NH<sub>4</sub><sup>+</sup> through remineralisation of organic N by 13 14 the heterotrophic community particularly in Phase I and Phase II. Although there is no indication of a high level of  $NH_4^+$  production above the variability in the data set, we presume 15 this bioavailable NH<sub>4</sub><sup>+</sup> would have been very quickly assimilated into particulate N in the N-16 17 limited plankton community. This rate of N regeneration probably limited net phytoplankton 18 growth such that significant phytoplankton biomass could not accumulate in the water column. Nevertheless, neither the readily available  $NH_4^+$  nor the nitrate pool were fully 19 exploited by the plankton assemblage with up to 50 nmol  $L^{-1}$  of nitrate and 170 nmol  $L^{-1}$  of 20 21  $NH_4^+$  remaining at the end of the study period on t43. In fact, nitrate concentrations continually increased throughout the experiment at an average net rate of 1 nmol N L<sup>-1</sup> day<sup>-1</sup> 22 (Fig. 1C) despite proportionally high phosphate availability. This suggests a small net 23 imbalance in N cycle processes and may be connected to ammonium inhibition of nitrate 24 25 uptake during spring-bloom decline and post-bloom period in the study area (Tamminen, 26 1995), leading to this small accumulation of nitrate in the water column.

No significant effect of  $CO_2$  was detected on the DON pool, nor DSi drawdown, or PON or BSi cumulative sinking fluxes (see also Paul et al. 2015 in this Special Issue). Likewise, if there was any difference in uptake of N from the N-rich DOM pool (N:P ~ 80:1) between  $CO_2$ treatments, we could not detect the small signal (nmol L<sup>-1</sup>) outside of the analytical precision (µmol L<sup>-1</sup>) of the DON measurements. Thus this organic N drawdown via regenerative production in diatoms in this study appeared to be either unaffected or immeasurable by simulated ocean acidification.

#### 1 5 Summary

2 Plankton biomass build-up in this study was limited by low inorganic N availability therefore 3 organic N pools were utilised supporting regenerative production during the more productive 4 period in Phase I, with diatoms benefitting from this N turnover. Estimated N<sub>2</sub>-fixation rates and abundances of the most dominant filamentous diazotroph, A. flos-aquae, remained very 5 6 low, therefore diazotrophs probably made only a minor contribution to overall N supply in 7 this plankton community. Hence we did not observe relief of N limitation and stimulation of a 8 summer plankton bloom by non-diazotrophic organisms. Indeed, dissolved inorganic nitrate 9 present increased throughout the experiment indicating higher supply than consumption, 10 despite a considerable phosphate excess present.

11 We detected no significant differences in N pool sizes between CO<sub>2</sub> treatments apart from the 12 PON pool. However, the detected positive effect of CO<sub>2</sub> on PON standing stocks was minor 13 (<3% difference in PON concentration). Thus N uptake rates were well balanced with supply 14 or any net CO<sub>2</sub>-related differences were too small to be detected in N pool sizes across the 15 range of simulated ocean acidification scenarios. In addition, we found no conclusive evidence from our data until t21 (N<sub>2</sub>-fixation rates, A. flos-aquae abundances, natural  $\delta^{15}$ N 16 abundances) that  $CO_2$  had a measurable impact on N inputs via diazotrophy. The absence of 17 18 any detectable effect may have been influenced by the low abundances of filamentous diazotrophic cyanobacteria in this study. However, the lack of response was consistent with 19 20 other studies of diazotrophic organisms in natural plankton communities where resource 21 competition with other plankton functional groups and top-down control may also play 22 important roles in mediating the physiological response of N<sub>2</sub>-fixing organisms.

23 Nonetheless, it appears that increased CO<sub>2</sub> may have slightly enhanced the ability of the N-24 limited plankton community in the Baltic Sea to exploit the low N sources available thereby 25 potentially explaining lower phosphate concentrations, higher particulate matter concentrations and Chl a observed under higher CO<sub>2</sub> (Paul et al., 2015). However, we have no 26 27 direct evidence of increased new N inputs via diazotrophy or changed N biogeochemistry 28 within the first three weeks and no conclusive indirect evidence from N pool sizes up to six 29 weeks after CO<sub>2</sub> manipulation. Therefore we conclude that elevated CO<sub>2</sub> had no observable impact on the N cycle in this summer Baltic Sea plankton community. 30

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1 Table 1. Summary of linear regression analyses of fCO<sub>2</sub> and nutrient stoichiometry, dissolved 2 silicate drawdown, abundance of large (>20 µm) dominant diatom species present 3 (Chaetoceros sp., Skeletonema marinoi), N2-fixation rates, stable nitrogen isotope natural 4 abundance, and particulate biogenic silica and particulate organic nitrogen concentrations. Numbers in bold indicate variable had a negative correlation with average  $fCO_2$ . Dashes 5 6 indicate no regression was completed to avoid any bias in the conclusions because either no 7 data or no complete data set is available. Asterisk (\*) indicates data and statistical analyses 8 from Paul et al. (2015). Degrees of freedom, n = 4.

9

<u>Variable</u>	Phase	p	F-statistic	$\underline{\mathbf{R}}^2$
N <sub>2</sub> -fixation rate	Ī	<u>0.764</u>	<u>0.104</u>	0.025
	II	<u></u>	<u></u>	<u></u>
	III	<u></u>	<u></u>	<u></u>
$\frac{\delta^{15}N \text{ in suspended}}{\text{particulate matter}}$	Ī	<u>0.417</u>	<u>0.819</u>	<u>0.170</u>
	II			
	III			==
<u>ð<sup>15</sup>N in sinking</u> particulate matter	Ī	<u>0.289</u>	<u>1.494</u>	<u>0.272</u>
	II			
	III			
<u>DSi drawdown</u>	Ī	<u>0.927</u>	<u>0.010</u>	0.002
	II	0.520	0.496	<u>0.110</u>
	III	<u>0.966</u>	<u>0.001</u>	<u>0.002</u>
<u>Chaetoceros</u> sp. abundance	Ī	<u>0.737</u>	<u>0.129</u>	<u>0.031</u>
	<u>II</u>	<u></u>		
	III	<u>0.075</u>	<u>5.726</u>	<u>0.589</u>
<u>Skeletonema</u> <u>marinoi abundance</u>	<u>I</u>	<u>0.772</u>	<u>0.097</u>	<u>0.024</u>
	<u>II</u>			<u></u>
	<u> </u>			<u></u>
Excess phosphate (P*)	<u>1</u>	<u>0.493</u>	<u>0.569</u>	0.125
	<u>II</u>	0.783	0.086	0.021
	<u> </u>	0.004	<u>37.560</u>	0.125
DIN:DIP (includes <u>NH4</u> <sup>+</sup> )	<u>1</u> 11	0.647	0.569	0.125
	<u>II</u> III	<u>0.550</u> 0.707	0.412	0.095
	<u> </u>	0.797	0.070	0.019
<u>Skeletonema</u> <u>marinoi abundance</u>	<u>1</u> П	<u>0.772</u>	0.097	<u>0.024</u>
	Π̈́			
Biogenic silica (BSi) <u>*</u>	I	0.070	0.601	6.032
	Ī	0.034	0.717	10.120
	ĪĪĪ	0.553	0.095	0.419
PON (total) *	Ī	0.668	0.051	0.214
	II	0.490	0.126	0.576
	III	0.001	0.940	<u>62.890</u>





Figure 1. Temporal development in A) calculated  $fCO_2$  using measured DIC and pH<sub>T</sub>, B) 3 chlorophyll a concentrations, C) dissolved inorganic nitrate concentrations, D) dissolved 4 5 inorganic phosphate concentrations over the study period, E) excess dissolved inorganic 6 phosphate concentrations (P\*) calculated according to Eqn. 1, F) measured dissolved 7 ammonium concentrations and G) suspended particulate organic nitrogen concentrations. Data for A – D, and F – G is from Paul et al. (2015). \* = p < 0.05, \*\* = p < 0.01 where red 8 9 indicates positive and blue a negative detected effect of  $fCO_2$ . Average treatment  $fCO_2$  was 10 calculated for each mesocosm between t1 and t43





3 Figure 2. Temporal development in A) dissolved organic nitrogen concentrations (DON), C) 4 dissolved silicate (DSi) drawdown and E) particulate biogenic silicate (BSi) concentrations 5 (data from Paul et al. (2015)), the abundances of the two dominant diatom species determined 6 by microscopy (B, D) and F), Fucoxanthin marker pigment concentrations (>20  $\mu$ m), a key 7 pigment in diatoms. Red asterisk denotes significant positive effect of CO<sub>2</sub> (\* = p < 0.05). 8





Figure 3. Variables indicating abundance and activity of filamentous diazotrophic 2 3 cyanobacteria: A) biomass of A. *flos-aquae* calculated from microscopy abundance data, B) N<sub>2</sub>-fixation rates determined by stable isotope incubations, C) carbon-normalised 4 5 Aphanizophyll marker pigment concentration (>20  $\mu$ m) relative as a proxy for A. flos-aquae abundance in the water column and D) in the sediment trap material, E) natural abundance 6  $\delta^{15}$ N of particulate organic nitrogen (PON) in the water column and F) natural abundance 7  $\delta^{15}N$  in the sinking particle organic nitrogen collected in the sediment trap determined by 8 9 analyses on an isotope ratio mass spectrometer. The green shaded area in B) between t23 and t43 indicates when contaminated <sup>15</sup>N-N<sub>2</sub> gas was used in incubations (see Dabundo et al. 10 11 2014).

#### 1 SUPPLEMENTARY MATERIALS

#### 2 Enrichment of mesocosms with <sup>15</sup>N-N<sub>2</sub> gas

Four of six mesocosms spanning the range of fCO2 treatments were enriched with the 3 isotopically labelled <sup>15</sup>N-N<sub>2</sub> gas to investigate the fate of newly fixed N in this plankton 4 community under future ocean acidification conditions. A similar approach to Mohr et al. 5 6 (2010), as described for the  $N_2$ -fixation incubations (see Section 2.2), was employed on a 7 larger scale. A total of approximately 1500 L of unfiltered seawater was collected from the 8 Baltic at ca. 10 m depth and pumped into the laboratory building at Tvärminne Zoological 9 Station. Mesocosm enrichment occurred in two pulses on t22 and t26. We added this in two steps because of the limited number of bags available for preparing the <sup>15</sup>N-N<sub>2</sub> enriched 10 seawater. For the first step, seawater was filtered and collected as for the N2-fixation 11 12 incubations in bags (thermoplastic polyurethane, ~30 L capacity) with a tap and a crimp 13 sealed septum (N20 grey butyl rubber plugs, Macherey and Nagel) on opposite ends of the 14 bag. The large physical effort required to dissolve the gas by 'bag-slapping', as commonly 15 done for small volumes using the method described by Mohr et al. (2010), led to a 16 modification of the enrichment method for the second enrichment step. Water was collected and degassed as previously described through the degassing membrane. Instead of collecting 17 18 the water directly after this step, the water then passed through a second membrane that was flooded with <sup>15</sup>N-N<sub>2</sub> gas and was connected to an overflow system which allowed monitoring 19 20 of gas dissolution (Fig. A). The high surface area in the membrane enhanced the labelled gas 21 dissolution. This enriched water was then pumped directly into the empty collection bags using a peristaltic pump without contact with the atmosphere. One complete cartridge of gas 22 (500 mL, nitrogen - <sup>15</sup>N-N<sub>2</sub>, 98 atom % <sup>15</sup>N, Sigma Aldrich, Lot no.: SZ1670V, SZ1423V, 23 CX0937) was added per bag through the septum. A total of 150 L of enriched seawater 24 prepared was added to four mesocosms (M3, M5, M6, M8), and 100 L unenriched filtered 25 26 seawater was added to the other two mesocosms (M1, M7) as isotope label controls on t22 27 and *t26*.



- 3 Figure S1. Diagram of set-up used for large-scale preparation of <sup>15</sup>N-N<sub>2</sub> enriched seawater
- 4 which was added to selected mesocosms.
- 5



Figure S2. A. flos-aquae carbon-normalised N2-fixation rates over the study period. Where

data points are missing before t9, rates were either below detection limit (0.15 nmol N  $L^{-1} d^{-1}$ ) 

or did not coincide with sampling for phytoplankton abundance counts. Green shaded area

between t23 and t43 indicates when contaminated <sup>15</sup>N-N<sub>2</sub> gas was used in incubations (see 

