

1 **Effects of CO₂ perturbation on phosphorus pool sizes and uptake in a**
2 **mesocosm experiment during a low productive summer season in the**
3 **northern Baltic Sea**

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40 **Abstract**

41 Studies investigating the effect of increasing CO₂ levels on the phosphorus cycle in natural
42 waters are lacking although phosphorus often controls phytoplankton development in many
43 aquatic systems. The aim of our study was to analyze effects of elevated CO₂ levels on
44 phosphorus pool sizes and uptake. The phosphorus dynamic was followed in a CO₂-manipulation
45 mesocosm experiment in the Storfjärden (western Gulf of Finland, Baltic Sea) in summer 2012
46 and was also studied in the surrounding fjord water. In all mesocosms as well as in surface
47 waters of Storfjärden, dissolved organic phosphorus (DOP) concentrations of 0.26±0.03 and
48 0.23±0.04 μmol l⁻¹, respectively, formed the main fraction of the total P-pool (TP), whereas
49 phosphate (PO₄) constituted the lowest fraction with mean concentration of 0.15 ±0.02 μmol l⁻¹
50 in the mesocosms and 0.17±0.07 μmol l⁻¹ in the fjord. Transformation of PO₄ into DOP appeared
51 to be the main pathway of PO₄ turnover. About 82% of PO₄ was converted into DOP whereby
52 only 18% of PO₄ was transformed into particulate phosphorus (PP). PO₄ uptake rates measured in
53 the mesocosms ranged between 0.6 and 3.9 nmol l⁻¹ h⁻¹. About 86% of them were realized by the
54 size fraction <3 μm. Adenosine triphosphate (ATP) uptake revealed that additional P was
55 supplied from organic compounds accounting for 25-27% of P provided by PO₄ only. CO₂
56 additions did not cause significant changes in phosphorus (P) pool sizes, DOP composition, and
57 uptake of PO₄ and ATP when the whole study period was taken into account. However,
58 significant short-term effects were observed for PO₄ and PP pool sizes in CO₂ treatments >1000
59 μatm during periods when phytoplankton biomass increased. In addition, we found significant
60 relationships (e.g., between PP and Chl*a*) in the untreated mesocosms which were not observed
61 under high *f*CO₂ conditions. Consequently, it can be hypothesized that the relationship between
62 PP formation and phytoplankton growth changed with CO₂ elevation. It can be deduced from the
63 results, that visible effects of CO₂ on P pools are coupled to phytoplankton growth when the
64 transformation of PO₄ into POP was stimulated. The transformation of PO₄ into DOP on the
65 other hand does not seem to be affected. Additionally, there were some indications that cellular
66 mechanisms of P regulation might be changed under CO₂ elevation changing the relationship
67 between cellular constituents.

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70 **Kea words**

71 ocean acidification, phosphorus cycling, DOP composition, PO₄ uptake, ATP uptake, Baltic Sea

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75 **1. Introduction**

76 Increasing emissions of anthropogenic CO₂ into the atmosphere and subsequent acidification of
77 the ocean can potentially affect the diversity of organisms and the functioning of marine
78 ecosystems (Eisler, 2012). The rise in atmospheric CO₂ concentrations was accelerated from
79 3.4±0.2 PgC yr⁻¹ in the 1980s to 4.0±0.2 PgC yr⁻¹ in the 2000s leading to increases of CO₂ in
80 ocean surface waters at a similar rate (IPCC, 2013). Atmospheric CO₂ is projected to rise to 750
81 - 1000 ppm and higher in 2100 (IPCC, 2001) corresponding with a decrease in open ocean pH by
82 0.3-0.5 units (Caldeira and Wickett, 2005) from the present level of ~8.1. Although this process
83 is of global significance and all parts of the oceans are at risk, there will be regional differences
84 in the degree of acidification (Borges et al., 2005). Thus, to determine the CO₂-related changes in
85 the oceans, multiple studies in different regions are required. Semi-enclosed coastal regions, such
86 as the Baltic Sea, react with higher changes in pH to CO₂ elevation than open ocean waters due
87 to high freshwater inputs resulting in a reduced buffer capacity (Orr, 2011).

88 In the Baltic Sea, several studies of CO₂ effects have been undertaken on the organism level of
89 fish (Frommel et al., 2013), zooplankton (Pansch et al., 2012; Vehmaa et al., 2012), macrophytes
90 (Pajusalu et al., 2013), benthic species (Hiebenthal et al., 2013; Stemmer et al., 2013), and
91 filamentous cyanobacteria (Czerny et al., 2009; Eichner et al., 2014; Wannicke et al., 2012).
92 Studies on the impacts of elevated CO₂ at the ecosystem level, however, have thus far been
93 limited to Kiel Bight in the western Baltic Sea (Engel et al., 2014; Rossoll et al., 2013; Schulz
94 and Riebesell, 2013), which may fundamentally differ from other parts of the Baltic Sea.

95 Next to nitrogen, phosphorus (P) controls the productivity of phytoplankton in the ocean (Karl,
96 2000; Sanudo-Wilhelmy et al., 2001; Tyrrell, 1999) and is a limiting factor in some regions
97 (Ammerman et al., 2003). The total phosphorus (TP) pool comprises phosphate (PO₄), dissolved
98 organic phosphorus (DOP), and particulate organic (POP) and inorganic (PIP) phosphorus. There
99 is a continuous transformation of phosphorus between these P species due to their uptake,
100 conversion, and release by organisms as well as by interaction with minerals. While PO₄ is the
101 preferred P-species of phyto- and bacterioplankton, DOP can become an important P source
102 when PO₄ is depleted (Llebot et al., 2010; Lomas et al., 2010). DOP includes nucleic acids,
103 phospholipids, and adenosine triphosphate (ATP) (Karl and Björkman, 2002) which are
104 structural and functional components of all living cells, but, can be also released into the
105 surrounding water.

106 In general, there is little knowledge on how the P cycle is affected by ocean acidification and
107 how related changes in P availability influence the response of organisms to CO₂ elevation. In
108 CO₂ manipulation experiments, particulate phosphorus dynamics were studied to determine

109 effects on C:P stoichiometry of phytoplankton (Riebesell and Tortell, 2011; Sugie and
110 Yoshimura, 2013) and PO₄ concentration dynamics to estimate its utilization (Bellerby et al.,
111 2008). CO₂ effects on phosphorus pool sizes and PO₄ uptake have so far been studied by Tanaka
112 et al. (2008) in the Raunefjorden, Norway and by Unger et al. (2013) and Endres et al. (2013) in
113 laboratory experiments with cultures of *Nodularia spumigena*. In order to reduce the gap of
114 knowledge, we studied the impact of elevated CO₂ on phosphorus pool sizes, the DOP
115 composition, and PO₄ uptake of a northern Baltic Sea plankton community. These measurements
116 provide important information on potential changes in P cycling under increasing CO₂ levels and
117 contribute to a better understanding of the P cycle in brackish water ecosystems.

118

119

120 **2. Material and methods**

121 **2.1 Experimental design and CO₂ manipulation**

122 The study was conducted in the northwestern Gulf of Finland, in the proximity of the Tvärminne
123 Zoological Station (TZS) (Fig. 1), between June 17 and August 4, 2012, using the KOSMOS
124 mesocosm system (Riebesell et al., 2013). Nine mesocosms (M1 - M9) were moored in the open
125 waters of the Storfjärden (5951.5 N, 2315.5 E) at a water depth of ~30 m. Only six of them were
126 included throughout the whole study period since leakages in the remaining three rendered them
127 unusable. Equipment and deployment procedures are described in detail by Paul et al. (2015b).
128 Briefly, polyurethane enclosure bags of 2 m in diameter and 18.5 m in length were mounted in
129 floating frames and lowered in such a way that ~17 m of each bag were immersed in the water
130 column and ~1.5 m remained above the water surface. Large organisms were excluded from the
131 mesocosms by a 3-mm mesh installed at the top and bottom of the bags before closure. The
132 mesocosms were deployed 10 days prior to CO₂ manipulation to rinse the bags and for full water
133 exchange. Sediment traps were mounted on the lower ends to close them water tight, while the
134 upper ends were raised above the water surface to prevent water entry during wave action. The
135 mesocosms were covered with a dome shaped roof to prevent nutrient input by bird and
136 potentially significant fresh water input by rain. Salinity gradients were removed by bubbling the
137 mesocosms with compressed air for 3.5 min, so that 5 days before the start of the experiment
138 (day -5) the water body was fully homogeneous.

139 CO₂ was injected at day 0 and the subsequent 4 days by pumping various quantities of 50-µm-
140 filtered and CO₂-enriched fjord water into seven of the mesocosms as described by Riebesell et
141 al. (2013). The intended CO₂ and pH gradients were reached after the last treatment on day 4.
142 Details are described in Paul et al. (2015b). For the two untreated (control) mesocosms, only

143 filtered fjord water was added to adjust the water volume to that of the treated mesocosms. To
144 compensate for outgassing, the CO₂ manipulation was similarly repeated in the upper 7 m layer
145 of the mesocosms on day 16.

146

147 **2.2 Sampling**

148 Daily sample collection started 3 days before the first CO₂ injection (day-3). Parallel samples
149 were taken from the surrounding fjord. Sampling over the entire 17 m depth was carried out
150 using an integrating water sampler (IWS HYDROBIOS -KIEL) that was lowered slowly on a
151 cable by hand. The sampling frequency differed depending on the parameter to be observed as
152 shown in the overview by Paul et al. (2015b).

153 Phosphorus pool parameters and uptake rates were determined every second day, except for
154 dissolved organic phosphorus (DOP) components, which were measured every 4 days.
155 Termination of the measurements varied due to logistical constrains. Thus, total phosphorus (TP)
156 and DOP were sampled only until day 29 whereas other parameters were sampled until day 43.

157 The collected water was filled in HCl-cleaned polyethylene canisters that had been pre-rinsed
158 with sample water. All containers were stored in the dark. Back on land, subsamples were
159 processed immediately for each P-analysis. The other analyses were carried out within a few
160 hours of sample collection and sample storage in a climate room at *in situ* temperature.

161

162 **2.3 Analytical methods**

163 **2.3.1 Temperature, salinity, and carbonate chemistry**

164 Measurements in the fjord and in each mesocosm were conducted using a CTD60M memory
165 probe (Sea and sun technology, Trappenkamp, Germany) lowered from the surface to a depth of
166 17 m at about 0.3 m s⁻¹ in the early afternoon (1:30 - 2:30 pm). For these parameters, depth-
167 integrated mean values are presented here.

168 The carbonate system is described in detail in Paul et al. (2015b). The pHT(total scale) was
169 determined using a spectrophotometric method (Dickson et al., 2007) on a Cary 100 (Varian)
170 and the dye m-cresol as indicator. Extinction was measured at 578 nm (E1) and 434 nm (E2) in a
171 10-cm cuvette. The pH was calculated from the ratio of E1 and E2 (Clayton and Byrne, 1993).

172 DIC was measured using a coulometric AIRICA system (MARIANDA, Kiel) measuring the
173 infrared absorption after N₂ purging of the sample and calibration with certified reference
174 material (CRM; Dr. A. Dickson, University of California, San Diego).

175 The *f*CO₂ was calculated from DIC, pHT, salinity and using the stoichiometric equilibrium
176 constant for carbonic acid of Mehrbach et al., (1973) as refitted by Lueker et al., (2000).

177

178 **2.3.2 Chlorophyll and inorganic nutrients**

179 Subsamples of 500 ml were filtered onto GF/F-filters. Chl a was extracted in acetone (90 %) in
180 plastic vials by homogenisation of the filters for 5 min. in a cell mill using glass beads. After
181 centrifugation (10 min., 800 x g, 4°C) the supernatant was analysed on a fluorometer (TURNER
182 10-AU) at an excitation of 450 nm and an emission of 670 nm to determine Chl a concentrations
183 (Jeffrey and Welschmeyer, 1997).

184

185 A segmented continuous-flow analyzer coupled with a liquid-waveguide capillary flow-cell
186 (LWCC) of 2 m length was used to determine phosphate (PO $_4$) and the sum of nitrite and nitrate
187 (NO $_2$ +NO $_3$) at nanomolar precision (Patey et al., 2008). The PO $_4$ determination was based on the
188 molybdenum blue method of Murphy and Riley (1962), and NO $_3$ +NO $_2$ on the method of Morris
189 and Riley (1963). PO $_4$ concentrations from the same subsample were also measured manually
190 using a 5-cm cuvette (Grasshoff et al., 1983). In most of the samplings PO $_4$ data obtained from
191 both methods did not differ significantly (paired t-test: p=0.262, t=1.127, n=109).

192

193 **2.3.3 Dissolved organic phosphorus (DOP)**

194 For the determination of DOP, duplicate 40-ml subsamples were filtered through pre-combusted
195 (6 h, 450 °C) glass fiber filters (Whatman GF/F) and stored in 50-ml vials (Falkon) at -20°C until
196 further processing. The thawed samples were oxidized in a microwave (MARSXpress, CEM,
197 Matthews, USA)(Johnes and Heathwaite, 1992) after the addition of potassium peroxydisulfate
198 in an alkaline medium (Bhaya et al., 2000). The P concentration, measured as PO $_4$ in a 10-cm
199 cuvette, represents the total dissolved phosphorus (DP) concentration. DOP was calculated as the
200 difference between the DP concentrations in the filtered and digested samples and the
201 corresponding PO $_4$ concentration analyzed as described above.

202

203 **2.3.4 Dissolved organic phosphorus compounds**

204 For all analyzed components, subsamples were pre-filtered through pre-combusted (6 h, 450°C)
205 filters (Whatman GF/F) to remove larger particles followed by filtration through 0.2- μ m
206 cellulose acetate filters to remove picoplankton. Subsamples were prepared for storage according
207 to the specific method used for each compound. After the analyses, the phosphorus content of
208 measured DOP compounds was summed and the amount subtracted from the total DOP
209 concentration. The difference is defined as the uncharacterized DOP.

210

211 Dissolved ATP

212 The method of (Björkman and Karl, 2001) adapted to Baltic Sea conditions (Unger et al., 2013),
213 was used to determine dissolved adenosine triphosphate (dATP). A $\text{Mg}(\text{OH})_2$ precipitate,
214 including the co-precipitated nucleotides, was obtained by treating 200 ml of the filtrate with 2
215 ml of 1 M NaOH (1% v/v). The precipitate was allowed to settle overnight and then centrifuged
216 at 1000 g for 15 min. The supernatant was discarded and the precipitate was transferred into 50-
217 ml Falcon tubes, centrifuged again (1.5 h, 1680 x g). The resulting pellet was dissolved by drop-
218 wise addition of 5 M HCl. The samples were frozen at -20°C until further processing. The pH of
219 the thawed samples was adjusted to 7.2 by the addition of TRIS buffer (pH 7.4, 20 mM). The
220 final volume was recorded. The dATP concentrations were measured in triplicate using the
221 firefly bioluminescence assay and a Sirius luminometer (Berthold Detection Systems Pforzheim,
222 Germany), as described by Unger et al. (2013). Standard concentrations were prepared as
223 described above, using aged Baltic Sea water and six ATP concentrations (adenosine 50-
224 triphosphate disodium salt hydrate, Sigma-Aldrich, A2383) ranging from 1 to 20 nmol l^{-1} . The
225 detection limit of the bioluminescence assay was 2.5 nmol l^{-1} . The fluorescence slope of the
226 standard concentrations was used to calculate dATP concentrations, correcting for the final
227 sample volume. The P-content of the dATP (dATP-P) was calculated by assuming that 1 mol of
228 ATP is equivalent to 3 mol P.

229

230 Dissolved phospholipids

231 The phosphate content of the dissolved phospholipids (PL-P) was analyzed using a modified
232 method of Suzumura and Ingall (2001, 2004). Briefly, 400-ml subsamples of the filtrate were
233 stored at -20°C until further processing. The samples were then thawed in a water bath at 30°C
234 and extracted twice with 100 ml of chloroform. The chloroform phase was collected,
235 concentrated to 5 ml in a rotary evaporator (Heidolph Instruments, Schwabach, Germany), and
236 then transferred into microwave tubes. The chloroform was completely evaporated by incubating
237 the tubes in a 60°C water bath overnight. After the addition of 20 ml of deionized water (Milli-Q,
238 Millipore), the samples were digested with potassium peroxydisulfate in alkaline medium and
239 microwaved as described for the DOP analysis. Six standard concentrations of phospholipids,
240 ranging from 0 to 125 $\mu\text{g l}^{-1}$, were prepared by adding the respective amounts of a stock solution
241 containing 5 mg of L-phosphatidyl-DL-glycerol sodium salt (PG, Sigma Aldrich, P8318) ml^{-1} to
242 the aged seawater. The detection limit was 0.8 nmol l^{-1} . The blanks contained only chloroform
243 and were processed as for the samples.

244

245 Dissolved DNA and RNA

246 Dissolved DNA and RNA (dDNA and dRNA) concentrations were determined according to Karl
247 and Bailiff (1989) and as described by Unger et al. (2013). For each sample, 200 ml of the
248 filtrate was gently mixed with the same volume of ethylene-diamine-tetracetic acid (EDTA, 0.1
249 M, pH 9.3, Merck, 1.08454) and 4 ml of cetyltrimethyl-ammonium bromide (CTAB, Sigma-
250 Aldrich, H5882) and stored frozen at -20°C for at least 24 h. After thawing the samples, the
251 precipitate was collected onto combusted (450°C , 6 h) glass fiber filters (25 mm, GF/F
252 Whatman), placed into annealed vials, and stored frozen at -80°C until further analysis.

253 DNA concentrations were measured using a fluorescence-spectrophotometer (Hitachi F 2000),
254 and RNA concentrations using a dual-beam UV/VIS-spectrophotometer U3010 (Hitachi).

255 Coupled standards (DNA + RNA) containing 1–10 μg DNA (Sigma Aldrich, D3779) l^{-1} and 20–
256 120 μg RNA (Sigma Aldrich, R1753) l^{-1} were prepared in aged seawater as described above. A
257 reagent blank served as the reference and aged seawater as the background control. The P-
258 contents of the DNA and RNA were calculated by multiplying the measured values by a factor
259 of 2.06 nmol P per μg dDNA and 2.55 nmol P per μg dRNA. The latter values were determined
260 by the microwave digestion of standard substrates.

261

262

263 **2.3.5 Particulate organic phosphorus, carbon, and nitrogen**

264 Particulate phosphorus (PP) was analyzed using two methods in parallel. In the “aqueous
265 method”, 40 ml of unfiltered subsamples were frozen at -20°C and analyzed as described for
266 DOP using the potassium peroxydisulfate digestion (Grasshoff et al., 1983). The measured PO_4
267 concentration represents total phosphorus (TP). PP is the difference between the total PO_4
268 concentration in the unfiltered digested sample and the sum of DOP+ PO_4 . In the “filter-method”,
269 500 ml subsamples were filtered onto pre-combusted GF/F-filters that were then placed into
270 Schott bottles containing 40 ml of deionised water. PP was digested to PO_4 by the addition of
271 oxidizing decomposition reagent (Oxisolv®, Merck) followed by heating in a pressure cooker
272 for 30 min. The PO_4 concentrations of the cooled samples were determined
273 spectrophotometrically according to Grasshoff et al. (1983). Paired t-test revealed significant
274 differences between two methods; however, the difference between the means of the filter
275 method and of the aqueous method ($0.19 \pm 0.03\text{mmol l}^{-1}$ and $0.16 \pm 0.04\mu\text{mol l}^{-1}$, respectively)
276 were near the detection limit ($0.02 \mu\text{mol l}^{-1}$) of the methods. Thus, solely the mean values
277 obtained from both measurements are used in the following.

278 Particulate carbon (PC) and nitrogen (PN) were analyzed by filtering 500 ml samples onto pre-
279 combusted (450°C, 6 h) glass fiber filters (Whatman GF/F), which were then stored frozen at
280 -20°C. PC and PN concentrations were measured by flash combustion of the dried (60°C) filters
281 using an EuroEA elemental analyser coupled with a Conflo II interface to a Finnigan Delta^{Plus}
282 mass spectrometer and include organic and inorganic matter.

283

284 **2.3.6 Phosphate and ATP uptake**

285 PO₄ uptake was measured by addition of radioactively labeled phosphate [³³P]PO₄ (specific
286 activity of 111 TBq mmol⁻¹, Hartmann Analytic GmbH) at concentrations of 50 pmol l⁻¹ to 50
287 ml subsamples, which were then incubated under laboratory light and the *in situ* temperatures for
288 ~2 h. For each mesocosm, three parallel samples and a blank were prepared. The blank was
289 obtained by the addition of formaldehyde (1% final concentration) 10 min before radiotracer
290 addition, in order to poison the samples. At defined time intervals within the incubation, 5-ml
291 subsamples were taken from each of the parallel samples and filtered onto polycarbonate (PC)
292 filters pre-soaked with a cold 20 mM PO₄ solution to prevent non-specific [³³P]PO₄ binding. The
293 filters were rinsed with 5 times 1 ml of particle-free bay water and placed in 6-ml scintillation
294 vials. Scintillation liquid (4 ml IrgaSafe; Perkin Elmer) was added and the contents of the vials
295 were mixed using a vortex mixer. After allowing the samples to stand for at least 2 h, the
296 radioactivity on the filters was counted in a Perkin Elmer scintillation counter. PC-filters of 0.2
297 µm and 3 µm pore sizes (Whatman and Millipore, respectively) were used to determine uptake
298 by the whole plankton community and the size fraction >3µm, respectively. Picoplankton uptake
299 was calculated as the difference between the activity on the 0.2-µm and 3-µm filters.

300
301 [³³P]ATP (specific activity of 111 TBq mmol⁻¹, Hartmann Analytic GmbH) was added to
302 triplicate 10-ml samples and a blank, each in a 20-ml vial, at a concentration of 50 pmol l⁻¹. The
303 samples were incubated in the dark at the *in situ* temperature for 1 h. The uptake was stopped by
304 addition of 200 µl of a cold 20 mM ATP solution to the samples, which were then filtered and
305 processed as described for the PO₄ uptake measurements.

306

307 **2.3.7 Bacterial production (BPP)**

308 Rates of bacterial protein production (BPP) were determined by incorporation of ¹⁴[C]-leucine
309 (¹⁴C-Leu, Simon and Azam, 1989) according to Grossart et al. (2006). Triplicates and a
310 formalin-killed control were incubated with ¹⁴C-Leu (7.9 GBq mmol⁻¹; Hartmann Analytic
311 GmbH, Germany) at a final concentration of 165 nmol l⁻¹, which ensured saturation of uptake

312 systems of both free and particle-associated bacteria. Incubation was performed in the dark at *in*
313 *situ* temperature (between 7.8°C and 15.8°C) for 1.5 h. After fixation with 2% formalin, samples
314 were filtered onto 5.0 µm (attached) nitrocellulose filters (Sartorius, Germany) and extracted
315 with ice-cold 5% trichloroacetic acid (TCA) for 5 min. Thereafter, filters were rinsed twice with
316 ice-cold 5% TCA, once with ethanol (96% v/v), and dissolved with ethylacetate for measurement
317 by liquid scintillation counting. Afterwards the collected filtrate was filtered on 0.2 µm (free-
318 living) nitrocellulose filters (Sartorius, Germany) and processed in the same way as the 5.0 µm
319 filters. Standard deviation of triplicate measurements was usually <15%. The sum of both
320 fractions (free-living bacteria and attached bacteria) is referred to total BPP. The amount of
321 incorporated ¹⁴C-Leu was converted into BPP by using an intracellular isotope dilution factor of
322 2 (Simon and Rosenstock, 1992). A conversion factor of 0.86 was used to convert the protein
323 produced into carbon (Simon and Azam, 1989).

324

325 **2.4 Statistical analyses**

326 The Grubbs test, done online (graphpad.com/quickcalcs/Grubbs1.cfm) was applied to identify
327 outliers in all data sets. The outliers were removed from further statistical analyses.

328 Spearman Rank correlations were carried out to describe the relationship between the
329 development of the parameters over time in the mesocosms and in the fjord using Statistica 6
330 software.

331 Short-term CO₂ effects on PP concentrations at days 0-2 and 23-43 between the CO₂ treatments
332 were verified with an ANCOVA analysis using the SPSS software. The “days” were treated as a
333 covariate interacting with the treatments. Paired t-test was applied to check the differences in
334 PO₄ concentrations between the treatments.

335

336

337 **3. Results**

338 **3.1 Development in the mesocosms**

339 **3.1.1 CO₂, pH, temperature and salinity**

340 The different mesocosms were characterized based on their averaged *f*CO₂ and pH values from
341 day 1 until day 43 (Fig.2a, b):

342 M1 365 µatm *f*CO₂, pH 8.08

343 M5 368 µatm *f*CO₂, pH 8.07

344 M7 497 µatm *f*CO₂, pH 7.95

345 M6 821 µatm *f*CO₂, pH 7.74

346 M3 1007 $\mu\text{atm } f\text{CO}_2$, pH 7.66

347 M8 1231 $\mu\text{atm } f\text{CO}_2$, pH 7.58

348 M1 and M5 were the untreated mesocosms and served as controls.

349 Temperature development in the mesocosms closely followed that in the fjord ranging from
350 7.82°C to 15.86°C. Based on this (compare Paul et al. 2015b for details), the experiment was
351 divided into four phases (Fig. 3): phase 0: day -3 to day 0; phase I: days 1–16, phase II: days 17–
352 30 and phase III: day 31 until the end of the measurements. Temperature dropped from 8.71°C to
353 7.82°C in phase 0 and rose from 8.07°C at the start of phase I to the maximum of 15.86°C by the
354 end of this phase. During phase II, the temperature decreased to 7.89°C interrupted by a short
355 reversal on days 22 and 23. During phase III, the temperature increased to 12.61°C (Table 1).

356 Salinity (5.69 ± 0.01) remained relatively stable in all mesocosms throughout the entire
357 experimental period (Fig. 3).

358

359 **3.1.2 Phytoplankton biomass**

360 Chlorophyll *a* (*Chla*) reached maximum concentrations of 2.06–2.48 $\mu\text{g } \Gamma^{-1}$ at day 5 (Fig. 4).
361 Average concentrations of $1.94\pm 0.23 \mu\text{g } \Gamma^{-1}$ in phase I exceeded those in phases II and III when
362 *Chla* decreased to a mean of $1.08\pm 0.16 \mu\text{g } \Gamma^{-1}$. The increase in *Chla* in the high CO_2 mesocosms
363 by $0.27\mu\text{g } \Gamma^{-1}$ in phase III was marginal for Baltic Sea summer conditions. According to Paul et
364 al. (2015b), this represents an increase of 24% which is a significant difference compared to the
365 controls.

366 We observed a significant relationship between *Chla* and PO_4 in the untreated and intermediate
367 treated mesocosms that diminished with increasing $f\text{CO}_2$ as indicated by lower p-values. The
368 statistical significance was lost in the highest $f\text{CO}_2$ mesocosms (Table 2).

369

370 **3.1.3 Phosphorus Pools**

371 **Total phosphorus (TP)** concentrations in the mesocosms ranged between 0.49 and 0.68 μmol
372 Γ^{-1} (Fig. 5a) during the experiment without statistically significant differences between the CO_2
373 treatments. Shortly after the bags were closed, the decline in TP concentrations began and
374 continued until the beginning of phase II. On average, TP concentrations decreased from
375 $0.63\pm 0.02 \mu\text{mol } \Gamma^{-1}$ on day -3 to $0.51\pm 0.01 \mu\text{mol } \Gamma^{-1}$ on day 21. Thereafter, the mean TP
376 remained constant at $0.54\pm 0.03 \mu\text{mol } \Gamma^{-1}$ until the end of the measurements. Thus, the loss of
377 phosphorus ($116 \pm 34 \text{ nmol } \Gamma^{-1}$) from the 17-m layer during the 29-day measurement period was
378 calculated to be $4.0 \text{ nmol } \Gamma^{-1} \text{ day}^{-1}$. The decline in TP can be explained by loss through
379 sedimentation of PP (Paul et al., 2015b).

380
381 **Particulate phosphorus (PP)** concentrations varied from 0.10 to 0.23 $\mu\text{mol l}^{-1}$ in all CO_2
382 treatments (Fig. 5b, Fig. 6). We expected that the decrease in TP was reflected in PP. However,
383 parallel changes occurred only periodically. PP concentrations increased during the first 5 days
384 after the bags were closed. This increase was stimulated by CO_2 additions from day 0 to day 2
385 (ANCOVA: $p=0.004$, $F=20.811$) (Fig. 7a). Subsequently, PP declined in parallel with TP until
386 day 21, albeit with a lower amount. Averaged over all mesocosms, TP decreased by 0.12 ± 0.03
387 $\mu\text{mol l}^{-1}$, whereas PP declined only by $0.06 \pm 0.01 \mu\text{mol l}^{-1}$ during this period. From day 23 until
388 the end of the measurements, PP remained at relatively constant concentrations; however, PP
389 concentrations in the high CO_2 treated mesocosms exceeded those in the other mesocosms
390 significantly (ANCOVA: $p<0.0001$, $F=11.99$) (Figs. 5b, 7). PP developed in parallel with PC.
391 The two parameters were positively correlated in the untreated and the intermediate CO_2
392 treatments, but not in the high CO_2 treatments (Table 2). Figures 3 and 6b show that the increase
393 in Chl*a* was delayed by 2–3 days compared to the increase in PP during the first growth event. A
394 correlation between PP and Chl*a* was detected only for the untreated mesocosms (Table 2).

395
396 **Dissolved organic phosphorus (DOP)** concentrations in the mesocosms ranged between 0.18
397 and 0.36 $\mu\text{mol l}^{-1}$ constituting 32–71% of the TP pool (Fig. 5c). DOP did not change
398 significantly in response to the CO_2 perturbations, and were similar to the concentrations in fjord
399 water. Concentrations of $\geq 0.3 \mu\text{mol l}^{-1}$ were measured on days 6 and 7 (phase I) and on day 23
400 (phase II); the high DOP value in the intermediate CO_2 treatment at day 19 was identified as an
401 outlier according to Grubbs test (Fig. 5c).

402 In phase I, DOP initially increased in parallel with Chl*a* and BPP but reached its maximum 1–2
403 days later, after which it decreased only marginally until the end of this phase, independent of
404 changes in BPP and Chl*a* (Fig. 8c, d). In phase II, the peak conformed to that of BPP. DOP
405 correlated with temperature only in the high $f\text{CO}_2$ mesocosms (Table 2). In addition, the
406 composition of DOP did not change with increasing CO_2 (Fig.10). The sum of RNA (~47%) plus
407 the unidentified fraction constituted 98–99% of the DOP pool whereas the other measured
408 compounds contributed only 1–2% (Table 3).

409
410 **Phosphate (PO_4)** concentrations ranged between 0.06 and 0.21 $\mu\text{mol l}^{-1}$, with deviations
411 between the mesocosms only in nanomolar range. The mean contribution of PO_4 to TP was
412 $25 \pm 6\%$, which was the lowest among all TP fractions (Fig. 6). From the start of the
413 measurements to day 13, PO_4 declined by $0.06 \mu\text{mol l}^{-1}$ (or $3.5 \text{ nmol l}^{-1} \text{ day}^{-1}$) from initial

414 values of $0.16 \pm 0.01 \mu\text{mol l}^{-1}$ (Fig. 5d). Subsequently, concentrations increased again, by an
415 average of $2.6 \text{ nmol l}^{-1} \text{ day}^{-1}$, until the end of the experiment. There were no significant
416 differences between CO_2 treatments until day 23, when high CO_2 concentrations led to slightly
417 lower PO_4 concentrations (Fig. 5d). Afterwards, PO_4 concentrations in the high $f\text{CO}_2$ mesocosms
418 were significantly lower than those in the untreated mesocosms ($t=6.51$, $p=0.0003$). This
419 observation is in accordance with the dynamics of PP and Chl*a* concentrations, which were
420 significantly elevated in the high CO_2 treatments. Thus, the transformation of PO_4 to POP via
421 stimulated biomass formation may have been promoted under high CO_2 conditions in phase III.
422 Since PO_4 was never fully exhausted, phosphorus limitation of phyto- and bacterioplankton can
423 be excluded. This interpretation is supported by the PC:PP ratios, which varied between 84.4 and
424 161.1 in all treatments (Paul et al., 2015b) deviating only slightly from the Redfield ratio.

425

426 3.1.4 Uptake of PO_4 and ATP

427 PO_4 turnover times of 1.5–8.4 days (mean 4.0 ± 1.2 days, $n=112$) in all mesocosms indicated no
428 dependency on the CO_2 treatment (Fig. 9a). Gross PO_4 uptake rates were in the range of 0.6–3.9
429 $\text{nmol l}^{-1} \text{ h}^{-1}$ (mean $1.7 \pm 0.6 \text{ nmol l}^{-1} \text{ h}^{-1}$, $n=112$), or 14.3–94.4 $\text{nmol l}^{-1} \text{ day}^{-1}$ (mean 41.3 ± 13.8
430 $\text{nmol l}^{-1} \text{ day}^{-1}$) (Fig. 9b, Table 4). The rates were highest on days 4 and 9 (phase I) and
431 decreased thereafter until day 15, followed by an increase to a mean maximum rate of 2.3 ± 0.5
432 $\text{nmol l}^{-1} \text{ h}^{-1}$ ($n=6$) at day 27. The size fraction $<3 \mu\text{m}$ was responsible for 59.1 to 98.4% of the
433 total PO_4 uptake (mean $86.5 \pm 7.6\%$) whereas the size fraction $>3 \mu\text{m}$ accounted for only 1.6–
434 40.9% (mean $13.5 \pm 7.4\%$). Thus, PO_4 was taken up mainly by picoplankton. However, only the
435 uptake rate by the size fraction $>3 \mu\text{m}$ was positively related to Chl*a* and inversely related to the
436 P content of the biomass (Table 2). Thus the PO_4 uptake was obviously stimulated when the
437 phytoplankton biomass increased and at simultaneous decrease of the cellular P. The relationship
438 between PO_4 uptake by this fraction and Chl*a* became evident only in the CO_2 -amended
439 conditions indicating that the interaction between P uptake, cellular P-content and growth of
440 phytoplankton was stimulated under elevated CO_2 conditions.

441 ATP turnover times of 0.2 to 3.6 days (mean 0.94 ± 0.74 days, $n=90$) were much shorter than the
442 PO_4 turnover times and did not vary between the treatments (Fig. 9c). Between 0.05 and 0.36
443 $\text{nmol ATP l}^{-1} \text{ h}^{-1}$ (mean $0.14 \pm 0.08 \text{ nmol l}^{-1} \text{ h}^{-1}$, $n=36$) were hydrolysed, corresponding to a P
444 supply of 0.14 and 1.08 $\text{nmol l}^{-1} \text{ h}^{-1}$ (mean $0.44 \pm 0.25 \text{ nmol l}^{-1} \text{ h}^{-1}$, $n=36$). Thus, phosphorus
445 additionally supplied from ATP accounted for ~25% of that provided by PO_4 . The picoplankton
446 size fraction ($<3 \mu\text{m}$) was responsible for 90–99% of ATP uptake, with only a marginal portion
447 (1.6–9.5%) attributable to the phytoplankton fraction $>3 \mu\text{m}$ (Table 4).

448

449 **3.2 Hydrography and pool sizes in the fjord**

450 Large variations in $f\text{CO}_2$ and pH occurred in fjord water during the period of investigation (Table
451 1). The relationship of $f\text{CO}_2$ with temperature and salinity indicated that the CO_2 conditions were
452 influenced predominantly by changes in the water masses, specifically by upwelling which
453 affected both the relationship of $f\text{CO}_2$ with PO_4 and probably the correlation of $f\text{CO}_2$ with $\text{Chl}a$
454 and PC (Table 2). $f\text{CO}_2$ ranged from 207 μatm (Fig. 2a) at days 12-16 when temperatures were
455 highest to 800 μatm at day 33 when deep water input occurred which was indicated by pH below
456 7.75.

457

458 **Chl***a* concentrations were between 1.12 and 5.46 $\mu\text{g l}^{-1}$ (mean $2.29 \pm 1.11 \mu\text{g l}^{-1}$; $n=38$), with
459 distinct phases correlating with temperature, salinity and pH. However, the *Chl**a* maximum
460 occurred at the beginning of phase II, which was 1-2 days after the maximum temperature.
461 Shortly thereafter, *Chl**a* decreased to its lowest level before it increased again, albeit only
462 marginally to 1.93 $\mu\text{g l}^{-1}$ during phase III (Fig. 4).

463

464 **TP** concentrations from day -3 until day 29 ranged between 0.54 and 0.70 $\mu\text{mol l}^{-1}$ (mean
465 $0.61 \pm 0.04 \mu\text{mol l}^{-1}$; $n=19$) (Figs. 5a, 6). With a general decreasing tendency, TP undulated with
466 a frequency of about 10 days in the period of phases 0 to the first half of phase I and of 6 days in
467 the second half of phase I to II. For the period under investigation, the TP fractions had the
468 following characteristics:

469 **PP** concentrations varied from 0.13 to 0.30 $\mu\text{mol l}^{-1}$ (mean $0.20 \pm 0.04 \mu\text{mol l}^{-1}$; $n=29$), thus
470 accounting for 23.4–51.8% (mean $34.7 \pm 7.9\%$; $n=19$) of the TP pool. The development of PP
471 over time did not follow that of TP (Fig. 5b). PP concentrations were highest between days 8 and
472 19, when the accumulation of PP in the biomass was reflected in declining C:P ratios from 180
473 to 107 (Paul et al. 2015b) and thereafter remained at the low ratio until the end of the
474 measurements. The PP increase in phase III occurred in parallel to *Chl**a* and to the PO_4 decrease
475 (Fig. 6). Thus PO_4 was transformed into PP via biomass production. The calculated P content of
476 phytoplankton was 0.05–0.15 (mean 0.1) $\mu\text{mol PP} (\mu\text{g Chl}a)^{-1}$.

477 **DOP** substantially contributed (26–45%) to the TP pool (Fig. 6). Concentrations ranged between
478 0.19 and 0.29 $\mu\text{mol l}^{-1}$ (mean $0.24 \pm 0.03 \mu\text{mol l}^{-1}$; $n=17$), with high concentrations occurring in
479 parallel to those of TP in phases I and II (Fig. 5c). The very low DOP value of 0.11 $\mu\text{mol l}^{-1}$, on
480 day 29, was an outlier (Grubbs test). For the whole study period, DOP concentrations correlated
481 positively with PP ($p=0.034$, $n=17$) and inversely with PO_4 concentrations ($p=0.005$, $n=17$). A

482 similar behavior between DOP and Chla was restricted to phases 0 and I, whereas the
483 relationship was inverse in phase II (Fig. 8b). As shown in Figure 8a, the DOP and BPP levels
484 alternated with the same rhythm, but inversely, in phases 0 and I and changed to a parallel
485 development in phase II. Statistical analysis was not feasible because DOP and BPP were not
486 always sampled on the same day and only very few data pairs were available.

487 **PO₄** concentrations ranged between 0.06 and 0.41 $\mu\text{mol l}^{-1}$ (mean $0.21 \pm 0.09 \mu\text{mol l}^{-1}$, $n=21$),
488 thus comprising $24.3 \pm 11.2\%$ ($n=21$) of the TP pool (Fig. 6). With a few exceptions, PO₄
489 concentrations declined from the beginning of the study period until the end of phase I and
490 increased during phase II and the beginning of phase III. These changes were caused by
491 upwelling of PO₄ enriched deep water of higher salinity and lower temperatures. The subsequent
492 decline in PO₄ between days 33 and 40 was caused by the stimulation of phytoplankton
493 production, as indicated by the increase in Chla concentration (Fig. 4).

494

495

496 **4. Discussion**

497 An increase in CO₂ in marine waters and the associated acidification may potentially have
498 multiple effects on organisms and biogeochemical element cycling (Gattuso and Hansson, 2011).
499 Reported findings indicate wide ranging responses, probably depending on the investigated
500 species and growth conditions. For example, CO₂ stimulation as well as lack of stimulation were
501 found for primary production and carbon fixation (Beardall et al., 2009; Boettjer et al., 2014),
502 DOC release (Engel et al., 2014; MacGilchrist et al., 2014) and phytoplankton growth (Riebesell
503 and Tortell, 2011). An interaction of CO₂ effects with phosphorus and iron availability has been
504 found by Sun et al. (2011) for a the diatom *Pseudo-nitzschia multiseriis* and by Yoshimura et al.
505 (2014) for a diatom dominated subarctic plankton community. Thus, responses of organisms and
506 ecosystems to enhanced CO₂ concentrations are complex and still poorly understood. The
507 present study is the first to determine the effects of increased CO₂ levels on phosphorus cycling
508 in a brackish water ecosystem.

509

510 **4.1 Response of P-pools and P-uptake to enhanced CO₂ in the mesocosms**

511 The Finish coast of the Gulf of Finland is one of the most important upwelling regions in the
512 Baltic Sea. During our investigation in 2012, surface temperatures, obtained from the NOAA
513 satellite (Siegel and Gerth, 2013) showed that upwelling persisted during the whole study period
514 but with varying intensity. The intensity of upwelling shaped the pattern of temperature not only

515 in the fjord but also in the mesocosms varying from 7.8 to 15.9°C. Such variations in temperature
516 influence the phosphorus transformation and interleave with CO₂ effects.

517 While nutrients were added in previous CO₂ enrichment experiments (Riebesell et al., 2008;
518 2013; Schulz et al., 2008), no amendments were undertaken in this study in order to be close to
519 natural conditions. Initial PO₄ concentrations of only 0.17 ± 0.01 μmol l⁻¹ were measured,
520 however, PO₄ was never exhausted (Figs. 5, 6). Cellular C:P and N:P ratios were close to the
521 Redfield ratio. Therefore, phosphorus limitation unlikely occurred in this experiment.
522 Simultaneous low nitrate and ammonium concentrations (Paul et al. 2015b) formed nutrient
523 conditions that benefit the growth of diazotrophic cyanobacteria. However, a cyanobacteria
524 bloom failed to appear, despite the low-level presence of *Aphanizomenon* sp. and
525 *Dolichospermum* sp. (Paul et al., 2015a) as potential seed stock. For Baltic Sea summer
526 conditions, the phytoplankton development with maximum Chl_a concentrations of 2.2–2.5 μg l⁻¹
527 remained relatively low with the highest contribution of cryptophytes and chlorophytes in phase
528 I and at the beginning of phase II. Picoplankton was mostly the dominating size fraction,
529 amounting ~20-70% of Chl_a in phase I and up to ~85% in phase III (Paul et al., 2015b).
530 However, a positive correlation of fCO₂ with Chl_a was observed only for the size fraction >20
531 μm. The abundance of diatoms that could be a part of this fraction increased from ~day 23 to day
532 30 and might have an influence on this relationship.

533 Against this background, the CO₂ perturbation did not cause significant changes in phosphorus
534 pool sizes, DOP composition, and P-uptake rates from PO₄ and ATP when the whole study
535 period was considered. However, small but yet significant short-term effects on PO₄ and PP pool
536 sizes were observed in phases I, III and partially in phase II (Fig. 7). CO₂ elevation stimulated
537 the formation of PP until day 3 (Fig. 5b) when chlorophytes, cyanobacteria, prasinophytes and
538 the pico-cyanobacteria started to grow (Paul et al., 2015b).

539 The effects of CO₂ addition on PO₄ and PP pool sizes were evident from day 23 onwards (Figs.
540 5b, 7). PO₄ concentrations were slightly, but significantly lower in the high CO₂ treatment than
541 in the untreated mesocosms, accompanied by significantly elevated PP concentrations. This
542 indicates that the transformation of PO₄ into PP was likely stimulated under high CO₂ conditions.
543 Since Chl_a was also elevated at similar PP:Chl_a ratios, the PO₄ taken up was used for new
544 biomass formation. However, the elevated transformation of PO₄ into PP was not reflected in the
545 PO₄ uptake rates which can be seen as gross uptake rates. But, an increase of PP, caused by
546 biomass formation, while the PO₄ uptake remained unchanged can only occur when the P release
547 from organisms is reduced. Thus, it is likely that not the gross uptake but rather the net uptake
548 was modified under CO₂ elevation.

549 While in phases II and III, high CO₂ levels caused a change in the PP and PO₄ pools for about 22
550 days, changes lasting only 2 days have been observed at the beginning of phase I (Fig.7a), but,
551 shorter effects cannot be excluded. Uptake and release are assumed to be continuous processes
552 and can alter the P pool sizes on timescales shorter than one day. Thus, variations and
553 differences in the treatments can be overseen at daily sampling. Unger et al. (2013) demonstrated
554 that an accelerated PO₄ uptake by the cyanobacterium *Nodularia spumigena* under elevated CO₂
555 incubations could only be observed during the first hours. Thereafter, the differences were
556 balanced and the same level of radiotracer labeling was reached in all treatments. An acceleration
557 in the formation of particulate P under CO₂ elevation without any changes of PO₄ turnover times
558 was also observed by Tanaka et al. (2008). They observed an increase of the PP amount and an
559 earlier appearance of the PP maximum under CO₂ elevation.

560 Correlations calculated by using the Spearman rank test between P pools or uptake rates and
561 other parameters for each mesocosm are presented in Table 2. The relationships between PP and
562 TP with Chl_a disappeared at elevated *f*CO₂, whereas correlations developed between PP and PC
563 as well as between the PO₄ uptake by phytoplankton in the >3 μm size class and the PP:Chl_a
564 ratio (Table 2). These shifts could be caused by changes in the phytoplankton composition
565 deduced from CO₂ effects on the pigment composition (Paul et al., 2015b).

566
567 Independent of the CO₂ treatment, TP decreased by 2.6 nmol l⁻¹ day⁻¹ in all mesocosms over the
568 course of the experiment, in agreement with the measured sedimentation rates (Paul et al.,
569 2015b). The strongest decrease (~3.2 nmol l⁻¹ day⁻¹) occurred during phase I. Of the total TP
570 removal during this phase (48 nmol l⁻¹), 84% (~40.5 nmol l⁻¹) could be explained by the
571 decrease in PP and 16% (~ 8 nmol l⁻¹) by changes in the dissolved pool. However, the PO₄
572 decline (~34.5 nmol l⁻¹) was stronger than that of the total dissolved P pool since DOP increased
573 in parallel by ~26.5 nmol l⁻¹. Thus, about 77% of the PO₄ reduction was retrieved as DOP and
574 remained in the dissolved P-pool as the main pathway of PO₄ transformation.

575 576 **4.2 Phosphorus dynamics in the Storfjärden**

577 Nutrients in upwelled water during our study were depleted in dissolved inorganic nitrogen and
578 enriched in PO₄, as reported for other upwelling areas of the Baltic Sea (Lass et al., 2010). Thus,
579 ammonium and NO₂+NO₃ concentrations in the surface water were only in the nanomolar range
580 (Paul et al., 2015b). PO₄ increased in parallel with the increase in salinity and decrease in
581 temperature. Maximum PO₄ concentrations of 0.33 μmol l⁻¹ and 0.42 μmol l⁻¹ (Figs 5, 6) were
582 observed at the end of the upwelling events in phases 0 and II, respectively. The correlation with

583 *Chla* and PP indicated that PO₄ was utilized during plankton growth in the subsequent relaxation
584 phases I and III. Due to PO₄ input into surface water, the phytoplankton community was unlikely
585 P-limited indicated by PC:PP ratios of 86–189 (mean 125, n=23) (Paul et al., 2015b). However,
586 the PO₄ availability might be not the only reason for the good P-nutritional status of the
587 plankton. It can be deduced from the long PO₄ turnover times in the mesocosms, where external
588 input was excluded, that the P demand of the plankton community might be low. The P content
589 deduced from PP:*Chla* ratios of 0.05–0.15 μmol P (μg *Chla*)⁻¹ was somewhat lower than those
590 observed during an upwelling event along the east coast of Gotland, where ratios between 0.1
591 and 0.2 μmol P (μg *Chla*)⁻¹ (Nausch et al., 2009) were estimated.
592 PP concentrations of 0.13–0.3 μmol l⁻¹ were in the range typically observed in the Baltic Proper
593 (Nausch et al., 2009; Nausch et al., 2012). However, PP concentrations in the Gulf of Finland
594 may reach higher values, as was the case in the summer of 2008, when the observed PP
595 concentration was 0.35 ± 0.07 μmol l⁻¹ (Nausch and Nausch, 2011).
596 DOP concentration of 0.27 ± 0.02 μmol l⁻¹ during our study was similar to that detected in the
597 Gulf of Finland in the summer of 2008 (Nausch and Nausch, 2011). In the Baltic Sea, DOP
598 exhibits vertical gradients with maximum concentrations in the euphotic surface layer and lower
599 than 0.1 μmol l⁻¹ at depths below 25 m. Thus, the observed DOP dynamics in surface water
600 during our study can be assumed to be the result of release, consumption and mineralization by
601 organisms or input from land. The relationship of DOP with *Chla* and BPP (Fig. 8) indicated that
602 the increased DOP concentrations in phase I may be due to release by phytoplankton
603 supplemented by bacterial release. DOP can be accumulated in water only when the release
604 exceeded the consumption or degradation. During phase II, phytoplankton biomass was low and
605 DOP release should thus be minor. Since the small mesozooplankton increased in the fjord
606 similar to those reported for the mesocosms in phases II and III (Paul et al. 2015b) DOP could be
607 released during grazing combined with the observed temporal offset of BPP and DOP maxima.
608 Thus, the observed DOP variations may be the result of processes in the surface water.

609

610

611 **5. Conclusions**

612 Surface water in Storfjärden showed highly variable *f*CO₂ conditions and reached levels of up to
613 800 μatm, which is similar to that expected in ca. 100 years from now. Deduced from the high
614 frequency of upwelling events, organisms experience elevated *f*CO₂ more or less regularly. Thus,
615 a general impact of *f*CO₂ on P pools and P uptake rates in the mesocosms could not be identified
616 for the overall period of investigation. However, short-term responses to *f*CO₂ elevation lasting

617 only few days were observed for the transformation of PO₄ into PP that was linked with
618 stimulation of phytoplankton growth. Although statistically significant, it is difficult to assess if
619 the differences between the treatments are of ecological relevance. Such short-term variations are
620 possible in the phosphorus dynamics since the pools size can be transformed within hours and
621 there changes are in the nanomolar concentration range. There are also indications that
622 relationships of P pool sizes or uptake with Chl*a* and PC can change as *f*CO₂ increases, but the
623 underlying mechanisms are still unclear. The transformation of PO₄ into DOP was not affected
624 by CO₂ elevation. It may be the major pathway of phosphorus cycling under hydrographical and
625 phytoplankton growth conditions as occurred in our experiment.

626

627

628 **Acknowledgement**

629 We are grateful to the KOSMOS team for their invaluable help with the logistics and
630 maintenance of the mesocosms throughout the experiment. In particular, we sincerely thank
631 Andrea Ludwig for organizing and coordinating the campaign and for the daily CTD
632 measurements. We appreciate the assistance of Jehane Ouriqua in the nutrient analysis and that
633 of many other participants who carried out the samplings. We also appreciate the collegial
634 atmosphere during the work and thank everyone who contributed to it. We would also like to
635 acknowledge the staff of the Tvärminne Zoological Station for their hospitality and support, for
636 allowing us to use the experimental facilities, and for providing CTD data for the summers of
637 2008–20011. Finally, we thank Jana Woelk for analysing the phosphorus samples in the IOW.
638 This study was funded by the BMBF project BIOACID II (FKZ 03F06550).

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Tables and figures

Table 1: Minimum, maximum and mean values of hydrographical parameters and $f\text{CO}_2$ for the different phases in the fjord. Temperatures in the mesocosms were identical with those in surrounding fjord water.

phase	min	max	mean
water temperature (°C)			
0	7.82	8.71	8.20
I	9.66	15.86	12.27
II	7.89	14.79	11.68
III	8.35	12.61	10.83
salinity			
0	5.72	5.85	5.78
I	5.46	5.85	5.65
II	5.67	6.04	5.82
III	5.9	6.05	5.98
pH			
0	8.09	8.23	8.16
I	8.11	8.30	8.17
II	7.81	8.30	8.00
III	7.75	7.93	7.83
$f\text{CO}_2$ (μatm)			
0	250	347	298
I	207	336	283
II	208	679	465
III	521	800	668

919 Table 2: Mesocosms in which the Spearman Rank correlation between P-pools or uptake rates and
 920 other parameters was significant. The relationship of PP with TP and Chl α was significant only in the
 921 untreated mesocosms while the correlation to PC was also significant in the mesocosms with
 922 intermediate CO $_2$ levels. DOP was related to temperature only in the high CO $_2$ treatments. Under high
 923 f CO $_2$ conditions, the PO $_4$ uptake in the size fraction >3 μ m correlated with Chl α and the P content of
 924 phytoplankton.

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Relationship between	f CO $_2$ (μ atm)	significant responses		
		r	p	n
PP-TP	365	0.599	0.008	18
	368	0.515	0.029	18
PP - Chl α	365	0.479	0.0130	25
	368	0.584	0.0022	25
PO $_4$ - Chl α	365	-0.832	<0.0001	21
	368	-0.756	0.0011	20
	497	-0.674	0.0008	21
	821	-0.524	0.0147	21
	1007	-0.634	0.0027	20
PP -PC	365	0.542	0.0061	24
	368	0.625	0.0011	24
	497	0.404	0.0490	24
	821	0.551	0.0052	24
DOP -temperature	1007	0.488	0.0470	17
	1231	0.525	0.0310	17
PO $_4$ uptake>3 μ m - Chl α	497	0.743	0.0056	12
	821	0.674	0.0081	14
	1231	0.476	0.0310	14
PO $_4$ uptake>3 μ m - POP/Chl α	497	-0.601	0.0380	12
	821	-0.631	0.0160	14
	1231	-0.626	0.0165	14

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935 Table 3: Contribution of different phosphorus components to DOP in the mesocosms and in the
 936 fjord.
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<i>f</i> CO ₂ (μatm)	contribution to DOP (%)					unidentified P
	ATP-P	PL-P	DNA-P	RNA-P	sum	
Fjord	0.7	0.7	0.04	69.4	70.84	29.16
365	0.7	0.5	0.03	44.1	45.33	54.67
368	0.6	0.5	0.03	46.9	48.03	51.97
497	0.6	0.4	0.04	49.5	50.54	49.46
821	0.6	0.4	0.03	41.8	42.83	57.17
1003	0.8	0.4	0.04	60.1	61.34	38.66
1231	0.5	0.4	0.03	48.6	49.53	50.47

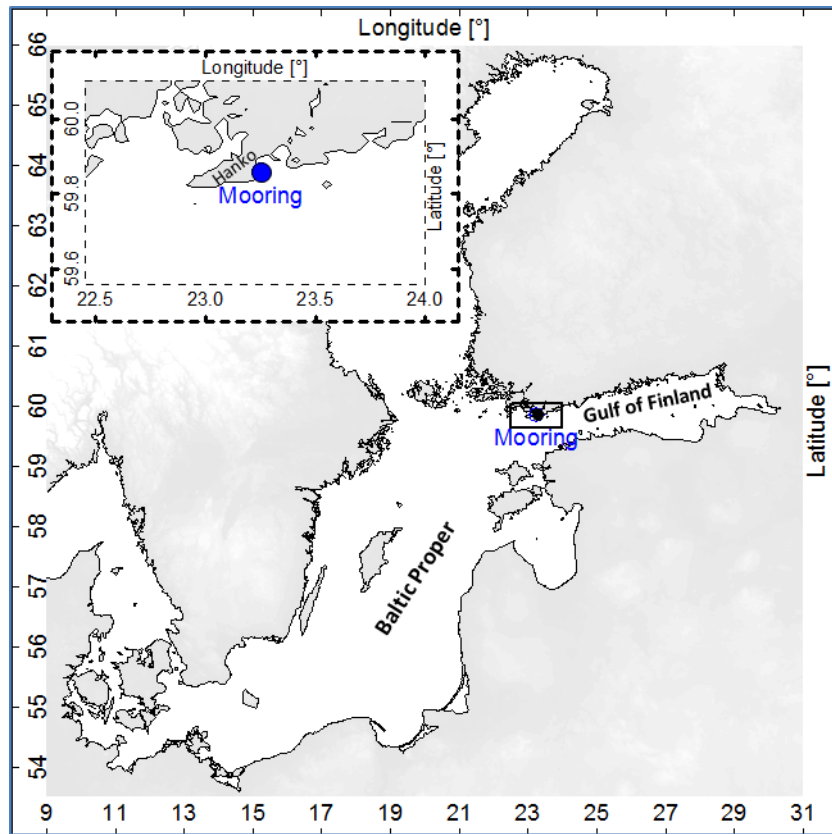
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Table 4: PO₄- and ATP uptake rates in the fjord and in the mesocosms. Minimum, maximum and mean values as well as the contribution of the size fraction <3 μm to the total activity are given for the whole period of investigation (each: n= 16 for PO₄ and n=6 for ATP uptake).

<i>f</i> CO ₂	total PO ₄ uptake (nmoll ⁻¹ h ⁻¹)			portion (%)	total ATP-P uptake (nmoll ⁻¹ h ⁻¹)			portion (%)
	min	max	mean	<3μm	min	max	mean	<3μm
Fjord	0.87	2.81	1.63 ± 0.58	76 ± 15	0.04	0.51	0.26 ± 0.15	92 ± 5
365	0.82	3.89	1.67 ± 0.82	81 ± 11	0.14	1.08	0.43 ± 0.33	96 ± 2
368	0.65	2.74	1.61 ± 0.58	86 ± 7	0.16	0.97	0.47 ± 0.27	96 ± 2
497	0.61	3.03	1.52 ± 0.59	86 ± 6	0.20	1.07	0.54 ± 0.28	96 ± 2
821	0.91	2.83	1.60 ± 0.59	88 ± 8	0.14	0.71	0.36 ± 0.21	97 ± 2
1003	0.67	3.79	1.73 ± 0.85	86 ± 6	0.22	0.69	0.39 ± 0.15	97 ± 1
1231	0.87	2.23	1.53 ± 0.43	87 ± 6	0.17	0.67	0.44 ± 0.17	97 ± 2

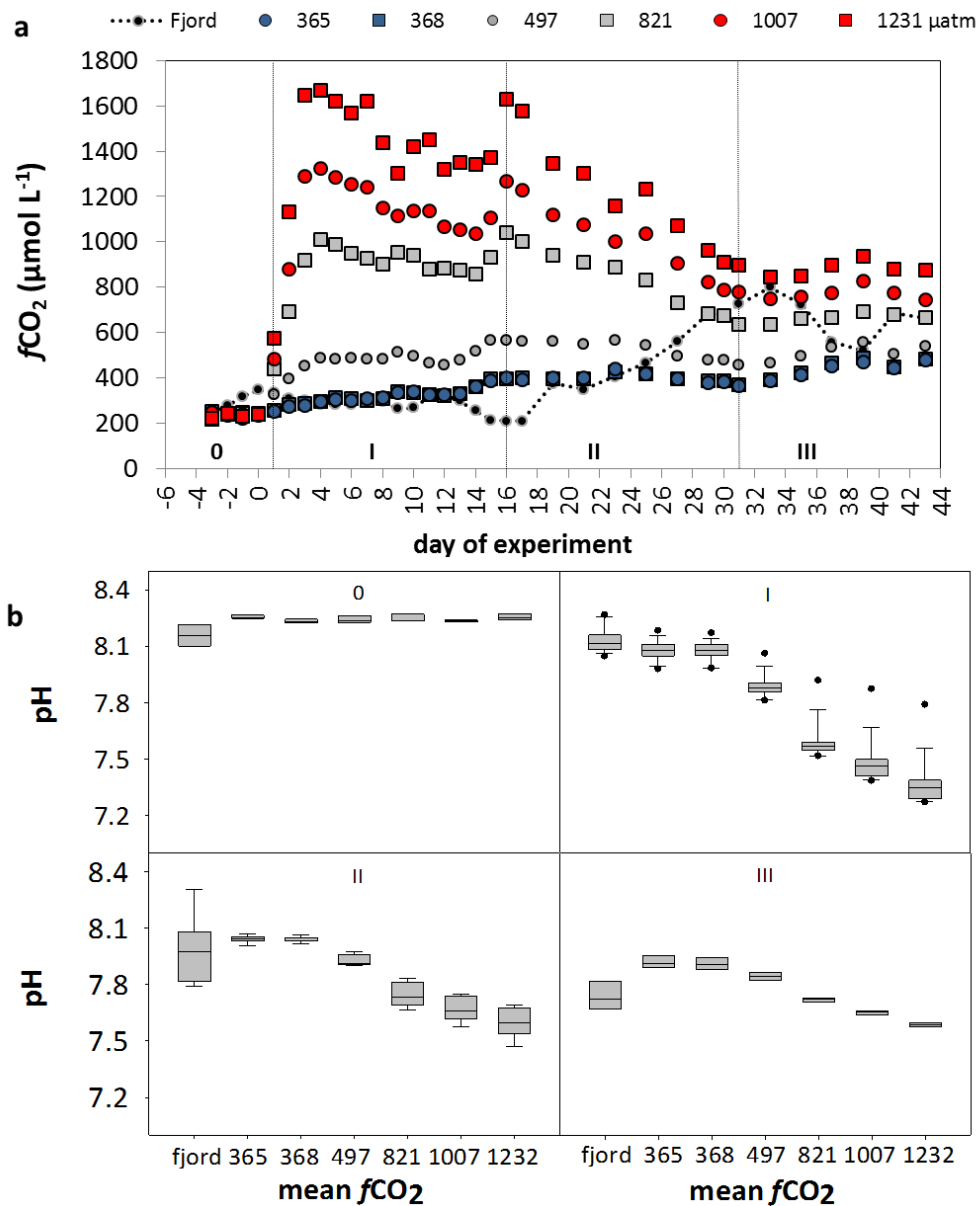
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Figure 1: The Baltic Sea and the location near the peninsula Hanko in the western Gulf of Finland where the mesocosms were deployed

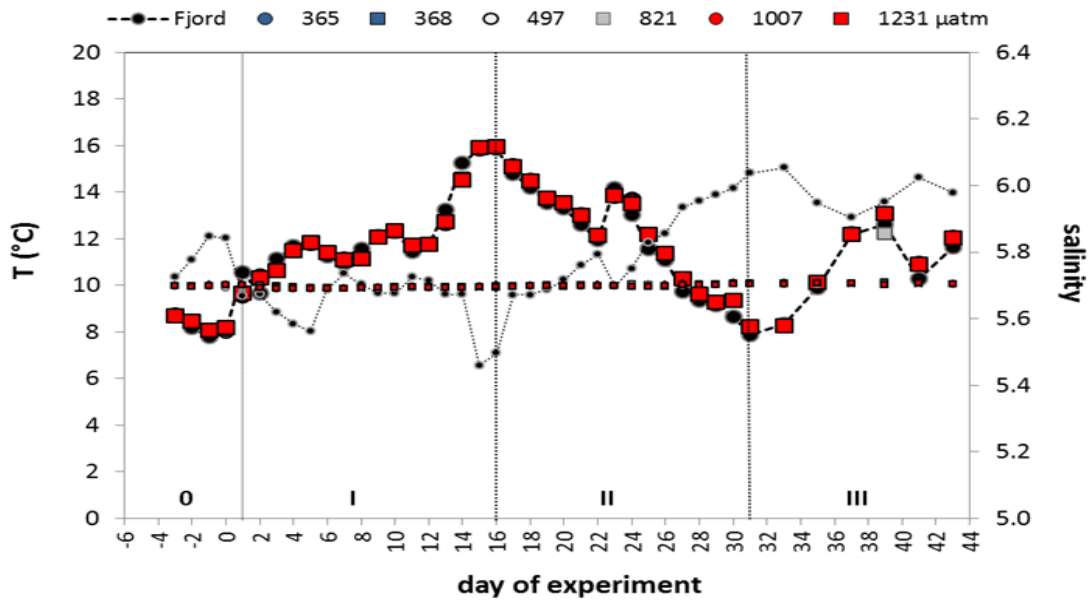
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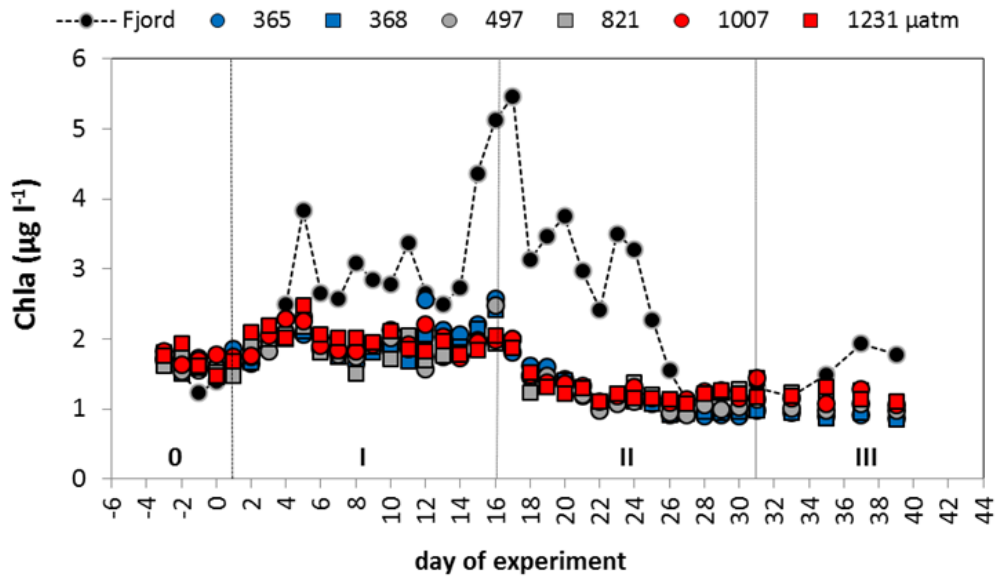
Figure 2a: $f\text{CO}_2$ values in the mesocosms and in the fjord throughout the experiment. Small black dots show the $f\text{CO}_2$ in the ambient fjord water. Treatment of the mesocosms with CO_2 saturated fjord water at the beginning of the experiment (days 0-4) created different $f\text{CO}_2$ levels in the mesocosms: blue symbols represents the untreated mesocosms, grey the intermediate, and red the high CO_2 treated mesocosms. The treatment was repeated at day 16.
Figure 2b: Corresponding pH ranges in the mesocosms during the four phases. Despite decreasing trend over time, a gradient between the mesocosms was kept over the whole period.

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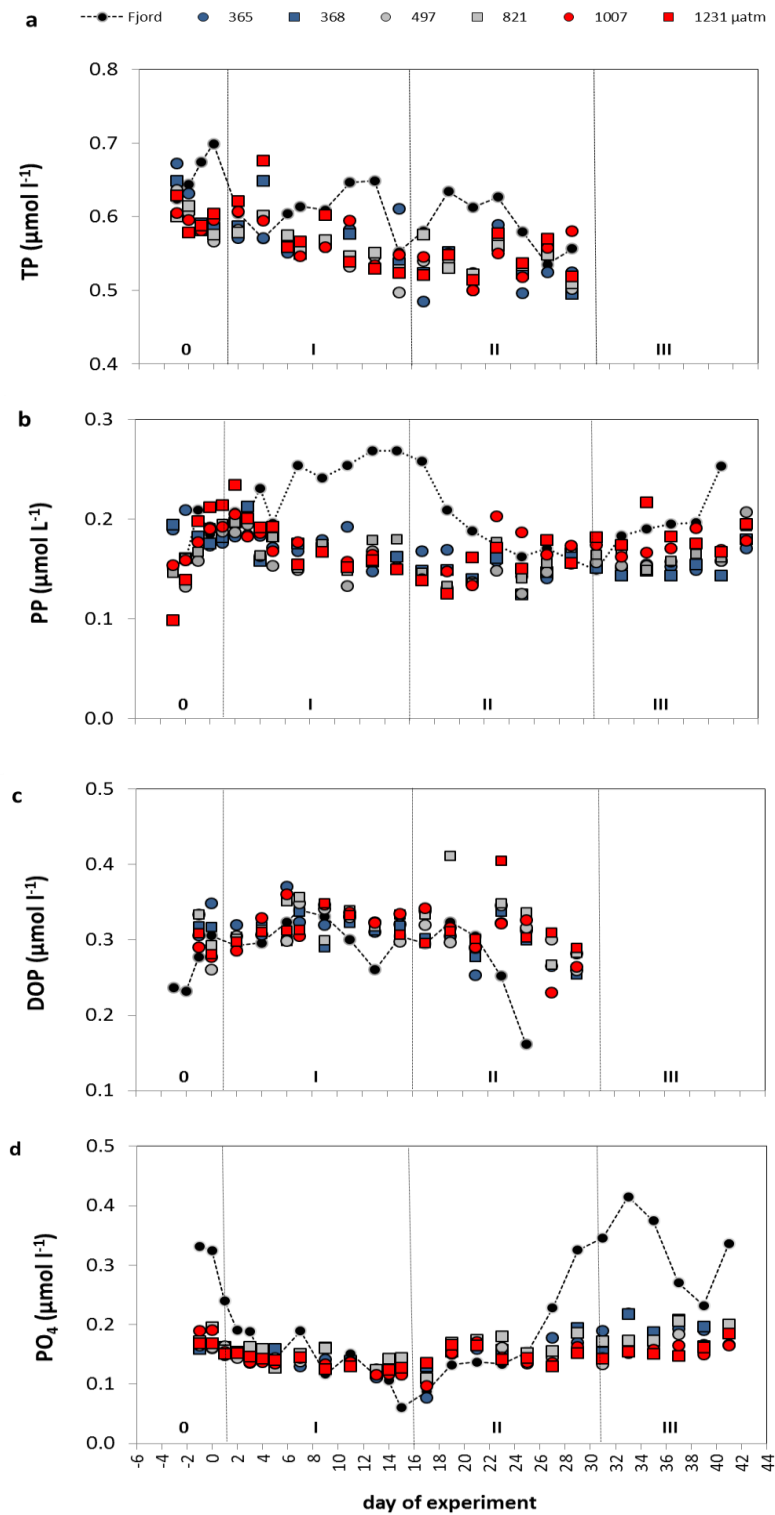
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Figure 3: Temperature and salinity averaged over the 17 m surface layer of the mesocosms and the fjord. The data were obtained from daily CTD casts. Large symbols represent temperature and the small symbols salinity. Fjord water is shown as black dots with broken line while blue symbols denote untreated, grey intermediate and red high $f\text{CO}_2$ levels in the mesocosms. According to the temperature regime, the experimental period can be divided into four phases (phases 0, I, II and III).



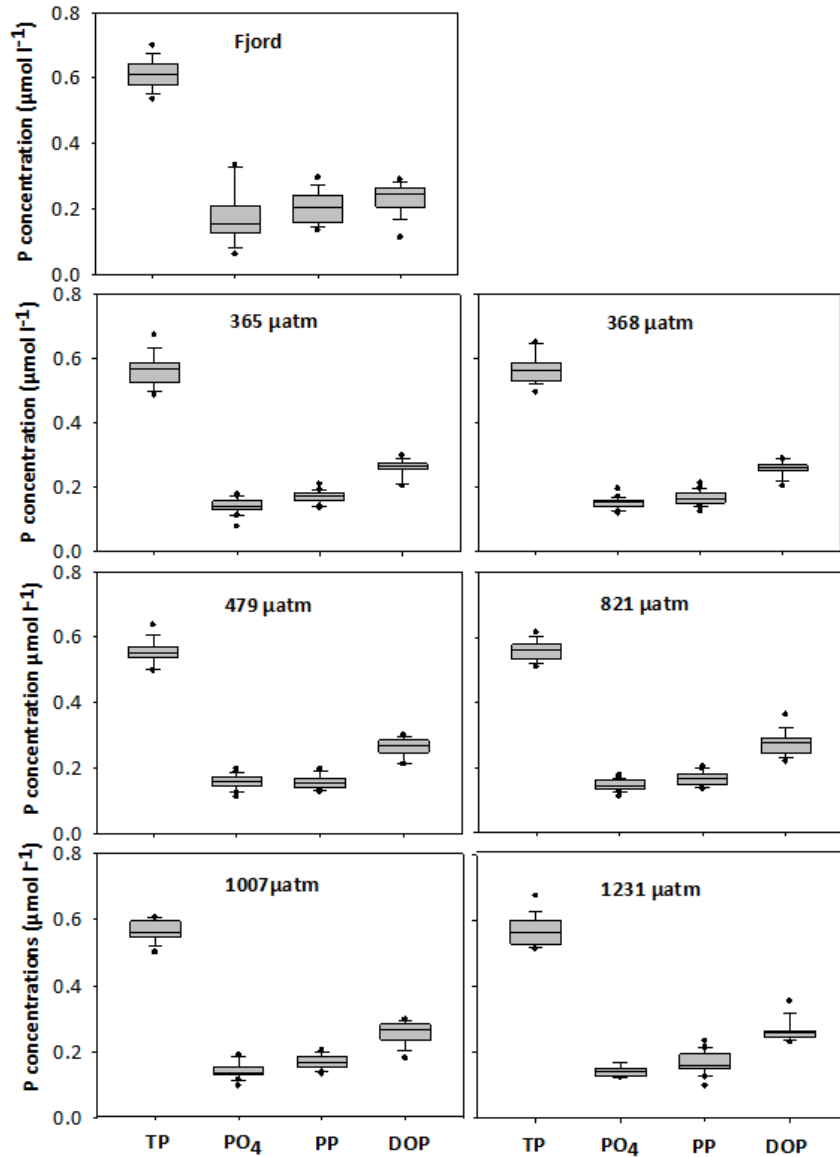
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 1057 Figure 4: Chla concentrations in fjord water and in the mesocosms with different $f\text{CO}_2$
 1058 conditions. The development over time can be divided into three phases as well. Blue represent
 1059 untreated, grey intermediate, and red highly treated $f\text{CO}_2$ levels. Black dots with dotted line are
 1060 the Chla concentrations in the fjord water.

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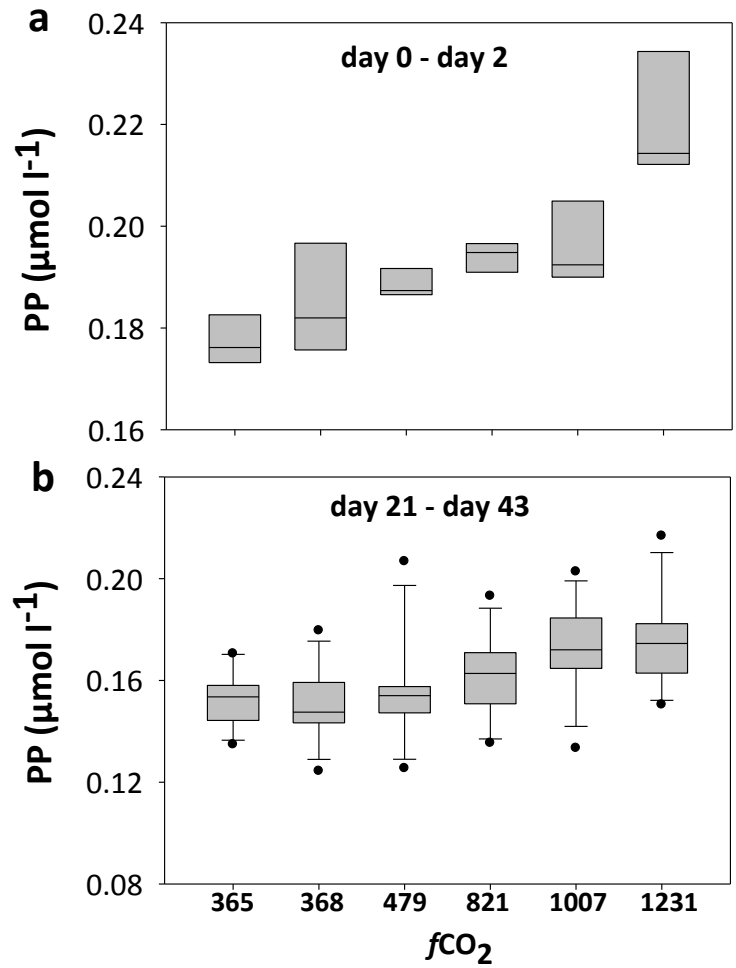
Figure 5a-d: Development of total phosphorus (TP) and the three measured P-fractions in fjord water (black dots with dotted line) and in the mesocosms over time. Blue represents untreated, grey intermediate and red high $f\text{CO}_2$ treatment levels.



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Figure 6: Contribution of the individual P-fractions to TP in fjord water and in the respective mesocosms. The data are averaged for the period when TP measurements were done (day -3 – day 29).

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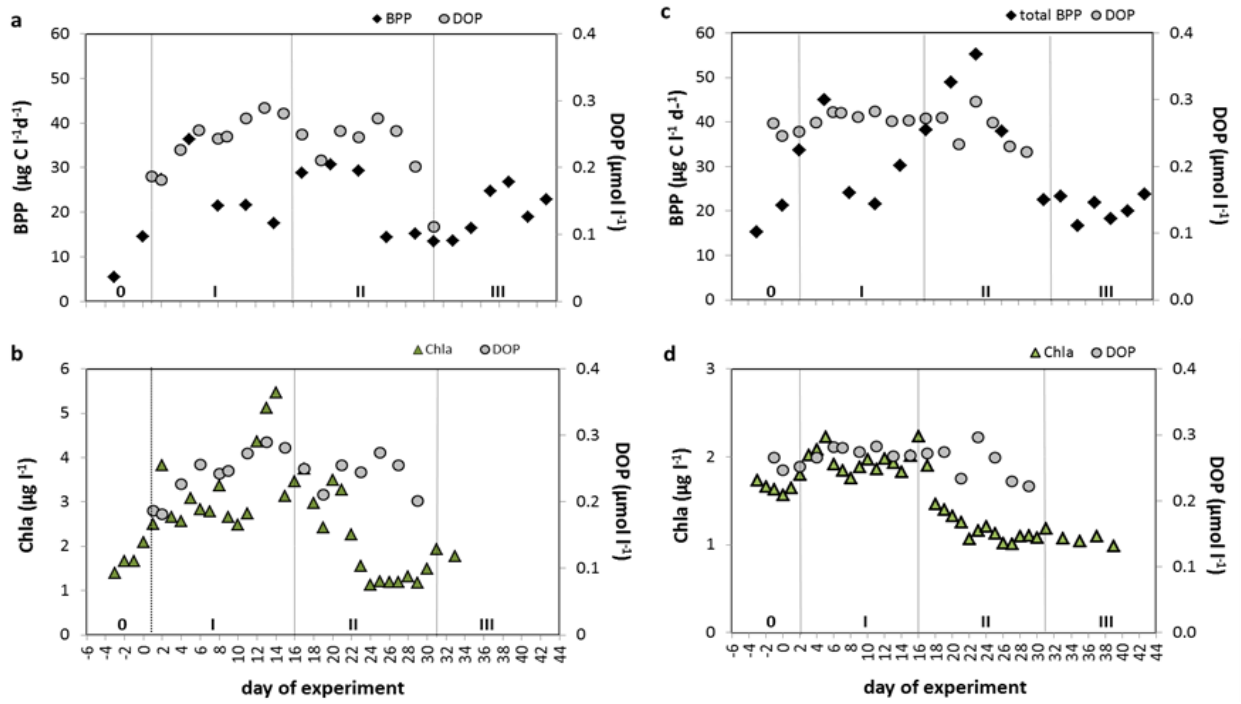


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Figure 7: PP concentration in the mesocosms during the initial phase from day 0 to day 2 (a) and from day 23 until the end (b) of experiment.

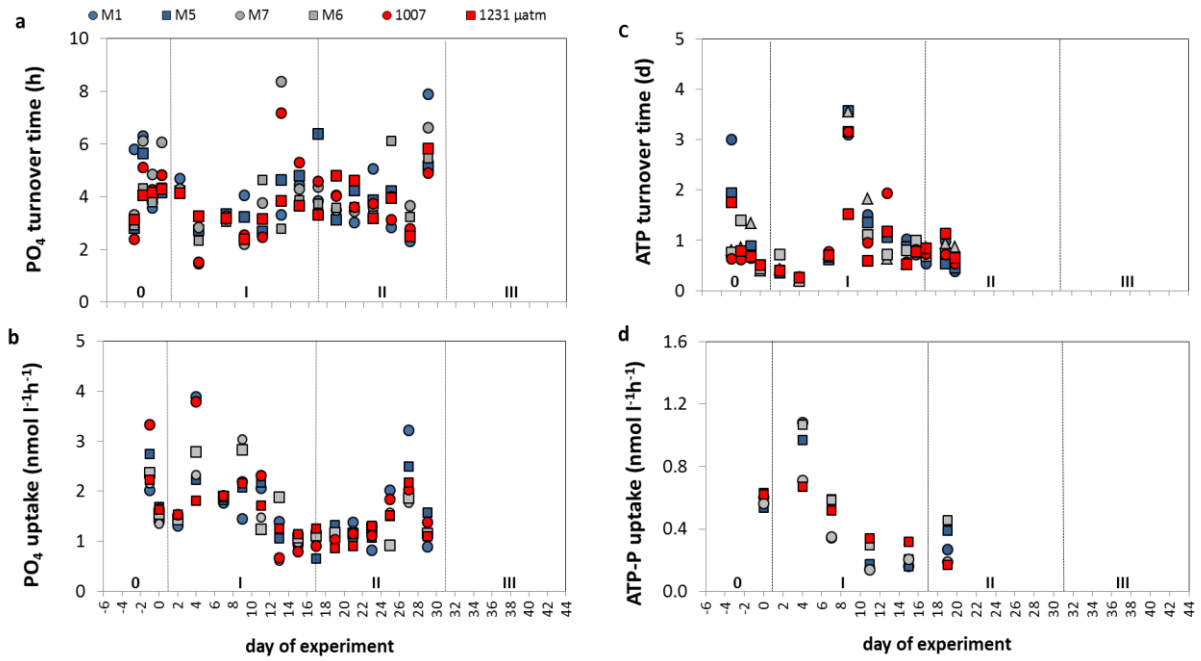
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Figure 8: Development of DOP in relation to bacterial production (BPP) and phytoplankton biomass (Chla) in the fjord (a, b) and in the mesocosms (c, d). For mesocosms, mean values averaged over all treatments are given.

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1156 Figure 9: Turnover times of PO_4 (a) and ATP (c) in the mesocosms as well as the respective
1157 uptake rates (b, d).

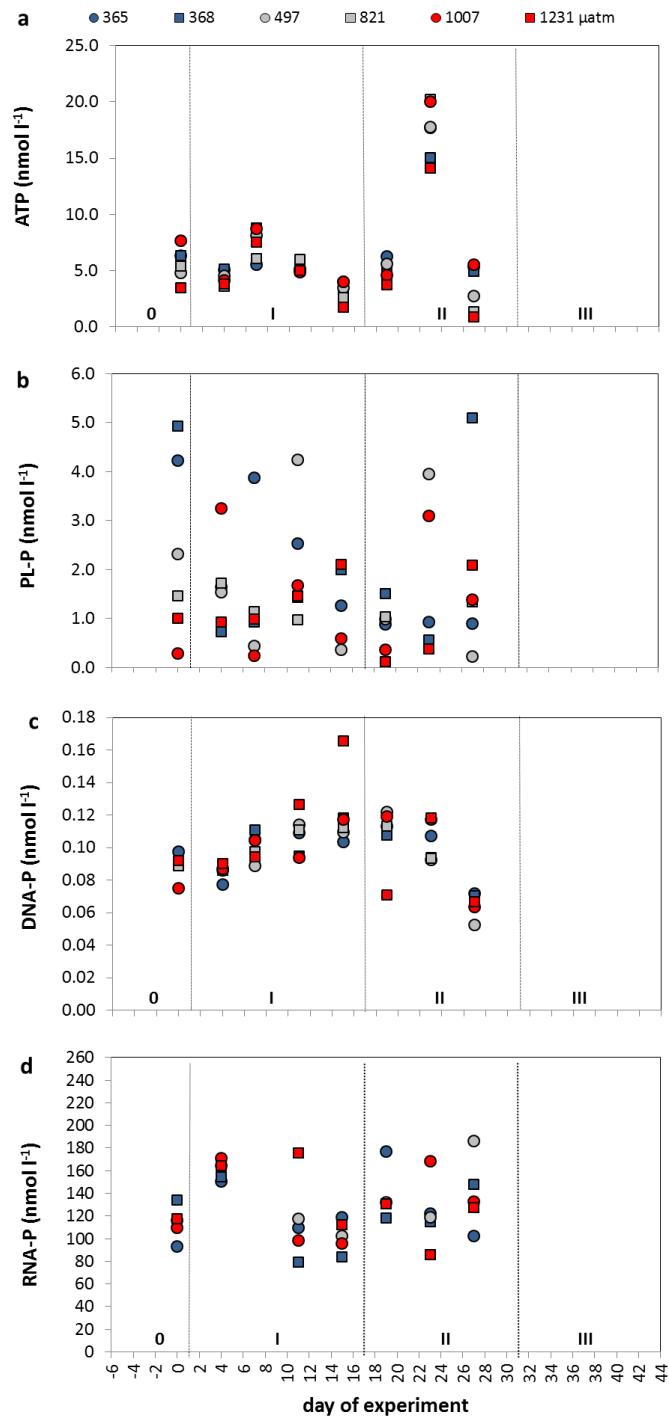
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Figure 10: Development of DOP compounds in the mesocosms and in the fjord from day 0 to day 27.