

General comments

I was very pleased to see that the authors incorporated many of the changes suggested by the three reviewers, most notably the sensitivity analysis. This has substantially improved the quality of the manuscript, and has moved away the paper from just being a comparison with the output from CMIP5 Earth System Models. The manuscript also clearly benefits from the improved model description, separating the Results and Discussion sections, and improved design of the tables and figures.

Personally, I disagree with the authors that their original Figure 7 didn't add anything to the manuscript, but can live with them taking it out as its original discussion contained some flaws. What still requires improvement is the description of the sensitivity analysis. Although the manuscript is definitely better structured compared to the first version and many typos have been removed, it is still left with some ambiguous statements and points that need clarification, mostly in the newly added parts.

*Due to the great comments from the reviewers the manuscript is significantly improved.

Specific comments

- Lines 26-28: do you mean 'sensitive' or rather 'most sensitive'? The current statement implies that pH and Ω_{ar} are not sensitive at all to other parameters; is this true? Also, add 'Of the parameters tested' at the beginning of the sentence; the statement now implies that all parameters were tested.

*We mean 'most sensitive'. Thank you for picking this up.

- Line 53: where does this 10% come from? Provide a reference. For example, Sarmiento and Gruber (2006) mention that currently ca. 1 out of 20 molecules of CO₂ that dissolve in seawater (i.e. 5%) will remain in this form.

*calculated from refit by Mehrback refit from Dickson and Millero, 1987

- Line 92: what are 'non-CO₂s'? Please explain

* "...which takes in emissions of CO₂, non-CO₂s, such as CH₄, N₂O and halocarbons, and aerosols..."

- Lines 94-96: If this is a new component of Hector compared to Hartin et al. (2015), then this can be detailed a bit more, for example by explaining this new algorithm in Appendix A.

*There are no new comments between Hector v1.0 and Hector v1.1. between the versions we updated the API for linking with external models, updated the backend R scripts, and some minor bug fixes.

- Line 116 (equation 3): shouldn't the left hand side of this equation be $F(i \rightarrow j)$? If not, how is this equation related to equation 2?

*yes, this has been corrected.

\[- Lines 126-129: The authors do mention now that all carbon in the ocean compartment is in the form of DIC and ratify this for DOC, but not for POC. Please add references showing how of ocean's carbon is in the form of POC and why it can be neglected in Hector.

*We added a small sentence addressing POC. “Dissolved organic matter is less than 2% of the total inorganic carbon pool, of which a small fraction is dissolved organic carbon (Hansell and Carlson, 2001) and particulate organic carbon less than 1% of the total carbon pool (Eglinton and Repeta, 2004).”

- Lines 146: why not state here that $k = 0.585 Sc^{-(1/2)} U^{10^2}$ and later simply define Tr as the product of α and k ?

*while this also is correct, we decided that the original description corresponds to the model code better, making it easier to track the flow of commands.

- Lines 169-174: I really like the inclusion of the sensitivity analysis and I’m perfectly fine with not testing all parameters in it. This description is, however, not detailed enough. What I miss here, however, is (a) which parameters were selected; (b) why you thought a priori that these parameters were important for the marine carbonate system; and (c) (this also holds for Table 7) it needs to be made clear where the selected parameters appear in Hector. For salinity and temperature this is quite clear, also because they are given in Table 1, but for e.g. albedo and beta (what does beta even stand for? I see it is defined in line 265, but it should be added to Table 7) this isn’t clear at all.

*This is a great comment by the reviewer. We have added throughout the text some significant improvements to the description of the parameters. For example, “We selected those parameters directly influencing atmospheric CO₂ concentrations (beta and Q₁₀), parameters involved in the calculation of temperature (albedo and climate sensitivity) and those parameters involved in the uptake of carbon in the surface ocean (ocean surface temperature, salinity, wind stress, and ocean circulation).”

- Lines 192-193: how exactly were the data annually averaged, e.g. via linear interpolation between sample times before averaging?

*The data was simply annually averaged. We appreciate the reviewer’s thoughtful point on doing a linear interpolation prior to averaging and we will implement this change in the future.

- Lines 193-195: were the settings in CO2SYS chosen such that they are compatible with the equations used in Hector to calculate the carbonate system? Please state e.g. which equilibrium constants were chosen here and whether Si and P data were included in these calculations (as opposed to Hector where they were not included).

*We updated the text to reflect these changes. “The carbonate parameters not found in Table 3 are computed from temperature, salinity, and the given carbonate parameter pairs using the CO2SYS software (Lewis and Wallace, 1998). The equilibrium constants (K1 and K2 from Mehrbach et al., 1973 refit by Dickson and Millero, 1987) and total phosphorus and silica set to zero where chosen to best match Hector.”

- Line 196: add “and assuming constant TA” as this is a crucial assumption in these calculations

*If TA was used as an input to CO2sys, than TA was not held constant.

- Line 173 (and section 3.2): a percentage change in pH is not physically meaningful as pH itself is presented on a logarithmic scale. Please give pH changes in absolute values, or give percentage changes of H⁺ concentration.

*We agree strongly with the reviewer. We have since changed our results to the percent change for [H⁺].

- Lines 211-212: only at high latitudes is this decline between 2100-2300 clearly visible; for the low latitudes both Hector and CMIP5 seem to show a very slight decrease. Rephrase this text.
*updated: “ Second, after 2100 the high latitude CMIP5 median begins to decline, while Hector rises and stabilizes.”

- Lines 269-270: what would be the underlying mechanism for these different responses? For example, why is pH in the high latitude surface ocean more sensitive to wind stress than at low latitudes? Please elaborate a bit on this in the discussion.

*Since we switched to percent change of H^+ the differences, we found that the high latitude was more sensitive to all changes in parameters at 2100 and 2300. This is most likely due to the fact that the high latitude ocean makes up only 15% of the global ocean.

- Lines 316-318: It is key to note that this upscaling is only valid if we assume a linear response. This should be added here. Moreover, I'm not convinced at all that it is realistic to expect such a linear response, you already see that in the fact that in Table 7 the percentage changes for +10% and -10% are not mirrored. I'd therefore suggest dropping this statement.

*the reviewer raises a good point and we have deleted this statement.

- Lines 331-333: I'd stress here that this is the case for the open ocean / on the global scale. Locally, in coastal areas, this may not be true and also cannot be shown with Hector.

*agree

- Appendix B (and section 4): nowhere in the manuscript it is mentioned how total borate is defined. This needs to be added to the Appendix. Is it calculated as a function of salinity? If so, it also makes up part of the system response when changing salinity; this should be mentioned in the Discussion.

*equation of borate was added to the Appendix.

- Table 5: Are the presented rates of change the average rates of change over those periods (i.e. 1850-2100 and 1850-2300) or the rates of change in these specific years (i.e. 2100 and 2300)? Please clarify in the figure caption. The same holds for Table 6; in this case it is specified in the text (lines 197-199) but for clarity it should be added to the caption.

*we added an extra column to the table with the year ranges

- Figure 2: I understand the big jump in CMIP5 mean DIC around 2100, switching from $n=15$ to $n=3$, but why is there a jump in mean DIC around ~1870?

*There are 4 models (2 versions of each GFDL and HadGEM) that do not start until 1859 and 1860. Therefore, when these models appear there is a jump in the graph.

Technical corrections

*We thank the reviewer for their careful edits of the manuscript. We have since made all necessary changes to the text.

- Line 24: change 'saturations' to 'saturation levels (Ω_{ar})' or 'saturation states (Ω_{ar})'. This definition is necessary as it comes back later in the abstract

- Lines 27-28: don't just give Q_{10} and beta here. If I read this without having read the article, I'd think beta refers to acid-base buffering capacity and Q_{10} to a temperature coefficient in the ocean. Define the parameters or just describe them.

- Line 47: the superscript in H^+ is missing.

- Line 49: remove 'total inorganic carbon' here.
- Lines 51-52: change "Therefore, because of the buffering capacity of the oceans" to something like "Because of this capacity of the ocean to buffer chemical changes"
- Line 62: "changes in ocean acidification" should be changed to "ocean acidification" or "changes in ocean chemistry"
- Lines 83-86: in my opinion, the use of commas is a bit strange; better use semicolons in between the various sections.
- Line 88: change 2.0 to 2; it is not a subsection
- Lines 112-114: Give the definition of j in equation 2 here.
- Line 119: this statement is somewhat misleading; also $[CO_3^{2-}]$ is a measurable carbonate system parameter; see e.g. Byrne and Yao (Marine Chemistry, 2008). Please rephrase.
- Line 156: change "solubility (K_{sp})" to "solubility product (K_{sp})"
- Lines 180-181: shouldn't it be 2004 and 2005, rather than 2005 and 2006 respectively as the border between historical and RCP8.5 experiments? At least that would be consistent with Table 4 (or should Table 4 say 'from 1850-2005' rather than 'from 1850-2004')
- Line 188: "over- or underestimation" (so add dash)
- Lines 197-199: add the Bates et al (2014) reference here
- Line 210: change "he" to "the"
- Line 233: delete comma after "Recent work"
- Line 239: delete comma after "Figure 5"
- Line 248: typo in "undersaturated"
- Line 255: change "under saturation" to "undersaturation"
- Lines 285-294: this section doesn't read smoothly. Please consider changing sentence order or rewrite this section.
- Line 290: change "under saturation" to "undersaturation"
- Line 303: something is missing here, perhaps 'it' after the semicolon.
- Lines 320: this is section 5, not section 6.
- Table 2: the second column (model name) doesn't add much; consider dropping it.
- Table 5: change "total pH" to "pH (total scale)"
- Figures 2-5: in the earlier version observations and model output were split in the legend. I prefer that over the current merged legends.
- Figure 1: I don't see any dashed arrows in this figure.
- Figure 6: remove the line explaining "Hector" in the legend. Add (a) and (b) to the respective plots.

Review - "Ocean acidification over the next three centuries using a simple global climate carbon-cycle model: projections and sensitivities."

General issue:

Throughout the manuscript there is a problem with references. These should be written either in the form "Hartin et al. (2015)" or in the form "(Hartin et al., 2015)". I have found errors on the following lines: 97, 122, 283, 348, 366, 399, and in addition most refs in Table 1 and all refs in Table 3. There could of course be other mistakes that I missed, so please check this carefully.

*Thank you for noticing this issue. We have carefully looked over all of the in-text citations and made all necessary changes.

Other comments:

*We thank the reviewer for their detailed comments below. We have addressed all of these within the manuscript.

Line 26-27: Beta and Q_{10} are not defined until line 265-267 (and by the way defined again on line 297-298). I think you should explain these parameters already in the abstract.

Line 47: The "+" in "H+" should be in superscript.

Line 69-73: The RCP and CMIP5 acronyms were already defined in the abstract, maybe that's sufficient.

Line 156 and 362: The "sp" in " K_{sp} " should be in subscript.

Line 210: "First, he carbon pools..." – change to "the carbon pools".

Line 233: "Recent work, suggests" – remove comma.

Line 236: "Twice" instead of "double".

Line 237: Change "decrease in pH of 7.96" to "decrease in pH to 7.96".

Line 255 and 290: Change "under saturation" to "undersaturation"?

Line 255-257: According to the text: "Even under a best-case scenario, RCP 2.6, low latitude pH will drop to 7.73 by 2100 and to 7.43 by 2300, with Ω_{Ar} saturations remaining well outside of present day values". However, according to Figure 6, pH in the best-case scenario is around 8.1 and 8.15 in 2100 and 2300 respectively, while the aragonite saturation steadily increases... so what happened here? – By the way, I would change "saturations" on line 256 to "saturation levels".

*Thank you for this. We have no idea where those numbers in the text came from but they have been updated accordingly.

Line 288: Pelejero et al. (2010) is not in the reference list.

Line 320: The Section number should be 5 and not 6.

Line 323: "study, calculates" – remove comma.

Line 328: "archive, includes" – remove comma.

Line 331-332: "Overall, we find that parameters directly involved changes in atmospheric [CO₂] have the most impact" – Strange sentence, I think a word is missing somewhere.

Line 369: Rongudai or Tongudai?

Line 376-379: I would change this section to: "C. Hartin designed and carried out the experiments. C. Hartin, B. Bond-Lamberty, and P. Patel developed the model code. A. Mundra processed data and

designed and prepared figures. C. Hartin prepared the manuscript with contributions from all co-authors.”

Line 386: According to the reference list, the Dore et al. paper is from 2009.

Line 414: Solid/dashed arrows: all arrows in this figure appear to be solid in my copy of the manuscript – maybe use different colors instead? * that was a typo, the arrows are curved and not dashed.

Review of: "Projections of ocean acidification over the next three centuries using a simple global climate carbon-cycle model" by C.A. Hartin et al.

General comments:

The paper presents a fast and competent model tool for future projections of the carbonate system. This is something I think is needed as complement to the more complex, computationally expensive earth system models. The study now includes a sensitivity study and the manuscript is improved by it. The paper is interesting, and presents a promising concept. It still needs some smaller corrections, but nothing major.

*We thank the reviewer for their detailed comments. We have made all necessary changes to the manuscript.

Specific comments:

I have no major specific comments at this stage in the review process.

Line-by-line corrections:

Line 27: beta and Q10 has not been previously defined here, please do so.

Line 47: superscript the + sign in H⁺.

Line 210: "he" should be "the"?

Lines 260-261: Is "set of parameters" and "the input parameters" the same parameters? Clarify in the text and please refer the reader to a table (table 7?) were they are listed or list them in the text. As of now, it is too hard to follow.

Lines 263-263: I don't understand the sentence. Are you comparing the range of omega and ph to atmospheric temperature range in your own model or another model or ... something else? Clarify.

Lines 267-268: I think it should be mentioned that also surface ocean temperature (table 7) has a rather large impact on omega. The effect of a 10% change in this parameter gives as large changes in omega as the Q10 change.

*We did add a few words about temperature as well to the aragonite sensitivities.

Line 298: "within the land" sounds weird to me, re-write to "on land" or something?

Line 303: The end of the sentence needs to be re-written.

Lines 332-333: Has this been found anywhere else, or is it new?

Line 388: Homogenize the size of the tables and improve presentation (unless that is done in the final layout anyway)

Line 415: Figure 1 has no dashed arrows that I can see.

1 **Ocean acidification over the next three centuries using a simple global climate carbon-cycle model:**
2 **projections and sensitivities.**

3

4 C. A. Hartin*, B. Bond-Lamberty, P. Patel and Anupriya Mundra

5

6 Pacific Northwest National Laboratory, Joint Global Change Research Institute at the University of

7 Maryland-College Park, 5825 University Research Court #3500, College Park, MD 20740, USA

8

9 *Corresponding author: Corinne.hartin@pnnl.gov

10

11 **ABSTRACT**

12 Continued oceanic uptake of anthropogenic CO₂ is projected to significantly alter the chemistry of the

13 upper oceans over the next three centuries, with potentially serious consequences for marine

14 ecosystems. Relatively few models have the capability to make projections of ocean acidification,

15 limiting our ability to assess the impacts and probabilities of ocean changes. In this study we examine

16 the ability of Hector v1.1, a reduced-form global model, to project changes in the upper ocean

17 carbonate system over the next three centuries, and quantify the model's sensitivity to parametric

18 inputs. Hector is run under prescribed emission pathways from the Representative Concentration

19 Pathways (RCPs) and compared to both observations and a suite of Coupled Model Intercomparison

20 (CMIP5) model outputs. Current observations confirm that ocean acidification is already taking place,

21 and CMIP5 models project significant changes occurring to 2300. Hector is consistent with the

22 observational record within both the high (>55°) and low latitude oceans (<55°). The model projects low

23 latitude surface ocean pH to decrease from preindustrial levels of 8.17 to 7.77 in 2100, and to 7.50 in

24 2300; aragonite saturations levels decrease from 4.1 units to 2.2 in 2100 and 1.4 in 2300 under RCP 8.5.

25 These magnitudes and trends of ocean acidification within Hector are largely consistent with the CMIP5
26 model outputs, although we identify some small biases within Hector’s carbonate system. Of the
27 parameters tested, Modeled changes in $[pH^+]$ are most sensitive to ~~those~~ parameters that directly affect
28 atmospheric CO₂ concentrations -(beta and Q₁₀ (terrestrial respiration temperature response) as well as
29 changes in ocean circulation, while changes in Ω_{Ar} saturation levels are sensitive to changes in ocean
30 salinity and Q₁₀. We conclude that Hector is a robust tool well-suited for rapid ocean acidification
31 projections, sensitivity analyses, and is capable of emulating both current observations and large scale
32 climate models under multiple emission pathways.

33 1. INTRODUCTION

34 Human activities have led to increasing anthropogenic emissions of greenhouse gases to the
35 atmosphere. In the first decade of the 21st century CO₂ emissions from anthropogenic sources and land
36 use changes accounted for ~9 Pg C yr⁻¹, with future emission projections of up to 28 Pg C yr⁻¹ by 2100
37 under ~~Representative Concentration Pathway 8.5 (RCP 8.5)~~ (Riahi et al., 2011). The world's oceans have
38 played a critical role in lessening the effects of climate change by absorbing 25-30% of the total
39 anthropogenic carbon emissions since 1750 (Le Quéré et al., 2013; Sabine et al., 2011).

40 In response to this increasing atmospheric burden of CO₂ and increasing oceanic uptake, the
41 oceans are experiencing both physical and biogeochemical changes: surface and deep water warming,
42 reduced subsurface oxygen, and a reduction in calcium carbonate saturation levels and pH (Doney,
43 2010). Mean surface ocean pH has already decreased by 0.1 units relative to preindustrial times
44 (Caldeira et al., 2003). If current emission trends continue, ocean acidification will occur at rates and
45 extents not observed over the last few million years (Feely et al., 2004; Feely et al., 2009; Kump et al.,
46 2009; Caldeira et al., 2003). Ocean acidification occurs when atmospheric CO₂ dissolves in seawater
47 (CO₂(aq)), forming carbonic acid (H₂CO₃), dissociating into carbonate (CO₃²⁻) and bicarbonate (HCO₃⁻),
48 and releasing protons (H⁺). The net effect of adding CO₂ to the system increases the concentrations of
49 [H₂CO₃], [HCO₃⁻], and [H⁺], while decreasing [CO₃²⁻] concentrations and lowering the pH. The sum of
50 [HCO₃⁻], [CO₃²⁻] and [CO₂*], where [CO₂*] = [CO₂(aq)] + [H₂CO₃] represents the ~~total inorganic carbon or~~
51 dissolved inorganic (DIC) of the system. As CO₂(aq) continues to increase in the ocean it reacts with
52 CO₃²⁻, forming HCO₃⁻, decreasing the fraction of CO₂ that can be readily absorbed by the oceans.
53 ~~Therefore, b~~Because of this ~~buffering~~ capacity of the ocean to buffer chemical changes, a doubling of
54 atmospheric [CO₂] will not correspond to a doubling of [CO₂*] but instead will result in an increase on
55 the order of 10% (Dickson and Millero, 1987). Due to both ~~these~~ chemical and physical changes (e.g.,
56 warming and stratification), the oceans may become less efficient in the uptake of anthropogenic CO₂ as

57 the climate continues to change (Sarmiento and Le Quéré, 1996; Matear and Hirst, 1999; Joos et al.,
58 1999; Le Quéré et al., 2010).

59 Numerous experiments and observations indicate that ocean acidification will have significant
60 effects on calcifying marine organisms. For example, the rate of coral reef building may decrease,
61 calcification rates of planktonic coccolithophores and foraminifera may be suppressed, and significant
62 changes in trophic level interactions and ecosystems may occur (Cooley and Doney, 2009; Silverman et
63 al., 2009; Fabry et al., 2008; Riebesell et al., 2000). Some coral reefs are believed to already be eroding
64 for parts of the year due to ~~changes in~~ ocean acidification (Yates and Halley, 2006; Albright et al., 2013).
65 Global surface pH is projected to drop by up to 0.33 units (Gehlen et al., 2014; Orr et al., 2005) and all
66 existing coral reefs will be surrounded by conditions well outside of preindustrial values and even
67 today's saturation levels (Ricke et al., 2013) under the RCP8.5 scenario.

68 These model projections of ocean acidification come primarily from Earth System Models
69 (ESMs) that integrate the interactions of atmosphere, ocean, land, ice and biosphere to estimate the
70 present and future state of the climate. ESMs are computationally expensive and typically run using
71 stylized experiments or a few ~~Representative Concentration Pathways (RCPs)~~ greenhouse gas
72 concentration trajectories used in the Intergovernmental Panel on Climate Change 5th Assessment
73 Report (IPCC, 2013). This generally limits ESM-based analyses to those scenarios. The occurrence of
74 ocean warming and acidification is consistent across ~~the Coupled Model Intercomparison Project~~
75 ~~(CMIP5)~~ ESMs, but their rates and magnitudes are strongly dependent upon the scenario (Bopp et al.,
76 2013).

77 An alternative to ESMs are reduced-form models, relatively simple and small models that can be
78 powerful tools due to their simple input requirements, computational efficiency, tractability, and thus
79 ability to run multiple simulations under arbitrary future emission pathways. This allows for the
80 quantification of arbitrary climate change scenarios, emulation of larger ESMs, as well as in-depth

81 parameter sensitivity studies and uncertainty analyses (Senior and Mitchell, 2000; Ricciuto et al., 2008;
82 Irvine et al., 2012).

83 Our goal of this study is to quantify how well Hector, a reduced-form model that explicitly
84 treats surface ocean chemistry, emulates the marine carbonate system of both observations and the
85 CMIP5 archive, and explore the parametric sensitivities to Hector's ocean outputs. The remainder of the
86 paper is organized as follows; section 2, a detailed description of Hector's ocean component, the data
87 sources and simulations run, section 3, results of the model comparison and sensitivity experiments and
88 lastly, section 4, a discussion of the results.

89

90 **2.0 Model Description – Hector**

91 Hector is open-source and available at <https://github.com/JGCRI/hector>. The repository
92 includes all model code needed to compile and run the model, as well as all input files and R scripts to
93 process its output. Hector is a reduced form climate carbon-cycle model, which takes in emissions of
94 CO₂, non-CO₂s (e.g., CH₄, N₂O and halocarbons, and aerosols), converts emissions to concentrations
95 where needed, calculates the global radiative forcing, and then global mean temperature change.
96 Hector contains a well-mixed global atmosphere, a land component consisting of vegetation, detritus,
97 and soil, and an ocean component. In this study we use Hector v1.1, with an updated ocean
98 temperature algorithm to better match the CMIP5 mean. For a detailed description of the land and
99 atmospheric components of Hector, please refer to Appendix A and Hartin et al. (2015).

100 **2.1 Ocean Component**

101 Hector's ocean component is based on work by Lenton (2000), Knox and McElroy (1984) and
102 Sarmiento and Toggweiler (1984). It consists of four boxes: two surface boxes (high and low latitude),
103 one intermediate, and one deep box. The cold high latitude surface box makes up 15% of the ocean
104 surface area, representing the subpolar gyres (> 55°), while the warm low latitude surface box (<55°)

105 makes up 85% of the ocean surface area. The temperatures of the surface boxes are linearly related to
 106 the global atmospheric temperature change, and are initialized at 2°C and 22°C for the high and low
 107 latitude boxes respectively. This temperature gradient sets up a flux of carbon into the cold high latitude
 108 box and a flux out of the warm low latitude box. The ocean-atmosphere flux of carbon is the sum of all
 109 the surface fluxes (F_i , $n=2$).

$$F_O(t) = \sum_{i=1}^n F_i(t) \quad (1)$$

110 Once carbon enters the high latitude surface box it is circulated between the boxes via
 111 advection and water mass exchange, simulating a simple thermohaline circulation. In this version of
 112 Hector we do not explicitly model diffusion; simple box-diffusion models and “HILDA” (e.g., Siegenthaler
 113 and Joos, 1992) type models are typically in good agreement with observations but are more
 114 computationally demanding than a simple box model (Lenton, 2000). The change in carbon of any
 115 ocean box i is given by the fluxes in and out (i), with $F_{atm \rightarrow i}$ as the atmospheric carbon flux of the two
 116 surface boxes:

$$\frac{dC_i}{dt} = \sum_{j=1}^{in} F_{j \rightarrow i} - \sum_{j=1}^{out} F_{i \rightarrow j} + F_{atm \rightarrow i} \quad (2)$$

117 The flux of carbon between the boxes is related to the transport ($T_{i \rightarrow j}$, $m^3 s^{-1}$) between i and j , the
 118 volume of i (V_i , m^3), and the total carbon in i (including any air-sea fluxes) (C_i , Pg C);

$$\frac{dC_{i \rightarrow j}}{dt} = \frac{T_{i \rightarrow j} * C_i(t)}{V_i} \quad (3)$$

119 Volume transports are tuned to yield an approximate flow of 100 Pg C from the surface high latitude box
 120 to the deep ocean box at steady state, simulating deep water formation.

121 Hector ~~includes calculates four measurable variables of the carbonate system in seawater:~~ DIC,
 122 total alkalinity (TA), ~~CO_3^{2-} , HCO_3^-~~ pCO_2 and pH , ~~any pair of which can be used to describe the entire~~
 123 ~~carbonate system.~~ DIC and TA are the two carbonate variables used to solve the rest of the carbonate

124 ~~system surface ocean pH and pCO₂ values.~~ These detailed carbonate chemistry equations are based on
125 numeric programs from Zeebe and Wolf-Gladrow, (2001) (Appendix B). We simplified the equations by
126 neglecting the effects of pressure, since we are only concerned with the surface ocean. Hector is run to
127 equilibrium in a perturbation-free mode, prior to running the historical period, ensuring that it is in
128 steady state (Hartin et al., 2015; Pietsch and Hasenauer, 2006). DIC ($\mu\text{mol kg}^{-1}$) in the surface boxes is a
129 function of the total carbon (Pg C) and the volume of the box. All carbon within the ocean component is
130 assumed to be inorganic carbon. Dissolved organic matter is less than 2% of the total inorganic carbon
131 pool, of which a small fraction is dissolved organic carbon (Hansell and Carlson, 2001) and particulate
132 organic carbon less than 1% of the total carbon pool (Eglinton and Repeta, 2004). Therefore, for
133 simplicity we chose not to include dissolved or particulate organic carbon within Hector.

134 TA is calculated at the end of model spinup (running to equilibrium in an a historical,
135 perturbation-free mode) and held constant ~~throughout the run going forward in time~~, resulting in 2311.0
136 $\mu\text{mol kg}^{-1}$ for the high latitude box and 2435.0 $\mu\text{mol kg}^{-1}$ for the low latitude box. These values are
137 within the range of open ocean observations, ~~of~~ 2250.0 – 2450.0 $\mu\text{mol kg}^{-1}$ (Key et al., 2004; Fry et al.,
138 2015). We assume negligible carbonate precipitation/dissolution and no alkalinity runoff from the land
139 surface to the open ocean. Alkalinity is typically held constant with time, ~~which is~~ a reasonable
140 assumption over several thousand years (Lenton, 2000; Zeebe and Wolf-Gladrow, 2001; Glotter et al.,
141 2014; Archer et al., 2009). On glacial-interglacial time scales alkalinity and the dissolution of CaCO₃
142 sediments is an important factor in controlling atmospheric [CO₂] (Sarmiento and Gruber, 2006), and
143 thus on these scales Hector will underestimate the oceanic CO₂ uptake.

144 Hector solves for pCO₂, pH (total scale), and [HCO₃⁻], [CO₃²⁻], and aragonite (Ω_{Ar}) and calcite
145 saturation ~~states~~ (Ω_{Ca}) in both the high and low latitude surface ocean boxes. pCO₂ is calculated from
146 the concentration of [CO₂^{*}] and the solubility of CO₂ in seawater, based on salinity and temperature.

147 [CO₂^{*}] is calculated from DIC and the first and second dissociation constants of carbonic acid from
148 Mehrbach et al. (1973), refit by Lueker et al. (2000) (Appendix B).

149 Carbon fluxes between the atmosphere and ocean are calculated (Takahashi et al., 2009):

$$F = k \alpha * \Delta pCO_2 = Tr * \Delta pCO_2 \quad (4)$$

150 where k is the CO₂ gas-transfer velocity, α is the solubility of CO₂ in seawater (K_0 , Appendix B), and the
151 ΔpCO_2 is the difference in pCO₂ between the atmosphere and ocean. The product of k and α results in
152 Tr, the sea-air gas transfer coefficient, where $Tr (g C m^{-2} month^{-1} \mu atm^{-1}) = 0.585 * \alpha * Sc^{-1/2} * U_{10}^2$, 0.585
153 is a unit conversion factor (from mol liter⁻¹ atm⁻¹ to g-C m⁻³ μatm^{-1} and from cm h⁻¹ to m month⁻¹) and Sc
154 is the Schmidt number. The Schmidt number (Appendix B) is calculated from Wanninkhof (1992) based
155 on the temperature of each surface box. The average wind speed (U_{10}) of 6.7 m s⁻¹ is the same over
156 both surface boxes (Table 1). pH (total scale), [HCO₃⁻], and [CO₃²⁻] are calculated using the [H⁺] ion and
157 solved for in a higher order polynomial (Appendix B).

158 Aragonite and calcite are the two primary carbonate minerals within seawater. The degree of
159 saturation in seawater with respect to aragonite (Ω_{Ar}) and calcite (Ω_{Ca}) is calculated from the product of
160 the concentrations of calcium [Ca²⁺] and carbonate ions [CO₃²⁻], divided by the solubility product (K_{sp}).
161 The [Ca²⁺] is based on equations from Riley and Tongudai (1967) at a constant salinity of 34.5. If $\Omega = 1$,
162 the solution is at equilibrium, and if $\Omega > 1$ ($\Omega < 1$) the solution is supersaturated (undersaturated) with
163 respect to the mineral.

$$\Omega = \frac{[Ca^{2+}] [CO_3^{2-}]}{K_{sp}} \quad (5)$$

164 2.2. Simulation and experiments

165 Hector is run under prescribed emissions from 1850-2300 for all four Representative
166 Concentration Pathways (RCP 2.6, RCP 4.5, RCP 6, RCP 8.5) (Moss et al., 2010; van Vuuren et al., 2007).
167 We compare how well Hector can emulate the carbonate system of the CMIP5 median. Our results

168 section will mainly focus on RCP8.5 exploring the response of the carbonate system under a high
169 emissions scenario.

170 We also ran a series of model sensitivity experiments to quantify how influential some of
171 Hector's parameter inputs are on its outputs (in particular, $[H^+]_{pH}$ and Ω_{Ar}). Such sensitivity analyses are
172 important to document model characteristics, explore model weaknesses, and determine to what
173 degree the model behavior conforms to our existing understanding of the ocean system. We do not
174 sample Hector's entire parameter space, a computationally demanding exercise, but instead choose a
175 list of the parameters that we expect, *a priori*, to be important in calculating the marine carbonate
176 system. We selected those parameters directly influencing atmospheric CO₂ concentrations (beta and
177 Q₁₀), parameters involved in the calculation of temperature (albedo and climate sensitivity) and those
178 parameters involved in the uptake of carbon in the surface ocean (ocean surface temperature, salinity,
179 wind stress, and ocean circulation). ~~We selected eight land and ocean parameters, varying each~~ These
180 parameters are varied by $\pm 10\%$ relative to the RCP8.5 control, and we compare the percentage change
181 from the reference and the perturbation cases in 2005, 2100, and 2300. The reference, RCP8.5, refers
182 to the tuned set of parameters found in Hector v1.1, resulting in Figures 2-6.

183 **2.3 Data Sources**

184 All RCP input emission data are available at <http://tntcat.iiasa.ac.at/RcpDb/>. Comparison data are
185 obtained from a suite of CMIP5 Earth System Models (Table 2) (Taylor et al., 2012). The CMIP5 output is
186 available from the Program for Climate Model Diagnostics and Intercomparison
187 (<http://pcmdi3.llnl.gov/esgcat/home.htm>). We took the 0-100m (depth) mean for all available CMIP5
188 data for six output variables, computing the monthly mean for all years in the historical (1850-2005) and
189 RCP 8.5 (2006-2300) experiments. All outputs were regridded to a standard 1-degree grid using bilinear
190 interpolation in CDO version 1.7.1rc1, and then high latitude (-90 to -55 and 55 to 90 degrees), low
191 latitude (-55 to 55), and global area-weighted means computed using R 3.2.4. All CMIP5 comparisons

192 used in this study are from model runs with prescribed atmospheric CO₂ concentrations. We
193 acknowledge that this is not a perfect comparison, as Hector is emissions-forced being compared to the
194 concentration-forced CMIP5 models, but very few of the latter were run under prescribed emissions.
195 We use a combination of root mean square error (RMSE), rates of change (Δ) and bias (degree of
196 systematic over- or underestimation) to characterize Hector's performance relative to the CMIP5
197 median.

198 We also compare Hector to a series of observational ocean data. Surface ocean observations of
199 DIC, pCO₂, pH, Ω_{Ar} , and Ω_{Ca} are from time-series stations in both the high and latitude oceans; Hawaii
200 Ocean Time Series (HOT), Bermuda Atlantic Time Series (BATS), the European Station for Time Series in
201 the Ocean at the Canary Islands (ESTOC), the Irminger Sea, and the Iceland Sea (Table 3) . The time-
202 series data are annually averaged over the upper 100m of the water column. The carbonate parameters
203 not found in Table 3 are computed from temperature, salinity, and the given carbonate parameter pairs
204 using the CO2SYS software (Lewis and Wallace, 1998). The equilibrium constants (K1 and K2 from
205 Mehrbach et al., 1973 refit by Dickson and Millero, 1987) and zero total phosphorus and silica where
206 chosen to best match Hector. Lastly, a longer record (1708 – 1988) of pH and Ω_{Ar} from Flinder's Reef in
207 the western Coral Sea, calculated from boron isotope measurements, is used in the comparison
208 (Pelejero et al., 2005). We use rates of change (Δ) from 1988-2014, which overlaps the ~~BATS and~~HOT
209 and BATS time series, to quantify how well Hector simulates the observed changes in the ocean
210 carbonate parameters (Table 6). (Dore et al., 2009; Bates et al., 2014).

211

212 **3. Results**

213 **3.1 Model and Observation Comparisons**

214 Hartin et al. (2015) conducted a thorough analysis of Hector v1.0 demonstrating that it can
215 reproduce the historical trends and future projections of atmospheric [CO₂], radiative forcing, and global

216 temperature change under the four RCPs. In this study we focus on the upper ocean high and low
217 latitude inorganic carbon chemistry under RCP 8.5.

218 Hector captures the trend in DIC concentrations for both the high and low latitude surface
219 ocean with a global RMSE average of $7.0 \mu\text{mol kg}^{-1}$ when compared to CMIP5 models over the historical
220 period (Table 4; Figure 2). We note that there is a systematic bias in both the high and low latitude
221 surface boxes when compared to CMIP5. First, the carbon pools of the surface boxes are initialized with
222 carbon values slightly higher than the median CMIP5 values. Second, ~~we note that~~ after 2100 the high
223 latitude CMIP5 median begins to decline, while Hector rises and stabilizes. Only 3 CMIP5 models ran out
224 to 2300, with one model driving the decline. Regardless, this offset only results in a <3% global
225 difference between CMIP5 and Hector.

226 Hector accurately tracks the pCO_2 in both the high and low latitude surface ocean with similar
227 rates of change from 1850-2300 (Figure 3). There is a low bias in Hector compared to CMIP5 models
228 after 2100, due to the low bias in projected atmospheric $[\text{CO}_2]$ within Hector over the same time period
229 (Hartin et al., 2015). We do find Hector to be in closer agreement with the observation record.

230 Figure 4 shows the high and low latitude surface pH of Hector compared to CMIP5 and
231 observations from BATS, HOT, ESTOC, Irminger Sea, Iceland Sea, and Flinders Reef. While the high
232 latitude surface pH is slightly higher than the CMIP5 models, Hector is more similar to high latitude
233 observations. Since the preindustrial, observations of surface ocean pH decreased by 0.08 units,
234 corresponding to a 24% increase in $[\text{H}^+]$ concentrations and an 8% decrease in $[\text{CO}_3^{2-}]$, similar to
235 numerous studies (Feely et al., 2004; Sabine et al., 2004; Caldeira et al., 2003; Orr et al., 2005) that
236 estimate an average global decrease in pH of 0.1 or a 30% increase in $[\text{H}^+]$.

237 The Flinder's Reef pH record provides a natural baseline to compare future trends in ocean
238 acidification. While we did not expect the model to match exactly, as this reef site is heavily influenced
239 by coastal dynamics and internal variability, rates of change from the preindustrial (1750) to 1988 are

240 the same (0.0002 yr^{-1}) for both Hector and Flinder's Reef. Over the limited observational record from
241 both the Pacific and Atlantic Oceans, Hector accurately simulates the decline in pH (-0.0017 yr^{-1})
242 compared to observations (Table 6). Other observations in the North Pacific show surface changes of
243 pH up to 0.06 units between 1991 and 2006 with an average rate of -0.0017 yr^{-1} (Byrne et al., 2010).
244 Recent work₇ suggests that the North Atlantic absorbed 50% more anthropogenic CO_2 in the last decade
245 compared to the previous decade, decreasing surface pH by 0.0021 (Woosley et al., 2016). Under RCP
246 8.5, Hector projects a decrease in low latitude pH of 8.17 in 1850 to 7.77 in 2100 and down to 7.5 by
247 2300, similar to CMIP5 (Table 5). At approximately 2050, atmospheric $[\text{CO}_2]$ is ~~double~~ twice the
248 preindustrial concentrations, corresponding to a decrease in pH ~~of to~~ 7.96. Shortly after this doubling,
249 pH values are well outside the natural variability found in Flinder's Reef.

250 Figure 5₇ illustrates the high and low latitude surface Ω_{Ar} . We only highlight Ω_{Ar} , as Ω_{Ca} is similar
251 to that of Ω_{Ar} . As with pH, Hector is slightly higher than the CMIP5 Ω_{Ar} median but closer to the
252 observational record. Hector accurately simulates the change in Ω_{Ar} (-0.0090 yr^{-1}) compared to
253 observations (Table 6). Repeated oceanographic surveys in the Pacific Ocean have observed an average
254 $0.34\% \text{ yr}^{-1}$ decrease in the saturation state of surface seawater with respect to aragonite and calcite over
255 a 14-year period (1991-2005) (Feely et al., 2012); the average decrease in Hector is between $0.19\% \text{ yr}^{-1}$
256 and $0.25\% \text{ yr}^{-1}$. Saturation levels of Ω_{Ar} decrease rapidly over the next 100 years in both the high and
257 low latitude. Hector accurately captures the decline in saturations states with low RMSE values for Ω_{Ar} .
258 Under RCP8.5 Hector projects that low latitude Ω_{Ar} will decrease to 2.2 by 2100 and down to 1.4 by
259 2300. The high latitude oceans will be undersaturated ~~understaturated~~ with respect to aragonite by
260 2100 and will drop down to 0.7 by 2300.

261 Lastly, Figure 6 highlights pH and Ω_{Ar} projections under all four RCPs from 1850 to 2300. Over
262 the last 20 years, both pH and Ω_{Ar} have declined sharply and will continue to decline under RCP 4.5, 6.0
263 and 8.5, outside of their preindustrial and present day values. These RCPs represent a range of possible

264 future scenarios, with ocean pH varying between 8.15 and 7.46 for the high latitude and Ω_{Ar} varying
265 between 1.94 and 0.60. High latitude Ω_{Ar} saturation levels presently are much lower than the low
266 latitude and reach under-saturation before the end of the century. Even under a best-case scenario, RCP
267 2.6, low latitude pH will drop to 8.077-73 by 2100 and to 8.127-43 by 2300, with Ω_{Ar} saturations s states
268 remaining well outside of present day values.

269 3.2 Model Parameter Sensitivity

270 ~~Parametric sensitivities are different between pH and Ω_{Ar} , and between the high and low~~
271 ~~latitude surface ocean boxes. Parametric sensitives are different between $[H^+]$ and Ω_{Ar} . We use $[H^+]$ to~~
272 ~~highlight changes in pH, as $pH = -\log[H^+]$. The reference, RCP8.5, refers to the tuned set of parameters~~
273 ~~found in Hector v1.1, resulting in Figures 2-6. Global pH is fairly insensitive to the values of the input~~
274 ~~parameters used, while Ω_{Ar} is slightly more sensitive (Table 7). For example, a 10% change in input~~
275 ~~parameters results in range from 0.0 – 0.21% in pH and 0.0 – 7.18% in Ω_{Ar} . In comparison a 10%~~
276 ~~parameter change results in a range from 0.0 – 10.3% in global atmospheric temperature change. In the~~
277 near term (from 2005-2100) the calculation of pH is sensitive to a combination of parameters, salinity
278 ocean circulation, and beta (terrestrial CO_2 fertilization), and wind stress, while on longer time scales (to
279 2300), $[H^+]$ pH is most sensitive to changes in Q_{10} (terrestrial respiration temperature response) and
280 ocean circulation (Table 7). Global Ω_{Ar} is most sensitive to changes in salinity in both the near and long
281 term. Similar to $[H^+]$ pH, Ω_{Ar} becomes more sensitive to changes in Q_{10} in the long term.

283 4. Discussion

284 The marine carbonate system is projected to undergo significant changes under the RCPs. pCO_2
285 and DIC are increasing rapidly as atmospheric $[CO_2]$ continues to rise under RCP 4.5, 6.0 and 8.5. pH, and
286 Ω_{Ar} are decreasing rapidly outside of observations and are projected to continue to decrease under all
287 scenarios (Figure 6). Only under RCP 2.6 do pH and Ω_{Ar} values begin to increase ~~back~~ towards present

288 day values. A lowering of Ω_{Ar} from approximately 4.0 to 3.0 is predicted to lead to significant reductions
289 in calcification rates in tropical reefs (Kleypas et al., 1999a; Silverman et al., 2009). By the end of the 21st
290 century ~~In agreement with Roy et al., (2015) and Ricke et al., (2013) by the end of the 21st-century (2072~~
291 ~~under RCP8.5) Hector projects that the~~ low latitude oceans Ω_{Ar} will drop below 3.0, well outside of the
292 preindustrial values of $\Omega_{Ar} > 3.5$, and. ~~At the end of the 21st Century, the~~ high latitude oceans Ω_{Ar} are
293 close to undersaturation ($\Omega < 1$) (Figure 6). These results agree with other studies that investigated
294 changes to the carbonate system (Roy et al., 2015; Ricke et al., 2013). ~~However, the threshold for~~
295 ~~biogenic carbonate precipitation is species dependent and may be significantly higher than 1.0 when~~
296 ~~combined with other factors. For example, some coral reef communities need to develop in waters with~~
297 ~~$\Omega_{Ar} > 3.3$ (Hoegh-Guldberg et al., 2007; Kleypas et al., 1999b) (Pelejero et al., 2010; Hoegh-Guldberg et~~
298 ~~al., 2007; Kleypas et al 1999).~~ Accounting for seasonal variations in the Ω_{Ar} saturation levels may move
299 ~~this~~ the time of under-saturation may move forward by up to 17 ± 10 years (Sasse et al., 2015). Due to
300 Hector's time step of 1 year, we may be overestimating the time when ocean acidification reaches a
301 critical threshold. We also note that other factors such as eutrophication, river discharge, and upwelling
302 will likely increase the probability that coastal regions will experience the effects of ocean acidification
303 sooner than the projected open ocean values in Hector (Ekstrom et al., 2015).

304 Using $[H^+]$ as a proxy for pH, we find that $[H^+]$ pH is sensitive to Q_{10} and ocean circulation. ~~fairly~~
305 ~~insensitive to most parametric changes, but in both the near and long term, pH is sensitive to~~
306 ~~parameters that indirectly affect atmospheric $[CO_2]$.~~ Changes in ~~both beta and~~ Q_{10} , (~~the terrestrial CO_2~~
307 ~~fertilization effect and~~ the respiration temperature response, ~~respectively~~) are responsible for the
308 ~~uptake and~~ release of carbon within-on the land. Uncertainties in the land carbon cycle have been
309 attributed to uncertainties in future CO_2 projections within the CMIP5 models (Friedlingstein et al.,
310 2014). Therefore, uncertainties in the land carbon cycle will also have implications for the marine
311 carbonate system. A 10% change in the thermohaline circulation parameter (Tt in Figure 1), representing

312 a portion of the high latitude to deep ocean circulation, results in ~3% change in [H⁺]. The dynamics of
313 ocean uptake of CO₂ are strongly dependent on this circulation of CO₂ laden waters from the surface
314 ocean to depth. CMIP5 models project a weakening in the Atlantic meridional overturning circulation by
315 an average of 36% under RCP8.5 by 2100 (Cheng et al., 2013). Therefore, changes in ocean circulation
316 may have implications on the marine carbonate system, influencing the ocean pH. We also find that the
317 high latitude surface ocean is more sensitive to parameter changes than the low latitude surface ocean.
318 The high latitude box makes up 15% of the global oceans in Hector and therefore, changes of the same
319 size are more easily felt in the high latitude box compared to the low latitude box. This may have direct
320 implications on future marine carbonate projections.

321 Global Ω_{Ar} saturation levels are most sensitive to changes in salinity. ~~Within Hector, s~~Salinity is
322 ~~used to directly involved in the~~ calculation [Ca²⁺] ~~and total boron (Appendix B); is used to determine~~
323 Ω_{Ar} . Typically the carbonate system is normalized to changes in salinity to understand the chemical
324 changes within the system, instead we show that Ω_{Ar} may be sensitive not only to future changes in
325 atmospheric [CO₂] but also sensitive to changes in precipitation and evaporation. This may be
326 important, as studies suggest significant changes in precipitation patterns under a changing climate
327 (Held and Soden, 2006; Liu and Allan, 2013).

328 ~~The dynamics of ocean uptake of CO₂ are strongly dependent on the rate of downward~~
329 ~~transport of CO₂ laden waters from the surface ocean to depth. Climate feedbacks on the carbonate~~
330 ~~system resulting from changes in ocean circulation are neglected in Hector, as the model holds ocean~~
331 ~~circulation constant in time. CMIP5 models project a weakening in the Atlantic meridional overturning~~
332 ~~circulation by an average of 36% under RCP8.5 by 2100 (Cheng et al., 2013). We investigate the~~
333 ~~sensitivity of the carbonate system to a change in ocean circulation by varying the thermohaline~~
334 ~~circulation parameter (T_s in Figure 1). This parameter represents a portion of the high latitude surface to~~
335 ~~the deep ocean circulation. A 10% change in ocean circulation (T_s) results in a <4% change in air-sea~~

336 ~~fluxes and moderate effects on surface pH and Ω_{Ar} . If we scale up from a 10% change in T_s to a 36%~~
337 ~~change, which is projected from the CMIP5 models, it may result in a roughly 14% change in the air-sea~~
338 ~~fluxes of carbon to the surface ocean and a 0.3% and 5.0% change in pH and Ω_{Ar} , respectively.~~

339

340 **6.5. Conclusions**

341 We developed a simple, open-source, object oriented carbon cycle climate model, Hector, that
342 reliably reproduces the median of the CMIP5 climate variables (Hartin et al., 2015). The ocean
343 component presented in this study calculates the upper ocean carbonate system (pCO_2 , DIC, pH, Ω_{Ar} ,
344 Ω_{Ca}). Under all four RCPs, pH and Ω_{Ar} decrease significantly outside of their preindustrial values
345 matching both observations and CMIP5. In the near future the open ocean and coral reef communities
346 are likely to experience pH and carbonate saturation levels unprecedented in the last 2 million years
347 (Hönisch et al., 2009).

348 This study is timely because the CMIP5 archive includes a large suite of ESMs that contained
349 dynamic biogeochemistry, allowing us to study future projections of the marine carbon cycle. Rather
350 than running the ESMs, we can use Hector to quickly emulate the global CMIP5 median for projection
351 studies under different emission pathways and sensitivity analyses of the marine carbonate system.

352 ~~Overall~~Within this study, we find that numerous parameters influence $[H^+]$ and Ω_{Ar} , with both being
353 sensitive to Q_{10} .~~have the most impact on future changes in ocean acidification.~~ Due to Hector's
354 simplistic nature and fast execution times, Hector has the potential to be a critical tool to the decision-
355 making, scientific, and integrated assessment communities, allowing for further understanding of future
356 changes to the marine carbonate system.

357 **Appendix A: Model Description – carbon cycle**

358 The carbon component in Hector contains three carbon reservoirs: a single well-mixed atmosphere,
359 a land component and an ocean component. The change in atmospheric carbon is a function of the
360 anthropogenic emissions (F_A), land-use change emissions (F_{LC}), and atmosphere-ocean (F_O) and
361 atmosphere-land (F_L) carbon fluxes. The default model time step is 1 year.

$$\frac{dC_{atm}(t)}{dt} = F_A(t) + F_{LC}(t) - F_O(t) - F_L(t) \quad (1)$$

362 The terrestrial cycle in Hector contains vegetation, detritus, and soil, all linked to each other and
363 the atmosphere by first-order differential equations. Vegetation net primary production is a function of
364 atmospheric CO_2 and temperature. Carbon flows from the vegetation to detritus to soil and loses
365 fractions of carbon to heterotrophic respiration on the way. An ‘earth’ pool debits carbon emitted as
366 anthropogenic emissions, allowing a continual mass-balance check across the entire carbon cycle.

367 Atmosphere-land fluxes at time t are calculated by:

$$F_L(t) = \sum_{i=1}^n NPP_i(t) - RH_i(t) \quad (2)$$

368 where NPP is the net primary production and RH is the heterotrophic respiration summed over user-
369 specified n groups (i.e., latitude bands, political units, or biomes) (Hartin et al 2015).

370

371 **Appendix B: Ocean Carbonate Chemistry**

372 The ocean’s inorganic carbon system is solved via a series of equations modified from Zeebe and Wolf-
373 Gladrow (2001). TA and DIC are used to calculate the other variables of the carbonate system:

$$DIC * \left(\frac{K_1}{[H^+]} + 2 \frac{K_1 K_2}{[H^+]^2} \right) = \left(TA - \frac{K_B B_T}{K_B + [H^+]} - \frac{K_W}{[H^+]} + [H^+] \right) * \left(1 + \frac{K_1}{[H^+]} + \frac{K_1 K_2}{[H^+]^2} \right) \quad (1)$$

374 This equation results in a higher order polynomial equation for H^+ , in which the roots (1 positive, 4
375 negative) are solved for. Once H^+ is solved for, pH , pCO_2 , HCO_3^- , and CO_3^{2-} can be determined. We

376 ignore the nonideality of CO₂ in air and therefore use the partial pressure of CO₂ instead of the fugacity
 377 of CO₂. Fugacity is slightly lower by ~0.3% compared to pCO₂ (Riebesell et al., 2009; Sarmiento and
 378 Gruber, 2006).

$$[CO_2^*] = \frac{DIC}{\left(1 + \frac{K_1}{[H^+]} + \frac{K_1 K_2}{[H^+]^2}\right)} \quad (2)$$

$$pCO_2 = \frac{[CO_2^*]}{K_H} \quad (3)$$

$$[HCO_3^-] = \frac{DIC}{\left(1 + \frac{[H^+]}{K_1} + \frac{K_2}{[H^+]}\right)} \quad (4)$$

$$[CO_3^{2-}] = \frac{DIC}{\left(1 + \frac{[H^+]}{K_2} + \frac{[H^+]^2}{K_1 K_2}\right)} \quad (5)$$

$$K_1 = \frac{[H^+][HCO_3^-]}{[CO_2]} \quad (6)$$

$$K_2 = \frac{[H^+][CO_3^{2-}]}{[HCO_3^-]} \quad (7)$$

379 K₁ and K₂ are the first and second acidity constants of carbonic acid from Mehrbach et al. (1973) and
 380 refit by Lueker et al. (2000).

$$K_B = \frac{[H^+][B(OH)_4^-]}{[B(OH)_3]} \quad (8)$$

381 K_B is the dissociation constant of boric acid from DOE (1994).

$$B = 416.0 * \frac{S}{35.0} \quad (9)$$

383 Total boron is from DOE (1994)

$$K_W = \frac{[H^+]}{[OH^-]} \quad (10)$$

384 K_w is the dissociation constant of water from Millero (1995).

$$K_{sp} = [Ca^{2+}] * [CO_3^{2-}] \quad (11)$$

385 K_{sp} of aragonite and calcite is calculated from Mucci, (1983).

386

387 For those equations with multiple coefficients:

388 1) K_H and K_0 are similar equations calculating Henry's constant or the solubility of CO_2 , but they return
389 different units ($mol\ kg^{-1}\ atm^{-1}$ and $mol\ L^{-1}\ atm^{-1}$) (see Weiss, 1974 for equations and coefficients). K_H
390 is used to solve pCO_2 while K_0 is used to solve air-sea fluxes of CO_2 .

391 2) The Schmidt number is taken from Wanninkhof (1992) for coefficients of CO_2 in seawater.

392 3) $[Ca^{2+}]$ ($mol\ kg^{-1}$) is calculated from Riley and Tongudai (1967).

393

394 **Acknowledgements**

395 This research is based on work supported by the US Department of Energy. The Pacific Northwest
396 National Laboratory is operated for DOE by Battelle Memorial Institute under contract DE-AC05-
397 76RL01830.

398

399 **Author Contributions**

400 C.Hartin designed and carried out the experiments. C. Hartin, B.Bond-Lamberty, and P.Patel
401 developed the model code. A.Mundra processed the data and prepared the figures. C.Hartin
402 prepared the manuscript with contributions from all co-authors.

403

404 **Table 1:** Description and values of ocean parameters in Hector [and parameters involved in the](#)
 405 [sensitivity analysis](#).

Description	Value	Notes
* Albedo ^t	-0.2 Wm⁻²	Constant global albedo from 1950 - 2300
Area of ocean	3.6e14 m ²	Knox and McElroy, (1984)
* Beta ^t	0.36	Terrestrial CO₂ fertilization Wullschleger et al. (1995)
* Climate Sensitivity ^t	3.0 °C	
Fractional area of HL	0.15	Sarmiento and Toggweiler, (1984)
Fractional area of LL	0.85	Sarmiento and Toggweiler, (1984)
* Q₁₀ ^t	2.0	Terrestrial respiration temperature response Davidson and Janssens (2006)
Thickness of surface ocean	100 m	Knox and McElroy, (1984)
Thickness of intermediate ocean	900 m	
Thickness of deep ocean	2677 m	Total ocean depth 3777m
Volume of HL	5.4e15 m ³	
Volume of LL	3.06e16 m ³	
Volume of IO	3.24e17 m ³	
Volume of DO	9.64e17 m ³	
Surface Area of HL	5.4e13 m ²	
Surface Area of LL	3.06e14 m ²	
Salinity HL and LL ^l	34.5	
Initial temperature of HL ^t	2.0 °C	

<u>Initial temperature of LL[†]</u>	<u>22.0 °C</u>	
*Thermohaline circulation (T _T) [†]	7.2e7 m ³ s ⁻¹	Tuned to give ~100 Pg C flux from surface to deep
*High latitude circulation (T _H)	4.9e7 m ³ s ⁻¹	Tuned to give ~100 Pg C flux from surface to deep
*Water mass exchange (intermediate to deep - E _{ID})	1.25e7 m ³ s ⁻¹	Lenton (2000); Knox and McElroy (1984)
*Water mass exchange (low latitude to intermediate - E _{IL})	2.08e7 m ³ s ⁻¹	Lenton (2000); Knox and McElroy (1984)
Wind speed HL and LL [†]	6.7 m s ⁻¹	Takahashi et al. (2009); Liss and Merlivat, (1986)

406 * Parameters contained within the input file.

407 [†] Parameters varied for the sensitivity analysis.

408 **Table 2:** CMIP5 ESM models used in this study containing ocean carbonate parameters. Ω_{Ar} , Ω_{Ca} were
 409 calculated from the model sea surface temperature, sea surface salinity, and CO_3 concentrations.

Model	Parameters (RCP 8.5)
BCC-cm1-1	pCO ₂ [*] , temperature
BNU-ESM	pCO ₂
CanESM2	DIC, pH, salinity
CESM1-BGC	CO ₃ , DIC, pH, salinity
CMCC-CESM	pCO ₂ , temperature, CO ₃ , DIC, pH, salinity
CNRM-CM5	CO ₃ , DIC
GFDL-ESM2G	pCO ₂ , temperature, pH, salinity
GFDL-ESM2M	pCO ₂ , temperature, CO ₃ , pH, DIC, salinity
GISS-E2-H-CC	pCO ₂ , temperature, DIC, salinity
GISS-E2-R-CC	pCO ₂ , temperature, DIC, salinity
HadGEM2-CC	pCO ₂ , temperature, CO ₃ , DIC, pH, salinity
HadGEM2-ES	pCO ₂ , temperature, CO ₃ [*] , DIC [*] , pH, salinity
IPSL-CM5A-LR	Temperature [*] , CO ₃ [*] , DIC [*] , pH [*] , salinity [*]
IPSL-CM5A-MR	Temperature, CO ₃ , DIC, pH, salinity
IPSL-CM5B-LR	Temperature, CO ₃ , DIC, pH, salinity
MIROC-ESM	pCO ₂ , temperature, salinity
MIROC-ESM- CHEM	pCO ₂ , temperature, salinity
MPI-ESM-LR	pCO ₂ [*] , temperature [*] , CO ₃ [*] , DIC [*] , pH [*] , salinity [*]
MPI-ESM-MR	pCO ₂ , temperature, CO ₃ , DIC, pH, salinity
MRI-ESM1	pCO ₂ , temperature
NorESM1-ME	pCO ₂ , temperature, CO ₃ , DIC, pH, salinity

410 * Variable output to 2300.

411 **Table 3:** Observational time-series information and carbonate parameters from each location.
 412

Time-Series Site	Location	Time-Series Length	Reference	Ocean Carbon Parameters	Data Access
BATS	Sargasso Sea	1988-2011	Bates, (2007)	TA, DIC	http://www.bios.edu/research/projects/bats
HOT	North Pacific	1988-2011	Dore et al., (2009)	TA, DIC, pH, pCO ₂ , Ω _{Ar} , Ω _{Ca}	http://hahana.soest.hawaii.edu/hot/hot_jgofs.html
ESTOC	Canary Islands	1995-2009	Gonzalez-Davila, (2009)	TA, pH, pCO ₂	http://www.eurosites.info/estoc.php
Iceland Sea	Iceland Sea	1985-2013	Olafsson, (2007a)	DIC, pCO ₂	http://cdiac.ornl.gov/oceans/Mooring/Iceland_Sea.html
Irminger Sea	Irminger Sea	1983-2013	Olafsson, (2007b)	DIC, pCO ₂	http://cdiac.ornl.gov/oceans/Mooring/Irminger_Sea.html
Flinders Reef	Coral Sea	1708-1988	Pelejero et al., (2005)	pH, Ω _{Ar}	ftp://ftp.ncdc.noaa.gov/pub/data/paleo/coral/west_pacific/great_barrier/flinders2005.txt

413

414 **Table 4:** Model validation metrics for the a) high latitude and b) low latitude ocean carbonate variables
 415 comparing Hector to CMIP5 from 1850-2005.
 416

a)	RMSE	R2	Bias
DIC	10.00	0.26	47.10
pCO ₂	2.65	0.98	-31.78
pH	0.004	0.975	0.061
Ω _{Ar}	0.01	0.98	0.37
Ω _{Ca}	0.02	0.98	0.58

b)	RMSE	R2	Bias
DIC	6.50	0.76	101.28
pCO ₂	3.43	0.98	-4.62
pH	0.004	0.966	0.025
Ω _{Ar}	0.02	0.97	0.36
Ω _{Ca}	0.03	0.97	0.53

417
 418
 419
 420
 421
 422

Table 5: Absolute values and rates of change per year (Δ) for the a) high and b) low latitude surface ocean from 1850-2100 and 2101-2300 under RCP 8.5 for DIC (μmol kg⁻¹), pCO₂ (μatm), ~~total~~ pH (total scale, unitless), Ω_{Ar} (unitless) and Ω_{Ca} (unitless).

a)	DIC			pCO ₂			pH			Ω _{Ar}			Ω _{Ca}		
	1850	2100	2300	1850	2100	2300	1850	2100	2300	1850	2100	2300	1850	2100	2300
Hector	2107.5	2258.1	2335.5	244.7	816.6	1732.1	8.23	7.76	7.46	2.2	1.0	0.6	3.5	1.5	0.9
Δ		0.602	0.387		2.29	4.58		-	-		-	-0.002		-0.008	-0.003
								0.0019	0.0015		0.0048				
CMIP5	2104.50	2175.79	2243.41	271.62	871.00	1903.82	8.17	7.70	7.38	1.82	0.75	0.44	2.90	1.20	0.70
Δ		0.285	0.34		2.40	5.16		-	-		-	-		-	-
								0.0019	0.0016		0.0012	0.0016		0.0068	0.0025

423

b)	DIC			pCO ₂			pH			Ω _{Ar}			Ω _{Ca}		
	1850	2100	2300	1850	2100	2300	1850	2100	2300	1850	2100	2300	1850	2100	2300
Hector	2073.9	2264.1	2357.6	294.7	879.6	1766.5	8.17	7.77	7.50	4.1	2.2	1.4	6.2	3.3	2.1
Δ		0.76	0.47		2.34	4.43		-	-		-	-		-	-0.006
								0.0016	0.0014		0.0076	0.0040		0.0116	
CMIP5	1997.57	2163.16	2298.89	290.47	930.92	1965.23	8.16	7.73	7.45	3.75	2.00	1.36	5.77	3.02	2.04
Δ		0.66	0.68		2.56	5.17		-	-		-	-		-	-0.0049
								0.0011	0.0014		0.0070	0.0032		0.0110	

424

425 **Table 6:** Trends and standard error for the carbonate system taken from Bates et al., (2014) and values
 426 calculated from Hector and CMIP5. ~~Global carbonate system parameters for Hector and CMIP5 from~~
 427 ~~1988-2014.~~
 428

	<u>Length of record</u>	DIC ($\mu\text{mol kg}^{-1}\text{ yr}^{-1}$)	pCO ₂ ($\mu\text{atm yr}^{-1}$)	pH (yr^{-1})	Ω_{Ar} (yr^{-1})
BATS	<u>1983-2014</u>	1.37 ± 0.07	1.69 ± 0.11	-0.0017 ± 0.0001	-0.0095 ± 0.0007
HOT	<u>1988-2014</u>	1.78 ± 0.12	1.72 ± 0.09	-0.0016 ± 0.0001	-0.0084 ± 0.0011
ESTOC	<u>1995-2014</u>	1.09 ± 0.10	1.92 ± 0.24	-0.0018 ± 0.0002	-0.0115 ± 0.0023
Iceland Sea	<u>1983-2014</u>	1.22 ± 0.27	1.29 ± 0.36	-0.0014 ± 0.0005	-0.0018 ± 0.0027
Irminger Sea	<u>1983-2014</u>	1.62 ± 0.35	2.37 ± 0.49	-0.0026 ± 0.0006	-0.0080 ± 0.0040
Hector	<u>1988-2014</u>	0.90	1.82	-0.0017	-0.0089
CMIP5	<u>1988-2014</u>	0.68	1.77	-0.0018	-0.0074

429
430

431 **Table 7:** Percentage change from reference (RCP8.5) for two Hector outputs a) global $[H^+]pH$ and b)
 432 global Ω_{Ar} for a $\pm 10\%$ change in eight model parameters. Results are shown for three years, 2005, 2100
 433 and 2300.
 434

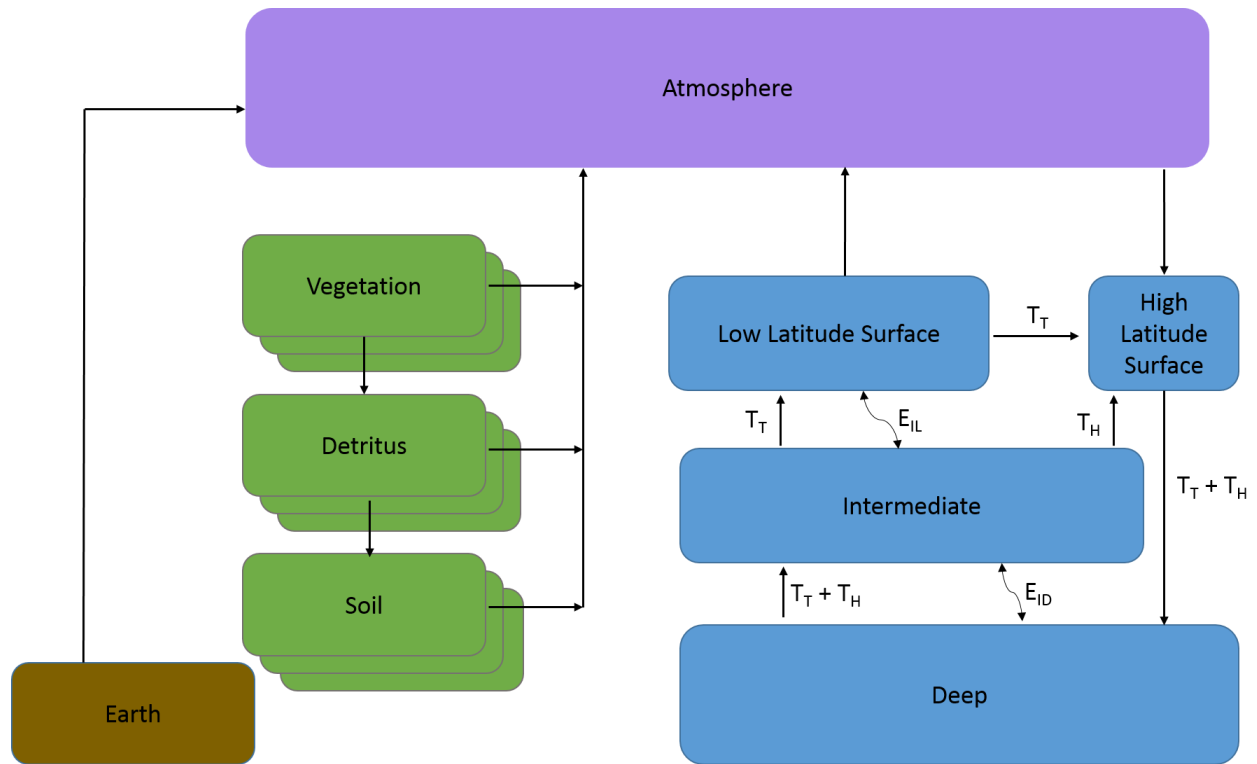
a) Year	Parameter	+10% change	-10% change
2005	Albedo	0.13	0.00
2100		0.59	0.00
2300		0.00	-0.31
2005	Beta (<u>terrestrial CO₂ fertilization</u>)	0.63	-0.50
2100		1.78	-1.78
2300		1.56	-1.88
2005	<u>Ocean</u> Circulation (<u>T_T</u>)	0.76	-0.76
2100		2.37	-1.78
2300		2.81	-3.44
2005	Q ₁₀ (<u>terrestrial respiration</u> <u>temperature response</u>)	-0.13	0.13
2100		-0.59	1.18
2300		-3.13	3.44
2005	Salinity	-0.88	1.51
2100		0.59	0.59
2300		1.88	-1.25
2005	Climate Sensitivity	-0.13	0.13
2100		-0.59	0.43
2300		-2.5	2.50
2005	Surface ocean temperature	0.00	0.00
2100		0.00	0.59
2300		0.62	0.31
2005	Wind Stress	-0.38	0.63
2100		-0.59	1.78
2300		-1.25	1.25

435

b) Year	Parameter	+10% change	-10% change
2005	Albedo	0.01	-0.00
2100		0.01	-0.01
2300		-0.00	0.00
2005	Beta (<u>terrestrial CO₂ fertilization</u>)	0.38	-0.40
2100		1.33	-1.34
2300		1.38	-1.35
2005	<u>Ocean</u> Circulation (<u>T_T</u>)	0.41	-0.45
2100		1.01	-1.05
2300		1.48	-1.55
2005	Q ₁₀ (<u>terrestrial respiration</u> <u>temperature response</u>)	-0.09	0.10
2100		-0.87	0.95
2300		-2.40	3.00
2005	Salinity	3.80	-4.28
2100		5.60	-5.89
2300		7.17	-7.18
2005	Climate Sensitivity	0.07	-0.07
2100		0.55	-0.56
2300		0.43	-0.27
2005	Surface ocean temperature	2.07	-1.99
2100		2.41	-2.29
2300		2.43	-2.27
2005	Wind Stress	-0.18	0.25
2100		-0.65	0.88
2300		-1.13	0.88

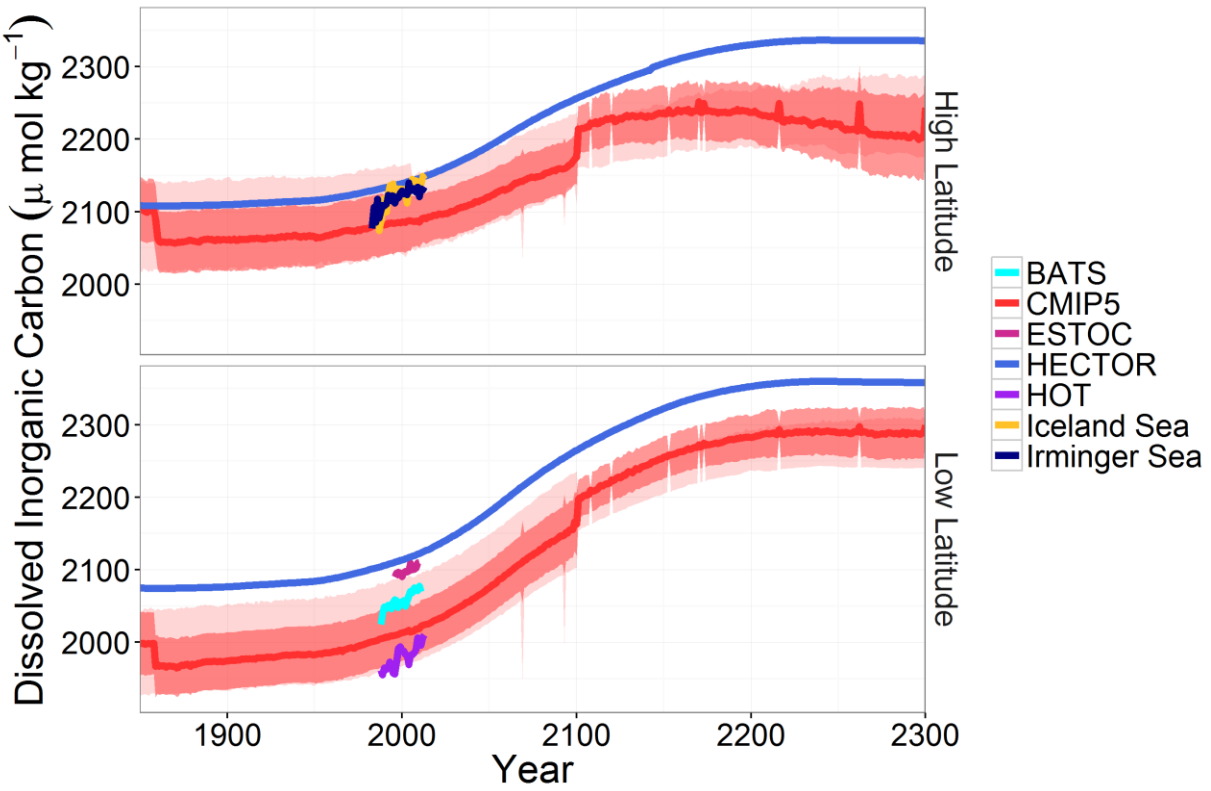
436

437 **Figure 1:** Representation of the carbon cycle in Hector. The atmosphere consists of one well-mixed box,
 438 connected to the surface ocean via air-sea fluxes of carbon. The terrestrial component consists of user
 439 defined biomes or regions for vegetation, detritus, and soil. The earth pool is continually debited to act
 440 as a mass balance check on the carbon cycle (Hartin et al., 2015). The ocean consists of four boxes, with
 441 advection (represented by **solid-straight** arrows) and water mass exchange (represented by **dashed**
 442 **curved** arrows) simulating thermohaline circulation. The marine carbonate system is solved for in the
 443 high and low latitude surface boxes. At steady state, there is a flux of carbon from the atmosphere to the
 444 the high latitude surface box, while the low-latitude surface ocean releases carbon to the atmosphere.
 445



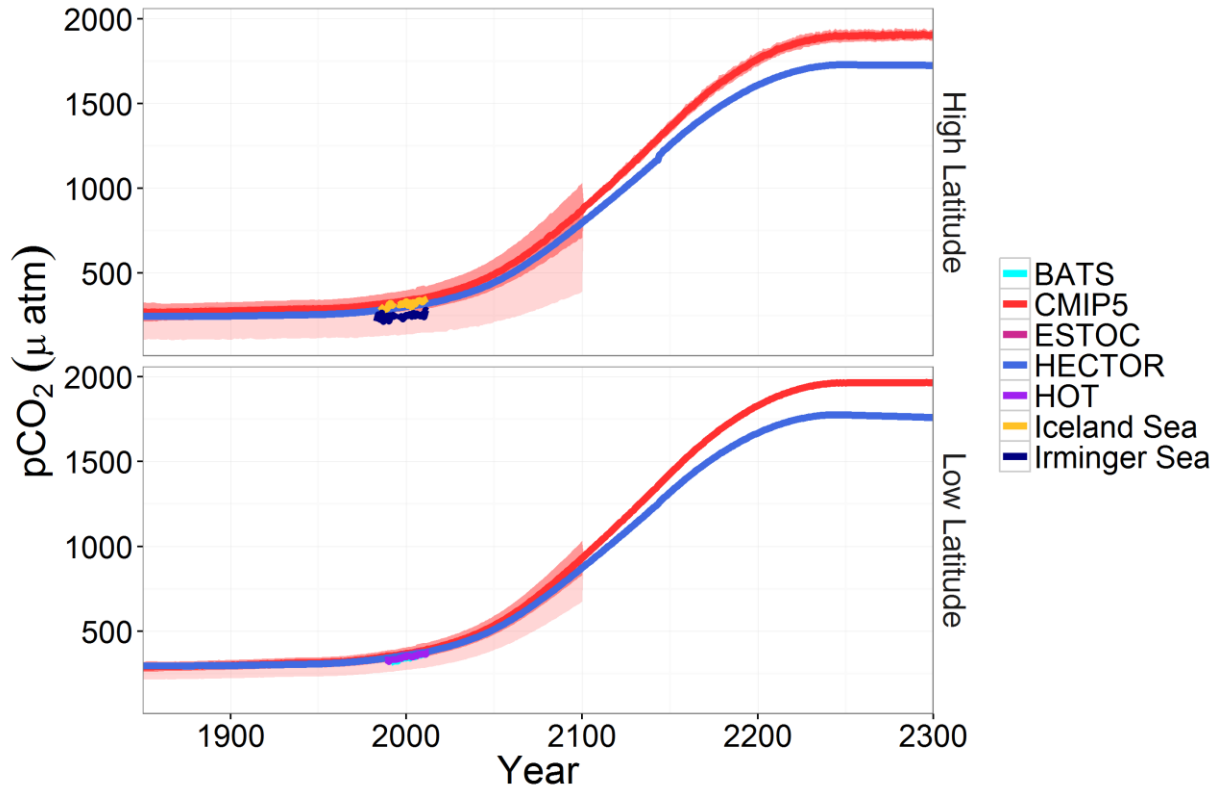
446

447 **Figure 2:** Dissolved inorganic carbon (DIC) for high (top) and low latitude (bottom) surface ocean under
 448 RCP 8.5; Hector (blue), CMIP5 median, standard deviation, and model range (red, $n = 15$ (18650-2100,
 449 with $n=4$ starting after 1860) and $n= 3$ (2101-2300)); and observations from BATS (teal), ESTOC (pink),
 450 HOT (purple), Iceland (yellow) and Irminger Sea (navy). Note a doubling of CO_2 from preindustrial
 451 values occurs around 2050.
 452



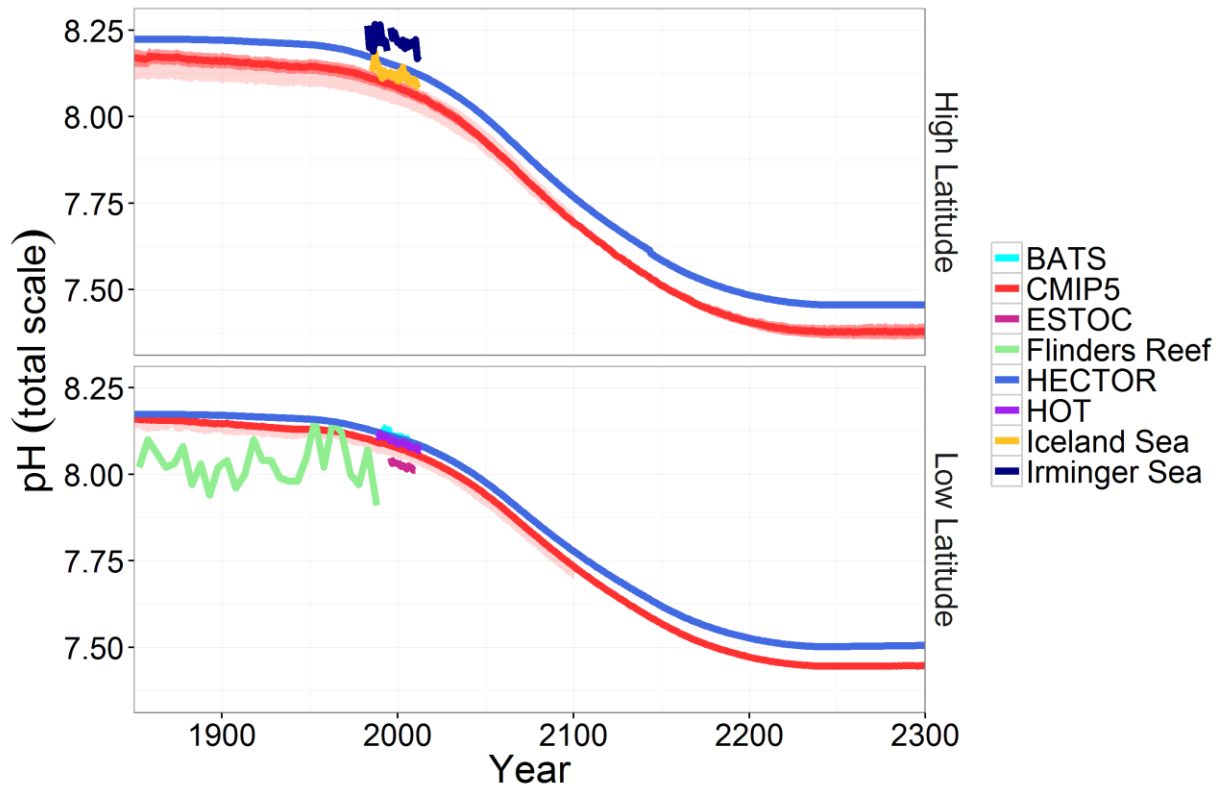
453

454 **Figure 3:** pCO₂ for high (top) and low latitude (bottom) surface ocean under RCP 8.5; Hector (blue),
455 CMIP5 median, standard deviation, and model range (red, *n* = 15 (1850-2100) and *n* = 2 (2101-2300));
456 and observations from BATS (teal), HOT (purple), ESTOC (pink), Iceland (yellow) and Irminger Sea (navy).



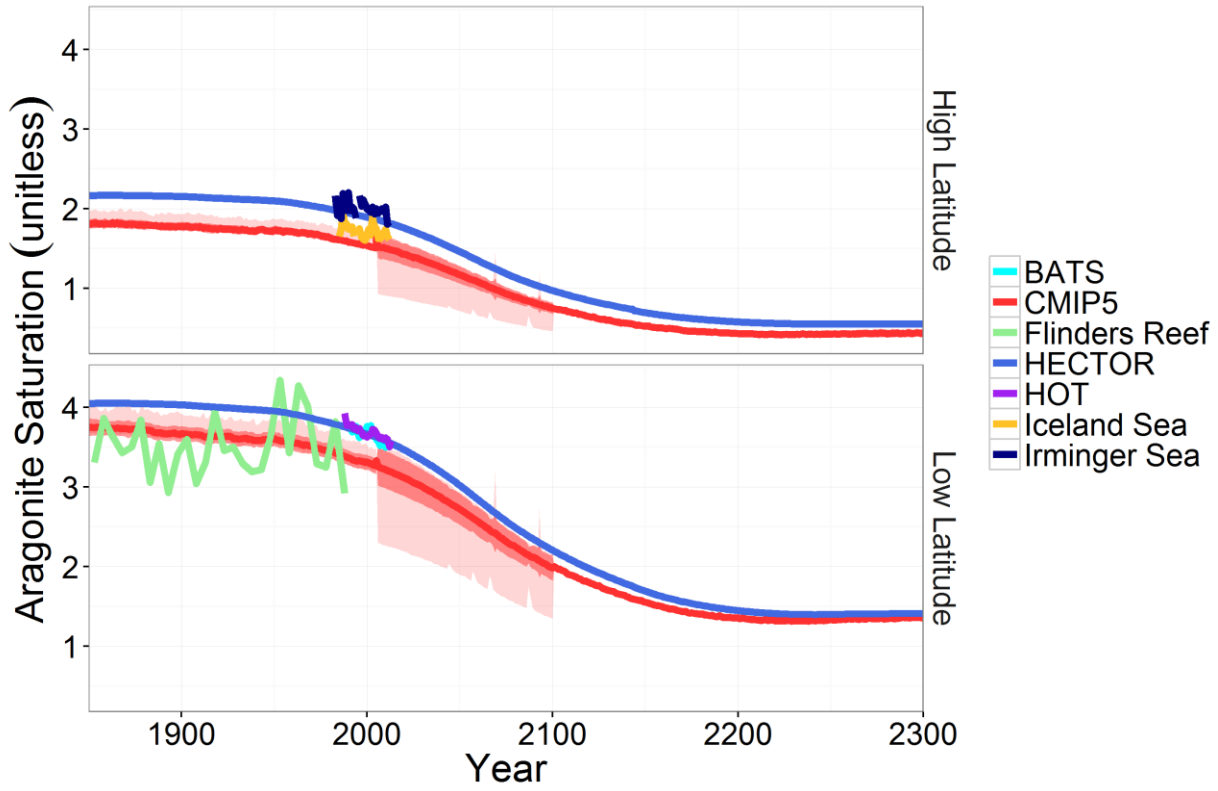
457

458 **Figure 4:** pH for high (top) and low latitude (bottom) surface ocean under RCP 8.5; Hector (blue), CMIP5
 459 median, standard deviation, and model range (red, $n = 13$ (1850-2100) and $n= 2$ (2101-2300)); and
 460 observations from BATS (teal), ESTOC (pink), HOT (purple) Flinder's Reef (green), Iceland (yellow) and
 461 Irminger Sea (navy).



462

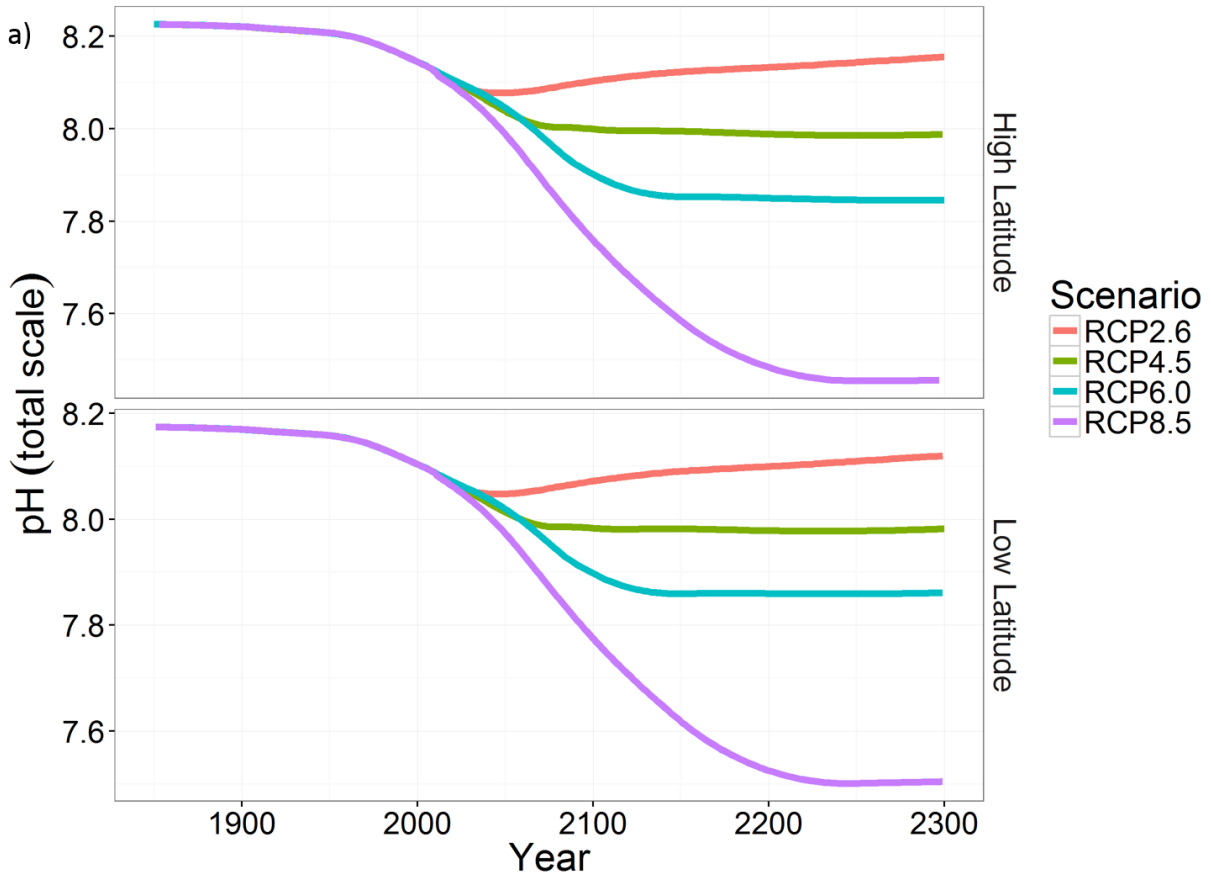
463 **Figure 5:** Aragonite saturation (Ω_{Ar}) for high (top) and low latitude (bottom) surface ocean under RCP
 464 8.5; Hector (blue), CMIP5 median, standard deviation, and model range (red, $n = 10$ (1850-2100) and $n=$
 465 2 (2101-2300)); and observations from BATS (teal), HOT (purple) and Flinder's Reef (green).



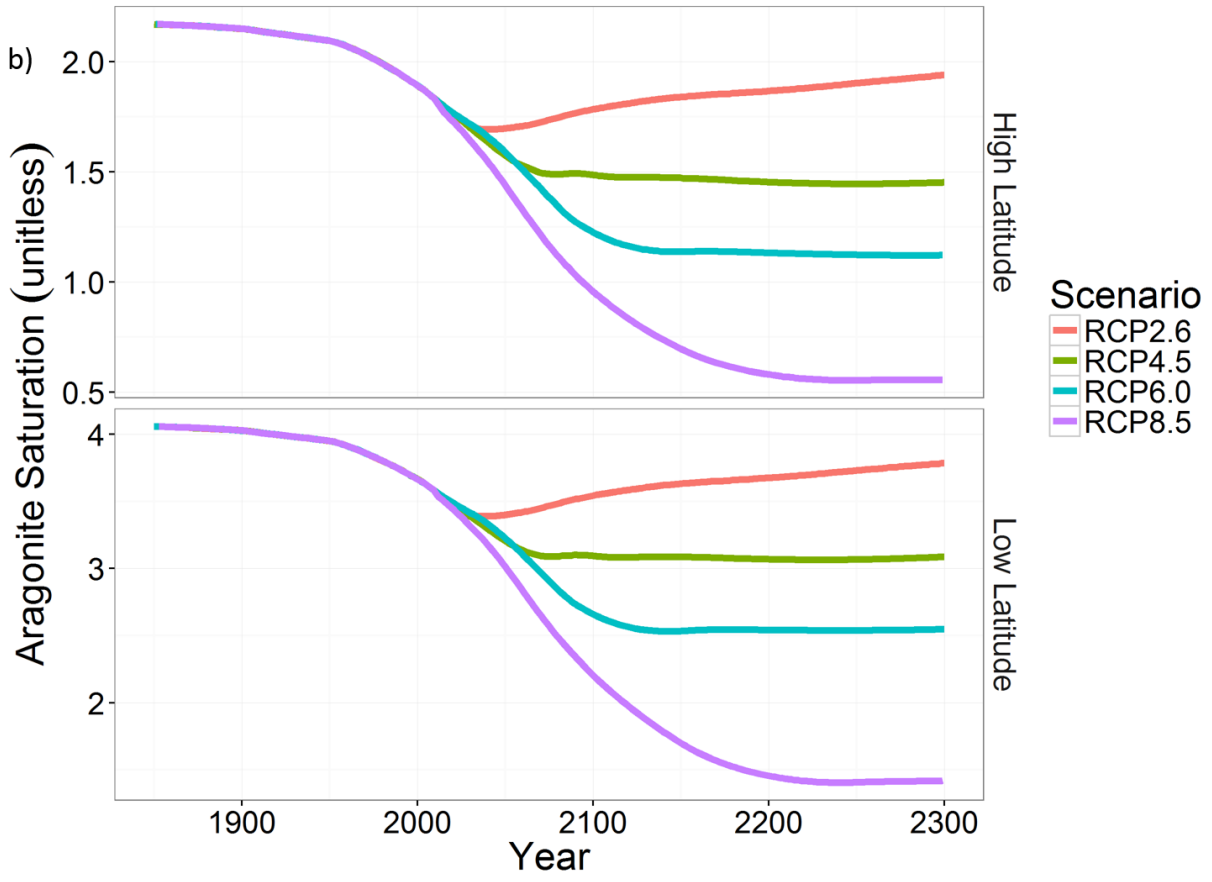
466

467

468 **Figure 6:** High and low latitude a) pH and b) aragonite saturation (Ω_{Ar}) time series for Hector from 1850-
469 2300 for RCP 2.6 (red), RCP 4.5 (green), RCP 6.0 (teal) and RCP 8.5 (purple). Note that even under a
470 strongly mitigated scenario (RCP 2.6), both Ω_{Ar} and pH at 2300 are still lower than preindustrial values.



471



472

473

474 REFERENCES

- 475 Albright, R., Langdon, C., and Anthony, K. R. N.: Dynamics of seawater carbonate chemistry, production,
476 and calcification of a coral reef flat, central Great Barrier Reef, *Biogeosciences*, 10, 6747-6758,
477 10.5194/bg-10-6747-2013, 2013.
- 478 Archer, D., Eby, M., Brovkin, V., Ridgwell, A., Cao, L., Mikolajewicz, U., Caldeira, K., Matsumoto, K.,
479 Munhoven, G., Montenegro, A., and Tokos, K.: Atmospheric Lifetime of Fossil Fuel Carbon Dioxide,
480 *Annual Review of Earth and Planetary Sciences*, 37, 117-134,
481 doi:10.1146/annurev.earth.031208.100206, 2009.
- 482 Bates, N. R.: Interannual variability of the oceanic CO₂ sink in the subtropical gyre of the North Atlantic
483 Ocean over the last 2 decades, *Journal of Geophysical Research: Oceans*, 112, C09013,
484 10.1029/2006JC003759, 2007.
- 485 Bates, N. R., Astor, Y. M., Church, M. J., Currie, K., Dore, J. E., Gonzalez-Davila, M., Lorenzoni, L., Muller-
486 Karger, F., Olafsson, J., and Santana-Casiano, J. M.: A time-series view of changing ocean chemistry due
487 to ocean uptake of anthropogenic CO₂ and ocean acidification, *Oceanography*, 27, 126-141,
488 <http://dx.doi.org/10.5670/oceanog.2014.16>, 2014.
- 489 Bopp, L., Resplandy, L., Orr, J. C., Doney, S. C., Dunne, J. P., Gehlen, M., Halloran, P., Heinze, C., Ilyina, T.,
490 Séférian, R., Tjiputra, J., and Vichi, M.: Multiple stressors of ocean ecosystems in the 21st century:
491 projections with CMIP5 models, *Biogeosciences Discuss.*, 10, 3627-3676, 10.5194/bgd-10-3627-2013,
492 2013.
- 493 Byrne, R. H., Mecking, S., Feely, R. A., and Liu, X.: Direct observations of basin-wide acidification of the
494 North Pacific Ocean, *Geophysical Research Letters*, 37, L02601, 10.1029/2009GL040999, 2010.
- 495 Caldeira, K., Jain, A. K., and Hoffert, M. I.: Climate Sensitivity Uncertainty and the Need for Energy
496 Without CO₂ Emission, *Science*, 299, 2052-2054, 10.1126/science.1078938, 2003.
- 497 Cheng, W., Chiang, J. C. H., and Zhang, D.: Atlantic Meridional Overturning Circulation (AMOC) in CMIP5
498 Models: RCP and Historical Simulations, *Journal of Climate*, 26, 7187-7197, 10.1175/JCLI-D-12-00496.1,
499 2013.
- 500 Cooley, S. R., and Doney, S. C.: Anticipating ocean acidification's economic consequences for commercial
501 fisheries, *Environmental Research Letters*, 4, 024007, 2009.
- 502 Davidson, E. A., and Janssens, I. A.: Temperature sensitivity of soil carbon decomposition and feedbacks
503 to climate change, *Nature*, 440, 165-173, 2006.
- 504 Dickson, A. G., and Millero, F. J.: A comparison of the equilibrium constants for the dissociation of
505 carbonic acid in seawater media, *Deep Sea Research Part A. Oceanographic Research Papers*, 34, 1733-
506 1743, [http://dx.doi.org/10.1016/0198-0149\(87\)90021-5](http://dx.doi.org/10.1016/0198-0149(87)90021-5), 1987.
- 507 DOE: Handbook of methods for the analysis of the various parameters of the carbon dioxide system in
508 sea water, edited by: Dickson, A. G., and Goyet, C., ORNL/CDIAC-74, 1994.
- 509 Doney, S. C.: The Growing Human Footprint on Coastal and Open-Ocean Biogeochemistry, *Science*, 328,
510 1512-1516, 10.1126/science.1185198, 2010.
- 511 Dore, J. E., Lukas, R., Sadler, D. W., Church, M. J., and Karl, D. M.: Physical and biogeochemical
512 modulation of ocean acidification in the central North Pacific, *Proceedings of the National Academy of
513 Sciences*, 106, 12235-12240, 10.1073/pnas.0906044106, 2009.
- 514 Eglinton, T. I., and Repeta, D. J.: Organic Matter in the Contemporary Ocean, *Treatise on Geochemistry*,
515 edited by: Holland, H. D., and Turekian, K. K., Elsevier, Amsterdam, 2004.
- 516 Ekstrom, J. A., Suatoni, L., Cooley, S. R., Pendleton, L. H., Waldbusser, G. G., Cinner, J. E., Ritter, J.,
517 Langdon, C., van Hooijdonk, R., Gledhill, D., Wellman, K., Beck, M. W., Brander, L. M., Rittschof, D.,
518 Doherty, C., Edwards, P. E. T., and Portela, R.: Vulnerability and adaptation of US shellfisheries to ocean
519 acidification, *Nature Clim. Change*, 5, 207-214, 10.1038/nclimate2508

520 <http://www.nature.com/nclimate/journal/v5/n3/abs/nclimate2508.html#supplementary-information>,
521 2015.

522 Fabry, V. J., Seibel, B. A., Feely, R. A., and Orr, J. C.: Impacts of ocean acidification on marine fauna and
523 ecosystem processes, *ICES Journal of Marine Science: Journal du Conseil*, 65, 414-432,
524 10.1093/icesjms/fsn048, 2008.

525 Feely, R. A., Sabine, C. L., Lee, K., Berelson, W., Kleypas, J., Fabry, V. J., and Millero, F. J.: Impact of
526 Anthropogenic CO₂ on the CaCO₃ System in the Oceans, *Science*, 305, 362-366,
527 10.1126/science.1097329, 2004.

528 Feely, R. A., Doney, S. C., and Cooley, S. R.: Ocean acidification: present conditions and future changes in
529 a high-CO₂ world, *Oceanography*, 22, 36-47, <http://dx.doi.org/10.5670/oceanog.2009.95>, 2009.

530 Feely, R. A., Sabine, C. L., Byrne, R. H., Millero, F. J., Dickson, A. G., Wanninkhof, R., Murata, A., Miller, L.
531 A., and Greeley, D.: Decadal changes in the aragonite and calcite saturation state of the Pacific Ocean,
532 *Global Biogeochemical Cycles*, 26, GB3001, 10.1029/2011GB004157, 2012.

533 Friedlingstein, P., Meinshausen, M., Arora, V. K., Jones, C. D., Anav, A., Liddicoat, S. K., and Knutti, R.:
534 Uncertainties in CMIP5 Climate Projections due to Carbon Cycle Feedbacks, *Journal of Climate*, 27, 511-
535 526, 10.1175/JCLI-D-12-00579.1, 2014.

536 Fry, C. H., Tyrrell, T., Hain, M. P., Bates, N. R., and Achterberg, E. P.: Analysis of global surface ocean
537 alkalinity to determine controlling processes, *Marine Chemistry*, 174, 46-57,
538 <http://dx.doi.org/10.1016/j.marchem.2015.05.003>, 2015.

539 Gehlen, M., Séférian, R., Jones, D. O. B., Roy, T., Roth, R., Barry, J., Bopp, L., Doney, S. C., Dunne, J. P.,
540 Heinze, C., Joos, F., Orr, J. C., Resplandy, L., Segsneider, J., and Tjiputra, J.: Projected pH reductions by
541 2100 might put deep North Atlantic biodiversity at risk, *Biogeosciences Discuss.*, 11, 8607-8634,
542 10.5194/bgd-11-8607-2014, 2014.

543 Glotter, M., Pierrehumbert, R., Elliott, J., Matteson, N., and Moyer, E.: A simple carbon cycle
544 representation for economic and policy analyses, *Climatic Change*, 126, 319-335, 10.1007/s10584-014-
545 1224-y, 2014.

546 Hansell, D. A., and Carlson, C. A.: Marine dissolved organic matter and the carbon cycle, *Oceanography*,
547 14, 41-49, 2001.

548 Hartin, C. A., Patel, P., Schwarber, A., Link, R. P., and Bond-Lamberty, B. P.: A simple object-oriented and
549 open-source model for scientific and policy analyses of the global climate system – Hector v1.0, *Geosci.*
550 *Model Dev.*, 8, 939-955, 10.5194/gmd-8-939-2015, 2015.

551 Held, I. M., and Soden, B. J.: Robust Responses of the Hydrological Cycle to Global Warming, *Journal of*
552 *Climate*, 19, 5686-5699, doi:10.1175/JCLI3990.1, 2006.

553 Hoegh-Guldberg, O., Mumby, P. J., Hooten, A. J., Steneck, R. S., Greenfield, P., Gomez, E., Harvell, C. D.,
554 Sale, P. F., Edwards, A. J., Caldeira, K., Knowlton, N., Eakin, C. M., Iglesias-Prieto, R., Muthiga, N.,
555 Bradbury, R. H., Dubi, A., and Hatziolos, M. E.: Coral Reefs Under Rapid Climate Change and Ocean
556 Acidification, *Science*, 318, 1737-1742, 10.1126/science.1152509, 2007.

557 Hönisch, B., Hemming, N. G., Archer, D., Siddall, M., and McManus, J. F.: Atmospheric Carbon Dioxide
558 Concentration Across the Mid-Pleistocene Transition, *Science*, 324, 1551-1554,
559 10.1126/science.1171477, 2009.

560 IPCC: Climate Change 2013: The Physical Science Basis. Contribution of Working Group I to the Fifth
561 Assessment Report of the Intergovernmental Panel on Climate Change, edited by: Stocker, T. F., D. Qin,
562 G.-K. Plattner, M. Tignor, S.K. Allen, J. Boschung, A. Nauels, Y. Xia, V. Bex and P.M. Midgley Cambridge
563 University Press, Cambridge, United Kingdom and New York, NY, USA, 1535 pp., 2013.

564 Irvine, P. J., Sriver, R. L., and Keller, K.: Tension between reducing sea-level rise and global warming
565 through solar-radiation management, *Nature Clim. Change*, 2, 97-100, doi:10.1038/nclimate1351, 2012.

566 Joos, F., Plattner, G.-K., Stocker, T. F., Marchal, O., and Schmittner, A.: Global Warming and Marine
567 Carbon Cycle Feedbacks on Future Atmospheric CO₂, *Science*, 284, 464-467,
568 10.1126/science.284.5413.464, 1999.

569 Key, R. M., Kozyr, A., Sabine, C. L., Lee, K., Wanninkhof, R., Bullister, J. L., Feely, R. A., Millero, F. J.,
570 Mordy, C., and Peng, T. H.: A global ocean carbon climatology: Results from Global Data Analysis Project
571 (GLODAP), *Global Biogeochemical Cycles*, 18, GB4031, 10.1029/2004GB002247, 2004.

572 Kleypas, J. A., Buddemeier, R. W., Archer, D., Gattuso, J.-P., Langdon, C., and Opdyke, B. N.: Geochemical
573 Consequences of Increased Atmospheric Carbon Dioxide on Coral Reefs, *Science*, 284, 118-120,
574 10.1126/science.284.5411.118, 1999a.

575 Kleypas, J. A., McManus, J. W., and Menez, L. A. B.: Environmental Limits to Coral Reef Development:
576 Where Do We Draw the Line?, *American Zoologist*, 39, 146-159, 10.1093/icb/39.1.146, 1999b.

577 Knox, F., and McElroy, M. B.: Changes in Atmospheric CO₂: Influence of the Marine Biota at High
578 Latitude, *J. Geophys. Res.*, 89, 4629-4637, 10.1029/JD089iD03p04629, 1984.

579 Kump, L. R., Bralower, T. R., and Ridgwell, A. J.: Ocean Acidification in Deep Time, *Oceanography*, 22, 94-
580 107, 2009.

581 Le Quéré, C., Takahashi, T., Buitenhuis, E. T., Rödenbeck, C., and Sutherland, S. C.: Impact of climate
582 change and variability on the global oceanic sink of CO₂, *Global Biogeochemical Cycles*, 24, GB4007,
583 10.1029/2009GB003599, 2010.

584 Le Quéré, C., Andres, R. J., Boden, T., Conway, T., Houghton, R. A., House, J. I., Marland, G., Peters, G. P.,
585 van der Werf, G. R., Ahlström, A., Andrew, R. M., Bopp, L., Canadell, J. G., Ciais, P., Doney, S. C., Enright,
586 C., Friedlingstein, P., Huntingford, C., Jain, A. K., Jourdain, C., Kato, E., Keeling, R. F., Klein Goldewijk, K.,
587 Levis, S., Levy, P., Lomas, M., Poulter, B., Raupach, M. R., Schwinger, J., Sitch, S., Stocker, B. D., Viovy, N.,
588 Zaehle, S., and Zeng, N.: The global carbon budget 1959–2011, *Earth Syst. Sci. Data*, 5, 165-185,
589 10.5194/essd-5-165-2013, 2013.

590 Lenton, T. M.: Land and ocean carbon cycle feedback effects on global warming in a simple Earth system
591 model, *Tellus B*, 52, 1159-1188, 10.1034/j.1600-0889.2000.01104.x, 2000.

592 Liu, C., and Allan, R. P.: Observed and simulated precipitation responses in wet and dry regions 1850–
593 2100, *Environmental Research Letters*, 8, 034002, 2013.

594 Lueker, T. J., Dickson, A. G., and Keeling, C. D.: Ocean pCO₂ calculated from dissolved inorganic carbon,
595 alkalinity, and equations for K₁ and K₂; validation based on laboratory measurements of CO₂ in gas and
596 seawater at equilibrium, *Marine Chemistry*, 70, 105-119, 2000.

597 Matear, R. J., and Hirst, A. C.: Climate change feedback on the future oceanic CO₂ uptake, *Tellus B*, 51,
598 722-733, 10.1034/j.1600-0889.1999.t01-1-00012.x, 1999.

599 Mehrbach, C., Culbertson, C. H., Hawley, J. E., and Pytkowicz, R. M.: Measurement of the apparent
600 dissociation constants of carbonic acid in seawater at atmospheric pressure, *Limnol. Oceanogr.*, 18, 897-
601 907, 1973.

602 Millero, F. J.: Thermodynamics of the carbon dioxide system in the oceans, *Geochimica et Cosmochimica*
603 *Acta*, 59, 661-677, 10.1016/0016-7037(94)00354-O, 1995.

604 Moss, R. H., Edmonds, J. A., Hibbard, K. A., Manning, M. R., Rose, S. K., van Vuuren, D. P., Carter, T. R.,
605 Emori, S., Kainuma, M., Kram, T., Meehl, G. A., Mitchell, J. F. B., Nakicenovic, N., Riahi, K., Smith, S. J.,
606 Stouffer, R. J., Thomson, A. M., Weyant, J. P., and Wilbanks, T. J.: The next generation of scenarios for
607 climate change research and assessment, *Nature*, 463, 747-756, doi:10.1038/nature08823, 2010.

608 Mucci, A.: The solubility of calcite and aragonite in seawater at various salinities, temperatures and at
609 one atmosphere pressure, *Amer. J. of Science*, 283, 781-799, 1983.

610 Orr, J. C., Fabry, V. J., Aumont, O., Bopp, L., Doney, S. C., Feely, R. A., Gnanadesikan, A., Gruber, N.,
611 Ishida, A., Joos, F., Key, R. M., Lindsay, K., Maier-Reimer, E., Matear, R., Monfray, P., Mouchet, A., Najjar,
612 R. G., Plattner, G.-K., Rodgers, K. B., Sabine, C. L., Sarmiento, J. L., Schlitzer, R., Slater, R. D., Totterdell, I.

613 J., Weirig, M.-F., Yamanaka, Y., and Yool, A.: Anthropogenic ocean acidification over the twenty-first
614 century and its impact on calcifying organisms, *Nature*, 437, 681-686, doi:10.1038/nature04095, 2005.
615 Pelejero, C., Calvo, E., McCulloch, M. T., Marshall, J. F., Gagan, M. K., Lough, J. M., and Opdyke, B. N.:
616 Preindustrial to Modern Interdecadal Variability in Coral Reef pH, *Science*, 309, 2204-2207,
617 10.1126/science.1113692, 2005.
618 Pietsch, S. A., and Hasenauer, H.: Evaluatin the self-initialization procedure for large-scale ecosystem
619 models., *Global Change Biology*, 12, 1-12, 2006.
620 Riahi, K., Rao, S., Krey, V., Cho, C., Chirkov, V., Fischer, G., Kindermann, G., Nakicenovic, N., and Rafaj, P.:
621 RCP 8.5—A scenario of comparatively high greenhouse gas emissions, *Climatic Change*, 109, 33-57,
622 10.1007/s10584-011-0149-y, 2011.
623 Ricciuto, D. M., Davis, K. J., and Keller, K.: A Bayesian calibration of a simple carbon cycle model: The role
624 of observations in estimating and reducing uncertainty, *Global Biogeochemical Cycles*, 22, GB2030,
625 10.1029/2006GB002908, 2008.
626 Ricke, K. L., Orr, J. C., Schneider, K., and Caldeira, K.: Risks to coral reefs from ocean carbonate chemistry
627 changes in recent earth system model projections, *Environmental Research Letters*, 8, 034003, dio:
628 10.1088/1748-9326/8/3/034003, 2013.
629 Riebesell, U., Zondervan, I., Rost, B., Tortell, P. D., Zeebe, R. E., and Morel, F. M. M.: Reduced
630 calcification of marine plankton in response to increased atmospheric CO₂, *Nature*, 407, 364-367, 2000.
631 Riebesell, U., Körtzinger, A., and Oschlies, A.: Sensitivities of marine carbon fluxes to ocean change,
632 *Proceedings of the National Academy of Sciences*, 106, 20602-20609, 10.1073/pnas.0813291106, 2009.
633 Riley, J. P., and Tongudai, M.: The major cation/chlorinity ratios in sea water, *Chemical Geology*, 2, 263-
634 269, [http://dx.doi.org/10.1016/0009-2541\(67\)90026-5](http://dx.doi.org/10.1016/0009-2541(67)90026-5), 1967.
635 Roy, T., Lombard, F., Bopp, L., and Gehlen, M.: Projected impacts of climate change and ocean
636 acidification on the global biogeography of planktonic Foraminifera, *Biogeosciences*, 12, 2873-2889,
637 10.5194/bg-12-2873-2015, 2015.
638 Sabine, C. L., Feely, R. A., Gruber, N., Key, R. M., Lee, K., Bullister, J. L., Wanninkhof, R., Wong, C. S.,
639 Wallace, D. W. R., Tilbrook, B., Millero, F. J., Peng, T.-H., Kozyr, A., Ono, T., and Rios, A. F.: The Oceanic
640 Sink for Anthropogenic CO₂, *Science*, 305, 367-371, 10.1126/science.1097403, 2004.
641 Sabine, C. L., Feely, R., Wanninkhof, R., Takahashi, T., Khatiwala, S., and Park, G.-H.: The global ocean
642 carbon cycle, In *State of the Climate in 2010, Global Oceans. Bull. Am. Meteorol. Soc.*, 92, S100-S108,
643 10.1175/1520-0477-92.6.S1, 2011.
644 Sarmiento, J. L., and Toggweiler, J. R.: A new model for the role of the oceans in determining
645 atmospheric PCO₂, *Nature*, 308, 621-624, 1984.
646 Sarmiento, J. L., and Le Quéré, C.: Oceanic Carbon Dioxide Uptake in a Model of Century-Scale Global
647 Warming, *Science*, 274, 1346-1350, 10.1126/science.274.5291.1346, 1996.
648 Sarmiento, J. L., and Gruber, N.: *Ocean Biogeochemical Dynamics*, edited by: Press, P. U., Princeton NJ,
649 2006.
650 Sasse, T. P., McNeil, B. I., Matear, R. J., and Lenton, A.: Quantifying the influence of CO₂ seasonality on
651 future aragonite undersaturation onset, *Biogeosciences*, 12, 6017-6031, 10.5194/bg-12-6017-2015,
652 2015.
653 Senior, C. A., and Mitchell, J. F. B.: The time-dependence of climate sensitivity, *Geophysical Research*
654 *Letters*, 27, 2685-2688, 10.1029/2000GL011373, 2000.
655 Siegenthaler, U., and Joos, F.: Use of a simple model for studying oceanic tracer distributions and the
656 global carbon cycle, *Tellus B*, 44, 186-207, 10.1034/j.1600-0889.1992.t01-2-00003.x, 1992.
657 Silverman, J., Lazar, B., Cao, L., Caldeira, K., and Erez, J.: Coral reefs may start dissolving when
658 atmospheric CO₂ doubles, *Geophysical Research Letters*, 36, n/a-n/a, 10.1029/2008GL036282, 2009.
659 Takahashi, T., Sutherland, S. C., Wanninkhof, R., Sweeney, C., Feely, R. A., Chipman, D. W., Hales, B.,
660 Friederich, G., Chavez, F., Sabine, C., Watson, A., Bakker, D. C. E., Schuster, U., Metzl, N., Yoshikawa-

661 Inoue, H., Ishii, M., Midorikawa, T., Nojiri, Y., Körtzinger, A., Steinhoff, T., Hoppema, M., Olafsson, J.,
662 Arnarson, T. S., Tilbrook, B., Johannessen, T., Olsen, A., Bellerby, R., Wong, C. S., Delille, B., Bates, N. R.,
663 and de Baar, H. J. W.: Climatological mean and decadal change in surface ocean pCO₂, and net sea–air
664 CO₂ flux over the global oceans, *Deep Sea Research Part II: Topical Studies in Oceanography*, 56, 554–
665 577, <http://dx.doi.org/10.1016/j.dsr2.2008.12.009>, 2009.

666 Taylor, K. E., Stouffer, R. J., and Meehl, G. A.: An Overview of CMIP5 and the Experiment Design, *Bulletin*
667 *of the American Meteorological Society*, 93, 485–498, 10.1175/BAMS-D-11-00094.1, 2012.

668 van Vuuren, D., Elzen, M. J., Lucas, P., Eickhout, B., Strengers, B., Ruijven, B., Wonink, S., and Houdt, R.:
669 Stabilizing greenhouse gas concentrations at low levels: an assessment of reduction strategies and costs,
670 *Climatic Change*, 81, 119–159, 10.1007/s10584-006-9172-9, 2007.

671 Wanninkhof, R.: Relationship between wind speed and gas exchange over the ocean, *Journal of*
672 *Geophysical Research: Oceans*, 97, 7373–7382, 10.1029/92JC00188, 1992.

673 Weiss, R. F.: Carbon dioxide in water and seawater: the solubility of a non-ideal gas, *Marine Chemistry*,
674 2, 203–215, 10.1016/0304-4203(74)90015-2, 1974.

675 Woosley, R. J., Millero, F. J., and Wanninkhof, R.: Rapid anthropogenic changes in CO₂ and pH in the
676 Atlantic Ocean: 2003–2014, *Global Biogeochemical Cycles*, 30, 70–90, 10.1002/2015GB005248, 2016.

677 Wullschleger, S. D., Post, W. M., and King, A. W.: On the Potential for a CO₂ Fertilization Effect in
678 Forests: Estimates of the Biotic Growth Factor, Based on 58 Controlled-Exposure Studies, *Biotic*
679 *Feedbacks in the Global Climate System: Will the Warming Feed the Warming?*, edited by: G.M., W., and
680 F.T., M., Oxford University, New York, 1995.

681 Yates, K., and Halley, R.: CO₂ concentration and pCO₂ thresholds for calcification and dissolution on
682 the Molokai reef flat, Hawaii, *Biogeosciences Discussions*, 3, 123–154, 2006.

683 Zeebe, R. E., and Wolf-Gladrow, D.: *CO₂ in Seawater: Equilibrium, Kinetics, Isotopes*, Elsevier,
684 Amsterdam, 2001.

685

686