We thank both referees for reviewing our revisions. Below is our response (in normal font) to the anonymous reviewer's comments (in *italic*).

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General comment:

I see that the authors have put in a lot of effort to address my concerns and that the manuscript has improved a lot. However, there are still some minor issues that should be addressed before publication.

Specific comments:

- *My concern about the time periods mentioned has not been satisfactorily addressed. While the authors have added two new sections where they describe time-periods, their mentioning of time-periods is rather imprecise. What exactly does it mean when the authors refer to preindustrial conditions in their model? Are they then considering the last 10-years of their spin-up runs or the first 10-years afterwards or does this include just one model year? Further, there are still several pictures and references to concentrations where no time-periods are mentioned. Do all concentrations and pictures throughout the manuscript refer to the same model years? If so, then please mention this. Still, it is beneficial to repeat the mentioning of the considered time-frame – this makes it easier to follow the manuscript.*

We consider the last year of the spin up for all analyses. Interannual variability is generally low in Earth System Models of Intermediate Complexity, as is the case in the Bern3D model. We add the following sentence to Section 3.1 The Bern3D model:

"The model is spun up to equilibrium under preindustrial conditions, with atmospheric CO_2 set to 278 ppm. The spin-up period is 4000 years without the sediment module and 50000 years with the sediment module. Remaining model drifts are negligible. **The last year of the spin-up period is considered for all analyses (note that unforced interannual variability is generally negligible in our model)."**

All concentrations, fluxes and pictures throughout the manuscript refer to this last year of the spin-up which reflects a preindustrial, climatological mean state. We find it sufficient to state this only once in this model section of the paper.

Also, the sentence "The gridded data-products from GLODAP and the World Ocean Atlas are derived from samples taken during the previous few decease" is rather vague. Both data-products mention a precise time-frame. So does, furthermore, Lee at al, 2001 and probably also other references. While this seems to be of little interest for the authors, there are probably other scientists who want to compare their results, where the specific time frame and the underlying conditions might be of interest.

The GLODAP data set consists of 9618 hydrographic stations collected on 95 cruises during the WOCE era between 1985 and 1999, and 2393 historical hydrographic stations

from 21 cruises occupied between 1972 and 1990. WOA09 represents objectively analyzed climatological fields with data entries from as early as 1900. The important point is that we are interested in a climatological mean state and not in interannual variability or trends. We therefore add the following and omit precise time frames:

"Total alkalinity data are from the GLODAP carbon climatology (Key et al., 2004) and salinity (Antonov et al., 2010), temperature (Locarnini et al., 2010), oxygen and phosphate (Garcia et al.2010a, b) data are from the World Ocean Atlas. **These gridded data-products represent objectively analyzed climatological fields of the respective oceanic variables and are based on samples taken during the previous decades.**"

Technical comments (referring to the revised manuscript):

- *p. 8, lines 250-255: While the equation for variable Keff appears now earlier in the text, the whole explanation for keff is now missing. Please insert it again!*

The reviewer appears to have overlooked the relevant text. The definition of the saturation product (eq. 6) is also required for the explanation of k_{eff} and the original explanation for k_{eff} follows below equation 6.

- *Table 2: Please refer to the depth of the export flux (75m, see Table 4).*

Done.

- *p. 4, line 111: please mention that this is the mean concentration from the GLODAP dataset.*

Done. "(TA, Eq. 1, mean concentration 2.427 mol m^{-3} **based on the regridded GLODAP dataset**)"

- *p. 4, line 114: please do not refer to the mean concentration here, but rather on page 4, line 132 and add which regression this mean concentration is associated with.*

Done. "From these different options, we accordingly chose the Global B3D linear regression **including 3 explanatory variables and global datasets** as robust regression for TA^0 **which yields a** mean concentration of $\sim$ 2.389 mol m^{-3}."

- *p. 5, line 133: please do not refer to the mean concentration here, but rather on page 5, line 139.*

Done. "We set $r_{Alk:OM}$ to 1.26 (Kanamori and Ikegami, 1982) and $r_{NO3:O2}$ to 16/170 (Anderson and Sarmiento, 1994) to uniformly, and globally link AOU changes to changes in TA (as in Feely et al., 2004; Koeve et al., 2014) **which yields a** mean concentration of -18 mmol m^{-3}."

- *p. 5, line 111: please mention that this mean concentration is achieved utilising the GLODAP dataset and your reference choices for TAr and TAo.*

Done. "The remaining signal, then, is TA$*$ (mean concentration $\sim$ 57 mmol m^{-3} **based on GLODAP and our reference choices to derive TA^0 and TA^r**)."

• *p. 19: The re-organisation of sentences and paragraphs makes this page now very difficult to read. I would suggest to re-arrange the sentences on this page again, so that everything about correlations, bias and RMSE is in one connected paragraph and not spread all over the page.*

Acknowledged. Paragraph is re-arranged.

• *p. 19, line 651: this is not "corresponding to the high correlation". Correlations are a measure of gradients/trends and RMSE are a measure of mean deviation, hence both measures do not correspond to each other but are independent. Please delete this part of the sentence. I am not sure why the RMSE is in percentage, as it is scale dependent and usually expressed as an absolute unit.*

Ok, first part deleted. Also, RMSE is now in absolute values between these illustrative runs.

"The average RMSE between the fast and slow dissolution scheme is 43.2 mmol m^{-3} and between the fast and constant is 24.5 mmol m^{-3} ."

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Many thanks for your comments.

A probabilistic assessment of calcium carbonate export and dissolution in the modern ocean

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Abstract. The marine cycle of calcium carbonate ($CaCO_3$) is an important element of the carbon cycle and co-governs the distribution of carbon and alkalinity within the ocean. However, $CaCO_3$ export fluxes and mechanisms governing $CaCO_3$ dissolution are highly uncertain. We present an observationally-constrained, probabilistic assessment of the global and regional $CaCO_3$ budgets. Parameters governing pelagic $CaCO_3$ export fluxes and dissolution rates are sampled using a Monte Carlo scheme to construct a 1000 member ensemble with the Bern3D ocean model. Ensemble results are constrained by comparing simulated and observation-based fields of excess dissolved calcium carbonate (TA*). The minerals calcite and aragonite are modelled explicitly and ocean–sediment fluxes are considered. For local dissolution rates either a strong or a weak dependency on $CaCO_3$ saturation is assumed. In addition, there is the option to have saturation-independent dissolution above the saturation horizon. The median (and 68 % confidence interval) of the constrained model ensemble for global biogenic $CaCO_3$ export is 0.90 (0.72–1.05) Gt-C yr^{-1}, that is within the lower half of previously published estimates (0.4–1.8 Gt-C yr^{-1}). The spatial pattern of $CaCO_3$ export is broadly consistent with earlier assessments. Export is large in the Southern Ocean, the tropical Indo–Pacific, the northern Pacific and relatively small in the Atlantic. The constrained results are robust across a range of diapycnal mixing coefficients and, thus, ocean circulation strengths. Modelled ocean circulation and transport time scales for the different setups were further evaluated with CFC11 and radiocarbon observations. Parameters and mechanisms governing dissolution are hardly constrained by either the TA* data or the current compilation of $CaCO_3$ flux measurements such that model realisations with and without saturation-dependent dissolution achieve skill. We suggest to apply saturation-independent dissolution rates in Earth System Models to minimise computational costs.

1 Introduction

The cycling of calcium carbonate ($CaCO_3$) forms an important component of the marine carbon cycle. It co-governs the surface-to-deep gradients of alkalinity and dissolved inorganic carbon (DIC) in the ocean (Volk and Hoffert, 1985), dominates deep ocean alkalinity fluxes, and influences the

surface fields of DIC and alkalinity, thereby providing the background conditions for the uptake of excess anthropogenic carbon from the atmosphere. However, the export and dissolution fluxes of $CaCO_3$ are highly uncertain. For example, current estimates of $CaCO_3$ export diverge by a factor of ~ 4 (0.4–1.8 Gt-C yr^{-1}, summarised in Berelson et al., 2007).

The $CaCO_3$ cycle is driven by calcifying organisms such as coccolithophorids, foraminifera or pteropods, which remove calcium (Ca^{2+}), alkalinity, and DIC from the pelagic surface ocean waters to form shells and structures of $CaCO_3$. These $CaCO_3$ particles are eventually exported out of the surface, gravitationally sink through the water column and dissolve at depth or get buried in ocean sediments. The formation and dissolution of $CaCO_3$ introduces a vertical gradient in alkalinity and DIC. These gradients have been sustained against the counteracting forces of physical mixing and transport, which would otherwise have removed these gradients. This redistribution of alkalinity and carbon by biogenic and physical transport affects the partitioning of carbon between the ocean and atmosphere and a reduction, expected under ongoing ocean acidification, or even a complete stop of $CaCO_3$ export tend to decrease atmospheric CO_2 by a few ppm, and in the extreme case by up to ~ 50 ppm, on century time scales (Heinze, 2004; Gangstø et al., 2011).

There are different mineral forms of $CaCO_3$ and solubility is higher for high-magnesium calcite (typically present in fish) and aragonite (typically present in free-swimming pelagic sea snails and sea slugs (pteropods)) than for calcite (typically present in calcifying algae (coccolithophorids)). Thermodynamic considerations suggest that dissolution of $CaCO_3$ particles occurs only when the product of the calcium and carbonate ion concentrations in the surrounding environment is below the saturation product. The saturation product of all minerals increases with increasing pressure (Mucci, 1983). At depth, respiration of organic matter additionally decreases the concentration of CO_3^{2-}. As a result, the bulk of the water in the deep ocean is undersaturated with respect to $CaCO_3$ minerals, generally enabling their dissolution, and oversaturated in the upper ocean, thermodynamically hindering their dissolution (Steinacher et al., 2009). Overall, the quantitative understanding of dissolution kinetics is, however, low and published estimates of saturation-dependent dissolution kinetic parameters range over several orders of magnitude (summarised in Sarmiento and Gruber, 2006). $CaCO_3$ dissolution may, nevertheless, still occur in the upper ocean in suitable, undersaturated microenvironments which would be present for instance in the guts of zooplankton, suspended organic aggregates, or fecal pellets (Bishop et al., 1980; Milliman et al., 1999; Jansen and Wolf-Gladrow, 2001). There are in fact several tracer-based studies reporting $CaCO_3$ dissolution above the saturation horizon of bulk seawater (Barrett et al., 2014; Feely et al., 2002, 2004; Sabine et al., 2002b; Chung et al., 2003). In a modelling study, Friis et al. (2006), nevertheless, demonstrated that the method which is often employed to derive these upper ocean dissolution rates (Berelson et al., 2007, see Discussion section on TA*-CFC age method), might not be applicable, because this method neglects physical transport and mixing of alkalinity. It is therefore still debated where and how fast settling $CaCO_3$ particles are dissolved in the water column. Another complication,

2

typically neglected in previous studies on open water $CaCO_3$ dissolution, is arising from ocean–

65   sediment interactions and the influence of associated burial and redissolution fluxes on alkalinity and carbon concentrations in the open ocean (Archer, 1996).

    Here we propose an alternative, probabilistic assessment of the global $CaCO_3$ budget. We consider explicitly the transport and mixing of alkalinity and account for ocean–sediment interactions to probabilistically constrain the $CaCO_3$ cycle with the observation-based distribution of the TA^*

70   tracer. TA^* reflects the imprint of the $CaCO_3$ cycle on alkalinity (see Sect. 2) and is therefore impacted by $CaCO_3$ export, water column dissolution, physical transport and mixing of the released alkalinity and DIC and ocean–sediment fluxes. For the probabilistic flux assessment, a large range of calcite and aragonite export and dissolution flux parameterisations are employed within our Earth System model of intermediate complexity (EMIC) – the Bern3D model – in a Monte Carlo setup

75   with 1000 members. The Bern3D model calculates the corresponding modelled steady-state tracer concentrations of TA^*. The most probable export and dissolution fluxes are then the ones resulting in modelled TA^* fields that match closely the observation-derived TA^* distribution. The aim is to assign uncertainty estimates that will be consistent with both the observations and model equations – a classical data assimilation problem (see Sect. 3.3). It is expensive to perform simulations with

80   interactive sediments (Heinze et al., 1999; Gehlen et al., 2006; Tschumi et al., 2011) due to long spin-up times required to bring ocean and sediments in equilibrium and we account a posteriori for the influence of sediment burial and dissolution fluxes on TA^* (see Sect. 3.3). This approach represents an alternative to the interpretation of concentrations without a physical model or the interpretation of flux/rate measurements within the water column, which are generally much sparser (~ 156 flux mea-

85   surements in the water column (Wilson et al., 2012) and ~ 56 benthic dissolution flux measurements (Berelson et al., 2007), globally) and more difficult to obtain. We apply the Wilson et al. (2012) data compilation of water column fluxes as an additional constraint for comparison. Additional sensitivity analyses with respect to vertical diffusion (k_{dia}, see Sect. 4) are illustrated. The results are compared and contrasted to data-based estimates of export and dissolution (as summarised in Berelson et al.,

90   2007). Finally, implications for the parameterisation of $CaCO_3$ dissolution in Earth System Models are discussed.

2   Observation-derived TA^*

Total alkalinity data are from the GLODAP carbon climatology (Key et al., 2004) and salinity (Antonov et al., 2010), temperature (Locarnini et al., 2010), oxygen and phosphate (Garcia et al.,

95   2010a, b) data are from the World Ocean Atlas. These ~~observational data~~ gridded data-products represent objectively analyzed climatological fields of the respective oceanic variables and are based on samples taken during the previous decades. They serve to split the alkalinity signal into its different physical and biogeochemical components such as TA^*, our target variable in the data as-

simulation. ~~The gridded data products from GLODAP and the World Ocean Atlas are derived from~~
~~samples taken during the previous few decades.~~ We first regridded all required, gridded datasets to
the Bern3D model grid (40x41x32 grid boxes) using the area-weighted regridding method of Ferret
before deriving the other properties.

TA^*, our target variable in the data assimilation, is a constructed tracer (Feely et al., 2002; Sabine
et al., 2002a; Chung et al., 2003; Koeve et al., 2014) to exclusively capture the imprint of $CaCO_3$
dissolution on alkalinity. TA^* is, in this sense, one of three components of measured total alkalinity
(TA, Eq. 1, mean concentration $2.427\,mol\,m^{-3}$ based on the regridded GLODAP dataset). TA^* can
be extracted from TA by accounting for preformed (TA^0, Eq. 2) and remineralised alkalinity (TA^r,
Eq. 3):

$$TA = TA^0 + TA^r + TA^*. \tag{1}$$

TA^0 ~~(mean concentration ~ 2.389)~~ is the background or preformed concentration, set at the ocean
surface, and mixed conservatively in the ocean interior. Accordingly, TA^r and TA^* are by definition
zero in the surface ocean. To describe TA^0 in the ocean interior, one relies on a multi-linear re-
gression relationship for surface ocean total alkalinity based on surface ocean salinity (S) and PO
(PO $= O_2 + r_{-O_2:PO_4} \cdot PO_4$, $r_{-O_2:PO_4} = 170$, Broecker, 1974) and sometimes surface ocean tem-
perature (T) as explanatory variables (all conservative variables, see Eq. (2), Gruber et al., 1996;
Sabine et al., 2002a; Feely et al., 2002).

$$TA^0 = a_0 + a_1 \cdot S + a_2 \cdot T + a_3 \cdot PO \tag{2}$$

The coefficients, a_i, are estimated from observations of TA (Key et al., 2004), S (Antonov et al.,
2010), T (Locarnini et al., 2010), and PO (Garcia et al., 2010a, b) in the surface ocean. The linear
regression fit is sometimes further subdivided to include only specific basins (Feely et al., 2002;
Koeve et al., 2014). Table 1 summarises different regressions, including previously published re-
gression estimates as well as new estimates by us calculated on the Bern3D model grid (B3D). The
root mean squared deviations (RMSE) between these fits are smaller than $0.00465\,mol\,m^{-3}$ (i.e.
smaller than $0.2\,\%$, excluding the Gruber et al. (1996) equation, which relies on an older database
from 1996). From these different options, we accordingly chose the Global B3D linear regression
including 3 explanatory variables and global datasets as robust regression for TA^0 which yields a
mean concentration of $\sim 2.389\,mol\,m^{-3}$.

TA^r ~~(mean concentration -18)~~ is linked stoichiometrically to the apparent oxygen utilisation
(AOU, Garcia et al., 2010b, Eq. 3) and accounts for decreases in TA due to the oxidation of organic
nitrogen, phosphorous, and sulfur (OM).

$$TA^r = r_{Alk:OM} \cdot r_{NO_3:-O_2} \cdot AOU = 1.26 \cdot 16/170 \cdot AOU \tag{3}$$

We set $r_{Alk:OM}$ to 1.26 (Kanamori and Ikegami, 1982) and $r_{NO_3:O_2}$ to 16/170 (Anderson and
Sarmiento, 1994) to uniformly, and globally link AOU changes to changes in TA (as in Feely et al.,

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2004; Koeve et al., 2014) which yields a mean concentration of $-18\,\mathrm{mmol\,m^{-3}}$. Wolf-Gladrow et al. (2007) propose an 8 % higher value for $r_{\mathrm{Alk:OM}}$ of 1.36, based on a different sulphur to carbon ratio. In addition, we note that AOU has been suggested to overestimate true oxygen utilisation by 20–25 % (Ito et al., 2004; Duteil et al., 2013). Accordingly, $\mathrm{TA^r}$ might be associated with an uncertainty of $\sim 20\,\%$.

The remaining signal, then, is $\mathrm{TA^*}$ (mean concentration $\sim 57\,\mathrm{mmol\,m^{-3}}$ based on GLODAP and our reference choices to derive $\mathrm{TA^0}$ and $\mathrm{TA^r}$), the changes in alkalinity due only to the $\mathrm{CaCO_3}$ cycle (Fig. 1). The global average RMSE of any of the described ways (16 in total) of accounting for $\mathrm{TA^r}$ (25 % lower AOU or different $r_{\mathrm{Alk:OM}}$) and $\mathrm{TA^0}$ (either 2 or 3 explanatory variables, and either global or regional) from our reference choices is $\sim 4\,\mathrm{mmol\,m^{-3}}$ (average RMSE 3.9, 3.8, 3.1 $\mathrm{mmol\,m^{-3}}$ in the Atlantic, Pacific and Indian Ocean, respectively, i.e. $\sim 7\,\%$). It is to note that this approach, inherent to its empirical nature, yields slightly negative $\mathrm{TA^*}$ values in some places. $\mathrm{TA^*}$ integrates to about 37.5 Pmol C or 75 Pmol Alk-equivalents of which $\sim 41\,\%$ come to lie above the calcite saturation horizon (similar to Koeve et al. (2014) who find 44.7 % of the $\mathrm{TA^*}$ inventory above the calcite saturation horizon). These estimates are robust across the different sources of uncertainty. $\mathrm{TA^*}$ concentrations are expressed in alkalinity equivalents throughout this paper and we do not divide $\mathrm{TA^*}$ concentrations by a factor of 2 as done in many observational studies which express $\mathrm{TA^*}$ in terms of carbon changes. $\mathrm{TA^*}$ inventories are in Pmol C and $\mathrm{CaCO_3}$ fluxes are given in carbon units ($\mathrm{Gt\text{-}C\,yr^{-1}}$, $\mathrm{mmol\text{-}C\,m^{-3}\,yr^{-1}}$, $\mathrm{mmol\text{-}C\,m^{-2}\,yr^{-1}}$).

3   The modelling framework

To constrain the alkalinity fluxes associated with $\mathrm{CaCO_3}$ cycling, we introduce different export and dissolution fluxes (with a total of 15 degrees of freedom) within the biogeochemistry component of the Bern3D dynamic ocean model. The parameters of the new formulations describing the export and dissolution fluxes of $\mathrm{CaCO_3}$ are varied using a Monte Carlo sampling method (McKay et al., 1979; Steinacher et al., 2013) and the resulting model ensemble, including 1000 members, is run to a pre-industrial steady state producing a broad range of solutions (see Sect. 3.2). The alkalinity fluxes associated with the $\mathrm{CaCO_3}$ cycle are then constrained by comparing observation-based and simulated $\mathrm{TA^*}$ data using a Bayesian approach following Steinacher et al. (2013); a skill score is assigned to each ensemble member and used as a weight for the computation of median values and probability density functions from the ensemble results. The sediment module is not included in the ensemble due to extensive computational cost (~ 150 vs. $\sim 8\,\mathrm{CPU\,h}$ for a single member run with and without the sediment, respectively). Instead, we re-run the model configurations which achieved the best skill scores with interactive sediments to account for $\mathrm{CaCO_3}$ burial and sediment redissolution a posteriori (see Sect. 3.3.3).

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3.1 The Bern3D Model

The Bern3D model couples a dynamic ocean, sea ice, an energy-moisture balance atmosphere,
a marine biogeochemical cycle, a dynamic global vegetation model, and an ocean sediment module (Müller et al., 2006; Tschumi et al., 2011; Ritz et al., 2011; Roth et al., 2014). Here an ocean
version with a horizontal resolution of 41 by 40 grid cells and 32 logarithmically-scaled vertical
layers is used (see also Roth et al., 2014). The horizontal resolution is the same for the components
atmosphere, ocean, sea ice, and sediments of the Bern3D model. Transport and mixing of tracers
in the ocean is based on Edwards et al. (1998) and Müller et al. (2006) as a three-dimensional frictional geostrophic model. The model has an isopycnal diffusion scheme and Gent–McWilliams parameterisation for eddy-induced transport (Griffies, 1998). The NCEP/NCAR monthly wind-stress
climatology (Kalnay et al., 1996) is prescribed at the surface. Air–sea gas exchange for CO_2 is implemented according to OCMIP-2 protocols (Najjar et al., 1999; Orr and Najjar, 1999). The global
mean air-sea transfer rate is reduced by 19% compared to OCMIP-2 to match observation-based estimates of natural and bomb-produced radiocarbon (Müller et al., 2008). A two-dimensional energy
moisture balance model represents the atmosphere (Ritz et al., 2011). The model is spun up to equilibrium under preindustrial conditions, with atmospheric CO_2 set to 278 ppm. The spin-up period
is 4000 years without the sediment module and 50 000 years with the sediment module. Remaining
model drifts are negligible. The last year of the spin-up period is considered for all analyses (note that
unforced interannual variability is generally negligible in our model). We implicitly neglect potential
changes in TA* over the industrial period by comparing model results for preindustrial conditions
with TA* data reconstructed from recent measurements. Such changes are negligible in simulations
with prescribed anthropogenic forcing in the Bern3D model.

The marine biogeochemical module computes the cycling of carbon, alkalinity, phosphate, iron,
oxygen, silica, and carbon isotopes. New production of organic material in the model is limited by
temperature, light, phosphate, and iron following Doney et al. (2006) as described by Parekh et al.
(2008) and Tschumi et al. (2011). One third (33 %) of the new production is exported out of the
euphotic zone (defined at 75 m) as particulate organic matter (POM), with the reminder contributing
to the dissolved organic matter pool. Within biogeochemically similar regions/provinces a parameter
termed rain ratio linearly scales the pattern of POM export to $CaCO_3$ export (silica-limitation is
not considered here). We define eight such regions, each assigned an independent value for the
export rain ratio parameter (mol inorganic carbon / mol organic carbon exported, later given in %
inorganic to organic carbon exported). These include the Pacific, Atlantic and Indian sections of
the Southern Ocean ($< 35°$ S, separated at $240°$ W, $63°$ W and $30°$ E), the tropical ($35°$ S to $30°$ N)
Pacific, Atlantic, and Indian Ocean, and the northern ($> 30°$ N) Atlantic (including the Arctic) and
Pacific Ocean. A global parameter (f_{calc}) determines how much of the total $CaCO_3$ export flux
represents calcite and how much represents aragonite ($1 - f_{calc}$). Abiotic $CaCO_3$ precipitation is
virtually absent in today's ocean and not considered (reviewed in Sarmiento and Gruber, 2006).

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205   In the model, we implemented TA* as an explicit, idealised tracer (Koeve et al., 2014). It captures the alkalinity equivalents of $CaCO_3$ dissolution whenever it occurs and mixes this signal, accordingly. TA* values are set to zero throughout the surface ocean.

In sensitivity simulations with the sediment module enabled, the flux of $CaCO_3$, and of other particles, reaching the seafloor is passed to the sediment module from where a fraction potentially

210   re-dissolves back into the water column. In simulations without the sediment module, the entire flux reaching the ocean floor re-dissolves back into the water column. Simulated TA* concentrations tend to be lower with the sediment module enabled than without sediment module, because a fraction of the $CaCO_3$ export flux is removed from the ocean and buried in the geosphere. The sediment diagenesis model (Heinze et al., 1999; Gehlen et al., 2006; Tschumi et al., 2011) features the same

215   horizontal resolution as the ocean model and 10 layers resolving the top 10 cm of the seafloor. It dynamically calculates the transport, remineralisation/redissolution and bioturbation of solid material within the top 10 cm of the seafloor as well as pore-water chemistry and diffusion as described in detail in Tschumi et al. (2011). Solutes diffuse over a boundary layer of 1 cm between the sediment column and the lowermost ocean grid cell. Four solid components ($CaCO_3$, opal, POM and clay)

220   and pore water substances (carbon and carbon isotopes, total alkalinity, phosphate, nitrate, oxygen and silicic acid) are modelled. The pore water carbonate ion concentration determines whether, and at which rate, $CaCO_3$ dissolves. Aragonite and calcite are not distinguished within the sediment module and $CaCO_3$ is assumed to be in the form of calcite. Any solid material that is pushed out of the diagenetic zone disappears into the subjacent diagenetically consolidated zone. During the

225   spin-up of the ocean–sediment model, the net loss of alkalinity, and other tracers such as carbon and nutrients, to the sediments is immediately replaced by corresponding riverine inputs which are distributed uniformly along the coastlines (and then taken to be part of TA^0). The riverine input is diagnosed at the end of the spin-up.

The model features the main water masses and mixing time scales of the ocean, an essential

230   prerequisite to realistically simulate TA* and other tracers. The simulated and observed distributions of the ventilation tracers Δ^{14}C and CFC11 are provided in the supplementary information along with a Taylor Diagram (Taylor, 2001) of CFC11, Δ^{14}C, temperature, salinity, DIC, TA, PO$_4$, oxygen, and TA* (Fig. A1 to A3). Globally, the correlation coefficient and standard deviation of the median relative to the standard deviation of the observations ($\sigma_{\text{rel.}}^{\text{obs.}}$) are 0.94 and 0.99 for Δ^{14}C and 0.91 and

235   0.98 for CFC11.

3.2 $CaCO_3$ dissolution within the water column

The dissolution equations for calcite and aragonite are implemented using equations with identical functional forms, but with different parameters for each mineral. In the following, we do not explicitly distinguish the two minerals to ease notation. The dissolution of calcite and aragonite below the

240   euphotic zone is assumed to be a function of the saturation state of the bulk seawater, Ω, and the par-

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ticle concentration per unit of water volume, $[CaCO_3]$ (see Gangstø et al. (2011) for a discussion):

$$\frac{d[CaCO_3]}{dt} = -k_{eff}(\Omega) \times [CaCO_3]. \tag{4}$$

k_{eff} denotes a first order rate constant for either calcite or aragonite. It is defined as:

$$k_{eff}(\Omega) = k_0 \times (1 - \Omega)^n \times H(\Omega - 1) + k_{bg}. \tag{5}$$

245   k_0 and k_{bg} are rates in units of 1/time. H is the Heaviside function, which is zero for supersaturated and 1 for undersaturated water. The saturation state is defined by the ratio of the product of calcium ion concentration times carbonate ion concentrations to the saturation product, K_{sp} (Mucci, 1983):

$$\Omega = \frac{[Ca^{2+}][CO_3^{2-}]}{K_{sp}}. \tag{6}$$

250   Ω values larger than one correspond to oversaturated and Ω values smaller than one to undersaturated conditions. Ω and thus k_{eff} are grid cell specific. At super-saturation, the dissolution rate k_{eff} equals the constant background rate k_{bg} (which can be zero). With increasing under-saturation, the dissolution rate increases towards its maximum value ($k_{bg} + k_0$). n is a unitless parameter and determines the deviation from linearity of this increase. For simplicity and to avoid the addition of further free

255 parameters, a constant sinking velocity, v, is assumed and identical for both calcite and aragonite particles. The flux profile of $CaCO_3$ then takes the form

$$F_{i,j}(z_k) = F_{i,j}(z_{k-1}) \cdot \exp\left(\frac{-k_{eff}(\Omega_{i,j,k})}{v} \cdot \Delta z_k\right) \tag{7}$$

where F is the downward flux of either calcite or aragonite particles per unit area evaluated at the bottom of each tracer grid cell at depth z_k. i, j, and k are grid cell indices indicating longitude, latitude,

260 and depth. Δz_k denotes grid cell height. The export flux is set equal to F ($z = 75\,\text{m}$) at the depth of the euphotic zone. Particles are dissolved instantaneously and sinking is not explicitly resolved in this formulation, reducing computational costs. v/k_{eff} is in units of m, and, if assumed constant, can be interpreted as the dissolution length scale, i.e. the depth at which the flux has decreased to $1/e$ (to $\sim 37\,\%$) of the export flux at 75 m. Any particles reaching the sea floor are dissolved completely in

265 the appropriate lowermost box, except when the sediment module is included.

    In Fig. 2 we illustrate three dissolution cases to explore the sensitivity of TA* to dissolution profiles which cover the sampled uncertainty in dissolution rate parameters. Table 2 (middle column, $k_{dia,\,ref}$) summarises the resulting TA* inventories and their relative distribution. The three dissolution rate profiles are selected to represent a case with constant, saturation-independent, dissolution

270 (constant) and two cases where this background dissolution rate is set to zero and $CaCO_3$ dissolves only below the saturation horizon. In the fast (slow) case, aragonite (chosen to represent 10 % of the total) and calcite (90 % of total export) dissolves quickly (slowly) below the saturation horizon. Aragonite and calcite dissolve within a few hundred meters below the saturation horizon in the fast case, while most $CaCO_3$ dissolves on the ocean floor in the slow case.

275 The choice of the dissolution rate profile has a substantial influence on the simulated TA^* inventory (Fig. 2, Table 2). The global TA^* inventory is $38\,\mathrm{Pmol\,C}$ for the case with a constant, saturation-independent dissolution and a rain ratio of $\sim 7\,\%$, which is close to the observation-derived inventory of $37\,\mathrm{Pmol\,C}$. The simulated inventory is 48 and $63\,\mathrm{Pmol\,C}$ for the fast and slow cases (where no dissolution occurs above the saturation horizon), respectively, and by that substantially higher. TA^*

280 accumulates in the deep ocean when all $CaCO_3$ is dissolved below the saturation horizon of aragonite and calcite and no dissolution is permitted above. The TA^* inventory and concentrations are sensitive to the choice of the dissolution rate profile, supporting our choice of TA^* as a target variable to constrain dissolution rates.

3.3 Ensemble simulations and metrics for skill assessment

285 #### 3.3.1 The Monte Carlo Ensemble

Following Steinacher et al. (2013) and Steinacher and Joos (2016) we run a 1000 member Latin Hypercube ensemble to constrain the export flux out of the surface ocean, and the dissolution of aragonite and calcite within the water column. Latin Hypercube sampling (McKay et al., 1979) is a statistical, Monte Carlo method to generate controlled random samples from a multidimensional

290 distribution (15 dimensions in our case). The defined parameter ranges are divided into equally probable intervals (1000 in our case). Random samples are then generated in each interval. This method ensures that the sampled values are representative of the real variability while minimising the number of required samples and thus the computational costs. We sample 15 parameters and apply uniform priors based on literature information. The free parameters are the eight rain-ratio parameters (prior

295 range: 0 to $18\,\%$ for the Pacific and South Indian regions, prior range: 0 to $16\,\%$ for the tropical Indian, prior range: 0 to $7\,\%$ for the North and tropical Atlantic, and prior range: 0 to $10\,\%$ for the South Atlantic), defining the total amount of $CaCO_3$ export for each of the six regions. The prior ranges of the rain ratios for the three Atlantic regions are limited to maximal 10%. This selection is based on results from previous ensemble setups that revealed an overestimation of TA^* for high rain

300 ratios in the Atlantic domain. Further we include the fraction f_{calc} (1–0.5) defining the split between aragonite and calcite export, and 3×2 parameters governing the dissolution kinetics for calcite and aragonite, respectively. These are k_0 (0.05–$10\,\mathrm{day}^{-1}$) and n (1–4), describing fast and slow dissolution kinetics as a function of undersaturation, and k_{bg}/v (0–$1/2500\,\mathrm{m}^{-1}$), the length scale associated with a constant, background dissolution rate acting both above and below the saturation horizon. v

305 is kept constant at $100\,\mathrm{m\,day}^{-1}$.

The observation-based saturation-state of the bulk seawater with respect to aragonite and calcite is prescribed for each model grid cell. It was calculated with the carbonate chemistry package seacarb (Gattuso et al., 2010) from GLODAP (Key et al., 2004) and World Ocean Atlas 2009 (WOA09) (Locarnini et al., 2010; Antonov et al., 2010). Seacarb calculates carbonate chemistry based on pressure,

310 temperature, salinity, alkalinity, DIC, silica and phosphate. This reduces computational costs as the carbonate chemistry package of the online model requires substantial computational time. Also, this avoids mismatches in the modelled and observation-based saturation states, which are also due to model deficiencies in the cycling of organic matter and physical transport. Mismatches in modelled and observed saturation states are particularly large in the North Pacific, where the modelled cal-

315 cite saturation horizon is up to $1.5\,\mathrm{km}$ too deep. The calcite saturation horizon is well represented in the South Pacific, Indian and Atlantic by the model. The results presented in Sect. 4 suggest that estimated $CaCO_3$ export production fields and dissolution rates are insensitive to the choice of the saturation field, because saturation-dependent and saturation independent parameterizations of dissolution yield similar TA^* fields.

320 ### 3.3.2 Skill scores

Global skill scores, S_m, are assigned to each member, m, of the Latin-Hypercube ensemble:

$$S_m = \exp\left(-0.5 \cdot \mathrm{MSE}^{\mathrm{rel}}\right). \tag{8}$$

$\mathrm{MSE}^{\mathrm{rel}}$ is the relative mean squared error of the simulated TA^* concentrations from member m with respect to observation-derived TA^*:

325 $$\mathrm{MSE}^{\mathrm{rel}} = \sum_j a_j \times \frac{\left(\mathrm{TA}_j^{*\mathrm{model}} - \mathrm{TA}_j^{*\mathrm{obs}} - \mathrm{TA}_j^{*\mathrm{sedcorr}}\right)^2}{\sigma^2}. \tag{9}$$

The sum includes all grid cells (indexed j). $\mathrm{TA}_j^{*\mathrm{model}}$ denotes simulated TA^* concentrations for ensemble member m and $\mathrm{TA}_j^{*\mathrm{obs}}$ observation-based TA^* concentrations estimated by using the Global B3D regression (Sect. 2). $\mathrm{TA}_j^{*\mathrm{sedcorr}}$ is a correction term arising from $CaCO_3$ burial in sediments, further explained below. a_j is the grid cell volume used as weight in the sum. σ^2 represents the com-

330 bined error of the observation-based TA^* estimates and of the model and sets the scale against which model deviations are evaluated. Model deviations from the observations are considered large or small relative to the magnitude of σ^2 (Schmittner et al., 2009). The total uncertainty in observation-derived, gridded TA^* data is difficult to estimate and includes uncertainties due to extrapolation of limited number of measurements, uncertainties in individual tracer measurements, and in the computation

335 of TA^* from tracer data. The error associated with the procedure to compute TA^* (average RMSE 3.9, 3.8, $3.1\,\mathrm{mmol\,m^{-3}}$ in the Atlantic, Pacific and Indian Ocean, respectively, see Sect. 2) is small compared to the model error (the best run achieves a RMSE of 11, 18, $18\,\mathrm{mmol\,m^{-3}}$ in the Atlantic, Pacific and Indian Ocean, respectively). Following Steinacher et al. (2013) and Schmittner et al. (2009), we estimate σ^2 as the (volume-weighted) variance of the model–data discrepancy for

340 the ensemble member with the lowest MSE (this variance is $275\,(\mathrm{mmol\,m^{-3}})^2$); this corresponds to $\mathrm{MSE}^{\mathrm{rel}}$ close to unity for the best-fitting ensemble member.

The skill scores S_m of the individual ensemble members are likelihood-type functions corresponding to a Gaussian distribution of the data-model discrepancy $(\mathrm{TA}^{*\mathrm{model}} - \mathrm{TA}^{*\mathrm{obs}} - \mathrm{TA}^{*\mathrm{sedcorr}})$

with zero mean and variance σ^2. S_m is an indication of the relative performance/credibility of each individual model configuration. Configurations which have relatively small deviations from the data are judged more probable than configurations which differ greatly from the observations.

S_m are used as weight to compute probability density functions (PDFs) and related measures such as the median (50th percentile) and the 16th and 84th percentiles defining the one standard deviation confidence interval (1σ) of the ensemble results. PDFs represent weighted and normalised histograms of the variables of interest. The normalisation is such that the integral over a PDF equals 1. A cubic spline interpolation is used to arrive at a continuous PDF from the discrete, normalised histogram. For the computation of median and confidence ranges the histograms are converted to cumulative distribution functions (CDFs). We interpolate linearly within the discrete CDFs to arrive at the chosen percentiles (i.e. CDF=(0.16, 0.5, 0.68)).

The above explanations apply for any simulated quantity of interest. In the following we will present PDFs, median values, and 1σ confidence ranges for aragonite and calcite export and dissolution as well as for tracer concentrations at individual grid cells or integrated over regions or the whole ocean. Spatial integrations are done for each ensemble member individually and before computing the PDFs and associated measures from the full ensemble.

3.3.3 A first order correction for ocean–sediment TA* fluxes

$CaCO_3$ burial removes alkalinity from the ocean water column and lowers concentrations and the overall TA* inventory relative to a run without the sediment module. Riverine input compensates this loss. This input of alkalinity is added to the surface ocean and by that to the part of the preformed alkalinity component (TA^0), leaving TA* unchanged. $CaCO_3$ export and dissolution within the water column and the corresponding fluxes of alkalinity and TA* remain largely unchanged between runs with and without sediments.

Ideally, the ensemble would be run fully interactively with the sediment module enabled to account for all important processes within the $CaCO_3$ cycle. However, this is computationally too expensive as the sediment module requires a long spin up to achieve equilibrium. A first order correction term TA$^{*\text{sedcorr}}$ that accounts for the influence of $CaCO_3$ burial and dissolution fluxes on TA* is estimated as follows. First, the skill scores are computed as described above with TA$^{*\text{sedcorr}}$ set to zero. Then, the 14 best ensemble members are selected and re-run with the sediment module enabled. The mean difference in the TA* fields between the simulations with and without sediments yields the sediment correction, TA$^{*\text{sedcorr}}$.

These simulations with sediments yield a mean global burial flux of $0.12\,\text{Gt-C}\,\text{yr}^{-1}$ (see Table 3: $0.011\,\text{Gt-C}\,\text{yr}^{-1}$ in the Atlantic, $0.062\,\text{Gt-C}\,\text{yr}^{-1}$ in the Pacific, and $0.042\,\text{Gt-C}\,\text{yr}^{-1}$ in the Indian Ocean). This is within the estimate by Feely et al. (2004) of 0.1–$0.14\,\text{Gt-C}\,\text{yr}^{-1}$.

The sediment burial correction on TA* is largest in the Pacific and smallest in the Atlantic (Figs. 3 and 4). The global TA* inventory in runs with the sediment is $6.9\,\text{Pmol}\,\text{C}$ lower compared to the

380   runs without the sediment (see Table 3: 0.9 Pmol C in the Atlantic, 4.6 Pmol C in the Pacific, and 1.4 Pmol C in the Indian Ocean). This reduction is equal to the inventory of the $TA^{*sedcorr}$ correction and ~ 16 and $\sim 20\,\%$ of the observation-derived TA^* inventory of the Pacific and Indian Ocean, respectively. TA^* concentrations are affected relatively uniformly below 1 km and differences in TA^* between simulations with and without sediment module tend to vanish toward the surface ocean

385   (Figs. 3 and 4). Correspondingly, the spatial patterns of TA^* are very similar for simulations with and without sediments; correlation coefficients are > 0.99. We expect that the correction will tend to increase the export flux of $CaCO_3$ in the optimisation to compensate for the loss by burial, but will not strongly affect dissolution parameters as the spatial patterns of TA^* remain similar with or without correction.

390   A few caveats apply to this first order estimate of $TA^{*sedcorr}$. First, our approach involves a few, likely minor, technical inconsistencies. Aragonite is not treated explicitly in the sediment module and all $CaCO_3$ is assumed to be in the form of calcite; this may tend to bias sediment burial high as calcite is less soluble. The saturation state within the sediments is computed interactively with modelled ocean boundary conditions; this may locally lead to inconsistencies as $CaCO_3$ dissolution

395   within the ocean water column is computed using the prescribed, observation-based saturation state. The alkalinity flux associated with organic matter remineralisation within the sediment is not explicitly distinguished and included in the flux of TA^* from sediments to the ocean; this results in a bias on order of 5 % in the redissolution flux and a negligible influence on the sediment correction.

4 Results

400 ### 4.1 Observation-derived vs. simulated TA^*

Reconstructed TA^* (Fig. 1, TA^{*obs}) is – by definition – close to zero at the ocean surface and correspondingly low within the well-ventilated North Atlantic Deep Waters, Antarctic Intermediate and Mode Waters, and within North Pacific Intermediate Waters. TA^* concentrations in the deep ocean are increasing with the age of water masses (see Fig. A2 of $\Delta^{14}C$ which is a proxy for water mass age)

405   as the dissolution of $CaCO_3$ continues to add TA^* along the flow path. TA^* concentrations are around 40 to 50 mmol m^{-3} in the deep Atlantic (Antarctic Bottom Water) and in the deep Southern Ocean and increase to 130 mmol m^{-3} in the northern Pacific. The reconstructed basin-mean profiles in the Pacific and Indian basin (Fig. 4) show strong gradients in the upper 1500 m and relatively uniform values below that. Concentrations are generally much lower in the Atlantic than in the Pacific.

410   The unconstrained model ensemble yields a large range of TA^* concentrations (Fig. 4, grey shading). The optimisation procedure greatly reduces this range in simulated TA^* to a comparably narrow confidence interval (Fig. 4, green shading representing the 68 % confidence interval). For example, basin-averaged concentrations in the deep Pacific (4000 m) range between 11 to 287 mmol m^{-3} in the unconstrained ensemble, while the corresponding 68 % confidence interval in TA^* is 73 to

415   122 mmol m^{-3} in the constrained ensemble. The selected a priori parameter ranges are therefore wide enough to result in a very broad range of TA* concentrations; the Bayesian optimisation framework confines this initial range around or close to the observation-based values.

The median field from the constrained model ensemble generally captures the observation-based TA* pattern (Fig. 5). The correlation coefficient (r) between the two fields are 0.83, 0.88, and 0.87 in
420   the Atlantic, Pacific, and Indian Ocean, respectively, and the RMSE between the two fields are 17.7, 23.8, 19.8 mmol m^{-3} for the respective basins. These deviations correspond, respectively, to 57, 28 and 28 % of their mean TA* concentration in each basin. Large positive deviations are found in the intermediate waters of the Pacific. This is also evident in the observation-derived basin-mean profiles of TA*, which generally fall within the 68 % confidence interval of the constrained model ensemble
425   (Fig. 4). In the Atlantic, the ensemble median concentrations are, on basin-average, higher than the reconstructed ones. In the Pacific and Indian Oceans, the median clearly overestimates TA* in the thermocline (the 68 % confidence range does not include the observations there) and somewhat underestimates TA* in the deep ocean. This is likely linked to known deficiencies in the model's circulation. Intermediate and mode waters (with low TA* concentrations) are not penetrating far
430   enough towards the equator. As a consequence, mixing of TA* depleted surface waters is too low and TA* concentrations are too high in the thermocline. Alternatively, we cannot exclude that dissolution may be overestimated in the thermocline of the Pacific and Indian Ocean. This data-model mismatch could potentially be reduced by introducing more than one background dissolution rate constant (k_{bg}) or a depth-dependent particle sinking velocity. However, this may simply mask deficiencies in
435   the circulation and we do not attempt such a solution. Generally, the correlation between observed and modelled TA* is remarkably high.

4.2   Probabilistic estimates of CaCO$_3$ and alkalinity fluxes

The estimated global median export flux of CaCO$_3$ at 75 m – with its 68 % confidence interval – is 0.90 (0.72–1.05) Gt-C yr^{-1} (Table 4 and Fig. 6). Basin-wide, we find CaCO$_3$ median export
440   fluxes (and 68 % confidence intervals) of 0.12 (0.078–0.17) Gt-C yr^{-1} from the Atlantic, 0.55 (0.39–0.68) Gt-C yr^{-1} from the Pacific, and 0.23 (0.14–0.32) Gt-C yr^{-1} from the Indian Ocean (Table 4 and Fig. 6). Regionally, largest export fluxes are simulated in the Southern Ocean sector of the Pacific, in the Pacific equatorial upwelling regions, and in the northwestern Pacific (Fig. 7). In the Indian Ocean export fluxes are highest in its eastern tropical regions and in its section of the Southern
445   Ocean. In the tropical and northern Atlantic export fluxes are generally low, consistent with the low TA* values in the bulk of NADW and AABW.

Median CaCO$_3$ export production per unit area in the constrained ensemble is considerably lower in the Atlantic sector of the Southern Ocean (135 mmol-C m^{-2} yr^{-1}) as compared to the Pacific (301 mmol-C m^{-2} yr^{-1}) and Indian (326 mmol-C m^{-2} yr^{-1}) sectors (Fig. 7 top). This is
450   attributable to the choice of the regional boundaries for the rain ratio regions and the assumption

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that the spatial pattern of export within a region is identical to the pattern simulated by the standard version of the model. The standard model yields relatively little zonal variation in the $CaCO_3$ export fluxes in the Southern Ocean in contrast to the data assimilation with lower than zonally-averaged export in the Atlantic. This reflects the much lower TA* reconstructed in the deep Atlantic as compared to the deep Pacific and Indian (Fig. 1). A large export in the Atlantic sector of the Southern Ocean tends to yield high simulated TA* concentrations in the Antarctic Bottom Water that fills the deep Atlantic. The Monte Carlo data assimilation therefore requires low $CaCO_3$ export in the Atlantic sector to minimize model-data mismatches in the deep Atlantic. It is difficult to correctly represent water mass formation and circulation in the Southern ocean and our model may be biased. A known bias is that the Atlantic Bottom Water circulation is too sluggish, also evidenced by simulated low radiocarbon signatures (Figs A2. A4). The influence of a potential bias in South Atlantic export on global $CaCO_3$ export is estimated to be relatively small; assuming the same (median) $CaCO_3$ export per unit area in the Atlantic sector as estimated for the Indian sector would yield $0.06\,\mathrm{Gt\text{-}C\,yr^{-1}}$ higher export than suggested by the ensemble median. While our Monte Carlo approach is suitable to estimate export fluxes over larger regions, the detailed spatial patterns in $CaCO_3$ export remain unconstrained.

Globally, the deposition flux on the respective deepest cells integrates to 0.16 (0.11–0.23) Gt-$\mathrm{C\,yr^{-1}}$, i.e., $\sim 18\,\%$ of the export flux (Fig. 7, Table 4). Local deposition depends on the local $CaCO_3$ export and on how much dissolution is occurring in the water column, which itself depends on the saturation state and on the depth of the water column. Particularly in the North Atlantic, along coastlines and in the Southern Ocean, high fluxes reach the ocean floor. These are dissolved into the water column in the ensemble setup without sediment module. Accordingly, $\sim 82\,\%$ of the global median $CaCO_3$ export dissolves in the water column. More specifically, $\sim 37\,\%$ of the $CaCO_3$ export dissolves in the upper water column above $1500\,\mathrm{m}$, $\sim 44\,\%$ below $1500\,\mathrm{m}$ depth, with the remaining $\sim 18\,\%$ dissolving at the sea floor (Table 4).

Average dissolution profiles for aragonite (red) and calcite (olive) in different ocean regions are displayed in Fig. 8. A peak in aragonite and calcite dissolution is located at or below the depth of the aragonite and calcite saturation horizon, respectively. These dissolution peaks are associated with the saturation-dependent dissolution rate coefficients (Eq. 5). As is the saturation horizon, these peaks are located deep down in the water column of the north and tropical Atlantic region (at 3–$4\,\mathrm{km}$ depth for aragonite), at intermediate depth of the South Atlantic, Indian, and Pacific ($1.5\,\mathrm{km}$ for aragonite) and at relatively shallow depth of the tropical and north Pacific region ($\sim 700\,\mathrm{m}$ for aragonite). As the calcite saturation horizon is found deeper in the water column than the aragonite saturation horizon, so are these dissolution peaks located deeper down in the water column for calcite than for aragonite.

Our constrained ensemble includes non-zero values for the background dissolution rate. Consequently, calcite and aragonite are dissolving throughout the water column, irrespective of the sat-

uration state in the Atlantic, Pacific and Indian Ocean. The percentage of the export flux which dissolves in waters supersaturated with respect to calcite are 72 % (60–80 %), 43 % (30–53 %), 68 %
490 (55–78 %) in the Atlantic, Pacific and Indian Ocean, respectively. We will further investigate in Sect. 4.4 to which extent the finding that a fraction of the $CaCO_3$ export dissolves above the calcite saturation horizon and our export and dissolution flux estimates are robust.

4.3 Sensitivity of results to ocean–sediment interactions, and circulation

4.3.1 Ocean-sediment interactions

495 As discussed in Sect. 3.3, $CaCO_3$ deposition on, burial within, and redissolution from ocean sediments affects TA^* concentrations mainly in the Pacific and Indian Ocean in our model. To assess uncertainties in $CaCO_3$ fluxes arising from uncertainties associated with the sediment correction, we set the sediment correction ($TA^{*sedcorr}$) of the TA^* field to zero to calculate potential skill scores (Eq. 9). This alternative case also illustrates the potential error due to the neglect of ocean–sediment
500 fluxes. Export fluxes of $CaCO_3$ are 2, 12, 8, and 8 % smaller in the Atlantic, Pacific, Indian and global ocean, respectively while the variance is about the same. This is not surprising as we have already noted (Sect. 3.3) that a certain burial flux tends to decrease the TA^* pattern uniformly, i.e. the simulations with and without the sediment correlate highly in terms of TA^*. The PDF of fluxes is therefore shifted to higher export fluxes, while the preference for dissolution parameters remains the
505 same. The $CaCO_3$ that dissolves above 1500 m is 4 % higher, 5 % lower, the same, and 4 % lower in the Atlantic, Pacific, Indian and global ocean, respectively, in the setup that neglects sediment fluxes compared to the standard setup.

4.3.2 Implications of different diapycnal diffusivities illustrated for three dissolution rate profiles

510 The diapycnal diffusivity, k_{dia} of the model is varied to probe uncertainties related to the magnitude of the ocean overturning circulation. k_{dia} is either set to 0.1 (low), 0.2 (standard), or $0.5 \, (\text{high}) \times 10^{-4} \, \text{m}^2 \, \text{s}^{-1}$. Increasing k_{dia} increases the strength of the overturning circulation, and deep ocean ventilation. Maximum Atlantic Meridional Overturning is 16, 18, and 23 Sv (Sverdrups), Southern Ocean overturning is -16, -14, and -15 Sv and maximal deep Pacific overturning is -13,
515 -14, and -20 Sv for the low, standard and high k_{dia} simulations, respectively. We compare the simulated (natural) $\Delta^{14}C$ of DIC and CFC11 distributions to their corresponding observations (Key et al., 2004) to evaluate the physical transport (see Appendix). Both are conservative tracers with a time information and indicative of the ventilation time scales of the thermocline and the deep ocean, respectively. The observation-based (Key et al., 2004) global mean (natural) $\Delta^{14}C$ of DIC is $-151\permil$.
520 The reference simulation achieves a mean (natural) $\Delta^{14}C$ of DIC of $-160\permil$. Correspondingly, ocean radiocarbon signatures become too low ($-176\permil$) with the low diapycnal mixing rate and

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too high with the high ($-126\%_o$) diapycnal mixing rate (see Fig. A4). Simulated surface-to-deep Δ^{14}C gradients are too low (high) relative to the observed gradients for the high (low) diapycnal diffusivity parameter, thereby indicating too fast (slow) surface-to-deep water exchange (Fig. A4). The global observation-based CFC11 inventory is estimated to 575 Mmol. The reference simulation yields 513 Mmol CFC11 as the mean inventory over the modelled period between 1990 and 2000. In the low mixing simulation, this inventory is even lower (479 Mmol) and in the high mixing simulation it is too high (631 Mmol). On a global scale, our reference choice is therefore in better agreement with these physical tracers (see Fig. A4).

We vary k_{dia} for three illustrative dissolution rate profiles introduced in Sect. 3.2 (see Fig. 2 and Table 2). In the low diapycnal diffusivity case, $CaCO_3$ export is 11 % lower and in the high diapycnal diffusivity case, $CaCO_3$ export is 30 % higher compared to the standard case. Larger overturning and mixing yields more nutrient input into the euphotic zone and thus more organic matter export. $CaCO_3$ export is independent of alkalinity in our model and does not depend on the choice of the dissolution rate. The simulated TA* patterns remain similar and correlation between the patterns is at least 0.93. Basin-average profiles in TA* vary little in the upper ocean and the deep Pacific ($< 14\,\mathrm{mmol\,m^{-3}}$) and modestly in the deep Atlantic ($< 44\,\mathrm{mmol\,m^{-3}}$) and deep Indian ($< 18\,\mathrm{mmol\,m^{-3}}$) when varying diapycnal diffusivity between 0.1 and $0.5\times10^{-4}\,\mathrm{m^2\,s^{-1}}$. Perhaps somewhat surprisingly, the simulated TA* inventory is relatively weakly affected by the choice of k_{dia}; the global TA* inventory varies by less than 7 % across the range of k_{dia} (Table 2). In other words, variations in the magnitude of ocean ventilation do hardly affect the TA* inventory for a given dissolution rate profile. Higher (lower) export under higher (lower) mixing compensate each other. These changes in TA* inventory are smaller than the influence of the sediment correction (see Table 3) or the choice of the dissolution profile. In conclusion, simulated TA* is only weakly affected by uncertainties in the diapycnal mixing coefficient.

The choice of the dissolution rate profile has a substantial influence on the simulated TA* inventory (Table 2). As mentioned previously, TA* accumulates in the deep ocean when all $CaCO_3$ is dissolved below the saturation horizon of aragonite and calcite and no dissolution is permitted above. This raises the question whether surface-to-deep transport is too slow in our model. As mentioned above, the radiocarbon signatures as well as CFC11 concentrations are on average close to observations (Figs. A3 and A2). Increasing ocean ventilation by increasing k_{dia} results in too young radiocarbon signatures and higher than observed CFC11 concentrations. However, it does not substantially reduce the overestimation of the TA* inventory in those cases.

4.4 How to parameterise $CaCO_3$ dissolution in an Earth System Model?

An important question is how to formulate the dissolution rate of calcite and aragonite particles in Earth System Models. Should the dissolution rate be a function of the simulated aragonite and calcite saturation state of the surrounding water? Should dissolution above the saturation horizon

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be permitted? In our ensemble, the degrees of freedom allow for either fast or slow dissolution kinetics as a function of saturation, and the option to also dissolve above the saturation horizon with

560 a constant dissolution rate. For a more systematic understanding of which parameterisations achieve high skill we return to the sensitivity simulations (Fig. 2).

Interestingly, the different dissolution schemes yield highly correlated patterns of TA*; the correlation coefficient between the global fields from the fast and slow dissolution scheme is 0.85, and between the fields from the fast and constant scheme is 0.88 in simulations with the same export

565 (Fig. 2). The high correlation between fast and slow dissolution schemes is not surprising as only 35% of the modern ocean's volume is actually undersaturated with respect to calcite. Differences therefore cannot emerge over much of the ocean and skill is therefore distributed across the range of parameter values sampled.

Figure 9 displays We analyse the relationship between model skill, background dissolution

570 rate dissolution parameterisation, $CaCO_3$ export, and TA* inventories in the model ensemble (see Fig. 9) to address these questions. To achieve a high skill (green-to-red coloured dots in the upper panels of Fig. 9), an individual ensemble member needs to reproduce the observation-based TA* inventory (dashed line) within a limited range. We identify an export range within which TA* can be reproduced skillfully (vertical green range in the upper panels of Fig. 9). The background

575 rate is responsible for this spread Surprisingly, a high skill is achieved across the range of different dissolution schemes applied and for a broad range of parameter values. In other words, neither the dissolution scheme nor its parameters are well constrained by the observation-based TA* field. This is illustrated by plotting the value of the background dissolution rate, k_{bg} (lower panels of Fig. 9) as a function of the TA* inventory and export. Generally, the higher the global $CaCO_3$ export flux,

580 the higher the background dissolution rate required to achieve a high skill. Likewise, lower export can be distributed skillfully without dissolution in supersaturated waters ($k_{bg} = 0$). Apparently, there are trade-offs between the magnitude of export and the applied dissolution parameterisation in terms of TA*, suggesting that export and dissolution parameters can only be constrained simultaneously within limits when using TA* as the only constraint. In other words, neither the dissolution scheme

585 nor its parameters are well constrained by

These findings are in line with the results from our sensitivity simulations (Fig. 2, Table 2). The parameterisation with dissolution above the saturation horizon (constant) yields the lowest TA* inventory, followed by the scheme with fast dissolution below the saturation horizon, and the scheme with dissolution near the ocean floor (slow). A high (low) export is thus required

590 for parameterisations with high (low) dissolution above the saturation horizon to simulate the observation-based TA* field. Corresponding to the high correlation, the RMSE between normalised observed and normalised modelled fields vary within a limited range (42 to 58%) for the three dissolution cases. Further, the different dissolution schemes yield highly correlated TA* fields; the correlation coefficient between the global fields from the fast and slow dissolution scheme is 0.84,

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595 and between the fields from the fast and constant scheme is 0.88 in simulations with the same $CaCO_3$ export (Fig. 2). The average RMSE between the fast and slow dissolution scheme is $43.2 \, \text{mmol m}^{-3}$ and between the fast and constant is $24.5 \, \text{mmol m}^{-3}$. The high spatial correlation in simulated TA^* and uncertainties in $CaCO_3$ export make it difficult to distinguish different dissolution parameterisations. The magnitude of $CaCO_3$ exports modulates absolute TA^* concentrations and thus model-

600 data bias and root mean square errors. Given these uncertainties, we cannot objectively determine the preferred dissolution scheme.

4.5 Flux measurements as additional constraint

Global sediment trap data represent another observational constraint of the $CaCO_3$ cycle. The database of Wilson et al. (2012, see Fig. A5) includes 156 measurements globally and is an up-

605 date of the Honjo et al. (2008) compilation. For comparison, we constrained the model ensemble using these sediment trap data instead of the TA^* data as target (Figs. A5 and A6). Skill-scores are calculated individually for the Pacific, Atlantic, and Indian Ocean. $CaCO_3$ export fluxes, dissolution profiles and parameters constrained with the sediment trap data are consistent within uncertainties with those constrained by the TA^* data. The sediment trap data, however, yield wider uncertainty

610 ranges, as illustrated in Fig. A6, and do therefore not permit us to reduce uncertainty ranges any further.

5 Discussion

5.1 Export of $CaCO_3$

Berelson et al. (2007) summarised current estimates of $CaCO_3$ export out of the euphotic zone

615 (based on models and data) to 0.4–$1.8 \, \text{Gt-C yr}^{-1}$ (spanning factor ~ 4). Our constrained median estimate of ~ 0.90 (ensemble range: 0.72–1.05) Gt-C yr^{-1} therefore lies on the lower part of these previously published estimates. At the end, these authors suggest that global $CaCO_3$ export must be higher than $1.6 \, \text{Gt-C yr}^{-1}$. This estimate is based on sediment trap data and other information constraining the flux to the deep ocean ($> 2000 \, \text{m}$) to $0.6 \pm 0.4 \, \text{Gt-C yr}^{-1}$ and results obtained

620 with the so-called TA^*-CFC age method suggesting an upper ocean dissolution of $1 \, \text{Gt-C yr}^{-1}$. The TA^*-CFC age method is heavily criticised by Friis et al. (2006) and tends to bias estimates systematically towards high values (Friis et al., 2006). This method and its shortcomings are further discussed in the next section on upper ocean dissolution. While our estimated flux to the deep ocean of 0.52 (0.43–0.61) Gt-C yr^{-1} is roughly consistent with the budget of Berelson et al. (2007), their

625 export estimate, and upper ocean dissolution, is in clear conflict with our results and those of other studies (Lee, 2001; Sarmiento and Gruber, 2006; Jin et al., 2006) that apply a range of different methodologies. We attribute this mismatch to deficiencies in the TA^*-CFC age method, implying that the export estimate by Berelson et al. (2007) is biased high.

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Global scale $CaCO_3$ export has been previously estimated (see Fig. 10). Jin et al. (2006) diag-
nosed a global $CaCO_3$ export of $1.1\,Gt\text{-}C\,yr^{-1}$ by restoring annual-mean potential alkalinity to ob-
servations (Key et al., 2004) in the euphotic zone and in the whole water column within the Modular
Ocean Model. Sediment burial and sediment-ocean fluxes are included implicitly in this approach.
They provide a relatively small uncertainty range of 0.8 to $1.2\,Gt\text{-}C\,yr^{-1}$ which falls within our
range. Most of the uncertainty is attributed to uncertainties in the alkalinity data by these authors.
Sarmiento and Gruber (2006) estimated $CaCO_3$ export ($\sim 0.5\,Gt\text{-}C\,yr^{-1}$) by combining satellite
NPP (as a mean of the three algorithms (Behrenfeld and Falkowski, 1997; Carr, 2002; Marra et al.,
2003)) with the organic particle export model of Dunne et al. (2005) and the rain-ratio estimate of
Sarmiento et al. (2002). Lee (2001) derived net $CaCO_3$ production ($\sim 0.92 \pm 0.3\,Gt\text{-}C\,yr^{-1}$) from
seasonal potential alkalinity decreases. Both approaches focus on information of the surface ocean
without taking advantage of information displayed in the full biogeochemical depth profile. In addi-
tion, Ridgwell et al. (2007) used an Ensemble Kalman Filter approach to assimilate total alkalinity
and phosphate data into their model and determined a global $CaCO_3$ export flux of $1.2\,Gt\text{-}C\,yr^{-1}$.
On average, these four studies yield a global mean $CaCO_3$ export of $0.93\,Gt\text{-}C\,yr^{-1}$, close to our
median estimate of $0.90\,Gt\text{-}C\,yr^{-1}$.

We find peaks in zonally-averaged export in the North Pacific, the tropical Pacific and Indian and
in the Southern Ocean and low export fluxes in the subtropics and the tropical Atlantic (Fig. 11).
This is in agreement with the results of Jin et al. (2006) and Lee (2001), with the exception that
Lee (2001) suggest little $CaCO_3$ export in all tropical regions. A significant $CaCO_3$ export in the
tropics is consistent with deep ocean sediment trap data (Francois et al., 2002; Berelson et al., 2007).
As noted by Jin et al. (2006), the low estimates of Lee (2001) for tropical regions is likely related
to small signals in seasonal alkalinity in the tropics, hampering their calculations. Concerning the
magnitude of the export, our zonally-averaged values are similar to these two studies in the Pacific
sector of the Southern Ocean, and smaller in the Atlantic and Indian Southern Ocean sectors as well
as in the North Pacific.

Interestingly, Lee (2001) find significant $CaCO_3$ export in the North Atlantic, in contrast to Jin
et al. (2006), Sarmiento and Gruber (2006), and this study. The relatively low North Atlantic export
suggested by these latter studies appears to be in conflict with the occurrence of coccolithophorid
blooms in this region (Brown and Yoder, 1994). On the other hand, the low TA* inventory in the
Atlantic (Fig. 1) argues for a limited export of $CaCO_3$ in this basin.

Carter et al. (2014) defined another alkalinity tracer, termed Alk*, that isolates the portion of the
alkalinity signal that varies in response to calcium carbonate cycling and exchanges with terrestrial
and sedimentary environments from the portion that varies in response to freshwater and organic
matter cycling. These authors compiled a riverine input of alkalinity into the low-latitude Atlantic
($> 40°\,S$ and $< 40°\,N$) equivalent to $0.057\,Gt\text{-}C\,yr^{-1}$, which is $\sim 40\%$ of the global continentally
derived alkalinity. Their Alk* tracer in the Atlantic has the lowest open-ocean surface concentrations

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despite this large riverine source. They conclude that these large riverine inputs must therefore be more than balanced by strong net $CaCO_3$ formation. For comparison, our estimates of $CaCO_3$ export between 35° S and 30° N in the Atlantic are 0.06 (0.018–0.1) Gt-C yr^{-1}. We note, however, that a direct comparison between these two studies remains difficult. The river input does not strictly have to be compensated by the export flux but rather by the burial flux. Also, the river input gets mixed and is subducted and transported southward by North Atlantic Deep Water and interpretation of concentrations without explicit consideration of transport and mixing is always difficult.

Regional patterns of $CaCO_3$ export (Fig. 10) vary among the different studies. In our approach $CaCO_3$ export fluxes are scaled by a rain ratio to simulated export of particulate organic carbon within each of the eight considered regions. Thus, the total export for the three Southern Ocean sectors, the Indian Ocean, the tropical and northern Pacific and the tropical and northern Atlantic is constrained by the TA* data, but not the pattern within each of these regions.

5.2 $CaCO_3$ dissolution in the upper ocean

The $CaCO_3$ leaving the surface ocean dissolves within the water column, at the sea floor or gets buried. Berelson et al. (2007) get to global dissolution rates of ~ 1 Gt-C yr^{-1} within upper level waters, clearly higher than our TA*-based estimate of 0.35 (0.26–0.46) Gt-C yr^{-1}. On a regional level, only two out of ten regional estimates by Berelson et al. (2007) are within our Latin-Hypercube uncertainty ranges; these are the estimates for the tropical Pacific (taken as 5° N to 5° S) and the low- and mid-latitude Atlantic (40° S to 40° N) (Table 5). High dissolution rates in the range of ~ 0.1 to 0.4 mmol-C m^{-3} yr^{-1} are estimated by Barrett et al. (2014) for a transect in the upper tropical and northern North Atlantic. These values are much larger than our estimates of order 0.01 mmol-C m^{-3} yr^{-1} for the upper tropical and northern Atlantic. These high estimates are based on the measured decrease of suspended $CaCO_3$ particles with depth multiplied with a $CaCO_3$ particle settling velocity of 80 m day^{-1} and neglecting any temporal trend in the $CaCO_3$ particle concentration. These estimates may be affected by uncertainties in the assumed particle settling velocity. $CaCO_3$ particles settling velocities are reported by Jansen et al. (2002), to vary greatly (0.15 to 3440 m day^{-1}) and to be typically order one m day^{-1} for coccolitophorides and several 100 m day^{-1} for foraminifera and pteropods. Our dissolution rates would be consistent with the measured depth gradient in suspended biogenic $CaCO_3$ particles for an average settling velocity of a few m day^{-1}.

As mentioned above, we link the differences between the estimates of Berelson et al. (2007) and this study to methodological problems (Friis et al., 2006) associated with the TA*-CFC age method that very likely introduce a high bias in the results of Berelson et al. (2007).

The TA*-CFC age method relies on deduced, observation-derived TA* concentrations and estimates of water mass age, typically derived from measurements of chlorofluorocarbons (CFCs) and their known atmospheric history. TA* concentrations are plotted against their CFC-age and a line is fitted to this data. The higher the TA* concentration for a given water mass, the more TA* must

have been added by dissolution to this particular water parcel according to this method. The slope of the relationship between TA* and age is in this sense the $CaCO_3$ dissolution rate (mol volume^{-1} time^{-1}).

705   The method has been criticised for its neglect of explicit transport and mixing processes. In particular, Friis et al. (2006) noted that TA* signals ended up above the saturation horizon in their model run, even though there was explicitly no dissolution allowed to occur there. This is confirmed by our sensitivity simulations with no dissolution above the saturation horizon (Fig. 2). This finding does not depend on the choice of the numerical model as mixing within the ocean must spread the TA*

710   signal within the ocean and establish a surface-to-deep gradient in TA* in the upper ocean even when all $CaCO_3$ dissolves at great depth. The TA*-CFC age method does not account for such processes and assigns dissolution rates in the waters above the saturation horizon irrespective of whether or not the signal stems from the deep ocean. Our approach to combine TA* data within an ocean transport model and the approach by Jin et al. (2006) or Lee (2001) avoid this shortcoming.

715   In addition, we notice, that in most regions the upper ocean TA* distribution remains remarkably similar within our ensemble, even for very different dissolution rate profiles (see Fig. 2). And in particular, that, for a given export, k_{slow} dissolution rate profiles, with most of the dissolution at or near the ocean bottom, tend to reach highest values of TA* both in the deep ocean and the thermocline (see also inventories in Table 2). In these cases, most of the TA* source is added to slowly ventilated

720   waters in the deep where it accumulates over time along deep water flow paths. This high TA* signal is being brought to the thermocline and eventually to the surface where TA* is reset to its preformed value of zero. As a result, a larger TA* gradient is established across the thermocline for deep compared to shallow dissolution. The TA*-CFC age method would therefore have a tendency to assign higher dissolution rates in the wrong cases. This further illustrates the difficulty to uniquely relate

725   upper ocean tracer concentrations to either dissolution or mixing processes.

5.3 $CaCO_3$ dissolution in the deep ocean

Turning to the deep ocean, Berelson et al. (2007) suggest based on sediment trap and benthic dissolution data that the particle flux below 2000 m is 0.6 ± 0.3 Gt-C yr^{-1}, sea floor dissolution for sites > 2000 m averages 0.4 ± 0.3 Gt-C yr^{-1}, and carbonate burial in deep marine sediments is 0.1 Gt-

730   C yr^{-1}. Our estimate of the particle flux below 1500 m is 0.52 (0.43–0.61) Gt-C yr^{-1}. Thus, the compilation of sediment trap data by Berelson et al. (2007) roughly supports our particle flux at 1500 m. However, the split between sea floor dissolution and open water dissolution in the deep is different. We estimate that most of the deep ocean particle flux dissolves within the water column (0.40 (0.32–0.47) Gt-C yr^{-1} below 1500 m). The steady state burial flux for the runs with

735   interactive sediments is 0.12 Gt-C yr^{-1}, corresponding to the burial of 1.95×10^{13} mol Alk, yr^{-1}, this is comparable in magnitude but smaller than the total alkalinity input by rivers estimated to be 2.3×10^{13} mol yr^{-1} by Carter et al. (2014) or the burial flux estimated by Dunne et al. (2012) of

$0.121 \, \text{Gt-C} \, \text{yr}^{-1}$. We note that $CaCO_3$ formation by coral reefs and burial in shallow coastal waters (Milliman, 1993; Vecsei and Berger, 2004) is not considered in our coarse resolution model.

5.4 Parameterisations of $CaCO_3$ within Earth System Models

Uncertainties appear too large to objectively determine well-defined parameter ranges for the dissolution rates of calcite and aragonite (judging from both the TA* and the flux data compilation). A good agreement between simulated and observed TA* fields can be achieved irrespective of whether dissolution is assumed to depend on the calcite or aragonite saturation state or whether dissolution rates are assumed to be constant throughout the water column. We recall that the computation of the saturation state by carbonate chemistry routines across all grid cells and for each model time step poses a considerable computational burden. For simplicity and to minimise computational costs, we therefore recommend to describe $CaCO_3$ dissolution by a constant dissolution rate in Earth System Models as long as these uncertainties exist. This yields an exponential particle flux profile when assuming constant settling velocities throughout the water column. This approach is used in previous studies (Archer and Maier-Reimer, 1994; Ridgwell et al., 2007). A shortcoming of the application of an exponential particle flux profile for $CaCO_3$ is that it is not easy to account for the potential influence on dissolution of changes in environmental variables, including a decrease in saturation state as expected under ongoing ocean acidification, or in the quality, form, and size distribution of exported $CaCO_3$ particles.

6 Summary and conclusions

Constraining the $CaCO_3$ cycle comes down to three fundamental questions: how much $CaCO_3$ is exported from the surface ocean, where is this $CaCO_3$ dissolved in the water column and how much is buried in sediments. Here, we setup a probabilistic framework to constrain the $CaCO_3$ budget within the Bern3D EMIC with observationally-based TA* as robust target variable. The saturation state of water with respect to calcite and aragonite is prescribed using observational estimates to provide realistic boundary conditions for the $CaCO_3$ dissolution parameterization. In addition to the uncertainty estimates obtained by our Bayesian Latin-Hypercube framework, we also consider uncertainties related to the choice of metrics to define the assimilation target, including flux measurements as alternative target variable, uncertainties in ocean transport, and ocean sediment interactions.

We estimate that 0.72–1.05 Gt-C with a best estimate of 0.90 Gt-C are exported out of the surface ocean each year in the form of biogenic $CaCO_3$. Of this about 37 % (0.33 (0.2–0.4) $\text{Gt-C} \, \text{yr}^{-1}$) are estimated to dissolve in the open water column above $1500 \, \text{m}$, and 44 % (0.40 (0.32–0.47) Gt-C yr^{-1}) in the open water column below, with the reminder (0.16 (0.113–0.23) $\text{Gt-C} \, \text{yr}^{-1}$) deposited on open ocean sediments, globally. Sensitivity simulations with interactive sediments sug-

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gest that about 30% of the deposition flux dissolves back into the ocean and 70% gets buried in consolidated sediments.

We find that the higher the export fluxes within the constrained, likely ranges, the more likely dissolution above the saturation horizon is needed to distribute TA* skillfully in the model. Different kinds of dissolution schemes (with and without dissolution above saturation) achieve realistic TA* distributions within the export ranges identified. Therefore, background dissolution above the saturation horizon cannot be ruled out from this Latin-Hypercube ensemble evaluated within the Bern3D EMIC. Future progress likely depends on a better observational characterisation of particle concentrations, size distribution, and settling within the water column. Physical transport and mixing reduces concentration gradients such that conceptually different dissolution schemes (e.g. dissolution permitted above saturation or not) cannot be distinguished statistically. This implies that concentrations cannot be used to infer dissolution rates directly. It also implies that dissolution parameterisations remain uncertain. For simplicity and to minimise computational costs, we suggest to use saturation-independent parameterisations of $CaCO_3$ dissolution within Earth System Models.

Appendix: Model evaluation and additional constraints

It is an essential prerequisite for the model to feature the main water masses and mixing time scales of the ocean to realistically simulate biogeochemical tracers such as TA*. In this Appendix, we graphically document ocean model performance by comparing simulated and observation-based distributions for a range of tracers. Results are for a pre-industrial steady state of the Bern3D ocean model configuration with a horizontal resolution of 40 by 41 grid cells and 32 vertical layers as coupled to an energy balance and sea ice module without sediment module. The atmospheric history of CFC11 is prescribed according to Bullister (2011). Here we do not account for potential changes in ocean circulation and CFC11 solubility over the industrial period.

Figure A1 provides a Taylor Diagram (Taylor, 2001) of CFC11, $\Delta^{14}C$, temperature, salinity, DIC, TA, PO_4, oxygen, and TA* of the standard model configuration with a constant rain ratio of 7 % and of the median TA* of the weighted model ensemble. Modelled and observed distributions of the ventilation tracers $\Delta^{14}C$ and CFC11 are compared along a section through the Atlantic, Southern Ocean and Pacific (Figs. A2 and A3). Simulated and observation-based basin-mean vertical gradients of CFC11 and natural $\Delta^{14}C$ are compared in Fig. A4 for three simulations where diapycnal mixing is set to a low, the standard, and a high value ($k_{dia} = (0.1, 0.2, 0.5) \times 10^{-4} \, m^2 \, s^{-1}$). Results bracket observation-based profiles with best agreement between model and observations for the standard setup.

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Table 1. Overview on suggested ways of regression for TA^0 either including two (S, PO) or three (S, PO, T) explanatory variables and either as a global or basin-wide fit. The last column shows the root mean square deviation relative to the Global B3D estimate.

|	Equation [ueq kg^{-1}]	Mean [mol m^{-3}]	RMSE from Global B3D [mol m^{-3}]
Gruber et al. (1996):	$TA^0 = (367.5 + 59.9\,\mathrm{psu}^{-1} \cdot S + 0.074\,\mathrm{kg\,umol}^{-1} \cdot PO)\,\mathrm{ueq\,kg}^{-1}$	2.377	0.012
Feely et al. (2002):	$TA^0 = 148.7 + 61.36 \cdot S + 0.0941 \cdot PO - 0.582 \cdot T_{\mathrm{pot}}$	2.39	0.0015
Friis et al. (2006):	as Feely et al. (2002)		
Global B3D 2V:	$TA^0 = (297.51 + 56.399\,\mathrm{psu}^{-1} \cdot S + 0.1259\,\mathrm{kg\,umol}^{-1} \cdot PO)\,\mathrm{ueq\,kg}^{-1}$	2.387	0.0022
Global B3D:	$TA^0 = (345.64 + 56.03\,\mathrm{psu}^{-1} \cdot S + 0.069\,\mathrm{kg\,umol}^{-1} \cdot PO - 0.9\,^\circ\mathrm{C}^{-1} \cdot T)\,\mathrm{ueq\,kg}^{-1}$	2.389	0
Regional B3D:		2.385	0.0047
Atlantic	$TA^0 = (688.15 + 44.97\,\mathrm{psu}^{-1} \cdot S + 0.129\,\mathrm{kg\,umol}^{-1} \cdot PO + 1.34\,^\circ\mathrm{C}^{-1} \cdot T)\,\mathrm{ueq\,kg}^{-1}$		
Pacific	$TA^0 = (381.05 + 55.26\,\mathrm{psu}^{-1} \cdot S +, 0.049\,\mathrm{kg\,umol}^{-1} \cdot PO - 1.20\,^\circ\mathrm{C}^{-1} \cdot T)\,\mathrm{ueq\,kg}^{-1}$		
Indian	$TA^0 = (637.70 + 47.38\,\mathrm{psu}^{-1} \cdot S + 0.078\,\mathrm{kg\,umol}^{-1} \cdot PO - 0.54\,^\circ\mathrm{C}^{-1} \cdot T)\,\mathrm{ueq\,kg}^{-1}$		

Gruber et al. (1996) is the original, global regression based on two explanatory variables (based on their 1996 database), Feely et al. (2002) and Friis et al. (2006) are both taken from Sabine et al. (2002b) also including temperature as an additional explanatory variable and fitted to a Pacific subset. Global B3D 2V is a global regression with two explanatory variables (based on WOA09 Locarnini et al., 2010; Antonov et al., 2010; Garcia et al., 2010a, b) on the Bern3D grid ($32 \times 40 \times 41$ grid cells), Global B3D includes three explanatory variables and Regional B3D further distinguishes each basin, separately.

Table 2. Results from sensitivity simulations. $CaCO_3$ export and TA^* inventories for different physical mixing (diapycnal mixing coefficient, k_{dia}) and $CaCO_3$ dissolution schemes. For these illustrative simulations, calcite and aragonite particles were assigned equal parameter values and 10 % of export is assumed to be in the form of aragonite. Fast: $k_0 = 10\,\mathrm{day}^{-1}$, $n = 1$, $f_{\mathrm{calc}} = 0.9$, $k_{\mathrm{bg}} = 0$; slow: $k_0 = 0.16\,\mathrm{day}^{-1}$, $n = 2$, $f_{\mathrm{calc}} = 0.9$, $k_{\mathrm{bg}} = 0$; constant: $k_{\mathrm{bg}}/v = 1/2900\,\mathrm{m}^{-1}$.

	$k_{\mathrm{dia,\ low}}$ $0.1 \times 10^{-4}\,\mathrm{m}^2\,\mathrm{s}^{-1}$	$k_{\mathrm{dia,\ ref}}$ $0.2 \times 10^{-4}\,\mathrm{m}^2\,\mathrm{s}^{-1}$	$k_{\mathrm{dia,\ high}}$ $0.5 \times 10^{-4}\,\mathrm{m}^2\,\mathrm{s}^{-1}$
Export at 75 m [Gt-C yr^{-1}]	0.727	0.818	1.071
TA^* Inventory [Pmol C]			
(fraction of which lies above $\Omega_{\mathrm{calc}} = 1$)			
Fast	47.81 (54 %)	47.87 (52 %)	44.5 (51 %)
Slow	62.49 (45 %)	63.01 (44 %)	58.93 (43 %)
Constant	38.12 (52 %)	38.35 (51 %)	35.44 (50 %)

Table 3. Mean sediment burial fluxes $[\text{Gt-C yr}^{-1}]$ and observation-based vs. simulated TA^* inventories [Pmol C]. $\text{TA}^{*\text{sedcorr}}$ are estimated by comparing output from the 14 Bern3D simulations that agree best with observation-based TA^* concentrations and with and without sediment module. The median estimate is from the constrained 1000 member ensemble (without sediment module).

	Atlantic	Pacific	Indian	Global
Burial flux $[\text{Gt-C yr}^{-1}]$	0.011	0.062	0.042	0.117
TA^* Inventory [Pmol C]				
TA^{*obs}	2.5	27.9	7.0	37.4
$\text{TA}^{*sedcorr}$	0.9	4.6	1.4	6.9
$\text{TA}^{*obs+sedcorr}$	3.4	32.4	8.4	44.1
$\text{TA}^{*median}$	5.4	31.1	8.7	45.5
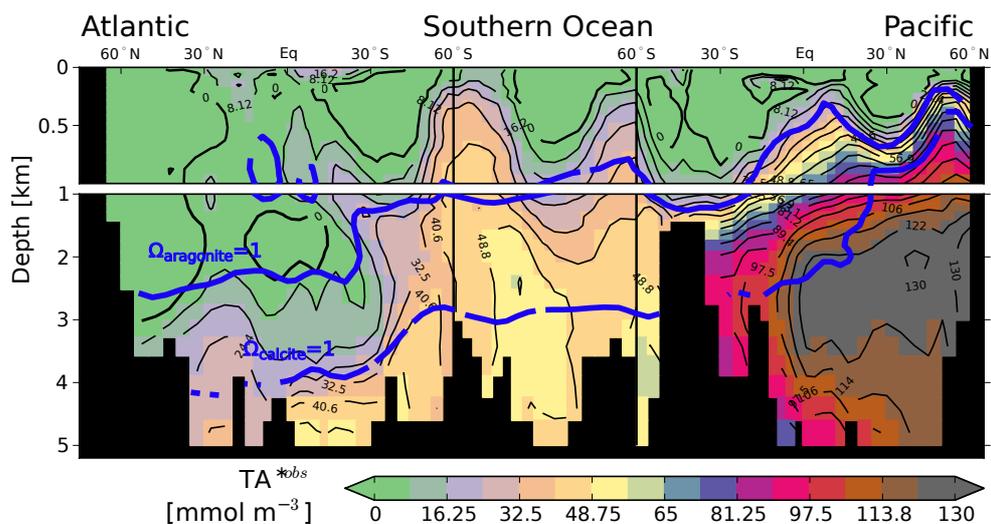
Figure 1. The TA^* tracer captures exclusively the influence of CaCO_3 dissolution on alkalinity. An observationally-based estimate of the TA^* distribution $[\text{mmol m}^{-3}]$ (based on GLODAP Key et al. (2004) and WOA09 Locarnini et al., 2010; Antonov et al., 2010; Garcia et al., 2010a, b) yielding an inventory of $\sim 37.5\,\text{Pmol C}$ of which $\sim 41\,\%$ comes to lie above the calcite saturation horizon. The calcite ($\sigma_{\text{calcite}}=1$) and aragonite ($\sigma_{\text{aragonite}}=1$) saturation horizon are shown by the blue lines. Displayed are results for a cross-section through the Atlantic (25° W), across the Southern Ocean (58° S) into the Pacific, and through the Pacific (175° W).

Table 4. Constrained fluxes (median and 68 % confidence interval, c.i.) of biogenic $CaCO_3$ [Gt-C yr^{-1}].

|	Atlantic median	c.i.	Pacific median	c.i.	Indian median	c.i.	Global median	c.i.
Export at 75 m								
70 – 30°N	0.021	[0.007-0.037]	0.081	[0.027-0.134]				
30°N – 35°S	0.061	[0.018-0.103]	0.307	[0.149-0.457]	0.143	[0.053-0.223]		
>35°S	0.041	[0.014-0.07]	0.174	[0.057-0.254]	0.088	[0.03-0.14]		
70°N – 90°S	0.121	[0.078-0.171]	0.549	[0.391-0.697]	0.23	[0.137-0.316]	0.897	[0.72-1.049]
Dissolution in waters shallower than 1500 m								
70 – 30°N	0.006	[0.002-0.011]	0.048	[0.015-0.086]				
30°N – 35°S	0.015	[0.004-0.03]	0.118	[0.049-0.2]	0.043	[0.016-0.08]		
>35°S	0.01	[0.003-0.021]	0.043	[0.013-0.082]	0.02	[0.006-0.042]		
70°N – 90°S	0.033	[0.018-0.054]	0.216	[0.125-0.32]	0.064	[0.032-0.11]	0.328	[0.202-0.444]
Deposition on sediments shallower than 1500 m								
70 – 30°N	0.003	[0.001-0.005]	0.008	[0.002-0.014]				
30°N – 35°S	0.003	[0.001-0.006]	0.008	[0.004-0.013]	0.015	[0.005-0.024]		
>35°S	0	[0.0-0.001]	0.007	[0.002-0.011]	0	[0.0-0.001]		
70°N – 90°S	0.007	[0.004-0.01]	0.023	[0.016-0.029]	0.016	[0.006-0.025]	0.045	[0.034-0.056]
Dissolution in waters deeper or equal to 1500 m								
70 – 30°N	0.008	[0.003-0.013]	0.014	[0.003-0.035]				
30°N – 35°S	0.03	[0.011-0.048]	0.133	[0.068-0.2]	0.053	[0.022-0.087]		
>35°S	0.023	[0.009-0.038]	0.085	[0.031-0.129]	0.043	[0.015-0.07]		
70°N – 90°S	0.06	[0.039-0.083]	0.233	[0.165-0.305]	0.097	[0.061-0.139]	0.395	[0.32-0.469]
Deposition on sediments deeper or equal to 1500 m								
70 – 30°N	0.003	[0.0-0.007]	0.002	[0.001-0.007]				
30°N – 35°S	0.009	[0.001-0.021]	0.031	[0.012-0.062]	0.019	[0.005-0.039]		
>35°S	0.005	[0.001-0.011]	0.02	[0.006-0.041]	0.016	[0.005-0.03]		
70°N – 90°S	0.018	[0.009-0.033]	0.056	[0.028-0.099]	0.035	[0.018-0.065]	0.116	[0.071-0.179]
Deposition on all sediments								
70 – 30°N	0.006	[0.001-0.012]	0.011	[0.003-0.021]				
30°N – 35°S	0.013	[0.002-0.027]	0.04	[0.017-0.073]	0.036	[0.011-0.061]		
>35°S	0.005	[0.001-0.012]	0.028	[0.008-0.051]	0.016	[0.005-0.03]		
70°N – 90°S	0.025	[0.014-0.043]	0.081	[0.049-0.126]	0.052	[0.026-0.087]	0.164	[0.113-0.227]

Table 5. Dissolution rates of biogenic $CaCO_3$ [Gt-C yr^{-1}] in the upper water column (200–1500 m depth levels) based on the TA^*-CFC age method as summarised in Berelson et al. (2007), and as constrained by TA^* data in our Latin-Hypercube ensemble (including the sediment correction, median and 68 % c.i.). The dissolution estimates by Berelson et al. (2007) were assigned an estimated uncertainty of $\sim 50\%$.

Location	Berelson et al. (2007)	This study median	c.i.
Atlantic			
>40°N	0.060	0.006	[0.002-0.011]
40°N - 40°S	0.010	0.02	[0.008-0.036]
>40°S	0.040	0.009	[0.003-0.017]
Pacific			
>40°N	0.070	0.039	[0.013-0.067]
40°N - 5°N	0.330	0.063	[0.032-0.099]
5°N - 5°S	0.020	0.035	[0.015-0.058]
5°S - 40°S	0.000	0.039	[0.019-0.066]
>40°S	0.160	0.034	[0.011-0.061]
Indian			
40°N - 40°S	0.160	0.062	[0.026-0.104]
>40°S	0.140	0.021	[0.009-0.038]
Total	1.0	0.346	[0.225-0.461]
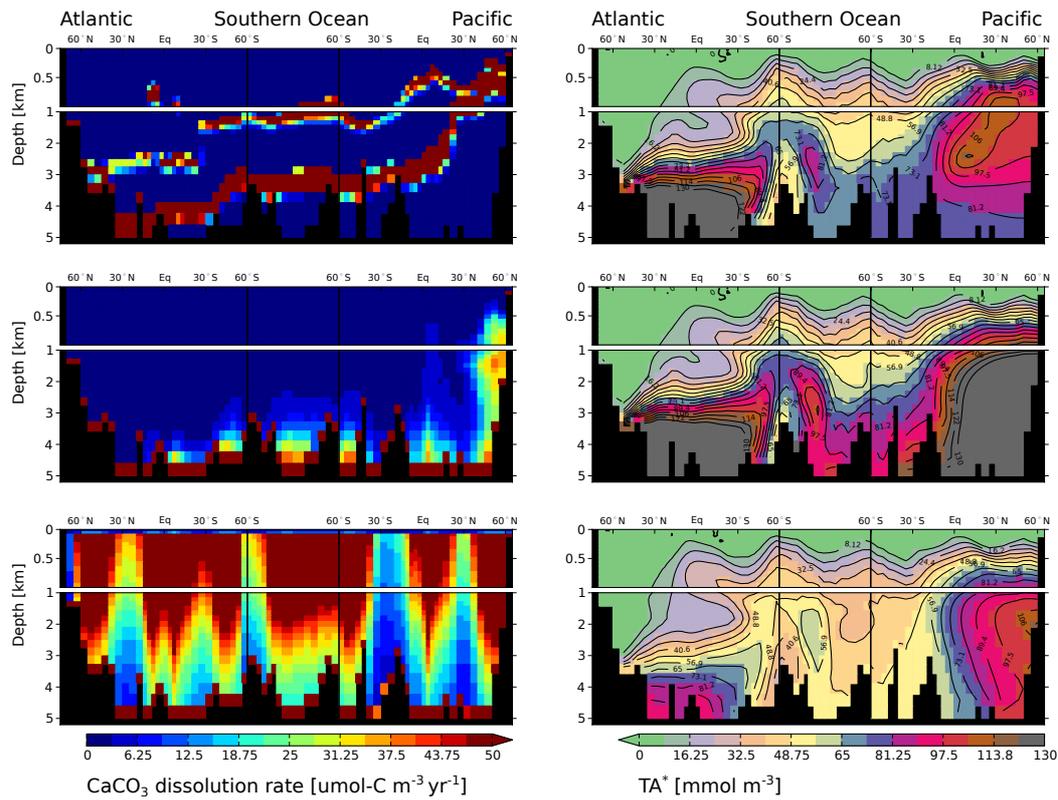
Figure 2. Results from sensitivity simulations applying three different illustrative dissolution schemes. Left column: $CaCO_3$ dissolution rates. Right column: the resulting steady state TA* for these three contrasting parameterisations of the $CaCO_3$ dissolution rate. The dissolution rate is set to increase fast (top row), slowly (middle row) with undersaturation of $CaCO_3$, or is set constant throughout the water column (bottom row). The standard version of the Bern3D model without sediment module is applied and the fraction of $CaCO_3$ export in the form of aragonite is set to 10 %. At least 43 % of the dissolution signal is simulated above the calcite saturation horizon irrespective of whether dissolution is allowed to occur above the saturation horizon (bottom row) or not (middle and top row). This points to the importance of physical transport in shaping the distribution of TA*.

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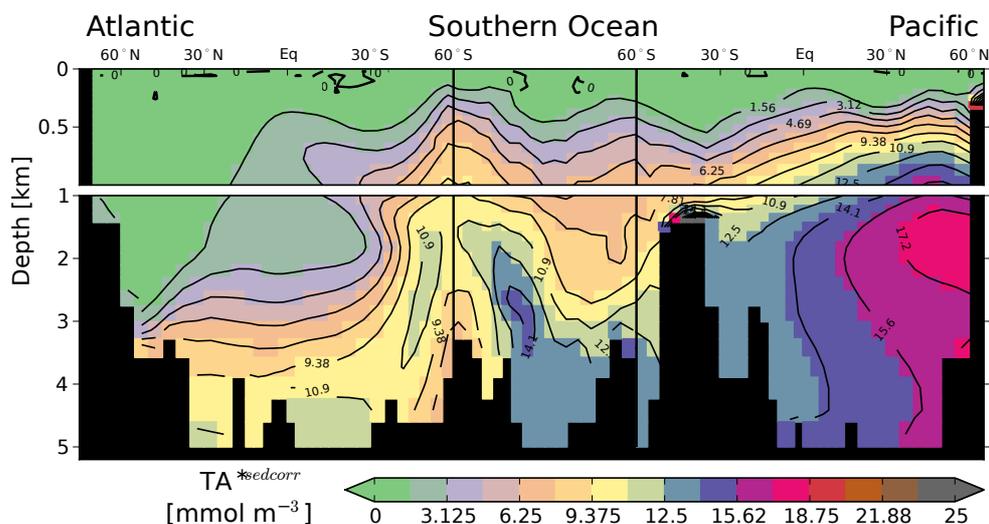
Figure 3. Influence of ocean–sediment fluxes on TA*. The mean offset in TA* between simulations with and without sediment module is shown for the 14 ensemble members with the highest skill. Displayed are results for a cross-section through the Atlantic ($25°$ W), across the Southern Ocean ($58°$ S) into the Pacific, and through the Pacific ($175°$ W). Note the colourbar.
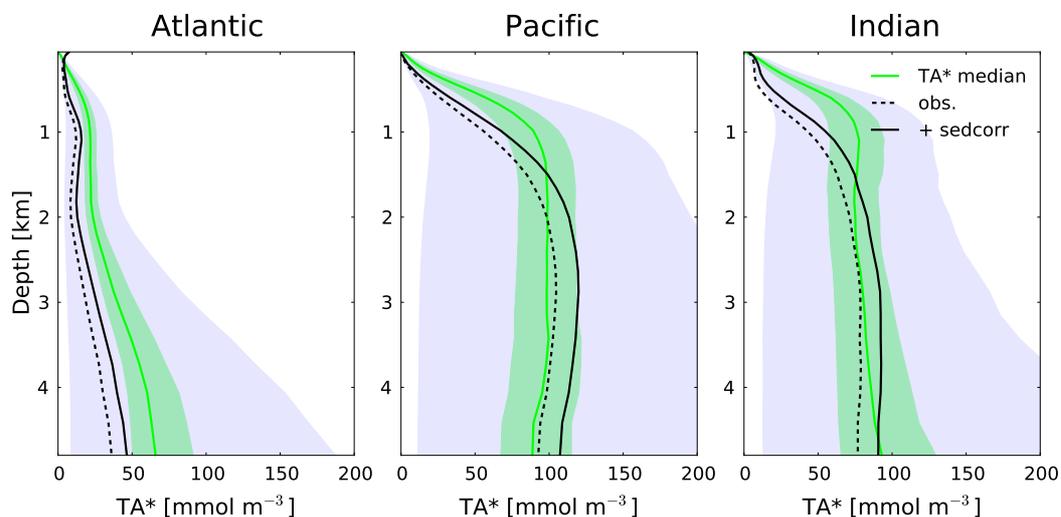
Figure 4. Model ensemble vs. observation-based basin-mean TA* profiles. The grey shading shows the unconstrained prior and the green shading shows the constrained (68 % confidence interval) distribution of the model ensemble. Lines represent the median of the constrained ensemble (green), observation-based TA* (black dashes), and observation-based TA* corrected for a sediment burial flux of 0.12 Gt-C yr^{-1} (black solid). The corresponding Southern Ocean sector is included in the averaging.
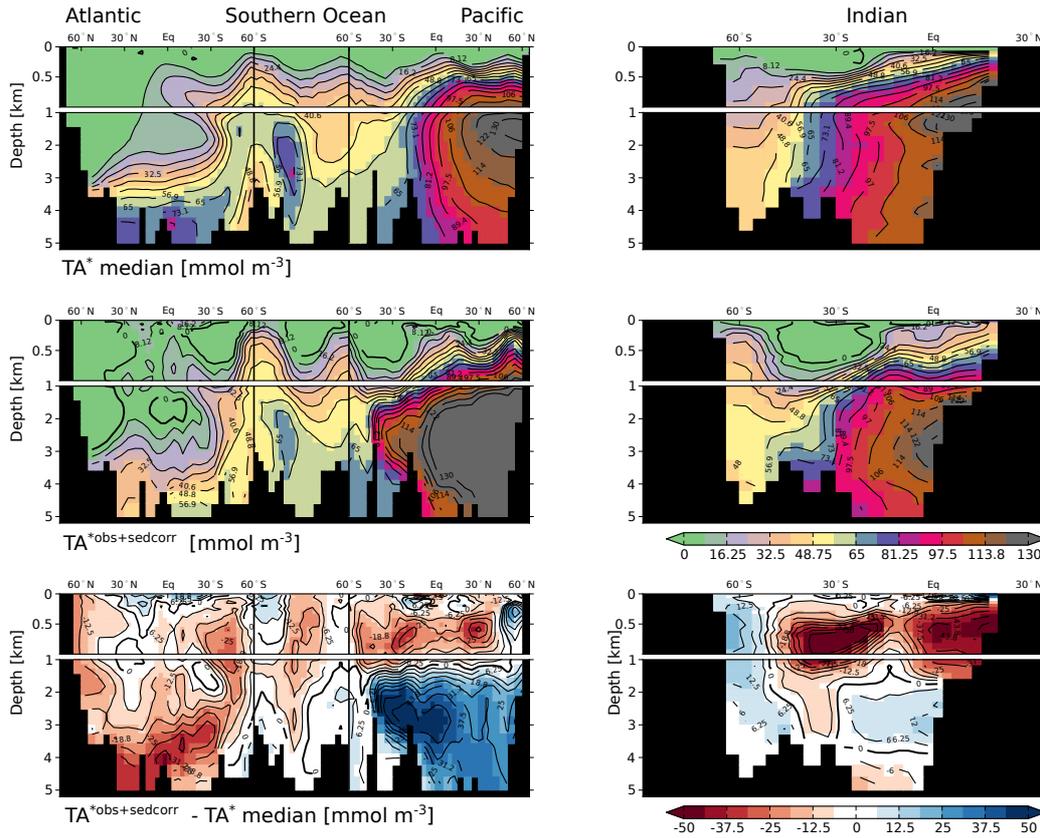
Figure 5. Observation-based vs. simulated TA*. Left column: modelled (median) TA*, observed TA*, and their difference in a cross-section through the Atlantic ($25°$ W), Southern Ocean ($58°$ S), and Pacific ($175°$ W). Right column: the same for a cross-section along $95°$ E in the Indian Ocean. The correlation coefficient, relative standard deviation ($\sigma_{\text{rel.}}^{\text{obs.}}$) and root mean square error (RMSE) between the simulated and observation-based fields are: 0.83, 1.09, and 17.7 mmol m^{-3} in the Atlantic; 0.88, 0.8, and 23.8 mmol m^{-3} in the Pacific; 0.87, 0.86, and 19.8 mmol m^{-3} in the Indian Ocean.
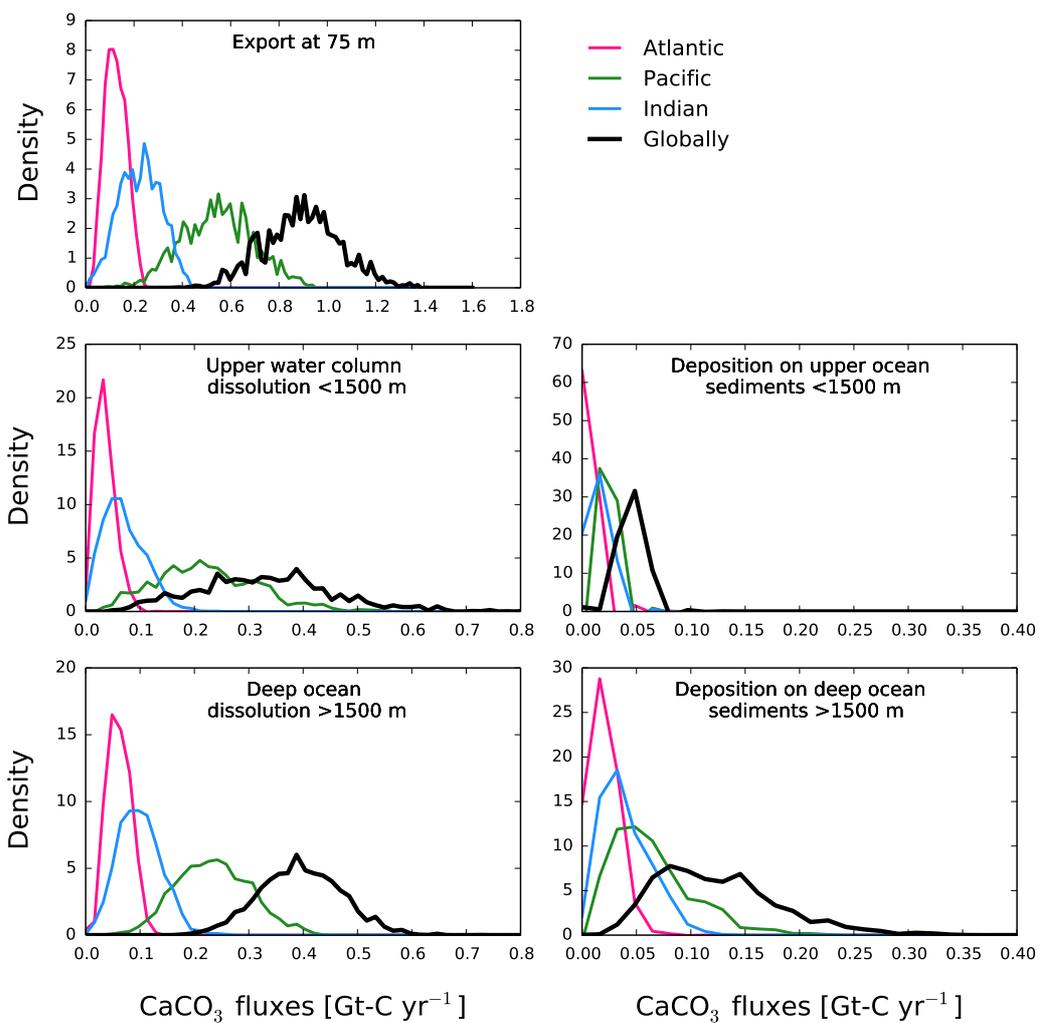
Figure 6. Probability density functions of basin-wide and global $CaCO_3$ fluxes as constrained by the observation-based TA* distribution (corrected for a sediment burial flux of $0.12\,\mathrm{Gt\text{-}C\,yr^{-1}}$). Note the different scaling of the x axes.

CaCO₃ Export [mmol-C m⁻² yr⁻¹]

Deposition [mmol-C m⁻² yr⁻¹]
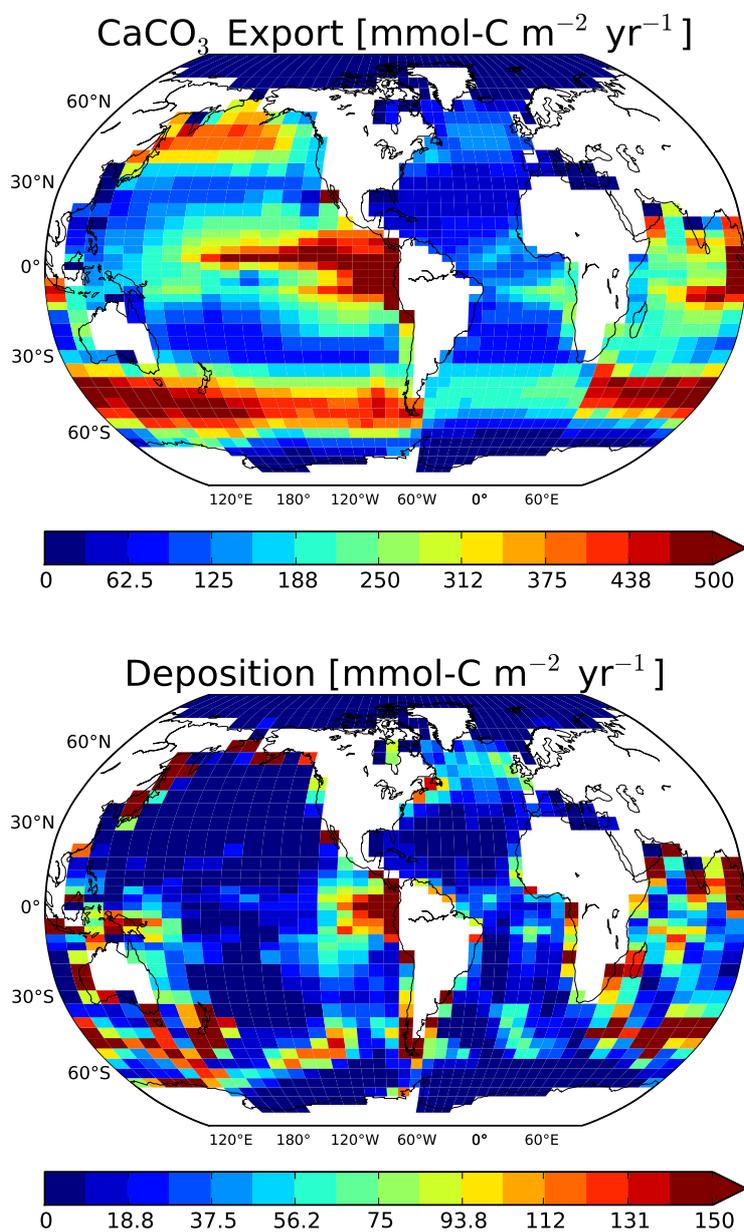
Figure 7. Top: median $CaCO_3$ export, globally 0.90 (0.72–1.05) $Gt\text{-}C\,yr^{-1}$. Bottom: median $CaCO_3$ fluxes reaching the ocean floor, globally 0.164 (0.113–0.227) $Gt\text{-}C\,yr^{-1}$. Note the different scaling of the colourbars.
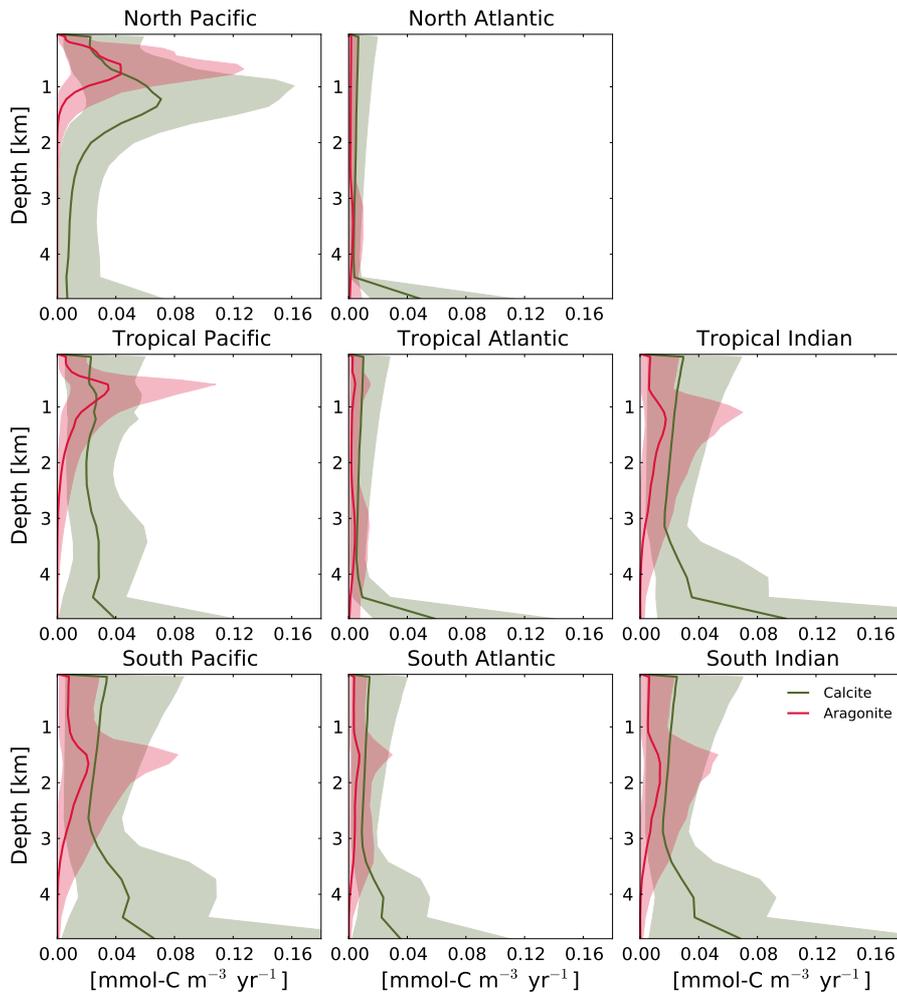
Figure 8. Constrained open-water dissolution rate profiles for aragonite (red) and calcite (olive). Ensemble medians (solid lines) and 68 % confidence intervals (shadings) are for spatial averages across individual regions and including only grid cells where the model water column extends to a depth of 5000 m.
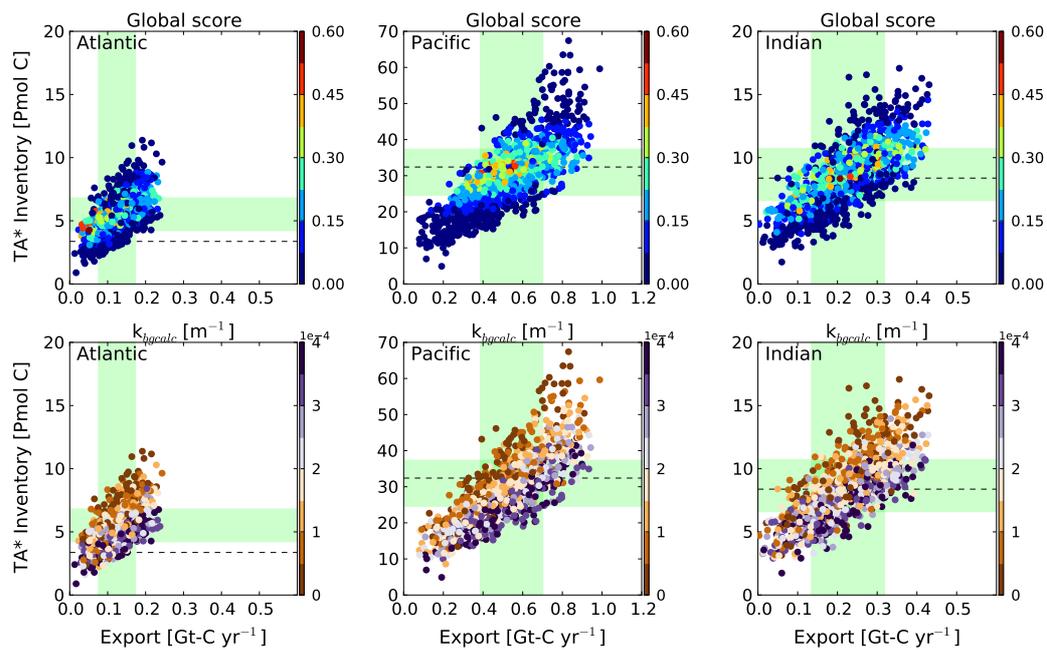
Figure 9. The simulated TA* inventory vs. the simulated $CaCO_3$ export of each ensemble member, coloured according to model skill (top) and background dissolution rate (bottom) for the Atlantic, Pacific, and Indian Ocean (columns). Each circle represents results from an individual simulation and its colour indicates the regional skill score (top) or the value of the background dissolution coefficient for calcite (bottom). Green shadings show the 68 % confidence range in basin-wide TA* inventories and $CaCO_3$ export of the constrained model ensemble and black-dashed lines indicate the estimated TA* inventories based on the observations including the sediment correction. Runs with low and high background dissolution can skillfully represent the TA* distribution. Runs with a higher $CaCO_3$ export tend to require a higher background dissolution to achieve a good skill.
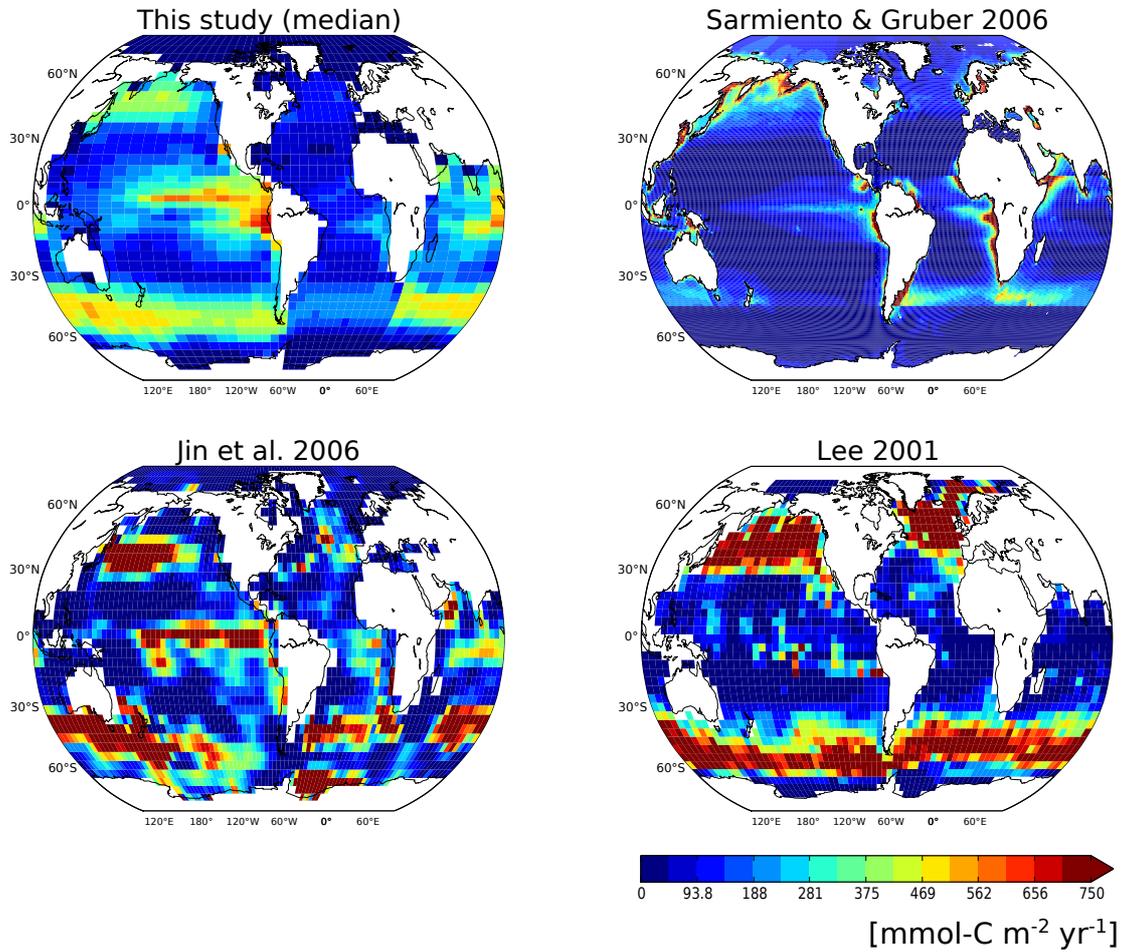
Figure 10. Observationally-constrained $CaCO_3$ export fields as estimated by different studies. Global total export is estimated to 0.5 by Sarmiento and Gruber (2006), to 1.14 by Jin et al. (2006), to 1.1 by Lee (2001), and to $0.90\,\mathrm{Gt\text{-}C\,yr^{-1}}$ by this study.
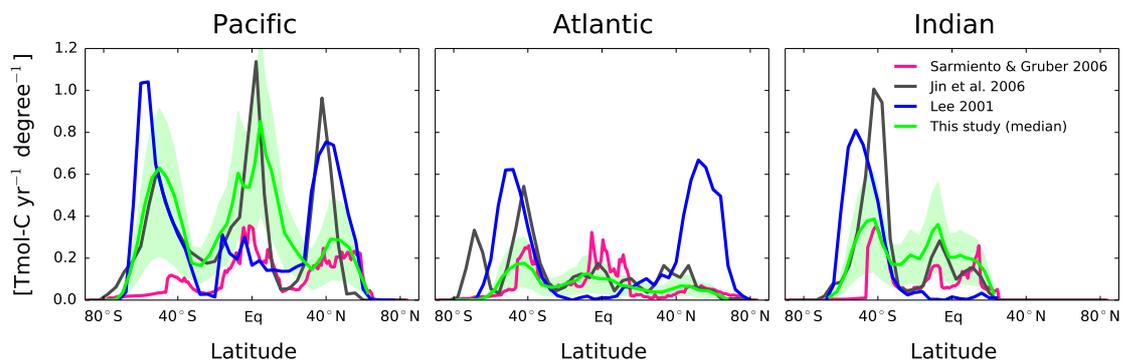

Figure 11. Zonally-integrated export fluxes of the constrained Bern3D-model ensemble (median and 68 % confidence interval) compared to estimates by Sarmiento and Gruber (2006), Jin et al. (2006) and Lee (2001).

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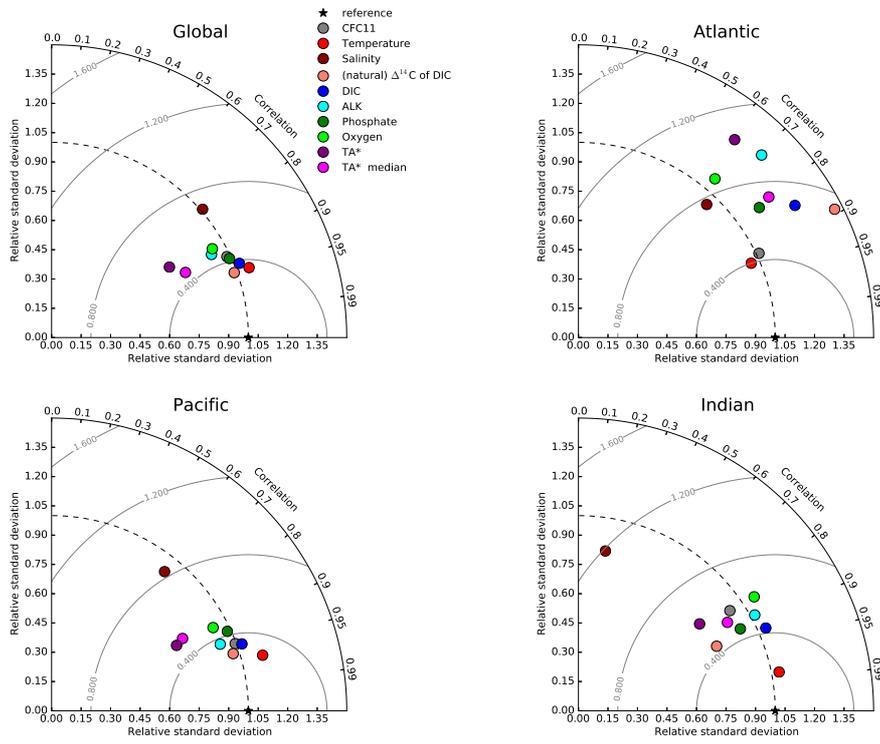
Figure A1. Taylor diagram (Taylor, 2001) for global and basin-wide volume-weighted oceanic tracer distributions as simulated by the Bern3D standard setup with $k_{\mathrm{dia}} = 0.2 \times 10^{-4} \, \mathrm{m^2 \, s^{-1}}$. Observation-derived fields are taken from GLODAP (Key et al., 2004) and the World Ocean Atlas 2009 (Locarnini et al., 2010; Antonov et al., 2010; Garcia et al., 2010a, b).
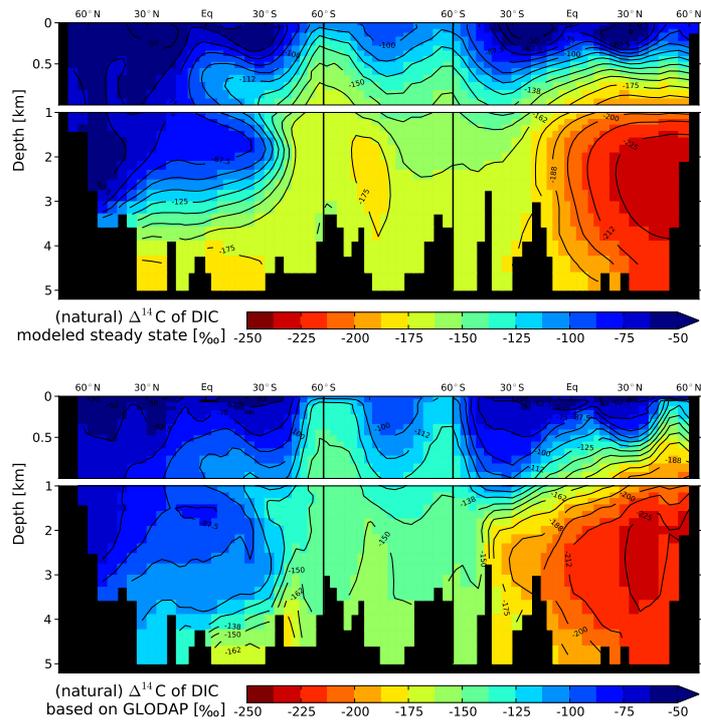
Figure A2. (Top) Distributions of (natural) Δ^{14}C of DIC [‰] as simulated with a standard k_{dia} of $0.2 \times 10^{-4}\,\mathrm{m}^2\,\mathrm{s}^{-1}$ at steady state and (bottom) as estimated from Δ^{14}C measurements (Key et al., 2004). The correlation coefficient, $\sigma_{\mathrm{rel.}}^{\mathrm{obs.}}$ and RMSEs are: 0.89,1.46, 23.77‰ in the Atlantic; 0.95, 0.97, 16.28‰ in the Pacific; 0.9, 0.78, 21.1‰ in the Indian Ocean. The section displayed is through the Atlantic (25° W), the Southern Ocean (58° S), and the Pacific (175° W).
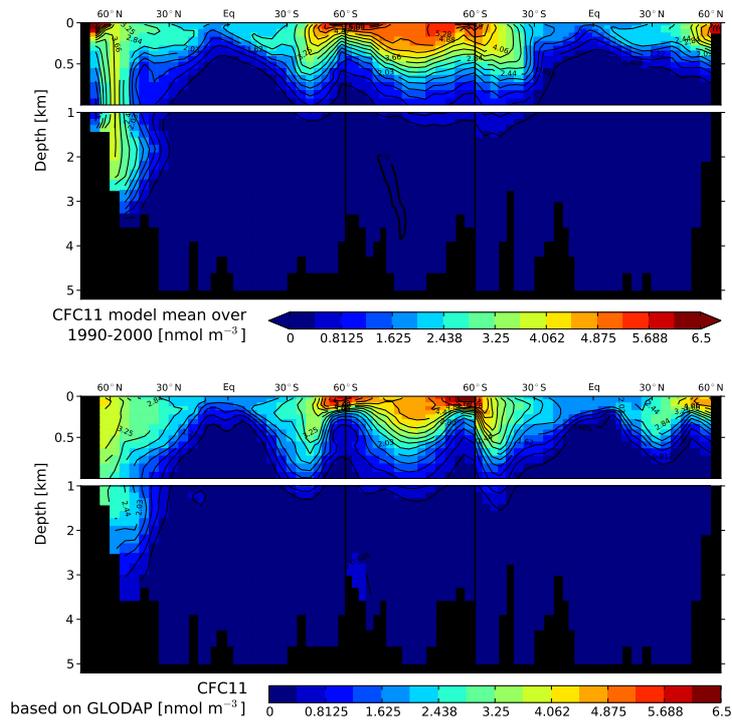
Figure A3. (Top) Distributions of CFC11 $[\mathrm{nmol\,m^{-3}}]$ as simulated with a standard k_{dia} of $0.2 \times 10^{-4}\,\mathrm{m^2\,s^{-1}}$ averaged over the model years 1990-2000 and (bottom) as observed (Key et al., 2004). The correlation coefficient, $\sigma_{\mathrm{rel.}}^{\mathrm{obs.}}$ and RMSEs are: 0.9, 1.01, 0.45 $\mathrm{nmol\,m^{-3}}$ in the Atlantic; 0.94, 1, 0.29 $\mathrm{nmol\,m^{-3}}$ in the Pacific; 0.83, 0.92, 0.54 $\mathrm{nmol\,m^{-3}}$ in the Indian Ocean. The section displayed is through the Atlantic ($25°\,\mathrm{W}$), the Southern Ocean ($58°\,\mathrm{S}$), and the Pacific ($175°\,\mathrm{W}$).
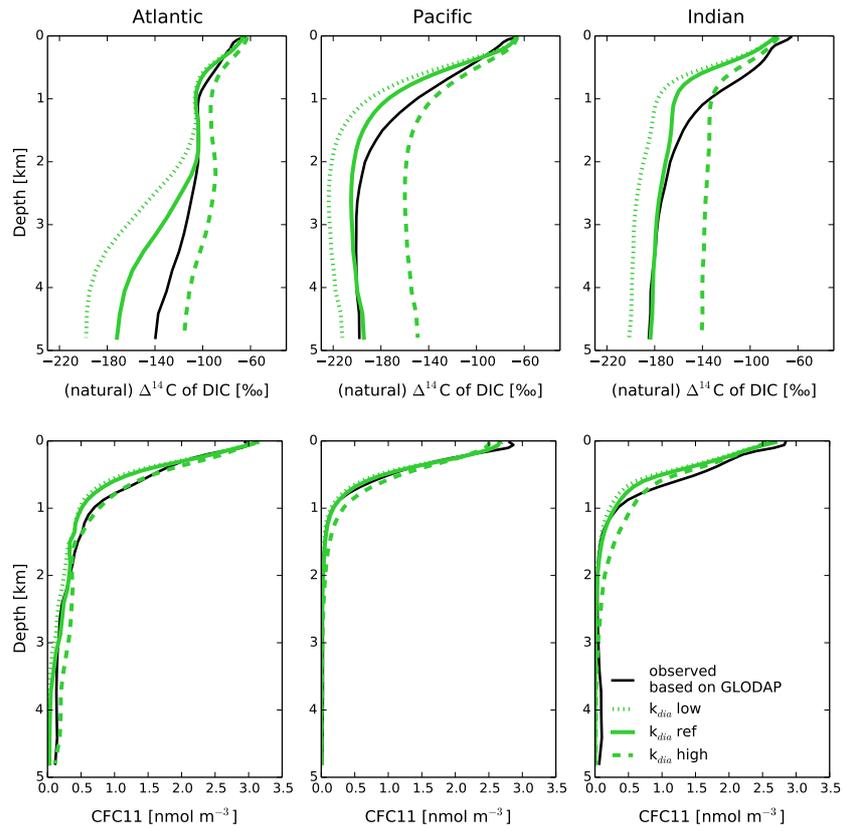
Figure A4. Basin-wide average profiles for (natural) Δ^{14}C of DIC [‰] at steady state (top) and for CFC11 [nmol m^{-3}] averaged over the model years 1990-2000 (bottom). Results are for simulations with a low (green dotted), standard (green solid), and high (green dashes) value of the diapycnal mixing coefficient k_{dia} (0.1, 0.2 and 0.5×10^{-4} m^2 s^{-1}). Observation-based estimates are in black. The corresponding Southern Ocean sector is included in the averaging.
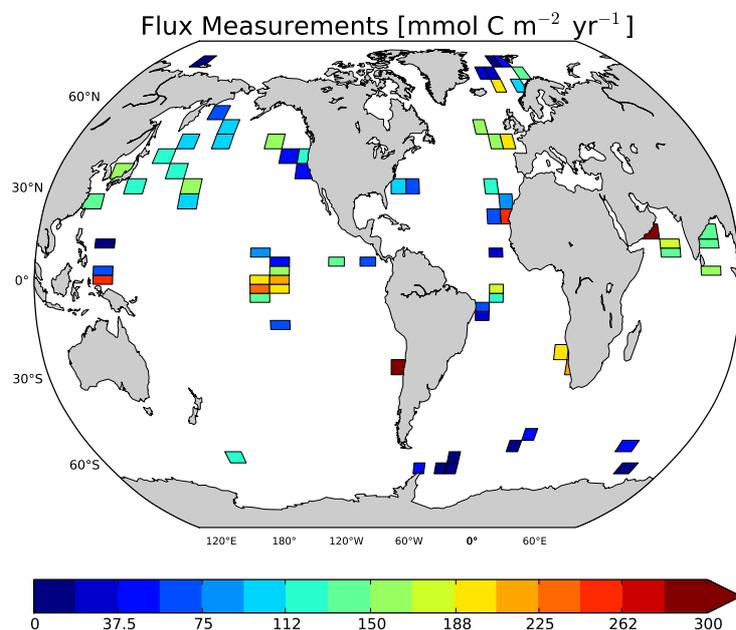
Figure A5. The global sediment trap data collection of Wilson et al. (2012, see their auxiliary material at doi:10.1029/2012GB004398) on the Bern3D grid. The measurements are located at different depths ($> 1500\,\mathrm{m}$). If more than one measurement was assigned to the same cell, the mean was chosen.
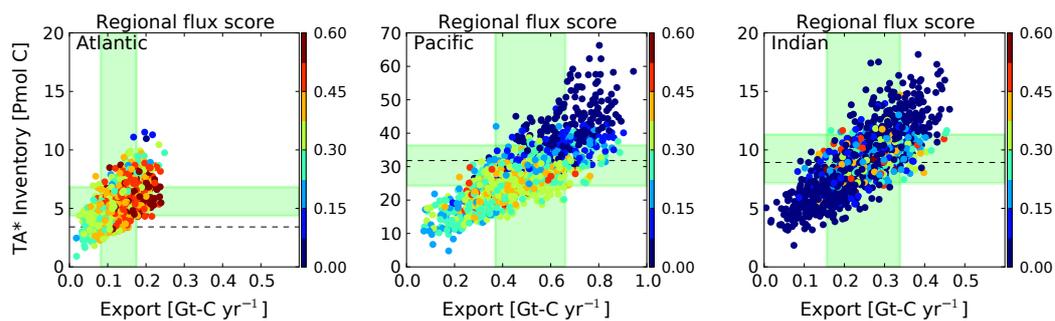
Figure A6. As Fig. 9 but coloured according to model skill with respect to the sediment trap database evaluated by basin. This target variable constrains export and dissolution to wider ranges (yellow to red colours) as compared to the TA* target (light green shading). Generally, high skill scores with respect to regional, sediment-corrected TA* are also associated with high skill scores with respect to regional fluxes. Only few models are in good agreement with the few flux measurements in the Indian Ocean.