

1 **Technical Note: Cost-efficient approaches to measure** 2 **carbon dioxide (CO₂) fluxes and concentrations in** 3 **terrestrial and aquatic environments using mini loggers**

4
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9 10 **Abstract**

11 Fluxes of CO₂ are important for our understanding of the global carbon cycle and greenhouse
12 gas balances. Several significant CO₂ fluxes in nature may still be neglected as illustrated by
13 recent findings of high CO₂ emissions from aquatic environments, previously not recognized
14 in global carbon balances. Therefore it is important to develop convenient and affordable
15 ways to measure CO₂ in many types of environments. At present, direct measurements of
16 CO₂ fluxes from soils or waters, or CO₂ concentrations in surface water, are typically labour
17 intensive or require costly equipment. We here present an approach with measurement units
18 based on small inexpensive CO₂ loggers, originally made for indoor air quality monitoring,
19 that were tested and adapted for field use. Measurements of soil-atmosphere and lake-
20 atmosphere fluxes, as well as of spatio-temporal dynamics of water CO₂ concentrations
21 (expressed as the equivalent partial pressure, $p\text{CO}_{2\text{aq}}$) in lakes and a stream network are
22 provided as examples. Results from all these examples indicate that this approach can provide
23 a cost- and labor efficient alternative for direct measurements and monitoring of CO₂ flux and
24 $p\text{CO}_{2\text{aq}}$ in terrestrial and aquatic environments.

25 26 **1 Introduction**

27 The carbon dioxide (CO₂) exchange across soil-atmosphere or water-atmosphere interfaces is
28 of fundamental importance for the global carbon cycle. Soil respiration returns substantial
29 amounts of the carbon fixed by plants to the atmosphere and contributes to the net ecosystem

1 exchange of carbon (Denman et al., 2007). Inland waters, including lakes, reservoirs and
2 rivers/streams are often showing a net emission of CO₂ from degradation or weathering
3 processes in surrounding soils, sediments and water columns (Aufdenkampe et al., 2011;
4 Battin et al., 2009). The inland water emissions has been estimated to 2.1 Pg yr⁻¹ (Raymond et
5 al., 2013) which is in the same order of magnitude as the estimated land carbon sink (2.6 Pg
6 yr⁻¹) (Denman et al., 2007).

7 Direct measurements of CO₂ fluxes across the soil-atmosphere and water-atmosphere
8 surface often rely on flux chamber (FC) measurements, representing a conceptually straight-
9 forward technique where the system in focus is covered by a chamber and the change in CO₂
10 over time in the chamber headspace is used to calculate the flux (Davidson et al., 2002).
11 Because of the heating inside soil chambers, and potentially rapid equilibration of the
12 chamber headspace for chambers on water, it is usually recommended to use short-term
13 deployments with repeated samplings during each deployment (e.g. sampling every 5th minute
14 for 30 minutes). For replicated and robust measurements it is also desired to perform repeated
15 deployments over extended periods. At the same time it is necessary to have multiple
16 measurement units to account for spatial variability. Therefore measurements accounting for
17 both spatial and temporal variability tend to be laborious if relying on manual sampling or
18 costly in terms of equipment if automated chamber systems are used.

19 Because direct flux measurements are time consuming, simpler alternatives have been
20 tried. For aquatic environments the CO₂ flux is often estimated from surface water
21 concentrations (usually expressed as equivalent partial pressure of CO₂ according to Henry's
22 Law; $p\text{CO}_{2\text{aq}}$) and the piston velocity (k) according to

$$23 \quad F = k \cdot K_{\text{H}} \cdot (p\text{CO}_{2\text{aq}} - p\text{CO}_{2\text{air}}) \quad (1)$$

24 where F is the flux between the water and the atmosphere (e.g. mol m⁻² d⁻¹), k is the piston
25 velocity (e.g. m d⁻¹; linked to the water turbulence and can be seen as the part of the water
26 column exchanging gas with the atmosphere per time unit), K_{H} is the Henry's Law constant
27 (e.g. mol m⁻³ atm⁻¹), and $p\text{CO}_{2\text{air}}$ is the partial pressure of CO₂ in the air above the water
28 surface ($p\text{CO}_{2\text{aq}}$ and $p\text{CO}_{2\text{air}}$ in units of atm) (Liss and Slater, 1974). Several ways to estimate
29 k from e.g. wind speed and various ways to measure water turbulence (for water bodies), or
30 slope (for running waters) have been used (Abril et al., 2009; Cole and Caraco, 1998; Gålfalk
31 et al., 2013; Raymond et al., 2013; Wallin et al., 2011), but although models may work well in
32 the systems where they were developed, extrapolations to other systems are uncertain (Borges

1 et al., 2004; Schilder et al., 2013; Wanninkhof, 1992). $p\text{CO}_{2\text{aq}}$ is typically either estimated
2 from pH and alkalinity or measured directly. The estimation of $p\text{CO}_{2\text{aq}}$ from pH and alkalinity
3 measurements is most common because of the large amounts of pH and alkalinity data
4 available from national monitoring (Raymond et al., 2013) but such indirect $p\text{CO}_{2\text{aq}}$ estimates
5 becomes unreliable at low alkalinity, at pH below 6, or at high levels of organic acids (e.g. in
6 humic waters) so direct measurements are desirable (Abril et al., 2015; Hunt et al., 2011).
7 Therefore direct measurements of fluxes and $p\text{CO}_{2\text{aq}}$ are needed to constrain the present
8 estimates of CO_2 fluxes (Abril et al., 2015). It should also be noted that $p\text{CO}_{2\text{aq}}$ is not solely
9 used for flux calculations - it a useful variable in itself for biogeochemical studies of aquatic
10 ecosystems, e.g. in assessments of ecosystem carbon metabolism.

11 The most common way to directly measure $p\text{CO}_{2\text{aq}}$ manually is by filling a large bottle
12 (1-2 L) completely with water, thereafter introducing a small headspace which is equilibrated
13 with the water by shaking, and then the headspace CO_2 concentration is measured (Cole et al.,
14 1994). Considering both indirect and direct approaches, there are presently data from
15 approximately 7900 water bodies and 6700 running water locations (Raymond et al., 2013).
16 However, these values typically represent snapshots in time for each system as monitoring of
17 temporal dynamics is demanding in terms of time or equipment. Daytime measurements
18 predominate in spite of expectations of higher $p\text{CO}_{2\text{aq}}$ during night when respiration
19 dominates over photosynthesis in many types of systems.

20 Due to the importance of CO_2 fluxes and concentrations, and the need to cover
21 temporal variability, a number of automated techniques have been developed. Apart from the
22 eddy covariance technique for large scale net fluxes, commercial automated flux chamber
23 systems to measure CO_2 flux from soil environments are available (e.g. www.li-cor.com). For
24 $p\text{CO}_{2\text{aq}}$, an increasing number of commercial systems have recently become available (e.g.
25 SAMI-CO₂, <http://sunburstsensors.com>, measures CO_2 indirectly via pH measurements in a
26 reagent solution; ProOceanus Mini-Pro CO_2 , <http://www.pro-oceanus.com>; Contros HydroC-
27 CO_2 , <http://www.contros.eu>). The costly components in those systems are typically the
28 instrumentation to measure and log CO_2 levels. For monitoring $p\text{CO}_{2\text{aq}}$ recent method
29 developments showed the possibility to have a near infrared CO_2 gas sensor (e.g. VAISALA
30 GMT220) under water by protecting it with a waterproof but gas permeable membrane
31 (Johnson et al., 2010). This technique is increasingly used and represents important progress,
32 while still being relatively expensive, accounting for both the CO_2 sensor and the separate

1 logger unit needed, and power consuming, requiring large and heavy batteries for long-term
2 remote use.

3 Recently flow-through equilibrators, has become increasingly used for $p\text{CO}_{2\text{aq}}$
4 measurements in various designs allowing remote or long term use (e.g. Abril et al., 2015;
5 Abril et al., 2006; Sutton et al., 2014). Water and air are pumped through the equilibrator
6 system and in some designs the gas is exchanged across a membrane surface while other
7 types of equilibrators are based on rapid direct gas exchange to an equilibrator headspace by
8 e.g. purging (Santos et al., 2012). A related approach is to pump air through gas permeable
9 tubing in the water (Hari et al., 2008). The air can be sampled by syringe or circulated through
10 an external infra-red gas analyzer.

11 A high cost of the measuring equipment means that only a few measurement units can
12 be afforded for simultaneous use, and thereby that information of spatial variability have to be
13 sacrificed. This is a severe limitation for constraining present estimates of CO_2 exchange
14 across land or water surfaces and the atmosphere. Low-cost equipment that can measure this
15 exchange over time at multiple well-constrained locations would be highly valuable. The aim
16 of this study was to test if low-cost CO_2 loggers developed for e.g. monitoring indoor air
17 quality and regulate ventilation in buildings, can also be used efficiently in environmental
18 research. These types of sensors typically do not have the same high performance and
19 sensitivity as the present commercial instruments for CO_2 measurements in environmental
20 science (e.g. by companies such as Los Gatos Research, Picarro, LI-COR, PP Systems, and
21 Quantek Instruments). However, if they are good enough for some environmental
22 applications, the lower cost, allowing for simultaneous deployment of a large number of
23 measurement units, would make such loggers highly beneficial.

24 We here present approaches to measure CO_2 fluxes and concentrations in nature using
25 small CO_2 logger that is positioned inside a chamber headspace. The cost of this type of CO_2
26 logger system is estimated to be <1-20 % of the alternative systems presently available and
27 used for environmental studies. Apart from testing logger performance under different
28 environmental conditions we provide examples of the following types of measurements:

- 29 – Fluxes between soil and atmosphere.
- 30 – Fluxes between lake surface water and the atmosphere.

- 1 – Measurements of surface water concentrations ($p\text{CO}_{2\text{aq}}$) by monitoring CO_2 in the
2 headspace of floating chambers in which the headspace CO_2 concentration was allowed
3 to be equilibrated with the water. This represents a new type of *in-situ* $p\text{CO}_{2\text{aq}}$
4 measurement supplementing the previous approaches having submerged sensors or
5 equilibrators, and where the issue of biofilm formation around submerged sensors is
6 avoided. These types of $p\text{CO}_{2\text{aq}}$ measurements were illustrated by measurements in a
7 lake and in a stream network.

8 We also provide detailed information on how to prepare loggers and on how to use them
9 under different conditions in the Supplement.

10

11 **2 The Material and methods**

12 **2.1 Logger description**

13 We used the ELG CO_2 logger made by SenseAir (www.senseair.se). It was chosen because of
14 promising specifications, including:

- 15 – CO_2 detection by non-dispersive infrared (NDIR) spectroscopy over a guaranteed range
16 of 0 - 5000 ppm (we discovered an actual linear range of 0 - 10 000 ppm; see below).
- 17 – Simultaneous logging of CO_2 , temperature, and relative humidity.
- 18 – Operating temperature range of 0 - 50 °C with temperature compensated CO_2 values.
- 19 – Full function at high humidity – from 0 - 99 % (non-condensing conditions).
- 20 – Includes an internal logger (5400 logging events), and adjustable measurement intervals
21 from 30 seconds to 0.5 years.
- 22 – Operated with 5.5 - 12 VDC (a small standard 9 V battery worked fine for extended
23 periods as long as the battery voltage is above 7.5 V) and has low power consumption
24 (depends on the measurement frequency, $\sim 250\mu\text{A}$ if 1 measurement/hour, $\sim 50\mu\text{A}$ in
25 sleep, $\sim 60\text{ mA}$ average during active measurement sequence ($\sim 12\text{s}$), see detailed
26 information at www.senseair.com).
- 27 – Quick and easy calibration by the user (see Supplement).

- 1 – Freely available user-friendly software for sensor control and data management (can be
2 downloaded at www.senseair.se).
 - 3 – Easily available documentation allowing supplementary modifications of the sensor for
4 field use.
 - 5 – Possibility to control one peripheral device connected to the logger (e.g. a pump).
- 6 More technical specifications and sensor documentation are available at the manufacturer's
7 web page (www.senseair.se).

8 **2.2 Sensor adaption for field use and initial calibration**

9 The loggers are sold as electrical board modules that are vulnerable to corrosion and do not
10 have suitable connectors for power supply, data communication, and calibration. Therefore
11 adaptations for field use had to be made. First, suitable connectors (power cable, data
12 communication cable, pins for calibration start/stop jumper, and pins for manual start/stop of
13 logging by jumper) were soldered onto the board. An UART data communication cable was
14 also made. Thereafter all parts of the board, except the connector pins, the temperature and
15 RH sensors and the CO₂ sensor membrane surface, were covered with several layers of
16 varnish for moisture protection. A detailed description on how to make all of this is available
17 in the Supplement.

18 The loggers were connected to power (individual 9V batteries for each logger) and
19 calibrated batch-wise in N₂ (representing zero CO₂ gas) by connecting the calibration pins
20 according to manufacturer instructions (zero calibration). Calibration is made repeatedly as
21 long as the jumpers are connected with improved results over time. Our typical procedure was
22 to run the zero calibration for approximately 3 hours. Alternative ways of calibration are also
23 possible as described in the Supplement, and were used when zero calibration was not
24 possible (e.g. in the field).

25 **2.3 Sensor performance tests**

26 Adequate sensor performance is a prerequisite for successful field use. Therefore we first
27 performed tests of calibration and linear measurement range (described below), and tests of
28 the influence of temperature and humidity on the measurements (explained in detail in the
29 Supplement).

1 **2.3.1 Test of calibration and linear measurement range**

2 After calibration, each sensor was tested by being set to log concentrations over time in a gas
3 tight box connected to a Los Gatos Research greenhouse gas analyzer (LGR; DLT-100) so
4 that the gas in the box with the batch of CO₂ loggers was continuously circulated through the
5 LGR instrument. CO₂ levels in the box were changed over time either by injection of standard
6 gases, or simply by breathing into the box to increase concentrations, or by putting an active
7 plant in the box to reduce CO₂ concentrations over time (by its photosynthesis). Thereby the
8 response of the loggers and the LGR to CO₂ levels ranging from 200 to 10 000 ppmv could be
9 compared.

10 **2.4 Field measurements**

11 Three types of field measurements were tried and are presented here as examples of how the
12 loggers can be used: (1) Flux measurements from soil, (2) flux measurements from water, and
13 (3) measurements of CO₂ concentration in water ($p\text{CO}_{2\text{aq}}$). The flux measurements were based
14 on monitoring of concentration changes over time with loggers placed in static flux chambers.
15 The $p\text{CO}_{2\text{aq}}$ measurements were also performed by measuring CO₂ concentrations inside a
16 chamber allowing the chamber headspace to reach equilibrium with the water, thereby making
17 headspace CO₂ concentrations reflect surface water concentrations according to Henry's Law.

18 For all these measurements the chambers used were made of plastic buckets (7.5 L
19 volume, 30 cm diameter) covered with reflective alumina tape to minimize internal heating.
20 This type of chamber has been shown to provide unbiased measurements of water–
21 atmosphere gas exchange (Cole et al., 2010; Gålfalk et al., 2013). The CO₂ loggers were
22 attached inside the chamber as shown in the Supplement (Figure S5). The battery was
23 protected by a gas tight plastic box. For the soil measurements the logger was left uncovered
24 in the chamber, but for measurements on water, protection against direct water splash as well
25 as condensation was needed. We tried the simplest possible approach by covering the sensor
26 with a plastic box having multiple 7 mm diameter holes drilled on one side to allow exchange
27 of air (see Figure S6). The air was forced to pass a plastic plate in the box before reaching the
28 logger to make some of the expected condensation occur on the plastic plate instead of on the
29 sensor itself. This way of protecting the sensor from condensation and splashing water could
30 potentially delay the response time if the air exchange between the chamber headspace and
31 the box is restricted, but a test described in the Supplement showed that this was not the case

1 in our type of measurements. The routines used for calibration and measurement validation,
2 including taking manual samples to check for potential sensor drift over time, are described in
3 the Supplement.

4 **2.4.1 Soil CO₂ flux measurements**

5 The soil flux measurements represented a simple test of logger suitability. The chambers were
6 put gently onto non-vegetated hardwood forest soil and the risk for extensive lateral gas
7 leakage was reduced by packing soil against the outer walls of the chamber. This procedure
8 does not correspond to common recommendations regarding soils chambers (e.g. having
9 preinstalled frames going into the soils) but shows if the loggers *per se* are suitable for soil
10 flux measurements regardless of what type of chamber is used. As traditional flux
11 measurements in soil chambers can be biased by the gas sampling (which can induce pressure
12 changes in the chamber disturbing the gas concentration gradients in the soil) (Davidson et al.,
13 2002), it is also favorable with a logger inside the chambers eliminating the need for gas
14 sampling during the flux measurement period. The headspace CO₂ concentrations were
15 logged over time at 2 minute intervals throughout measurement periods of 40 minutes. The
16 change in headspace CO₂ content over time was calculated by the common gas law
17 considering chamber volume and area, and represented the measured fluxes. In our tests new
18 measurement periods were started by simply lifting the chamber for a few minutes to vent the
19 headspace and then replacing the chamber on the soil.

20 **2.4.2 Aquatic CO₂ flux measurements**

21 For aquatic flux measurements, floating chambers were put on a small boreal forest
22 lake. In the examples presented here, CO₂ fluxes during morning and evening were measured
23 over 4 days. The logger unit was started indoors before going to the lake and measurements
24 were made every 6th minute throughout the whole 4-day period. Fluxes were calculated from
25 the change in CO₂ content over time in the chamber headspace. To start a new measurement
26 the chamber was lifted, vented for five minutes, and then replaced on the water. This venting
27 procedure was made morning and evening generating two flux estimates per day valid for the
28 period right after venting and restarting the measurements. After the 4-day period the
29 chambers were taken from the lake and data was downloaded from the logger when back in
30 the laboratory. We also performed additional flux measurements on a pond at the Linköping
31 University Campus using both data from the CO₂ logger inside a chamber, and from manual

1 samples taken by syringe from the same chamber which were analyzed by gas
2 chromatography. This comparison was made to verify that the change in headspace CO₂
3 content over time measured with loggers corresponded to traditional manual measurements.

4 **2.4.3 Surface water $p\text{CO}_{2\text{aq}}$ measurements**

5 Our $p\text{CO}_{2\text{aq}}$ measurements are based on the principle that after a floating chamber headspace
6 has equilibrated with the water, the measured partial pressure of CO₂ in the chamber
7 headspace will represent this surface water $p\text{CO}_{2\text{aq}}$. In this way $p\text{CO}_{2\text{aq}}$ can be measured in a
8 chamber headspace without any submerged sensors being in risk of damage from water
9 intrusions or resulting in bias from biofilms on the submerged sensor surface. On the other
10 hand the $p\text{CO}_{2\text{aq}}$ response in a chamber headspace will be delayed due to the equilibration
11 time which will depend on the piston velocity (k) and chamber dimensions. The response time
12 can potentially be shortened by mixing of the headspace or the surface water under the
13 chamber by installing fans or by pumping. We evaluated the effect of equilibration time
14 during a diel measurement cycle with and without fans and pumps (no notable effect
15 observed) and performed additional modeling accounting for a greater range of k -values and
16 testing effects of reducing the chamber volume to area ratio. A comparison between $p\text{CO}_{2\text{aq}}$
17 from instantaneous chamber headspace measurements and bottle headspace extractions were
18 also made. The details of the evaluation and comparison is presented in detail in the Results
19 and Discussion below and in the Supplement. Based on the outcome we here focused on
20 exploring the use of the $p\text{CO}_{2\text{aq}}$ chamber units further without any fans/pumps because we
21 wanted to first try the simplest and most power-efficient approach. As peripheral devices can
22 conveniently be connected and controlled by the loggers, addition of fans or pumps is
23 practically easy to explore further in cases when needed based on specific research questions.
24 In general the tests and examples provided here represent a start and we expect that future
25 users will develop additional ways to use the loggers presented.

26 We made environmental $p\text{CO}_{2\text{aq}}$ measurements in several ways including:

- 27 (a) Test of spatio-temporal variability in a large shallow lake (Tämnaren, Uppsala,
28 Sweden). Here seven units were deployed for approximately 2 days with a logging
29 interval of 5 min, near the North and South shores and at the center of the lake,
30 respectively (Fig. 1).

- 1 (b) Test of a 20 day deployment with a 1 h logging interval at a small shallow boreal lake
2 (in the Skogaryd Researach Catchment, Vänersborg, Sweden).
- 3 (c) Test of measuring stream $p\text{CO}_{2\text{aq}}$ at 14 locations in a stream network (Skogaryd,
4 Vänersborg, Sweden) over a 24 h period with a logging interval of 1 min.

5

6 **3 Results and discussion**

7 **3.1 Test of calibration, linear response range, and influence of temperature** 8 **and humidity**

9 The results of the sensors were always well correlated with LGR results (Fig. 2). Above 7000
10 ppmv the LGR response started to become non-linear but the CO_2 loggers kept a linear
11 response up to 10 000 ppmv (confirmed also by additional analyses using gas
12 chromatography). The combined influence of temperature and humidity was found to be
13 small, causing an error $< 7.6\%$ (see Supplement). Logger drift over time was not notable in
14 the tests and examples provided here, but is expected during long-term use (the manufacturer
15 estimate a drift of 50 ppmv per year under indoor conditions). It is therefore recommended to
16 collect occasional manual samples for drift check and correction (see Supplement) and to
17 recalibrate the loggers frequently.

18 **3.2 Flux measurements**

19 Examples of results from the flux measurements are shown in Fig. 3. Clear and consistent
20 linear responses of CO_2 concentrations over time in the chambers, being suitable for
21 calculation of fluxes, were collected with very limited effort in both terrestrial and aquatic
22 environments. The work primarily consisted of starting the units, deploying chambers,
23 flushing the chamber headspace at desired time intervals to restart measurements, and
24 downloading the data. The calculation of the flux is based on the slope of the CO_2 change in
25 the chamber headspace during the deployment. Thus, a flux measurement is based on a
26 relative CO_2 change which is not sensitive to moderate drift or to exact absolute values.
27 Nevertheless, as a part of our general measurement routines, occasional manual measurements
28 were taken before flushing the chamber for sensor validation and drift correction (no drift
29 correction was needed for any data presented in this study).

1 The approach to place a CO₂ logger inside each chamber leads to several new
2 advantages for flux measurements including:

- 3 (1) It allows chambers to be individual units that can be distributed much more widely than
4 a system where the chambers are connected by tubing to one single external analyzer.
5 This is important for capturing spatial variability and not being restricted to a limited
6 area around a gas analyzer.
- 7 (2) Substantial time is saved by eliminating the need for manual sampling and subsequent
8 sample handling and analyses. This allows much more time to be spent on better
9 coverage of spatial or temporal variability in the fluxes or on accessory measurements.

10 The low cost of each flux chamber unit together with the time saving per unit adds substantial
11 value even for short term, non-automated flux measurement efforts. The same work effort
12 normally needed for manual flux measurements (including not only sampling but also sample
13 preservation and manual sample analyses) with one chamber could now yield flux
14 measurements from more than 10 chambers with logger units inside.

15 The fluxes obtained for the soils were 2534-2954 mg C m⁻² d⁻¹ (Fig. 3a), which
16 corresponds well with the previous range found for soil fluxes in corresponding environments
17 (Raich and Schlesinger, 1992). The lake fluxes measured were 216-666 and 364-427 mg C m⁻²
18 d⁻¹ (Fig. 3b and 3c, respectively), which also is well within the range previously found in
19 aquatic ecosystems (Selvam et al., 2014; Trolle et al., 2012). The flux data from the logger
20 inside the chamber were nearly identical with data from manual sampling and gas
21 chromatography analysis (Fig. 3c). Thus, given their low price and suitable sensitivity, these
22 chamber-logger units seem highly useful in most types of flux chamber measurements and
23 have the potential to substantially increase the data generation per work effort.

24 **3.3 pCO_{2aq} measurements**

25 The pCO_{2aq} values in all the examples were in the expected range of 200 to >10 000 found in
26 various types of waters (Abril et al., 2014; Marotta et al., 2009; Raymond et al., 2013; Selvam
27 et al., 2014). The most common traditional methods to measure pCO_{2aq} are the alkalinity-pH
28 method and the bottle headspace equilibration technique (the latter from here on called the
29 bottle method). The superiority of the bottle method compared to the alkalinity-pH method
30 has already been thoroughly addressed (Abril et al., 2015). Therefore we here focus on
31 comparing the bottle and the pCO_{2aq} chamber (i.e. chamber equilibrator) approaches.

1 The principle behind the $p\text{CO}_{2\text{aq}}$ chamber approach is exactly the same as the principle for the
2 bottle method and constitute the fundamental principle behind Henry's Law, e.g. that gas
3 exchange between a confined gaseous headspace and a connected water volume will
4 eventually approach an equilibrium at which the headspace concentration or partial pressure
5 corresponds with the concentration in the water near the water-headspace interface. So in
6 essence the methods are similar. There are however at least three reasons to believe that
7 instantaneous $p\text{CO}_{2\text{aq}}$ measurements from the common bottle headspace extraction and our
8 $p\text{CO}_{2\text{aq}}$ chamber technique are not always identical:

- 9 (1) The headspace to water volume ratio affects the measurements as the CO_2 transferred to
10 the headspace could reduce the amount of CO_2 left in the water if the water volume is too
11 small, resulting in underestimated $p\text{CO}_{2\text{aq}}$ values. This can bias the bottle values
12 depending on the headspace and water volumes and this is why it is often recommended
13 to use a large bottle (1-2 L) and a small headspace (25-50 ml) in the bottle method. Even
14 if following this recommendation, the headspace to water volume ratio is much smaller
15 for the $p\text{CO}_{2\text{aq}}$ chamber approach (e.g. a few L of headspace versus many m^3 or even
16 large parts of the mixed water layer of a lake) which should therefore be more accurate in
17 this regard. Fortunately, the bottle method bias is in most cases small (about 5 % for a 20
18 °C scenario with a 1 L bottle, a 50 ml headspace, and no available bicarbonate that can
19 buffer the loss of CO_2 to the headspace) and can be corrected for but it is not always clear
20 if such corrections are made.
- 21 (2) For the bottle approach, the transfer of water into large bottles without risk of losing
22 volatile solutes is not trivial. Water pumping and transfer from water samplers can cause
23 degassing. Hence the water sampling can result in loss of CO_2 causing underestimation of
24 the real $p\text{CO}_{2\text{aq}}$. In the $p\text{CO}_{2\text{aq}}$ chamber approach, there is no water sampling and the risk
25 of water sampling bias is therefore removed.
- 26 (3) Another reason that numbers may not be identical is the potential delayed response of the
27 $p\text{CO}_{2\text{aq}}$ in the chamber while the bottle approach gives a snapshot value valid for the
28 sampled water volume. This delay differs depending on the piston velocity (k ; see Fig. 4)
29 and means that day time CO_2 values in the $p\text{CO}_{2\text{aq}}$ chambers may be influenced by the
30 higher $p\text{CO}_{2\text{aq}}$ from the previous night, thereby overestimating the instantaneous day-time
31 $p\text{CO}_{2\text{aq}}$. Accordingly, night time CO_2 values in the chamber may underestimate the
32 instantaneous night $p\text{CO}_{2\text{aq}}$ by influence from lower daytime $p\text{CO}_{2\text{aq}}$.

1 Essentially, all the three points above show that single $p\text{CO}_{2\text{aq}}$ chamber measurements,
2 representing a longer time period, are not directly comparable with instantaneous bottle
3 values, and makes it likely that chamber $p\text{CO}_{2\text{aq}}$ values measured during day time should be
4 slightly higher than corresponding bottle $p\text{CO}_{2\text{aq}}$ measurements. This is also what we find
5 when comparing single daytime $p\text{CO}_{2\text{aq}}$ samples from chambers and bottles (Fig. 5). The
6 difference seems to increase with $p\text{CO}_{2\text{aq}}$ levels which is what would be expected if the bias is
7 caused by loss from sampling (point 2 above) or by a strong diel cycling (point 3 above).

8 We find that while the principles behind both the bottle and the chamber approach are
9 robust, there may be a delayed response of the $p\text{CO}_{2\text{aq}}$ chamber depending on k (Fig. 4). Thus
10 single snapshot measurements from the chambers during daytime can be overestimated (see
11 Figure 5). However, the daily averages from the $p\text{CO}_{2\text{aq}}$ chambers were representative under a
12 wide range of k scenarios (in Fig. 4 the mean daily $p\text{CO}_{2\text{aq}}$ chamber values were on an
13 average 97% of the real values; range 92-99 %). There is also potential to speed up the
14 temporal response of the $p\text{CO}_{2\text{aq}}$ chambers by changing the chamber design (decreasing the
15 volume and increasing the area; see also Fig. 6). Another way to speed up the response time
16 would be to let the logger control a pump that draws air from the logger box and releases it
17 just below the water surface under the chamber, resulting in surface water purging favouring
18 rapid equilibration. This adaption could easily be made but requires a larger battery for long-
19 term use.

20 The time of initial equilibration after deployment may be long at low k values (Figure
21 6). For example, in a water body at wind speeds below 0.6 m s^{-1} (corresponding to k values
22 lower than 0.5 m d^{-1} using one common wind speed- k model; Cole and Caraco, 1998) the
23 equilibration time is > 10 hours given the volume to area ratio of our chambers (Fig. 6). As
24 stated above, this limits the use of the chamber $p\text{CO}_{2\text{aq}}$ approach for diel variability,
25 particularly during the first period after deployment. The delay in the chamber response when
26 being near equilibrium levels is much shorter also at k values, making diel variability possible
27 to distinguish although with a delay and hampered amplitude requiring careful consideration
28 (Fig. 4).

29 The measurements from chambers with equilibrated headspace revealed large spatial
30 differences in $p\text{CO}_{2\text{aq}}$ with synchronous temporal variability on the big lake (Fig. 7). Data
31 from a long-term deployment (20 days) showed a consistent diel pattern with increasing
32 $p\text{CO}_{2\text{aq}}$ during night and decreasing levels during the day as expected (Fig. 4 and above

1 discussion). The long-term tests showed that our passive approach with a protective box to
2 avoid condensation in the logger measurement cell worked well for 1-2 weeks. Over time
3 moisture seemed to accumulate in the sensor protection box and consequently unrealistic high
4 peaks caused by water condensation inside the measurement cell, often reaching the
5 maximum value (10 000 ppm; Fig. 8a), were noted more frequently with time. This effect
6 disappeared once conditions in the chamber favored drying of the sensor and the sensors
7 survived occasional condensation with maintained performance. The occurrence of
8 condensation events increased with increasing temperature difference between day and
9 nighttime temperatures and therefore the condensation events were more common on the
10 sunlit lake surfaces than on waters in the shadow (e.g. the streams described below). To
11 remove the condensation data peaks we adopted a simple data filtering routine that removed
12 data points that were more than 10% higher than the ± 4 hour median relative to the data point
13 (Fig. 8b). This filtering procedure to remove data influenced by condensation becomes
14 inefficient if condensation events are too frequent. We therefore suggest to routinely drying
15 the logger indoors overnight every 7-14 days (depending on the local conditions) of
16 deployment. Given the low price, the loggers can simply be replaced with a separate set of dry
17 units to avoid losing data while the loggers are drying. For longer deployments where weekly
18 or biweekly visits are not possible, more advanced measures to prevent condensation should
19 be considered. Potentially, silica gel in the sensor protection box could delay extensive
20 influence of condensation events. As the loggers can control one peripheral unit it would also
21 be possible to equip the system with a larger battery and a pump that draws air to the sensor
22 through a desiccant removing water vapor. Another potential alternative to prevent
23 condensation is to heat the measurement cell a few degrees above the surrounding air if there
24 is enough power.

25 The logger units were also found highly suitable for logging $p\text{CO}_{2\text{aq}}$ in streams (Fig.
26 9). By tethering the units on the streams, equilibrium time is reduced by the turbulence
27 induced around the chamber edges. (While this is a problem for stream flux measurements, it
28 is beneficial for $p\text{CO}_{2\text{aq}}$ measurements with our approach.) Further, the low price of our units
29 allows the use of a greater number of units compared to other approaches, which is an
30 advantage for monitoring $p\text{CO}_{2\text{aq}}$ at multiple points in e.g. a stream network for doing CO_2
31 mass balances and for studying the regulation of $p\text{CO}_{2\text{aq}}$ over large scales. Fig. 9 provides an
32 example where 14 units were used simultaneously in a stream network and where spatio-
33 temporal variability over 24 h revealed (1) significant spatial differences between locations in

1 the catchment, providing indications of different CO₂ export from soils and also of local hot
2 spots for CO₂ emissions, and (2) how a rain event and an associated change in discharge
3 influenced the temporal dynamics of $p\text{CO}_{2\text{aq}}$.

4 **4 Conclusions**

5 We conclude that the approach to measure and log CO₂ fluxes and $p\text{CO}_{2\text{aq}}$ presented here can
6 be an important supplement to previously presented approaches. When focusing on high
7 temporal resolution of $p\text{CO}_{2\text{aq}}$ (response time of minutes), the previous approaches with
8 submersible sensors (e.g. (Johnson et al., 2010) or rapid equilibrators systems connected to
9 CO₂ analyzers (e.g. Abril et al., 2006; Frankignoulle et al., 2001) are probably preferred. In
10 such cases, the Senseair CO₂ logger may be suitable for use together with equilibrator
11 systems. The chamber approach described here provides a cost- and labor-efficient multi-
12 measurement point alternative for (i) easy flux measurements and (ii) $p\text{CO}_{2\text{aq}}$ measurements
13 which are not biased by potential biofilms on submersed equipment, and where delayed
14 response times for $p\text{CO}_{2\text{aq}}$ are acceptable (the delay is shorter at higher turbulence/piston
15 velocity and can be estimated from the data obtained from the initial part of the deployment
16 showing how quickly water-headspace equilibrium is reached).

17 While well constrained CO₂ fluxes are critical for the global carbon balance, the previous
18 estimates are uncertain in terms of spatio-temporal variability and flux regulation. For aquatic
19 environments CO₂ fluxes are often based on indirect measurements demonstrated to
20 frequently be highly biased (Abril et al., 2015). Hence there is a need to rapidly improve the
21 situation and increase the global availability of high quality data based on direct CO₂
22 measurements. We believe the presented measurement approaches with small logger units are
23 affordable, efficient, user friendly, and suitable for widespread use – thereby having potential
24 to be important tools in future CO₂ studies.

25 **Associated content**

26 Supplementary material including a manual on how to build and use the described CO₂ logger
27 units, details about some of our tests, and advice on the practical use of the loggers are
28 available.

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9

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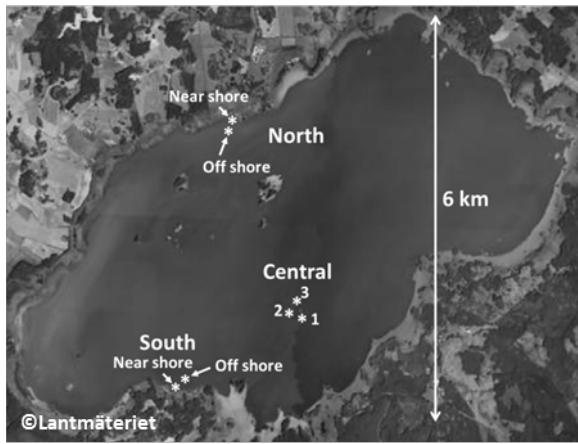
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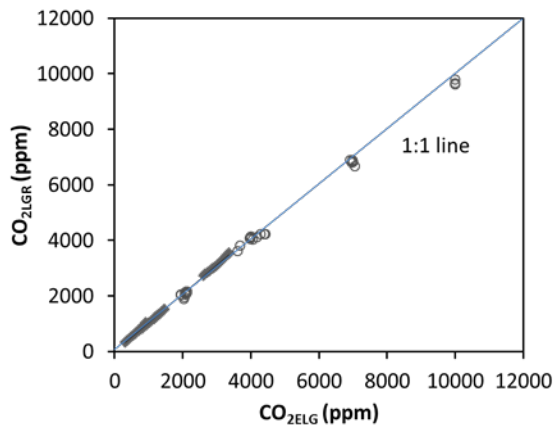
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 2 Figure 1. Map indicating the locations of the chambers on the lake Tännaren. The map is
 3 published with permission from Lantmäteriet, Sweden according to agreement i2012/898 with
 4 Linköping University.

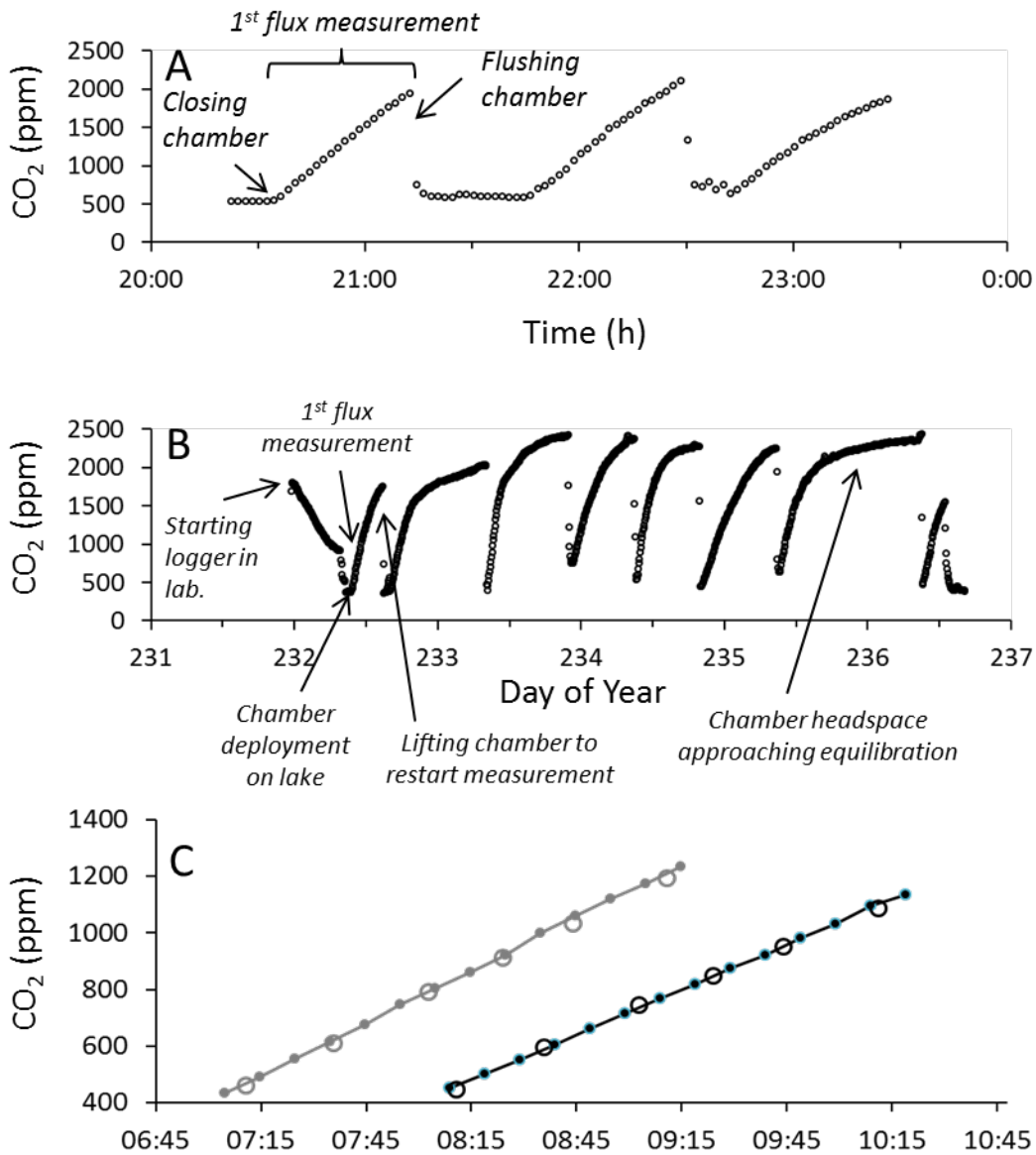
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6
 7 Figure 2. Comparison of CO₂ mixing ratio (ppm) measured with a Los Gatos Research
 8 greenhouse gas analyzer (LGR; DLT100) and the CO₂ logger by Senseair (ELG).
 9 Measurements were made with ELG loggers from two different batches at two separate
 10 occasions (diamonds forming bold lines and circles, respectively). The ELG have a maximum
 11 limit at 10 000 ppm in its present configuration. The LGR is affected by saturation/quenching
 12 effects in the measurement cell starting at 6000 ppm explaining the slight offset compared to
 13 the 1:1 line.

14

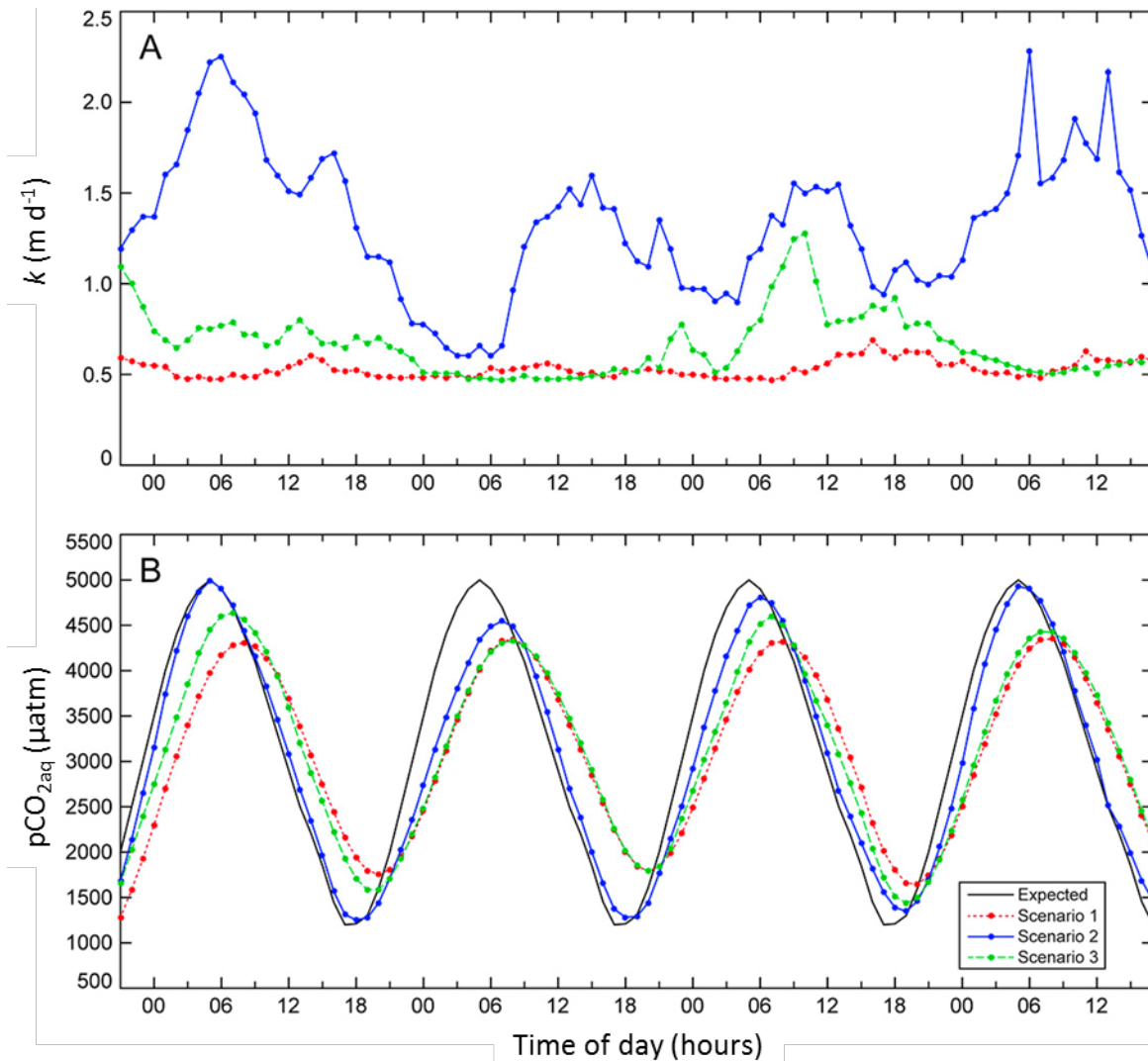
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1
 2 Figure 3. Examples of CO₂ measurements by loggers inside flux chambers. Panel (a) shows
 3 changes in CO₂ concentration with time inside a chamber (used to calculate fluxes) due to
 4 soil CO₂ efflux in three repeated experiments. Panel (b) shows logger raw data from eight
 5 repeated measurements on a small wind sheltered boreal lake using a floating chamber. The
 6 different work steps in this example are indicated in the figure. In this example chamber
 7 deployments were restarted manually at low temporal frequency due to additional parallel
 8 field work and depending on priorities such measurements can be made at much higher
 9 frequency. The CO₂ logger can also be used in automatic chambers (Duc et al., 2013). Panel
 10 (c) shows a comparison between data from CO₂ loggers inside two floating chambers on a
 11 pond (solid lines with dots) and manual samples taken from the same chambers and analyzed

1 by gas chromatography (circles). Gray and black symbols denote the two different
2 measurements.

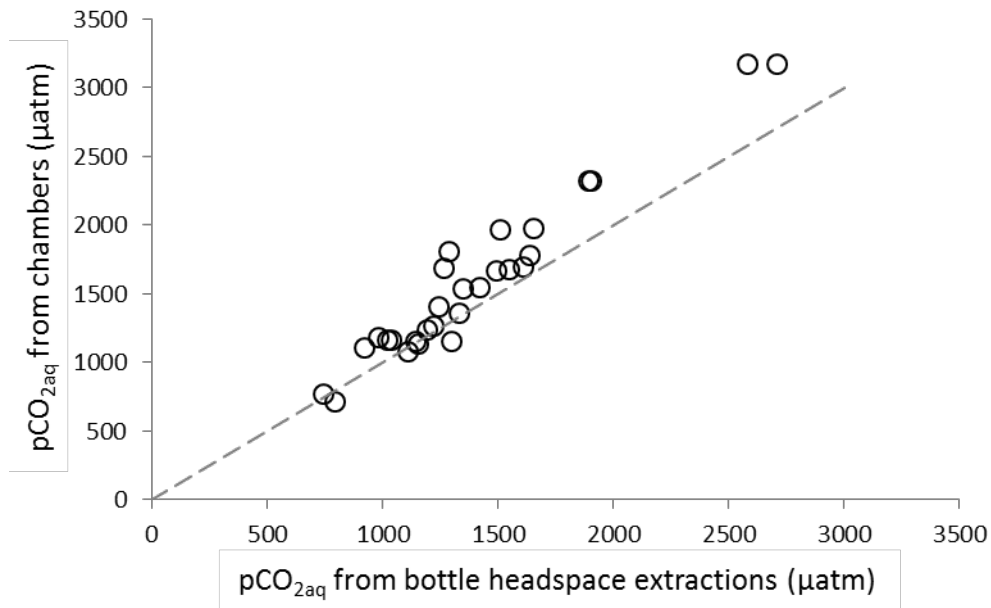
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5 Figure 4. Example where k values (piston velocity; see text) were calculated from wind speed
6 according to (Cole and Caraco, 1998) for three real scenarios with different diel variability
7 (Panel A), and then used to model the diel pattern in $p\text{CO}_{2\text{aq}}$ chambers of the type we used
8 compared to the expected cases based on instantaneous $p\text{CO}_{2\text{aq}}$ levels (Panel B). The expected
9 case is fictive but inspired by levels found for a pond with large diel variability (Natchimuthu
10 et al., 2014).

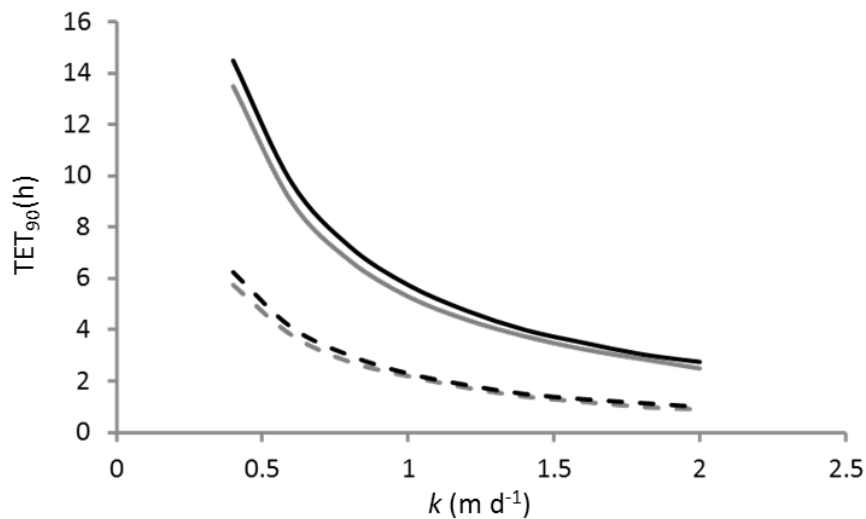
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2 Figure 5. Comparison between instantaneous day-time measurements from $p\text{CO}_{2\text{aq}}$ chambers
 3 (allowed to reach equilibrium) and traditional bottle headspace extractions (1025 ml total
 4 volume, 50 ml headspace, not corrected for the enclosing a limited amount of inorganic
 5 carbon in the bottle; see text). R^2 for a linear regression is 0.94. The dashed line is the 1:1 line
 6 (see above text for discussion of the deviation from this line).

7



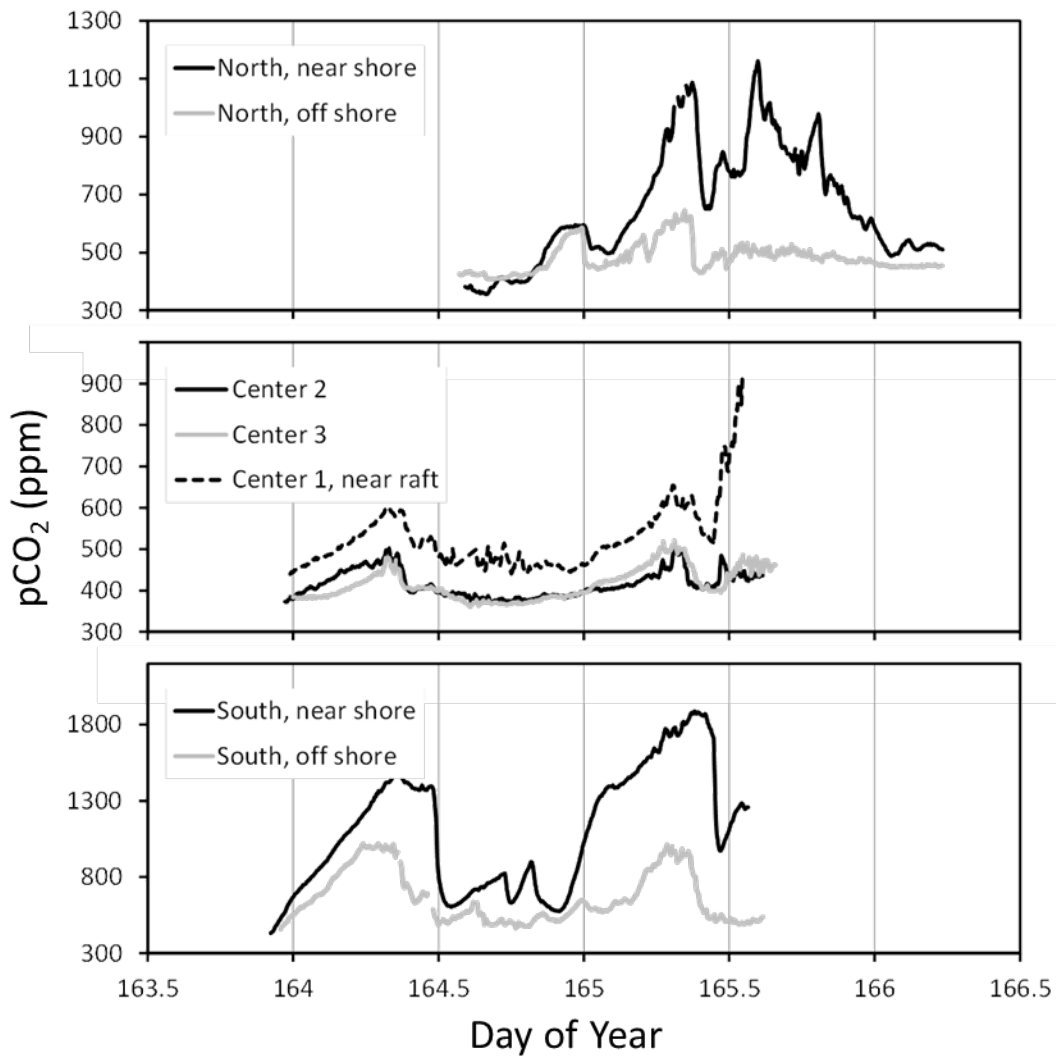
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2 Figure 6. Theoretical equilibration time to within 90% (TET_{90}) of the true pCO_{2aq} after
 3 deploying the described chambers (solid lines) at different piston velocities (k), a temperature
 4 of 20 °C, and a pCO_{2aq} of 2000 μatm (grey) or 8000 μatm (black). The dashed lines show
 5 TET_{90} for chambers with a two times higher area to volume ratio compared to the chambers
 6 we used. Another way to speed up equilibration time is by mixing the water below the
 7 chambers (see text above).

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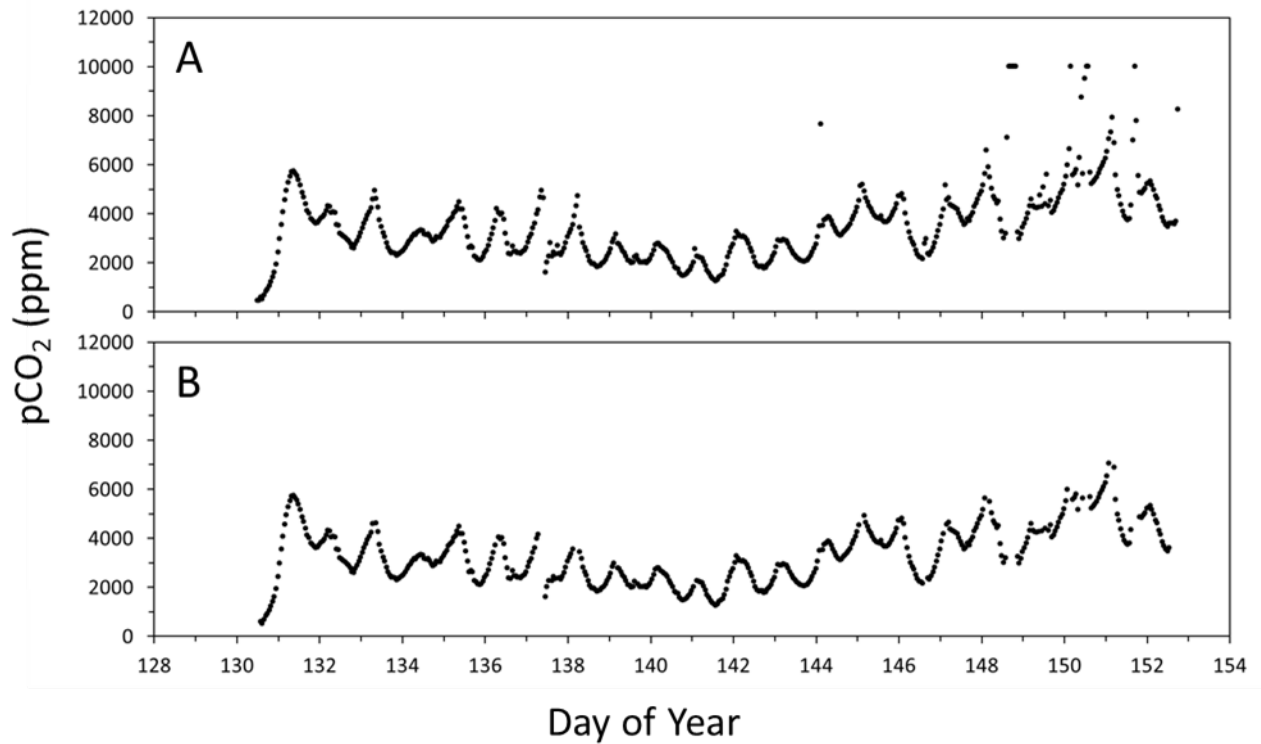


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2 Figure 7. Illustration of spatial variability of $p\text{CO}_{2\text{aq}}$ (expressed as mixing ratio – ppm) in a
 3 large shallow (mean depth 2 m) lake revealed by seven CO_2 logger-chamber units. The
 4 locations of each chamber are indicated in Figure 1. See text for details. Note different y axis
 5 scales and that this lake was wind exposed with variable wind conditions during the
 6 measurement period.

7

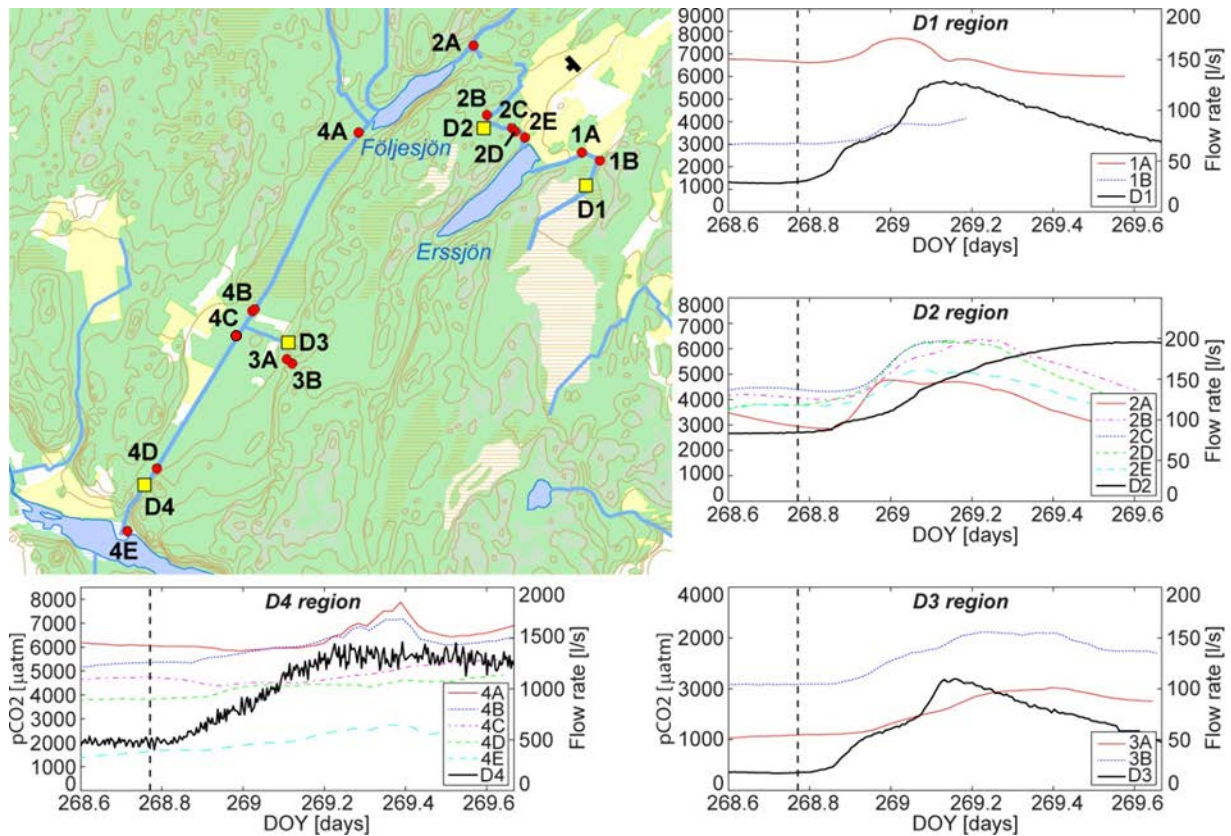
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2 Figure 8. Example of long-term monitoring of $p\text{CO}_{2\text{aq}}$ at 1h intervals in a small shallow boreal
 3 wetland pond (mean depth 1 m). Panel **A** shows raw data indicating spikes in the data most
 4 likely due to condensation events (or possibly related with animals temporary visiting the
 5 chambers; insects, frogs, etc), particularly towards the end of the deployment. Panel **B** shows
 6 the same data as in **A** after a simple filtering procedure removing data points that were more
 7 than 10% greater than the -4 to +4 h median of surrounding the data point.

8



1
 2 Figure 9. Example of 24 h of data from 14 CO₂ logger-chamber units placed on the main
 3 streams in a catchment stream network to log stream $p\text{CO}_{2\text{aq}}$. Yellow squares (D1 – D4)
 4 denote water discharge stations representing stream regions and the water flows from D1 to
 5 D4 with the D3 stream being a tributary entering the main stream upstream of D4. The red
 6 dots represent the CO₂ logger-chamber units. Data (with the initial time of chamber
 7 equilibration removed) are displayed region-wise in the sub-panels together with the
 8 measured discharge. A rain event caused an increase in the discharge half way during the
 9 measurement period which seems related with increased $p\text{CO}_{2\text{aq}}$ in most locations. DOY
 10 denotes day or the year. The map is published with permission from Lantmäteriet, Sweden
 11 according to agreement i2012/898 with Linköping University.

12