



*Supplement of*

## **Changes in dissolved iron deposition to the oceans driven by human activity: a 3-D global modelling study**

**S. Myriokefalitakis et al.**

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**Table S1.** Emissions used in TM4-ECPL for a) present (year 2008), b) past (year 1850) and c) future (year 2100) simulations. All emissions are in Tg yr<sup>-1</sup>, except for NOx (Tg-N yr<sup>-1</sup>), NH<sub>3</sub> (Tg-N yr<sup>-1</sup>), SO<sub>x</sub> (Tg-S yr<sup>-1</sup>) and DMS (Tg-S yr<sup>-1</sup>).

Species		Biomass Burning	Anthropogenic	Aircrafts	Ships	Biogenic	Soils	Oceans
ACETONE	1850	6.46E-02	4.23E-01			3.59E+01		
	2008	4.23E+00	2.64E+00	3.03E-07				
	2100	5.09E+00	2.73E+00	2.80E-07				
BC	1850	2.62E-01	1.05E+00		8.05E-04			
	2008	2.62E+00	5.39E+00	8.39E-06	1.24E-01			
	2100	3.17E+00	1.11E+00	7.32E-06	1.43E-02			
BENZENE	1850	4.83E-01	7.51E-01		8.05E-04			
	2008	4.83E+00	6.31E+00	4.90E-06	1.16E-01			
	2100	5.89E+00	2.15E+00	4.52E-06	4.80E-02			
C <sub>2</sub> H <sub>4</sub>	1850	6.35E-01	1.24E+00		2.21E-03	1.75E+01		1.20E+00
	2008	6.35E+00	6.87E+00	3.04E-06	2.57E-01			
	2100	7.76E+00	2.24E+00	2.80E-06	1.06E-01			
C <sub>2</sub> H <sub>6</sub>	1850	3.17E-01	5.93E-01		1.01E-03	3.01E-01		8.00E-01
	2008	3.17E+00	2.87E+00	4.37E-06	1.50E-01			
	2100	3.96E+00	8.09E-01	4.03E-06	6.22E-02			
C <sub>3</sub> H <sub>6</sub>	1850	6.26E-01	9.14E-01		2.56E-03	1.66E+01		1.30E+00
	2008	6.25E+00	6.41E+00	4.64E-06	2.98E-01			
	2100	7.68E+00	2.27E+00	4.29E-06	1.23E-01			
C <sub>3</sub> H <sub>8</sub>	1850	1.96E-01	2.42E-01		1.96E-01	3.01E-02		1.00E+00
	2008	1.96E+00	3.47E+00	4.87E-06	4.17E-01			
	2100	2.54E+00	1.26E+00	4.50E-06	1.73E-01			
C <sub>4</sub> H <sub>10</sub>	1850	9.07E-02	4.56E-01		6.27E-03	6.09E-02		6.80E+00
	2008	9.07E-01	3.32E+01	4.01E-05	8.79E-01			
	2100	1.11E+00	1.96E+01	3.70E-05	3.64E-01			
CH <sub>2</sub> O	1850	5.87E-01	1.45E-01			4.33E+00		
	2008	5.86E+00	3.39E+00	7.54E-07				
	2100	7.21E+00	1.43E+00	6.97E-07				
CH <sub>3</sub> CHO	1850	4.86E-01	3.16E-01			2.04E+01		
	2008	4.85E+00	1.80E+00	2.07E-06				
	2100	6.00E+00	5.95E-01	1.91E-06				
CH <sub>3</sub> OH	1850	1.17E+00	6.14E-01			1.38E+02		
	2008	1.17E+01	4.87E+00	2.90E-06				
	2100	1.42E+01	2.85E+00	2.68E-06				
CO	1850	4.61E+01	6.30E+01		7.74E-03	8.37E+01		2.00E+01
	2008	4.60E+02	5.99E+02	6.55E-04	1.15E+00			

	2100	5.49E+02	2.34E+02	5.11E-04	2.29E-01			
<b>MEK</b>	1850	0.00E+00	0.00E+00	0.00E+00	0.00E+00	6.98E-01	8.00E+00	
	2008			1.04E-06				
	2100			9.58E-07				
	1850	7.74E-03	7.74E-03		7.74E-03			
<b>NOx</b>	2008	5.48E+00	2.67E+01	1.44E-03	5.14E+00			
	2100	6.39E+00	7.48E+00	1.84E-03	9.47E-01			
	1850	2.33E+00	4.63E+00		9.05E-04			
<b>POA</b>	2008	2.33E+01	1.39E+01	2.52E-06	1.29E-01			
	2100	2.82E+01	3.46E+00	2.20E-06	1.42E-02			
	1850	1.93E-01	1.01E+00		3.60E-02			
<b>SOx</b>	2008	1.93E+00	4.69E+01	1.14E-04	5.22E+00			
	2100	2.21E+00	7.70E+00	1.00E-04	5.04E-01			
	1850	1.42E-01	3.55E-01		5.71E-04	1.41E+00	8.00E+00	
<b>TOLUENE</b>	2008	1.42E+00	7.22E+00	7.36E-06	6.71E-02			
	2100	1.73E+00	4.70E+00	6.80E-06	2.78E-02			
	1850	3.71E-02	2.57E-01		1.56E-03			
<b>XYLENE</b>	2008	3.70E-01	1.18E+01	6.69E-06	2.06E-01			
	2100	4.51E-01	7.66E+00	8.52E-02	6.18E-06			
	1850	1.73E+00	1.45E+00			3.25E+00	8.00E+00	
<b>CH<sub>3</sub>COOH</b>	2008	1.73E+01	7.46E+00					
	2100	2.08E+01	2.33E+00					
	1850	4.33E-01	3.62E-01			3.25E+00	8.00E+00	
<b>HCOOH</b>	2008	4.32E+00	1.87E+00					
	2100	5.21E+00	5.83E-01					
	1850	1.47E-04						
<b>DMS</b>	2008	1.47E-03					1.90E+01	
	2100	1.76E-03						
	1850	5.87E-01						
<b>GLY</b>	2008	5.86E+00					2.00E+01	
	2100	7.21E+00						
	1850	9.20E-01	5.40E+00					
<b>NH3</b>	2008	9.20E+00	3.41E+01				8.00E+00	
	2100	1.00E+01	5.15E+01					
<b>ISOPRENE</b>	2008					5.20E+02		1.00E+00
<b>TERPENE</b>	2008					1.04E+02		2.00E-04

**Table S2.** Aqueous-phase Fe-chemical mechanism used in the TM4-ECPL

Reactions	$K_{298}$ (M <sup>-n+1</sup> s <sup>-1</sup> )	Ea/R (K)	References
$\text{Fe}^{2+} + \text{H}_2\text{O}_2 \rightarrow \text{Fe}^{3+} + \text{OH} + \cdot\text{OH}$	$5.24 \times 10^1$	5050	Kremer, 2003
$\text{Fe}^{2+} + \text{O}_2^- + 2\text{H}^+ \rightarrow \text{Fe}^{3+} + \text{H}_2\text{O}_2$	$1.00 \times 10^7$	5050	Rush and Bielski, 1985
$\text{Fe}^{2+} + \text{HO}_2 + \text{H}^+ \rightarrow \text{Fe}^{3+} + \text{H}_2\text{O}_2$	$1.20 \times 10^6$	5050	Jayson et al., 1973b
$\text{Fe}^{2+} + \text{OH} \rightarrow \text{Fe(OH)}^{2+}$	$4.60 \times 10^8$	1100	Christensen and Sehested, 1981
$\text{Fe}^{2+} + \text{NO}_3 \rightarrow \text{Fe}^{3+} + \text{NO}_3^-$	$8.00 \times 10^6$		Pikaev et al., 1974
$\text{Fe}^{2+} + \text{NO}_2 + \text{H}^+ \rightarrow \text{Fe}^{3+} + \text{HONO}$	$3.10 \times 10^4$		Epstein et al., 1982
$\text{Fe}^{2+} + \text{O}_3 (+ \text{H}_2\text{O}) \rightarrow {}^*\text{Fe(OH)}^{2+} + \text{OH} + \text{O}_2$	$8.20 \times 10^5$		Logager et al., 1992
$\text{Fe}^{3+} + h\nu (+ \text{H}_2\text{O}) \rightarrow \text{Fe}^{2+} + \text{OH} + \text{H}^+$	$6.41 \times 10^{-6}$		Benkelberg and Warneck, 1995
$\text{Fe(OH)}^{2+} + h\nu \rightarrow \text{Fe}^{2+} + \text{OH}$	$4.51 \times 10^{-3}$		Benkelberg and Warneck, 1995
$\text{Fe(OH)}_2^{+} + h\nu \rightarrow \text{Fe}^{2+} + \text{OH} + \cdot\text{OH}$	$5.77 \times 10^{-3}$		Benkelberg et al., 1991
$\text{Fe(OH)}^{2+} + \text{O}_2^- \rightarrow \text{Fe}^{2+} + \text{O}_2 + \cdot\text{OH}$	$1.50 \times 10^8$		Rush and Bielski, 1985
$\text{Fe(OH)}^{2+} + \text{HO}_2 \rightarrow \text{Fe}^{2+} + \text{O}_2 + \text{H}_2\text{O}$	$1.30 \times 10^5$		Ziajka et al., 1994
$\text{Fe}^{3+} + \text{SO}_4^{2-} \rightarrow \text{Fe}(\text{SO}_4)^+$	$3.20 \times 10^3$		Jayson et al., 1973b
$\text{Fe}(\text{SO}_4)^+ \rightarrow \text{Fe}^{3+} + \text{SO}_4^{2-}$	$2.70 \times 10^1$		Jayson et al., 1973b
$\text{Fe}(\text{SO}_4)^+ + h\nu \rightarrow \text{Fe}^{2+} + \text{SO}_4^-$	$6.43 \times 10^{-3}$		Lin et al., 2014
$\text{Fe}^{3+} + \text{C}_2\text{O}_4^{2-} \rightarrow \text{Fe}(\text{C}_2\text{O}_4)^+$	$7.50 \times 10^6$		Lin et al., 2014
$\text{Fe}(\text{C}_2\text{O}_4)^+ \rightarrow \text{Fe}^{3+} + \text{C}_2\text{O}_4^{2-}$	$3.00 \times 10^{-3}$		Lin et al., 2014
$\text{Fe}(\text{C}_2\text{O}_4)^+ + \text{C}_2\text{O}_4^{2-} \rightarrow \text{Fe}(\text{C}_2\text{O}_4)_2^-$	$1.89 \times 10^4$		Lin et al., 2014
$\text{Fe}(\text{C}_2\text{O}_4)_2^- \rightarrow \text{Fe}(\text{C}_2\text{O}_4)^+ + \text{C}_2\text{O}_4^{2-}$	$3.30 \times 10^{-3}$		Lin et al., 2014
$\text{Fe}(\text{C}_2\text{O}_4)^+ + \text{O}_2^- \rightarrow \text{Fe}(\text{C}_2\text{O}_4) + \text{O}_2$	$1.00 \times 10^6$		Sedlak and Hoigne, 1993
$\text{Fe}(\text{C}_2\text{O}_4)^+ + \text{HO}_2 \rightarrow \text{Fe}(\text{C}_2\text{O}_4) + \text{O}_2 + \text{H}^+$	$1.20 \times 10^5$		Sedlak and Hoigne, 1993
$\text{Fe}(\text{C}_2\text{O}_4)_2^- + \text{O}_2^- \rightarrow \text{Fe}(\text{C}_2\text{O}_4)_2^{2-} + \text{O}_2$	$1.00 \times 10^6$		Sedlak and Hoigne, 1993
$\text{Fe}(\text{C}_2\text{O}_4)_2^- + \text{HO}_2 \rightarrow \text{Fe}(\text{C}_2\text{O}_4)_2^{2-} + \text{O}_2 + \text{H}^+$	$1.20 \times 10^5$		Sedlak and Hoigne, 1993
$\text{Fe}(\text{C}_2\text{O}_4) + \text{H}_2\text{O}_2 \rightarrow \text{Fe}(\text{C}_2\text{O}_4)^+ + \text{OH} + \cdot\text{OH}$	$5.24 \times 10^4$		Sedlak and Hoigne, 1993
$\text{Fe}(\text{C}_2\text{O}_4)_2^- + h\nu (+ \text{O}_2) \rightarrow \text{Fe}^{2+} + \text{C}_2\text{O}_4^{2-} + 2\text{CO}_2 + \text{O}_2^-$	$2.47 \times 10^{-2}$		Lin et al., 2014
Equilibrium	Keq (mol kg <sup>-1</sup> )		
$\text{Fe}^{3+} + \text{H}_2\text{O} \leftrightarrow \text{Fe}(\text{OH})^{2+} + \text{H}^+$	$1.10 \times 10^{-4}$		Ervens et al., 2003
$\text{Fe}(\text{OH})^{2+} + \text{H}_2\text{O} \leftrightarrow \text{Fe}(\text{OH})_2^+ + \text{H}^+$	$1.40 \times 10^{-7}$		Ervens et al., 2003

**Table S3.** Number of measurements (N), mean correlation coefficient ( $r$ ), normalised mean bias ( $NMB$ ), normalised mean error ( $NME$ ) and root mean square error ( $RMSE$ ) values calculated for total Iron (TFe) and dissolved Iron (DFe) as presented in Fig 5..

Figure	TM4-ECPL	N	R	NMB (%)	NME (%)	RMSE (ng-Fe/m <sup>3</sup> )
<b>Fig. 5a</b>	<b>TFe</b>	412	0.06	2406	147	6196
<b>Fig. 5b</b>	<b>DFe</b>	412	0.06	67	84	29
<b>Fig. 5c</b>	<b>TFe</b>	452	0.4	16	89	760

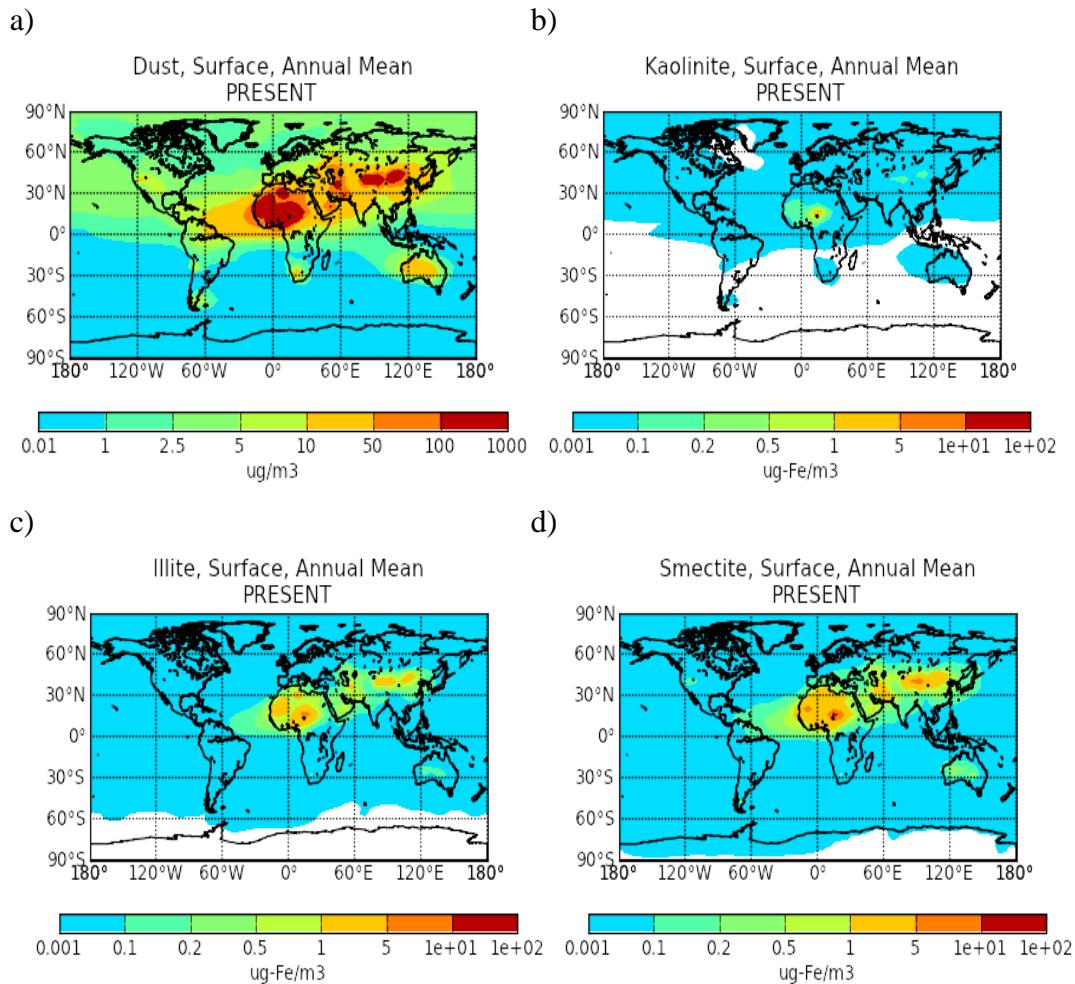
$$R = \left[ \frac{\frac{1}{N} \sum_{i=1}^N (O_i - \bar{O})(P_i - \bar{P})}{\sigma_O \sigma_P} \right] \quad (\text{eq. S1})$$

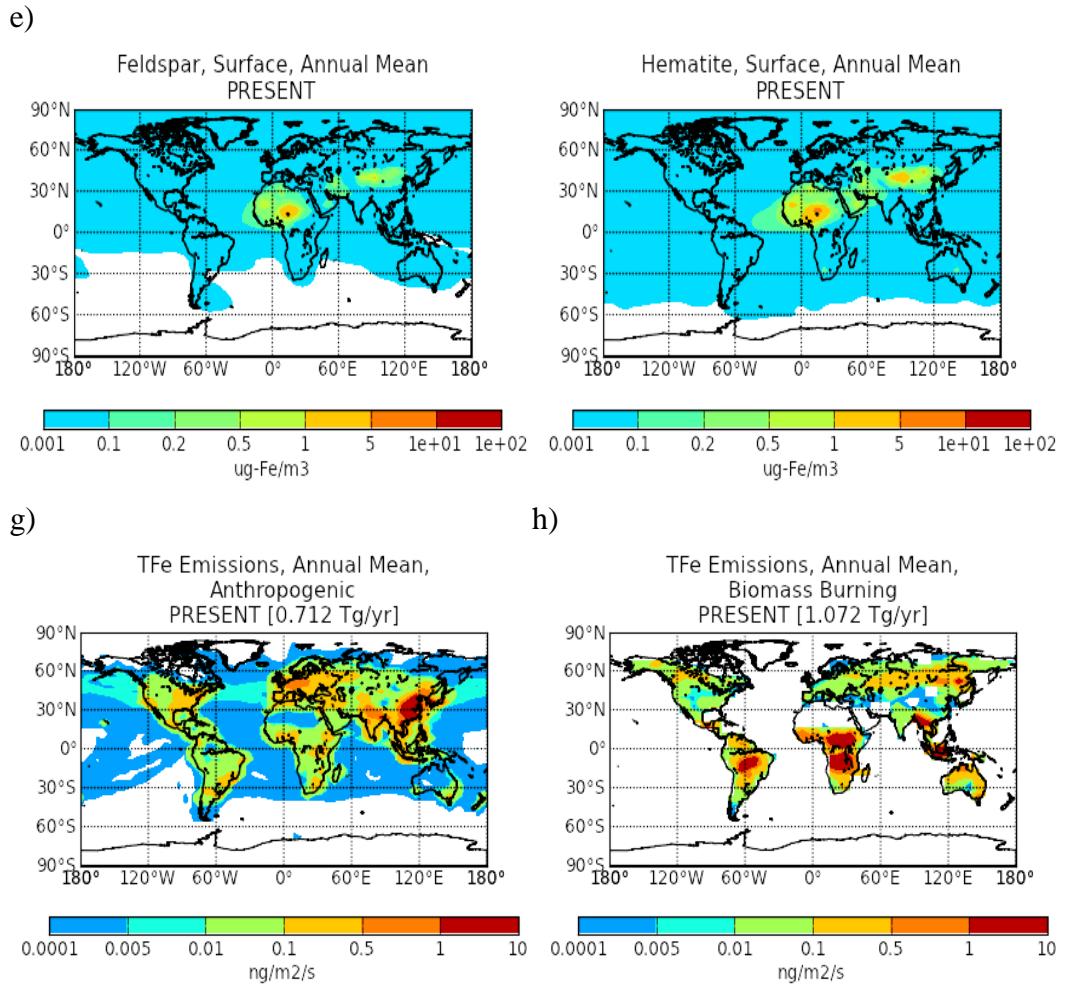
$$NMB = \frac{\sum_{i=1}^N (M_i - O_i)}{\sum_{i=1}^N O_i} \times 100 \quad (\text{eq. S2})$$

$$RMSE = \sqrt{\frac{1}{N} \sum_{i=1}^N (P_i - O_i)^2} \quad (\text{eq. S3})$$

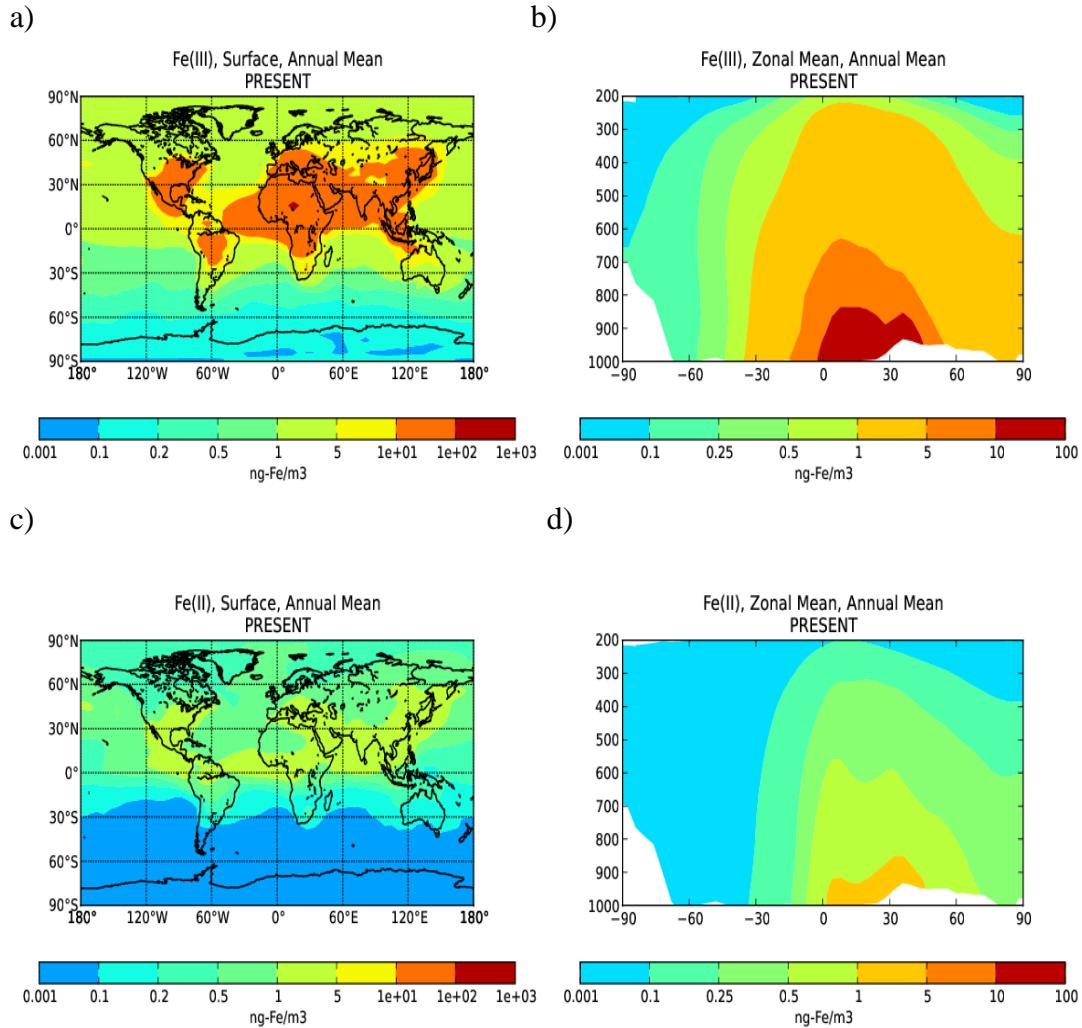
$$NME = \frac{\sum_{i=1}^N |M_i - O_i|}{\sum_{i=1}^N O_i} \times 100 \quad (\text{eq. S4})$$

**Figure S1.** Annual mean surface concentrations of a) dust (in  $\mu\text{g m}^{-3}$ ) and TFe in iron-containing minerals, b) Kaolinite, c) Illite, d) Smectite, e) Hematite, f) Feldspars g) in anthropogenic aerosols (from fossil fuel including shipping emissions) and h) in biomass burning aerosols (in  $\mu\text{g-Fe m}^{-3}$ ) as calculated by TM4-ECPL for the present atmosphere.

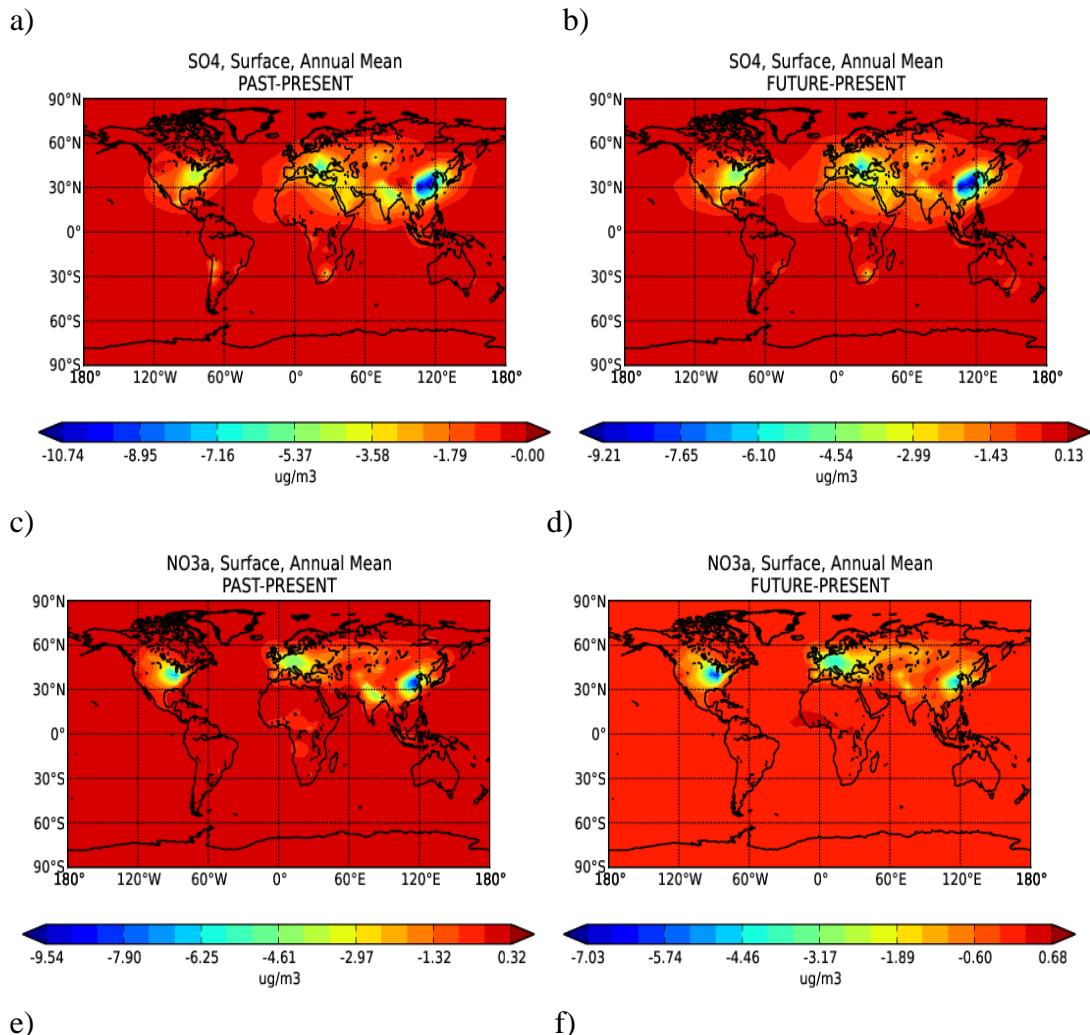


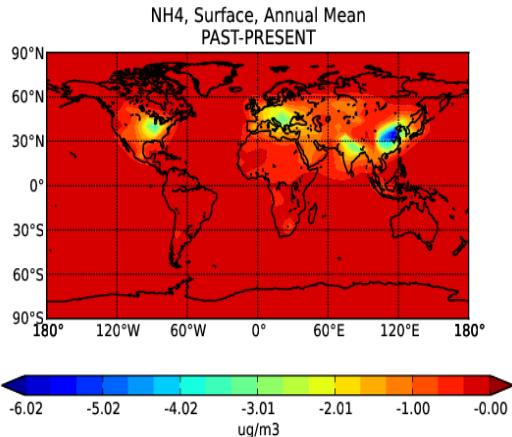


**Figure S2.** Calculated annual mean concentrations for the present atmosphere for the dissolved Fe fractions in ng-Fe m<sup>-3</sup>: Fe(III) (a,b) and Fe(II) (c,d) for surface (left panels: a,c) and zonal mean (b,d).

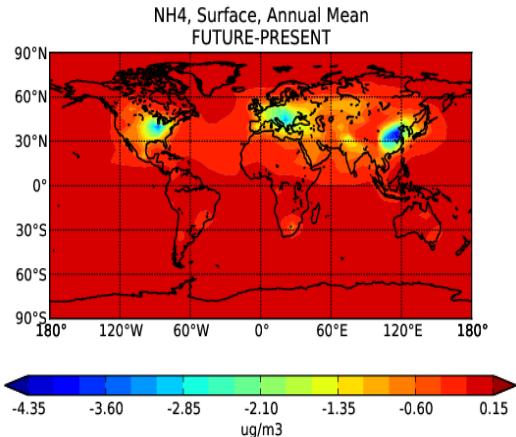


**Figure S3.** Absolute differences (in  $\mu\text{g m}^{-3}$ ) of surface concentrations calculated by TM4-ECPL using the emissions of PAST (left panels: a, c, e, g) and FUTURE (right panels: b, d, f, h) from those calculated using the present-day emission scenario (PRESENT) for a, b)  $\text{SO}_4^{2-}$ ; c, d)  $\text{NO}_3^-$ ; e, f)  $\text{NH}_4^+$  and g, h) for OXL.

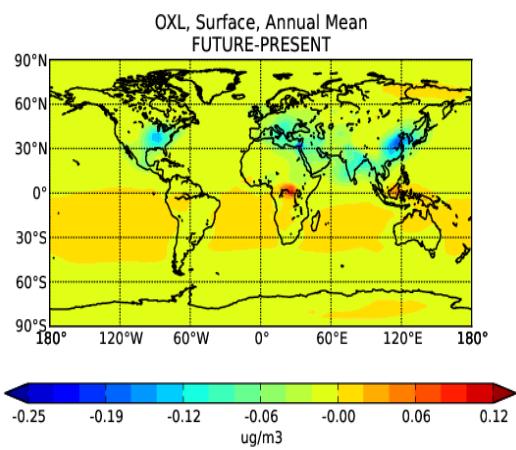
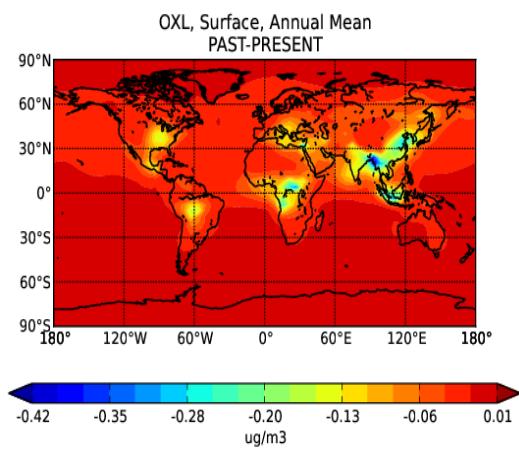




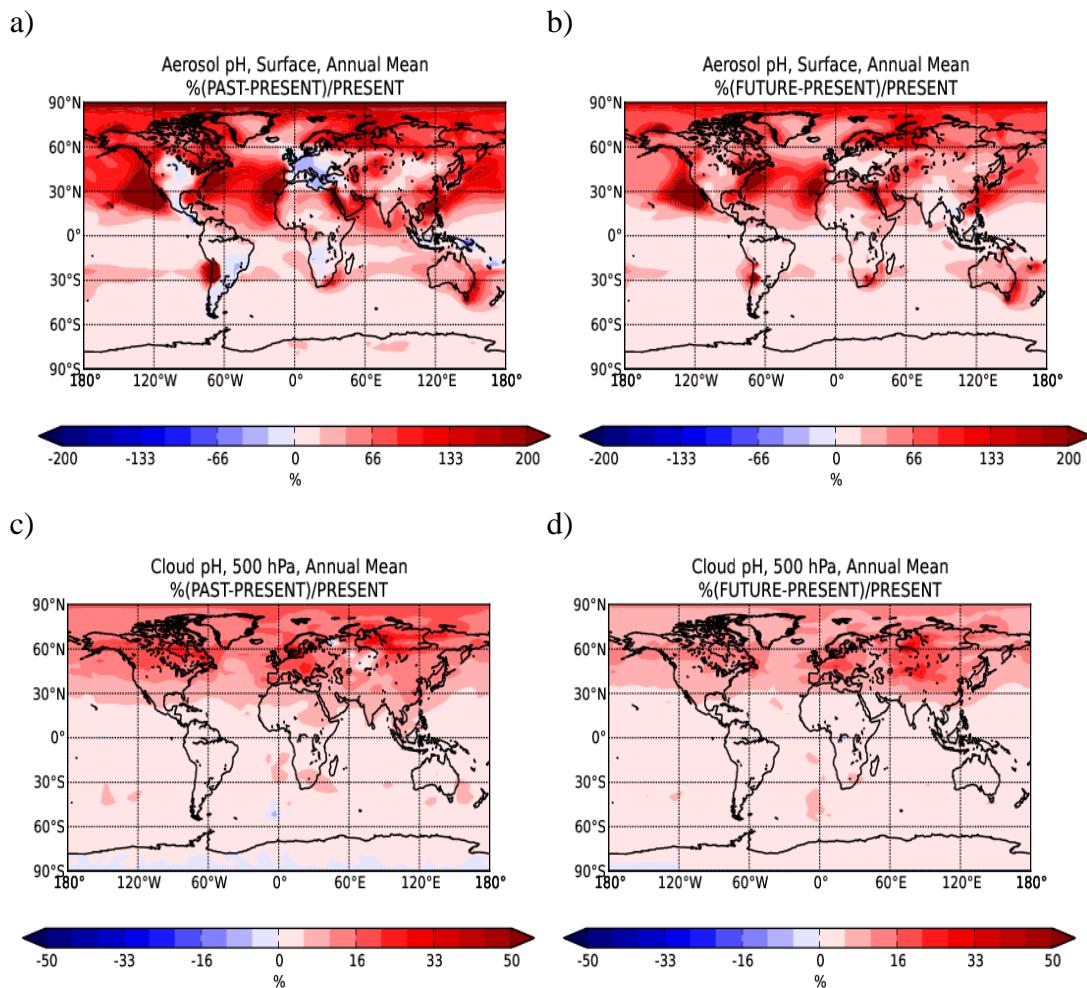
g)



h)

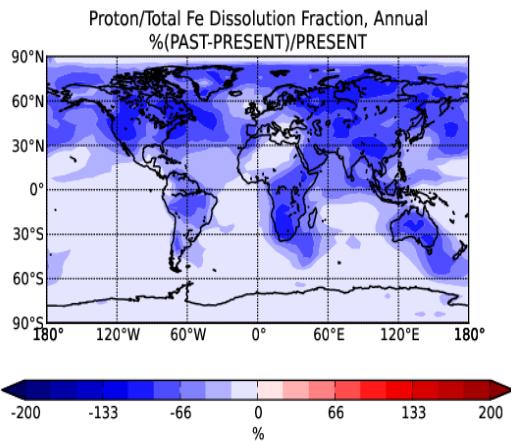


**Figure S4.** Percentage changes (%) in pH calculated by TM4-ECPL using the emissions of PAST (left panels: a, c) and FUTURE (right panels: b, d) simulations from those calculated using the present-day emission scenario (PRESENT) for surface aerosol water (a,b) and cloud water (c,d) at 500hPa.

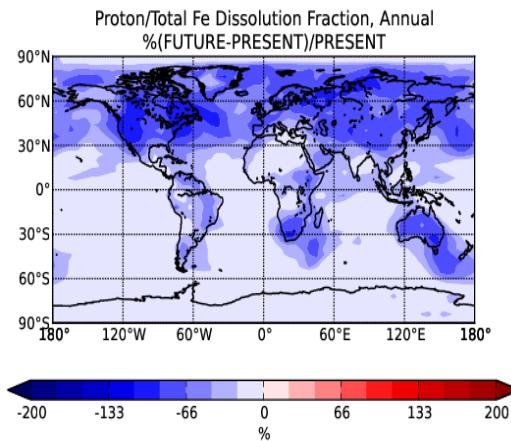


**Figure S5.** The surface percentage differences of PAST (left panels: a,c) and FUTURE (right panels: b,d) simulations from the PRESENT day simulation for a,b) Proton/Total Mineral Dissolution and c,d) Ligand/Total Mineral Dissolution.

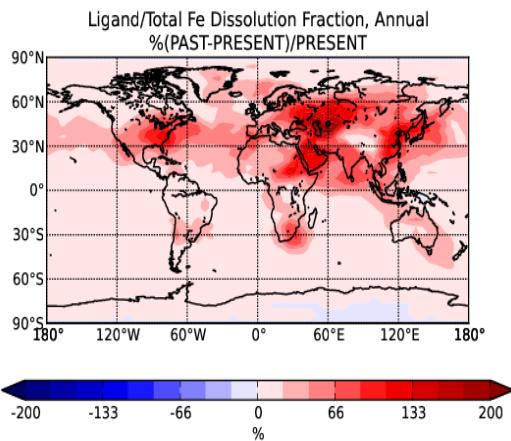
a)



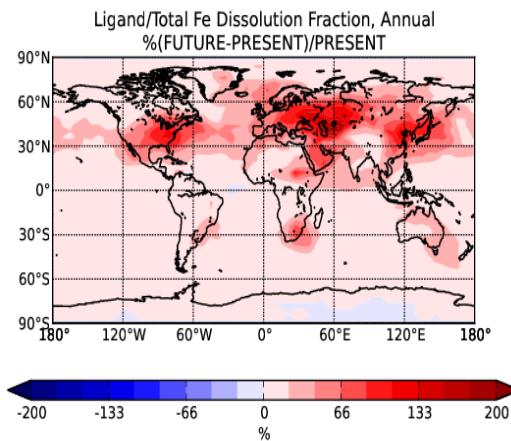
b)



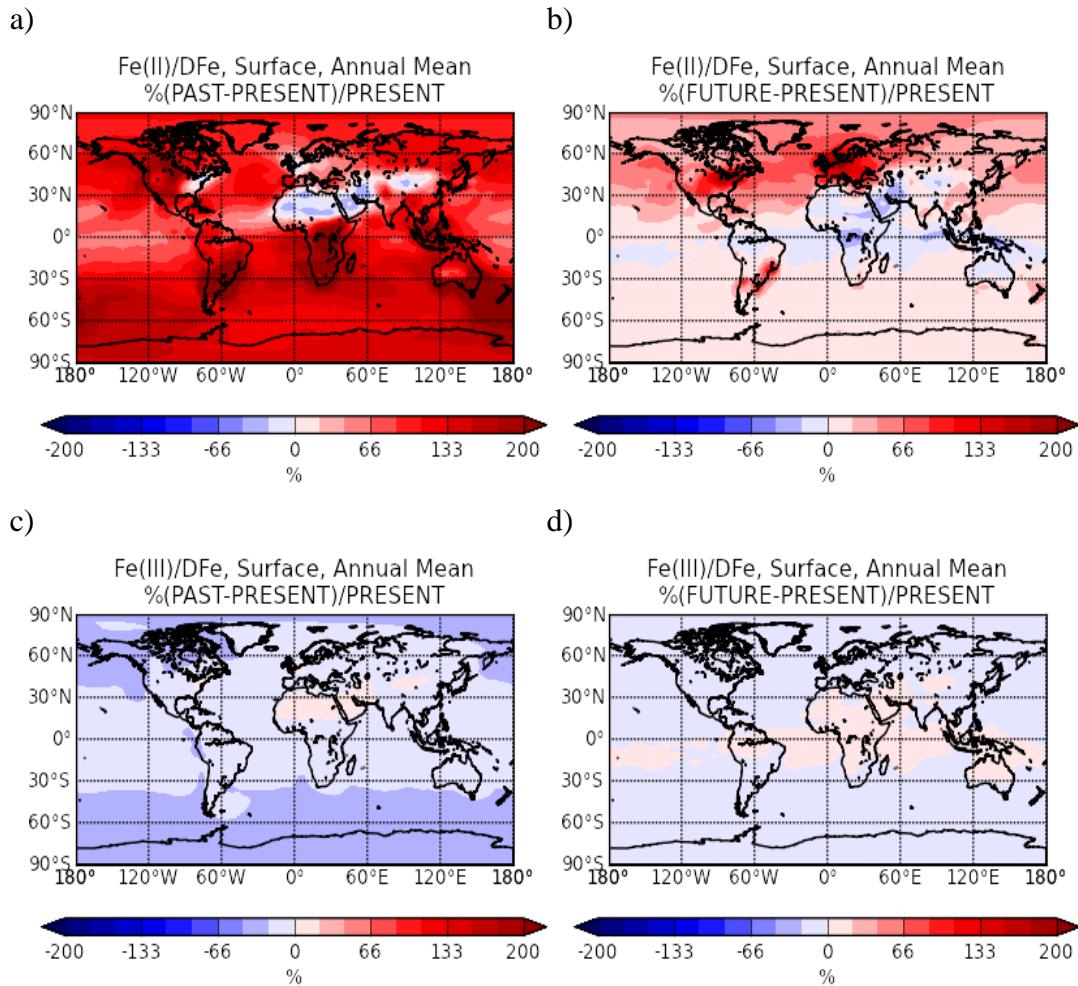
c)



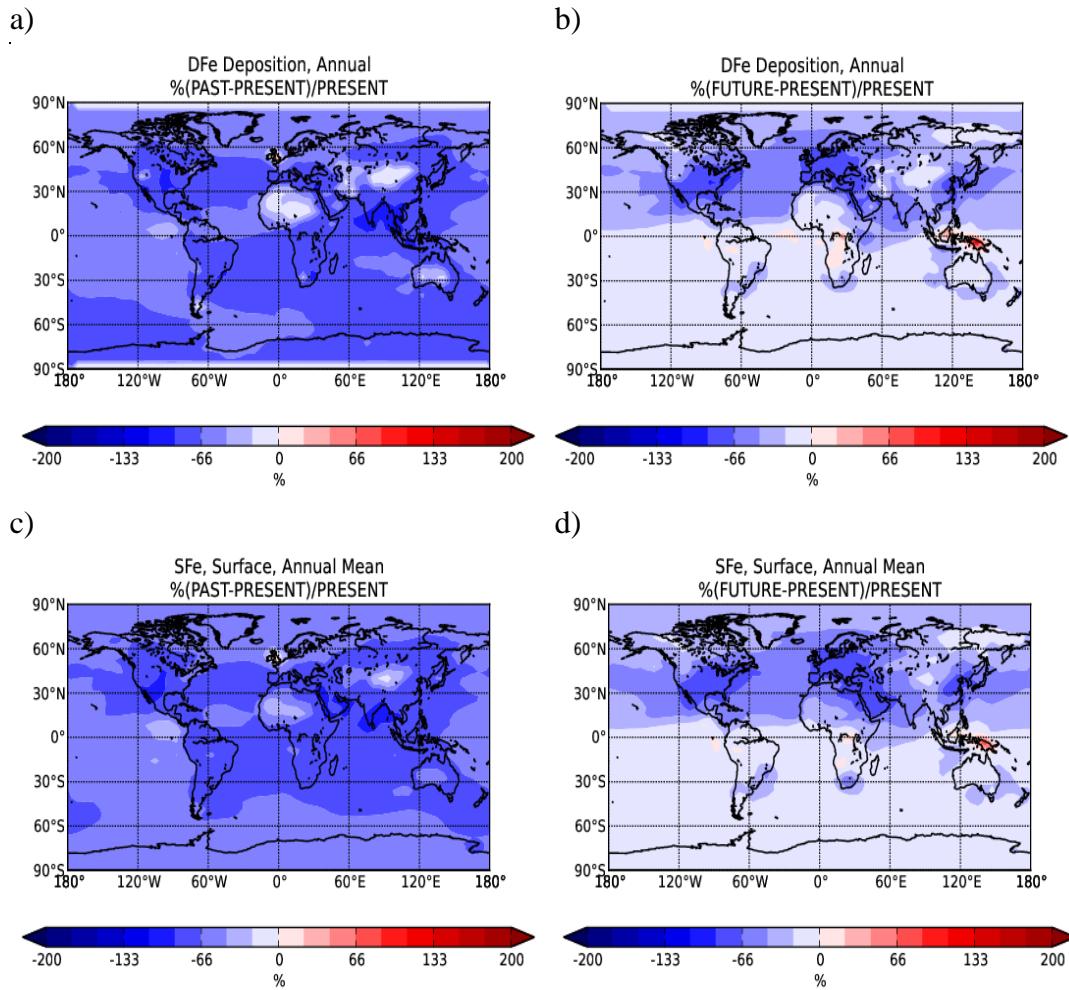
d)



**Figure S7.** The surface percentage differences (%) of PAST (left panels: a,c) and FUTURE (right panels: b, d) simulations from the PRESENT day simulation for a,b) Fe(II)/DFe and c, d) Fe(III)/DFe, as calculated by TM4-ECPL.

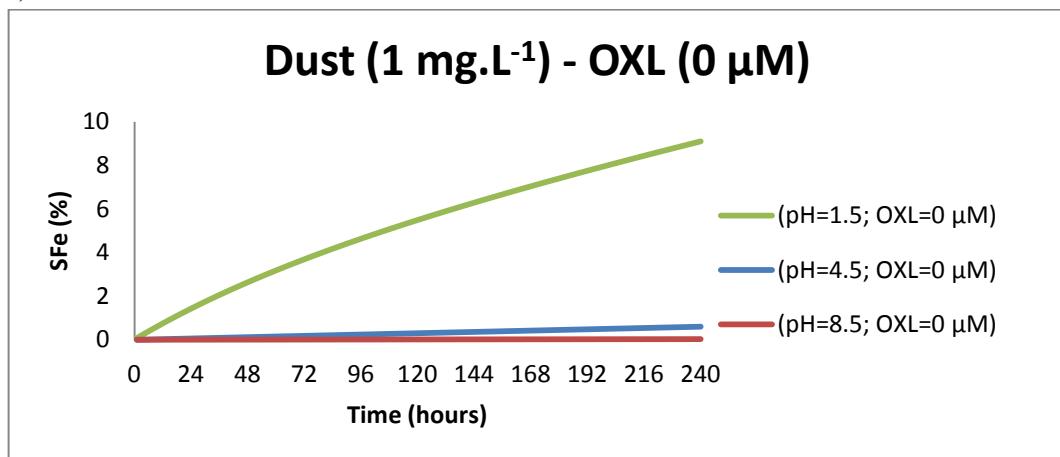


**Figure S8.** The percentage differences (%) of PAST (left panels: a,c) and FUTURE simulations (right panels: b,d) from the PRESENT simulation for a,b) DFe deposition and c,d) Iron Solubility (SFe), as calculated by TM4-ECPL.

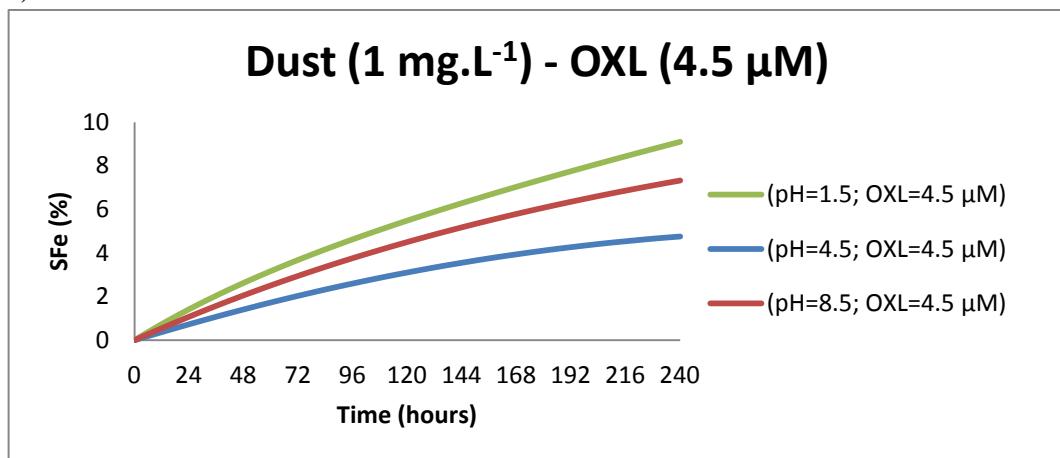


**Figure S9.** The percentage iron dissolution (SFe) predicted from the chemical scheme used for this study for dust concentrations  $1 \text{ mg L}^{-1}$  and pH values of 1.5, 4.5 and 8.5 for OXL concentrations of a)  $0 \mu\text{M}$ , b)  $4.5 \mu\text{M}$  and c)  $8 \mu\text{M}$ .

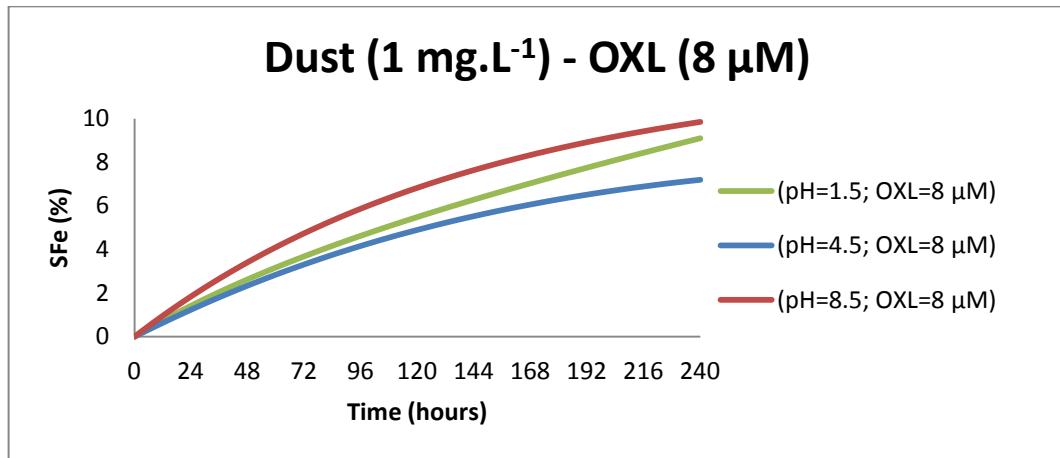
a)



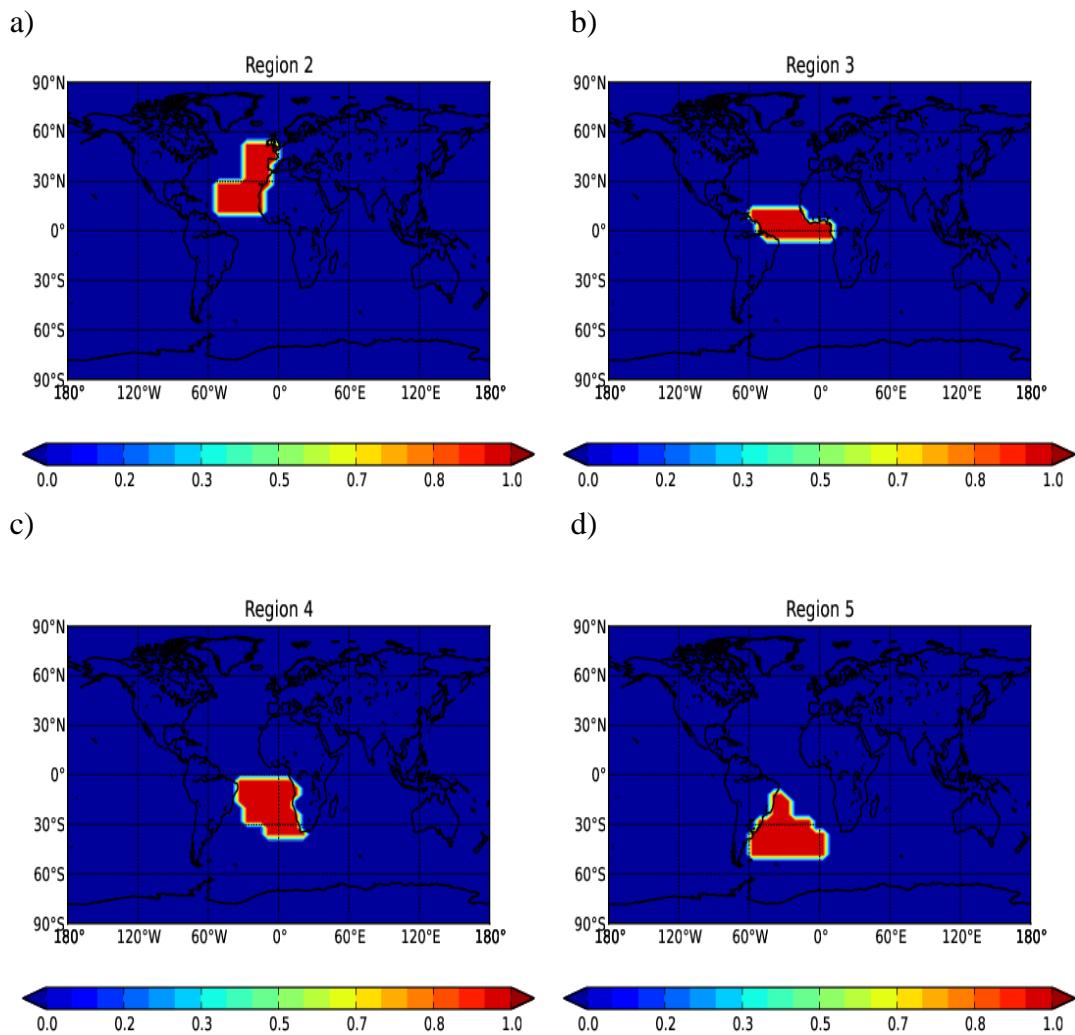
b)



c)



**Figure S10.** Map of the deposition regions (2-5) (red colour) as adopted by Baker et al. (2013) and used for the calculations of Fig. 7.



## References

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