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changes in a future  
climate change  
scenario**

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# Resilience to temperature and pH changes in a future climate change scenario in six strains of the polar diatom *Fragilariopsis cylindrus*

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## Abstract

The effects of ocean acidification and increased temperature on physiology of six strains of the polar diatom *Fragilariopsis cylindrus* from Greenland were investigated. Experiments were performed under manipulated pH levels (8.0, 7.7, 7.4, and 7.1) and different temperatures (1, 5 and 8 °C) to simulate changes from present to plausible future levels. Each of the 12 scenarios was run for 7 days, and a significant interaction between temperature and pH on growth was detected. By combining increased temperature and acidification, the two factors counterbalanced each other, and therefore no effect on the growth rates was found. However, the growth rates increased with elevated temperatures by ~20–50 % depending on the strain. In addition, a general negative effect of increasing acidification on growth was observed. At pH 7.7 and 7.4, the growth response varied considerably among strains. However, a more uniform response was detected at pH 7.1 with most of the strains exhibiting reduced growth rates by 20–37 % compared to pH 8.0. It should be emphasized that a significant interaction between temperature and pH was found, meaning that the combination of the two parameters affected growth differently than when considering one at a time. Based on these results, we anticipate that the polar diatom *F. cylindrus* will be unaffected by changes in temperature and pH within the range expected by the end of the century. In each simulated scenario, the variation in growth rates among the strains was larger than the variation observed due to the whole range of changes in either pH or temperature. Climate change may therefore not affect the species as such, but may lead to changes in the population structure of the species, with the strains exhibiting high phenotypic plasticity, in terms of temperature and pH tolerance towards future conditions, dominating the population.

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# 1 Introduction

The Arctic Ocean is currently experiencing fast environmental changes, such as warming and sea ice loss, as well as sea ice and ecosystem structure changes due to natural and anthropogenic factors (Arrigo, 2014; Nicolaus et al., 2012; Turner and Overland, 2009). According to some models, the average sea surface temperature (SST) in some areas of the global ocean will increase by 1–4 °C over the next 100 years (Alley et al., 2007; Feng et al., 2008), with the largest changes happening in the Arctic (Gradinger, 1995; Hansen et al., 2010). At high latitudes above the Arctic Circle, the average surface air warming rate was found to be about 0.7 °C per decade (~ 6 °C by the end of the 21st century), which will have a strong impact on the SST of the Arctic Ocean (Comiso, 2010). These changes may impact algal communities via changes in physical forcing, biogeochemical cycling, and food web interactions due to loss of habitat (Boras et al., 2010; Fountain et al., 2012; Johannessen and Miles, 2011; Melnikov, 2005). Higher temperatures may intensify heterotrophic processes in sea ice, via increased grazing rates and nutrient regeneration (Melnikov, 2009). Earlier melting of snow cover may accelerate the timing of ice algal blooms, but it is difficult to predict their impact; and mismatching in timing between the phytoplankton production and the reproductive cycle of key Arctic secondary producers could have negative consequences for the entire lipid-driven Arctic marine ecosystem (Søreide et al., 2010). Recent studies on ocean surface warming suggest increased phytoplankton productivity as a consequence of increased temperatures (Feng et al., 2009; Mock and Hoch, 2005; Torstensson et al., 2012). Mock and Hoch (2005) reported that given enough time, the polar diatom *Fragilariopsis cylindrus* could efficiently adjust its photosynthesis to diverse temperatures. Similarly, Torstensson et al. (2012) showed that an elevated temperature (from 0.5 to 4.5 °C) increased the growth rate of the benthic/sea ice diatom *Navicula directa*.

Next to rapid changes in the ocean surface temperature and their consequences on the marine ecosystem, ocean acidification is expected to occur relatively fast in the Arctic environment. The major reasons are its unique features, such as cold and relatively

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fresh surface waters which promote high CO<sub>2</sub> solubility (Yamamoto et al., 2012). According to Alley et al. (2007), the atmospheric partial pressure of CO<sub>2</sub> ( $p\text{CO}_2$ ) is likely to exceed 700 parts per million (ppm) by the year 2100. In the open oceans, where phytoplankton biomass and primary productivity is usually low, this will be accompanied by a seawater pH decline from a global preindustrial level of  $\sim 8.2$  to about 7.8 (Alley et al., 2007; Orr et al., 2005; Yamamoto et al., 2012), with low seasonal variability (Feely et al., 2009). However, in coastal ecosystems pH displays large seasonal and diurnal fluctuations due to high primary production, respiration, upwelling and water residence time (Duarte et al., 2013; Thaisen et al., 2015).

To date, experimental data on phytoplankton tolerance to decreasing pH and rising SST are scarce and mostly only available for phytoplankton from temperate coastal waters. Berge et al. (2010) investigated the tolerance of eight temperate phytoplankton species from four groups (dinoflagellates, cryptophytes, diatoms, prymnesiophytes) to lowered pH, and showed that marine phytoplankton was, in general, resistant to climate change in terms of ocean acidification. Similarly, Nielsen et al. (2011) reported that the investigated coastal plankton communities from temperate regions were unaffected by projected 21 century changes in pH and free CO<sub>2</sub>. Iglesias-Rodriguez et al. (2008) reported increased calcification and primary production of the coccolithophore haptophyte *Emiliania huxleyi* at elevated CO<sub>2</sub> concentrations. On the other hand, Riebesell et al. (2000) and Feng et al. (2008) showed decreasing calcification rates and malformed coccoliths of the same species at increasing acidification. A recent study on ocean acidification in the polar areas showed negative effects on growth rates of the brine algal community, when exposed to pH below 7.6 (McMinn et al., 2014). Likewise, Torstensson et al. (2012) reported somewhat reduced growth rates of the polar diatom *Navicula directa* at increased  $p\text{CO}_2$  levels (960 ppm; pH  $\sim 7.7$ ).

Experimental data on combined effects of elevated temperatures and decreased pH on the growth of phytoplankton from polar waters remain limited and poorly understood (Slagstad et al., 2011). Most studies investigating climate effects on phytoplankton use only one strain as representative of a species despite it being well documented that

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species are genetically and physiologically diverse. Therefore, conclusions based on single strains could potentially be misleading. The aim of the present study was to simulate pH and temperature changes from present to probable future levels, to be able to evaluate their potential impact on the growth of the polar diatom species *Fragilariopsis cylindrus* (Grunow) Krieger, based on six strains of the species. *F. cylindrus* is one of the most widespread and common diatoms in polar and sub-polar regions (Kang and Fryxell, 1992; Lundholm and Hasle, 2008), and an important species in terms of biomass and primary production during spring blooms in the Arctic Sea (von Quillfeldt, 2000). It is common in pack ice as well as in the water column throughout the year (Kang et al., 1993; Kang and Fryxell, 1992), although its relative abundance considerably decreases after late spring (von Quillfeldt, 2000).

## 2 Materials and methods

### 2.1 Cultures

Water samples were collected from Disko Bay (69°11 N, 53°31 W) on the west coast of Greenland. Six different clonal strains (D3G1, D4D11, D10A12, D5A4, D8F4 and D8G3) of *Fragilariopsis cylindrus* were isolated into clonal cultures in April (D3G1 – 23 April 2011, D4D11 and D10A12 – 26 April 2011, D5A4 – 29 April 2011) and May (D8F4 and D8G3 – 7 May 2011) 2011 by isolating single cells or single chains. The strains were cultured at 4 °C at 20–30  $\mu\text{mol photons m}^{-2} \text{s}^{-1}$  following a light : dark cycle of 16 : 8 h, and the medium used was L1 (Guillard and Hargraves, 1993) based on autoclaved 0.2  $\mu\text{m}$  filtered seawater with a salinity of 33.

#### 2.1.1 Experimental setup

The experiments were carried out at three different temperatures, 1, 5, and 8 °C, and four different pH treatments, pH 8.0, 7.7, 7.4, and 7.1. The experiments were designed to ensure that the cells were kept in the exponential growth phase; hence, for the total

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duration of the experiments, maximum growth rates were observed. The first set of experiments with six strains (D3G1, D4D11, D10A12, D5A4, D8F4 and D8G3) was carried out at 5 °C and all four pH treatments. Based on these results, the second set of experiments was carried out with a reduced number of strains (D3G1, D4D11 and D10A12) at 8 °C and with all the pH treatments, and the last set with the same reduced number of strains at 1 °C. The cells were exposed to 90–100  $\mu\text{mol photons m}^{-2} \text{s}^{-1}$ , following a light : dark cycle of 16 : 8 h.

For acclimation, each of the six strains was grown in L1 medium, based on 0.2  $\mu\text{m}$  filtered seawater, with a pH value of 8.0, at a temperature of 5 °C, and at a light intensity of 90–100  $\mu\text{mol photons m}^{-2} \text{s}^{-1}$  for two days. Three of the strains were acclimated to 8 and 1 °C, in steps of  $\sim 2$  °C per day. After two days of acclimation, pH of the cultures was lowered to pH 7.7, 7.4 and 7.1 in steps of 0.3 pH units every 24 h by addition of strongly acidified L1 medium (pH  $0.49 \pm 0.02$ ). The pH level of the acidified L1 medium was lowered by using gaseous  $\text{CO}_2$  (Air Liquid Denmark A/S. UN 1013 Carbon Dioxide, Class 2, 2A, ADR). During the days of acclimation to different temperatures and pH treatments, the strains were grown in 65 mL flasks with L1 medium and exposed to the same light intensity (90–100  $\mu\text{mol photons m}^{-2} \text{s}^{-1}$ ) following a light : dark cycle of 16 : 8 h. The L1 growth medium was selected to ensure nutrient-replete conditions for the algal cultures during the experiment (Lundholm et al., 2004).

The experimental flasks (65 mL) were inoculated with a cell concentration of 1000 cells  $\text{mL}^{-1}$  and L1 medium of the pH-specific value. All experiments were performed in triplicates. For enumeration of cells, 2 mL were withdrawn and fixed with 30  $\mu\text{L}$  of acidic Lugol's solution (2 % final concentration). Before sub-sampling, each flask was gently rotated vertically at least 15 times to ensure that the cells were equally distributed. Sub-sampling was carried out at approximately 10 a.m. every day, starting with those grown at pH 7.1 and followed by those at 7.4, 7.7 and 8.0. Volumes removed for sub-sampling were replaced with equal volumes of adjusted L1 medium. To avoid large fluctuations of pH, the cultures were diluted on a daily basis with pH-specific media. If desired pH was not obtained after dilution, a few drops of acidified L1 medium were added to

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lower pH of the samples. The pH level was measured before and after dilution. For cell counting, an inverted light microscope (OLYMPUS CKX31, 100× magnification) and a Sedgewick-Rafter chamber were used, and a minimum of 400 cells from each sample was counted, corresponding to a deviation of  $\pm 10\%$  using 95% confidence limits (Utermöhl, 1958). Sampling was initiated on day 3 to allow the experimental cultures to acclimate to the experimental conditions and to overcome the initial lag phase (day 0–2). Thus, the time period from day 0 to day 3 was considered as part of the acclimation period and not included in the results.

Temperature and pH were measured using a WTW pH 340i pH-meter with a SenTix 41 electrode, with a sensor detection limit of 0.01. The pH electrode was calibrated weekly (2 point calibration) using Sentron buffers of pH 7.0 and 10.0 dilutions.

### 2.1.2 Dissolved inorganic carbon and nutrients

The concentration of dissolved inorganic carbon (DIC) in fresh media (all four pH treatments) was measured in triplicate. Measurements were done using an infrared gas analyzer (IRGA) and a bicarbonate standard solution ( $2 \text{ mmol L}^{-1}$ ), as described in Nielsen et al. (2007). The concentration of carbon species (bicarbonate ion  $\text{HCO}_3^-$ , carbonate ion  $\text{CO}_3^{2-}$ , and dissolved carbon dioxide  $\text{CO}_2$  (aq) and carbonic acid  $\text{H}_2\text{CO}_3$ ) in the media was calculated from pH, salinity, temperature and DIC, using the CO2SYS.XLS program (*set of constants*: K1, K2 from Mehrbach et al., 1973 refit by Dickson and Millero, 1987;  $\text{KHSO}_4$  from Dickson, 1990; *pH scale* as seawater scale ( $\text{mol kg}^{-1}$ -SW), Lewis and Wallace, 2014).

Samples ( $3 \text{ mL} \times 50 \text{ mL}$ ) for measurements of inorganic nutrients (nitrate  $\text{NO}_3^-$ , phosphate  $\text{PO}_4^{3-}$  and silicate  $\text{Si(OH)}_4$ ) were taken from L1 medium (pH 8.0) and frozen immediately. The samples were analyzed at the Institute for Bioscience, University of Aarhus, following procedures of Hansen and Koroleff (2007).

### 2.1.3 Maximum growth rates

Assuming exponential growth of the cells, the maximum growth rates were calculated from the logarithmic curves of cell growths (logarithmic cumulative cell concentrations vs. days) using the equation

$$\ln N_1 = \ln N_0 + \mu(t_1 - t_0), \quad (1)$$

where  $N_0$  and  $N_1$  are the number of cells at time  $t_0$  and  $t_1$ .

The maximum growth rate for a given strain and pH treatment at a specific temperature was calculated employing linear regression for the steepest part of the growth curve. Linear regression was carried out for each replica of the strain at a given treatment, and the mean of maximum growth rates of the three replicates at a given treatment was taken as the maximum growth rate for that combination of strain and treatment. The temperature coefficients,  $Q_{10}$  for the growth rates, as a consequence of increased temperature by  $10^\circ\text{C}$ , were calculated according to the equation

$$Q_{10} = \left( \frac{\mu_{T_2}}{\mu_{T_1}} \right)^{\left( \frac{10}{T_2 - T_1} \right)}, \quad (2)$$

where  $\mu_{T_1}$  and  $\mu_{T_2}$  are the maximum growth rates at temperatures  $T_1 = 1^\circ\text{C}$  and  $T_2 = 8^\circ\text{C}$ .

### 2.2 Molecular characterization

All six *F. cylindrus* strains (D3G1, D4D11, D10A12, D5A4, D8F4 and D8G3) were used for molecular characterization of ITS1, 5.8S and ITS2 (ITS – Internal transcribed spacer) of the nuclear rDNA. Cells of each of the six strains were concentrated and frozen. DNA extractions, sequencing and alignment followed Lundholm and Hasle (2008).

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Strain D10A12 showed the overall highest growth rates at the highest temperature (8 °C). At pH 7.1, significant differences were found between the growth rates at 1 °C and the two higher temperatures (5 and 8 °C). At pH 7.4 and 8.0, significant differences were observed among all temperatures, whereas at pH 7.7, significant differences were found between 8 °C and the two lower temperatures ( $P < 0.05$ ; Fig. 1a, Fig. S2a). In strains D4D11 and D3G1, significant differences were found between the growth rates for all combinations of treatment (pairwise comparisons,  $P < 0.05$ ; Fig. 1b and c, Fig. S2b and c).

### 3.1.2 Growth vs. pH – at three different temperatures

A general negative effect of increased acidification at three different temperatures on the growth rates was observed (Fig. 2).

At 5 °C, the maximum growth rates were highest in strains D5A4 and D10A12, irrespective of the pH treatment. The maximum growth rates of the three other strains (D4D11, D8F4 and D8G3) were approximately 50 % smaller than those of D5A4 and D10A12 for every pH value, but approximately twice as high as the lowest growth rates observed in strain D3G1 (Fig. 2b, Table S1 in the Supplement). Within the pH 7.1 treatment, significant differences were observed among the growth rates of all the strains ( $P < 0.05$ ), except between D8G3 and D4D11, and D5A4 and D10A12 ( $P > 0.05$ ; Fig. S3-Ila). Within the pH 7.4 and 7.7 treatments, significant differences between the growth rates of each pair of the strains were observed ( $P < 0.05$ ), except between D8G3 and D4D11 ( $P > 0.05$ ; Fig. S3-Ilb-c). Within the 8.0 pH treatment, significant differences were observed among the growth rates of all the strains ( $P < 0.05$ ; Fig. S3-Ild).

Overall we found a decrease in growth rates from pH 8.0 to pH 7.1 at 5 °C (Fig. 2b, Table S1), yet with variation among strains. In strain D8F4, the highest maximum growth rates were observed at pH 8.0, and gradually lower growth rates were observed with increasing acidification. The maximum growth rates in strains D4D11 and D5A4 overall decreased with increasing acidification, although a slight increase at pH 7.4 was ob-



significant differences were observed among the growth rates for all combinations of pH treatments ( $P < 0.05$ ). Significant differences among the growth rates for all combinations of the pH treatments except for the pH combination 7.4–7.7 ( $P > 0.05$ ) were observed in strains D10A12 and D4D1 (pairwise comparisons,  $P < 0.05$ ; Fig. S3-III).

### 3.2 Experimental temperature, pH, DIC and nutrients

Temperature and pH in the experimental treatments fluctuated minimally around the designated values ( $< 0.05^\circ\text{C}$  and  $< 0.03$  units, respectively) (Fig. 3, Table 2). The DIC concentrations increased with decreasing pH of the medium. In all treatments, the concentration of  $\text{HCO}_3^-$  exceeded 90 % of the total inorganic carbon, with the highest share being observed at pH 7.1, and the lowest at pH 8.0. The concentration of  $\text{CO}_2$  (aq) and  $\text{H}_2\text{CO}_3$  decreased with increasing pH, from  $199.8 \pm 4.3 \mu\text{mol L}^{-1}$  at pH 7.1 to  $24.3 \pm 1.4 \mu\text{mol L}^{-1}$  at pH 8.0, whereas the concentration of  $\text{CO}_3^{2-}$  increased from  $16.9 \pm 0.4 \mu\text{mol L}^{-1}$  at pH 7.1 to  $96.7 \pm 4.9 \mu\text{mol L}^{-1}$  at pH 8.0 (Table S4). The concentrations of nutrients  $\text{NO}_3^-$ ,  $\text{PO}_4^{3-}$  and  $\text{Si(OH)}_4$  from L1 medium (pH 8.0) were  $523.04 \pm 5.70 \mu\text{M}$ ,  $30.06 \pm 0.85 \mu\text{M}$  and  $47.44 \pm 4.03 \mu\text{M}$ , respectively, which fitted the Si : N : P = 16 : 16 : 1 ratio of marine diatoms (Justić et al., 1995).

### 3.3 Molecular identification

The sequences of ITS1, 5.8S and ITS2 of all six strains were identical to each other and also identical to strain Real9 of *F. cylindrus* in Genbank with accession number EF660056, confirming their identity and similarity.

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## 4 Discussion and conclusions

### 4.1 Growth of multiple strains of the Arctic diatom *Fragilariopsis cylindrus* at different temperatures and pH

*Fragilariopsis cylindrus* is an ecologically important polar sea-ice and phytoplankton species, and as a model organism it may help us improve the understanding of the consequences resulting from changes in the atmospheric CO<sub>2</sub> concentration and concurrent SST rise in high-latitude environments. By manipulating temperature and pH levels in laboratory experiments, plausible future climate change scenarios were simulated. Throughout the experiment, the temperature and pH of the experimental treatments fluctuated minimally (< 0.05 °C and < 0.03 units, respectively; Table 2), making the treatments clearly separate from each other, and thus enabling an evaluation of the combined effects of ocean acidification and temperature on several strains of a microalgal species.

#### 4.1.1 Combined effects of temperature and pH on growth of the multiple *F. cylindrus* strains

Future marine phytoplankton will not be exposed solely to a decrease in pH but also to other concurrent changes such as increased SST, which is why it is important to consider cumulative effects of multiple climate stressors (e.g. present study; Schlüter et al., 2014; Xu et al., 2014). This study showed a statistically significant interaction between pH and temperature on the growth of all *F. cylindrus* strains cultivated at four pH treatments and three temperatures (two-way ANOVA,  $P < 0.05$ ). An overall positive effect of increased temperature and alkalinisation on the growth rates at the same time was detected for *F. cylindrus*, despite the variability in strain-specific responses (Figs. 1 and 2).

The variation in the growth rates within a single species suggests variation in evolutionary potential within species (Beaufort et al., 2011; Langer et al., 2009), which is

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why it is important to take intra-specific diversity into account when trying to understand the physiology and evolution of natural populations (Collins et al., 2014). This study showed that different strains of *F. cylindrus* can be affected by climate change in different ways. At a temperature of 5 °C and different pH treatments, some strains experienced positive, negative or no effects when treated with specific pH treatment. In contrast, a more general pattern in growth rates was observed at 1 and 8 °C at specific pH treatments – (1) the growth rates of all three strains increased with alkalinity (from pH 7.1 to 8.0), and at the same time (2) all three strains exhibited highest growth rates at pH 8.0 which further increased with elevated temperature (from 1 to 8 °C). These two observed patterns illustrate that the combination of both parameters counterbalances each other. Here is why: if the growth rates of the three strains observed at pH 8.0 and temperature 1 °C, which represent the present conditions in the Arctic environments, are compared with the growth rates obtained at pH 7.7 and temperature 5 °C, which are the conditions expected by the year 2100, no effect of the elevated temperature and acidification can be found (e.g. D3G1  $\mu = 0.22 \pm 0.00$  and  $\mu = 0.23 \pm 0.00$ , respectively). Similar results were observed for *F. cylindrus* strain from the Antarctic when exposed to pH 7.8 and temperature of 6 °C (Xu et al., 2014).

In contrast, if one parameter is examined at a time, a general positive effect of increased temperature (see Sect. 4.1.2), and a general negative effect of increased acidification (see Sect. 4.1.3) is found. However, one has to take into consideration that the largest variability was found among the strains. Some strains showed better performance than others when cultivated in the same conditions, indicating that these strains may display high resilience to the changes in pH and temperature predicted for the 21 century (e.g. present study; Kremp et al., 2012; Langer et al., 2009). Climate change may therefore lead to alterations in strain composition, with the strains exhibiting high phenotypic plasticity, in terms of temperature and pH tolerance, dominating the population. To our knowledge, this is the first study reporting the intra-specific variability of a phytoplankton species from the polar environments in response to elevated temperatures and ocean acidification.

#### 4.1.2 Effects of temperature on growth of the multiple *F. cylindrus* strains

A change of temperature had significant effects on *F. cylindrus* growth rates (Fig. 2). The increasing growth rates correlated to elevated temperatures despite the variability in strain-specific responses. One of the fastest growing strains (D10A12) generally exhibited 50 % higher growth rates than the other strains, irrespective of the treatment. The growth rates were always highest at 8 °C regardless of pH. The slowest growing strain (D3G1) displayed the biggest differences in growth rates when cultured at elevated temperatures, with ~ 50 % higher growth rates at 8 °C compared to those at 1 °C. The fastest growing strain (D10A12) exhibited ~ 20 % higher growth rates at 8 °C compared to 1 °C. The resulting  $Q_{10}$  values were found to be strain-specific, ranging from 1.29 to 3.35 (1–8 °C; Table 1). These results are in agreement with findings on *F. cylindrus* exhibiting increased growth rates at elevated temperatures by approximately 45 % ( $Q_{10} = 2.28$ ) in the temperature range of –1–7 °C (Mock and Hoch, 2005). Likewise, Torstensson et al. (2012) reported that at elevated temperatures (from 0.5 to 4.5 °C) the growth rates of the benthic/sea ice diatom *Navicula directa* increased by approximately 43 %. As similar data on polar phytoplankton are restricted, the present study provides important information for evaluating the effects of temperature increase in polar areas. *Fragilariopsis cylindrus*, a polar microalgal representative, is well-adapted to a wide range of temperatures due to the phenotypic variation among strains and temperature adaptation of individual strains. We found that increased temperatures had greatest impact on the slowest growing strain (Table 1, Fig. 2).

The noteworthy variability in strain-specific responses, with growth rates varying up to ~ 65 % suggests that some strains perform better when exposed to perturbations in the environment than others (Fig. 2, Tables S1–3). This high phenotypic plasticity in terms of temperature tolerance could be explained by the large temperature fluctuations that occur concomitantly with changes in solar irradiance, which phytoplankton experiences on daily and seasonal bases in polar environments. In Disko Bay, the spring SST vary from  $-1.1 \pm 0.3$  to  $2.3 \pm 2.0$  °C (based on 4 year temperature data provided

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by DiskoBasis/Arctic Station, Faculty of Science, University of Copenhagen), and from  $-1.8$  to  $6.7$  °C during the year (Hansen et al., 2012) with an average SST of  $1.8 \pm 1.2$  °C (March–December; DiskoBasis/Arctic Station). Similar intra-specific variation has also been observed among strains of the diatom *Skeletonema marinoi* from two geographic areas. Kremp et al. (2012) reported that *S. marinoi* strains from the North Sea, which also experience large temperature fluctuations in their natural environment, uniformly exhibited higher growth rates at elevated temperatures (at 24 compared to 20 °C), indicating that the ability to adjust to varying temperature is advantageous for species. In contrast, *S. marinoi* strains from the temperate Adriatic Sea, where the temperature is known to be more stable, responded to elevated temperatures in different ways, with some strains being unaffected and others being positively or negatively affected by temperature changes (Kremp et al., 2012). The present study confirms previous notion of high intra-specific variability within species, and emphasizes, that this variation might be even larger than the variation observed due to changing environmental factors, stressing the need for several strains when exploring the environmental effects on species.

#### 4.1.3 Effects of pH on growth of the multiple *F. cylindrus* strains

Acidification results in both decreasing pH and increasing CO<sub>2</sub> concentration. Generally, rising CO<sub>2</sub> is considered to facilitate photosynthetic carbon fixation by some phytoplankton groups (Riebesell, 2004). The direct effect of changes in environmental pH is less clear but recent studies have shown that it can affect intracellular pH and membrane potential, as well as enzyme activity (McMinn et al., 2014). In the present study, the concentration of DIC and the carbon species at pH 8.0 corresponded well to the concentrations found in the ocean surface;  $\sim 2$  mmol L<sup>-1</sup> DIC, with  $\sim 90$  % HCO<sub>3</sub><sup>-</sup>,  $\sim 9$  % CO<sub>3</sub><sup>2-</sup>, and  $\sim 1$  % CO<sub>2</sub> (aq) and H<sub>2</sub>CO<sub>3</sub> (Feely et al., 2009; Riebesell, 2004). DIC increased with increased acidification from  $\sim 2.15$  mmol L<sup>-1</sup> at pH 8.0 to  $\sim 2.65$  mmol L<sup>-1</sup> at pH 7.1 (Table S4), which is in agreement with the predictions by Feely et al. (2009).

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The present study found a general negative effect of increasing acidification on *F. cylindrus* growth, and similarly to what was observed for temperature, a significant variability among the strains was observed (Fig. 1). A decrease in pH to 7.7, which is the expected global change in pH by the end of this century, and further decrease to 7.4, negatively affected growth of most of the strains. Reduced growth rates by 2–23 % for a pH of 7.7 and 4–29 % for 7.4, were observed (compared to pH 8.0). On the other hand, some strains were unaffected by the increased seawater acidity, and some of them were even positively affected (the growth rates increased up to 15 % for a pH of 7.7 and 25 % for a pH of 7.4 compared to pH 8.0). A further decrease in pH to 7.1 reduced the growth rates of most of the strains by 20–37 %, as compared to those at pH 8.0. Similarly, McMinn et al. (2014) reported that the brine algal community experienced significantly reduced growth rates at pH 7.6, and when treated at pH 7.2, the growth decreased by 50 % compared to a pH of 8.0. Reduced growth rates due to increased acidification were also observed in *Navicula directa*. At a  $p\text{CO}_2$  level of 960 ppm (corresponding to pH  $\sim$  7.7), the diatom experienced reduced growth by 5 % as compared to 380 ppm (corresponding to pH  $\sim$  8.1) (Torstensson et al., 2012). Similar to our results, a recent study on ocean acidification in the Arctic marine ecosystem found decreased growth rates of the polar phytoplankton community when exposed to increased seawater acidity; community growth rates gradually decreased with increasing acidification (from 8.0 to 7.4) with a greater reduction at pH 7.1, where the community experienced reduced growth rates by 55 % as compared to pH 8.0 (Thoisen et al., 2015).

To date, studies on phytoplankton responses to ocean acidification have mainly been focused on temperate or tropical regions, and only a few studies have been carried out in polar regions (e.g. present study; McMinn et al., 2014; Thoisen et al., 2015; Torstensson et al., 2012). However, increased  $p\text{CO}_2$  ( $\sim$  1000 ppm, which corresponds to pH  $\sim$  7.7) affected tropical phytoplankton communities, which were found to respond with decreased primary production by 7–36 % (Gao et al., 2012). In contrast, Berge et al. (2010) and Nielsen et al. (2010) showed that the growth of coastal marine phytoplankton was, in general, unaffected by ocean acidification (pH  $\sim$  7.0–8.5 and 7.6–8.0,

respectively). They speculated that common natural pH fluctuations in coastal regions made phytoplankton more pH-tolerant in these areas and therefore growth was not affected. Similar pH fluctuations were observed in the Arctic coastal waters (Disko Bay) during the spring bloom in 2012, with a pH gradient of 7.5–8.3. The pH fluctuation was found to be caused by the transition from the polar night period and the dominating respiration processes (pH ~ 7.5) to the polar day period with the increasing phytoplankton biomass and concomitant photosynthesis (Thoisen et al., 2015). The present study showed that *F. cylindrus* is generally well-adapted to acidification down to ~ pH 7.4, although with notable strain-specific response variability. Some strains (e.g. D10A12, D4D11, D5A4, and D8G3) were slightly affected by the lower pH (–10% <  $\mu$  < +10%), whereas others (e.g. D3G1, D8F4) responded with greater reduction in the growth rates (< 29%). Similar observations on the strain-specific response were also reported among other phytoplankton species (Kremp et al., 2012; Langer et al., 2009). Thus, these observations suggest that shifts in dominance among strains due to ocean acidification might be expected.

Long-term adaptation to environmental parameters of *F. cylindrus* strains was not considered in this study. However, the adaptation is expected to increase phenotypic plasticity (Schlüter et al., 2014) of the strains, and therefore the ability of the species to adapt to future climate conditions should increase even more.

**The Supplement related to this article is available online at doi:10.5194/bgd-12-4627-2015-supplement.**

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**Table 1.** The  $Q_{10}$  values (1–8 °C) for D10A12, D4D11 and D3G1 strains were calculated according to Eq. (2), based on the mean maximum growth rates displayed in Table S2 and Table S3.

	D10A12	D4D11	D3G1
7.1	1.29	2.42	3.35
7.4	1.54	2.06	2.34
7.7	1.48	2.45	2.81
8.0	1.46	2.36	2.43

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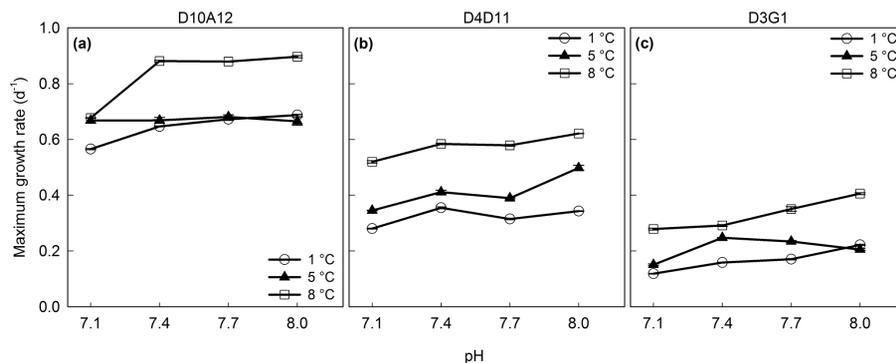
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**Table 2.** The average temperatures  $\pm$  SD ( $^{\circ}$ C) and pH values  $\pm$  SD in the experimental treatments from day 0 to 7;  $n^a = 12$ ;  $n^b = 24$ .

	Average temperatures ( $^{\circ}$ C)		
	$1.4 \pm 0.6$ $^{\circ}$ C <sup>a</sup>	$5.4 \pm 0.1$ $^{\circ}$ C <sup>b</sup>	$7.7 \pm 0.2$ $^{\circ}$ C <sup>a</sup>
Average pH values	$7.10 \pm 0.01$ <sup>a</sup>	$7.09 \pm 0.02$ <sup>b</sup>	$7.11 \pm 0.01$ <sup>a</sup>
	$7.37 \pm 0.01$ <sup>a</sup>	$7.38 \pm 0.02$ <sup>b</sup>	$7.40 \pm 0.01$ <sup>a</sup>
	$7.64 \pm 0.01$ <sup>a</sup>	$7.67 \pm 0.02$ <sup>b</sup>	$7.68 \pm 0.02$ <sup>a</sup>
	$7.91 \pm 0.01$ <sup>a</sup>	$7.95 \pm 0.01$ <sup>b</sup>	$7.95 \pm 0.03$ <sup>a</sup>

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**Figure 1.** The mean maximum growth rates ( $d^{-1}$ ) of strains D5A4, D10A12, D4D11, D8G3, D8F4 and D3G1 cultivated at (a) 1°C, (b) 5°C and (c) 8°C, and all four pH treatments. Error bars represent  $\pm$  one SD.

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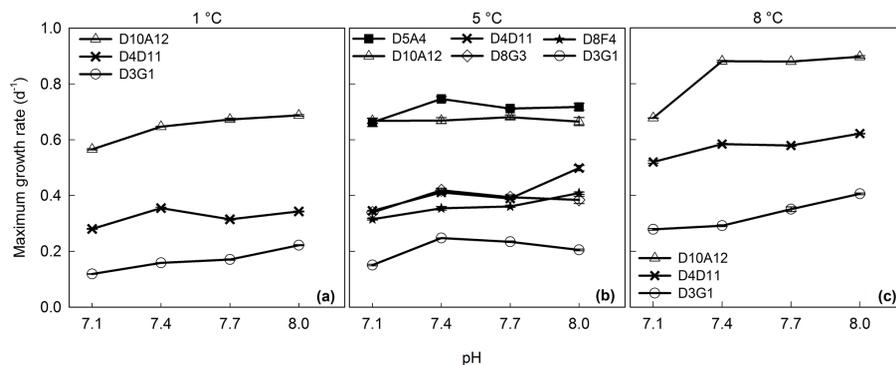
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**Figure 2.** The mean maximum growth rates ( $d^{-1}$ ) of strains D5A4, D10A12, D4D11, D8G3, D8F4 and D3G1 cultivated at **(a)** 1°C, **(b)** 5°C and **(c)** 8°C, and all four pH treatments. Error bars represent  $\pm$  one SD.

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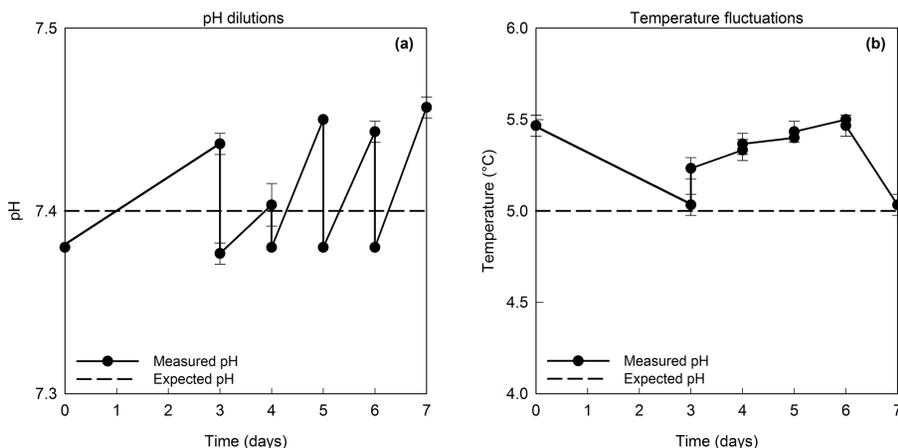
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**Figure 3.** (a) An example of pH dilutions in the pH treatment 7.4 at 5 °C, shown as a function of time. (b) An example of temperature fluctuations in the treatment with pH 7.4, shown as a function of time. The first three days represent the acclimation period and are not included in the results. Error bars represent  $\pm$  one SD.

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