

1 **Hydrologic controls on DOC, As and Pb export from a polluted peatland – The**
2 **importance of heavy rain events, antecedent moisture conditions and hydrological**
3 **connectivity**

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1 **Abstract**

2 Bogs can store large amounts of lead (Pb) and arsenic (As) from to atmospheric deposition of
3 anthropogenic emissions. Pb and As are exported along with dissolved organic carbon (DOC)
4 from these organic-rich systems, but it is not yet clear which hydrological (pre-)conditions
5 favor their export. This study combines a one year monitoring of precipitation, bog water
6 level and pore water concentration changes with bog discharge and DOC, iron, As and Pb
7 stream concentrations. From this data annual DOC, As, and Pb exports were calculated.
8 Concentrations ranged from 5 mg L⁻¹ to 30 mg L⁻¹ for DOC, 0.2 µg L⁻¹ to 1.9 µg L⁻¹ for As,
9 and 1.3 µg L⁻¹ to 12 µg L⁻¹ for Pb with highest concentrations in late summer. As and Pb
10 concentrations significantly correlated with DOC concentrations. Fluxes depended strongly
11 on discharge, as 40% of As and 43% of Pb were exported during 10% of time with highest
12 discharge, pointing out the over-proportional contribution of short-time high discharge events
13 to annual As, Pb and DOC export. Exponential increase in element export from the bog is
14 explained by connection of additional DOC, As and Pb pools in the acrotelm during water
15 table rise, which is most pronounced after drought. Pb, As and DOC concentrations in pore
16 water provide evidence of an increase of the soluble Pb pool as soon as the peat layer gets
17 hydrologically connected, while DOC and As peak concentrations in runoff lag behind in
18 comparison to Pb. Our data indicates a distinct bog-specific discharge threshold of 8 L s⁻¹,
19 which is thought to depend mainly on the bogs size and drainage conditions. Above this
20 threshold element concentration do not further increase and discharge gets diluted. Combining
21 pore water and discharge data shows that As and Pb exports are not only dependent on the
22 amount of precipitation and discharge, but on the frequency and depth of water table
23 fluctuations. Comparing the annual bog As and Pb export with element inventories indicates
24 that As is much more mobilized than Pb, with annual fluxes accounting for 0.85 ‰ and 0.27
25 ‰ of total As and Pb inventory, respectively.

1 **1 Introduction**

2 Peatlands provide important services in the environmental system. They play an important
3 role in the storage of carbon and other elements such as heavy metals. On one hand the
4 filtering function of peatlands partly prevents the input of these pollutants to streams. On the
5 other hand, peatlands can release heavy metals as soluble DOM-metal complexes. Even
6 though, acting as major carbon sink, peatlands are also the greatest source of dissolved
7 organic carbon (DOC) to the aquatic system (Aitkenhead et al., 1999). Several studies
8 demonstrated runoff, water level fluctuations and temperature to be the most important
9 controls on terrestrial DOC production and export (Freeman et al., 2001; Hongve et al., 2004;
10 Köhler et al., 2008; Seibert et al., 2009; Laudon et al., 2011). Recently, numerous studies
11 describe the emerging problem of rising DOC levels in streams and lakes especially in
12 catchment areas comprising peatlands (e.g. Worrall et al., 2004; Monteith et al., 2007), which
13 might cause severe problems for aquatic biota and in drinking water production (Chow et al.,
14 2003). Besides the observed general increase in concentrations, which has been attributed to
15 the decrease in acid rain deposition or temperature rise (Freeman et al., 2001; Evans et al.,
16 2005; Monteith et al., 2007), a flushing effect, which produces high instantaneous DOC loads
17 during high discharge events after a summer dry period has been described (e.g. Worrall et al.,
18 2002; Clark et al., 2007). Even though temporally limited, the high concentrations might
19 affect the aquatic system even stronger. This effect gains more importance through the
20 predicted climate change with enhanced dry periods and more frequent heavy rain events
21 (IPCC, 2013). How these high DOC concentrations after re-wetting of the peatlands surface-
22 near layer built up is not yet fully understood. Fenner and Freeman (2011) proposed that the
23 microbial DOC production is stimulated after a drought period by reduced concentrations of
24 phenolic compounds through an enhanced phenol oxidase activity. Moreover, release of DOC
25 adsorbed to Fe-(oxy)hydroxides through Fe-(oxy)hydroxides dissolution after re-wetting and

1 decreasing redox potential has shown to be an important process for DOC dynamics in
2 peatlands (Bauer and Blodau, 2009; Knorr, 2013; Riedel et al., 2013), as well as DOC
3 solubilization through pH rise (Grybos et al., 2009). Similarly, Clark et al. (2012) described a
4 decrease of DOC solubility due to acidification during droughts.

5 Although the general focus in most studies is set on changing DOC dynamics, it is also
6 known that some elements are mainly exported by transport with DOC to the discharging
7 stream. Lead (Pb) strongly binds to organic matter (Tipping, 1998; Rothwell et al., 2007) and
8 high DOC concentrations increase Pb mobility (Jordan et al., 1997). In spite of the high
9 affinity of arsenic (As) to iron(oxy)hydroxides, recent mechanistic studies revealed a strong
10 binding of As to sulfhydryl groups of organic matter in the anaerobic peat layer (Langner et
11 al., 2011, 2014). Due to the low pH (<4.5 pH) and the low amount of mineral phases in peat, a
12 correlation of DOC and As concentrations in bog draining streams has been observed
13 (Rothwell et al., 2009; Neubauer et al., 2013). Moreover, As is known to be subjected to post-
14 depositional mobilization due to water level fluctuations and resulting redox changes (Blodau
15 et al., 2008; Rothwell et al., 2009; Langner et al., 2014).

16 Bogs, which receive element inputs exclusively by atmospheric deposition, and also other
17 peatland types have the potential of accumulating As and Pb in the peat. The anthropogenic
18 deposition rate of those two elements largely exceeds pre-industrial background fluxes
19 (Shotyk, 1998; Bindler, 2006). Mining activity, fossil fuel combustion, especially emissions
20 from burning of leaded gasoline, often resulted in high As and Pb concentrations in peat
21 layers, which developed over the past centuries. Based on observed metal concentrations
22 some peatlands especially those influenced by mining areas would even have to be classified
23 as highly contaminated soils.

1 The near surface layer of a bog, the acrotelm, is the most active part, and conducts most of the
2 bogs discharge (Evans et al., 1999; Holden and Burt, 2003) and DOC production (Clark et al.,
3 2008). Due to high porosity, the acrotelm has a high hydraulic conductivity and is
4 characterized by recurrent water table drawdowns and aeration resulting in higher biological
5 activity. Moreover, this surface-near peat layer, which often hosts large amounts of
6 atmospheric derived metals, is the main source of metal release from ombrotrophic peatlands.

7 Up to now, most studies on heavy metal release from contaminated peatlands have focused on
8 discharge, only. Studies connecting DOC export with *in-situ* pore water chemistry
9 measurements are rare or lacking in regard of Pb and As. Clark et al. (2005) found a strong
10 correlation between peat soil solution DOC and stream water DOC concentrations over a ten-
11 year record. Many field or laboratory studies focused on DOC production in view of water
12 level, redox-state, acidification, temperature or microbial conditions (e.g. Grybos et al., 2009;
13 Clark et al., 2012; Evans et al., 2012) without a connection to discharge measurements or they
14 were conducted in mineral soil environments (e.g. Kokfelt et al., 2009; Singh et al., 2014, for
15 Pb see Vinogradoff et al., 2005). In regard of As, Blodau et al. (2008) found a higher release
16 of As in minerogenic peat mesocosms after a drought period, which is in line with leaching
17 experiments described by Tipping et al. (2003). Rothwell et al. (2009) observed higher As
18 concentrations during late summer storm flow after the summer dry period in a bog
19 catchment.

20 The aim of this study is to gain further understanding in the generation and controls of Pb and
21 As export from ombrotrophic peatlands. If the export of Pb and As is mainly controlled by
22 DOC production and transport, higher metal concentrations and fluxes are expected during
23 periods of elevated DOC concentrations. Moreover, export dynamics should strongly depend
24 on peatland hydrology. This study tries to unravel the importance of hydrologic conditions
25 like antecedent bog water levels, as well as precipitation, temperature and pH for the export of

1 As and Pb. We expect that these factors mainly control hydrologic pathways, DOC
2 production and the hydraulic connection of different As and Pb pools. We thus aim to
3 investigate to which extent As and Pb export is constrained by supply of DOC and related
4 soluble metal organic complexes or discharge quantities.

5 To tackle these questions we chose a bog, which is heavily contaminated by As and Pb
6 through historic mining activities and is known to exhibit high As and Pb concentrations in
7 the acrotelm. We continuously monitored direct bog discharge at a first order stream to gain
8 information of the DOC, Pb and As export dynamics dependent on precipitation. To combine
9 discharge generation and element concentrations with (pre-)conditions within the peat
10 catchment, we sampled pore water to investigate time- and depth resolved DOC and metal
11 release patterns and monitored temperature, precipitation and water level at the bog. For an
12 estimation of the overall export and mobilization potential from the bog we calculated fluxes
13 and peat As and Pb inventories. To examine the importance of storm flow events we
14 conducted high resolution measurements occasionally during snowmelt and rain events in
15 spring and fall.

16

17 **2 Materials and Methods**

18 **2.1 Study Area**

19 The Odersprung bog (OS, 52°46.383'N, 10°33.816'E; 800 m asl, 1500 mm mean annual
20 precipitation and 5°C mean annual temperature) is an ombrotrophic peatland located within
21 the nature protection area in the Harz Mountains in Northwestern Germany. The treeless part
22 of the bog covers an area of about 17 ha and has a mean peat depth of 3 m (Beug et al., 1999).
23 Vegetation is dominated by *Sphagnum magellanicum* and *Sphagnum rubellum*. *Eriophorum*
24 *angustifolium* and *Molina caerulea* occur less frequently (Baumann, 2009). The bog is

1 drained by a small erosion rill, which originates within the bog. Discharge sampling was
2 conducted at the rill outflow of the bog where all water is exclusively received from the bog.
3 Former studies on the effect of past local mining activities in the Harz Mountains reported Pb
4 and As peat concentrations of up 2300 mg kg⁻¹ and 100 mg kg⁻¹, respectively, which is by a
5 factor of 10,000 (Pb) higher than background values (Biestler et al., 2012).

6 **2.2 Sampling and Field Measurements**

7 Seasonal discharge sampling was conducted from April 2013 to November 2013 covering the
8 time from snowmelt to begin of snowfall. Water samples were taken in a six day interval by
9 an automated water sampler (ISCO autosampler 3700), equipped with 0.5 L PE bottles and a
10 Teflon hose, which was automatically rinsed with sample water prior to each sampling.
11 Further grab samples as well as pH and electric conductivity measurements were conducted
12 every two to three weeks. For grab samples new 50 mL PE tubes were used and previously
13 rinsed with sample twice before sampling. For all sampling techniques blank controls were
14 run. Several storm flow events were sampled at high frequency (every 3 hours) over a period
15 of three days, respectively. At all sampled events the catchment was free of ice and snow.

16 Pore water samples at the Odersprung bog were taken using a suction sampler described in
17 Broder et al. (2012). In short, the sampler consisted of PE-sinter slides (5x0.5x1 cm) inserted
18 in a 3.5 m long rod. Slides were connected by tubing to a stop-cock above the peatland
19 surface. Samples were taken by means of PE syringes and transferred to new PE tubes, which
20 were rinsed with sample before. Sampling resolution was 20 cm until 60 cm depth and 30 cm
21 until 210 cm depth. The suction sampler was installed within the catchment area of the
22 erosion rill (Fig. 1). Close to the pore water sampler a peat core was extracted using a Russian
23 peat corer (Eijkelkamp Agrisearch Equipment). Cores were collected down to 300 cm depth,
24 sliced in 2.5 cm sections, frozen and freeze dried. For the calculation of As and Pb inventories

1 in the acrotelm 15 short cores (30 cm depth) were sampled by means of a Wardenaar peat
2 profile sampler (Eijkelkamp Agrisearch Equipment).

3 Discharge at the bog outlet was quantified using a V-notch weir. Stage was recorded every 15
4 min by a water level logger (Odyssey dataflow systems) for the recalculation to the actual
5 discharge. Water level at the bog site was monitored in close proximity to the pore water
6 sampler using PVC piezometer tubes of 4 cm diameter, fully slotted until 120 cm depth and a
7 water level logger (Odyssey dataflow systems). Temperature, relative humidity and
8 precipitation (using a tipping-bucket rain gauge) were monitored (tinytag tgp 4500 and 4810,
9 Gemini) at the site.

10 **2.3 Laboratory Analyses**

11 Water samples were filtered with 0.45 μm nylon filters (Merck, Millipore) within the
12 laboratory and stored at 4°C. All water samples were analyzed for total dissolved organic
13 carbon by a thermo-catalytic total carbon analyzer (Analytik Jena multi N/C 2100S) using the
14 NPOC method. As, Fe and Pb concentrations were determined by an ICP-MS (Agilent 7700).
15 Instrumental drift and quality was checked by certified reference materials (SPS-SW 1 and
16 SLRS-4, riverine water, National Research Council, Canada). Detection limits for Fe, As and
17 Pb were 0.01 $\mu\text{g/L}$, 0.001 $\mu\text{g/L}$ and 0.001 $\mu\text{g/L}$, respectively.

18 All peat samples were freeze dried, milled and further analyzed with a resolution of 2.5 cm
19 until 50 cm depth and 10 cm resolution below. Peat As and Pb concentrations were
20 determined by an energy-dispersive miniprobe multielement analyzer (EMMA-XRF, see
21 Cheburkin and Shotykh, (1996)). Carbon and nitrogen concentrations in peat were measured
22 using an elemental analyzer (Euro EA 3000, HEKAtech).

23 **2.4 Flux calculations and As and Pb inventory estimates**

1 DOC, As and Pb annual fluxes were estimated by calculating the total annual load from
2 analyzed water samples and the continuous discharge record using method 5 according to
3 Walling and Webb (1985) and Littlewood (1992):

$$4 \text{ Load} = K \times Q_r \left(\frac{\sum_{i=1}^n C_i Q_i}{\sum_{i=1}^n Q_i} \right) \quad (1)$$

5 Where K is a conversion factor (here number of seconds per year), Q_r here the annual mean
6 discharge from the continuous record, C_i the instantaneous concentration, Q_i the instantaneous
7 discharge and n the number of samples ($n = 112$). Load estimates were divided by catchment
8 size to gain fluxes in $\text{g ha}^{-1}\text{a}^{-1}$.

9 The rill catchment area is mainly covered by an open and treeless part of the bog (8 ha, see
10 Fig. 1). Catchment boundaries were determined by topographic conditions and the occurrence
11 of perennial streams, which required including a part of a hillslope with shallow podsollic
12 forest soils. However, it was perceived that small pools build up between the domed bog and
13 forested areas, which drain downslope during rain events, and do not discharge into the bog
14 (see also Fig. 1). Therefore, the chemical characteristic from the bog stream should not be
15 biased by forest soil draining water. Flux calculations from high-frequency concentration
16 measurements during snowmelt and discharge events were calculated according to method 3
17 (Walling and Webb, 1985), where instantaneous loads were calculated by the measured
18 concentration and the integrated discharge data of the preceding sampling interval (3h) to
19 obtain a robust discharge estimate. To further improve flux estimates the annual record was
20 separated either by season or by hydrograph and was additionally calculated by method 2
21 (after Walling and Webb, 1985). Standard error and 95% confidence limits of flux
22 calculations were conducted after Hope et al. (1997). As and Pb inventories for the peatland
23 catchment were calculated from the median value of the upper 30 cm As and Pb inventories
24 of all 16 peat cores to estimate an inventory per unit area of the bog surface. Uncertainties of

1 the inventory calculations were estimated by the standard deviation from the mean of all As
2 and Pb inventories of the 16 peat cores.

3

4 **3 Results and Discussion**

5 **3.1 Hydrologic conditions**

6 The year 2013 was characterized by strong snow melt and several intense rain events (Fig. 2).
7 The precipitation record before day of the year (doy) 100 (mid of April) and after doy 289
8 (end of October) was affected by snowfall, which produces exceptionally high precipitation
9 values that distorts quantification as we used a tipping-bucket. At two other occasions the rain
10 gauge got clogged, thus for a few high flow discharges and water level rises no rainfall was
11 recorded (doy 150-162 and doy 256-270). During the snow-free record from mid-April to
12 mid-October the catchment received 537 mm rain. Monthly precipitation totals recorded
13 highest rainfall in fall, while a longer dry period prevailed in summer.

14 In total, 36 discharge events with high flow could be distinguished after the end of snowmelt.
15 High flow events partly followed shortly after a preceding event, especially in fall where rain
16 was frequent. The two main discharge events of the 2013 record triggered by a longer rain fall
17 period, occurred in spring and peaking at a discharge of 44 L s^{-1} and 55 L s^{-1} . The three
18 largest events in fall peaked at $23\text{-}24.5 \text{ L s}^{-1}$. High discharge values at the beginning of the
19 record indicate that snowmelt contributed an important portion to the annual discharge (Fig.
20 2). As a quantification of the complete snowmelt discharge was impossible, calculated annual
21 element fluxes might be underestimated (Dyson et al., 2011). A typical event hydrograph
22 exhibits a steep rising and falling limb with a slow decline as the event subsides. Setting a low
23 flow limit at 0.3 L s^{-1} by thorough visual hydrograph separation, low flow occurred during
24 more than 28% of the time record but contributed only 2.5% of total discharge. This is

1 indicative of an event runoff regime with negligible groundwater contribution. In fact,
2 constant base flow over longer periods could never be observed, as flow steadily declined
3 without rain fall until it almost ceased when the accessible water pool in the acrotelm was
4 depleted. Therefore, we do not refer to base flow conditions, but rather use the term low flow
5 as no steady groundwater component seem to contribute to the bog's discharge. The flashy
6 hydrograph response fits to a rapid runoff generation by surface-near and surface flow. The
7 dominance of this kind of runoff generation is similar to previously reported flashy regimes at
8 blanket bogs (Evans et al., 1999; Holden and Burt, 2003).

9 Lag times from the beginning of rainfall to discharge were short with a mean response time of
10 $1.3 \text{ h} \pm 0.5 \text{ h}$. Shortest lag times occurred when discharge was still high at the beginning of a
11 new high flow event similar as has been observed for a blanket bog (Daniels et al., 2008).
12 This instantaneous response indicates that at wet preconditions surface runoff prevails as this
13 is the most rapid runoff component.

14 Despite high rainfall events in summer, only small corresponding discharge events were
15 recorded. Here, the water pool of the bog's surface-near layer is recharged before discharge
16 takes place. The process is clearly depicted in Fig. 2 where the water level rose almost
17 immediately after beginning of rainfall at day 205. Over the entire record the water table
18 ranged between 3.5 cm above and 36 cm below the peat surface with a median depth of 10.5
19 cm. The lowest water level occurred during the summer drought (36 cm depth) after 20 days
20 without precipitation.. From the 36 discharge events 22 could be classified as following wet
21 and 13 as following dry preconditions based on the median water level of 10.5 cm as
22 threshold value. Estimated runoff to precipitation ratios for discharge events were lowest for
23 those events with exceptionally dry preconditions following the summer drought. In general,
24 dry precondition events exhibited lower runoff to precipitation ratios than wet precondition
25 events.

1 **3.2 DOC, As and Pb stream concentrations**

2 DOC release patterns from the Odersprung bog were similar than those reported from other
3 peatlands (e.g. Worrall et al., 2002; Laudon et al., 2004; Clark et al., 2008). Lowest DOC
4 concentrations (5 mg L^{-1}) were observed in the middle of April during snowmelt when
5 dilution was highest, whereas highest DOC concentrations ($\sim 30 \text{ mg L}^{-1}$) occurred in late
6 summer to fall when DOC production was highest and mostly low flow conditions occur (Fig.
7 2). DOC concentration decreased thereafter until begin of winter and the end of the annual
8 record. This general seasonal trend with lower concentrations in winter and spring and
9 increasing concentrations during summer can be attributed to reasons unrelated to discharge,
10 but to the seasonal temperature dynamics and therefore, enhanced biological activity, organic
11 matter decomposition and solubility. The low DOC concentrations during snowmelt can be
12 attributed to a dilution effect, but also to a smaller extent to the low DOC production and
13 storage during winter (Dyson et al., 2011).

14 Concentrations of As ranged from $0.2 \text{ } \mu\text{g L}^{-1}$ to $1.9 \text{ } \mu\text{g L}^{-1}$, whereas Pb concentrations were
15 significantly higher ($1.3 \text{ } \mu\text{g L}^{-1}$ to $12 \text{ } \mu\text{g L}^{-1}$) and in some cases even exceeded the WHO
16 threshold value for drinking water ($10 \text{ } \mu\text{g L}^{-1}$; WHO, 2011). Highest Pb concentrations
17 occurred during a rain event in fall, which is discussed further below. Concentrations of As
18 showed the same seasonal trend as observed for DOC with highest concentrations in summer
19 (Fig. 2). This trend was recognizable for Pb as well, but less pronounced (Fig. 2). Both, As
20 and Pb concentrations significantly correlated with DOC (r^2 : 0.96 and 0.87, respectively),
21 which supports the assumption that both elements are mainly transported along with DOC
22 within and out of this organic-rich system. Pb is known to strongly bind to organic matter
23 (Tipping, 1998). As forms soluble DOM-complexes (Buschmann et al., 2006), but also shows
24 sorption to iron(hydr)oxides or formation of As-Fe-NOM colloids or complexes (Ritter et al.,
25 2006; ThomasArrigo et al., 2014). Accordingly, Fe concentrations significantly correlate with

1 DOC and As (Fig. 1; r^2 Fe/DOC: 0.86; r^2 Fe/As: 0.80). At the bog's outflow, acidic conditions
2 (< pH 4.5) prevail over the whole sampling period. Under these conditions iron-DOM
3 complexes such as a ternary complex of As-Fe-DOM, dominate the soluble fraction (Tipping
4 et al., 2002; Lofts et al., 2008; Neubauer et al., 2013). Exports of As, Fe, and DOC thus seem
5 to be strongly linked and it is likely that both elements are mobilized by the same processes
6 and originate from the same source area within the bog.

7 **3.3 DOC, As and Pb concentrations during storm flows**

8 Three high frequency samplings of storm flows were conducted at the bog outlet in 2013. The
9 first spring event started after a longer low flow period and a low water level within the bog
10 of around 13 cm below surface, indicating dry preconditions (doy 137; 18/5/13; Fig. 3). The
11 second event followed shortly after the first, but with wet preconditions (doy 150; 31/5/13).
12 The third sampled storm flow event occurred in fall with again dry preconditions and a water
13 level of around 12 cm below surface (doy 251; 9/9/13; Fig. 3). In general, concentration
14 changes of all elements during high discharge events were rather small compared to the
15 annual variations indicating that high precipitation and increased runoff do not necessarily
16 cause dilution, but seem to mobilize or connect additional pools of DOC, As and Pb.

17 At the first event, DOC and As concentration started to decrease immediately after the onset
18 of high flow from 26.8 mg L⁻¹ and 1.5 µg L⁻¹ to 21.8 mg L⁻¹ and 1.0 µg L⁻¹, respectively at the
19 end of the event. No flushing effect with increasing stream water concentrations was visible
20 (Fig. 3). Pb concentrations however, immediately increased with rising water level and
21 discharge. Water levels rose from 13 to 6 cm depth, while Pb concentrations increased from 5
22 µg L⁻¹ to 7 µg L⁻¹. With the falling limb of the hydrograph Pb concentrations also decreased
23 to concentrations of about 4 µg L⁻¹. The reason for the different behavior of Pb compared to
24 As and DOC is unknown. The increase in Pb concentrations with increasing water level

1 indicates that there is no dilution but that additional Pb pools are mobilized as surface-near
2 peat layers get hydrologically connected to discharge. It is still unknown how the Pb pool
3 evolves during the winter months, probably by mineralization of organic matter in the
4 acrotelm. In case of DOC concentrations, which did not increase with increasing water level,
5 we assume that a potentially mobile DOC pool had not yet been built up in spring when
6 temperature and thus biological productivity were still low. The role of redox induced
7 formation or dissolution of mineral Fe-phases for As mobilization during water level changes
8 is not yet understood. One explanation could be that As and DOC, as well as As-organic
9 complexes are adsorbed to precipitated Fe-oxides when water levels are low. The re-
10 establishment of anoxic conditions after water level rise causes a reductive dissolution of Fe-
11 oxide and a release of DOC and As. However, these processes are assumingly too slow to be
12 effective within a storm flow event (Grybos et al., 2007). This hypothesis will be further
13 discussed in the pore water section further below.

14 The second storm flow event exhibited low concentrations compared to the annual
15 concentration record at this site, starting with concentrations of 13 mg L^{-1} DOC, $2 \text{ } \mu\text{g L}^{-1}$ Pb
16 and $0.7 \text{ } \mu\text{g L}^{-1}$ As, when low flow conditions had just re-established after a greater rain event.
17 During the low flow regime, concentrations slightly increased until the onset of the high
18 discharge event. DOC and As concentrations peaked at 18 mg L^{-1} and $1 \text{ } \mu\text{g L}^{-1}$, respectively at
19 the onset of the event, while Pb concentrations only peaked 3 hours later coinciding with
20 maximum flow and a water level rise from 9.5 cm to 5.6 cm depth. Regardless of further
21 smaller discharge peaks, DOC, As and Pb stream concentrations decreased to former low
22 flow concentration levels until the end of the discharge event. Even though a distinct rise in
23 concentrations of DOC, As and Pb caused by this rain event was evident, concentration
24 ranges for all three elements were low and of the same magnitude as lower concentrations
25 before the first event in May. This indicates a strong depletion of all three elements through

1 the directly preceding event, which exhibited the highest recorded discharge of the whole
2 record. The obvious exhaustion of potentially mobile DOC, As and Pb here, induces an
3 element export constrained by production rates. As low flow established again after this rain
4 event, concentrations started to slowly rise again as it has been observed under low flow just
5 before the rain event.

6 At the fall event concentrations before the event were higher than in spring with 33.5 mg L^{-1}
7 DOC, $8 \text{ } \mu\text{g L}^{-1}$ Pb and $1.5 \text{ } \mu\text{g L}^{-1}$ As. During this event, DOC, Pb and As behave differently
8 (Fig. 3). DOC concentrations first decreased at the onset of the rising limb of the hydrograph
9 and rising water level within the bog, but thereafter DOC concentrations started to rise until
10 discharge peaked (Fig. 3) and remained high at 35 mg L^{-1} to 37 mg L^{-1} until discharge
11 increased even further. Unfortunately, the automated sampler failed to sample a part of the
12 following main event, but DOC concentrations after the event were again much lower (27 mg
13 L^{-1} , day 255) than concentrations before the whole event. The As concentration dynamics
14 during this fall event followed that of DOC with a peak concentration of $1.9 \text{ } \mu\text{g L}^{-1}$. For Pb,
15 the first decrease in concentration was not apparent, but an immediate rise to higher
16 concentrations and a second discharge peak resulted also in distinctly higher Pb
17 concentrations of up to $11.9 \text{ } \mu\text{g L}^{-1}$. After the event As and Pb concentrations were slightly
18 lower than before the event ($6 \text{ } \mu\text{g L}^{-1}$ Pb and $1.2 \text{ } \mu\text{g L}^{-1}$ As). In line with the different
19 discharge peaks, the water level responded immediately to rain fall with a stepwise rise up to
20 a level of 1 cm above the peat surface. DOC and As concentrations evidently peaked shortly
21 after the discharge peak and even decreased with increasing discharge, which can be
22 indicative of a fast runoff component with less intense DOC and As mobilization or dilution.
23 However, Pb concentrations were not diluted, but increased with first discharge peak, which
24 is contradicting the pattern observed for DOC and As. Similarly, Rothwell et al. (2007) also
25 observed variable peak dynamics for Pb amongst other metals over several fall discharge

1 events. Summarizing the response of Pb stream water concentrations for all three rain events,
2 Pb concentrations seem to respond immediately to water level rise within the bog with an
3 increasing in concentrations. As fast surface runoff at the beginning of the event can be ruled
4 out as a reason for the initial decrease in As and DOC stream concentrations, the different
5 behavior of the three elements indicates that DOC and As seemed to be controlled by
6 different mobilization processes than Pb. Concentrations of DOC, As and Pb thus do not
7 respond entirely similar to rain events. This difference in the dynamics cannot be explained
8 based on our data set and prompts the question why the strong correlation of DOC with As
9 and Pb concentrations observed for the annual low resolution record is not valid for the rain
10 events.

11 The weak flushing effect during the spring events indicates a lack of supply of DOC and As
12 caused by a low biological productivity early in the year. The stronger flushing effect of Pb
13 will be discussed further below in relation to the results of pore water analysis. The fall event
14 followed after the summer dry period and the time of highest productivity. When bog water
15 level rises to the surface, the entire previously aerated peat layer gets hydrologically
16 connected and contributes to near-surface flow. Therefore, a much greater amount of DOC,
17 As and Pb can be mobilized and exported by such discharge events resulting in the highest
18 discharge concentrations of all three elements of that year. High frequency sampling revealed
19 a great dependency of DOC, As and Pb discharge concentrations on pool exhaustion within
20 the bog. This is evident by decreasing concentrations during an event, but also a decline in
21 DOC, As and Pb concentrations, when rain events follow up over a short time interval. This
22 exhaustion effect can be explained by a lack of supply of readily mobile element pools, when
23 peat layers are hydrologically connected over a longer time period and constant bog water
24 levels. These results are congruent with Rothwell et al. (2007), who sampled subsequent

1 discharge events at a bog draining stream in fall. In addition, the observed high concentration
2 levels of DOC, As, and Pb highlight the role of seasonal dynamics of productivity.

3 Figure 4 displays an element concentration to discharge (c/Q) plot for DOC, As and Pb. Here,
4 it is apparent that above a threshold of discharge of about 8 L s^{-1} concentrations of DOC, As
5 and Pb decrease or remain at constant. This indicates that at a discharge above 8 L s^{-1} element
6 concentrations get diluted, most likely because all available pools have been connected. As
7 this threshold is distinct and uniform for all three elements, pool exhaustion might be less
8 likely, as the pool size is expected to differ over the vegetation period with a longer supply in
9 summer and fall. Also, mobilization processes are probably different for DOC, Pb and As
10 depending on binding types, which result in different responses depending on moisture and
11 temperature preconditions of discharge events. However, rain water concentrations of DOC,
12 As and Pb are much lower than pore water concentrations and hence dilution by precipitation
13 should affect those elements all in the same way, as was observed here. Assuming that surface
14 runoff takes place after saturation of the entire peat layer, the particular discharge at which
15 surface runoff commences and dilution takes place should be similar over the whole year.
16 This conclusion also implies that at this bog, surface runoff is generated after peat saturation
17 and not through infiltration excess.

18 **3.4 Peat decomposition and solid phase As, Pb and Fe concentrations**

19 Solid phase Pb and As concentration exhibited a similar general trend with depth with higher
20 contents in the uppermost meter (Fig. 5 A). Pb concentrations peaked in 40 and 72.5 cm depth
21 with $1,200 \text{ mg kg}^{-1}$ and 706 mg kg^{-1} , respectively. These two peaks has been reported before
22 for other bogs in the Harz Mountains (Biester et al., 2012) and can be clearly related to
23 mining activities in the past. Kemptner and Frenzel (2000) related those peaks to regional
24 mining activities in the 12th/13th century and 17th century by ^{14}C dating in a bog within 5 km

1 distance from our sampling site. Due to the strong historic mining influence the recent
2 decrease in Pb concentrations in the uppermost cm was very pronounced. Also below 75 cm
3 Pb concentrations sharply decreased to non-anthropogenic background levels and $< 20 \text{ mg kg}^{-1}$
4 below 100 cm depth. Concentrations of As were also highest in the uppermost 75 cm and
5 show peaks at similar depths as Pb with maximum concentrations of 65 mg kg^{-1} and 38 mg
6 kg^{-1} at 30 cm and 72.5 cm depth. Peak concentration of Pb and As were high, even compared
7 to other contaminated sites (e.g. Rothwell et al., 2009). While Pb is no redox-sensitive
8 element, As might be mobilized after deposition and gets enriched in the surface-near layer
9 due to redox changes, especially, when strong water level fluctuations prevail (Rothwell et al.,
10 2010). As As(V) associates with Fe-(hydr)oxides, As mobility is known to be controlled by
11 the reductive dissolution of Fe-(hydr)oxides, which causes the release of As into pore waters
12 and would lead to an enrichment in the surface-near layer along with precipitated Fe-
13 (hydroxides). Fe enrichment at the redox boundary occurs through the upward diffusion of
14 dissolved Fe(II) and precipitation where oxic conditions prevail. The Fe depth profile here
15 showed enrichment in the uppermost peat layer with a peak concentration of 5.5 g kg^{-1} at 5
16 cm depth and decreasing concentrations down to 1.3 g kg^{-1} at 32.5 cm depth. Lowest Fe
17 concentrations were found below 100 cm depth with concentrations less than 1.3 g kg^{-1} . In
18 general Fe concentrations were low compared to other peatlands (Riedel et al., 2013) and
19 much lower than in studies describing Fe as an important factor for DOC and As retention.
20 Concentrations of As showed no enrichment in the upper peat layer as did Fe, indicating that
21 As is not coupled to redox induced changes of Fe phases here. We assume that As in our bog
22 is predominately bound to organic matter similar as observed in other peatlands, where As is
23 mainly bound to organic matter by reduced organic sulphur groups under reducing conditions
24 (Langner et al., 2011).

1 In previous studies peat decomposition has been found to enrich particular element
2 concentrations through mass loss (Biester et al., 2003; Biester et al., 2004; Biester et al.,
3 2014). Figure 5 A displays C/N ratios of the organic matter as proxy for peat decomposition
4 (Kuhry and Vitt, 1996; Broder et al., 2012; Biester et al., 2014). In the upper 12.5 cm C/N
5 ratios were high (88-128) with a distinct decrease to a value of 54 in 15 cm depth (Fig. 5 A).
6 Further below, C/N ratios remained low between 23 - 50 with higher values between 70 cm
7 and 150 cm (50-100). This indicates that the upper 15 cm are less decomposed, while with
8 decreasing C/N with depth the degree of decomposition increases with again a lower degree
9 of decomposition between 70 - 150 cm depth. A straight influence of decomposition on As
10 and Pb distribution was not found here and concentrations were rather determined by
11 enhanced deposition rates or in case of Fe by redox processes than by peat decomposition.

12 **3.5 Pore water DOC, As and Pb concentrations**

13 DOC concentrations in the pore water profile ranged from 20 mg L⁻¹ to 250 mg L⁻¹ (Fig. 5).
14 Highest concentrations were found at the down most sample in 225 cm depth. DOC
15 concentrations in the upper 50 cm, where most discharge is generated, were low in spring
16 under wet conditions (doy 147: 34-41 mg L⁻¹; Fig. 6) and much higher in fall after rewetting
17 following the dry summer period with concentrations of 119 mg L⁻¹ in 20 cm depth and 100
18 mg L⁻¹ in 40 cm depth (Fig. 6, doy 254). In late fall concentrations decreased again, but were
19 still much higher than in spring. This is in line with the recorded seasonal trend of DOC
20 concentrations at the discharging stream and the fact that DOC stream concentrations are
21 dependent on temperature and microbial activity to built up potentially mobile pools (Clark et
22 al., 2005), but also to the hydraulic connectivity of the surface-near peat layers. Normally,
23 within a couple of weeks after rewetting acidity is consumed in pore waters through anaerobic
24 respiration processes, like sulfate- or iron reduction (Fenner and Freeman, 2011). This should
25 favor DOC solubility through higher pH, but also by release of formerly Fe-(hydr)oxid-bound

1 DOM (Grybos et al., 2009). Moreover, anaerobic conditions suppress peat mineralization and
2 the described enzymatic latch of phenol oxidase activity favors DOC production after
3 rewetting (Fenner and Freeman, 2011). In line with these assumptions, DOC concentrations
4 were highest several weeks after the summer drought in fall, when the water level had fully
5 recovered again (doy 254, Fig. 6). However, the lower DOC concentrations measured during
6 this rewetting, at an intermediate water level depth of 14 cm, probably reflect a limitation in
7 DOC supply and an exhaustion of the DOC pool through export, as the sampling just
8 followed two rain events, which already had triggered high DOC stream concentrations.

9 During the summer drought, the amount of pore water was too low to obtain sufficient sample
10 volume for DOC measurements at 20 cm depth. At 40 cm depth DOC concentrations during
11 drought and following rewetting (doy 254) were higher than in spring and during rewetting
12 (doy 236). On the one hand, this might contradict the concept of Fenner and Freeman (2011)
13 or Clark et al. (2012) who stated low DOC concentrations during drought due to microbial
14 limitations by drought stress and acidification by drought-induced oxidation processes. On the
15 other hand, the measured low pH (< 4.5 pH), in all pore water samples at our site might
16 indicate that the drought acidification effect is not pronounced here, as pH in our peat is
17 generally low. Moreover, due to the low Fe concentrations in our peat, the effect of DOC
18 immobilization through binding to Fe-(hydr)oxides during drought and aeration as proposed
19 elsewhere (Riedel et al., 2013) is probably low at our study site. Accordingly, due to the low
20 pH and the low amount of Fe-(hydr)oxides at our site suppression of DOC production by
21 drought events seems to be of low importance here and probably in ombotrophic peatlands in
22 general.

23 Pore water As and Pb concentrations in the Odersprung bog ranged from 1.2 $\mu\text{g L}^{-1}$ to 3.8 μg
24 L^{-1} and 0.5 $\mu\text{g L}^{-1}$ to 8.4 $\mu\text{g L}^{-1}$, respectively (Fig. 5). Both, As and Pb concentrations were
25 highest in the uppermost sample at 20 cm depth. While As was steadily decreasing and

1 leveled out in 70 cm depth, Pb concentrations peaked again around 60 cm depth with 4-5 μg
2 L^{-1} Pb before it leveled out in about 120 cm depth. Concentration profiles throughout the year
3 show only small changes in the uppermost samples. Lowest As pore water concentrations at
4 20 cm and 40 cm depth occurred at the first sampling in spring (Fig. 6, doy 136) with 2.5 μg
5 L^{-1} and 2.1 μg L^{-1} , respectively. At the end of the summer drought As concentrations
6 remained at around 3 μg L^{-1} (doy 203), while Pb pore water concentrations decreased from
7 about 6 μg L^{-1} (20 cm depth) and 3 μg L^{-1} (40 cm depth) to lowest annual concentrations of
8 3.3 μg L^{-1} and 1.7 μg L^{-1} Pb during summer drought, respectively. While highest Pb
9 concentrations (8.4 μg L^{-1}) were measured at the beginning of fall, (Fig. 6, doy 236) when the
10 water level had not yet fully recovered, highest As concentrations occurred after the complete
11 rewetting of the bog in September congruently with DOC concentrations (doy 254). Pb
12 concentrations at 40 cm depth were constant with exception of a slight decrease during
13 summer drought (doy 311: ~ 3 μg L^{-1}). The very high Pb concentrations measured during
14 water table recovery might be due to mobilization of a Pb pool, which built up during drought
15 in the most reactive surface-near peat layer by microbial decomposition or mineralization
16 processes. This assumption might also explain the decoupling of DOC and Pb export apparent
17 at discharge events after dry preconditions, as a readily solubilized Pb pool is easily flushed.
18 Moreover, it explains the absence of the initial dilution effect as seen in DOC stream
19 concentrations. Furthermore, the DOC pool increases after drought, when wet conditions
20 prevail again, explaining the lag time of DOC peak concentrations in pore waters.

21 The decreasing Pb concentrations with depth allow a determination of the discharge water
22 source. The hydrologic response of the bog discharge was characteristic for a rapid near-
23 surface runoff. Comparing absolute concentrations in pore water and discharging water, the
24 high Pb concentrations of more than 10 μg L^{-1} in runoff at the fall rain event can only be
25 generated in the uppermost part of the bog, where Pb concentrations were highest. Following

1 this assumption also the low flow concentrations must have been generated in the upper part
2 of the peat profile as Pb concentrations below 1 m depth were too low to generate the
3 observed Pb concentrations in discharge. The variable Pb concentrations at different flow
4 conditions are attributed to different water levels, which affect variable parts of the Pb pool in
5 the acrotelm and are thus able to mobilize different amounts of Pb. This does also indicate
6 that, in line with the hydrologic discharge response, low flow discharge is generated from
7 upper peat layers and not from seeping water originating from the deep peat sections.

8 **3.6 Fluxes and inventory estimations**

9 While DOC and As concentrations were highest in summer and seemed to be more dependent
10 on temperature, DOC, As and Pb fluxes were not controlled by the concentration changes
11 over the year, but mainly by discharge (Fig. 7). This dependency is valid up to a discharge of
12 8 L s^{-1} , while at higher discharge dilution is observed. Highest element fluxes occurred during
13 high discharge events in spring and fall and during snowmelt. Flux calculations show that
14 10% of the monitoring time span, when highest discharge was recorded, contributed 39 %, 40
15 % and 43 % of the annual DOC, As and Pb export. Similar patterns have been described by
16 Koehler et al. (2009) for DOC fluxes from a blanket bog in Ireland. Clark et al. (2007)
17 calculated that 10 % of the monitoring time span including highest discharge contributed 50
18 % of the annual DOC export and Hinton et al. (1997) estimated 41-57 % of annual DOC
19 export by the 10 % time of highest discharge. When separating the annual hydrograph record
20 in storm flow and low flow conditions (boundary at 0.3 L s^{-1} by visual examination), 72 % of
21 the record exhibited high discharge conditions, which contributed 97.6 % of total annual
22 discharge and about 96 % of annual As, Pb and DOC export..

23 This flux calculations implicate that high discharge events contribute over-proportional to
24 element exports in a short time period. While low flow conditions generate high element

1 concentrations, greater discharge not primarily dilutes element concentrations, but connects
2 additional pools to discharge, which results in higher fluxes. Concluding, as different DOC,
3 As, and Pb pools, i.e. different peat layers, correspond to different hydrologic conditions, also
4 different linkages to element exports are likely. If low flow discharge is generated in the
5 lower acrotelm, not only general pool sizes are different, but also mobilization conditions, i.e.
6 different pH, redox, DOC quality. Further insight into differences between low flow and high
7 flow export dynamics might be gained through Pb isotope determinations (see Klaminder et
8 al. 2008), a determination of DOC age or DOC characterization.

9 The importance of high discharge events to Pb, As and DOC exports gets more relevant in
10 view of changing climate conditions with more frequent heavy rain events in this region.
11 Furthermore, that it is crucial at which time of the year rain events will occur in the future. A
12 data separation by season indicates that fall contributes to the largest extend to annual element
13 export followed by spring and similar ranges in summer and snowmelt (Table 1). This pattern
14 is in line with the assumption that higher microbial activity on the one hand, and elevated
15 discharge, which flushes the acrotelm on the other hand favors DOC, As and Pb export.. The
16 lower fluxes during summer are attributed to prevailing low flow, dry conditions. Figure 8
17 displays discharge to element loading plots separated by season and high discharge event
18 sampling. It highlights higher loadings at the same discharge volume dependent on season by
19 a steeper increase of the regression line. In general, element loadings are lowest at snowmelt
20 and increase over the year with highest loadings during the fall event. At the end of the year,
21 loadings decrease again. While this general trend is similar for DOC and As, Fig. 8 also
22 shows that Pb loadings during snowmelt and the second spring event are quite similar and that
23 concentrations are less dependent on discharge volume, indicated by lower R^2 values of the
24 regression lines.

1 The annual DOC export from the investigated bog catchment can be estimated to 155 kg C ha^{-1}
2 $\text{a}^{-1} \pm 67 \text{ kg C ha}^{-1} \text{ a}^{-1}$ (Table 2) and is similar to values from other studies (Worrall et al.,
3 2003; Koehler et al., 2009). As the upstream catchment is completely snow covered during
4 winter, including the stream outlet at the bog, no flux quantification during winter is possible.
5 It can be speculated though, that fluxes are probably low due to low temperatures and low
6 flow conditions during winter season.

7 Aqueous exports of As and Pb were calculated to $7.8 \pm 3.0 \text{ g As ha}^{-1} \text{ a}^{-1}$ and $39 \pm 16 \text{ g Pb ha}^{-1}$
8 a^{-1} . Annual Pb export was in the same range as values reported by Rothwell et al. (2011) with
9 annual aqueous Pb export of $55 \pm 18 \text{ g Pb ha}^{-1} \text{ a}^{-1}$ from a contaminated eroded ombrotrophic
10 peatland. The authors also highlighted the importance of particulate Pb export for peatland
11 systems, which needs to be considered when estimating total Pb export. The calculated
12 aqueous As export from our site were much lower than those reported by Rothwell et al.
13 (2011) ($47.1 \pm 9.9 \text{ g As ha}^{-1} \text{ a}^{-1}$), in spite of higher As contents in our peat. This might be due
14 to the eroded nature of their sampled bog and the prevailing post-depositional mobilization of
15 As in peat due accelerated water table drawdowns and a consequent binding to Fe-(hydr)oxides
16 (Rothwell et al., 2010).

17 As and Pb inventories were calculated based on the median element contents in the upper
18 30 cm of the 16 analyzed cores and were 0.91 g As m^{-2} ($0.64\text{-}1.17 \text{ g m}^{-2}$) and $13.98 \text{ g Pb m}^{-2}$
19 ($7.47\text{-}24.02 \text{ g Pb m}^{-2}$), respectively (Table 2). Element contents in the upper 30 cm varied
20 over a wide range pointing out the importance of a multi-core approach for calculating
21 peatland element inventories. Rothwell et al. (2010) calculated a storage of $0.19\text{-}0.44 \text{ g As m}^{-2}$
22 and $12.2\text{-}13.5 \text{ g Pb m}^{-2}$ in another contaminated peatland. While our calculated Pb inventories
23 are similar, As inventories are higher than those reported by Rothwell et al. (2010). When
24 referring annual fluxes to the bog area of the catchment $0.04 \text{ kg As a}^{-1}$ and 0.2 kg Pb a^{-1} were
25 exported in 2013 by the bog drainage. This equals to 0.85 % and 0.27 % of the calculated As

1 and Pb inventories respectively (Table 2). Even though conclusions based on a one year data
2 set are limited and uncertainties are high, it becomes clear that As seemed to be more mobile
3 than Pb. Neglecting future element deposition and changing export conditions it would take
4 more than 1000 years to deplete As pools and more than 3000 years for Pb pools stored
5 within this polluted peatland. Furthermore, the observed element content variation among all
6 cores might also implicate differences in pore water concentration levels within the stream
7 catchment.

8

9 **4. Conclusions**

10 Our results highlight the importance of comprehensive field studies to gain further
11 understanding in generation and controls of element exports from peatlands. The combination
12 of pore water and discharge data showed that As and Pb exports are not only dependent on the
13 amount of precipitation and discharge, but on the frequency and depth of water table
14 fluctuations, and the extend of pool connectivity in the acrotelm. This has been demonstrated
15 by higher As and Pb concentrations and exponential increase of export at high discharge
16 events, especially after a longer dry period and higher temperature. A distinct bog-specific
17 discharge threshold of 8 L s^{-1} was observed, which indicates a connection of all available
18 pools. This threshold presumably depends mainly on the size of the bog and drainage
19 conditions. Significant correlations of annual As, Pb and DOC concentrations in discharge
20 hints to transport of As and Pb as organic complexes with decoupling of Pb from DOC
21 concentrations at storm events following dry preconditions. Comparing the annual bog As and
22 Pb export with element inventories indicates that As is much stronger mobilized than Pb, with
23 annual fluxes accounting for 0.85 ‰ and 0.27 ‰ of total As and Pb inventory, respectively.
24 Results also point out the over-proportional contribution of high discharge events to As, Pb

1 and DOC annual export. The challenge for prospective research is to unravel the
2 biogeochemical effects of short term water level fluctuations on trace element and DOC
3 mobilization processes within peatlands to further improve parameterization of peatland
4 catchment models.

5

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11

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1 **Tables**

2 **Table 1:** Seasonal distribution of discharge, DOC, As and Pb export fluxes at the Odersprung
3 catchment.

| | Record time | Q | DOC export | As export | Pb export |
|----------|-------------|------|------------|-----------|-----------|
| | % | % | % | % | % |
| Snowmelt | 7.6 | 19.8 | 7.4 | 11.1 | 7.2 |
| Spring | 21.3 | 25.5 | 19.2 | 15.3 | 19.1 |
| Summer | 35.6 | 10.8 | 12.5 | 10.2 | 12.0 |
| Fall | 35.6 | 47.5 | 60.8 | 63.4 | 61.7 |

4

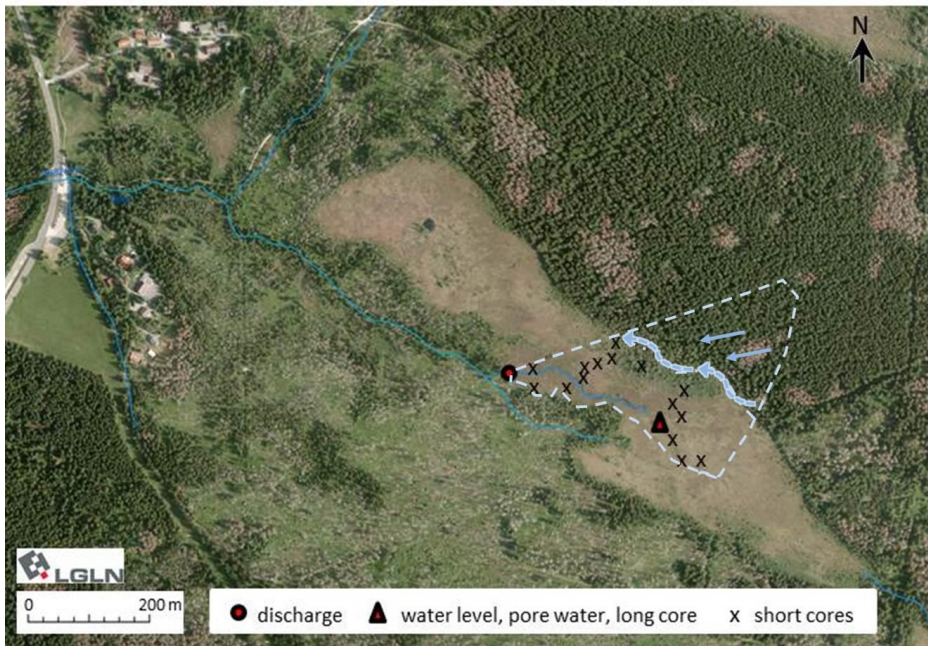
1 **Table 2:** Estimated mean DOC, As and Pb export in 2013 and element inventories in the
 2 upper 30 cm peat. Uncertainties were estimated by the standard deviation of flux calculations
 3 and element inventories calculated of each core.

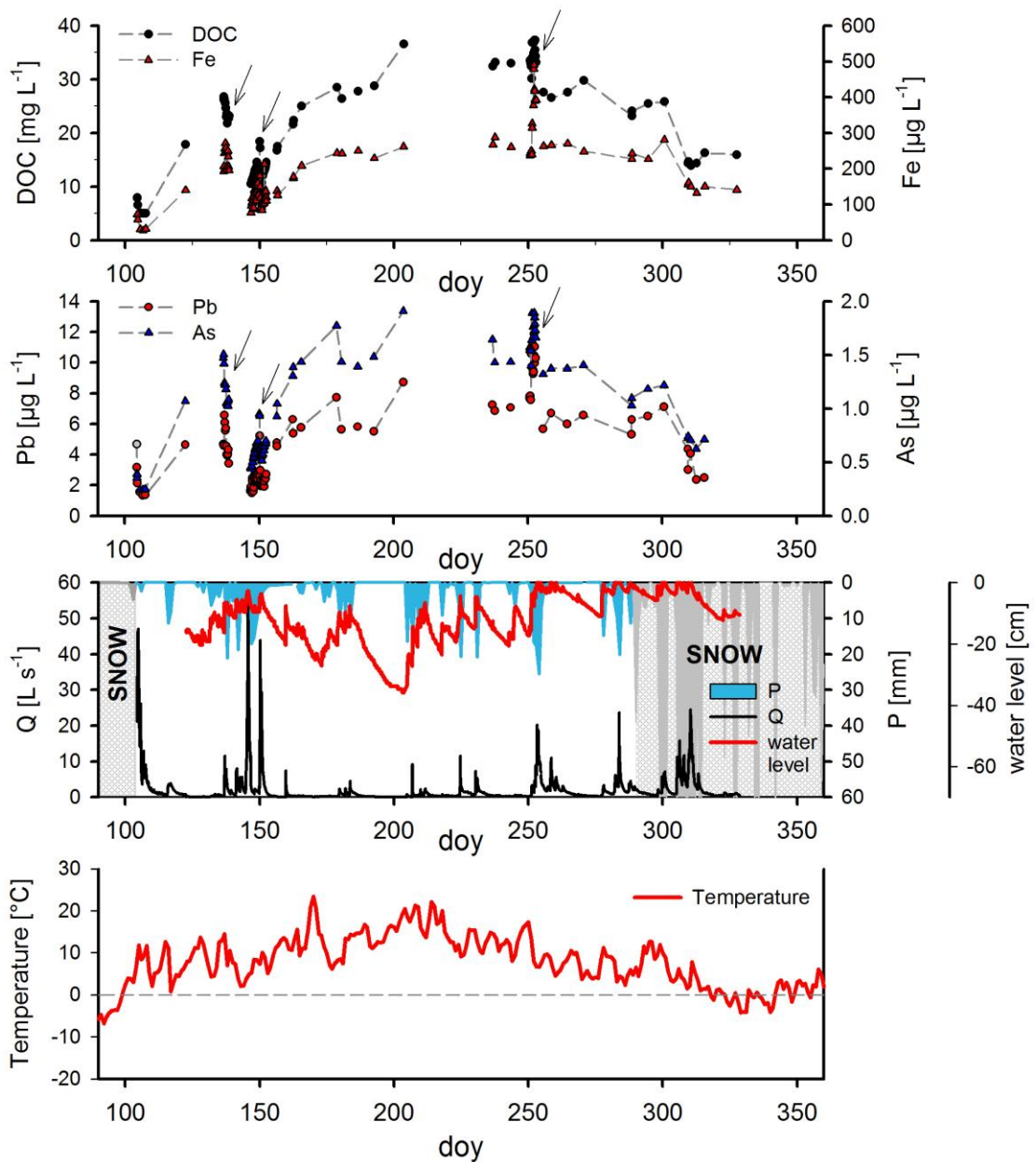
| | Mean annual fluxes | | Peat inventory ^a | | Export |
|------------------|------------------------------------|--------------------------------------|-----------------------------|--|----------------|
| | g ha ⁻¹ a ⁻¹ | kg bog ⁻¹ a ⁻¹ | g m ⁻² | kg ha ⁻¹ | ‰ of inventory |
| DOC ^b | 154.79*10 ³ | 773.93 | 16,689 | 166,89*10 ³ | 0.93 |
| uncertainty | 88-221*10 ³ | 441-1107 | (15,336-17,417) | (153.36*10 ³ - 174.170*10 ³) | (0.89-1.01) |
| As | 7.8 | 0.04 | 0.91 | 9.1 | 0.85 |
| uncertainty | 4.8-10.8 | 0.02-0.06 | (0.64-1.17) | (6.4-11.7) | (0.66-1.22) |
| Pb | 39.2 | 0.2 | 13.98 | 139.8 | 0.27 |
| uncertainty | 23.1-55.3 | 0.12-0.28 | (7.47-24.02) | (74.7-240.2) | (0.16-0.51) |

4 ^a upper 30 cm ^btotal C for peat inventory, respectively

5

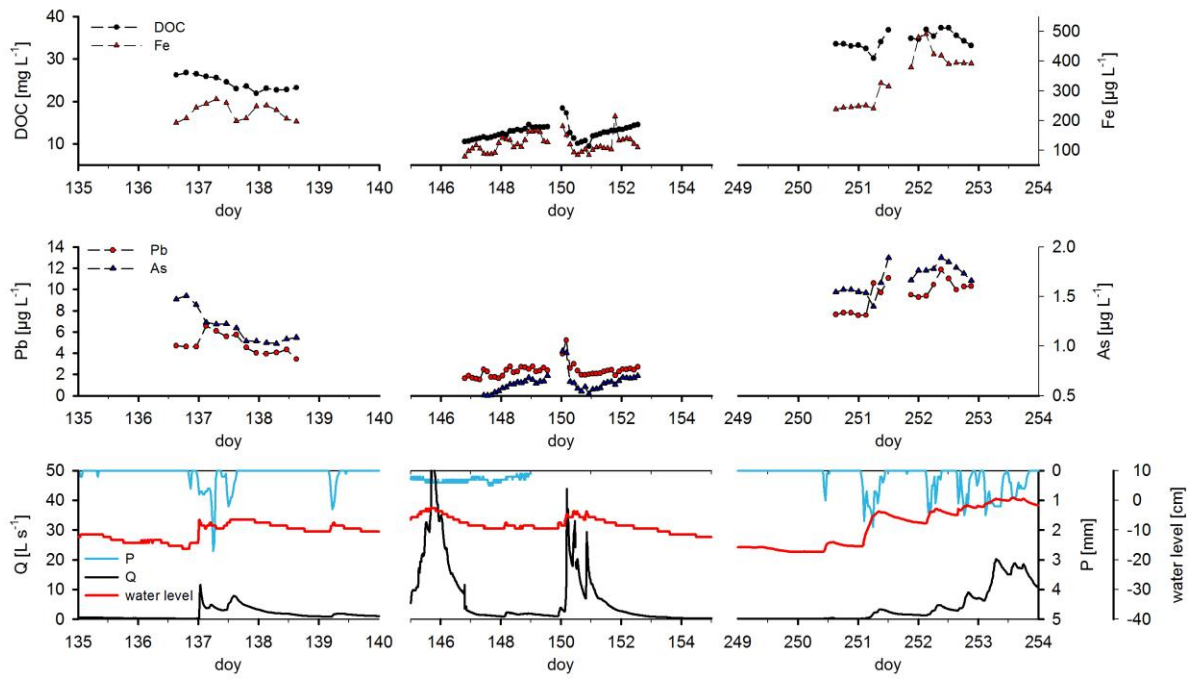
1 **Figures**





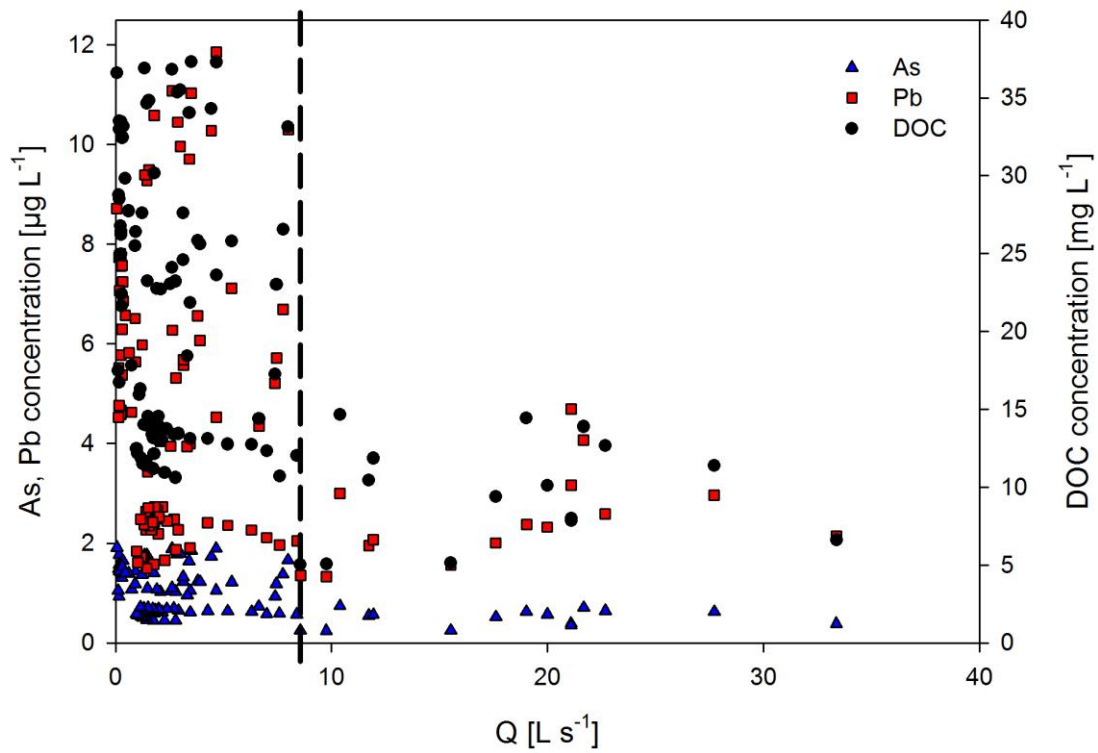
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2 **Fig. 2:** Annual concentration record of DOC, Fe, Pb and As (top) at the bog outlet. Annual
 3 records of bog water level, daily precipitation, discharge and temperature (low). Sampled
 4 discharge events are highlighted by black arrows. Note that winter precipitation as snow
 5 cannot be quantified.



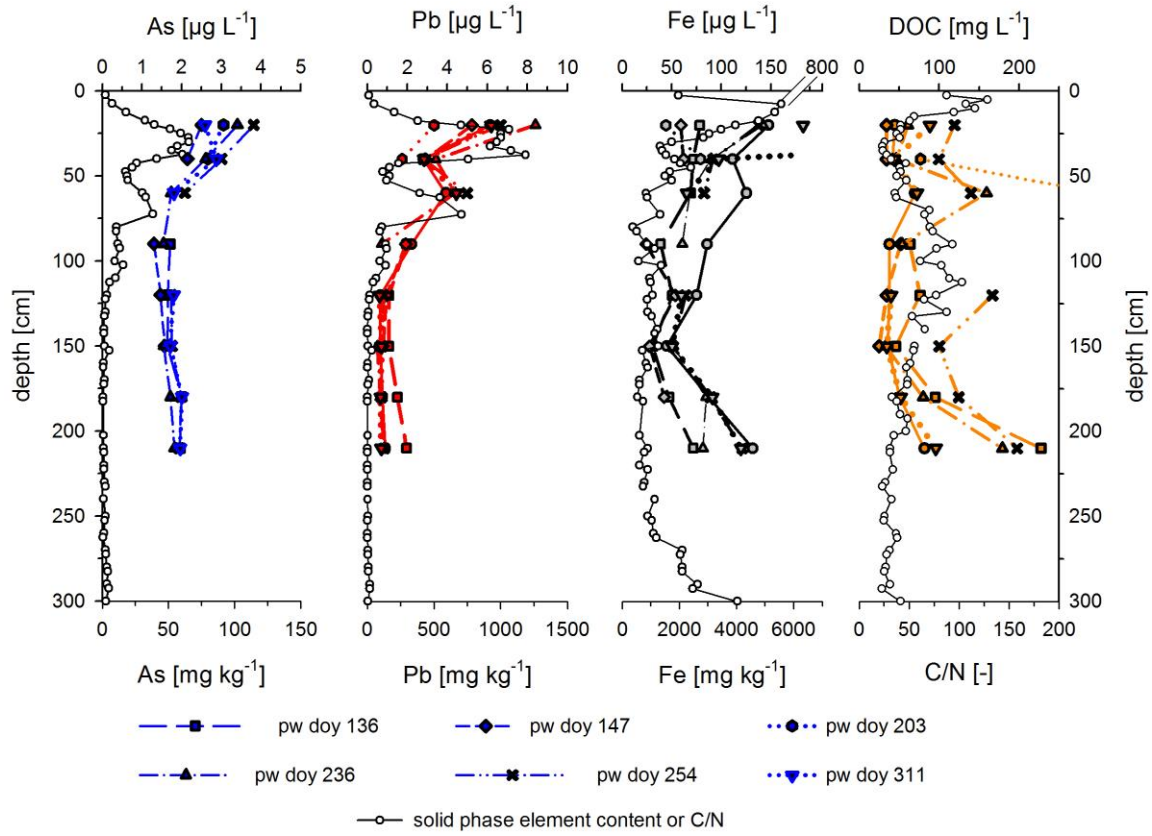
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2 **Fig. 3:** High resolution DOC, As and Pb concentrations during recorded spring (day 135-
 3 155) and fall (day 250-260) events. Records of bog water level, precipitation (30 min
 4 resolution) and discharge (low).



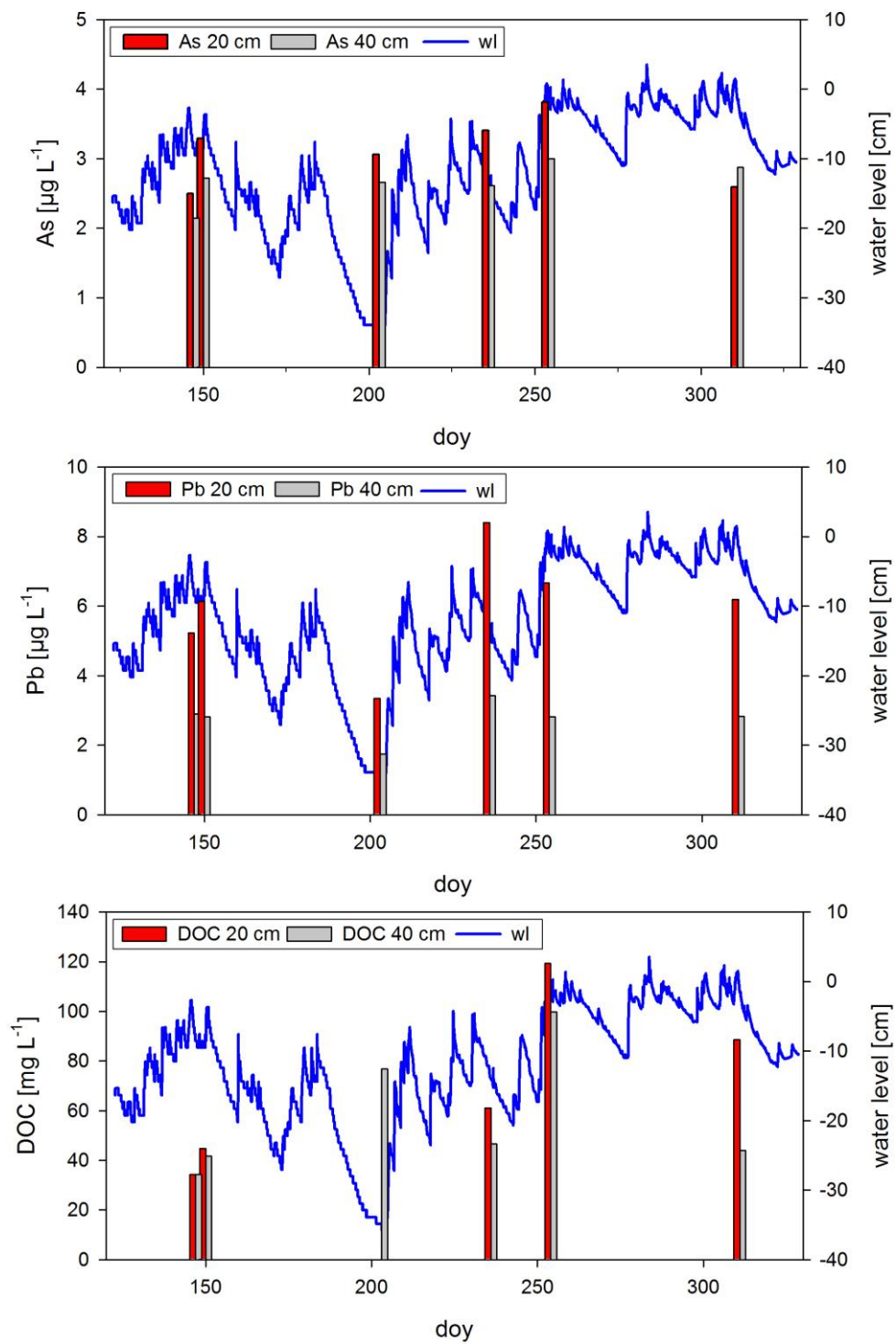
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2 **Fig. 4:** DOC, Pb and As stream concentration to discharge (Q) plot (c/Q). The black dashed
 3 line indicates the threshold discharge value for a compulsive dilution effect.



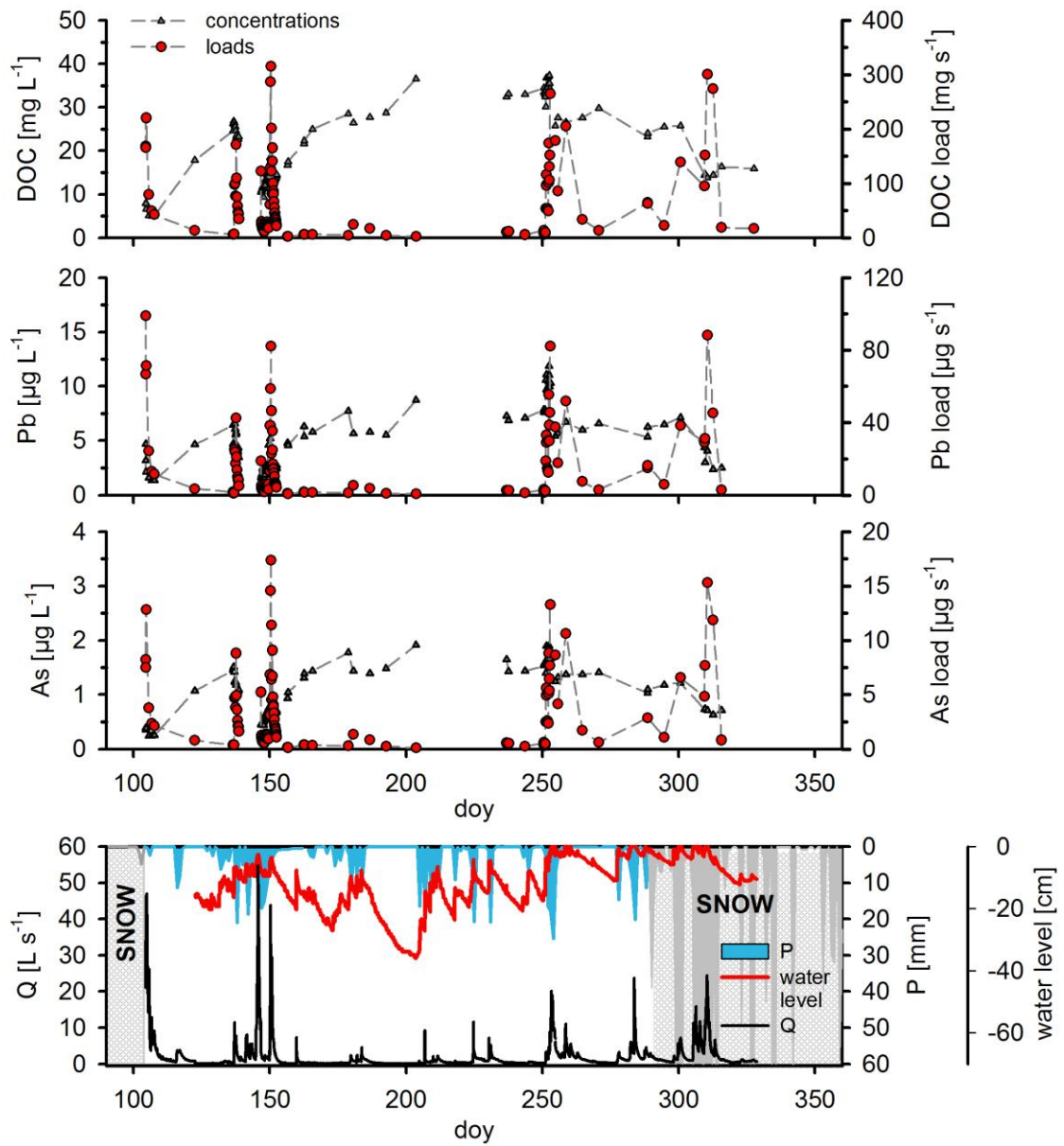
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2 **Fig. 5:** A) As and Pb contents of the long peat core taken at the Odersprung bog and carbon
 3 nitrogen ratios (C/N) as indicators for peat decomposition. B) Pore water concentrations of As
 4 (left, blue), Pb (middle, red) and DOC (right, orange) along a depth profile (250 cm).



1

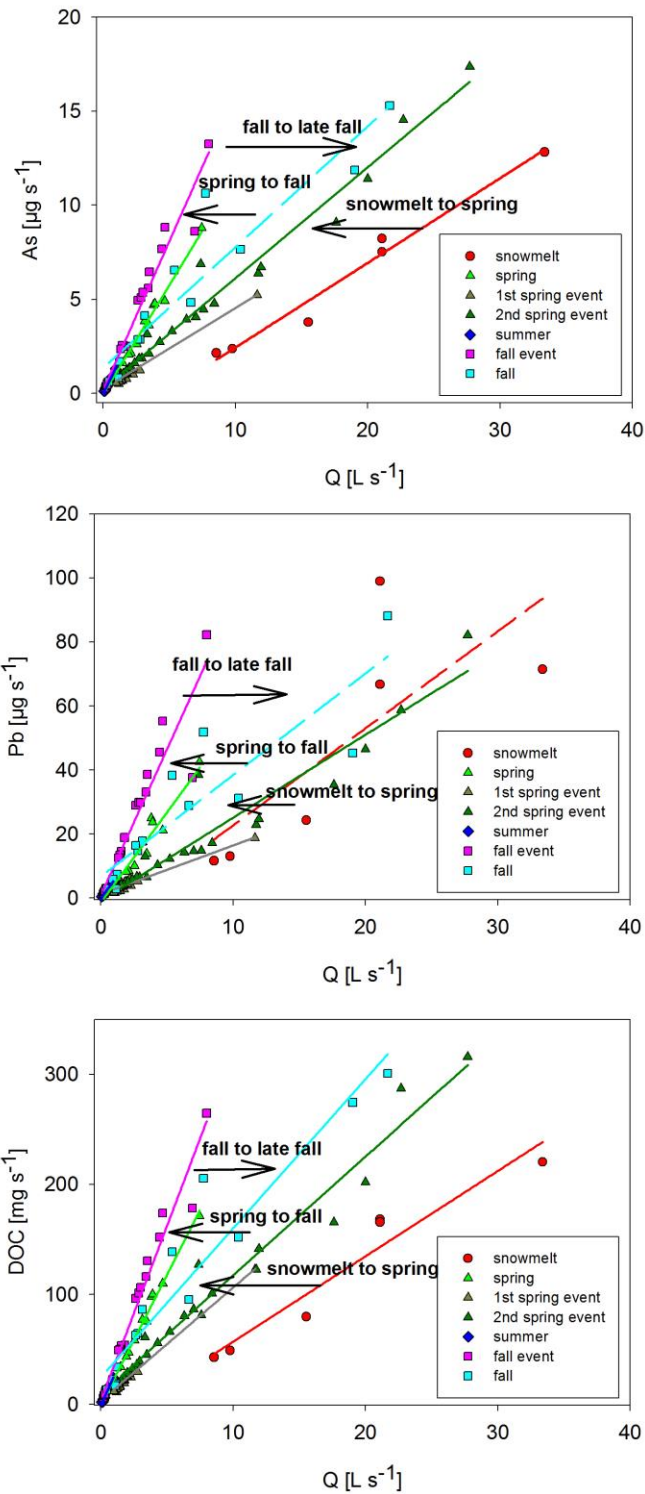
2 **Fig. 6:** Pore water concentrations of As (top), Pb (middle) and DOC (low) in 20 cm and 40
 3 cm depth over time. The blue line indicates the recorded bog water level (wl).



1

2 **Fig. 7:** Concentration record and instantaneous loads of DOC, Pb and As at the bog outlet.

3 Annual records of bog water level, precipitation and discharge (low).



1

2 **Fig. 8:** As (top), Pb (middle) and DOC (low) loadings to discharge (Q) plots divided by
 3 season and event sampling. Regression lines with $R^2 > 0.90$ are plotted with a solid line, $R^2 <$
 4 0.9 are plotted with a dashed line.