

1 **Microbial carbon recycling: an underestimated process**
2 **controlling soil carbon dynamics. Part II) a C₃-C₄ vegetation**
3 **change field labelling experiment**

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13

1 **Abstract**

2

3 The mean residence times (MRT) of different compound classes of soil organic matter (SOM)
4 do not match their inherent recalcitrance to decomposition. One reason for this is the
5 stabilisation within the soil matrix, but recycling, i.e. the reuse of “old” organic material to
6 form new biomass may also play a role as it uncouples the residence times of organic matter
7 from the lifetime of discrete molecules in soil.

8 We analysed soil sugar dynamics in a natural 30 years old labelling experiment after a wheat-
9 maize vegetation change to determine the extent of recycling and stabilisation by assessing
10 differences in turnover dynamics between plant and microbial derived sugars: While plant
11 derived sugars are only affected by stabilisation processes, microbial sugars may be subject to
12 both, stabilisation and recycling. To disentangle the dynamics of soil sugars, we separated
13 different density fractions (free particulate organic matter (fPOM), light occluded particulate
14 organic matter ($\leq 1.6 \text{ g cm}^{-3}$; oPOM_{1.6}), dense occluded particulate organic matter ($\leq 2 \text{ g cm}^{-3}$;
15 oPOM₂) and mineral-associated organic matter ($> 2 \text{ g cm}^{-3}$; Mineral)) of a silty loam under
16 long term wheat and maize cultivation. The isotopic signature of neutral sugars was measured
17 by high pressure liquid chromatography coupled to isotope ratio mass spectrometry
18 (HPLC/IRMS), after hydrolysis with 4 M Trifluoroacetic acid (TFA).

19 While apparent mean residence times (MRT) of sugars were comparable to total organic
20 carbon in the bulk soil and mineral fraction, the apparent MRT of sugar carbon in the oPOM
21 fractions were considerably lower than those of the total carbon of these fractions. This
22 indicates that oPOM formation was fuelled by microbial activity feeding on new plant input.
23 In the bulk soil, MRT of the mainly plant derived xylose were significantly lower than those
24 of mainly microbial derived sugars like galactose, rhamnose, fucose, indicating that recycling
25 of organic matter is an important factor regulating organic matter dynamics in soil.

26

1 **1 Introduction**

2 For several decades, it was assumed that the molecular structure accounts for the rate of
3 decomposition of different organic compounds in soils, i.e. compounds of high chemical
4 recalcitrance were assumed to be selectively preserved (Stevenson, 1994). However, the use
5 of compound specific isotope analysis provided new understanding of soil organic matter
6 (SOM) dynamics. As an example, lignin, a compound of high chemical recalcitrance, has
7 shorter mean residence times (MRT) than labile compounds like sugars or proteins (Amelung
8 et al., 2008; Gleixner et al., 2002; Kiem and Kögel-Knabner, 2003; Schmidt et al., 2011). The
9 main mechanisms for the long persistence of these labile compounds in soil are stabilisation
10 on the one hand, i.e. protection of organic matter from mineralization either by reduced
11 accessibility for microorganisms caused by physical protection (by mineral interaction or
12 occlusion within soil aggregates) or chemical recalcitrance (Six et al., 2002; Sollins et al.,
13 1996; von Lützow et al., 2006), and microbial recycling on the other, i.e. the reuse of “old”
14 organic compounds by microorganisms (Gleixner et al., 2002; Sauheitl et al., 2005). The latter
15 leads to an underestimation of the actual turnover dynamics but overestimates the persistence
16 of single molecules as a whole within the SOM. Although these different underlying
17 mechanisms have been proposed quite a while ago, their relevance in different soils and soil
18 horizons, especially concerning the importance of stabilisation versus microbial recycling,
19 still remain unclear. First studies on polar membrane lipids of microorganisms in marine
20 sediments suggest a strong underestimation of recycling in our current view on carbon
21 dynamics in soils and sediments (Takano et al. 2010). However, knowledge about soils
22 especially microbially active topsoils are still missing. Therefore, assessing the importance of
23 stabilisation and recycling for the persistence of organic matter in soil will improve the
24 understanding of the carbon cycle and close an important knowledge gap.

25 However, the pool of SOM is highly complex and intractable to analyse as a whole. Thus, we
26 examined the fate of sugars; an important compound class of the SOM that is involved in
27 almost all biological processes in soils, the MRT of which do not match their low biochemical
28 recalcitrance (Gleixner et al., 2002; Derrien et al., 2006; Derrien et al., 2007). Sugars in soils
29 are commonly classified according to their main origin into plant (arabinose (ara),
30 xylose(xyl)) or microbial derived sugars (galactose (gal), mannose (man), rhamnose (rha),
31 fucose (fuc)) (Oades, 1984; Moers et al., 1990). While turnover dynamics of plant derived

1 sugars should mainly be governed by stabilisation processes, the turnover dynamics of
2 microbial sugars may be influenced by both, stabilisation and recycling.

3 The MRT of bulk and sugar carbon were examined in density fractions to elucidate turnover
4 dynamics in SOM pools with different degrees of degradation and protection. While free
5 particulate organic matter (fPOM) represents an only partly degraded SOM pool with fast
6 turnover, occluded particulate organic matter (oPOM) and mineral associated organic matter
7 correspond to pools that are more preserved from microbial attacks and show slow turnover
8 (John et al., 2005; Golchin et al., 1994b). The study was made on a field experiment located
9 in Rothalmünster with natural ^{13}C labelling by a vegetation change from C3 (wheat) to C4
10 vegetation (maize).

11 We hypothesise that MRT of plant and microbial sugar carbon will be different as the
12 mechanisms controlling their turnover dynamics are different: turnover of microbial derived
13 sugars should be mainly ruled by recycling whereas the turnover of plant derived sugars is
14 ruled by stabilisation.

15

16 **2 Materials and Methods**

17 **2.1 Study Site**

18 Soil samples were collected from the long-term field experiment at “Höhere Landbauschule”
19 Rothalmünster, Bavaria, Germany (N 48° 21' 47'', E 13° 11' 46''). The mean annual
20 temperature is 9.2 °C and the mean annual precipitation is 757 mm. Soil samples (Ap-horizon
21 & E-horizon) were taken in April 2011 from (i) a continuous maize plot (*Zea mays L.*)
22 established in 1979 on a former grassland plot until 1970 followed by wheat cultivation until
23 1978 and (ii) a continuous wheat plot (*Triticum aestivum L.*) established in 1969. Previous
24 vegetation on the wheat plot was grassland. The soil at the two sites was classified as a
25 stagnic Luvisol (IUSS Working Group WRB, 2014), derived from loess. Soil texture is silty
26 loam (11% sand, 73% silt, 16% clay). More details about the soil properties can be found in
27 John et al. (2005) and Ludwig et al. (2005).

28

1 **2.2 Density fractionation**

2 Density fractionation of soil was performed according to John et al. (2005). Briefly, 10 g of
3 soil were weighed into a 50 mL centrifuge tube and filled with 40 mL 1.6 g cm^{-3} sodium
4 polytungstatesolution (SPT, Sometu, Berlin). The tube was gently shaken 5 times by hand and
5 allowed to settle for 30 min. Afterwards the solution was centrifuged for 40 min at 3700 rpm.
6 The supernatant including floating materials was filtered with polyamide membrane filters
7 ($0.45 \text{ }\mu\text{m}$, Sartorius Göttingen) using vacuum and washed with distilled water to gain the
8 fPOM. Residual soil was re-suspended in 25 ml SPT (1.6 g cm^{-3}) and 18 glass pearls (4 mm
9 diameter) were added, the solution was then shaken for 16 hours at 60 movements per minute
10 to break up the aggregates. Subsequently, the solution was centrifuged 40 min at 3700 rpm,
11 vacuum filtered ($0.45 \text{ }\mu\text{m}$) and washed with distilled water to obtain the occluded particulate
12 organic matter (oPOM_{1.6}). The residual soil was re-suspended with 25 mL SPT using a
13 density of 2 g cm^{-3} , shaken for 10 min at 100 rpm and centrifuged (40 min at 3700 rpm). To
14 obtain the occluded particulate organic matter with a density of $1.6\text{-}2 \text{ g cm}^{-3}$ (oPOM₂), the
15 supernatant was vacuum-filtered and washed with distilled water. The remaining fraction
16 (Mineral) was washed three times with 20 mL water to remove SPT. Each time, the sample
17 was centrifuged and the supernatant discarded. All fractions were dried at $40 \text{ }^\circ\text{C}$.

18

19 **2.3 Sugar analysis**

20 Sugars were extracted and purified using a modified procedure based on Amelung et al.
21 (1996) and Amelung and Zhang; (2001). For extraction, sub-samples containing
22 approximately $0.5\text{-}5 \text{ }\mu\text{g C}$ (depending on the availability of the respective fraction) were
23 hydrolysed with 10 mL 4 M TFA at $105 \text{ }^\circ\text{C}$ for 4 hours. Afterwards the samples were filtered
24 through a glass fibre filter (Minisart GF, Sartorius, Göttingen, Germany) and dried by rotary
25 evaporation ($40 \text{ }^\circ\text{C}$, 50 hPa). In contrast to Amelung et al. (1996), the pre-dried samples were
26 re-dissolved in 0.5 mL water and evaporated to dryness for 3 times to remove all traces of
27 TFA (which impedes chromatographic separation, see Basler and Dyckmans (2013)). Then,
28 the samples were re-dissolved in approximately 3 mL water and passed through 4 g Dowex
29 X8 cation exchange resin (Sigma Aldrich, Steinheim, Germany) and 5 g Serdolit PAD IV
30 adsorption resin (Serva Electrophoresis GmbH, Heidelberg, Germany) for purification. Sugars
31 were eluted by adding 8 times 2 mL water. The eluate was freeze-dried and stored at $-18 \text{ }^\circ\text{C}$

1 until analysis. For HPLC-IRMS analysis the samples were dissolved in 3 mL water and
2 transferred into measurement vials.

3 The TFA extraction method is known to effectively extract hemi-cellulosic sugars (Amelung
4 et al., 1996) but cellulose is not cleaved by this method. The results presented here thus only
5 refer to non-cellulosic sugars and substantially underestimate the total sugar contribution of
6 plants SOM.

7

8 **2.4 Isotopic analysis**

9 Isotopic composition and total carbon content of plant material, bulk soil and density fractions
10 was analysed by EA-IRMS. The compound specific isotope analysis of the monosaccharides
11 was performed using a high-pressure liquid chromatography system (Sykam,
12 Fürstfeldbruck, Germany) coupled to an isotope ratio mass spectrometer (Delta V
13 Advantage, Thermo Scientific, Bremen, Germany) via an interface (LC-Isolink, Thermo
14 Scientific, Bremen, Germany) as described by Basler and Dyckmans (2013). Shortly, the
15 chromatographic column (Carbo Pac 20, Dionex) was held at 10 °C and a 0.25 mM NaOH
16 solution was used as mobile phase at a flow rate of 250 $\mu\text{L min}^{-1}$.

17

18 **2.5 Chloroform-Fumigation-Extraction**

19 Microbial Biomass (C_{mic}) was determined by the fumigation extraction method (Brookes et
20 al., 1985; Vance et al., 1987). K_2SO_4 concentrations were adapted for isotopic analysis
21 (Engelking et al., 2008). Briefly, a sub-sample of 20 g moist soil was separated into two
22 portions of 10 g. One soil sub-sample was directly extracted as described below. One portion
23 was placed in a desiccator with ethanol free CH_3Cl at 25 °C for 24 h. For extraction, soil
24 samples were shaken with 60 mL 0.05 M K_2SO_4 for one hour and subsequently filtered
25 (Whatman 595 $\frac{1}{2}$). The dissolved organic carbon was analysed using a TOC analyser multi
26 C/N® 2000 (Analytik Jena, Jena, Germany). For stable isotope measurements freeze-dried
27 aliquots were analysed by EA-IRMS. The isotopic signature of the microbial biomass was
28 calculated as follows:

$$29 \quad \delta^{13}\text{C}_{mic} = \frac{(\delta^{13}\text{C}_F \cdot C_F) - (\delta^{13}\text{C}_{nF} \cdot C_{nF})}{(C_F - C_{nF})} \quad (1)$$

1 where $\delta^{13}C_F$ and $\delta^{13}C_{nF}$ are the isotopic signatures of the fumigated and non-fumigated
2 extracts and C_F and C_{nF} are the extracted carbon content [mg kg^{-1}] of the fumigated and non-
3 fumigated soil samples. Carbon extracted from non-fumigated samples represents the K_2SO_4
4 extractable C fraction (exC).

5 **2.6 Estimations of maize-derived carbon and turnover times**

6 Under the assumption that the maize and wheat sites have a similar history and similar C
7 dynamics and fractionation during decomposition is comparable for wheat and maize plant
8 material., the proportion of maize-derived carbon in bulk soil and density fractions was
9 calculated according to (Balesdent and Mariotti, 1996; Derrien et al. 2006):

$$10 \quad f = \frac{(\delta_{sample} - \delta_{reference})}{(\delta_{maize} - \delta_{wheat})} \quad (2)$$

11 where f is the relative proportion of maize-derived carbon, δ_{sample} is the $\delta^{13}\text{C}$ value of the
12 maize plot sample, $\delta_{reference}$ presents the measured ^{13}C value of the corresponding wheat plot
13 samples, and δ_{maize} and δ_{wheat} are the ^{13}C values of the crop residues of maize (-13.2‰) and
14 wheat (-27.5‰). The resulting difference of 14.3 between wheat and maize plants was used
15 for all fractions (bulk material and individual sugars) because the determination especially of
16 mainly microbial derived sugars in plant material was very difficult.

17 The error of maize contribution percentage calculated from error propagation was below 10 %
18 for all samples, which is in the range of the standard error calculated from the replications.

19 Assuming steady state conditions and homogeneous soil fractions which can be described
20 with a single pool model (Six and Jastrow, 2002), the MRT is calculated according to Derrien
21 and Amelung (2011):

$$22 \quad MRT = \frac{1}{k} \quad (3),$$

23 Where the time constant k is calculated from the following equation:

$$24 \quad f = 1 - \exp^{(-kt)} \quad (4)$$

25 where t is the time of maize cultivation.

1 Since conditions like fertilization and C contents of the soil remained about the same after the
2 change from C3 to C4 vegetation, and the conversion was from other cereal crops (wheat) to
3 maize, which are very similar with respect to biochemical nature, soil inputs, location of soil
4 inputs, decay rates and decay products, the system approximates a steady-state system
5 (Balesdent and Mariotti, 1996) as required. It is well known that the assumption that MRT of
6 soil organic carbon can be described by a single pool model is a rough simplification since it
7 is a complex mixture of SOM with different stability and turnover even if the isolated soil
8 fractions are one step towards homogeneity, especially concerning POM fractions. Therefore,
9 we used the term “apparent” MRT. In addition it has to be noted that we refer to the MRT of
10 the carbon in individual molecules and not of the intact molecules as a whole.

11 **2.7 Statistical analysis**

12 Analysis of Variance (ANOVA) with ensuing Post-hoc test (Tukey) were conducted to detect
13 differences among the sugars within a soil fraction (bulk soil, density fractions) and among
14 individual sugars of different soil fractions. Statistical analysis were made using R 3.0.2 (R
15 Core Team, 2013).

16 **3 Results**

17 **3.1 Carbon and sugar content in soil, density fractionations and plant** 18 **material**

19 The recovery of carbon after density fractionation of the wheat and maize plots was about
20 90% in the Ap-horizon and about 86% in the E-horizon. Between 79 and 89% of total
21 recovered carbon was found in the Mineral in the investigated soils (Fig. 1). The oPOM₂
22 fraction accounted for 7 and 10% of the carbon found in the Ap-horizon and for 4 and 9% in
23 the E-horizon of the wheat and maize plot, respectively. Less carbon was found in the
24 oPOM_{1,6} fractions (between 3 and 5%) and the free particulate organic matter (fPOM; 2-3%).
25 The contribution of sugar carbon to total carbon in oPOM_{1,6} was between 5 and 8%. Higher
26 contributions were observed in the oPOM₂ with 11 to 15% (data not shown). The general
27 sugar distribution in the bulk soil fraction was glc>gal>man=ara=xyl>rha>fuc and was
28 slightly different in the POM fractions, where ara and xyl occurred in higher proportions than
29 gal and man (Table 1).

1 In the plant, sugars were dominated by xyl with about 44 mg C g⁻¹ (wheat) and 30 mg C g⁻¹
2 (maize), followed by ara and glc with about 8 mg C g⁻¹ (wheat) and 6 mg C g⁻¹ (Table 1). The
3 other sugars each contributed 4 mg C g⁻¹ or less. The extracted sugars accounted for 20% and
4 8% of total carbon and in the wheat and maize plants, respectively.

5 **3.2 Contribution of maize-derived Carbon to the sugars in different soil** 6 **fractions**

7 In general, the contribution of maize-derived carbon in the varying density fractions decreased
8 in the order fPOM>oPOM₂>Mineral>oPOM_{1,6}. The proportion of maize-derived carbon in
9 bulk soil was around 40% in the Ap and 30% in the E-horizon (Fig. 2). The apparent MRT of
10 carbon calculated from this data ranged between 25 (fPOM, Ap) and 119 (oPOM_{1,6}, E) years
11 (Table 2). The contribution of maize to the exC was within the range of the bulk soil, whereas
12 the proportion of maize in Cmic was twice as high as in the bulk soil (Fig. 2). The proportions
13 of maize-derived carbon in individual sugars showed a distinct pattern (Fig.2): In the bulk
14 soil, the highest proportion of maize-derived carbon was observed in xyl (~70% in Ap, 56%
15 in E). The other sugars showed maize-derived carbon proportions in the range of the bulk soil
16 of about 37% in Ap and 30% in E with the exception of ara, fuc and gal in E with only 25%
17 maize contribution. Bulk fPOM had maize contributions of 88 and 78% in the Ap and E-
18 horizon, respectively. Maize contribution for all sugars in both horizons was close to 100%
19 and thus the fPOM fraction was not evaluated further. In the oPOM_{1,6} fraction, the proportions
20 of maize-derived carbon of individual sugars were two or three times higher than for total
21 carbon in this fraction (Fig. 2A and B). In the oPOM_{1,6} fraction of Ap, xyl and man showed
22 the highest percentages (~85%) of maize-derived carbon, followed by glc (77%) and ara, rha
23 and gal (about 50%). The lowest percentage of maize-derived carbon was found for fuc
24 (~30%) in the Ap-horizon. In the E horizon, all sugars contained about 55% maize-derived C
25 and showed no significant differences (p<0.05), but there was still a trend towards higher
26 percentages of maize-derived carbon in xyl and man as compared to the other sugars.

27 In the oPOM₂ fraction, the highest percentages of maize-derived carbon in the sugars of all
28 fractions were observed with about 77% and 65% in the Ap and E-horizon. In the oPOM₂
29 fraction no significant difference in maize contribution among the sugars was observed
30 (p<0.05) in both horizons, but a trend of higher values for xyl (88%) and lower values for rha
31 (58%) were found for the Ap-horizon (Fig. 2).

1 In the Mineral, the percentages of maize-derived carbon in the Ap-horizon showed no
2 significant difference to the bulk soil fraction and amounted about 52% of maize-derived
3 carbon. Xylose showed the highest values with 66% and man and ara showed the smallest
4 percentages (44%). In the Mineral of the E-horizon, the maize percentages were about 37%
5 and showed no significant difference to the bulk soil (Fig.2). Xyl and man showed the highest
6 percentages (~50%) of maize-derived carbon, followed by ara, glc, fuc and gal with about
7 25%. The calculated MRT for the sugar carbon in density fractions (Table 2) showed values
8 from 14 years (xyl in oPOM₂ Ap-horizon) to 152 years (man, in bulk soil E-horizon).

9

10 **4 Discussion**

11 Carbon content increased with decreasing density of the fractions concomitant with
12 decreasing organo-mineral associations, similar to earlier findings on the same (John et al.,
13 2005) and other soils (Baisden et al., 2002; Golchin et al., 1994a). The fPOM fractions
14 contained between 2% and 8 % of total carbon and the major part (86%) were found in the
15 Mineral fraction The relative contribution of sugars to bulk carbon was 8% in the Ap-horizon
16 and around 7% in the E horizon in agreement with values reported by Cheshire (1979),
17 Derrien et al. (2006) and Guggenberger et al. (1994). The proportions of sugar carbon in the
18 POM fractions decreased in the order oPOM₂>fPOM>oPOM_{1,6} in both horizons. This
19 corroborates the ¹³C NMR analysis on the same soil, which revealed decreasing O-alkyl
20 carbon content (representing e.g. sugars) in oPOM_{1,6} as compared to oPOM₂, whereas alkyl-
21 carbon content (representing lipids, fatty acids, plant aliphatic polymers) increased (Helfrich
22 et al., 2006). The ratio of alkyl to O-alkyl carbon has been reported to provide an indicator of
23 decomposition, as O-alkyl carbon rich substances are more easily accessible and thus
24 preferentially decomposed and more recalcitrant compounds accumulate (Golchin et al.,
25 1994b; Baldock et al., 1997). Consequently, the higher sugar contribution in oPOM₂ as
26 compared to oPOM_{1,6} probably indicates a higher degree of decomposition in the oPOM_{1,6}
27 fraction. This supports the concept of Golchin et al. (1994a), who suggest that the fresh,
28 carbohydrate rich POM is utilised by microorganisms with concurrent increase of organo-
29 mineral associations (→ oPOM₂) and the formation of aggregates. Within the aggregates,
30 decomposition proceeds and labile compounds become more and more depleted. In turn,
31 microbial activity decreases and less binding agents are produced and binding to mineral
32 particles is decreased (decreased density → oPOM_{1,6}). Due to reduced microbial activity and

1 decreasing production of binding agents the aggregates become unstable and finally disrupt,
2 and new aggregates may develop if fresh plant or microbial debris is available to fuel
3 microbial activity.

4 In the density fractions the apparent MRT of bulk carbon increased in the order
5 fPOM<oPOM₂<Mineral<oPOM_{1,6} in both soil depths, which is in line with studies by John et
6 al. (2005) and Rethemeyer et al. (2005) on the same soil and corroborates the concept of
7 Golchin et al. (1994a) of the aggregate hierarchy described above. Although the oPOM_{1,6}
8 fraction had the highest proportion of C3 carbon, the sugars in the oPOM_{1,6} fractions were
9 much younger than the bulk fraction, but in range with the oPOM₂ fraction and the microbial
10 biomass. This indicates that the microbial activity leading to aggregate formation also in the
11 “old” oPOM_{1,6} fraction is fuelled from relatively fresh assimilates and shows the importance
12 of microbial activity to form binding agents, as mentioned before by Oades (1984).
13 Corroborating, the apparent MRT of sugar carbon in both oPOM fractions is comparable to
14 the apparent MRT of the microbial biomass carbon in both soil horizons.

15 Man, as a microbial derived sugar showed considerably higher incorporation of maize-derived
16 carbon similar to xyl in the oPOM fractions although the contribution of man by plants was
17 very little. A possible explanation could be fungal activity, as it is known that fungi feed
18 mainly on the recent vegetation (Hobbie et al., 2002; Kramer and Gleixner, 2006).
19 Additionally, mannan, a mannose polymer, is abundant in exo-polysaccharides and cell walls
20 of fungi (Osaku et al., 2002; Stribley and Read, 1974; Bowman and Free, 2006) and the
21 involvement of fungal activity in soil aggregate formation was highlighted in several studies
22 (Chenu, 1989; Caesar-Tonthat, 2002; Tisdall and Oades, 1982). In the oPOM fractions of the
23 E-horizon (especially oPOM_{1,6}) man was much less influenced by maize-derived carbon
24 compared to Ap; this may indicate a reduced importance of fungal activity to oPOM
25 formation in the subsoil or at least no distinct allocation of maize-derived carbon through the
26 hyphal network to the subsoil.

27 Xyl had the highest percentages of maize-derived carbon in all soil fractions and depths,
28 owing to the high input of xyl from plant material (mainly from hemicellulose). Additionally,
29 root exudates provided a further small source of xyl as shown by Derrien et al. (2004) and, in
30 turn, roots and their exudates promote aggregate formation (Six et al., 2004; Oades, 1984). In
31 contrast ara, which has also been described as mainly plant derived (Oades, 1984), showed
32 smaller percentages of maize-derived carbon in all density fractions compared to xyl. One

1 could assume that ara and xyl, as sugars of the same origin, were subject to the same
2 dynamics, and more specifically, a similar mole ratio of ara to xyl in plants and oPOM was
3 expected. In the oPOM fractions, however, the ratio of ara to xyl increased (as compared to
4 the plants) and, in addition, the percentages of maize-derived carbon of ara were not
5 significantly different from sugars derived mainly from microorganisms (fuc, rha and gal).
6 This indicates that in our soil, ara and xyl dynamics are not ruled by the same factors. It has
7 been shown that both sugars are highly abundant in plant material at a molar ratio of 1:3
8 (ara:xyl) or higher (Boschker et al., 2008; Glaser et al., 2000; Moers et al., 1990; Oades,
9 1984), however we found much less contribution of ara than xyl by both wheat and maize
10 plants with a molar ratio of 1:5 (Table 1). On the other hand, both ara and xyl are produced by
11 microbial biomass (Muramaya, 1988; Cheshire, 1977; Coelho et al., 1988; Basler et al.) and
12 we therefore assume that in this study, ara was much more influenced by microbial
13 production than xyl and its high mean age in oPOM_{1.6} and oPOM₂ (28 to 48 years), was
14 considerably influenced by microbial activity (and substrate recycling). This also indicates
15 that the formation of oPOM fractions is predominantly based on microbial activity and not
16 plant input in the first place. In contrast to ara, the dynamics of xyl were dominated by plant
17 input and recycling seems to play a minor role.

18 Taken together the finding of substantially higher MRT for carbon of microbial sugars
19 (influenced by both, stabilisation and substrate recycling), compared to that of plant derived
20 sugars (the turnover dynamics of which are dominated by stabilisation processes) indicates,
21 that the mean age of SOM is strongly influenced by substrate recycling and that stabilisation
22 processes do not play a dominant role for SOM dynamics.

23

24 **5 Conclusion**

25 This study provides new insight in the dynamics of soil sugars, as an important compound of
26 SOM. Our data show that the reuse of organic matter is of high importance for soil sugar
27 dynamics and is largely responsible for high MRT of sugar carbon in soil. Stabilisation
28 processes on the other hand seem to play only a minor role for the persistence of sugars in
29 soil, as only xyl dynamics were dominated by stabilisation. Moreover, we could show that
30 microbial activity fuelled by fresh organic matter plays an important role in aggregate
31 formation, corroborate the concept of Golchin et al. (1994a). However, the mechanisms of
32 recycling i.e. intact re-utilization versus intensive metabolization and incorporation in

1 modified compounds remain unclear based on compound specific isotope analysis only.
2 However, combining compound specific isotope analysis with position-specific labeling help
3 disentangling the processes underlying the carbon recycling (Apostel et al. 2015; Dippold and
4 Kuzyakov 2013). Ultimately, our findings highlight the importance of recycling processes for
5 SOM dynamics on the molecular as well as the aggregate level.

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1 **References**

- 2 Amelung, W., Brodowski, S., Sandhage-Hofmann, A., and Bol, R.: Combining Biomarker
3 with Stable Isotope Analyses for Assessing the Transformation and Turnover of Soil Organic
4 Matter, in: *Advances in Agronomy*, Vol 100, Adv. Agron., Elsevier Academic Press Inc, 155–
5 250, 2008.
- 6 Amelung, W., Cheshire, M. V., and Guggenberger, G.: Determination of neutral and acidic
7 | sugars in soil by capillary- gas-liquid chromatography after trifluoroacetic acid hydrolysis,
8 *Soil Biol. Biochem.*, 28, 1631–1639, 1996.
- 9 Amelung, W. and Zhang, X.: Determination of amino acid enantiomers in soils, *Soil Biol.*
10 *Biochem.*, 33, 553–562, 2001.
- 11 Apostel, C., Dippold, M., and Kuzyakov, Y.: Biochemistry of hexose and pentose
12 transformations in soil analyzed by position-specific labeling and ¹³C-PLFA, *Soil Biol.*
13 *Biochem.*, 80, 199–208, doi:10.1016/j.soilbio.2014.09.005, 2015.
- 14 Baisden, W. T., Amundson, R., Cook, A. C., and Brenner, D. L.: Turnover and storage of C
15 and N in five density fractions from California annual grassland surface soils, *Global*
16 *Biogeochem. Cy.*, 16, doi:10.1029/2001GB001822, 2002.
- 17 Baldock, J. A., Oades, J. M., Nelson, P. N., Skene, T. M., Golchin, A., and Clarke, P.:
18 Assessing the extent of decomposition of natural organic materials using solid-state ¹³C
19 NMR spectroscopy, *Aust. J. Soil Res.*, 35, 1061, doi:10.1071/S97004, 1997.
- 20 Balesdent, J. and Mariotti, A.: Measurement of soil organic matter turnover using
21 Measurement of soil organic matter turnover using ¹³C natural abundance, in: *Mass*
22 *spectrometry of soils*, Boutton, T. W., Yamasaki, S.-i. (Eds.), Books in soils, plants, and the
23 environment, M. Dekker, New York, 83–111, 1996.
- 24 Basler, A., Dippold, M., Helfrich, M., and Dyckmans, J.: Recycling versus Stabilisation of
25 soil sugars– a long-term laboratory incubation experiment, *Biogeosciences*, submitted.
- 26 Basler, A. and Dyckmans, J.: Compound-specific delta C-13 analysis of monosaccharides
27 from soil extracts by high-performance liquid chromatography/isotope ratio mass
28 spectrometry, *Rapid Commun. Mass Spectrom.*, 27, 2546–2550, doi:10.1002/rcm.6717, 2013.
- 29 Boschker, H., Moerdijk-Poortvliet, T., van Breugel, P., Houtekamer, M., and Middelburg, J.
30 J.: A versatile method for stable carbon isotope analysis of carbohydrates by high-

1 performance liquid chromatography/isotope ratio mass spectrometry, *Rapid Commun. Mass*
2 *Spectrom.*, 22, 3902–3908, 2008.

3 Bowman, S. M. and Free, S. J.: The structure and synthesis of the fungal cell wall, *Bioessays*,
4 28, 799–808, doi:10.1002/bies.20441, 2006.

5 Brookes, P., Landman, A., Pruden, G., and Jenkinson, D. S.: Chloroform fumigation and the
6 release of soil nitrogen: A rapid direct extraction method to measure microbial biomass
7 nitrogen in soil, *Soil Biol. Biochem.*, 17, 837–842, doi:10.1016/0038-0717(85)90144-0, 1985.

8 Caesar-Tonthat, T. C.: Soil binding properties of mucilage produced by a basidiomycete
9 fungus in a model system, *Mycol Res*, 106, 930–937, doi:10.1017/S0953756202006330,
10 2002.

11 Chenu, C.: Influence of a fungal polysaccharide, scleroglucan, on clay microstructures, *Soil*
12 *Biol. Biochem.*, 21, 299–305, doi:10.1016/0038-0717(89)90108-9, 1989.

13 Cheshire, M. V.: Origins and stability of soil polysaccharide, *J. Soil Sci.*, 28, 1–10,
14 doi:10.1111/j.1365-2389.1977.tb02290.x, 1977.

15 Cheshire, M. V.: Nature and origin of carbohydrates in soils, Academic Pr, London, 216 pp.,
16 1979.

17 Coelho, R. R., Linhares, L. F., and Martin, J. P.: Sugars in hydrolysates of fungal melanins
18 and soil humic acids, *Plant Soil*, 106, 127–133, doi:10.1007/BF02371204, 1988.

19 Derrien, D. and Amelung, W.: Computing the mean residence time of soil carbon fractions
20 using stable isotopes: impacts of the model framework, *Eur. J. Soil Science*, 62, 237–252,
21 doi:10.1111/j.1365-2389.2010.01333.x, 2011.

22 Derrien, D., Marol, C., Balabane, M., and Balesdent, J.: The turnover of carbohydrate carbon
23 in a cultivated soil estimated by ¹³C natural abundances, *Eur. J. Soil Science*, 57, 547–557,
24 2006.

25 Derrien, D., Marol, C., and Balesdent, J.: The dynamics of neutral sugars in the rhizosphere of
26 wheat. An approach by C-13 pulse-labelling and GC/C/IRMS, *Plant Soil*, 267, 243–253,
27 2004.

28 Derrien, D., Marol, C., and Balesdent, J.: Microbial biosyntheses of individual neutral sugars
29 among sets of substrates and soils, *Geoderma*, 139, 190–198, 2007.

1 Dippold, M. and Kuzyakov, Y.: Biogeochemical transformations of amino acids in soil
2 assessed by position-specific labelling, *Plant Soil*, 373, 385–401, doi:10.1007/s11104-013-
3 1764-3, 2013.

4 Engelking, B., Flessa, H., and Joergensen, R. G.: Formation and use of microbial residues
5 after adding sugarcane sucrose to a heated soil devoid of soil organic matter, *Soil Biol.*
6 *Biochem.*, 40, 97–105, doi:10.1016/j.soilbio.2007.07.009, 2008.

7 Glaser, B., Turrion, M. B., Solomon, D., Ni, A., and Zech, W.: Soil organic matter quantity
8 and quality in mountain soils of the Alay Range, Kyrgyzia, affected by land use change, *Biol.*
9 *Fertil. Soils*, 31, 407–413, 2000.

10 Gleixner, G., Poirier, N., Bol, R., and Balesdent, J.: Molecular dynamics of organic matter in
11 a cultivated soil, *Org. Geochem.*, 33, 357–366, 2002.

12 Golchin, A., Oades, J. M., Skjemstad, J. O., and Clarke, P.: Soil structure and carbon cycling,
13 *Aust. J. Soil Res.*, 32, 1043–1068, doi:10.1071/SR9941043, 1994a.

14 Golchin, A., Oades, J. M., Skjemstad, J. O., and Clarke, P.: Study of free and occluded
15 particulate organic-matter in soils by solid-state C-13 CP/MAS NMR-spectroscopy and
16 scanning electron-microscopy, *Aust. J. Soil Res.*, 32, 285–309, doi:10.1071/SR9940285,
17 1994b.

18 Guggenberger, G., Christensen, B. T., and Zech, W.: Land-use effects on the composition of
19 organic matter in particle-size separates of soil: I. Lignin and carbohydrate signature, *Eur. J.*
20 *Soil Science*, 45, 449–458, doi:10.1111/j.1365-2389.1994.tb00530.x, 1994.

21 Helfrich, M., Ludwig, B., Buurman, P., and Flessa, H.: Effect of land use on the composition
22 of soil organic matter in density and aggregate fractions as revealed by solid-state C-13 NMR
23 spectroscopy, *Geoderma*, 136, 331–341, 2006.

24 Hobbie, E. A., Weber, N. S., Trappe, J. M., and van Klinken, Gert J.: Using radiocarbon to
25 determine the mycorrhizal status of fungi, *New Phytol*, 156, 129–136, doi:10.1046/j.1469-
26 8137.2002.00496.x, 2002.

27 IUSS Working Group WRB: World reference base for soil resources 2014: International soil
28 classification system for naming soils and creating legends for soil maps, World soil resources
29 reports, FAO, Rome, Online-Ressource, 2014.

1 John, B., Yamashita, T., Ludwig, B., and Flessa, H.: Storage of organic carbon in aggregate
2 and density fractions of silty soils under different types of land use, *Geoderma*, 128, 63–79,
3 doi:10.1016/j.geoderma.2004.12.013, 2005.

4 Kiem, R. and Kögel-Knabner, I.: Contribution of lignin and polysaccharides to the refractory
5 carbon pool in C-depleted arable soils, *Soil Biol. Biochem.*, 35, 101–118, 2003.

6 Kramer, C. and Gleixner, G.: Variable use of plant- and soil-derived carbon by
7 microorganisms in agricultural soils, *Soil Biol. Biochem.*, 38, 3267–3278,
8 doi:10.1016/j.soilbio.2006.04.006, 2006.

9 Ludwig, B., Helfrich, M., and Flessa, H.: Modelling the Long-Term Stabilization of Carbon
10 from Maize in a Silty Soil, *Plant Soil*, 278, 315–325, doi:10.1007/s11104-005-8808-2, 2005.

11 Moers, M. E. C., Baas, M., Deleeuw, J. W., Boon, J. J., and Schenck, P. A.: Occurrence and
12 origin of carbohydrates in peat samples from a red mangrove environment as reflected by
13 abundances of neutral monosaccharides, *Geochim. Cosmochim. Ac.*, 54, 2463–2472, 1990.

14 Muramaya, S.: Microbial synthesis of saccharides in soils incubated with ¹³C-labelled
15 glucose, *Soil Biol. Biochem.*, 20, 193–199, doi:10.1016/0038-0717(88)90036-3, 1988.

16 Oades, J. M.: Soil organic matter and structural stability: mechanisms and implications for
17 management, *Plant Soil*, 76, 319–337, 1984.

18 Osaku, C. A., Sasaki, G. L., Zancan, G. T., and Iacomini, M.: Studies on neutral
19 exopolysaccharides produced by the ectomycorrhiza *Thelephora terrestris*, *FEMS Microbiol*
20 *Lett*, 216, 145–149, doi:10.1111/j.1574-6968.2002.tb11428.x, 2002.

21 R Core Team: R: A Language and Environment for Statistical Computing, R Foundation for
22 Statistical Computing, Vienna, Austria, 2013.

23 Rethemeyer, J., Kramer, C., Gleixner, G., John, B., Yamashita, T., Flessa, H., Andersen, N.,
24 Nadeau, M.-J., and Grootes, P. M.: Transformation of organic matter in agricultural soils:
25 radiocarbon concentration versus soil depth, *Geoderma*, 128, 94–105,
26 doi:10.1016/j.geoderma.2004.12.017, 2005.

27 Sauheitl, L., Glaser, B., and Bol, R.: Short-term dynamics of slurry-derived plant and
28 microbial sugars in a temperate grassland soil as assessed by compound-specific delta C-13
29 analyses, *Rapid Commun. Mass Spectrom.*, 19, 1437–1446, 2005.

- 1 Schmidt, M. W., Torn, M. S., Abiven, S., Dittmar, T., Guggenberger, G., Janssens, I. A.,
2 Kleber, M., Kögel-Knabner, I., Lehmann, J., Manning, David A C, Nannipieri, P., Rasse, D.
3 P., Weiner, S., and Trumbore, S. E.: Persistence of soil organic matter as an ecosystem
4 property, *Nature*, 478, 49–56, doi:10.1038/nature10386, 2011.
- 5 Six, J., Bossuyt, H., Degryze, S., and Denef, K.: A history of research on the link between
6 (micro)aggregates, soil biota, and soil organic matter dynamics, *Soil Till. Res.*, 79, 7–31,
7 doi:10.1016/j.still.2004.03.008, 2004.
- 8 Six, J., Conant, R. T., Paul, E. A., and Paustian, K.: Stabilization mechanisms of soil organic
9 matter: Implications for C-saturation of soils, *Plant Soil*, 241, 155–176,
10 doi:10.1023/A:1016125726789, 2002.
- 11 Six, J. and Jastrow, J. D.: Organic matter turnover, in: *Encyclopedia of soil science*, Lal, R.
12 (Ed.), Dekker, New York, NY, 936–942, 2002.
- 13 Sollins, P., Homann P., and Caldwell BA.: Stabilization and destabilization of soil organic
14 matter: mechanisms and controls, *Geoderma*, 74, 65–105, doi:10.1016/S0016-
15 7061(96)00036-5, 1996.
- 16 Stevenson, F. J.: *Humus chemistry: Genesis, composition, reactions*, 2nd ed., Wiley, New
17 York, xiii, 496, 1994.
- 18 Stribley, D. P. and Read, D. J.: The Biology of Mycorrhiza in the Ericaceae. III. Movement of
19 Carbon-14 from Host to Fungus, *New Phytol*, 73, 731–741, doi:10.1111/j.1469-
20 8137.1974.tb01301.x, 1974.
- 21 Takano, Y., Chikaraishi, Y., Ogawa, N. O., Nomaki, H., Morono, Y., Inagaki, F., Kitazato,
22 H., Hinrichs, K.-U., and Ohkouchi, N.: Sedimentary membrane lipids recycled by deep-sea
23 benthic archaea, *Nature Geosci*, 3, 858–861, doi:10.1038/ngeo983, 2010.
- 24 Tisdall, J. M. and Oades, J. M.: Organic matter and water-stable aggregates in soils, *J. Soil
25 Sci.*, 33, 141–163, doi:10.1111/j.1365-2389.1982.tb01755.x, 1982.
- 26 Vance, E., Brookes, P., and Jenkinson, D. S.: An extraction method for measuring soil
27 microbial biomass C, *Soil Biol. Biochem.*, 19, 703–707, doi:10.1016/0038-0717(87)90052-6,
28 1987
- 29 von Lützow, M. von, Kögel-Knabner, I., Ekschmitt, K., Matzner, E., Guggenberger, G.,
30 Marschner, B., and Flessa, H.: Stabilization of organic matter in temperate soils: mechanisms

1 and their relevance under different soil conditions - a review, *Eur. J. Soil Science*, 57, 426–
2 445, doi:10.1111/j.1365-2389.2006.00809.x, 2006.

Table 1. Carbon content [mg C g⁻¹(dw)] and sugar content [mg C g⁻¹(dw)] in bulk soil, soil density fractions and wheat and maize plants. Latin letters (a-e) within one row indicate significant differences (p<0.05) among the different sugars within one fraction. Greek letters (α-δ) within one column indicate significant differences among different fractions for individual sugars. Means and standard error.

Fraction	Carbon	Fuc	Rha	Ara	Xyl	Glc	Gal	Man
continious Wheat plot (Ap)	mg Cg ⁻¹ bulk	mg Cg ⁻¹ fraction						
oPOM _{1,6} (n=5)	0.28 ±0.01	0.68 ±0.27 ^{ad}	0.91 ±0.28 ^{acd}	2.68 ±0.6 ^{ab}	6.08 ±1.4 ^{aa}	7.42 ±1.95 ^{aa}	2.24 ±0.51 ^{abc}	2.23 ±0.49 ^{abcd}
oPOM ₂ (n=5)	0.71 ±0.06	0.37 ±0.03 ^{ac}	1.39 ±0.5 ^{abc}	2.70 ±0.1 ^{ab}	5.83 ±1.01 ^{aa}	6.37 ±2.29 ^{aa}	1.85 ±0.56 ^{ab}	3.19 ±0.84 ^{ab}
mineral (n=3)	9.51 ±1.02	0.05 ±0.01 ^{bd}	0.09 ±0.01 ^{bcd}	0.15 ±0.01 ^{bc}	0.15 ±0.01 ^b	0.26 ±0.03 ^{ba}	0.18 ±0.02 ^{ab}	0.16 ±0.05 ^b
bulk (n=3)	12.06 ±0.8	0.03 ±0.00 ^{bc}	0.07 ±0.00 ^{bc}	0.13 ±0.02 ^b	0.14 ±0.01 ^b	0.27 ±0.27 ^{ba}	0.16 ±0.00 ^{ab}	0.14 ±0 ^b
continious Wheat plot (E)								
oPOM _{1,6} (n=5)	0.25 ±0.05	0.42 ±0.19 ^{ab}	0.60 ±0.32 ^{ab}	2.38 ±0.91 ^{ab}	3.36 ±1.25 ^{aa}	5.38 ±1.87 ^{aa}	1.93 ±0.85 ^{ab}	2.06 ±0.66 ^{ab}
oPOM ₂ (n=5)	0.29 ±0.04	0.40 ±0.09 ^{ac}	0.88 ±0.23 ^{abc}	1.90 ±0.42 ^{abc}	2.96 ±0.7 ^{ab}	4.39 ±0.76 ^{aa}	1.90 ±0.37 ^{abc}	1.81 ±0.34 ^{abc}
mineral (n=3)	5.90 ±0.01	0.03 ±0.01 ^{bd}	0.08 ±0.01 ^{bcd}	0.07 ±0.01 ^{bc}	0.07 ±0 ^{bcd}	0.14 ±0.01 ^{ba}	0.10 ±0.01 ^{ab}	0.09 ±0.01 ^b
bulk (n=3)	7.86 ±0.26	0.03 ±0.00 ^{be}	0.04 ±0.01 ^{bde}	0.07 ±0.01 ^{bcd}	0.06 ±0.02 ^{bcd}	0.16 ±0.02 ^{ba}	0.10 ±0.02 ^{ab}	0.08 ±0.03 ^b
continious Maize plot (Ap)								
oPOM _{1,6} (n=5)	0.49 ±0.02	0.76 ±0.3 ^{ae}	1.17 ±0.28 ^{ac}	3.22 ±0.67 ^{ab}	5.85 ±1 ^{aa}	8.61 ±0.67 ^{aa}	3.04 ±0.56 ^{aa}	2.90 ±0.34 ^{ab}
oPOM ₂ (n=5)	1.15 ±0.09	0.50 ±0.02 ^{ac}	0.86 ±0.09 ^{acd}	2.61 ±0.36 ^{ab}	5.39 ±1.17 ^{aab}	6.00 ±0.33 ^{ba}	2.56 ±0.22 ^{ab}	2.28 ±0.23 ^{abc}
mineral (n=3)	9.31 ±1.51	0.03 ±0.00 ^{be}	0.09 ±0.02 ^b	0.13 ±0.03 ^{abc}	0.13 ±0.02 ^{ab}	0.27 ±0.05 ^{ga}	0.17 ±0.04 ^{ab}	0.16 ±0.03 ^{ab}
bulk (n=3)	12.51 ±0.38	0.04 ±0.00 ^{bf}	0.09 ±0.01 ^b	0.15 ±0.01 ^b	0.16 ±0.02 ^b	0.36 ±0.02 ^{ga}	0.20 ±0.01 ^{ab}	0.17 ±0.01 ^b
continious Maize plot (E)								
oPOM _{1,6} (n=5)	0.42 ±0.02	0.46 ±0.09 ^{ac}	0.92 ±0.09 ^{ae}	3.63 ±0.21 ^{abc}	5.76 ±0.93 ^{ab}	9.44 ±0.51 ^{aa}	3.25 ±0.24 ^{acd}	3.03 ±0.17 ^{ad}
oPOM ₂ (n=5)	0.82 ±0.04	0.48 ±0.02 ^{ad}	0.82 ±0.08 ^{ac}	2.48 ±0.34 ^b	4.56 ±0.9 ^{ab}	5.42 ±0.36 ^{ba}	2.47 ±0.23 ^b	2.09 ±0.21 ^b
mineral (n=3)	7.75 ±0	0.03 ±0.00 ^{bc}	0.08 ±0.01 ^b	0.07 ±0 ^{cd}	0.10 ±0.01 ^{dc}	0.14 ±0.01 ^{ga}	0.07 ±0.01 ^{ab}	0.09 ±0.01 ^{bc}
bulk (n=3)	11.10 ±0.24	0.04 ±0.00 ^b	0.07 ±0.00 ^{ef}	0.11 ±0.01 ^{cd}	0.10 ±0.01 ^{dc}	0.25 ±0.02 ^{da}	0.16 ±0.02 ^{da}	0.13 ±0.02 ^{bc}
Plants								
wheat (n=9)	417.09 ±28.33	n.d.	1.07 ±0.73	7.57 ±0.72	44.16 ±4.66	8.82 ±1.18	3.66 ±0.5	1.89 ±0.44

maize (n=9)	431.93 ±32.03	n.d.	1.14 ±0.27	5.85 ±0.52	29.66 ±2.95	5.64 ±0.41	2.88 ±0.35	n.d.
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Table 2. Calculated MRT of total and individual sugar carbon in density fractions and bulk soil. Means and standard error (n=4).

fractions	horizon	bulk C	MRT [years]					
			Ara	Xyl	Fuc	Rha	Gal	Man
oPOM _{1.6}	Ap	97 ± 3	44 ± 4	17 ± 3	65 ± 24	54 ± 6	39 ± 2	17 ± 2
	E	119 ± 6	48 ± 5	28 ± 4	55 ± 9	59 ± 15	51 ± 6	33 ± 3
oPOM ₂	Ap	37 ± 6	28 ± 4	14 ± 4	30 ± 4	43 ± 14	28 ± 4	30 ± 5
	E	61 ± 8	31 ± 2	23 ± 5	28 ± 2	37 ± 12	41 ± 9	37 ± 3
Mineral	Ap	64 ± 3	57 ± 6	29 ± 2	51 ± 4	48 ± 7	47 ± 7	58 ± 10
	E	98 ± 3	102 ± 11	45 ± 3	87 ± 7	60 ± 6	88 ± 7	54 ± 3
bulk soil	Ap	68 ± 4	80 ± 13	27 ± 1	77 ± 15	77 ± 22	71 ± 2	70 ± 11
	E	85 ± 4	115 ± 12	41 ± 8	104 ± 19	72 ± 8	138 ± 32	152 ± 32

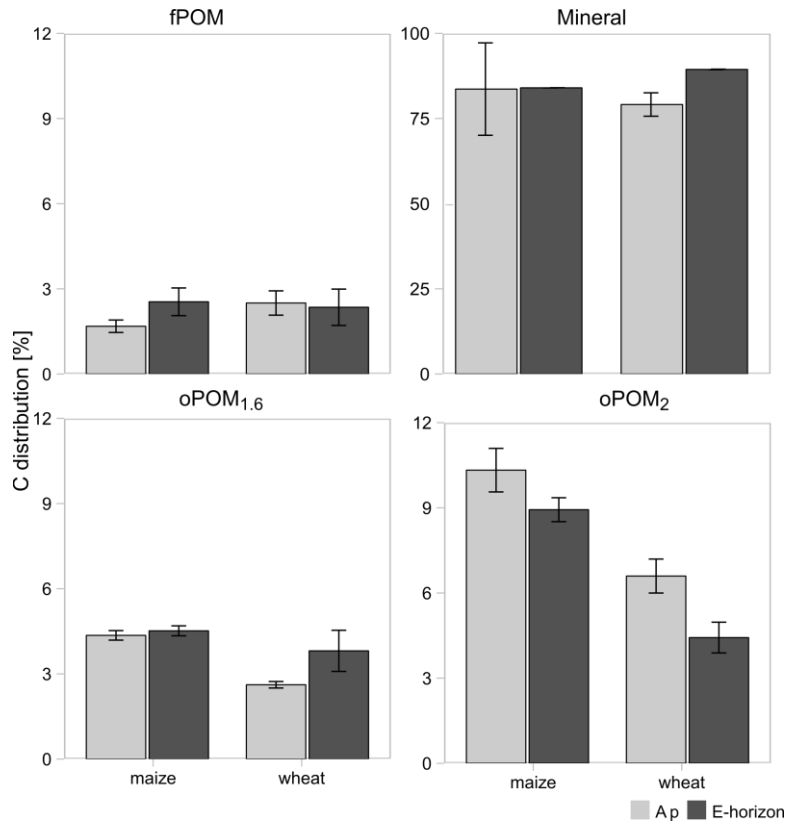
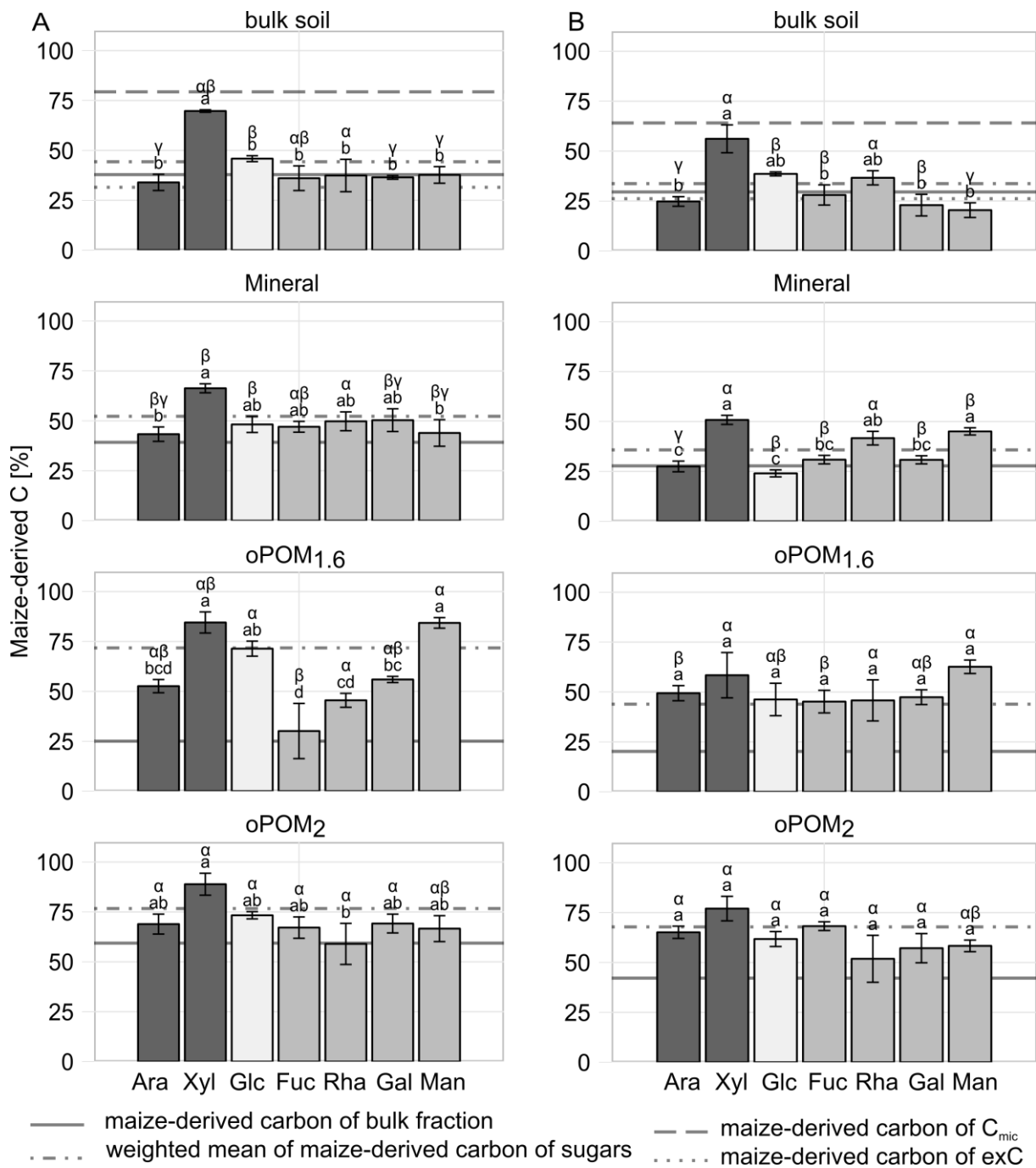


Figure1. Organic Carbon distribution in the investigated density fractions. Means and standard error (n=5).



1
2 Figure 2. Maize contribution to sugars in bulk soil, Mineral, oPOM_{1.6} and oPOM₂ fractions in
3 the (A) Ap (0-30cm) and (B) E-horizon (30-45cm). Latin letters (a-d) indicate significant
4 differences ($p < 0.05$) among the individual sugars within one fraction. Greek letters (α - γ)
5 indicate significant differences among different fractions for individual sugars. Means and
6 standard error ($n = 4$).

7
8