

1 **Methane Distribution, Fluxes, and Budget in the East**
2 **China Sea and Yellow Sea**

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16

17 **Abstract**

18 We measured dissolved methane (CH₄) concentrations, saturations, and air-sea
19 fluxes and sediment-water fluxes during five cruises in March, May, August, October,
20 and December of 2011 in the East China Sea (ECS) and the Yellow Sea (YS). CH₄
21 concentrations had obvious spatial and seasonal variability due to the complex mixing
22 of different water masses and other variables. Maximum CH₄ concentration, sea-air
23 and sediment-water fluxes all occurred during summer. CH₄ concentration decreased
24 gradually from the coastal area to the open sea, and high levels of CH₄ generally
25 appeared near the Changjiang Estuary and outside the Hangzhou Bay. In early spring

1 and winter, CH₄ in the shelf region had a uniform distribution from the surface to the
2 bottom, while it increased gradually with depth in other seasons. Subsurface CH₄
3 maximum occurred at a depth of about 200 m in the slope region during May, October,
4 and December. The CH₄ levels at the bottom were generally higher than at the surface,
5 and this was enhanced during summer due to the occurrence of hypoxia in the bottom
6 waters. Changjiang Diluted Water, Kuroshio, and Taiwan Warm Current Water
7 affected the geographic distribution of CH₄ in the ECS, and these water bodies
8 contributed about 3.45, 2.97, 14.60 mol·s⁻¹ of CH₄ to the ECS during summer and
9 2.11, 5.28, 5.20 mol·s⁻¹ CH₄ during winter, respectively. Sediment was also a
10 significant source of dissolved CH₄ in the ECS, and we estimated the average
11 sediment-water CH₄ flux of the ECS and YS as about 1.06 and 0.73 μmol·m⁻²·d⁻¹,
12 respectively. We used a box model to calculate preliminarily the CH₄ budget in the
13 ECS, which suggests that the main CH₄ sources in the ECS were *in situ* CH₄
14 formation in water column and sediment emissions. Air-sea exchange was the major
15 external sink of CH₄ in the ECS. We estimated total annual CH₄ emission from the
16 ECS and YS to be about 4.09×10⁹ mol/yr, hence the ECS and YS are active areas for
17 CH₄ production and emission.

18

19 **1 Introduction**

20 Methane (CH₄) is an active atmospheric trace gas that is responsible for about 23%
21 of the global greenhouse effect, and also participates in atmospheric chemistry and the
22 biogeochemical cycle of global carbon (Crutzen and Zimmermann, 1991; Quay et al.,
23 1999; IPCC, 2013). The global atmospheric CH₄ has increased significantly since the
24 industrial revolution, and was reported as 1803±2 ppb in 2011, which is about 2.5-fold
25 higher than that (722±25 ppb) in 1750 (IPCC, 2013). The continuing rise of
26 atmospheric CH₄ indicates an imbalance of sources and sinks. Natural sources are
27 responsible for about 40% of global CH₄ emission (Shakhova et al., 2010), among
28 which the ocean accounts for only about 0.4-5.0% of the total emission (Crutzen,
29 1991; Bange et al., 1994; Reeburgh, 2007). Although most of oceanic CH₄ are

1 oxidized by O₂ and sulfate in both aerobic and anaerobic environment before emission,
2 ocean still releases about 11-18 Tg CH₄ yr⁻¹ (Bange et al., 1994) into the air. In marine
3 ecosystems, CH₄ distributions and emissions have large spatial and temporal
4 variations (Bange et al., 1994, 2004; Kock et al., 2008; Forster et al., 2009; Zhang et
5 al., 2004, 2008a). Shelf areas and estuaries were estimated to contribute about 75% to
6 the global oceanic CH₄ emissions, although they just cover a small part of the world's
7 oceans (Bange et al., 1994). However, this estimate still has great uncertainties due to
8 large spatial and seasonal variations and limited area coverage.

9 The East China Sea (ECS) and the Yellow Sea (YS) together forms an important
10 marginal sea of the northwestern Pacific Ocean. This is one of the largest continental
11 shelves in the world, with a total surface area of 1.2×10^6 km². The ECS opens its
12 north to the YS with the straight line from the northern tip of the mouth of the
13 Changjiang (Yangtze River) toward the Jeju Island as the boundary (the blue dashed
14 line in Fig. 1). The ECS stretches south to the Taiwan Strait, and is adjacent to the
15 western Pacific along its east edge. Its total area is about 7.7×10^5 km², and the average
16 water depth is about 72 m. This wide and river-dominated shelf receives large
17 amounts of fresh water (9.03×10^{11} m³·year⁻¹), sediment (4.14×10^8 t·year⁻¹) (Wang et
18 al., 2008), and nutrients (Zhang et al., 1996; Gao et al., 2012) from the Changjiang.
19 The hydrographic characters of this region are also influenced greatly by a circulation
20 system including the Kuroshio, Tsushima Warm Current, Yellow Sea Warm Current
21 on the eastern boundary of the shelf, the Coastal Currents along the western side and
22 the Taiwan Warm Current dispersing out in the middle shelf (Su, 1998; Zhang et al.,
23 2007; Yuan et al., 2008).

24 So far oceanic CH₄ emission is poorly quantified, owing to lack of the temporal
25 continuity and areas coverage (Bange et al., 2009). In particular, data on CH₄ in
26 coastal areas of China are scattered and fragmentary. In early 1990s, Tsurushima et al.
27 (1996) observed the horizontal and vertical distribution of CH₄ along the section “PN
28 line” in the ECS. Rehder and Suess (2001) surveyed the distribution of CH₄ in surface
29 waters along the main path of Kuroshio, which only covered a small part of areas in

1 the East China and South China Sea. Zhang et al. (2004) reported distributions and
2 fluxes of CH₄ in the ECS and the YS in spring 2001, and they also estimated CH₄
3 input from the Changjiang and CH₄ export from the shelf to Kuroshio water. Yang et
4 al. (2010) reported the seasonal variation of CH₄ concentrations and air-sea fluxes in
5 the north YS during 2006 and 2007. Ye et al. (2015) and Zhang et al. (2008a) reported
6 CH₄ distributions and fluxes in the ECS in summer, but the former emphasized the
7 influence of hypoxia on CH₄ distribution in the bottom water, while the later mainly
8 introduced the high spatial variability of CH₄ depth profiles along four sections.
9 Although these results give us a glimpse of CH₄ in coastal waters of China, they are
10 still far from understanding the seasonal variations of CH₄ distribution and emission,
11 and quantifying CH₄ sources and sinks in these areas. Thus more data on CH₄ in
12 coastal and shelf waters of China are still needed to further enrich the global oceanic
13 CH₄ database, and to understand the biogeochemical cycle of CH₄ in the shelf areas
14 and their regional contribution to global oceanic CH₄ emission.

15 In this paper, we characterized the spatial distribution and seasonal variation of
16 dissolved CH₄ in the ECS and YS based on data collected during five cruises in 2011,
17 and identified factors that affected these patterns. We also estimated fluxes at the
18 sea-air and sediment-water interfaces to determine the amount of CH₄ released by the
19 ECS into the atmosphere and the amount that escaping from sediments into the water
20 column. We used a box model to calculate the preliminary CH₄ budget in the ECS,
21 identify the main sources and sinks of dissolved CH₄ in this area, and estimate the
22 contribution of different sources to CH₄ in ECS quantitatively. The ultimate aims of
23 this research are to provide a deep understanding of dissolved CH₄ in the ECS, and to
24 improve our knowledge of CH₄ cycle in the shelf region.

25

26 **2 Materials and methods**

27 **2.1 Seawater sampling and analysis**

28 Five cruises were conducted in the ECS and YS during 2011 to collect data at

1 different locations (Fig. 1, Table 1). Seawater samples were collected at different
2 depths using 5-L or 8-L Niskin bottles mounted to a Sea-Bird CTD rosette. Surface
3 waters were collected at a depth of ~2 m, and bottom waters were typically collected
4 at ~3 m above the seafloor. Two subsamples for CH₄ determinations were transferred
5 from the Niskin bottles into glass vials (~117 mL) using a silicone tube. After
6 overflow of approximately 1.5- to 2-fold of bottle volume, 1 mL of a saturated
7 solution of HgCl₂ was added to inhibit microbial activity. Then, the sample bottle was
8 immediately sealed with a butyl rubber stopper and an aluminum cap (to exclude
9 excess water) and stored upside down in a dark box (Zhang et al., 2008a). All water
10 samples were analyzed after return to the laboratory, within 60 days after collection
11 (Zhang et al., 2004). Salinity and temperature data were measured by the CTD, and
12 oxygen data from CTD profiles were calibrated with oxygen measurements from the
13 Winkler titration method (Bryan et al., 1976). Wind speeds were measured by the
14 shipboard automatic weather stations at about 10 m above the sea surface.

15 Dissolved CH₄ from seawater samples was measured using a gas-stripping method
16 and a GC-14B gas chromatograph (Shimadzu, Japan) with a flame ionization detector
17 (FID) (Zhang et al., 2004). FID responses were calibrated using known volumes of
18 CH₄ standards (2.02, 4.18, and 50.0 ppmv, Research Institute of China National
19 Standard Materials). There was a linear relationship between FID response and CH₄
20 concentration, so a multi-point calibration method was used to determine CH₄
21 concentration based on chromatographic peak area. The precision of this method was
22 better than 3% during the routine analysis of the seawater samples (Zhang et al.,
23 2004).

24 **2.2 Sediment sampling and incubation experiments**

25 The emission of CH₄ from sediments was measured by the closed chamber
26 incubation method (Barnes and Owens, 1999). Sediment samples were collected by a
27 box corer at different sampling stations (Fig. 1, red triangles). Only samples with
28 undisturbed sediment surfaces were used. At each station, 15 sediment cores were
29 collected using plexiglass tubes that had openings on both ends (i.d.= 5 cm, height =

1 30 cm), with the bottoms sealed using air-tight rubber bungs. Just prior to the
2 beginning of flux measurements, ambient bottom water was added carefully (with no
3 gas headspace), and then the core was capped with a plexiglass top that had gas-tight
4 O-ring seals and two sampling ports. All cores were arranged around a central shaft
5 which was supported by magnets that rotated at 60 rpm, and were placed in a
6 water-filled tank that was held at ambient room temperature using a recirculating
7 water temperature-controlled bath. Ten glass bottles filled with ambient bottom water
8 were placed in the same tank and used as blank. Cores were incubated in the dark for
9 24~48 h. Three overlying water samples were collected at 4-h to 8-h intervals,
10 transferred into 56.5 mL glass bottles, and treated with 0.5 mL HgCl₂ to inhibit
11 microbial activity. At the same time, two bottled water samples were also treated with
12 0.5 mL HgCl₂ as a blank. The CH₄ concentrations of all samples were measured by
13 the gas-stripping method described above. Sediment-water CH₄ flux was estimated
14 from the slope of the CH₄ increase in the overlying water as a function of time. The
15 effect of temperature discrepancy (usually 0.8-10.5°C) on the CH₄ emission rate from
16 sediments was corrected by the Arrhenius empirical equation (Aller et al., 1985; Song
17 et al., 2015). When temperature increases by 10 °C, the chemical reaction rate (here
18 referring to CH₄ production and consumption rate) will increase by 2~4 times. We
19 took 3 times for calibrating the calculation of sediment-water CH₄ fluxes accordingly.
20 Use of the Arrhenius equation for temperature correction is usually reasonable and
21 acceptable when an incubation experiment is not conducted at the *in situ* temperature
22 (e.g. Aller et al., 1985; Song et al., 2015).

23 **2.3 Saturation and sea-to-air flux calculations**

24 The saturation (R , %) and sea-to-air flux (F , $\mu\text{mol}\cdot\text{m}^{-2}\cdot\text{d}^{-1}$) of CH₄ were calculated
25 by the following formulas:

$$26 \quad R (\%) = C_{obs} / C_{eq} \times 100 \quad (1)$$

$$27 \quad F = k \times (C_{obs} - C_{eq}) \quad (2)$$

28 where C_{obs} is the observed concentration of dissolved CH₄ and C_{eq} is the

1 air-equilibrated seawater CH₄ concentration, calculated from the *in situ* temperature
2 and salinity and the solubility data of Wiesenburg and Guinasso (1979). Atmospheric
3 CH₄ was not measured in this study. Instead, an annual mean atmospheric CH₄ mixing
4 ratio of 1902 ppb at three observation stations near the East China Sea (LLN, TAP and
5 SDZ) in 2011, from the NOAA/ESRL Global Monitoring Division in situ program
6 (<http://www.esrl.noaa.gov/gmd>), was used for all calculations. Although seasonal
7 variations in atmospheric CH₄ mixing ratios were detected in these stations due to
8 terrestrial influence, they make a minor difference in the results of sea-air fluxes. In
9 formula 2, k stands for the gas transfer coefficient, which is a function of wind speed
10 and Schmidt number (Sc), generally estimated by the empirical equations. Various
11 empirical equations were published to estimate k . Nightingale et al. (2000) reported an
12 evaluation of sea-air gas exchange in coastal ocean, and the corresponding value lies
13 near the median of extensive methods and models (Cockenpot et al., 2015).
14 Wanninkhof (2014) improved the methodology and updated the relationship between
15 gas exchange and wind speed based on his previous research over the last two decades
16 (Wanninkhof, 1992). The new relationship between k and Sc was constructed using
17 the modified global ocean ¹⁴C inventories and improved wind speed products, and it
18 can be well applied to the gas exchange study at the intermediate winds of 4-15 m/s.
19 Both methods (hereafter N2000 and W2014) were chosen to calculate sea-air fluxes in
20 this paper.

21 **3 Results**

22 **3.1 Hydrography of the ECS**

23 The hydrography of the ECS is highly variable due to the influence of three main
24 water masses including the Changjiang Diluted Water, Taiwan Warm Current Water,
25 and Kuroshio (Su, 1998; Li and Su, 2000; Zhang et al., 2007; Qi et al., 2014). These
26 influences are evident in the temperature-salinity (T-S) curves along all sections
27 during May, October, and December (Figs. 2 and 3). The salinities at stations near the
28 Changjiang Estuary (*i.e.* C0, B1 in May; P01, F03 in October; F02, P01 in December)
29 were less than 32, due to the influence of Changjiang Diluted Water (CDW). In

1 contrast, the Kuroshio affected the T-S relationships at stations on the continental
2 slope (*i.e.* CJ, D1, D9, ED and G8 in May; P12, E10 in October; E10 in December).
3 Moreover, water columns of these stations could be divided into four layers according
4 to their T-S relationships: (*i*) Kuroshio surface water (KSW) was present at a depth of
5 0~50 m, and this water had high temperature and salinity; (*ii*) Kuroshio subsurface
6 water (KSSW) was at a depth of 50~300 m, and this water had a lower temperature
7 than the surface water; (*iii*) Kuroshio intermediate water (KIW) was present at a depth
8 of 300~800 m, and had lower salinity and temperature than the upper water; and (*iv*)
9 Kuroshio deep water (KDW) was at 800 m and lower, and this high-density water had
10 a temperature of about 5°C.

11 The Taiwan Warm Current Water (TWCW) also affected the mid-shelf of the ECS,
12 as indicated by the high temperature and a moderate salinity 33.0~34.5 psu in the
13 southwestern region of the ECS (stations T01, T03 in May; Z01, Z03 in October; Z02,
14 Z03 in December). The shelf water mixed with water from the CDW, TWCW, KSW,
15 and KSSW, and its salinity was 32~ 34 psu. High CH₄ levels were usually present in
16 areas impacted by the CDW, and low CH₄ levels were present in the water from the
17 Kuroshio Current.

18 **3.2 Seasonal variations of CH₄ in the ECS and YS**

19 Table 2 shows the range and average of temperature, salinity, and CH₄
20 concentrations in surface and bottom waters of the ECS and YS during the five cruises
21 of 2011. Considering the large spatial difference among different cruises, we divided
22 the sampling region into four subareas based on the boundary of the ECS and YS
23 (blue dashed line in Fig. 1), the line of salinity 30 (green curve in Fig. 1) and the
24 200m depth line (red curve in Fig. 1). They were the YS, the Changjiang Estuary
25 ($S \leq 30$), the ECS shelf ($S > 30$, $\text{depth} \leq 200$ m) and the ECS slope ($S > 30$, $\text{depth} > 200$ m).
26 The ECS shelf was surveyed during all cruises and was chosen to compare the
27 average surface and bottom CH₄ concentrations during different seasons (Fig. 4). It
28 can be seen that both surface and bottom CH₄ concentrations had obvious seasonal
29 variations, with the highest level occurring in summer and the lowest level in early

1 spring (March). The salinity during August was the lowest due to the surge of
2 Changjiang runoff in summer. CH₄ concentration in the Changjiang (135.3±36.9
3 nmol·L⁻¹) was usually about 10-30 times higher than that in the ECS, and this
4 enhanced CH₄ concentration in the ECS together with high CH₄ production with
5 increasing temperature. During the five cruises, bottom CH₄ concentrations were
6 usually higher than those at the surface, especially during summer.

7

8 **3.3 Geographic Distribution of CH₄ in the YS and ECS**

9 Figure 3 shows the geographic distributions of temperature, salinity, and CH₄ in
10 surface and bottom waters of the YS and ECS in 2011. Two cruises were in spring
11 (March, Fig. 3a; May, Fig. 3b). During May, surface and bottom water temperature
12 increased gradually from north to south, and temperature was relatively low (2~3°C)
13 in the bottom water on the edge of the ECS continental shelf. Surface and bottom
14 salinity increased gradually from the Changjiang Estuary to the southeast. Dissolved
15 CH₄ concentrations in surface and bottom waters gradually declined from the
16 Changjiang Estuary towards the open sea during spring. High CH₄ concentrations in
17 the surface water appeared near Changjiang Estuary (T05: 29.67 nmol·L⁻¹ in March;
18 C0: 21.38 nmol·L⁻¹ in May) due to the influence of the Changjiang Diluted Water.
19 CH₄ concentrations in the bottom were slightly higher than in the surface, and highest
20 levels were observed at F03 (30.63 nmol·L⁻¹) and F04 (19.58 nmol·L⁻¹) during March
21 and at B1 (17.81 nmol·L⁻¹) and D1 (20.01 nmol·L⁻¹) during May. In the southeastern
22 continental shelf of the ECS, CH₄ concentrations were relatively low (about 2~3
23 nmol·L⁻¹), mainly due to the influence of the CH₄-depleted Kuroshio surface water.
24 This is consistent with the results reported by Rehder and Suess (2001) and
25 Tsurushima et al. (1996).

26 Temperature in the surface and bottom waters increased from north to south during
27 August (Fig. 3c). Salinity had a similar trend with spring, but CDW had an obvious
28 extension in the ECS during summer, and surface salinity was below 32 at most

1 regions of the continental shelf. Dissolved CH₄ increased with increasing temperature
2 and freshwater discharge during summer (about 33,484 m³·s⁻¹, more than 2-times than
3 during spring; *Changjiang Sediment Bulletin*, 2011). The mean surface and bottom
4 CH₄ concentrations were 8.21 nmol·L⁻¹ and 11.88 nmol·L⁻¹, respectively. Just as in
5 spring, high CH₄ concentrations in the surface and bottom waters were present near
6 the Changjiang Estuary and outside Hangzhou Bay. Particularly, high bottom CH₄
7 concentrations (10.51~12.48 nmol·L⁻¹) were observed in the high turbidity zone of
8 the Changjiang Estuary (~122°00'-122°20'E, 30°50'-31°15'N; Shen, 2012), together
9 with low oxygen concentrations (2.10~2.82 mg·L⁻¹). Besides, high temperature during
10 summer may lead to water stratification, which prevents dissolved CH₄ in bottom
11 waters from diffusing into upper waters, and thereby further enhance the CH₄
12 accumulation in the bottom water.

13 During October, surface seawater temperature and salinity increased gradually from
14 northwest to southeast (Fig. 3d). Bottom temperatures in the ECS were almost all in
15 the range of 19~22°C, but there was a cold bottom center (below 10°C) on the
16 southeastern region of the YS. Water with high temperature and salinity at the
17 southeastern corner of the studied area might have been affected by the northward
18 branch of the Kuroshio. CH₄ concentrations during autumn were significantly lower
19 than during summer. Bottom CH₄ concentrations of the entire ECS shelf were
20 relatively high (above 8.0 nmol·L⁻¹), especially at coastal areas, *i.e.* the Changjiang
21 Estuary (P01, P03), the mouth of Hangzhou Bay (T05), and the surrounding waters of
22 Jeju Island (A10, D07). On the contrary, CH₄ concentrations were quite low (about 3
23 nmol·L⁻¹) in the southeastern part of the ECS continental shelf due to the influence of
24 CH₄-depleted Kuroshio surface water.

25 The geographic distributions of surface and bottom temperature and salinity during
26 December (Fig. 3e) were similar to that recorded during October, but CDW only had a
27 slight influence. The CH₄ level of the whole ECS and YS during December was far
28 below the levels during August and October, and the average CH₄ concentration in the
29 surface waters (4.07 nmol·L⁻¹) was slightly lower than that in the bottom waters (4.53

1 nmol·L⁻¹). Surface CH₄ concentrations in the YS were slightly higher than those in the
2 ECS, and high CH₄ concentrations occurred in the southern YS near Cheju Island,
3 while sporadically high levels of CH₄ in the bottom waters mainly occurred near the
4 continental slope in the ECS. To sum up, CH₄ concentrations in the surface and
5 bottom waters of the YS and ECS during winter were uniform and stable, and were
6 3~5 nmol·L⁻¹ in most regions.

7 **3.4 Depth Distribution of CH₄ in the ECS**

8 Section PN (red solid line in Fig. 1) extends from the Changjiang Estuary southeast
9 towards the Okinawa Trough and across the CDW and the mainstream of the
10 Kuroshio. Section E (purple solid line in Fig. 1) lies across the entire shelf of the ECS,
11 and extends from the coastal area of Zhejiang and Fujian provinces east towards the
12 Okinawa Trough. We used these representative sections to study the hydrological and
13 chemical characteristics of the ECS. Here we choose the top buoyant water (depth less
14 than 200 m) to analyze the depth distribution of dissolved CH₄ on the ECS shelf.

15 Figure 5 shows the depth distributions of temperature, salinity, and CH₄ along
16 section PN during March, May and October, and section E during December.
17 Seawater temperature and salinity gradually increased with distance from the shore,
18 but the depth profiles had seasonal variations. During early spring (March, Fig. 5a)
19 and winter (December, Fig. 5d), the water column in the middle of ECS shelf was
20 almost well-mixed in the top 100 m, and temperature and salinity along section PN
21 were nearly uniform from the surface to the bottom; however, the depth profiles of
22 temperature and salinity were stratified during late spring (May, Fig. 5b) and autumn
23 (October, Fig. 5c). Perennial stratification in the water column occurred in the
24 Changjiang Estuary, while water column stratification in the middle shelf began to
25 occur during late spring, faded during the fall and disappeared completely during
26 December.

27 Correspondingly, dissolved CH₄ concentrations along section PN and section E
28 gradually decreased with distance from the shore, and the maximum CH₄

1 concentration of surface water was near the shore. During March and December, CH₄
2 concentrations were relatively uniform from surface to bottom, but they increased
3 gradually with depth during May and October. Particularly, high bottom CH₄ values
4 were usually observed at stations close to the continental shelf, especially in the shelf
5 break area. The bottom CH₄ concentration at P09 during October reached 12.16
6 nmol·L⁻¹, almost 2-fold higher than the surface level.

7 Figure 6 shows depth profiles of seawater temperature, salinity, and CH₄
8 concentrations at stations CJ, P12, and E10 (red circles in Fig. 1) located at the
9 continental slope where the mainstream of the Kuroshio flows northeastward along
10 the 200 m isoline. The mixed layers at the sloping region ranged between 30 and 120
11 m in depth and became deeper from spring to winter, below the mixed layer water
12 temperature decreased gradually. Salinity showed maximum at around 100-200 m,
13 then declined sharply and reached minimum at about 500 m, below which a slight
14 increase occurred with depth. All CH₄ concentrations increased initially with depth,
15 and reached maximum at around 200 m, then decreased to yield a minimum at 500 m.
16 The first CH₄ peak (3-6 nM) may be explained by advective supply from the adjacent
17 continental slope, where bottom waters usually contain high levels of CH₄ (Ye et al.,
18 2015). The second CH₄ peak occurred at a depth of 600 m during May and at 800 m
19 during October. CH₄ concentrations further increased below 800 m during May and
20 December, suggesting the existence of CH₄ sources in sediments.

21 **3.5 Sediment-water CH₄ fluxes in the YS and ECS**

22 Sediment-water fluxes of CH₄ from the ECS and YS had an obvious seasonal
23 variation, with the maximum occurring in summer (Fig. 7) during which the flux was
24 about 2-times higher than the other seasons. CH₄ was emitted from the sediments at
25 most stations and acted as a net source of CH₄ in the water column. CH₄ release from
26 sediments also had obvious spatial and temporal variation. The sediment incubation
27 experiments (“sample” in supplementary Figure 1) at P01 indicated that the CH₄
28 concentration in the overlying waters increased linearly with incubation time (t)
29 (March: [CH₄] = 0.59×t + 6.33, r² = 0.73; October: [CH₄] = 0.19×t + 2.64, r² = 0.85;

1 December: $[\text{CH}_4] = 0.23 \times t + 4.37$, $r^2 = 0.95$), and the sediment-water CH_4 flux at P01
2 was $1.93 \mu\text{mol} \cdot \text{m}^{-2} \cdot \text{d}^{-1}$ during March, $0.72 \mu\text{mol} \cdot \text{m}^{-2} \cdot \text{d}^{-1}$ during October and 1.60
3 $\mu\text{mol} \cdot \text{m}^{-2} \cdot \text{d}^{-1}$ during December, respectively. Station C1 (near the Changjiang river
4 mouth) had a sediment-water CH_4 flux of $2.94 \mu\text{mol} \cdot \text{m}^{-2} \cdot \text{d}^{-1}$ during August, much
5 higher than any other sites; this might be ascribed to the low oxygen concentration in
6 the overlying water ($3.25 \text{ mg} \cdot \text{L}^{-1}$ based on Winkler titration) and the rich organic
7 carbon in the sediment (Lin et al., 2002; Kao et al., 2003).

8 Due to the limit of weather and lab resources, our sediment collection in August
9 2011 mainly covered the coastal areas. To avoid spatial bias induced by our sampling
10 area, we added the sediment-water CH_4 fluxes obtained from a cruise in August 2013
11 (Figure 1, stations labelled by the black star) to provide data for the shelf and slope
12 regions. As estimated, average sediment-water CH_4 flux from the ECS and YS was
13 about 1.06 and $0.73 \mu\text{mol} \cdot \text{m}^{-2} \cdot \text{d}^{-1}$ in 2011, respectively. Based on their surface areas
14 (about 7.7×10^5 and $3.8 \times 10^5 \text{ km}^2$), the annual CH_4 emission from sediments of the
15 ECS and YS in 2011 was about 2.98×10^8 and $1.01 \times 10^8 \text{ mol}$, respectively. This
16 indicated that sediments were an important source of CH_4 in the bottom waters.
17 However, the estimate of CH_4 released from the sediment had great uncertainties
18 because of the scant sampling stations in each survey and the high spatial variation
19 of sediment-water CH_4 flux.

20 **3.6 Sea-air CH_4 fluxes in the YS and ECS**

21 To eliminate the influence of large spatial difference, we divided the sampling
22 region into four subareas, estimated the area using grid method and calculated the
23 sea-air fluxes with N2000 and W2014 for each subarea. The CH_4 saturation and
24 average area-weighted sea-air flux density were shown in Table 3. CH_4 saturation in
25 the YS had an obvious seasonal variation, with that in autumn much higher than in
26 spring. However, Yang et al. (2010) reported that average surface CH_4 saturation in
27 the YS was highest ($515.2\% \pm 231.5\%$) during August. Surface CH_4 saturations in the
28 ECS also showed seasonal variation, with the maximum CH_4 saturation occurring in
29 summer, followed by late spring, autumn, winter and early spring. These results were

1 consistent with the results reported by Ye et al. (2015). Highest CH₄ saturation was
2 observed at the Changjiang Estuary. We recorded the highest CH₄ saturation at station
3 T05 (1007%) in March, station C0 (858%) in May, and station E01 (1558%) in
4 August. In general, the surface waters of the YS and ECS were all oversaturated with
5 atmospheric CH₄, except for some sporadic stations during spring. Thus, the YS and
6 the ECS were net sources of atmospheric CH₄.

7 Sea-air CH₄ fluxes calculated using the N2000 equation were quite similar to those
8 estimated from the W2014 equation, and they also showed seasonal and spatial
9 variations in the wide ECS shelf, with the highest CH₄ flux occurring in the late
10 spring and the lowest in early spring. In August, although the estuary only covered 25%
11 of total observation area, it was responsible for about 46% of total CH₄ emission from
12 ECS. It indicated that sea-air exchange of CH₄ in coastal areas was extremely intense,
13 much greater than on the shelf area. Sea-air CH₄ flux density in the shelf water of the
14 ECS was about 11.61 μmol·m⁻²·d⁻¹ during summer, which was much higher than the
15 previous research (merely 2.81–6.89 μmol·m⁻²·d⁻¹) by Zhang et al. (2008a). According
16 to N2000 and W2014 equation, the annual average area-weighted sea-air CH₄ flux
17 density of the ECS and YS was about 9.75 μmol·m⁻²·d⁻¹ during 2011. Based on the
18 areas of the YS (~380000 km²) and ECS (~770000 km²), we estimated the total CH₄
19 emission from them as 4.09×10⁹ mol (about 0.065 Tg) during 2011. Bange (1994)
20 estimated the global oceanic CH₄ emission was 11–18 Tg CH₄·year⁻¹, so the YS and
21 ECS accounted for about 0.45% of the global oceanic emission. This value was much
22 higher than its corresponding area proportion of 0.32%, indicating that the YS and
23 ECS are active areas for CH₄ production and emission.

24 **4 Discussion**

25 **4.1 Factors influencing CH₄ distribution in the ECS and YS**

26 The concentration, saturation, and sea-air and sediment-water fluxes of CH₄ in the
27 ECS and YS all had obvious seasonal variations. Mean CH₄ saturation (R_{CH₄}) had a
28 linear correlation with mean water temperature (T) in the surface water (R_{CH₄} =

1 13.91×T – 30.10, $r^2 = 0.77$; Supplementary Figure 2) and bottom water ($R_{CH_4} =$
2 33.50×T – 225.03, $r^2 = 0.76$; Supplementary Figure 2) during different seasons.
3 Average sediment-water fluxes of CH₄ showed a weak correlation with average T
4 ($F_{CH_4}=0.06T+0.03$, $R^2=0.3$), suggesting that CH₄ emission from the sediments
5 increases with temperature. It was reported that CH₄ production rate increases with
6 temperature in the range of 0-30°C (Liikanen et al., 2002; Glissmann et al., 2004).
7 Besides, the high temperature may enhance the relative abundance and diversity of
8 methanogenic communities (Høj et al., 2008; Metje and Frenzel, 2005).
9 Yvon-Durocher et al. (2014) reported seasonal variations of CH₄ emissions from
10 diverse ecosystems using meta-analysis, and showed that CH₄ emissions increased
11 significantly with seasonal increases of temperature due to increase of CH₄ production
12 from methanogens and anaerobic microbial communities. Our results were consistent
13 with these previous studies, and supported the view that water temperature played an
14 important role in regulating the seasonal variation and distribution of CH₄ in the ECS
15 and YS.

16 The sediment incubation experiments and the depth profile of CH₄ in the ECS all
17 indicated that sediment release was a significant source of CH₄ in bottom waters,
18 especially at the shelf break area. Lin et al. (1992) found that organic carbon
19 concentrations in bottom sediments increased across the shelf break, and were greatest
20 in sediments at depths of 1000-1500 m in the ECS, suggesting that high CH₄ in
21 bottom waters may come from high CH₄ production and subsequent release from the
22 organic-rich sediments. Previous studies also showed that many submarine mud
23 volcanoes and hydrothermal vents occur along the continental slope of the ECS (Zhao
24 et al., 2006; Kawagucci et al., 2011). Methane-containing fluid was episodically
25 vented, then transported to the water column. CH₄ in the hydrothermal fluid has been
26 considered as a thermogenic origin and the CH₄ level may be 10⁴-10⁷ times higher
27 than those in the ambient seawater (Kawagucci et al., 2011). When released from the
28 vent, it forms buoyant plumes rapidly (Tsunogai et al., 2000). During the dilution,
29 CH₄ can be oxidized quickly in the plume due to the microbial oxidation (De Angelis

1 et al., 1993). The residual CH₄ spreads into the upper seawater. Hence, CH₄ release
2 from sediments, mud volcanoes and hydrothermal fluids may also influence the CH₄
3 distribution in the bottom waters, especially at the continental slope.

4 CH₄ formation and consumption in the water column are also important factors that
5 influence the distribution of CH₄ in the ECS and YS. Subsurface methane maxima
6 were observed in this study, which have been considered to be a common
7 phenomenon in the open sea (Reeburgh, 2007). Early studies demonstrated that
8 advective transport of CH₄ or *in situ* CH₄ production by microbes in anoxic
9 microenvironments led to excess CH₄ in the mixed layer (Scranton and Brewer, 1977;
10 De Angelis and Lee, 1994; Karl and Tilbrook, 1994). More recent results suggest that
11 under conditions with rich oxygen and specific nutrient limitation, a variety of
12 methyl-rich organic phosphorus or sulfur compounds were all likely to be utilized by
13 microorganisms and served as precursors of CH₄ production (Karl et al., 2008; Damm
14 et al., 2008; Zindler et al., 2013; Florez-Leiva et al., 2013). The bottom water
15 incubation experiments (“blank” in section 2.2 and in Supplementary Figure 1) at
16 some stations (L1 in May, J1 in August, A10 and A02 in October and R07 in
17 December) in this study indicated that CH₄ concentration increased linearly with
18 incubation time. This suggested that *in situ* CH₄ production might be another source
19 of the excess CH₄ in the bottom waters. Moreover, CH₄ production in the water
20 column can be enhanced by hypoxia (Ye et al., 2015). Extremely high bottom CH₄
21 values (20-26 nM) were found near the Changjiang Estuary and outside Hangzhou
22 Bay during August 2011, together with low oxygen level (2.00-4.00 mg/L) and high
23 particulate organic carbon flux (3900-7300 mg C·m⁻²·d⁻¹; Hung et al., 2013). Both
24 high bottom water production and sediment release may contribute to this.

25 Riverine input plays an important role in regulating the distribution of CH₄ in the
26 YS and ECS. CH₄ concentration in the river water of Changjiang was about 10-40
27 times higher than that of the ECS (Zhang et al., 2004; 2008), and the freshwater
28 discharge from the Changjiang during summer was much greater than during winter
29 (Table 4), therefore the widespread dispersal of Changjiang plume had a great impact

1 on CH₄ distribution, especially in summer. Similar to the CDW, the Kuroshio and
2 TWCW had different effects on the ECS during summer and winter. During summer,
3 water input from the TWCW ($2.39 \times 10^6 \text{ m}^3 \cdot \text{s}^{-1}$) was much greater than that from the
4 Kuroshio ($0.89 \times 10^6 \text{ m}^3 \cdot \text{s}^{-1}$) (Zhang et al., 2007), and the CH₄ level of the TWCW
5 ($6.11 \text{ mol} \cdot \text{L}^{-1}$, Ye et al., 2015) was almost two-fold of that in the Kuroshio (2.91
6 $\text{mol} \cdot \text{L}^{-1}$). Thus, the TWCW had a greater influence on CH₄ distribution in the ECS
7 than the Kuroshio during summer, while they had comparable influence during winter
8 due to similar water discharges and CH₄ concentrations. Thus, the mixing of different
9 water masses and their seasonal variations directly impact CH₄ distribution in the
10 ECS.

11 Submarine groundwater discharge (SGD) has been recognized as an important
12 pathway for material transport to the marine environments (Burnett et al., 2006). It
13 usually contains high levels of CH₄, and can serve as an important CH₄ source for the
14 coastal ocean (Bugna et al., 1996; Corbett et al., 2000; Santos et al., 2009; Lecher et
15 al., 2015). SGD is a mixture of fresh groundwater and recirculated seawater, and there
16 is no CH₄ data available for either of them for the ECS. We collected some
17 groundwater samples as fresh groundwater end-member along the coast of the ECS in
18 December 2011 and July 2012. During December 2011, groundwater samples were
19 collected at twenty-five sites along the Jiangsu and Zhejiang coastal region
20 ($120.857 \sim 121.896^\circ \text{E}$, $30.124 \sim 30.956^\circ \text{N}$). CH₄ concentrations varied significantly at
21 these stations with a great range of 33~61602 nM and a median of 271 nM (Zhang et
22 al., unpublished data). During July 2012, eight groundwater samples for CH₄
23 measurement were collected along the coast of the ECS ($121.371 \sim 121.934^\circ \text{E}$,
24 $30.733 \sim 31.976^\circ \text{N}$). CH₄ concentrations ranged between 138 and 3428 nM with a
25 median of 758 nM. CH₄ concentrations in fresh groundwater end-member along the
26 coast of the ECS were much higher than those in the seawater. Hence submarine
27 ground water discharge might be an important CH₄ source for the East China Sea and
28 could influence its CH₄ distribution.

29 **4.2 Preliminary estimate of CH₄ budget in the ECS**

1 In order to quantify the contributions of different sources and sinks to dissolved
2 CH₄ in the ECS, CH₄ budget was estimated preliminary based on data presented here
3 and collected from previous research. Zhang et al. (2007) re-estimated the nutrient
4 budget of the ECS using a box model during summer and winter. According to mass
5 conservation, the water and salinity balance of the ECS can be expressed as:

$$6 \quad \sum_{i=1}^n WFi + \Delta Q = 0 \quad (3)$$

$$7 \quad \sum_{i=1}^n WFi \times Ci + \Delta Mi = 0 \quad (4)$$

8 where WFi represents the water flux of inputs (+) and outflows (-) over the shelf; ΔQ
9 is the water mass distinction caused by sea level change in the ECS; Ci is the salinity
10 or value of a certain element for a known water mass; and ΔMi is the increase or
11 decrease of the given element during exchange at the sediment-water and sea-air
12 interface. The inflow to the ECS Shelf includes water from the Taiwan Strait
13 (TWCW), riverine input from the land-mass (of which the Changjiang accounts for
14 about 90–95% or more), and incursion of the Kuroshio from north of Taiwan and over
15 the broad shelf mainly composed of KSW (~25%) and KSSW (~75%) (Zhang et al.,
16 2007). The exchange between the ECS and the YS is taken into account as well. The
17 outflow refers to currents through the Tsushima/Korea Strait (Zhang et al., 2007).

18 As with the shelf water budget in the ECS (Zhang et al., 2007), we estimated the
19 budget of dissolved CH₄ for summer and winter. The Kuroshio and TWCW were
20 hardly observed during our summer cruise because of the limited investigation region,
21 so the corresponding data were from the literature (Ye et al., 2015) and unpublished
22 data for the ECS that was collected in July 2013. Data for riverine CH₄ were from our
23 on-going monitoring project at station Xuliujing (121°2′ E, 31°46′ N), which is the
24 most downstream main channel station at the Changjiang. Considering that the sea-air
25 fluxes values from W2014 and N2000 relationships were quite similar, we took those
26 estimated by W2014 for budget calculation. Submarine groundwater discharge (SGD)
27 data for the ECS is rather limited. Gu et al. (2012) estimated the SGD in the
28 Changjiang effluent plume to be about $(0.2-1.0) \times 10^9 \text{ m}^3 \text{ d}^{-1}$ during summer and no

1 SGD data is available for the whole ECS in literature yet. Hence it is difficult to
2 estimate the CH₄ input to ECS via SGD quantitatively. Fortunately, Prof. Jinzhou Du
3 from East China Normal University provided us their update and unpublished SGD
4 data obtained using Ra isotopes as tracers for the ECS (including submarine fresh
5 water discharge and recirculated saline groundwater discharge), which is $0.68 \times 10^8 \text{ m}^3$
6 d^{-1} and $0.46 \times 10^9 \text{ m}^3 \text{ d}^{-1}$ in dry and wet season, respectively (Du et al., unpublished
7 data, personal communication). It is hard to determine CH₄ concentrations in the
8 recirculated seawater, so we assume that CH₄ concentrations in the fresh groundwater
9 and recirculated saline groundwater are the same. Median CH₄ concentration was
10 chosen for calculation since it is less susceptible to abnormally high CH₄
11 concentrations observed in the groundwater. Hence, CH₄ fluxes via submarine
12 groundwater discharge were estimated by multiplying the median CH₄ concentration
13 by the SGD, which yielded a flux of 0.21 mol/s and 4.01 mol/s for winter and summer,
14 respectively. SGD is a mixture of fresh groundwater and recirculated seawater, among
15 which the latter could account for 90% of the discharge or more (Burnett et al., 2006),
16 but considering CH₄ concentration in the porewaters ($\sim 0.45 \text{ } \mu\text{M}$, Zhang et al.,
17 unpublished data) of the ECS usually is much lower than those in fresh groundwater,
18 the above estimation of CH₄ flux via submarine groundwater discharge may be
19 overestimated to some extent.

20 Table 4 lists the discharges of various water masses and their CH₄ concentrations,
21 and Figure 8 shows a preliminary estimate of the CH₄ budget in the ECS. We ignored
22 the effects of evaporation and precipitation because of unavailable data and the low
23 water fluxes of these processes. During summer, the TWCW transported $14.60 \text{ mol} \cdot \text{s}^{-1}$
24 of CH₄ to the ECS via Taiwan Strait, the Kuroshio transported $2.97 \text{ mol} \cdot \text{s}^{-1}$ of CH₄,
25 the CH₄ input from CDW is $3.45 \text{ mol} \cdot \text{s}^{-1}$, and sediment released $14.70 \text{ mol} \cdot \text{s}^{-1}$ of CH₄.
26 Groundwater might contribute $4.01 \text{ mol} \cdot \text{s}^{-1}$ CH₄ to the ECS, which is comparable to
27 the input via river runoff. However, CH₄ emission at the sea-air interface reached
28 $138.40 \text{ mol} \cdot \text{s}^{-1}$ and CH₄ export from the ECS was $18.12 \text{ mol} \cdot \text{s}^{-1}$. Consequently, to
29 maintain a balance of the CH₄ in the ECS, the rate of *in situ* net CH₄ formation (i.e.

1 CH₄ production-CH₄ oxidation) should be 116.73 mol·s⁻¹, which contributed more
2 than 70% of the CH₄ sources in summer.

3 During winter, the Kuroshio imported 5.28 mol·s⁻¹ of CH₄ into the ECS (1.8-fold
4 more than in summer), and the TWCW imported 5.20 mol·s⁻¹ of CH₄ (1/3 of that in
5 summer). Although the winter discharge of the Changjiang was almost equal with the
6 water inflow YS, CH₄ input by Changjiang (2.11 mol·s⁻¹) was significantly higher
7 than that from the YS (0.06 mol·s⁻¹). CH₄ emission from sediments was 8.38 mol·s⁻¹
8 and CH₄ input from groundwater was about 0.21 mol/s during winter (only 5% of that
9 in summer). CH₄ release from the ECS into the atmosphere was about 81.28 mol·s⁻¹
10 and the CH₄ export out of the ECS was 13.14 mol·s⁻¹. Thus, the net CH₄ production
11 rate of seawater during winter was inferred to be 73.18 mol·s⁻¹ to balance the CH₄ loss
12 by sea-air exchange and outflow.

13 However, the above results were merely a rough estimate with large uncertainties.
14 Firstly, there were potential errors involved in the measurements and calculations of
15 sea-air fluxes. Secondly, the estimation of CH₄ input via SGD was quite crude due to
16 the limited available data about SGD and groundwater CH₄. Thirdly, we performed
17 sediment incubations only at several stations during each survey, and the results were
18 far from representative of sediment emission from the whole ECS due to large spatial
19 and seasonal variations. Finally, some CH₄ sources and sinks were neglected in this
20 estimation. For example, aerobic CH₄ oxidation has been shown to be a substantial
21 sink of CH₄, however, no estimates of methane oxidation extents and rates were made
22 in the water column here. Seepage of thermogenic methane from the sediments was
23 also ignored due to lack of data. Hence the role of net in situ microbial production
24 might be exaggerated by estimating it as the difference between sources and sinks.
25 Although our method of estimation was not perfect, we demonstrated a variety of CH₄
26 sources and sinks for the ECS and roughly estimated their relative contribution. It
27 suggests that in situ production in the water column and sediment emissions are major
28 CH₄ sources in the marginal shelf seas, while sea-to-air release was the major external
29 sink of CH₄ in the ECS. Groundwater might be an important source of CH₄ in the

1 ECS, especially in wet seasons. Hence more research work on CH₄ production and
2 consumption as well as CH₄ in groundwater discharge is needed to further understand
3 the CH₄ budget in the ECS.

4 **5 Conclusions**

5 CH₄ distribution and emission in the ECS and YS had obvious spatial and seasonal
6 variations, and were also influenced by various factors, including mixing of different
7 water masses, water temperature, freshwater input, sediment release, hydrothermal
8 seepage and oxygen levels in the water column. We estimated the CH₄ budget of the
9 ECS using a box model, and the results indicated that in situ seawater production and
10 sediment release might be the major CH₄ source, while sea-air exchange was the
11 major external sink of CH₄ in the ECS. Groundwater might be an important source of
12 CH₄ in the ECS, especially in wet seasons. The ECS and the YS together was
13 estimated to release about 4.09×10^9 mol of CH₄ per year into the atmosphere, which
14 accounts for about 0.45% of the global oceanic emission and was much higher than its
15 corresponding area proportion of 0.32%. Hence the YS and ECS were active areas for
16 CH₄ production and emission.

17

18 **Author contributions**

19 M.-S. Sun collected and analyzed water samples in October and December 2011. G.-L.
20 Zhang designed the sampling strategy. M.-S. Sun and G.-L. Zhang prepared the
21 manuscript. X.-P. Cao collected and analyzed water samples in March, May, and
22 August 2011. X.-Y. Mao provided hydrological data for March, October and
23 December of 2011. J. Li provided hydrological data for May and August of 2011.
24 W.-W. Ye provided some data for calculation of the methane budget of the East China
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26

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1 Table 1. Characteristics of the five cruises in the East China Sea and the Yellow Sea.

2

Survey period	Research vessel	CTD	Station no.
13 Mar. - 10 Apr., 2011	Dong Fang Hong 2	Sea-Bird 911 plus	42
11 May - 7 June, 2011	Experiment 3	Sea-Bird 911 plus	54
10-30 August, 2011	Bei Dou	Sea-Bird 917	38
16 Oct. - 8 Nov., 2011	Dong Fang Hong 2	Sea-Bird 911 plus	55
20 Dec. 2011-7 Jan., 2012	Dong Fang Hong 2	Sea-Bird 911 plus	46

3

1 Table 2. Temperature, salinity, and CH₄ concentrations in surface and bottom waters
 2 of the Yellow Sea and East China Sea during five cruises in 2011. Numbers indicate
 3 ranges and average \pm SD (SD gives the average difference between the average value
 4 and the individual values).

5

Season	Depth	Temperature (°C)	Salinity (psu)	CH ₄ (nmol·L ⁻¹)
March	surface	4.32~19.03 (10.68 \pm 3.06)	29.79~34.60 (33.16 \pm 1.08)	2.39~29.67 (4.47 \pm 4.09)
	bottom	5.48~16.13 (10.06 \pm 2.40)	32.00~34.49 (33.26 \pm 0.75)	2.63~30.63 (5.10 \pm 4.97)
May	surface	16.40~26.17 (21.57 \pm 3.00)	30.45~34.59 (33.14 \pm 1.40)	1.88~26.39 (6.04 \pm 5.58)
	bottom	4.00~23.92 (15.78 \pm 4.62)	24.21~34.90 (33.07 \pm 1.99)	1.31~30.36 (6.41 \pm 5.11)
August	surface	21.93~28.25 (25.91 \pm 1.45)	22.57~33.99 (30.22 \pm 2.72)	3.71~33.62 (8.21 \pm 6.02)
	bottom	9.01~24.99 (20.75 \pm 2.98)	28.69~34.47 (33.02 \pm 1.62)	6.20~26.22 (11.88 \pm 4.59)
October	surface	17.89~26.18 (21.91 \pm 2.26)	28.06~34.46 (32.84 \pm 1.47)	2.44~13.52 (5.03 \pm 1.68)
	bottom	4.12~24.19 (17.38 \pm 5.43)	30.87~34.67 (33.41 \pm 1.23)	2.50~15.24 (7.51 \pm 2.93)
December	surface	7.94~23.46 (15.23 \pm 4.28)	29.87~34.70 (33.33 \pm 1.22)	3.01~6.03 (4.07 \pm 0.63)
	bottom	7.84~23.45 (14.83 \pm 4.08)	30.91~34.73 (33.48 \pm 1.07)	3.03~10.20 (4.53 \pm 1.33)

1 Table 3. Surface CH₄ saturation, sea-air CH₄ fluxes and the average area-weighted flux density in the Yellow Sea and East China Sea.

2

Region	Months	Area ratio	R(CH ₄)/%	U ₁₀ /m·s ⁻¹	F(W2014) (μmol·m ⁻² ·d ⁻¹)	F(N2000) (μmol·m ⁻² ·d ⁻¹)	average area-weighted flux density (μmol·m ⁻² ·d ⁻¹)
YS	Mar.	25%	94-148 (117)	2.2-12.1 (8.3)	-0.56-9.59 (2.31)	-0.55-9.09 (2.23)	0.57
	May	12.7%	94-161 (123)	6.4-10.3 (8.3)	-0.37-3.49 (2.05)	-3.8-3.64 (2.05)	0.26
	Aug.	---	---	---	---	---	---
	Oct.	41.2%	101-376 (218)	1.6-9.9 (5.1)	0.04-30.47 (6.42)	0.04-29.59 (6.59)	2.68
	Dec.	42.9%	105-238 (146)	2.9-13.6 (7.2)	0.31-16.58 (3.87)	0.39-15.52 (3.86)	1.62
Estuary	Mar.	0.1%	(1007)	(0.2)	(0.05)	(0.33)	1.9×10 ⁻⁴
	May	3.1%	161-858 (490)	1.3-11.3 (5.8)	0.15-50.49 (26.54)	0.27-48.23 (27.00)	0.83
	Aug.	26.1%	172-1558 (578)	1.9-8.9 (5.7)	2.25-118.23 (27.22)	3.06-120.19 (28.34)	7.25
	Oct.	2.2%	(558)	(8.7)	(51.27)	(50.69)	0.11
	Dec.	---	---	---	---	---	---
ECS shelf	Mar.	75%	91-340 (139)	0.3-13.5 (7.9)	-1.85-52.57 (4.67)	-1.75-49.61 (4.52)	3.45
	May	76.3%	87-1049 (252)	1.2-23.6 (9.0)	-4.30-138.96 (16.36)	-3.99-129.83 (15.79)	12.27
	Aug.	73.9%	195-528 (287)	3.3-8.4 (6.2)	1.66-24.99 (11.40)	2.00-26.09 (11.82)	8.58
	Oct.	52.9%	128-317 (215)	0.8-11.9 (7.3)	0.72-27.51 (10.50)	0.85-26.26 (10.42)	5.53
	Dec.	54.8%	128-213 (163)	3.7-14.1 (8.9)	1.26-23.49 (8.42)	1.47-22.15 (8.15)	4.54
ECS slope	Mar.	---	---	---	---	---	---
	May	7.9%	93-157 (128)	4.5-15.2 (9.7)	-0.97-17.47 (4.81)	-0.94-16.19 (4.55)	0.37
	Aug.	---	---	---	---	---	---
	Oct.	3.7%	186-211 (199)	8.4-10.1 (9.2)	11.44-12.84 (12.14)	11.37-12.44 (11.90)	0.44
	Dec.	2.3%	(232)	(11.7)	(25.84)	(24.59)	0.58

1 Table 4. Data used for calculation of the CH₄ budget in the East China Sea.

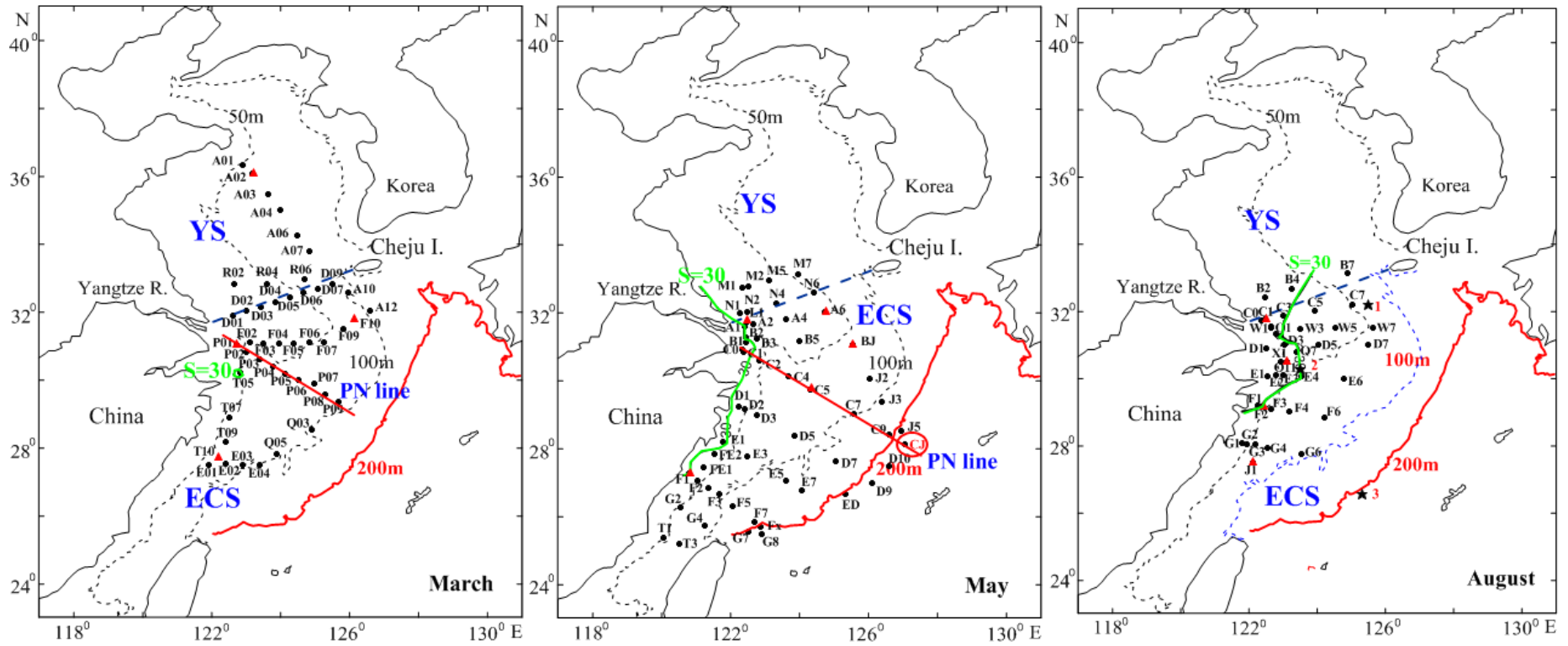
2

Parameter	Summer	Winter
Water exchange between ECS and YS (Sv)	-0.009 ^a	0.013 ^a
Evaporation (Sv)	-0.0068 ^a	-0.010 ^a
Rainfall (Sv)	0.031 ^a	0.0062 ^a
Terrestrial input (Sv)	0.0393 ^a	0.0122 ^a
Taiwan Strait water (Sv)	2.39 ^a	1.22 ^a
Kuroshio water (Sv)	0.89 ^a	1.81 ^a
Outflow of ECS (Sv)	-3.33 ^a	-3.05 ^a
Groundwater discharge (m ³ ·s ⁻¹)	5300	790
Avg. CH ₄ concentration in Changjiang (nM)	87.90 ^b	173.26 ^b
Avg. CH ₄ concentration in TWCW (nM)	6.11 ^c	4.26
Avg. CH ₄ concentration in Kuroshio (nM)	3.34 ^c	2.91
Avg. CH ₄ concentration in YS (nM)	6.56 ^d	4.33
Avg. CH ₄ concentration in shelf of ECS (nM)	5.44 ^c	4.31
Avg. sea-air CH ₄ flux (μmol·m ⁻² ·d ⁻¹)	15.53	9.12
Avg. sediment-water CH ₄ flux (μmol·m ⁻² ·d ⁻¹)	1.65	0.94
Median groundwater CH ₄ concentration (nM)	758	271

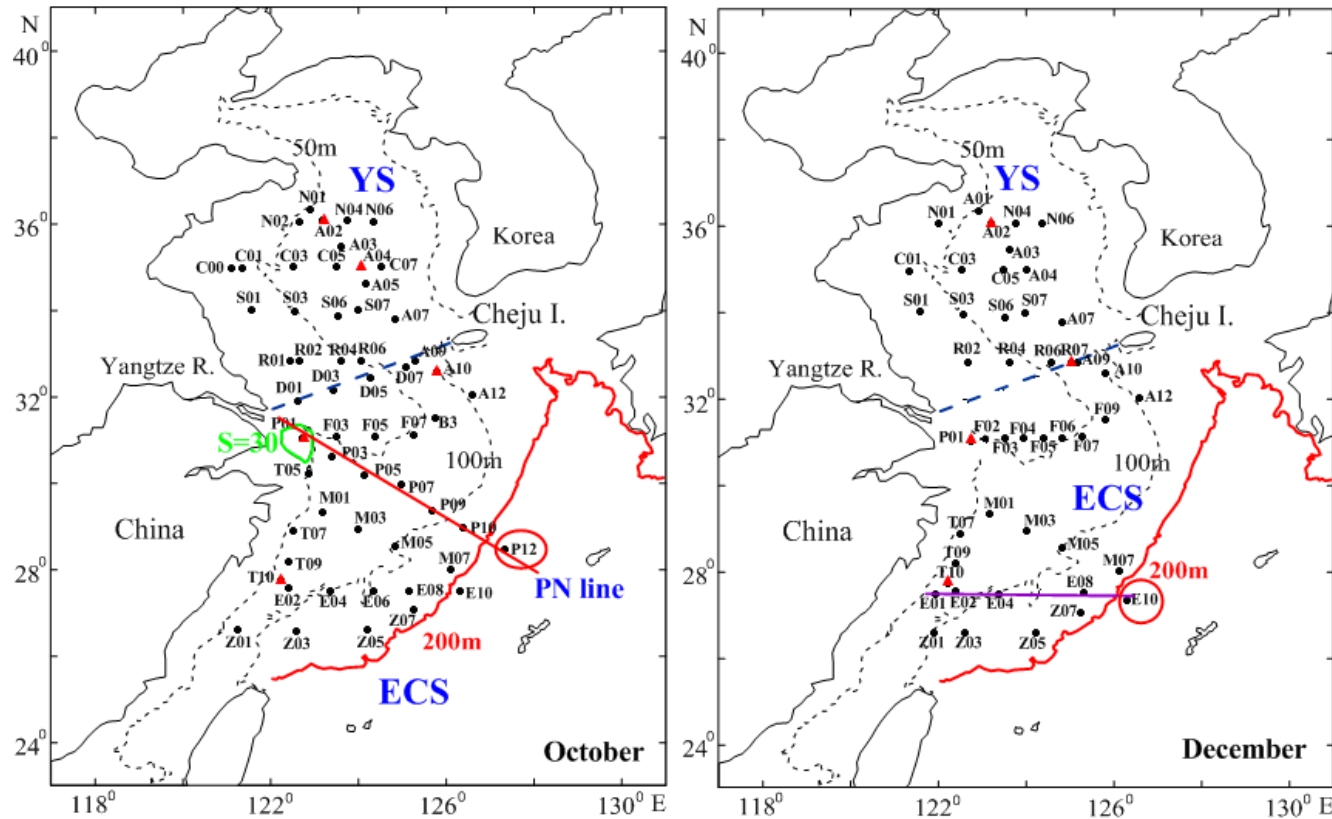
3

4 Note: Positive values represent water import from an external source and negative
 5 values represent water export from the ECS. 1 Sv = 10⁶ m³·s⁻¹; a. data from Zhang et
 6 al. (2007); b. unpublished data from an on-going monitoring project at Xuliujing
 7 station in the lower Changjiang; c. data from Ye et al. (2015); d. unpublished data
 8 from observations in July 2013.

1 **Figure captions:**

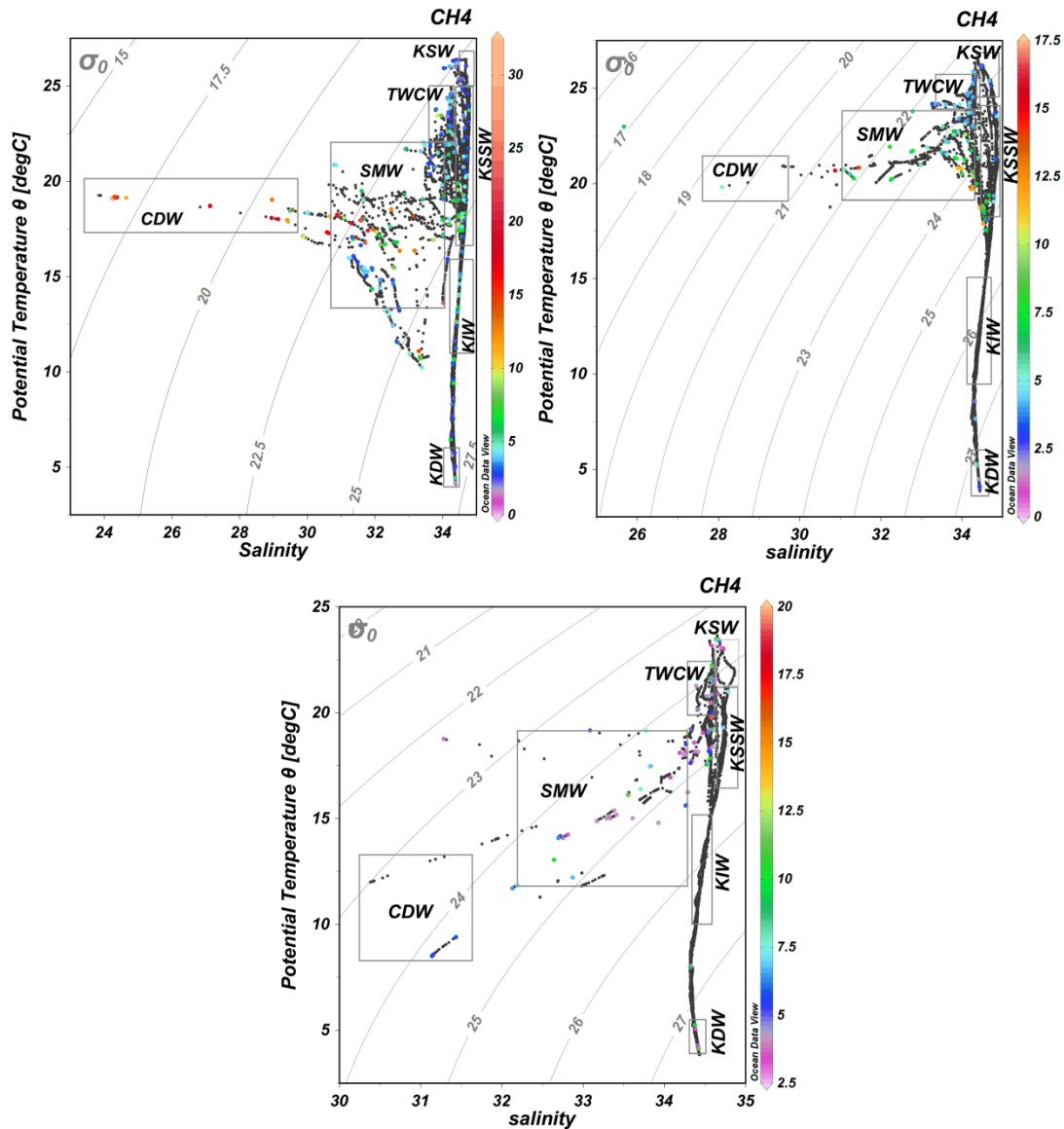


2



1
 2 Fig. 1. Sampling locations in the East China Sea and the Yellow Sea during the five cruises in 2011. Black solid dots: seawater sampling
 3 locations; red solid triangles: sediment sampling locations; red solid line: PN line; blue dashed line: boundary between the East China Sea and
 4 the Yellow Sea; purple solid line: section E; red solid circle: the station depth more than 1000 m; red curve: the depth of 200 m line; the red
 5 curve: the salinity of 30 line; black star in August: sediment incubation stations (1, 2, 3) in August 2013.

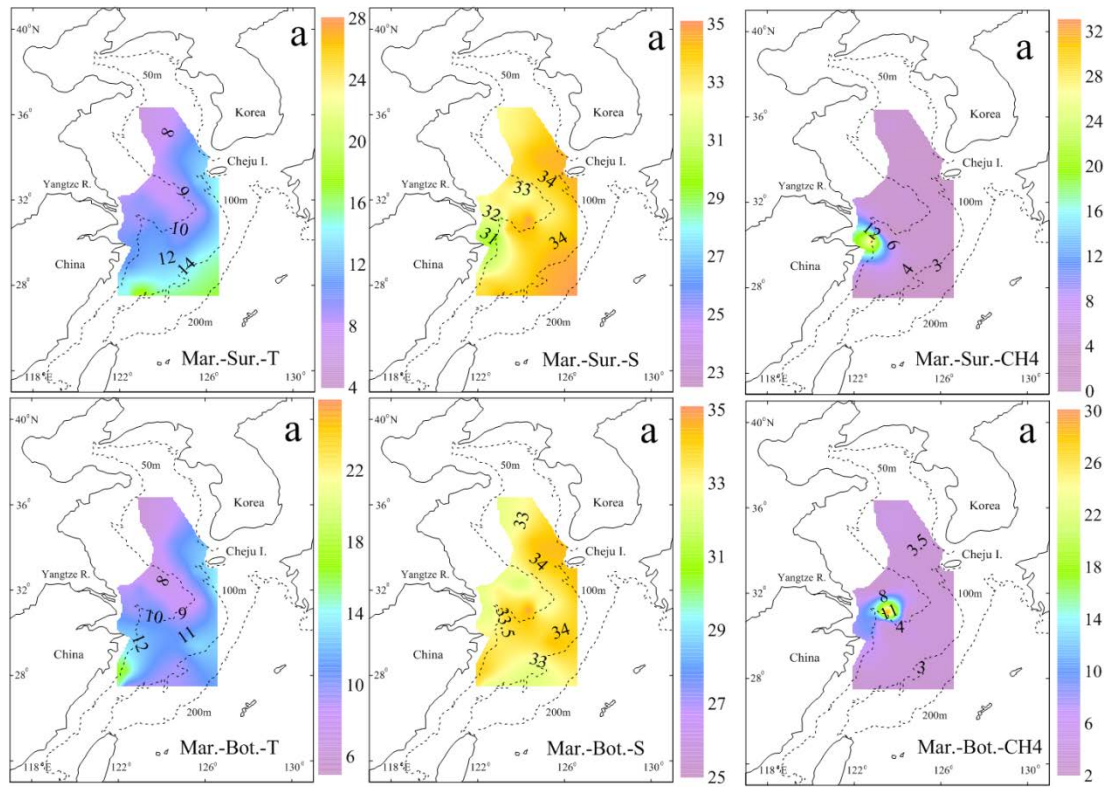
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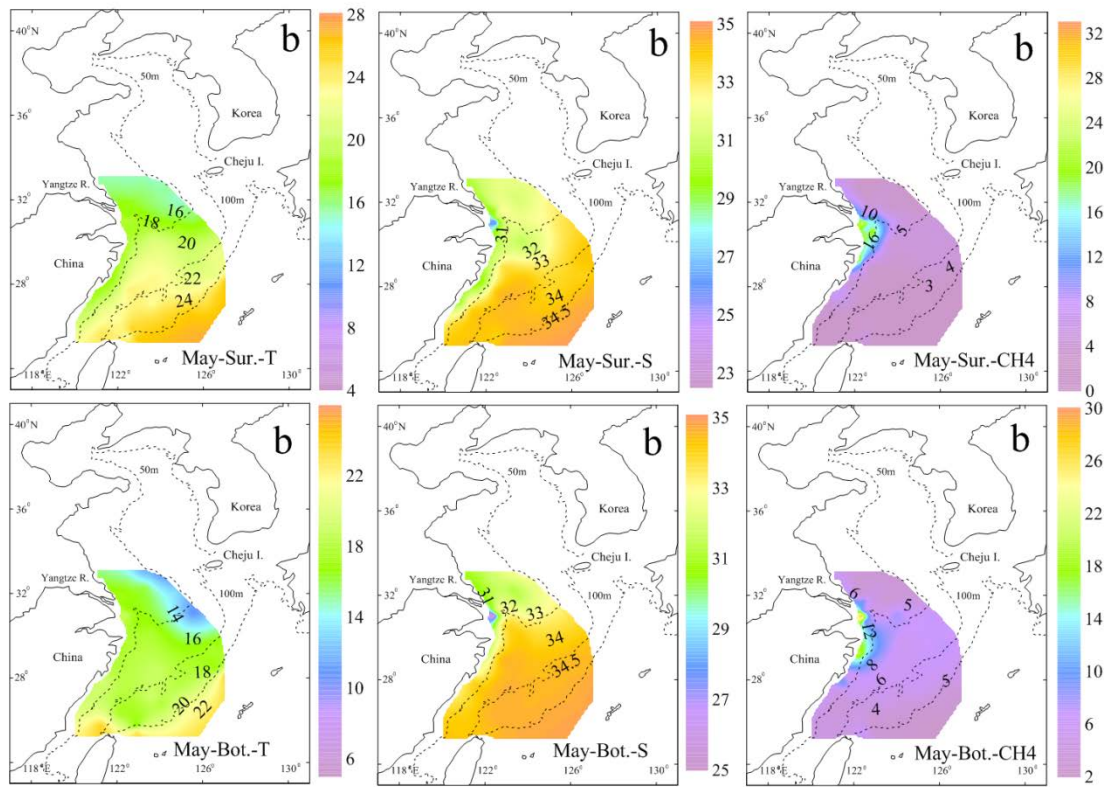
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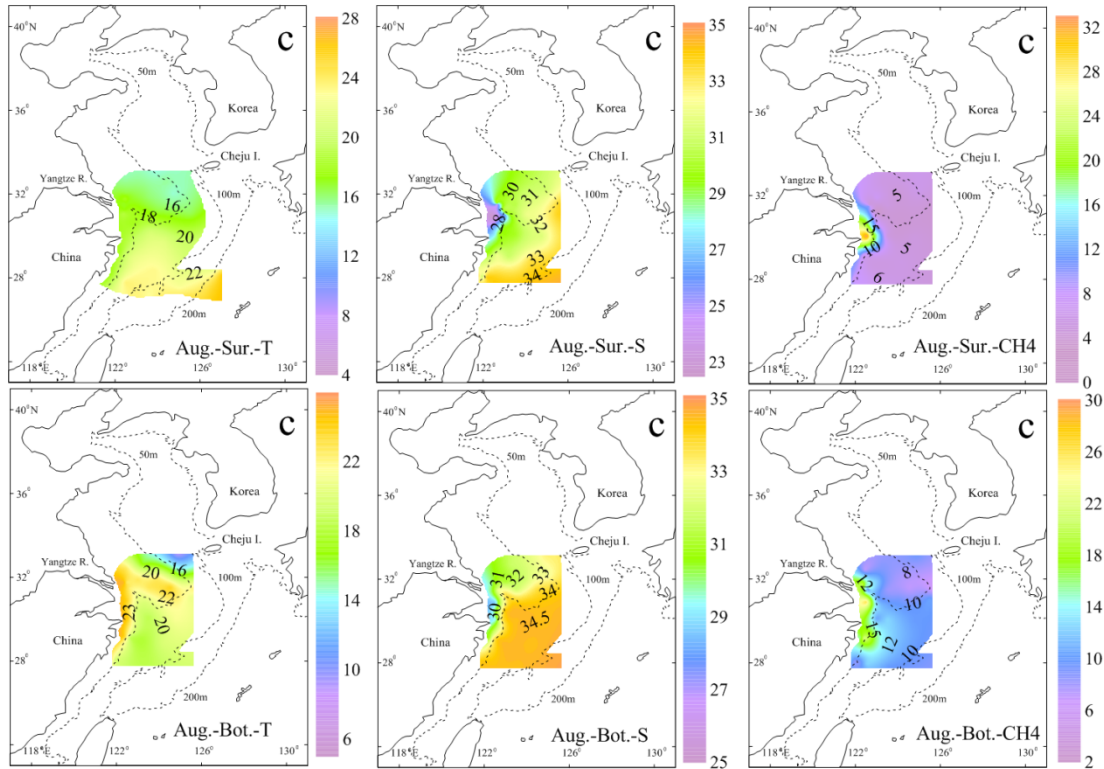
3 Fig. 2. Temperature-salinity diagrams and CH₄ concentrations in the East China Sea
4 during May, October, and December of 2011. The dominant water masses are
5 classified as previously described (Li and Su, 2000; Qi et al., 2014) and indicated by
6 rectangular outlines. CDW: Changjiang Diluted Water; SMW: Shelf Mixed Water;
7 KSW: Kuroshio Surface Water; KSSW: Kuroshio Subsurface Water; KIW: Kuroshio
8 Intermediate Water; KDW: Kuroshio Deep Water; TWCW: Taiwan Warm Current
9 Water. Black dots: non-CH₄ sampling point; Color dots: CH₄ sampling points, with
10 concentrations indicated by color scale.

1

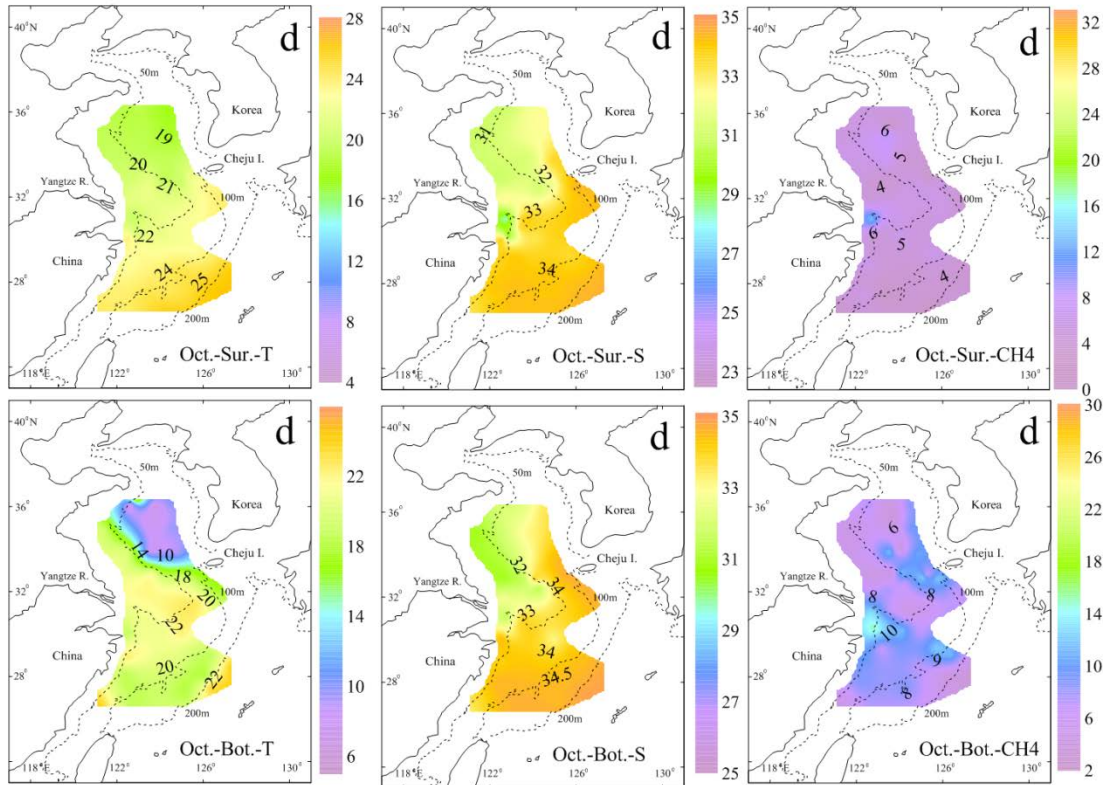


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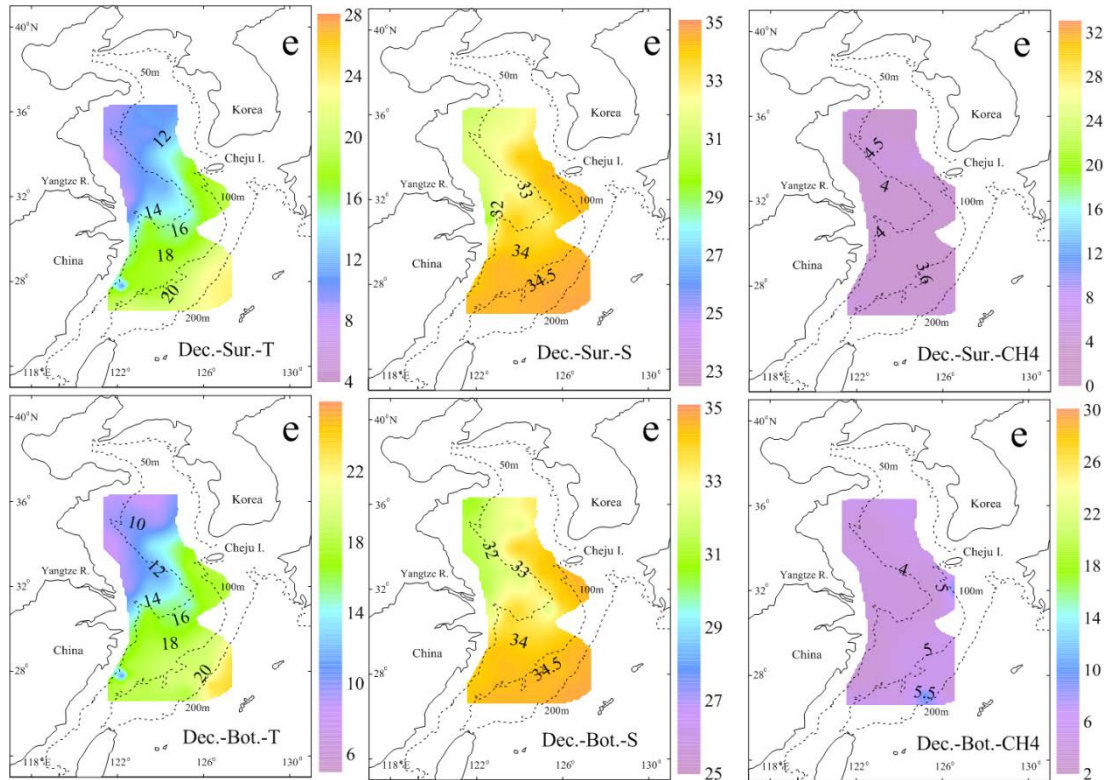




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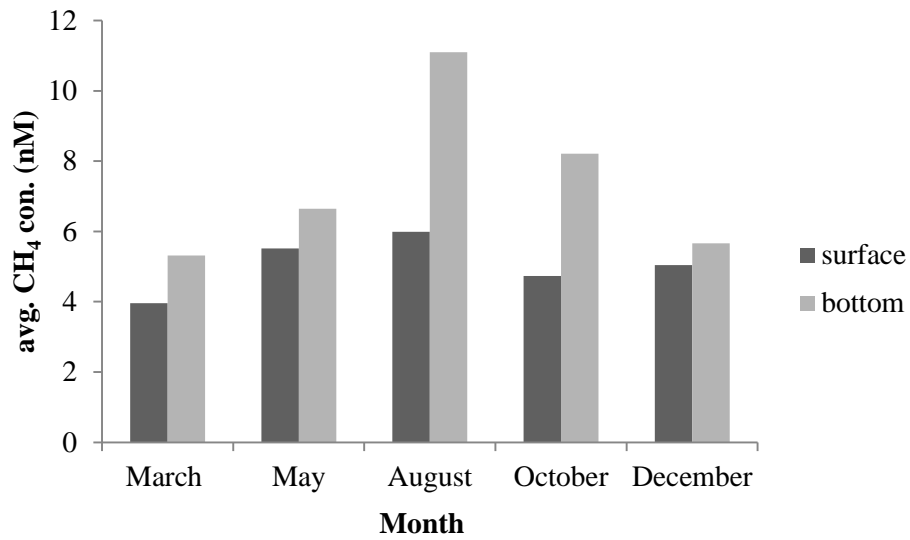


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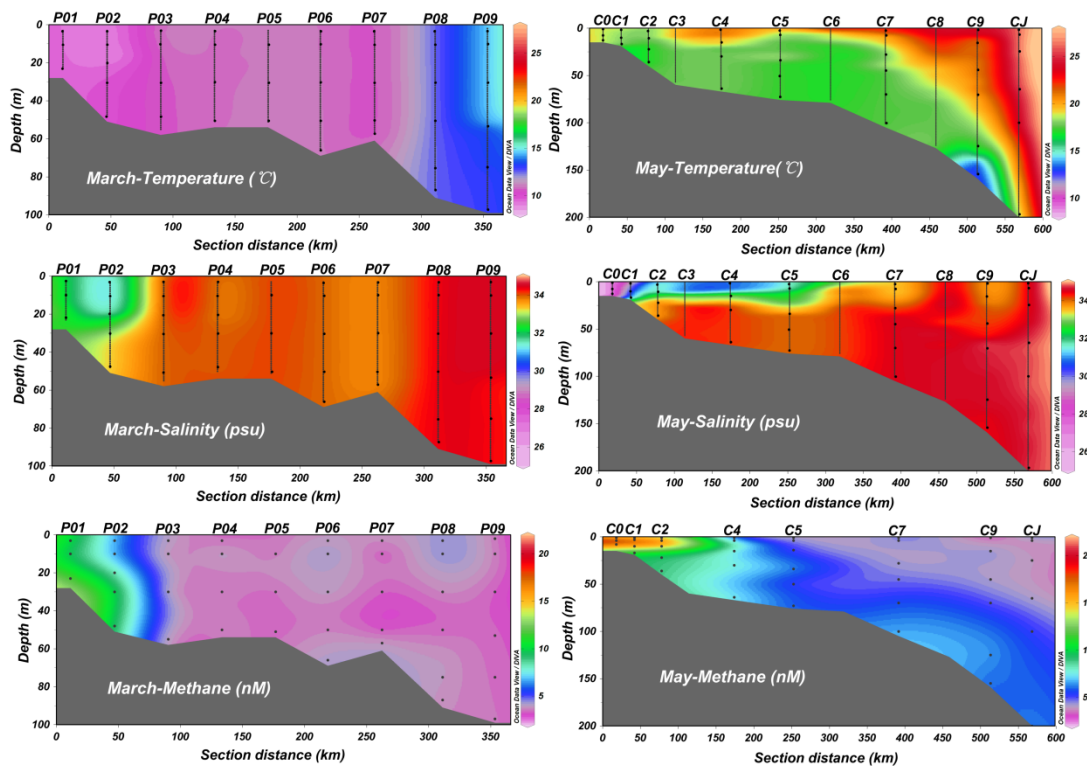
3 Fig. 3. Geographical distribution of temperature (°C), salinity (psu), and CH₄
 4 (nmol·L⁻¹) in surface and bottom waters of the study area during March (a), May (b),
 5 August (c), October (d), and December (e) of 2011.

6

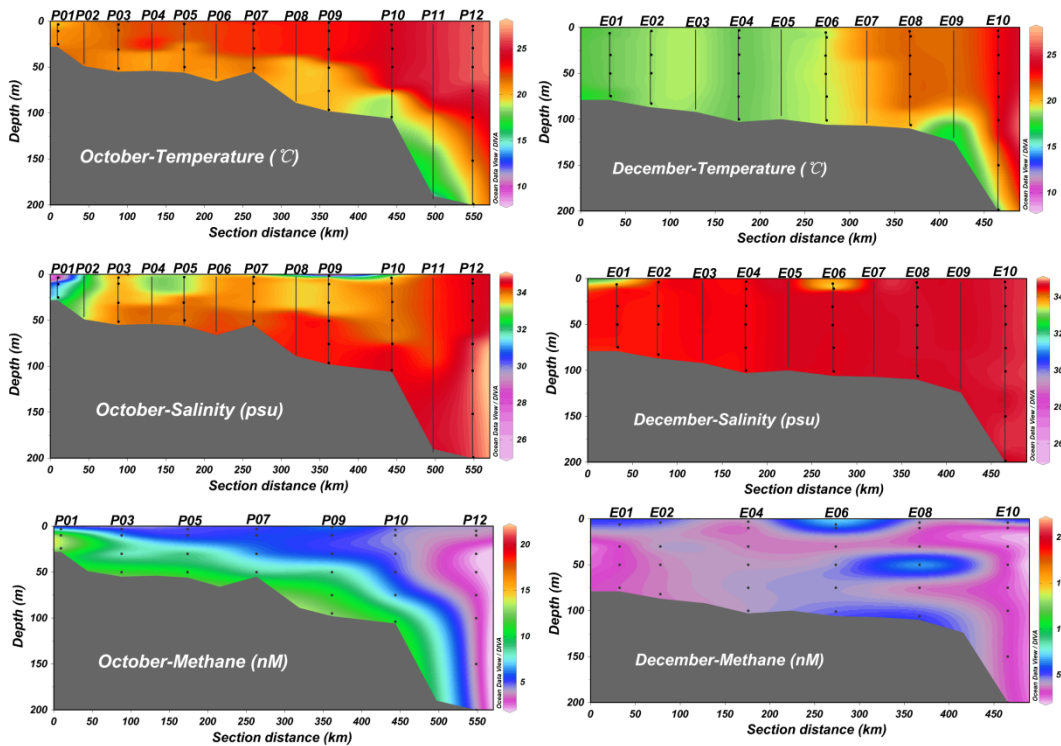


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2 Fig. 4 Comparison of the average surface and bottom CH₄ concentrations in the ECS
3 shelf during different seasons

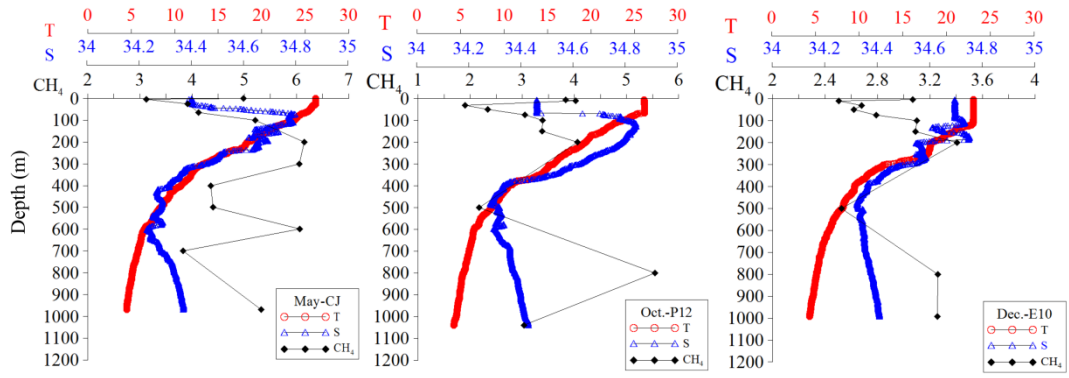


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2

3 Fig. 5. Depth distributions of salinity (psu), temperature (°C), and CH₄ (nmol·L⁻¹)
 4 along section PN during March (a), May (b) and October (c) and section E during
 5 December (d).

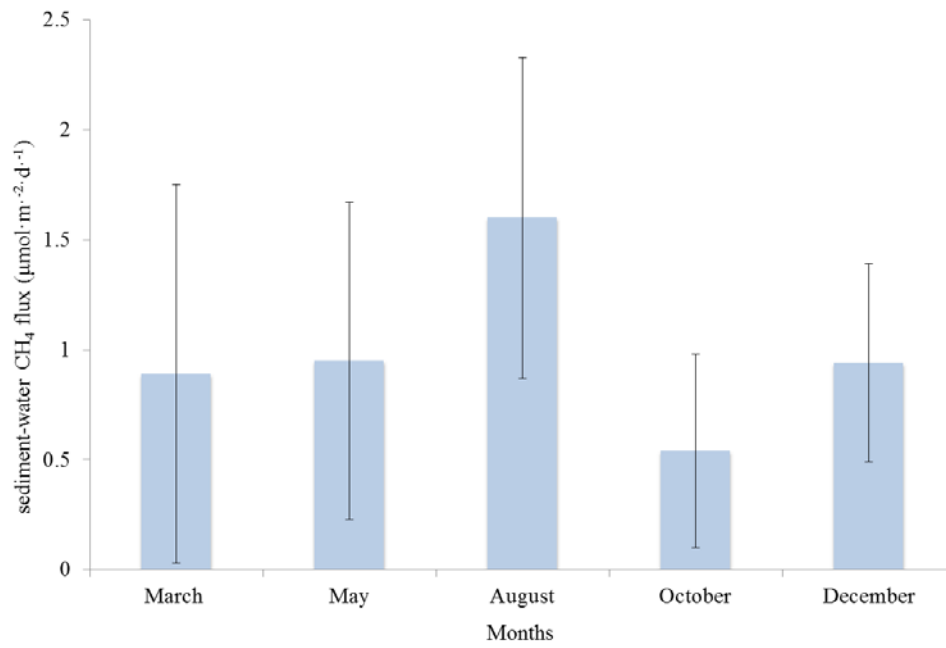


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2

3 Fig. 6. Depth profiles for CH₄ (nmol·L⁻¹), salinity (psu), and temperature (°C) at
 4 station CJ in May, station P12 in October, and station E10 in December.

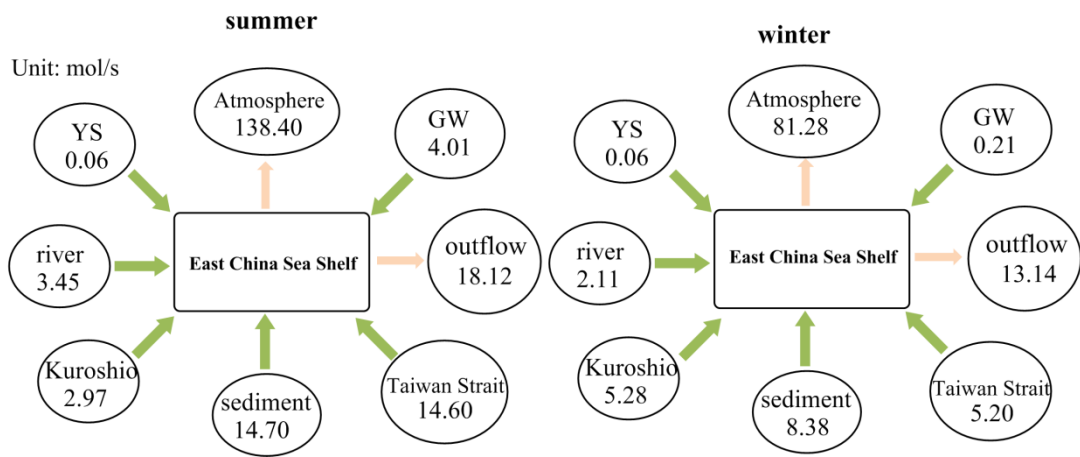
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2

3 Fig. 7. Seasonal variation of sediment-water CH₄ fluxes from the East China Sea and
4 Yellow Sea.

1



2

3

4

Fig. 8. CH₄ budget of the East China Sea during summer and winter.

5