1 Dear editors and reviewers:

2

Below is a detailed response to comments made by the referee's on our paper 'Spatial variability of diploptene  $\delta^{13}$ C values in thermokarst lakes: the potential to analyse the complexity of lacustrine methane cycling'.

6

7 We would like to thank the reviewers for their efforts and constructive comments and we 8 believe that they have improved the manuscript. Revisions to the manuscript are given in 9 detail below and can be seen in the marked up version of the manuscript included here. 10 Quotes from the manuscript incorporating the changes are in red.

11

12 Please note, we would like to alter an author affiliation. This is an extra alteration to the 13 manuscript.

- 14
- 15 Sincerely
- 16 Kim Davies on behalf of all the authors.
- 17
- 18 **Referee #1**
- 19 General comments (GC)
- 20

# **GC 1:** The title of the manuscript makes me think there is potential for using $\delta$ 13C on diploptene for reconstructing historical estimates of lake methane ebullition...but after

- 23 reading the manuscript, I am not so sure that is true
- 24

R1: We agree that the title could have been misleading. We have altered it to more accuratelydescribe the paper.

- 27
- 28 New title:

<sup>29</sup> 'Spatial variability of diploptene  $\delta^{13}$ C values in thermokarst lakes highlights the complexity 30 of tracing the lacustrine methane cycle'

31

32 GC2: The authors should be careful to not suggest that the proposed method appears 33 more valuable than it really is (at least at our current state of knowledge). Specifically, I 34 think the abstract is more positive about using  $\delta$ 13C-diploptene observations for 35 reconstructing past methane oxidation in lakes than the manuscript supports.

36

37 Response 2: We have changed the abstract to better reflect the uncertainty associated with the 38 results. Specifically, we have added a sentence about the need for further research before the 39  $\delta^{13}$ C values of diploptene could be considered for reconstructing past methane oxidation. The

- alterations make it clear to the reader that the results are complicated and they reveal the
   potential variability which can be found in contemporary sediments even in a small lake.
- 3

4 GC3: I think this manuscript can be published, as long as the authors are clear that this is a 5 very preliminary study, which basically tells us that with more research,  $\delta$ 13Cdiploptene 6 might become useful as a biomarker for informing methane oxidation history in lake 7 sediments. But for now, we don't know enough to reliably apply it.

8

9 Response 3: We very much agree with this and this is a fundamental point that the manuscript
10 is trying to make. We have included changes to the abstract and the conclusion which should
11 make these points much clearer.

- 12
- 13 Sentences altered/added:

14 The diploptene  $\delta^{13}$ C values highlight strong within-lake variability but presently, there is no 15 clear pattern in this variability that can be linked to thermokarst specific methane emissions.

16

17 This study, whilst preliminary, highlights the need for further research and implies that at this 18 stage, single-value, down-core records of hopanoid isotopic signatures are unlikely to be

19 secure indicators of changing methane flux at the whole-lake scale.

20

However the results were highly variable and suggest that, like methane production, MO is highly complex, both in terms of its spatial distribution and in relation to the type of substrate available. A single model for thermokarst lakes is unlikely to capture all patterns present at both the inter-lake and intra-lake level, and as the data stand, there is a large amount of variability which cannot be linked to specific types of methane production.

26

We conclude that given the current data, further research should be completed in order to
understand the variability in d13c diploptene values prior to utilisation of this method for the
reconstruction of methane cycling in lacustrine systems.

- 30
- 31 Specific comments (SC)
- 32

SC1: Diploptene is mispelled TWICE in the abstract. 'Diplotene' is something
 completely different (and is not a chemical).

- 35
- 36 R1: We have fixed these grammatical errors.

37

38 SC2: In the abstract: 'Using  $\delta$ 13C-diploptene as a proxy for methane oxidation activity,

39 we suggest the observed differences in methane oxidation levels among sites within the

40 two lakes could be linked to differences in source area of methane production (e.g. age

1 2	and type of organic carbon) and bathymetry as it relates to varying oxycline depths and changing pressure gradients.'
3 4	Okbut as noted in the manuscript, there was no radiocarbon dating in one of the lakes. So it seems that the suggestion of age differences is premature
5	
6	R2: We have removed the reference to chronology in the abstract and altered the wording.
7	
8	SC3: Section 5.3: "A crucial outcome of this study is the large variability seen in the
9 10 11	$\delta$ 13C values of diploptene across small spatial distances. This is an important finding, as often whole lakes can be represented by a single sampling site in palaeoenvironmental studies."
12	
13 14 15 16 17	I agree with this! But it undermines some of the conclusions of the manuscript, especially the last statement of the conclusions: "We conclude that diploptene biomarkers have considerable potential to help reconstruct patterns of methane cycling in lakes and, with certain caveats, particularly attention to context, past methane dynamics."
18	
19 20 21	Isn't it more true that this study raises MANY cautions that must be resolved before $\delta$ 13C-diploptene values can be used to 'reconstruct patterns of methane cycling and past methane dynamics'?
22	
23 24 25	I don't see how the results in this study do much more than show that sometimes the $\delta$ 13C-diploptene values make sense with current observations of methane ebullition and methane oxidizing bacteria biomass, and sometimes they don't (e.g. Figure 4)
26	
27 28 29	R3: We agree that this final sentence in the conclusions was out of place and did not fit with what the data are showing. We have replaced this sentence with one that highlights that more work should be done.
30	
31	New sentence:
32 33 34	We conclude that given the current data, further research should be completed in order to understand the variability in d13c diploptene values prior to utilisation of this method for the reconstruction of methane cycling in lacustrine systems.
35	
36	Minor Comments (MC)
37	
38 39	MC1: 'The connections between methane production" This sentence should be split into two sentences, probably after 'not well understood'
40	

1	R1: This sentence has been amended suggested
2	
3 4 5	MC2: page 12164 "in the food web to the incorporation of carbon from of methane" extra word: remove "of"
6	R2: 'of' has been removed
7	
8	MC3: page 12166: odd to cite personal observation from the first author
9	
10	R3: the reference to personal observation has been removed
11	
12 13	MC4: "actively thermokarsting" Don't verb nouns unless absolutely necessary.
14	R4: The sentence has been changed to: 'actively thawing and eroding'
15	
16 17	MC5: page 12171: It's too bad that there is no radiocarbon data for Ace Lake.
18 19	R5: We would have liked to provide a high resolution chronology for both lakes, however, due to way in which the project was funded, there was no money available.
20	
21 22	MC6: page 12176: "palaeoenvironmental investigations take into lake type" should be "palaeoenvironmental investigations take into account lake type"
23	
24	R6: The word 'account' has been added.
25	
26 27	Figure comments (FC)
27 28 29	FC1: Figure 2: unit "Mg" should be "mg"
2) 30	R1: Mg has been changed to 'mg'
31	
32 33 34	FC2: Figure 3: What is the basis of the bin sizes for the bubble counts on this figure? The first interval has size 3, then a bin size of 6, then a bin size of 5, then a bin size of 2, then a bin size of 93! Why?
35	
36	R2: The bin sizes were designed to show the variability in the data. We agree however, that

37 they didn't make sense. The bin sizes have been amended and now follow a logical pattern.

- During the editing process, it was noticed that one of the data points was missing so it has
   been added to the diagram.
- 3

FC3: Figures 3 and 4: The compass rose arrow is WRONG on one of these figures. I
don't know which is wrong, but they both cannot be true.

6 Also, the shoreline of Ace Lake looks rather different in Figure 3 and Figure 4. Why?

7

8 R3: The lake outline in figure 3 had a slightly different orientation to the lake outline in the
9 other diagrams. For ease, figure 3 has been altered so that it fits with the style of the rest of
10 the diagrams. This means the north arrow is also fixed.

- 11
- 12 **Referee #2**

13

14 Davies and co-authors present a study on spatial variability of diplopteneo13C 15 measurements of the bacterial biomarker diploptene in thermokarst lakes together with methane monitoring, and that has the potential to assess past carbon cycling in northern 16 17 high latitudes. However, I think the paper still needs some work before publication. The 18 critical issue is that the authors showed totally 14 diploptene  $\delta$ 13C values within the 19 same zone (TK zone or Centre section) from two lakes, but with big variations. Although 20 the authors proposed several reasons for the large variability in the  $\delta 13C$  values of 21 diploptene across small spatial distances, maybe more data would be useful.

22

23 Response: We thank the reviewer for their comments. Whilst we agree that more data would 24 be very helpful there is no option for processing anymore samples. The samples were run as a 25 preliminary dataset utilising a very limited PhD budget which has been exhausted and 26 therefore no more funds exist. In order to address any issues with the size of the dataset, we 27 have highlighted the preliminary nature of the data in the manuscript throughout the abstract, 28 discussion and conclusion. However, it should be pointed out that the study was set up so as 29 to achieve repeat samples in each zone and that this is a relatively high sample resolution 30 when considering the field in which we are interested in trying to apply the technique (palaeo 31 reconstructions).

32

# 33 SC1: In the abstract, line 13: diplotene should be diploptene. The same as line 18.

- 34
- 35 R1: We have fixed these grammatical errors.
- 36

SC2: In the abstract, line 21: (e.g., age and type of organic carbon) and. . ... I think the
 authors just showed one age and it seems they didn't discuss anything about type of
 organic carbon in the paper. The authors should clarify this.

40

41 R2: We have removed the reference to chronology in the abstract and altered the wording.

SC3: 3.2 Methane monitoring: the authors mentioned methane δ13C and δD, but didn't
 show/discuss them in the paper.

4

1

5 R3: The mention of  $\delta D$  has been removed from the manuscript. The  $\delta 13C$  we refer to can be 6 found in table 1 and is used as part of the mixing model.

7

8 SC4: 3.5 Mass balance equation: ". . .δ13C hetero\_hopane is the δ13C value of the
9 hopanoids derived from heterotrophic bacteria. ...", So please specify which hopanoids
10 in the paper because a lot of hopanoids are derived from bacteria

11

R4: From this comment, we can see that the reference to other types of hopanoids is confusing. We have altered the wording of the mixing model to better reflect what we were trying to show. We have added a sentence in the introduction to help convey the theory behind the mixing model.

16

The mixing model is trying to understand the contribution of MOB to the diploptene signal,
where diploptene is derived from both MOB and heterotrophic bacteria. Here, shifts towards
more negative d13c values would suggest a greater contribution of MOB to the diploptene

- 20 signal.
- 21
- 22 Added sentences:

In particular, the compound diploptene (17  $\beta$ (H), 21  $\beta$ (H)-hop-22 (29)-ene), is a hopanoid hydrocarbon derived from a range of bacterial sources. however due to the utilisation of methane as a carbon source, the d13c values of diploptene derived from MOB will be more negative than if it were derived from other heterotrophic bacteria which utilise organic carbon from vegetation.

28

29 The  $\delta^{13}$ C values of diploptene derived from heterotrophic bacteria will primarily reflect the 30 substrate carbon which in this instance will be organic material and not methane. These 31 values are therefore unlikely to vary; however a ~2 to 4‰ shift can occur during lipid 32 biosynthesis (Pancost and Sininghe Damsté 2003, and references therein).

33

34 SC5: Results section of Line 24-25(P12171): '. . .. . .Diploptene δ13C values in the
35 thermokarst zone of Ace L. are similar to those of the lake centre at Smith,...', I
36 couldn't see they are similar.

37

38 R5: we have removed this sentence

- 39
- 40 SC6: Line 14 in the 5.2 section, it is fig3 or fig.4?

1	
2	R6: This should be figure 4, we have amended this.
3	
4 5	SC7: 5.2 section and Table 3: It is also not very clear that MOB biomass has large variations across the small distances. For example, at the TK zone of Ace Lake, sample at and at are clean but the difference of MOB biomass is around 200/. If it is because of
0 7	microbial community, so give more evidence.
8	
9	
10	Referee #3
11	
12	SC1: In the abstract diploptene is misspelled twice.
13	
14	R1: We have fixed these grammatical errors.
15	
16	SC2: Page 12163 Line 4, What is a "bight"
17 18	$R_2$ : We have included a brief definition of a bight in the text Briefly, a bight is a curve in the
19	coastline or a bay formed from such a curve.
20 21	SC3: Page 12164 Line 2-4, "potential confounding factor" this seems potential pretty important, what impact could this have on your results.
22	
23 24	R3: This factor is discussed in section 5.2, however we have added more detail in the introduction.
25	
26	Sentence added:
27	Methane production and oxidation that occurs in the near-surface sediments will represent a
28	background level which is likely to be found in many contemporary lake settings and the
29	amount should be lower than that derived from thermokarst specific sources. We might expect
30	some level of depletion in $\delta^{13}$ C values due to near-surface production but crucially, if $\delta^{13}$ C
31	values of diploptene are to be used as a proxy for past methane production, we would expect
32	thermokarst specific methane production that is being oxidised would have much lower $\delta^{13}C$
33	values than background methane oxidation.
34	SC4: Introduction, it is not clear exactly what patterns you would you expect to see in

35 diploptene  $\delta$ 13C under the scenarios discussed.

1 R4: We have added a number of sentences which make it clearer what patterns we would

- 2 expect to see.
- 3 Sentences added:

In particular, the compound diploptene (17  $\beta$ (H), 21  $\beta$ (H)-hop-22 (29)-ene), is a hopanoid hydrocarbon derived from a range of bacterial sources. however due to the utilisation of methane as a carbon source, the d13c values of diploptene derived from MOB will be more negative than if it were derived from other heterotrophic bacteria which utilise organic carbon from vegetation.

9

10 Therefore if MOB are present in the sediments of thermokarst lakes, we would expect to see 11 depleted  $\delta^{13}$ C values of diploptene.

- 12 We might expect some level of depletion in  $\delta^{13}$ C values due to near-surface production but
- 13 crucially, if  $\delta^{13}$ C values of diploptene are to be used as a proxy for past methane production,
- 14 we would expect thermokarst specific methane production that is being oxidised would have
- 15 much lower  $\delta^{13}$ C values than background methane oxidation.

# 16 SC5: Page 12168, Line 5, Any particular reason for using the 1-2cm sediment slice?

- 17 R5: The 0-1cm sediment slice was more variable is sample size due to the sediment-water
- 18 interface, therefore the 1-2cm slice is still well oxygenated and was more likely to represent
- 19 the same level across all samples.

# 20 SC6: Page 12167, Line 25, Don't include the δD analytical error if you don't include any

- **δD data.**
- 22 R6: The mention of  $\delta D$  has been removed from the manuscript.

# 23 SC7: Page 12170 Line 11, You give a potential range of 0-30‰ what value did you use, is

- 24 this the 10‰ you discuss earlier, please clarify.
- R7: We have added sentences to clarify what we did. The results are presented as a rangewhich incorporate both the minimum and maximum possible fractionation factor, therefore
- 27 d13c<sub>MOB\_dip\_min</sub> will represent the lowest possible value given maximum fractionation (30%).
- 28 Sentences added:
- 29 In order to incorporate this large range, we used both the minimum and maximum value of
- 30 fractionation (0 and 30‰) to show different scenarios rather than assuming a single value.
- 31 This should also cover any potential variation due to differing  $\delta^{13}C_{methane}$ . Therefore the

1 equation was calculated twice, once using  $\delta^{13}C_{mob\_dip\_min}$  and once using  $\delta^{13}C_{mob\_dip\_max}$ .

3 SC8: Overall the calculation of diploptene  $\delta$ 13C seems pretty vague with a lot of 4 estimates, this is ok, isotopes can be messy, but the discussion of these choices and the 5 variation/uncertainty they introduce could be more clearly discussed, especially give the 6 high variability and inconsistency of your results and the claims that this method could 7 be used to do historical reconstructions.

8

9 R8: We are unsure if the reviewer is referring to the d13c values we have of the diploptene 10 from the sediments or the d13c values inferred to calculate MOB concentrations. If referring 11 to the latter, then we agree that they are messy and deliberately vague as we did not want the 12 estimates to seem more robust than we can actually calculate. We have included further 13 sentences to clarify the purpose of the mixing model.

14

We have also pointed out more clearly that this area of research needs much more
development before it could be used for reconstructions.

18 Sentences added:

19 By developing this mixing model and considering, in more detail, the potential end member 20 values for the  $\delta^{13}$ C values of diploptene derived from different sources (MOB and other 21 heterotrophic bacteria) we can get a semi-quantitative idea of the distribution patterns of 22 MOB across the samples.

These estimates have a large degree of uncertainty associated with them and we note thatthere are some important caveats to using this mixing model.

25

SC9: Line 12171 Line 6, How many bubbles were sampled for  $\delta$ 13C, there are no error values listed, which seems to suggest only a single sample was analyzed at each site. If that is the case, there is not much you can infer from this one number; especially considering how your diploptene  $\delta$ 13C data shows just how spatially variable  $\delta$ 13C is in this system. 1

R9: The value from Ace Lake represents an average from across 5 seep locations in the thermokarst zone whilst the value from Smith Lake is taken from a single seep. We agree that the number of samples should be increased but in this instance, no more samples can be taken. These values have been used for the mixing model and we agree that it is likely that these values will be variable, however, we hoped that by using a minimum and maximum value for fractionation we would have incorporated a large amount of variation. We have included a sentence in the mixing model section which discusses this further.

9

#### 10 Added sentences:

The  $\delta^{13}C_{\text{methane}}$  is the measured value of methane captured at seep locations in the thermokarst 11 12 zones at each lake. As the value is based on a limited number of data (n1 and n5 for Smith L. 13 and Ace L. respectively), it is likely there will be more variation than is seen in the model. 14 Furthermore MOB can be significantly depleted in comparison to the source carbon they utilise (Whiticar 1999); isotopic differences can be as large 30‰ (Jahnke et al., 1999). In 15 16 order to incorporate this large range, we used both the minimum and maximum value of fractionation (0 and 30‰) to show different scenarios rather than assuming a single value. 17 This should also cover any potential variation due to differing  $\delta^{13}C_{methane}$ . Therefore the 18 equation was calculated twice, once using  $\delta^{13}C_{mob\_dip\_min}$  and once using  $\delta^{13}C_{mob\_dip\_max.}$ 19 20

SC10: Methods: Sample size, replication, sampling location information needs to be clearly covered in the methods section. This information needs to be included for all analyses, not just diploptene  $\delta$ 13C, although I couldn't even find sample size information for diploptene  $\delta$ 13C in the methods section (it is mentioned later in te manuscript).

R10: we have included a table which shows the sample weights and made reference to figure2 which shows the sampling locations for the sediment cores.

28 Table included:

SampleSamplesizeNo(dry g)

Smith		
	1	0.2596
	2	0.2206
	3	0.3584
	4	0.1486
	5	0.1942
	6	0.5654
	7	0.3841
	8	0.2024
	9	0.3386
	10	0.2185
Ace		
	a1	1.3427
	a2	0.6812
	a3	0.5758
	a4	0.808

1

SC11: It looks like Ace lake was only sampled in the TK zone whereas Smith Lake was also sampled away from the TK zone. This is unfortunate, since it really limits the ability to distinguish potential impacts of thermokarst activity from other spatial differences within/between lakes.

6 R11: We agree that this is a shame. The centre of Ace L. is much deeper than at Smith L. 7 (~9m) and we wanted to try to reduce the number of other potential factors which could 8 influence the d13c values. Furthermore, due to the limited number of samples which could be 9 run, we felt it would be more beneficial to get repeat samples from within zones. We have 10 included a sentence in the methods to point this out.

11

12 Sentence added:

13 In order to remove water depth as a confounding variable and to increase the number of

14 replications in each zone, Ace L. was not sampled as it was much deeper than Smith L. centre

15 (~9m).

SC12: The Figures & Results sections make it difficult to fully assess the variability of the diploptene δ13C data, in the text only the min/max values for each site is listed (no average +/- std dev so you can't tell if there is juts a few outliers or the data is evenly spread out) and then the figures just show 10‰ increments.

6

1

R12: We have included further sentences in the results section to highlight the standard
deviation and the data more clearly. We have also added more standard deviation values to
table 3 to make the variation clearer.

- 10
- 11 Table 3  $\delta^{13}$ C values of diploptene at the study sites. The values are an average of three
- 12 replicates. The standard deviation of these replicates and of each zone and across all samples
- 13 is also given.

		Sample	$\delta^{13}C_{dip}$	Sample		Standard
		Number	(‰)	replicate		Deviation
				standard		(SD)
				Deviation		
				(SD)		
Ace						
	TK	a1	-50.1	1.5		
	zone					
		a2	-58.5	2.0		
		a3	-53.1	0.4		
		a4	-68.2	0.1	TK zone	8.0
Smith						
	Centre	1	-51.4	2.7		
		2	-48.3	0.0		
		3	-56.8	N/A		

		4	-49.2	1.0		
		5	-46.9	1.8		
		6	-48.0	0.1	Centre	3.6
	ТК	7	-38.8	0.3		
	zone					
		8	-40.9	0.2		
		9	-42.9	0.1		
		10	N/A	N/A	TK zone	2.0
					Total	7.8
1						
2						
3						
4						
5						
6						
7						
8						
9						
10						
11						
12						
13						
14						

1	Spatial variability of diploptene $\delta^{13}$ C values in thermokarst
2	lakes: the potential to analyse the complexity of lacustrine
3	methane cycling highlights the complexity of tracing the
4	lacustrine methane cycle
5	
6 7	K. L. Davies <sup>1</sup> *, R. D. Pancost <sup>2,3</sup> , M. E. Edwards <sup>1</sup> K. M. Walter Anthony <sup>4</sup> , P. G. Langdon <sup>1</sup> and L. Chaves Torres <sup>52</sup>
8	[1] Geography & Environment, University of Southampton, SO17 1BJ, United Kingdom
9 10	[2] Organic Geochemistry Unit, School of Chemistry, University of Bristol, BS8 1TS, United Kingdom
11	[3] Cabot Institute, University of Bristol, BS8 1 UJ, United Kingdom
12 13	[4] Water & Environment Research Centre, University of Alaska Fairbanks, Fairbanks, Alaska, 99775, USA
14	[5] Biogeosciences Research Group, Institute for the Sciences of Nature, Territory and
15	Renewable Energies (INTE-PUCP), Pontifical Catholic University of Peru, Av. Universitaria
16	1801, San Miguel, Lima 32, Peru
17	*Correspondence to: Kimberley Davies, kld206@soton.ac.uk
18	
19	
20	
21	
22	
23	
24	
25	
26	

1

#### 2 Abstract

3 Cryospheric changes in northern high latitudes are linked to significant greenhouse gas flux to 4 the atmosphere, including methane release that originates from organic matter decomposition 5 in thermokarst lakes. The connections between methane production in sediments, transport 6 pathways and oxidation are not well understood. and Tthis has implications for any attempts 7 to reconstruct methane production from sedimentary archives. We assessed m Methane oxidising bacteria were used to represent methane oxidation as represented by methane 8 9 oxidising bacteria across the surface sediments of two interior Alaska thermokarst lakes in relation to methane emissions via ebullition (bubbling). The bacterial biomarker diploptene 10 was present and had low  $\delta^{13}$ C values (lower than -38‰) in all sediments analysed, suggesting 11 methane oxidation was widespread. The most <sup>13</sup>C-depleted diploptene was found in the area 12 of highest methane ebullition emissions in Ace Lake ( $\delta^{13}$ C diploptene values between -68.2 13 14 and -50.1%), suggesting a positive-potential link between methane production, oxidation, and emission in this area. In contrast, significantly less depleted diploptene  $\delta^{13}$ C values (between -15 42.9 and -38.8‰) were found in the area of highest methane ebullition emissions in Smith 16 Lake. Lower  $\delta^{13}$ C values of diploptene were found in the central area of Smith Lake (between 17 -56.8 and -46.9‰), where methane ebullition rates are low but methane diffusion appears 18 high. Using  $\delta^{13}$ C-diploptene as a proxy for methane oxidation activity, we suggest the 19 20 observed differences in methane oxidation levels among sites within the two lakes could be 21 linked to differences in the level of methane diffusing from the sediments, the source area of methane production (e.g. surface versus deep sedimentsage and type of organic carbon) and 22 23 bathymetry as it relates to varying oxycline depths and changing pressure gradients, although these theories need to be tested. As a result, methane oxidation is highly lake dependent. The 24 25 diploptene  $\delta^{13}$ C values also highlight strong within-lake variability but presently, there is no 26 clear pattern in this variability that can be linked to thermokarst specific methane emissions, This study, whilst preliminary, highlights the need for further research and implies<del>ying</del> that at 27 this stage, single-value, down-core records of hopanoid isotopic signatures are not-unlikely to 28 29 be secure indicators of changing methane flux at the whole-lake scale.

30

#### 1 1 Introduction

2 Thermokarst and thermokarst-affected lakes (those formed and/or influenced by thaw and 3 collapse of ice-rich ground) are now recognized as important past and present sources of 4 methane flux to the atmosphere (Shirokova et al., 2012; Walter et al., 2006, 2008; Wik et al., 5 2013). Under current scenarios of projected future climate warming in regions sensitive to 6 thaw (Colins et al., 2013), these lakes are expected to remain a source of methane emissions 7 to the atmosphere (Vincent et al., 2013). Predictions of the future contribution they will make 8 to the dynamic global carbon cycle and any estimations of past emission rates are largely 9 based on measurements recorded over the last 15 years (e.g. Brosius et al., 2012; Walter 10 Anthony et al., 2014). Long-term (i.e. Holocene) variations in lake-derived methane flux to 11 the atmosphere and changes in emissions during discrete climatic events in the past are less 12 well understood (but see Walter Anthony et al., 2014; Walter et al., 2007b). A proxy for past 13 gas flux from lakes would be an important development in better understanding long term 14 carbon cycling, but we are far from understanding within-lake methane dynamics well enough for such a proxy to yet be reliable. 15

The broad term 'thermokarst lakes' encompasses a complex range of lakes types associated with different geographical and geomorphological settings in permafrost regions. Methane production within these lakes and fluxes to the atmosphere vary with lake type. Walter et al. (2008) and Brosius et al. (2012) divide thermokarst lakes into two main categories: yedoma lakes and non-yedoma lakes, where yedoma refers to late Pleistocene deposits of organic- and ice-rich silt, typically several or more metres deep (Zimov et al., 2006; Schirrmeister et al., 2013).

23 Methane production in thermokarst lakes can be classified by production type: production that 24 occurs in anoxic surface sediments, as is common in most freshwater lakes and reservoirs, and production that occurs in deeper sediments, especially along the boundary of the "thaw bulb", 25 which is specific to thermokarst lakes (Figure 1). Anoxia is caused by oxygen depletion 26 27 associated with microbial decomposition of organic matter. Anoxic conditions are enhanced 28 by thermal stratification in the water column and/or by rapid sedimentation that buries labile 29 organic material before it can be processed at the sediment surface. A common trait of 30 thermokarst lakes is methane production via mineralisation of organic carbon from sources 31 not found in other lakes. For example, methane emissions can occur where thermokarst-32 induced erosion leads to large-scale slumping of banks into the littoral zone; material is

typically of Holocene age, but may be older (Figure 1). As well as the production from 1

2 slumped material, vedoma lakes may feature high methane emissions related to the microbial

processing of older, labile carbon in the deep thaw bulb (talik, i.e., an area of thawed 3

4 permafrost sediment underneath the lake). Walter Anthony and Anthony (2013) suggest that

5 yedoma thermokarst lakes typically produce more methane than non-yedoma thermokarst

lakes owing to a higher availability of labile carbon in thick, thawed yedoma sequences. 6

7 Once produced, methane can be transported to the atmosphere through a number of pathways:

8 ebullition (bubbling), turbulent diffusion and plant mediated transport (Bastviken, 2004).

9 Several studies have focused on these emission pathways, assessing methane production and

10 emission levels in freshwater environments (e.g. Bastviken, 2004; Bastviken et al., 2011;

11 Delsontro et al., 2011; Joyce and Jewell, 2003).

12 Thermokarst-specific methane ebullition seeps have been observed and measured using GPS mapping and submerged bubble traps and described as persistent, spatially explicit fluxes at 13 14 the water-air interface (Sepulveda-Jauregui et al., 2014; Walter et al. 2006, 2008; Walter Anthony and Anthony 2013). Ebullition seeps are thought to be fairly stable due to the 15 16 development of conduits or 'bubble tubes' (Greinert et al., 2010; Scandella et al., 2011), 17 which are point sources from which methane is emitted to the atmosphere repeatedly at the 18 sediment-water interface. Nearly always, such seeps are densest near to actively eroding lake 19 margins, which we call the "thermokarst zone". Here, methanogenesis is high due to 20 thermokarst-specific sources of methane production: thawing of fresh talik and bank collapse (Figure 1; Kessler et al., 2012). Walter Anthony et al. (2010) postulate that most methane 21 22 production that is specific to thermokarst lakes is transported to the atmosphere via seep 23 ebullition (due to high rates of methane production in dense, thick talik sediments), although 24 the diffusive flux component can be relatively high in older, more stable thermokarst lakes 25 that have accumulated Holocene-aged organic carbon in near-surface sediments. 26 Less work has focused on methane production in surficial sediments of thermokarst lakes,

27 dissolution and diffusion of methane from the sediments to the water column, and resultant

28 diffusive emission, particularly in thermokarst zones. This paper reports an analysis of carbon

29 isotopes in sedimentary bacterial biomarkers in relation to different forms of atmospheric

30 methane flux from two lakes near Fairbanks, Alaska, with the aim of improving our

understanding of methane cycling in thermokarst lake systems and assessing the effectiveness 31

32 of biomarkers as a proxy for methane cycling in lakes.

# 1.1 The link between methane ebullition and methane diffusion from sediments

3 A significant fraction of methane produced in lake sediments can be oxidized and recycled 4 within the lake, processes that offset methane emissions. Methane that has diffused from the 5 sediments is subject to aerobic microbial oxidation by bacteria (Bastviken et al., 2002; 6 Liebner and Wagner, 2007; Trotsenko and Khmelenina, 2005). Aerobic methane oxidation (MO) is thought to considerably reduce methane emissions from water bodies (Reeburgh. 7 8 2007). Methane Ooxidation studies in lakes have mostly been carried out under stratified 9 water column conditions (Bastviken et al., 2002; Kankaala et al., 2006). As with diffusive 10 methane flux (Sepulveda-Jauregui et al., 2015), little work has focused on aerobic MO in 11 thermokarst lakes (Martinez-Cruz et al., 2015). Understanding the link between MO and 12 observed fluxes is crucial for developing a proxy for past methane production in thermokarst 13 lakes.

14 In studies based on deep marine environments there is a correlation between widespread 15 methane, released via cold seeps through sediments, and MO, as indicated by the presence and  $\delta^{13}$ C values of specific bacteria and compounds (Elvert et al., 2001a; Pancost et al., 2001, 16 2000b). In these environments both anaerobic (Alperin and Hoehler, 2010; Briggs et al., 17 18 2011) and aerobic (Birgel and Peckmann, 2008; Elvert and Niemann, 2008) methane 19 oxidation processes have been identified and are important for mediating methane flux to the 20 atmosphere. As well as a link between methane ebullition seeps and methane diffusion in 21 deep marine settings, a study carried out in a shallow (9m) near-shore bight (a curved bay) 22 linked the formation of bubble tubes with increased methane diffusing from the sediments (Martens and Klump, 1980), the argument being that bubble tubes create an increased surface 23 24 area that enhances methane diffusion, even though the methane transported via ebullition is 25 taken directly to the atmosphere and is not subject to oxidation. While derived from different 26 environments than thermokarst lakes, the deep and shallow marine results suggest a positive 27 relationship between transport via ebullition and methane diffusion from sediments, which may also occur in thermokarst lakes. 28

He et al. (2012) provide evidence that suggests a possible correlation between a coal-bed

30 sourced methane ebullition seep and MO in the non-yedoma thermokarst lake, L. Qalluuraq,

31 Alaska. The highest MO potentials occurred near the coal-bed sourced ebullition seep and

32 were associated with the presence of type I MOB in the sediments at the seep location. He et

1 al. (2012) also observed high spatial variability of MO potentials and methanotroph 2 communities and highlighted the need for further investigation of MO in thermokarst lakes. In contrast, based on  $\delta^{13}$ C and  $\delta$ D stable isotope values and radiocarbon ages of methane in 3 4 bubbles, Walter et al. (2008) and Walter Anthony et al. (2014) suggest that methane emitted 5 by ebullition originating in deep thaw-bulb sediments by-passes aerobic MO and that the 6 majority of deep-sourced methane is transported through ebullition seeps as opposed to 7 escaping sediments via diffusion. We therefore have two contrasting conceptual models 8 (hypotheses): an enhancement model and a by-pass model. In the enhancement model, the 9 thermokarst zone of a lake, where ebullition seeps are most abundant, would have higher 10 levels of deep-sourced methane diffusion from sediments when compared with "quiescent" areas that are virtually ebullition seep free. In the by-pass model, where diffusion of deep-11 12 sourced methane out of sediments is thought to be minimal, we expect no difference between thermokarst-zone and lake-centrediffusion of deep sourced methane from sediments, or, 13 14 conceivably, less diffusion in the seep rich area. A potential confounding factor is diffusion of methane that is formed in near-surface sediments, which can have variable and contrasting 15 16 patterns across lakes, independent of spatial patterns of ebullition seeps. Methane production 17 and oxidation that occurs in the near-surface sediments will represent a background level 18 which is found in many contemporary lake settings however the amount should be lower than 19 that derived from thermokarst specific sources. We might expect some level of depletion in  $\delta^{13}$ C values due to near-surface production but crucially, if  $\delta^{13}$ C values of diploptene are to be 20 used as a proxy for past methane production, we would expect that if thermokarst specific 21 methane production is being oxidised, this would have much lower  $\delta^{13}$ C values than even 22 background MO. 23

24 Past methane emissions may be addressed qualitatively by using indirect proxies, for 25 example, features related to the cycle of methane through the lacustrine food web. Biogenic methane has highly depleted  $\delta^{13}$ C values (usually -850 to -850%), Whiticar, 1999), depending 26 on the methane production pathway and substrate availability. These depleted  $\delta^{13}$ C values can 27 be traced through the food web, for example, in low-level heterotrophs such as invertebrates. 28 Previous studies have linked depletion in the  $\delta^{13}$ C values at various stages in the food web to 29 the incorporation of carbon from of methane oxidising bacteria (MOB; van Hardenbroek et 30 31 al., 2010; Jones and Grey, 2011; Sanseverino et al., 2012). Recent studies have demonstrated 32 that some chironomid (non-biting midge) taxa utilise MOB as a food source within lakes

1 (Deines et al., 2007; van Hardenbroek et al., 2010). In thermokarst lakes, depleted  $\delta^{13}C$ 

2 values in larvae and fossil head capsules have been linked to increased methane flux (van

3 Hardenbroek et al., 2012). Wooller et al. (2012) also interpret negative shifts in  $\delta^{13}$ C values of

4 fossil chironomids and daphnia as an increase in methane availability.

5 MOB have been identified in sediments from a wide range of terrestrial and aquatic

6 environments. They are known to synthesise a number of specific compounds that can be

7 isolated. In particular, the compound diploptene (17  $\beta$ (H), 21  $\beta$ (H)-hop-22 (29)-ene), <u>is</u> a

8 hopanoid hydrocarbon derived from a range of bacterial sources., however due to the

9 <u>utilisation of methane as a carbon source, the  $\delta^{13}$ C values of diploptene derived from MOB</u>

10 will be more negative than if it were derived from other heterotrophic bacteria which utilise

11 organic carbon from vegetation. In marine sediments, diploptene has been identified as a

12 methanotrophic biomarker via <u>low-negative</u>  $\delta^{13}$ C values <u>in marine sediments and as well as in</u>

13 microbial mats associated with methane seeps (Elvert et al., 2001b; Pancost et al., 2000a,

14 2000b) as well as and Holocene peat (van Winden et al., 2010; Zheng et al., 2014). Diploptene

and the related diplopterol have been used to establish past patterns of MO from marine

16 sediment records (Jahnke et al., 1999; Pancost et al., 2000a) as well as lake sediments

17 (Spooner et al., 1994; Schouten et al., 2001), and peat deposits (Kip et al., 2010; van Winden

18 et al., 2012; Zheng et al., 2014). Therefore if MOB are present in the sediments of

19 thermokarst lakes, we would expect to see depleted  $\delta^{13}$ C values of diploptene.

To oxidise methane effectively, MOB require access to dissolved methane in sediments and 20 21 lake water. The assumption is, therefore, that isotopic depletion at or near the base of the food 22 web indicates oxidation of dissolved methane. The extent to which isotopic signals can be 23 used as a proxy for past methane ebullition flux in thermokarst lakes depends on the 24 relationship between ebullition and diffusion and the sensitivity of the isotope signal to 25 changing methane supply. In order to investigate these issues, we applied the approach used to identify MO at deep marine vents and seeps-lipid biomarkers from bacteria-to different 26 27 areas associated with known ebullition emission patterns in two Alaskan lakes. MOB are a 28 more direct proxy for methane than organisms higher in the food chain, and their use should 29 allow a better understanding of methane diffusion from sediments, particularly in areas of ebullition seeps. The presence and  $\delta^{13}$ C values of diploptene were used firstly to establish if 30 MO was occurring at levels detectable by biomarkers, and secondly to assess the degree of 31

1 MO observed in areas characterized by different modes of methane production and transport

2 to the atmosphere.

#### 3 2 Regional context & Study sites

4 Yedoma-like deposits that are similar to those described in, and common to, Siberia 5 (Schirrmiester et al 2011) can be found in Interior Alaska. These sediments can have a 6 relatively high organic content (Péwé, 1975) and are rich in excess ice. Thermokarst lakes that 7 develop in landscapes dominated by these deposits have been placed into the yedoma or non-8 yedoma types (as described above) in previous studies (Walter et al., 2008; Brosius et al., 9 2012; Sepulveda-Jauregui et al., 2015). Two lakes were sampled in April 2011 and July 2012 10 (Figure 2). Ace L. represents a yedoma-type lake (Sepulveda-Jauregui et al., 2015), where the permafrost soils surrounding the lake and eroding into the lake along its NE margin are 11 12 predominantly yedoma. Smith L. is classified as a non-yedoma lake in which Holocene-aged deposits are likely the main source of organic matter fuelling methane production. 13 Smith L. (64°51'55.92"N, 147°52'0.70"W; figure 2) is a shallow ( $\leq 4$  m), productive lake 14 15 located in Interior Alaska. It has a gentle bathymetric profile with average water depths between 1-3m. The lake is not subject to a strong fetch or high energy inflow or outflow. 16 17 Observations during the ice-free periods suggest high primary productivity, with blue/green algal blooms predominant throughout the summer months (KLD, personal observation). The 18 19 lake likely originated by thermokarst processes (Alexander and Barsdate, 1971); comparisons 20 of lake shorelines between the 1950s and today suggest that segments of the southern and 21 western margins have been actively thermokarsting actively thawing and eroding during 22 recent decades, and tilting trees currently lining the margin of a bay on the southeast shore are 23 further evidence of localized thermokarst. Smith Lake is a useful study site as its shallow 24 profile reduces the potential of production or storage of methane due to stratification. Ace L. (64°51'45.49N, 147°56'05.69W) is part of the Ace-Deuce Lake system (Alexander and 25 26 Barsdate, 1974) situated within an area covered by the Pleistocene Gold Hill and Goldstream 27 loess formations (Pewe 1975). Ace L. is thermokarst in origin and formed through the 28 thawing of ice bodies in the loess. The Ace-Deuce Lake system has high nutrient levels, and 29 therefore Ace Lake can be described as a eutrophic lake with a strong seasonal nutrient cycle 30 (Alexander and Barsdate, 1974).

#### 1 3 Methods

#### 2 **3.1 Establishing sample regions**

3 Walter Anthony and Anthony (2013) defined the 'thermokarst' zone for a number of lakes, 4 and we continue to use this definition here. The thermokarst zone was the region of active 5 thermokarst margin expansion observed using historical aerial photographs obtained during the past 60 years. In most lakes, the density of ebullition seeps is higher in thermokarst zones 6 7 compared to non-thermokarst zones (Walter Anthony and Anthony, 2013). In Ace and Smith 8 L., ebullition emissions were quantitatively monitored through a combination of winter-time 9 ice-bubble surveys and bubble-trap flux measurements via previous studies (Sepulveda-10 Jauregui et al., 2015) and our own summertime bubble counts (figure 2). We obtained surface 11 sediment cores well within the zone boundaries and as close to observed ebullition seep 12 locations as possible (figure 2). At Ace L., bubble counts may have been underrepresented 13 due to fetch-mediated surface turbulence disturbing visual counts of bubbles. However this 14 was an issue at all count sites, such that, any error encountered will be associated with the overall scale of emissions measured and not with bias between zones. In order to remove 15 16 water depth as a confounding variable and to increase the number of replications in each zone,

17 Ace L. was not sampled as it was much deeper than Smith L. centre (~9m).

#### 18 **3.2 Methane monitoring**

19 Ebullition gas samples were collected from seep locations in the thermokarst zone (n1 and n5 for Smith L, and Ace L. respectively) in the manner described in Walter Anthony et al. (2012) 20 21 for determination of bubble methane concentration, stable isotope analyses, and radiocarbon 22 dating. Gases were collected from submerged bubble traps into 60-ml glass serum vials 23 following Walter et al. (2008), sealed with butyl rubber stoppers, and stored under refrigeration in the dark until analysis in the laboratory. We measured methane concentration 24 25 using a Shimadzu 2014 equipped with an FID at the Water and Environmental Research Centre at University of Alaska Fairbanks (UAF). We determined  $\delta^{13}C_{CH4}$ , using a Finnegan 26 27 Mat Delta V, and  $\delta D_{CH4}$  on a Delta XP at Florida State University. Subsamples of gas were 28 combusted to CO<sub>2</sub>, purified, and catalytically reduced to graphite (Stuiver and Polach, 1977), and the  ${}^{13}C/{}^{12}C$  isotopic ratios were measured by accelerator mass spectrometry at the Woods 29 Hole Oceanographic Institution's National Ocean Sciences AMS Facility. Stable isotope 30 compositions are expressed in  $\delta$  (‰) = 103 ((R<sub>sample</sub>/R<sub>standard</sub>)-1), where R is <sup>13</sup>C/<sup>12</sup>C or D/H 31

1 and standards refers to the Vienna Pee Dee Belemnite (VPDB) and Vienna Standard Mean 2 Ocean Water (VSMOW), respectively. The analytical errors of the stable isotopic analysies 3 was are  $\pm 0.1 \% \delta^{13}$ C and  $\pm 1.0 \% \delta$ D. We express radiocarbon data as percent modern 4 carbon pmC (%) = ((<sup>14</sup>C/<sup>12</sup>C)<sub>sample</sub>/(<sup>14</sup>C/<sup>12</sup>C)standard) x 100, which is the percentage of 5 <sup>14</sup>C/<sup>12</sup>C ratio normalized to  $\delta^{13}$ C = -25‰ and decay corrected relative to that of an oxalic 6 standard in 1950 (Stuiver and Polach, 1977).

#### 7 3.3 Biomarker analysis

8 Surface sediment samples were retrieved using a gravity corer and the 0-5cm sequence was 9 extruded at 1-cm resolution and retained for analysis; the 1-2 cm slice was subsampled for biomarker analysis. Sample sizes can be found in table 1. -Two sequential extractions were 10 performed upon the samples. The first step was a modified Bligh and Dyer extraction (Bligh 11 and Dyer, 1959). Briefly, buffered water was prepared adjusting a solution of 0.05M KH<sub>2</sub>PO<sub>4</sub> 12 13 in water to pH 7.2 through the addition of NaOH pellets. Subsequently, a monophasic solvent 14 mixture was made up with buffered water, CHCl<sub>3</sub> and MeOH (4:5:10 v/v). Samples were 15 sonicated in Bligh-Dyer solvent mixture for 15 minutes and then centrifuged at 3000 rpm for 16 5 minutes. Supernatant was collected in a round bottom flask. This step was repeated twice 17 and all supernatants were combined and dried to obtain the total lipid extraction (TLE) 18 labelled TLE1. Post-extraction sediment residues were air-dried. The Bligh and Dyer post-19 extraction residues were sonicated in DCM for 15 minutes and then centrifuged at 3000 rpm 20 for 5 minutes. This step was repeated first with DCM:MeOH (1:1, v/v) and then with MeOH. 21 Supernatants were combined after every step of sonication-centrifugation to obtain TLE2. 22 Both TLE1 and TLE2 were then combined to yield the final TLE. 23 The TLE was split into three fractions of increasing polarity using silica flash column

24 chromatography (Oba et al., 2006; Pitcher et al., 2009). Silica gel columns (0.5 g, 60 Å

25 particle size) were prepared and conditioned with 4 ml of *n*-hexane:ethyl acetate (3:1, v/v).

Fractions were eluted with 3 ml of *n*-hexane:ethyl acetate (3:1, v/v) to obtain the simple lipid

27 fraction, 3 ml of ethyl acetate to obtain glycolipids and 10ml of MeOH to obtain

- 28 phospholipids. The simple lipid fraction was further split into neutral lipid and the fatty acid
- 29 fractions. The organic phase was then collected into a round bottom flask and Na<sub>2</sub>SO<sub>4</sub>
- 30 anhydrous was added until complete removal of water. Silica gel columns (again, 0.5 g, 60 Å
- 31 particle size) were prepared and conditioned with 4 ml of the recently prepared CHCl<sub>3</sub> sat

- 1 solution. The simple lipid fraction was then loaded onto the column and subsequently, the
- 2 neutral lipid fraction was eluted with 9 ml of CHCl<sub>3</sub> sat. Finally, the neutral lipids were
- 3 separated into apolar and polar lipid fractions. Columns were prepared with approximately 0.5
- 4 g of activated alumina (Al<sub>2</sub>O<sub>3</sub>) and compounds eluted with 4 ml of *n*-hexane:DCM (9:1, v/v)
- 5 and 3 ml of DCM:MeOH (1:2, v/v) to yield the two fractions, respectively. Here, we focus on
- 6 analyses of the neutral lipid apolar fraction.

# 7 3.4 Compound identification and Compound-specific $\delta^{13}$ C isotope analysis

- 8 GC-MS analyses were performed using a Thermoquest Finnigan Trace GC and MS. The GC
- 9 was fitted with an on-column injector and the stationary phase was CP Sil5-CB. Detection
- 10 was achieved with electron ionization (source at 70 eV, scanning range 50-580 Daltons). The
- 11 temperature program consisted of three stages: 70-130 °C at 20 °C/min rate; 130-300 °C at 4
- 12 °C/min; and 300 °C, temperature held for 10 min.
- 13 Gas chromatography combustion isotope ratio mass spectrometry (GC-IRMS) was performed
- 14 using a ThermoScientific Trace GC Ultra coupled to a Conflo IV interface and DeltaV mass
- 15 Spectrometer. The GC conditions and program were the same as for GC-MS analyses.
- 16 Calibration was achieved using CO<sub>2</sub> reference gas of known isotopic composition and sample
- 17  $\delta$ 13C values were expressed against the standard VPDB. All measurements were performed
- 18 in duplicate.

# 19 **3.5 Mass Balance equation**

- 20 A carbon isotopic mass balance equation (Equation 1), or two-part mixing model, was
- 21 developed to evaluate the contribution of MOB to the total bacterial biomass, and therefore,
- 22 the relative amount of oxidation occurring at each sample location. <u>By developing this</u>
- 23 mixing model and considering in more detail the potential end member values for the  $\delta^{13}C$
- 24 values of diploptene derived from different sources (MOB and other heterotrophic bacteria)
- 25 we can get a semi-quantitative idea of the distribution patterns of MOB across the samples.
- 26 The resulting end member values are given in table 24. The equation is as follows:

27 
$$f_{\text{mob}} = \frac{\delta^{13} C_{\text{dip}\_\text{sample}} - \delta^{13} C_{\text{hetero}\_\text{hopanedip}}}{\delta^{13} C_{\text{mob}\_\text{hopanedip}} - \delta^{13} C_{\text{hetero}\_\text{hopanedip}}}$$
(1)

28  $f_{mob}$  is the fraction of diploptene generated by MOB and  $\delta^{13}C_{dip\_sample}$  is the stable carbon 29 isotopic composition of diploptene in a given sample.  $\delta^{13}C_{hetero\_hopane\_dip\_}$  is the inferred  $-\delta^{13}C$ 

1 value of the hopanoids diploptene- if it were derived solely from heterotrophic bacteria, the 2 inferred other primary source of this hopane hopanoids in this setting. , and It is expressed as the  $\delta^{13}C_{\text{bacterial biomass}} - \Delta^{13}C_{\text{biosynthesis}}$  (~4%). The  $\delta^{13}C$  values of diploptene derived from 3 Hheterotrophic bacteria -will primarily reflect the values of the substrate carbon which in this 4 5 instance will be organic material and not methane. These values are therefore unlikely to vary; however a  $\sim 2$  to 4‰ shift can occur during lipid biosynthesis (Pancost and Sininghe 6 Damsté 2003, and references therein).  $\delta^{13}C_{\text{mob-hopane-dip}}$  is the likely value of the diploptene 7 hopanoids if it were derived solely from MOB. It is calculated from the  $\delta^{13}C_{\text{methane}}$  minus the 8 9 fractionation that occurs during carbon uptake by methanotrophs (0-30%; Jahnke et al., 1999) minus the biosynthetic fraction during lipid synthesis ( $\Delta^{13}C_{\text{biosynthsis}}$ ; ~10%). The  $\delta^{13}C_{\text{methane}}$ 10 is the measured value of methane captured at seep locations in the thermokarst zones at each 11 lake. As the value is based on a limited number of data (n1 and n5 for Smith L. and Ace L. 12 respectively), it is likely there will be more variation than is seen in the model. Furthermore 13 MOB can be significantly depleted in comparison to the source carbon they utilise (Whiticar 14 15 1999); isotopic differences can be as large 30‰ (Jahnke et al., 1999). In order to incorporate this large range, we used both the minimum and maximum value of fractionation (0 and 30‰) 16 17 to show different scenarios rather than assuming a single value. This should also cover any potential variation due to differing  $\delta^{13}C_{\text{methane}}$  Therefore the equation was calculated twice, 18 once using  $\delta^{13}C_{mob\_dip}$  min and once using  $\delta^{13}C_{mob\_dip}$  max. 19 20 With little information available on the fractionation of hopanoids during their biosynthesis by MOB, we assumed a conservative value of 10% for our study. Four end-member values 21 were calculated, taking into account maximum and minimum extremes for  $\delta^{13}C_{dip}$  and 22

23  $\delta^{13}C_{hetero}$  (Table 42). A threshold of 10% was used arbitrarily to identify the point at which we 24 considered MOB to be contributing to the diploptene signal.

#### 25 4 Results

Early-winter ice-bubble surveys combined with bubble-trap measurements of ebullition flux
and bubble methane concentration revealed that ebullition seeps occur with high density in

- 28 the thermokarst zone (2.27 seeps  $m^2$  and 4.2 seeps  $m^2$  for Smith L. and Ace L., respectively)
- 29 compared to the rest of the lake (0.35 seeps  $m^2$  and 0.67 seeps  $m^2$  for Smith L. and Ace L.,
- 30 respectively). Seep ebullition values in the thermokarst bays were 85 and 151 mg  $CH_4 m^2 d^1$
- 31 for Smith L. and Ace L., respectively (Figure 2). In the rest of lake (lake centre and non-
- 32 thermokarst margins) seep ebullition was 6 and 20 mg  $CH_4 \text{ m}^{-2} \text{ d}^{-1}$  for Smith L. and Ace L.,

- 1 respectively. The  $\delta^{13}$ C values for methane in bubbles collected from seeps in the thermokarst
- 2 zones were -60.9‰ and -64.6‰ for Smith Lake and Ace L., respectively. At Smith L., the
- 3 radiocarbon age of methane in ebullition bubbles collected adjacent to the margin was ~2ka,
- 4 indicating a dominant Holocene carbon source (likely decomposing near-surface peat). No
- 5 radiocarbon dates of methane were available at Ace L.
- Diploptene was detected in all but one of the samples analysed (Table 23; figure 3). This
  sample was not part of further analysis. <u>The values ranged from -68.2 to -38.8%</u> and had an
  overall standard deviation of 7.8%.
- 9 In the Ace L. thermokarst zone, diploptene values ranged from the lowest value for the whole
- 10 dataset of -68.2% to -50.1%. The most negative value was found at the greatest water depth

11 (3.2m) and was the only sample that does not lie within 1 standard deviation of the mean for

12 this thermokarst zone. However, another sample at the same depth was far less depleted (-

- 13 50.1‰), which suggests the low  $\delta^{13}$ C value is not explained by water depth. In Smith L.,
- 14 diploptene  $\delta^{13}$ C values ranged from -56.8‰ to -38.8‰.
- 15 Samples from the centre and edge of Smith L. (n=6, n=3 respectively) were compared and a
- 16 Mann-Whiney U test applied (H0: diploptene  $\delta^{13}$ C values are not different). The values for
- 17 Smith L. indicates that the MOB proportional contributions to the total bacterial communities
- 18 differed significantly between the two sample zones, values from the thermokarst zone of
- 19 Smith L. being higher (-42.9 to -38.8%) than those in the lake centre (-56.8 to -46.9%).
- 20 Diploptene  $\delta^{13}$ C values in the thermokarst zone of Ace L. are similar to those of the lake
- 21 centre at Smith, and values from the Smith thermokarst zone are higher than both of these.
- 22 Thermokarst zone diploptene  $\delta^{13}$ C values at Ace L<sub>2</sub>ake were more negative than those at
- 23 Smith <u>Lake L.</u> by at least 10‰, despite methane  $\delta^{13}$ C values being less than 5‰ different.
- 24 However, the samples in the thermokarst zone of Ace L. and the centre of Smith L. (n=4, n=6
- 25 respectively) were not significantly different according to a Mann Whitney U test.
- 26 The potential contributions of MOB, under different end-member assumptions, to the
- 27 diploptene signal are shown in Table  $\frac{34}{2}$ . The minimum and maximum possible contributions
- range from 19 to 85%, 7 to 27% and 19 to 63% for Ace L. thermokarst zone, Smith L.
- 29 thermokarst zone and Smith centre, respectively.

#### 1 5 Discussion

2 **5.1** Distribution of ebullition seeps

The spatial distribution of ebullition seeps at Ace L. and Smith L. adheres to the general
pattern of seep occurrences as described in other studies, (Walter Anthony and Anthony,

5 2013), in that the highest density of methane ebullition seeps were found in the thermokarst6 zone.

#### 7 5.2 The presence and spatial variability of MOB

The  $\delta^{13}$ C values of diploptene ranged from -68.2 to -38.8-‰ (Figure <u>34</u>), values similar to 8 9 those that have been previously highlighted- as evidence for methanotrophy in lacustrine sediments (-64‰ to -55‰; Spooner et al., 1994; Naeher et al., 2014), marine sediments 10 11 (-62‰ to -35‰; Freeman et al., 1994; Thiel et al., 2003) and in wetlands (-40‰ to -30‰ to; van Winden et al., 2010; Zheng 2014). Therefore, we conclude that diploptene  $\delta^{13}$ C values are 12 13 reflecting the presence of MOB bacteria in lake sediments. The lowest values in Ace L. are 14 among the lowest reported for lacustrine (or other terrestrial) systems, suggesting a relatively 15 high degree of methanotrophy in those sites. In the thermokarst zone at Ace L., the diploptene values were highly variable but all suggested some degree of MO was occurring, and the 16 17 fraction of diploptene derived from MOB was >10% even under the most conservative 18 assumptions (Table 34).

19 The results of the mixing model suggest that MOB can contribute anywhere between 7-83% 20 of the diploptene production across all sampled areas (Table 34). These estimates have a large 21 degree of uncertainty associated with them and -Wwe note that there are some important caveats to using this mixing model. Crucially, diploptene is not derived from all bacteria nor 22 23 even all methanotrophic bacteria (Rohmer et al., 1987). Nor does it likely occur in constant biomass-to-lipid ratios in those organisms from which it can derive, such that extrapolations 24 25 from a diploptene mass balance to inferring bacterial biomass distributions should be done 26 cautiously. They are best considered semi-quantitative. Nonetheless, a MOB contribution to 27 total biomass of ~10 to 80% is similar to that derived from other studies (11-80%; Bastviken et al. 2003; Sundh et al. 2005; Kankaala et al. 2006). Regardless of absolute MOB estimates, 28 29 our data show that the centre of Smith L. and the thermokarst zone at Ace L. have the highest 30 proportion of MOB in the total bacterial biomass.

- The data presented here allow us to develop, alongside other studies, models of methane
   production and emission pathways in thermokarst lakes.
- 3 At Ace L., MOB biomass was high relative to other samples collected in this study and in the
- 4 context of previous studies. Ace L. is a 'yedoma-type' lake and has a high methane ebullition
- 5 flux (151 mg CH<sub>4</sub> m<sup>2</sup> d<sup>1</sup>), likely derived from older (e.g. Pleistocene), deeper sediments in the
- 6 talik bulb (Walter et al., 2008; Sepulveda-Jauregui 2015). Given the coincidence of high
- 7 bubble counts and high estimated MOB biomass, it could be assumed that the supply of
- 8 dissolved methane and therefore MO is high in the thermokarst zone and this methane might
- 9 <u>be derived from thermokarst specific sources</u>.

10 Ace L. appears to be representative of the enhancement model, whereby methane ebullition flux from bubble tubes increase the amount of methane diffusion from the sediments. In Ace 11 12 L., and by extension other yedoma-type thermokarst lakes, where methane is produced in deep sediments the increased contact time with sediment (both over distance and time taken 13 14 for bubbles to reach the sediment-water interface) may allow for increased methane diffusion in adjacent sediments. Alternatively, thermokarst erosion of yedoma-type permafrost is also 15 16 known to supply nitrogen and phosphorus to lakes (Walter Anthony et al. 2014), enhancing primary production, which in turn can fuel methanogenesis and MO from contemporary 17 18 (atmospheric) carbon (Martinez-Cruz et al., 2015). We cannot definitively distinguish 19 between these alternatives since the carbon utilised by MOB observed in Ace L. could be derived from deep. <sup>14</sup>C-depleted methane and/or from shallow-sediment, contemporary 20 21 methane. However it could be argued that even if the methane that is being oxidised is from 22 near-surface sediments, the high level of production is due to the lake type (yedoma) and the thawing and eroding margins. This might be a common pattern in these types of lakes and 23 could be reflected in the  $\delta^{13}$ C values of diploptene, however this needs to be tested with 24 further research. 25

- 26 Within the thermokarst zone at Smith L. the  $\delta^{13}$ C values of diploptene were less variable
- 27 (range: 10‰) than the Ace L. thermokarst zone (18‰) and the  $\delta^{13}$ C values were overall more
- enriched (-42.9 to -38.8‰). In fact, the thermokarst zone in Smith L. had the lowest
- 29 proportion of MOB for the entire dataset, with a MOB contribution to diploptene being
- 30 equivocal for most of these samples. Conversely, samples from the centre of Smith L. had
- 31 diploptene  $\delta^{13}$ C values that were similar to those of the Ace<u>L</u> thermokarst zone. The
- 32 differences between the centre and the thermokarst zone could arise from alterations in the

microbial community that manifest as different MOB expressions of hopanoids, for example, 1 2 Smith L. thermokarst zone MOB might not be biosynthesising diploptene or its precursor. Alternatively, there may be differences in the balance of MO for energy versus biomass 3 production. Another explanation for the difference in  $\delta^{13}$ C values, which could be validated 4 5 through further investigation, could be due to differences in the methane production pathways as highlighted by Walter et al. (2008). The higher  $\delta^{13}$ C values of diploptene could be due to 6 7 more enriched methane formed through acetate fermentation. The most direct interpretation 8 given the currently dataset, however, is that MOB are more abundant in the centre of the lake 9 than at the thermokarst margin and, by extension, more MO is taking place in the lake centre. 10 Given the pattern of high MO in the centre of Smith L. and less MO at the edge but more flux 11 to atmosphere via ebullition, it seems that Smith methane dynamics are more akin to those of 12 'clastic' lakes or other, non-thermokarst boreal lakes (e.g. Bastviken et al., 2004). The patterns at Smith L. also suggest that methane dynamics in the thermokarst zone follow the 13 14 by-pass model in which methane ebullition is an independent process that interacts weakly 15 with the lacustrine system.

Overall, the Smith thermokarst zone had lower methane ebullition rates (85 mg  $CH_4 \text{ m}^{-2} \text{ d}^{-1}$ ) 16 and less negative  $\delta^{13}$ C of methane as measured from ebullition flux (-60.9‰) than Ace L. It 17 is possible that this methane is not produced in the talik, but in near-surface sediments likely 18 19 derived from peat slumping at the margin. This is supported by the late Holocene radiocarbon 20 date of ebullition seep methane. The large size of the sediment blocks and the early stage of 21 decomposition of the organic material that slump into the lake may mean there is less exposed 22 substrate surface area and less methane production, as compared to yedoma-lake production from the fine-grained and more labile sediments. Production in shallower sediments (and 23 often shallow water depths) means reduced partial pressure and faster release of bubbles from 24 25 the sediment. Here, if bubble tubes initiate in shallower sediments (that are shallower than the talik bulb but deeper than the anoxic near-surface sediments) and the overall number, size and 26 27 intensity of bubble tubes is reduced, then the connection between ebullition and diffusion could be decoupled. 28

29 Whether there is a reliable connection between ebullition flux and high diffusion in the

- 30 thermokarst zone is still to be determined. Currently, as the data stand, it is difficult to
- 31 decipher a clear pattern that can be linked to thermokarst specific methane production., but

<u>T</u>the results of this novel but preliminary study highlight the need to continue research in this
 area.

#### 3 5.3 Assessing past and current carbon cycling in thermokarst lakes

A crucial outcome of this study is the large variability seen in the  $\delta^{13}$ C values of diploptene 4 across small spatial distances which cannot be linked to specific types of methane production 5 (e.g. near-surface or deeper, thermokarst specific -production). This is an important finding, 6 as often whole lakes can be represented by a single sampling site in palaeoenvironmental 7 studies. Such large fluctuations in  $\delta^{13}$ C values in surface sediments, which were taken as 8 9 replicates (e.g. repeat samples from the same zone within a lake), highlight the need for caution when interpreting shifts in  $\delta^{13}$ C values through time (i.e., down a single sediment 10 core). 11

While the differences in diploptene  $\delta^{13}$ C values between chosen study zones discussed above 12 are statistically significant, the sample number is small, and this topic could benefit from 13 further sampling. There is a large degree of heterogeneity in the values in all three study 14 15 areas. Interestingly, previous studies of MOB in lake sediments also show large variability in 16 bacterial communities across small spatial extents (Kankaala et al., 2006). This could have 17 implications for interpretation of not only biomarkers but also other geochemical records. For 18 example, it is unclear how high spatial and temporal variability in MOB biomass affects the 19 isotopic composition of consumers higher in the food web. The biological and geochemical 20 connections between MOB and higher trophic organisms need to be better understood in 21 order to interpret past methane emissions.

#### 22 6 Conclusions

23 A primary aim of our research was to contribute towards the understanding of the links 24 between methane production, transport and recycling in thermokarst lakes. Diploptene  $\delta^{13}$ C 25 values were used as a proxy for MO that could to test whether these can be linked to variations in methane supply via diffusion in thermokarst lakes. Diploptene was present in 26 almost all samples and its  $\delta^{13}$ C values were highly variable. A two-part mixing model 27 highlighted potential variation in total MOB biomass with almost no MOB contributing to 28 29 bacterial biomass in some samples but forming over half the total bacterial population in 30 others. However the results were highly variable and suggest that, *L*like methane production, MO is highly complex, both in terms of its spatial distribution and in relation to the type of 31

1 substrate available. A single model for thermokarst lakes is unlikely to capture all patterns 2 present at both the inter-lake and intra-lake level, and as the data stand, there is a large amount of variability which cannot be linked to specific types of methane production.- Thus, it is 3 crucial that interpretation of diploptene  $\delta^{13}$ C values (and other MO proxies) in 4 5 palaeoenvironmental investigations take into account lake type (e.g., yedoma or non-yedoma) and possible spatial heterogeneity in methane production pathways. Moreover, future work 6 should examine localized spatial variability of MO within lakes and how spatial variation is 7 8 integrated temporally, as this may critically affect observed down-core patterns of biomarkers 9 and their isotopic signals. We conclude that given the current data, further research should be 10 completed in order to understand the variability in d13c diploptene values prior to utilisation of this method for the reconstruction of methane cycling in lacustrine systems. We conclude 11 that diploptene biomarkers have considerable potential to help reconstruct patterns of methane 12 13 eveling in lakes and, with certain caveats, particularly attention to context, past methane 14 dynamics.

15

### 16 Acknowledgements

- 17 This research was supported by a NERC grant (NE/K000233/1) to M. Edwards and
- 18 P. Langdon, a QRA new researcher's award to K. Davies and a PhD Scholarship to
- 19 K. Davies from Geography and Environment, University of Southampton. We
- 20 gratefully acknowledge field and equipment assistance from Nancy Bigelow,
- 21 Charlotte Clarke, Rob Collier and Ben Gaglioti, and permission from the owners to
- work at Ace Lake. Mark Dover (Cartography Unit, G&E) made valuable
- 23 improvements to the figures.
- 24

#### 25 References

- 26 Alexander, V. and Barsdate, R. J.: Physical Limnology, Chemistry and Plant Productivity of a
- 27 Taiga Lake, Int. Revue ges. Hydrobiol, 56(6), 825–872, 1971.
- 28 Alexander, V. and Barsdate, R. J.: Limnological Studies of a Subarctic Lake System, Int.
- 29 Revue ges. Hydrobiol, 59(6), 737–753, 1974.

- 1 Alperin, M. J. and Hoehler, T. M.: Anaerobic methane oxidation by archaea/sulfate-reducing
- 2 bacteria aggregates: 1. Thermodynamic and physical constraints, American Journal of Science,
- 3 309(10), 869–957, doi:10.2475/10.2009.01, 2010.
- 4 Bastviken, D.: Methane emissions from lakes: Dependence of lake characteristics, two
- regional assessments, and a global estimate, Global Biogeochemical Cycles, 18(4), GB4009,
  doi:10.1029/2004GB002238, 2004.
- 7 Bastviken, D., Ejlertsson, J. and Tranvik, L.: Measurement of methane oxidation in lakes: a
- 8 comparison of methods., Environmental science & technology, 36(15), 3354–61 [online]
- 9 Available from: http://www.ncbi.nlm.nih.gov/pubmed/12188365, 2002.
- 10 Bastviken, D., Ejlertsson, J., Sundh, I. and Tranvik, L.: Methane as a source of carbon and
- energy for lake pelagic food webs, Ecology, 84(4), 969–981, doi:10.1890/0012-
- 12 9658(2003)084[0969:MAASOC]2.0.CO;2, 2003.
- Bastviken, D., Tranvik, L. J., Downing, J. A., Crill, P. M. and Enrich-prast, A.: Freshwater
  Methane Emissions Offset the Continental Carbon Sink, Science, 331, 50, 2011.
- 15 Birgel, D. and Peckmann, J.: Aerobic methanotrophy at ancient marine methane seeps: A
- 16 synthesis, Organic Geochemistry, 39(12), 1659–1667, doi:10.1016/j.orggeochem.2008.01.023,
- 17 2008.
- 18 Bligh, E. G. and Dyer, W. J.: A rapid method of total lipid extraction and purification,
- 19 Canadian journal of biochemistry and physiology, 37(8), 911–917, 1959.
- 20 Briggs, B. R., Pohlman, J. W., Torres, M., Riedel, M., Brodie, E. L. and Colwell, F. S.:
- 21 Macroscopic biofilms in fracture-dominated sediment that anaerobically oxidize methane.,
- 22 Applied and environmental microbiology, 77(19), 6780–7, doi:10.1128/AEM.00288-11, 2011.
- 23 Brosius, L. S., Walter Anthony, K. M., Grosse, G., Chanton, J. P., Farquharson, L. M.,
- 24 Overduin, P. P. and Meyer, H.: Using the deuterium isotope composition of permafrost
- 25 meltwater to constrain thermokarst lake contributions to atmospheric CH 4 during the last
- 26 deglaciation, Journal of Geophysical Research: Biogeosciences, 117(G1), n/a–n/a,
- 27 doi:10.1029/2011JG001810, 2012.
- 28 Colins, M., Knutti, R., Arblaster, J., Dufresne, J. L., Fichefet, T., Friedlingstein, P., Gao, X.,
- 29 Gutowski, W. J., Johns, T., Krinner, G., Shongwe, M., Tebaldi, C., Weaver, A. J. and Wehner,
- 30 M.: Long-term Climate Change: Projections, Commitments and Irreversibility, in Climate
- 31 Change 2013: The Physical Science Basis. Contribution of the working group I to the Fifth
- 32 Assessment Report of the intergovernmental Panel on Climate Change, edited by T. F.
- 33 Stocker, D. Qin, G.-K. Plattner, M. Tignor, S. K. Allen, J. Boschung, A. Nauels, X. Y, V. Bex,
- 34 and P. M. Midgley, Cambridge University Press, Cambridge, United Kingdom and New York,
- 35 USA., 2013.
- 36 Deines, P., Bodelier, P. L. E. and Eller, G.: Methane-derived carbon flows through methane-
- 37 oxidizing bacteria to higher trophic levels in aquatic systems., Environmental microbiology,
- 38 9(5), 1126–34, doi:10.1111/j.1462-2920.2006.01235.x, 2007.

- 1 Delsontro, T., Kunz, M. J., Kempter, T., Wehrli, B. and Senn, D. B.: Spatial Heterogeneity of
- 2 Methane Ebullition in a Large Tropical Reservoir, Environmental science & technology, 45,
- 3 9866–9873, 2011.
- 4 Elvert, M. and Niemann, H.: Occurrence of unusual steroids and hopanoids derived from
- 5 aerobic methanotrophs at an active marine mud volcano, Organic Geochemistry, 39(2), 167–
- 6 177, doi:10.1016/j.orggeochem.2007.11.006, 2008.
- 7 Elvert, M., Greinert, J., Suess, E. and Whiticar, M. J.: Carbon Isotopes of Biomarkers Derived
- 8 from Methane Oxidizing Microbes at Hydrate Ridge, Cascadia Convergent Margin, in
- 9 Natural Gas Hydrates: Occurrence, Distribution, and Detection, Volume 124, edited by C. K.
- 10 Paull and W. D. Dillon, pp. 115–129, Wiley Online Library., 2001a.
- 11 Elvert, M., Whiticar, M. . and Suess, E.: Diploptene in varved sediments of Saanich Inlet:
- 12 indicator of increasing bacterial activity under anaerobic conditions during the Holocene,
- 13 Marine Geology, 174(1-4), 371–383, doi:10.1016/S0025-3227(00)00161-4, 2001b.
- 14 Freeman, K. H., Wakeham, S. G. and Hayes, J. M.: Predictive isotopic biogeochemistry:
- 15 hydrocarbons from anoxic marine basins., Organic geochemistry, 21(6-7), 629–44, 1994.
- 16 Greinert, J., Lewis, K. B., Bialas, J., Pecher, I. a., Rowden, a., Bowden, D. a., De Batist, M.
- 17 and Linke, P.: Methane seepage along the Hikurangi Margin, New Zealand: Overview of
- 18 studies in 2006 and 2007 and new evidence from visual, bathymetric and hydroacoustic
- 19 investigations, Marine Geology, 272(1-4), 6–25, doi:10.1016/j.margeo.2010.01.017, 2010.
- 20 Van Hardenbroek, M., Heiri, O., Grey, J., Bodelier, P. L. E., Verbruggen, F. and Lotter, A. F.:
- 21 Fossil chironomid d13C as a proxy for past methanogenic contribution to benthic food webs
- 22 in lakes?, Journal of Paleolimnology, 43(2), 235–245, doi:10.1007/s10933-009-9328-5, 2010.
- 23 Van Hardenbroek, M., Heiri, O., Parmentier, F. J. W., Bastviken, D., Ilyashuk, B. P., Wiklund,
- J. a., Hall, R. I. and Lotter, A. F.: Evidence for past variations in methane availability in a
- 25 Siberian thermokarst lake based on d13C of chitinous invertebrate remains, Quaternary
- 26 Science Reviews, 1–11, doi:10.1016/j.quascirev.2012.04.009, 2012.
- 27 He, R., Wooller, M. J., Pohlman, J. W., Quensen, J., Tiedje, J. M. and Leigh, M. B.: Shifts in
- 28 identity and activity of methanotrophs in arctic lake sediments in response to temperature
- changes., Applied and environmental microbiology, 78(13), 4715–23,
- 30 doi:10.1128/AEM.00853-12, 2012.
- 31 Jahnke, L. L., Summons, R. E., Hope, J. M. and Des Marais, D. J.: Carbon isotopic
- 32 fractionation in lipids from methanotrophic bacteria II : The effects of physiology and
- 33 environmental parameters on the biosynthesis and isotopic signatures of biomarkers,
- 34 Geochimica et Cosmochimica Acta, 63(1), 79–93, 1999.
- Jones, R. I. and Grey, J.: Biogenic methane in freshwater food webs, , 213–229,
   doi:10.1111/j.1365-2427.2010.02494.x, 2011.
- 37 Joyce, J. and Jewell, P. W.: Physical Controls on Methane Ebullition from Reservoirs and
- 38 Lakes, Environmental and Engineering Geoscience, IX(2), 167–178, 2003.

- 1 Kankaala, P., Huotari, J., Peltomaa, E., Saloranta, T. and Ojala, A.: Methanotrophic activity in
- 2 relation to methane efflux and total heterotrophic bacterial production in a stratified, humic,
- 3 boreal lake, Limnology and Oceanography, 51(2), 1195–1204, doi:10.4319/lo.2006.51.2.1195,
- 4 2006.
- 5 Kessler, M. a., Plug, L. J. and Walter Anthony, K. M.: Simulating the decadal- to millennial-
- 6 scale dynamics of morphology and sequestered carbon mobilization of two thermokarst lakes
- 7 in NW Alaska, Journal of Geophysical Research: Biogeosciences, 117(G2), n/a–n/a,
- 8 doi:10.1029/2011JG001796, 2012.
- 9 Liebner, S. and Wagner, D.: Abundance, distribution and potential activity of methane
- 10 oxidizing bacteria in permafrost soils from the Lena Delta, Siberia., Environmental
- 11 microbiology, 9(1), 107–17, doi:10.1111/j.1462-2920.2006.01120.x, 2007.
- 12 Martens, C. P. S. and Klump, J. V. A. L.: Biogeochemical cycling in an organic-rich coastal
- marine basin-I. Methane sediment-water exchange processes, Geochimica et Cosmochimica
   Acta, 44, 471–490, 1980.
- 15 Martinez-Cruz, K., Sepulveda-Jauregui, A., Walter Anthony, K. and Thalasso, F.: Geographic
- 16 and seasonal variation of dissolved methane and aerobic methane oxidation in Alaskan lakes,
- 17 Biogeosciences Discussions, 12(5), 4213–4243, doi:10.5194/bgd-12-4213-2015, 2015.
- 18 Naeher, S., Niemann, H., Peterse, F., Smittenberg, R. H., Zigah, P. K. and Schubert, C. J.:
- 19 Tracing the methane cycle with lipid biomarkers in Lake Rotsee (Switzerland), Organic
- 20 Geochemistry, 66, 174–181, doi:10.1016/j.orggeochem.2013.11.002, 2014.
- 21 Oba, M., Sakata, S. and Tsunogai, U.: Polar and neutral isopranyl glycerol ether lipids as
- biomarkers of archaea in near-surface sediments from the Nankai Trough, Organic
  geochemistry, 37(12), 1643–1654, 2006.
- 24 Pancost, R. ., Hopmans, E. . and Sinninghe Damsté, J. .: Archaeal lipids in Mediterranean
- 25 cold seeps: molecular proxies for anaerobic methane oxidation, Geochimica et Cosmochimica
- 26 Acta, 65(10), 1611–1627, doi:10.1016/S0016-7037(00)00562-7, 2001.
- 27 Pancost, R. D., Damsté, J. S. S., De, S., Maarel, M. J. E. C. Van Der and Gottschal, J. C.:
- 28 Biomarker Evidence for Widespread Anaerobic Methane Oxidation in Mediterranean
- 29 Sediments by a Consortium of Methanogenic Archaea and Bacteria, Applied and
- 30 environmental microbiology, 66(3), 1126–1132, doi:10.1128/AEM.66.3.1126-
- 31 1132.2000.Updated, 2000a.
- 32 Pancost, R. D., Geel, B. Van, Baas, M. and Sinninghe Damsté, J. S.: d13C values and
- 33 radiocarbon dates of microbial biomarkers as tracers for carbon recycling in peat deposits,
- 34 Geology, 28(7), 663–666, doi:10.1130/0091-7613(2000)28<663, 2000b.
- 35 Péwé, T. L.: Quaternary geology of Alaska,, 1975.
- 36 Pitcher, A., Hopmans, E. C., Schouten, S. and Damsté, J. S. S.: Separation of core and intact
- 37 polar archaeal tetraether lipids using silica columns: insights into living and fossil biomass
- 38 contributions, Organic Geochemistry, 40(1), 12–19, 2009.

- 1 Reeburgh, W. S.: Oceanic methane biogeochemistry., Chemical reviews, 107(2), 486–513,
- 2 doi:10.1021/cr050362v, 2007.
- 3 Sanseverino, A. M., Bastviken, D., Sundh, I., Pickova, J. and Enrich-Prast, A.: Methane
- 4 carbon supports aquatic food webs to the fish level., PloS one, 7(8), e42723,
- 5 doi:10.1371/journal.pone.0042723, 2012.
- 6 Scandella, B. P., Varadharajan, C., Hemond, H. F., Ruppel, C. and Juanes, R.: A conduit
- dilation model of methane venting from lake sediments, Geophysical Research Letters, 38(6),
   n/a-n/a, doi:10.1029/2011GL046768, 2011.
- 9 Schirrmeister, L., Froese, D., Tumskov, V., Grosse, G. and Wetterich, S.: Yedoma,
- Encyclopedia of Quaternary Science, 3, 542–552, doi:10.1016/B978-0-444-53643-3.00106-0,
  2013.
- 12 Sepulveda-Jauregui, a., Walter Anthony, K. M., Martinez-Cruz, K., Greene, S. and Thalasso,
- 13 F.: Methane and carbon dioxide emissions from 40 lakes along a north–south latitudinal
- 14 transect in Alaska, Biogeosciences Discussions, 11(9), 13251–13307, doi:10.5194/bgd-11-
- 15 13251-2014, 2014.
- 16 Sepulveda-Jauregui, a., Walter Anthony, K. M., Martinez-Cruz, K., Greene, S. and Thalasso,
- 17 F.: Methane and carbon dioxide emissions from 40 lakes along a north–south latitudinal
- 18 transect in Alaska, Biogeosciences, 12(11), 3197–3223, doi:10.5194/bg-12-3197-2015, 2015.
- 19 Shirokova, L. S., Pokrovsky, O. S., Kirpotin, S. N., Desmukh, C., Pokrovsky, B. G., Audry, S.
- 20 and Viers, J.: Biogeochemistry of organic carbon, CO2, CH4, and trace elements in
- 21 thermokarst water bodies in discontinuous permafrost zones of Western Siberia,
- 22 Biogeochemistry, 113(1-3), 573–593, doi:10.1007/s10533-012-9790-4, 2012.
- 23 Spooner, N., Rieley, G., Collister, J. W., Lander, I. M., Cranwell, I. P. A. and Maxwell, J. R.:
- 24 Stable carbon isotopic correlation of individual biolipids in aquatic organisms and a lake
- bottom sediment, Organic Geochemistry, 21(6/7), 823–827, 1994.
- Stuiver, M. and Polach, H. A.: Discussion; reporting of C-14 data., Radiocarbon, 19(3), 355–
  363, 1977.
- 28 Sundh, I., Bastviken, D. and Tranvik, L. J.: Abundance, activity, and community structure of
- 29 pelagic methane-oxidizing bacteria in temperate lakes, Applied and environmental
- 30 microbiology, 71(11), 6746–6752, 2005.
- 31 Thiel, V., Blumenberg, M., Pape, T., Seifert, R. and Michaelis, W.: Unexpected occurrence of
- hopanoids at gas seeps in the Black Sea, Organic Geochemistry, 34(1), 81–87,
- 33 doi:10.1016/S0146-6380(02)00191-2, 2003.
- Trotsenko, Y. a and Khmelenina, V. N.: Aerobic methanotrophic bacteria of cold ecosystems.,
  FEMS microbiology ecology, 53(1), 15–26, doi:10.1016/j.femsec.2005.02.010, 2005.
- 36 Vincent, W. F., Laurion, I., Pienitz, R. and Anthony, K. M. W.: Climate Impacts on Arctic
- 37 Lake Ecosystems, in Climatic Change and Global Warming of Inland Waters: Impact and

- 1 Mitigation for Ecosystems and Societies, edited by C. R. Goldman, M. Kumagai, and R. D.
- 2 Robarts, John Wiley & Sons, Ltd., 2013.
- 3 Walter Anthony, K. M. and Anthony, P.: Constraining spatial variability of methane
- ebullition seeps in thermokarst lakes using point process models, Journal of Geophysical
- 5 Research: Biogeosciences, 118(July), 1015–1034, doi:10.1002/jgrg.20087, 2013.
- 6 Walter Anthony, K. M., Zimov, S. a, Grosse, G., Jones, M. C., Anthony, P. M., Chapin, F. S.,
- 7 Finlay, J. C., Mack, M. C., Davydov, S., Frenzel, P. and Frolking, S.: A shift of thermokarst
- 8 lakes from carbon sources to sinks during the Holocene epoch., Nature, 511(7510), 452–6,
- 9 doi:10.1038/nature13560, 2014.
- 10 Walter, K. M., Zimov, S. A., Chanton, J. P., D., V. and Chapin III, F. S.: Methane bubbling
- 11 from Siberian thaw lakes as a positive feedback to climate warming., Nature, 443(7107), 71–5,
- 12 doi:10.1038/nature05040, 2006.
- 13 Walter, K. M., Smith, L. C. and Chapin, F. S.: Methane bubbling from northern lakes: present
- 14 and future contributions to the global methane budget., Philosophical transactions. Series A,
- 15 Mathematical, physical, and engineering sciences, 365(1856), 1657–76,
- 16 doi:10.1098/rsta.2007.2036, 2007a.
- 17 Walter, K. M., Edwards, M. E., Grosse, G., Zimov, S. A. and Chapin, F. S.: Thermokarst
- lakes as a source of atmospheric CH4 during the last deglaciation., Science (New York, N.Y.),
  318(5850), 633–6, doi:10.1126/science.1142924, 2007b.
- 20 Walter, K. M., Chanton, J. P., Chapin, F. S., Schuur, E. a. G. and Zimov, S. a.: Methane
- 21 production and bubble emissions from arctic lakes: Isotopic implications for source pathways
- and ages, Journal of Geophysical Research, 113, doi:10.1029/2007JG000569, 2008.
- 23 Whiticar, M. J.: Carbon and hydrogen isotope systematics of bacterial formation and
- 24 oxidation of methane, Chemical Geology, 161(1-3), 291–314, doi:10.1016/S0009-25 25/11(99)00092-3 1999
- 25 2541(99)00092-3, 1999.
- 26 Wik, M., Crill, P. M., Varner, R. K. and Bastviken, D.: Multiyear measurements of ebullitive
- 27 methane flux from three subarctic lakes, Journal of Geophysical Research: Biogeosciences,
- 28 118(April), n/a–n/a, doi:10.1002/jgrg.20103, 2013.
- 29 Van Winden, J. F., Kip, N., Reichart, G.-J., Jetten, M. S. M., Camp, H. J. M. O. Den and
- 30 Damsté, J. S. S.: Lipids of symbiotic methane-oxidizing bacteria in peat moss studied using
- 31 stable carbon isotopic labelling, Organic Geochemistry, 41(9), 1040–1044,
- 32 doi:10.1016/j.orggeochem.2010.04.015, 2010.
- 33 Wooller, M. J., Pohlman, J. W., Gaglioti, B. V., Langdon, P., Jones, M., Walter Anthony, K.
- 34 M., Becker, K. W., Hinrichs, K.-U. and Elvert, M.: Reconstruction of past methane
- 35 availability in an Arctic Alaska wetland indicates climate influenced methane release during
- 36 the past ~12,000 years, Journal of Paleolimnology, 48(1), 27–42, doi:10.1007/s10933-012-
- **37 9591-8**, 2012.

- 1 Zheng, Y., Singarayer, J. S., Cheng, P., Yu, X., Liu, Z., Valdes, P. J. and Pancost, R. D.:
- Holocene variations in peatland methane cycling associated with the Asian summer monsoon
- 2 3 system., Nature communications, 5, 4631, doi:10.1038/ncomms5631, 2014.

	<u>Sample</u> <u>No</u>	<u>Sample size</u> (dry g)
<u>Smith</u>		
	<u>1</u>	<u>0.2596</u>
	<u>2</u>	<u>0.2206</u>
	<u>3</u>	<u>0.3584</u>
	<u>4</u>	<u>0.1486</u>
	<u>5</u>	<u>0.1942</u>
	<u>6</u>	0.5654
	<u>7</u>	<u>0.3841</u>
	<u>8</u>	<u>0.2024</u>
	<u>9</u>	<u>0.3386</u>
	<u>10</u>	0.2185
Ace		
	<u>a1</u>	<u>1.3427</u>
	<u>a2</u>	<u>0.6812</u>
	<u>a3</u>	<u>0.5758</u>
	<u>a4</u>	<u>0.808</u>

 Table 1. Freeze dried sample weights for samples from Ace L. and Smith L.

- <u>Table 2.</u> Mixing model end member values and  $\delta^{13}$ C values of the primary variables used to 1
- calculate the proportion of MOB at each sample point.  $\delta^{13}C_{bulk}$  is the average bulk sediment 2
- value from each lake,  $\pm$  indicates the standard deviation of the  $\delta^{13}C_{bulk}$ . MOB and 3
- heterotrophic bacteria have been assumed to have maximum levels of lipid biosynthesis 4
- occurring (10 and 4‰ respectively).  $\delta^{13}C_{\text{mob-hopane_dip_min}}$  is the estimated minimum stable isotope value given the  $\delta^{13}C$  value of methane at each lake and the maximum potential 5
- 6
- fractionation of carbon by MOB.  $\delta^{13}C_{\text{mob-hopane_dip_max}}$  is the estimated value of MOB with no 7
- fractionation during assimilation.  $\delta^{13}C_{hetero\_dip\_hopane\_max}$  is the maximum estimated stable 8 isotope value of heterotrophic bacteria if no fractionation is occurring during assimilation and 9
- the bulk sediment is +1.0 standard deviation (S.D.) from the mean at each lake.  $\delta^{13}C_{hetero-}$
- 10 hopane min represents the minimum value for heterotrophic hopanes given maximum possible 11
- fractionation during assimilation and if bulk sediment is -1.0 S.D from the mean. 12

	$\delta^{13}C_{bulk}$		$\delta^{13}C_{mob\_dip\_hopane\_min}$	$\delta^{13}C_{mob\_dip\_hopane\_max}$	$\delta^{13}C_{hetero\_dip\_hopane\_min}$	$\delta^{13}C_{hetero\_dip\_hopajne\_max}$
	(‰)	±	(‰)	(‰)	(‰)	(‰)
Ace	-30.8	2.1	-104.6	-74.6	-36.9	-32.7
Smith	-29.3	0.8	-100.9	-70.9	-34.1	-32.5
13						
14						
15						
16						
17						
18						
19						
20						
21						
22						
23						
24						
25						
26						
27						

Table  $\frac{23}{5}\delta^{13}$ C values of diploptene at the study sites. The values are an average of three

replicates. The standard deviation of these replicates and of each zone and across all samples

		Sample	$\delta^{13}C_{dip}$	<u>Sample</u>		<u>Standard</u>
		Number	(‰)	replicate		<b>Deviation</b>
				<u>s</u> tandard		<u>(SD)</u>
				Deviation		
				(SD)		
Ace						
	TK	a1	-50.1	1.5		
	zone					
		a2	-58.5	2.0		
		a3	-53.1	0.4		
		a4	-68.2	0.1	TK zone	<u>8.0</u>
Smith						
	Centre	1	-51.4	2.7		
		2	-48.3	0.0		
		3	-56.8	N/A		
		4	-49.2	1.0		
		5	-46.9	1.8		
		6	-48.0	0.1	<u>Centre</u>	<u>3.6</u>
	TK	7	-38.8	0.3		
	zone					
		8	-40.9	0.2		
		9	-42.9	0.1		
		10	N/A	N/A	TK zone	<u>2.0</u>
					<u>Total</u>	<u>7.8</u>

\_

1Table 34. Estimated contribution of MOB to the diploptene signal. Calculations assume2fractionation due to biosynthesis of 10% for MOB and 4% for heterotrophic bacteria.  $f_{mob_min}$ 3was calculated assuming the highest fractionation for both MOB and heterotrophs (30 and44% respectively).  $f_{mob_max}$  assumes no fractionation during assimilation.  $f_{mob_average}$  was5calculated using average  $\delta^{13}$ C values for  $\delta^{13}C_{mob-hopane}$  and  $\delta^{13}C_{hetero-hopane.}$ 

		Sample Number	$f_{mob\_min}$	$f_{mob\_max}$	$f_{mob\_average}$
Ace					
	TK zone	a1	0.19	0.42	0.28
		a2	0.32	0.62	0.43
		a3	0.24	0.49	0.33
		a4	0.46	0.85	0.61
Smith					
	Centre	1	0.26	0.49	0.34
		2	0.21	0.41	0.28
		3	0.34	0.63	0.45
		4	0.23	0.44	0.30
		5	0.19	0.37	0.26
		6	0.21	0.40	0.28
	TK zone	7	0.07	0.17	0.11
		8	0.10	0.22	0.14
		9	0.13	0.27	0.18

Figure 1. Illustration of methane production zones and emission pathways in lakes alongside 1 2 thermokarst-specific zones and pathways. 1) Surface sediment ebullition zone. Methane that is produced in the anoxic surface sediments is released via ebullition, usually near the margins 3 (Bastviken et al., 2004). (2) Surface sediment diffusion zone. Methane is produced in the 4 5 anoxic surface sediments and diffuses in the sediments above and into the water column. Some of this methane will reach the water surface-air interface but a large amount is likely to 6 7 be oxidised by MOB (Kankaala et al., 2006). This process is common in many lakes also. (3) 8 Talik zone. Methane is produced in the deeper talik sediments underneath the lake and is 9 released via ebullition seeps (Walter et al. 2008). Often this is a higher flux and is more 10 constant than surface sediment ebullition. This production zone and pathway is a thermokarst-11 specific process. (4) Slump zone. Methane production in the surface sediments is increased 12 due to the introduction of large volumes of slumped sediments. This methane is also released 13 via ebullition seeps. Often, the flux from these ebullition seeps is higher than surface sediment ebullition but not as high as talik ebullition. This process might occur in any lakes that have 14 dynamic margins and high erosion rates; however, it is likely that this process is most 15 common in thermokarst lakes due to the melting of permafrost, so it is termed thermokarst-16 specific. Red ?'s indicate where methane diffusion from the sediments has not been studied 17 18 in detail.

19



21

1 Figure 2. Locations of the study lakes in Alaska and the sample points within each lake. The

- 2 red (Ace L.) and blue (Smith L.) bars indicate the flux values as measured at an individual
- 3 ebullition seep within a given area of the lake.
- 4



- 1 Figure 3. Bubble counts at Ace Lake. A tally of all bubbles that broke at the water surface
- 2 within a 2 m radius of the sample location. The thermokarst zone is found at the top of the
- 3 lake and has the highest bubble counts.
- 4



6



- 1 Figure 4. Diploptene  $\delta^{13}$ C values at Smith Lake and Ace Lake. In general the most depleted
- 2 values are found in Ace and in the centre of Smith. The Thermokarst zone at Smith L. has the
- 3 least depleted values for the whole dataset

