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**Anaerobic methane
oxidation in Black
Sea sediments**

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Regulation of anaerobic methane oxidation in sediments of the Black Sea

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Abstract

Anaerobic oxidation of methane (AOM) and sulfate reduction (SRR) were investigated in sediments of the western Black Sea, where methane transport is controlled by diffusion. To understand the regulation and dynamics of methane production and oxidation in the Black Sea, rates of methanogenesis, AOM, and SRR were determined using radiotracers in combination with pore water chemistry and stable isotopes. On the shelf of the Danube paleo-delta and the Dnjepr Canyon, AOM did not consume methane effectively and upwards diffusing methane created an extended sulfate-methane transition zone (SMTZ) that spread over more than 2.5 m and was located in formerly limnic sediment. Measurable AOM rates occurred mainly in the lower part of the SMTZ, sometimes even at depths where sulfate seemed to be unavailable. The inefficiency of methane oxidation appears to be linked to the limnic history of the sediment, since in all cores methane was completely oxidized at the limnic-marine transition. The upward tailing of methane was less pronounced in a core from the deep sea in the area of the Dnjepr Canyon, the only station with a SMTZ close to the marine deposits. Sulfate reduction rates were mostly extremely low, and in the SMTZ were even lower than AOM rates. Rates of bicarbonate-based methanogenesis were below detection limit in two of the cores, but $\delta^{13}\text{C}$ values of methane indicate a biogenic origin. The most depleted $\delta^{13}\text{C}$ -signal was found in the SMTZ of the core from the deep sea, most likely as a result of carbon recycling between AOM and methanogenesis.

1 Introduction

The process of anaerobic oxidation of methane (AOM) is widespread in continental margin sediments and occurs in a variety of different environments, including diffusive sediments (Iversen and Jørgensen, 1985), sediments containing shallow gas accumulations (Niemann et al., 2005; Treude et al., 2005b), gas-hydrate bearing sediments (Treude et al., 2003; Joye et al., 2004; Orcutt et al., 2004), and mud volcanoes (Nie-

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mann et al., 2006), In the western part of the Black Sea, which is dominated by an extensive shelf in the northwest and by river deltas of the Danube and Dnjepr rivers, the sediment contains large amounts of methane, and numerous active methane seeps (Popescu et al., 2001). These seeps are mainly located above the gas hydrate stability zone, at the transition from the continental shelf to the upper slope, and are common in the area of the Danube canyon and the Dnjepr paleo-delta (Naudts et al., 2006). A unique feature of methane seeps in the anoxic part of the Black Sea are carbonate chimneys that grow from the sediment up into the water column and which are overgrown by thick bacterial mats mediating AOM (Michaelis et al., 2002; Treude et al., 2005a). The abundance of methane in the Black Sea sediments results from the favorable conditions for preservation of organic matter because of slow degradation under anoxic conditions (Brumsack, 1989; Arthur and Dean, 1998). Since the concentrations of electron acceptors such as oxygen, nitrate, and metal-ions like Fe(III) and Mn(IV) are typically low or absent in Black Sea sediments below the anoxic water column (Thamdrup et al., 2000), sulfate reduction and bicarbonate-based methanogenesis are the dominant terminal anaerobic pathways of organic matter degradation (Jørgensen et al., 2004). The majority of the methane is therefore expected to be biogenic, as confirmed by stable isotope analyses (Amouroux et al., 2002; Michaelis et al., 2002).

The stratified sediments of the Black Sea consist of limnic clays deposited before 7500 yr BP, which are overlain by a microlaminated organic-rich sapropel that marks the transition to marine coccolith ooze deposited after the intrusion of sea water from the Mediterranean (Ross et al., 1970; Calvert et al., 1991). Where free gas does not reach the surface but is dissolved in the pore water at depth, methane diffuses upwards into the sulfate zone, where it is oxidized in combination with sulfate reduction. In most marine shelf sediments the overlapping methane and sulfate profiles form a distinct sulfate-methane transition zone (SMTZ) (Devol and Anderson, 1984; Niewöhner et al., 1998) in which AOM and sulfate reduction occur. Earlier studies of the SMTZ in the Black Sea revealed that the methane profile is tailing up towards the sediment surface (Reeburgh et al., 1991; Jørgensen et al., 2001), thus creating an extended SMTZ. This

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broad zone of methane and sulfate coexistence indicates that microbial turnover of methane is for some reason particularly sluggish (Jørgensen et al., 2001) and therefore provides an interesting opportunity to study the regulation of SRR and AOM rates and the controls on AOM distribution.

5 In Black Sea sediments AOM provides the energy substrate for 7–18% of the total sulfate reduction and is the main source of H₂S formation at depth (Jørgensen et al., 2004). The occurrence of a deep sink for H₂S below the SMTZ, which is supposed to be related to a high content of reactive iron in the formerly limnic sediments, might play a role in regulating SRR and AOM activity in Black Sea sediments.

10 Among the questions addressed in this study were, a) how is methane oxidation influenced by the unique environmental conditions in the Black Sea and b) what are the reasons for the sluggish AOM rates which lead to methane tailing in these sediments. It is further investigated how SRR and AOM rates are distributed in the extended SMTZ, and what role methanogenesis plays in this environment.

15 2 Methods

2.1 Sample collection

Sediment cores were sampled with a gravity corer (GC) during cruise 317/3 with RV Poseidon in October 2004 at three stations in the western Black Sea (Fig. 1, Table 1). The 5-m long cores were cut into 1-m sections on deck. At the top of each 1-m section
20 a methane sample was immediately taken to check for the potential loss of dissolved methane before detailed sub-sampling. All core sections were sub-sampled for chemical analyses and for microbial process rate measurements at the latest 24 h after retrieval of the core.

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2.2 Chemical analyses

2.2.1 Methane concentrations

Methane concentrations were measured in 3 cm³ sediment samples collected with 5 ml cut-off syringes and sealed in serum vials that contained 6 ml NaOH (2.5 % w/v). The samples were shaken and stored to achieve equilibrium between the slurry and the headspace. The methane concentration in the headspace was determined by gas chromatography (5890A, Hewlett Packard) using a packed stainless steel Porapak-Q column (6 ft., 0.125 in., 80/100 mesh, Agilent Technology) and a flame ionization detector. The column temperature was 40°C, and helium was used as a carrier gas at a flow rate of 30 ml min⁻¹. The methane profile from sub-samples was compared to the samples taken immediately from the 1-m sections and showed that the loss of methane during storage was not significant in the SMTZ.

2.2.2 Sulfate concentrations

Sulfate concentrations were measured in 5 cm depth intervals from pore water collected by pressure-filtration (Millipore 0.45 μm) with nitrogen (Iversen and Jørgensen, 1985; Fossing et al. 1990). 1 ml of the pore water was directly fixed in 0.25 ml ZnCl₂ (2% w/v) to trap dissolved sulfide and prevent oxidation. The concentration of sulfate in the pore water was analyzed by non-suppressed anion exchange chromatography (Waters 510 HPLC Pump; Waters IC-Pak 50×4.6 mm anion exchange column; Waters 430 Conductivity detector). Isophtalic acid (1 mM, pH 4.6) in methanol (10% v/v) was used as eluant. The detection limit for sulfate was 0.2 mM.

2.2.3 Hydrogen sulfide

Hydrogen sulfide was determined from the pore water samples fixed with ZnCl₂ (2% w/v) by the diamine complexation method described by Cline (1969). The sulfide concentration was measured by spectrophotometry at a wavelength of 670 nm.

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2.2.4 Dissolved inorganic carbon (DIC)

The concentration of DIC in pore water was measured on samples sealed in headspace-free glass vials poisoned with HgCl_2 (0.25 mM), and stored at 10°C. The samples were analyzed by flow injection (Hall and Aller, 1992) using HCl (30 mM) and NaOH (10 mM) as eluants and a conductivity detector (VWR scientific, model 1054).

2.2.5 Density and porosity

Samples for density and porosity were taken in 10 ml cut-off syringes and 8 cm³ of sediment was weighed before and after drying at 60°C. The density was calculated as the wet weight per cm³ and the porosity was determined from the loss of water per cm³ after the sediment was completely dried.

2.2.6 Acetate concentrations

Concentrations of acetate were measured from undiluted squeezed pore water that was stored frozen and thawed immediately before measurement. Samples were analyzed on a Dionex[®] ICS-2000 Ion Chromatography System equipped with a Dionex[®] AS50 autosampler at 4°C. Determinations of VFA species were carried out using a Dionex[®] Anion Self-Regenerating Suppressor (ASRS[®]-ULTRA II 4-mm) unit in combination with a conductivity detector.

2.2.7 Dissolved iron

The concentration of Fe^{2+} in pore water was analyzed with Ferrozine (1 g L⁻¹ in 50 mM HEPES buffer, pH 7) according to Stookey (1970) by spectrophotometry at 562 nm.

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2.2.8 Total reactive iron

The total reactive iron in the sediment was extracted with dithionite-citrate-acetic-acid (Canfield, 1989). The dithionate extracts were analyzed for total iron (Fe^{2+} and Fe^{3+}) with Ferrozine plus 1% (w/v) hydroxylamine hydrochloride.

5 2.2.9 Diffusional fluxes

Fluxes of methane, sulfate and sulfide were calculated from the slopes of the concentration profile into the SMTZ and the respective diffusion coefficients according to Fick's first law of diffusion:

$$J = \phi \cdot D_s \cdot \frac{dC}{dz}$$

10 where J is the diffusive flux [$\text{mmol m}^{-2} \text{d}^{-1}$], ϕ is the porosity, D_s is the diffusion coefficient in the sediment [$\text{cm}^2 \text{d}^{-1}$], and dC/dz is the concentration gradient [$\mu\text{mol cm}^{-4}$]. Diffusion coefficients at the respective in situ temperature and salinity were calculated from Schulz and Zabel (2000), corrected for porosity of the sediment with $n=2$ for $\phi < 0.7$ and $n=3$ for $\phi \leq 0.7$ according to Iversen and Jørgensen (1993):

$$15 D_s = \frac{D}{1 + n(1 - \phi)}$$

2.2.10 Stable isotopes

For stable isotope analyses gases were stripped from pore water according to McAullife (1971). The gas was stored as a headspace in inverted crimp top Wheaton[®] vials by displacement of a preservative solution that consisted of KCl 10% (w/v) in de-ionized water adjusted to pH 1 using HCl. Analysis of $^{13}\text{C}/^{12}\text{C}$ ratios was conducted by isotope ratio monitoring gas chromatography mass spectrometry (GC-C-IRMS) using a Varian 3400[®] GC coupled to a Thermoelectron XP[®] mass spectrometer via a Gas Bench[®]

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interface. Methane was separated on a PLOT Q capillary column (0.32 mm×30 m) and combusted to CO₂ at 1000°C in a ceramic reactor containing Cu and Pt wires. A high purity blend of 1% O₂ in helium was fed into the reactor at ~0.1 ml min⁻¹ to ensure quantitative conversion of CH₄ to CO₂. The H₂O produced was removed using a Nafion[®] membrane. Accuracy and precision of δ¹³C-CH₄ analysis by this method were both better than ±0.2‰ based upon replicate analysis of a BOC alpha gravimetric CH₄ standard. Stable isotope ratios are reported in the standard δ¹³C notation relative to Vienna Pee Dee Belemnite (VPDB).

2.3 Microbial process rates

2.3.1 Anaerobic oxidation of methane (AOM)

Sediment for AOM measurements was either a) taken for each 5 cm depth interval with glass tubes in three parallel 5 cm³ samples and sealed headspace-free with butyl stoppers (P771GC), or b) sampled in acrylic core liners with injection holes at 2 cm depth intervals (P806GC and P824GC). ¹⁴C-methane tracer (1.35 KBq) was injected into each sample and incubated for 20–24 h at the in situ temperature of the SMTZ. After incubation the sediment was transferred to glass vials containing 25 ml NaOH (2.5% w/v) and suspended completely to stop microbial activity. To determine the detection limit of the rate measurement, five control samples per core were stopped immediately after tracer injection. To calculate AOM rates, the methane concentration in each sample was analyzed by gas chromatography and the pool of ¹⁴C-methane was measured by combustion of the headspace and scintillation counting. The produced ¹⁴C-CO₂ in the sediment was extracted through acidic diffusion, trapped in scintillation vials with phenylethylamine, and the radioactivity was counted (Treude et al., 2003).

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2.3.2 Sulfate reduction rate (SRR)

The samples for SRR measurements were taken as described for AOM, injected with ^{35}S -sulfate tracer (500 kBq), and incubated for 20–24 h at in situ temperature. The microbial activity was stopped by mixing with 20 ml ZnAc (20% w/v). Samples were analyzed by the cold distillation method described by Kallmeyer et al. (2004), where the total radiolabelled reduced inorganic sulfur (TRIS) is determined in relation to the total radioactive sulfate pool (all units in decays per minute) that remained in the sample. SRR in $\text{nmol cm}^{-3} \text{d}^{-1}$ was calculated using the following equation (Jørgensen, 1978):

$$\text{SRR} = [\text{SO}_4^{2-}] \cdot \frac{{}^{35}\text{S-TRIS}}{{}^{35}\text{S-TRIS} + {}^{35}\text{S-SO}_4^{2-}} \text{d}^{-1} \cdot 1.06$$

where $[\text{SO}_4^{2-}]$ is the sulfate concentration in mM and 1.06 is an estimated fractionation factor between ^{35}S and the normal isotope ^{32}S . Because of high variability of the remaining sulfate tracer that was recovered in the samples for core P771GC the SRR was calculated using the injected tracer activity minus produced $^{35}\text{S-TRIS}$ as the $^{35}\text{S-SO}_4^{2-}$ pool.

Measurements of AOM and SRR were considered to be above the detection limit when the produced $^{14}\text{C-CO}_2$ or $^{35}\text{S-TRIS}$ exceeded the mean production in the zero time controls plus 2 times their standard deviation (Ferdelman et al., 1999).

2.3.3 Bicarbonate methanogenesis

Bicarbonate methanogenesis rates were measured by injecting ^{14}C -bicarbonate (activity 38 kBq) at 2-cm depth intervals into subcores taken in acrylic core liners. After incubation for 6 to 24 h at in situ temperature the incubations were terminated by transferring 2 cm slices of the subcores to glass vials (30 ml) containing 7 ml of NaOH (1 M). In the laboratory, the vial headspace was flushed (carrier gas 95% N_2 : 5% O_2 at 70 ml/min for 20 min) through a CO_2 -trap (Supelco, UK) and over copper oxide at 800°C

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in a furnace (Carbolite, UK) to oxidize any produced $^{14}\text{CH}_4$ to $^{14}\text{CO}_2$. The $^{14}\text{CO}_2$ was trapped in Optiphase HiSafe-3 and β -phenethylamine (93:7), and measured in a scintillation counter (Perkin Elmer, UK). Activity rates were calculated from the label turnover times applied to the relevant cold-pool concentrations of DIC.

5 2.3.4 Acetate methanogenesis

To determine acetate methanogenesis rates the samples were obtained as for bicarbonate methanogenesis, but were injected with ^{14}C -acetate tracer (activity 20 kBq) at 2-cm depth intervals and also incubated for 6 to 24 h at in situ temperature. Samples were processed the same way as described above for bicarbonate methanogenesis.
10 Rates were calculated from the label turnover times and the relevant cold-pool concentrations of acetate.

2.4 Microbiology

2.4.1 Acridine orange direct counts (AODC)

AODC was used to determine the total number of microorganisms. A 1 cm^3 sample of sediment was preserved in a furnaced serum vial containing 9 ml formaldehyde (2% v/v in artificial seawater, filter sterilized $0.2\ \mu\text{m}$). Three replicate sub-samples (5–25 μl) were stained for 3 min with 50 μl acridine orange (0.1% w/v) in 10 ml formaldehyde (2% v/v in artificial seawater, filter sterilized $0.1\ \mu\text{m}$) and vacuum filtered through a black polycarbonate membrane filter ($0.22\ \mu\text{m}$). Paraffin oil mounted membrane filters
20 were viewed under incident UV illumination with a Zeiss Axioskop epifluorescence microscope at X1000. Both unattached cells and cells attached to particles were counted and the number of attached cells was doubled to account for cells hidden from view (Goulder, 1977). Dividing and divided cells were counted separately to provide an index of the growth potential of the populations.

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3 Results

3.1 Pore water concentrations

Depth profiles of methane and sulfate concentrations as substrates of AOM mediated by sulfate reduction and the products sulfide and DIC from the three gravity cores are presented in Fig. 2. In core P771GC the marine deposit of laminated coccolith ooze at the top was followed at depth by a sapropel layer that changed into grey limnic clay at 70 cm. Below 160 cm depth the clay was marbled with darker sediment and contained black grains of greigite and dark laminations at 200 cm depth (Table 1).

The core was sampled in the Danube Canyon area at a water depth of 680 m between stations 6 and 7 of the transect described by Jørgensen et al. (2004). Similar to the findings of those authors the methane profile of P771GC also showed an extended tailing up towards the sediment surface. The highest methane concentrations of ~ 1.2 mM were measured at 360 cm depth below which methane had been outgassing upon core recovery, as indicated by cracks in the sediment observed below 400 cm depth. The upwards diffusive flux of methane was $19 \mu\text{mol m}^{-2} \text{d}^{-1}$ in the lower part of the SMTZ where methane was incompletely oxidized. A flux of $4 \mu\text{mol m}^{-2} \text{d}^{-1}$ remained between 70 and 232 cm. The methane concentration only approached zero at ~ 70 cm depth, at the boundary between the marine and limnic sediments (Table 1).

Sulfate concentrations decreased nearly linearly from 25 mM at the top of the core to a depth of ~ 340 cm, with a sulfate flux into the SMTZ of $149 \mu\text{mol m}^{-2} \text{d}^{-1}$. Yet, sulfate was not entirely depleted below this depth but a constant concentration of ~ 1 mM apparently remained below 350 cm. Sulfide concentrations in core P771GC were low for a methane rich environment and did not exceed 0.2 mM. The peak of H_2S abundance at ~ 100 cm depth was located at the top of the limnic sediment in the upper part of the SMTZ, at a notably shallow depth in comparison to the methane and sulfate profiles. H_2S diffusion occurred both upwards and downwards in the sediment, with similar fluxes towards the sediment surface ($0.12 \text{ nmol m}^{-2} \text{d}^{-1}$) and towards deeper sediment ($0.13 \text{ nmol m}^{-2} \text{d}^{-1}$). DIC concentrations were also rather low, increasing steadily from

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the top of the core to maximum values of ~ 8 mM at 270 cm depth, coinciding with the horizon where H_2S was depleted.

Core P806GC was collected on the shelf edge of the palaeo Dnjepr Delta, on the flank of a ridge that was lined with methane seeps at the top (Naudts et al., 2006).

5 The sediment contained a dense shell layer beneath the sapropel at ~ 10 cm depth and a second shell layer at 80 to 140 cm depth (Table 1). Below the second shell layer the sediment consisted of silty fine-grained sand that contained gas cracks from 220 cm downwards, consistent with the methane profile which indicated outgassing below ~ 200 cm depth. The methane flux into the SMTZ was $98 \mu\text{mol m}^{-2} \text{d}^{-1}$, which is
10 much higher than in core P771GC. The methane and sulfate profiles exhibited a similar pattern of an extended SMTZ with methane tailing up towards the sediment surface.

The sulfate profile was linear from the top of the core down to a depth of 180 cm, with a flux of $145 \mu\text{mol m}^{-2} \text{d}^{-1}$ into the SMTZ, 1.5 times higher than the methane flux. The residual pool of sulfate below the SMTZ was smaller than in core P771GC,
15 as sulfate concentrations below 200 cm were approximately at the detection limit of 0.2 mM. The H_2S concentration at the top of the core (0.2 mM) was similar to that of core P771GC, but the sulfide fluxes were higher ($1.3 \text{ mmol m}^{-2} \text{d}^{-1}$ towards the top and $1.8 \text{ mmol m}^{-2} \text{d}^{-1}$ towards depth) because of the higher H_2S peak of ~ 2 mM that occurred in the lower part of the SMTZ. The linear increase of DIC concentration again
20 reached maximum values in the zone of sulfide depletion at ~ 200 cm depth and, like the H_2S concentrations, they were also much higher than in core 771GC.

Core P824GC sampled in the deeper part of the Black Sea on mid-slope contained a thick sapropel layer with laminations that changed into uniform grey clay at 120 cm depth (Table 1). The methane profile did not show such a pronounced upwards tailing
25 as at the other two sites, and resembled more the typical concave upwards pattern observed in most marine environments, forming a distinct SMTZ at 110 to 170 cm depth. The methane flux into the SMTZ was $126 \mu\text{mol m}^{-2} \text{d}^{-1}$, the highest of the three sites. Sediment cracks indicating methane super-saturation were observed from >280 cm depth although the methane concentration profile would suggest outgassing

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from >220 cm depth.

The pore water sulfate profile was very similar to that measured at P806GC, with a concentration of ~ 15 mM at the top of the core and a sulfate flux into the SMTZ of $301 \mu\text{mol m}^{-2} \text{d}^{-1}$. Similar to P771GC sulfate was not completely depleted below 168 cm depth, but concentrations of <0.6 mM remained and decreased only gradually with depth.

DIC concentrations increased with depth down through the marine deposits at the top of the core attaining a maximum concentration of 14 mM at the bottom of the SMTZ. In the limnic part of the sediment DIC decreased slightly to ~ 12 mM.

The pore water concentration of H_2S was relatively constant at ~ 1 mM in the upper 20–100 cm of the core, decreasing only at the very top of the core. In relation to methane and sulphate, the H_2S maximum of 3 mM at 200 cm depth was situated deeper than in the other two cores, even below the SMTZ, and was depleted at a depth of 330 cm.

The concentrations of iron in the pore water and the solid phase of core P824GC are presented in Fig. 3. Dissolved Fe^{2+} showed a steep increase below a depth of ~ 430 cm, forming a distinct iron diffusion front at this depth. In the solid phase, reactive iron concentrations of up to $45 \mu\text{mol cm}^{-3}$ were measured in the top 100 cm of the core, with a local minimum at the transition from the marine to the limnic sediment. Below the sapropel layer, reactive iron concentrations were $<40 \mu\text{mol cm}^{-3}$, with lowest values occurring between 200 and 300 cm depth. Reactive iron increased below 300 cm, with a steep increase below 390 cm reaching up to $300 \mu\text{mol cm}^{-3}$ at 430 cm depth. Consistent with this distribution of reactive iron, black bands of FeS occurred in the sediment between depths of 390 cm and 440 cm.

3.2 Microbial process rates and stable isotopes

Rates of AOM, SRR and methanogenesis measured at station P771GC are shown in Fig. 4. Rates of AOM were low in the top 0–175 cm, below which rates increased to reach a maximum of $5.6 \text{ nmol cm}^{-3} \text{d}^{-1}$ at 250 cm depth. A distinct AOM peak was

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not observed. Instead, rates of ~ 0.5 to $4 \text{ nmol cm}^{-3} \text{ d}^{-1}$ were measured throughout the lower part of the core. As discussed below, however, it is not clear whether these values represent actual AOM activities.

Two peaks of sulfate reduction were detected, one occurring at 200 cm depth ($3\text{--}7 \text{ nmol cm}^{-3} \text{ d}^{-1}$) and the other at 350 cm depth ($\sim 9 \text{ nmol cm}^{-3} \text{ d}^{-1}$) near the SO_4^{2-} depletion boundary. Low SRR were also detected between these peaks, yet the pattern of SRR did not reflect the profile of AOM. The lack of high rates at the top of the core (cf. Weber et al., 2001) is probably due to the loss of the more active surface sediment during gravity coring.

Bicarbonate-based methanogenesis was analyzed throughout the core, but rates were not detectable. Instead, acetate-based methanogenesis was found in the sulfate zone in the upper 100 cm of the core, but these rates were extremely low, with a maximum of $4.3 \times 10^{-4} \text{ nmol cm}^{-3} \text{ d}^{-1}$. At the depth where AOM occurred, acetate-based methanogenesis rates were below the detection limit of the radiotracer method. The occurrence of acetate-based methanogenesis at 50–100 cm was accompanied by high acetate concentrations of up to $30 \mu\text{M}$, whereas other VFAs, such as formate or lactate, had concentrations below $10 \mu\text{M}$ throughout the core.

The methane isotopic signal was most negative below the SMTZ, consistent with a preferential depletion of $^{12}\text{C-CH}_4$ during AOM, thus changing the $\delta^{13}\text{C}$ of the methane from -88‰ in the deep limnic sediment to -78‰ within the SMTZ. The $\delta^{13}\text{C-CH}_4$ values above the inflection point at 210 cm suggest a change in either the rate or the process by which methane is being consumed above this depth. The $\delta^{13}\text{C-CO}_2$ values, controlled by AOM as well as by SRR, did not reflect this trend and varied around values of -30‰ .

The total number of cells at the top of the core was 2.5×10^8 per cm^3 and decreased with depth, most steeply in the upper 70 cm.

The distribution of microbial process rates in core P806GC showed distinct peaks of AOM and SRR at similar depth (Fig. 5). AOM activity occurred in a ~ 70 cm wide zone with a maximum rate of $\sim 2.4 \text{ nmol cm}^{-3} \text{ d}^{-1}$ at 218 cm. The peak of SRR at the bottom

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of the SMTZ ($\sim 0.8 \text{ nmol cm}^{-3} \text{ d}^{-1}$) was much lower than the AOM rates at the same depth.

Bicarbonate-based methanogenesis rates were highest just above the AOM zone with a maximum of $0.06 \text{ nmol cm}^{-3} \text{ d}^{-1}$, and a rate of $0.015 \text{ nmol cm}^{-3} \text{ d}^{-1}$ at $\sim 200 \text{ cm}$ depth, in the zone of maximum rates of AOM and sulfate reduction. Similar to P771GC, very low rates of acetate-based methanogenesis were detected, mostly in the sulphate zone in the top 100 cm of the core. Additional acetate-based methanogenesis activity occurred together with bicarbonate-based methanogenesis at 140 cm to 210 cm depth, with maximum methane production above the AOM zone. Rates of microbial processes were not determined below 230 cm depth.

The concentrations of acetate, formate and lactate were typically below $10 \mu\text{M}$ in most of core P806GC, with the exception of higher lactate values (up to $27 \mu\text{M}$) at 118 cm depth and acetate abundance of $\sim 10 \mu\text{M}$ at 240 cm depth below the zone of AOM.

The most negative $\delta^{13}\text{C-CH}_4$ values (-73.6 to -72.8‰) were measured between 184 and 214 cm depth. A minor $\delta^{13}\text{C}$ -enrichment (up to -66.9‰) occurred at shallower depth, most likely as a consequence of preferential loss of $^{12}\text{C-CH}_4$ during AOM. The significant amount of methane remaining above the SMTZ is consistent with the relatively small positive shift in $\delta^{13}\text{C-CH}_4$ values. In addition, $\delta^{13}\text{C-CH}_4$ values were not available for deeper sediments where higher methane concentrations can be found and, consequently, the contrast between $\delta^{13}\text{C}$ values and methane above the SMTZ and in deeper sediments may be greater than these data suggest. Notably, $\delta^{13}\text{C}$ values of DIC were not unusually negative indicating a lack of contribution of AOM-derived CO_2 to the pore water pool.

The total number of microbial cells at the surface of the core was comparable to P771GC and decreased mainly in the top sediment. Below 36 cm depth, cell numbers remained almost constant at $\sim 3 \times 10^7 \text{ cells cm}^{-3}$. A slight decrease in cell numbers was observed only in the zone of AOM and sulfate reduction.

At station P824GC (Fig. 6), the profiles of methane and sulfate formed a distinct

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SMTZ from 130 to 170 cm depth, a zone which also showed increased microbial activity. Maximum AOM rates ($4.5 \text{ nmol cm}^{-3} \text{ d}^{-1}$) occurred at 150 cm, and coincided with increased sulfate reduction rates ($1.2 \text{ nmol cm}^{-3} \text{ d}^{-1}$ at 162 cm) at about the same depth. The AOM rates were measured down to 290 cm where a maximum rate of $\sim 13 \text{ nmol cm}^{-3} \text{ d}^{-1}$ was found. However, similar to core P771GC, it is not clear whether these rates reflect actual AOM activity.

Heterotrophic sulfate reduction in the top of the core exhibited low rates ($<1 \text{ nmol cm}^{-3} \text{ d}^{-1}$) with a minimum at 150–200 cm. In addition to the methane-dependent SRR peak in the SMTZ, a second increase in $^{35}\text{S-SO}_4^{2-}$ reduction was detected at 280 cm depth, but since sulfate might not be available at this depth it is not certain whether this turnover of radiotracer represents actual sulfate reduction.

Bicarbonate-based methanogenesis rates for core P824GC are not available, but acetate-based methanogenesis rates were very low (maximum $6.7 \times 10^{-5} \text{ nmol cm}^{-3} \text{ d}^{-1}$). In contrast to cores P771GC and P806GC, detectable rates did not occur in the upper part of the cores but instead between 130 cm and 200 cm depth. VFAs were measured throughout the core at concentrations $<20 \mu\text{M}$.

Below the SMTZ, methane had $\delta^{13}\text{C}$ values of -85 to -81‰ . At the SMTZ there was a sudden shift to highly negative $\delta^{13}\text{C}$ values reaching -97‰ at 160 cm depth. This extreme ^{13}C -depletion at low concentrations of methane in the SMTZ occurred typically methane is $\delta^{13}\text{C}$ -enriched as a result of AOM. The highly negative $\delta^{13}\text{C-CH}_4$ values coincided partly with a negative shift in $\delta^{13}\text{C-DIC}$ values just above the maximum in DIC concentration.

The total cell counts in core P824GC did not show the steep decrease of cell numbers at the top of the core as it was observed at the other two sites. The cell numbers decreased more gradually, exhibiting slight maxima at 165 cm and 340 cm depth, of which the shallower maximum coincided with enhanced rates of AOM and sulfate reduction.

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4 Discussion

Anaerobic oxidation of methane, sulfate reduction, and methanogenesis were investigated in three diffusive sedimentary environments of the Black Sea, where the SMTZ was situated in the limnic part of the sediment. In cores P771GC and P806GC the SMTZs were fully located in the limnic deposits, while in core P824GC it coincided with the boundary between the Holocene marine sapropel and the underlying lacustrine clay.

4.1 AOM and SRR

The most noticeable feature of the methane profiles in the Black Sea is that AOM does not consume the entire methane flux in a well-defined zone of major AOM activity. This apparent inefficiency results in an extended zone of methane and sulfate co-occurrence with tailing of pore water methane far up towards the sediment surface. This feature of a gradual oxidation of methane had already been observed previously in gravity cores from the western Black Sea (Reeburgh et al., 1991; Jørgensen et al., 2001) and was re-confirmed by the methane profile in core P771GC. It seems to be a common pattern that is also present in methane-rich sediments in other parts of the Black Sea, as suggested by its occurrence in core P806GC from an area of the Dnjev Canyon. Jørgensen et al. (2001) described the methane-tailing as “sluggish” AOM. Microbial process rates in core P806GC determined by radiotracer experiments were indeed low, especially for SRR. However, in diffusion dominated systems at other continental margins AOM rates are mostly similarly low, and peak rates from the Black Sea, in particular in core P771GC with a very pronounced methane tailing, were in the same range ($3\text{--}6 \text{ nmol cm}^{-3} \text{ d}^{-1}$) as maximum AOM rates measured in the North Sea (Niemann et al., 2005) and Skagerrak (Parkes et al., 2007), where methane tailing did not occur.

The combined presence of methane and sulphate in an extended SMTZ should facilitate AOM activity and sulfate reduction also in shallower horizons. Yet, these processes

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were not evenly distributed within the SMTZ but occurred mainly in the lower part of the SMTZ. In contrast to the SMTZ of typical methane-rich marine sediments, where SRR and AOM rates occur in a distinct zone defined by the availability of the substrates, the AOM activity in cores P771GC and P824GC was spread over a broad sediment interval and was also detected at depths where sulfate was restricted to background values or was even not measurable. The occurrence of AOM below the main zone of activity is very unusual and questionable. In P824GC, where samples were collected throughout the core, the turnover of radioactive $^{14}\text{C-CH}_4$ showed that the major activity occurred at 140–160 cm depth (Fig. 7b) and that the rates at greater depths do not result from an increased fraction of methane turnover, but instead from the higher methane concentrations in the deeper part of the core. Detection of $^{14}\text{C-CO}_2$ at ~290 cm depth could be a problem of defining the background radioactivity associated with the tracer incubation method at sites with low AOM activity. However, in P824GC there was also a peak in SRR activity at the same depth as the deep AOM activity, and sulfate-reducing bacteria were reported to be present in this horizon (Leloup et al., 2006). It therefore remains unclear whether AOM rates below 180 cm depth reflect actual activity. The measurements of $^{14}\text{CO}_2$ production in the lower part of core P771GC did show increased fractional $^{14}\text{C-CH}_4$ turnover (Fig. 7a). Whereas AOM rates in the deeper part of this core might be overestimated because of high methane concentrations, it appears likely that AOM is not confined to a distinct zone of activity.

A good correlation between AOM and SRR was observed in the shallowest core P806GC, in agreement with the observation of Jørgensen et al. (2001) who noted that the coupling of both processes is less distinct at greater water depth. The measured rates of AOM and SRR were confined to a narrow zone around 180 cm depth. However, this close coupling between AOM and SRR did not result in the formation of a well defined SMTZ or an effective oxidation barrier for methane diffusing into the AOM zone.

Despite relatively high rates of AOM and SRR in P771GC and the close coupling of these rates in P806GC, only core P824GC did not exhibit a pronounced tailing of methane. It was the only station with a distinct SMTZ, similar to the typical SMTZ found

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in most methane-bearing ocean margin sediments (Niewöhner et al., 1998; Hensen et al., 2003). The sulfate profile and the lower part of the methane profile were very similar in P806GC and P824GC and also the rates did not differ sufficiently to provide an explanation why tailing occurred in one core but not in the other. Thus, methane tailing cannot simply be explained by low rates of AOM in the major AOM zone. Apparently the consumption of methane at concentrations <0.5 mM, at the upper end of the AOM zone around ~ 160 cm depth, was much less efficient in P806GC compared to P824GC and resulted in methane tailing. The reason why AOM is inefficient at <0.5 mM CH_4 at P806GC and P771GC, whereas such concentrations are further depleted by methanotrophic archaea at P824G, is not know.

The key difference between P824GC and the cores P771GC and P806GC is the location of the SMTZ relative to the transition from marine to limnic sediment. In P771GC and P806GC it occurred deep within the limnic sediment, whereas the SMTZ in core P824GC occurred close to the marine-limnic boundary. Moreover, tailing of methane in all cores disappeared exactly at the depth where methane reached the boundary of the marine deposits. It might be characteristic that the location of the SMTZ in the limnic layers of the sediment is the reason for the low rates and the sluggish turnover of methane. Tailing of methane has also been observed at other sites where buried limnic sediments are overlain by marine deposits, e.g. the Baltic Sea, (Iversen and Jørgensen, 1985; Piker et al., 1998; Treude et al., 2005b) or Saanich Inlet (Devol and Anderson, 1984), although the tailing is not as pronounced as in the Black Sea. At those locations AOM takes place in the marine deposits and the sediments of the Black Sea are thus unusual because the entire SMTZ is located in the limnic sediment layers.

The significance of the marine-limnic transition is also evident in the SRR profile of core P824GC, the only core to show sulfate reduction activity based on organic matter degradation in the top 110 cm. The drop in organic matter-related SRR coincided with the shift from sapropel to grey clay, marking the transition from marine to limnic sediment at this depth. The core was sampled in an area of mass wasting where slides from the central divide of the Dnjepr canyon have been identified (Naudts et al., 2006),

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and the thickness of the marine deposits at this site might be an indication that some of the sediment originated from such a submarine slide. The apparent absence of sulfate reduction based on sediment organic matter at the top of cores P771GC and P806GC is probably due to the loss of surface sediment during gravity coring since previous investigations of SRR in the Black Sea have reported high sulfate reduction rates confined to the top ~10 cm of sediment (Albert et al., 1995; Dando and Niven, 1998; Weber et al., 2001). In core P806GC, this surface SRR may also have been low because the sediment was physically eroded on the flanks of the ridge where the core was collected. This suggestion is supported by a very thin accumulation of marine deposit on top of the limnic sediment at this site.

It is noticeable that SRR took place throughout the marine deposit of P824GC, but not, or only with low rates, in the limnic sediment. With the increase in salinity in the Black Sea some 8000 years ago, sulfate became available in the formerly limnic sediments and sulfate is therefore not limiting for SRR today. More important than sulfate availability might be the organic matter content. In the limnic period during the last glaciation, the influx of organic matter from rivers was low and the limnic sediments are today very low in organic material whereas the marine deposits are very rich.

The burial rate and the degradation rate constant of organic matter deposited during the late Pleistocene to Holocene periods control the amount and depth distribution of organic substrate available today for sulfate reduction or methanogenesis. In the absence of bioirrigation under the modern anoxia of the Black Sea, the depth in the sediment at which sulfate becomes depleted appears not to be controlled by surface activity but rather by the methane flux which determines the methane-dependent sulfate reduction in the SMTZ (Jørgensen et al., 2004). The upwards diffusive flux of methane is, in turn, a result of the burial of degradable organic matter in the geological past and, although methanogenesis may take place in low-organic deposits, these have a great depth extension and thereby a significant accumulated methane production.

The phylogenetic diversity of sulfate reducing bacteria in the limnic sediment layers could not be distinguished from the diversity in the marine deposits (Leloup et al.,

2006), and also the relative abundance of SRB was comparable to other diffusive systems with a more efficient AOM zone (Parkes et al., 2007). Although very different kinds of organisms are included in the total cell counts, there is no indication that the low rates are based on a lack of the organisms mediating the processes (Leloup et al., 2006).

4.2 Sulfate dynamics and H₂S

In core P771GC sulfate was apparently not depleted entirely at depth, but the influence of the methane related sulfate reduction created a linear sulfate profile in the upper part of the core. Compared to the SRR and AOM rates modeled by Jørgensen et al. (2001) at the neighbouring stations, the subsurface tracer determined rates were almost two orders of magnitude higher, and probably overestimate the actual activity.

AOM mediated by sulfate reduction usually takes place at the bottom of the sulfate zone, but because sulfate was not depleted at the bottom of the core it might be possible that continuing low SRR facilitate AOM below 330 cm depth. We detected sulfate concentrations of up to 1 mM below 330 cm but it is not yet clear whether the presence of this sulfate is an artifact that originated from re-oxidation of sulfide during processing of the sediment and pore water extraction, and was originally not present in the sediment, or whether it constitutes an existing sulfate pool. It is unlikely that the sulfate diffused down from the sediment surface because the sulfate turnover time is less than 10 years in the SMTZ and there was no gradient that could drive diffusion below 330 cm. Therefore, the presence of sulfate in these deep sediment layers would imply a deep sulfate source, possibly based on the re-oxidation of iron-sulfides to sulfate (Neretin et al., 2004). The observation of these authors, that H₂S is trapped by iron from the formerly limnic sediment layers, is consistent with the disappearance of H₂S at depth in all three cores of our study. The diffusion front of dissolved iron usually coincides with the depletion of H₂S and black bands in the sediment generally mark this sulfidization (Jørgensen et al., 2004). In core P824GC the concentration of dissolved iron started to increase only 60 cm below the depth where sulfide disap-

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peared, and it is not known what causes this separation. The increase of reactive iron in the solid phase started below the depth of H₂S depletion indicating that Fe²⁺ dissolved from iron sources below and diffuses upwards, where it precipitates presumably as pyrite, in addition to the pyrite formed during the sulfur-limited stage >9800 yr ago (Jørgensen et al., 2004). Originally reduced iron had probably been present in the entire limnic sediment, but was trapped by ongoing sulfidization, as more sulfate became available with increasing salinity, and so the sulfidization front moved downwards after the Holocene/Pleistocene transition (Jørgensen et al., 2004; Neretin et al., 2004). This penetration of the sulfidization front was similar at all three sites, with sulfide depletion ca 200 cm below the sapropel layer.

As a result of the sulfide sink in deeper sediment layers, the thermodynamic yield available for microorganisms that carry out sulfate reduction coupled to AOM is very favourable at this depth. That might favour the occurrence of these processes mainly at the bottom of the SMTZ, even though both methane and sulfate are present over a broad zone.

4.3 Methanogenesis and stable isotopes

The role of bicarbonate-based methanogenesis in the cores presented is not entirely clear. The presence of detectable sulfate in the entire 500 cm long gravity core P771GC might be an explanation why no bicarbonate methanogenesis was detectable in this core. Yet, the methane in the Black Sea has been reported to be of biogenic origin (Amouroux et al., 2002, Michaelis et al., 2002), which is consistent with the $\delta^{13}\text{C-CH}_4$ values of -100% to -80% measured in our cores.

Biogenic methane formation typically yields $\delta^{13}\text{C-CH}_4$ values of -100% to -60% (Whiticar and Faber, 1986). A positive shift in $\delta^{13}\text{C}$ values in residual methane subjected to AOM (or aerobic methane oxidation) is commonly observed and results from the fractionation of carbon isotopes by methane-consuming microorganisms. A resulting ^{13}C -enrichment of methane in the main AOM zone was observed in P771GC and P806GC, similar to other AOM sites (e.g. Whiticar and Faber, 1986; Alperin et al.,

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1988). In core P824GC, in contrast, $\delta^{13}\text{C}\text{-CH}_4$ values became more negative at the top of the SMTZ. Seifert et al. (2006) suggested for AOM mats from the Black Sea that such a trend may result from methanogenesis rates exceeding those of AOM. However, in core P824GC it seems unlikely that bicarbonate-based methanogenesis rates at 160 cm depth are greater than AOM rates.

Although rates of methanogenesis are low in all three cores it appears that the greater content of organic matter in core P824GC and its proximity to the main zone of AOM activity results in greater recycling of $\delta^{13}\text{C}$ -depleted CO_2 derived from methane oxidation. This effect has been previously described by Borowski et al. (1997) and can yield highly negatively $\delta^{13}\text{C}\text{-CH}_4$ values (e.g. $< -100\text{‰}$) as a result of the normal C-isotope fractionation associated with bicarbonate-based methanogenesis (~ 60 to 70‰) being coupled with ^{13}C -depleted CO_2 produced from AOM. It remains unclear why recycling of AOM-derived CO_2 is more prominent in some sediments than in others. Data from the three Black Sea gravity cores suggest that availability of electron donors from degradation of local sedimentary organic matter may be a prerequisite for extensive C-recycling in the vicinity of the AOM zone. Noteworthy is that $\delta^{13}\text{C}\text{-DIC}$ values are comparatively negative in cores P771GC and P824GC, but anomalous ^{13}C -depletion of methane does not occur in core P771GC where the SMTZ lies entirely within the organic-matter-poor limnic sediments. In all three cores, maxima in DIC concentration coincide with the main SMTZ, thus diffusion of AOM-derived CO_2 both above and below the main zone of AOM activity is possible. Yet, only in core P824GC does the ^{13}C -poor DIC contribute to production of methane directly in the AOM zone, having highly negatively $\delta^{13}\text{C}$ values. Methane having ^{13}C values of -97 to -95‰ occurs just above the transition from limnic grey clay to marine sapropel in core P824GC.

A further distinction worth noting in the three cores is that methane-tailing exists only where recycling of AOM-produced CO_2 does not appear to occur. In core P806GC the peak of the bicarbonate-based methanogenesis measured in the SMTZ was located above the AOM zone and is consistent with the enrichment of $\delta^{13}\text{C}\text{-CO}_2$ at ~ 160 cm and the less negative $\delta^{13}\text{C}\text{-CH}_4$ signal at this depth due to the enhanced production of

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$^{12}\text{C-CH}_4$. But $\delta^{13}\text{C}$ values are not nearly as negative as from core P824GC indicating that a direct recycling of the produced $^{12}\text{C-CO}_2$ does not occur in P806GC. Thus, bicarbonate methanogenesis above the methane oxidation zone might be discussed as the reason for methane tailing at site P806GC, but this would not be consistent with the data from P771GC, where tailing also occurred. It seems, however, possible that the efficiency of AOM may be linked to degradation of organic matter within the sedimentary matrix.

Acetate methanogenesis was found in all three cores, with highest rates just below the transition between the marine and limnic sediment, but the rates were far lower than the AOM rates, and much too low to account for significant methane production in the depth intervals studied. The absence of measurable rates of methanogenesis combined with stable isotope evidence for a biogenic origin of the methane suggests that the main methanogenic zone is located below the sampling depth of these cores and that rates of methanogenesis in the sediment are extremely low and difficult to detect with the method used. Due to the gradual decrease of degradability of buried organic matter with depth and age, methanogenesis rates should be lower than the overlying SRR, yet high enough to sustain the methane flux into the SMTZ. The methane flux results from the integrated methane production over a deep methanogenic zone, reaching far down into the sediment. As an example, a methanogenesis rate of $1 \text{ pmol cm}^{-3} \text{ d}^{-1}$ over a 20 m interval would be sufficient to maintain the observed methane flux of $19 \mu\text{mol m}^{-2} \text{ d}^{-1}$ from P771GC. Integrated rates of measured AOM ($45 \mu\text{mol m}^{-2} \text{ d}^{-1}$) exceed the calculated methane flux and should therefore be sufficient to oxidize methane completely.

5 Conclusion

The incomplete oxidation of methane in the major AOM zone and resulting methane tailing is a common feature of diffusion dominated methane rich sediments of the western Black Sea. The reason for the sluggish oxidation seems to be related with the

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location of the SMTZ inside the sediment with a limnic history. Tailing of methane was less pronounced in core P824GC, where the SMTZ was located close to the marine-limnic boundary and where the limnic sediment was covered by a thick layer of marine deposits. In contrary to the typical increase in $\delta^{13}\text{C-CH}_4$ due to $\delta^{12}\text{C CH}_4$ utilization in the AOM zone the stable isotopes became lighter in core P824GC suggesting that recycling of $\delta^{13}\text{C}$ -depleted CO_2 , derived from AOM, occurred in this core in contrast to P771GC and P806GC. Such a methane cycling in the AOM zone might be based on higher levels of organic matter at the site of P824GC. In contrary, at the sites where the SMTZ is located in the originally limnic sediments it might be suggested that methanogenesis above the AOM zone could result in a tailing of the methane profile, which would be consistent with the data from P806GC but could not be confirmed at P771GC. The importance of the limnic-marine boundary was further indicated by the disappearance of methane at this transition in all cores, as well as the restriction of heterotrophic SRR to the marine deposits in P824GC.

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Table 1. Characteristics of the three sites and the gravity cores studied.

Core ID	Location	Latitude [Deg./Min]	Longitude [Deg./Min]	Water depth [m]	Marine/limnic transition	Depth of $\rho\text{CH}_4 > 1$ bar	Depth of black bands
P771GC	Paleo-Danube	43°37.83'	30°09.69	683	~70 cm	400 cm	200 cm
P806GC	Paleo-Dnjepr	44°46.83'	31°59.30'	205	~10 cm	220 cm	–
P824GC	Paleo-Dnjepr	44°39.06	32°01.00'	1014	~120 cm	280 cm	390 cm

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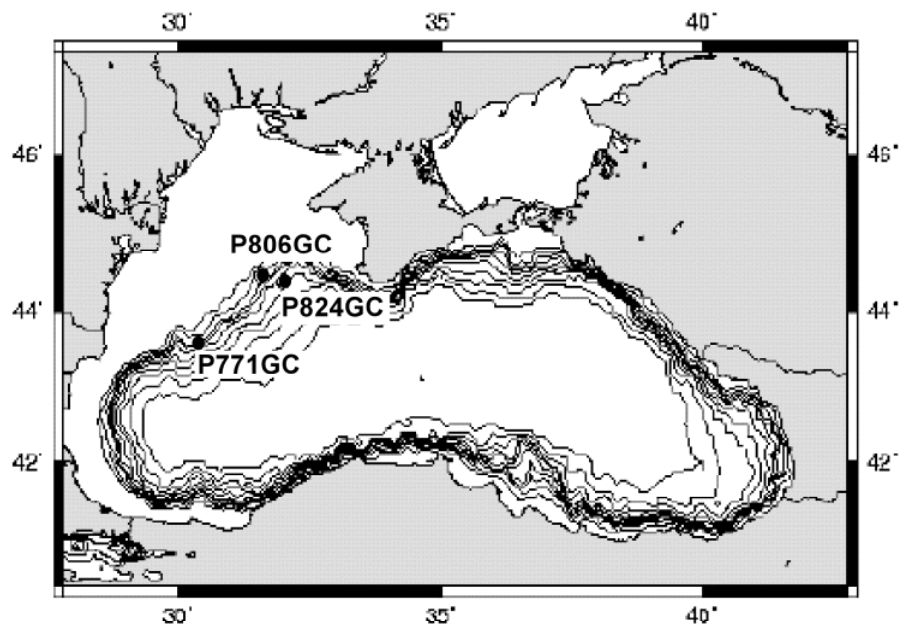


Fig. 1. Map of the Black Sea with the three different sampling sites of the gravity cores.

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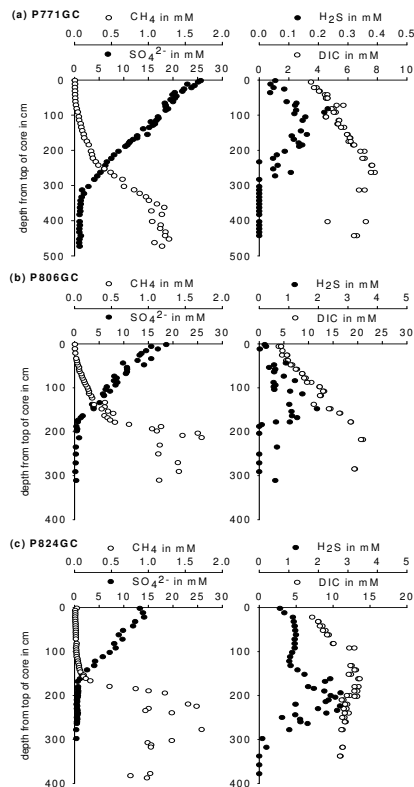


Fig. 2. Pore water concentration profiles of CH₄, SO₄²⁻, H₂S and DIC in gravity cores from the three stations. The top of the core is not identical with the sediment surface because some sediment might have been lost by coring.

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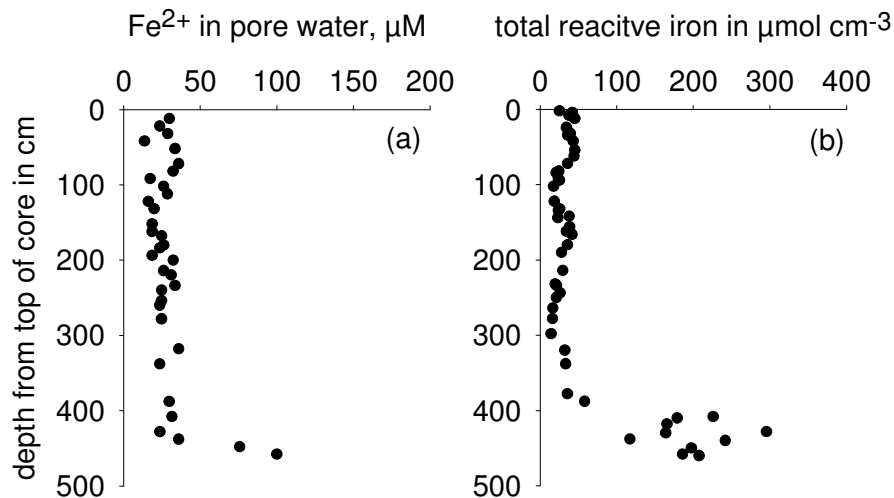


Fig. 3. Iron concentrations in core P824GC: (a) dissolved iron in the pore water; (b) total reactive iron in the solid phase.

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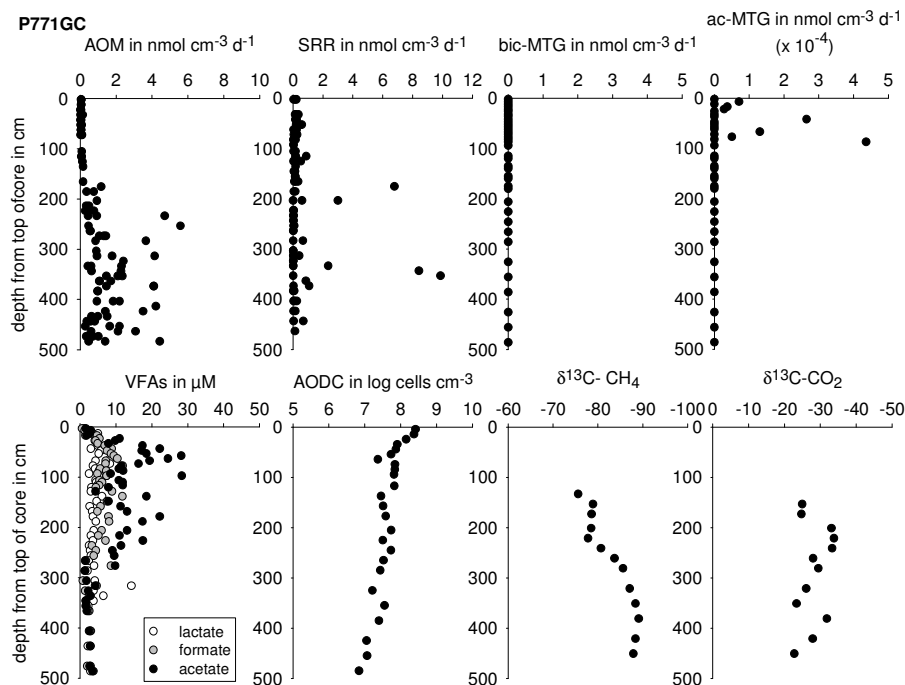


Fig. 4. Depth profiles from core P771GC of AOM, SRR, methanogenesis from bicarbonate (bic-MTG), methanogenesis from acetate (ac-MTG), volatile fatty acids (VFA), acridine orange direct cell counts (AODC), and $\delta^{13}\text{C}$ stable isotope composition of methane and bicarbonate.

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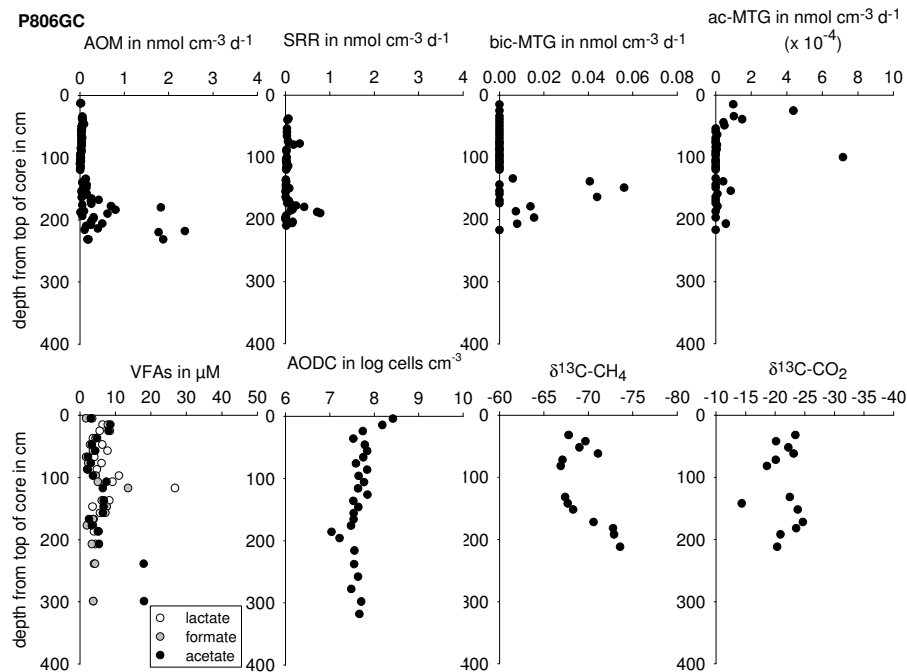


Fig. 5. Depth profiles from core P806GC, as described for Fig. 4.

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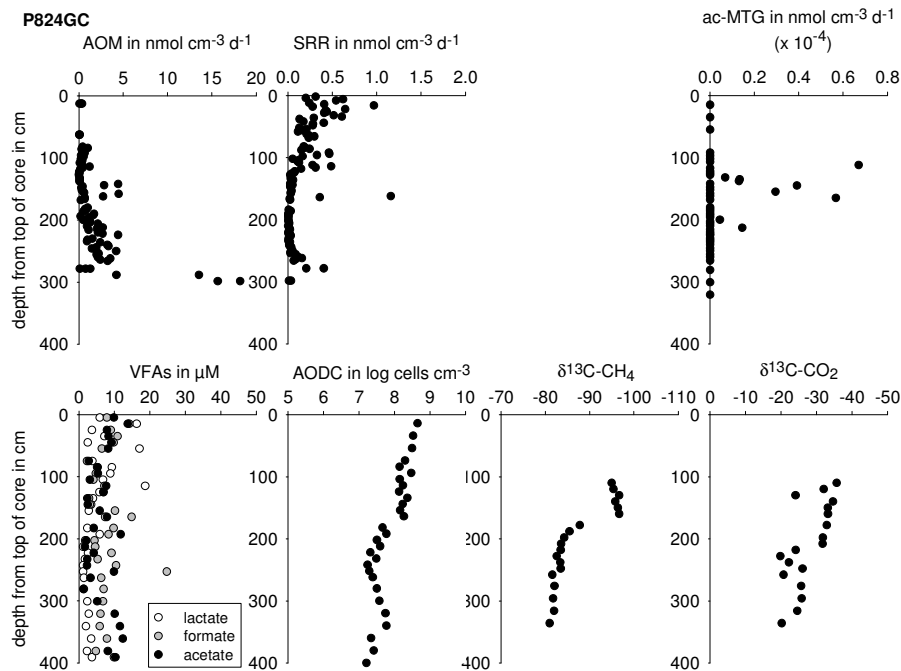


Fig. 6. Depth profiles from core P824GC, as described for Fig. 4.

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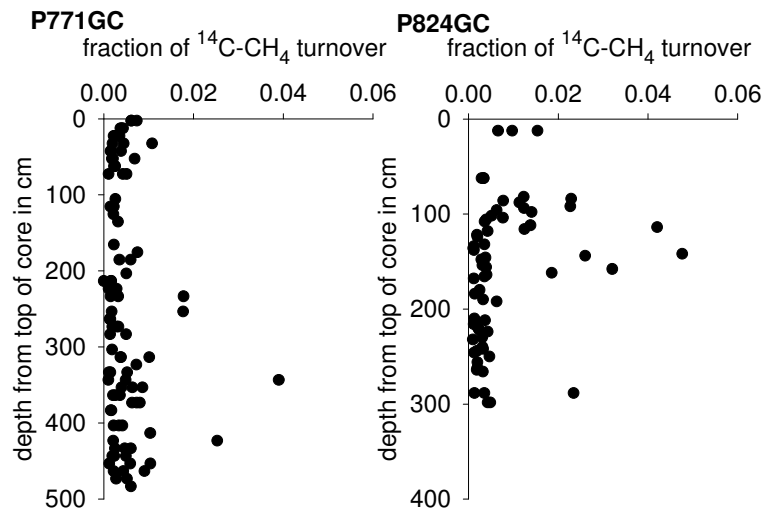


Fig. 7. Tracer turnover of ^{14}C -labeled methane, calculated as the proportion of the total ^{14}C -pool recovered as $^{14}\text{C}\text{-CO}_2$ for cores P771GC and P824GC.

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