

Interactive comment on “Contrasting effects of temperature and winter mixing on the seasonal and inter-annual variability of the carbonate system in the Northeast Atlantic Ocean” by C. Dumousseaud et al.

C. Dumousseaud et al.

eric@noc.soton.ac.uk

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Reply to anonymous referee #2: We would like to thank the reviewer for his/her valuable comments which helped us improve the manuscript. Our responses to the specific comments and technical corrections are listed below:

The authors make the conclusion that the cold winters lead to higher productivity during the following growing season, in agreement with several recent publications. By comparing two situations (cold winter/higher spring productivity and warm winter/lower

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spring productivity) they further argue for an expected lower uptake of CO₂ by the ocean, due to a stronger warming from winter to summer and decreased solubility of CO₂ in seawater despite a stronger intensity of the phytoplankton bloom. This is in contrast with previous findings and it highlights the importance of local forcings like wind-speed for the computation of Air-Sea CO₂ fluxes. I'm not confident whether this case of study could be generalized as a general trend for the future since the presented dataset of 24 months cannot also compare the effects of strong wind intensity following a cold winter or the effect of lower wind intensity following a warm winter. Hence, I suggest the last two sentences of the conclusion are rephrased in this sense.

- Thank you, we have modified the text. However, the values in Table 3 show that the temperature difference between years led to a greater influence on CO₂ flux (difference of 1.4 mmol m⁻² d⁻¹ between the two summers) than did wind speed (0.4 mmol m⁻² d⁻¹) and therefore our results highlight the importance of temperature (solubility) rather than wind speed. We acknowledge in the manuscript that the large variability in the observed CO₂ fluxes was associated with various factors including winter mixing, sea temperature and wind intensity. All these factors are themselves expected to be affected by future climate change.

Beside this, the dataset provided by Dumousseaud is presented as unique for carbonate chemistry and coccolithophore abundances. Nevertheless, for the later, only 3 values are presented: 0.9₁₀₆ cells L⁻¹ in May 2006, 0.4₁₀₆ cells L⁻¹ in July 2006 and a base line lower than 0.1₁₀₆ L⁻¹ for the remainder of the cruises. This does not constitute, in my opinion, a "unique" dataset for coccolithophore abundance in the English Channel and the Bay of Biscay, and the introduction should be adapted accordingly. - This has been modified.

1. A very detailed protocol is provided by the authors for the sampling and the determination of the carbonate parameters (sections 2.2 and 2.3), pointing out the accuracy and the precision of the measurements of TA, for example. However, it is not clear to me whether TA and DIC samples were filtered or not, since the presence of par-

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ticulate calcium carbonate phases (in the form of coccolithophore remains, coccoliths or forams) in the water column is a major interference for the titration of TA and the determination of DIC by coulometry. - We agree that in periods of large blooms the filtration of the samples can be important. However, the coccolithophores abundances observed in this study were low compared to previous years or other studies. The sea water samples were not filtered as the filtration process is likely to introduce detrimental effects on the DIC analysis. It has now been specified in the text that the samples were unfiltered.

2. The authors use their TA values from the Pride of Bilbao transect with external values of $f\text{CO}_2$ (calculated from $x\text{CO}_2$ obtained using an equilibrator-based system and infra-red absorption during the Santa Maria cruises) in order to correct DIC by a factor 2 % in 2007. They applied this correction “despite no evident problem with the analysis” (L. 149) was stated except the agreement was wrong for the first 6 months of 2007 (from February to July 2007). Such a correction of DIC, in my opinion, is fuzzy and the text poorly justifies its application to the 2007 dataset. This part needs to be clarified. The speciation of dissolved carbonates in seawater and the computation of seawater $f\text{CO}_2$ (or $p\text{CO}_2$) can be deduced with accuracy from the couple (TA, DIC), along with nutrients, T and S measurements with the CO2SYS program. Therefore, I suggest to the authors to compare, on the figure 7, the $p\text{CO}_2$ retrieved from the Santa Maria transects to those retrieved from the Pride of Bilbao transects (without the 2 % correction on DIC) in order to validate the dataset presented here. - The correction has been clarified in the text. However, for clarity reasons, we do not think that adding extra values to the graph in Figure 7 will benefit the manuscript.

3. “The calculated DIC values were compared against the measured values and showed good agreement ($\pm 4\text{--}6 \mu\text{mol kg}^{-1}$), with the exception of the February, April, May and June 2007 data” (L. 146-148). This period corresponds to the occurrence of coccolithophore blooms in the Western English Channel and the Bay of Biscay (e.g. Garcia-Soto et al., Evolution and structure of a shelf coccolithophore bloom in the West-

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ern English Channel, Journal of Plankton Research, Vol.17 iss. 11 pp.2011-2036, 1995) during which large amounts of particulate CaCO₃ can be released into the water column. In the case where DIC and TA were not filtered to remove CaCO₃, this would help explaining that aberrant values were found at this period. - *E. huxleyi* blooms in the North Atlantic area mainly occur in the period between May and July. Satellite images also confirmed the low *E. huxleyi* numbers observed during the two years of our study compared with the previous years when bloom conditions with high cell numbers were observed.

4. "Seawater samples were pre-filtered through a 200 μm mesh to prevent zooplankton grazing during the filtration" (L. 158-159). Which was the volume filtered for coccolithophore abundance determination? Could you show the time series of coccolithophore abundances? It would be helpful to compare also time series of coccolithophore abundances with remote sensing water-leaving radiance or remote sensing derived CaCO₃ concentration. - The volume has now been specified in the text. However, due to the low coccolithophore abundances observed throughout the time of the study (confirmed by satellite images), the authors consider that the comparison of the abundances with remote-sensing observations would not benefit the manuscript.

5. It is stated by the authors that "the wind-speed data used for the calculation of the gas transfer velocity was (were?) obtained from the MET Office Gascogne Buoy hourly data and averaged for each crossing" (L. 191-192). This buoy is located 45.201 N 5.000 W, off Bordeaux. As I understand the text, a single $_$ is derived from this location and applied to the entire transect "crossing eight regions of different oceanographic characteristics" (L. 90). Wouldn't it be more appropriate to use synoptic wind-speed data, for example using wind-speed fields along the transects to derive $_$ rather than a single value? - Thank you. We have now used the QuikSCAT daily wind speed data monthly averaged for each of the six regions studied. The discussion has been modified accordingly and the wind data observed for each region is shown in Figure 7.

6. The equation (L. 300) proposed by the authors "in order to estimate the maximum

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6, C4367–C4372, 2010

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TA drawdown (Δ TA) expected from the coccolithophore abundances (C)” (L. 294-300) is an estimate of the particulate inorganic carbon (PIC) associated to the abundance of coccospheres in the surface waters and not an estimate of the impact of biogenic calcification on seawater TA, as stated by the authors. To be exact in the computation of PIC, the abundance of suspended coccoliths released continuously by *E. huxleyi* should be added to this value since the coccolith:coccosphere ratio can reach up to 175 in the Western English Channel (Garcia-Soto et al., 1995). Calcification is a continuous process (with a specific rate) that occurs along the life cycle of the coccolithophore blooms (3-4 weeks in Garcia-Soto et al., 1995), during which HCO_3^- is converted into calcite, which lowers the TA by 2 μmol as 1 μmol of CaCO_3 is produced. The drawdown of TA (Δ TA) is the integration over the watermass and over the time of several processes such as calcification and dissolution of CaCO_3 and is related to the recent history of the watermass, and not on the standing stock of PIC. The authors, here, make the confusion between rates and standing stocks and suggest a weak influence of coccolithophore calcification on the carbonate chemistry of the Bay of Biscay. The mixing of cold and deep nutrient-rich waters in surface due to the occurrence of tidal waves in the Northern Bay of Biscay (see Fig. 3 in Wollast and Chou, 2001) brings alkalinity to the surface and lateral advection of waters with a different history with respect to coccolithophores are other cases where the Δ TA cannot be stated as a robust descriptor of the influence of coccolithophore calcification in surface waters. - For reasons of brevity, the calculation of the TA variation estimated from the coccolithophore abundances has been removed as it does not represent a key point in the manuscript.

7. It is stated “that the seasonal distribution of TA was not primarily controlled by production and dissolution of calcium carbonate (confirmed by the low coccolithophore abundances observed), but by uptake and supply of nitrate, (: : :) and freshwater inputs or removal such as mixing, precipitation, evaporation or river inputs ” (L. 304-307). The uptake of nutrients by phytoplankton, in agreement with the nutrient- H^+ -compensation principle (Wolf-Gladrow et al., 2007), increases the TA and, therefore, cannot explain the drawdown of TA observed by the authors in April 2006 and 2007 in the Bay of

Biscay, except if calcification concomitantly occurs with primary production, the general feature of coccolithophore blooms. - The authors agree with this. What we aimed to convey in the text is that it is the overall seasonal distribution that is affected by these various factors, and not the TA drawdown observed in April. The text has been modified in order to clarify this.

8. A C:N ratio of 8.4 is used through the text (from L. 380 on), for which no reference is given. The computation of the _DIC:_Nitrate ratio in Table 3 roughly gives values ranging from 6.4 (close to the Redfield C:N ratio) to 9.4 (provided that the _Nitrate is given in _mol kg^{-1} instead of _mol L^{-1} , as referred in Table 3, and in the sake of homogeneity with the scale of figure 2c in _mol kg^{-1}) with an average of 8.0 for the entire zone of study. Such a value above the Redfield stoichiometry is a strong indication for C-overconsumption in this area. This is an important conclusion that should be pointed out later in the conclusion. - Thank you, the text has been modified in order to give the range of DIC/Nitrate ratio observed and the choice of the 8.0 ratio has been clarified (we are thankful to the reviewer for pointing this mistake as the ratio of 8.4 previously used corresponded to the 2005/2006 values averaged). The error in the nitrate units in Figure 2c has been corrected.

9. "The historical time-series of the monthly NAO index showed a dominant negative phase for the period between March 2005 and November 2006 (: : :), corresponding with the enhanced winter MLD observed in the Bay of Biscay during this period, and a positive phase during the winter of 2006/2007, corresponding with the shallower winter MLD during this period". I suggest the NAO index should also be presented on the figure 5. For a better reading of the correspondence between MLD and T, the width of the figure 5b should be the same as fig 5a. - Thank you, the Figure 5 has been modified. However, the winter NAO index was used (combination of Dec/Jan/Feb/Mar), and therefore adding the values to the figure would not benefit the figure. The text has now been clarified and values were added in the text in section 3.6.

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