

Interactive  
Comment

***Interactive comment on “Can Mg isotopes be used to trace cyanobacteria-mediated magnesium carbonate precipitation in alkaline lakes?” by L. S. Shirokova et al.***

**L. S. Shirokova et al.**

oleg@lmtg.obs-mip.fr

Received and published: 12 October 2011

The first major concern of this reviewer is that we based our conclusion that Mg isotopes cannot be used to track biologically mediated precipitation of hydrous Mg-carbonates on the results from experiments with one genus of bacteria (*Synechococcus* sp.). We think that these results can be extended to all cyanobacteria in general. Our recent work on another cyanobacteria, *Gloeocapsa* sp. performed in experimental Mg, HCO<sub>3</sub><sup>-</sup> - bearing solutions yielded very similar isotopic shift between carbonate mineral and aqueous solution in case of biologically-induced and abiotic (inorganic) precipitation (Mavromatis et al., 2011, GCA, in press). The cellular organization of two

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

Discussion Paper



cyanobacterial species is totally different: *Synechococcus* sp. is represented by individual cells in the form of cocci whereas as *Gloeocapsa* sp. Form large associates of 2 to 5 cells enclosed together in a rigid polysaccharidic capsule. In addition, the two species exhibit dramatically different features of biocalcification: *Synechococcus* sp. possesses a self-protection mechanism against uncontrolled encrustation (calcification) at the cell surface (Martinez et al., 2010) which is absent in *Gloeocapsa* sp. (Bundeleva et al., 2011, and Bundeleva, 2011). Finally, other bacterial species such as *Plankthothrix* sp., and another strain of *Synechococcus* sp. (Mavromatis et al., 2010, 2011) yielded the  $\text{Mg}^{2+}$  – liquid during hydrous Mg carbonate precipitation very similar to that of Salda lake culture (*Synechococcus* sp.B8901) and *Gloeocapsa* sp. Given that all studied cyanobacterial species largely present in Salda lake water and stromatolites (this study and Braithwaite and Zedler (1994, 1996) we feel that the conclusion is applied to dominant types of cyanobacteria in modern stromatolites of alkaline lakes. We added the necessary explanation in the Discussion as recommended by the reviewer.

Bundeleva, I.A.: Modélisation expérimentale de la précipitation des minéraux carbonates lors de l'activité bactérienne. These Doctorat de l'université de Toulouse, 273 pp, 2011. Bundeleva, I.A., Shirokova, L.S., Pokrovsky, O.S., Bénéth, P., Balor, S.: Experimental modeling of calcium carbonate precipitation by cyanobacteria *Gloeocapsa* sp. Geobiology, submitted, 2011. Martinez, R. E., Gardes, E., Pokrovsky, O. S., Schott, J. and Oelkers, E. H.: Do photosynthetic bacteria have a protective mechanism against carbonate precipitation at their surfaces? *Geochim. Cosmochim. Acta* 74, 1329-1337, 2010. Mavromatis, V., Pearce, C., Shirokova, L.S., Bundeleva, I., Pokrovsky, O.S., Bénéth, P., Schott, J.: Magnesium isotope fractionation during inorganic and cyanobacteria-induced hydrous magnesium carbonate precipitation. *Geophysical Research Abstracts*, EGU General Assembly 2010, Vol. 12, EGU2010-11616-1, 2010. Mavromatis, V., Pearce, C., Shirokova, L. S., Bundeleva, I. A., Pokrovsky, O. S., Bénéth, P. and Oelkers, E.H.: Magnesium isotope fractionation during inorganic and cyanobacteria-induced hydrous magnesium carbonate precipitation, *Geochim. Cosmochim. Acta*, in press, 2011. Mavromatis, V., Shirokova, L.S., Bundeleva, I.,

C3550

**BGD**

8, C3549–C3552, 2011

Interactive  
Comment

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

Discussion Paper



Pokrovsky, O.S., Bénézech, P., Oelkers, E.H., Gerard, E.: Biomineralization of hydrous Mg carbonates in the Salda Lake, SE Turkey: New insights from stable Mg isotopes. EGU, Vienna, Geophysical Research Abstracts 2011, Vol. 13, 1598, 2011.

The second remark of this reviewer is that the abstract does not adequately portray all that is useful in this paper. He/she suggested editing the abstract via including the information from the last three pages of the text, and we revised the abstract accordingly. We also moved up the final sentence of the abstract as recommended.

General technical comments: We corrected “Mg-carbonates” instead of “magnesium carbonates” ;  $\delta^{26}\text{Mg}_{\text{mineral-solution}}$  instead of  $\delta^{26}\text{Mg}_{\text{solid-solution}}$ , defined and used throughout the text the term “stromatolite” as recommended. The reviewer inquired about the possibility of incorporating the electronic supplementary material into the body of the paper. Biogeosciences allows having large dataset published as ESM and we would like to use this option for presenting the primary chemical analyses data which are not interesting for general reader. Both portrait and landscape figures can be incorporated in the manuscript should it be published in the final version.

We carefully edited the number of decimal places for  $\delta^{26}\text{Mg}$  in tables and corrected as recommended. Note that when presenting the range of isotopic values, we tried to minimize the number of decimal places. Given that the 2 sigma value is always reported in Tables 2 and 3, these numbers are rigorously defined.

The reviewer suggest to use the terms “more positive” and “more negative” instead of “heavier” and “lighter” and we corrected the text accordingly.

We converted Figures 9 and 10 in black and white as recommended.

We took into account all specific technical comments noted by reviewer. We added references where requested, changed “mineral carbonation” to “mineral  $\text{CO}_2$  sequestration”. We explained that the most significant stromatolite growth occurs in warmer waters and at higher solar radiation during the summer and as such, water samples

**BGD**

8, C3549–C3552, 2011

Interactive  
Comment

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

Discussion Paper



were collected during September 2010 when the water temperature was the highest. As recommended, we reported the range of delta values for the internal and external parts of stromatolites and the littoral sand, and included the chemical formula for brucite.

The reviewer suggested to “to speculate on dypingite saturation state even if we can’t calculate it because it is a major phase precipitated in their experiments.” This is useful remark; however, the first precipitated phase is nesquehonite, not dypingite. As such, the major solution changes occur during nesquehonite formation whose supersaturation index is discussed. *Chroococcales* sp. was a misprint, we meant *Synechococcus* sp.

We corrected all noted misprints in Tables and Figures.

We thank anonymous reviewer No 1 for very constructive and insightful review.

---

Interactive comment on Biogeosciences Discuss., 8, 6473, 2011.

**BGD**

8, C3549–C3552, 2011

---

Interactive  
Comment

Full Screen / Esc

Printer-friendly Version

Interactive Discussion

Discussion Paper

