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Snowpack concentrations and estimated fluxes of volatile organic compounds in a boreal forest

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Pan	Title Page					
	Abstract	Introduction				
7	Conclusions	References				
	Tables	Figures				
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Abstract

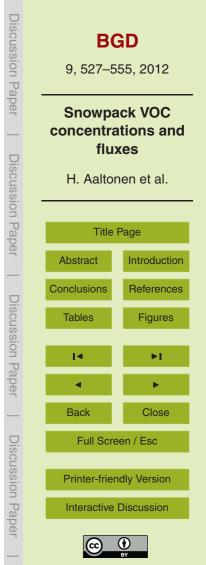
Soil forms an important source for volatile organic compounds (VOCs), but in boreal forests these fluxes and their seasonal variations have not been characterized in detail, especially wintertime fluxes, which are almost completely unstudied. In this study, we measured the VOC concentrations inside a snowpack in a boreal Scots pine (*Pinus sylvestris* L.) forest in southern Finland, using adsorbent tubes and air samplers installed permanently in the snow profile. Based on the VOC concentrations at three heights inside the snowpack, we estimated the fluxes of these gases. We measured 20 VOCs from the snowpack, monoterpenes being the most abundant group with concentrations varying from 0.11 to 16 µg m⁻³. Sesquiterpenes and oxygen-containing monoterpenes were also detected. Inside the pristine snowpack, the concentrations of terpenoids decreased from the soil surface towards the snow surface, suggesting soil as being the source for terpenoids. Forest damages resulting from heavy snow loading during the measurement period increased the terpenoid concentrations dramatically,

especially in the upper part of the snowpack. The results show that soil processes are also active and efficient VOC sources during winter and that natural or human disturbance can increase forest floor VOC concentrations substantially. Our results stress the importance of soil as a source of VOCs during the season when other biological sources, basically plants, have lower activity.

20 **1** Introduction

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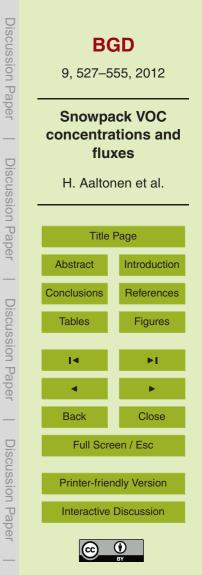
Volatile organic compound (VOC) emissions in boreal forests at the branch and canopy levels have been rather well characterized (Hakola et al., 2003, 2006; Ruuskanen et al., 2005), in contrast to soil VOC fluxes. Wintertime atmospheric VOC concentrations and fluxes from the canopy (Hakola et al., 2003, 2009; Lappalainen et al., 2009) are better known than those from soil, which are almost completely unstudied. VOC emissions from boreal forest soil are highest in the spring and autumn (Hellén et al., 2006; Aaltonen et al., 2011), but the processes behind the seasonal variations remain



uncertain. Most likely they are related to biological activity either above or below the soil surface. Since the soil surface is covered by snow during a substantial part of the year in the Boreal Zone, we conducted terpenoid concentration measurements inside the snowpack. So far, few studies reporting measurements of snowpack gas concen-

- ⁵ trations have been published (e.g. Helmig et al., 2009; Kos and Ariya, 2010; Ariya et al., 2011), and these studies have been focused mainly on VOCs other than terpenoids. The air chemistry in the troposphere (Kulmala et al., 2000), in which VOCs also take part, stresses the importance of better understanding of wintertime forest floor VOC processes.
- ¹⁰ During the active growing season, the above- and belowground parts of plants, as well as the diverse and effective soil microbial populations maintained by the forest ecosystem, are a source of forest floor and soil VOCs (Janson, 1993; Hayward et al., 2001; Asensio et al., 2008; Leff and Fierer, 2008; Bäck et al., 2010). In winter, biological processes contribute to these fluxes, such as seasonal activity of plants,
- ¹⁵ change due to limiting environmental factors and, most importantly, temperature and light availability. Root exudation plays a major role in the forest carbon cycle. Mycorrhizal hyphae, which are important contributors to decomposition of soil organic matter, are largely dependent on recent photosynthetate emitted from the roots (Högberg et al., 2008). Mycorrhizal fungi emit substantial amounts of VOCs, the amount and qual-
- ity being highly species-specific (Bäck et al., 2010). In late autumn and in winter when the photosynthesis of the trees is low, the lack of easily available energy sources (root exudates) may also decrease the decomposing activity in the soil, but most likely temperature is the overwhelming factor that affects the activity of decomposing organisms in the soil (Davidson et al., 2006; Pumpanen et al., 2008; Schindlbacher et al., 2008;
- Vesala et al., 2010). In addition to VOC sources in the soil, microbes living inside and on the snowpack may produce VOCs, but also act as sinks for them (Helmig et al., 2009; Ariya et al., 2011).

In boreal areas, leaf turnover is an important process involved in winter tolerance and nutrient retranslocation from older to younger needles. In deciduous species, leaf fall is

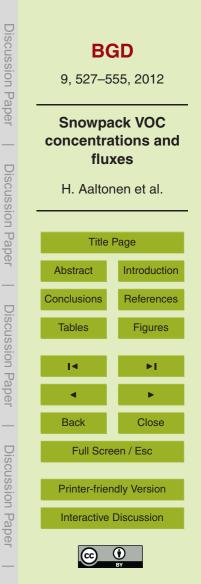


regulated by light availability and temperature, inducing senescence in a rather limited period of time in autumn. For evergreen foliage, the litterfall occurs in a less specific time. VOCs from fallen needles and other litter may be released during winter as a result of decomposition and physical degradation. Due to freezing and thawing cycles, the physical breakdown/degradation of litter is more important during winter than other seasons.

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Chamber techniques have traditionally been applied for measurement of forest floor VOC fluxes and for other greenhouse gases as well. However, difficulties caused by low wintertime VOC fluxes and fluxes passing the chamber due to the high porosity of

- snow complicate the use of chambers for snowpack VOC flux measurements. Thus, we applied a gradient method for measurements, with custom-made samplers collecting air samples from three levels within the snowpack. This method is optimal for low fluxes, causing a smaller disturbance for the gases measured, and the flux calculation is based on concentration differences between the sampling layers. The gradient
- ¹⁵ method also has some disadvantages, such as the need for additional information on environmental variables (e.g. temperature, humidity/moisture, porosity of the media) and its sensitivity to diffusivity of the target compound(s), especially when the diffusion coefficients for many compounds are poorly available and difficult to determine (Pumpanen et al., 2009).
- We measured the VOC concentrations, more specifically the terpenoids, inside the snowpack in a boreal forest during two subsequent winters. With these measurements, we wanted to improve our estimates of the importance of the winter season to annual VOC emissions in a boreal forest ecosystem. We assumed that the biologically active surface soil produces higher snowpack VOC concentrations close to the soil and that
- the formation of VOCs is dependent on soil temperature, the driving force of soil biological activity. Based on the concentrations of terpenoids inside the snowpack, we estimated the fluxes of these gases in the snowpack between the soil and the atmosphere.



2 Methods

2.1 Measurement site

We measured the snowpack VOC concentrations from a Scots pine (*Pinus sylvestris* L.) forest at the SMEAR II (Station for Measuring Forest Ecosystem-Atmosphere Relations II) station (61°51′ N, 24°17′ E, 180 m a.s.l.), located in the vicinity of the Hyytiälä Forestry Field Station in southern Finland (Hari and Kulmala, 2005). The forest stand at the SMEAR II station is 46 years old and dominated by Scots pine (> 60% of the trees); some Norway spruce (*Picea abies* (L.) H. Karst.), aspen (*Populus tremula* L.) and birch (*Betula* L. spp.) also grow in the forest. The stand height is ~18 m and the canopy is open, with an average tree density of ~1370 stems (diameter-at-breast height ≥5 cm) per hectare (Ilvesniemi et al., 2009). The soil above the homogeneous bedrock is Haplic podzol in glacial till, with an average depth of 0.5–0.7 m.

2.2 VOC profile measurements

We performed the snowpack VOC profile measurements during winters 2008–2009 and 2009-2010 (Table 1), in periods when the soil was covered by at least 10 cm of 15 snow. The VOC concentrations were measured from three permanently installed gas profile collectors consisting of three partially perforated circular polytetrafluoroethylene (PTFE) tubes (collector), where the lowest tube was placed at the ground, the middle tube at a height of 10–15 cm, while the uppermost was 20 cm higher than the middle tube. The PTFE tubes were 4 m long, with a 10 mm outer diameter and 1 mm wall 20 thickness. The central part had two 0.7 m-long sectors with perforation, and both ends and the middle sector were without perforation (Fig. 1). Samples were collected by circulating air from the sampling tubes through a Tenax-Carbopack-B adsorbent tube at a flow rate of $\sim 100 \,\mathrm{ml\,min^{-1}}$, using portable pumps. Between samplings, both ends of the PTFE tubes were closed. During summer 2009, the collectors were removed 25 from the field, but reinstalled in the same positions in autumn. Chemically inert PTFE



was chosen for the material of the collectors to avoid surface reactions deleterious to sample quality. The tubes were installed at selected heights with thin wires (\emptyset 2 mm) to minimize the heating of dark surfaces by sunlight.

Samplings were performed approximately monthly, each time consisting of four 15 min sampling periods (60 min total), with 15 min gaps between them. The air content inside the collector (~0.21) alone would not have been sufficient for analysis; thus, the air samplings were prolonged to 60 min. The samplings were divided into four periods to equilibrate the VOC concentrations between the collector and the surrounding snow.

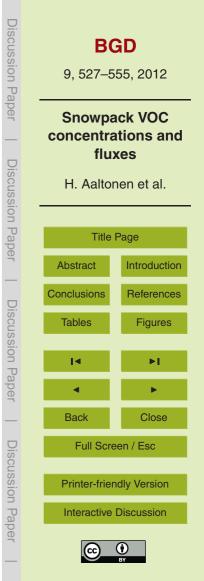
The samples in the adsorbent tubes were analysed in the laboratory, using a thermodesorption instrument (Perkin-Elmer TurboMatrix 650; PerkinElmer, Waltham, MA, USA) attached to a gas-chromatograph (Perkin-Elmer Clarus 600) with a massselective detector (Perkin-Elmer Clarus 600T). The sample tubes were desorbed at 300 °C for 5 min, cryofocused in a Tenax cold trap (-30 °C) prior to injecting the analytes into the column by rapidly heating the cold trap (40 °C min⁻¹) to 300 °C. The mass detector used enables simultaneous full scan and singular ion monitoring. Five-point calibration standards in methanol solutions were used. The standards were injected into the sampling tubes and the methanol was flushed away before the analysis. The

detection limits varied from 0.04 ng to 0.60 ng per tube and the overall uncertainties

of sampling and analysis, calculated from parallel samples, were 12%, 10–40% and 33–52% for isoprene, monoterpenes and sesquiterpenes, respectively.

2.3 Supportive data

Snow depth was measured weekly at seven points, starting from the first snowfall and ending after the snowpack was completely melted. The snow water equivalent was measured every second week from the seven points for estimating the diffusivity of the ²⁵ snowpack. Since the snow water content was measured for the full snowpack only, we used the SNOWPACK model to estimate the snow porosity throughout the snowpack during the VOC measurements. SNOWPACK is a one-dimensional model for snowpack structure, mass and energy balance, developed at the Swiss Federal Institute for



Snow and Avalanche Research (SLF) for avalanche-warning purposes. SNOWPACK is a predictive model that uses Lagrangian finite elements to solve for heat and mass transfer, stresses and strains within the snow cover. The model is physically based: energy balance, mass balance, phase changes, water and water vapour movement

- 5 are included, and the layer calculations are based on snow microstructure (crystal size and form, bond size, number of bonds per crystal). A complete description of the model can be found in Bartelt and Lehning (2002), Lehning et al. (2002a, b). SNOWPACK has been validated in several studies under varying climatic conditions (e.g. Lehning et al., 1998; Lundy et al., 2001; Rasmus et al., 2007).
- Air and humus layer temperatures (PT-100 resistance thermometer) and air-pressure 10 data (Pressure Indicator DPI 260, Druck Ltd, Leicester, UK) were needed for flux calculations; the data were provided by the SMEAR II station (Hari and Kulmala, 2005). The humus layer temperatures were also valuable data for estimating the biological activity of the soil.
- During the second winter of measurements (2009–2010), heavy snow loads caused 15 substantial forest damages by breaking treetops and branches, as well as by felling whole trees. In spring 2010, the damages (more specifically the species composition, stem volume and basal area-at-breast height of the damaged trees) at the SMEAR II area were measured.

2.4 Flux calculations

For calculation of the VOC fluxes, based on the profile concentration measurements, we used Fick's law of diffusion as follows:

$$F = -D\left(\frac{\delta C}{\delta z}\right),\tag{1}$$

where F is the gas flux $(ngm^{-2}h^{-1})$, D is the diffusivity of certain compound(s) in air (m² h⁻¹), δC the difference in gas concentrations (ng m⁻³) and δz the distance

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(m) between adjacent collectors. The diffusivity, taking into account the properties of porous media, was estimated by:

$$D = \phi \tau D_i \left(\frac{P_0}{P}\right) \left(\frac{T}{T_0}\right)^{1.75},$$

modified from Seok et al. (2009), using a uniform temperature exponent (1.75) with Eq. 5. φ is the snowpack porosity and τ the tortuosity, D_i is the diffusion coefficient of certain compound(s), which is scaled to the predominant conditions of pressure (*P*) and temperature (*T*). Since we had temperature measurements of the air and humus layer, the temperature values for the snow profile were interpolated by assuming that these measurements represented both ends of the snowpack. Using the SNOWPACK

- model, we validated this method for calculating the profile temperatures by assuming that the change was linear inside the snowpack, which showed good agreement, except for sunny springdays. The fixed temperature exponent 1.75 is a theoretically determined coefficient used also for calculations of diffusion coefficients (Fuller et al., 1969).
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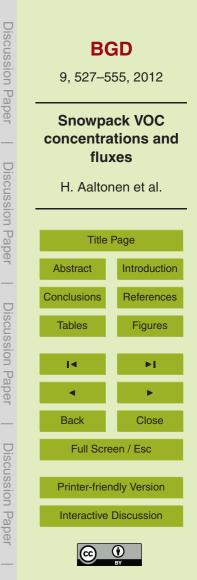
Snowpack porosity (ϕ) was calculated by the equation

$$\phi = 1 - (h_w/0.917/h_s),$$

where h_w is the snow water equivalent (m), h_s the snowpack depth (m) and 0.917 describes the relationship between the densities of ice and water. Tortuosity (τ) is calculated from porosity according to Duplessis and Masliyah (1991) by the equation

$$\tau = \frac{1 - (1 - \phi)^{\frac{2}{3}}}{\phi}.$$
 (4)

Experimentally determined diffusion coefficients for terpenoids are seldom available. Helmig et al. (2003) determined the diffusion coefficients for some sesquiterpenes, but for monoterpenes only some theoretical estimates are available (van Roon et al.,



(2)

(3)

2005). Thus, we estimated the diffusion coefficients by the equation originally formed by Fuller et al. (1969) and modified by Poling et al. (2000) as follows:

$$D_{AB} = \frac{0.00143T^{1.75}}{P\left\{2\left[\left(\frac{1}{M_A}\right) + \left(\frac{1}{M_B}\right)\right]^{-1}\right\}\left[(\Sigma_v)_A^{\frac{1}{3}} + (\Sigma_v)_B^{\frac{1}{3}}\right]^2},$$

where D_{AB} is a binary diffusion coefficient (cm² s⁻¹), *T* temperature (K), *P* pressure (bar), M_A and M_B molecular weights of compounds A and B (g mol⁻¹). Σ_v represents the summed atomic/molecular diffusion volumes, which were 15.9 for carbon, 2.31 for

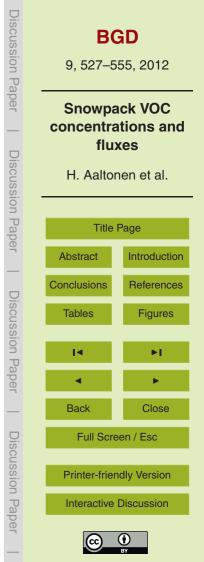
hydrogen and 19.7 for air (Fuller et al., 1969). Since many of the mono- and sesquiterpenes contain one or more ring structures, but Fuller et al. (1969) determined only the effects of aromatic and heterocyclic rings on diffusivity (structural diffusion volume

- -18.3 for both these of ring structures), ignoring all other types of ring structures, we subtracted 18.3 once from the summed diffusion volumes. The compounds within the monoterpene group are isomeric, as are those within the sesquiterpenes, and the diffusion coefficients of the various isomeric forms are equal. It is possible that the VOCs undergo chemical reactions inside the snowpack before they reach the snow surface,
- ¹⁵ but since these reactions are poorly known, we could not implement them in the calculations. For further calculations the unit of D_{AB} was converted as m² h⁻¹.

3 Results

3.1 Environmental conditions

The air temperatures of these two sampling winters (1 November–30 April) differed greatly at the SMEAR II station (Fig. 2). In 2008–2009, except for a cold period in late March, the temperatures were always higher than the 30-year average, which is -4°C for period 1 November–30 April and -6.9°C for the winter months (Drebs et al., 2002).



(5)

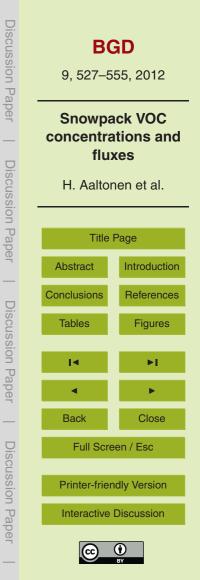
However, throughout the following midwinter 2009–2010 the temperatures were notably lower than the average. Exceptionally, the air temperatures in winter 2009–2010 were continuously below zero for over 3 months, beginning from the early December. The difference between winters was also wide in the humus layer temperatures. During the first winter, the temperature was above zero most of the time, dropping below

- the first winter, the temperature was above zero most of the time, dropping below zero only occasionally, whereas in winter 2009–2010 temperatures were below zero continuously from mid-December to early March (Fig. 2). In the latter winter, the humus temperatures dropped close to -3° C, while in the first winter it never dropped even to -1° C.
- ¹⁰ These two winters also differed, regarding the amount of snow (Fig. 2). During winter 2008–2009 the snow depth was throughout the winter clearly below the 30-year average (Drebs et al., 2002), while in winter 2009–2010 it was very close to the average. In winter 2008–2009, the uppermost profile collectors were barely covered by snow only during the time of the thickest snowpack, the average maximum snowpack thickness of the 20 was a partial bains 47 cm (Drebs et al., 2022).
- of the 30-year period being 47 cm (Drebs et al., 2002). The depth of the snow in winter 2009–2010 was approximately twice that in the previous winter, being ~60 cm at maximum. The maximum depth in February–March was slightly above the average, but at the beginning and end of the snow cover period, the snow depth was lower than the average. The water equivalent of snow was also over two-fold higher in 2009–2010
 than in 2008–2009, being ~10 % of the snow depth during both winters.

During the forest damages in winter 2009–2010, approximately 12% of the trees fell at the SMEAR II stand, and practically all of these (> 99%) were Scots pines. In addition to fallen trees, a large unmeasured volume of different sizes of branches and needles dropped on and inside the snowpack.

25 3.2 Snowpack VOC concentrations

We detected 20 biogenic VOCs from the snowpack, 11 of which were monoterpenes and oxygen-containing monoterpenes, 7 sesquiterpenes and 2 hemitepenoids (Table 2). The most abundant VOC group was the monoterpenes, the average



concentrations varying from 0.13 to $4.4 \,\mu g \,m^{-3}$ during the first winter and from 0.11 to $16 \,\mu g \,m^{-3}$ during the second winter (Fig. 3). The concentration data, which were used also for flux estimations, from two profiles harmed during the forest damages in the second measurement winter were omitted from these averages. Δ^3 -carene and α -⁵ pinene were the predominant compounds, followed by β -pinene, terpinolene, limonene and camphene. During the first winter, α -pinene had the highest concentrations, but in the middle of the second winter, Δ^3 -carene clearly predominated. The average sesquiterpene concentrations were generally less than half of the monoterpene concentrations (Fig. 4), $1.4 \,\mu g \,m^{-3}$ being the highest concentration in the first winter and 0.82 $\mu g \,m^{-3}$ in the second. Occasionally, the sesquiterpene concentrations were close to the detection limit, but on the first measurement day 27 November 2008 no sesquiter-

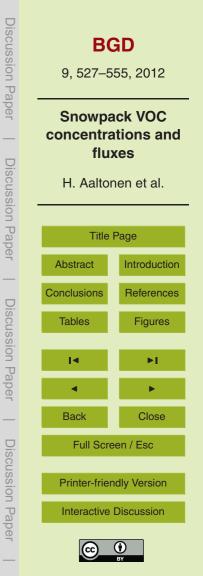
penes were found. Isoprene was detected in the first samplings of both winters, but its concentrations were negligible, reaching to $0.40 \,\mu g \,m^{-3}$ at maximum (Table 2).

During the first winter and the first 2 months of the second winter, the monoterpene concentrations were clearly highest at the ground level and decreased rapidly towards the snow surface (Fig. 3). During the latter half of the second winter, the bottom-totop order of monoterpene concentrations changed in two out of three profiles. With the sesquiterpenes, the trend of decreasing concentrations towards the snow surface persisted more or less throughout the measurement period (Fig. 4). At the end of the

²⁰ second winter, the monoterpene concentrations, especially β -pinene, Δ^3 -carene and terpinolene, increased dramatically (from 1000- up to 1700-fold) in these two profiles. With other monoterpenes, the increase varied from 10- to 200-fold and with sesquiterpenes the increase in concentrations was approximately 10-fold.

The monoterpene concentrations in the lowest collectors were on average about

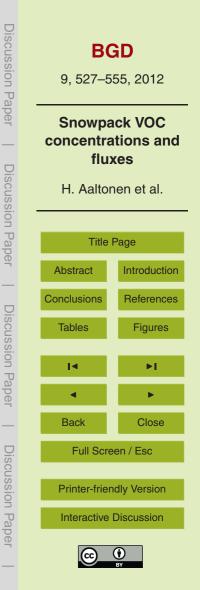
²⁵ 10-fold higher (5.3–11) than in the middle collector throughout the first winter, but in the second winter the ratio dropped to approximately 1 (0.36–1.5). However, for the sesquiterpenes the ratio was more similar in both winters (1.8–8.8), with no change between winters.



3.3 Wintertime fluxes of VOCs

We estimated the wintertime forest floor VOC fluxes, based on our measurements of VOC concentrations in the snowpack profiles. The snowpack in winter 2008–2009 did not fully cover the topmost collector during any measurement and was even less than

- ⁵ 20 cm thick during the first two measurements. Therefore, for the first two measurement times we calculated fluxes only from the ground level to the middle collector and for the two latter times the flux values from the middle to the topmost collectors were merely indicative. Since the diffusion calculation (Eq. 1) was parameterized for snow and it did not take into account convective transport, it was not meaningful to estimate the transport in air. The forest damages in midwinter 2009–2010 destroyed the structure of two out of three profiles (those with the high terpenoid concentrations), thus from
- the last two measurements we obtained the concentration data for the flux calculations from one plot only.
- The monoterpene fluxes from the ground level to the middle of the profile varied from 10 to 490 ng m⁻² h⁻¹, during winter 2008–2009 (Fig. 3), the fluxes from the middle to the top profile being very low (26 ng m⁻² h⁻¹ at maximum). During the latter winter, negative fluxes were also observed, the values ranging from -580 to 590 ng m⁻² h⁻¹ for the lower level and -300 to 670 ng m⁻² h⁻¹ for the upper level. The sesquiterpene concentrations were clearly lower than those of the monoterpenes, as were also the sesquiterpene fluxes (Fig. 4). In the first winter, fluxes from the ground to the middle of the snow profile varied between 0.19 and 22 ng m⁻² h⁻¹, except for one plot on 5 February 2009, when the flux was 220 ng m⁻² h⁻¹. The fluxes from the middle of the profile to the top were practically zero (-0.96-0.08 ng m⁻² h⁻¹). During the second winter, the fluxes were -0.36-83 ng m⁻² h⁻¹ from the soil surface to the collector in the middle of the profile and -71-4.9 ng m⁻² h⁻¹ from the middle to the topmost collector.

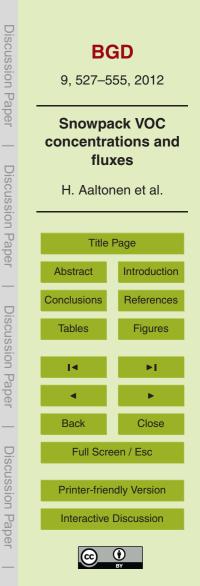


4 Discussion

4.1 VOC concentrations

The snowpack VOC concentrations closer to the soil surface were usually higher than those from the snow-atmosphere interface. During the first winter of measurements ⁵ this phenomenon was clear, but not during the second winter. Nor was the assumed linkage evident between the soil temperature, associated with soil biological activity and wintertime biogenic VOC emissions. During the latter colder winter, the snowpack VOC concentrations were always higher, even before the forest damages incurred by the snow load.

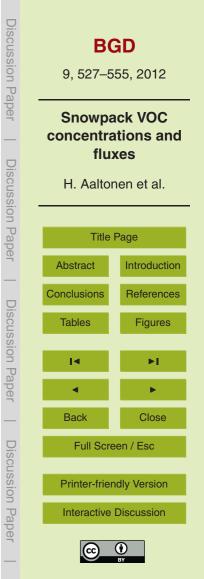
- ¹⁰ When these snowpack VOC concentrations were compared with our belowground VOC measurements (unpublished) conducted during the snow-free period, the magnitudes, excluding the exceptionally high concentration inside the snowpack during late winter 2009–2010, were truly equal. This emphasizes the significance of the snow cover period to the annual VOC emissions from the soil and highlights the role of soil
- as a wintertime VOC source when other biological sources are less active. The Scots pine branch monoterpene emissions peak in early spring and midsummer and are minimal during the midwinter period, whereas the sesquiterpenes are only seen during midsummer in the branch emissions (Tarvainen et al., 2005; Hakola et al., 2006). Pine emissions consist of several compounds, but the most common ones are α -pinene
- ²⁰ and Δ^3 -carene. They were the two most abundant monoterpenes in this study and our summertime measurements at the same site as well (Aaltonen et al., 2011). Both of these compounds have carbon-carbon double bonds and thus form secondary organic aerosols by reactions with ozone (O₃) (Hatfield et al., 2011). Some other monoterpenes (e.g. limonene) and the sesquiterpenes we observed are more susceptible to
- oxidation by O_3 , but their concentrations were in general very low. However, inside the snowpack the O_3 reactions are negligible, and the reactivity with O_3 is not realized before the compound is released to the atmosphere.



In our measurements, the terpenoid concentrations often dropped quickly towards the snow surface, a phenomenon also evident in the snow profile VOC measurements done by Helmig et al. (2009) for compounds originating from the soil. The terpenoid concentrations in the air were often clearly lower than the concentrations inside the snowpack. The sharply dropping concentrations close to the snow surface were probably due to the different diffusion constants of the snow surface and the air, resulting in more efficient transport between the snow and air close to the snow surface than in the deeper layers.

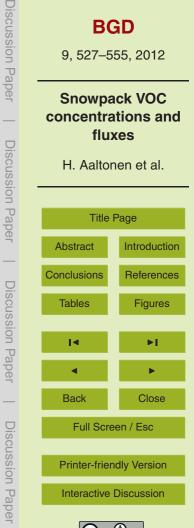
4.2 Sources and sinks of snowpack VOCs

- The clearly decreasing VOC concentrations from the ground level towards the snow surface during most of the measurement periods suggest that the VOC source was located either below ground or on the soil surface (litter). One probable source of the VOCs is the active decomposition below the snowpack, which is supported by the finding of Kähkönen et al. (2001) that the decomposition process does not cease even slightly at below-zero temperatures. Bowling et al. (2009) also found clear evidence of microbial activity below the snowpack by measuring similar stable carbon isotope compositions (δ¹³C) of soil respiration during winter and summer. Some terpenoids, however, showed contrasting trends in concentrations, i.e. the concentrations were al-
- ways higher in the upper snow layers. Compounds having clearly the highest concentrations in the uppermost collector included 1,8-cineol and linalool, whose emissions are strongly light-dependent and thus originate from photosynthetic tissues (Staudt et al., 1997; Tarvainen et al., 2005). Another example of compounds with a top-to-bottom flux is nopinone, which is an oxidation product of β -pinene (Holzinger et al., 2005). O₃ concentrations are low inside the snowpack (Zeller and Hehn, 1995), while OH may be
- ²⁵ more abundant, at least under sunny conditions (Anastasio et al., 2007; Beyersdorf et al., 2007). However, light levels on boreal forest floors during winter are very low. Thus, for oxidative reactions the snow-air interface, including the top layers of the snowpack, is more important than the deeper layers of the snowpack.



Snowpack concentrations of organic compounds are not affected only by soil-living microbes, but also by microbes living inside the snowpack. Several studies (e.g. Amato et al., 2007; Ariya et al., 2011) have reported various microbial groups and species living and actively functioning in snow. These microbes – algae, fungi and bacteria – may
 ⁵ either produce or use VOCs, or affect the chemical reactions in which the VOCs participate. During winter and especially during spring, dissolved organic carbon (DOC) from trees accumulates inside and on the snowpack. DOC itself, as well as DOC decomposed by snow-living microbes, may act as notable sources of snowpack VOC fluxes in spring. However, this is a totally unidentified VOC source and needs further studies.

- Heavy snow loading at the end of the second winter caused extensive forest damage at the study site. Treetops, branches and needles fell, as did some pine tops partly or fully in two profiles. This was likely the reason for the dramatic increase in snow monoterpene concentrations observed in March 2010. Damaged parts of trees, especially needles and young branches, may emit high amounts of terpenoids from the
- ¹⁵ abundant reservoirs in needles and woody tissue (Staudt et al., 1997; Ghirardo et al., 2010). Haapanala et al. (2011) measured clear increases in monoterpene emissions from stumps and logging residue after timber felling, the sesquiterpene and isoprene emissions remaining at low levels, supporting our assumption of forest damages being the main reason for the dramatic increase in snowpack monoterpene concentrations.
- The proportions of different monoterpenes inside the snowpack changed concomitantly with the increased total concentrations, suggesting a sudden change in the VOC source, most probably from "normal" litter decomposition or biological sources in the soil to damaged tree organs. Most likely the additional VOCs originated from physical processes (crushing of the needles and branches and evaporation of easily volatile
- ²⁵ compounds from the plant tissue) and not from biological decomposition processes, which could not play such a significant role in winter when the temperature of the snowpack was almost constantly below 0°C. In the collector that was left undamaged, the magnitude of the concentrations remained similar as before. Concomitantly with the increase in VOC concentrations in the snowpack, the highest concentrations were



no longer measured from the ground level, but instead in the middle or top collectors. In addition to the change in VOC source from soil to fallen tree litter, the profile structures may have been harmed by falling branches, i.e. it is possible that the collectors were shifted from their initially installed places closer to each other. Nevertheless, the

⁵ concentrations in the damaged collectors were far above those measured from the soil surface (Aaltonen et al., 2011) or from inside the soil (unpublished). Thus, inevitably the concentration increase after the snow damage can be traced to fallen branches and needles. This stresses the importance of occasional damage periods, such as windthrows, forestry operations and snow damage, to the stand-level VOC flux.

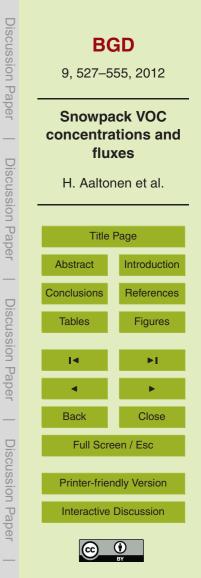
10 4.3 VOC fluxes

Since the monoterpene concentrations in ambient air (indicated by values from the topmost collectors which were above the snow surface) were quite low in winter 2008–2009, the fluxes were upwards throughout the season. The extensive forest damages in midwinter 2009–2010 apparently increased the monoterpene concentrations in the air over the SMEAR II forest, and in the last two measurements the snowpack may have acted as a sink for these compounds. Since the sources of the various VOCs are unknown, it is difficult to estimate whether the snow properties more determine the partitioning and movements of some compounds. However, the sesquiterpene concentrations and estimated fluxes were less affected by the forest damages than the

²⁰ monoterpenes. Moreover, in addition to the possible different sources of these compounds, the snow physical properties may have also played a role in this phenomenon.

These wintertime monoterpene fluxes measured inside the pine forest snowpack at the SMEAR II station were on average at least one magnitude lower than those we measured with chambers during a snow-free period (Aaltonen et al., 2011). The

25 sesquiterpene fluxes measured from the snowpack were more or less similar to those measured by Aaltonen et al. (2011) during the snow-free time, since those fluxes are also low in summer. While the reactions of the VOCs inside the snowpack are poorly understood and thus omitted from our flux calculations, the calculated fluxes must be



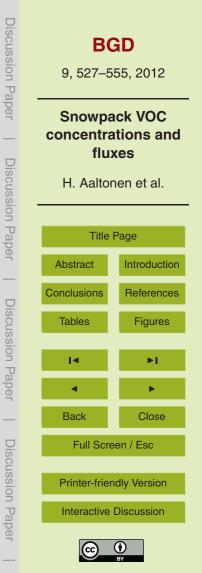
considered more as estimates than exact values. However, as discussed already in chapter 4.1, the concentrations of the main oxidants, OH and O_3 , inside the snowpack are presumably low, except for the surface layer of the snowpack (Zeller and Hehn, 1995; Anastasio et al., 2007; Beyersdorf et al., 2007), as are the temperature and light levels, all of which reduce the reactivity of VOCs.

4.4 Importance of snow physical properties

5

The physical properties of snow may have substantial effects on the diffusion of volatiles in the snowpack. The density of snow is one of the most important factors controlling gas transport in the snowpack (Seok et al., 2009) and fortunately quite easy to measure. Vertical movement of gases may be blocked by the ice layers formed by freezing after periods of above-zero temperatures. The second winter of our measurements included a long period with temperatures constantly below zero, and thus the snowpack presumably was mostly homogenous, soft snow. The SNOWPACK model runs resulted in over 50 % air volume fractions for the snowpack during all the mea-

- ¹⁵ surement days in both winters and also for cases when the snow grain type modelled showed ice formation inside the snowpack. Thus, the ice layers would probably never have fully stopped the gas transport inside the snowpack during our measurements. However, the possible thick ice layers inside the snowpack may have affected the distribution of gas fluxes by causing horizontal gas movements and the release of gases into
- the atmosphere through occasional holes, such as along tree trunks. Climate change is predicted to increase wintertime temperatures at high latitudes more than during other seasons, which could make the snow consistency less permeable. This type of change will stresses the importance of springtime to forest floor VOC emissions even more (see Hellén et al., 2006; Aaltonen et al., 2011), because the blocked compounds
- are rapidly released during and after snowmelt. The spring peak in aerosol particle formation events (Dal Maso et al., 2005) will probably also increase, if the springtime VOC burst during snowmelt will become more intense.



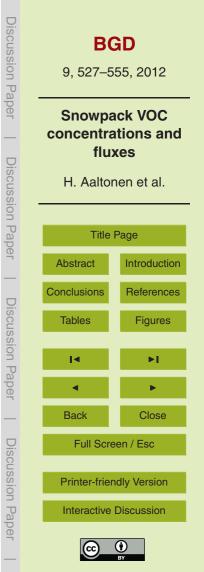
The physical properties of the snowpack can also be affected by biological functioning. Ariya et al. (2011) observed that microbes living in the snow may significantly influence snow morphology by making cavities and thus increasing the specific snow surface area. The cavities Ariya et al. (2011) observed in snow crystals were ~2 μ m in diameter and most probably made by fungal hyphae or by bacterial filaments. For reactive gases the increased specific snow surface area means further opportunities for chemical reactions.

5 Conclusions

5

Snowpacks in boreal pine forests contain substantial concentrations of terpenoids, especially monoterpenes. In a pristine snowpack, these concentrations decrease towards the snow surface, suggesting that soil is the source for terpenoids. When the activity of plant roots is low, the most probable actual source is decomposition of litter. The hard winter during the measurement period caused substantial forest damage and changes in snowpack monoterpene concentrations and distributions. Both the radically

- increased monoterpene concentrations as well as the change observed in compound proportions suggest that the crushed needles and fallen branches and trees can act as strong sources, even when they are frozen. Even though biological and physical factors are difficult to separate out, it is clear that snowpack physical properties play a role in the diffusion efficiency of gas fluxes.
- This study shows that decomposition processes in the soil are active and also an efficient VOC source during winter and that natural or human disturbances can cause high VOC emissions from nonactive biomass. Our results stress the importance of the soil as a source of VOCs throughout the year, and especially during the season when the activities of other biological sources have lowered and the atmospheric lifetime of
- ²⁵ VOCs is extended. These new results of forest floor wintertime VOC emissions can be used for modelling the ecosystem-level total VOC emissions more accurately.



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tre of Excellence programme CRAICC. Financial support by the EU projects NitroEurope and 5 Infrastructure for Measurements of the European Carbon Cycle (IMECC) is gratefully acknowledged.

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Discussion Paper

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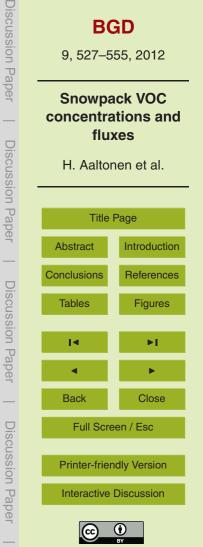
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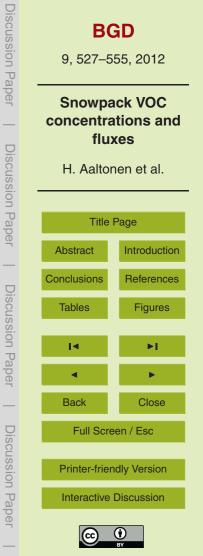
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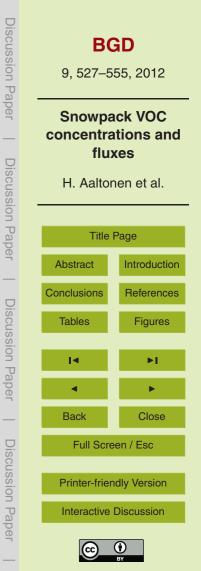
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Table 1. Sampling dates and respective air temperatures (min-max), snowpack depths and snow water equivalents during winters 2008–2009 and 2009–2010.

Sampling dates	Air temperature °C	Snowpack depth cm	Snow water equivalent mm
27 November 2008	-3.7 to 4.0	8	12
5 February 2009	-5.8 to -3.3	17	34
3 March 2009	-6.4 to -1.3	27	63
1 April 2009	0.3 to 5.5	25	67
12 January 2010	-10.8 to -4.5	29	39
2 February 2010	-8.4 to -8.0	35	57
1 March 2010	-1.4 to -0.3	53	110
24 March 2010	-2.3 to 2.3	55	142

Table 2. Average concentrations of the terpenoids observed in different snowpack profiles during winters 2008–2009 and 2009–2010.

		2008–2009			2009–2010	
$ng m^{-3}$	0 cm	15 cm	30 cm	0 cm	15 cm	30 cm
Hemiterpenoids						
isoprene	82	24	11	17	400	12
methyl butenol	20	10	110	13	15	13
Monoterpenes						
<i>α</i> −pinene	1700	130	74	2800	2800	1400
β –pinene	35	6,0	18	140	210	190
bornylacetate	130	40	24	93	29	78
camphene	550	49	15	640	530	200
Δ^3 -carene	660	85	39	1900	2300	1400
limonene	74	75	63	80	190	320
linalool	9,3			20		25
nopinone	9,4	9,0	12	8,8	19	21
p-cymene	12	7,7	9,5	23	30	33
terpinolene	41	6,5	9,2	26	64	50
1,8-cineol	2,3	3,6	34		7,6	12
Total monoterpenes	3224	412	299	5731	6180	3728
Sesquiterpenes						
α -humulene	72	42	21	100	11	120
alloaromadendrene/farnesene	73	27	16	75	12	71
aromadendrene	140	25	14	89	7,1	90
β –caryophyllene	87	24	17	93	35	54
iso-longifolene	100	25	5,5	96	7,6	62
longicyclene	69	19	9,5	74	11	71
longifolene	480	100	25			
Total sesquiterpenes	1021	262	108	527	84	467

BGD 9, 527-555, 2012 **Snowpack VOC** concentrations and fluxes H. Aaltonen et al. **Title Page** Introduction Abstract Conclusions References Tables Figures 14 < Back Close Full Screen / Esc **Printer-friendly Version** Interactive Discussion

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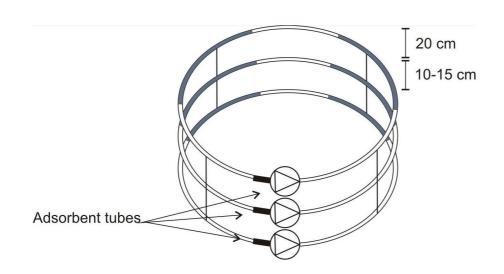
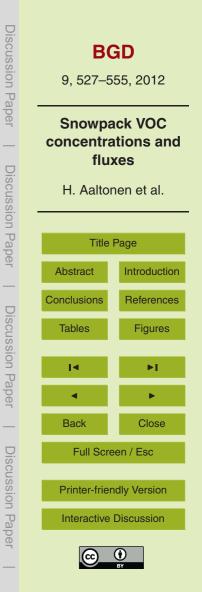


Fig. 1. Schematic figure of VOC snow profile. Grey areas represent perforated sectors.



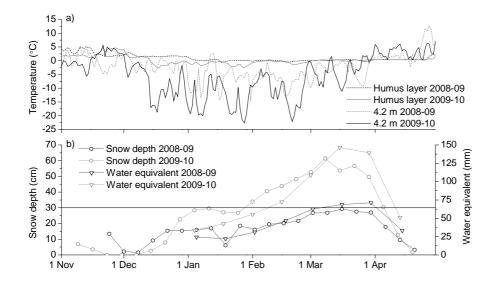


Fig. 2. Temperature, snow depth and snow water equivalent at SMEAR II station during winters 2008–2009 and 2009–2010. (a) temperature in humus layer and at 4.2 m height, (b) snow depth and snow water equivalent. Horizontal line in panel b) shows the height of the uppermost collectors.

