

1 **Ecological Controls on N₂O Emission in Surface Litter and Near-surface**
2 **Soil of a Managed Grassland: Modelling and Measurements**

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ABSTRACT

9 Large variability in N₂O emissions from managed grasslands may occur because most emissions
10 originate in surface litter or near-surface soil where variability in soil water content (θ) and temperature
11 (T_s) is greatest. To determine whether temporal variability in θ and T_s of surface litter and near-surface
12 soil could explain that in N₂O emissions, a simulation experiment was conducted with *ecosys*, a
13 comprehensive mathematical model of terrestrial ecosystems in which processes governing N₂O
14 emissions were represented at high temporal and spatial resolution. Model performance was verified by
15 comparing N₂O emissions, CO₂ and energy exchange, and θ and T_s modelled by *ecosys* with those
16 measured by automated chambers, eddy covariance (EC) and soil sensors at an hourly time-scale during
17 several emission events from 2004 to 2009 in an intensively managed pasture at Oensingen,
18 Switzerland. Both modelled and measured events were induced by precipitation following harvesting
19 and subsequent fertilizing or manuring. These events were brief (2 – 5 days) with maximum N₂O
20 effluxes that varied from $< 1 \text{ mg N m}^{-2} \text{ h}^{-1}$ in early spring and autumn to $> 3 \text{ mg N m}^{-2} \text{ h}^{-1}$ in summer.
21 Only very small emissions were modelled or measured outside these events. In the model, emissions
22 were generated almost entirely in surface litter or near-surface (0 – 2 cm) soil, at rates driven by N
23 availability with fertilization vs. N uptake with grassland regrowth, and by O₂ supply controlled by litter
24 and soil wetting relative to O₂ demand from microbial respiration. In the model, NO_x availability
25 relative to O₂ limitation governed both the reduction of more oxidized electron acceptors to N₂O and the
26 reduction of N₂O to N₂, so that the magnitude of N₂O emissions was not simply related to surface and

27 near-surface θ and T_s . Modelled N_2O emissions were found to be sensitive to defoliation intensity and
28 timing which controlled plant N uptake and soil θ and T_s prior to and during emission events. Reducing
29 LAI remaining after defoliation to one-half that under current practice and delaying harvesting by 5 days
30 raised modelled N_2O emissions by as much as 80% during subsequent events and by an average of 43%
31 annually. Modelled N_2O emissions were also found to be sensitive to surface soil properties. Increasing
32 near-surface bulk density by 10% raised N_2O emissions by as much as 100% during emission events and
33 by an average of 23% annually. Relatively small spatial variation in management practices and soil
34 surface properties could therefore cause the large spatial variation in N_2O emissions commonly found in
35 field studies. The global warming potential from annual N_2O emissions in this intensively managed
36 grassland largely offset those from net C uptake in both modelled and field experiments. However
37 model results indicated that this offset could be adversely affected by suboptimal land management and
38 soil properties.

39

40

INTRODUCTION

41 The contribution of managed grasslands to reducing atmospheric greenhouse gas (GHG)
42 concentrations through net uptake of CO_2 (Ammann et al., 2005) may be at least partially offset by net
43 emissions of N_2O (Conant et al., 2005, Flécharde et al., 2005). These emissions may be substantial, with
44 N_2O emission factors of as large as 3% measured in intensively managed grasslands with fertilizer rates
45 of 25 - 30 g N m⁻² y⁻¹ (Imer et al., 2013; Rafique et al., 2011) These emissions are highly variable
46 temporally and spatially because they are determined by complex interactions among short-term weather
47 events (warming, precipitation), land management practices (N amendments, defoliation), and soil
48 properties (e.g. bulk density, water retention). The N_2O driving these emissions in managed grasslands is
49 thought to be generated within the upper 2 cm of the soil profile (van der Weerden et al., 2013) and in
50 surface litter left by grazing or harvesting (Pal et al., 2013) so that diurnal heating and precipitation
51 events that cause rapid warming and wetting of the litter and soil surface may cause large but brief
52 emission events. These events are thought to be driven by increased demand for electron acceptors by
53 nitrification and denitrification, and reduced supply of O_2 by which these demands are preferentially
54 met, and therefore increased demand for alternative acceptors NO_3^- , NO_2^- and N_2O by autotrophic
55 nitrifiers and heterotrophic denitrifiers.

56 The magnitude of N₂O emission events in managed grasslands generally increases with the
57 amount of N added as urine, manure or fertilizer, and with the intensity of defoliation by grazing or
58 cutting (Ruzjerez et al. 1994). Thus Imer et al. (2013) found a negative correlation between LAI and
59 N₂O emissions at intensively managed grasslands in Switzerland. The increase in emissions with
60 defoliation has been attributed to increased urine and manure deposition and soil compaction with
61 defoliation by grazing, and to slower uptake of N and water by slower-growing plants with defoliation
62 by harvesting (Jackson et al., 2015). Both N additions and defoliation are thought to raise these
63 emissions by increasing the supply of NH₄⁺ and NO₃⁻ to autotrophic nitrifiers and heterotrophic
64 denitrifiers. This increase raises the demand for alternative e⁻ acceptors by these microbial populations if
65 the supply of O₂, the preferred e⁻ acceptor, fails to meet demand, as may occur when soil water content
66 (θ) after defoliation rises with precipitation or reduced transpiration. This supply is governed by physical
67 and hydrological properties (porosity, water retention) of the near-surface soil. Consequently land use
68 practices and soil properties must be considered when estimating N₂O emissions from managed
69 grasslands.

70 Recognition of the effects of precipitation events, N amendments and soil properties on N₂O
71 emissions has led to empirical models in which annual emission inventories are calculated directly from
72 annual precipitation and N inputs (Lu et al., 2006), or monthly emission events are calculated from
73 monthly precipitation, air temperature T_a , and θ (Flécharde et al., 2007). However the soil depth at which
74 most emitted N₂O is generated (0 – 2 cm) is much shallower than that at which θ used in these models is
75 measured (5 – 10 cm) (Flécharde et al., 2007), and the soil temperature T_s at this depth may differ from T_a
76 This is particularly so for grasslands in which N additions are necessarily left on the soil surface without
77 incorporation. Thus large N₂O emissions may be caused by surface wetting from precipitation on dry
78 soils following fertilizer application, so that deeper θ is sometimes found to be of little explanatory value
79 in empirical models (Flécharde et al., 2007). Furthermore the response of denitrification to θ has been
80 found in experimental studies to rise sharply with T_s , likely through the combined effects of T_s on
81 increasing demand and reducing supply of O₂ at microbial microsites (Craswell, 1978). The interaction
82 between T_s and θ on N₂O emissions is clearly apparent in the meta-analysis of N₂O emissions from
83 European grasslands by Flécharde et al. (2007). This interaction has been represented in empirical models
84 by fitting interdependent threshold values of T_s and θ above which emissions have been measured in
85 field experiments (Smith and Massheder, 2014). However a more robust simulation of this interaction on

86 N₂O emissions should be built from basic biological and physical processes that are independent of site-
87 specific measurements.

88 Process models used to simulate N₂O emissions from managed grasslands must therefore
89 explicitly represent the effects of short-term weather events on near-surface T_s and θ , as well as the
90 effects of N additions and defoliation on near-surface NH_4^+ and NO_3^- . These models must also
91 explicitly represent the effects of mineral N, T_s and θ , and of soil physical and hydrological properties,
92 on the demand for vs. supply of O₂ and alternative e⁻ acceptors NO_3^- , NO_2^- and N₂O, and on the
93 oxidation-reduction reactions by which these e⁻ acceptors are reduced. However earlier process models
94 have usually simulated N₂O emissions as T_s -dependent functions of nitrification and denitrification
95 rates, modified by texture-dependent functions of water-filled pore space (WFPS) (e.g. Li et al., 2005).
96 In some models additional empirical functions of T_s (Chatskikh et al., 2005), or of T_s and WFPS
97 (Schmid et al., 2001), are used to calculate the fraction of nitrification that generates N₂O, and the
98 fraction of heterotrophic respiration R_h that drives denitrification (Schmid et al., 2001), thereby avoiding
99 the explicit simulation of O₂ and its control on N₂O emissions. A more detailed summary of functions of
100 mineral N, T_s and WFPS currently used to model N₂O emissions is given in Fang et al. (2015).
101 These functions have many model-dependent parameters and function independently of each other, so
102 that key interactions among reduced C and N substrates, T_s and θ on N₂O production may not be
103 simulated. In none of these approaches are the oxidation-reduction reactions by which N₂O is generated
104 or consumed explicitly represented. Furthermore the effects of defoliation and surface litter on N₂O
105 emissions have not been considered in earlier process models.

106 Process models used to simulate N₂O emissions must also accurately represent the key processes
107 of C cycling that drive those of N cycling from which N₂O is generated and consumed. These include
108 gross and net primary productivity (GPP and NPP) which drive mineral N uptake and assimilation with
109 plant growth. GPP and consequent plant growth also drive autotrophic respiration (R_a), the below-
110 ground component of which contributes to soil O₂ demand. NPP drives litterfall and root exudation,
111 which in turn drive heterotrophic respiration (R_h) that also contributes to litter and soil O₂ demand, and
112 thereby to demand for alternative e⁻ acceptors which drive N₂O generation. Heterotrophic respiration
113 also drives key N transformations such as mineralization/immobilization, thereby controlling availability
114 of these alternative e⁻ acceptors. Land use practices such as defoliation from grazing or harvesting, and

115 soil properties such as porosity and water retention, alter these key C cycling processes, and thereby
116 N₂O emissions. Therefore these emissions are best simulated by comprehensive ecosystem models.

117 In the mathematical model *ecosys*, the effects of weather and N amendments on T_s , θ , and
118 mineral N, and hence on the demand for vs. supply of O₂, NO₃⁻, NO₂⁻ and N₂O, and thereby on N₂O
119 emissions, are simulated by explicitly coupling the transport processes with the oxidation – reduction
120 reactions by which these e⁻ acceptors are known to be generated, transported and consumed in soils
121 (Grant and Pattey, 1999, 2003, 2008; Grant et al., 2006; Metivier et al., 2009). The development of
122 model algorithms for these processes was guided by two key principles:

- 123 (1) all algorithms in the model must represent physical, biochemical and biological processes
124 studied in basic research programs (e.g. convective-diffusive transport, oxidation-reduction
125 reactions) so that these algorithms can be parameterized independently of the model
- 126 (2) this parameterization must be conducted at spatial and temporal scales smaller than those of
127 prediction (in this case seasonal N₂O fluxes) so that site-specific effects on predicted values are
128 not incorporated into the algorithms, limiting their robustness.

129 These principles are designed to avoid as much as possible the use of site- and model-specific
130 algorithms that may lack application in sites and models other than those for which they were
131 developed. Although models based on these principles appear complex, they can be better constrained
132 than simpler models because they are parameterized from independent experiments. The resulting detail
133 that application of these principles brings to the model enables better constrained tests of model output
134 against more comprehensive and diverse site data than are possible with simpler models.

135 In an extension of earlier work with *ecosys*, we propose that temporal and spatial variation in
136 N₂O emissions from an intensively managed grassland can be largely explained from the modelled
137 effects of N amendments (fertilizer, manure), plant management (e.g. harvest intensity and timing), soil
138 properties (e.g. bulk density) and weather (T_s , precipitation events) on the demand for vs. supply of O₂,
139 NO₃⁻, NO₂⁻ and N₂O in surface litter and near-surface soil (0 – 2 cm). Testing this explanation requires
140 frequent measurements to characterize the large temporal variation in N₂O emissions found in managed
141 ecosystems. Such measurements were recorded from 2004 to 2009 using automated chambers in

142 intensively managed grass-clover grassland at Oensingen, Switzerland, and used here to test our
143 modelled explanation of these fluxes.

144

145 **MODEL DEVELOPMENT**

146

147 **General Overview**

148 The hypotheses for N₂O oxidation-reduction reactions and their coupling with gas transport in
149 *ecosys* are represented in Fig. 1 and described further below with reference to equations and definitions
150 listed in Appendices A, C, D, E, H of the Supplement (indicated by square brackets in the text below,
151 e.g. [H1] refers to Eq. 1 in Appendix H), as well as in earlier papers (Grant and Pattey, 1999, 2003,
152 2008; Grant et al., 2006; Metivier et al., 2009). These hypotheses are part of a larger model of soil C, N
153 and P transformations (Grant et al., 1993a,b), coupled to one of soil water, heat and solute transport in
154 surface litter and soil layers, which are in turn components of the comprehensive ecosystem model
155 *ecosys* (Grant, 2001).

156

157 **Mineralization and Immobilization of Ammonium by All Microbial Populations**

158 Heterotrophic microbial populations m (obligately aerobic bacteria, obligately aerobic fungi,
159 facultatively anaerobic denitrifiers, anaerobic fermenters, acetotrophic methanogens, and obligately
160 aerobic and anaerobic non-symbiotic diazotrophs) are associated with each organic substrate i ($i =$
161 animal manure, coarse woody plant residue, fine non-woody plant residue, particulate organic matter, or
162 humus). Autotrophic microbial populations n (aerobic NH₄⁺ and NO₂⁻ oxidizers, hydrogenotrophic
163 methanogens and methanotrophs) are associated with inorganic substrates. These populations grow with
164 energy generated from coupled oxidation of reduced dissolved C (DOC) by heterotrophs, or of mineral
165 N (NH₄⁺ and NO₂⁻) by nitrifiers, and reduction of e- acceptors O₂ and NO_x. These populations decay
166 according to first-order rate constants with provision for internal recycling of limiting nutrients (N, P).
167 During growth, each functional component j ($j =$ nonstructural, labile, resistant) of these populations
168 seeks to maintain a set C:N ratio by mineralizing NH₄⁺ ([H1a]) from, or by immobilizing NH₄⁺ ([H1b])
169 or NO₃⁻ ([H1c]) to, microbial nonstructural N. Nitrogen limitations during growth may cause C:N ratios
170 to rise above set values, and greater recovery of microbial N from structural to nonstructural forms to

171 reduce N loss during decay, but at a cost to microbial function. These transformations control the
 172 exchange of N between organic and inorganic states, and hence affect the availability of alternative e^-
 173 acceptors for nitrification and denitrification.

174

175 **Oxidation of DOC and Reduction of Oxygen by Heterotrophs**

176 Constraints on heterotrophic oxidation of DOC imposed by O_2 uptake are solved in four steps:

- 177 1) DOC oxidation under non-limiting O_2 is calculated from active biomass, DOC concentration, and an
 178 Arrhenius function of T_s [H2],
- 179 2) O_2 reduction to H_2O under non-limiting O_2 (O_2 demand) is calculated from 1) using a set respiratory
 180 quotient [H3],
- 181 3) O_2 reduction to H_2O under ambient O_2 is calculated from radial O_2 diffusion through water films of
 182 thickness determined by soil water potential [H4a] coupled with active uptake at heterotroph surfaces
 183 driven by 2) [H4b]. O_2 diffusion and active uptake is calculated for each heterotrophic population
 184 associated with each organic substrate, allowing [H4] to calculate lower O_2 concentrations at
 185 microbial surfaces associated with more biologically active substrates (e.g. manure, litter). Localized
 186 zones of low O_2 concentration (hotspots) are thereby simulated when O_2 uptake by any aerobic
 187 population is constrained by O_2 diffusion to that population. O_2 uptake by each heterotrophic
 188 population also accounts for competition for O_2 uptake with other heterotrophs, nitrifiers, roots and
 189 mycorrhizae, calculated from its O_2 demand relative to those of other aerobic populations.
- 190 4) DOC oxidation to CO_2 under ambient O_2 is calculated from 2) and 3) [H5]. The energy yield of DOC
 191 oxidation drives the uptake of additional DOC for construction of microbial biomass $M_{i,h}$ according to
 192 construction energy costs of each heterotrophic population [A21]. Energy costs of denitrifiers are
 193 larger than those of obligately aerobic heterotrophs, placing denitrifiers at a competitive disadvantage
 194 for growth and hence DOC oxidation that declines with greater use of e^- acceptors other than O_2 .

195

196 **Oxidation of DOC and Reduction of Nitrate, Nitrite and Nitrous Oxide by Denitrifiers**

197 Constraints imposed by NO_3^- availability on DOC oxidation by denitrifiers are solved in five
 198 steps:

- 199 1) NO_3^- reduction to NO_2^- under non-limiting NO_3^- is calculated from electrons demanded by DOC
 200 oxidation to CO_2 but met by O_2 reduction to H_2O because of diffusion limitations to O_2 supply, and
 201 hence transferred to NO_3^- [H6],

- 202 2) NO_3^- reduction to NO_2^- under ambient NO_3^- is calculated from 1), accounting for relative
 203 concentrations and affinities of NO_3^- and NO_2^- [H7],
- 204 3) NO_2^- reduction to N_2O under ambient NO_2^- is calculated from demand for electrons not met by NO_3^-
 205 reduction in 2), accounting for relative concentrations and affinities of NO_2^- and N_2O [H8],
- 206 4) N_2O reduction to N_2 under ambient N_2O is calculated from demand for electrons not met by NO_2^-
 207 reduction in 3) [H9],
- 208 5) additional DOC oxidation to CO_2 enabled by NO_x reduction in 2), 3) and 4) is added to that enabled
 209 by O_2 reduction from [H5], the energy yield of which drives additional DOC uptake for construction
 210 of $M_{i,n}$. This additional uptake offsets the disadvantage incurred by the larger construction energy
 211 costs of denitrifiers.

212

213 **Oxidation of Ammonia and Reduction of Oxygen by Nitrifiers**

214 Constraints on nitrifier oxidation of NH_3 imposed by O_2 uptake are solved in four steps:

- 215 1) substrate (NH_3) oxidation under non-limiting O_2 is calculated from active biomass, NH_3 and CO_2
 216 concentrations, and an Arrhenius function of T_s [H11],
- 217 2) O_2 reduction to H_2O under non-limiting O_2 is calculated from 1) using set respiratory quotients [H12],
- 218 3) O_2 reduction to H_2O under ambient O_2 is calculated from radial O_2 diffusion through water films of
 219 thickness determined by soil water potential [H13a] coupled with active uptake at nitrifier surfaces
 220 driven by 2) [H13b]. O_2 uptake by nitrifiers also accounts for competition for O_2 uptake with
 221 heterotrophic DOC oxidizers, roots and mycorrhizae,
- 222 4) NH_3 oxidation to NO_2^- under ambient O_2 is calculated from 2) and 3) [H14]. The energy yield of NH_3
 223 oxidation drives the fixation of CO_2 for construction of microbial biomass $M_{i,n}$ according to
 224 construction energy costs of nitrifier populations.

225

226 **Oxidation of Nitrite and Reduction of Oxygen by Nitrifiers**

227 Constraints on nitrifier oxidation of NO_2^- to NO_3^- imposed by O_2 uptake [H15 - H18] are solved
 228 in the same way as are those of NH_3 [H11 - H14]. The energy yield of NO_2^- oxidation drives the fixation
 229 of CO_2 for construction of microbial biomass $M_{i,o}$ according to construction energy costs of each nitrifier
 230 population.

231

Oxidation of Ammonia and Reduction of Nitrite by Nitrifiers

Constraints on nitrifier oxidation of NH_3 imposed by NO_2^- availability are solved in three steps:

- 1) NO_2^- reduction to N_2O under non-limiting NO_2^- is calculated from electrons demanded by NH_3 oxidation but not accepted for O_2 reduction to H_2O because of diffusion limitations to O_2 supply, and hence transferred to NO_2^- [H19],
- 2) NO_2^- reduction to N_2O under ambient NO_2^- and CO_2 is calculated from 1) [H20], competing for NO_2^- with denitrifiers [H8] and nitrifiers [H18],
- 3) additional NH_3 oxidation to NO_2^- enabled by NO_2^- reduction in 2) [H21] is added to that enabled by O_2 reduction from [H14]. The energy yield from this oxidation drives the fixation of additional CO_2 for construction of $M_{i,n}$.

Uptake of Ammonium and Reduction of Oxygen by Roots and Mycorrhizae

- 1) NH_4^+ uptake by roots and mycorrhizae under non-limiting O_2 is calculated from mass flow and radial diffusion between adjacent roots and mycorrhizae [C23a] coupled with active uptake at root and mycorrhizal surfaces [C23b]. Active uptake is subject to inhibition by root nonstructural N:C ratios [C23g] where nonstructural N is the active uptake product, and nonstructural C is the CO_2 fixation product transferred to roots and mycorrhizae from the canopy.
- 2) O_2 reduction to H_2O is calculated from 1) plus oxidation of root and mycorrhizal nonstructural C under non-limiting O_2 using a set respiratory quotient [C14e],
- 3) O_2 reduction to H_2O under ambient O_2 is calculated from mass flow and radial diffusion between adjacent roots and mycorrhizae [C14d] coupled with active uptake at root and mycorrhizal surfaces driven by 2) [C14c]. O_2 uptake by roots and mycorrhizae also accounts for competition with O_2 uptake by heterotrophic DOC oxidizers, and autotrophic nitrifiers, calculated from their O_2 demands relative to those of other populations.
- 4) oxidation of root and mycorrhizal nonstructural C to CO_2 under ambient O_2 is calculated from 2) and 3) [C14b],
- 5) NH_4^+ uptake by roots and mycorrhizae under ambient O_2 is calculated from 1), 2), 3) and 4) [C23b].

Cation Exchange and Ion Pairing of Ammonium

261 A Gapon selectivity coefficient is used to solve cation exchange of NH_4^+ vs. Ca^{2+} [E10] as
262 affected by other cations [E11] – [E15] and CEC [E16]. A solubility product is used to equilibrate
263 soluble NH_4^+ and NH_3 [E24] as affected by pH [E25] and other solutes [E26 – E57].
264

265 **Soil Transport and Surface - Atmosphere Exchange of Gaseous Substrates and Products**

266 Exchange of all modelled gases γ ($\gamma = \text{O}_2, \text{CO}_2, \text{CH}_4, \text{N}_2, \text{N}_2\text{O}, \text{NH}_3$ and H_2) between aqueous
267 and gaseous states is driven by disequilibrium between aqueous and gaseous concentrations according to
268 a T_s -dependent solubility coefficient, constrained by a transfer coefficient based on air-water interfacial
269 area that depends on air-filled porosity [D14 – D15] (Fig. 1). These gases undergo convective-dispersive
270 transport through soil in gaseous [D16] and aqueous [D19] states driven by soil water flux and by gas
271 concentration gradients. Dispersive transport is controlled by gaseous diffusion [D17] and aqueous
272 dispersion [D20] coefficients calculated from gas- and water-filled porosity. Exchange of all gases
273 between the atmosphere and both gaseous and aqueous states at the soil surface are driven by
274 atmosphere - surface gas concentration differences and by boundary layer conductance above the soil
275 surface, calculated from wind speed and from structure of vegetation and surface litter [D15].
276

277 **FIELD EXPERIMENT**

278

279 **Site description**

280 The Oensingen field site is located in the central Swiss lowlands ($7^\circ 44'E, 47^\circ 17'N$) at an altitude
281 of 450 m. The climate is temperate with an average annual rainfall of about 1100 mm and a mean air
282 temperature of 9.5°C . The soil is classified as a Eutri-Stagnic Cambisol developed on clayey alluvial
283 deposits, key properties of which are given in Table 1. Prior to the experiment, the field site was
284 managed as a ley-arable rotation. In December 2000, the field was ploughed and left in fallow until
285 11 May 2001. The field was then sown with a grass-clover mixture typical for permanent grassland
286 under intensive management. The field was ploughed again on 19 December 2007, left in fallow until
287 5 May 2008, when it was tilled and re-sown with the same grass-clover mix as in 2001. The period
288 of study extended from sowing in 2001 to the end of 2009, during which the field was cut between
289 three and five times per year and harvested as hay, silage or fresh grass, fertilized two to three times
290 per year with manure as liquid cattle slurry and two to three times per year with mineral fertilizer as

291 ammonium nitrate (NH_4NO_3) pellets, for an average annual N application of 23 g N m^{-2} . All key
292 management operations during this period are summarized in Table 2.

293

294 **Soil, plant and meteorological measurements**

295 Soil θ and T_s were recorded continuously using TDR (Time Domain Reflectometry, ThetaProbe
296 ML2x, Delta-T Devices, Cambridge, UK) and thermocouples at 5, 10, 30 and 50 cm for θ and at 2, 5,
297 10, 30 and 50 cm for T_s . Leaf area index (LAI) was measured weekly with an optical leaf area meter
298 (LI-2000, Li-Cor, Lincoln, NB, USA). Plants were collected every 2 to 4 weeks and the samples were dried
299 for 48 h at 80°C , weighed and analyzed for C, N, P and K by using an elemental analyzer. Hourly
300 climatic data were recorded continuously with an automated meteorological station, including air
301 temperature ($^\circ\text{C}$), rainfall (mm), relative humidity (%), global radiation (W m^{-2}) and windspeed (m
302 s^{-1}).

303

304 **Nitrous oxide flux measurements**

305 N_2O fluxes were measured with a fully automated system consisting of up to eight stainless steel
306 chambers ($30 \text{ cm} \times 30 \text{ cm} \times 25 \text{ cm}$) (Flechard et al., 2005, Felber et al., 2014) fixed on PVC frames
307 permanently inserted 10-cm deep into the soil. The positions of the chambers were changed about every
308 two months. During measurements, the lids of the chambers were sequentially closed for 15 min. every
309 2 hours to allow N_2O accumulation in the chamber headspace. During closure the chamber atmosphere
310 was recirculated at a rate of 1000 ml min^{-1} through polyamide tube lines (4-mm ID) to analytical
311 instruments installed in a temperature-controlled field cabin adjacent to the field plots (10 m) and then
312 back to the chamber headspace. Until autumn 2006 concentrations of N_2O , CO_2 and H_2O in the head
313 space were measured once per minute with an INNOVA 1312 photoacoustic multi-gas analyzer
314 (INNOVA Air Tech Instruments, Ballerup, Denmark; www.innova.dk). Interferences in the
315 measurements caused by overlaps in the absorption spectra of the different gases and by temperature
316 effects were corrected with a calibration algorithm described in detail by Flechard et al (2005). In
317 autumn 2006 the system was changed to the gas filter correlation technique for N_2O (Model 46C,
318 Thermo 279 Environmental Instruments Inc., Sunnyvale, CA, USA). This system was calibrated every 8
319 hours using certified standard gas mixtures (Messer Schweiz AG, Lenzburg, Switzerland) (Felber et al.
320 2014).

321

322 These measurements were used to calculate N₂O fluxes from the rate of change in concentration
323 by using a linear or non-linear approach determined by the HMR R-package (Pedersen et al., 2010). The
324 first three of the fifteen 1-min. measurements were omitted from the flux calculation to exclude gas
325 exchange during closing that did not result from changes in emission/production in the soil. This
326 procedure caused a mean increase of about 30% in the fluxes compared to values published in Fléchar
327 et al. (2005) and Ammann et al. (2009), which were evaluated using linear regression. Fluxes from all
328 chambers were averaged over 4-hourly intervals and resulting values attributed to the mid-points of the
329 intervals. Standard errors of these averages were calculated from all fluxes measured during each
330 interval, and thus included both spatial and temporal variation. The fluxes measured from 2002 to 2003
331 were summarized in Fléchar et al. (2005). Those from 2004 to 2007 were re-evaluated from values
332 described in Ammann et al. (2009). Those from 2008 and 2009 were reprocessed from the EU-Project
333 NitroEurope-IP database using the HMR algorithm.

334

335 **CO₂ and Energy Flux Measurements**

336 CO₂ and energy fluxes were measured by an eddy covariance (EC) system consisting of three-
337 axis sonic anemometers (models R2 and HS, Gill instruments, Lymington, UK) and an open-path
338 infrared CO₂/H₂O gas analyzer (model LI-7500, Li-Cor, Lincoln, USA). The EC system used in this
339 study is described in Ammann et al. (2007). The EC tower was located in the centre of the field (52m
340 x 146m), whereas the chambers were located in the south east corner. For most meteorological
341 conditions, the chambers were not within the footprint of the EC towers, although for the main wind
342 directions 80% or more of the footprint was within the field (Neftel et al. 2008). The management of
343 the entire field was uniform throughout the experiment.

344

345 **MODEL EXPERIMENT**

346

347 *Ecosys* was initialized with the biological properties of plant functional types (PFTs)
348 representing the ryegrass and clover planted at Oensingen. These properties were identical to those in an
349 earlier study (Grant et al., 2012) except for a perennial rather than annual growth habit. These PFTs
350 competed for common resources of radiation, water and nutrients, based on their vertical distributions of
351 leaf area and root length driven by uptake and allocation of C, N and P in each PFT. *Ecosys* was also
352 initialized with the physical and chemical properties of the Eutri-Stagnic Cambisol at Oensingen (Table

353 1). The model was then run from model dates 1 Jan. 1931 to 31 Dec. 2000 under repeating sequences of
354 land management practices and continuous hourly weather data (radiation, T_a , RH, wind speed and
355 precipitation) recorded at Oensingen from 1 Jan. 2001 to 31 Dec. 2007 (i.e. 10 cycles of 7 years). This
356 run was long enough for C, N and energy cycles in the model to attain equilibrium under the Oensingen
357 site conditions well before the end of the spinup run. The modelled site was plowed on 19 Dec. 2000,
358 terminating all PFTs.

359 The model run was then continued from model dates 1 Jan. 2001 to 31 Dec. 2009 under
360 continuous hourly weather data recorded at Oensingen from 1 Jan. 2001 to 31 Dec. 2009 with the
361 same PFTs and land management practices as those at the field site listed in Table 2. For each manure
362 application in the model, an irrigation of 4 mm was added to account for the water in the slurry. For
363 each harvest in the model, the fraction of canopy LAI to be cut (usually 0.85 – 0.95) was calculated
364 from measurements of LAI before and after the corresponding harvest in the field. In *ecosys*, leaves of
365 each PFT are aggregated into a common canopy which is dynamically resolved into a selected
366 number of layers (10 in this case) of equal LAI for calculating irradiance interception. The leaf
367 fraction to be cut was removed from successive leaf layers from the top of the combined canopy
368 downwards until the cumulative removal attained the set fraction, so that the LAI cut from each PFT
369 depended on the leaf area of the PFT in these layers. Of the phytomass cut with the LAI, 0.76 was
370 removed as harvest and the remainder was added to surface litter, as determined in the intensively
371 managed grassland at Oensingen by Amman et al. (2009). N₂O emissions modelled from 2004
372 through 2009 were compared with those measured by the automated chambers by regressing log-
373 transformed 4-hour averages of modelled on measured values during each year of the study, and also
374 by regressing total emissions modelled vs. measured during emission events following each fertilizer
375 or manure application. These comparisons were supported by ones with thermistor and TDR
376 measurements of T_s , θ , and with EC measurements of CO₂ and energy exchange.

377

378 **Model Sensitivity Studies**

379 Modelled N₂O emissions may be affected by three general sources of uncertainty in model
380 inputs: land management practices, soil properties and model parameters. To examine the possible
381 effects of some different land management practices on N₂O emissions, the model run from 2001 to
382 2009 (field) was repeated with (1) increased harvest intensity in which canopy LAI remaining after
383 each harvest was reduced to one-half of those in the first run (1/2), and (2) increased harvest intensity

384 with each harvest delayed by 5 days ($1/2 + 5d$). These alternative practices caused canopy regrowth
385 and hence N uptake to be slower during emission events following subsequent manure and fertilizer
386 applications.

387

388 To examine the possible effects of spatial variability in soil properties on N_2O emissions, the
389 model run from 2001 to 2009 (field) was repeated with bulk density (BD) of the upper 3 cm in the soil
390 profile (Table 1) increased by 5% or 10%. These larger BDs reduced soil porosity in the upper 3 cm
391 of the soil, thereby slowing gas exchange with the atmosphere, particularly when the soil was wet
392 (Fig. 1). All other soil properties used in the model remained unchanged (Table 1).

393

394 To examine an effect of uncertainty in model parameterization, the model run from 2001 to
395 2009 (field) was repeated with the values of two key parameters governing N_2O emissions, the
396 Michaelis-Menten constants for reduction of O_2 (K_{O_2} in [H4]) or of NO_3^- and NO_2^- (K_{NO_x} in [H7],
397 [H8] and [H20]), halved or doubled from those used in the model. Halving or doubling K_{O_2} hastened
398 or slowed the reduction of O_2 by nitrifiers and denitrifiers and hence slowed or hastened the transfer
399 of electrons to reduce NO_2^- and NO_3^- during nitrification and denitrification. Halving or doubling
400 K_{NO_x} hastened or slowed the reduction of NO_2^- by nitrifiers and of NO_3^- and NO_2^- by denitrifiers All
401 other parameters in the model remained unchanged.

402

403

404

RESULTS

405

LAI Modelled vs. Measured from 2002 to 2009

407 Accurate modelling of ecosystem C cycling and hence N_2O emissions requires accurate
408 modelling of plant growth as determined by land management practices. LAI modelled and measured
409 from 2002 to 2009 rose rapidly from low values remaining in spring and after each harvest (Table 1)
410 to $4 - 6 \text{ m}^2 \text{ m}^{-2}$ before the next harvest, except during 2003 (Fig. 2). Regrowth of LAI in *ecosys* was
411 driven by plant nonstructural C, N and P pools replenished from storage reserves remobilized after
412 harvests, and from products of current C, N and P uptake, those of C being governed by irradiance
413 interception calculated from regrowing LAI. Regrowth in the model was less rapid than that measured
414 in 2009 (Fig. 2) because more frequent cutting forced more frequent replenishment of plant

415 nonstructural C, N and P pools which gradually depleted storage reserves and hence slowed
416 subsequent regrowth. Hence rates of regrowth modelled after harvests were affected by harvest timing
417 and intensity, as represented by the fractions of LAI removed at harvest.

418

419 **N₂O Fluxes Modelled vs. Measured from 2004 to 2009**

420 During peak emissions, standard deviations of N₂O fluxes measured within each 4-hourly
421 interval were found to be as much as 85% relative to mean values. These deviations were largely
422 attributed to small-scale spatial variation in land management (manure and fertilizer application,
423 surface litter from harvesting) and in soil properties (bulk density, water retention), which was not
424 represented in the model run, rather than to temporal variation in environmental conditions (θ , T_s)
425 which was represented in the model run. Therefore only a limited fraction of variation in the
426 measured values was amenable to correlation with modelled values. Consequently slopes and
427 coefficients of determination (R^2) from regressions of modelled on measured log-transformed fluxes
428 varied from 0.5 to 1.0 and from 0.1 to 0.5 respectively, while intercepts remained close to zero (Table
429 3a). However ratios of mean squares for regression vs. error (F) were highly significant ($P < 0.001$) in
430 all years of the study, indicating some agreement in the timing and magnitude of modelled and
431 measured emission events. Improved agreement would require that more detailed information about
432 land management and soil properties at each chamber site be provided to the model.

433

434 **Daily-Aggregated N₂O Fluxes Modelled vs. Measured from 2004 to 2009**

435 Daily aggregations of both measured and modelled N₂O emissions indicated that emission
436 events during the study period were confined to intervals of no longer than 5 days when precipitation
437 followed manure or fertilizer applications (Fig. 3). Outside of these intervals emissions remained very
438 small except for a period of emissions modelled, but not measured, after manure application in
439 autumn 2006 (Fig. 3c) and measured, but not modelled, before fertilizer application in spring 2008
440 (Fig. 3e).

441

442 The largest emissions followed manure applications in July and August, but their magnitudes
443 did not vary with the amount of manure N applied. For example, emissions during an event in August
444 2009 (244 vs. 185 mg N m⁻² measured vs. modelled in Fig. 3f) were greater than those during an
445 event in July 2007 (86 vs. 112 mg N m⁻² measured vs. modelled in Fig. 3d) which in turn were greater

446 than those during an event in July 2005 (54 vs. 96 mg N m⁻² measured vs. modelled in Fig.2b).
 447 However manure N application preceding the event in August 2009 (4.5 g N m⁻²) was less than that in
 448 July 2007 (6.7 g N m⁻²) which in turn was less than that in July 2005 (8.5 g N m⁻²) (Table 2), so that
 449 smaller applications were followed by greater emissions, precluding a simple emission factor for
 450 manure N application.

451

452 The magnitude of emission events following fertilizer application also varied. For example,
 453 emissions during an event in late August 2007 (105 vs. 82 mg N m⁻² measured vs. modelled in Fig.
 454 3d) were greater than those during events in September 2004 (24 vs. 2 mg N m⁻² measured vs.
 455 modelled in Fig 2a) and 2005 (6 vs. 11 mg N m⁻² measured vs. modelled in Fig. 3b), although the
 456 fertilizer N applications of 3.0 g N m⁻² preceding each event were the same (Table 2). These
 457 differences in emissions indicated important differences in ecological controls imposed by
 458 environmental conditions (θ and T_s) and plant management during each event.

459

460 The standard deviations of ~85% relative to the mean values of fluxes measured within each 4-
 461 hourly interval during emission events was used to estimate an uncertainty in daily-aggregated fluxes
 462 of *ca.* 30%. Uncertainty in daily fluxes measured during emission events was smaller than the several-
 463 fold differences among the events indicating that the magnitude of these events likely differed
 464 significantly. Regressions of modelled on measured total emissions during the events following each
 465 fertilizer or manure application from 2004 to 2009 (Fig. 3) gave better agreement than did those of the
 466 4-hourly averaged fluxes (Table 3b), indicating that modelling the precise timing of fluxes during
 467 these events remains a challenge.

468

469 **Relationships between N₂O Fluxes and Environmental Conditions during Selected Emission** 470 **Events**

471 Environmental conditions measured and modelled from harvest to the end of the two largest
 472 emission events following manure applications in July 2007 (Fig. 3d) and August 2009 (Fig. 3f) were
 473 examined in greater detail to investigate relationships among near-surface T_s , θ , aqueous gas
 474 concentrations, and surface fluxes of energy, CO₂ and N₂O (Figs. 4 and 5). In July 2007, several small
 475 precipitation events wetted and cooled the soil between harvesting on DOY 187 and manure
 476 application on DOY 194 (Fig. 4a,b). The soil then dried during several days without precipitation and

477 warmed with reduced shading from defoliation (Fig. 2) until DOY 200, after which the soil wetted
 478 with further precipitation and cooled with increased shading from plant regrowth (Fig. 4a,b). The
 479 higher θ measured during this period (Fig. 4b) may have been caused by difficulties in maintaining
 480 calibration of the TDR probes over long periods in the high-clay soil at Oensingen (Table 1). This
 481 higher θ was not likely caused by overestimated evapotranspiration because modelled LE fluxes,
 482 reduced by low LAI after harvesting but increasing with subsequent regrowth, were close to those
 483 measured (Fig. 4c), suggesting that total water uptake was accurately modelled. Comparison of
 484 modelled and measured θ was further complicated by soil cracking which altered infiltration at low θ .
 485 The effects of θ -dependent macroporosity on preferential flow are explicitly modelled in *ecosys*, but
 486 have not yet been tested in detail.

487

488 CO_2 influxes were also reduced by low LAI after cutting, but recovered to pre-cut levels by the
 489 end of the emission event (Fig. 4d), driving rapid regrowth of LAI (Fig. 2). Large CO_2 effluxes
 490 measured and modelled after manure application indicated rapid R_h and hence O_2 demand that
 491 persisted for several days.. Influxes measured in the field were reduced from those in the model for
 492 several days after manure application, suggesting temporary interference of CO_2 fixation by the
 493 manure application which was not accounted for in the model.

494

495 Litterfall from plant growth [C18, C19] and cutting, as well as from manure application caused
 496 a litter layer of 1 – 2 cm to develop on the soil surface in the model. During the N_2O emission event
 497 from DOY 200 to DOY 205 in 2007 (Fig. 3d), several precipitation events (Fig. 4a) wetted the
 498 modelled surface litter and near-surface soil (layers 1 and 2 in Table 1) (Fig. 4e) without increasing θ
 499 at 5 cm (Fig. 4b). This surface wetting slowed gas exchange with the atmosphere, sharply reducing
 500 aqueous O_2 concentrations [$\text{O}_{2(s)}$] (Fig. 4f) and thereby raising aqueous N_2O concentrations [$\text{N}_2\text{O}_{(s)}$]
 501 (Fig. 4g). Between precipitation events, drying of the surface litter and near-surface soil in the model
 502 allowed recovery of [$\text{O}_{2(s)}$] and forced declines in [$\text{N}_2\text{O}_{(s)}$]. These rises and declines in [$\text{N}_2\text{O}_{(s)}$] drove
 503 rises and declines in N_2O emissions that tracked those measured in the chambers (Fig. 4h). These
 504 emissions rose immediately with the onset of precipitation on DOY 200 (Fig. 4a) before wetting
 505 occurred at 5 cm (Fig. 4b), indicating that emissions were driven by surface wetting (Fig. 4e)
 506 combined with rapid O_2 demand (Fig. 4d). The net generation of N_2O modelled in each soil zone,
 507 calculated from $[\text{H8}] + [\text{H20}] - [\text{H9}]$, indicated that 0.21 of surface emissions originated in the surface

508 litter and the remainder in the 0 – 1 cm soil layer as indicated by higher $[\text{N}_2\text{O}_{(s)}]$ (Fig. 4g), while the
509 deeper soil layers were a very small net sink of N_2O . Rises and declines in $[\text{N}_2\text{O}_{(s)}]$ also drove rises
510 and declines in N_2 emissions that persisted until DOY 205, after which more rapid mineral N uptake
511 with recovering plant growth, driven by rising LAI (Fig. 2) and hence CO_2 influxes (Fig. 4d), caused
512 both emissions to return to background levels (Fig. 4h).

513

514 In 2009, a period of low precipitation with soil drying and warming occurred between
515 harvesting in late July and manure application on DOY 218 in early August, followed by heavy
516 precipitation with soil wetting and cooling on DOY 220 (Fig. 5a,b). LE effluxes and CO_2 influxes
517 declined sharply with LAI after cutting, and did not recover to pre-cut levels by the end of the
518 subsequent emission event on DOY 224 (Fig. 5c,d), indicating a slow recovery of plant growth.
519 Slurry application caused brief surface wetting on DOY 218 (Fig. 5e) and heavy precipitation on
520 DOY 220 caused prolonged soil wetting at the surface (Fig. 5e) and at 5 cm (Fig. 5b). Wetting caused
521 declines in $[\text{O}_{2(s)}]$ (Fig. 5f) and thereby rises in $[\text{N}_2\text{O}_{(s)}]$ (Fig. 5g) that were sustained over 3 days.
522 These rises drove particularly rapid N_2O emissions in the model which were consistent in magnitude
523 with those measured in the chambers (Fig. 5h). Diurnal variation modelled with soil warming and
524 cooling (Fig. 5a) was not apparent in the measurements, although modelled values remained within
525 the large uncertainty of the measured values during the emission event. These large emissions were
526 enabled in the model by slow plant uptake of manure N (Table 2) caused by the slow recovery of
527 plant CO_2 uptake and hence growth after cutting (Fig. 5d). The rises in $[\text{N}_2\text{O}_{(s)}]$ also drove rises in
528 modelled N_2 emissions (Fig. 5h). Emissions declined with surface litter drying on DOY 223 (Fig. 5e)
529 which allowed surface $[\text{O}_{2(s)}]$ to rise (Fig. 5f) and $[\text{N}_2\text{O}_{(s)}]$ to fall (Fig. 5g) while θ at 5 cm remained
530 high (Fig. 5b), again indicating that N_2O emissions were largely determined by ecological controls in
531 the surface litter and soil. The net generation of N_2O modelled in each soil zone indicated that 0.48 of
532 surface emissions originated in the surface litter, 0.48 in the 0 – 1 cm soil layer and 0.05 in the 1 – 3
533 cm soil layer, while the deeper soil layers were a very small net sink of N_2O , as indicated by near-
534 surface gradients of $[\text{N}_2\text{O}_{(s)}]$ (Fig. 5g).

535

536 Greater N_2O emissions were modelled and measured during the event in August 2009 than in
537 July 2007 (Fig. 5h vs. Fig. 4h), in spite of smaller N addition (Fig. 3f vs. Fig. 3d; Table 2) and similar
538 θ and T_s modelled and measured at 5 cm (Fig. 5a,b vs. Fig. 4a,b). These greater emissions were

539 attributed in the model to (1) earlier and heavier precipitation after manure application (2 days after
 540 application in Fig. 5a vs. 6 days in Fig. 4a), and (2) slower recovery of CO₂ fixation after defoliation,
 541 indicated by slower rises in diurnal amplitude of CO₂ fluxes (Fig. 5d vs. Fig. 4d). Heavier
 542 precipitation in 2009 vs. 2007 drove sustained vs. intermittent surface and near-surface wetting (Fig.
 543 5e vs. Fig. 4e) and hence sustained vs. intermittent declines in [O_{2(s)}] and rises in [N₂O_(s)] (Fig. 5f,g
 544 vs. Fig. 4f,g). Slower recovery of CO₂ fixation after cutting in 2009 vs. 2007 slowed removal of added
 545 NH₄⁺ and NO₃⁻ from soil. This slower removal, combined with the shorter period between manure
 546 application and precipitation, left larger NO₃⁻ concentrations ([NO₃⁻]) in litter and surface soil to drive
 547 N₂O production following precipitation [H7]. These model findings indicated the importance to N₂O
 548 emissions of surface and near-surface θ after precipitation, and of plant management (intensity and
 549 timing of defoliation in relation to N application) and its effect on subsequent plant CO₂ fixation and
 550 N uptake.

551

552 **Effects of Intensity and Timing of Defoliation on N₂O Emission Events**

553 Increasing harvest intensity and delaying harvest dates slowed LAI regrowth modelled after
 554 harvests (Fig. 6). The effects of this slowing on N₂O emissions during selected events modelled after
 555 subsequent fertilizer and manure applications were examined under diverse θ and T_s (Figs. 7, 8).
 556 Following manure application on DOY 194 in 2006 (Table 2), slower LAI regrowth from increasing
 557 and delaying defoliation slowed the recovery of CO₂ fixation (Fig. 7a) and of NH₄⁺ uptake (Fig. 7b),
 558 allowing more nitrification of manure N and hence greater surface [NO₃⁻] (Fig. 7c). Slower LAI
 559 regrowth (Fig. 6) also reduced shading and ET, raising T_s (Fig. 7d) and θ (Fig. 7e). N₂O emissions
 560 modelled under field management remained small because of soil drying, in spite of high T_s ,
 561 consistent with measurements (Fig. 3c, Fig. 7f). Increases in emissions modelled with slower LAI
 562 regrowth, particularly from delayed harvesting (Fig. 7f), were attributed to slower N uptake (Fig. 7b)
 563 and hence larger [NO₃⁻] in litter and surface soil (Fig. 7c), and to warmer and wetter soil (Fig. 7d,e)
 564 which increased O₂ demand while reducing O₂ supply.

565

566 Following a similar manure application on DOY 194 in 2007 (Table 2; Fig. 6), slower LAI
 567 regrowth from increasing and delaying defoliation also caused reductions in CO₂ fixation (Fig. 7g),
 568 which slowed NH₄⁺ and NO₃⁻ uptake (Fig. 7h), allowing more nitrification of manure N and hence
 569 greater [NO₃⁻] (Fig. 7i). Lower LAI also caused increases in T_s (Fig. 7j) and θ (Fig. 7k). Emissions

570 modelled and measured under field management in 2007 (Fig. 7l) were greater than those in 2006
 571 (Fig. 7f), in spite of lower T_s (Fig. 7j vs. Fig. 7d), because near-surface wetting from several
 572 precipitation events (Fig. 4a,e) reduced $[O_{2(s)}]$ and increased $[N_2O_{(s)}]$ (Fig. 4f,g). Emissions modelled
 573 with increased and delayed harvesting rose from those with field harvesting as the emission event
 574 progressed (Fig. 7l) because elevated $[NO_3^-]$ from the manure application persisted longer during the
 575 event (Fig. 7i).

576

577 Following fertilizer application on DOY 259 in 2005 (Table 2), modelled and measured
 578 emissions remained small after soil wetting (Fig. 8f) because lower T_s (Fig. 8d) slowed soil
 579 respiration after wetting, manifested as smaller measured and modelled CO_2 effluxes (Fig. 8a), and so
 580 slowed demand for e^- acceptors. Under these conditions, increasing and delaying defoliation had little
 581 effect on modelled N_2O emissions (Fig. 8f), while CO_2 fixation (Fig. 8a) and N uptake (Fig. 8b) were
 582 only slightly reduced and surface NO_3^- only slightly increased (Fig. 8c). Following the same fertilizer
 583 application on DOY 240 in 2007, modelled and measured emissions were greater than those in 2005
 584 (Fig. 8l) because soils were warmer (Fig. 8j) with more rapid respiration (Fig. 8g), and because
 585 fertilizer application and subsequent wetting occurred sooner after cutting (Table 2). Consequently
 586 recovery of CO_2 fixation was less advanced (Fig. 8g), reducing cumulative N uptake (Fig. 8h) and
 587 leaving larger $[NO_3^-]$ to drive N_2O generation during the event (Fig. 8h). However reducing LAI
 588 remaining after each harvest did not raise N_2O emissions after this application (Fig. 8l), because
 589 slower LAI regrowth from earlier harvests had reduced primary productivity and consequently
 590 litterfall and hence the mass of the surface litter from which much of the emitted N_2O was generated.
 591 Consequently more intense harvests could cause surface litter later in the year to decline to levels at
 592 which N_2O generation modelled in the litter was reduced.

593

594 **Annual Productivity, N_2O Emissions and the Effects of Defoliation Intensity and Timing**

595 In the model, plant management practices affected LAI regrowth (Fig. 6), CO_2 fixation, N
 596 uptake, and hence soil $[NO_3^-]$ and N_2O emissions (Figs. 7,8). These effects were summarized at an
 597 annual time scale in Table 4. Modelled and EC-derived gross primary productivity (GPP) remained
 598 close to $2000 \text{ g C m}^{-2} \text{ y}^{-1}$ during most years except with low precipitation in 2003 and replanting in
 599 2008, indicating a highly productive ecosystem with rapid C cycling and hence rapid demand for e^-
 600 acceptors (Table 4). Larger modelled vs. measured GPP caused larger modelled vs. measured NEP in

601 2003, 2005 and 2007. Harvest removals in the model varied with NEP except during replanting in
602 2008, but tended to exceed those recorded in the field, particularly with low EC-derived NEP in 2005
603 and 2006. Modelled values were determined in part by the assumed constant harvest efficiency of
604 0.76. Including C inputs from manure applications, modelled and estimated net biome productivity
605 (NBP) were positive except during replanting in 2008, indicating that this intensively managed
606 grassland was a C sink unless replanted. Average annual NBP modelled vs. measured from 2002 to
607 2009 was 30 vs. 58 g C m⁻², with the lower modelled value attributed to greater modelled harvest
608 removals, particularly in 2006.

609

610 Slower LAI regrowth from increasing and delaying defoliation (Fig. 6) reduced modelled GPP,
611 R_e and hence NEP by 5 - 10% during years with greater productivity. However increasing and
612 delaying defoliation did not much affect harvest removals because reduced NEP was offset by greater
613 harvest intensity, so that NBP was reduced except with replanting in 2008.

614

615 Annual N₂O emissions were estimated from chamber measurements for each year of the study
616 by scaling the mean measured fluxes to annual values. These values are presented in Table 4 as upper
617 boundaries for annual emissions because flux measurements from which means were calculated were
618 more frequent during emission events. A lower boundary for annual emissions was also estimated in
619 Table 4 by replacing missing flux measurements with zero. Average lower and upper boundaries for
620 annual emissions estimated from 2002 to 2009 were 0.220 and 0.355 g N m⁻² respectively vs. an
621 average annual emission in the model of 0.260 g N m⁻² (Table 4). Modelled emissions were nearer to
622 upper boundaries during years with lower measured emissions (2003, 2004, 2006), and to lower
623 boundaries during years with higher measured emissions (2007, 2008, 2009). There was no significant
624 correlation between annual N inputs and measured or modelled emissions. Although annual emissions
625 in the model were close to 1% of annual N inputs during most years, they were greater in 2008 and
626 2009 in spite of smaller N inputs because of the large emission events modelled after summer
627 applications of fertilizer and manure (Fig. 3e,f; Fig. 5h). Annual N inputs (Table 4), supplemented by
628 3 – 6 g N m⁻² y⁻¹ modelled from symbiotic fixation by clover [F1 – F26]), were only slightly larger
629 than annual N removals with harvesting, supplemented by losses of 2 – 3 g N m⁻² y⁻¹ from all other
630 gaseous and aqueous emissions (N₂ from denitrification, NH₃ from volatilization, NO₃⁻ from
631 leaching). Consequently residual soil NO₃⁻, while present in the model, did not accumulate during the

632 study period, and so did not drive increasing N₂O emissions with sustained N applications. Modelled
 633 and measured annual N₂O emissions, if expressed in C equivalents (~130 g C g N⁻¹), largely offset net
 634 C uptake expressed as NBP (Table 4).

635
 636 Increasing harvest intensity and delaying harvest dates had little effect on annual N₂O
 637 emissions modelled during the first two years after planting in 2001 and 2008, but raised them
 638 substantially thereafter (2003 – 2007) (Table 4). During this period, annual emissions rose by an
 639 average of 24% with increased harvest intensity, and by an average of 43% with increased harvest
 640 intensity combined with delayed harvest dates. These increases were attributed to reduced N uptake,
 641 and to increased T_s and θ (Figs. 7, 8).

642

643 **Effects of increased bulk density on N₂O emissions**

644 Increasing near-surface (0 – 3 cm) soil BD by 5% or 10% at the beginning of 2001 in the model
 645 reduced [O_{2(s)}] after rainfall events and slowed recovery of [O_{2(s)}] during subsequent drying as shown
 646 following the manure application in July 2007 (Fig. 9a) and the fertilizer application in late August
 647 2007 (Fig. 9c). These reductions caused increases in modelled N₂O effluxes that varied during
 648 emission events (Fig. 9b,d). Effluxes modelled with increases of 10% in near-surface BD were at times
 649 double those modelled without (e.g. DOY 201 and 240 in Fig. 9), indicating that relatively small
 650 changes in soil surface properties could at times cause large changes in emissions. The effects of
 651 increased BD on modelled T_s , θ , CO₂ exchange, crop production and N uptake during these events
 652 were small (results not shown). Increasing near-surface BD by 10% raised annual N₂O emissions by
 653 amounts that increased with annual precipitation from *ca.* 10% in drier years (e.g. 2003) to *ca.* 50% in
 654 wetter (e.g. 2006) (Table 5).

655

656 **Effects of Changes in K_{O_2} and K_{NO_x} on N₂O emissions**

657 Lowering K_{O_2} to one-half that used in *ecosys* reduced annual N₂O emissions modelled from
 658 2004 to 2009 by 16% to an average of 0.218 g N m⁻² y⁻¹, near the average lower boundary of the
 659 measured values (Table 5). Raising K_{O_2h} to double that used *ecosys* increased these emissions by 28%
 660 to an average of 0.334 g N m⁻² y⁻¹, near the average upper boundary of the measured values. Lowering
 661 K_{NO_x} to one-half that used in *ecosys* increased annual N₂O emissions modelled from 2004 to 2009 by
 662 30% to an average of 0.338 g N m⁻² y⁻¹, near the average upper boundary of the measured values

663 (Table 5). Raising K_{NO_x} to double that used *ecosys* reduced these emissions by 27% to an average of
664 $0.189 \text{ g N m}^{-2} \text{ y}^{-1}$, near the average lower boundary of the measured values. In years with lower annual
665 emissions (2003, 2004, 2006 in Table 4), the lower K_{O_2} or higher K_{NO_x} gave modelled values that were
666 closer to measured values. However in years with higher annual emissions (2008 and 2009 in Table 4),
667 the higher K_{O_2} or lower K_{NO_x} gave modelled values that were closer.

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DISCUSSION

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Modelled vs. Measured N_2O Emissions

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Most N_2O emission events measured from 2004 to 2009 were simulated within the range of measurement uncertainty, estimated to be about 30% of mean daily values (Fig. 3). However some deviations between modelled and measured N_2O emissions were apparent, such as the larger emissions modelled in autumn 2006 (Fig. 3c) and the smaller emissions modelled in spring 2008 (Fig. 3e). These deviations may be attributed to uncertainties in both the measurements and the model. In the automated measurement system, the static chambers were rotated about every two months among fixed positions in a corner of the field. During these periods, surface conditions in the chamber could deviate from the mean field conditions represented in the model. However we do not have an explanation for the very small emissions measured after the three manure slurry applications 2006. The chambers had been removed before the applications and were reinstalled within two hours, during which the cut grass was removed so that the surface litter in the chambers may have been reduced from that outside. In the model, emissions following manure or fertilizer applications were sensitive to the amount of surface litter as noted earlier. The absence of emission events measured after slurry applications in 2006 was unusual (Fig. 3) given the large precipitation that year (Table 4), demonstrating that large variability at small spatial scales inevitably affects these measurements. Such variability adversely affects agreement between modelled and measured emissions (Table 3).

During spring 2008 sustained emissions of about $5 \text{ mg N m}^{-2} \text{ d}^{-1}$ were measured by the chambers in the absence of any manure or fertilizer applications (Fig. 3e). These emissions were related to the ploughing of the field to a depth of 25cm in December 2007 (Table 2) which hastened soil organic matter decomposition, and hence N mineralization that increased mineral N substrate for

694 nitrification and denitrification, and possibly microbial nitrifier and denitrifier populations. These
695 increases must remain conjectural as the Oensingen study did not include stratified analysis of N₂O
696 production factors (e.g. microbial biomass, potential denitrification) within the chamber soils.
697 Although *ecosys* simulates hastened SOM decomposition with tillage (Grant et al., 1998), large
698 amounts of above- and below-ground plant litter with relatively high C:N ratios were incorporated in
699 the model with tillage in December 2007 which slowed net N mineralization and hence accumulation
700 of mineral N products in the model during spring 2008. Consequently modelled N₂O emissions
701 remained small until mineral N was raised by fertilizer applications in July (Fig. 3c).

702

703 **Modelling Controls on N₂O Emissions by Litter and Near-Surface θ and T_s**

704 In the model, almost all the N₂O emissions originated in the surface litter and in the near-
705 surface (0 – 1 cm) soil layer, so that emissions were strongly controlled by litter and near-surface θ
706 and T_s (Figs. 3 – 4). This model finding is consistent with the experimental finding of Pal et al. (2013)
707 from ¹⁵N enrichment studies that approximately 70% of N₂O measured during emission events in a
708 managed grassland originated in the surface litter. Similarly van der Weerden et al. (2013) inferred
709 from diurnal variation in T_s and N₂O emissions measured after urine amendments on a managed
710 grassland that N₂O production was at or near the soil surface (0 - 2 cm). Also Flécharde et al. (2007)
711 inferred in a meta-analysis of N₂O emissions from grasslands in Europe that θ measured at 5 cm was
712 not in some cases an adequate scaling factor for N₂O source strength because N₂O production and
713 emission took place at or near the soil surface. *Ecosys* simulated little net production, and even a
714 small net consumption, of N₂O in soil below 2 cm during emission events, as may be inferred from
715 peak [N₂O_(s)] modelled in the 0 – 1 cm soil layer and much lower [N₂O_(s)] modelled in the 1 – 3 cm
716 soil layer below (Figs. 3g and 4g). This model finding was consistent with the experimental finding of
717 Neftel et al. (2000) that N₂O concentrations below near-surface soil layers in a managed grassland
718 remained below atmospheric values during emission events, from which they inferred that any N₂O
719 generated at depths greater than ~3 cm would not likely reach the soil surface. Thus attempts to relate
720 N₂O emissions to T_s and θ measured at greater depths than 3 cm in grasslands are unlikely to be
721 informative if these differ from near-surface values. These emissions should rather be related to
722 conditions in the litter and near-surface soil, which need to be better characterized in future studies.

723

724 Consequently modelled N₂O emissions were highly sensitive to surface wetting and drying (e.g.
725 Fig. 4e,h) modelled from precipitation vs. ET (e.g. Fig. 4a,c), or to surface warming and cooling (e.g.
726 Fig. 8j,l) modelled from surface energy balance (e.g. Fig. 4c). The sensitivity to surface wetting and
727 drying was modelled from the effects of θ on air- vs. water-filled porosity and hence on diffusivity of
728 gases in gaseous [D17] and aqueous [D20] phases, and on gaseous volatilization - dissolution transfer
729 coefficients and hence gas exchange between gaseous and aqueous phases [D14, D15]. These
730 transfers controlled O₂ supply, and hence demand for alternative e⁻ acceptors as the O₂ supply fell
731 below O₂ demand, which drove N₂O generation from denitrification [H6 – H8] and nitrification
732 [H19]. The control of O₂ supply on e⁻ acceptors used in nitrification thereby simulated the effect of
733 WFPS on the fraction of N₂O generated during nitrification identified by Fang et al. (2015) as
734 necessary to modelling N₂O emissions, while avoiding the model-specific parameterization needed in
735 simpler models. The sensitivity to surface wetting in *ecosys* enabled sharp rises in N₂O emissions to
736 be modelled from surface litter and near-surface soil after small precipitation events during DOY 200
737 - 201 in 2007 (Fig. 4a,h), and after slurry application during DOY 218 in 2009 (Fig. 5a,h), even when
738 the soil at 5 cm remained dry (Fig. 4b; Fig. 5b). Such rises were consistent with the experimental
739 findings of Flécharde et al. (2007) that precipitation on dry soil can cause substantial N₂O emissions
740 after fertilizer application in grasslands.

741

742 The sensitivity to surface warming and cooling was modelled from the effects of T_s on
743 diffusivity of gases in gaseous [D17] and aqueous [D20] phases, and on solubility of gases and hence
744 exchange of gases between gaseous and aqueous phases [D14, D15], both parameterized from basic
745 physical relationships independently from the model. These transfers controlled [O_{2(s)}] in the surface
746 litter and soil (Figs. 3f and 4f), and hence O₂ uptake by aerobic heterotrophs [H4] and autotrophs
747 [H13] through a Michaelis-Menten constant [H4b, H13b]. The sensitivity to surface warming and
748 cooling was also modelled from the effects of T_s on SOC oxidation [H2] and hence O₂ demand by
749 aerobic heterotrophs [H3], and on NH₄⁺ and NO₂⁻ oxidation [H11, H15] and hence O₂ demand by
750 aerobic autotrophs [H12, H16]. These effects were driven by a single Arrhenius function used for all
751 biological transformations [A6] parameterized from basic research conducted independently from the
752 model. Under sustained high surface θ , this combination of physical and biological processes drove
753 large diurnal variation in N₂O emissions modelled with diurnal surface warming and cooling during
754 emission events (e.g. DOY 221 in Fig. 5h, DOY 243 in Fig. 8l), as observed experimentally by van

755 der Weerden et al. (2013). By explicitly simulating the diverse processes that determine N₂O
756 emissions, *ecosys* could model the large sensitivity of emissions to T_s without the use of
757 unrealistically large parameters for temperature sensitivity inferred from controlled temperature
758 studies of N₂O emissions (e.g. Dobbie and Smith, 2001). This large sensitivity to T_s has been
759 inadequately represented in simpler models, causing underestimation of large emissions measured
760 from warm soils (e.g. Saggar et al., 2004). At a seasonal time scale, higher T_s could cause large
761 increases in N₂O emissions modelled with comparable θ after the same fertilizer application (Fig. 8l
762 vs. Fig. 8f). However the effects of T_s on N₂O emissions were dominated by those of θ during surface
763 wetting and drying (e.g. Figs. 4h, 7l).

764

765 Values of both θ and T_s thus determined O₂ demand not met by O₂ uptake which drove demand
766 for alternative e⁻ acceptors by heterotrophic denitrifiers [H6] and autotrophic nitrifiers [H19]. This
767 demand drove the sequential reduction of NO₃⁻, NO₂⁻ and N₂O to NO₂⁻, N₂O and N₂ respectively by
768 heterotrophic denitrifiers [H7, H8, H9], and the reduction of NO₂⁻ to N₂O by autotrophic nitrifiers
769 [H20]. The consequent production of N₂O (Fig. 4g, Fig. 5g) and N₂ drove emissions of both N₂O and
770 N₂ (Fig. 4h, Fig. 5h) through volatilization [D14, D15] and through gaseous and aqueous diffusion
771 [D16, D19]. Ratios of N₂O and N₂ emissions in *ecosys* (Fig. 4h, Fig. 5h) were not parameterized as
772 done in other models, but rather were determined by relative affinities determined from basic research
773 [H8, H9], and by environmental conditions. When demand from heterotrophic denitrifiers for
774 alternative e⁻ acceptors was small relative to their availability, the preferential reduction of more
775 oxidized e⁻ acceptors generated larger emissions of N₂O [H7, H8] relative to N₂ [H9]. Such
776 conditions occurred during the early part of an emission event when surface [NO₃⁻] rose with
777 nitrification of fertilizer or manure NH₄⁺ after application (e.g. DOY 200 – 201 in Fig. 4h). However
778 when demand for alternative e⁻ acceptors was large relative to their availability, this same reduction
779 sequence forced more rapid reduction of N₂O to N₂ and hence smaller emissions of N₂O relative to
780 N₂. Such conditions occurred during the later part of emission events when surface [NO₃⁻] declined
781 with plant uptake (e.g. DOY 202 – 205 in Fig. 4h and DOY 222 in Fig. 5h), or when greater surface
782 wetting reduced O₂ supply (e.g. DOY 220 in Fig. 5h). This greater demand for alternative e⁻ acceptors
783 with wetting provided a process-based explanation for declines in N₂O emissions frequently found at
784 higher θ in field studies (e.g. Rafique et al., 2011) without explicit parameterization of N₂O:N₂ ratios.

785

786 Nitrification and denitrification were also driven by the concentrations of NH_4^+ [H11], NO_3^-
 787 [H7], NO_2^- [H8, H15, H20] and N_2O [H9] relative to Michaelis-Menten constants evaluated from
 788 basic research. The concentrations of NH_4^+ and NO_3^- in *ecosys* were increased by N additions from
 789 manure and fertilizer N applications (Table 2), and by net mineralization soil organic N from
 790 oxidation of litterfall, manure and SOM [A26] as indicated by soil CO_2 effluxes. These concentrations
 791 were reduced by root uptake of NH_4^+ and NO_3^- [C23] and consequent plant N assimilation with
 792 growth, indicated by more rapid CO_2 fixation with time after cutting (Figs 3 – 4 and Figs. 6 - 7). In
 793 the model, more rapid CO_2 fixation drove more rapid production of nonstructural C, and hence more
 794 rapid exchange of nonstructural C and N between canopy and roots [C50], and so hastened root active
 795 N uptake by increasing R_a driving root growth [C14b], and by hastening removal of N uptake
 796 products and hence reducing their inhibition of active uptake [C23g]. The diversity of controls on key
 797 substrates for N_2O generation suggests that robust simulations of N_2O emissions require
 798 comprehensive ecosystem models in which these controls are fully represented.

799

800 **Modelling Effects of Defoliation Intensity and Timing on N_2O Emissions**

801 The control of NH_4^+ and NO_3^- availability by root N uptake indicated that plant management
 802 practices determining uptake would thereby affect N_2O emissions. In the model, increasing harvest
 803 intensity and delaying harvest dates both slowed N uptake (Fig. 7b,h and Fig. 8b,h) by slowing the
 804 recovery of LAI (Fig. 6) and CO_2 fixation (Fig. 7a,g and Fig. 8a,g). Both thereby increased [NO_3^-]
 805 (Fig. 7c,i and Fig. 8c,i), T_s (Fig. 7d,j and Fig. 8d,j) and θ (Fig. 7e,k and Fig. 8e,k), raising N_2O
 806 effluxes modelled during most emission events (Fig. 7f,l and Fig. 8f,l), and hence annually (Table 4).
 807 This model finding was consistent with the field observations of Jackson et al. (2015) that increased
 808 N_2O emissions after defoliation in grasslands were caused by reduced uptake of N and water by
 809 slower-growing plants.

810

811 The effects of defoliation on N_2O emissions during modelled emission events were similar to,
 812 or greater than, those of T_s and θ (e.g. Fig. 7f,l), consistent with the experimental finding of Imer et al.
 813 (2013) that plant management, as represented by its effects on LAI, had a larger effect on N_2O fluxes
 814 than did the environment, as represented by T_a , at an intensively managed grassland in Switzerland.
 815 Reducing LAI remaining after harvest by one-half and delaying harvest by 5 days had little effect on
 816 modelled harvest removals (Table 4), suggesting that N_2O emissions from managed grasslands are

817 more sensitive to plant management practices than are yields. Intensity and timing of harvests should
818 therefore be selected to avoid slow regrowth of LAI following N additions by avoiding excessive
819 defoliation and by allowing as much time as possible between defoliation and subsequent fertilizer or
820 manure application. Neftel et al. (2010) reported enhanced N_2O emissions after cuts in managed
821 grassland and hypothesized that a simple mitigation option would be to optimize the timing of the
822 fertilizer applications. To our knowledge this option has not been systematically investigated.

823

824 **Modelling Effects of Soil Bulk Density on N_2O Emissions**

825 The small increases in near-surface BD included in this study were typical of those arising from
826 natural variation in soil properties or from compaction by vehicular traffic during field management
827 operations. In the model, these increases reduced soil porosity and hence gaseous diffusivity [D17]
828 which slowed O_2 uptake from the atmosphere [D15] and O_2 transfer through the soil profile [D16].
829 Consequent reductions in near-surface [$O_{2(s)}$] (Fig. 9a,c) slowed O_2 reduction by denitrifiers [H4] and
830 nitrifiers [H13], forcing more rapid e^- transfer to NO_3^- by denitrifiers [H6] and to NO_2^- by nitrifiers
831 [H19] and hence more rapid emissions of N_2O following applications of manure (Fig. 9b) and fertilizer
832 (Fig. 9d).

833

834 In a study of soil compaction effects on N_2O emissions from a fertilized agricultural field in a
835 climate similar to that at Oensingen, Bessou et al. (2010) found that increasing the BD of the upper 30
836 cm of the soil profile by *ca.* 15% raised annual N_2O emissions measured with automated chambers by
837 at least 50% during each of two growing seasons. These rises were similar to those modelled with a
838 smaller increase in BD of the upper 3 cm during the wettest year of this study (Table 5). During
839 emission events, Bessou et al. (2010) measured peak fluxes from compacted soil that were double
840 those from uncompacted, as also modelled here (Fig. 9b,d).

841

842 The detailed algorithms from which *ecosys* was constructed enabled increases in N_2O
843 emissions from surface compaction to be simulated from specified changes to surface BD, a
844 measureable site characteristic, without further model parameterization. The marked increases in N_2O
845 emissions modelled with these increases in BD (Table 5) indicated that some of the large spatial
846 variation in these emissions commonly found in field measurements could arise from relatively small
847 variation in physical properties of near-surface soil. In future studies of N_2O emissions, near-surface

848 soil properties could be determined at each measurement site to establish the extent to which variation
849 in these properties are associated with those in emissions.

850

851 **Modelling Effects of K_{O_2} and K_{NO_x} on N_2O Emissions**

852 The value of K_{O_2} used in *ecosys* (=2 μM) was taken from the upper range of values determined
853 experimentally for intact cells of heterotrophic bacteria by Longmuir (1954). Halving or doubling K_{O_2}
854 changed modelled N_2O emissions (Table 5) by amounts similar to uncertainty in measured emissions
855 expressed as lower and upper boundaries of likely values (Table 4), although the doubled value of K_{O_2}
856 was larger than those derived from experiments. The value of K_{NO_x} used in *ecosys* (=100 μM) was
857 within the range of values determined experimentally by Yoshinari et al. (1977). As for K_{O_2} , halving or
858 doubling K_{NO_x} changed modelled N_2O emissions (Table 5) by amounts similar to uncertainty in
859 measured emissions expressed as lower and upper boundaries of likely values (Table 4). The halved
860 value of K_{NO_x} was closer to those measured by Betlach and Tiedje (1981) and Khalil et al. (2007) while
861 the doubled value was closer to that measured by Klemedtsson et al. (1977). These changes indicate
862 that key parameters used in process models must be capable of being constrained by accurate
863 evaluation in independent experiments.

864

865

866 **CONCLUSIONS**

867

868 N_2O emissions modelled in this managed grassland originated in the surface litter and upper 2
869 cm of the soil profile. The shallow origin of these emissions enabled *ecosys* to simulate the response of
870 measured emissions to changes in near-surface θ and T_s during brief emission events when rainfall
871 followed manure or mineral fertilizer applications. Measurements of θ and T_s used to estimate N_2O
872 emissions from managed grasslands should therefore be taken in surface litter and near-surface soil (0
873 – 2 cm), rather than deeper in the soil profile (5 – 10 cm) as is currently done.

874

875 N_2O fluxes modelled during emission events were greater when grassland regrowth and hence
876 mineral N uptake was slower following harvest and subsequent N application. The control of N_2O
877 emissions by grassland N uptake indicated that N_2O emissions from managed grassland could be

878 increased by harvesting practices and fertilizer timing that resulted in slower regrowth during periods
879 when emission events are most likely to occur. N₂O fluxes modelled during emission events rose
880 sharply with small increases in surface BD, indicating the importance of avoiding surface compaction
881 in fields to which large amounts of N are applied.

882

883 The basic and comprehensive approach to model development in *ecosys* allowed diverse
884 responses of N₂O emissions to changes in weather (T_s , θ), land management and soil properties to be
885 modelled from specified changes to readily measured inputs with parameters constrained by basic
886 experiments conducted independently of the model rather than derived from site-specific observations.
887 This approach enabled concurrent, well-constrained tests of model performance against a diverse set of
888 field measurements, and so is expected to confer robustness to the modelling of these emissions under
889 different climates, soils and land uses in future studies.

890

891

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892

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1019 **Table 1.** Key soil properties of the Eutri-Stagnic Cambisol at Oensingen as used in *ecosys*.

Depth	BD [¶]	TOC	TON	FC [†]	WP [†]	K _{sat} [†]	pH	Sand [‡]	Silt [‡]	Clay [‡]	CF
m	Mg m ⁻³	g kg ⁻¹	g kg ⁻¹	m ³ m ⁻³	m ³ m ⁻³	mm h ⁻¹		g kg ⁻¹	g kg ⁻¹	g kg ⁻¹	m ³ m ⁻³
0.01	1.21	27.2	2.9	0.38	0.22	3.4	7	240	330	430	0
0.03	1.21	27.2	2.9	0.38	0.22	3.4	7	240	330	430	0
0.07	1.21	27.2	2.9	0.38	0.22	3.4	7	240	330	430	0
0.13	1.24	27.2	2.9	0.39	0.23	3.4	7	240	330	430	0
0.28	1.28	20.2	2.1	0.40	0.24	2.4	7	180	380	440	0
0.6	1.28	11.6	1.1	0.40	0.24	1.4	7	180	380	440	0
0.7	1.28	11.6	1.1	0.40	0.24	1.4	7	180	380	440	0
0.9	1.28	9	0.9	0.40	0.24	1.4	7	180	380	440	0
1.5	1.28	6	0.6	0.40	0.24	1.4	7	180	380	440	0.1

1020 [¶]abbreviations BD: bulk density, TOC and TON: total organic C and N, FC: field capacity, WP: wilting
1021 point, K_{sat}: saturated hydraulic conductivity, CF: coarse fragments.

1022 [‡]BD, TOC and texture were determined from soil cores taken in 2001 and 2006. Details are given in
1023 Leifeld et al. (2011).

1024 [†]FC, WP and K_{sat} were estimated from BD, TOC and texture according to Saxton et al. (1996) and
1025 Saxton and Rawls (2006).

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Table 2. Plant and soil management operations at the Oensingen intensively managed grassland from 2001 to 2009.

Year	Plant Management		Soil Management					
	Date	Management	Date	Management	Amount (g m ⁻²)			
					NH ₄ ⁺	NO ₃ ⁻	ON	OC
2001			07 May	tillage				
			10 May	tillage				
	11 May	planting	15 June	mineral fertilizer	1.5	1.5		
	1 July	harvest	12 July	mineral fertilizer	1.5	1.5		
	8 Aug.	harvest	16 Aug.	mineral fertilizer	1.15	1.15		
	12 Sep.	harvest						
	31 Oct.	harvest						
2002			12 Mar.	mineral fertilizer	1.5	1.5		
	15 May	harvest	22 May	manure slurry	4.2		2.8	31.2
	25 June	harvest	1 July	mineral fertilizer	1.75	1.75		
	15 Aug.	harvest	18 Aug.	manure slurry	5.9		5.3	49.6
	18 Sep.	harvest	30 Sep.	mineral fertilizer	1.5	1.5		
	07 Dec.	harvest						
2003			18 Mar.	manure slurry	5.9		5.3	61.1
	30 May	harvest	02 June	mineral fertilizer	1.5	1.5		
	04 Aug.	harvest	18 Aug.	manure slurry	6.3		1.9	19.0
	13 Oct.	harvest						
2004			17 Mar.	manure slurry	5.0		1.5	19.5
	11 May	harvest	17 May	mineral fertilizer	1.5	1.5		
	25 June	harvest	01 July	manure slurry	5.5		0.5	9.9
	28 Aug.	harvest	31 Aug.	mineral fertilizer	1.5	1.5		
	03 Nov.	harvest						
2005			29 Mar.	manure slurry	6.7		3.1	42.0
	10 May	harvest	17 May	mineral fertilizer	1.5	1.5		
	27 June	harvest	05 July	manure slurry	5.0		3.5	59.6
	29 Aug.	harvest	16 Sep.	mineral fertilizer	1.5	1.5		
	24 Oct.	harvest						
2006	24 May	harvest						
	05 July	harvest	13 July	manure slurry	4.7		1.4	12.5
	12 Sep.	harvest	27 Sep.	manure slurry	4.4		1.3	13.6
	26 Oct.	harvest	30 Oct.	manure slurry	6.4		3.2	57.8

2007			03 Apr.	manure slurry	5.2		4.6	75.1
	26 Apr.	harvest	03 May	mineral fertilizer	1.5	1.5		
	06 July	harvest	13 July	manure slurry	4.9		1.8	45.9
	23 Aug.	harvest	28 Aug.	mineral fertilizer	1.5	1.5		
	11 Oct.	harvest	24 Oct.	manure slurry	4.6		3.0	38.9
	19 Dec.	terminate	19 Dec.	plowing				
2008			01 May	tillage				
			04 May	tillage				
	05 May	planting						
	01 July	harvest	10 July	mineral fertilizer	1.5	1.5		
	29 July	harvest	07 Aug.	mineral fertilizer	1.5	1.5		
	08 Sep.	harvest	19 Sep.	manure slurry	2.9		0.5	8.6
	07 Nov.	harvest						
2009			07 Apr.	mineral fertilizer	1.5	1.5		
	01 May	harvest	12 May	manure slurry	4.4		1.6	26.0
	16 June	harvest	06 Aug.	manure slurry	3.3		1.2	19.0
	29 July	harvest						
	07 Sep.	harvest	15 Sep.	mineral fertilizer	6.5(urea)			
	20 Oct.	harvest						

Table 3: Intercepts (*a*), slopes (*b*) coefficients of determination (R^2), ratios of mean squares for regression vs. error (F) and number of data pairs from regressions of (a) log-transformed 4-hour averages of N_2O fluxes ($mg\ N\ m^{-2}\ h^{-1}$) modelled vs. measured during each year from 2004 to 2009, and (b) total N_2O fluxes ($mg\ N\ m^{-2}$) modelled vs. measured during emission events following each fertilizer or manure application from 2004 to 2009 (see Fig. 3) at the Oensingen intensively managed grassland.

Year	<i>a</i>	<i>b</i>	R^2	F^\dagger	<i>n</i>
(a)					
2004	$1.25 \pm 0.88 \times 10^{-5}$	0.49 ± 0.06	0.08	69	818
2005	$1.63 \pm 0.43 \times 10^{-5}$	0.59 ± 0.03	0.24	368	1173
2006	$4.28 \pm 0.44 \times 10^{-5}$	1.04 ± 0.08	0.14	155	948
2007	$1.21 \pm 0.33 \times 10^{-5}$	0.67 ± 0.02	0.35	989	1794
2008	$1.44 \pm 0.51 \times 10^{-5}$	0.44 ± 0.03	0.08	157	1703
2009	$-0.03 \pm 0.25 \times 10^{-5}$	0.71 ± 0.02	0.49	1574	1614
(b)					
2004 - 2009	$28 \pm 9\ mg\ N\ m^{-2}$	0.67 ± 0.13	0.54	27	23

[†] All values of F were highly significant ($P < 0.001$).

Table 4. Annual gross primary productivity (GPP), ecosystem respiration (R_e), net ecosystem productivity (NEP = GPP - R_e), harvest, net biome productivity (NBP) and N₂O emissions derived from EC or chambers and modelled (M) with current land management (Table 2), and with defoliation increased so that LAI remaining after harvesting was reduced by one-half (1/2), with defoliation increased and delayed by 5 days (1/2 + 5d). Positive values indicate uptake, negative values emissions.

Year		2002	2003	2004	2005	2006	2007	2008	2009
Precip.(mm)		1478	817	1158	966	1566	1328	1188	1004
MAT (°C)		9.56	9.58	8.92	8.67	9.30	9.59	9.30	9.48
GPP	EC	2159	1773	2058	1766	1817	2102	1455	2119
(g C m ⁻² y ⁻¹)	M: current	2214	1836	2220	2111	1953	2539	1419	1852
	: 1/2	2064	1764	2054	1969	1865	2285	1305	1705
	: 1/2 + 5d	2014	1774	2076	1966	1771	2277	1225	1686
R_e	EC	-1490	-1558	-1541	-1565	-1577	-1684	-1450	-1657
(g C m ⁻² y ⁻¹)	M: current	-1560	-1421	-1704	-1679	-1680	-1935	-1366	-1373
	: 1/2	-1457	-1345	-1569	-1572	-1579	-1714	-1212	-1259
	: 1/2 + 5d	-1458	-1350	-1541	-1517	-1519	-1679	-1183	-1235
NEP	EC	669	215	517	201	240	418	5	462
(g C m ⁻² y ⁻¹)	M: current	654	415	516	432	273	604	53	479
	: 1/2	607	419	485	397	286	571	93	446
	: 1/2 + 5d	556	414	535	449	252	598	42	451
Harvest	field	462	241	401	247	232	448	293	532
(g C m ⁻² y ⁻¹)	M: current	570	314	525	460	421	690	308	487
	: 1/2	561	360	465	497	455	678	314	484
	: 1/2 + 5d	537	353	579	513	446	686	262	473
C inputs		81	80	29	102	84	160	9	45
NBP	field	288	54	145	56	92	130	-279	-25
(g C m ⁻² y ⁻¹)	M: current	165	181	20	74	-64	74	-246	37
	: 1/2	127	139	49	2	-85	53	-212	7
	: 1/2 + 5d	101	141	-15	38	-110	72	-211	23
N inputs		27.6	22.5	18.5	24.3	21.4	30.1	9.4	20.0
N ₂ O	chamber								
(g N m ⁻² y ⁻¹)	upper bound	-0.130	-0.050	-0.060	-0.230	-0.020	-0.280	-0.480	-0.510
	lower bound	-0.450	-0.180	-0.180	-0.320	-0.060	-0.350	-0.620	-0.680
	M: current	-0.302	-0.209	-0.183	-0.193	-0.220	-0.281	-0.326	-0.366
	: 1/2	-0.269	-0.215	-0.250	-0.249	-0.318	-0.312	-0.335	-0.318
	: 1/2 + 5d	-0.284	-0.234	-0.347	-0.352	-0.273	-0.348	-0.327	-0.395

Table 5. Annual N₂O emissions modelled with current field management (Table 2) and soil properties (Table 1) (current), with soil bulk density (BD) increased by 5% and 10% to a depth of 3 cm, and with the Michaelis-Menten constants for reduction of O₂ (K_{O_2}) and of NO₃⁻ and NO₂⁻ (K_{NO_x}) halved or doubled from those used in the model.

Year		2002	2003	2004	2005	2006	2007	2008	2009
Precip.(mm)		1478	817	1158	966	1566	1328	1188	1004
MAT (°C)		9.56	9.58	8.92	8.67	9.30	9.59	9.30	9.48
N ₂ O (g N m ⁻² y ⁻¹)	current	-0.302	-0.209	-0.183	-0.193	-0.220	-0.281	-0.326	-0.366
	BD + 5%	-0.352	-0.213	-0.218	-0.199	-0.309	-0.332	-0.358	-0.372
	BD + 10%	-0.334	-0.235	-0.231	-0.236	-0.336	-0.374	-0.424	-0.371
	$K_{O_2} \times 0.5$	-0.250	-0.179	-0.154	-0.159	-0.160	-0.216	-0.276	-0.349
	$K_{O_2} \times 2.0$	-0.390	-0.263	-0.221	-0.247	-0.315	-0.385	-0.381	-0.468
	$K_{NO_x} \times 0.5$	-0.382	-0.261	-0.265	-0.267	-0.262	-0.378	-0.432	-0.457
	$K_{NO_x} \times 2.0$	-0.234	-0.163	-0.126	-0.132	-0.126	-0.208	-0.232	-0.288

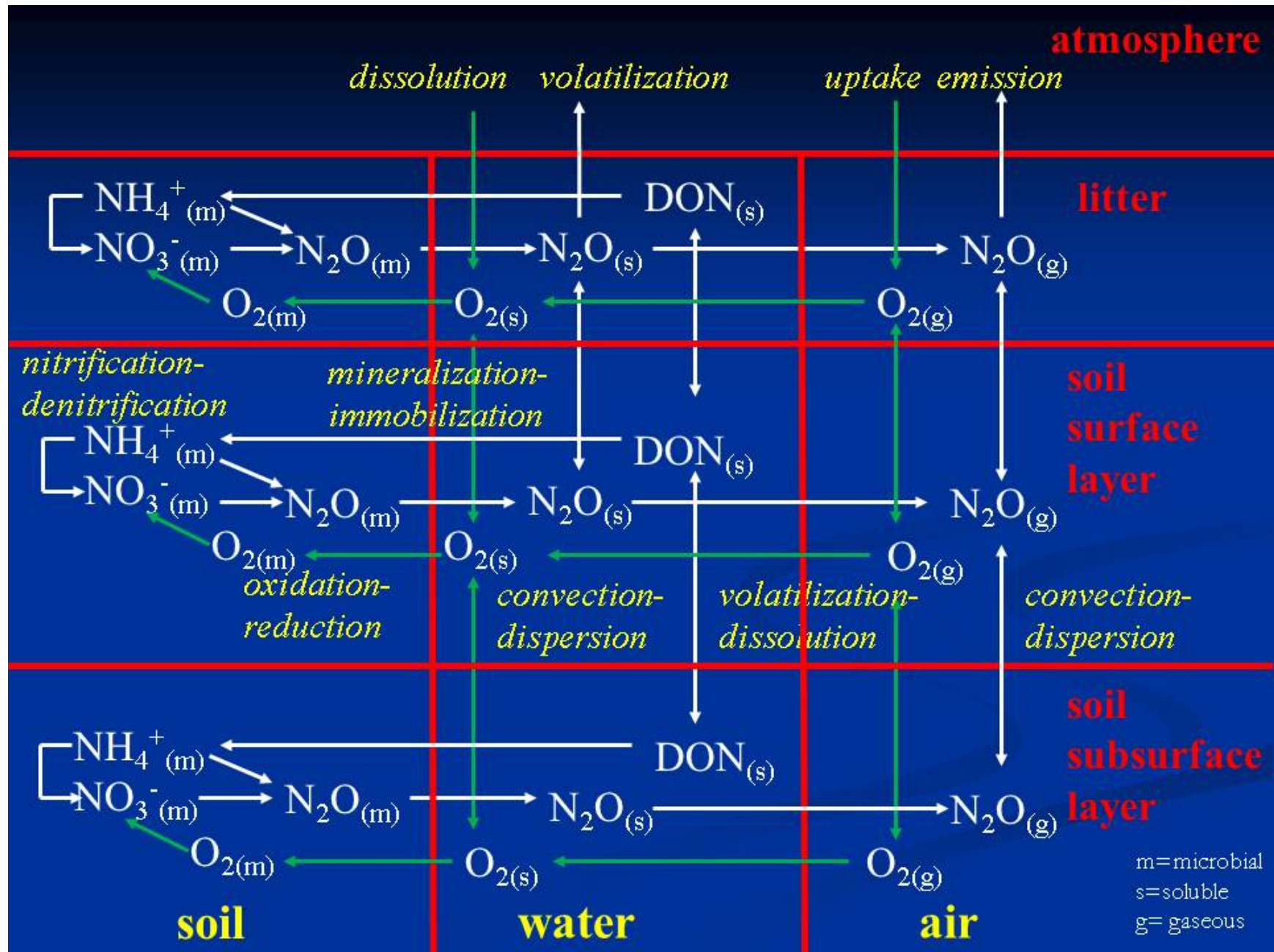


Fig. 1: Summary of key processes governing generation and emission of N_2O as represented in *ecosys*.

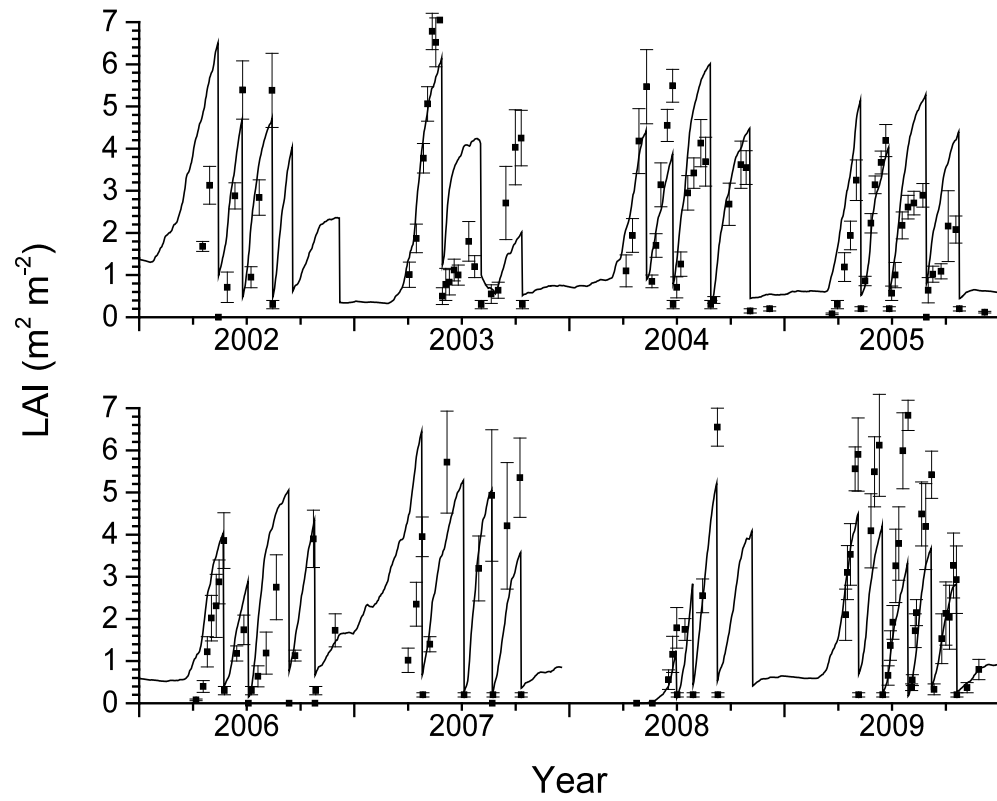


Fig. 2. LAI measured (symbols) and modelled (lines) from 2002 through 2009 at the Oensingen intensively managed grassland.

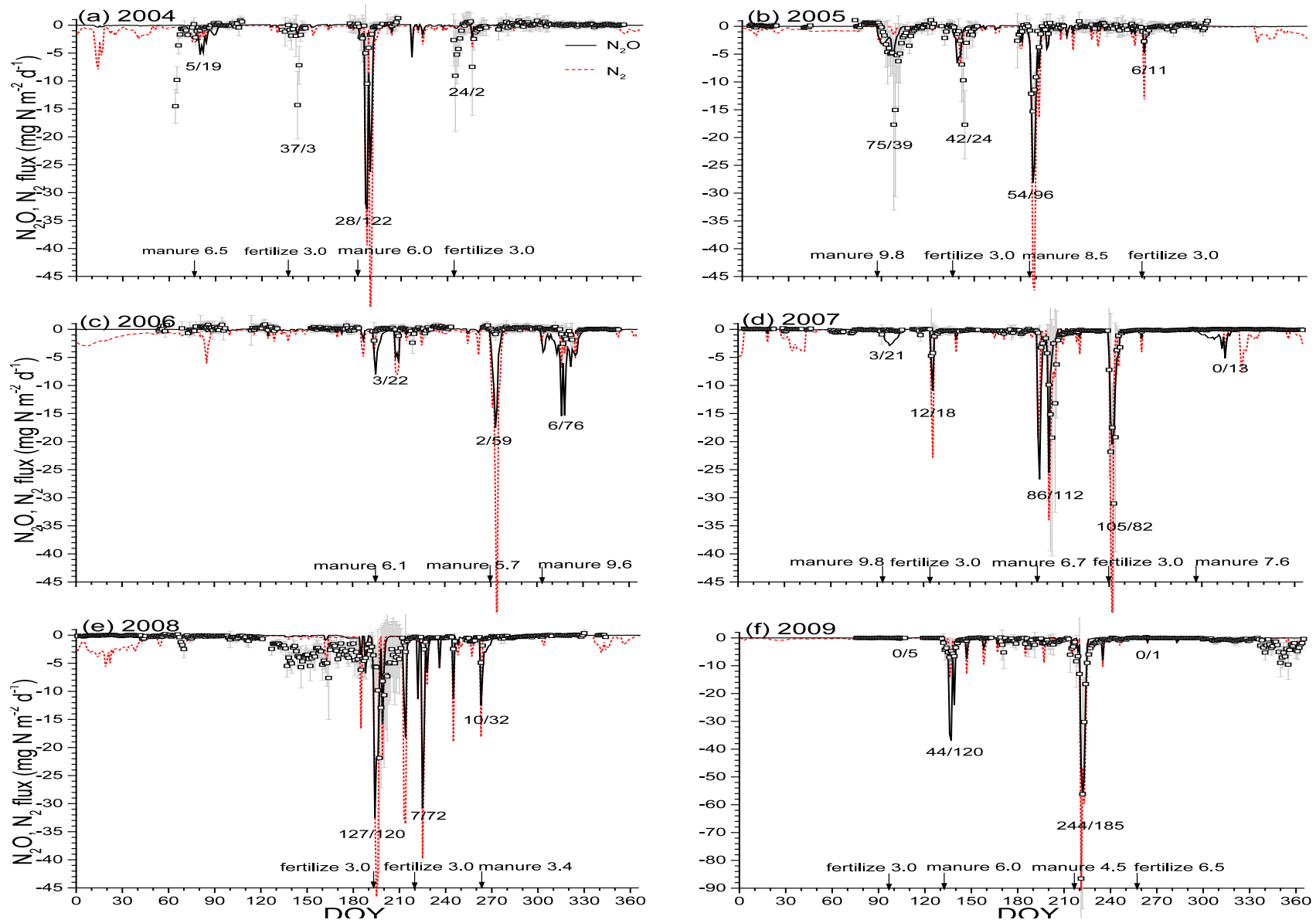


Fig. 3. Daily-aggregated N_2O emissions measured (symbols) and N_2O and N_2 emissions modelled (lines) from 2004 through 2009 at the Oensingen intensively managed grassland. Numbers above and beside each fertilizer or manure addition indicate total measured/modelled N_2O -N emitted during emission events (mg N m^{-2}), and total N applied (g N m^{-2}). Negative values indicate effluxes to the atmosphere.

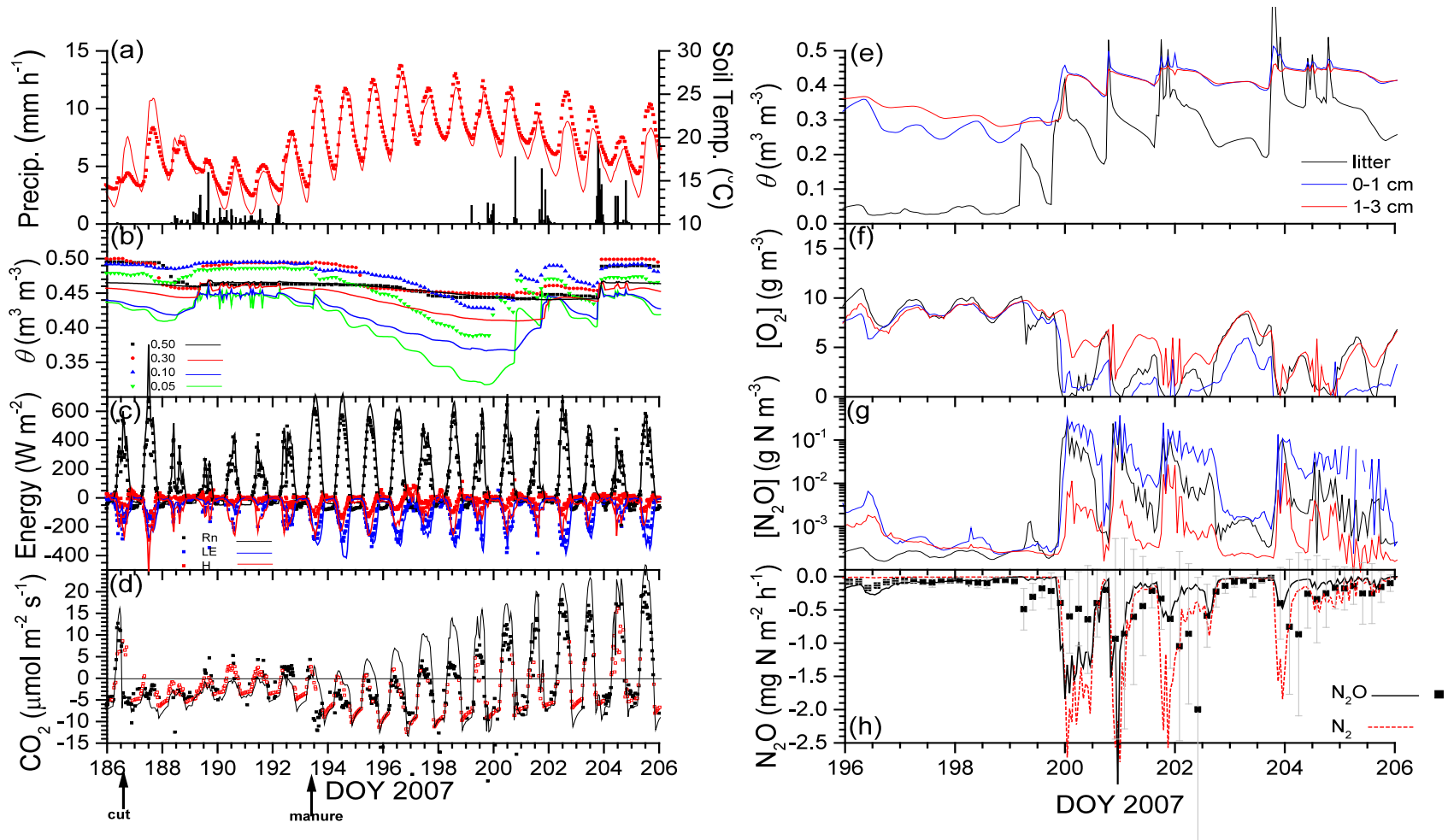


Fig. 4. (a) Precipitation and soil temperature at 0.05 m, (b) soil water content (θ) at 0.05, 0.10, 0.30 and 0.50 m, (c) energy and (d) CO₂ fluxes measured (closed symbols), gap-filled (open symbols) and modelled (lines) during 20 days from harvest (cut) to the end of the emission event following manure application (manure) in July 2007. (e) θ , (f and g) aqueous concentrations of O₂ and N₂O modelled in the surface litter and at 0.01 and 0.02 m in the soil, and (h) N₂O and N₂ fluxes measured (symbols) and modelled (lines) during the last 10 days of this period when the emission event occurred. For fluxes, positive values represent influxes to the soil, negative values effluxes to the atmosphere.

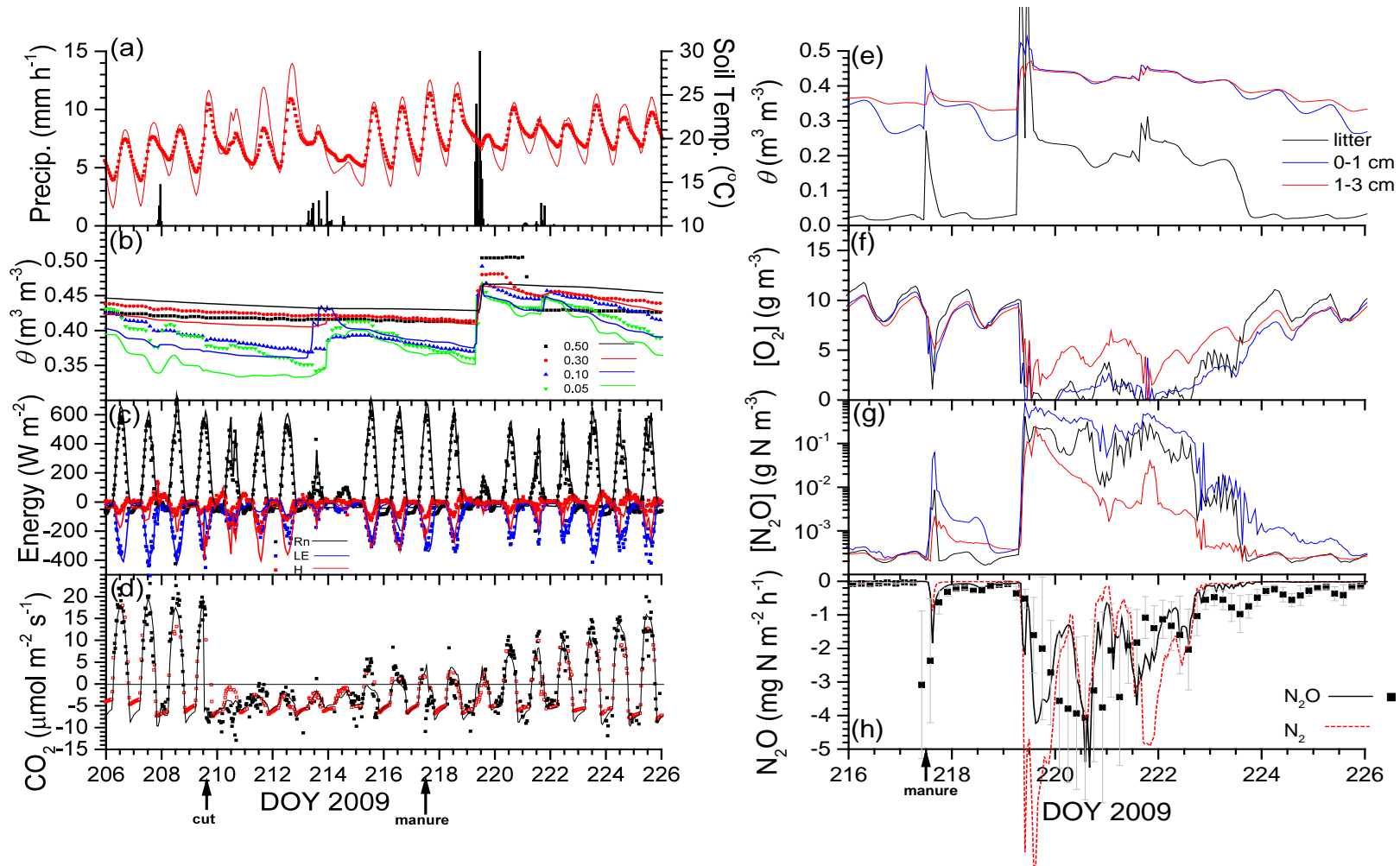


Fig. 5. (a) Precipitation and soil temperature at 0.05 m, (b) soil water content (θ) at 0.05, 0.10, 0.30 and 0.50 m, (c) energy and (d) CO₂ fluxes measured (closed symbols), gap-filled (open symbols) and modelled (lines) during 20 days from harvest (cut) to the end of the emission event following manure application (manure) in August 2008. (e) θ , (f and g) aqueous concentrations of O₂ and N₂O modelled in the surface litter and at 0.01 and 0.02 m in the soil, and (h) N₂O and N₂ fluxes measured (symbols) and modelled (lines) during the last 10 days of this period when the emission event occurred. Positive flux values represent influxes to the soil, negative values effluxes to the atmosphere.

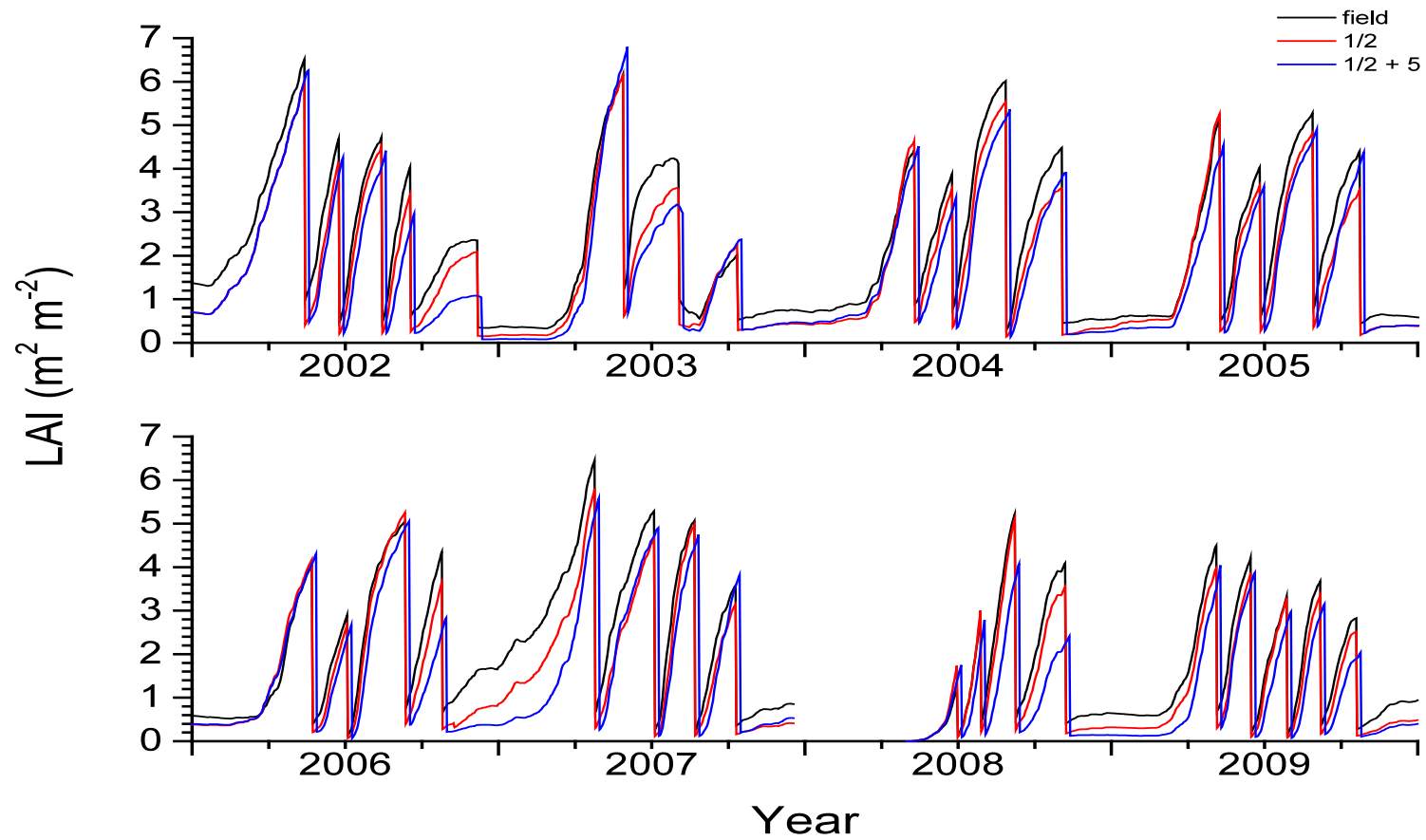


Fig. 6. LAI modelled from 2002 through 2009, with LAI after each cut reduced to one-half of that estimated from the field experiment without or with a delay of 5 days at the Oensingen intensively managed grassland.

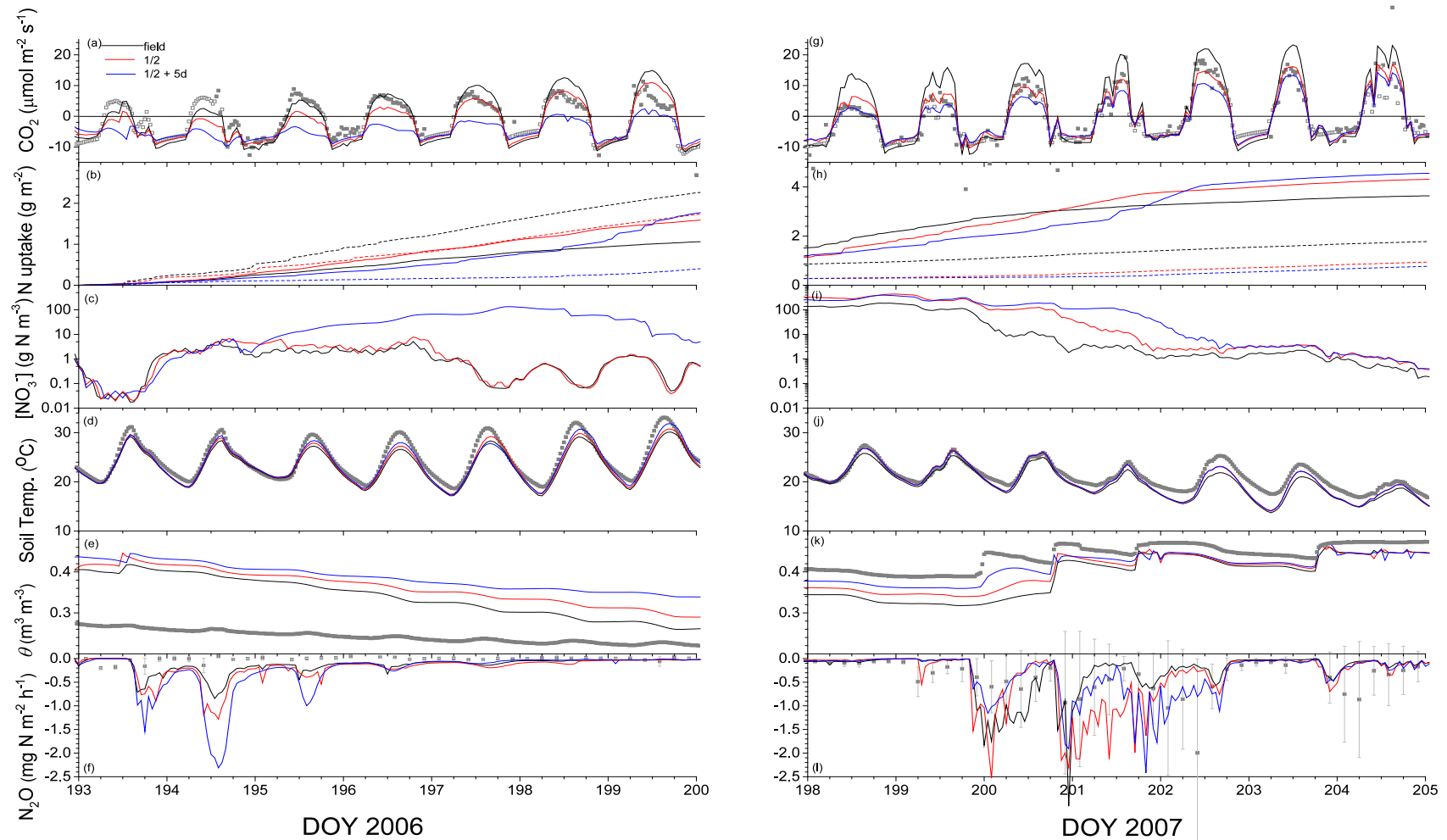


Fig. 7. (a,g) CO_2 fluxes, (b,h) cumulative NH_4^+ (dashed) and NO_3^- (solid) uptake since manure application, (c,i) aqueous NO_3^- concentrations at 0 – 1 cm, (d,j) T_s and (e,k) θ at 5 cm, and (f,l) N_2O fluxes measured (symbols) and modelled (lines) with LAI after each cut reduced to one-half of that estimated from the field experiment without or with a delay of 5 days during emission events following manure applications on DOY 194 in (a-f) 2006 and (g-l) 2007 (see Table 2). For fluxes, positive values represent influxes to the soil, negative values effluxes to the atmosphere.

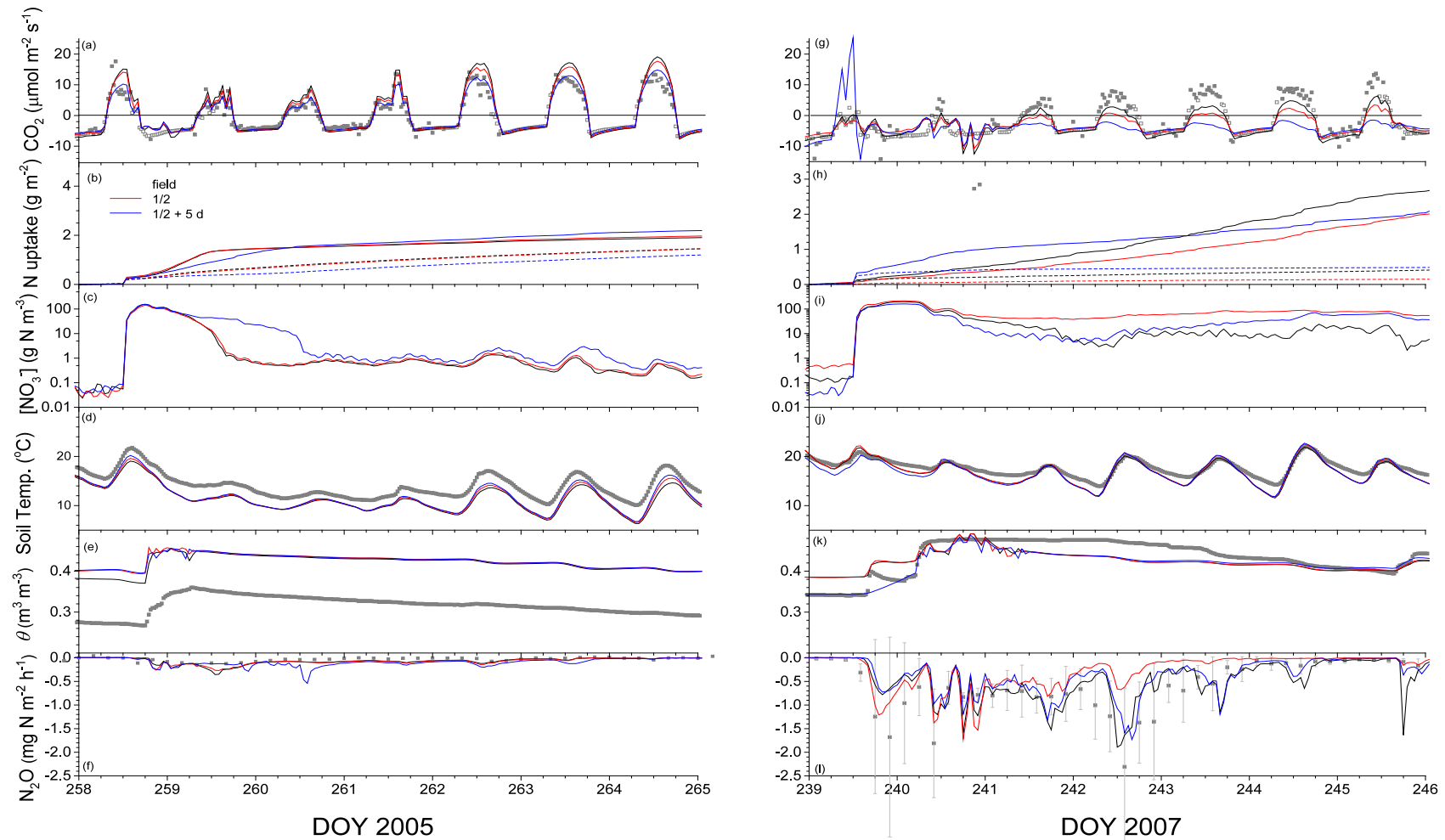


Fig. 8 (a,g) CO_2 fluxes, (b,h) cumulative NH_4^+ (dashed) and NO_3^- (solid) uptake since fertilizer application, (c,i) aqueous NO_3^- concentrations at 0 – 1 cm, (d,j) T_s and (e,k) θ at 5 cm, and (f,l) N_2O fluxes measured (symbols) and modelled (lines) with LAI after each cut reduced to one-half of that estimated from the field experiment without or with a delay of 5 days during emission events following fertilizer applications on DOY 259 in 2005 (a-f) and DOY 240 in 2007 (g-l) (see Table 2). For fluxes, positive values represent influxes to the soil, negative values effluxes to the atmosphere.

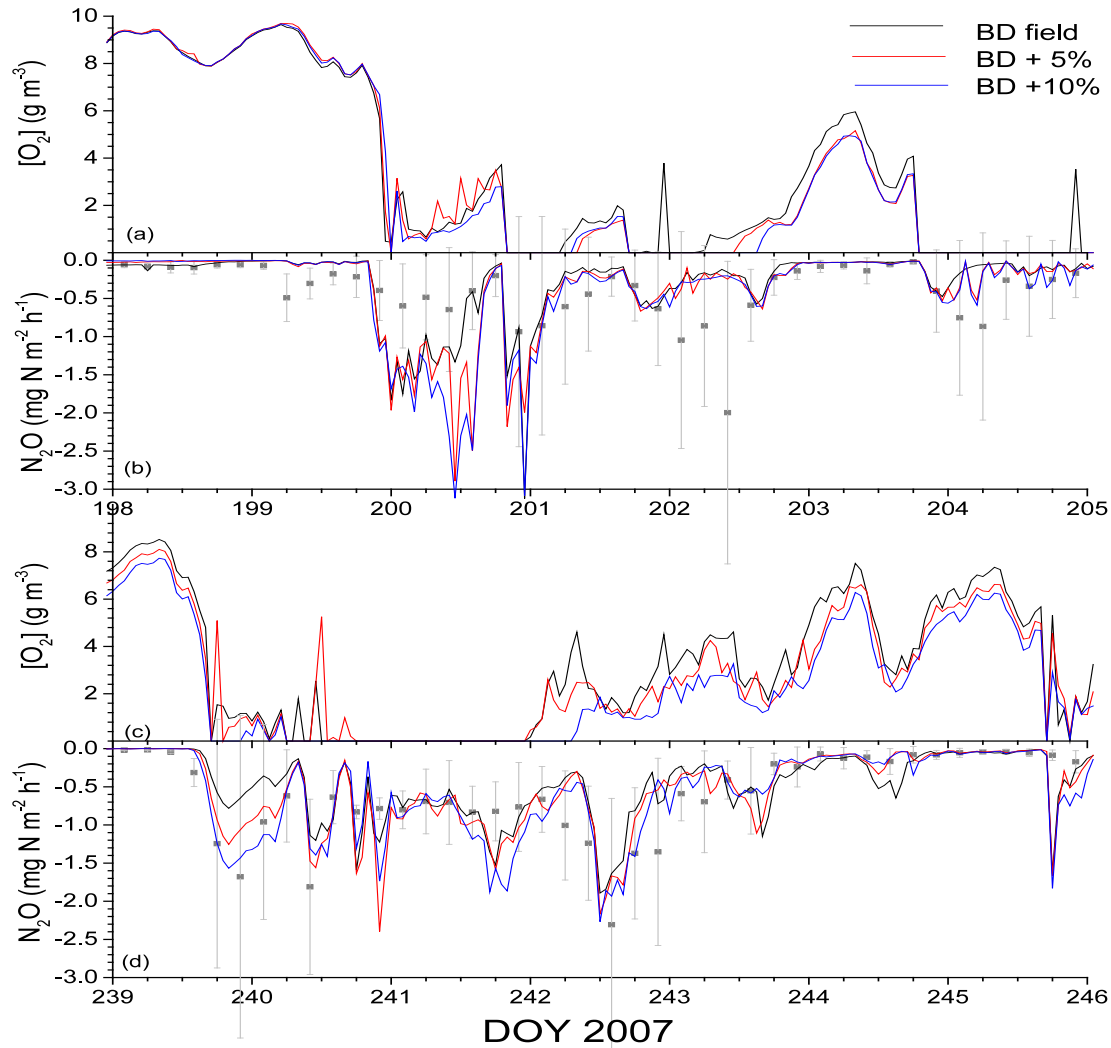


Fig. 9. (a,c) Aqueous O_2 concentrations, and (b,d) N_2O fluxes measured (symbols) and modelled (lines) with bulk density (BD) from field measurements, and with BD raised by 5% or 10% following (a,b) manure application on DOY 194 and (c,d) fertilizer application on DOY 240 in 2007 (see Table 2). For fluxes, positive values represent influxes to the soil, negative values effluxes to the atmosphere.