

Dear Editor,

On behalf of all co-authors, I would like to thank you very much for handling the paper and the reviewers for their comments which help us to improve the manuscript in terms of presentation as well as scientific contents. We very much appreciate the effort and amount of time spent by the reviewers to review this manuscript. We agree with most of the concerns of the reviewers and addressed in the revised manuscript. Three major suggestions by the reviewers are as follows:

1. Clumped isotope effect associated with photosynthesis especially in relation to the findings in Eiler and Schauble (2004) and explore the ways to reconcile the findings from the two studies.
2. Application of Keeling plot for source identification when the source and sink coexist. Discussions about δ^{47} results are mostly concluded to "unknown" enzymatic reaction during photosynthesis. Therefore, any quantitative discussion, such as estimating individual fluxes from/to the urban CO_2 , is not offered.
3. The discussion about the clumped isotope effect is mostly qualitative. The reviewer suggested to construct a simple (semi-) quantitative model to simulate the evolution of the concentration and isotopic composition of CO_2 in the greenhouse experiments.

Our response to the first query is as follows: We will elaborately discuss the effect of photosynthesis on the clumped isotope signatures in the residual CO_2 and compare our findings with that of the Eiler and Schauble (2004). We are gathering more data at leaf level which will help to understand the effect of photosynthesis on clumped isotopes, and the results will be presented in a future publication. Meanwhile we will link more the ambient CO_2 results to the greenhouse data which we have learned a lot more thank to a better controlled environment.

To reply to the second query, we agree with the reviewer that the identification of source using Keeling plot in the cases where both source and sinks co-exists is not valid. However, in most of the cases either source or sink is dominant. e.g., in the case of greenhouse, day time is dominated by photosynthesis and night time by respiration. In the revised manuscript, we will also take the photosynthesis into account and do the appropriate calculation to identify the source.

To reply the third query, we want to assure that more in-depth discussion will be included in the revised manuscript. Some simple modelling works were carried out with the traditional isotopes; we will extend this for clumped isotopes in the revised manuscript.

We have also made substantial modifications following reviewers' comments and suggestions. We already published another manuscript recently where we discussed all details about the data quality, details about the clumped isotope measurements including CO_2 purification and cited at appropriate places.

Below, please find our point-by-point response to referee's comments (referee's comments are in italics).

Sincerely yours,

Mao-Chang Liang
Academia Sinica

Anonymous Referee #1

This study provides excellent dataset for almost all of CO₂ isotopologues in the atmosphere.

Air samples were collected quite extensively, from open ocean, coasts, mountain, forest, grassland, sub-urban and urban traffic. Moreover, closed terrarium experiment and collecting exhaust from cars were conducted as well. Research plan and obtained results are very nice. While authors provides very valuable dataset, the individual discussion seems not always nice. My major comments on their discussion are; 1) They apply Keeling plot to most cases for source identification. If the case is simple two-source mixing, Keeling plot must be effective. However, this is generally not applicable for the case that source and sink coexist, except that both are the same isotopic composition (fractionation) and fluxes. I guess greenhouse experiment and grassland observation may be the cases. When Keeling analysis does work well, then authors seek the reason of inconsistency and develop some discussion. Some of these discussions are not so effective. Authors should pay attention that Keeling plot is not a universal tool. 2) On a related matter of 1), developed discussions about D₄₇ results are mostly concluded to "unknown" enzymatic reaction during photosynthesis. Therefore, any quantitative discussion, such as estimating individual fluxes from/to the urban CO₂, is not offered. Another approaches may be possible, I guess.

We thank the reviewer for appreciating the data. We agree with the points raised by the reviewer and answered the query at the beginning where we summarised the major points. We will provide a detail assessment in the revised manuscript.

I think this manuscript is worth-publishing to the journal Biogeosciences after addressing specific comments supplied as a separate file. Specific comments involve these issues, too.

Please also note the supplement to this comment:

<http://www.biogeosciences-discuss.net/bg-2016-106/bg-2016-106-RC3-supplement.pdf>

We agree with most of the comments and modified the manuscript accordingly. Point to point reply of the queries of the reviewer are given below.

L56, 85, 110, 301, 427, 482, 583: Authors used a term "bulk" for d₁₃C and d₁₈O, implying that clumped isotope (D₄₇) is not bulk isotopic composition. To my knowledge, the term "bulk" is often used to distinguish between "weighted-average (bulk) isotope ratio of a material" and "compound-specific isotope ratio of a material", such as d₁₃C for "organic matter" versus "protein, lipids, sugar, etc."; or "weighted-average (bulk) isotope ratio of a compound" versus "position-specific isotope ratio of a compound" such as d₁₃C of long-chain

hydrocarbon for "all carbon" or "1,2,3,4,...,n th carbon". In this sense, D₄₇ of CO₂ is also "bulk", or D₄₇ is neither part of d₁₃C or d₁₈O but a integration of d₁₃C and d₁₈O to some extent. Thus I think authors should avoid using the term "bulk" for d₁₃C and d₁₈O. Instead, "conventional", "traditional" or without any adjective may be better.

We agree with the terminology of the reviewer and modified in the revised manuscript.

L68: "Evapotranspiration" should be replaced to "transpiration."

Done

L68-69: This sentence is out of context. I guess it may follow the sentence of L58-62.

Modified in the revised manuscript

L76-79: Contextually, this sentence should describe about d₁₈O. But this sentence mentions general characteristics of CO₂, not only d₁₈O. Revise or move it to more appropriate place. In addition, I request one or more references to mention that present biogeochemical models remain inconclusive.

We agree with the reviewer, this is a general statement and removed from the revised manuscript

L85-87: *"..limited because of the challenge.." Somewhat strange. "..limited due to the demand of very high precision.." or "..limited but several challenges have conducted to apply it to the atmospheric study.." might be more suitable.*

The sentence is modified

L92: *"..have similar time-scales for the isotope exchange between CO₂ and water.."*

The sentence is modified

L91-94: *I agree that effect of photosynthesis and respiration on clumped isotope has not been studied well, but I disagree that corresponds to d18O as well. At least, their backgrounds are not equal.*

We agree with the reviewer about d18O and modified the sentence accordingly

L108-109: *One or more references are necessary.*

Additional references are provided

L117-119: *This is concluding remark. Move it to conclusion.*

Done

L123: *Delete "amu"*

Done

L130: *2 L; 2 atmospheric pressure*

Done

L130-133: *I could not understand collection procedures well. Was the flask flushed out prior to sample collection without dehumidifier before collection? I believe such kind of pre-process for flushing should be done with identical condition to sample collection. How long did you take for actual sample collection except for pre-flushing?*

Yes, flasks were flushed out prior to sampling for ~10 minutes and flushing was done through the perchlorate (dehumidifier) column. The flasks were equipped with two stopcocks and after flushing the end stopcock was closed and allowed the pressure to build to 2 atm and then isolated by closing the other stopcock. This is described briefly in the revised manuscript. We also refer the details to our previously published papers such as Liang and Mahata (2015).

L139: *What is "systematic analyses"?*

“Systematic” refers to the study performed systematically, i.e., more regular and intensive sampling. To remove possible confusion, the word systematic is removed.

L141-142; L146-147: *Just to recommend, "..5 m high. It was closed at least one day before each experiment and the ventilation was kept as minimum as possible."*

Done

L150-155: Add the height of the canopy.

Done

L155-157: Add each sampling height above sea level.

Done

L169-174: If you used a vacuum line, add which process is in vacuo. If not, I'm sorry.

Yes, we used vacuum line. CO₂ was extracted from air using a glass vacuum line connected to a turbo molecular pump by cryogenic technique. The vacuum line as well as the sample flask connection assembly including its head space was pumped to high vacuum before starting the CO₂ extraction. The details are mentioned in the revised manuscript. We also refer the details to our previously published papers such as Liang and Mahata (2015).

L194-196: Specify the names of the standard (VPDB, VSMOW, etc., for each).

Done

L212: What is "this limit"?

The limit here refers to the full scrambling state. In this revised version, we replaced the term by “random distribution.”

L217-219: Just to recommend, "Masses 48 and 49 were monitored to confirm isobaric interferences due to contamination of hydrocarbons (Ghosh.)."

Modified the sentence

L221-233: Refer Yoshida et al. (2013) RCM27, 207-215, for the evidence of independence from d47 on D47.

Dependence of d47 on D47 varies from mass spectrometer to mass spectrometer. Therefore, this is not relevant, we will discuss this in the supplementary of the revised manuscript.

L235, 237: I am not so sure whether this term is really appropriate or not, however "empirical transfer function" is often based on the field observation, such as marine foraminifer community structure versus habitat temperature. Authors obtained a relation experimentally, thus I think "reference frame equation", "laboratory equation" or "local equation" should be more appropriate instead of empirical transfer function.

Though “empirical transfer function” is used by Dennis et al. (2011), we agree with the reviewer that the “reference frame equation” is more appropriate.

L237-239: Authors need not to discuss in detail, but should compare their results with former study.

The reference frame equation varies between mass spectrometer to mass spectrometer, even it differs for a given mass spectrometer at different time. It is known to the community.

L245: The 1-sigma values of d13C and d18O seem too large whereas that of d47 seems in agreement with previous studies (Table S1). Huntington et al. (2009) described that d13C or d18O uncertainties were roughly an order of magnitude better than d47, because of those higher abundance. Actually, Yoshida et al. (2013) showed these lower uncertainties accordingly. To my knowledge, in any way, if one measures d13C with [44] signal of 12V and integration time of 2.5 hour, the standard deviation may be better than 0.01 permil, not only for single gas but also for several aliquots. Actually, results of CO2 digested from carbonates (Table 1) are similar accordingly. Do you have any idea why uncertainties of cylinder CO2 became so high, or d47 uncertainty became lower relatively?

For carbonates, it is possible to achieve a std. dev. of 0.01 (Table 1). For air CO2 (compressed cylinder air or atmospheric air), handling/purification worsen the precision. Though efforts have been put (see Liang and Mahata, 2015, for example), the best precision we can get so far for d13C and d18O is ~0.05 per mil. The precision we agree that is not sufficient for CO2 long term monitoring, but is sufficient for the current study. Possible cause is likely that slight fractionations during the extraction cause this variation in d13C and d18O. However, this possible fractionation does not impair the D47 analysis.

L250-252: Add references for demonstrating poor consensus.

Dennis et al., (2011), the inter-laboratory comparison shows D47 values NBS-19 from 0.373 to 0.404‰ for three laboratories, reference added in the revised manuscript.

L254-255: Not only showing deviations from expected temperature, specify the reproducibility of D47 thermometry.

Done

L267-272: Lack of data source of temperature at South China Sea.

Actual measurements during sample collection, mentioned in the revised manuscript.

L276: Diurnal variation..

Corrected

L282: Define Keeling plot and describe its purpose before the first use for readers from different fields.

A brief description of Keeling plot and purpose is incorporated in the revised manuscript.

L288: What is expected (potential) contamination of anthropogenic CO2 in the greenhouse?

The potential contaminants are the ambient air with significant anthropogenic components which was found absent from [CO2] and all the isotope signatures.

L296: What does "daytime" correspond? Daytime on 12th October? Or other three days?

It is from morning 9 am to evening 5 pm, statement is modified in the revised manuscript.

L297-299: The criteria of separation between weak/strong for photosynthesis or respiration in Fig. 4 is quite unclear. It seems very arbitrary. Define it clearly, otherwise delete this sentence and Fig. 4.

By weak photosynthesis we wanted to mean that the photosynthetic activity was reduced artificially. This was done by covering the greenhouse with a double layered black cloth on a dark cloudy day. This is more clearly explained in the revised manuscript.

3.2: Catalytic converter in the exhaust plays a role to convert CO to CO₂. Is there any possibility this catalytic reaction may affect d18O value as same as D47, not only by exchanging oxygen with water?

The change in d18O in the exhaust was also observed (Sec. 3.2). We are not aware of any process other than exchange of oxygen isotopes between CO₂ and condensed water which can cause the change in the d18O or D47 of the exhaust CO₂.

3.3: This section should be divided into each field and reorganize to avoid confusion. For example, marine (including SCS and coastal sites), urban (Roosevelt Road), sub-urban (AS), grassland (NTU) and mountain. I guess authors might confuse a bit. For example, the relations between CO₂ (1/CO₂) and d18O as well as d13C and d18O for grassland are significant (regressions were done with data from Table 5), unlike its statement found in

L346-349. Incidentally, the order to explain d13C and d18O results is marine, urban, sub-urban, grassland, mountain then forest. On the other hand, that to explain D47 results is marine, sub-urban, grassland, forest, mountain then urban. Easy to confuse.

We reorganized the section and presentation is consistent in the revised manuscript.

L314-328, L368-372: These should be reorganized as a separate section "marine CO₂" for example.

Done

L330-333, L386-390: These should be reorganized as a separate section "urban CO₂" for example.

Done

L333-339, L372-376: These should be reorganized as a separate section "sub-urban CO₂" for example.

Done

L339-349, L376-379: These should be reorganized as a separate section "grassland CO₂" for example.

Done

L353-357, L379-384: These should be reorganized as a separate section "forest CO₂" for example.

Done

L314-328: The analysis based on the Keeling plot and subsequent source identification may be problematic. First, authors did not clarify whether the ocean of the study area/period is source or sink of CO₂. Second, data range both of CO₂ and d13C are narrow and number of data is limited, thus intercept of regression line must have large uncertainties. Therefore, some sentences from L324 to 328 and associated discussion in Section 4 may not be so meaningful. Moreover, authors should consider marine air interacts with ocean surface layer

(mixed layer), not with deep ocean directly. The inconsistency between opaque Keeling intercept and $d_{13}C$ value from unconnected deep ocean is not surprising at all.

We agree with the reviewer about the application of Keeling plot with a few data points covering a small range. The region is a net source of CO₂ in the atmosphere, discussed in the revised manuscript. We put less emphasis on the Keeling plots over the ocean in the revised manuscript.

L331: 39.32 instead of 39.319

Done

L332-333: *The average $d_{18}O$ value is not different significantly from that of grassland, thus this explanation is partly incorrect.*

The mean values are different though the uncertainty associated with the values is large. $d_{13}C$ values are significantly different, but it is difficult to conclude based on $d_{18}O$ as mentioned in the later part of the section. The statements are modified in the revised manuscript.

L344-345: *I agree with this conclusion, however not by the result from Keeling plot, but by strong relations of CO₂- $d_{18}O$ and $d_{13}C$ - $d_{18}O$ as mentioned above. D47 result may support this, thus I would like to emphasize that all results from same field should be described at once (in same block), should not be separated. However, this kind of concluding remark is supposed to be in the discussion.*

We agree that this should be discussed as a block in the discussion, modified in the revised manuscript.

L346-349: *I totally disagree with this sentence. Authors should verify data again.*

Away from, for example, significant anthropogenic sources, due to presence of a variety of water sources (leaf water, soil water, etc), correlation between $1/[CO_2]$ and $d_{18}O$ is always not observable.

L358-367: *This block and Fig. 6 may not be necessary.*

This paragraph along with Fig. 6 has been removed from the revised manuscript.

4: *The section and order of description is inconsistent with Results. This prevents readers from moving on smoothly. Consider above mentioned comment and reorganization.*

We thank the reviewer for the suggestion. This section is totally reorganized in the revised manuscript.

L400-418: *These blocks should move to introduction.*

Removed from the revised manuscript

L422: *"biological" instead of "biogeochemical"*

Done

L437-446: *The obtained fractionation factor of -15.3, which is significantly different from expected C₃-type fractionation, clearly demonstrated that this calculation is not applicable to the photosynthesis-respiration coexisting process. Authors should consider the different approaches. For example, assuming constant respiration rate for whole day (applying night time respiration rate to daytime), then obtaining gross productivity.*

We agree with the reviewer that the calculation should include respiration also. We modified our calculation assuming a constant respiration and presented the estimate in the revised manuscript.

L446-454: Describe how consistent with previous studies. Consider same calculation mentioned above.

Calculation is modified as per the suggestion and the calculated d18O discrimination has been compared with previous studies in the revised manuscript.

L455-489: Authors demonstrated that d18O of respired CO₂ is out of equilibrium with ambient temperature (water is supposed to have constant value, thus disequilibrium is due to temperature variation). If so, D₄₇ of respired CO₂ must be always out of equilibrium as well unless d13C is disequilibrium in a same manner (difficult to postulate due to the different fractionation process). However, authors mentioned that respired CO₂ is in equilibrium with temperature because data in the early morning or night-time show close to equilibrium. This is a contradiction in principle. With keeping this contradiction, authors developed further discussion with respect to catalytic reaction. I cannot say whether the discussion is correct or not, however, I can say authors ignores a significant contradiction in the same block. Temperature change during night-time and cloudy (sun-shaded in addition) daytime were small whereas sunny days had wide range of temperature. Simply considering, larger magnitudes of disequilibrium during sunny daytime may be attributed this large temperature variation. Alternatively or additionally, authors had better consider that air temperature may be different from body temperature inside leaves. Plants have homeostatic function with respect to temperature, a transpiration. CO₂ is respired inside the leaf in partial isotope equilibrium with body temperature, not ambient temperature. I believe authors could develop much more deep and quantitative discussion with data shown in this study, before measuring clumped isotope of O₂.

We did not say that d18O of respired CO₂ is out of equilibrium. We only showed that the respired CO₂ is in thermodynamic equilibrium with the leaf and soil water using the obtained D₄₇ values. We agree with the reviewer that the plant body temperature could be different from the air temperature but with progress of the day we expect change in the D₄₇ values. As stated in the later part of this section, this needs to be tested at leaf level which we are planning and hopefully, the results will help to understand/model the effect of photosynthesis on the D₄₇ values.

L469: Remove "we believe"

Done

L474: Yeung et al., 2015).

Done

L490-498: This block should move to Summary.

Done

L501-513: As mentioned above, I find it difficult to understand why authors would like to link atmospheric CO₂ to respired CO₂ in the deep ocean. I think this is unnecessary, and recommend to remove entire this block.

We agree with the reviewer to remove this paragraph from the revised manuscript

4.3: As mentioned above, authors had better consider the possibility of catalytic reaction between CO and CO₂ at the converter.

Yes, reaction between CO and CO₂ inside the catalytic converter at the temperature of the converter could also lead to the change in the D47 values, though this would not change in the d18O values as the source of O₂ in both CO and CO₂ is the atmospheric O₂. This is discussed in the revised manuscript.

4.4: Authors gave f , anthropogenic contribution, in the two-source component equation from the difference between observed (urban) and marine CO₂. This assumption ignores photosynthetic uptake or influence of other sources completely. Authors should get f by solving simultaneous equations based on the concentration and isotopic composition, conversely, then discuss. This approach may be more purposeful, quantitative and premised (why isotope study is needed).

We agree with the reviewer that a more quantitative estimate for CO₂ cycling fluxes between reservoirs is possible. However we note that for example, atmospheric transport, that we mentioned at the end of the section, can easily interfere the calculation (box model interpretation, for example). This is the main reason that we give a more quantitative assessment for the greenhouse data, but not ambient CO₂ data.

5 or new 4.6: A trial to estimate individual fluxes of combustion, respiration and photosynthesis for C3 and C4, respectively, from/to the urban (or sub-urban) CO₂ is very welcome by using [CO₂], d13C, d18O and D47.

Please see the previous response. We agree that the multiple CO₂ isotopologues can help to constrain the CO₂ fluxes of combustion, respiration and photosynthesis for C3 and C4, etc. However, incomplete knowledge on meteorological influence and lack of systematic dataset around the region prevent us from full assessment. From the available data presented, we showed that D47 behaves differently from [CO₂], d13C, and d18O. To minimize regional and/or global interference (due to atmospheric transport, for example), we use greenhouse as a testbed for assessing the associated biological CO₂ fluxes. For combustion, there are other tracers more useful than the presented CO₂ isotopologues, such as VOCs and 14C.

Fig. 1: Detail map of collection site in the Taipei city is desirable instead of right panel. Coastal and mountain sites can be involved into the left panel.

Done

Fig. 3C: Although there appears a fair negative relation between d18O and D47 in Figs. 3A, B and D, coordinated rapid drops subsequent increases of these values are found on 4th August (3C) as well as 31st July. Do you have any idea what happened at these periods?

Actually the correlation is significant only in Figure 3D. The reason for the rapid decrease in the D47 values in the early in response to photosynthesis is not very clear. We are doing more study at leaf level to identify the possible cause.

Fig. 4: As mentioned above, the criteria to separate A and B is unclear.

Here we wanted to show that D47 values are similar to that expected thermodynamically when respiration is strong and photosynthesis is weak but not the other way round. This is elaborated in the revised manuscript

Fig. 5: Data from urban site should be added. Ocean and coastal sites can be merged.

Done

Fig. 6: This figure is unnecessary (see above).

Removed from the revised manuscript

Fig. 7: Reorganize (rearrange) according to the order of results and discussions.

Done

New Fig. 9?: The summarizing diagram for individual fluxes (schematic box diagram) is welcome.

We agree that a summarizing diagram of individual flux will enhance the presentation. However, with the present data it will be too early to assign D47 values to individual fluxes. We will keep this suggestion in mind and try in future with more data.

Table 2: Add relative humidity if available.

The data from nearest weather stations are added.

Anonymous Referee #2

The manuscript "Clumped isotopes in near surface atmospheric CO₂ over land, coast and ocean in Taiwan and its vicinity" provided a valuable dataset of clumped isotopes in atmospheric CO₂ and the authors did a good job. For the comments please see the attached file.

We thank the reviewer for appreciating our effort. All the reviewers queries from the pdf and modifications/changes made in the revise manuscript are listed below. Also the other minor suggestions such as changing present/past tenses in the sentences, deleting/adding texts in the manuscript will be made in the revised manuscript.

Line 28: The sentences should be in past tense.

Done

Line 32: Not clear which processes. mention them i.e. photosynthesis, fossil fuel combustion ...

The different processes are photosynthesis, respiration, local anthropogenic emissions, modified in the revised manuscript.

Line 33: Split the sentence

The sentence is modified

Line 34: Restructure the sentence: for example, the contribution of various sources of CO₂ on D47 ...

The sentence is restructured

Line 41: Split the sentence

Done

Line 61: Split the sentence because it is hard to follow what you mean. Maybe: ... ocean and landbiosphere. The photosyn... ¹³C in plants is higher than ...

The sentence is divided into two for making it simple and easily understandable

Line 63: It is not clear what you mean. You should explain how photosynthesis and respiration may change ¹⁸O of CO₂ in vicinity of plants, if it is what you wanted to say. Is there any discrimination against ¹⁸O during assimilation of CO₂ for photosynthesis which may lead to enrichment or depletion in CO₂ besides the leaves? In the next sentence your

explanation just shows enrichment because of evapotranspiration but what is the effect of photosynthesis? Would be this isotopic discrimination due to evapotranspiration against ^{18}O still present if the plant was not under water stress at all?

The statements are modified as follows:

$\delta^{18}\text{O}$ is used for partitioning net CO_2 terrestrial fluxes between soil respiration and exchange with the plant leaves, the exchange is enhanced by the presence of carbonic anhydrase in plants and soils (Francey and Tans, 1987; Farquhar and Lloyd, 1993; Yakir and Wang, 1996; Ciais et al., 1997; Peylin et al., 1999; Murayama et al., 2010; Welp et al., 2011). This is because $\delta^{18}\text{O}$ of CO_2 fluxes originated from soil respiration are different from that exchanged with the leaf water. $\delta^{18}\text{O}$ in soil water reflect the $\delta^{18}\text{O}$ value of the local meteoric water while leaf water is relatively enriched due to transpiration.

Line 69: need reference

Appropriate references are included

Line 71: This sentence should be in line 62 before ^{18}O is used for partitioning ...

Done

Line 79: You mean reservoirs with different ^{18}O ?

Yes, the statement is modified in the revised manuscript

Line 85: Split the sentence. You mixed many things together.

Done

Line 86-96: Very well! This makes your study unique and valuable.

We thank reviewer for appreciating the work

Line 271: Materials and Methods is good.

Thank you

Line 277: The lowest CO_2 concentration, $[\text{CO}_2]$ and the highest ...

Corrected

Line 296: equilibrium with what? split the sentence.

Thermodynamic equilibrium with the leaf and soil water, sentence modified and split.

Line 300: my suggestion: The correlation between D_47 and CO_2 ... was observed only when the photosynthesis was weak.

Suggestion implemented

Line 302: very good finding.

Thank you

Line 399: This paragraph can be deleted. It is not discussing any of the observation and measurements.

This paragraph has been removed from the revised manuscript

Line 404: The sentences after "however" are not kind of discussion. I did not get why they should be mentioned here.

These three paragraphs have been removed from the revised manuscript.

Line 418: I think the whole these 3 paragraphs should be deleted. It is not clear what you wanted to say. Even if it was like an introduction for the discussion (which is not really necessary) you should follow to emphasis on the main issues respectively to what you will mention later for example effect of photosynthesis on D47, antropogenic effects in urban regions,

These three paragraphs have been removed from the revised manuscript.

Line 477: Split the sentence

Done

Line 505: put the reference here and split the sentences.

Done

Line 507: It is better to mention the intercept value here

The intercept value is mentioned now

Line 511: write the value

Done

Line 521: refer to the fig. or value here

Done

Line 529: So D47 values in CO₂ over oceans at nights should show no deviation from thermodynamic equilibrium. Is that true? How would be this effect in coastal areas where because of shallow water aquatic plants may live as well?

Yes, there should not be any deviation in the D47 in night also. The effect of photosynthesis on clumped isotopes is observable when photosynthesis is very strong e.g., in a confined greenhouse. Probably effect is present everywhere but not detectable with the measurement precision. Therefore, in the coastal areas we expect similar D47 values as observed over the open ocean unless there is a significant CO₂ is contributed from the other sources such vehicle and industrial emissions.

Line 559: It seems logical but how? Do you have an estimation of isotopic composition of condensed water? How CO₂ isotopic composition can change? I mean CO₂ will dissolve in water but how its isotopic composition can change?

Unfortunately we do not have any measurement of the d18O value of the condensed water but it is expected to be similar to the atmospheric O₂ plus the fractionation associated with the condensation (atmospheric O₂ is used for combustion). CO₂ readily exchanges oxygen isotopes when comes in contact with water, here probably a partial exchange takes place causing the deviation from the expected d18O and D47 values.

Line 562: Split the sentence! It is hard to follow you.

The sentence is split

Line 566: mention the temperature

Temperature of the catalytic converter as well as the possible mechanism has been explained in the revised manuscript.

Line 583: reference needed

Done

Line 589: split the sentence

Done

Line 604: Can it be also less anthropogenic contribution?

It can also be due to underestimation of the anthropogenic CO₂ at the sampling spot. The regional background [CO₂] here could be lower than that assumed and the actual anthropogenic fraction of CO₂ could be higher than that assumed here. Discussed in the revised manuscript

Line 620: split the sentence.

Done

Line 625: How could be anthropogenic effects in a dense and isolated forest area?

It is very unlikely to have anthropogenic CO₂ in an isolated place, but we did not neglect a priori. Later using D47 we showed that there is no anthropogenic CO₂ in the forest site.

Line 628: I think still you cannot exclude the photosynthesis effect. Furthermore, the humidity in such condition should be high which may modify the effect of respiration and may affect the temperature. You need to discuss about these issues.

We agree that we cannot exclude photosynthetic effect. The issue is discussed more elaborately in the revised manuscript.

Line 648: You need to discuss here about the photosynthetic effect on D47 values, especially that you took the samples just above the grasses.

We agree with this suggestion and discussed more on this in the revised manuscript.

Line 657: The fig. 8 is really a good job.

Thank you

Line 671: explain it here also how and through which mechanism.

The explanation is also given here

Line 678: You did not discuss about this before. Delet this or explain the relation in the text. Why not "atmospheric CO₂ budget" instead?

This is removed from the revised manuscript and instead "atmospheric CO₂ budget" is used

Anonymous Referee #3

This manuscript reports new measurements of clumped isotope compositions of atmospheric CO₂ collected from different environments and settings. Studies of clumped isotope composition of atmospheric CO₂ were among the first applications of clumped isotope methods, but have received less attention in recent years

compared to other applications. It's great to see another focused study on this subject. The dataset presented in this study is quite extensive, and mostly confirms the major findings from previous studies. However, the conclusion the authors draw regarding the effect of photosynthesis on the clumped isotope composition of CO₂ differs significantly from previous studies, and could potentially open many research opportunities. Overall, this manuscript improves our understanding of the various controls on the clumped isotope composition of atmospheric CO₂, and can help future efforts to better constrain the atmospheric CO₂ budgets. I have several specific comments about this manuscript, as detailed below, and would recommend these issues be addressed prior to publication.

We thank the reviewer for appreciating the work. Effect of photosynthesis on the clumped isotopes of atmospheric CO₂ is an interesting finding and will be discussed more elaborately in the revised manuscript. We agree with all the concerns of the reviewer and addressed them in the revised manuscript.

Major comments:

1. Separation of N₂O from CO₂. A GC column was used to separate N₂O from CO₂ in this study. The authors showed a reasonable separation of the two in Fig. S2, but didn't mention the exact CO₂ trapping time in their experiments. It's possible the CO₂ yield was compromised in order to achieve the optimal separation of N₂O. The authors need to provide more details and discuss how the compromised yield and/or residual N₂O might affect their clumped isotope data.

We collect CO₂ during 14 – 23 min. Yes CO₂ yield was slightly compromised in order to achieve the optimal separation of N₂O. The recovery of CO₂ was always greater than 95 %. The residual N₂O is checked comparing the sample and reference 48 and 49 signals. If value of D48 is large (>5) we did not consider the measurement, either we clean the sample again or just through. When the signal voltage for 49 between sample and reference is more than 0.3 mV we also neglected those numbers. We have discussed this in a recently published article (Laskar et al. Terra Nova, 2016) and cited at appropriate place.

2. Photosynthesis effect. In their greenhouse experiments, the authors observed that the clumped isotope compositions of CO₂ were higher than what expected from thermodynamic equilibrium when photosynthesis was active. This finding is very intriguing and differs from what observed in previous studies (e.g. Eiler and Schauble 2004), where the clumped isotope compositions of CO₂ residual to photosynthesis were shown to generally decrease.

Please see a previous response where the issue is discussed.

a. Given the importance of this finding, I think the authors need to provide D48 and D49 data of their measurements to show that the elevated D47 values were not related to any contamination issues. More generally, the authors are encouraged to include all their raw clumped isotope measurement data in the electronic supplementary material of their manuscript, which is becoming a convention in the clumped isotope community.

We will include the raw data including D48 and D49 in the revised manuscript.

b. The authors need to expand their discussion about the clumped isotope effects associated with photosynthesis they observed, especially in relation to the findings in Eiler and Schauble (2004), and explore ways to reconcile the findings from the two studies.

Please see a previous response where the issue is discussed.

c. The authors did a nice job estimating the carbon and oxygen isotope fractionations associated with photosynthesis in their greenhouse experiments. But their discussion about the clumped isotope effect is mostly qualitative. The authors might want to construct a simple (semi-)quantitative model to simulate the evolution of the concentration and isotopic composition of CO₂ in their greenhouse experiments. Such a

model might enable them to quantitatively estimate the clumped isotope effects associated with photosynthesis, which would be an important contribution of this study.

Please see the previous response where the issue is discussed.

Minor comments:

1. Line 440: the authors neglected the daytime respiration when estimating the isotope effects associated with photosynthesis. They need to provide evidence to support this approach.

We agree that this is a crude assumptions, we will redo the calculations incorporating the day time respiration in the revised manuscript.

2. In section 4.1, the authors estimated the rates of respiration, photosynthesis, and CO₂-water exchange in their greenhouse experiments, in the unit of molecules cm⁻² s⁻¹. But it's not entirely clear how those values were derived. More details are needed.

We will provide more details about the calculations in the revised manuscript.