

**Associate Editor Decision: Publish subject to minor revisions (Editor review)** (14 Oct 2016) by Dr. Jean-Pierre Gattuso

Comments to the Author:

Dear Author,

Thank you for submitting a revised version of your manuscript submitted to Biogeosciences, which can be accepted for publication after minor revision. When submitting the revised version, please let me know which of the changes were not implemented, if any, and why. This will speed up final acceptance.

I look forward to seeing this paper published and thank you for considering Biogeosciences to publish these very interesting results.

Best regards,  
Jean-Pierre Gattuso  
BG editor

**Dear Dr. Gattuso,**

**Thank you for your assistance moving our manuscript closer to publication. Below are attached your recommendations, our changes and/or an explanation of why the change was not implemented.**

**All the best,  
~Amy Maas**

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- The referees asked to report the size range. In your reply, you say that the size range is already reported, as mass. Mass is not size! I appreciate that some sizes are not available but you could provide the measurements you have (length or diameter, depending on the shape of the species considered).

**Unfortunately, measurements of size were not made at the time of the respiration experiments, in the interest of minimizing any effect of handling of animals. Following the experiments, the specimens were immediately frozen for later molecular work; some of these individuals have already been used for analyses of gene expression, and we are highly reluctant to thaw any of those that remain as this would make them unusable for future molecular work. Thus, this change was not implemented.**

- I wonder how the percent saturation was calculated. For example, in the abstract you mention 130  $\mu\text{mol/kg}$  is 10%. Using the standard equation of Garcia & Gordon (1992, L&O), the  $\text{O}_2$  saturation at  $S=35$ ,  $T=5$ , depth=200 m, is 307.892  $\mu\text{mol/kg}$ . Hence, 130  $\mu\text{mol/kg}$  would correspond to 42% of the saturation value.

**We had an internal discussion about this percent saturation attribution as well. As you know, water that is fully saturated with oxygen only has 21%, the rest being made up of nitrogen. To order gas of the concentration that is appropriate for the Pacific we thus ordered 10% oxygen. This is ~48% of what would be oxygen saturation (similar to your calculation, minus the fact that we did not take into**

account the 200 m depth, just the salinity and temperature as the experiments were run at sea level). As this appears to be confusing, for clarity, we have gone back through and re-expressed the % as relative to oxygen saturation rather than total gas composition for both 21% (100% saturated) and 10% (48% saturated).

- Mention in section 2.2 that pH was measured on the total scale

**The information of pH in total scale has already been provided on line 200 in section 2.2. We have added the abbreviation to draw attention to the scale.**

- 224 and elsewhere: always use the subscript "T" when an absolute pH value is given, including in the heading of Table 2. The heading of that table should also provide the unit for temperature.

**The changes have been made.**

- Abbreviate "hours by "h"

**The change has been made**

- 272: ranged between 15 and 50 ml ... and 8 to 20 ml (similar changes needed line 323)

**The changes have been made (as well as in line 273).**

- 337-338: the symbol for mole is "mol", not "M". But in this sentence mole could be spelled out.

**The "M"s have been changed to moles.**

- 337: indicate for which species Mayzaud reported this respiratory quotient of 0.8.

**The information has been added**

- Please list citations chronologically throughout the manuscript.

**The change has been made**

- References need to be formatted as described in the instructions to authors.

**The references have been carefully checked to meet the formatting requirements of Biogeosciences.**

- Biogeosciences strongly promotes the full availability of the data sets reported in the papers that it publishes in order to facilitate future data comparison and compilation as well as meta-analysis. This can be achieved by uploading the data sets in an existing database and providing the link(s) in the paper. Alternatively, the data sets can be published, for free, alongside the paper as supplementary

information. The ascii (or text) format is preferred for data and any format can be handled for movies, animations etc...

**The respiration data is available online via the DOI provided in the text. The carbonate chemistry measurements and calculations associated with the experiments have now been included as supplementary data. The environmental chemistry measurements for the Pacific are available in BCO-DMO (Chu et al 2016) and the Atlantic data is in preparation for paper submission and will be available in BCO-DMO after paper acceptance.**

1 **The metabolic response of thecosome pteropods from the North**  
2 **Atlantic and North Pacific Oceans to high CO<sub>2</sub> and low O<sub>2</sub>**

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32 **Abstract.** As anthropogenic activities directly and indirectly increase carbon dioxide (CO<sub>2</sub>) and  
33 decrease oxygen (O<sub>2</sub>) concentrations in the ocean system, it becomes important to understand  
34 how different populations of marine animals will respond. Water that is naturally low in pH, with  
35 a high concentration of carbon dioxide (hypercapnia) and a low concentration of oxygen, occurs  
36 at shallow depths (200-500 m) in the North Pacific Ocean, whereas similar conditions are absent  
37 throughout the upper water column in the North Atlantic. This contrasting hydrography provides  
38 a natural experiment to explore whether differences in environment cause populations of  
39 cosmopolitan pelagic calcifiers, specifically the aragonitic-shelled pteropods, to have a different  
40 physiological response when exposed to hypercapnia and low O<sub>2</sub>. Using closed-chamber end-  
41 point respiration experiments, eight species of pteropods from the two ocean basins were  
42 exposed to high CO<sub>2</sub> (~800 μatm) while six species were also exposed to moderately low O<sub>2</sub>  
43 (~~1048%~~ saturated, or ~130 μmol kg<sup>-1</sup>) and a combined treatment of low O<sub>2</sub>/high CO<sub>2</sub>. None of  
44 the species tested showed a change in metabolic rate in response to high CO<sub>2</sub> alone. Of those  
45 species tested for an effect of O<sub>2</sub>, only *Limacina retroversa* from the Atlantic showed a response  
46 to the combined treatment, resulting in a reduction in metabolic rate. Our results suggest that  
47 pteropods have mechanisms for coping with short-term CO<sub>2</sub> exposure and that there can be  
48 interactive effects between stressors on the physiology of these open ocean organisms that  
49 correlate with natural exposure to low O<sub>2</sub> and high CO<sub>2</sub>; these are considerations that should be  
50 taken into account in projections of organismal sensitivity to future ocean conditions.

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59 **Key Words: ocean acidification, zooplankton, respiration**

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61 **1. Introduction**

62 Ocean acidification, a result of the dissolution of anthropogenically-produced carbon dioxide  
63 (CO<sub>2</sub>) into sea water, is increasingly considered to be one of the most pervasive human changes  
64 to the marine system (Halpern et al., 2008; Doney et al., 2009; Gruber, 2011). The pH of the  
65 ocean surface has already dropped by ~0.1 units relative to preindustrial levels and is predicted  
66 to drop another 0.3-0.4 pH units in the next one hundred years (Haugan and Drange, 1996; Bopp  
67 et al., 2013; IPCC, 2013). As CO<sub>2</sub> dissolves in the ocean, it causes changes in seawater carbonate  
68 chemistry, notably increasing hydrogen ion concentration and decreasing the concentration of  
69 carbonate ions. As a consequence of the changing equilibria, there is a reduction in pH and in the  
70 saturation state of calcium carbonate (CaCO<sub>3</sub>), including the biogenic forms of calcite and  
71 aragonite. In some regions, as ocean acidification continues, the water becomes undersaturated  
72 and corrosive, meaning that, in the absence of compensating biological action, conditions will  
73 favor the dissolution of the CaCO<sub>3</sub> found in the shells and skeletons of calcifying organisms,  
74 with aragonite being more sensitive than calcite (Millero, 2007).

75 Ocean acidification, therefore, impacts calcifying species on multiple fronts. Changes in  
76 environmental pH can modify the acid-base balance of intra- and extracellular fluids of marine  
77 organisms, which may result in reduced fitness or outright mortality (Seibel and Walsh, 2001;  
78 Seibel and Fabry, 2003; Widdicombe and Spicer, 2008). Perturbations of seawater carbonate  
79 chemistry can also affect the ability of some calcifying animals to create and maintain calcium  
80 carbonate structures with implications for energetics, survival, competition and biogeochemical  
81 export (Riebesell et al., 2000; Fabry et al., 2008; Ries et al., 2009). Understanding the long-term  
82 effects of this increase in ocean acidity on both organisms and ecosystems has, therefore, become  
83 of great concern. Important and outstanding research goals are to understand how changing CO<sub>2</sub>  
84 impacts current populations and to predict whether these populations will be able to adapt to the  
85 rate and severity of the rising anthropogenic CO<sub>2</sub> inputs (e.g. Sunday et al., 2011; Dam, 2013;  
86 Kelly and Hofmann, 2013).

87 One approach to understanding the response of marine animals to acidification is to  
88 examine places where animals already experience conditions of elevated CO<sub>2</sub> (hypercapnia). By  
89 comparing individuals that inhabit regions of high CO<sub>2</sub> with those that never experience high  
90 levels naturally, insight can be gained into the potential for adaptation of species to high CO<sub>2</sub>  
91 over evolutionary timescales. The ocean chemistry of the northwest Atlantic and the northeast

92 Pacific Oceans provides such a natural experiment. High CO<sub>2</sub> concentrations are generally  
93 absent from the upper water column in the Atlantic (Wanninkhof et al., 2010). In contrast there  
94 currently are hypercapnic conditions, where the water is undersaturated with respect to aragonite,  
95 in the upper water column in parts of the Pacific.

96         The source of hypercapnia in the Pacific Ocean is a combined result of ocean circulation  
97 coupled with the biological processes, leading the old deep waters of the Pacific to be some of  
98 the most CO<sub>2</sub> rich in the ocean (Broecker et al., 1982). On top of this natural process, ocean  
99 acidification also plays a role: the pH of the upper water column in the North Pacific is  
100 decreasing by ~0.002 pH units per year (Byrne et al., 2010; Chu et al., 2016), similar to the  
101 global average of 0.0022 pH units per year (Williams et al., 2015). Such a change corresponds to  
102 a total CO<sub>2</sub>, or dissolved inorganic carbon (DIC), increase of 1–2 μmol kg<sup>-1</sup> yr<sup>-1</sup> (Peng et al.,  
103 2003; Sabine et al., 2008; Sabine and Tanhua, 2010; Chu et al., 2016). Although the surface  
104 waters in these regions are typically well oxygenated and with a pH > 8, animals that live at or  
105 migrate to depth experience increasingly low oxygen (O<sub>2</sub>), pH, under-saturation with respect to  
106 calcium carbonate, and elevated CO<sub>2</sub> (Seibel, 2011). Historically these regions, which occur in  
107 many ocean basins, were in fact known more for their low O<sub>2</sub> than for their high CO<sub>2</sub> and were  
108 termed oxygen minimum zones (OMZs). These carbon maximum/oxygen minimum zones are  
109 extensive in the North Pacific Ocean, whereas similar conditions are rare in much of the Atlantic  
110 (Paulmier et al., 2011). Closely related taxa and cosmopolitan species in these two regions  
111 therefore experience very different pH levels as well as CO<sub>2</sub> and O<sub>2</sub> concentrations in their  
112 normal distribution. Independent from high CO<sub>2</sub>, the reduced O<sub>2</sub> at depth in these OMZs has a  
113 profound impact on zooplankton distribution (i.e.: Wishner et al., 2008; Escribano et al., 2009;  
114 Maas et al., 2014) and can have important implications for the physiology of zooplankton  
115 (Childress and Seibel, 1998; Rosa and Seibel, 2008; Seibel, 2011).

116         Thecosome pteropods are an interesting group for investigating planktonic exposure and  
117 response to hypercapnia and low O<sub>2</sub>. Broadly distributed throughout the open ocean, species of  
118 thecosomes found in shallow waters of temperate and polar seas can become a numerically  
119 dominant member of the zooplankton community (van der Spoel, 1967; Hunt et al., 2008;  
120 Bednaršek et al., 2012a). As such, they can be an important part of the food chain (Armstrong et  
121 al., 2005; Hunt et al., 2008; Karnovsky et al., 2008), and contribute substantially to carbon flux  
122 (Fabry and Deuser, 1991; Noji et al., 1997; Bauerfeind et al., 2009; Manno et al., 2010). Bearing

123 thin shells of aragonite, one of the less stable forms of biogenic calcium carbonate, the  
124 calcification of thecosomes has been shown to be impacted by exposure to conditions replicating  
125 the projected changes in surface water pH and saturation state of the future ocean in the next 100  
126 years (Comeau et al., 2009; Lischka et al., 2011; Manno et al., 2012). Furthermore, recent  
127 assessments have shown that their shells are degraded in upwelling and polar regions  
128 characterized by under-saturated conditions with respect to aragonite (Bednaršek et al., 2012b;  
129 Bednaršek et al., 2014; Bednarsek and Ohman, 2015). Studies of metabolism and behavior,  
130 however, reveal a complex sensitivity to pH, dependent upon natural pre-exposure and the  
131 presence of interactive stressors (Comeau et al., 2010; Maas et al., 2012b; Manno et al., 2012;  
132 Seibel et al., 2012).

133 Previous work has shown that some tropical and sub-tropical thecosome species undergo  
134 diel vertical migrations into persistent and pronounced regions of low O<sub>2</sub> and hypercapnia in the  
135 Eastern tropical North Pacific. These species showed no change in metabolic rate (O<sub>2</sub>  
136 consumption) when exposed to high CO<sub>2</sub> (1000 µatm), revealing the ability of some groups of  
137 thecosome to maintain aerobic metabolism in acidified waters for short periods of time. The one  
138 species in the region that does not migrate, however, responded with a suppression of  
139 metabolism when exposed to high CO<sub>2</sub> (Maas et al., 2012b). This work in the Eastern tropical  
140 North Pacific provides evidence that there may be the potential for environmental adaptation of  
141 thecosomes to high CO<sub>2</sub>, but provides no insight into the combined effects of CO<sub>2</sub> with low O<sub>2</sub>.  
142 Although research into this topic is underway for other calcifying organisms in coastal habitats  
143 (Melzner et al., 2013; Gobler et al., 2014), in the open ocean our understanding remains limited.

144 The objective of this study, therefore, was to compare the effect of high CO<sub>2</sub> and low O<sub>2</sub>  
145 on thecosome pteropods from the northwest Atlantic and the northeast Pacific Oceans. One of  
146 the benefits of this comparison is that there are a number of species of thecosomes that have  
147 cosmopolitan distributions occupying both basins and that are known to be diel vertical  
148 migrators (Table 1; van der Spoel, 1967; Bé and Gilmer, 1977). Thus populations in the Pacific  
149 would naturally experience hypercapnia and low O<sub>2</sub> in their daytime deep habitat in the Pacific,  
150 while in contrast, those from the Atlantic would rarely experience the same environmental  
151 stressors. The taxonomy of thecosomes has recently begun to be revisited using molecular and  
152 paleontological tools (i.e. Hunt et al., 2010; Jennings et al., 2010; Janssen, 2012; Maas et al.,  
153 2013) and there is growing evidence of cryptic speciation for some pteropod groups (Gasca and

154 Janssen, 2014; BurrIDGE et al., 2015). It thus should be noted that the inter-basin comparisons  
155 performed here may be of cryptic congeners rather than conspecific populations. Using these  
156 organisms, which are presumably adapted to their local conditions, we can test whether species  
157 or congeners exhibit a population-specific physiological response to these environmental  
158 conditions indicative of different sensitivities.

159

## 160 **2. Methods**

161 Thecosome pteropods caught during cruises to the northwest Atlantic and northeast Pacific were  
162 exposed aboard ship to manipulated conditions of moderately high CO<sub>2</sub> and/or low O<sub>2</sub> for short  
163 durations (< 18 h). After this exposure their metabolic rates were measured and then compared to  
164 determine whether there were species- or region-specific responses to the treatments.

### 165 **2.1 Sampling**

166 Animals were collected on two cruises, the first on August 7<sup>th</sup> – September 1<sup>st</sup> 2011 in the  
167 northwest Atlantic aboard the R/V *Oceanus*, and the second in the northeast Pacific from August  
168 9<sup>th</sup> – September 18<sup>th</sup> 2012 aboard the R/V *New Horizon*. The majority of the sampling in the  
169 Atlantic took place along a three-part ‘z’-shaped transect running between 35°N 52°W and 50°N  
170 42°W, as well as at sites during transit to and from port (Fig. 1). The first portion of this cruise  
171 track corresponded to a segment of the World Ocean Circulation Experiment / Climate and  
172 Ocean: Variability, Predictability and Change project (WOCE/CLIVAR) line A20. In the North  
173 Pacific the main sampling took place along a two-part transect running between 50°N 150°W  
174 and 33.5°N 135°W, corresponding to a portion of WOCE/CLIVAR line P17N, as well as at sites  
175 during transit to and from port (Fig. 1).

176 Sampling was part of a larger interdisciplinary project employing a suite of tools to  
177 explore the natural distribution and hydrographic environment of the thecosomes. The sampling  
178 design included underway measurements of hydrography, carbonate chemistry and multi-  
179 frequency acoustic backscattering. Comprehensive sampling of the water column was conducted  
180 at pre-determined stations using a depth-stratified 1-m<sup>2</sup> Multiple Opening/Closing Net and  
181 Environmental Sensing System with 150 µm mesh nets (MOCNESS; Wiebe et al., 1985), a  
182 towed broadband echosounder, video plankton recorder casts, and profiles with a 24-place 10-L  
183 Niskin bottle rosette and associated conductivity, temperature and depth (CTD) package. This  
184 CTD was equipped with dual temperature and conductivity sensors, a Digiquartz pressure sensor,

185 a SBE43 dissolved oxygen sensor, a biospherical underwater photosynthetically active radiation  
186 (PAR) sensor with surface reference, a Wet Labs C-Star transmissometer (660 nm wavelength),  
187 and a Wet Labs ECO-AFL fluorometer.

188 Hydrographic profiles associated with this study were collected of temperature, O<sub>2</sub> and  
189 salinity using the CTD-Rosette-Niskin bottle package at stations along the main survey transects  
190 (Fig. 1). Where CTD casts were unavailable, at stations conducted during the transits to and from  
191 port, an expendable bathythermograph (XBT) was deployed to determine the temperature of the  
192 water column. Bottle samples of carbonate parameters, nutrients, and other parameters were  
193 collected at selected water depths using the CTD-Rosette package.

## 194 **2.2 Environmental Carbonate Chemistry**

195 Discrete pH samples were directly collected from the 10-L Niskin bottles into 10 cm cylindrical  
196 optical cells and measured within 4 h of collection (Clayton and Byrne, 1993; Dickson et al.,  
197 2007). These pH samples were analyzed spectrophotometrically on an Agilent 8453  
198 spectrophotometer at a control temperature ( $25.0 \pm 0.1^\circ\text{C}$ ) following the method detailed in  
199 Dickson (2007) and Clayton and Byrne (1993) using m-cresol purple as the indicator. The pH  
200 results in total scale ( $\text{pH}_T$ ) have been corrected for indicator impurity (Liu et al., 2011) and  
201 indicator perturbation to seawater samples. The pH measurements have a precision better than  
202 0.001 and an accuracy of  $\sim 0.002$ .

203 Nutrient samples (nitrate/nitrite, phosphate, silicate, and ammonia) were collected in 20  
204 mL plastic bottles after filtration through a 0.22 $\mu\text{m}$  Pall capsule filter and kept frozen until  
205 analysis. Nutrient samples were analyzed either at the WHOI Nutrient Analytical Facility or the  
206 University of California, Santa Barbara, using a Lachat Instruments QuickChem 8000 four-  
207 channel continuous flow injection system, following standard colorimetric methods approved by  
208 U.S. Environmental Protection Agency.

209 Discrete samples were also taken for dissolved inorganic carbon (DIC) and total  
210 alkalinity (TA). These were collected in 250 mL Pyrex borosilicate glass bottles after being  
211 filtered with a 0.45  $\mu\text{m}$  in-line capsule filter and poisoned with saturated mercuric chloride  
212 (Dickson et al., 2007). DIC samples were analyzed on a DIC auto-analyzer (AS-C3, Apollo  
213 SciTech, Bogart, USA) via sample acidification, followed by non-dispersive infrared CO<sub>2</sub>  
214 detection (LiCOR 7000: Wang and Cai, 2004; Wang et al., 2013). The instrument was calibrated  
215 with certified reference material (CRM) from Dr. A.G. Dickson at the Scripps Institution of

216 Oceanography. The DIC measurements have a precision and accuracy of  $\pm 2.0 \mu\text{mol kg}^{-1}$ . TA  
217 measurements were made with an Apollo SciTech alkalinity auto-titrator, a Ross combination  
218 pH electrode, and a pH meter (ORION 3 Star) based on a modified Gran titration method with a  
219 precision and accuracy of  $\pm 2.0 \mu\text{mol kg}^{-1}$  (Wang and Cai, 2004).

220 The remaining water column carbonate system parameters, including aragonite saturation  
221 state and  $\text{pCO}_2$  were calculated from DIC- $\text{pH}_T$  pairs at in situ nutrient, temperature, salinity and  
222 pressure using the software CO2Sys (Pierrot et al., 2006) and the dissociation constants of  
223 Mehrbach et al. (1973), refitted by Dickson and Millero (1987), and the  $\text{KHSO}_4$  dissociation  
224 constant from Dickson (1990). Depths for  $\text{pH}_T=7.7$ ,  $\text{pCO}_2=800 \mu\text{atm}$  and aragonite saturation  
225 state of 1 were then linearly interpolated using the closest available measurements.

226 Surface water  $\text{pCO}_2$  was continuously measured throughout both cruises using an  
227 automated underway system (Model 8050, General Oceanics Inc., USA) based on headspace air-  
228 seawater equilibration followed by infrared detection (LiCOR 7000). This system was calibrated  
229 every 1-2 hours with three  $\text{CO}_2$  gas standards traceable to World Meteorological Organization  
230  $\text{CO}_2$  Mole Fraction Scale. These underway  $\text{pCO}_2$  measurements have a precision and accuracy of  
231  $\sim \pm 1 \mu\text{atm}$ . Measurements made by the underway system provide insight into the surface  
232 carbonate chemistry parameters at stations made in transit where bottle samples were not  
233 collected.

### 234 **2.3 Specimen Capture**

235 Thecosome species were chosen for physiological study opportunistically as they appeared in net  
236 samples at successive stations. Species were targeted specifically for their abundance and the  
237 likelihood of their presence in both ocean basins and only adult individuals were used. Most  
238 individuals were collected with a 1-m diameter, 150- $\mu\text{m}$  mesh Reeve net with a  $\sim 25$  L cod-end in  
239 the Atlantic and a similar 1-m diameter, Reeve net equipped with 330- $\mu\text{m}$  mesh in the Pacific.  
240 Use of the Reeve net with its large and heavy cod-end in combination with slow haul rates  
241 (typically 5-10  $\text{m min}^{-1}$ ) allowed for gentle collection of the delicate thecosomes, consistently  
242 supplying animals in good condition with undamaged shells and external mantle appendages.  
243 Net tows were made at night when animals were expected to congregate at shallow depths, were  
244  $\sim 1$  h in duration in an effort to minimize the handling time of the organisms, and reached a  
245 maximal depth between 100–150 m. Depths were targeted that had a high chlorophyll *a* peak  
246 during CTD casts, high acoustic backscattering on the echosounder, and/or where thecosomes

247 had been abundantly sampled at the same station using the MOCNESS. Occasionally,  
248 individuals of less abundant species were collected from the nets of the MOCNESS for  
249 physiological study, but only if their shells were undamaged and they were swimming normally.

250 Post-capture, individuals were transferred to filtered water in densities of  $< 15 \text{ ind. L}^{-1}$   
251 and kept for at least 8 h in temperature controlled waterbaths to allow for gut clearance.  
252 Temperatures for experimentation (20, 15 or  $10^{\circ}\text{C}$ ) were chosen to be generally representative of  
253 the waters from which the animals were sampled, based on the vertical distributions and  
254 hydrographic conditions documented with the stratified MOCNESS sampling. Chosen  
255 temperatures were typically the average of the water temperature between 25-100 m, although in  
256 the middle section of the Atlantic cruise experimental temperatures were reflective of the 25–50  
257 m average due to the particularly shallow vertical distribution of the dominant species (*Limacina*  
258 *retroversa*) sampled in this region. This was to ensure that experiments were occurring at  
259 physiologically relevant and, presumably, natural temperatures for each species. After gut  
260 clearance, individuals that were in good condition (i.e., swimming and with shell intact) were  
261 used for oxygen consumption experiments.

#### 262 **2.4 Experimental Exposures and Oxygen Consumption Rate**

263 Post-gut clearance, healthy animals were put into separate glass syringe respiration chambers,  
264 one individual per chamber, with a known volume of  $0.2 \mu\text{m}$  filtered seawater and  $25 \text{ mg L}^{-1}$   
265 each of streptomycin and ampicillin. This minimized the microbial respiration effects on the  
266 measurements of carbonate chemistry and  $\text{O}_2$  consumption rates by pteropods during the  
267 experiments. The inclusion of antibiotics, a method which has previously been used with  
268 thecosomes to prevent bacterial growth in respiration experiments (Maas et al., 2012a), was  
269 shown during the Pacific cruise to have no effect on the  $\text{O}_2$  consumption of at least *Limacina*  
270 *helicina*, for the exposure durations associated with these experiments (Howes et al., 2014). The  
271 volume of water in the treatments was chosen to complement the size of the organism and  
272 temperature of the experiment and ranged between 15- and 50 mL in 2011 and 8- and 20 mL in  
273 2012. For every 3- to 5 treatment chambers, a “control” respiration chamber (experimental  
274 seawater with antibiotics and without pteropods) was set up to monitor microbial activity and to  
275 provide water for characterization of the starting conditions.

276 Filtered seawater for experimental exposures was collected during both cruises in batches  
277 at approximately weekly intervals from the surface; experimental water thus began with

278 chemical properties (notably including TA, DIC, pH, as well as salinity) reflective of the local  
279 environment and was then manipulated to modify CO<sub>2</sub> and/or O<sub>2</sub> concentrations. Manipulations  
280 were achieved by bubbling 1 L batches of collected seawater with gas mixes (certified accurate  
281 to ± 2%) for 45–60 min with one of two oxygen (~~21~~100% and ~~10~~48% of O<sub>2</sub> saturation) levels  
282 crossed with two CO<sub>2</sub> (nominally 380 μatm and 800 μatm) levels. At the time of the experiment,  
283 surface air pCO<sub>2</sub> conditions were on average ca. 380 ppm, dictating our ambient (i.e., low  
284 carbon, LC) conditions. In 2011 the ambient condition (~~~21~~100% of O<sub>2</sub> saturation and 380 μatm  
285 CO<sub>2</sub>) was achieved by bubbling with an ambient clean air line, while in 2012 it was achieved by  
286 a certified 380 ppm gas mix.

287 The experimentally modified concentrations mimic the CO<sub>2</sub> and O<sub>2</sub> levels that would be  
288 experienced by thecosomes within the top 400 m of the Pacific Ocean, and reflect the average  
289 projected atmospheric CO<sub>2</sub> level for the open ocean in the year 2100 (A2 emissions scenario,  
290 IPCC, 2007). This resulted in four total treatments: low (i.e., ambient) CO<sub>2</sub>, high oxygen  
291 (LC/HO) representative of current ambient surface ocean conditions; high carbon, high oxygen  
292 (HC/HO), replicating what we expect the average future surface ocean to resemble; low CO<sub>2</sub>,  
293 low oxygen (LC/LO); and high carbon, low oxygen (HC/LO), which is similar to what  
294 organisms in the Pacific would experience during a diel vertical migration into the local oxygen  
295 minimum zone. The goal of this design was to allow us to compare directly the Atlantic and  
296 Pacific thecosomes to see whether exposure to 800 μatm pCO<sub>2</sub> and/or ~~10~~48% of O<sub>2</sub> saturation  
297 resulted in different outcomes. The level of low O<sub>2</sub> chosen for this study was well above the  
298 threshold that has been designated as stressful for non-specialized metazoan life (< 2 mg O<sub>2</sub> L<sup>-1</sup>  
299 or 60 μmol O<sub>2</sub> kg<sup>-1</sup>; Vaquer-Sunyer and Duarte, 2008), in order to test the non-lethal effect of  
300 moderately low O<sub>2</sub> on individuals from the two ocean basins. Calculations based on the salinity  
301 and temperature of the water indicated that bubbling with ~~10~~48% of O<sub>2</sub> saturation achieved  
302 conditions of ~~10~~48–~~13~~62% of O<sub>2</sub> saturation by the start of experiments. Subsequent analyses (see  
303 below) also confirmed that intended CO<sub>2</sub> concentrations were achieved for all treatments within  
304 reasonable ranges, with the exception of the LC/LO Atlantic treatment. In this case, the gas  
305 cylinder was evidently improperly mixed by the manufacturer and analyses suggested a ca. 100  
306 ppm CO<sub>2</sub> concentration. The results for this treatment are still presented but should be  
307 interpreted as a distinct treatment.

308 Oxygen consumption was measured following similar techniques as described in Marsh  
309 and Manahan (1999). Briefly, at the conclusion of the experiment water was withdrawn from  
310 treatment or control chambers using an airtight 500  $\mu$ L Hamilton syringe and injected past a  
311 Clarke-type microcathode (part #1302, Strathkelvin Instruments, North Lanarkshire, United  
312 Kingdom) attached to an O<sub>2</sub> meter (part #782) in a water-jacketed injection port (part #MC100).  
313 This was done three times, allowing the reading to stabilize for at least 30 seconds before a  
314 measurement was taken. Generally, the change in oxygen consumption was between 3–25% of  
315 the control value. In high oxygen experiments, if the oxygen level fell below 70% of air  
316 saturation they were excluded from the analysis.

317 Following exposure, animals were removed from the chamber, blotted dry and frozen in  
318 liquid nitrogen. These individuals were later weighed using a microbalance ( $\pm$  0.0001 g) and the  
319 resulting mass specific O<sub>2</sub> consumption rates are reported in  $\mu$ moles (g wet weight)<sup>-1</sup> h<sup>-1</sup>. Wet  
320 weights are here used as they are more relevant for physiological understanding of animal  
321 function (Childress et al., 2008) but dry weights can be estimated from these using the wet  
322 weight to dry weight relationships developed previously for pteropods (Ikeda, 2014). To  
323 replicate the duration of exposure that would be experienced by most thecosomes in the Pacific  
324 undergoing a daily migration to depth, the experiments were targeted to last 6–to 12 h. In  
325 practice, experiments ranged from 6–to 18 h for normoxic and 3–to 10 h for low O<sub>2</sub> trials. This  
326 variation in duration resulted from balancing the need to elicit a measureable change in O<sub>2</sub>  
327 concentration with preventing extreme O<sub>2</sub> depletion of the chambers (< 6% oxygen saturation)  
328 and accounting for multiple species of variable size and metabolic rate.

## 329 **2.5 Experimental Carbonate Chemistry**

330 Carbonate chemistry of the treatments was characterized in most cases via measurements of DIC  
331 and TA of experimental seawater, unless indicated otherwise. The process of measuring the O<sub>2</sub> in  
332 the treatments used up a large portion of the water and then the chamber was unsealed and  
333 disturbed to remove the animal, rendering it impractical to measure the carbonate chemistry  
334 directly from the respiration chambers. DIC measurements were thus taken from the control  
335 syringes within 18 h of the end of each experiment and used to represent the starting point of the  
336 carbonate chemistry conditions the animals experienced. Water samples were allowed to come to  
337 room temperature (> 6 h) before analysis. DIC was measured using the same system as that used  
338 for the hydrographic characterization (see above). Estimates of the effect of CO<sub>2</sub> production via

339 respiration in treatment chambers on DIC were made using a respiratory quotient of 0.8 M-mole  
340 of CO<sub>2</sub> per 1 M-mole of O<sub>2</sub> consumed (calculated using *Sagitta elegans*; Mayzaud, 1976) to  
341 characterize the ending conditions of the experiments.

342 Due to the small volumes of water in the experimental chambers, it was not possible to  
343 measure both DIC and TA from the control syringes. Instead, TA samples intended to be  
344 representative of the starting experimental conditions were collected via siphoning from each  
345 batch of collected surface water. These samples were subsequently measured based on the  
346 analytical method described above (Wang and Cai 2004). TA of experimental water was  
347 assumed to have been constant over the course of each experiment as water was filtered (0.2 μm)  
348 and antibiotic treated (thus microbial activities were kept at minimum). Although pteropod  
349 aerobic respiration, excretion, and calcification within a respiration chamber could influence TA,  
350 these are presumed to have not had a significant influence over the time scales in question.

351 In some instances, however, measured TA from the batches of experimental water was  
352 substantially dissimilar to that of the surface measurements made from nearby in situ surface  
353 bottle samples collected with the CTD (> 20 μmol kg<sup>-1</sup>; see section 3.3). Calculated pCO<sub>2</sub> values  
354 in these cases were also significantly different from batches of experimental water collected from  
355 other locations, but bubbled with the same CO<sub>2</sub> gas tank. These differences are more than 10  
356 times the measurement precision/accuracy and 5 times the uncertainty of duplicate sampling and  
357 measurements during the cruises. They are also beyond the likely level of TA variation due to  
358 differences in sampling location (geographic and in depth) between the in situ bottle samples and  
359 experimental water batches and rather are likely a consequence of the difficulties associated with  
360 cleanly siphoning the experimental water batches (i.e., contamination during sampling). For  
361 completeness, the carbonate chemistry system parameters for the experimental water, including  
362 aragonite saturation state and pCO<sub>2</sub>, are reported based on calculations using DIC-TA pairs using  
363 both the experimental TA and the in situ measurements from the CTD bottle samples; in those  
364 cases where the TA measurements diverged substantially (> 20 μmol kg<sup>-1</sup>), however, we base  
365 our interpretations on the in-situ measured TA at nearby CTD stations instead of the values of  
366 experimental water. In those circumstances where batch water was taken from test stations  
367 during transit to/from the main study regions and CTD bottle data were unavailable, the  
368 experimental TA was checked using calculated TA values using DIC from the LC/HO treatments  
369 and pCO<sub>2</sub> from the underway measurements.

## 2.6 Statistics

Oxygen consumption rates were tested for significant differences between groups using SPSS. Univariate General Linear Models (GLM) were conducted to determine the effect of CO<sub>2</sub> level, O<sub>2</sub> level, and their interactive effect using the log transformed oxygen consumption with log transformed wet mass as a covariate separately for each species (2 factor design; “CO<sub>2</sub> × O<sub>2</sub>”). In the Atlantic this full factorial design was confounded by the incorrect gas mixture so each treatment was tested independently (1 factor design; “treatment”). Species that were collected during both years/basins, and experiments conducted on species at multiple temperatures, were analyzed separately so that the effect of variations in mass between seasons and the changes in metabolic rate at different temperatures would not confound the analysis. The datasets were tested for normality and homoscedasticity and, in cases where significance was found in the GLM they were explored with Bonferroni pairwise post-hoc comparisons.

For some species the temperature of experimentation was different among stations within a basin. For analyses with these species when comparing species between ocean basins, we applied a standard temperature coefficient (Q<sub>10</sub>) to compare across temperatures. The adjusted rates (R<sub>f</sub>) were calculated at 15°C using a Q<sub>10</sub> of 2 according to the equation:

$$R_f = R_i * \left( Q_{10}^{\left( \frac{15 - T_i}{10} \right)} \right)$$

where R<sub>i</sub> is the original metabolic rate measured at the original temperature (T<sub>i</sub>). Although previous work with thecosomes has shown that Q<sub>10</sub> is species-specific (Seibel et al., 2007; Maas et al., 2011; Maas et al., 2012a), for many of the species used in this study there are no published estimates of Q<sub>10</sub>. Thus, this coefficient value was chosen as it is mid-range for the published Q<sub>10</sub> of non-polar thecosome species as recently compiled by Ikeda (2014; 1.3-2.7) and is consistent with estimates of average Q<sub>10</sub> for marine ectotherms, which typically fall between 2-3 (Hochachka and Somero, 2002; Seibel and Drazen, 2007).

## 3. Results

### 3.1 Specimen Capture

Following currently accepted morphology-based taxonomy, adult individuals from a total of eight species of pteropods were collected over the course of the two cruises for physiological studies. Only relatively large adult specimens were used in respiration trials, in part to avoid any confounding effects of ontogeny and in part to ensure a measurable change in oxygen levels. We

400 collected two species of thecosome pteropods exclusively from the Atlantic, *Limacina retroversa*  
401 (Fleming, 1823), a subpolar species, which is absent from the North Pacific, and *Diacria*  
402 *trispinosa* (Blainville, 1821), which can be found in temperate and tropical regions of the  
403 Atlantic, Pacific and Indian Oceans. Although present in both the North Atlantic and Pacific, the  
404 polar to sub-polar species *Limacina helicina* (Phipps, 1774), was only sampled in the Pacific  
405 transect. Collections of this species consisted of intermixed formae, the high spiraled *Limacina*  
406 *helicina helicina acuta* (van der Spoel, 1967), the lower spiraled *Limacina helicina helicina*  
407 *pacifica* (van der Spoel, 1967), and a forma that bore resemblance to both in a mixed  
408 morphology. Since both the assemblage and morphology of these formae were mixed they were  
409 tested as one population/species. In both ocean basins we collected *Styliola subula* (Quoy and  
410 Gaimard, 1827), *Cavolinia inflexa* (Lesueur, 1813) and *Clio pyramidata* (Linnaeus, 1767).  
411 There is some morphological and molecular evidence that *Cuvierina columnella* (Rang, 1827) is  
412 actually multiple distinct species, now including *Cuvierina atlantica* and *Cuvierina pacifica*  
413 (Janssen, 2005; Burrige et al., 2015), and we tested individuals of these species from their  
414 respective ocean basins.

### 415 3.2 Hydrography

416 Two hydrographic regimes were evident along the North Pacific study transect (Table 2; Fig. 2).  
417 The northern-most stations (50°N 150°W to 47 °N 144.6°W; stations T2-T7, 3-7; Fig. 1) were  
418 coldest, with temperatures between 25-100 m ranging from 5-10°C. At these stations O<sub>2</sub> fell  
419 below ~~1048%~~ saturated (~130 μmol kg<sup>-1</sup>) at depths less than ~250 m, pH fell below 7.7 at depths  
420 less than 130 m, and pCO<sub>2</sub> had already reached 800 μatm by ~200 m. Individuals in this area  
421 experienced an  $\Omega_{Ar} = 1$  between 160-185 m, well within the typical diel vertically migratory  
422 range of both of the species found in the region (*C. pyramidata* and *L. helicina*). At stations from  
423 more southern latitudes (47 °N 144.6°W to 33.5°N 135°W; stations 15-34, T9-T10; Fig. 1),  
424 temperatures at depths between 25-100 m were higher, ranging between 10-17°C, representative  
425 of the transition zone into the North Pacific Gyre. Along this portion of the transect O<sub>2</sub>  
426 concentration consistently fell below ~~1048%~~ saturated by depths of 340 and 400 m. The depth at  
427 which pH<sub>T</sub> fell below 7.7 increased gradually from ~150 to 230 m as latitude decreased.  
428 Correspondingly, the depth at which pCO<sub>2</sub> in this area reached 800 μatm was 330 to 440 m, and  
429 the aragonite saturation horizon 330 m to 430 m depth. The depth at which species would  
430 experience a pH<sub>T</sub> below 7.7 was within the inhabited depth range known from the literature for

431 all of the species tested in this portion of the study region, but only the species *Clio pyramidata*,  
432 with a typical vertical range of 0-500 m (Table 2), would be likely to experience ~~48%~~ of O<sub>2</sub>  
433 saturation, 800  $\mu\text{atm}$  pCO<sub>2</sub> and aragonite under-saturation in its typical distribution (Table 1).

434 In contrast to the Pacific, along the entire Atlantic transect O<sub>2</sub> concentration was above  
435  $\sim 200 \mu\text{mol kg}^{-1}$  (~~572%~~ saturation) in the top 500 m, while pCO<sub>2</sub> never reached 800  $\mu\text{atm}$  and  
436 aragonite under-saturation never occurred throughout the top 1000 m. There were three dominant  
437 hydrographic regimes in the Atlantic (Table 2; Fig. 2). In the northeastern part of the sampling  
438 region (50°N 42°W to 44.9 °N 42°W; stations 21-31; Fig. 1), where the Gulf Stream meets the  
439 Labrador Current, average temperatures at 25-100 m were near 15°C and pH<sub>T</sub> only fell below 7.7  
440 at depths exceeding 400 m. Similarly, in the southwest part of the sampling region (from 42°N  
441 52°W to 36°N 52°W; stations 3-13; Fig. 1), corresponding to the Sargasso Sea and through the  
442 Gulf Stream, pH<sub>T</sub> only fell below 7.7 at depths exceeding 450 m, although the upper water  
443 column was warmer, with average temperatures of 20°C. There was a third water mass type,  
444 typical of colder fresher shelf waters, at station 32 and in an intrusion off the Grand Banks at  
445 stations 17 and 19. Stations conducted in this water were typified by a temperature and salinity  
446 anomaly with temperatures below 5°C from 25-100 m and a salinity signature < 33, contrasting  
447 significantly with the surface salinities of the northern portion ( $\sim 34$ ) and southern portion ( $\sim 36$ )  
448 of the Atlantic transect. As a consequence, these stations contained water of the lowest pH, with  
449 surface waters reaching 7.7 at depths shallower than 200 m.

### 450 **3.3 Carbonate Chemistry of Experiments**

451 Bubbling with CO<sub>2</sub> levels of  $\sim 380$  and  $\sim 800$  ppm resulted in a distinct separation of carbonate  
452 chemistry between treatments during the experiments in both oceans (Table 3). Due to pre-  
453 existing differences in the carbonate chemistry of the seawater collected in each ocean, TA  
454 differed between the two basin treatments. In the Atlantic the DIC of the ambient CO<sub>2</sub> treatments  
455 ranged from 2030-2090  $\mu\text{mol kg}^{-1}$  and the high CO<sub>2</sub> treatments from 2140-2220  $\mu\text{mol kg}^{-1}$ , with  
456 an average difference between treatments of similar temperature and salinity of 132  $\mu\text{mol kg}^{-1}$ .  
457 Surface TA in the region decreased from  $\sim 2370 \mu\text{mol kg}^{-1}$  in the southern part of the transect to  
458 2300  $\mu\text{mol kg}^{-1}$  in the northern latitudes. In the Pacific the DIC of the ambient CO<sub>2</sub> treatment  
459 ranged from 1930-2020  $\mu\text{mol kg}^{-1}$  and the high CO<sub>2</sub> treatment from 2030-2110  $\mu\text{mol kg}^{-1}$ , with  
460 an average difference of 90.7  $\mu\text{mol kg}^{-1}$  between the treatments. Surface TA in this basin was

461 2150  $\mu\text{mol kg}^{-1}$  in the most northern collection and had increased to 2200  $\mu\text{mol kg}^{-1}$  by the  
462 transect mid-point.

463 Calculations of  $\text{pCO}_2$  based on these measurements of DIC and TA suggested that target  
464  $\text{pCO}_2$  levels were generally attained and were consistent between the two cruises, with the  
465 exception of the LC/LO treatment in the Atlantic. In this case, there was a substantial deviation  
466 from the intended  $\text{pCO}_2$ , suggesting values ranging from 99-139  $\mu\text{atm}$  in contrast to a range of  
467 311-391  $\mu\text{atm}$  for the LC/HO in the Atlantic and 283-409  $\mu\text{atm}$  for LC/HO and 295-397  $\mu\text{atm}$  in  
468 the LC/LO in the Pacific. Evidently, this indicates improper mixing of the gas concentration in  
469 the Atlantic LC/LO gas cylinder by the manufacturer. The calculations for the high  $\text{CO}_2$   
470 treatments were more consistent between cruises, with  $\text{pCO}_2$  for the HC/HO being 585-868  $\mu\text{atm}$   
471 and the HC/LO being 755-783 in the Atlantic, while in the Pacific the HC/HO treatment was  
472 between 520-740  $\mu\text{atm}$  and the HC/LO 546-766  $\mu\text{atm}$ . The variability in calculated  $\text{pCO}_2$  values  
473 likely represents variations in bubbling time, temperature, and the degree to which the water  
474 reached saturation relative to the gas mixtures.

475 As a consequence of the natural differences in seawater carbonate chemistry, in particular  
476 the TA differences between two ocean basins, there were inherent differences in the aragonite  
477 saturation state between the Pacific and Atlantic treatments (Table 3). In the Atlantic,  $\Omega_{\text{Ar}}$  of the  
478 ambient  $\text{CO}_2$  treatment ranged from 2.4-3.5, except for the LC/LO treatment ( $\Omega_{\text{Ar}}$  4.0-5.5), which  
479 was bubbled with an incorrect gas mixture as discussed above. In comparison, in the Pacific the  
480 ambient  $\text{CO}_2$  condition had a lower range of  $\Omega_{\text{Ar}}$  (2.2-2.4) for both the LC/HO and the LC/LO  
481 treatments. The experimental conditions of the high  $\text{CO}_2$  treatments reached their lowest value in  
482 the middle part of the transect ( $\Omega_{\text{Ar}} = 1.2$  at mid-latitudes; Table 3), where cold northern waters  
483 of low salinity were encountered. Experimental  $\Omega_{\text{Ar}}$  had a range of 1.5-2.0 for the rest of the  
484 transect in the Atlantic. The values of experimental  $\Omega_{\text{Ar}}$  were lower overall in the Pacific,  
485 although the high  $\text{CO}_2$  treatments also never reached under-saturation ( $\Omega_{\text{Ar}}$  1.3-1.8). In general,  
486 the manipulation of carbonate chemistry in this study successfully created two distinct ranges for  
487 both  $\text{pCO}_2$  and aragonite saturation state ( $\Omega_{\text{Ar}}$ ).

488 It is important to acknowledge that the production of  $\text{CO}_2$  via respiration of the organisms  
489 within the chambers would modify the carbonate chemistry of the treatments over the duration of  
490 the experiments. Based on the average respiration rate, we estimate an average DIC production  
491 of  $\sim 18.0 \mu\text{mol kg}^{-1}$  by the end of an experiment. Applying such a change to the experimental

492 conditions in the northeast Pacific, where seawater is more sensitive to changes in DIC due to a  
493 lower buffering capacity compared to the Atlantic (i.e., a worst case scenario),  $\Omega_{Ar}$  would only  
494 change by  $<0.1$  in both the LC and HC experimental chambers over the course of the respiration  
495 experiments. Although this is an appreciable effect, we nonetheless retain a wide separation  
496 between the ambient and high CO<sub>2</sub> treatments and in no cases would the treatments reach under-  
497 saturation as a consequence of this biological activity. As such, for simplicity the results reported  
498 in Table 3 do not include this correction for respiration.

### 499 **3.4 Oxygen Consumption Rate**

#### 500 **3.4.1 Effect of CO<sub>2</sub>**

501 Varying availability and abundances of the different thecosome pteropod species in the net  
502 samples precluded all species being exposed to the full factorial design but individuals of all  
503 species were tested under the low CO<sub>2</sub>, high oxygen (LC/HO) and high carbon, high oxygen  
504 (HC/HO) treatments (Fig. 3, Table 4). To explore differences in metabolism attributable to a  
505 response to CO<sub>2</sub>, the log transformed wet mass was used in a GLM as a covariate comparing the  
506 log transformed oxygen consumption (response variable) under low and high CO<sub>2</sub> conditions;  
507 each population within a species that was sampled in both basins or run at multiple experimental  
508 temperatures, was examined separately. There was no significant effect of CO<sub>2</sub> for any species in  
509 either basin.

#### 510 **3.4.2 Effect of basin**

511 Following this assessment, we were interested in determining whether there were  
512 between basin differences in metabolic rate. As such we ran a GLM using log transformed  
513 metabolic rates for the three species that were found in both basins, normalized to 15 °C to  
514 account for differences in experimental temperature by applying a standard temperature  
515 coefficient. With the log-transformed wet mass as a covariate, we tested for an effect of basin,  
516 CO<sub>2</sub> and an interactive term. *Clio pyramidata* had a similar metabolic rate between basins. In  
517 contrast, *Cavolinia inflexa* ( $F_{1,20}=10.358$ ,  $p=0.004$ ) and *Styliola subula* ( $F_{1,23}=11.817$ ,  $p=0.002$ )  
518 both had a significantly lower metabolic rate in the Pacific, although no interactive effect of CO<sub>2</sub>.

#### 519 **3.4.2 Effect of O<sub>2</sub>**

520 For the species where enough individuals were collected to provide experimental  
521 replicates to explore the interactive effects of CO<sub>2</sub> and O<sub>2</sub> we also ran a species and basin  
522 specific GLM exploring the effect of treatment (Fig. 3, Table 5). *Clio pyramidata*, the only

523 species we were able to test in both basins showed no significant effect of high CO<sub>2</sub>, low O<sub>2</sub> or  
524 the interactive treatment in either basin. In the Pacific, *L. helicina* and *C. inflexa* similarly  
525 showed no significant change in metabolic rate as a consequence of any of the treatments. In  
526 contrast, in the Atlantic, there was a significant effect of treatment for *L. retroversa* and a  
527 Bonferroni post-hoc analysis comparing the treatments found that the high CO<sub>2</sub>, low O<sub>2</sub> (HC/LO)  
528 treatment was significantly lower than all other treatments (Fig. 4;  $F_{3,38}=17.836$ ,  $p<0.001$ ; a  
529 ~60% reduction in the average mass specific metabolic rate in comparison with the LC/HO  
530 treatment; Table 4). *Cuvierina atlantica* was tested at both 15 and 20 °C in the Atlantic, so to  
531 make comparisons among these experiments a temperature coefficient was applied and rates  
532 were normalized to 15 °C, after which no significant effect of any treatment was found for this  
533 species.

534

#### 535 **4. Discussion**

536 This study reveals that short term exposure to low O<sub>2</sub> and high CO<sub>2</sub>, similar to what would be  
537 experienced by individuals in the Pacific during diel vertical migration, does not influence the  
538 oxygen consumption of adult individuals of most of the thecosome pteropod species examined  
539 from either the Atlantic or Pacific. The only species that had a significant change in respiration  
540 in response to any of the treatments was *Limacina retroversa* from the Atlantic, which responded  
541 to the combined effect of low O<sub>2</sub> and high CO<sub>2</sub> with a reduction in oxygen consumption rate.

#### 542 **4.1 Experimental Design**

543 A factor that should be considered when interpreting our results is the dynamic hydrographic  
544 conditions that the animals experience naturally between and within the ocean basins.  
545 Thecosomes of multiple species were found at a range of temperatures, salinities and carbonate  
546 chemistries, meaning that they experienced a range of pH and aragonite saturation states in their  
547 natural habitat. When comparing animals from multiple locations, we chose to use local water in  
548 order to replicate these natural conditions and to manipulate exclusively the CO<sub>2</sub> concentration,  
549 as this is the factor that is changing due to anthropogenic activity. This approach, however, does  
550 not control for the other parameters of the carbonate chemistry system, which will vary between  
551 regions. Despite this fact, there was a clean distinction between treatments, notably in terms of  
552 aragonite saturation state as well as CO<sub>2</sub> concentration, which provides insight into the effect of  
553 moderate short duration exposure to CO<sub>2</sub>.

554 It is also important to note that the individuals of *L. helicina* from the Pacific experiments  
555 did occasionally have very high mortality during the period prior to experimentation (>80% at  
556 transit station T2 and T5, decreasing substantially to the northwest and along the main Pacific  
557 transect). These individuals, which are polar/sub-polar organisms and are typically found  
558 between -2 to 10 °C (Lalli and Gilmer, 1989), were collected from water that was likely near the  
559 upper limit of their optimal temperatures although alternate possibilities are that these were a  
560 population reaching senescence, or that they were collected in a hydrographic regime with low  
561 food availability. Animals collected from these sites that were used in subsequent respiration  
562 experiments may therefore have been taken from an already stressed population and should be  
563 recognized as such.

#### 564 **4.2 Carbon Dioxide Effect**

565 Hydrographic profiles collected in the Pacific coincident to sampling of thecosomes  
566 indicate that organisms in the northern portion of the study region would experience conditions  
567 of high CO<sub>2</sub> and low O<sub>2</sub> in the upper ~450 m of their distribution (Chu et al., 2016). Based on  
568 previous knowledge of the vertical distributions of the thecosomes used in this study, only the  
569 species *Clio pyramidata* would ever experience a pH<sub>T</sub> below 7.7 and none of the thecosomes  
570 studied would experience 800 µatm pCO<sub>2</sub> or under-saturation within their vertical range in the  
571 Atlantic study region and (Table 1). Despite these environmental differences, we found no  
572 significant effect of increasing CO<sub>2</sub> alone on the respiration rates of any of the species from  
573 either ocean basin. These results increase the published evidence that short term (6-18 h)  
574 exposure to enhanced CO<sub>2</sub> without synergistic stressors has no significant effect on the metabolic  
575 rate of many species of thecosome pteropods. Thus far, there are only two species that have been  
576 documented to show a change in metabolism based on exposure to manipulated CO<sub>2</sub> alone:  
577 *Limacina antarctica* (789-1000 µatm, 24 h: Seibel et al., 2012) and *Diacria quadridentata* (1000  
578 µatm, 6-18 h: Maas et al., 2012b). The metabolic rates of all other species yet studied, including  
579 *Hyalocylis striata*, *Clio pyramidata*, *Diacavolinia longirostris*, *Creseis virgula* (6-18 h: Maas et  
580 al., 2012b), and *Limacina helicina* (24 h: Comeau et al., 2010), were not significantly affected by  
581 short term exposure to high CO<sub>2</sub>, although the latter species showed an increase in metabolic rate  
582 when high CO<sub>2</sub> was combined with high temperatures. Our results, which increase the  
583 geographic coverage for *L. helicina* and *C. pyramidata* and provide the first data about the

584 species *C. pacifica*, *C. atlantica*, *L. retroversa*, *D. trispinosa*, *C. inflexa* and *S. subula*, corroborate  
585 these earlier findings.

586 One interpretation of these results is that physiological responses may have occurred, but  
587 involved the reallocation of resources to different tissues or metabolic pathways; this  
588 redistribution could serve to maintain the thecosome total energy budget, and subsequently  
589 would not significantly change the metabolic rate of the individuals. A transcriptomic study done  
590 with individuals of *Clio pyramidata* as a companion project to the present work in fact suggested  
591 that expression of some genes was influenced by CO<sub>2</sub> exposure even though metabolic rate was  
592 not (Maas et al., 2015), perhaps suggesting some re-allocation among energetic demands. If this  
593 is the case it indicates that, to some degree, the short-term exposure to high CO<sub>2</sub> concentration is  
594 within the physiological tolerance of the tested species. Alternative hypotheses are that the  
595 duration of exposure was too short or the severity of the CO<sub>2</sub> treatment too minimal to elicit a  
596 measurable response. It is possible, for example, that some processes, like biomineralization,  
597 may be influenced by high CO<sub>2</sub>, but only after a longer exposure duration. Finally, it may be that  
598 changes in respiration rate were subtle, requiring a much greater sample size to identify in light  
599 of biological variability, but exploration of this hypothesis would require a dedicated experiment  
600 to collect more individuals and likely a smaller number of species.

601 This possible tolerance to short term CO<sub>2</sub> exposure may be due to the fact that within  
602 their distribution or diel migrational range there are conditions, or perhaps seasons, where the  
603 natural hydrography causes many species of thecosome to experience conditions of high  
604 CO<sub>2</sub>/low pH, and the species are therefore adapted to this range of exposure. The Arctic species  
605 *L. helicina* and subarctic species *L. retroversa*, for instance, are thought to inhabit waters which  
606 have been shown to reach a concentration of > 950 μatm CO<sub>2</sub> and to be undersaturated with  
607 respect to aragonite during the winter season in Kongsfjord, Svalbard (Lischka and Riebesell,  
608 2012). These conditions are pervasive throughout the upper water column, meaning that *L.*  
609 *helicina* and *L. retroversa*, which are not strong diel migrators, would experience seasonal under-  
610 saturation in these polar regions. The more temperate and tropical open ocean thecosomes,  
611 including *C. pyramidata*, *C. inflexa* and *S. subula* are all currently believed to be circumglobal  
612 and most, to varying degrees, diel migratory (Table 1; van der Spoel, 1967; Bé and Gilmer,  
613 1977). Populations are therefore likely to encounter high CO<sub>2</sub> in sub-surface waters in regions  
614 associated with OMZs, including much of the North Pacific and off the coast of Northern Africa.

615 The ability to cope with high CO<sub>2</sub> for short durations may have been selected for over time as a  
616 natural consequence of the types of unavoidable environmental variability experienced by these  
617 planktonic populations.

### 618 **4.3 Low O<sub>2</sub> and Combined Effects**

619 In the Pacific Ocean, none of the species for which we had enough individuals to perform the  
620 low O<sub>2</sub> study (*L. helicina*, *C. pyramidata*, and *C. inflexa*) had a significant change in metabolic  
621 rate under low (~~1048%~~ saturated) O<sub>2</sub>, even when combined with enhanced CO<sub>2</sub>. These results  
622 indicate that the O<sub>2</sub> levels were above the concentration below which these species can no longer  
623 sustain their routine metabolic activity (Pcrit; Hochachka and Somero, 2002) and that any  
624 changes in physiology associated with the treatments required no increased energetic expenditure  
625 or metabolic reduction. As subsurface waters throughout the cruise were frequently below  
626 ~~1048%~~ of O<sub>2</sub> saturation (< ~130 μmol kg<sup>-1</sup>), this indicates that these species may be naturally  
627 adapted to coping with low O<sub>2</sub> conditions.

628 In the Atlantic, examination of the effects of low O<sub>2</sub> is confounded by an unfortunate and  
629 accidentally low level of CO<sub>2</sub> (~130 μatm) in the LC/HO treatment (Table 3). Tests of the effect  
630 of high CO<sub>2</sub> (HC/HO) and the interactive (HC/LO) treatments nonetheless remain valid, and for  
631 *L. retroversa*, exposure to HC/LO caused a large and significant reduction in metabolic rate.  
632 Suppression in metabolic rate is a common tactic for surviving unfavorable conditions (Guppy  
633 and Withers, 1999; Seibel, 2011). Although metabolic depression is generally survivable in the  
634 short term, over longer time scales there are often implications for growth, reproduction and  
635 survival (reviewed in: Pörtner, 2010; Seibel, 2011). In the Atlantic, our measured in situ O<sub>2</sub>  
636 levels were never below 15% (~200 μmol kg<sup>-1</sup>). In contrast with the other species studied, which  
637 in at least some portions of their geographic range are occasionally found in association with  
638 subsurface low O<sub>2</sub> combined with hypercapnia, *L. retroversa* lives exclusively in the sub-polar  
639 North Atlantic Ocean and the Southern Circumpolar Current. As such this is the only species in  
640 this study in which no population is likely to experience conditions of low O<sub>2</sub> and high CO<sub>2</sub>  
641 together naturally anywhere in its distribution. Its inability to maintain metabolic rate during this  
642 interactive exposure may be a short-term metabolic response to environmental conditions that are  
643 unsustainable over longer time periods. As a consequence of the very low CO<sub>2</sub> in the LC/LO  
644 treatment, it is impossible to determine whether the metabolic suppression for *L. retroversa* in  
645 the HC/LO was in response to reduced O<sub>2</sub> availability alone or to the interactive effect of low O<sub>2</sub>

646 with high CO<sub>2</sub>. In the LC/LO treatment any change in respiration due to low O<sub>2</sub> could have been  
647 masked by a change in the energy budget as a response to the low (equivalent to pre-industrial  
648 atmospheric conditions) levels of CO<sub>2</sub>. The results suggest that further work in the Atlantic is  
649 warranted to disentangle these stressors and to determine whether the observed change in  
650 metabolic rate was solely a consequence of O<sub>2</sub> availability or truly a synergistic effect.

651 Interestingly, although the temperature coefficients were not species-specific and may  
652 not, therefore, perfectly normalize the dataset, one trend revealed by their use was a significant  
653 difference in the normalized metabolic rates between individuals of the species *S. subula* and *C.*  
654 *inflexa* from the Atlantic and Pacific Oceans. The comparatively lower metabolic rates from the  
655 Pacific may be a real response to the lower availability of O<sub>2</sub> for aerobic metabolism. Having a  
656 slower routine rate of O<sub>2</sub> consumption may be the result of a more efficient respiratory  
657 mechanism or an adaptation for coping with occasional exposures to the relatively high CO<sub>2</sub> and  
658 low O<sub>2</sub> conditions found in the northeast Pacific Ocean.

659

## 660 **5. Conclusions**

661 Thecosomes pteropods are thought to be some of the most sensitive of the oceanic zooplankton  
662 species to acidification. The responses we documented in the face of short-term CO<sub>2</sub> exposure  
663 and low O<sub>2</sub> reveal interesting patterns about basin scale differences in sensitivity, possibly  
664 associated with adaptation to local environmental conditions. Importantly, our results indicate  
665 that short-term exposure to high CO<sub>2</sub> does not have an effect on the respiration rate of multiple  
666 species of temperate and sub-polar thecosome species from both the North Atlantic and Pacific  
667 Oceans, irrespective of recent likely environmental exposure. The lack of effect of CO<sub>2</sub> as a  
668 single-stressor on metabolic rate in adult organisms of various species has been seen in a number  
669 of studies (reviewed in: Dupont et al., 2010; Kroeker et al., 2013), although there are many other  
670 metrics that have been shown to be more consistently affected. As such, thecosomes may have  
671 physiological coping mechanisms that allow them to maintain their energy budget for short  
672 periods of time in the face of high CO<sub>2</sub> via the re-allocation of their energetic resources. Over  
673 longer time periods, however, this could reduce their scope for growth and reproduction,  
674 negatively impacting the fitness of the population as has been demonstrated with other marine  
675 calcifiers (i.e.: Stumpff et al., 2011; Dupont et al., 2013; Melzner et al., 2013). Testing these  
676 hypotheses remains difficult as thecosomes are hard to maintain in captivity and there are no

677 published studies of individuals kept fed and exposed to CO<sub>2</sub> in laboratory conditions for long  
678 durations (reviewed in: Howes et al., 2014; Thabet et al., 2015). Keeping individuals well fed is  
679 a critical factor since high food availability has been suggested to modulate the effect of high  
680 CO<sub>2</sub> exposure in both thecosomes (Seibel et al., 2012) and other calcifying species (Thomsen et  
681 al., 2013). Comparative short-term studies of wild caught animals such as the present  
682 experiments, therefore, currently give us the best insight into the sensitivity of these open-ocean  
683 populations, and the ability to predict how they will respond to the expected changes in the ocean  
684 environment.

685 Furthermore, although adult individuals may show no change in metabolic rate, there is  
686 evidence that juvenile stages of many calcifying species are typically more sensitive to CO<sub>2</sub>  
687 exposure (i.e. Connell et al., 2013; Waldbusser et al., 2015) and emerging evidence supports the  
688 idea that eggs, veligers and juveniles of *L. retroversa* and *L. helicina* are more vulnerable to  
689 acidification than adults (Lischka et al., 2011; Thabet et al., 2015; Manno et al., 2016). Thus,  
690 although adults may be capable of surviving short-term exposure, as acidity in surface waters  
691 increases there may be population level stress due to ontogenetic sensitivity.

692 These findings also draw attention to the consequences of the high degree of vertical  
693 variability in the open ocean environment, with animals in the Pacific found migrating between  
694 deep waters, undersaturated with respect to aragonite, and the surface (Lawson, unpublished  
695 data; Maas et al., 2012b; Chu et al., 2016). Recent studies in the California Current system  
696 indicate that thecosome shells show signs of in situ dissolution when associated with waters that  
697 are undersaturated with respect to aragonite (Bednaršek et al., 2014; Bednarsek and Ohman,  
698 2015). Although our short duration experiments do not directly address the effect of longer-term  
699 exposure to high CO<sub>2</sub>, it does remind us that as open ocean environments respond to  
700 anthropogenic change there may be vertical refugia from ocean acidification stress as well as  
701 regions where animals may already experience high CO<sub>2</sub>. As surface waters acidify, the ability to  
702 endure short-duration exposure and to migrate in both the Atlantic and Pacific populations may  
703 provide mechanisms for mitigating detrimental effects of acidification exposure. The potential  
704 compression of vertical habitat associated with the shoaling of the aragonite compensation depth,  
705 however, may have implications for predator/prey interactions, carbon pumping and other  
706 ecosystem functions (Seibel, 2011; Bednarsek and Ohman, 2015). Furthermore, it is clear that  
707 thecosome shells are highly sensitive to dissolution (Comeau et al., 2012; Lischka and Riebesell,

708 2012; Manno et al., 2012) and there could be fitness and ecological consequences of dissolution  
709 in regions with vertical variation in carbonate chemistry.

710 Finally, as concerns about increasing CO<sub>2</sub> drive further explorations of comparative  
711 organismal physiology in the marine system, it is important to recognize that often the exposure  
712 of animals to increased CO<sub>2</sub> will occur in concert with expanding regions of low O<sub>2</sub>. This has  
713 been explored in the coastal environment where the interaction of acidification with  
714 eutrophication and associated low O<sub>2</sub> is comparatively well studied (Cai et al., 2011; Melzner et  
715 al., 2013) and in theoretical frameworks (Pörtner, 2010; Gruber, 2011; Sokolova, 2013).  
716 Experiments in the open ocean environment, however, are only beginning to be conducted and  
717 their implications explored. This study suggests that to make accurate predictions about how  
718 populations will respond to climate change and adequately understand the factors affecting  
719 organismal response, further investigations of the interactive effects of low O<sub>2</sub> and hypercapnia  
720 should consider natural environmental variability, population biogeography and phylogenetic  
721 sensitivity.

722 **Data availability**

723 Cruise data for the project is available via the National Science Foundation’s Biological and  
724 Chemical Oceanography Data Management Office (BCO-DMO) under the project “Horizontal  
725 and Vertical Distribution of Thecosome Pteropods in Relation to Carbonate Chemistry in the  
726 Northwest Atlantic and Northeast Pacific” (<http://www.bco-dmo.org/project/2154>). The raw data  
727 for the respiration experiments are included in this deposition (DOI: 10.1575/1912/6421). [The](#)  
728 [raw data for the carbonate chemistry of the manipulations are included as supplementary data.](#)

730 **Author contributions**

731 A. Maas and G. Lawson designed the experiments. All co-authors participated in oceanographic  
732 cruises and collection of samples. A. Maas conducted all of the experiments and statistical  
733 analyses. Z.A. Wang advised on the manipulation of carbonate chemistry and provided the  
734 measurements of both the hydrographic and experimental conditions. A. Maas prepared the  
735 manuscript with contributions from both co-authors.

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1005

1006 Table 1: Environmental preferences and diel vertical migratory patterns for the species used in  
 1007 this study based on previously published data (Bé and Gilmer, 1977; Lalli and Gilmer, 1989).  
 1008 Data includes published full ranges at which organisms have been found, as well as previous  
 1009 authors' estimates of the preferred (optimal) ranges of each species, when available. Note that  
 1010 these are based on relatively sparse observations of broadly distributed species, many of which  
 1011 may be cryptic congeners, and thus should be treated as estimates.

Species	(optimal) temp (°C)	(optimal), depth (m)	migrator?
<i>Cuvierina atlantica</i>	18 to 26	100-250	possible
<i>Cuvierina pacifica</i>	Only recently established as a separate species, the habits are assumed to be similar to the Atlantic congener.		
<i>Cavolinia inflexa</i>	16 to 28	0-250	no
<i>Clio pyramidata</i>	7 to 27	(0-500), <1500	yes
<i>Limacina helicina</i>	(-2 to 10)	(50-100), <300	possible
<i>Limacina retroversa</i>	(7 to 12)	(20-30), < 150	possible
<i>Styliola subula</i>	(18 to 22)	50-300	yes
<i>Diacria trispinosa</i>	9 to 28	30-200	no

1012

1013 Table 2: The hydrography and location for each station where animals for experiments were  
1014 collected. Each basin was characterized by multiple hydrographic regimes (see text and Fig 2);  
1015 transitions between regimes are denoted by dashed horizontal lines. At stations along the main  
1016 transect the depth (m) at which O<sub>2</sub> decreased below 130 μmol O<sub>2</sub> kg<sup>-1</sup> (~~~1048%~~ saturated), the  
1017 average temperature from 25-100 m (°C) and the average salinity from 25-100 m were derived  
1018 from CTD casts. At a few stations (denoted via <sup>a</sup>) in the Atlantic there was warm water at the  
1019 surface and cold fresher water below. The only species in this region, *Limacina retroversa*, has  
1020 an optimum temperature between 7-12 °C (Bigelow, 1924) and was generally found above 50 m  
1021 (Lawson, unpublished data). At these sites the average temperature and salinity is reported first  
1022 for between 25-100 m and then also for 25-50 m to reflect the conditions likely experienced by  
1023 the pteropods. pCO<sub>2</sub> and Ω<sub>Ar</sub> were calculated from measured pH<sub>T</sub> and DIC bottle samples. We  
1024 interpolated linearly the depths (m) at which the pH<sub>T</sub> decreased below 7.7, pCO<sub>2</sub> reached 800  
1025 μatm, and aragonite saturation (Ω<sub>Ar</sub>) reached 1 from the discrete measurements at adjacent  
1026 depths. At stations conducted while in transit to the main study transects (denoted by prefix T)  
1027 the average temperature from 25-100 m (°C) was documented from XBT casts. At these transit  
1028 stations no O<sub>2</sub> or carbonate chemistry data were available (noted with a dash). The species  
1029 caught at each station and used in this study are demarcated with a star (\*).  
1030

Year	Station	Latitude (°N)	Longitude (°W)	average temp (°C) 25-100 m	average salinity 25-100 m	depth of 130 µmol O <sub>2</sub> kg <sup>-1</sup>	depth of pH <sub>T</sub> 7.7	depth of 800 µatm	depth of Ω <sub>AT</sub> = 1	<i>C. atlantica</i>	<i>C. pacifica</i>	<i>C. inflexa</i>	<i>C. pyramidata</i>	<i>L. helicina</i>	<i>L. retroversa</i>	<i>S. subula</i>	<i>D. trispinosa</i>
2011 Atlantic	32	49.1	-44.3	5.3, 9.0	34.4, 34.0	NA	74.1	NA	NA						*		
	31	50.0	-42.0	14	35.8	NA	385.4	NA	NA								*
	30	49.6	-41.9	14.1	35.8	NA	452.8	NA	NA	*							*
	26	47.5	-42.0	13.3	35.2	NA	644.9	NA	NA	*			*				
	24	46.5	-42.0	14.5	35.5	NA	453.9	NA	NA	*			*				
	21	44.9	-42.0	16.5	36.2	NA	501.1	NA	NA				*				*
	19	44.0	-44.9	4.9, 11.2	33.4, 32.9	NA	181.0	NA	NA						*		
	17	43.0	-47.8	1.8, 8.1	33.2, 32.8	NA	143.1	NA	NA						*		
	13	40.9	-52.0	20.7	36.5	NA	756.7	NA	NA	*		*					*
	10	47.5	-52.0	19.4	35.9	NA	466.9	NA	NA	*		*					*
	8	38.5	-52.0	22.8	36.5	NA	805.7	NA	NA	*		*					*
	3	36.0	-52.0	21.4	36.6	NA	937.7	NA	NA	*							
2012 Pacific	T2	45.6	-128.5	-	-	-	-	-	-				*	*			
	T3	46.6	-133.5	-	-	-	-	-	-					*			
	T4	47.7	-138.5	6.4	-	-	-	-	-				*				
	T5	45.7	-129.8	10.0	-	-	-	-	-					*			
	T6	46.6	-134.9	9.5	-	-	-	-	-					*			
	T7	47.6	-140.2	8.6	-	-	-	-	-				*				
	3	49.0	-148.2	6.2	32.7	209	128.9	193.7	168.5								
	6	47.5	-145.6	7.1	32.7	235	108.3	199.2	159.1				*	*			
	7	47.0	-144.6	7.8	32.7	256	131.0	214.0	185.1				*				
	15	43.1	-138.1	10.9	32.9	363	199.5	368.2	334.8				*				
	18	41.5	-135.8	13.7	33.0	340	147.3	331.7	380.6				*				
	21	39.9	-135.0	12.7	33.1	348	162.0	332.2	302.8		*						
	24	38.6	-135.0	14.7	33.3	402	222.8	411.8	372.7		*		*				
	30	35.6	-135.0	16.2	33.3	349	200.7	437.8	425.1		*	*	*				
	32	34.4	-135.1	16.5	33.3	348	202.9	439.2	432.0		*	*	*				
34	33.6	-135.0	17.4	34.0	368	233.3	370.1	352.4			*	*			*		
T9	33.7	-133.6	17.0	-	-	-	-	-		*	*	*					
T10	33.8	-133.2	15.9	-	-	-	-	-		*	*	*					

Table 3: Carbonate chemistry during manipulation experiments. The manipulation experiments were conducted at multiple temperatures (T.) and salinities (S.) based on the conditions the organisms were caught in. As described in more detail in the text, DIC measurements were made of water drawn from the control chambers while TA was measured for batches of experimental water (denoted as xpt. TA). In situ TA (i.s. TA), based on nearby CTD bottle sampling at the surface, is also shown. At test stations conducted while in transit to/from the main study regions, where bottle samples of in situ TA were unavailable, underway pCO<sub>2</sub> values and the LC/HO DIC were used to calculate in situ TA (denoted with \*). In some instances, measurements of experimental TA differed by >20 μmol kg<sup>-1</sup> from nearby in situ measurements of surface TA. This difference greatly exceeds expected variability based on measurement uncertainty and spatial (geographic and vertical) offsets in the locations of experimental water collection relative to the nearest CTD cast; in these circumstances, the experimental TA was likely erroneous due to sampling issues (e.g., contamination). For completeness, and to aid in identification of erroneous experimental TA values, calculations of carbonate chemistry parameters, including aragonite saturation state ( $\Omega_{Ar}$ ) and pCO<sub>2</sub> were made based on DIC and both experimental TA and in situ TA. In further data analysis and interpretation, calculations based on experimental TA are given preference except those few instances where experimental TA differed from in situ by >20 μmol kg<sup>-1</sup> (bold denotes preferred calculations). Calculated saturation state and pCO<sub>2</sub> are reported as the average and standard deviation per batch of water. Note that the LC/LO gas tank in 2011 (in italics) appears to have been improperly mixed by the manufacturer as calculations suggested it contained a much lower CO<sub>2</sub> level than the intended 380 μatm; it should consequently be considered an entirely separate treatment from the 2011 LC/HO (where CO<sub>2</sub> levels were based on bubbling with an ambient air line).

	Treatment	T. °C	S.	i.s. TA ( $\mu\text{mol kg}^{-1}$ )	xpt. TA ( $\mu\text{mol kg}^{-1}$ )	DIC ( $\mu\text{mol kg}^{-1}$ )	i.s. $\Omega_{\text{Ar}}$	i.s. pCO <sub>2</sub> ( $\mu\text{atm}$ )	xpt. $\Omega_{\text{Ar}}$	xpt. pCO <sub>2</sub> ( $\mu\text{atm}$ )	
2011	<del>380 <math>\mu\text{atm CO}_2</math></del>	10	33	2300.3	<b>2307.3</b>	2094.4	2.3 ± 0.2	336.2 ± 37.7	<b>2.4 ± 0.2</b>	<b>324.8 ± 35.8</b>	
Atlantic	<del>21% <math>\text{O}_2</math> LC/HO</del>	15	33	2300.3	<b>2307.3</b>	2066.5	2.6 ± 0.7	404.5 ± 172.7	<b>2.7 ± 0.7</b>	<b>390.8 ± 164.5</b>	
		15	35	<b>2296.4</b>	2354.5	2066.4	<b>2.5 ± 0.1</b>	<b>382.3 ± 20.4</b>	3.1 ± 0.1	297.7 ± 14.3	
		20	34	2353.4*	<b>2345.8</b>	2028.6	3.6 ± 0.2	302.8 ± 31.6	<b>3.5 ± 0.2</b>	<b>311.6 ± 32.9</b>	
		20	34	2366.0	<b>2367.2</b>	2077.5	3.3 ± 0.1	363.1 ± 23.2	<b>3.3 ± 0.1</b>	<b>361.4 ± 23.1</b>	
	<del>380 <math>\mu\text{atm CO}_2</math></del>	10	33	2300.3	<b>2307.3</b>	1919.7	4.0	139.0	<b>4.1</b>	<b>135.5</b>	
	<del>10% <math>\text{O}_2</math> LC/LO</del>	15	33	2300.3	<b>2307.3</b>	1774.8	5.5 ± 0.6	101.2 ± 23.9	<b>5.6 ± 0.6</b>	<b>99.0 ± 23.3</b>	
		15	35	<b>2296.4</b>	2354.5	1852.7	<b>4.6</b>	<b>139.2</b>	5.3	116.1	
	<del>800 <math>\mu\text{atm CO}_2</math></del>	10	33	2300.3	<b>2307.3</b>	2219.7	1.2 ± 0.2	779.9 ± 114.0	<b>1.2 ± 0.2</b>	<b>742.4 ± 106.8</b>	
	<del>21% <math>\text{O}_2</math> HC/HO</del>	15	33	2300.3	<b>2307.3</b>	2208.0	1.3	908.7	<b>1.4</b>	<b>867.8</b>	
		15	35	<b>2296.4</b>	2354.5	2139.5	<b>1.9</b>	<b>585.2</b>	2.4	434.4	
		20	34	2353.4*	<b>2345.8</b>	2176.9	2.1 ± 0.1	651.8 ± 23.4	<b>2.1 ± 0.1</b>	<b>678.2 ± 24.8</b>	
		20	34	2366.0	<b>2367.2</b>	2212.7	1.9 ± 0.4	786.0 ± 196.0	<b>1.9 ± 0.4</b>	<b>780.9 ± 194.2</b>	
	<del>800 <math>\mu\text{atm CO}_2</math></del>	15	33	2300.3	<b>2307.3</b>	2186.2	1.5 ± 0.2	788.7 ± 157.6	<b>1.5 ± 0.2</b>	<b>754.9 ± 148.3</b>	
	<del>10% <math>\text{O}_2</math> HC/LO</del>	15	35	<b>2296.4</b>	2354.5	2179.6	<b>1.5 ± 0.3</b>	<b>782.9 ± 164.6</b>	2.0 ± 0.3	558.2 ± 103.9	
2012	<del>380 <math>\mu\text{atm CO}_2</math></del>	10	32.1	2151.9*	<b>2142.8</b>	1934.8	2.2 ± 0.1	285.2 ± 21.4	<b>2.3 ± 0.1</b>	<b>283.0 ± 21.2</b>	
Pacific	<del>21% <math>\text{O}_2</math> LC/HO</del>	10	33.5	2208.0	<b>2222.7</b>	2001.9	2.4 ± 0.6	302.2 ± 100.9	<b>2.4 ± 0.6</b>	<b>303.3 ± 101.4</b>	
		15	32.5	<b>2182.6*</b>	2095.7	1983.4	<b>2.2 ± 0.0</b>	<b>388.1 ± 5.5</b>	1.4 ± 0.0	646.7 ± 11.5	
		15	33.5	2208.0	<b>2222.7</b>	2020.8	2.3 ± 0.2	407.7 ± 52.1	<b>2.3 ± 0.2</b>	<b>409.1 ± 52.4</b>	
		<del>380 <math>\mu\text{atm CO}_2</math></del>	10	32.5	<b>2182.6*</b>	2095.7	1973.9	<b>2.3 ± 0.1</b>	<b>295.5 ± 20.0</b>	1.4 ± 0.1	489.2 ± 41.2
		<del>10% <math>\text{O}_2</math> LC/LO</del>	15	33.5	2208.0	<b>2222.7</b>	2017.5	2.3	3956.0	<b>2.3</b>	<b>397.4</b>
	<del>800 <math>\mu\text{atm CO}_2</math></del>	10	32.1	2151.9*	<b>2142.8</b>	2026.3	1.4 ± 0.1	525.0 ± 35.0	<b>1.4 ± 0.1</b>	<b>519.7 ± 34.5</b>	
	<del>21% <math>\text{O}_2</math> HC/HO</del>	10	33.5	2208.0	<b>2222.7</b>	2120.6	1.3	628.2	<b>1.3</b>	<b>631.2</b>	
		15	32.5	<b>2182.6*</b>	2095.7	2031.7	<b>1.8 ± 0.1</b>	<b>527.6 ± 50.9</b>	1.0 ± 0.1	952.4 ± 115.1	
		15	33.5	2208.0	<b>2222.7</b>	2112.2	1.4 ± 0.2	736.0 ± 96.0	<b>1.4 ± 0.2</b>	<b>739.4 ± 96.6</b>	
	<del>800 <math>\mu\text{atm CO}_2</math></del>	10	32.5	<b>2182.6*</b>	2095.7	2066.5	<b>1.4 ± 0.1</b>	<b>545.5 ± 65.1</b>	0.8 ± 0.1	1056.0 ± 151.6	
	<del>10% <math>\text{O}_2</math> HC/LO</del>	15	33.5	2208.0	<b>2222.7</b>	2118.3	1.4	762.4	<b>1.4</b>	<b>766.0</b>	

Table 4: The average wet mass (mass; g) and mass-specific oxygen consumption rate (MO<sub>2</sub>;  $\mu\text{mol O}_2 \text{ g}^{-1} \text{ h}^{-1}$ )  $\pm$  the standard error (SE) for each treatment (Treat.) and species. The numbers of replicates (N) per treatment are reported and the species are arranged by temperature (Temp; °C) as well as the year and basin of collection.

Year	Temp.	Species	Treat.	N	mass	$\pm$ SE	MO <sub>2</sub>	$\pm$ SE		
2011 Atlantic	10	<i>Limacina retroversa</i>	LC/HO	12	.00281	0.00037	10.33	1.17		
			HC/HO	13	.00284	0.00031	10.10	0.56		
			LC/LO	9	.00274	0.00026	8.12	0.66		
			HC/LO	9	.00377	0.00053	4.21	0.55		
	15	<i>Clio pyramidata</i>	LC/HO	10	.01944	0.00408	7.81	0.71		
			HC/HO	8	.01410	0.00435	8.55	1.48		
			LC/LO	9	.02363	0.00867	6.63	1.21		
			HC/LO	8	.03945	0.00467	6.99	0.45		
		<i>Cuvierina atlantica</i>	LC/HO	8	.04493	0.00264	5.05	0.63		
			LC/LO	10	.04636	0.00252	3.25	0.28		
			HC/LO	10	.05040	0.00219	4.29	0.37		
			<i>Diacria trispinosa</i>	LC/HO	8	.03718	0.00316	4.44	0.56	
	20	<i>Cuvierina atlantica</i>	LC/HO	9	.01876	0.00396	4.31	0.85		
			HC/HO	9	.01683	0.00284	4.53	1.13		
		<i>Cavolinia inflexa</i>	LC/HO	8	.00626	0.00104	14.30	1.48		
			HC/HO	4	.00508	0.00049	13.81	1.39		
<i>Styliola subula</i>		LC/HO	10	.00400	0.00038	13.96	1.80			
		HC/HO	8	.00289	0.00035	15.95	0.87			
		2012 Pacific	10	<i>Limacina helicina</i>	LC/HO	7	.00140	0.00026	5.26	1.17
					HC/HO	8	.00149	0.00021	5.51	0.69
LC/LO	6				.00300	0.00058	4.91	0.69		
HC/LO	10				.00296	0.00038	7.18	1.45		
<i>Clio pyramidata</i>	LC/HO		9	.02646	0.00258	5.43	0.45			
	HC/HO		8	.02355	0.00369	4.39	0.60			
	LC/LO		14	.01459	0.00185	5.58	0.81			
	HC/LO		12	.01250	0.00245	5.72	1.14			
15	<i>Cuvierina pacifica</i>		LC/HO	4	.01829	0.00563	3.41	0.56		
			HC/HO	7	.02130	0.00636	3.53	0.57		
	<i>Cavolinia inflexa</i>	LC/HO	5	.01330	0.00062	3.53	0.44			
		HC/HO	8	.01556	0.00149	3.34	0.41			
		LC/LO	4	.01405	0.00185	2.41	0.33			
		HC/LO	2	.01855		3.98				
	<i>Styliola subula</i>	LC/HO	6	.00360	0.00044	5.30	1.20			
		HC/HO	4	.00220	0.00029	7.73	2.14			
<i>Clio pyramidata</i>		LC/HO	4	.03020	0.0037	3.82	0.66			
		HC/HO	5	.02904	0.00329	3.21	0.27			

Table 5: Statistical results of the univariate general linear models (GLM) for each species were analyzed separately by year and are listed relative to the temperature of the experiment (Temp.; °C). For species studied at multiple temperatures (denoted by \*), the metabolic rates were adjusted to 15°C using a  $Q_{10} = 2$  to allow for direct comparison. The effect of the independent factors of CO<sub>2</sub> level (CO<sub>2</sub>), O<sub>2</sub> level (O<sub>2</sub>), their interactive effect (Int.) and the covariate of mass were analyzed in regards to the metabolic rate and reported as *p*-values for the Pacific (mean mass specific metabolic rate values found in Table 4). For the Atlantic, each treatment was tested as independent (Treat.) due to the accidentally low CO<sub>2</sub> condition in the LC/LO gas mixture. We report whether the data met the assumption of normality of the residuals with Shapiro-Wilk (norm.; for *p* under 0.05 the assumption is not met) and heterogeneity of variance (var.; for *p* under 0.05 the assumption is not met) and denote in bold where the dataset did not fully meet these statistical assumptions. Note that for the sole case where the treatment or CO<sub>2</sub> effect was significant (*L. retroversa*) all assumptions were met.

Year	Temp.	Species	Effect on metabolic rate							
			CO <sub>2</sub>	O <sub>2</sub>	Int.	Treat.	Mass	norm.	var.	
2011 Atlantic	10	<i>Limacina retroversa</i>					<0.001	<0.001	0.542	0.522
	15	<i>Clio pyramidata</i>					0.295	<0.001	0.079	0.263
		<i>Cuvierina atlantica</i> *					0.174	<0.001	0.972	< <b>0.001</b>
		<i>Diacria trispinosa</i>	.731					<0.001	0.802	0.885
		<i>Cavolinia inflexa</i>	.677					.008	0.498	0.876
		<i>Styliola subula</i>	.791					.040	.922	<b>0.014</b>
2012 Pacific	10	<i>Limacina helicina</i>	.464	.323	.914		.007	<b>0.045</b>	<b>0.026</b>	
	15	<i>Clio pyramidata</i> *	.255	.156	.726		.018	< <b>0.001</b>	0.068	
		<i>Cuvierina pacifica</i>	.709				<0.001	0.639	0.357	
		<i>Cavolinia inflexa</i>	.309	.717	.219		.113	0.581	0.28	
		<i>Styliola subula</i>	.763				.668	0.353	0.325	

## Figure legends

**Figure 1: Cruise tracks and animal sampling.** Thecosomes were collected during the night at stations along the main survey transect (solid line) and at stations during transit (dashed line) during cruises to the northwest Atlantic in 2011 and northeast Pacific in 2012. The shapes correspond to the species caught at each station and used in this study. Blue (10 °C), grey (15 °C) and red (20 °C) boxes around the station numbers (#) correspond to the temperature that was representative of 25-100 m at each station (Table 2) and used in the experiments with animals from that station.

**Figure 2: Hydrography of sampling regions.** Hydrographic profiles of stations representative of the specific water mass types from the northern (P-T5, P-6, A-26), middle (P-18, A-19) and southern (P-32, A-8) portions of the Pacific (P) and Atlantic (A) study transects (station locations: Fig. 1). At station P-T5, the temperature profile (grey) was from an XBT cast because no CTDs were conducted during transits. For all stations along the main transects, left-hand plots show temperature (grey), salinity (black) and oxygen (black dotted) measured via sensors on the CTD and binned to 1 m depth intervals. Middle plots show TA (black) and DIC (grey) from discrete bottle samples (dots show depths of bottle samples). Right-hand plots show pCO<sub>2</sub> (black) and aragonite saturation state ( $\Omega_{Ar}$ ; grey) calculated based on TA and DIC measurements.

**Figure 3: Thecosome respirometry.** Mean metabolic rate and standard error ( $\mu\text{mol O}_2 \text{ g}^{-1} \text{ h}^{-1}$ ) of thecosomes exposed to low (i.e., ambient) CO<sub>2</sub> and normal levels of O<sub>2</sub> (light blue; LC/HO), high CO<sub>2</sub> and normal O<sub>2</sub> levels (dark blue; HC/HO), low CO<sub>2</sub> and low O<sub>2</sub> (light red; LC/LO), or high CO<sub>2</sub> and low O<sub>2</sub> (dark red; HC/LO). The species and temperature of the experiment are reported below the x-axis. Significance is reported based on a basin, species, and temperature specific GLM which tested for the effect of treatment on O<sub>2</sub> consumption with a Bonferroni post-hoc analysis (Table 5). In the Atlantic analysis each treatment was tested independently, while in the Pacific CO<sub>2</sub> and O<sub>2</sub> were treated as factors. For each species and temperature, treatments are reported as non-significant (N.S.) or, in the case of significance, by letters that indicate which treatments are statistically similar (same letter) or different (different letter) at a p-value < 0.05.

Note that for *C. atlantica* the metabolic rates of individuals respired at 20° C were converted to 15°C using a temperature coefficient of 2 (see methods) for this GLM analysis.

**Figure 4:** Log transformed metabolic rates ( $\mu\text{mol O}_2 \text{ h}^{-1}$ ) for *L. retroversa* at 10 °C, not normalized to mass, plotted against the log transformed wet mass (mg) of individuals exposed to low CO<sub>2</sub> and normal levels of O<sub>2</sub> (black circles; LC/HO), high CO<sub>2</sub> and normal O<sub>2</sub> levels (dark grey diamonds; HC/HO), low CO<sub>2</sub> and low O<sub>2</sub> (white circles; LC/LO), or high CO<sub>2</sub> and low O<sub>2</sub> (light grey diamonds; HC/LO).