



1 **The metabolic response of thecosome pteropods from the North**  
2 **Atlantic and North Pacific Oceans to high CO<sub>2</sub> and low O<sub>2</sub>**

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32 **Abstract.** As anthropogenic activities, notably the burning of fossil fuels, increase carbon  
33 dioxide (CO<sub>2</sub>) and result in a decrease in oxygen (O<sub>2</sub>) concentrations in the ocean system, it  
34 becomes important to understand how different populations of marine animals will respond.  
35 Water that is naturally lower in pH, with a high concentration of carbon dioxide (hypercapnia)  
36 and a low concentration of oxygen, occurs at shallow depths (200-500 m) in the North Pacific  
37 Ocean, whereas similar conditions are absent throughout the upper water column in the North  
38 Atlantic. This contrasting hydrography provides a natural experiment to explore whether  
39 differences in environment cause populations of cosmopolitan pelagic calcifiers, specifically the  
40 aragonitic-shelled pteropods, to have a different physiological response when exposed to  
41 hypercapnia and low O<sub>2</sub>. Using closed-chamber end-point respiration experiments, eight species  
42 of pteropods from the two ocean basins were exposed to high CO<sub>2</sub> (~800 μatm) while six species  
43 were also exposed to moderately low O<sub>2</sub> (10%, or ~130 μmol kg<sup>-1</sup>) and a combined treatment of  
44 low O<sub>2</sub>/high CO<sub>2</sub>. None of the species tested showed a change in metabolic rate in response to  
45 high CO<sub>2</sub> alone. Of those species tested for an effect of O<sub>2</sub>, only *Limacina retroversa* from the  
46 Atlantic showed a response to the combined treatment, resulting in a reduction in metabolic rate.  
47 Our results suggest that pteropods have mechanisms for coping with short-term CO<sub>2</sub> exposure  
48 and suggest that there can be interactive effects between stressors on the physiology of these  
49 open ocean organisms that correlate with natural exposure to low O<sub>2</sub> and high CO<sub>2</sub>; these are  
50 considerations that should be taken into account in projections of organismal sensitivity to future  
51 ocean conditions.

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60 **Key Words: ocean acidification, zooplankton, respiration**

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## 62 1. Introduction

63 Ocean acidification, a result of the dissolution of anthropogenically produced carbon dioxide  
64 ( $\text{CO}_2$ ) into sea water, is increasingly considered to be one of the most pervasive human changes  
65 to the marine system (Doney et al., 2009; Gruber, 2011; Halpern et al., 2008). The pH of the  
66 ocean surface has already dropped by  $\sim 0.1$  units relative to preindustrial levels and is predicted  
67 to drop another 0.3-0.4 pH units in the next one hundred years (Bopp et al., 2013; Haugan and  
68 Drange, 1996; IPCC, 2013). As  $\text{CO}_2$  dissolves in the ocean, it causes changes in seawater  
69 carbonate chemistry, notably increasing hydrogen ion concentration and decreasing the  
70 concentration of carbonate ions. As a consequence of the changing equilibria, there is a reduction  
71 in pH and in the saturation state of calcium carbonate ( $\text{CaCO}_3$ ), including the biogenic forms of  
72 calcite and aragonite. As ocean acidification continues, eventually the water becomes  
73 undersaturated and corrosive, meaning that, in the absence of compensating biological action,  
74 conditions will favor the dissolution of the  $\text{CaCO}_3$  found in the shells and skeletons of calcifying  
75 organisms, with aragonite being more sensitive than calcite (Millero, 2007).

76 Ocean acidification, therefore, impacts calcifying species on multiple fronts. Changes in  
77 environmental pH can modify the acid-base balance of intra- and extracellular fluids of marine  
78 organisms, which may result in reduced fitness or outright mortality (Seibel and Fabry, 2003;  
79 Seibel and Walsh, 2001; Widdicombe and Spicer, 2008). Changes in  $\text{CaCO}_3$  saturation state can  
80 also affect the ability of some calcifying animals to create and maintain calcium carbonate  
81 structures with implications for energetics, survival, competition and biogeochemical export  
82 (Fabry et al., 2008; Riebesell et al., 2000; Ries et al., 2009). Understanding the long-term effects  
83 of this increase in ocean acidity on both organisms and ecosystems has, therefore, become of  
84 great concern. Important and outstanding research goals are to understand how changing  $\text{CO}_2$   
85 impacts current populations and to predict whether these populations will be able to adapt to the  
86 rate and severity of the rising anthropogenic  $\text{CO}_2$  inputs (e.g. Dam, 2013; Kelly and Hofmann,  
87 2013; Sunday et al., 2011).

88 One approach to understanding the response of marine animals to increasing acidification  
89 is to examine places where animals already experience conditions of elevated  $\text{CO}_2$   
90 (hypercapnia). By comparing individuals that inhabit regions of high  $\text{CO}_2$  with those that never  
91 experience high levels naturally, insight can be gained into the potential for adaptation of species  
92 to high  $\text{CO}_2$  over evolutionary timescales. The ocean chemistry of the northwest Atlantic and the



93 northeast Pacific Oceans provides such a natural experiment. High CO<sub>2</sub> concentrations are  
94 generally absent from the upper water column in the Atlantic (Wanninkhof et al., 2010). In  
95 contrast there currently are hypercapnic conditions, where the water is undersaturated with  
96 respect to aragonite, in the upper water column in parts of the Pacific.

97         The source of hypercapnia in the Pacific Ocean is a combined result of ocean circulation  
98 coupled with the biological processes, leading the old deep waters of the Pacific to be some of  
99 the most CO<sub>2</sub> rich in the ocean (Broecker et al., 1982). On top of this natural process, ocean  
100 acidification also plays a role: the pH of the upper water column in the North Pacific is  
101 decreasing by on the order of 0.001–0.002 pH units per year (Byrne et al., 2010), corresponding  
102 to a total CO<sub>2</sub>, or dissolved inorganic carbon (DIC), increase of 1–2 μmol kg<sup>-1</sup> yr<sup>-1</sup> (Peng et al.,  
103 2003; Sabine et al., 2008; Sabine and Tanhua, 2010). Although the surface waters in these  
104 regions are typically well oxygenated and with a pH > 8, animals that live at or migrate to depth  
105 experience increasingly low oxygen (O<sub>2</sub>), pH, undersaturation with respect to calcium carbonate  
106 and elevated CO<sub>2</sub> (Seibel, 2011). Historically these regions, which occur in many ocean basins,  
107 were in fact known more for their low O<sub>2</sub> than for their high CO<sub>2</sub> and were termed oxygen  
108 minimum zones (OMZs). These carbon maximum/oxygen minimum zones are extensive in the  
109 North Pacific Ocean, whereas similar conditions are rare in much of the Atlantic (Paulmier et al.,  
110 2011). Closely related taxa and cosmopolitan species in these two regions therefore experience  
111 very different pH levels, CO<sub>2</sub> and O<sub>2</sub> concentrations in their normal distribution. Independent  
112 from high CO<sub>2</sub>, the reduced O<sub>2</sub> at depth in these OMZs has a profound impact on zooplankton  
113 distribution (i.e.: Escribano et al., 2009; Maas et al., 2014; Wishner et al., 2008) and can have  
114 important implications for the physiology of zooplankton (Childress and Seibel, 1998; Rosa and  
115 Seibel, 2008; Seibel, 2011).

116         Thecosome pteropods are an interesting group for investigating planktonic exposure and  
117 response to hypercapnia and low O<sub>2</sub>. Broadly distributed throughout the open ocean, species of  
118 thecosomes found in shallow waters of temperate and polar seas can become a numerically  
119 dominant member of the zooplankton community (Bednaršek et al., 2012a; Hunt et al., 2008; van  
120 der Spoel, 1967). As such, they can be an important part of the food chain (Armstrong et al.,  
121 2005; Hunt et al., 2008; Karnovsky et al., 2008), and contribute substantially to carbon flux  
122 (Bauerfeind et al., 2009; Fabry and Deuser, 1991; Manno et al., 2010; Noji et al., 1997). Bearing  
123 thin shells of aragonite, one of the less stable forms of biogenic calcium carbonate, the



124 calcification of thecosomes has been shown to be impacted by exposure to conditions replicating  
125 the projected changes in surface water pH and saturation state of the future ocean in the next 100  
126 years (Comeau et al., 2009; Lischka et al., 2011; Manno et al., 2012). Furthermore, recent  
127 assessments have shown that their shells already bear the mark of acidification in upwelling and  
128 polar regions characterized by under-saturated conditions with respect to aragonite (Bednaršek  
129 et al., 2014a, b; Bednarsek and Ohman, 2015; Bednaršek et al., 2012b). Studies of metabolism  
130 and behavior, however, reveal a complex sensitivity to pH, dependent upon natural pre-exposure  
131 and the presence of interactive stressors (Comeau et al., 2010a; Maas et al., 2012a; Manno et al.,  
132 2012; Seibel et al., 2012).

133 Previous work has shown that some tropical and sub-tropical thecosome species undergo  
134 diel vertical migrations into persistent and pronounced regions of low O<sub>2</sub> and hypercapnia in the  
135 Eastern tropical North Pacific. These species showed no change in metabolic rate (O<sub>2</sub>  
136 consumption) when exposed to high CO<sub>2</sub> (1000 µatm), revealing the ability of some groups of  
137 thecosome to maintain aerobic metabolism in acidified waters for short periods of time. The one  
138 species in this region that does not migrate, however, responded with a suppression of  
139 metabolism when exposed to high CO<sub>2</sub> (Maas et al., 2012a). This work in the Eastern tropical  
140 North Pacific provides evidence that there may be the potential for environmental adaptation of  
141 thecosomes to high CO<sub>2</sub>, but provides no insight into the combined effects of CO<sub>2</sub> with low O<sub>2</sub>.  
142 Although research into this topic is underway for other calcifying organisms in coastal habitats  
143 (Gobler et al., 2014; Melzner et al., 2013), in the open ocean our understanding remains limited.

144 The objective of this study, therefore, was to compare the effect of high CO<sub>2</sub> and low O<sub>2</sub>  
145 on thecosome pteropods from the northwest Atlantic and the northeast Pacific Oceans. One of  
146 the benefits of this comparison is that there are a number of species of thecosomes that have  
147 cosmopolitan distributions occupying both basins and that are known to be diel vertical  
148 migrators (Table 1; Bé and Gilmer, 1977; van der Spoel, 1967). Thus populations in the Pacific  
149 would naturally experience hypercapnia and low O<sub>2</sub> in their daytime deep water habitat in the  
150 Pacific, while in contrast, those from the Atlantic would rarely experience the same deep water  
151 environmental stressors. Using these organisms, which are presumably adapted to their local  
152 conditions, we can test whether species exhibit a population-specific physiological response to  
153 these environmental conditions indicative of different sensitivities.

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## 155 2. Methods

156 Thecosome pteropods caught during cruises to the northwest Atlantic and northeast Pacific were  
157 exposed aboard ship to manipulated conditions of moderately high CO<sub>2</sub> and/or low O<sub>2</sub> for short  
158 durations (< 18 h). After this exposure their metabolic rates were measured and then compared to  
159 determine whether there were species- or region-specific responses to the treatments.

### 160 2.1 Sampling

161 Animals were collected on two cruises, the first on August 7<sup>th</sup> – September 1<sup>st</sup> 2011 in the  
162 northwest Atlantic aboard the R/V *Oceanus*, and the second in the northeast Pacific from August  
163 9<sup>th</sup> – September 18<sup>th</sup> 2012 aboard the R/V *New Horizon*. The majority of the sampling in the  
164 Atlantic took place along a three-part ‘z’-shaped transect running between 35°N 52°W and 50°N  
165 42°W, as well as at sites during transit to and from port (Fig. 1). The first portion of this cruise  
166 track corresponded to a segment of the World Ocean Circulation Experiment / Climate and  
167 Ocean: Variability, Predictability and Change project (WOCE/CLIVAR) line A20. In the North  
168 Pacific the main sampling took place along a two-part transect running between 50°N 150°W  
169 and 33.5°N 135°W, corresponding to a portion of WOCE/CLIVAR line P17N, as well as at sites  
170 during transit to and from port (Fig. 1).

171 Sampling was part of a larger interdisciplinary project employing a suite of tools to  
172 explore the natural distribution and hydrographic environment of the thecosomes. The sampling  
173 design included underway measurements of hydrography, carbonate chemistry and multi-  
174 frequency acoustic backscatter. Comprehensive sampling of the water column was conducted at  
175 pre-determined stations using a depth-stratified 1-m<sup>2</sup> Multiple Opening/Closing Net and  
176 Environmental Sensing System with 150 μm mesh nets (MOCNESS; Wiebe et al., 1985), a  
177 towed broadband echosounder, video plankton recorder casts, and profiles with a 24 10-L Niskin  
178 bottle rosette and associated conductivity, temperature and depth (CTD) package.

179 Hydrographic profiles associated with this study were collected of temperature, O<sub>2</sub> and  
180 salinity using the CTD-Rosette-Niskin bottle package at stations along the main survey transects  
181 (Fig. 1). This CTD was equipped with dual temperature and conductivity sensors, a Digiquartz  
182 pressure sensor, a SBE43 dissolved oxygen sensor, a biospherical underwater photosynthetically  
183 active radiation (PAR) sensor with surface reference, a Wet Labs C-Star transmissometer (660  
184 nm wavelength), and a Wet Labs ECO-AFL fluorometer. Where CTD casts were unavailable, at  
185 stations conducted during the transits to and from port, an expendable bathythermograph (XBT)



186 was deployed to determine the temperature of the water column. Bottle samples of carbonate  
187 parameters, nutrients, and other parameters were collected at selected water depths using the  
188 CTD-Rosette package.

## 189 **2.2 Environmental Carbonate Chemistry**

190 Discrete pH samples were directly collected from the 10-L Niskin bottle into 10 cm cylindrical  
191 optical cells and measured within 4 h of collection (Clayton and Byrne, 1993; Dickson et al.,  
192 2007). These pH samples were analyzed spectrophotometrically on an Agilent 8453  
193 spectrophotometer at a control temperature ( $25.0 \pm 0.1^\circ\text{C}$ ) following the method detailed in  
194 Dickson (2007) and in Clayton and Byrne (1993) using m-cresol purple as the indicator. The pH  
195 results in total scale have been corrected for indicator impurity (Liu et al., 2011) and indicator  
196 perturbation to seawater samples. The pH measurements have a precision better than 0.001 and  
197 an accuracy of  $\sim 0.002$ .

198 Nutrient samples (nitrate/nitrite, phosphate, silicate, and ammonia) were collected in 20  
199 mL plastic bottles after filtration through a 0.22 $\mu\text{m}$  Pall capsule filter and kept frozen until  
200 analysis. Nutrient samples were analyzed either at the WHOI Nutrient Analytical Facility or the  
201 University of California, Santa Barbara, using a Lachat Instruments QuickChem 8000 four-  
202 channel continuous flow injection system, following standard colorimetric methods approved by  
203 U.S. Environmental Protection Agency.

204 Discrete samples were also taken for dissolved inorganic carbon (DIC) and total  
205 alkalinity (TA). These were collected in 250mL Pyrex borosilicate glass bottles after being  
206 filtered with a 0.45  $\mu\text{m}$  in-line capsule filter and poisoned with saturated mercuric chloride  
207 (Dickson et al., 2007). DIC samples were analyzed on a DIC auto-analyzer (AS-C3, Apollo  
208 SciTech, Bogart, USA) via sample acidification, followed by non-dispersive infrared  $\text{CO}_2$   
209 detection (LiCOR 7000: Wang et al., 2013; Wang and Cai, 2004). The instrument was calibrated  
210 with certified reference material (CRM) from Dr. A.G. Dickson at the Scripps Institution of  
211 Oceanography. The DIC measurements have a precision and accuracy of  $\pm 2.0 \mu\text{mol kg}^{-1}$ . TA  
212 measurements were made with an Apollo SciTech alkalinity auto-titrator, a Ross combination  
213 pH electrode, and a pH meter (ORION 3 Star) based on a modified Gran titration method with a  
214 precision and accuracy of  $\pm 2.0 \mu\text{mol kg}^{-1}$  (Wang and Cai, 2004).

215 The remaining water column carbonate system parameters, including aragonite saturation  
216 state and  $\text{pCO}_2$  were calculated from DIC-pH pairs at in situ nutrient, temperature, salinity and



217 pressure using the software CO2Sys (Pierrot et al., 2006) and the dissociation constants of  
218 Mehrbach et al. (1973), refitted by Dickson and Millero (1987), and the  $\text{KHSO}_4$  dissociation  
219 constant from Dickson (1990). Depths for  $\text{pH}=7.7$ ,  $\text{pCO}_2=800 \mu\text{atm}$  and aragonite saturation  
220 state of 1 were then linearly interpolated using the closest available measurements.

221 Surface water  $\text{pCO}_2$  was continuously measured throughout both cruises using an  
222 automated underway system (Model 8050, General Oceanics Inc., USA) based on headspace air-  
223 seawater equilibration followed by infrared detection (LiCOR 7000). This system was calibrated  
224 every 1-2 hours with three  $\text{CO}_2$  gas standards traceable to World Meteorological Organization  
225  $\text{CO}_2$  Mole Fraction Scale. These underway  $\text{pCO}_2$  measurements have a precision and accuracy of  
226  $\sim \pm 1 \mu\text{atm}$ . Measurements made by the underway system provide insight into the carbonate  
227 chemistry parameters at stations made in transit where bottle samples were not collected.

### 228 2.3 Specimen Capture

229 Thecosome species were chosen for physiological study opportunistically as they appeared in net  
230 samples at successive stations. Species were targeted specifically for their abundance and the  
231 likelihood of their presence in both ocean basins. Most individuals were collected with a 1-m  
232 diameter, 150- $\mu\text{m}$  mesh Reeve net with a  $\sim 25 \text{ L}$  cod-end in the Atlantic and a similar 1-m  
233 diameter, Reeve net equipped with 330- $\mu\text{m}$  mesh in the Pacific. Use of the Reeve net with its  
234 large and heavy cod-end in combination with slow haul rates (typically  $5\text{-}10 \text{ m min}^{-1}$ ) allowed  
235 for a gentle collection of the delicate thecosomes, consistently supplying animals in good  
236 condition with undamaged shells and external mantle appendages. Net tows were made at night  
237 when animals were expected to congregate at shallow depths, were  $\sim 1 \text{ h}$  in duration in an effort  
238 to minimize the handling time of the organisms, and reached a maximal depth between 100–150  
239 m. Depths were targeted that had a high chlorophyll  $a$  peak during CTD casts, high acoustic  
240 backscattering on the echosounder, and/or where thecosomes had been abundantly sampled at  
241 the same station using the MOCNESS. Occasionally, individuals of less abundant species were  
242 collected from the nets of the MOCNESS for physiological study, but only if their shells were  
243 undamaged and they were swimming normally.

244 Post-capture, individuals were transferred to filtered water in densities of  $< 15 \text{ ind. L}^{-1}$   
245 and kept for at least 8 h in temperature controlled waterbaths to allow for gut clearance.  
246 Temperatures for experimentation (20, 15 or  $10^\circ\text{C}$ ) were chosen to be generally representative of  
247 the waters from which the animals were sampled, based on the vertical distributions and





248 hydrographic conditions documented with the stratified MOCNESS sampling. Chosen  
249 temperatures were typically the average of the water temperature between 25-100 m, although in  
250 the middle section of the Atlantic cruise experimental temperatures were reflective of the 25–50  
251 m average due to the particularly shallow vertical distribution of the dominant species (*Limacina*  
252 *retroversa*) sampled in this region. This was to ensure that experiments were occurring at  
253 physiologically relevant and, presumably, natural temperatures for each species. After gut  
254 clearance, individuals that were in good condition (i.e., swimming and with shell intact) were  
255 used for oxygen consumption experiments.

#### 256 **2.4 Experimental Exposures and Oxygen Consumption Rate**

257 Post-gut-clearance, healthy individuals were put into separate glass syringe respiration chambers  
258 with a known volume of 0.2  $\mu\text{m}$  filtered seawater and 25  $\text{mg L}^{-1}$  each of streptomycin and  
259 ampicillin. This minimized the microbial respiration effects on the measurements of carbonate  
260 chemistry and  $\text{O}_2$  consumption rates by pteropods during the experiments. The inclusion of  
261 antibiotics, a method which has previously been used with thecosomes to prevent bacterial  
262 growth in respiration experiments (Maas et al., 2012b), was shown during the Pacific cruise to  
263 have no effect on the  $\text{O}_2$  consumption of at least *Limacina helicina*, for the exposure durations  
264 associated with these experiments (Howes et al., 2014). The volume of water in the treatments  
265 was chosen to complement the size of the organism and temperature of the experiment and  
266 ranged between 15-50 mL in 2011 and 8-20 mL in 2012. For every 3-5 treatment chambers, a  
267 “control” respiration chamber (experimental seawater with antibiotics and without pteropods)  
268 was set up to monitor microbial activity and to provide water for characterization of the starting  
269 conditions.

270 Filtered seawater for experimental exposures was collected during both cruises in batches  
271 at approximately weekly intervals from the surface; experimental water thus began with  
272 chemical properties (notably including TA, DIC, pH, as well as salinity) reflective of the local  
273 environment and was then manipulated to modify  $\text{CO}_2$  and/or  $\text{O}_2$  concentrations. Manipulations  
274 were achieved by bubbling 1 L batches of collected seawater with gas mixes (certified accurate  
275 to  $\pm 2\%$ ) for 45–60 min with one of two oxygen (21 and 10%  $\text{O}_2$ ) levels crossed with two  $\text{CO}_2$   
276 (nominally 380 ppm and 800 ppm) levels. At the time of the experiment, surface air  $\text{pCO}_2$   
277 conditions were on average ca. 380 ppm, dictating our ambient (LC) conditions. In 2011 the



278 ambient condition ( $\sim 21\%$   $O_2$  and  $380 \mu\text{atm } CO_2$ ) was achieved by bubbling with an ambient  
279 clean air line, while in 2012 it was achieved by a certified 380 ppm gas mix.

280 The experimentally modified concentrations mimic the  $CO_2$  and  $O_2$  levels that would be  
281 experienced by thecosomes within the top 400 m of the Pacific Ocean, and reflect the average  
282 projected atmospheric  $CO_2$  level for the open ocean in the year 2100 (A2 emissions scenario,  
283 IPCC, 2007). This resulted in four total treatments: low (i.e., ambient)  $CO_2$ , high oxygen  
284 (LC/HO) representative of current ambient surface ocean conditions; high carbon, high oxygen  
285 (HC/HO), replicating what we expect average future surface oceans to resemble; low  $CO_2$ , low  
286 oxygen (LC/LO); and high carbon, low oxygen (HC/LO), which is similar to what organisms in  
287 the Pacific would experience during a diel vertical migration into the local oxygen minimum  
288 zone. The goal of this design was to allow us to compare directly the Atlantic and Pacific  
289 thecosomes to see whether exposure to  $800 \mu\text{atm } pCO_2$  and/or  $10\% O_2$  resulted in different  
290 outcomes. The level of low  $O_2$  chosen for this study was well above the threshold that has been  
291 designated as stressful for non-specialized metazoan life ( $< 2 \text{ mg } O_2 \text{ L}^{-1}$  or  $60 \mu\text{mol } O_2 \text{ kg}^{-1}$ ;  
292 Vaquer-Sunyer and Duarte, 2008), in order to test the non-lethal effect of moderately low  $O_2$  on  
293 individuals from the two ocean basins. Calculations based on the salinity and temperature of the  
294 water indicated that bubbling with  $10\% O_2$  achieved conditions of 10–13%  $O_2$  saturation at the  
295 start of experiments. Subsequent analyses (see below) also confirmed that intended  $CO_2$   
296 concentrations were achieved for all treatments within reasonable ranges, with the exception of  
297 the LC/LO Atlantic treatment. In this case, the gas cylinder was evidently improperly mixed by  
298 the manufacturer and analyses suggested a ca. 100 ppm  $CO_2$  concentration. The results for this  
299 treatment are still presented but should be interpreted as a distinct treatment.

300 Oxygen consumption was measured following similar techniques as described in Marsh  
301 and Manahan (1999). Briefly, at the conclusion of the experiment water was withdrawn from  
302 treatment or control chambers using an airtight 500  $\mu\text{L}$  Hamilton syringe and injected past a  
303 Clarke-type microcathode (part #1302, Strathkelvin Instruments, North Lanarkshire, United  
304 Kingdom) attached to an  $O_2$  meter (part #782) in a water-jacketed injection port (part #MC100).  
305 This was done three times, allowing the reading to stabilize for at least 30 seconds before a  
306 measurement was taken. Generally, the change in oxygen consumption was between 3–25% of  
307 the control value. In high oxygen experiments, if the oxygen level fell below 70% of air  
308 saturation they were excluded from the analysis. Animals were removed from the chamber,



309 blotted dry and frozen in liquid nitrogen. These individuals were later weighed using a  
310 microbalance ( $\pm 0.0001$  g) and the resulting mass specific  $O_2$  consumption rates are reported in  
311  $\mu\text{moles (g wet weight)}^{-1} \text{ h}^{-1}$ . Wet weights are here used as they are more relevant for  
312 physiological understanding of animal function (Childress et al., 2008) but dry weights can be  
313 estimated from these using the wet weight to dry weight relationships developed previously for  
314 pteropods (Ikeda, 2014). To replicate the duration of exposure that would be experienced by  
315 most thecosomes in the Pacific undergoing a daily migration to depth, the experiments were  
316 targeted to last 6–12 h. In practice, experiments ranged from 6–18 h for normoxic and 3–10 h for  
317 low  $O_2$  trials. This variation in duration resulted from balancing the need to elicit a measureable  
318 change in  $O_2$  concentration with preventing extreme  $O_2$  depletion of the chambers ( $< 6\%$  oxygen  
319 saturation) and accounting for multiple species of variable size and metabolic rate.

## 320 **2.5 Experimental Carbonate Chemistry**

321 Carbonate chemistry of the treatments was characterized in most cases via measurements of DIC  
322 and TA of experimental seawater, unless indicated otherwise. The process of measuring the  $O_2$  in  
323 the treatments used up a large portion of the water and then the chamber was unsealed and  
324 disturbed to remove the animal, rendering it impractical to measure the carbonate chemistry  
325 directly from the respiration chambers. DIC measurements were thus taken from control syringes  
326 within 18 h of the end of each experiment and used to represent the starting point of the  
327 carbonate chemistry conditions the animals experienced. Water samples were allowed to come to  
328 room temperature ( $> 6$  h) before analysis. DIC was measured using the same system as that used  
329 for the hydrographic characterization (see above). Estimates of the effect of  $CO_2$  production via  
330 respiration in treatment chambers on DIC were made using a respiratory quotient of 0.8 M of  
331  $CO_2$  per 1 M of  $O_2$  consumed (Mayzaud, 1976) to characterize the ending conditions of the  
332 experiments.

333 Due to the small volumes of water in the experimental chambers, it was not possible to  
334 measure both DIC and TA from the control syringes. Instead, TA samples intended to be  
335 representative of the starting experimental conditions were collected via siphoning from each  
336 batch of filtered and antibiotic-treated water. These samples were subsequently measured based  
337 on the analytical method described above (Wang and Cai 2004). TA of experimental water was  
338 assumed to have been constant over the course of each experiment as water was filtered ( $0.2 \mu\text{m}$ )



339 and antibiotic treated (thus microbial activities were kept at minimum), and aerobic respiration  
340 does not change TA in a significant way.

341 In some instances, however, measured TA from experimental water was substantially  
342 dissimilar to that of the surface measurements made from nearby in-situ surface bottle samples  
343 collected with the CTD ( $> 20 \mu\text{mol kg}^{-1}$ ; see section 3.3). Calculated  $\text{pCO}_2$  values in these cases  
344 were also significantly different from batches of experimental water collected from other  
345 locations, but bubbled with the same  $\text{CO}_2$  gas tank. These differences are more than 10 times the  
346 measurement precision/accuracy and 5 times the uncertainty of duplicate sampling and  
347 measurements during the cruises. They are also beyond the likely level of TA variation due to  
348 differences in sampling location (geographic and in depth) between the in situ bottle samples and  
349 experimental water batches and rather are likely a consequence of the difficulties associated with  
350 cleanly siphoning the experimental water batches (e.g., contamination during sampling). For  
351 completeness, the carbonate chemistry system parameters for the experimental water, including  
352 aragonite saturation state and  $\text{pCO}_2$ , are reported based on calculations using DIC-TA pairs using  
353 both the in situ and experimental TA; in those cases where the TA measurements diverged  
354 substantially ( $> 20 \mu\text{mol kg}^{-1}$ ), however, we base our interpretations on the in-situ measured TA  
355 at nearby CTD stations instead of the values of experimental water. In those circumstances  
356 where batch water was taken from test stations and CTD bottle data were unavailable, the  
357 experimental TA was checked using calculated TA values using DIC from the LC/HO treatments  
358 and  $\text{pCO}_2$  from the underway measurements.

## 359 2.6 Statistics

360 Oxygen consumption rates were tested for significant differences between groups with  
361 Bonferroni pairwise post-hoc comparisons using SPSS. Univariate General Linear Models  
362 (GLM) were conducted to determine the effect of  $\text{CO}_2$  level,  $\text{O}_2$  level, and their interactive effect  
363 using the log transformed oxygen consumption with log transformed wet mass as a covariate  
364 separately for each species (2 factor design; “ $\text{CO}_2 \times \text{O}_2$ ”). In the Atlantic this full factorial design  
365 was confounded by the incorrect gas mixture so each treatment was tested independently (1  
366 factor design; “treatment”). Species that were collected during both years/basins, and  
367 experiments conducted on species at multiple temperatures, were analyzed separately so that the  
368 effect of variations in mass between seasons and the changes in metabolic rate at different  
369 temperatures would not confound the analysis.



370 For some species the temperature of experimentation was different among stations within  
371 a basin. For analyses with these species when comparing species between ocean basins, we  
372 applied a standard temperature coefficient ( $Q_{10}$ ) to compare across temperatures. The adjusted  
373 rates ( $R_f$ ) were calculated at 15°C using a  $Q_{10}$  of 2 according to the equation:

$$374 \quad R_f = R_i * \left( Q_{10}^{\left( \frac{15 - T_i}{10} \right)} \right)$$

375 where  $R_i$  is the original metabolic rate measured at the original temperature ( $T_i$ ). Although  
376 previous work with thecosomes has shown that  $Q_{10}$  is species-specific (Maas et al., 2011; Maas  
377 et al., 2012b; Seibel et al., 2007), for many of the species used in this study there are no  
378 published estimates of  $Q_{10}$ . Thus, this coefficient value was chosen as it is mid-range for the  
379 published  $Q_{10}$  of non-polar thecosome species as recently compiled by Ikeda (2014; 1.3-2.7) and  
380 is consistent with estimates of average  $Q_{10}$  for marine ectotherms, which typically fall between  
381 2-3 (Hochachka and Somero, 2002; Seibel and Drazen, 2007).

382

### 383 3. Results

#### 384 3.1 Specimen Capture

385 Following currently accepted taxonomy, individuals from a total of eight species were collected  
386 over the course of the two cruises for physiological studies. The taxonomy of thecosomes has  
387 recently begun to be revisited using molecular and paleontological tools (i.e. Hunt et al., 2010;  
388 Janssen, 2012; Jennings et al., 2010; Maas et al., 2013), however, and there is growing evidence  
389 of cryptic speciation for some pteropod groups (Burrige et al., 2015; Gasca and Janssen, 2014).  
390 It thus should be noted that these inter-basin comparisons may be of cryptic congeners rather  
391 than conspecific populations.

392 We collected two species of thecosome pteropods exclusively from the Atlantic,  
393 *Limacina retroversa* (Fleming, 1823), a subpolar species, which is absent from the North Pacific,  
394 and *Diacria trispinosa* (Blainville, 1821), which can be found in temperate and tropical regions  
395 of the Atlantic, Pacific and Indian Oceans. Although present in both the North Atlantic and  
396 Pacific, the polar to sub-polar species *Limacina helicina* (Phipps, 1774), was only sampled in the  
397 Pacific transect. Collections of this species consisted of intermixed formae, the high spiraled  
398 *Limacina helicina helicina acuta* (van der Spoel, 1967), the lower spiraled *Limacina helicina*  
399 *helicina pacifica* (van der Spoel, 1967), and a forma that bore resemblance to both in a mixed



400 morphology. Since both the assemblage and morphology of these formae were mixed they were  
401 tested as one population/species. In both ocean basins we collected *Styliola subula* (Quoy and  
402 Gaimard, 1827), *Cavolinia inflexa* (Lesueur, 1813) and *Clio pyramidata* (Linnaeus, 1767).  
403 There is some morphological and molecular evidence that *Cuvierina columnella* (Rang, 1827) is  
404 actually multiple distinct species, now including *Cuvierina atlantica* and *Cuvierina pacifica*  
405 (Burridge et al., 2015; Janssen, 2005), and we tested individuals of these species from their  
406 respective ocean basins.

### 407 3.2 Hydrography

408 Two hydrographic regimes were evident along the North Pacific study transect (Table 2; Fig. 2).  
409 The northern-most stations, including portions of the transit from port and stations from 50°N  
410 150°W to 47°N 144.6°W were coldest, with the temperatures between 25-100 m ranging from  
411 5-10°C. In this area O<sub>2</sub> fell below 10% (~130 μmol kg<sup>-1</sup>) by 250 m. In this northern part of the  
412 transect, pH fell below 7.7 by 130 m, and pCO<sub>2</sub> had already reached 800 μatm by ~200 m.  
413 Individuals in this area experienced an  $\Omega_{Ar} = 1$  between 160-185 m, well within the typical diel  
414 vertically migratory range of both of the species found in the region (*C. pyramidata* and *L.*  
415 *helicina*). At stations from more southern latitudes, from 47°N 144.6°W to 33.5°N 135°W,  
416 temperatures at depths between 25-100 m were higher, ranging between 10-15°C, representative  
417 of the transition zone into the North Pacific Gyre. Along this portion of the transect O<sub>2</sub>  
418 concentration consistently fell below 10% by depths between 340 and 400 m. The depth at which  
419 pH fell below 7.7 increased gradually from ~150 to 230 m as latitude decreased. Similarly, the  
420 depth at which pCO<sub>2</sub> in this area reached 800 μatm deepened from 330 to 440 m, and the  
421 aragonite saturation horizon transitioned from 330 m to 430 m depth. The depth at which species  
422 would experience a pH below 7.7 was within the inhabited depth range known from the literature  
423 for all of the species tested in this study region, but only the species *Clio pyramidata* likely  
424 experienced 10% O<sub>2</sub>, 800 μatm pCO<sub>2</sub> and aragonite undersaturation in its typical distribution in  
425 this portion of the Pacific transect (Table 1).

426 In contrast to the Pacific, along the entire Atlantic transect O<sub>2</sub> concentration was above  
427 ~200 μmol kg<sup>-1</sup> in the top 500 m, while pCO<sub>2</sub> never reached 800 μatm and aragonite  
428 undersaturation never occurred throughout the top 1000 m. There were three dominant  
429 hydrographic regimes in the Atlantic (Table 2; Fig. 2). In the northeastern part of the sampling  
430 region (50°N 42°W to 44.9°N 42°W), where the Gulf Stream meets the Labrador Current,



431 average temperatures at 25-100 m were near 15°C and pH only fell below 7.7 at depths  
432 exceeding 400 m. Similarly, in the southwest part of the sampling region (from 42°N 52°W to  
433 36°N 52°W), corresponding to the Sargasso Sea and through the Gulf Stream, pH only fell below  
434 7.7 at depths exceeding 450 m, although the upper water column was warmer, with average  
435 temperatures being 20°C. There was a third water mass type, typical of colder fresher shelf  
436 waters, at station 32 and in an intrusion off the Grand Banks at stations 17 and 19. This water  
437 was typified by a temperature and salinity anomaly with average temperatures falling below 5°C  
438 from 25-100 m and a salinity signature < 33, contrasting significantly with the surface salinities  
439 of the northern portion (~34) and southern portion (~36) of the Atlantic transect. As a  
440 consequence, these stations contained water of the lowest pH, with surface waters reaching 7.7 at  
441 depths shallower than 200 m. Based on previous knowledge of the vertical distributions of the  
442 thecosomes used in this study, only the species *Clio pyramidata* would ever experience a pH  
443 below 7.7 in this overall Atlantic study region and none of the thecosomes studied would  
444 experience 800  $\mu\text{atm}$  pCO<sub>2</sub> or under-saturation within their vertical range (Table 1).

### 445 3.3 Carbonate Chemistry of Experiments

446 Bubbling with CO<sub>2</sub> levels of ~380 and ~800 ppm resulted in a distinct separation of carbonate  
447 chemistry between treatments during the experiments in both oceans (Table 3). Due to pre-  
448 existing differences in the carbonate chemistry of the seawater collected in each ocean, TA  
449 concentrations were different between the two basin treatments. In the Atlantic the DIC of the  
450 ambient CO<sub>2</sub> treatments ranged from 2030-2090  $\mu\text{mol kg}^{-1}$  and the high CO<sub>2</sub> treatments from  
451 2140-2220  $\mu\text{mol kg}^{-1}$ , with an average difference between treatments of similar temperature and  
452 salinity of 132  $\mu\text{mol kg}^{-1}$ . Surface TA in the region decreased from ~2370  $\mu\text{mol kg}^{-1}$  in the  
453 southern part of the transect to 2300  $\mu\text{mol kg}^{-1}$  in the northern latitudes. In the Pacific the DIC of  
454 the ambient CO<sub>2</sub> treatment ranged from 1930-2020  $\mu\text{mol kg}^{-1}$  and the high CO<sub>2</sub> treatment from  
455 2030-2110  $\mu\text{mol kg}^{-1}$ , with an average difference of 90.7  $\mu\text{mol kg}^{-1}$  between the treatments.  
456 Surface TA in this basin was 2150  $\mu\text{mol kg}^{-1}$  in the most northern collection and had decreased  
457 to 2200  $\mu\text{mol kg}^{-1}$  by the transect mid-point.

458 Calculations of pCO<sub>2</sub> based on these measurements of DIC and TA suggested that target  
459 pCO<sub>2</sub> levels were generally attained and were consistent between the two cruises, with the  
460 exception of the LC/LO treatment in the Atlantic. In this case, there was a substantial deviation  
461 from the intended pCO<sub>2</sub>, suggesting values ranging from 99-139  $\mu\text{atm}$  in contrast to a range of





462 311-391  $\mu\text{atm}$  for the LC/HO in the Atlantic and 283-409  $\mu\text{atm}$  for LC/HO and 295-397  $\mu\text{atm}$  in  
463 the LC/LO in the Pacific. Evidently, this indicates improper mixing of the gas concentration in  
464 the Atlantic LC/LO gas cylinder by the manufacturer. The calculations for the high  $\text{CO}_2$   
465 treatments were more consistent between cruises, with the HC/HO being 585-868  $\mu\text{atm}$  for  $\text{pCO}_2$   
466 and the HC/LO being 755-783 in the Atlantic, while in the Pacific the HC/HO treatment was  
467 between 520-740  $\mu\text{atm}$  and the HC/LO 546-766  $\mu\text{atm}$ . The variability in calculated  $\text{pCO}_2$  values  
468 likely represents variations in bubbling time, temperature, and the degree to which the water  
469 reached saturation relative to the gas mixtures. The variability within each distinct treatment may  
470 also reflect, to some degree, what pteropods may experience under that particular mean  
471 condition, i.e. low vs. high  $\text{CO}_2$ .

472 As a consequence of the natural differences in seawater carbonate chemistry, in particular  
473 the TA differences between two ocean basins, there were inherent differences in the aragonite  
474 saturation state between the Pacific and Atlantic treatments (Table 3). In the Atlantic  $\Omega_{\text{Ar}}$  of the  
475 ambient  $\text{CO}_2$  treatment ranged from 2.4-3.5, except for the LC/LO treatment ( $\Omega_{\text{Ar}}$  4.0-5.5), which  
476 was bubbled with an incorrect gas mixtures as discussed above. Comparatively, in the Pacific the  
477 ambient  $\text{CO}_2$  condition had a lower range of  $\Omega_{\text{Ar}}$  (2.2-2.4) for both the LC/HO and the LC/LO  
478 treatments. The experimental conditions of the high  $\text{CO}_2$  treatments in the Atlantic only  
479 approached under-saturation in the middle part of the transect ( $\Omega_{\text{Ar}} = 1.2$  at mid-latitudes; Table  
480 3), where cold northern waters of low salinity were encountered and  $\Omega_{\text{Ar}}$  had a range of 1.5-2.0  
481 for the rest of the transect in the Atlantic. The values of  $\Omega_{\text{Ar}}$  were lower overall in the Pacific,  
482 although the high  $\text{CO}_2$  treatments also never reached under-saturation ( $\Omega_{\text{Ar}}$  1.3-1.8). The  
483 manipulation of carbonate chemistry in general successfully created two distinct ranges for both  
484  $\text{pCO}_2$  and aragonite saturation state ( $\Omega_{\text{Ar}}$ ) in this study.

485 It is important to acknowledge that the production of  $\text{CO}_2$  via respiration of the organisms  
486 within the chambers would modify the carbonate chemistry of the treatments over the duration of  
487 the experiments. Based on the average respiration rate, and using a respiratory quotient of 0.8  
488 (Mayzaud, 1976), we estimate an average DIC production of  $\sim 18.0 \mu\text{mol kg}^{-1}$  by the end of an  
489 experiment. Applying such a change to the experimental conditions in the northeast Pacific,  
490 where seawater is more sensitive to changes in DIC due to a lower buffering capacity compared  
491 to the Atlantic (i.e., a worst case scenario),  $\Omega_{\text{Ar}}$  would only change by  $<0.1$  in both the LC and  
492 HC experimental chambers over the course of the respiration experiments. Although this is an





493 appreciable effect, we nonetheless retain a wide separation between the ambient and high CO<sub>2</sub>  
494 treatments and in no cases would the treatments reach under-saturation as a consequence of this  
495 biological activity. As such, for simplicity the results reported in Table 3 do not include this  
496 variability.

### 497 **3.4 Oxygen Consumption Rate**

#### 498 **3.4.1 Effect of CO<sub>2</sub>**

499 Varying availability and abundances of the different thecosome pteropod species in the net  
500 samples precluded all species being exposed to the full factorial design but individuals of all  
501 species were tested under the low CO<sub>2</sub>, high oxygen (LC/HO) and high carbon, high oxygen  
502 (HC/HO) treatments (Fig. 3, Table 4). To explore differences in metabolism attributable to a  
503 response to CO<sub>2</sub>, the log transformed wet mass was used in a GLM as a covariate comparing the  
504 log transformed oxygen consumption (response variable) under low and high CO<sub>2</sub> conditions;  
505 each population within a species that was sampled in both basins or run at multiple experimental  
506 temperatures, was examined separately. There was no significant effect of CO<sub>2</sub> for any species in  
507 either basin.

#### 508 **3.4.2 Effect of basin**

509 Following this assessment, we were interested in determining whether there were  
510 between basin differences in metabolic rate. As such we ran a GLM using log transformed  
511 metabolic rates for the three species that were found in both basins, normalized to 15 °C to  
512 account for differences in experimental temperature by applying a standard temperature  
513 coefficient. With the log-transformed wet mass as a covariate, we tested for an effect of basin,  
514 CO<sub>2</sub> and an interactive term. *Clio pyramidata* had a similar metabolic rate between basins. In  
515 contrast, *Cavolinia inflexa* ( $F_{1,20}=10.358$ ,  $p=0.004$ ) and *Styliola subula* ( $F_{1,23}=11.817$ ,  $p=0.002$ )  
516 both had a significantly lower metabolic rate in the Pacific, although no interactive effect of CO<sub>2</sub>.

#### 517 **3.4.2 Effect of O<sub>2</sub>**

518 For the species where enough individuals were collected to provide experimental  
519 replicates to explore the interactive effects of CO<sub>2</sub> and O<sub>2</sub> we also ran a species and basin  
520 specific GLM exploring the effect of treatment (Fig. 3, Table 5). *Clio pyramidata*, the only  
521 species we were able to test in both basins showed no significant effect of high CO<sub>2</sub>, low O<sub>2</sub> or  
522 the interactive treatment in either basin. In the Pacific, *L. helicina* and *C. inflexa* similarly  
523 showed no significant change in metabolic rate as a consequence of any of the treatments. In



524 contrast, in the Atlantic, there was a significant effect of treatment for *L. retroversa* and a  
525 Bonferroni post-hoc analysis comparing the treatments found that the high CO<sub>2</sub>, low O<sub>2</sub> (HC/LO)  
526 treatment was significantly lower than all other treatments (Fig. 4A;  $F_{3,38}=17.836$ ,  $p<0.001$ ; a  
527 ~60% reduction in the average mass specific metabolic rate in comparison with the LC/HO  
528 treatment; Table 4). *Cuvierina atlantica* was tested at both 15 and 20 °C in the Atlantic, so to  
529 make comparisons among these experiments a temperature coefficient was applied and rates  
530 were normalized to 15 °C, after which no significant effect of any treatment was found for this  
531 species.

532

#### 533 4. Discussion

534 This study reveals that short term exposure to low O<sub>2</sub> and high CO<sub>2</sub>, similar to what would be  
535 experienced by individuals in the Pacific during diel vertical migration, does not influence the  
536 oxygen consumption of most of the thecosome pteropod species examined from either the  
537 Atlantic or Pacific. The only species which had a significant change in respiration in response to  
538 any of the treatments was *Limacina retroversa* from the Atlantic, which responded to the  
539 combined effect of low O<sub>2</sub> and high CO<sub>2</sub> with a reduction in oxygen consumption rate.

#### 540 4.1 Experimental Design

541 A factor that should be considered when interpreting our results is the dynamic hydrographic  
542 conditions that the animals experience naturally between and within the ocean basins.  
543 Thecosomes of multiple species were found at a range of temperatures, salinities and carbonate  
544 chemistries, meaning that they experienced a range of pH and aragonite saturation states in their  
545 natural habitat. When comparing animals from multiple locations, we chose to use local water in  
546 order to replicate these natural conditions and to manipulate exclusively the CO<sub>2</sub> concentration,  
547 as this is the factor that is changing due to anthropogenic activity. This approach, however, does  
548 not control for the other parameters of the carbonate chemistry system, which will vary between  
549 regions. Despite this fact, there was a clean distinction between treatments, notably in terms of  
550 aragonite saturation state as well as CO<sub>2</sub> concentration, that provides insight into the effect of  
551 moderate short duration exposure to CO<sub>2</sub>.

552 It is also important to note that the individuals of *L. helicina* from the Pacific experiments  
553 did occasionally have very high mortality during the period prior to experimentation (>80% at  
554 transit station T2 and T5, decreasing substantially to the northwest and along the main Pacific



555 transect). These individuals, which are polar/sub-polar organisms and are typically found  
556 between -2 to 10 °C (Lalli and Gilmer, 1989), were collected from water that was likely near the  
557 upper limit of their optimal temperatures. Animals collected from these sites that were used in  
558 subsequent respiration experiments may therefore have been taken from an already stressed  
559 population of individuals and should be recognized as such.

#### 560 4.2 Carbon Dioxide Effect

561 Hydrographic profiles collected in the Pacific coincident to sampling of thecosomes, indicate  
562 that organisms in the northern portion of the study region would experience conditions of high  
563 CO<sub>2</sub> and low O<sub>2</sub> in the upper ~450 m of their distribution (Chu et al., in review), unlike in the  
564 Atlantic. Despite these environmental differences, we found no significant effect of increasing  
565 CO<sub>2</sub> alone on the respiration rates of any of the species from either ocean basin. These results  
566 increase the published evidence that short term (6-18 h) exposure to enhanced CO<sub>2</sub> without  
567 synergistic stressors has no significant effect on the metabolic rate of many species of thecosome  
568 pteropods. Thus far, there are only two species that have been documented to show a change in  
569 metabolism based on exposure to manipulated CO<sub>2</sub> alone: *Limacina antarctica* (789-1000 µatm,  
570 24 h: Seibel et al., 2012) and *Diacria quadridentata* (1000 µatm, 6-18 h: Maas et al., 2012a). The  
571 metabolic rates of all other species yet studied, including *Hyalocylis striata*, *Clio pyramidata*,  
572 *Diacavolinia longirostris*, *Creseis virgula* (6-18 h: Maas et al., 2012a), and *Limacina helicina*  
573 (24 h: Comeau et al., 2010a), were not significantly affected by short term exposure to high CO<sub>2</sub>,  
574 although the latter species showed an increase in metabolic rate when high CO<sub>2</sub> was combined  
575 with high temperatures. Our results, which increase the geographic coverage for *L. helicina* and  
576 *C. pyramidata* and provide the first data about the species *C. pacifica*, *C. atlantica*, *L. retroversa*,  
577 *D. trispinosa*, *C. inflexa* and *S. subula*, corroborate these earlier findings.

578 One interpretation of these results is that physiological responses may have occurred, but  
579 involved the reallocation of resources to different tissues or metabolic pathways; this  
580 redistribution could serve to maintain the thecosome total energy budget, and subsequently  
581 would not significantly change the metabolic rate of the individuals. A transcriptomic study done  
582 with individuals of *Clio pyramidata* as a companion project to the present work in fact suggested  
583 that expression of some genes was influenced by CO<sub>2</sub> exposure even though metabolic rate is not  
584 (Maas et al., 2015), perhaps suggesting some re-allocation among energetic demands. If this is  
585 the case it indicates that, to some degree, the short-term exposure to high CO<sub>2</sub> concentration is



586 within the physiological tolerance of the tested species. Alternative hypotheses are that the  
587 duration of exposure was too short or the severity of the CO<sub>2</sub> treatment too minimal to elicit a  
588 measurable response. It is possible, for example, that some processes, like biomineralization,  
589 may be influenced by high CO<sub>2</sub>, but only after a longer exposure duration. Finally, it may be that  
590 changes in respiration rate were subtle, requiring a much greater sample size to identify in light  
591 of biological variability, but exploration of this hypothesis would require a dedicated experiment  
592 to collect more individuals and likely a smaller number of species.

593 This possible tolerance to short term CO<sub>2</sub> exposure may be due to the fact that within  
594 their distribution or diel migrational range there are conditions, or perhaps seasons, where the  
595 natural hydrography causes many species of thecosome to experience conditions of high  
596 CO<sub>2</sub>/low pH, and the species are therefore adapted to this range of exposure. The Arctic species  
597 *L. helicina* and subarctic species *L. retroversa*, for instance, are thought to inhabit waters which  
598 have been shown to reach a concentration of > 950 μatm CO<sub>2</sub> and to be undersaturated with  
599 respect to aragonite during the winter season in Kongsfjord, Svalbard (Lischka and Riebesell,  
600 2012). These conditions are pervasive throughout the upper water column, meaning that *L.*  
601 *helicina* and *L. retroversa*, which are not strong diel migrators, would experience seasonal under-  
602 saturation in these polar regions. The more temperate and tropical open ocean thecosomes,  
603 including *C. pyramidata*, *C. inflexa* and *S. subula* are all currently believed to be circumglobal  
604 and most, to varying degrees, diel migratory (Table 1; Bé and Gilmer, 1977; van der Spoel,  
605 1967). Populations are therefore likely to encounter high CO<sub>2</sub> in sub-surface waters in regions  
606 associated with OMZs, including much of the North Pacific and off the coast of Northern Africa.  
607 The ability to cope with high CO<sub>2</sub> for short durations may have been selected for over time as a  
608 natural consequence of the types of unavoidable environmental variability experienced by these  
609 planktonic populations.

#### 610 4.3 Low O<sub>2</sub> and Combined Effects

611 In the Pacific Ocean, none of the species for which we had enough individuals to perform the  
612 low O<sub>2</sub> study (*L. helicina*, *C. pyramidata*, and *C. inflexa*) had a significant change in metabolic  
613 rate under low (10%) O<sub>2</sub>, even when combined with enhanced CO<sub>2</sub>. These results indicate that  
614 the O<sub>2</sub> levels were above the concentration below which these species can no longer sustain their  
615 routine metabolic activity (Pcrit; Hochachka and Somero, 2002) and that any changes in  
616 physiology associated with the treatments required no increased energetic expenditure or



617 metabolic reduction. As subsurface waters throughout the cruise were frequently below 10% O<sub>2</sub>  
618 (< ~130 μmol kg<sup>-1</sup>), this indicates that these species may be naturally adapted to coping with low  
619 O<sub>2</sub> conditions.

620 In the Atlantic, examination of the effects of low O<sub>2</sub> is confounded by an unfortunate and  
621 accidentally low level of CO<sub>2</sub> (~130 μatm) in the LC/HO treatment (Table 3). Tests of the effect  
622 of high CO<sub>2</sub> (HC/HO) and the interactive (HC/LO) treatments nonetheless remain valid, and for  
623 *L. retroversa*, exposure to HC/LO caused a large and significant reduction in metabolic rate.  
624 Suppression in metabolic rate is a common tactic for surviving unfavorable conditions (Guppy  
625 and Withers, 1999; Seibel, 2011). Although metabolic depression is generally survivable in the  
626 short term, over longer time scales there are often implications for growth, reproduction and  
627 survival (reviewed in: Pörtner, 2010; Seibel, 2011). In the Atlantic, our measured in situ O<sub>2</sub>  
628 levels were never below 15% (~200 μmol kg<sup>-1</sup>). In contrast with the other species studied, which  
629 in at least some portions of their geographic range are occasionally found in association with  
630 subsurface low O<sub>2</sub> combined with hypercapnia, *L. retroversa* lives exclusively in the sub-polar  
631 North Atlantic Ocean and the Southern Circumpolar Current. As such this is the only species in  
632 this study in which no population is likely to experience conditions of low O<sub>2</sub> and high CO<sub>2</sub>  
633 together naturally anywhere in its distribution. Its inability to maintain metabolic rate during this  
634 interactive exposure may be a short-term metabolic response to environmental conditions that are  
635 unsustainable over longer time periods. As a consequence of the very low CO<sub>2</sub> in the LC/LO  
636 treatment, it is impossible to determine whether the metabolic suppression for *L. retroversa* in  
637 the HC/LO was in response to reduced O<sub>2</sub> availability alone or to the interactive effect of low O<sub>2</sub>  
638 with high CO<sub>2</sub>. In the LC/LO treatment any change in respiration due to low O<sub>2</sub> could have been  
639 masked by a change in the energy budget as a response to the low (equivalent to pre-industrial  
640 atmospheric conditions) levels of CO<sub>2</sub>. The results suggest that further work in the Atlantic is  
641 warranted to disentangle these stressors and to determine whether the observed change in  
642 metabolic rate was solely a consequence of O<sub>2</sub> availability or truly a synergistic effect.

643 Interestingly, although the temperature coefficients were not species-specific and may  
644 not, therefore, perfectly normalize the dataset, one trend revealed by their use was a significant  
645 difference in the normalized metabolic rates between species such as *S. subula* and *C. inflexa*  
646 from the Atlantic and Pacific Oceans. The comparatively lower metabolic rates from the Pacific  
647 may be a real response to the lower availability of O<sub>2</sub> for aerobic metabolism. Having a slower



648 routine rate of O<sub>2</sub> consumption may be the result of a more efficient respiratory mechanism or an  
649 adaptation for coping with occasional exposures to the relatively high CO<sub>2</sub> and low O<sub>2</sub> conditions  
650 found in the northeast Pacific Ocean.

651

## 652 5. Conclusions

653 Thecosomes pteropods are thought to be some of the most sensitive of the oceanic zooplankton  
654 species to acidification. The responses we documented in the face of short-term CO<sub>2</sub> exposure  
655 and low O<sub>2</sub> reveal interesting patterns about basin scale differences in sensitivity, possibly  
656 associated with adaptation to local environmental conditions. Importantly, our results indicate  
657 that short-term exposure to high CO<sub>2</sub> does not have an effect on the respiration rate of multiple  
658 species of temperate and sub-polar thecosome species from both the North Atlantic and Pacific  
659 Oceans, irrespective of recent likely environmental exposure. The lack of effect of CO<sub>2</sub> as a  
660 single-stressor on metabolic rate in adult organisms of various species has been seen in a number  
661 of studies (reviewed in: Dupont et al., 2010; Kroeker et al., 2013), although there are many other  
662 metrics that have been shown to be more consistently affected. As such, thecosomes may have  
663 physiological coping mechanisms that allow them to maintain their energy budget for short  
664 periods of time in the face of high CO<sub>2</sub> via the re-allocation of their energetic resources. Over  
665 longer time periods, however, this could reduce their scope for growth and reproduction,  
666 negatively impacting the fitness of the population as has been demonstrated with other marine  
667 calcifiers (i.e.: Dupont et al., 2013; Melzner et al., 2013; Stumpp et al., 2011). Testing these  
668 hypotheses remains difficult as thecosomes are hard to maintain in captivity and there are no  
669 published studies of individuals kept fed and exposed to CO<sub>2</sub> in laboratory conditions for long  
670 durations (reviewed in: Howes et al., 2014; Thabet et al., 2015). Keeping individuals well fed is  
671 a critical factor since high food availability has been suggested to modulate the effect of high  
672 CO<sub>2</sub> exposure in both thecosomes (Seibel et al., 2012) and in other calcifying species (Thomsen  
673 et al., 2013). Comparative short-term studies of wild caught animals such as the present  
674 experiments, therefore, currently give us the best insight into the sensitivity of these open-ocean  
675 populations, and the ability to predict how they will respond to the expected changes in the ocean  
676 environment.

677 These findings also draw attention to the consequences of the high degree of vertical  
678 variability in the open ocean environment, with animals in the Pacific found migrating between



679 deep waters, undersaturated with respect to aragonite, and the surface (Lawson, unpublished  
680 data; Chu et al., in review; Maas et al., 2012a). Recent studies in the California Current system  
681 indicate that thecosome shells show signs of in situ dissolution when associated with water  
682 masses that are undersaturated with respect to aragonite (Bednaršek et al., 2014b; Bednarsek and  
683 Ohman, 2015). Although our short duration experiments do not directly address the effect of  
684 longer-term exposure to high CO<sub>2</sub>, it does remind us that as open ocean environments respond to  
685 anthropogenic change there may be vertical refugia from OA stress as well as regions where  
686 animals may already experience high CO<sub>2</sub>. As surface waters acidify, the ability to endure short-  
687 duration exposure and to migrate in both the Atlantic and Pacific populations may provide  
688 mechanisms for mitigating detrimental effects of acidification exposure. The potential  
689 compression of vertical habitat associated with the shoaling of the aragonite compensation depth,  
690 however, may have implications for predator/prey interactions, carbon pumping and other  
691 ecosystem functions (Bednarsek and Ohman, 2015; Seibel, 2011). Furthermore, it is clear that  
692 thecosome shells are highly sensitive to dissolution (Comeau et al., 2012; Lischka and Riebesell,  
693 2012; Manno et al., 2012) and there could be fitness and ecological consequences of dissolution  
694 in regions with vertical variation in carbonate chemistry.

695 Finally, as concerns about increasing CO<sub>2</sub> drive further explorations of comparative  
696 organismal physiology in the marine system, it is important to recognize that often the exposure  
697 of animals to increased CO<sub>2</sub> will occur in concert with expanding regions of low O<sub>2</sub>. This has  
698 been explored in the coastal environment where the interaction of acidification with  
699 eutrophication and associated low O<sub>2</sub> is comparatively well studied (Cai et al., 2011; Melzner et  
700 al., 2013), and in theoretical frameworks (Gruber, 2011; Pörtner, 2010; Sokolova, 2013).  
701 Experiments in the open ocean environment, however, are only beginning to be conducted and  
702 their implications explored. This study suggests that to make accurate predictions about how  
703 populations will respond to climate change and adequately understand the factors affecting  
704 organismal response, further investigations of the interactive effects of low O<sub>2</sub> and hypercapnia  
705 should consider natural environmental variability, population biogeography and phylogenetic  
706 sensitivity.





707 **Data availability**

708 Cruise data for the project is available via BCO-DMO under the project “Horizontal and Vertical  
709 Distribution of Thecosome Pteropods in Relation to Carbonate Chemistry in the Northwest  
710 Atlantic and Northeast Pacific” (<http://www.bco-dmo.org/project/2154>). The raw data for the  
711 respiration experiments are included in this deposition (DOI: 10.1575/1912/6421).

712

713 **Author contributions**

714 A. Maas and G. Lawson designed the experiments. All co-authors participated in oceanographic  
715 cruises and collection of samples. A. Maas conducted all of the experiments and statistical  
716 analyses. Z.A. Wang advised on the design of the carbonate chemistry analysis and provided the  
717 measurements of both the hydrographic and experimental conditions. A. Maas prepared the  
718 manuscript with contributions from all co-authors.

719

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994 Table 1: Environmental preferences and diel vertical migratory patterns for the species used in  
 995 this study based on previously published data (Bé and Gilmer, 1977; Lalli and Gilmer, 1989).  
 996 Data includes published full ranges at which organisms have been found, as well as previous  
 997 authors' estimates of the preferred (optimal) ranges of each species when available. Note that  
 998 these are based on relatively sparse observations of broadly distributed species, many of which  
 999 may be cryptic congeners, and thus should be treated as estimates.

Species	(optimal) temp (°C)	(optimal), depth (m)	migrator?
<i>Cuvierina atlantica</i>	18 to 26	100-250	possible
<i>Cuvierina pacifica</i>	Only recently established as a separate species, the habits are assumed to be similar to the Atlantic congener.		
<i>Cavolinia inflexa</i>	16 to 28	0-250	no
<i>Clio pyramidata</i>	7 to 27	(0-500), <1500	yes
<i>Limacina helicina</i>	(-2 to 10)	(50-100), <300	possible
<i>Limacina retroversa</i>	(7 to 12)	(20-30), < 150	possible
<i>Styliola subula</i>	(18 to 22)	50-300	yes
<i>Diacria trispinosa</i>	9 to 28	30-200	no

1000



1001 Table 2: The hydrography and location for each station where animals for experiments were  
1002 collected. At stations along the main transect the depth (m) at which O<sub>2</sub> decreased below 130  
1003 μmol O<sub>2</sub> kg<sup>-1</sup> (~10%) and the average temperature from 25-100 m (°C) were derived from CTD  
1004 casts. At a few stations (denoted via <sup>a</sup>) in the Atlantic there was warm water at the surface and  
1005 cold water below. The only species in this region, *Limacina retroversa*, has an optimum  
1006 temperature between 7-12 °C (Bigelow, 1924) and was generally found above 50 m (Lawson,  
1007 unpublished data). At these sites the average temperature is reported first for between 25-100 m  
1008 and then also for 25-50 m to reflect the conditions likely experienced by the pteropods. pCO<sub>2</sub> and  
1009 Ω<sub>Ar</sub> were calculated from measured pH and DIC bottle samples. We interpolated linearly the  
1010 depths (m) at which the pH decreased below 7.7, pCO<sub>2</sub> reached 800 μatm, and aragonite  
1011 saturation (Ω<sub>Ar</sub>) reached 1 from the discrete measurements at adjacent depths. At stations  
1012 conducted while in transit to the main study transects (denoted by prefix T) the average  
1013 temperature from 25-100 m (°C) was documented from XBT casts. At these transit stations no  
1014 O<sub>2</sub> or carbonate chemistry data were available (noted with a dash). The species caught at each  
1015 station and used in this study are demarcated with a star (\*).  
1016





Year	Station	Latitude (°N)	Longitude (°W)	average temp 25-100 m	depth of 130 μmol O <sub>2</sub> kg <sup>-1</sup>	depth of pH 7.7	depth of 800 μatm	depth of Ω <sub>Ar</sub> = 1	<i>C. atlantica</i>	<i>C. pacifica</i>	<i>C. inflexa</i>	<i>C. pyramidata</i>	<i>L. helicina</i>	<i>L. retroversa</i>	<i>S. subula</i>	<i>D. trispinosa</i>
2011 Atlantic	32	49.1	-44.3	5.3, 9.0	NA	74.1	NA	NA	*					*	*	
	31	50.0	-42.0	14	NA	385.4	NA	NA							*	
	30	49.6	-41.9	14.1	NA	452.8	NA	NA								
	26	47.5	-42.0	13.3	NA	644.9	NA	NA		*	*	*				
	24	46.5	-42.0	14.5	NA	453.9	NA	NA		*	*	*				
	21	44.9	-42.0	16.5	NA	501.1	NA	NA							*	
	19	44.0	-44.9	4.9, 11.2	NA	181.0	NA	NA						*		
	17	43.0	-47.8	1.8, 8.1	NA	143.1	NA	NA						*		
	13	40.9	-52.0	20.7	NA	756.7	NA	NA		*	*	*			*	*
	10	47.5	-52.0	19.4	NA	466.9	NA	NA		*	*	*			*	*
2012 Pacific	8	38.5	-52.0	22.8	NA	805.7	NA	NA	*	*	*					
	3	36.0	-52.0	21.4	NA	937.7	NA	NA	*							
	T2	45.6	-128.5	-	-	-	-	-					*			
	T3	46.6	-133.5	-	-	-	-	-					*			
	T4	47.7	-138.5	6.4	-	-	-	-		*						
	T5	45.7	-129.8	10.0	-	-	-	-		*			*			
	T6	46.6	-134.9	9.5	-	-	-	-		*			*			
	T7	47.6	-140.2	8.6	-	-	-	-		*			*			
	3	49.0	-148.2	6.2	209	128.9	193.7	168.5								
	6	47.5	-145.6	7.1	235	108.3	199.2	159.1		*	*	*	*			
7	47.0	-144.6	7.8	256	131.0	214.0	185.1		*	*	*	*				
15	43.1	-138.1	10.9	363	199.5	368.2	334.8		*	*	*	*				
18	41.5	-135.8	13.7	340	147.3	331.7	380.6		*	*	*	*				
21	39.9	-135.0	12.7	348	162.0	332.2	302.8		*	*	*	*				
24	38.6	-135.0	14.7	402	222.8	411.8	372.7		*	*	*	*				
30	35.6	-135.0	16.2	349	200.7	437.8	425.1		*	*	*	*				
32	34.4	-135.1	16.5	348	202.9	439.2	432.0		*	*	*	*			*	
34	33.6	-135.0	17.4	368	233.3	370.1	352.4		*	*	*	*			*	
T9	33.7	-133.6	17.0	-	-	-	-		*	*	*	*				
T10	33.8	-133.2	15.9	-	-	-	-		*	*	*	*				



Table 3: Carbonate chemistry during manipulation experiments. The manipulation experiments were conducted at multiple temperatures (T.) and salinities (S.) based on the conditions the organisms were caught in. As described in more detail in the text, DIC measurements were made of water drawn from the control chambers while TA was measured for batches of experimental water (denoted as xpt. TA). In situ TA (i.s. TA), based on nearby CTD bottle sampling at the surface, is also shown. At test stations, where bottle samples of in situ TA were unavailable, underway pCO<sub>2</sub> values and the LC/HO DIC were used to calculate in situ TA (denoted with \*). In some instances, measurements of experimental TA differed by >20 μmol kg<sup>-1</sup> from nearby in situ measurements of surface TA. This difference greatly exceeds expected variability based on measurement uncertainty and spatial (geographic and vertical) offsets in the locations of experimental water collection relative to the nearest CTD cast; in these circumstances, the experimental TA was likely erroneous due to sampling errors (e.g., contamination). For completeness, and to aid in identification of erroneous experimental TA values, calculations of carbonate chemistry parameters, including aragonite saturation state ( $\Omega_{Ar}$ ) and pCO<sub>2</sub> were made based on DIC and both experimental TA and in situ TA. In further data analysis and interpretation, calculations based on experimental TA are given preference except those few instances where experimental TA differed from in situ by >20 μmol kg<sup>-1</sup> (bold denotes preferred calculations). Calculated saturation state and pCO<sub>2</sub> are reported as the average and standard deviation per batch of water. Note that the LC/LO gas tank in 2011 (in italics) appears to have been improperly mixed by the manufacturer as calculations suggested it contained a much lower CO<sub>2</sub> level than the intended 380 μatm; it should consequently be considered an entirely separate treatment from the 2011 LC/HO (were CO<sub>2</sub> levels were based on bubbling with an ambient air line).



Treatment	T. °C	S.	i.s. TA ( $\mu\text{mol kg}^{-1}$ )	xpt. TA ( $\mu\text{mol kg}^{-1}$ )	DIC ( $\mu\text{mol kg}^{-1}$ )	i.s. QAr	i.s. pCO <sub>2</sub> ( $\mu\text{atm}$ )	xpt. QAr	xpt. pCO <sub>2</sub> ( $\mu\text{atm}$ )	
2011 Atlantic	380 $\mu\text{atm CO}_2$ 21% O <sub>2</sub>	10	33	2300.3	2307.3	2094.4	2.3 ± 0.2	336.2 ± 37.7	2.4 ± 0.2	324.8 ± 35.8
		15	33	2300.3	2307.3	2066.5	2.6 ± 0.7	404.5 ± 172.7	2.7 ± 0.7	390.8 ± 164.5
		15	35	2296.4	2354.5	2066.4	2.5 ± 0.1	382.3 ± 20.4	3.1 ± 0.1	297.7 ± 14.3
	380 $\mu\text{atm CO}_2$ 10% O <sub>2</sub>	20	34	2353.4*	2345.8	2028.6	3.6 ± 0.2	302.8 ± 31.6	3.5 ± 0.2	311.6 ± 32.9
		20	34	2366.0	2367.2	2077.5	3.3 ± 0.1	363.1 ± 23.2	3.3 ± 0.1	361.4 ± 23.1
		10	33	2300.3	2307.3	1919.7	4.0	139.0	4.1	135.5
	800 $\mu\text{atm CO}_2$ 21% O <sub>2</sub>	15	33	2300.3	2307.3	1774.8	5.5 ± 0.6	101.2 ± 23.9	5.6 ± 0.6	99.0 ± 23.3
		15	35	2296.4	2354.5	1852.7	4.6	139.2	5.3	116.1
		10	33	2300.3	2307.3	2219.7	1.2 ± 0.2	779.9 ± 114.0	1.2 ± 0.2	742.4 ± 106.8
		15	33	2300.3	2307.3	2208.0	1.3	908.7	1.4	867.8
		15	35	2296.4	2354.5	2139.5	1.9	585.2	2.4	434.4
		20	34	2353.4*	2345.8	2176.9	2.1 ± 0.1	651.8 ± 23.4	2.1 ± 0.1	678.2 ± 24.8
	800 $\mu\text{atm CO}_2$ 10% O <sub>2</sub>	20	34	2366.0	2367.2	2212.7	1.9 ± 0.4	786.0 ± 196.0	1.9 ± 0.4	780.9 ± 194.2
		15	33	2300.3	2307.3	2186.2	1.5 ± 0.2	788.7 ± 157.6	1.5 ± 0.2	754.9 ± 148.3
		15	35	2296.4	2354.5	2179.6	1.5 ± 0.3	782.9 ± 164.6	2.0 ± 0.3	558.2 ± 103.9
10		32.1	2151.9*	2142.8	1934.8	2.2 ± 0.1	285.2 ± 21.4	2.3 ± 0.1	283.0 ± 21.2	
10		33.5	2208.0	2222.7	2001.9	2.4 ± 0.6	302.2 ± 100.9	2.4 ± 0.6	303.3 ± 101.4	
15		32.5	2182.6*	2095.7	1983.4	2.2 ± 0.0	388.1 ± 5.5	1.4 ± 0.0	646.7 ± 11.5	
2012 Pacific	380 $\mu\text{atm CO}_2$ 10% O <sub>2</sub>	15	33.5	2208.0	2222.7	2020.8	2.3 ± 0.2	407.7 ± 52.1	2.3 ± 0.2	409.1 ± 52.4
		10	32.5	2182.6*	2095.7	1973.9	2.3 ± 0.1	295.5 ± 20.0	1.4 ± 0.1	489.2 ± 41.2
		15	33.5	2208.0	2222.7	2017.5	2.3	3956.0	2.3	397.4
	800 $\mu\text{atm CO}_2$ 21% O <sub>2</sub>	10	32.1	2151.9*	2142.8	2026.3	1.4 ± 0.1	525.0 ± 35.0	1.4 ± 0.1	519.7 ± 34.5
		10	33.5	2208.0	2222.7	2120.6	1.3	628.2	1.3	631.2
		15	32.5	2182.6*	2095.7	2031.7	1.8 ± 0.1	527.6 ± 50.9	1.0 ± 0.1	952.4 ± 115.1
800 $\mu\text{atm CO}_2$ 10% O <sub>2</sub>	15	33.5	2208.0	2222.7	2112.2	1.4 ± 0.2	736.0 ± 96.0	1.4 ± 0.2	739.4 ± 96.6	
	10	32.5	2182.6*	2095.7	2066.5	1.4 ± 0.1	545.5 ± 65.1	0.8 ± 0.1	1056.0 ± 151.6	
	15	33.5	2208.0	2222.7	2118.3	1.4	762.4	1.4	766.0	



Table 4: The average wet mass (mass; g) and mass-specific oxygen consumption rate ( $MO_2$ ;  $\mu\text{mol O}_2 \text{ g}^{-1} \text{ h}^{-1}$ )  $\pm$  the standard error (SE) for each treatment (Treat.) and species. The number of individuals (N) per treatment are reported and the species are arranged by temperature (Temp; °C) as well as the year and basin of collection.

Year	Temp.	Species	Treat.	N	mass	$\pm$ SE	$MO_2$	$\pm$ SE	
2011 Atlantic	10	<i>Limacina retroversa</i>	LC/HO	12	.00281	0.00037	10.33	1.17	
			HC/HO	13	.00284	0.00031	10.10	0.56	
			LC/LO	9	.00274	0.00026	8.12	0.66	
		15	<i>Clio pyramidata</i>	HC/LO	9	.00377	0.00053	4.21	0.55
				LC/HO	10	.01944	0.00408	7.81	0.71
				HC/HO	8	.01410	0.00435	8.55	1.48
	20	<i>Cuvierina atlantica</i>	LC/LO	9	.02363	0.00867	6.63	1.21	
			HC/LO	8	.03945	0.00467	6.99	0.45	
			LC/HO	8	.04493	0.00264	5.05	0.63	
			LC/LO	10	.04636	0.00252	3.25	0.28	
			HC/LO	10	.05040	0.00219	4.29	0.37	
	2012 Pacific	10	<i>Diacria trispinosa</i>	LC/HO	8	.03718	0.00316	4.44	0.56
				HC/HO	10	.03589	0.0027	4.09	0.51
			<i>Cuvierina atlantica</i>	LC/HO	9	.01876	0.00396	4.31	0.85
HC/HO				9	.01683	0.00284	4.53	1.13	
15		<i>Cavolinia inflexa</i>	LC/HO	8	.00626	0.00104	14.30	1.48	
			HC/HO	4	.00508	0.00049	13.81	1.39	
		<i>Styliola subula</i>	LC/HO	10	.00400	0.00038	13.96	1.80	
			HC/HO	8	.00289	0.00035	15.95	0.87	
2012 Pacific	10	<i>Limacina helicina</i>	LC/HO	7	.00140	0.00026	5.26	1.17	
			HC/HO	8	.00149	0.00021	5.51	0.69	
			LC/LO	6	.00300	0.00058	4.91	0.69	
		15	<i>Clio pyramidata</i>	HC/LO	10	.00296	0.00038	7.18	1.45
				LC/HO	9	.02646	0.00258	5.43	0.45
				HC/HO	8	.02355	0.00369	4.39	0.60
	15	<i>Cuvierina pacifica</i>	LC/LO	14	.01459	0.00185	5.58	0.81	
			HC/LO	12	.01250	0.00245	5.72	1.14	
			LC/HO	4	.01829	0.00563	3.41	0.56	
			HC/HO	7	.02130	0.00636	3.53	0.57	
		<i>Cavolinia inflexa</i>	LC/HO	5	.01330	0.00062	3.53	0.44	
			HC/HO	8	.01556	0.00149	3.34	0.41	
			LC/LO	4	.01405	0.00185	2.41	0.33	
			HC/LO	2	.01855		3.98		
<i>Styliola subula</i>	LC/HO	6	.00360	0.00044	5.30	1.20			
	HC/HO	4	.00220	0.00029	7.73	2.14			
	LC/HO	4	.03020	0.0037	3.82	0.66			
	HC/HO	5	.02904	0.00329	3.21	0.27			



Table 5: Statistical results of the univariate general linear models (GLM) for each species were analyzed separately by year and are listed by the temperature of the experiment (Temp.; °C). For species studied at multiple temperatures (denoted by \*), the metabolic rates were adjusted to 15°C using a  $Q_{10} = 2$  to allow for direct comparison. The effect of the independent factors of CO<sub>2</sub> level (CO<sub>2</sub>), O<sub>2</sub> level (O<sub>2</sub>), their interactive effect (Int.) and the covariate of mass were analyzed in regards to the metabolic rate and reported as *p*-values for the Pacific (mean mass specific metabolic rate values found in Table 4). For the Atlantic, each treatment was tested as independent (Treat.) due to the accidentally low CO<sub>2</sub> condition in the LC/LO gas mixture.

Year	Temp.	Species	Effect on metabolic rate				
			CO <sub>2</sub>	O <sub>2</sub>	Int.	Treat.	Mass
2011 Atlantic	10	<i>Limacina retroversa</i>				<0.001	<0.001
	15	<i>Clio pyramidata</i>				0.295	<0.001
		<i>Cuvierina atlantica</i> *				0.174	<0.001
		<i>Diacria trispinosa</i>	.731				<0.001
		<i>Cavolinia inflexa</i>	.677				.008
		<i>Styliola subula</i>	.791				.040
2012 Pacific	10	<i>Limacina helicina</i>	.464	.323	.914		.007
	15	<i>Clio pyramidata</i> *	.255	.156	.726		.018
		<i>Cuvierina pacifica</i>	.709				<0.001
		<i>Cavolinia inflexa</i>	.309	.717	.219		.113
		<i>Styliola subula</i>	.763				.668



## Figure legends

**Figure 1: Cruise tracks and animal sampling.** Thecosomes were collected during the night at stations along the main survey transect (solid line) and at stations during transit (dashed line) during cruises to the northwest Atlantic in 2011 and northeast Pacific in 2012. The shapes correspond to the species caught at each station and used in this study. Blue (10 °C), grey (15 °C) and red (20 °C) boxes around the station numbers (#) correspond to the temperature that was representative of 25-100 m at each station (Table 2) and used in the experiments with animals from that station.

**Figure 2: Hydrography of sampling regions.** Hydrographic profiles of stations representative of the specific water mass types from the northern (P-T5, P-6, A-26), middle (P-18, A-19) and southern (P-32, A-8) portions of the Pacific (P) and Atlantic (A) study transects (station locations: Fig. 1). At station P-T5, the temperature profile (grey) was from an XBT cast because no CTDs were conducted during transits. For all stations along the main transects, left-hand plots show temperature (grey), salinity (black) and oxygen (black dotted) measured via sensors on the CTD and binned to 1 m depth intervals. Middle plots show TA (black) and DIC (grey) from discrete bottle samples (dots show depths of bottle samples). Right-hand plots show pCO<sub>2</sub> (black) and aragonite saturation state ( $\Omega$ ; grey) calculated based on TA and DIC measurements.

**Figure 3: Thecosome respirometry.** Mean metabolic rate and standard error ( $\mu\text{mol O}_2 \text{ g}^{-1} \text{ h}^{-1}$ ) of thecosomes exposed to low (i.e., ambient) CO<sub>2</sub> and normal levels of O<sub>2</sub> (light blue; LC/HO), high CO<sub>2</sub> and normal O<sub>2</sub> levels (dark blue; HC/HO), low CO<sub>2</sub> and low O<sub>2</sub> (light red; LC/LO), or high CO<sub>2</sub> and low O<sub>2</sub> (dark red; HC/LO). The species and temperature of the experiment are reported below the x-axis. Significance is reported based on a basin, species, and temperature specific GLM which tested for the effect of treatment on O<sub>2</sub> consumption with a Bonferroni post-hoc analysis. In the Atlantic analysis each treatment was tested independently, while in the Pacific CO<sub>2</sub> and O<sub>2</sub> were treated as factors. For each species and temperature, treatments are reported as non-significant (N.S.) or, in the case of significance, by letters that indicate which treatments are statistically similar (same letter) or different (different letter) at a p-value < 0.05.



Note that for *C. atlantica* the metabolic rates of individuals respired at 20° C were converted to 15°C using a temperature coefficient of 2 (see methods) for this GLM analysis.

**Figure 4:** Log transformed metabolic rates ( $\mu\text{mol O}_2 \text{ h}^{-1}$ ) for *L. retroversa* at 10 °C, not normalized to mass, plotted against the log transformed wet mass (mg) of individuals exposed to low CO<sub>2</sub> and normal levels of O<sub>2</sub> (black circles; LC/HO), high CO<sub>2</sub> and normal O<sub>2</sub> levels (dark grey diamonds; HC/HO), low CO<sub>2</sub> and low O<sub>2</sub> (white circles; LC/LO), or high CO<sub>2</sub> and low O<sub>2</sub> (light grey diamonds; HC/LO).









