

**This file contains:**

- a point-by-point response to the referees, which includes a description of the relevant changes;
- a marked-up manuscript version, i.e., the manuscript with Track Changes activated.

Dear editor,

We are pleased to present our revised version of the manuscript “The fate of fixed nitrogen in marine sediments with low organic loading: an in situ study”. This version incorporates the suggestions provided by the two anonymous reviewers, whom we would like to thank for their detailed analysis of our work. We think that, with their suggestions and criticism, this revised version of the paper has substantially improved. We have addressed the reviewers’ comments point-by-point and we presented our answers (in blue) below their original comments (in black).

Anonymous Referee #1

General Comments The authors present a study in which they quantified the fate of fixed nitrogen in sediments of a cold, oligotrophic system. The authors used  $^{15}\text{N}$  tracers and a combination of in situ incubations using a benthic lander and ex situ sediment core and slurry incubations. The authors are the first to simultaneously measure rates of denitrification, anammox, and DNRA in oligotrophic sediments. They accomplish this using in situ lander incubations, which are logistically difficult to perform, but may actually provide more accurate estimates of in situ rates than traditional core or slurry incubations. The authors found that denitrification dominated  $\text{N}_2$  production, but anammox bacteria were also active, accounting for 18-26% of  $\text{N}_2$  production. The authors also measured detectable DNRA and found that DNRA rates were highest, and comparable to denitrification rates, at the shallow coastal station. A sediment nitrogen budget was constructed and indicated that, despite the  $\text{N}_2$  production measured at the stations, the primary fate of sediment organic nitrogen in the summer is recycling and efflux as TDN back into the overlying water. Lastly, this study compared concentrations of ladderane lipids, a biomarker for anammox bacteria, to anammox rates and found no correlation between the two. These datasets are sparse in the literature, so this is an informative contribution to the scientific community studying anammox.

Overall, I think the authors addressed important questions related to sediment nitrogen cycling that will be of interest to many readers of this journal. The paper is very well written and organized clearly. I am comfortable with the conclusions and support publication of this manuscript with minor edits, as detailed below.

**We would like to thank the reviewer for their detailed and insightful analysis of our work. We really appreciate their efforts in improving the manuscript, and we are happy to read their acknowledgment of the manuscript merits.**

Specific Comments p.1, line 12 insert “the” before “global”

**Edit made.**

p.2, line 5 delete “to” before “ ~ 45%”

**Edit made.**

p.2, line 8 define the abbreviation “DNRA” the first time it’s used in the text body

**We have explained the acronym “DNRA”, as well as “N”, “GOB” and “TDN” at their first use in the text body, although these acronyms were already defined in the Abstract.**

p.2, line 13 insert “the” before “electron”

**Edit made.**

p.2, lines 23-24 It would be helpful if you mention briefly the link between Mn and anammox, since it is related to your hypotheses and your interpretation of your results.

**We have presented the possible link between high Mn concentrations and high anammox contribution/low denitrification contribution.**

p.2, line 28 define the abbreviation “GOB” the first time it’s used  
**Abbreviation explained.**

p.3, line 1 I suggest replacing “happen” with “occur”  
**Edit made.**

p. 3, line 7 suggested change: “. . .we hypothesize that we will measure low benthic N cycling rates. . .”  
**In our opinion it does not sound correct to use the future tense in this context. The past tense in English is often used to talk about hypotheses. For consistency, we have left the verb at the past tense here.**

p.3, line 9 change to “porewater,” (one word) to be consistent with the rest of the text  
**Edit made.**

p.4, lines 27-28 It would be helpful here if you could define what the average (or range of) water height(s) above the sediment surface was for the lander incubations. No need to list it for every incubation, just give the reader an idea of how much water volume was involved in these incubations.  
**We have now included the range of the incubated water volumes.**

p.6, line 28 Is the 75uM concentration for the sum of  $15\text{NH}_4^+$  +  $14\text{NO}_3^-$  or for each of the N species?  
**We have now specified that it refers to each of the N species.**

p.7, lines 18-20 For clarity, I suggest you present the r-IPT equations from Risgaard-Petersen et al. (2003) so that readers who are unfamiliar with them can understand how you get from  $p_{29\text{N}_2}$  and  $p_{30\text{N}_2}$  and  $r_a$  to  $p_{14}$ . This will also give you a chance to define  $p_{14}$  explicitly, and describe how it represents  $\text{N}_2$  produced without the  $15\text{N}$  addition, i.e., actual  $\text{N}_2$  production. Many unfamiliar with IPT think that the added  $15\text{NO}_3^-$  will stimulate denitrification and that those rates are included in your results, when in actuality the IPT approach allows one to separate  $p_{14}$  (actual) from total  $\text{N}_2$  production from  $15\text{N}$  and  $14\text{N}$  (potential).  
**The r-IPT equations from Risgaard-Petersen et al. (2003) have been added to the text in order to explain how, for example,  $p_{14}/lan$  was calculated. For conciseness, however, we have decided not to present the calculations also for  $p_{14}/wc$  and  $p_{14}/sl$  as the reader has now sufficient information to understand that they were calculated in the same fashion as for  $p_{14}/lan$  but from sediment core incubations (water phase and slurry phase, respectively).**

Eqn. 2 Somewhere here in the text describing eqn. 2 you should state clearly that  $p_{14}sl$  includes both water and sediment  $p_{14}$ .  
**We have added in the text above Eq. 2 that slurried phase means water plus sediment.**

p. 7, lines 25-26 I understand why you have to use the same  $F_{wc}$  measured in 2014 for the 2013 calculations. “You don’t have the sediment core incubations from 2013. I’m just not convinced that the  $F_{wc}$  values would be consistent from 2013 to 2014. Your rates (denitrification, anammox,  $\text{O}_2$ , TDN, etc.) as well as OPD show year-to-year variability, so it would not be surprising to me if the  $F_{wc}$  values were variable. Perhaps here (or elsewhere) you could defend this assumption in a bit more detail and discuss the potential implications for your calculated rates?”  
**We acknowledge the referee for raising an important point here. We cannot indeed exclude that  $F_{wc}$  in 2013 could have been slightly lower than those we measured in 2014 because the oxygen penetration depths were**

higher in 2013 than in 2014 (Table 1), suggesting that  $N_2$  production happened deeper in the sediment. A small underestimation of in situ  $N_2$  production in 2013 cannot be excluded. We have added this argument in the Results when we present our measured  $F_{wc}$  values.

p.8, line 1 Since you use the term “ra” here, and it’s a widely used term to describe the contribution of anammox to total  $N_2$  production, I suggest you use it throughout the rest of the text and tables/figures.

**Edit made throughout the text and in Fig. 6.**

p.8, lines 15-16 I have read this section multiple times, and I still am unsure what this sentence means. I think you’re saying that you have to use the  $F_{wc}$  calculated from the  $p_{14}$  values for this  $NH_4^+$  calculation. If  $p_{15}NH_4^+$  was not detected in just one of the incubations (GOB1-3), why couldn’t you use the  $p_{15}NH_4^+$  fluxes from all of the other incubations? At least they’re still related to the parameter you’re working with ( $NH_4^+$ ). How will using the  $F_{wc}$  derived from the  $p_{14}$  values affect the calculated  $NH_4^+$  rates?

**As stated in the Results, DNRA could not be determined in sediment core incubations because  $p^{15}NH_4^+$  was not detectable in our time courses at three (GOB1, GOB2, GOB3) out of four stations. In sediment core incubations,  $p^{15}NH_4^+$  could only be detected at station RA2. Using the  $F_{wc}$  from  $p^{15}NH_4^+$  at RA2 for all the other stations would result in much higher, and probably unrealistic, DNRA rates than those we estimated now by using  $F_{wc}$  from whole core’s  $p_{14}$ . However, we agree with the referee that using  $F_{wc}$  from  $p_{14}$  may not be 100% representative of the in situ, actual rate. Thus, in the Discussion, we have further highlighted the fact that our up-scaled DNRA rate may represent an underestimation of the actual DNRA rates.**

p. 10, line 13 Insert “from” after “ranged”

**Edit made.**

p. 10, line 24 The sentence “Between the four stations. . .>GOB3.” reads awkwardly. I suggest changing to “Downcore  $NH_4^+$  concentrations were greatest in RA2, followed by GOB2 >= GOB1>GOB3.”

**Edit made.**

p.11, line 5 Replace “GOB3” with “GOB2”

**Thank you for this important correction. We have now replaced GOB3 with GOB2.**

p. 11, lines 13-15 The sentence “The facts that . . .supported by DON.” is awkwardly worded, making it difficult to understand its meaning.

**We agree with the referee and have now rephrased this all sentence.**

p.11, lines 17-18 reword to “. . .with  $F_{wc}$  values of 0.26, 0.23. . .”

**Edit made.**

p.11, line 22 At the end of this paragraph, I suggest you present the ra values from the slurry incubations (also include in Table 2), since that’s really the main point of doing the slurries. It’s fine to keep the data in Figure 6 since it’s relevant to the discussion of the other NRPs. But I think the data should be first introduced here to make it clear where that data come from.

**We have now added the ra values at the end of the paragraph and in Table 2.**

p.12, lines 19-22 The sentence “The  $^{15}N$  isotope pairing technique. . .is low..” is awkwardly worded, making it difficult to follow.

**We have now split this sentence into two shorter sentences.**

p. 13, line 7 Delete “for” before “potential”

**Edit made.**

p. 13, line 18 Replace “Alike” with “Like”

**Edit made.**

p.13, line 32 Why did you pool all of the data from 0-4cm for the ladderane concentrations to compare to the rate values? The values are highly variable from surface to 4cm. Did you try just using data from anoxic sediments, where anammox may have been occurring? Or depths where NO<sub>2</sub><sup>-</sup> and NH<sub>4</sub><sup>+</sup> were present? I wonder if you would have seen a better correlation between the ladderane concentrations and the rates. It would be helpful to include some discussion of this.

**In the previous version of the manuscript we pooled the ladderane data from 0-4 cm because this is the approach that is often used in literature. However, we agree with the referee that this is not the best approach when values are highly variable, as in our situation. We have now correlated the potential anammox rates with (1) the average ladderane abundances and (2) with the ladderane abundances in the anoxic sediments layer (1.5-3.5 cm), which coincides with the layer sampled for anoxic slurry experiments for anammox potential. We have also performed correlations between ladderane abundances and environmental parameters (temperature, water depth, salinity, etc.). Yet, we did not see any statistically significant correlations. We have now added this information to the Results. In the Discussion we have now explained that these non-significant correlations might be due to the low number of observations or to differences in microbial spatial heterogeneity between the sediment used for the lipid analysis and the sediment used for the 15N incubations.**

p. 14, lines 17-18 You mention here that H<sub>2</sub>S was never detected in the sediment porewater, but you do not present that data anywhere. I suggest you mention it briefly in the results section since you took the time to describe the microsensors method.

**We have added a sentence in the Results to explain that H<sub>2</sub>S concentrations were below detection limits.**

p. 15, line 5 Replace “process” with “proceeds”

**Edit made.**

p. 15, line 7 Replace “being” with “at”

**Edit made.**

p. 15, line 13 Delete “eventually”

**Edit made.**

p. 15, line 16 Replace “upscale” with “scale up”

**Edit made.**

p. 15, lines 15-17 I’m unsure what conservative method you are referring to. It would be helpful to explain briefly here since it’s important enough to bring up in your discussion.

**We have now explained what source of error we refer to.**

p. 16, line 3 Reword to “The removal rate and the recycling rate were constrained by. . .”

**Edit made.**

p. 16, line 10 Replace “prove” with “suggest”

**Edit made.**

p. 16, line 14 Replace “basin-wise” with “basin-wide”

**Edit made.**

p. 16, lines 17-19 You briefly mention the contribution of DNRA to the TDN flux here, but I think it would be helpful to present the data in Fig. 7 so that the reader can get a feel of interstation variability.

**We have added the ranges of the relative contribution of DNRA to the TDN fluxes in Figure 7.**

Figure 2 Make sure to note which symbols are N vs. C (black vs. white).

**Edit made.**

Figure 6 (c) The y-axis labeled “AAO contribution” should be changed to “ra”, as discussed above. Also, the caption for panel (a) should replace “Shaded” with “Hatched” so as not to be confused with the gray shaded bars (2014).

**Edits made. Thank you.**

Figure 7 In the caption, replace “nitrogen cycling” with “TDN efflux” since that’s more accurate.

**Edit made.**

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Anonymous Referee #2

The authors present a high-quality dataset on nitrogen cycling in coastal sediments with a low carbon loading. The manuscript is generally well written and based on a high-quality dataset comprising in situ flux measurements, incubations experiments to partitioning nitrogen flows and some basic background data (ladderane lipids as biomarker for Anammox, burial of nitrogen using  $^{210}\text{Pb}$  excess, etc).. The conclusions are largely confirming our existing view of nitrogen biogeochemistry in low carbon coastal sediments and such present a useful addition to the literature. I suggest the authors to articulate their DON flux findings a little more.

**We appreciate the reviewer’s acknowledgement of the merits of our work, and we thank her/him for her/his insightful and useful comments. We also believe that the results concerning the DON data are intriguing and novel. We have put more emphasis on the high contribution of the DON flux to the total efflux of fixed nitrogen. This high DON export from the sediments to the water column may be a reason for the high activity of bacterioplankton and the dominance of heterotrophy vs. autotrophy found in the waters of the Gulf of Bothnia (Algesten et al. 2004). We discussed these aspects on page 16 and 17; section 4.3.**

Although the writing is generally clear, some fine tuning and precision of wording would improve this very good manuscript further. - insert hyphens for multi-word adjectives: e.g. bottom-water salinity.

**Edits made.**

- one the one and on the other hand always come together

**Edits made.**

- sometimes the logic of sentences needs improvement, e.g. p3, l. 9-10: pore-water chemistry is the result of N cycling processes; anammox biomarker reflect cycling processes but do not control it, etc.etc. Another example: p. 12, l. 25: our rates therefore represent in situ conditions. Rate reported are representative for the in situ rates. Rates do not represent conditions.

**We appreciated these corrections. We have reworded the text accordingly.**

Oligotrophic marine sediments: is that the right term? Water column ecosystems are considered eutrophic or oligotrophic, but sediments are usually classified as low or high carbon loading systems. Nutrient concentrations are quite high in sediment, including the ones reported here. Moreover, can you use the term oligotrophic for sediments with an oxygen penetration depth of less than 2 cm? Not convincing. > 75% of the seafloor has larger OPD.

**We agree with the reviewer that “oligotrophic” is not the most suitable adjective to describe marine sediments, although it is commonly used in literature. We have changed the title to: “The fate of fixed nitrogen in marine sediments with low organic loading: an in situ study”. For consistency, we have also changed the term “oligotrophic sediments” into “low-organic sediments” throughout the text.**

The authors emphasize somewhat the peculiarities of low temperature conditions, e.g. p. 2, l. 19, but are all deep-sea systems not cold. Consequently there are quite some studies on DNRA in cold systems along ocean margins. Rewrite the text. Moreover, why should temperature matter so much? A permanently cold system will function well, in the end supply of oxidants and reduced substances set the stage.

**Temperate coastal sediments, except for those of the high Arctic/Antarctic, have seasonal temperature variations that may affect biogeochemical processes. In other cold Baltic Sea sediments, for example, temperature was shown to significantly affect nitrogen cycling processes and the partitioning between denitrification and DNRA rates (Bonaglia et al. 2014). Moreover, DNRA bacteria isolated in Arctic fjord sediments had their highest optimal growth rate at 18 °C, while denitrifiers had their optima at 0 °C (Canion et al. 2013).**

**Even in the permanently cold (< 10 °C) GOB sediments we have temperature fluctuations, distinguishing them from the Arctic and deep-sea sediments. To date, we are not aware of any single study reporting on significant DNRA activity in year-round cold sediments, either from coastal environments or the open sea. This is further corroborated by the study just published by McTigue et al. (2016), which showed that denitrification was one to two orders of magnitude greater than DNRA in Alaskan Arctic shelf sediments. Thus, one of the main messages of our paper is that significant DNRA activity cannot be excluded *a priori* in cold, oligotrophic systems.**

The material and methods section is very detailed and sometime too much detailed knowledge is expected from the reader: all the abbreviations, etc. Perhaps a few lines on explaining the principle of the approaches would better guide the reader through the details.

**We believe that it is preferable to describe Methods in details rather than omitting important steps of the operations in this type of scientific works with novel and complex experimental setups. However, in the revised manuscript, we have shortened the <sup>210</sup>Pb and ladderane parts, whose protocols are already been described in details by others. We also briefly introduced the main principles behind each of these methodologies.**

On page 8, it is mentioned that C and N were measured before and after HCL treatment. Two remarks: (1) this is the wrong reference because Verardo et al. used sulfurous acid rather than HCl and (2) communicate to the reader that you report only total nitrogen and organic carbon in this manuscript. You made the right choice of not using Norg because of acidification artifacts.

The procedure by Verardo et al. was referenced because of the type of detector used (flash combustor by a Carlo Erba elemental analyzer). We have specified that we slightly modified the sample preparation method and that only the  $C_{org}$  and N data are presented in the paper. We have also motivated why we excluded the  $N_{org}$  data presentation. Thanks for this feedback.

Burial rates are based on sediment burial rates inferred from  $^{210}Pb$  excess measurements. Although you touch upon the issue of bioturbation in the material and methods sections and conclude that you can ignore it, later on you present visual faune observations suggesting otherwise. Communicate to the reader that burial rates may be inflated because of bioturbation, in particular at stations.. Even better show the  $^{210}Pb$  excess profiles in the appendix/supplementary info.

**We exclude that in this type of sediments bioturbation may have biased burial rates. The macrofaunal organisms retrieved in the benthic chambers and in the sediment cores were almost exclusively specimens of *Monoporeia affinis*, a small amphipod that was found either swimming in the water column or colonizing the upper 3-4 cm of the sediment. The abundances of the deep burrower *Marenzelleria* spp. were negligible and their effect on the  $^{210}Pb$  distribution was therefore minimal. Moreover, macrofauna was completely absent at RA2 and at the GOB stations sediments were laminated below 5-6 cm depth, which clearly exclude particle mixing below that depth. We have specified these points both in the Methods (Page 9, Line 9) and in the Results (Page 10, Lines 10-17).**

- Minor corrections: - p. 1, l. 12: on the global  
**Edit made.**

- p. 1, l. 13: most scientific investigations have increased the last few years because the scientific community has grown. Reformulate.  
**We have edited the first two sentences of the Abstract, which reads much better now.**

- P. 1, l. 17: burial rates were not experimentally determined: they were inferred from  $^{210}Pb$  excess observations  
**Edit made.**

- P. 1, l. 24: clarify here that you mean total dissolved fixed nitrogen.  
**Edit made.**

- P. 2, l. 26: southern and central Baltic Sea are among the : : :  
**Edit made.**

- P. 3, l. 2: but do not report anammox  
**Edit made.**

- P. 4, l. 30: control or output?  
**The correct word here is output.**

- P. 8, l. 11: an dimensionless linear sorption coefficient  
**Edit made.**

- P.10, l. 19: depth-interval weighted average porosities?  
**Edit made.**



- P. 12, l. 15: give the most accurate..

**Edit made.**

- P. 13, l. 17-19: why this role of latitude: is this the cause? I guess that coastal-deep-sea gradient is more important than latitudinal.

**We agree with the reviewer that in this context lower latitude is not really the cause of lower anammox rates. We have thus removed this comparison. Thank you for these remarks.**

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#### **REFERENCES:**

- Algesten, G., J. Wikner, S. Sobek, L. J. Tranvik, and M. Jansson. 2004. Seasonal variation of CO<sub>2</sub> saturation in the Gulf of Bothnia: Indications of marine net heterotrophy. *Global Biogeochem. Cycles* **18**.
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- Canion, A., O. Prakash, S. J. Green, L. Jahnke, M. M. M. Kuypers, and J. E. Kostka. 2013. Isolation and physiological characterization of psychrophilic denitrifying bacteria from permanently cold Arctic fjord sediments (Svalbard, Norway). *Environ. Microbiol.* **15**: 1606-1618.
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# The fate of fixed nitrogen in marine sediments with low organic loading: an in situ study

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**Abstract.** Over the last decades, the impact of human activities on the global nitrogen (N) cycle has drastically increased. Consequently, benthic N cycling has mainly been studied in anthropogenically impacted estuaries and coasts, while its understanding in oligotrophic systems is still scarce. Here we report on benthic solute fluxes and on rates of denitrification, anammox and dissimilatory nitrate reduction to ammonium (DNRA) studied by in situ incubations with benthic chamber landers during two cruises to the Gulf of Bothnia (GOB), a cold, oligotrophic basin located in the northern part of the Baltic Sea. Rates of N burial were also inferred to investigate the fate of fixed N in these sediments. Most of the total dissolved fixed nitrogen (TDN) diffusing to the water column was composed of organic N. Average rates of dinitrogen (N<sub>2</sub>) production by denitrification and anammox (range 53–360 μmol N m<sup>-2</sup> d<sup>-1</sup>) were comparable to those from Arctic and subarctic sediments worldwide (range 34–344 μmol N m<sup>-2</sup> d<sup>-1</sup>). Anammox accounted for 18–26 % of the total N<sub>2</sub> production. Absence of free hydrogen sulfide and low concentrations of dissolved iron in sediment pore water suggested that denitrification and DNRA were driven by organic matter oxidation rather than chemolithotrophy. DNRA was as important as denitrification at a shallow, coastal station situated in the northern Bothnian Bay. At this pristine and fully oxygenated site, ammonium regeneration through DNRA contributed more than one third to the TDN efflux and accounted, on average, for 45 % of total nitrate reduction. At the offshore stations, the proportion of DNRA in relation to denitrification was lower (0–16 % of total nitrate reduction). Median value and range of benthic DNRA rates from the GOB were comparable to those from the southern and central eutrophic Baltic Sea and other temperate estuaries and coasts in Europe. Therefore, our results contrast with the view that DNRA is negligible in cold and well-oxygenated sediments with low organic carbon loading. However, the mechanisms behind the variability in DNRA rates between our sites were not resolved. The GOB sediments were a major source (237 kt y<sup>-1</sup>, which corresponds to 184 % of the external N load) of fixed N to the water column through recycling mechanisms. To our knowledge, our study is the first to document the simultaneous contribution of denitrification, DNRA, anammox and TDN recycling combined with in situ measurements.

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## 1 Introduction

Excess of fixed nitrogen (N) accumulating in aquatic ecosystems due to planktonic dinitrogen (N<sub>2</sub>) fixation, discharge of wastewater and runoff of fertilizers, constitutes one of the greatest eutrophication issues in coastal waters (Howarth and Marino, 2006), ultimately leading to water column anoxia and biodiversity loss (Vaquer-Sunyer and Duarte, 2008). Coastal, estuarine and shelf sediments contribute ~45 % of the global N loss due to the two anaerobic, microbial processes denitrification and anammox (Seitzinger et al., 2006). Denitrification is the stepwise reduction of nitrate to nitrous oxide and N<sub>2</sub>, while anammox produces N<sub>2</sub> through ammonium oxidation coupled to nitrite reduction. These two processes help in counteracting eutrophication by permanent removal of fixed N from the system. A third N reducing process, dissimilatory nitrate reduction to ammonium (DNRA), leads to recycling and preservation of fixed N in the system, and can ultimately increase the occurrence of algal blooms and exacerbate eutrophication if stimulated at the expense of the N<sub>2</sub>-producing pathways (An and Gardner, 2002; Bonaglia et al., 2014a).

In sediments with high sulfide concentrations, DNRA is generally of major importance (Bonaglia et al., 2014a; Christensen et al., 2000; De Brabandere et al., 2015). This is likely because sulfide is used as the electron donor in the nitrate reduction process which is carried out by large sulfur bacteria that proliferate in these conditions (Jørgensen and Nelson, 2004). DNRA dominates nitrate reduction also in sediments with high organic carbon (C) loading in tropical (Dong et al., 2011) and subtropical (An and Gardner, 2002) estuarine sediments, where high C:N ratios would favor this process over denitrification and anammox (Burgin and Hamilton, 2007; Kraft et al., 2014). It has recently been proposed that reduced iron can serve as an alternative electron donor for DNRA in estuarine sediments (Robertson et al., 2016). The few experimental studies conducted in permanently cold (< 10 °C), oligotrophic marine systems have suggested that the role of DNRA is negligible, while denitrification and anammox have been considered the main nitrate/nitrite (as NO<sub>3</sub><sup>-</sup> + NO<sub>2</sub><sup>-</sup> = NO<sub>x</sub><sup>-</sup>) reduction processes (NRPs) (Crowe et al., 2012; Gihring et al., 2010). However, bacterial assimilation has also been found to be an important nitrate removal pathway in cold, low-organic Arctic sediments (Blackburn et al., 1996). In marine systems, anammox is, in relation to denitrification, generally more important in deep environments particularly in manganese (Mn)-rich sediments, where the lower contribution of denitrification may be due to competition with Mn reduction for organic substrates (Engström et al., 2005; Trimmer et al., 2013). Yet, factors responsible for the relative partitioning between the three NRPs in low-organic sediments are still obscure.

Southern and central Baltic Sea are among the most eutrophic marine areas in the world, due to large inputs of nutrients, extended thermohaline stratification and limited water circulation (Elmgren, 2001). However, the northern part of the Baltic Sea — the Gulf of Bothnia (GOB) — is still relatively unaffected by anthropogenic nutrient loading because of low population density and extensive forest coverage in its catchment area (Pettersson et al., 1997), which prevent the risk of dense planktonic blooms similarly to the Arctic coastal zones (Billen et al., 2011). Planktonic primary production in the waters of the GOB is considered to be mainly limited by phosphorus (P), while dissolved inorganic N is generally abundant

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(Rolf and Elfving, 2015; Tamminen and Andersen, 2007). In contrast to the southern Baltic, the GOB water column is well-oxygenated suggesting that anaerobic NRPs occur exclusively in the sediments. To our knowledge, only two studies report rates of benthic denitrification in the GOB, but do not report anammox and DNRA activity (Stockenberg and Johnstone, 1997; Tuominen et al., 1998).

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5 This study targeted rates of NRPs in sediments of the GOB by in situ benthic lander incubations supported by additional on-deck incubations, as the main aim was to better understand processes of the N cycling in this relatively unexplored Baltic Sea basin. The oligotrophic, cold waters of the GOB make this basin an ideal environment to study the relative importance of individual NRPs under pristine conditions. Based on reported chemical data, we hypothesized low benthic N cycling rates dominated by denitrification and anammox, with the latter process prevailing at high Mn concentrations. Pore-water chemistry, anammox biomarkers and C:N ratios were also characterized as they may help explaining variations in rates of N cycling processes. To investigate the fate of fixed N, N burial and efflux from the sediment were also quantified.

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## 2 Methods

### 2.1 Study area and sampling

15 The GOB, which by area and volume represents the second largest basin of the Baltic Sea after the Baltic Proper, is the northernmost Baltic basin and is further divided into the Bothnian Bay (BB) and the Bothnian Sea (BS) (Fig. 1). The areas of the BB and BS cover 36,260 km<sup>2</sup> and 64,886 km<sup>2</sup>, with mean depths of 41 and 66 m, respectively (Leppäranta and Myrberg, 2009). Both basins are normally covered by ice during winter for on average 120 and 60 days in the BB and BS, respectively (Håkansson et al., 1996). Bottom-water salinity decreases from 6 in the southern section of the BS to 2 in the northern section of the BB. Due to sills and archipelago areas in the south of the BS, the GOB remains largely isolated from the density-stratified waters of the Baltic Proper. As such, and because of the low productivity and weak stratification of its water masses, the GOB is generally well-oxygenated throughout the year, and hypoxia has not significantly affected the GOB in the last centuries (Savchuk, 2013). The entire BB and the offshore waters of the BS are considered oligotrophic and P-limited in their current state (Billen et al., 2011; Tamminen and Andersen, 2007).

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25 The four sampling stations (RA2, GOB1, GOB2, and GOB3) are located along a bottom-water salinity gradient, on a north-south transect across the GOB (Fig. 1). RA2 is a shallow coastal station situated just outside the mouth of the Råne River, while GOB1, GOB2 and GOB3 are offshore stations. RA2, GOB1 and GOB2 are in the BB, while GOB3 is in the BS.

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30 The stations were visited during two research expeditions, in June 2013 and in July 2014, with the only exception of station RA2, which was visited only during the second expedition (Table 1). In both years, in situ benthic chamber incubations were performed and sediment samples were collected for determination of various parameters. In 2014, we also performed on-deck incubations of sediment cores and anoxic slurries.

## 2.2 Sediment properties

For analysis of sediment physico-chemical properties and pore water, seafloors were sampled by means of a modified box corer (28 x 28 cm internal diameter) (Blomqvist et al., 2015) and by a Gemini corer (9 cm internal diameter). Both samplers provided nearly undisturbed sediment surfaces.

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5 At each station, one Gemini core was sliced at intervals of 0.5 cm down to 2 cm depth; at intervals of 1 cm from 2 to 6 cm depth; and at intervals of 2 cm from 6 to 20 cm depth. Each sediment slice was split and one half frozen for organic geochemistry parameters, and the other refrigerated for later determination of water content and sediment accumulation rate (see below for analysis details). A second Gemini core was sliced at the same intervals to obtain pore-water nutrient profiles. The sediment slices were centrifuged at 670 g (2500–3000 rpm) for 15 min and the supernatant was immediately filtered using 0.45 µm polyethersulfone (PES) filters then stored dark and refrigerated until analysis. In 2013,  $\text{NH}_4^+$  and the sum of  $\text{NO}_2^- + \text{NO}_3^-$  ( $\text{NO}_x^-$ ) were determined, while in 2014 the three dissolved inorganic nitrogen (DIN) species ( $\text{NH}_4^+$ ,  $\text{NO}_3^-$  and  $\text{NO}_2^-$ ) and **total dissolved nitrogen (TDN)** were analyzed (see below for details).

Two small plastic liners (4.6 cm internal diameter) were inserted into the box core sample and sub-sampled for on deck microelectrode profiling of dissolved oxygen and sulfide concentrations. At least three to five microprofiles were measured in each of the two sediment cores using a Clark-type oxygen microsensor (OX-50, Unisense) and a **hydrogen sulfide** microsensor (H<sub>2</sub>S-50, Unisense) mounted onto a **double-headed motorized micromanipulator (MM33-2, Unisense)**, using a vertical resolution of 100 µm. Sulfide microprofiles were carried out down to 5 cm depth, while the O<sub>2</sub> microprofiles were stopped immediately below the depth where O<sub>2</sub> was exhausted. An overlying water column of 4 cm was left in the sediment core and circulated by a gentle flow of air towards the water surface from an angle of ~45 ° in order to obtain a stable diffusive boundary layer during measurements. Before measurements, OX-50 was calibrated using a 2-point calibration procedure, while H<sub>2</sub>S-50 was calibrated daily in fresh anoxic Na<sub>2</sub>S solutions according to the manufacturer's recommendation.

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## 2.3 In situ benthic chamber incubations

The two Gothenburg benthic landers (big and small landers) were deployed at each station to measure benthic solute fluxes and N cycling process rates (Brunnegård et al., 2004; De Brabandere et al., 2015; Ståhl et al., 2004). The big and small landers are equipped with four and two box-shaped (20 cm x 20 cm) incubation chamber modules, respectively. The landers' chambers enclosed sediment together with the overlying water (**range 8–12 L water**), which was constantly stirred by a horizontal paddle wheel positioned centrally in each chamber (Tengberg et al., 2004). The chambers' lids were closed 2.5–3 h after the benthic lander was deployed on the sediments to assure proper ventilation before incubation started. Physico-chemical conditions inside each chamber, as well as in the ambient bottom water just outside chambers, were monitored with an oxygen optode (3830 or 3835, Aanderaa) and a salinity sensor (3919A, Aanderaa). Both optodes and conductivity sensors

had temperature output. Each chamber was equipped with ten 60 mL syringes for solution injection and water sampling. Three of the six chambers were incubated for O<sub>2</sub> and TDN flux determination (Ståhl et al., 2004). In these chambers, 10 min after lid closed, the first syringe injected 60 mL distilled water, corresponding to 0.5–1 % of the chamber volume. Chamber volumes were calculated from the resulting decrease in salinity. Nine water samples were collected by syringe withdrawal at regular intervals during an incubation period of 29.5–33 h after the lid closed. After recovery of the landers, the syringe samples were filtered (0.45 µm, PES) and stored refrigerated until analysis for TDN concentrations immediately after each cruise.

The rest of the chambers were used for incubation with <sup>15</sup>NO<sub>3</sub><sup>-</sup> tracer for determination of denitrification and DNRA rates following the protocol by De Brabandere et al., (2015), with minor modification. In each of these chambers: 10 min after the lid closed a syringe withdrew a first water sample for nutrient analysis. Ten min later, 60 mL of a 12 mM <sup>15</sup>NO<sub>3</sub><sup>-</sup> solution (prepared by dissolving Na<sup>15</sup>NO<sub>3</sub> 99.4 atom % (Sigma–Aldrich) in distilled water) was injected by the second syringe to reach a final <sup>15</sup>NO<sub>3</sub><sup>-</sup> concentration of ~70 µM in the chamber. After another 10 min the third syringe withdrew a second water sample for nutrient analysis, from which <sup>15</sup>NO<sub>3</sub><sup>-</sup> amendment could be calculated. Seven water samples were collected by syringe withdrawal at regular intervals during an incubation period of 29–32.5 h after the initial operations were concluded. After recovery of the landers, the seven syringe samples from each of the <sup>15</sup>NO<sub>3</sub><sup>-</sup>-amended chambers were sampled first by filling a series of 12 mL Exetainers (Labco) to which 100 µL of a 37 % formaldehyde solution was added to stop biological activity. The Exetainers were stored upside down in a fridge for later analysis of the isotopic composition of N<sub>2</sub>. From the same syringe samples a second aliquot was filtered (PES, 0.45 µm) and split into two plastic vials, one for analysis of DIN, and one for <sup>15</sup>NH<sub>4</sub><sup>+</sup> analysis. The nutrient vials were stored dark and refrigerated until analysis immediately after each cruise, while the <sup>15</sup>NH<sub>4</sub><sup>+</sup> vials were immediately frozen upright.

#### 2.4 Sediment core incubations

In 2014, two sediment box core casts were sampled by inserting 15 plastic liners (4.6 cm internal diameter, 30 cm length) and collecting half sediment and half water for sediment core incubations with the addition of <sup>15</sup>NO<sub>3</sub><sup>-</sup> to determine rates of denitrification and DNRA (De Brabandere et al., 2015; Nielsen, 1992). These incubations were used to determine the relative accumulation of <sup>15</sup>N<sub>2</sub> and <sup>15</sup>NH<sub>4</sub><sup>+</sup> in the sediment and the overlying water, respectively, in order to correct rates from benthic chamber incubations, as this procedure does not take into account the <sup>15</sup>N<sub>2</sub> and <sup>15</sup>NH<sub>4</sub><sup>+</sup> fractions trapped in the sediment (Nielsen, 1992). Since large chambers better capture the spatial heterogeneity of the sediment (Glud and Blackburn, 2002), we use the chamber-based rates as the best estimates of activity in situ.

The sediment cores were transferred into a 25 L incubation tank that was previously filled with ambient bottom water, situated in a temperature-controlled room kept at bottom water temperature (Table 1). The cores were left uncapped for 6 h, during which the water phase of each core was stirred with a magnetic bar driven by an external magnet at 60 rpm.

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Subsequently, 5 mL of a 200 mM  $^{15}\text{NO}_3^-$  solution (prepared by dissolving  $\text{Na}^{15}\text{NO}_3$  99.4 atom % (Sigma–Aldrich) in distilled water) was added to the water tank in order to reach a  $^{15}\text{NO}_3^-$  concentration of  $\sim 70 \mu\text{M}$ . Before and after this addition of  $^{15}\text{NO}_3^-$  solution, triplicate water samples were collected from the tank, filtered (PES,  $0.45 \mu\text{m}$ ) and stored dark and refrigerated for later  $\text{NO}_3^-$  analysis in order to calculate the final  $^{15}\text{NO}_3^-$  amendment. The incubation started after a lag time of up to 14 h, which was necessary to homogeneously mix the added nitrate with the endogenous nitrate and to establish a linear production of  $^{15}\text{N}_2$  within the sediment as inferred from the oxygen penetration depth (Dalsgaard et al., 2000). At the beginning of the incubation the cores were capped with rubber stoppers so that no air bubbles formed and the water was mixed by externally driven magnetic bars. Triplicate cores were sampled at regular intervals during the incubation, which lasted 12 h at RA2 and 22 h at GOB1, GOB2 and GOB3. The  $\text{O}_2$  concentration, monitored in a control core with an optode (3830, Aanderaa), did not decrease by more than 20 % of the initial value during the incubation time. The incubation was terminated by uncapping each core and sampling its water phase with a syringe. An aliquot was transferred into a 12 mL Exetainer to which 100  $\mu\text{L}$  of a 37 % formaldehyde solution was added. A second aliquot was filtered (PES,  $0.45 \mu\text{m}$ ), placed into a plastic vial, and immediately frozen for  $^{15}\text{NH}_4^+$  analysis. Subsequently, the water phase and upper 7–9 cm of sediment were blended into slurry. Slurry samples were collected in 12 mL Exetainers to which 200  $\mu\text{L}$  of a 37 % formaldehyde solution was added. The Exetainers were stored upside down in a fridge until later analysis of  $\text{N}_2$  isotopic compositions. An additional sample of the slurry was taken from each core, centrifuged (670 g for 10 min), filtered (PES,  $0.45 \mu\text{m}$ ), placed into a plastic vial and immediately frozen for  $^{15}\text{NH}_4^+$  analysis.

## 2.5 Anoxic slurry incubations

Anoxic slurry incubations amended with  $^{15}\text{NO}_3^-$  and  $^{15}\text{NH}_4^+$  were performed in order to estimate the contribution of anammox to total  $\text{N}_2$  production during the 2014 expedition (Risgaard-Petersen et al., 2003; Thamdrup and Dalsgaard, 2002). The experiment followed the procedure described in Bonaglia et al., (2014b). Briefly, the oxic layers were removed from two Gemini cores and the 2-cm thick sediment layers below were extruded and homogenized in a glass bottle filled with helium (He). 100 ml of this sediment was transferred to a second glass bottle filled with 900 mL filtered (PES,  $0.45 \mu\text{m}$ ), anoxic bottom water. This slurry was bubbled with He for 10 min to remove any oxygen that entered during previous operations, and was dispensed through a Viton™ tubing into a series of 12 mL Exetainers, each containing a 4 mm glass bead. The bottle was shaken vigorously while filling the Exetainers maintaining the slurry homogeneous throughout dispensing. The Exetainers (n=36) were filled completely and directly capped to avoid air bubbles. The samples were pre-incubated for up to 16 h on a rotating stirrer in order to remove any residual oxygen and nitrate. After pre-incubation, 15 Exetainers received 100  $\mu\text{L}$  of an anoxic 9 mM  $^{15}\text{NO}_3^-$  solution (final  $^{15}\text{NO}_3^-$  concentration:  $\sim 75 \mu\text{M}$ ); 15 Exetainers received 100  $\mu\text{L}$  of an anoxic 9 mM  $^{15}\text{NH}_4^+ + ^{14}\text{NO}_3^-$  solution (final  $^{15}\text{NH}_4^+$  and  $^{14}\text{NO}_3^-$  concentrations:  $\sim 75 \mu\text{M}$  each); and 6 Exetainers received no tracer and were used as control. Triplicate vials from each treatment (n=9) were sampled directly with a syringe by inserting a 3 mL He headspace. The subsample was centrifuged (670 g for 10 min), filtered (PES,  $0.45 \mu\text{m}$ ), placed into a plastic vial and stored cold for later  $\text{NO}_3^-$  and  $\text{NH}_4^+$  analysis, from which the label percentage of the  $^{15}\text{N}$ -

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compounds could be calculated. Formaldehyde (200  $\mu$ L, 37 % solution) was injected into each of the He headspace Exetainers and mixed. The rest of the samples (n=27) were incubated on the rotating stirrer for up to 8 h at station RA2 and up to 18 h at stations GOB1, GOB2 and GOB3. Triplicate vials from the  $^{15}\text{N}$  treatments were sacrificed at regular intervals during incubation by injecting 100  $\mu$ L of 37 % formaldehyde. At the last time point, triplicate vials from the control treatment were also sacrificed to verify that no  $^{15}\text{N}$ -tracer contamination had occurred. All Exetainers were stored upside down in a fridge until analysis of isotopic compositions of  $\text{N}_2$ .

## 2.6 Laboratory analyses and rate calculations

Concentrations of TDN,  $\text{NH}_4^+$ ,  $\text{NO}_3^-$  and  $\text{NO}_2^-$  were determined colorimetrically on a segmented flow nutrient analyzer system (OI Analytical, Flow Solution IV). The dissolved organic nitrogen (DON) concentration was calculated as the concentration of TDN minus the concentrations of  $\text{NH}_4^+$  and  $\text{NO}_x^-$ . Benthic fluxes from lander incubations were calculated as the net concentration change of solutes ( $\text{O}_2$  and TDN) per area and time (Ståhl et al., 2004). The concentrations were corrected for the dilution that occurs when ambient bottom water enters the chamber as samples are withdrawn from the incubated chamber water. Benthic fluxes were calculated by multiplying the slope value of the regression of the concentration values versus time with the height of the incubated water column, which was estimated from the chamber volume. Fluxes were evaluated by the protocol described in Ekeröth et al., (2016), and considered to be significant if the  $p$ -value of the linear regression was  $\leq 0.05$ .

The isotopic composition of the  $\text{N}_2$  samples from the denitrification and anammox experiments were determined by headspace analysis using gas chromatography-isotope ratio mass spectrometry (GC-IRMS, DeltaV plus, Thermo) (De Brabandere et al., 2015). Slopes of the linear regression of  $^{29}\text{N}_2$  and  $^{30}\text{N}_2$  concentration against time were used to calculate production rates of labeled  $\text{N}_2$  ( $p^{29}\text{N}_2$  and  $p^{30}\text{N}_2$ , respectively). Rates of  $\text{N}_2$  production in benthic lander incubations ( $p_{14}lan$ ), and in the water column ( $p_{14}wc$ ) and slurried phase (= water + sediment) ( $p_{14}sl$ ) of sediment core incubations were calculated based on the revised-isotope pairing technique (r-IPT) by Risgaard-Petersen et al., (2003). For clarity,  $p_{14}lan$  was calculated as:

$$p_{14}lan = 2 \times r_{14} (p^{29}\text{N}_2lan + p^{30}\text{N}_2lan (1 - r_{14})) \quad (1)$$

where  $p^{29}\text{N}_2lan$  and  $p^{30}\text{N}_2lan$  are the production rates of labeled  $\text{N}_2$  from the lander incubations and  $r_{14}$  was calculated as:

$$r_{14} = \frac{(1 - ra) \times R_{29} - ra}{(2 - ra)} \quad (2)$$

where  $ra$  is the contribution of anammox to  $\text{N}_2$  production (see below) and  $R_{29}$  is the ratio between  $p^{29}\text{N}_2lan$  and  $p^{30}\text{N}_2lan$ . The parameters  $p_{14}wc$  and  $p_{14}sl$  were calculated likewise and their equations are not presented here for conciseness.



As  $p_{14}lan$  takes into account only the  $N_2$  diffusing to the water column, the in situ or actual  $N_2$  production rate ( $p_{14}$ ) was calculated using the following formula:

$$p_{14} = p_{14}lan / F_{wc}$$

(3)

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5 where  $F_{wc}$  is the fraction of  $N_2$  production diffusing to the water column of sediment core incubations calculated as:

$$F_{wc} = p_{14}wc / p_{14}sl$$

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The in situ  $N_2$  production rate was partitioned into  $N_2$  production coupled to nitrification ( $p_{14}n$ ) and  $N_2$  production depending on the water column nitrate ( $p_{14}w$ ) according to Risgaard-Petersen et al., (2003).

Moved down [1]: Sediment core incubations were only carried out in 2014, but because sediment geometries and zonation were similar in the two years we expected the same  $F_{wc}$ , consistent with De Brabandere et al. (2015).

10 The contribution of anammox ( $ra$ ) to the total  $N_2$  production — necessary to calculate  $p_{14}$  (Risgaard-Petersen et al., 2003) — was estimated from anoxic slurry incubations. As significant rates of DNRA were measured in benthic lander incubations (see below), a correction to account for  $^{15}NH_4^+$  production in Exetainers incubated with  $^{15}NO_3^-$  was performed based on Song et al., (2013).  $F_A$ , the fraction of  $^{15}NH_4^+$  in the ammonium pool in the nitrate reduction zone, was estimated from the concentrations of  $^{15}NH_4^+$  and  $^{14}NH_4^+$  at the final time point (tenth syringe) from benthic lander incubations.  $F_A$  in these incubations was  $\leq 0.06$ . The in situ anammox rate (AAO) was calculated by multiplying  $p_{14}$  by  $ra$ . The in situ denitrification rate (DEN) was calculated as the  $p_{14}$  minus AAO.

Concentrations of labeled ammonium ( $^{15}NH_4^+$ ) were quantified after oxidation of  $NH_4^+$  to  $N_2$  with hypobromite (Warembourg, 1993). Samples were purged with helium for 10 min, treated with alkaline hypobromite, and analyzed by the headspace technique as described above. Slopes of the linear regression of  $^{15}NH_4^+$  concentration against time were used to calculate production rates of labeled ammonium ( $p^{15}NH_4^+$ ) after correction for adsorption assuming a dimensionless linear sorption coefficient of 1 (De Brabandere et al., 2015). DNRA rates in benthic lander incubations (DNRA $lan$ ) were calculated according to Christensen et al., (2000). Similar to  $N_2$  production rates, the in situ DNRA rate was calculated from the formula:

$$DNRA = DNRAlan / F_{wc}$$

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25 where  $F_{wc}$  was used because  $p^{15}NH_4^+$  was not detected in the time courses from sediment core incubations at GOB1–3 (see Results).

Sediment samples used for water content determination were dried at 70 °C for 4–9 days until reaching a constant weight. Porosity was calculated from the water content assuming a dry sediment density of 2.65 g mL<sup>-1</sup>. The dried sediment was

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ground into a homogeneous powder and was analyzed for total carbon (C) and total nitrogen (N), as well as organic carbon ( $C_{org}$ ) and organic nitrogen ( $N_{org}$ ) after treatment with HCl fumes, with a Carlo Erba NA-1500 elemental analyzer modifying the procedure by Verardo et al. (1990). Only  $C_{org}$  and N data were presented in this study.  $N_{org}$  data were not shown because of acidification artifacts.

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Frozen sediment samples used for the analysis of ladderane lipids and the sediment accumulation rate (SAR) were weighed, freeze-dried and weighed again. The freeze-dried sediment was manually homogenized into fine powder. Gamma emitting radioisotopes for  $^{210}\text{Pb}$  dating were analyzed by gamma spectrometry following the protocol by Cutshall et al. (1983). For each sediment profile, the excess  $^{210}\text{Pb}$  activity concentration ( $\text{Bq kg}^{-1}$ ) was plotted versus the cumulative sediment mass ( $\text{g cm}^{-2}$ ) centered at the middle of each sediment layer. As bioturbation was absent (RA2), or low and restricted to the three top sediment cm (GOB1–3), an exponential regression curve of the form  $e^{-kx}$  was fitted to the data, where x is the cumulative sediment depth ( $\text{g cm}^{-2}$ ) and k is a constant. Using the half-life of  $^{210}\text{Pb}$  (22.4 years), SAR ( $\text{g cm}^{-2} \text{y}^{-1}$ ) was determined from  $(\ln(2)/22.4)/k$ . The N burial rate was calculated for each station by multiplying the SAR with the average sedimentary N content at the depth of 14–20 cm, i.e. where it had reached a stable value.

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Deleted: Briefly, 15 mL of sediment powder was packed into a Petri dish plastic container, weighed and placed on a high purity germanium (Ge) detector with a beryllium window for 1–2 days. For the surface sediment samples with a restricted dry sample mass a well-type Ge detector was used. At the counting period, sample self-absorptions were analyzed according to Cutshall et al. (1983). The  $^{210}\text{Pb}$  excess activities in the samples were obtained by subtracting the  $^{226}\text{Ra}$  activity from the gross  $^{210}\text{Pb}$  activity.

For analysis of ladderane lipids, biomarkers for anammox bacteria, ground sediment was extracted using a modified method (additional extraction step using dichloromethane (DCM)) of Matyash et al. (2008). The target biomarker, a  $\text{C}_{20}$ -[3]-ladderane monoether attached to a phosphocholine (PC) head group, was analyzed using high performance liquid chromatography electrospray ionization tandem mass spectrometry (HPLC-ESI-MS/MS, Thermo). Separation was performed using reverse phase chromatography on a Kinetex XB C18,  $1.7\mu\text{m}$ ,  $100\text{\AA}$  column (Phenomenex, USA) based on the methods of Lanekoff and Karlsson (2010) and Zhu et al. (2013). The PC- $\text{C}_{20}$ -[3]-ladderane monoether in the samples was quantified using a PC- $\text{C}_{20}$ -[3]-ladderane monoether standard, purified from anammox cell biomass from an anammox waste water treatment reactor at Syvab, Himmerfjärdsverket, Grödinge, Sweden.

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Deleted: The standard was purified by adapting the semi-preparative HPLC method of Jaeschke et al. (2009) to reverse phase. Standard purity was determined by measuring the P content by spectrophotometry (Evolution 260, Thermo) and comparing this to the calculated P molar weight of the standard (Jaeschke et al., 2009).

## 2.7 Mass balance calculations and data analysis

A benthic mass balance was calculated to assess whether the sediments were sources or sinks of N to/from the water column. It was calculated assuming steady state conditions where N deposition on the sediment surface equals the sum of burial,  $\text{N}_2$  production, and the TDN efflux from the sediment. At each station, average rates of burial,  $\text{N}_2$  production and TDN exchange were calculated for 2013 and 2014, and the average biannual values were summed to calculate N deposition. The relative contribution (%) of each of the three processes was calculated by dividing each process by N deposition. Basin-wide annual N removal and recycling (in  $\text{kt N y}^{-1}$ ) were calculated by scaling up the respective average rate to the GOB area reported in section 2.1.

If not stated otherwise in the text, measurements are reported as average  $\pm$  standard error (SE). Detection limits of in situ rates were estimated as the median value of  $2 \times \text{SE}$  of the significant rates (De Brabandere et al., 2015), and were  $3.4 \mu\text{mol N}$

$\text{m}^{-2} \text{d}^{-1}$  for DNRA and  $29 \mu\text{mol N m}^{-2} \text{d}^{-1}$  for  $p_{14}$ . Statistical tests were performed in order to detect differences in solute fluxes and process rates between stations. Homogeneity of variance of the dataset was checked using Cochran's test. One way analysis of variance tests were performed. When the differences in the mean values among stations were greater than would be expected by chance, pairwise multiple comparisons among stations were performed by the Tukey test. Correlations between process rates and environmental factors were tested using the Pearson's Correlation Coefficient ( $r$ ). The level of significance was always set to  $p < 0.05$ . Statistical analyses were performed with SigmaPlot 13.0 (Systat Software, CA, USA).

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### 3 Results

#### 3.1 Sediment properties and macrofauna

Sediment at GOB1 was dark olive brown in the top 3 cm, where few burrows of the amphipod *Monoporeia affinis* were visible against the walls of the plastic liners. Sediment was light olive brown from 3 to 6 cm depth and characterized by brown/black laminations below 6 cm depth. Sediment at GOB2 was similar to GOB1 but with grey/black laminations instead. GOB3 sediment was light olive brown until 6 cm depth, followed by a grey clay layer. Both GOB2 and GOB3 sediments presented a sparse number of burrow structures of *M. affinis* down to 3 cm, and faint burrow traces of the polychaete *Marenzelleria* spp. down to 6-7 cm depth. RA2 sediment was fluffy and olive brown-colored in the top cm, followed by a black/dark olive brown layer. No burrows could be found at RA2. Relative abundances of macrofaunal taxa in the GOB sediments are presented in Supplementary Fig. S1.

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Sediment porosities were highest at RA2, followed by GOB2, GOB1 and GOB3, and depth-integrated average porosities ranged from 0.82 to 0.93 at the four sampling sites (Supplementary Fig. S2). Sedimentary  $C_{\text{org}}$  and N generally decreased with sediment depth, except for GOB3 in 2014 where they increased slightly below 2.5 cm depth (Fig. 2). At RA2, both  $C_{\text{org}}$  and N were constant until 4.5 cm, below which they started decreasing. The  $C_{\text{org}}$  and N content at the sediment surface (top 2 cm) were 4–5 % and 0.4–0.5 % dry weight, respectively, except at GOB3 where they were ~2.5 % and ~0.25 % dry weight. Sediment  $C_{\text{org}}:N$  molar ratios increased slightly with depth at all stations except at RA2, where they were relatively constant (Supplementary Fig. S3). Sediment surface  $C_{\text{org}}:N$  ratios were 12–14 at all stations, except of GOB3 where they were lower (10–12).

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Pore-water hydrogen sulfide ( $\text{H}_2\text{S}$ ) concentrations were always below the detection limit (0.3  $\mu\text{M}$ ) of the method. Pore-water nitrite ( $\text{NO}_2^-$ ) concentrations were low at all stations ( $< 0.8 \mu\text{M}$ ), while nitrate ( $\text{NO}_3^-$ ) or the nitrate–nitrite sum ( $\text{NO}_x^-$ ) showed a concentration peak (6–18  $\mu\text{M}$ ) between the sediment surface and down to 1.25 cm. Ammonium ( $\text{NH}_4^+$ ) and dissolved organic nitrogen (DON) concentrations in the pore water increased with sediment depth at all stations (Fig. 3).

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Downcore  $\text{NH}_4^+$  concentrations were highest at RA2, followed by GOB2  $\geq$  GOB1 > GOB3. An  $\text{NH}_4^+$  depleted surface layer

(<1.5  $\mu\text{M}$ ) was absent at RA2 but present and increasing in thickness from north to south at the GOB stations, ranging from 0.75 cm at GOB1 to 2.5 at GOB3. GOB2 had the highest DON concentrations, ranging from 406 to 677  $\mu\text{M}$  below 5.5 cm depth. DON was the dominant form of total dissolved nitrogen (TDN) in the pore water at all stations except for RA2, where  $\text{NH}_4^+$  was the dominant species in the TDN pool.

5 The sediment accumulation rates (SAR) at the four stations varied between 0.022 (GOB3) and 0.032  $\text{g cm}^{-2} \text{y}^{-1}$  (GOB1) (Table 1). Burial rates of N ranged from 0.098 (GOB1) to 0.15  $\text{mmol N m}^{-2} \text{d}^{-1}$  (RA2) (Table 1). The oxygen penetration depth (OPD) was higher in 2013 than in 2014 (Table 1). RA2 had lower OPD (0.24 cm) than the other three stations, where it ranged from 0.80 to 1.9 cm (Table 1).

10 The targeted ladderane lipid anammox biomarker (PC-C<sub>20</sub>-[3]-ladderane monoether) could be detected at all stations at abundances that were highly variable, both spatially and temporally (Fig. 4). PC-C<sub>20</sub>-[3]-ladderane monoethers were more abundant in the top sediment layer and decreased with depth, with the exception of GOB2 in 2014, where a subsurface peak was recorded (Fig. 4). Higher abundances were present in 2013 compared to 2014 at GOB1 and GOB2, and the opposite was observed at GOB3. Average abundances in the top 4 cm of sediment were in the range ~~34-1073~~  $\text{pg g}^{-1}$  sediment dry weight (DW) equivalent to ~~6-177~~  $\text{pg g}^{-1}$  sediment wet weight (WW) (Table 2). We could not find any significant correlation ( $p > 0.05$ ) between ladderane abundances and environmental parameters (temperature, salinity, water depth, OPD).

### 3.2 Benthic exchange of solutes and rates of N cycling processes

20 Total oxygen uptake (TOU) varied between -5.0 (GOB2, 2013) and -11.3  $\text{mmol O}_2 \text{m}^{-2} \text{d}^{-1}$  (RA2, 2014) (Fig. 5a) and was significantly higher at RA2 than at the other stations (ANOVA,  $p < 0.001$ ). Separate fluxes of  $\text{NH}_4^+$  and  $\text{NO}_x^-$  were mainly non-significant (data not shown) contrarily to the TDN fluxes, which were found to be significant and varied between 0.1  $\text{mmol N m}^{-2} \text{d}^{-1}$  at GOB1 in 2014 and 0.9  $\text{mmol N m}^{-2} \text{d}^{-1}$  at RA2 (Fig. 5b). The TDN efflux was mainly supported by DON because the contribution of DIN to the TDN flux was minor, and the pore-water concentrations just below the sediment surface were dominated by DON. TDN flux was non-significant at GOB3 in 2014. RA2 had a significantly higher TDN flux than GOB1 (2014) and GOB3 (ANOVA, Tukey pair-wise test,  $p < 0.001$ ).

25 Sediment core incubations revealed that most of the  $^{15}\text{N}_2$  produced was retained in the sediment during incubation, with  $F_{\text{wc}}$  values of 0.26, 0.23, 0.33 and 0.21 at GOB1, GOB2, GOB3 and RA2, respectively. Sediment core incubations were only carried out in 2014, but because sediment geometries and zonations were similar in the two years we expected consistent  $F_{\text{wc}}$ . However,  $F_{\text{wc}}$  values in 2013 could have been slightly lower than those we measured in 2014 because OPDs were greater in 2013 than in 2014, implying that  $\text{N}_2$  production happened deeper down in the sediment. Thus, we cannot totally exclude underestimation of in situ  $\text{N}_2$  production rates in 2013.

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Anoxic slurries incubated with  $^{15}\text{NH}_4^+$  and  $^{14}\text{NO}_3^-$  revealed anammox activity in all the investigated sediments, with potential rates ranging from 0.2 to 0.8  $\text{nmol N g}^{-1} \text{ WW h}^{-1}$  at GOB1 and GOB3, respectively (Table 2). Potential denitrification rates (from  $^{15}\text{NO}_3^-$ -amended anoxic slurry incubations) were higher than anammox and varied from 1.0 to 3.5  $\text{nmol N g}^{-1} \text{ WW h}^{-1}$  at GOB1 and GOB3, respectively (Table 2). Anammox potential rates did not correlate with either average ladderane abundances ( $r = 0.62, p > 0.05$ ) or with ladderane abundances from the anoxic layer (1.5–3.5 depth) of the sediments ( $r = 0.70, p > 0.05$ ). Values of  $ra$  were 18 % at RA2, 19 % at GOB1 and GOB3, and 26 % at GOB2, meaning that denitrification was more important than anammox at all stations (Table 2). There was no statistically significant difference (ANOVA,  $p = 0.167$ ) in  $ra$  between different stations.

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Trends of in situ  $\text{N}_2$  production (53–360  $\mu\text{mol N m}^{-2} \text{ d}^{-1}$ ), denitrification (43–297  $\mu\text{mol N m}^{-2} \text{ d}^{-1}$ ), and anammox (10–63  $\mu\text{mol N m}^{-2} \text{ d}^{-1}$ ), were similar between stations and rates were highest at RA2, and progressively lower at GOB2, GOB1 and GOB3 (Fig. 6a, b, c).  $\text{N}_2$  production was mainly sustained by  $\text{NO}_3^-$  produced within the sediment by nitrification ( $p_{14n}$ ), which contributed 92, 90, 87 and 83 % at GOB2, GOB1, RA2 and GOB3, respectively (Fig. 6a). Thus,  $\text{N}_2$  production dependent on water column  $\text{NO}_3^-$  ( $p_{14w}$ ) was minimal. Rates of total  $\text{N}_2$  production,  $p_{14n}$ ,  $p_{14w}$  and denitrification were significantly higher at RA2 (ANOVA,  $p < 0.01$ ) than at the other stations. Anammox rates at RA2 and GOB2-2014 were significantly higher than at GOB3 (ANOVA, Tukey pair-wise test,  $p < 0.01$ ).

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At GOB1–3, DNRA could not be determined in sediment core incubations because  $p^{15}\text{NH}_4^+$  was not detectable in the time courses, and this is consistent with the higher detection limits of core incubations compared to lander incubations. Thus, the in situ DNRA rates were calculated using  $F_{wc}$  values from  $^{15}\text{N}_2$  accumulation in core incubations (see above), which may have caused an underestimation of the in situ DNRA rates because  $\text{NH}_4^+$  diffusion to the water column is slower than  $\text{N}_2$  diffusion. DNRA was not detectable in situ at GOB3, which is consistent with low to undetectable efflux of TDN measured in benthic chambers without  $^{15}\text{NO}_3^-$  addition at this station (Fig. 5b, 6d). DNRA rates were lower than denitrification rates at GOB1 and GOB2 (ranging from 8 to 22  $\mu\text{mol N m}^{-2} \text{ d}^{-1}$ ) (Fig. 6d). Station RA2 displayed DNRA rates which were comparable to the denitrification rates,  $266 \pm 90$  and  $297 \pm 49 \mu\text{mol N m}^{-2} \text{ d}^{-1}$ , respectively. RA2 had significantly higher DNRA rates than the other stations (ANOVA,  $p < 0.001$ ).

## 4 Discussion

### 4.1 Coexistence of multiple nitrate/nitrite reduction processes

We measured rates of  $\text{N}_2$  production and DNRA by means of benthic chamber lander incubations, which give the most accurate results of benthic transformation processes (Glud and Blackburn, 2002; Tengberg et al., 1995), and by correcting these rates with the relative anammox contribution ( $ra$ ) determined in parallel anoxic slurry incubations (Risgaard-Petersen et al., 2003). The in situ incubations indicated that denitrification, DNRA and anammox all contributed significantly to NRPs

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in the GOB sediments. To our knowledge, our study is the first that detects simultaneous activity of denitrification, DNRA, and anammox with in situ measurements. In anaerobic slurry incubations, the  $^{15}\text{N}$  isotope pairing technique (IPT) may be inaccurate when all three NRPs coexist because DNRA reduces  $^{15}\text{NO}_3^-$  to  $^{15}\text{NH}_4^+$ , which may affect the isotope distributions of anammox and denitrification. This is particularly significant when DNRA activity is high and the background concentration of  $^{14}\text{NH}_4^+$  is low (Song et al., 2016). At the stations where DNRA was detected,  $F_A$  (the fraction of  $^{15}\text{NH}_4^+$  in the ammonium pool) was 1.7 % at GOB1 and GOB2, and 5.6 % at RA2 which would lead to a simultaneous < 5 % overestimation of denitrification rate and underestimation of anammox rate (Song et al., 2016). Although this represents a small inaccuracy, we accounted for the influence of coupled DNRA-anammox during calculations. Rates reported are therefore representative for the in situ conditions.

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Previous attempts to discern between the three NRPs in low-organic marine sediments involved the use of ex situ sediment core incubations (Crowe et al., 2012; Gihring et al., 2010). However, this method may not be representative of the in situ biogeochemical conditions when N cycling rates are low, e.g., at cold, low productivity sites. In Arctic sediments, DNRA was indeed not detected (< 50  $\mu\text{mol N m}^{-2} \text{d}^{-1}$ ) (Gihring et al., 2010). Benthic chamber incubations have 2.7 and 4.5 fold lower detection limits for denitrification and DNRA rate measurements than sediment core incubations, respectively (De Brabandere et al., 2015). This is due to core-to-core variation compared to repeated sampling of the same chamber. In addition, ex situ experiments generally involve heavy manipulation of the sediments, which can lead to alteration of sediment structures and infaunal biomass when small cores are used for sampling and incubation (Glud and Blackburn, 2002).

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Rates of NRPs were dominated by denitrification (rate 43–297  $\mu\text{mol N m}^{-2} \text{d}^{-1}$ ; median 91  $\mu\text{mol N m}^{-2} \text{d}^{-1}$ ), which was in the lower range of rates previously measured in the GOB (0–940  $\mu\text{mol N m}^{-2} \text{d}^{-1}$  (Stockenberg and Johnstone, 1997; Tuominen et al., 1998)). However, those previous studies could not take into account potential anammox activity and may not be representative of the in situ rate (Risgaard-Petersen et al., 2003). Denitrification rates from this study were in the lower range of rates found in the adjacent Baltic Proper and Gulf of Finland (38–1619  $\mu\text{mol N m}^{-2} \text{d}^{-1}$ ; median 170  $\mu\text{mol N m}^{-2} \text{d}^{-1}$  (Bonaglia et al., 2014a; Deutsch et al., 2010; Hietanen and Kuparinen, 2008; Jäntti et al., 2011)). However, our rates were comparable to those reported from Arctic and subarctic sediments (33–340  $\mu\text{mol N m}^{-2} \text{d}^{-1}$  (Gihring et al., 2010; Rysgaard et al., 2004; Seitzinger and Giblin, 1996)). Anammox rates (6–75  $\mu\text{mol N m}^{-2} \text{d}^{-1}$ ; median 27  $\mu\text{mol N m}^{-2} \text{d}^{-1}$ ) were higher than those measured in coastal areas of the Gulf of Finland and the central Baltic Proper (0–38  $\mu\text{mol N m}^{-2} \text{d}^{-1}$ ; median 11  $\mu\text{mol N m}^{-2} \text{d}^{-1}$  (Bonaglia et al., 2014a; Hietanen and Kuparinen, 2008)), but comparable to rates of Arctic sediments from Greenland and Svalbard (1–92  $\mu\text{mol N m}^{-2} \text{d}^{-1}$  (Gihring et al., 2010; Rysgaard et al., 2004)) and those of Celtic and Irish Sea sediments (2–46  $\mu\text{mol N m}^{-2} \text{d}^{-1}$  (Jaeschke et al., 2009)). It thus seems that the GOB sediments behave similarly to Arctic and subarctic sediments in terms of global N loss and partitioning between denitrification and anammox. Like Arctic sediments,

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denitrification might be limited by the availability of organic matter, which ~~could~~ increase the contribution of anammox to N loss ~~compared to sediments from the central Baltic Sea~~.

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DNRA rates found in this study (0–266  $\mu\text{mol N m}^{-2} \text{d}^{-1}$ ; median 19  $\mu\text{mol N m}^{-2} \text{d}^{-1}$ ) were in the lower range of rates reported in previous studies from anthropogenically impacted Baltic Sea estuaries and hypoxic shelf sediments (7–1060  $\mu\text{mol N m}^{-2} \text{d}^{-1}$ ; median 23  $\mu\text{mol N m}^{-2} \text{d}^{-1}$  (Bonaglia et al., 2014a; Jäntti et al., 2011)), albeit median values between these two environments were comparable. Interestingly, these comparisons suggest that the overall importance of DNRA in Baltic Sea sediments is similar across the gradient of trophic conditions. The only successful measurement of DNRA in Arctic, subarctic and boreal sediments to date describes rate of DNRA of only 0.12  $\mu\text{mol N m}^{-2} \text{d}^{-1}$  in the Lower St. Lawrence Estuary (Crowe et al., 2012). Tropical and subtropical estuaries with high  $C_{\text{org}}$  and nitrate concentrations, ~~however~~, have been shown to be dominated by DNRA at rates up to 27,288  $\mu\text{mol N m}^{-2} \text{d}^{-1}$  (An and Gardner, 2002; Dong et al., 2011). Temperate estuarine and coastal sediments investigated by the  $^{15}\text{N}$ -labeling technique have in situ rates that are intermediate between those extreme values (0–130  $\mu\text{mol N m}^{-2} \text{d}^{-1}$  (Christensen et al., 2000; Rysgaard et al., 1996)), and are comparable to our range.

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The presence of anammox bacteria was further substantiated by the ~~quantification~~ of ladderane lipids, ~~specific biomarkers of~~ anammox bacteria (Sinninghe Damste et al., 2002). ~~In our study we specifically targeted the intact PC-monoether ladderane as it represents a more accurate proxy for living anammox cells than core ladderane lipids~~ (Boumann et al., 2006; Brandsma et al., 2011; Jaeschke et al., 2009). ~~Ladderane abundances did not correlate with potential anammox rates. However, heterogeneity in microbial spatial distribution between the sediment used for the lipid analysis and the sediment used for the anammox incubations combined with the small sample size could affect this correlation.~~ Our findings contrast with those of Bale et al., (2014), who observed a good agreement between intact PC-monoether ladderane abundance, potential anammox rates, 16S rRNA and hszA gene copy abundance in North Sea sediment. However, other studies have also shown that the abundances of PC-monoether ladderane can be weak indicators of anammox activity (Jaeschke et al., 2009), possibly due to the fact that the PC-monoether ladderanes may be degradation lysis products of intact polar lipids (Brandsma et al., 2011). The abundance of PC-monoether ladderanes generally decreased with sediment depth indicating that living anammox bacteria were mainly distributed close to the sediment surface, as previously reported for shallow sediments off northwest Africa and in the Irish Sea (Jaeschke et al., 2010; Jaeschke et al., 2009). The range of abundance of PC-monoether ladderanes found in this study (~~34–1073~~  $\text{pg g}^{-1}$ ) was up to two orders of magnitude higher than that from Celtic and Irish Sea sediments (0–60  $\text{pg g}^{-1}$  (Jaeschke et al., 2009)), and that from north-west African sediments (< 1–30  $\text{pg g}^{-1}$  (Jaeschke et al., 2010)), but comparable to the range reported in organic-rich, muddy sediments from the North Sea (100–1250  $\text{pg g}^{-1}$  (Lipsewers et al., 2014)).

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## 4.2 Control factors of nitrate/nitrite reduction processes

In marine sediments, denitrification and DNRA are mainly driven by the oxidation of organic C and hydrogen sulfide (Canfield et al., 2005). In the sediments of the Gulf of Bothnia we consider a major dependence of denitrifiers and DNRA bacteria on hydrogen sulfide unlikely as this compound never accumulated in the pore water at any of the investigated stations. It has recently been shown that reduced dissolved iron ( $\text{Fe}^{2+}$ ) can serve as electron donor for DNRA both in synthetic groundwater (Coby et al., 2011) and in seasonally hypoxic estuarine sediments (Robertson et al., 2016). In those sediments, DNRA was most active in sites with high pore-water  $\text{Fe}^{2+}$  concentrations ( $>100 \mu\text{M}$ ) in the top mm below the sediment surface (Robertson et al., 2016). We also consider this possibility unlikely for the sediments of the GOB as concentrations of pore-water  $\text{Fe}^{2+}$  in the top 2 cm of sediment were more than one order of magnitude lower than this, ranging from 3 to  $12 \mu\text{M}$  (Hannah S. Weber, unpublished). In particular, RA2, the station with the highest rates of DNRA, had pore-water  $\text{Fe}^{2+}$  concentrations of only  $5 \mu\text{M}$  in the top sediment layer.

The absence of dissolved sulfide and low concentrations of  $\text{Fe}^{2+}$  in the GOB sediments rather indicate that the quantity and/or quality of organic matter exerted considerable influence on denitrification and DNRA, and suggest that the apparent electron donor for nitrate reduction was organic C and not an inorganic substrate. Rates of heterotrophic nitrate respiration were generally low because of the oligotrophic nature of this ecosystem, which provides low organic C loading to the sediments. Nitrate respiration by denitrification and DNRA only accounted for  $<1\%$  of the total C mineralization inferred from the TOU at GOB3 and  $\sim 2\%$  at GOB1 and GOB2, which is comparable to estuarine Baltic Proper sediments where rates were limited by labile C supply (Bonaglia et al., 2014a). At station RA2 nitrate respiration made up  $5\%$  of the total C mineralization, and this higher percentage is consistent with the lower OPD at this station than at the other stations (Table 1). With similar  $C_{\text{org}}$  and N contents among stations, the relatively higher contribution of nitrate respiration at RA2 might be explained by process dependence on organic C quality rather than quantity, because at the offshore stations high remineralization rates in the water column render the sinking material more refractory (Algesten et al., 2004).

In continental shelf sediments with high reactive Mn content, it has been shown that denitrification is outcompeted by Mn reduction as C remineralization proceeds, which renders anammox relatively more important than denitrification (Thamdrup, 2012; Trimmer et al., 2013). In the Bothnian Bay there was a clear increase in surface sediment Mn content (dithionite-extractable) with water depth, at  $47$ ,  $66$  and  $158 \mu\text{mol g}^{-1} \text{DW}$  at RA2, GOB1 and GOB2, respectively (Hannah S. Weber, unpublished). However,  $r_a$  and rates did not differ significantly between these three stations, indicating that the effect of Mn content on nitrate reduction in the GOB was much weaker than that reported from the deep Skagerrak, where denitrifiers were almost totally outcompeted by Mn-reducers at Mn contents of  $270\text{--}421 \mu\text{mol g}^{-1}$  (Trimmer et al., 2013). In benthic environments where phytoplanktonic organic matter *sensu* Redfield ( $C:N \approx 6.6$ ) is oxidized with nitrate as electron acceptor through NRPs, the predicted  $r_a$  is  $29\%$  (Thamdrup, 2012). However, C:N ratios in the GOB were always higher than Redfield's ratio, which may explain the lower  $r_a$  ( $18\text{--}26\%$ ) to total  $\text{N}_2$  production that we found in the GOB sediments.

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DNRA contribution to total nitrate reduction ranged from 7 to 63 % in the subarctic, oligotrophic Bothnian Bay. Because of the conservative method used to **scale up** the DNRA rates from the chamber incubations ( $F_{we}$  derived from the  $p_{14}$  values and not from the  $p^{15}\text{NH}_4^+$ , see Results), these DNRA contributions are likely underestimates. Yet, to our knowledge, our study is the first that demonstrates significant, even predominant (in one lander's chamber at RA2) DNRA rates in an oligotrophic environment. These results are in contrast with what has been proposed in the recent literature, i.e., that DNRA is negligible in oligotrophic systems of cold and well-oxygenated waters (Crowe et al., 2012; Gihring et al., 2010). DNRA was of major significance at the shallow, **low-organic** site, where the relatively high C:N ratios of ~14 might provide a competitive advantage for DNRA bacteria vs. denitrifiers (Hardison et al., 2015; Kraft et al., 2014). Strains of the bacterium *Shewanella* spp., a genus that performs DNRA, were recently isolated in Arctic fjord sediments and their highest optimal growth rate was at 18 °C, while denitrifiers had their optima at 0 °C (Canion et al., 2013). We thus speculate that DNRA bacteria may have the capacity to increase their activity when the temperatures increased by ~6 °C from the winter to the summer, which was the case at RA2. It is clear from these results that further understanding of the controlling factors of heterotrophic DNRA in oligotrophic environments is necessary and such studies should focus on kinetic experiments coupled to molecular analysis of DNRA bacteria.

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#### 4.3 Fixed N loss vs. recycling in Gulf of Bothnia sediments

We estimated the N mass balance for the benthic GOB environment assuming that N deposition on the sediment surface equals the sum of burial,  $\text{N}_2$  production and total dissolved N efflux from the sediment (see Methods for calculations). The relative contribution of the N turnover mechanisms was comparable between the investigated stations (Fig. 7). Only 36–46 % of the particulate organic N that sinks to the GOB sediments was permanently lost from the system. We estimated that bacterial  $\text{N}_2$  production and N burial together removed on average 160 kt  $\text{N y}^{-1}$  from the entire basin, which is 1.2-fold higher than the total external N load (Savchuk, 2005). The removal rate **and the recycling rate** were constrained by the low rates of N cycling processes at GOB3. The calculated flux of unsupported  $^{210}\text{Pb}$  was substantially lower at GOB3 (59  $\text{Bq m}^{-2} \text{y}^{-1}$ ) than at the other three stations (121–230  $\text{Bq m}^{-2} \text{y}^{-1}$ ), which may indicate that this station temporally exhibited accumulation and erosion behavior. Unlike the other stations, there was no clear decrease of  $\text{C}_{\text{org}}$  or N content down-core at GOB3, and the C:N ratios were relatively constant throughout the sediment profiles. Interestingly, in the 2014 profiles, both  $\text{C}_{\text{org}}$  and N content increased slightly with depth at this station. C:N ratios generally increase when organic matter is decomposed due to preferential remineralization of N compared to C (Canfield et al., 2005), so these observations may **suggest a decrease in** organic matter degradation in the last few decades at this station.

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The GOB sediments recycled 54–64 % of the deposited N to the water column in the form of TDN, which is close to the recycling percentage reported from **anthropogenically impacted** sediments, such as those in the shallow Yangtze Estuary (Deng et al., 2015) and the North Sea coast (Billen, 1978). Thus, it appears that in the summer the majority of the particulate organic N that sinks to the GOB sediments returns to the water column. **Notably, the DON flux was the main component of**

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the total flux of TDN in the GOB. High benthic export of DON to the water column may be a reason for the high activity of bacterioplankton and the dominance of heterotrophy vs. autotrophy, which was found in the waters of the GOB (Algesten et al., 2004).

5 The calculated basin-wide recycling rate of  $237 \text{ kt N y}^{-1}$  was 1.8-fold higher than the terrestrial N load plus the N input from atmosphere (Savchuk, 2005). This high internal N recycling into the benthic–pelagic system of the GOB sustains water column primary production and may be a contributing factor to the strong P limitation of this basin (Rolff and Elfving, 2015). In particular, DNRA in the shallow coastal sediment of the Bothnian Bay sustains up to 37 % of the N recycled from the sediment to the water column, suggesting that this microbial process must be taken into account in N budgets not only in  
10 euxinic, but also in oligotrophic systems. Although the assumption of steady state contains uncertainties as deposition and rates of N cycling may vary on a seasonal scale, our calculations give an indication of the relative importance of N loss vs. recycling in the GOB sediments during the period of maximal primary production (Klais et al., 2011).

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## Tables

**Table 1: Main site parameters obtained at the investigated sites. SAR is the sediment accumulation rate and OPD is the average oxygen penetration depth.**

Station	Sampling season	Coordinates	Depth (m)	Temperature (°C)	Salinity	SAR ( $\text{g cm}^{-2} \text{y}^{-1}$ )	Burial rate ( $\text{mmol N m}^{-2} \text{d}^{-1}$ )	OPD (cm)
GOB1	June 2013	23°23.7' E, 65°11.5' N	86	2.6	3.5	0.032	0.098	1.3
	July 2014			1.9	3.4			1.1
GOB2	June 2013	21°59.5' E, 64°11.6' N	111	1.4	4.0	0.031	0.132	1.9
	July 2014			4.9	4.1			1.3
GOB3	June 2013	18°33.2' E, 62°07.1' N	91	2.6	6.0	0.022	0.124	1.6
	July 2014			3.2	6.2			0.80
RA2	July 2014	22°26.8' E, 65°43.8' N	12.5	8.4	2.6	0.029	0.155	0.24



5 **Table 2: Average rates of potential denitrification and anammox from anoxic slurry incubation expressed per gram of wet sediment with associated standard errors (SE, n=15). Values of *ra*, and average abundances of PC-monoether ladderanes in the top 4 cm of sediment with associated SE (n=4). DW is sediment dry weight and WW is sediment wet weight.**

Station	Sampling season	Denitrification (nmol N g <sup>-1</sup> h <sup>-1</sup> )		Anammox (nmol N g <sup>-1</sup> h <sup>-1</sup> )		<i>ra</i> (%)	Ladderanes (pg g <sup>-1</sup> DW)		Ladderanes (pg g <sup>-1</sup> WW)	
		average	SE	average	SE		average	SE	average	SE
GOB1	June 2013						1073	932	177	149
	July 2014	1.04	0.06	0.24	0.02	19	267	152	51	28
GOB2	June 2013						264	148	33	17
	July 2014	1.86	0.07	0.59	0.02	26	34	14	6	2
GOB3	June 2013						70	22	18	4
	July 2014	3.47	0.21	0.79	0.02	26	365	128	129	52
RA2	July 2014	1.50	0.03	0.29	0.00	18	57	22	7	3

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Figures

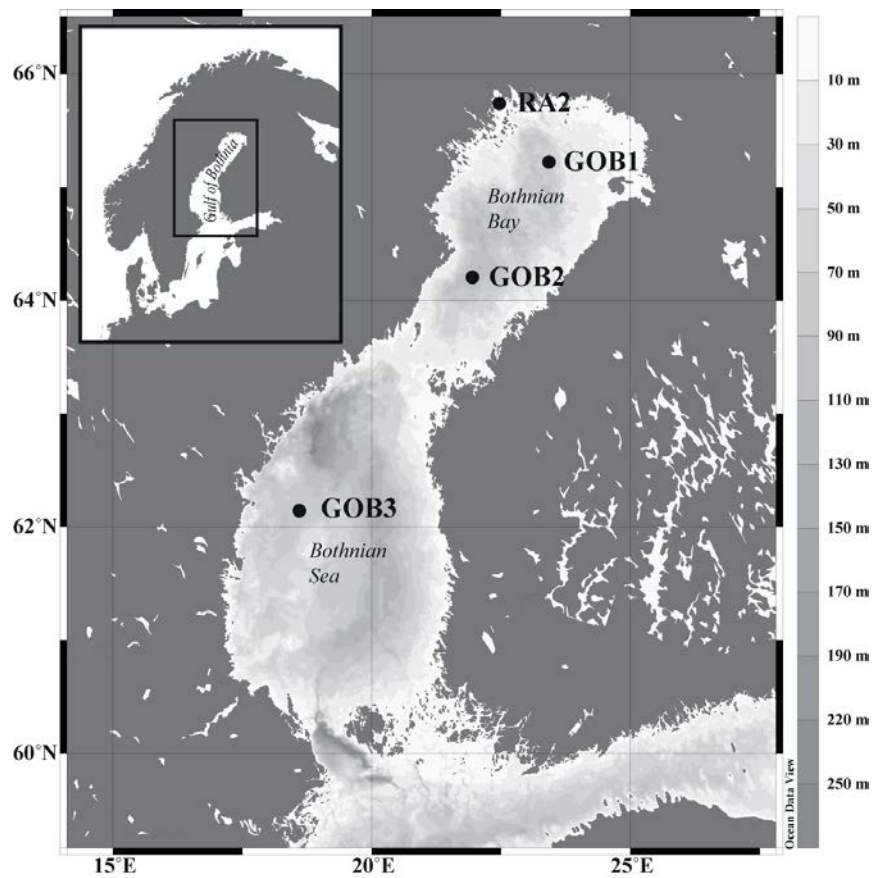


Figure 1: Location of the four sampling stations in the Gulf of Bothnia and bathymetry of its two sub-basins, the

5 Bothnian Bay and the Bothnian Sea.

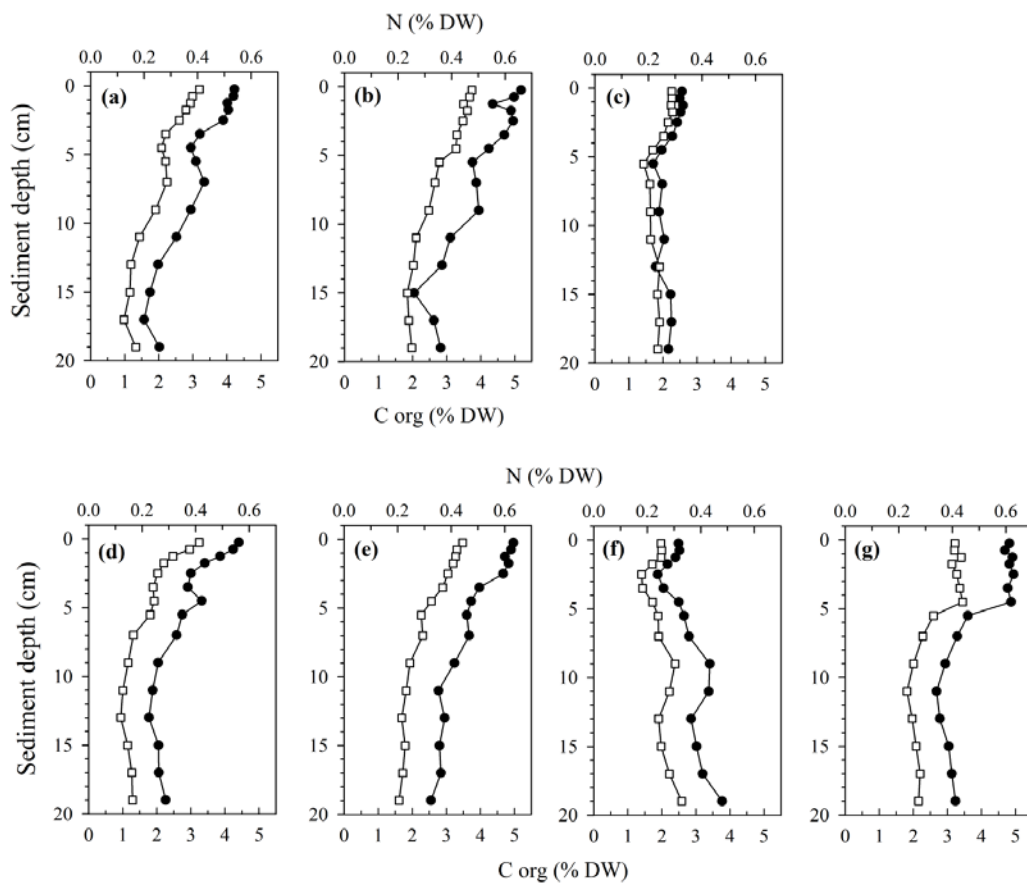


Figure 2: Profiles of organic carbon ( $C_{org}$ , represented by black dots) and nitrogen (N, represented by white squares) content in the sedimentary solid phase at station GOB1 (a), GOB2 (b) and GOB3 (c) in 2013; and GOB1 (d), GOB2 (e), GOB3 (f), and RA2 (g) in 2014. All values are expressed as  $C_{org}$  or N percentage of sediment dry weight (DW).

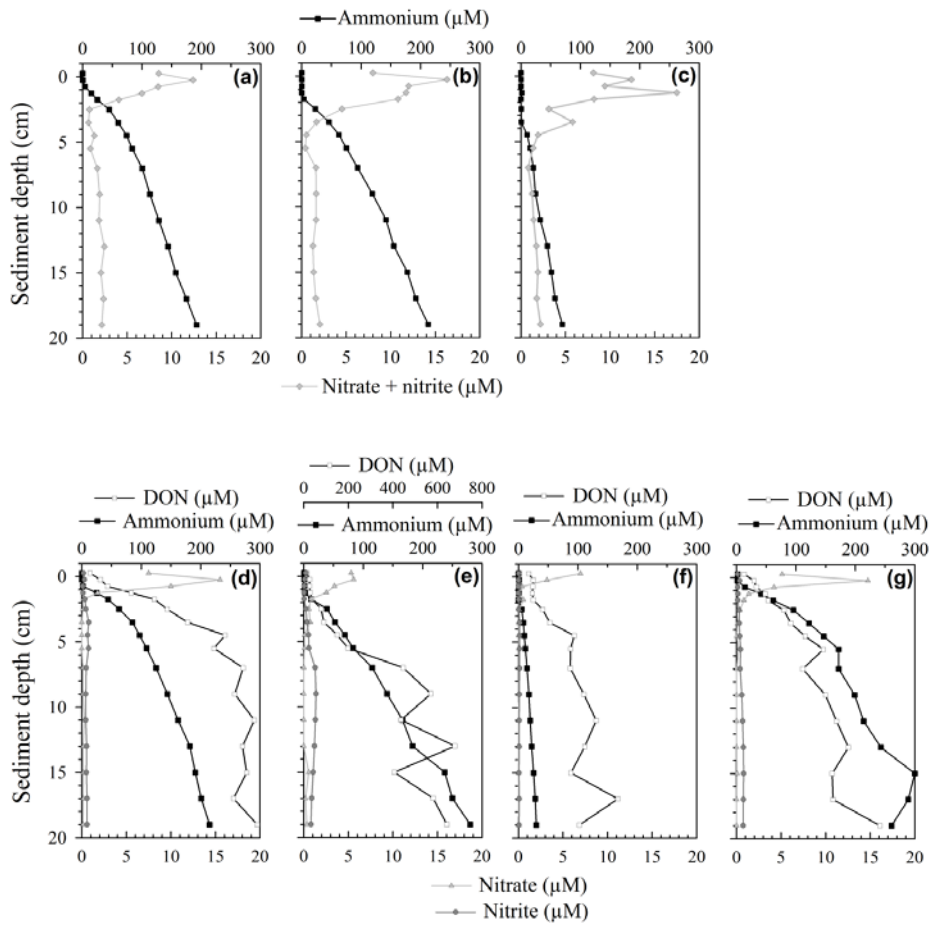


Figure 3: Pore-water concentration profiles of ammonium, nitrate, nitrite and dissolved organic nitrogen (DON) as a function of depth in the sediment of GOB1 (a), GOB2 (b), and GOB3 (c) in 2013; and GOB1 (d), GOB2 (e), GOB3 (f), and RA2 (g) in 2014. Values on top of each profile represent bottom-water samples. Note a second x-axis with different scale for (e).

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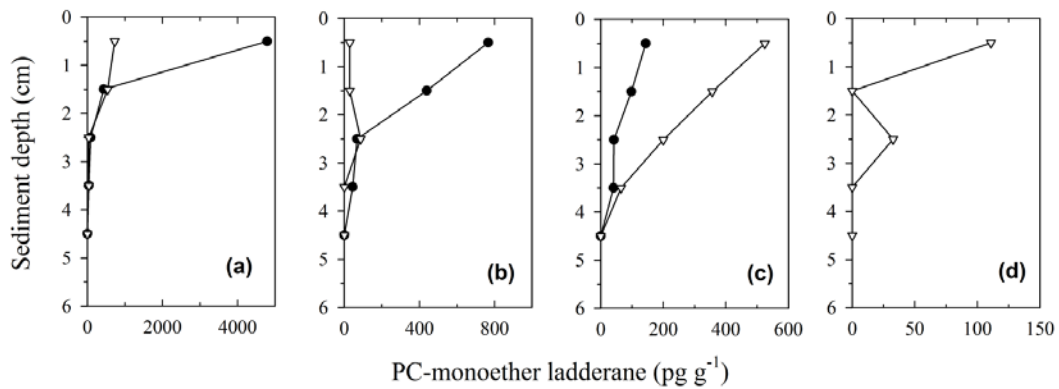
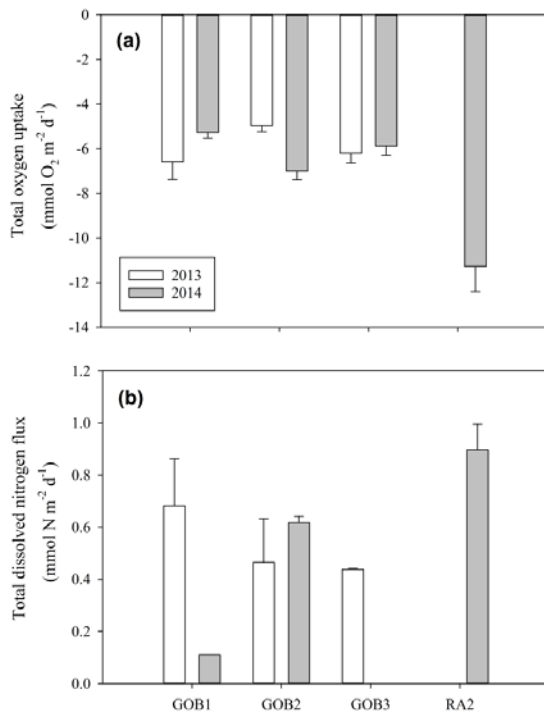


Figure 4: Abundance profiles of PC-C<sub>20</sub>-[3]-ladderane monoether (pg g<sup>-1</sup> sediment dry weight) as a function of depth at GOB1 (a); GOB2 (b); GOB3 (c); RA2 (d). Black dots represent values from 2013 while white triangles represent

5 | values from 2014. Please note different scales on the x-axes.



**Figure 5: Total oxygen uptake (TOU) by the sediment (a), and flux of total dissolved nitrogen (TDN) across the sediment-water interface (b) measured at the four stations by in situ incubations with benthic chamber landers. Bars represent average value and error bars represent standard errors. White bars and grey bars represent 2013 and 2014 rates, respectively. At station GOB3, TDN fluxes in 2014 were non-significant.**

5 rates, respectively. At station GOB3, TDN fluxes in 2014 were non-significant.

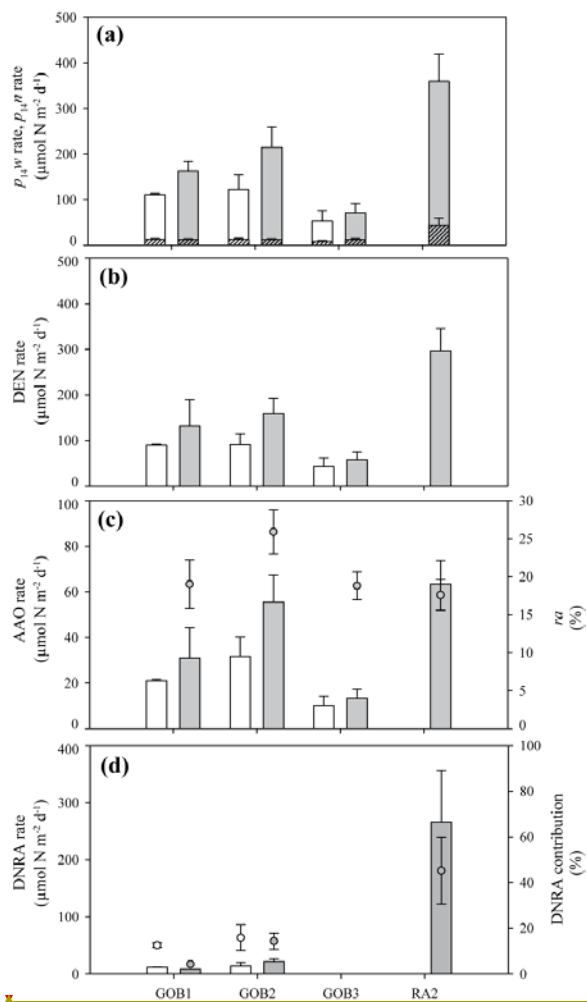
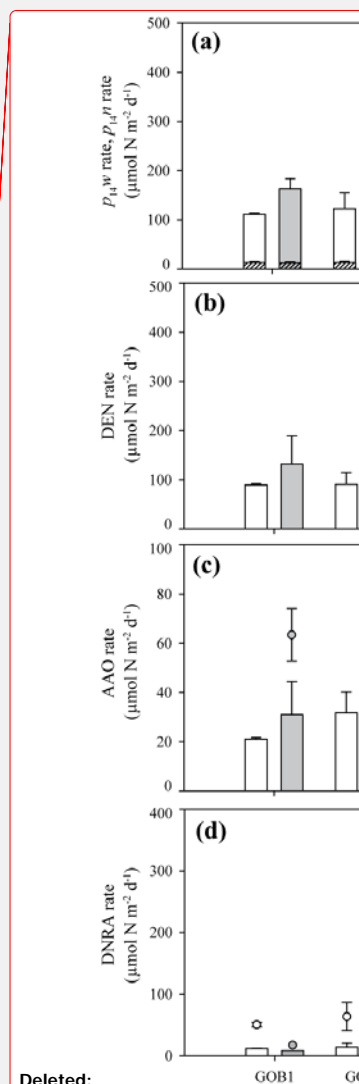


Figure 6: Average rates of  $\text{NO}_x^-$ -reducing processes measured at the four stations by in situ incubations with benthic chamber landers amended with  $^{15}\text{N}$ -nitrate, where white and gray bars represent 2013 and 2014 rates, respectively, with associated standard errors: (a) **Hatched** bars represent rates of  $\text{N}_2$  production depending on the water column nitrate ( $p_{14w}$ ) while **unhatched** bars represent rates of  $\text{N}_2$  production coupled to nitrification ( $p_{14n}$ ); (b) Bars represent total denitrification (DEN) rates; (c) Bars represent total anammox (AAO) rates and refer to the left Y-axis while



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grey dots represent percentages of  $r_a$  and refer to the right Y-axis; (d) Bars represent total DNRA rates and refer to the left Y-axis. White and grey dots represent contribution of DNRA to total nitrate reduction (DEN rate + DNRA rate) and refer to the right Y-axis for year 2013 and 2014, respectively. DNRA rates at GOB3 were not significant.

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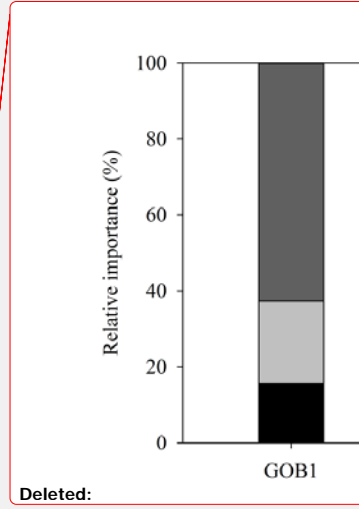
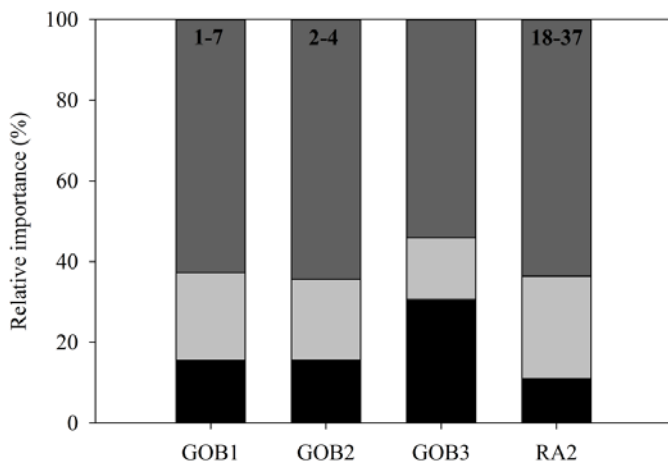


Figure 7: Relative importance in percentage of sediment burial (black bars), N<sub>2</sub> production (light grey bars) and **TDN efflux** (dark grey bars) as N sink and source at the four stations. **Range values represent the contribution (%) of DNRA to the TDN efflux.** Processes are expressed on a molar (N) scale. See text for calculations.

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