

1 **The effects of different environmental factors on biochemical composition of particulate**
2 **organic matters in Gwangyang Bay, South Korea**

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16 **Abstract**

17 Biochemical composition of particulate organic matter (POM) through phytoplankton
18 photosynthesis is important to determine food quality for planktonic consumers as well as
19 physiological conditions of phytoplankton. Major environmental factors controlling for the
20 biochemical composition were seasonally investigated in Gwangyang Bay which has only natural
21 conditions (e.g., no artificial dams) in South Korea. Water samples for the biochemical compositions
22 were obtained from three different light depths (100%, 30%, and 1%) mainly at 3 sites in Gwangyang
23 bay from April 2012 to April 2013. Different biochemical classes (carbohydrates [CHO], proteins
24 [PRT], and lipids [LIP]) were extracted and then the concentrations were determined by the optical
25 density measured with a spectrophotometer. The highest and lowest of PRT compositions among the
26 three biochemical classes were in April 2012 (58.0%) and August 2012 (21.2%), whereas the highest
27 and lowest LIP compositions were in August 2012 (49.0%) and April 2012 (24.8%), respectively.
28 CHO composition was recorded high in January 2013 and maintained above 25% during the study
29 period. The calorific contents of food material (FM) ranged from 1.0 Kcal m⁻³ to 6.1 Kcal m⁻³ (annual
30 mean ± S.D. = 2.8 Kcal m⁻³ ± 1.1 Kcal m⁻³). Based on Pearson's correlation coefficient analysis, a
31 major governing factor for biochemical composition of POM was dissolved inorganic nitrogen
32 loading from river-input in Gwangyang bay. In conclusion, relatively larger amount of FM and higher
33 calorific contents of POM found in this study compared to other regions reflected good nutritive
34 conditions for sustaining productive shellfish and fish populations in Gwangyang bay. Continuous
35 observations are needed for monitoring marine ecosystem response to potential environmental
36 perturbations in Gwangyang bay.

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38 **Key words:**

39 Particulate organic matter, biochemical composition, phytoplankton, nitrogen source

40 **1. Introduction**

41 Particulate organic matter (POM) mostly from phytoplankton photosynthesis in the euphotic
42 layer is an important food source for planktonic consumers in water columns (Cauwet, 1978) and their
43 biochemical contents reaching the benthic environments are largely utilized by benthic organisms
44 (Nelson and Smith, 1986; Rice et al., 1994). Therefore, POM is an essential link between surface and
45 benthic ecosystems (Graf, 1992). Previous studies showed that the biochemical composition of the
46 POM such as protein (PRT), lipid (LIP) and carbohydrate (CHO) levels could provide useful
47 information on the nutritional value which is potentially available to consumers (Mayzaud et al, 1989;
48 Navarro et al., 1993; Navarro and Thompson, 1995). However, previous studies mainly focused on
49 the occurrence in the different patterns of biochemical composition of POM. It is noteworthy to
50 investigate how biochemical composition of POM responds to changes in various environmental
51 factors, such as nutrients, light, temperature, and salinity and to assess food quantity for higher trophic
52 levels.

53 The coastal areas represent one of the world's most vital aquatic resources, supporting and
54 providing food resources and habitats for large numbers of fish and shellfish species (Kwak et al.,
55 2012; Wetz and Yoskowitz, 2013; references therein). In Gwangyang bay, the southern coast of Korea
56 (Fig. 1), coastal fisheries and shellfish farming have been prevalence. Over the past decades, the bay
57 have become industrialized such as the construction of steel mill company, power plant and industrial
58 complex and environmental disturbances have been predicted. Also, estuaries have a high short-term
59 variability depending on many episodic events, such as freshwater inputs, tidal cycles (neap-spring),
60 and wind (storms) (Cloern and Nichols, 1985). These anthropogenic forces and environmental
61 changes drastically affect the estuarine habitat properties which can cause different biochemical
62 compositions of POM. Unfortunately, little information is yet available on the biochemical
63 composition of POM in the bay, South Korea. Hence, this study tested the question of the main
64 environmental factors determining seasonal variation and of biochemical composition POM and

65 assessed quantity of food material (FM) in the bay. Physical (temperature, salinity, irradiance, river-
66 input and rainfall data), chemical (nutrients), and biological (chlorophyll-*a* [chl-*a*], particulate organic
67 carbon [POC] and nitrogen [PON]) parameters were measured in order to both characterize the origin
68 of POM and understand their effects on the biochemical composition of POM. The aims of this study
69 were to: (1) investigate seasonal variation of biochemical composition of POM, (2) identify the origin
70 of POM, and (3) determine a major governing environmental factor for biochemical composition of
71 POM in Gwangyang Bay, Korea.

72 **2. Materials and methods**

73 **2.1. Study site and sampling procedure**

74 The study site was located in Gwangyang Bay (34.9 ° N, 127.8 ° E), the southern coast of
75 Korea (Fig. 1). The total area of the bay is 230 km² at mean sea level (Kang et al., 2003). The bay is
76 characterized by semidiurnal tides with a maximal range of about 4.8 m at spring tides (Korea
77 Hydrographic and Oceanographic Administration). Freshwater flows into the bay from the Seomjin
78 River at the northern part of the bay (mean flow 27 m³ s⁻¹ and annually 1.9 × 10⁹ t during the study
79 period; the National Institute of Environmental Research) and seawater enters through the narrow
80 southern channel (Yeosu Channel).

81 To obtain data for seasonal variation of POM in the euphotic depth, the field samplings were
82 undertaken at 3 stations of the bay (St.1 or St. 2A, St. 4, and St. 5; see Fig. 1) on a seasonal basis April,
83 June, August, and October in 2012 and January and April in 2013. St. 1 was changed to St. 2A after
84 April 2012 because of logistic problems. Both stations have similar environmental conditions at a
85 relatively close distance. Using a 5 L Niskin water sampler, water samples were collected at different
86 depths of 3 light intensities (100%, 30%, and 1% of surface irradiances; hereafter 3 light depths)
87 ~~which were determined by a secchi disk~~ and transferred to brown sample bottles which were
88 previously washed with a solution of 0.1 N HCl. The different 3 light depths were determined by a

89 secchi disk using vertical attenuation coefficient ($K_d = 1.7/\text{secchi depth}$) from Poole and Atkins (1929)
90 which have been applied globally.

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91 To obtain *in situ* physical parameters, water temperature and salinity were measured with
92 YSI-30 (YSI incorporated) and photosynthetically active radiation (PAR) was measured onboard
93 during the cruise. PAR was measured one time per each cruise at every 30 seconds during the
94 incubation hours for primary productivity by a quantum sensor (LI-190SA, LI-COR) with a data
95 logger (LI-1400, LI-COR) on deck. Since the main purpose of the PAR measurements was calculating
96 hourly primary productivity executed for 4~5 hours during day time around local noon time, the
97 irradiance values in this study might be not representative for our sampling periods. Rainfall and
98 river-input data during the study period were obtained from the Korea Meteorological Administration
99 (<http://www.kma.go.kr/index.jsp>) and the National Institute of Environmental Research
100 (<http://water.nier.go.kr/main/mainContent.do>). For relationships between river-input and other factors,
101 river-inputs were integrated from 20 days prior to our sampling dates since phytoplankton
102 productivity is recovered after 20 days after rainfall in Gwangyang Bay according to Min et al. (2011).

103 **2.2. Chl-*a* and major inorganic nutrient analysis**

104 In order to determine chl-*a* concentration, water samples from 3 light depths were filtered
105 through 25 mm GF/F (Whatman, 0.7 μm) which were kept frozen immediately and returned to the
106 laboratory at Pusan National University, Korea for a further analysis. The filters for chl-*a*
107 concentration were extracted in 90% acetone in a fridge (4 $^{\circ}\text{C}$) for 24 h and centrifuged for 20
108 minutes at 4000 rpm. Using a fluorometer (Turner Designs, 10-AU) which had been calibrated with
109 commercially purified chl-*a* preparations, chl-*a* concentrations were measured and calculated (Parsons
110 et al., 1984). Water samples for inorganic nutrient concentrations from surface and bottom waters
111 were obtained from Niskin bottles. The samples were kept frozen (-70 $^{\circ}\text{C}$) and sent for analysis to the
112 laboratory in the East Sea Fisheries Research Institute (QUAATRO, Seal Analytical).

113 **2.3. Particulate organic carbon and nitrogen analysis**

114 The water samples (300 ml) for POC, PON, and $\delta^{13}\text{C}$ of POM were collected from surface at
115 the 3 stations at every sampling time. The water samples were filtered through pre-combusted (450 °C,
116 4 h) 25 mm GF/F (Whatman, 0.7 μm) using a low vacuum pressure less than 5 in. Hg. The filters for
117 POC, PON, and $\delta^{13}\text{C}$ values were preserved frozen (-20 °C) for further analysis at home laboratory.
118 For stable isotope analysis, the preserved filters were acidified by concentrated hydrochloric acid
119 fumes overnight to remove carbonate (Hama et al., 1983) and the abundances of ^{13}C and ^{15}N and
120 the total amounts of POC and PON were determined using a Thermo Finnigan Delta + XP mass
121 spectrometer at the stable isotope laboratory of the University of Alaska Fairbanks, USA.

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122 **2.4. Biochemical composition analysis**

123 The water samples for the biochemical composition (carbohydrates, proteins, and lipids) of
124 POM were collected from 3 light depths. The water samples were filtered through 47 mm GF/F
125 (Whatman, 0.7 μm pore), which were immediately frozen at -70 °C and preserved for biochemical
126 composition analysis at the home laboratory.

127 *Protein analysis*

128 Protein (PRT) concentrations were assessed according to a modified method of Lowry et al.
129 (1951). The filters for PRT analysis were transferred into 12 mL centrifuge tubes with 1 mL DH_2O ,
130 respectively. The filters were grounded (using a glass rod) in the tubes with a 5 ml alkaline copper
131 solution (a mixture of 2% Na_2CO_3 in 0.1 N NaOH with 0.5% $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ in 1 % sodium or
132 potassium tartrate; 50:1, v/v). The solutions for PRT concentrations were mixed well (using a vortex)
133 and allowed to stand for 10 min at room temperature in the hood. After 10 min, 0.5 mL of diluted
134 Folin-Ciocalteu phenol reagent (1:1, v/v) was added into the solution, mixed occasionally with a
135 vortex mixer, and allowed to sit for 1 h 30 min. The solutions with a blue color were centrifuged at
136 3,000 rpm for 10 min. Absorbance of the supernatant was measured at 750 nm. Bovine Serum

137 Albumin (2 mg mL^{-1} , SIGMA) was used as a standard for the PRT concentration.

138 *Lipid analysis*

139 Lipid (LIP) concentrations were extracted according to a column method modified from
140 Bligh and Dyer (1959), and Marsh and Weinstein (1966). The filters for LIP analysis were transferred
141 into 16 mL glass tubes with 3 mL of chloroform-methanol (1:2, v/v). The filters in the tubes were
142 grounded, and then the mixtures were mixed using a vortex mixer. For LIP extraction, glass tubes
143 with samples were stored in the fridge ($4 \text{ }^{\circ}\text{C}$) to prevent the solvents from evaporating. After 1 h, the
144 solvents were centrifuged at 2,000 rpm for 10 min and the supernatants were collected and stored in
145 new tubes. This extraction procedure was performed once again immediately. When the extractions
146 were completed, 4 mL of DH_2O was added to the solution in the new tubes, and the solution was
147 homogenized using a vortex mixer. After mixing, the tubes were centrifuged at 2,000 rpm for 10 min,
148 and the solvents were separated into two phases (the chloroform phase for lipids and methanol +
149 DH_2O phase). The methanol + DH_2O phase was removed from the solvent using a Pasteur pipette.
150 The chloroform phase was placed in a dry oven at $40 \text{ }^{\circ}\text{C}$ for 48 h. After it totally dried for
151 carbonization analysis (Marsh and Weinstein 1966), 2 mL of H_2SO_4 was added to the tubes and they
152 were placed in a heating block at $200 \text{ }^{\circ}\text{C}$ for 15 min. After this heating procedure, the tubes were
153 quickly placed in a water bath at room temperature; 3 ml of DH_2O was added to the tubes and the
154 solvents were homogenized (with a vortex mixer) and stood for 10 min or until all bubbles had
155 disappeared. Absorbance of the supernatant was measured at 375 nm. Tripalmitin solutions were used
156 as a standard for the LIP concentration.

157 *Carbohydrate analysis*

158 Carbohydrate (CHO) concentrations were measured according to Dubois et al. (1956). The
159 POM samples for carbohydrate analysis were transferred individually into 15 mL polypropylene (PP)
160 tubes. After 1 mL of DH_2O was added to the PP tubes, the samples were grounded using a glass rod.

161 One ml of 5 % phenol for CHO extraction was added additionally, and the solutions were allowed to
162 stand for 40 min at room temperature in the hood. After the extraction, 5 mL of sulfuric acid (H₂SO₄)
163 was added to the solutions, mixed using a vortex mixer, and allowed to stand for 10 min. The
164 solutions with an orange-yellow color were centrifuged at 3,500 rpm for 10 min. Absorbance of the
165 supernatant was measured at 490 nm using UV spectrophotometer (Labomed, Germany). D (+) -
166 glucose solutions (1 mg mL⁻¹, SIGMA) were used as a standard for the CHO concentration.

167 **2.5. Statistical analyses and calorific value calculation**

168 Statistical tests were carried out using the statistic software “SPSS” (*t*-test, ANOVA and
169 Pearson’s Correlation Coefficient). The level of significance was set at $p < 0.05$. The calorific value
170 (Kcal g⁻¹) of the food material (FM) (FM was defined by Danovaro et al. (2000); PRT + LIP + CHO
171 concentrations; hereafter FM) and the calorific content of FM (Kcal m⁻³ = Kcal g⁻¹ × g FM m⁻³) were
172 calculated using the Winberg (1971) equation.

173 **3. Results**

174 **3.1. Seasonal distribution and variation of environmental factors and chl-*a* concentrations**

175 The values of environmental factors were summarized in Table 1. The temperature ranged
176 from 5.5 °C to 26.1 °C. ~~The temperature increased from April to August (the highest temperature in~~
177 ~~August 2012 at St. 4: 26.1 °C) and decreased from August to January (the lowest temperature in~~
178 ~~January 2013 at St. 2A: 5.5 °C). T and~~ the salinity ranged from 14.5 ‰ to 32.9 ‰ ~~during our sampling~~
179 ~~period. Generally, the salinity increased from St. 1 or St. 2A to St. 5.~~ Relatively lower salinity, which
180 is mainly affected by fresh water input from the Seomjin River, was observed at St. 2A. The annual
181 average euphotic depth was 6.5 ± 3.4 m, ranging from 2 m to 12 m.

182 The highest nutrient concentrations were measured in April 2012, when the concentrations of
183 NO₂ + NO₃, SiO₂, NH₄, and PO₄ were above 5.0 μM, 2.0 μM, and 0.2 μM, respectively, except at 1%

184 light depth at St. 4. All inorganic nutrients except SiO₂ were nearly depleted in August 2012 (Table 1).
185 During the rest of our study period, NO₂ + NO₃ and SiO₂ concentrations were observed with similar
186 decreasing patterns from St.1 or St. 2A to St. 5. NH₄ concentrations averaged from October 2012 to
187 April 2013 were 1.1 μM ± 0.4 μM, ranging from 0.5 μM to 1.9 μM. PO₄ concentrations (average ±
188 S.D. = 0.1 ± 0.1 μM) ranged from 0 to 0.4 μM ~~throughout the water columns at all stations except at~~
189 ~~St. 2A in April 2012~~ during the study period.

190 Surface irradiance averaged from each measurement for 4-5 hours ranged from 167.9 ±
191 133.5 to 1593.3 ± 414.5 μmols m⁻² s⁻¹ (average ± S.D.) from April 2012 to April 2013. The highest
192 and lowest irradiance were recorded in April 2013 and April 2012, respectively. Chl-*a* concentrations
193 in the euphotic depth ranged from 0.8 μg L⁻¹ to 14.2 μg L⁻¹ during the study period (annual mean ±
194 S.D. = 3.4 μg L⁻¹ ± 2.8 μg L⁻¹; Table 1).

195 Monthly rainfall and river-input in the study location ranged from 15.6 mm to 559.0 mm
196 (annual mean ± S.D. = 151.0 mm ± 155.5 mm) and 42.3 to 447.2 x 10⁶ t (annual mean = 144.4 x 10⁶ t),
197 respectively (Table 2). Rainfall and river-input were recorded as high during summer and low during
198 winter (Table 2). ~~Average~~

199 ~~irradiance during our incubation hour ranged from 167.9 ± 133.5 to 1593.3 ± 414.5 μmols m⁻²~~
200 ~~s⁻¹ (average ± S.D.) from April 2012 to April 2013. The highest and lowest irradiance were recorded~~
201 ~~in April 2013 and April 2012, respectively.~~

202 ~~Chl-*a* concentrations in the euphotic depth ranged from 0.8 μg L⁻¹ to 14.2 μg L⁻¹ during the~~
203 ~~study period (annual mean ± S.D. = 3.4 μg L⁻¹ ± 2.8 μg L⁻¹; Table 1). There were no significant~~
204 ~~differences of chl-*a* concentrations among 3 light depths and spatial distribution. However, there was~~
205 ~~seasonal variation of chl-*a* concentrations during study period. Chl-*a* concentrations were increased~~
206 ~~from April to August and decreased from August to October in 2012 and increased slightly again in~~
207 ~~January and April 2013.~~

208 **3.2. $\delta^{13}\text{C}$ values and carbon to nitrogen ratios of POM**

209 $\delta^{13}\text{C}$ values of sea surface POM ranged from - 23.1 ‰ to - 16.5 ‰ and the annual average
210 $\delta^{13}\text{C}$ value was -20.9 ‰ (S.D. = ± 3.2 ‰). The annual average carbon to nitrogen (C:N) ratio of POM
211 was 7.0 ± 0.4 (average \pm S.D.), ranging from 6.8 to 7.7 (Table 3).

212 **3.3. Seasonal variation of biochemical composition**

213 The contents of CHO, PRT, and LIP of POM in the water column ~~ranged from were~~ $14.2\text{-}\mu\text{g}$
214 ~~L^{-1} to $412.3\text{ }\mu\text{g L}^{-1}$ ($129.5 \pm 87.2\text{ }\mu\text{g L}^{-1}$), from $22.8\text{-}\mu\text{g L}^{-1}$ to $382.4\text{ }\mu\text{g L}^{-1}$ ($155.0 \pm 73.3\text{ }\mu\text{g L}^{-1}$), and~~
215 ~~from $21.4\text{-}\mu\text{g L}^{-1}$ to $401.4\text{ }\mu\text{g L}^{-1}$ ($154.9 \pm 78.9\text{ }\mu\text{g L}^{-1}$), respectively (Table 4). The FM contents of~~
216 ~~POM ranged from $170.9\text{ }\mu\text{g L}^{-1}$ to $915.7\text{ }\mu\text{g L}^{-1}$ ($435.5 \pm 175.5\text{ }\mu\text{g L}^{-1}$).~~ ~~Since there were no significant~~
217 ~~differences in biochemical concentrations of POM and FM among 3 light depths and spatial~~
218 ~~distributions, On monthly basis, we averaged each biochemical compound and FM from every depths~~
219 ~~and stations (Fig. 2) on monthly basis. The biochemical compositions varied seasonally.~~ The CHO
220 and LIP concentrations increased from April to August and decreased from August to October in 2012.
221 In contrast, the PRT concentrations decreased from April to October in 2012 and increased from
222 October in 2012 to April in 2013. The seasonal pattern of FM concentrations was similar to the
223 pattern of chl-*a* concentrations ($r = -0.36, p < 0.05$, Pearson's Correlation Coefficient).

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224 In order to estimate the biochemical composition as food quality, we obtained relative
225 contributions of each biochemical concentration of POM to FM, based on percentage basis. The
226 biochemical composition of each class (CHO, PRT and LIP) ~~ranged from were~~ 8.3% to 59.1% , ~~from~~
227 ~~6.8% to 74.9% and from 9.4% to 68.3% , respectively (annual mean \pm S.D. of CHO, PRT, and LIP~~
228 ~~composition = $26.4 \pm 9.4\%$, $37.8 \pm 16.1\%$, and $35.7 \pm 13.9\%$, respectively; Table 4).~~ ~~We found the~~
229 ~~seasonal variation of biochemical composition based on monthly basis of biochemical composition~~
230 ~~(Fig. 2). To illustrate these variations of biochemical composition of POM, the highest and lowest~~
231 ~~PRT compositions were in April 2012 and August 2012. In contrast to PRT compositions, the highest~~

232 ~~and lowest LIP compositions were in August 2012 and April 2012. CHO composition was recorded~~
233 ~~high in January 2013, but to compare CHO composition to PRT and LIP composition, CHO~~
234 ~~composition was not strong varied during the study period.~~

235 **3.4. Seasonal variations of the calorific values and contents of FM**

236 The calorific values and contents of FM ~~ranged from were~~ 5.4 ~~Kcal g⁻¹ to~~ 7.9 Kcal g⁻¹
237 (annual mean \pm S.D. = 6.6 Kcal g⁻¹ \pm 0.6 Kcal g⁻¹) and 1.0 ~~Kcal m⁻³ to~~ 6.1 Kcal m⁻³ (annual mean \pm
238 S.D. = 2.8 Kcal m⁻³ \pm 1.1 Kcal m⁻³), respectively (Table 4). The calorific values ~~of FM~~ had no
239 apparent seasonal pattern, whereas the calorific contents ~~of FM~~ had a seasonal pattern similar to the
240 seasonal variation of FM concentrations.

241 **3.5. Relationship between biochemical pools and environmental conditions**

242 Relationships between biochemical pools and environmental conditions were performed
243 using Pearson's correlation matrix. Based on the results, we found a significant, positive relationships
244 between PRT composition and river-input ($r = 0.84$, $p < 0.01$, Table 5, Fig. 3) and PRT composition
245 and dissolved nitrogen concentrations (NH₄ : $r = 0.69$, $p < 0.01$; NO₂+NO₃ : $r = 0.54$, $p < 0.01$, Table
246 5). Lipid composition had an inverse relationships with river-input ($r = -0.63$, $p < 0.01$) and dissolved
247 nitrogen concentrations (NH₄ : $r = -0.59$, $p < 0.01$; NO₂+NO₃ : $r = -0.53$, $p < 0.01$). These
248 relationships led to a significant reverse relationship between PRT composition and LIP composition
249 ($r = -0.81$, $p < 0.01$, Fig. 4). PRT composition was negatively correlated with temperature ($r = -0.52$, p
250 < 0.01), whereas LIP composition was positively correlated with temperature ($r = 0.72$, $p < 0.01$).
251 ~~There were no significant relationships between PRT composition and irradiance and LIP composition~~
252 ~~and irradiance.~~

253 **4. Discussion and conclusion**

254 **4. 1. Environmental conditions and chl-*a* concentration**

255 The annual average chl-*a* concentration during the research period was 3.4 $\mu\text{g L}^{-1}$ (S.D.=
256 $\pm 2.8 \mu\text{g L}^{-1}$) with a range from 0.8 to 14.2 $\mu\text{g L}^{-1}$ which is in a similar range of chl-*a* concentrations
257 reported previously in Gwangyang bay, although it varied across different seasons and sampling
258 depths (Cho et al., 1994; ~~Choi~~ [Choi and Noh et al.](#), 1998; Lee et al., 2001a; Kwon et al., ~~2001~~2002; Jang
259 et al., 2005; Yang et al., 2005; Beak et al., 2011; Min et al., 2011; Beak et al., 2015). Previous studies
260 reported that chl-*a* concentration was influenced mainly by salinity, temperature, and nutrients (nitrate
261 and phosphate) depending on freshwater input from the Seomjin River. Our results in this study were
262 similar to former studies ($r = 0.34$ and -0.41 , $p < 0.05$, $n = 48$ and 28 for salinity and NH_4 ,
263 respectively). However, high chl-*a* concentrations were previously recorded in spring and fall,
264 whereas the highest concentrations were observed in summer (August 2012) from this study. In fact,
265 Baek et al. (2015) reported that high chl-*a* concentrations were found in summer similarly, although
266 there was difference between environmental factors and chl-*a* concentrations as compared with our
267 results. The high levels of chl-*a* were observed with high nutrient concentrations and low salinity
268 levels in the surface water by Baek et al. (2015), whereas the high values existed with low nutrient
269 concentrations and high salinity levels in our results.

270 Despite this dissimilarity of environmental factors with high chl-*a* concentrations, we also
271 found the highest chl-*a* concentrations observed in summer. According to Shaha and Cho (2009),
272 there is a tendency with increasing precipitation and river-input in Gwangyang Bay during summer.
273 This trend could increase loading nutrients from freshwater for maintaining phytoplankton growth in
274 summer. In addition, a strong light intensity during summer could be favorable for phytoplankton
275 growth since our study area was extremely turbid conditions during almost all seasons due to
276 freshwater discharge and a strong spring-neap tidal oscillation. As a result, the combination of these
277 factors is believed to enhance chl-*a* concentration and primary production of phytoplankton during
278 summer in Gwangyang Bay.

279 **4. 2. POM characterization**

280 In general, POM consists of a mixture of living as well as detritus materials (phytoplankton,
281 bacteria, zooplankton, fecal pellets, terrestrial matters, etc.) originating from freshwater and estuarine
282 and marine environments. POM samples can be characterized or determined for source of the major
283 contributor(s). The C:N ratio generally ranges between 6 and 10 for phytoplankton, whereas the ratios
284 are between 3 and 6 for zooplankton and bacteria (Savoye et al, 2003; references therein). For
285 terrestrial organic matters, the C:N ratios are normally over 12 (Savoye et al, 2003; references therein).
286 Therefore, it is useful to classify phytoplankton from heterotrophs and terrestrial materials (Lobbes et
287 al., 2000; Savoye et al., 2003; Lee and Whitledge, 2005). In this study, the mean C:N ratios of POM
288 was 7.0 (S.D. = \pm 0.4), which indicates that this POM is mainly phytoplankton (Table 3). However,
289 the ~~original~~ C:N ratio ~~must~~ ~~can~~ be ~~used with caution~~ ~~changed~~ ~~because of its variation in~~ ~~caused by the~~
290 ~~process of POM degradation~~ ~~biochemical alterations~~ (Savoye et al, 2003). For example, PON is
291 preferentially degraded compared to POC of phytoplankton, which causes an increase of the C:N ratio
292 (Thornton and McManus, 1994; Savoye et al, 2003). ~~In contrast,~~ ~~Terrestrial organic matters~~ ~~(with~~
293 ~~high C:N ratios)~~ colonized by bacteria ~~(with~~ ~~low C:N ratios)~~ ~~could~~ ~~lowers~~ their initial high C:N ratio
294 (Savoye et al, 2003; references therein). Therefore, similar C:N ratios of POM could be produced by
295 degraded phytoplankton and bacteria-colonized terrestrial organic matters (Lancelot and Billen 1985;
296 Savoye et al, 2003). In addition to C:N ratios, $\delta^{13}\text{C}$ of POM can be ~~alternatively~~ used for determining
297 their origin. Kang et al. (2003) reported that the mean $\delta^{13}\text{C}$ signature of phytoplankton in Gwangyang
298 Bay was -20.8 ‰ (S.D. = \pm 1.1 ‰). In this study, our mean $\delta^{13}\text{C}$ signature of POM was -20.9 ‰ (S.D.
299 = \pm 3.2‰), which also indicates that POM was mostly phytoplankton during the study periods (Table
300 3). ~~However, some large contributions of benthic microalgae were seasonally found in our samples~~
301 ~~with relatively higher $\delta^{13}\text{C}$ values on August and October 2012 (Table 3). According to Kang et al.~~
302 ~~(2003), the mean $\delta^{13}\text{C}$ value of benthic microalgae is approximately -14.1 ‰ in Gwangyang Bay.~~
303 Based on our C:N ratio and $\delta^{13}\text{C}$ value in this study, we confirmed that our POM samples were
304 primarily comprised of phytoplankton ~~(seasonally benthic microalgae)~~ in Gwangyang Bay. ~~This is~~
305 ~~interesting that river-derived terrestrial organic matters were not important component of the POM in~~

306 Gwangyang Bay with a large river runoff. Indeed, several previous studies reported a small fraction of
307 terrestrial particulate matter in the same bay as well as in the southeastern coastal bays in Korea
308 (Kang et al., 1993; Lee et al., 2001b; Kwon et al., 2002). Currently, we do not have solid mechanisms
309 for the low contribution of terrestrial organic matters. A further investigation is needed for this
310 paradoxical process.

311 **4. 3. Environmental conditions and biochemical pools**

312 Biochemical pools of POM originating from phytoplankton are influenced by various
313 environmental factors, such as temperature, salinity, nutrients, and light conditions (Morris et al., 1974;
314 Smith and Morris, 1980; Rivkin and Voytek, 1987; Boëchat and Giani, 2008; Cuhel and Lean, 1987;
315 Mock and Kroon, 2002; Khotimchenko and Yakoleva, 2005; Ventura et al., 2008; Sterner et al. 1997).
316 In this study, significant relationships were found between environmental conditions and biochemical
317 pools, especially PRT and LIP (Table 5). Temperature was positively and negatively correlated with
318 LIP and PRT. Previous studies reported that higher temperature stress mainly affects nitrogen
319 metabolism (Kakinuma et al., 2006) which is related to significant decrease of PRT with increases of
320 LIP and CHO content (Tomaselli et al., 1988; Oliveira et al., 1999). In a high temperature-stressed
321 condition of phytoplankton, the decrease in PRT content is related to breakdown of protein structure
322 and interference with enzyme regulators (Pirt, 1975), whereas LIP is predominant because LIP is more
323 closely associated with cell structure such as thickened cell walls (Smith et al., 1989; Kakinuma et al.,
324 2001, 2006). Our results are in agreement with other works, as described above.

325 The relationships between nutrients and biochemical pools could be explained by nutrient
326 limitation and the characteristics of each biochemical compound. A combination of nutrient
327 concentrations and ratios can be used to assess nutrient limitation (Dortch and Whitlege, 1992; Justić
328 et al., 1995). Dortch and Whitlege (1992) suggested that nutrient limitations are existed in the
329 Mississippi river plume and Gulf of Mexico, if the dissolved inorganic phosphorus (DIP), dissolved
330 inorganic nitrogen (DIN), and dissolved silicon (DSi) concentrations in water column are less than 0.2,

331 1.0 and 2.0 μM , respectively, depending on the half-saturation constant (K_s) that the threshold value is
332 required for the uptake and growth of phytoplankton (Eppley et al., 1969; Fisher et al 1988).- In
333 addition, molar ratios of the DIN:DIP ~~and~~, DSi:DIN and DSi:DIP ~~–~~can be indicators of nutritional
334 status and the physiological behavior of phytoplankton (Redfield et al., 1963; Goldman et al., 1979;
335 Elrifi and Turpin, 1985; Dorch and Whiteledge 1992; Roelke et al. 1999). According to Dortch and
336 Whitledge (1992), the following criteria of their molar ratios were (a) DIN:DIP ratio < 10 and
337 DSi:DIN ratio > 1 for nitrogen (N) limitation; (b) DIN:DIP ratio > 30 and DSi:DIP ratio > 3 for
338 phosphorus (P) limitation; (c) DSi:DIN ratio < 1 and DSi:DIP ratio < 3 for silicate (Si) limitation. In
339 this study, nutrient limitation conditions were observed by absolute nutrient concentrations or/and
340 their molar ratios depending on seasons (Table 6). Previous studies of biochemical composition in
341 relation to nutrient limitation reported that PRT production of phytoplankton was enhanced under
342 abundant N conditions (Fabiano et al., 1993; Lee et al., 2009). In contrast, LIP production and storage
343 were dominant (Shifrin and Chisholm, 1981; Harrison et al., 1990) and PRT contents decreased
344 (Kilham et al., 1997; Lynn et al., 2000; Heraud et al., 2005) under N-depleted conditions. High LIP
345 contents have also been detected in phytoplankton under P or/and Si limitation (Lombardi and
346 Wangersky, 1991; Lynn et al. 2000; Heraud et al., 2005; Sigee et al., 2007). Under N or P-limited
347 conditions, triglyceride content (energy storage) increases and shifts from PRT to LIP metabolism
348 since proteins are nitrogenous compounds whereas LIP and CHO are non-nitrogenous substrates
349 (Lombardi and Wangersky, 1991; Smith et al., 1997; Takagi et al., 2000). In our study, Si and P
350 concentrations may not significantly impact on biochemical composition of phytoplankton. Si
351 concentrations were almost above 2.0 μM except in April 2013 during the study period. P limitation
352 was observed based on the absolute concentration and molar ratios during study period. However,
353 under P limitation, phytoplankton can relocate the cellular P pool to maintain their P requirements for
354 the maximum growth rate (Cembella et al., 1984; Ji and Sherrell, 2008). In this respect, we suggest
355 that DIN could be significantly impact on biochemical composition of phytoplankton in our study
356 area. DIN was initially believed to be the most important limiting factor for phytoplankton growth in

357 marine ecosystems (Ryther and Dunstan, 1971; Howarth, 1988). In fact, DIN was strongly positively
358 correlated with PRT composition, whereas it was negatively correlated with LIP composition. The
359 most of DIN loading came from freshwater input of the Seomjin River (Table 5, river-input vs NH₄
360 and NO₂+NO₃; r = 0.91 and 0.55, p < 0.01, respectively) influences on PRT and LIP synthesis and
361 subsequently macromolecular composition of phytoplankton. As a result, the amount of river-input
362 was also strongly correlated with PRT composition (Table 5 and Fig. 3). Therefore, DIN is an
363 important controlling factor for biochemical composition, especially PRT and LIP composition of
364 phytoplankton in Gwangyang bay.

365 Although irradiance is also known for an important governing factor for biochemical
366 composition, irradiance was not ~~significantly-statistically~~ correlated with biochemical pools in this
367 study (Table 5). We measured ~~irradiance-PAR~~ during our short incubation time (4~5h) for
368 phytoplankton productivity as a parallel study. Since ~~T~~this short time of measured irradiance can be
369 largely variable by a local weather, it might be not enough to reflect and detect the change of
370 biochemical composition in phytoplankton with irradiance. The irradiance between April 2012 and
371 April 2013 was largely different (approximately 10 times lower in April 2012 than in April 2012;
372 Table 1). Increasing synthesis of proteins is found as light intensity decrease because a relatively
373 lower irradiance saturation level is required for protein synthesis than that of other biochemical
374 components (Lee et al., 2009; Suárez and Marañón, 2003; Morris et al., 1974, 1978). Consistently, the
375 protein compositions were significantly higher in April 2012 than in April 2013 (t-test, p < 0.01; Fig.
376 2) in this study. The proteins accounted approximately 62 % and 37 % of biochemical compositions in
377 April 2012 and April 2013, respectively. However, the main reason for no consistent relationships
378 between irradiance and biochemical components along seasons might be the PAR measurements as
379 discussed previously in this study.

380 The structure and composition of phytoplankton assemblages and species could have a
381 significant influence on the seasonal variation of biochemical composition. Although we did not

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382 conduct a study of phytoplankton community structure, there is seasonal succession of phytoplankton
383 community structure in the bay. Previous studies showed that the dominant phytoplankton community
384 was diatoms and dominant diatom species were *Skeletonema spp.* during summer and winter in
385 Gwangyang bay (Choi et al., 1998; Baek et al., 2015). Kim et al. (2009) also reported that diatom and
386 dinoflagellate communities have experienced a considerable change because of increased nutrient
387 loadings from both domestic sewage and industrial pollution during summer. Therefore, the seasonal
388 change of phytoplankton species composition and community structure could lead to determining
389 different biochemical pools on seasonal basis.

390 However, other studies in different regions reported that environmental conditions, such as
391 temperature, nutrients and irradiance are more important controlling factors in biochemical
392 composition than variation of phytoplankton community and species composition (Lindqvist and
393 Lingnell, 1997; Suárez and Marañón, 2003). In this study, we also concluded that DIN from river-
394 input was a primary governing factor for the seasonal variation of biochemical composition of
395 phytoplankton in Gwangyang Bay as discussed above.

396 **4.4. Total FM and energy content of POM in a global context**

397 The annual average of FM was $434.5 \mu\text{g L}^{-1}$ (S.D. = $\pm 175.5 \mu\text{g L}^{-1}$) in this study. Since there
398 were no comparable data available in South Korea, we compared our results with other regions (Table
399 7), although they were conducted in different seasons and sampling depths. PRT contents in this study
400 were as high as in the Ross Sea (Fabiano and Pusceddu, 1998; Fabiano et al., 1999a), the Amundsen
401 Sea (Kim et al., [20152016](#)) and the Humboldt Current System (Isla et al., 2010). A similar range of
402 LIP contents was observed in Bedford Basin (Mayzaud et al., 1989), Yaldad Bay (Navarro et al., 1993)
403 and the Humboldt Current System (Isla et al., 2010). CHO contents were comparatively higher in this
404 study than other studies except Bedford Basin (Mayzaud et al., 1989) and Yaldad Bay (Navarro et al.,
405 1993). One of the highlights is that the calorific contents of FM [in this study](#) were generally higher
406 than those of other areas except several regions. The FM values were comparatively higher than other

407 | regions such as the northern Chuekchi Sea (Kim et al., [20142015](#); Yun et al., [20142015](#)), Ross Sea
408 | (Fabiano et al., 1996; Fabiano and Pusceddu, 1998; Fabiano et al., 1999a; Pusceddu et al., 1999),
409 | Amundsen Sea (Kim et al., [20152016](#)) and the northern part of the East/Japan Sea (Kang et al.,
410 | unpublished) or similar to the Humboldt Current System which is known as an important spawning
411 | sites for pelagic fishes and the highest abundance of anchovy eggs (Isla et al., 2010). Actually, the
412 | southern coastal sea (including our study area) in Korea represents calm seas, an indented coastline,
413 | and numerous bays, which have high diversities of habitat for fishes and shellfishes (Kwak et al.,
414 | 2012) and give a favorable condition for mariculture (Kwon et al., 2004). The high quantity of FM
415 | and the calorific contents of POM found in this study reflected good nutritive conditions of primary
416 | food materials mainly provided by phytoplankton for the maintenance of productive shellfish and fish
417 | populations in Gwangyang bay.

418 | **5. Summary and Conclusion**

419 | ~~This study is the first report that was investigated the biochemical composition of POM on~~
420 | ~~seasonal basis in Gwangyang Bay, South Korea and we determined major controlling factors for~~
421 | ~~biochemical composition which is influenced by various environmental factors (Morris et al., 1974;~~
422 | ~~Smith and Morris, 1980; Rivkin and Voytek, 1987; Boëchat and Giani, 2008; Cuhel and Lean, 1987;~~
423 | ~~Mock and Kroon, 2002; Khotimchenko and Yakoleva, 2005; Ventura et al., 2008; Sterner et al. 1997).~~
424 | ~~Among different factors, temperature was positively correlated with LIP whereas negatively~~
425 | ~~correlated with PRT in this study (Table 5), which is consistent with previous works. In addition, we~~
426 | ~~found that PRT and LIP compositions were strongly correlated with DIN loading largely depending on~~
427 | ~~the amount of river input from the Seomjin river which influences on PRT and LIP synthesis and~~
428 | ~~subsequently macromolecular composition of phytoplankton in Gwangyang bay. The concentrations~~
429 | ~~and the calorific contents of FM found in this study were relatively higher in comparison to previous~~
430 | ~~studies in various regions, which reflecting that POM (mainly from phytoplankton) provides a good~~
431 | ~~nutritive condition to maintain this highly productive estuarine ecosystem in Gwangyang bay.~~

432 ~~Recently, significant environmental perturbations in their watersheds and externally from~~
433 ~~climatic forcings have been reported in various estuaries (Wetz and Yoskowitz, 2013). More intense~~
434 ~~but less frequent tropical cyclones are expected over the coming century (e.g., Elsner et al., 2008;~~
435 ~~Knutson et al., 2010) and many changes in drought and flood cycles have been proceeding globally~~
436 ~~(e.g., Min et al., 2011; Pall et al., 2011; Trenberth and Fasullo, 2012; Trenberth, 2012). The~~
437 ~~cumulative effects of these perturbations could alter the quantity and quality of biochemical~~
438 ~~composition of POM and cause subsequent changes in ecosystem structure and trophic dynamics in~~
439 ~~estuaries (Cloern, 2001; Paerl et al., 2006; Rabalais et al., 2009; Wetz and Yoskowitz, 2013).~~
440 ~~Therefore, continuous field measurements and observations on biochemical composition of POM as~~
441 ~~food quality are needed to monitor for better understanding future response of marine ecosystem on~~
442 ~~potential environmental perturbations in Gwangyang Bay.~~

443

444 **Acknowledgements**

445 ~~We thank the anonymous reviewers who greatly improved an earlier version of the~~
446 ~~manuscript.~~ This research was supported by "Long-term change of structure and function in marine
447 ecosystems of Korea" funded by the Ministry of Oceans and Fisheries, Korea.

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736 **Table captions**

737 Table 1. Environmental factors and chl-*a* concentrations in Gwangyang bay during the research period

738 (- : no data).

739 Table 2. Monthly patterns of Rrainfall and river input.

740 Table 3. $\delta^{13}\text{C}$ values and C:N ratios of POM at surface in Gwangyang bay ~~(surface)~~.

741 Table 4. Biochemical concentrations and composition, calorific values and contents in Gwangyang

742 bay (- : no data).

743 Table 5. Significant correlation coefficient (*r*) among proteins (PRT), lipids (LIP) and environmental

744 factors (ns ; no significance, **: $p < 0.01$). River-inputs were integrated from 20 days prior to

745 our sampling dates.

746 Table 6. Observed nutrient limitations during the study period.

747 Table 7. Comparison of biochemical quantity of POM, FM and the calorific contents.

748 **Figure captions**

749 Fig. 1. Sampling location in Gwangyang bay, Korea ; Maps of Korea (a), Southern Coastal Sea (b)
750 and main sampling stations (c).

751 Fig. 2. Seasonal variation of biochemical composition in Gwangyang bay.

752 Fig. 3. The positive relationship between river-input and protein composition. River-inputs were
753 integrated from 20 days prior to our sampling dates.

754 Fig. 4. The inverse relationship between lipid compositions and protein compositions.

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Table 1. Environmental factors and chl-*a* concentrations in Gwangyang Bay during the research period (- : no data).

Year	Date	Irradiance ($\mu\text{mols m}^{-2} \text{s}^{-1}$)	Station	Light depth (%)	Temperature ($^{\circ}\text{C}$)	Salinity (‰)	Depth (m)	NH ₄ (μM)	NO ₂ +NO ₃ (μM)	SiO ₂ (μM)	PO ₄ (μM)	Chl- <i>a</i> ($\mu\text{g L}^{-1}$)	
2012	April	167.9 \pm 133.5 (average \pm S.D.)	St.1	100	13.9	14.5	0	3.6	56.4	26.0	80.9	1.89	
				30	13.3	25.6	1	-	-	-	-	1.95	
				1	13.5	28.0	3	2.4	16.0	9.8	0.2	2.08	
			St.4	100	15.0	24.4	0	2.6	15.1	16.3	0.2	1.81	
				30	13.6	31.4	1	-	-	-	-	-	
				1	12.3	32.9	5	1.9	2.1	2.1	0.1	2.03	
			St.5	100	12.6	31.7	0	3.1	9.5	7.1	0.3	2.07	
				30	12.3	31.6	1	-	-	-	-	-	
				1	12.2	32.4	5	3.0	6.4	5.1	0.3	2.04	
			June	1158.1 \pm 627.6	St.2A	100	22.9	27.6	0	-	-	-	-
	30	22.8				27.6	1	-	-	-	-	-	0.76
	1	22.9				28.7	3	-	-	-	-	-	0.76
	St.4	100			23.6	31.5	0	-	-	-	-	-	1.00
		30			22.6	31.9	3	-	-	-	-	-	1.67
		1			22.1	32.3	11	-	-	-	-	-	1.02
	August	1320.0 \pm 316.9	St.4	100	25.8	30.6	0	0.1	0.1	10.6	0.1	8.11	
				30	25.7	31.6	2	-	-	-	-	-	8.49
				1	25.7	31.7	8	0.1	0.1	11.9	0.1	5.99	
			St.5	100	25.6	31.6	0	0.7	0.3	8.2	0.0	14.20	
				30	26.1	31.5	2	-	-	-	-	-	9.85
1				25.7	31.7	8	0.1	0.1	10.1	0.1	3.19		
October	-	St.2A	100	20.6	29.8	0	1.4	3.0	11.3	0.1	1.07		
			30	20.5	29.8	1	-	-	-	-	-	1.30	
			1	21.9	30.2	3	1.3	1.3	8.1	0.1	1.24		
		St.4	100	20.9	30.3	0	1.6	3.1	14.0	0.1	2.69		
			30	20.7	30.3	1	-	-	-	-	-	2.93	
			1	20.6	30.6	5	1.1	0.6	7.4	0.1	1.74		
		St.5	100	19.1	30.4	0	1.0	0.4	6.5	0.1	2.47		
			30	18.5	30.5	2	-	-	-	-	-	1.98	
			1	18.1	30.4	8	1.2	0.2	5.3	0.0	2.20		

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Table 1. (continued)

Year	Date	Irradiance ($\mu\text{mol s}^{-2} \text{ m}^{-2} \text{ s}^{-1}$)	Station	Light depth (%)	Temperature ($^{\circ}\text{C}$)	Salinity (‰)	Depth (m)	NH ₄ (μM)	NO ₂ +NO ₃ (μM)	SiO ₂ (μM)	PO ₄ (μM)	Chl-a ($\mu\text{g L}^{-1}$)
2013	January	297.4 ± 310.5	St.2A	100	5.5	20.5	0	0.5	4.2	4.0	0.1	1.39
				30	7.0	28.0	1	-	-	-	-	1.52
				1	7.3	29.4	4	0.5	3.7	3.6	0.1	1.48
			St.4	100	7.7	31.1	0	1.0	3.8	3.4	0.1	2.79
				30	7.4	31.3	4	-	-	-	-	3.41
				1	7.3	32.8	12	0.6	3.1	2.5	0.0	5.37
			St.5	100	6.3	31.8	0	0.8	3.3	2.6	0.1	5.79
				30	6.6	31.9	3	-	-	-	-	5.25
				1	6.4	32.5	11	1.0	3.0	3.6	0.2	5.33
	April	1593.3 ± 414.5	St.2A	100	14.3	26.2	0	1.9	3.7	3.1	0.1	1.81
				30	14.4	27.5	1	-	-	-	-	1.72
				1	14.3	29.1	3	1.5	2.5	2.3	0.1	2.06
			St.4	100	14.7	32.0	0	1.6	2.0	2.5	0.1	2.24
				30	15.3	32.0	1	-	-	-	-	4.41
			1	15.2	32.6	5	1.5	1.7	1.6	0.1	7.39	
St.5	100	16.1	31.9	0	1.1	1.3	1.3	0.1	4.39			
	30	16.1	32.0	3	-	-	-	-	5.22			
	1	16.6	32.3	11	1.1	0.7	1.0	0.1	5.90			

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Table 2. Monthly rainfall and river input.

Year	Date	Rainfall (mm)	River input (10 ⁶ t)
2012	April	195.5	149.4
	May	44.4	148.9
	June	69.6	42.3
	July	235.8	223.3
	August	559.0	228.9
	September	360.1	447.2
	October	38.0	98.5
	November	52.5	83.4
	December	96.7	89.4
2013	January	15.6	79.3
	February	116.4	94.6
	March	79.9	91.5
	April	99.1	100.3

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Table 3. $\delta^{13}\text{C}$ values and C:N ratios of POM in Gwangyang Bay (surface).

Year	Date	$\delta^{13}\text{C}$ (‰)	C:N (molar:molar)
2012	April	-22.8 ± 2.9	7.0 ± 1.2
	June	-23.1 ± 1.3	6.8 ± 0.2
	August	-16.5 ± 2.4	6.7 ± 0.5
	October	-17.1 ± 0.9	6.9 ± 0.6
2013	January	-22.5 ± 0.6	7.7 ± 0.6
	April	-23.1 ± 0.2	6.8 ± 0.7
(average \pm S.D.)		-20.9 ± 3.2	7.0 ± 0.4

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Table 4. Biochemical concentrations and composition, calorific values and contents in Gwangyang Bay (- : no data).

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Year	Date	Station	Light depth (%)	CHO ($\mu\text{g L}^{-1}$)	PRT ($\mu\text{g L}^{-1}$)	LIP ($\mu\text{g L}^{-1}$)	FM ($\mu\text{g L}^{-1}$)	CHO/FM (%)	PRT/FM (%)	LIP/FM (%)	Kcal g ⁻¹	Kcal m ⁻³
2012	April	St.1	100	45.0	144.2	22.9	212.1	21.2	68.0	10.8	5.6	1.2
			30	53.1	218.6	51.9	323.6	16.4	67.6	16.0	5.9	1.9
			1	53.1	220.4	84.2	357.6	14.8	61.6	23.5	6.2	2.2
		St.4	100	14.2	128.1	28.6	170.9	8.3	74.9	16.7	6.1	1.0
			30	50.0	155.1	21.4	226.5	22.1	68.5	9.4	5.6	1.3
			1	20.2	146.0	37.3	203.5	9.9	71.8	18.3	6.1	1.2
		St.5	100	60.2	198.0	143.0	401.2	15.0	49.3	35.7	6.7	2.7
			30	132.4	198.0	42.8	373.2	35.5	53.1	11.5	5.5	2.0
			1	146.7	265.3	210.0	622.1	23.6	42.7	33.8	6.5	4.1
	June	St.2A	100	170.7	99.7	233.5	503.8	33.9	19.8	46.3	6.9	3.5
			30	135.5	108.0	251.9	495.4	27.3	21.8	50.9	7.2	3.5
			1	163.5	85.0	225.1	473.7	34.5	17.9	47.5	6.9	3.3
		St.4	100	99.1	44.6	199.5	343.2	28.9	13.0	58.1	7.4	2.5
			30	133.4	142.4	203.5	479.3	27.8	29.7	42.4	6.8	3.3
			1	91.6	110.8	232.3	434.6	21.1	25.5	53.5	7.3	3.2
	August	St.4	100	69.3	73.9	213.5	356.7	19.4	20.7	59.9	7.6	2.7
			30	61.2	56.5	173.8	291.5	21.0	19.4	59.6	7.6	2.2
			1	127.2	77.9	162.2	367.3	34.6	21.2	44.2	6.8	2.5
		St.5	100	155.5	289.4	204.7	649.6	23.9	44.6	31.5	6.4	4.2
			30	412.3	102.0	401.4	915.7	45.0	11.1	43.8	6.6	6.1
			1	83.3	22.8	228.3	334.4	24.9	6.8	68.3	7.9	2.6
	October	St.2A	100	71.0	82.2	104.1	257.3	27.6	32.0	40.5	6.7	1.7
			30	42.7	62.4	100.3	205.4	20.8	30.4	48.8	7.2	1.5
			1	74.3	111.6	98.5	284.4	26.1	39.2	34.6	6.5	1.9
St.4		100	51.6	105.2	105.3	262.2	19.7	40.1	40.2	6.8	1.8	
		30	119.4	121.9	144.4	385.6	31.0	31.6	37.4	6.6	2.5	
		1	78.5	169.0	134.4	381.9	20.6	44.2	35.2	6.6	2.5	
St.5		100	37.2	70.0	86.5	193.6	19.2	36.1	44.7	7.0	1.4	
		30	42.3	92.5	112.0	246.7	17.2	37.5	45.4	7.1	1.7	
		1	33.9	108.4	97.3	239.7	14.2	45.2	40.6	6.9	1.7	

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Table 4. (continued)

Year	Date	Station	Light depth (%)	CHO ($\mu\text{g L}^{-1}$)	PRT ($\mu\text{g L}^{-1}$)	LIP ($\mu\text{g L}^{-1}$)	FM ($\mu\text{g L}^{-1}$)	CHO/FM (%)	PRT/FM (%)	LIP/FM (%)	Kcal g^{-1}	Kcal m^{-3}	
2013	January	St.2A	100	150.3	139.3	115.5	405.2	37.1	34.4	28.5	6.1	2.5	
			30	347.0	131.1	109.2	587.3	59.1	22.3	18.6	5.4	3.2	
			1	331.3	127.1	-	-	-	-	-	-	-	-
		St.4	100	171.6	164.0	-	-	-	-	-	-	-	-
			30	183.5	168.7	139.7	491.9	37.3	34.3	28.4	6.1	3.0	
			1	115.9	182.3	107.1	405.2	28.6	45.0	26.4	6.2	2.5	
		St.5	100	113.6	212.0	133.4	459.0	24.7	46.2	29.1	6.3	2.9	
			30	264.1	204.8	120.5	589.4	44.8	34.8	20.4	5.7	3.4	
			1	99.3	195.5	104.2	399.0	24.9	49.0	26.1	6.2	2.5	
	April	St.2A	100	237.7	262.9	189.9	690.5	34.4	38.1	27.5	6.1	4.2	
			30	185.5	308.0	198.7	692.3	26.8	44.5	28.7	6.3	4.3	
			1	274.8	382.4	180.3	837.5	32.8	45.7	21.5	5.9	4.9	
		St.4	100	115.0	141.9	181.4	438.4	26.2	32.4	41.4	6.8	3.0	
			30	116.4	187.0	191.0	494.5	23.5	37.8	38.6	6.7	3.3	
			1	205.2	222.1	185.7	612.9	33.5	36.2	30.3	6.2	3.8	
St.5	100	160.4	176.3	289.1	625.7	25.6	28.2	46.2	7.0	4.4			
	30	146.9	217.8	253.3	618.0	23.8	35.2	41.0	6.8	4.2			
	1	171.3	204.9	272.6	648.8	26.4	31.6	42.0	6.8	4.4			

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Table 5. Significant correlation coefficient (r) among proteins (PRT), lipids (LIP) and environmental factors (ns ; no significance, **, $p < 0.01$).

Variables	r	p	n
%PRT × Temp.	- 0.52	**	46
%LIP × Temp.	0.72	**	46
%PRT × NH ₄	0.69	**	28
%LIP × NH ₄	-0.59	**	28
%PRT × NO ₂ +NO ₃	0.54	**	28
%LIP × NO ₂ +NO ₃	-0.53	**	28
%PRT × River-input	0.84	**	46
%LIP × River-input	-0.63	**	46
NH ₄ × River-input	0.91	**	28
NO ₂ +NO ₃ × River-input	0.55	**	28
%PRT × %LIP	-0.81	**	46
%PRT × Irradiance	-0.22	ns	39
%LIP × Irradiance	0.24	ns	39

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Table 6. Observed nutrient limitations during the study period (nd ; not detected).

Year	Date	Based on absolute concentrations (μM)				Based on molar ratios			
		DIN	SiO_2	PO_4	Limitation	DIN:DIP	DSi:DIP	DSi:DIN	Limitation
2012	April	20.3 ± 20.2	11.1 ± 8.8	13.6 ± 32.9	nd	56.8 ± 45.5	37.5 ± 36.9	0.6 ± 0.2	P
	June	-	-	-	-	-	-	-	-
	August	0.4 ± 0.4	10.2 ± 1.5	0.1 ± 0.0	N, P	9.8 ± 14.2	173.4 ± 56.5	42.7 ± 23.7	N
	October	2.7 ± 1.5	8.8 ± 3.3	0.1 ± 0.0	P	40.4 ± 20.8	142.2 ± 74.0	3.6 ± 0.8	P
2013	January	4.2 ± 0.4	3.3 ± 0.6	0.1 ± 0.1	P	69.5 ± 63.1	50.6 ± 41.4	0.8 ± 0.1	P
	April	3.4 ± 1.3	2.0 ± 0.8	0.1 ± 0.0	Si, P	27.1 ± 8.9	15.5 ± 5.5	0.6 ± 0.1	nd

Table 7. Comparison of biochemical quantity of POM, FM and the calorific contents.

	Regions (depth)	PRT ($\mu\text{g l}^{-1}$)	LIP ($\mu\text{g l}^{-1}$)	CHO ($\mu\text{g l}^{-1}$)	FM ($\mu\text{g l}^{-1}$)	Kcal m^{-3} (average \pm S.D.)	Authors
Arctic regions	Gwangyang Bay, South Korea (Euphotic depth)	23-382	21-401	14-412	171-916	2.8 \pm 1.1	This study
	Bedford Basin, Canada(2.5 m)	200-650	130-440	160-630	660-1570		Mayzaud et al. (1989)
	Logy Bay, Newfoundland (6 m)	80-740	20-75	8-120	130-1030	2.7 \pm 2.8	Navarro & Thompson (1995)
	The Northern Chukchi Sea, 2011 (Euphotic depth)	1-86	50-105	22-147	94-246	1.0 \pm 0.2	Kim et al. (2014)
	The Northern Chukchi Sea, 2012 (Euphotic depth)	9-183	37-147	16-253	90-373	1.2 \pm 0.2	Yun et al. (2014)
Antarctic regions	Pacific Sector Antarctic Ocean (0-1500 m)	14-100	3-60	3-66	25-220		Tanoue (1985)
	Off Princess Astrid Coast, Antarctica (0-100m)	24-200	15-174	22-147	148-393		Dhargalkar et al. (1996)
	Ross Sea, Antarctica (10m)	11-402	91	91-187	193-680	2.6 \pm 1.8	Fabiano and Pusceddu (1998)
	Ross Sea, Antarctica (0-200 m)	40-406	18-115	22-251	110-660		Fabiano et al. (1999a)
	Terra Nova Bay, Antarctica (0-750 m)	10-620	2-77	8-144	19-885	1.3 \pm 1.0	Fabiano et al. (1996)
	Terra Nova Bay, Antarctica (under pack ice)	96-201	38-112	10-68	145-382	1.7 \pm 1.1	Pusceddu et al. (1999)
	Amundsen Sea (Euphotic depth)	6-396	13-37	3-216	43-639	1.2 \pm 0.8	Kim et al. (2015)
	W-Mediterranean (0-200 m)	72-105	37-51	33-88	143-246		Fabiano et al. (1984)
Other regions	W-Mediterranean submarine cave (10m)	4-77	4-104	1-75	15-220	0.4 \pm 0.2	Fichez (1991)
	Mediterranean seagrass (4 m)	25-135	50-180	40-110	125-395		Danovaro et al. (1998)
	Ligurian Sea (10 m) NW-Mediterranean	32-107	21-140	21-131	74-378	1.5 \pm 1.4	Danovaro & Fabiano (1997)
	Mediterranean (30m)	70-90	90-110	10-20	177-213	1.4 \pm 0.2	Modica et al. (2006)
	Cretan Sea (0-1500 m)	7-92	4-63	13-149	54-200	0.6 \pm 0.2	Danovaro et al. (2000)
	Bay of Biscay, 2000 (0-30m)	109-2426	26-2037	2-345	961 (a.v.)	6.7 \pm 5.0	Díaz et al. (2007)
	Yaldad Bay, Chile (10 cm a.b.)	300-2250	30-560	50-1050	3310-2960	10.0 \pm 10.9	Navarro et al. (1993)
	The Humboldt current system, Northern Chile (5-89m)	40-470	60-390	70-510	24-1282	3.5 \pm 3.3	Isla et al. (2010)
	Magellan Strait (0-50m)	60-150	30-70	20-40	110-256	1.0 \pm 0.5	Fabiano et al. (1999b)
	The northern part of the East Sea (Euphotic depth)	28-425	12-180	19-206	109-810	1.5 \pm 0.6	Kang et al. (unpublished)

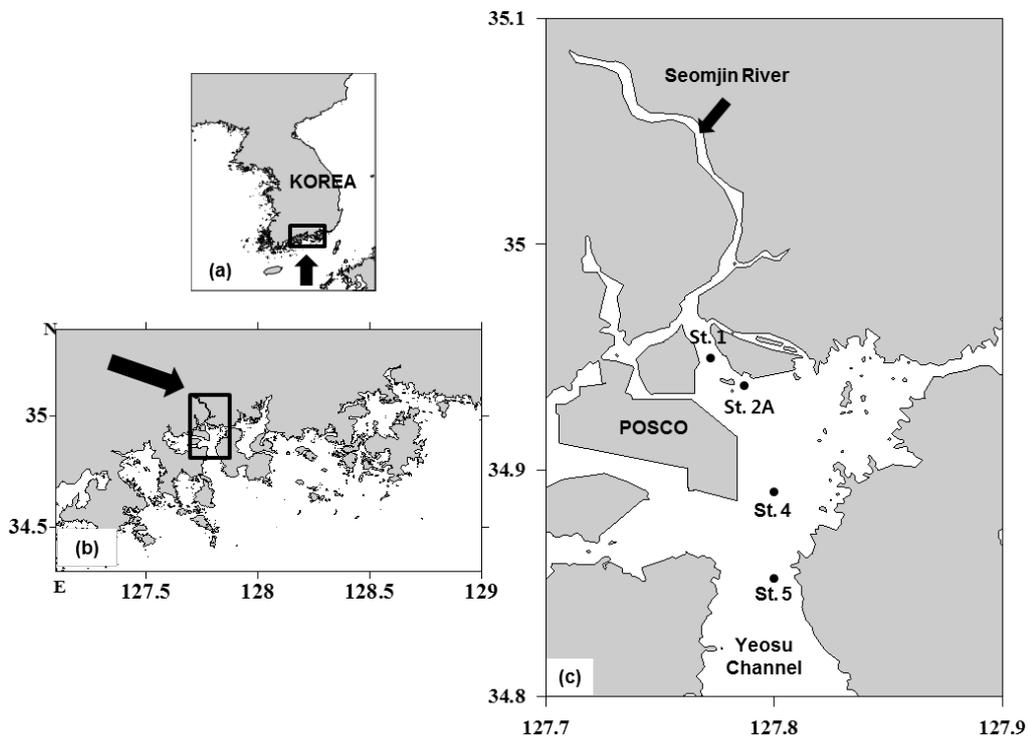


Fig. 1. Sampling location in Gwangyang Bay, Korea ; Maps of Korea (a), Southern Coastal Sea (b) and main sampling stations (c).

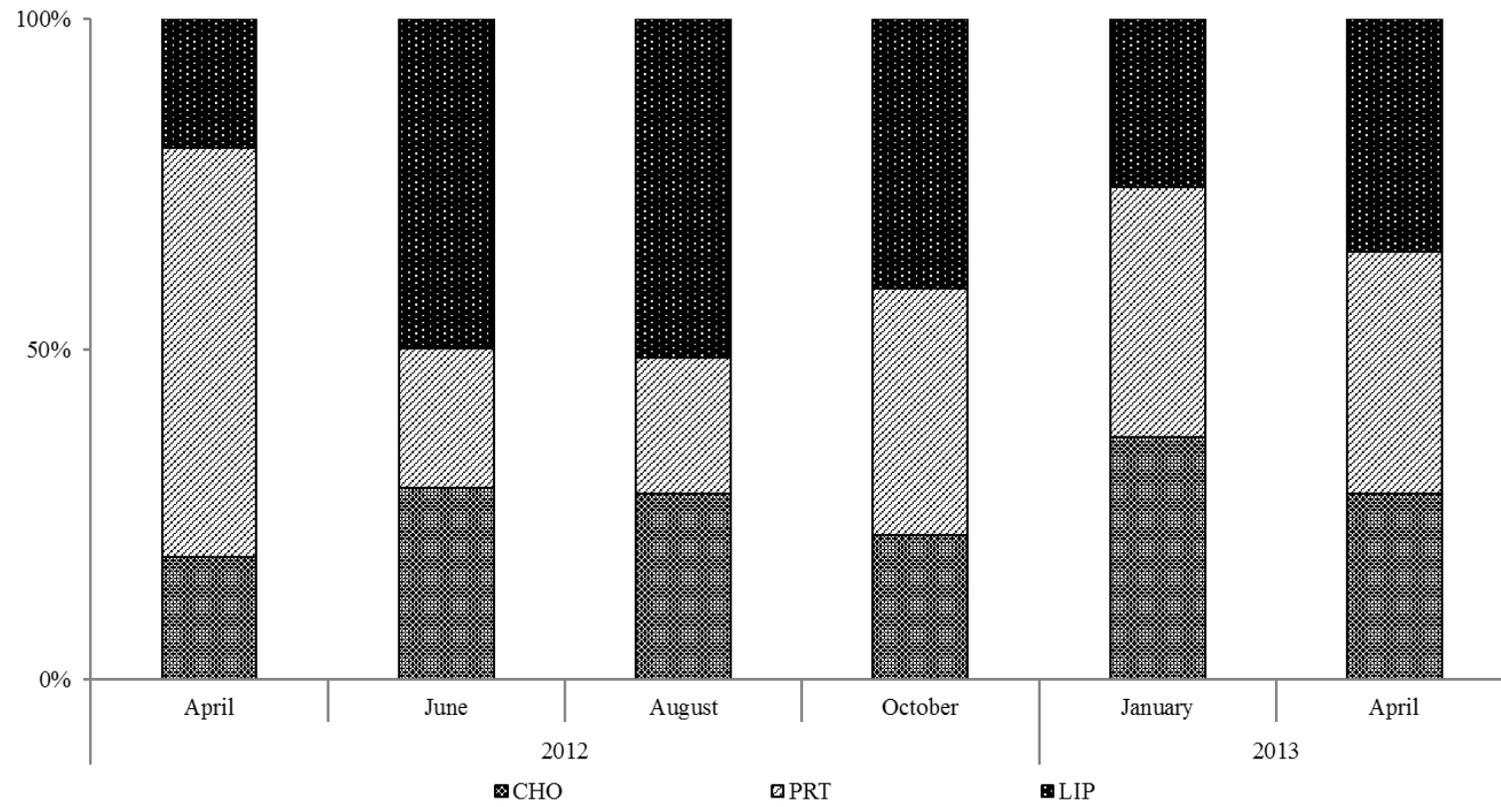


Fig. 2. Seasonal variation of biochemical composition in Gwangyang Bay.

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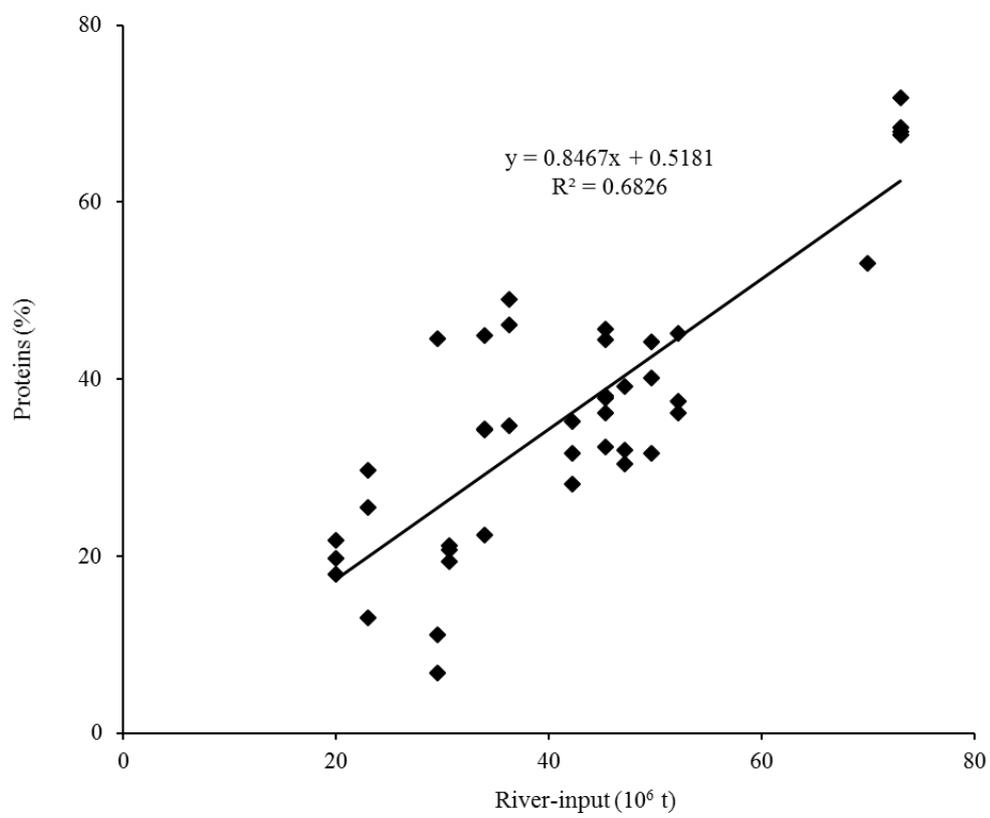


Fig. 3. The positive relationship between river-input and protein composition.

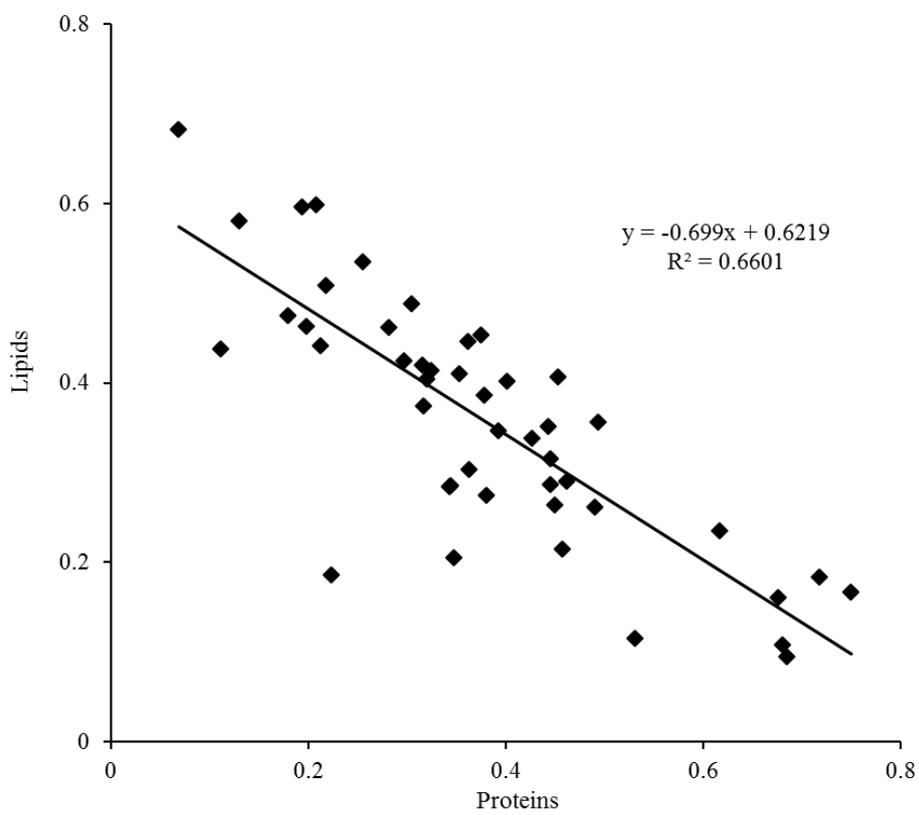


Fig. 4. The inverse relationship between lipid compositions and protein compositions.