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Interactive comment on "Nitrous oxide and methane in two tropical estuaries in a peat-dominated region of North-western Borneo" by D. Müller et al.

D. Müller et al.

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Received and published: 23 March 2016

We thank Anonymous Referee 1 for his helpful comments and suggestions. Our detailed answers can be found below.

This paper describes CH4 and N2O distributions in two tropical estuaries. Current literature for estuarine CH4 and N2O concentrations is still limited, and this type of study is significant for our scientific understanding and relevant to BG. The paper is well written and reads easily. However, there are several issues that need to be addressed prior to publication.

1) The dataset of CH4 and N2O concentrations does not cover the region of low

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salinity (0-5), where high CH4 and N2O might be expected. Furthermore, no sample was taken at the river-end member. Hence it is hard to take a full picture of trace gas variation in the whole estuary system.

The correlation of N_2O with salinity during the dry season indicates that the freshwater end-member might indeed exhibit the highest N_2O during the dry season. We calculated the expected river end-member using the correlation of N_2O with salinity during the dry season and found concentrations of 9.1 nM, (Lupar, r = 0.5), 9.3 nM (Saribas, r = 0.8) and 15 nM (Saribas tributary, r = 0.9). For CH_4 , this is a bit harder to do, as we did not observe a correlation between methane and salinity. While we agree that the lack of data in the low salinity region deserves a more thorough discussion, which we will include in the revised manuscript in sections 4.2 and 4.3, our main conclusions can be maintained despite the lack of data for the upper estuaries:

- 1) Eutrophication did not lead to enhanced N_2O . Data from the river end-member would not provide additional information, as eutrophication was not observed in the river end-member.
- 2) DIN was a poor predictor of N_2O . This was mainly inferred from Figure 2b), a river end-member data point is not likely to change this overall observation.
- 3) Postulation of additional N_2O and CH_4 sources during the wet season. This conclusion is based mainly on the observation of high N_2O and CH_4 values at salinities 10-20. Of course, it would be extremely interesting to see if a similar observation could be made in the freshwater region.

2) Page 5, line 13: Does the DO sensor calibrated with Winkler titration method? The FDO 925 sensor was calibrated by the manufacturer (WTW, Germany). According to the manufacturer, user calibration is not required for the specified lifetime of the sensor. Nevertheless, a routine function check was performed in water vapor saturated air, using the check and calibration vessel (FDO (R) check) that was provided with the sensor. This information will be added in the revised manuscript.

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3) Page 6, line 25: I think the authors should give a little more detail about the k600 they used. For example, were they measured in situ or calculated by widely used wind-speed related relationships in the literature?

The k600 that we used were derived from CO_2 fluxes measured with a floating chamber in 2014. The measurements are described in detail in Müller et al., 2015a. From simultaneous measurements of the CO_2 flux and the water and air pCO_2 , we derived kCO_2 and, ultimately, k600 for these estuaries. In the revised manuscript, we will change the following sentence:

For k, we used k600 values that were reported for the Lupar and Saribas estuaries in Müller et al. (2015a).

to

For k, we used k600 values that were derived for the Lupar and Saribas estuaries using the floating chamber method (Müller et al., 2015a). Floating chamber measurements were conducted at several locations along the estuaries during the wet season campaign and averaged over the spatial extent of the individual estuaries. We argued in Müller et al. (2015a) that the k600 values determined in this way are more appropriate than commonly used wind speed parameterizations, which neglect the influence of tidal currents and the water flow velocity.

4) Table 3 showed that dissolved CH4 in the Lupar estuary was low and undersaturated (2nM and 88%). What's possible reasons for this?

This seems to be an artifact. The reasons why we think so are the following: 1.) There are only two datapoints directly following each other which indicate undersaturation. 2.) With the exception of CO, all gas concentrations that we could retrieve from the respective spectra indicate undersaturation or atmospheric equilibrium values (N_2O , CH_4 and CO_2). This indicates that atmospheric air was in the measurement cell. 3.) No measurements were taken during the 20 minutes preceding the undersaturation values. A background measurement (empty cell) had been performed before that, and

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the cell was filled with atmospheric air afterwards before the measurements of equilibrator air continued. If we reconsider these facts, it seems likely that atmospheric air was still in the cell. We would therefore exclude the two data points from the analysis. The numbers will be corrected in the revised manuscript.

5) Table 4 didn't show k600 for dry season, and the authors should explain what k600 was used for the calculation for dry season. k600 for Saribas tributary and Lupar estuary are almost twice of that for Saribas. Do the authors have any idea of the reason?

We used the same k600 for the dry and wet season. In the revised manuscript, we will add a more detailed description of the k600 values used in the methods section (see comment 3).

We discussed the variability of the k600 values in Müller et al., 2015a and after adding more detailed information in the methods section, the reader will be referred to that publication. There, we reasoned that the strong currents in the Lupar estuary are responsible for the relatively higher k600. The higher gas exchange velocity in the Saribas tributary if compared to the Saribas is consistent with the notion that the gas exchange velocity decreases with increasing stream order (Raymond et al., 2012).

6) Figure 1: Scales should be added and the South China Sea should be located on the map.

The Figure will be revised accordingly.

7) Figure 2: N2O vs salinity, it was shown that there are great N2O peaks during wet seasons between the areas with salinity of 12-15, suggesting a significant N2O source. The authors should discuss this in the text.

On page 10, lines 23-27 of the discussion paper, we tried to combine several lines of evidence as to where this N_2O comes from. Our argument is as follows. 1) There are sources in the estuary (i.e., salinities 10-20) in the wet season which we did not observe in the dry season. 2) Another feature of N_2O that we observed in the wet but not in the dry season was the co-variation of N_2O with CH_4 . 3) In the wet season,

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both N_2O and CH_4 varied with the tides, with higher concentrations during low tide. We inferred from 1) that there are additional sources of N_2O in the wet season, from 2) that they are sources of both N_2O and CH_4 and from 3) that either tidal creeks or the estuarine sediments constitute this additional source. We will modify the sections 4.2 and 4.3 in the revised manuscript in order to provide a more specific and detailed discussion.

References

Müller et al. (2015a): Fate of peat-derived carbon and associated CO2 and CO emissions from two Southeast Asian estuaries. Biogeosciences Discussions 12:8299-8340, doi: 10.5194/bgd-12-8299-2015

Raymond et al. (2012): Scaling the gas transfer velocity and hydraulic geometry in streams and small rivers. Limnology and Oceanography: Fluids and Environments 2:41-53, doi: 10.1215/21573689-1597669

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We thank Anonymous Referee 2 for the positive evaluation of our manuscript. Our replies to the detailed comments can be found below.

This paper presents new information on seasonal differences in N2O and CH4 fluxes in mangrove environments. It furthers our understanding of the role of natural factors such as salinity, DO and DOC on greenhouse gas production in a generally sparsely-sampled region. There are a few minor issues to address:

Spelling/Grammar: This paper is very well written, there were only a few issues that I could find.

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1. Page 8, line 10: "N2O was correlated with DOC (Fig. 2a), whereas this correlation was strong..." I think this should be "N2O was correlated with DOC (Fig. 2a), *and* this correlation was strong..." or just "N2O was correlated with DOC (Fig. 2a); this correlation was strong..."

This will be corrected in the revised manuscript (suggestion 2).

2. Page 9, like 17: "...the Lupar and Saribas rivers are no blackwater rivers..." Is this meant to say "...are *not* blackwater rivers..." It's technically correct either way, but the first phrasing sounds more colloquial.

We will change this to the more formal phrasing in the revised manuscript.

3. Page 10, line 15: "Either, a source of N2O exists on the continental shelf..." This comma is unnecessary.

The comma will removed from the sentence in the revised manuscript.

Other Comments:

1. This paper refers to k-value calculations derived from floating chamber experiments (covered in an earlier paper, Müller et al., 2015). It might be a good idea to make it clear in the methods section that these k values are from floating chambers, to distinguish this k calculation from the more common technique of estimating k-values from equations well-known in the literature.

We agree, we will replace the following sentence:

For k, we used k600 values that were reported for the Lupar and Saribas estuaries in Müller et al. (2015a).

with

For k, we used k600 values that were derived for the Lupar and Saribas estuaries using the floating chamber method (Müller et al., 2015a). Floating chamber measurements were conducted at several locations along the estuaries during the wet season campaign and averaged over the spatial extent of the individual estuaries. We argued in Müller et al. (2015a) that the k600 values determined in this way are more appropri-

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ate than commonly used wind speed parameterizations, which neglect the influence of tidal currents and the water flow velocity.

- 2. On a related point, gas transfer velocities can be temporally and spatially heterogeneous. Were the floating chamber measurements made near the field site? Were they made upstream, downstream, or along the length of the estuaries? A brief mention of the location or timing of the floating chamber measurements might give a better idea of the precision of this approach to k-value calculation. We will add this information in the methods section as indicated in the reply to your comment above.
- 3. The authors thoroughly document the source of the atmospheric mixing ratios of N2O and CH4 (which are needed in order to calculate water-to-air fluxes). However, from what I understand, the N2O flux totals are likely going to be much more sensitive to the choice of k-value. Is there some reason why the k-value would be particularly sensitive to the atmospheric mixing value (say if it were 325.25 ppb rather than 325.15 ppb)? Considering that the local air mixing ratio could be slightly greater or less than the Mauna Loa value, it might be good to mention that this isn't a large source of error in the calculation.

The k600 values were derived from CO_2 fluxes measured with a floating chamber and simultaneous measurements of the pCO_2 in water and in the air. That is, when we determined the k600 in the first place, we used atmospheric concentrations measured at the site and not mixing ratios determined elsewhere. Therefore, for the k600 values themselves, the error for $pCO2^{air}$ is not significantly larger than the instrument's uncertainty. Nevertheless, as you mentioned in comment 2 as well, the k600 can be temporally and spatially variable. Therefore, with respect to the calculated N_2O and CH_4 fluxes, k600 still causes the largest uncertainty. We did a simple error propagation calculation on the average values, assuming an error of 60% for k, 0.5% for K_0 , 2% for pCH_4^{water} and 0.2% for pN_2O^{water} (disequilibrium error, Johnson 1999) and 1% for pCH_4 and pN_2O in air. As a result, the uncertainty in k accounts for 96% of the flux un-

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certainty. Therefore, we think our approach of reporting the flux with the % uncertainty of k600 is justified. We will change the caption of Table 4 to

k600 values and median N_2O and CH_4 areal and total fluxes from the Lupar, Saribas and the Saribas tributary. The uncertainties relate to the maximum variability of the k600 value, as the k600 uncertainty propagation was responsible for approximately 96% of the flux uncertainty. Values are given for the spatial extent of the rivers that was covered in 2013.

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List of relevant changes in the manuscript

Pages and line numbers refer to the marked-up version of the manuscript.

- Methods: Details about the oxygen sensor calibration (p 5, line 14-15) and the k600 values used (p.6, lines 27-31) were added.
- Results: Since the apparent undersaturation was an artifact, the two datapoints were excluded from the analysis. As a consequence, a few numbers were corrected in Tables 2,3 and 4 and in the text (p. 8, 1. 3-4 & 22, p.9, 1. 17-18).
- Discussion: A few discussion points were extended or made more explicit. This includes the additional N2O sources in the wet season (p. 10, 1.25-26, p.11, l. 1-10), the lack of data for the upper estuary (p.10, 1.30-p.11, l. 1 and 22-23) and the explanation for the low CH4 values (p. 11,1. 25-26).
- References: A few references were added to support the discussion (Abril et al. 2007, Barnes and Owens 1998, de Angelis and Scranton 1993, de Wilde and de Bie 2000, Robinson et al. 1998). One study was updated (Müller et al., 2015a: previously a discussion paper, now finally published and referred to as Müller et al. 2016).
- Tables: The caption of Table 4 was changed, numbers in the Tables were corrected as indicated above.
- Figures: The map (Fig 1) was modified according to Referee 1's suggestions; all other Figures were slightly modified (two datapoints were removed as indicated above).
- Grammar, phrasing and spelling: Changes were made in p. 2 l. 16, p. 7 l. 19, p. 88, l. 11 & 15, p. 9. l. 29, p 10, l. 25-26, p 27 l. 5

Nitrous oxide and methane in two tropical estuaries in a peatdominated region of North-western Borneo

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Abstract. Estuaries are sources of nitrous oxide (N2O) and methane (CH4) to the atmosphere. However, our present knowledge of N₂O and CH₄ emissions from estuaries in the tropics is very limited because data is scarce. In this study, we present first measurements of dissolved N₂O and CH₄ from two estuaries in a peat-dominated region of north-western Borneo. Two campaigns (during the dry season in June 2013 and during the wet season in March 2014) were conducted in the estuaries of the rivers Lupar and Saribas. Median N₂O concentrations ranged between 7.2 and 12.3 nmol L⁻¹ and were higher in the marine end-member (13.0 ± 7.0 nmol L⁻¹). CH₄ concentrations were low in the coastal ocean (3.6 ± 0.2 nmol L⁻¹) 1) and higher in the estuaries (medians between 12.2 and 64.0 nmol L-1). The respiration of abundant organic matter and presumably anthropogenic input caused a slight eutrophication, which did not lead to hypoxia or enhanced N2O concentrations, however. Generally, N₂O concentrations were not related to dissolved inorganic nitrogen concentrations. Thus, the use of an emission factor for the calculation of N₂O emissions from the inorganic nitrogen load leads to an overestimation of the flux from the Lupar and Saribas estuaries. N₂O was negatively correlated with salinity during the dry season, which suggests a riverine source. In contrast, N₂O concentrations during the wet season were not correlated with salinity but locally enhanced within the estuaries, implying that there were additional estuarine sources during the wet (i.e. monsoon) season. Estuarine CH₄ distributions were not driven by freshwater input but rather by tidal variations. Both N₂O and CH₄ concentrations were more variable during the wet season. We infer that the wet season dominates the variability of the N₂O and CH₄ concentrations and subsequent emissions from tropical estuaries. Thus, we speculate that any changes in the Southeast Asian monsoon system will lead to changes in the N2O and CH4 emissions from these systems. We also suggest that the ongoing cultivation of peat soil in Borneo is likely to increase N₂O emissions from these estuaries, while the effect on CH4 remains uncertain.

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1. Introduction

Nitrous oxide (N_2O) and methane (CH_4) are greenhouse gases whose global warming potentials exceed that of carbon dioxide (CO_2) by far (a factor of 265 for N_2O and 28 for CH_4 on a 100 year time horizon, Myhre et al., 2013). Thus, an assessment of the natural and anthropogenic sources and sinks as well as the formation pathways of N_2O and CH_4 is essential to understand present Earth's climate variability and to predict its future development. The world's oceans, including its coastal zones, are sources of N_2O and CH_4 to the atmosphere and play a major role in the global budget of atmospheric N_2O , but only a minor role in the global budget of atmospheric CH_4 (Ciais et al., 2013). Rivers and estuaries are considered hotspots for the production and emission of both N_2O and CH_4 (Bange, 2006; Bastviken et al., 2011; Borges et al., 2015; Murray et al., 2015; Seitzinger and Kroeze, 1998).

In aquatic systems, N_2O is mainly formed as a byproduct during nitrification ($NH_4^+ \rightarrow NO_3^-$) and to minor extent as an intermediate during denitrification ($NO_3^- \rightarrow N_2O \rightarrow N_2$) (Freing et al., 2010). In both processes, the yield of N_2O strongly depends on the concentration of ambient oxygen (O_2) (e.g. Bange, 2008). Both nitrification and denitrification are microbial processes and occur in the water column and in sediments (e.g. Bange, 2008). CH_4 is formed during the decomposition of organic material by microbial methanogenesis (e.g. Reeburgh, 2007; Valentine, 2011; Ferry, 2010). Since CH_4 formation requires strictly anaerobic conditions, it is produced in anoxic environments as found in sediments, in the interior of suspended particles or in zooplankton guts (see e.g. Reeburgh, 2007; Valentine, 2011). Additionally, CH_4 is oxidized under aerobic and anaerobic conditions in the water column and in the sediments, respectively (e.g., Valentine, 2011). On the continental shelf so-called geological CH_4 can be released directly to the water column through mud volcanoes, via groundwater input or seeping at pockmark structures (e.g., Valentine, 2011). Alternative aerobic CH_4 formation pathways in the nutrient poor (oligotrophic) surface ocean have been suggested (see e.g. Karl et al., 2008; Damm et al., 2010; Zindler et al., 2013), but they are negligible in nutrient rich (eutrophic) coastal systems.

In this study, we present first measurements of dissolved N₂O and CH₄ from the estuaries of the rivers Lupar and Saribas, which are located in north-western Borneo (Sarawak, Malaysia, Southeast Asia). These measurements are complemented by dissolved N₂O and CH₄ measurements from a cruise on the adjacent coastal shelf of the South China Sea. The Lupar and Saribas rivers drain catchments which are covered by tropical peatlands, parts of which have been converted to oil palm plantations or other crops. This study aimed at investigating the effect of the carbon-rich peatlands on the N₂O and CH₄ estuarine distributions and emissions and the potential impact of estuarine eutrophication. Two intensive sampling campaigns took place during the dry and wet seasons in June 2013 and March 2014, respectively, and provide a first conception of the seasonality in these systems.

2. Materials and Methods

2.1 Study Area

Our study was conducted in Sarawak, the largest Malaysian state, which is located in the north of the island of Borneo (see Fig. 1a). About 12 % of the area of Sarawak is covered by peatlands (Chai, 2005), approximately 41 % of which have been converted to oil palm plantations (SarVision, 2011). Sarawak has a tropical climate with a mean annual air temperature of 26.1°C in the capital Kuching and average (1961-1990) annual rainfall of 4101 mm yr⁻¹ (DWD, 2007). Between November and February, Sarawak experiences enhanced rainfall due to the Northeast monsoon, while June and July are the driest months (DWD, 2007).

Two macrotidal estuaries which enclose Malaysia's largest peat dome, the Maludam peninsula, are in the focus of this study. The catchment areas of the Lupar and Saribas rivers are 6558 km² and 1943 km², respectively (Lehner et al., 2006), with approximate discharges of 490 m³ s⁻¹ and 160 m³ s⁻¹ (Müller et al., 20165a). The Lupar catchment comprises mainly the division of Sri Aman, which has a population of approximately 95,000. The Saribas basin belongs largely to the Betong division, with a population of approximately 110,000 (SPU, 2012). Major settlements along the Lupar River and estuary are Sri Aman, Lingga and Sebuyau (Fig. 1b). Important settlements along the Saribas River and estuary are Betong, Pusa and Beladin. Oil palms are being cultivated in both catchments (Fig. 1b).

In the Lupar and Saribas estuaries, sampling took place during two campaigns in June 2013 and March 2014. Our sampling strategy is described in detail in Müller et al. (20165a). In that study, we showed that precipitation during June 2013 and March 2014 did not deviate much from the historical average, so that we considered the measurements in June 2013 representative of the dry season and those in March 2014 representative of the wet season. Therefore, we refer to the two campaigns as "MLD dry" (June 2013) and "MLD wet" (March 2014). Note that the peak of the monsoon season is between December and February, so the monsoon recedes in March, which is why our statements about seasonality are conservative. We complement this dataset with measurements that were performed during the cruise SO218 – SHIVA with the German research vessel (R/V) *Sonne* (15-29 November 2011) (Quack and Krüger, 2013). The cruise started in Singapore and went along the Sarawakian coastline to end in Manila, Philippines (see Fig. 1a). For this study, we selected those samples that were taken in the coastal ocean off north-western Sarawak (latitude <2.5, 110.5<longitude<111.5) in order to complement our dataset from the Lupar and Saribas estuaries.

2.2 Water chemistry

In the Lupar and Saribas inner and outer estuaries, we sampled 20 stations during the dry season and 23 during the wet season. Water samples were taken from approximately 1 m below the water surface. Samples were taken for dissolved organic carbon (DOC), dissolved inorganic (nitrogenous) nutrients (DIN = NO_3 , NO_2 , and NH_4), salinity, water temperature and O_2 :

- Samples for DOC were filtered (pore size 0.45 μm) and acidified with 21 % phosphoric acid until the pH had dropped below 2. DOC concentrations were determined by high temperature combustion and subsequent measurement of resultant CO₂ with a non-dispersive infrared detector (NDIR) (EPA method 415.1). Details are given in Müller et al. (20165a).
- Inorganic nutrient samples were filtered through a Whatman glass microfibre filter (pore size 0.7 μm), preserved
 with a mercuric chloride (HgCl₂) solution and frozen until analysis. Concentrations of NO₃-, NO₂- and NH₄+ were
 determined spectrophotometrically (Hansen and Koroleff, 1999) at a wavelength of 540 nm with a Continuous Flow
 Analyzer (Alliance, Austria).
- 3. Salinity and temperature were measured with a CastAway CTD at each station during the MLD cruises. Additionally, we measured conductivity using a TetraCon 925 conductivity sensor (WTW, Germany). We converted conductivity to salinity using the equations from Bennett (1976). During SO218, salinity and temperature were measured continuously with a thermosalinograph on board.
- 4. Dissolved oxygen (DO) was measured using a Multi3420 with an FDO 925 oxygen sensor (WTW, Germany). A routine function check in water vapor saturated air was conducted prior to the measurements using the check and calibration vessel that was provided with the sensor.

2.3 N2O and CH4 measurements

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Our setup during the MLD cruises is described in detail in Müller et al. (20165a). Surface water was pumped through a Weiss equilibrator (Johnson, 1999) at a rate of approximately 20 L min⁻¹. The headspace air was analyzed using an in-situ Fourier Transform InfraRed (FTIR) trace gas analyzer (University of Wollongong, Australia). This instrument allows for the continuous and simultaneous measurements of several trace gas species, such as N₂O and CH₄ as well as CO₂ and CO (Müller et al., 20165a) with high accuracy and precision over a wide range of concentrations (Griffith et al., 2012). Spectra were averaged over 5 min and dry air mole fractions were retrieved using the software MALT5 (Griffith, 1996). The gas dry air mole fractions were corrected for pressure, water and temperature cross-sensitivities (Hammer et al., 2013). The CO₂ and CO data obtained from these measurements have been reported by Müller et al., (20165a). Here, we present the N₂O and CH₄ measurements. Calibration was achieved by measuring a suite of gravimetrically prepared gas mixtures (Deuste Steininger) ranging from 324 to 3976 ppb N₂O and 1.8 to 239 ppm CH₄ in synthetic air, which were calibrated against the World Meteorological Organization (WMO) reference scale at the Max-Planck-Institute for Biogeochemistry in Jena, Germany. The error associated with the FTIR retrieval is usually small. In ambient air, the total uncertainties reported by Hammer et al. (2013) are 0.084 nmol mol⁻¹ for N₂O and 0.25 nmol mol⁻¹ for CH₄, corresponding to approximately 0.01 % and 0.03 %. The larger source of uncertainty is a potentially remaining disequilibrium between water and headspace in the equilibrator, which can cause an error of <0.2 % for N₂O and 2 % for CH₄ (Johnson, 1999).

Water temperature in the equilibrator and in the water as well as ambient air temperature and pressure were monitored as described in Müller et al. (201 $\underline{65a}$). N₂O and CH₄ partial pressures were calculated. Since the sample air was dried before

entering the FTIR analyzer, we corrected for the removal of water vapor (Dickson et al., 2007). N_2O molar concentrations were calculated from N_2O fugacity and solubility (K_0) according to Weiss and Price (1980). CH_4 molar concentrations were calculated from CH_4 partial pressure and solubilities were derived from the equations given by Wiesenburg and Guinasso (1979).

5 During the R/V Sonne cruise SO218, surface sea water was continuously supplied from the ship's hydrographic shaft (moon pool) using a submersible pump at about 4 m water depth. N₂O and CH₄ samples were taken in triplicates, preserved with HgCl₂ and analyzed in the lab using headspace equilibration and gas chromatography. Details about the analytic procedures can be found in Walter et al. (2006) and Bange et al. (2010). The average of the three samples was calculated and data was discarded if the standard deviation exceeded 30 % of the average value. For N₂O, two additional data points were taken from 0 the surface N₂O concentrations determined in depth profiles.

Atmospheric mixing ratios of N₂O were taken from the Mauna Loa (Hawaii) monitoring station of the NOAA/ESRL halocarbons in situ program (Dutton et al., 2015) which was the nearest atmospheric N₂O monitoring station in the northern hemisphere. Atmospheric N₂O in Mauna Loa averaged 325.15 ppb in November 2011, 326.26 ppb in June 2013 and 327.08 ppb in March 2014. Atmospheric CH₄ was derived from the NOAA/ESRL GMD Carbon Cycle Cooperative Global Air Sampling Network (Dlugokencky et al., 2014). The nearest atmospheric CH₄ monitoring station was Bukit Kototabang, Indonesia. Unfortunately, CH₄ data from NOAA/ESRL were not available for March 2014, so we estimated the atmospheric CH₄ during that month from the value reported for March 2013 and an annual growth rate of 4 ppb between 2004-2013 at Bukit Kototabang. Atmospheric CH₄ averaged 1841.64 ppb in November 2011, 1798.64 ppb in June 2013 and a value of 1879.35 ppb was derived for March 2014.

0 2.4 Flux estimation

In order to calculate N_2O and CH_4 flux densities F (in nmol m⁻² s⁻¹) across the water-air interface, we used the thin film model, which reads

$$F = kK_0(pGas_{water} - pGas_{air})f, (1)$$

where k is the gas exchange velocity (m s⁻¹), K_o ist the solubility of N₂O and CH₄ in seawater (mol L⁻¹ atm⁻¹, see Sect. 2.3), $pGas_{water}$ is the partial pressure derived from the equilibrator measurements (natm), $pGas_{air}$ is the partial pressure of the gas in air (natm) as measured at the atmospheric monitoring stations (see Sect. 2.3), and f is a conversion factor from L⁻¹ to m⁻³. For k, we used k_{600} values that were derived for the Lupar and Saribas estuaries using the floating chamber method (Müller et al., 2016). Floating chamber measurements were conducted at several locations along the estuaries during the wet season campaign and averaged over the spatial extent of the individual estuaries. We argued in Müller et al. (2016) that the k_{600} values determined in this way are more appropriate than commonly used wind speed parameterizations, which neglect the influence of tidal currents and the water flow velocity. For k, we used k_{600} values that were reported for the Lupar and Saribas estuaries in Müller et al. (2015a). In-situ k was calculated based on the Schmidt numbers of N₂O and CH₄, which relates the kinematic viscosity to the diffusivity of the gas in water. Kinematic viscosity was calculated according to Siedler

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and Peters (1986), the diffusivity of N₂O was computed using Eq. 2 in Bange et al. (2001) and the diffusivity of CH₄ was calculated with the formula given in Jähne et al. (1987). The annual areal flux density was estimated as the average of the wet and dry season values (for the spatial extent that was covered during the 2013 cruise, see Müller et al., 20165a). The total N₂O and CH₄ fluxes (t N yr⁻¹ and t C yr⁻¹) were calculated assuming an estuarine surface area of 220 km² for the Lupar and 102 km² for the Saribas estuary (Müller et al., 20165a). For N₂O, we compare this estimate to one derived using the DIN export and the emission factor for estuaries suggested in Mosier et al., (1998) (0.0025 kg N₂O-N / kg N leaching and runoff). The DIN export was calculated from river discharge (see Sect. 2.1) and the median DIN concentration.

3. Results

3.1 Water chemistry

- In the estuaries of Lupar and Saribas, salinity ranged from 0-30.6 in the dry season and from 0-31.0 in the wet season. N₂O and CH₄ concentration data are available for salinities of 4.3-26.5 (MLD dry) and 6.9-26.4 (MLD wet). For the coastal ocean off north-western Sarawak, N₂O measurements covered salinities between 31.3 and 32.7 and CH₄ measurements between 32.2 and 32.7. DO saturation in the estuaries ranged between 63.6 to 94.6 % (MLD dry) and 79.0-100.4 % (MLD wet) (Müller et al., 20165a).
- 15 The total DIN concentrations were already published in Müller et al. (20165a). Here, we report the concentrations of the different inorganic nitrogen species. Generally, DIN concentrations were quite low, but locally enhanced. NO₃ ranged between 6.3 μmol L⁻¹ and 36.3 μmol L⁻¹ in the dry season and between 2.8 and 17.9 μmol L⁻¹ in the wet season. A maximum value of 84.0 μmol L⁻¹ was observed during MLD dry approximately 20 km offshore in north-western direction from Sebuyau. There, an influence from the Lupar river plume, but possibly also from the close-by Sadong Reviver was detected (salinity = 22.2).
 - NO_2^- concentrations ranged between 0.1 μ mol L^{-1} and 0.6 μ mol L^{-1} in the dry season and between <0.1 μ mol L^{-1} and 2.3 μ mol L^{-1} in the wet season. NH_4^+ ranged between <0.1 μ mol L^{-1} and 2.6 μ mol L^{-1} in the dry season, whereas a maximum value of 8.0 μ mol L^{-1} was observed at one station in the Lupar estuary. In the wet season, NH_4^+ was higher, ranging between 0.2 μ mol L^{-1} to 7.8 μ mol L^{-1} . However, overall, DIN concentrations were higher in the dry season than in the wet season (Müller et al., $\frac{2015a}{2016}$, Table 1).
 - On the Lupar River, we determined average DIN concentrations of $5.1~\mu$ mol L⁻¹ (2013) and $5.3~\mu$ mol L⁻¹ (2014) upstream of the town of Sri Aman, which can be considered unpolluted. This value is 2-6 times lower than the DIN concentrations in the Lupar estuary. On the Saribas River, a slightly enhanced value of $18.6~\mu$ mol L⁻¹ (2013) was measured at salinity 0 outside the town of Betong, so it cannot be considered unpolluted, yet it was lower than the DIN concentrations in the estuary during that season (Table 1). We estimated that together, Lupar and Saribas deliver 6086 t N as DIN to the South China Sea every year (Table 1).

3.2 N₂O

Dissolved N₂O concentrations in the water ranged from 6.9 nmol L⁻¹ to 13.4 nmol L⁻¹ during MLD dry in June 2013 (corresponding to saturations of 103 to 184 %) and from 6.3-6 nmol L⁻¹ to 116.8 nmol L⁻¹ during MLD wet in March 2014 (93-102 to 1679 % saturation). Most of the time, N₂O was close to atmospheric equilibrium. However, local enhancements were observed: During both the dry and the wet season, the Saribas tributary exhibited markedly higher N₂O concentrations than the Saribas main river (Fig. 3a,b, Table 2). During the wet season, N₂O concentrations were more variable (higher standard deviation (SD)) and higher maximum concentrations were observed (Table 2). Although the mean N₂O concentrations in the Saribas estuary and Saribas tributary were higher in the wet season as well, no difference was found with regards to the medians (Table 2).

- Salinity and N_2O concentrations were correlated in the dry season (r = -0.62), but uncorrelated in the wet season (Fig. 2c). By the town of Sebuyau at the Lupar river mouth, where the Sebuyau Reiver flows into the South China Sea, N_2O was enhanced with concentrations of up to 116.8 nmol L^{-1} in the wet season (not shown in Fig. 2.3.4). The data from SO218 revealed enhanced N_2O concentrations offshore during November 2011 (see Fig. 2c), with a median of 13.0 \pm 7.0 nmol L^{-1} (218 \pm 119 % saturation, see Table 2).
- N₂O was correlated with DOC (Fig. 2a)₂, whereas this correlation was strong in the dry season (r=0.87) and weak during the wet season (r=0.38). N₂O concentrations were not correlated with DIN (Fig. 2b), NO₃, NH₄ (not shown) or DO (Fig. 2d). During the dry season, we found no link between N₂O concentrations and tidal variations (Fig. 4a). In the wet season, however, N₂O exhibited slightly higher concentrations during low tide and its spatiotemporal variation resembled that of CH₄ (Fig. 4b).

20 3.3 CH₄

Dissolved CH₄ concentrations ranged from 5.2 nmol L⁻¹ to 59 nmol L⁻¹ during MLD dry in 2013 (228 to 2782 % saturation) and from 2.03.7 nmol L⁻¹ to 135 nmol L⁻¹ during MLD wet in 2014 (88-168 to 6003 % saturation) in the Lupar and Saribas estuaries and was spatially variable. The highest value was measured by the town of Sebuyau during MLD wet, whereas the highest median concentrations were detected in the Saribas estuary during the dry season and in the Saribas tributary during the wet season (Fig. 3c,d; Table 3). In general, no seasonal pattern could be identified. Mean and median CH₄ concentrations in the Lupar estuary were comparable in the dry and wet season. In the Saribas estuary, CH₄ concentrations were higher during the dry season but on the Saribas tributary, they were higher during the wet season (Table 3). However, higher maximum concentrations were observed in the wet season in both estuaries and the tributary (Table 3).

CH₄ concentrations were not correlated with salinity (Fig. 5c). A relatively low marine end-member concentration was determined during the SO218 cruise: With a median of 3.6 ± 0.2 nmol L⁻¹ (176 ± 9 % saturation), CH₄ concentrations in the coastal ocean were only slightly enhanced in comparison to the atmospheric equilibrium concentration (see Table 3).

CH₄ was not correlated with DOC (Fig. 5a), DO (Fig. 5d), DIN or suspended particulate matter (not shown), but increased with increasing pCO₂ (Fig. 5b). This relationship was stronger in the wet season (r=0.55) than in the dry season (r=0.15). One striking feature was that CH₄ showed a strong response to the tides (Fig. 4). This is visible for most of the data, even though we changed location during the measurements. Tidal and spatial variations are overlapping in Fig. 4, but the tidal variation seems to dominate. This is confirmed by one stationary measurement that we conducted over night at one station on the Saribas tributary in 2014 (Fig. 4b): CH₄ peaked 4-fold during low tide if compared to the concentration level at high tide.

3.4 N₂O and CH₄ flux densities

Median N_2O and CH_4 flux densities are listed in Table 4. Both the highest N_2O flux density and the highest CH_4 flux density were computed for the Saribas tributary. During both seasons, N_2O flux densities from the Saribas tributary were up to one order of magnitude higher than from the Lupar and the Saribas estuaries. Annual average N_2O flux densities were low for both Lupar and Saribas, amounting to 1.3 ± 0.3 mmol m^{-2} yr⁻¹ and 1.9 ± 1.6 mmol m^{-2} yr⁻¹, respectively. The N_2O flux density from the Saribas tributary was one order of magnitude higher (12.0 ± 7.5 mmol m^{-2} yr⁻¹).

The CH₄ flux density observed on the Saribas tributary during the wet season was approximately five times higher than the flux densities from Lupar or Saribas during any season. As a result, the annual average flux from the Saribas tributary (89.2 \pm 55.2 mmol m⁻² yr⁻¹) was approximately four times as high as those computed for the Lupar and Saribas estuaries, which were comparable (22.4- \pm 5.4 mmol m⁻² yr⁻¹ and 23.0 \pm 19.1 mmol m⁻² yr⁻¹, respectively).

In total, 4 ± 1 t N₂O-N yr⁻¹ and $\frac{59}{60} \pm 17$ t CH₄-C yr⁻¹ were emitted from the Lupar estuary and 3 ± 2 t N₂O-N yr⁻¹ and 28 ± 25 t CH₄-C yr⁻¹ from the Saribas (Table 4). Using the emission factor of Mosier et al., (1998), we obtained 11 t N₂O-N yr⁻¹ for the Lupar and 4 t N₂O-N yr⁻¹ for the Saribas estuary.

4. Discussion

4.1 Eutrophication in the Lupar and Saribas estuaries

Blackwater rivers and their estuaries usually have very low nutrient concentrations (<1 μ mol L⁻¹ NO₃⁻, Kselik and Liong 2004; <5 μ mol L⁻¹ NO₃⁻, Alkhatib et al., 2007). Although the Lupar and Saribas rivers are not blackwater rivers, they have several blackwater tributaries (Kselik and Liong 2004; Müller et al., 2015a2016), and 30.5 % and 35.5 % of their catchments are covered by peat (Müller et al., 2015a2016). Therefore, rather low nutrient concentrations were expected. However, several villages and smaller towns are found along the shore of both estuaries, and there is cultivation of sago and oil palm in the catchments (Fig. 1b).

Estuarine DIN concentrations were higher than in the unpolluted freshwater end-member of the Lupar Rriver, indicating that the estuary was indeed slightly eutrophic during the time of our measurements. This eutrophication can be attributed both to the release of DIN during respiration of organic matter, which was shown to be pronounced in the Lupar and Saribas

estuaries (Müller et al., 2015a2016), especially in the dry season (see discussion below), and to anthropogenic input. These processes were also identified as important sources of inorganic nutrients in the Siak River, a eutrophic blackwater river in central Sumatra, Indonesia (Baum and Rixen, 2014). The DIN concentrations in the Lupar and Saribas estuaries were similar to those reported by Baum and Rixen (2014) for the Siak River.

Eutrophication can lead to enhanced estuarine primary production and consequently to hypoxia. This was not observed. Although organic matter was respired in the estuaries (Müller et al., 2015a2016), oxygen depletion was relatively moderate in the surface water due to a quick replenishment from the overlying air, as suggested by the high gas exchange velocity (Table 4). Since we did not measure oxygen profiles, we can only speculate about the oxygen levels in the bottom water. The high turbulence in the water likely prevented stratification and promoted the ventilation of the water column and, therefore, may have prevented the development of anoxic bottom waters.

4.2 N₂O

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Dissolved N_2O concentrations were mostly close to atmospheric equilibrium concentrations (i.e. 100% saturation), which is in line with other comparable studies in the tropics and subtropics (Richey et al., 1988; Zhang et al., 2010; Rao and Sarma 2013; Borges et al., 2015). High N_2O concentrations were reported for eutrophic and hypoxic coastal waters in the western Indian continental shelf (up to 533 nmol L^{-1} , Naqvi et al., 2000) and for the Peruvian upwelling (up to 986 nmol L^{-1} , Arevalo-Martinez et al., 2015). Similarly, N_2O concentrations in the subtropical Brisbane estuary were higher than in the Lupar and Saribas (median concentrations between 7.2 and 12.3 nmol L^{-1}), ranging between 9.1 and 45 nmol L^{-1} (Musenze et al., 2014).

The low N_2O concentrations in the Lupar and Saribas estuaries are not surprising. Even though these estuaries were slightly eutrophic, the DIN concentrations were still below the average for tropical non-blackwater rivers (Baum and Rixen, 2014). For estuaries with low DIN concentrations, low N_2O concentrations are expected (Zhang et al., 2010). Interestingly, enhanced N_2O was measured offshore during cruise SO218. Either, a source of N_2O exists on the continental shelf, or the enhanced N_2O concentrations are due to interannual variability: The SO218 cruise took place in 2011, while the measurements in the Lupar and Saribas estuaries were performed in 2013 and 2014. Ultimately, the SO218 measurements were performed at the onset of the monsoon season, so possibly, seasonal variability plays a role-as well. This is considered most likely, as it is in line with the tendency towards higher and more variable N_2O concentrations during MLD wet if compared to MLD dry in the Saribas tributary.

Stronger oxygen depletion and higher DIN concentrations suggest higher respiration rates during the dry season (Müller et al., 2015a2016). In accordance with this, N₂O was well correlated with DOC and salinity during the dry season, suggesting that it originated mainly from respiratory activity in freshwaterthe upper estuary. In this study, samples were not collected for salinities<4, so we cannot make this claim with certainty. However, previous studies suggest that the highest N₂O concentrations are usually encountered at low salinities in the vicinity of the turbidity maximum (Robinson et al., 1998, Barnes and Owens, 1999, Zhang et al., 2010). On the other hand, N₂O maximum concentrations along estuaries may also

occur at intermediate salinities (de Wilde and de Bie, 2000). This seems to have been the case during our wet season campaign, where we observed the highest N₂O concentrations at salinities between 10-20 (Fig. 2c). N₂O concentrations were not correlated with salinity during the wet season. In contrast, during the wet season, a correlation of N₂O with salinity was not observed and the correlation with DOC was weak, suggesting that additional sources of N₂O existed in the estuary and obscured these relationships. The nature of this additional estuarine source is indicated by the covariation of In the wet season, N₂O showed the same spatial variations as and CH_{4.7} which were The variability of both gases was linked to the tidal cycle, which was not the case for N₂O during MLD dry (see Fig. 4). This implies that N₂O and CH₄ had the same source during the wet season and that this source was located within the estuary at intermediate salinities. In line with the tidal variability, we propose production of both N₂O and CH₄, e.g., production in estuarine sediments or in tidal creeks as a common source of both gases.

The most striking feature of the spatial distribution of N_2O was its strong variability, with enhanced concentrations in the Saribas tributary during both seasons, indicating a local source. The most obvious candidate for a point source is anthropogenic, i.e. sewage. However, NO_3^- , NO_2^- , and NH_4^+ were not enhanced in the same way as N_2O , suggesting that DIN concentrations are a poor predictor for estuarine N_2O concentrations (see discussion below).

5 4.3 CH₄

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Similar to N₂O, dissolved CH₄ concentrations were relatively moderate. For example, while CH₄ concentrations in the partially hypoxic Pearl River estuary ranged between 23 and 2984 nmol L⁻¹ (Chen et al., 2008), median CH₄ concentrations in our study area varied between 4-64 nmol L⁻¹. Similarly, the CH₄ concentration range reported by Musenze et al. (2014) for the Brisbane estuary, Australia, was substantially higher (31-578 nmol L⁻¹) than in this study, although both the DOC concentration range and the DO saturation range were similar. This is surprising, as the peat-draining tributaries are extremely oxygen-depleted and contain large amounts of organic matter (Müller et al., 2015b). These conditions are usually suitable for CH₄ production. Although we do not have data on the river end-member CH₄ concentration, we can infer from However, the lack of correlation between CH₄ and salinity implies that freshwater input was not the main source of CH₄ in the estuaries. Since CH₄ is a poorly soluble gas, a large fraction might be released to the atmosphere before reaching the estuary and the coastal ocean. Another possible explanation is methane oxidation, which was shown to remove large fractions of CH₄ in some estuaries (deAngelis and Scranton 1993, Abril et al. 2007). In addition, salt intrusion may inhibit CH₄ production in the estuaries. Although CH₄ concentrations tended to be higher during the wet season, a clear seasonal pattern of CH₄ concentrations was not apparent, which is in line with observations at other tropical (Barnes et al., 2006; Teodoru et al., 2014) and subtropical (Musenze et al., 2014) sites.

The generally positive relationship between CH₄ and pCO₂ has been observed in other tropical aquatic systems (Teodoru et al., 2014, Borges et al., 2015) and is indicative of organic matter decomposition as a source of both gases. The strong response of CH₄ to tidal variations indicates that CH₄ is produced in the sediments and released when the hydrostatic pressure drops during falling tide. The tidal variability of CH₄ (and N₂O in the wet season) can also be interpreted as

indication of the contribution of intertidal sediments and tidal creeks to the CH₄ concentrations in these estuaries. The importance of tidal creeks (Middelburg et al., 2002) and tidal pumping (Barnes et al., 2006; Borges and Abril, 2011) for CH₄ concentrations in estuaries is widely recognized.

4.4 N₂O and CH₄ flux densities and emissions

With the exception of the Saribas tributary ($F_{annual} = 12.0 \pm 7.5 \text{ mmol m}^2 \text{ yr}^{-1}$), N₂O flux densities were quite low (1.3-1.9) mmol m⁻² yr⁻¹) and at the lower end of the range reported for Indian estuaries (-0.4-5.2 mmol m⁻² yr⁻¹, Rao and Sarma, 2013). CH₄ flux densities (22.45-89.2 mmol m⁻² yr⁻¹) were within the ranges reported for other tropical sites: Koné et al. (2010), for example, determined flux densities of 28.5-123.4 mmol m⁻² yr⁻¹ for stratified lagoons of Ivory Coast; Shalini et al. (2006) report flux densities of 19.7-102.2 mmol m⁻² yr⁻¹ for Pulicat lake, India; and Biswas et al. (2007) measured CH₄ flux densities between 0.7 and 49 mmol m⁻² yr⁻¹ in the estuaries of the Sundarban mangrove ecosystem. However, we caution that the comparison suffers from the different approaches to determining the gas exchange velocity k. Koné et al. (2010), Shalini et al. (2006), Biswas et al. (2007) and Rao and Sarma (2013) used empirical equations relating k to wind speed, while we used estimates based on floating chamber measurements. We refer to our discussion in Müller et al. (2015a2016), where we showed that our estimates derived by floating chamber measurements yielded higher values than if we had used empirical relationships with wind speed. We argued that floating chamber measurements offered a better representation of the actual flux than gas exchange models that rely on empirical relationships with wind speed, which were initially derived for the open ocean (Wanninkhof, 1992) and do not consider current-induced turbulence as a driver of gas exchange. Musenze et al. (2014) used empirical models for both the wind-related and the current-related gas exchange velocity and added them up to derive flux estimates for the Brisbane estuary. Not surprisingly, the fluxes they report were much higher than in the aforementioned studies, with 0.8-28.0 mmol N₂O m⁻² yr⁻¹ and 7.5 to 636 mmol CH₄ m⁻² yr⁻¹. Compared with these estimates, the N2O and CH4 fluxes from Lupar and Saribas are quite low, which is consistent with the lower N2O and CH4 concentrations as discussed above.

On the basis of our floating chamber measurements (see Müller et al, $\frac{2015a2016}{2015}$) we computed an overall annual flux of 7 t N_2O -N for the Lupar and Saribas estuaries. This is only half of the estimate (15 t N_2O -N yr⁻¹) computed using the Mosier et al. (1998) emission factor. The reason is obviously the missing relationship between DIN concentrations and N_2O emissions in the studied system. Although across estuaries, increasing DIN is generally associated with increasing N_2O (Murray et al., 2015), the relationship is not linear: Zhang et al. (2010) found a polynomial relationship between DIN and N_2O across estuaries in different geographic locations. The emission factor from Mosier et al. (1998), in contrast, is based on constant N_2O /DIN ratio. Therefore, it is not surprising that for low DIN environments, as our study site, the use of emission factors leads to a considerable bias. In line with the data presented here, Borges et al. (2015) did not find a relationship between N_2O and DIN from measurements in several tropical and sub-tropical African rivers and estuaries. This could indicate that this relationship is not applicable in tropical systems and demonstrates the need of further studies in low latitudes to improve estimates of global N_2O emissions from estuaries.

4.5 Implications for future land use change

Nutrient loads to estuaries are predicted to increase in the future in most of Southeast Asia (Seitzinger and Kroeze, 1998). Our results imply that eutrophication in peat-dominated estuaries does not automatically lead to enhanced N_2O emissions. However, the localized elevation of N_2O concentrations, as observed on the Saribas tributary, suggests that estuarine N_2O concentrations might be impacted by local anthropogenic sources. Indeed, it has been shown that N_2O emissions from peat soils depend on land use and that cultivated sites generally exhibit higher N_2O fluxes to the atmosphere (Hadi et al., 2000). It is likely that this behavior is mirrored in the aquatic systems as well.

As CH₄ emissions from peatlands depend largely on the water table (Couwenberg et al., 2010), CH₄ fluxes from peatlands are enhanced under sago (Melling et al., 2005) and rice (Couwenberg et al., 2010) and reduced under oil palm (Melling et al., 2005). The cultivation of sago in the catchment of the Saribas tributary might therefore partially explain the observed high CH₄ fluxes from the Saribas tributary. The future development of CH₄ dynamics in estuaries in this region in dependence of land use change is hard to predict, as there are potentially counteracting effects of the conversion of peatlands to oil palm and other industrial crops, which require different agricultural practices (flooding versus drainage).

5. Conclusions

Overall, we found that the two tropical estuaries of the rivers Lupar and Saribas in a peat-dominated region in Malaysia were small to moderate sources of N₂O and CH₄ to the atmosphere. DIN concentrations were slightly enhanced compared to the unpolluted riverine end-member. This eutrophication did not lead to hypoxia or to enhanced N₂O concentrations. DIN was generally a poor predictor of N₂O, which provides further evidence that the use of emission factors for the calculation of N₂O fluxes from tropical estuaries is inappropriate. Although predictions about the future development of the Southeast Asian monsoon are highly uncertain and locally variable, chances are that rainfall will moderately increase in this region (Christensen et al., 2013). Our results suggest that this may increase the N₂O and probably also the CH₄ emissions from estuaries, as the concentrations of both gases were more variable during the wet season, with higher maximum concentrations of both gases and additional sources of N₂O in the estuaries. For our study area, additional sampling at the peak of the monsoon season would be desirable in order to consolidate these statements. Yet, our results provide a first conception of the seasonality in these systems and underline the fact that time series measurements (seasonal sampling) are vital for the understanding of N₂O and CH₄ fluxes from aquatic systems in monsoonal regions.

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Tables

				N export (t yr ⁻¹)			
		Min	Max	Mean	Median	SD	
Lupar	dry	20.9	30.8	23.6	22.2	3.6	4532
	wet	11.9	21.7	18.2	19.7	3.2	
Saribas	dry	25.5	37.6	31.0	30.0	6.1	1554
	wet	10.0	18.0	14.0	14.0	5.6	
Saribas	dry	20.5	30.0	24.3	22.4	5.0	n.d.
tributary	wet	10.9	18.2	13.0	11.4	3.5	

Table 1: Estuarine DIN concentration medians for the spatial extent that was covered in 2013 and the estimated inorganic N export to the South China Sea.

		Concentrations (nmol L ⁻¹)					Saturations (%)				
		Min	Max	Mean	Median	SD	Min	Max	Mean	Median	SD
Lupar	dry	7.2	9.3	7.7	7.6	0.4	106	116	108	108	1
	wet	6.3 <u>6.6</u>	13.8	7.7	7.2	1.0	93 102	208	116	109	15
Saribas	dry	7.3	8.4	7.9	7.9	0.2	110	118	114	113	2
	wet	8.2	19.3	9.0	8.7	1.2	118	592	134	126	46
Saribas	dry	8.1	13.4	11.8	12.3	1.5	117	184	165	169	19
tributary	wet	9.1	24.4	13.5	11.9	4.3	130	340	189	169	59
Coastal ocean		10.1	27.3	15.8	13.0	7.0	168	462	266	218	119

Table 2: N₂O concentrations and saturations in the Lupar, Saribas and the Saribas tributary and in the coastal ocean. Values for the estuaries are given for the spatial extent of the rivers that was covered in 2013.

	Concentrations (nmol L ⁻¹)					Saturations (%)					
		Min	Max	Mean	Median	SD	Min	Max	Mean	Median	SD
Lupar	dry	7.3	42.3	18.7	10.6	12.2	339	1910	837	486	520
	wet	2.0 3.7	61.2	24. 3 4	13.5	20.3	88 168	2799	1100 1104	614 <u>615</u>	915
Saribas	dry	8.3	58.8	28.2	25.6	15.5	397	2782	1331	1188	727
	wet	9.0	68.5	15.1	12.2	9.1	401	3050	671	545	395
Saribas	dry	5.2	53.3	20.6	12.4	16.9	228	2458	942	554	784
tributary	wet	23.3	113.9	63.2	64.0	27.4	1040	5058	2773	2786	1198
Coastal ocean		3.3	3.7	3.5	3.6	0.2	166	188	177	176	9

Table 3: CH₄ concentrations and saturations in the Lupar, Saribas and the Saribas tributary and in the coastal ocean. Values for the estuaries are given for the spatial extent of the rivers that was covered in 2013.

		k ₆₀₀	Flux density		Annual av	erage flux	Flux	
		(cm h ⁻¹)	(nmol m ⁻² s ⁻¹)		density (mm	nol m ⁻² yr ⁻¹)	(t N yr ⁻¹ and t C	
							yr ⁻¹)	
			N ₂ O	CH ₄	N ₂ O	CH ₄	N ₂ O	CH ₄
Lupar	dry	n.d.	0.04 ± 0.01	0.59 ± 0.14	1.3 ± 0.3	22. 4 — <u>5</u> ±	4 ± 1	59 - <u>60</u> ±
	wet	20.5 ± 4.9	0.04 ± 0.01	0. 83 _ <u>84</u> _±		5.4		17
1				0.20				
Saribas	dry	n.d.	0.04 ± 0.03	1.01 ± 0.84	1.9 ± 1.6	23.0 ± 19.1	3 ± 2	28 ± 25
	wet	13.2 ± 11.0	0.08 ± 0.07	0.45 ± 0.37				
Saribas	dry	n.d.	0.39 ± 0.24	0.81 ± 0.50	12.0 ± 7.5	89.2 ± 55.2	n.d.	n.d.
tributary	wet	23.9 ± 14.8	0.37 ± 0.23	4.84 ± 3.00				

Table 4: k_{600} values and median N_2O and CH_4 areal and total fluxes from the Lupar, Saribas and the Saribas tributary. The uncertainties relate to the maximum variability of the k_{600} value, as the k_{600} uncertainty propagation was responsible for approximately 96% of the flux uncertainty. Values are given for the spatial extent of the rivers that was covered in 2013, k_{600} values and median N_2O and CH_4 areal and total fluxes from the Lupar, Saribas and the Saribas tributary. The uncertainties relates to the maximum variability of the k_{600} value. Values are given for the spatial extent of the rivers that was covered in 2013.

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Figures

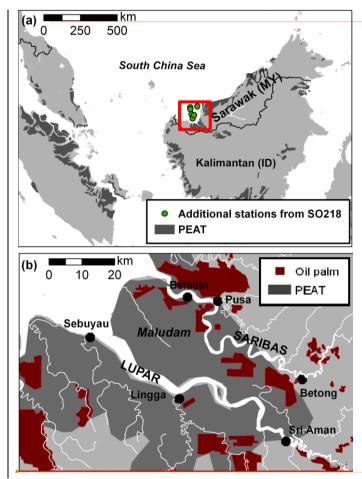
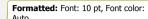


Figure 1: Map of the study area showing (a) the location of Sarawak on the island of Borneo and the cruise track of the R/V *Sonne* in November 2011 (SO218). (b) Close-up map of the Lupar and Saribas estuaries, enclosing the Maludam peninsula, showing the major settlements along the rivers and estuaries. Peat areas are indicated by the dark grey area, oil palm plantations as of CIFOR (2014) are shown in red.

Comment [d1]: Figure 1 was revised (scale bars were inserted and the South China Sea was labelled)

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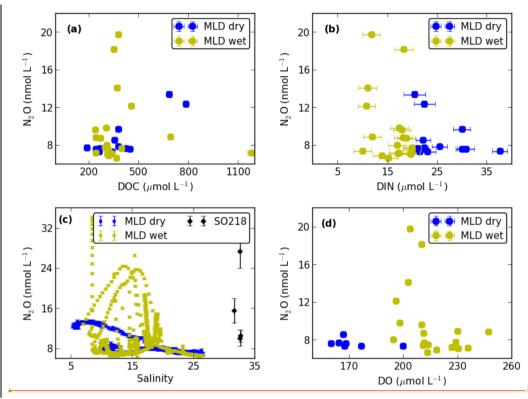


Figure 2: Relationship of nitrous oxide concentrations with (a) dissolved organic carbon (DOC), (b) dissolved inorganic nitrogen (DIN), (c) salinity and (d) dissolved oxygen (DO). MLD refers to the campaigns on the Lupar and Saribas estuaries in the dry and wet season, respectively, SO218 denotes data from the R/V *Sonne* cruise.

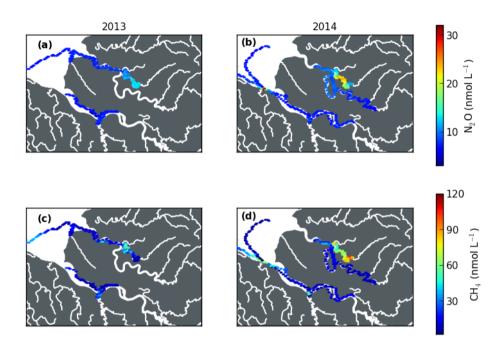
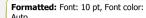


Figure 3: Dissolved N_2O (a+b) and CH_4 (c+d) concentrations measured during the 2013 (left) and 2014 (right) MLD campaigns.



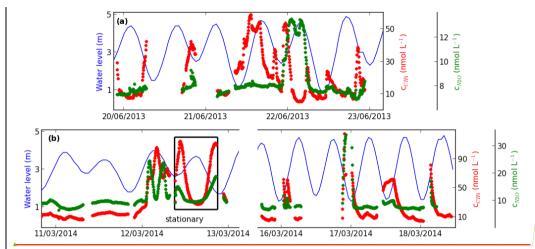
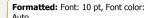


Figure 4: Time series of methane and nitrous oxide concentrations in the Lupar and Saribas estuaries measured during the dry season (2013) campaign (upper panel) and the wet season campaign (lower panel) and the water level as predicted for Pulau Lakei (+1 h for Lupar and Saribas). Spatial, temporal and tidal variations are overlapping in the Figure. One stationary measurement, as recorded on the Saribas Rriver in 2014, is denoted with the black box. Note the discontinuous time axis in the lower panel.



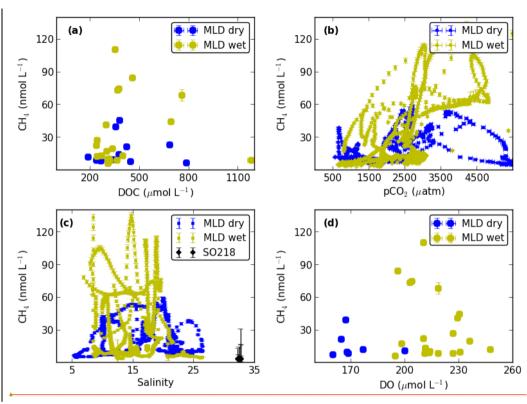


Figure 5: Relationship of methane concentrations with (a) dissolved organic carbon (DOC), (b) pCO₂, (c) salinity and (d) dissolved oxygen (DO). MLD refers to the campaigns on the Lupar and Saribas estuaries in the dry and wet season, respectively, SO218 denotes data from the R/V *Sonne* cruise.