

The riverine source of CH₄ and N₂O from the Republic of Congo, Western Congo Basin

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Abstract. We discuss concentrations of dissolved CH₄, N₂O, O₂, NO₃⁻ and NH₄⁺, and emission fluxes of CH₄ and N₂O for river sites in the Western Congo Basin, Republic of Congo (ROC). Savannah, swamp forest and tropical forest
15 samples were collected from the Congo main stem and seven of its tributaries during November 2010 (41 samples; “wet season”) and August 2011 (25 samples; “dry season”; CH₄ and N₂O only). Dissolved inorganic nitrogen (DIN: NH₄⁺ + NO₃⁻; wet season) was dominated by NO₃⁻ (63 ± 19% of DIN). Total DIN concentrations (1.5-45.3 μmol L⁻¹) were consistent with the near absence of agricultural, domestic and industrial sources for all three land types. Dissolved O₂ (wet season) was mostly under-saturated in swamp forest (36 ± 29%) and tropical forest (77 ± 36%)
20 rivers but predominantly super-saturated in savannah rivers (100 ± 17%). The dissolved concentrations of CH₄ and N₂O were within the range of values reported earlier for sub-Saharan African rivers. Dissolved CH₄ was found to be super-saturated (11.2 - 9553 nmol L⁻¹; 440-354444%), whereas N₂O ranged from strong under-saturation to super-saturation (3.2-20.6 nmol L⁻¹; 47-205%). Evidently, rivers of the ROC are persistent local sources of CH₄ and can be minor sources or sinks for N₂O. During the dry season the mean and range of CH₄ and N₂O concentrations were quite
25 similar for the three land types. Wet and dry season mean concentrations and ranges were not significant for N₂O for any land type or for CH₄ in savannah rivers. The latter observation is consistent with seasonal buffering of river discharge by an underlying sandstone aquifer. Significantly higher wet season CH₄ concentrations in swamp and forest rivers suggest that CH₄ can be derived from floating macrophytes during flooding and/or enhanced methanogenesis in adjacent flooded soils. Swamp rivers also exhibited both low (47%) and high (205%) N₂O
30 saturation but wet season values were overall significantly lower than in either tropical forest or savannah rivers, which were always super-saturated (103-266%) and for which the overall means and ranges of N₂O were not significantly different. In swamp and forest rivers O₂ saturation co-varied inversely with CH₄ saturation (log %) and positively with % N₂O. A significant positive correlation between N₂O and O₂ saturation in swamp rivers was coincident with strong N₂O and O₂ under-saturation, indicating N₂O consumption during denitrification in the sediments. In savannah rivers persistent N₂O super-saturation and a negative correlation between N₂O and O₂ suggest
35 N₂O production mainly by nitrification. This is consistent with a stronger correlation between N₂O and NH₄⁺ than between N₂O and NO₃⁻. Our ranges of values for CH₄ and N₂O emission fluxes (33-48705 μmol CH₄ m⁻² d⁻¹; 1-67 μmol N₂O m⁻² d⁻¹) are within the ranges previously estimated for sub-Saharan African rivers but they include uncertainties deriving from our use of “basin-wide” values for CH₄ and N₂O gas transfer velocities. Even so,
40 because we did not account for any contribution from ebullition, which is quite likely for CH₄ (at least 20%), we consider our emission fluxes for CH₄ to be conservative.

1 Introduction

Methane (CH₄) and nitrous oxide (N₂O) accounted for 17% and 6% respectively, of the total atmospheric radiative forcing by well-mixed greenhouse gases in 2011 (Myhre et al., 2013). CH₄ also impacts tropospheric oxidising capacity, O₃ and OH radical and is a source of stratospheric O₃ (Myhre et al., 2013) while N₂O is the largest cause of stratospheric O₃ loss, via NO production (Ravishankara et al., 2009). Since the onset of the industrial revolution, tropospheric CH₄ and N₂O have substantially increased but their growth rates have varied. The early 1980s to the mid-2000s saw an overall decline in tropospheric CH₄ growth (Dlugokencky et al., 2009; 2011). This has been ascribed to declining fossil fuel emissions and/or variations in the OH radical sink (Rice et al., 2016; Schaefer et al., 2016), punctuated by episodic events such as the 1991-1992 Pinatubo eruption and an intense 1997-1998 El Niño (Nisbet et al., 2016). However, since 2007 increased growth has been sustained. Recent evidence from isotopic studies (Nisbet et al., 2016) and a box-model (Schaefer et al., 2016) ascribes this to increased biogenic emissions, particularly in the tropics, where they have been linked to expanding tropical wetlands in response to positive rainfall anomalies (Nisbet et al., 2016), and/or growing emissions from agricultural sources (Schaefer et al., 2016). The mean CH₄ tropospheric dry mole fraction in 2011, 1803 ± 2 ppbv, was 150% above the pre-industrial value (Hartmann et al., 2013). Of particular note for N₂O, the availability of sufficiently reliable data on the anthropogenic components of river, estuary and coastal zone sources resulted in their classification being changed in the IPCC AR4 synthesis, from “natural” to “anthropogenic” (Denman et al., 2007). A small but significant seasonal to inter-annual variability in N₂O growth rate may reflect climate-driven changes in soil N₂O (Thompson et al., 2013). The current rate of N₂O growth is 0.73 ± 0.03 ppb yr⁻¹ and its tropospheric dry mole fraction in 2011, 324 ± 0.1 ppbv, was ~20% above its pre-industrial value (Hartmann et al., 2013).

The evidence base for freshwater ecosystems (streams, rivers, lakes, and reservoirs) as important tropospheric CH₄ and N₂O sources is small but increasing. The global freshwater CH₄ source could be $\sim 10^{13}$ - 10^{14} g CH₄ yr⁻¹ (Bastviken et al., 2011; Kirschke et al., 2013; Stanley et al., 2016). This order of magnitude range largely reflects a discrepancy between “top-down” approaches based on atmospheric inversions (e.g. Kirschke et al., 2013) and “bottom-up” estimates that necessitate the upscaling of freshwater observations (e.g. Bastviken et al., 2011). For example, atmospheric constraints on top-down budgets imply that some component emissions of bottom-up approaches may be overestimates (Saunio et al., 2016). Notwithstanding this uncertainty, the global freshwater CH₄ source can be evaluated in the light of natural and total global CH₄ source estimates of 179 - 484×10^{12} g CH₄ yr⁻¹ and 526 - 852×10^{12} g CH₄ yr⁻¹ respectively (Kirschke et al., 2013). Quantifying the global freshwater contribution thus has high inherent uncertainty but based on these estimates it could be ~ 2-56 % of natural CH₄ emissions and ~1-19% of total CH₄ emissions. Converting these to CO₂ equivalents based on a 100 year Global Warming Potential gives $\sim 0.65 \times 10^{15}$ g C (CO₂ equivalent) yr⁻¹ (Bastviken et al., 2011), a significant offset to the combined terrestrial and oceanic carbon sink $\sim 5.5 \times 10^{15}$ g C yr⁻¹ (Le Quéré et al., 2015). A global estimate of river N₂O emissions based on microbial production from agriculturally-derived nitrogen is $\sim 6.8 \times 10^{11}$ g N₂O yr⁻¹, around 10% of the total global anthropogenic N₂O source, but because this involved upscaling emissions from entirely within the contiguous United States (Beaulieu et al., 2011), it too must be highly uncertain.

Tropical river systems in Africa include some of the world’s largest, together contributing ~12% of both global freshwater discharge (Valentini et al., 2014) and river surface area (Raymond et al., 2013). Borges et al (2015b) recently reported annual emissions ~ 3 - 4×10^{12} g CH₄ and $\sim 10^{10}$ g N₂O for twelve large river systems in sub-Saharan Africa, including the three largest by catchment area (Congo, Niger, Zambezi). Notably, their CH₄ estimate is five times higher than was previously attributed to all tropical rivers (Bastviken et al., 2011) and both estimates are significant at the continental scale given that reported total African emissions are $\sim 66 \pm 35 \times 10^{12}$ g CH₄ yr⁻¹ and $3.3 \pm 1.3 \times 10^{12}$ g N₂O yr⁻¹ (Valentini et al., 2014).

The potential scale of CH₄ and N₂O emissions from tropical freshwaters and their attendant uncertainties warrant further investigation. In this paper we present and discuss concentrations of dissolved CH₄, N₂O, O₂, NO₃⁻ and NH₄⁺, and corresponding CH₄ and N₂O emissions for river sites in savanna, swamp forest and tropical forest. Samples were collected along the Congo main stem and in seven of its tributary systems in the Western Congo Basin, Republic of Congo, during November 2010 and August 2011.

2 Study site and sample locations

The ~4700 km long Congo River (Fig. 1) has an equatorial location that affords it a bimodal hydrological regime. Maximal flows are in December and May and minimal flows are in August and March (Coynel et al., 2005). The Congo Basin (9°N - 14°S; 11° - 31°E) is the largest hydrological system in Central Africa, covering ~3.8 × 10⁶ km² (~ 12% of the total African land mass; Fig. 1) and incorporating the world's fourth largest wetland area ~3.6 × 10⁵ km² (Laporte et al., 1998). The Congo's annual freshwater discharge is the world's second largest at ~1300 km³ (Borges et al., 2015b), 50% of all freshwater flow from Africa to the Atlantic Ocean. Rivers and streams in the Congo Basin have a total open water surface area ~2.7 × 10⁴ km² (Raymond et al., 2013). The climate is warm (mean annual temperature 24.8 ± 0.8 °C) and humid with an annual rainfall ~1800 mm (Laraque et al., 2001).

We sampled the Congo main stem, seven of its tributary rivers and some of their sub-tributaries, at sites within the Republic of Congo (ROC: area 3.4 × 10⁵ km²), in the western Congo Basin (Figure 1). Individual catchment areas, freshwater discharge rates and rainfall are listed in Table 1. Around 50% of the ROC land area is classified as tropical forest, with the remainder classified as either swamp or savannah in approximately equal proportion (Clark and Decalo, 2012). Sampling sites were selected to represent each of these three land cover types (Fig. 1), which were georeferenced to the World Geodetic System 1984 (WGS84) and intersected with the highest level sub-watershed polygons defined by the HYDRO1K global hydrological dataset (U.S. Geological Survey, 2000). This enabled assigning the fractional cover for each land cover type, and hence the dominant land cover type, to the areas immediately surrounding each sampling location. Swamp includes both temporally and permanently inundated areas of "forest", with vegetation adapted to poorly drained, anaerobic soils (Mayaux et al., 2002). For all three land cover types the mean annual temperature range (period 1990-2012) is ~1-3°C. Temperatures are lowest (~22-24°C) in July-August and highest (~25-26°C) in March-April (<http://sdwebx.worldbank.org/climateportal/>). For savannah the average monthly rainfall during July-May (1990-2012) is ~120-260 mm, typically being maximal in October-November, but < 40 mm falls during June-August (<http://sdwebx.worldbank.org/climateportal/>). For forest and swamp the annual range in monthly rainfall is less pronounced. Both have two discernable rainfall maxima, during April-May and October-November (~150-240 mm month⁻¹), and a minimum in June-August (~40-120 mm month⁻¹) (<http://sdwebx.worldbank.org/climateportal/>).

ROC swamp and forest (Fig. 1) broadly correspond to the westernmost part of the "Cuvette Centrale" (Central Basin). This is a large shallow depression composed mainly of dense, humid forest and extending from approximately 15°W to 25°W and 5°N to 4°S. Its central western part remains flooded throughout the rainy seasons. Rivers sampled in this region (Sangha, Likouala-aux-Herbes, Likouala, Lengoue, Mambili: Fig. 1, Table 1) drain predominantly sandy or clayey quaternary deposits. The Kouyou basin (Fig. 1) borders the 'Batéké Plateaux', a 600-700m relief sandstone formation to the south, intersected by dry valleys and covering much of the southern ROC. Here, bushy savannah is intersected by the Alima, Nkéni and Léfini rivers. Due to water storage in an underlying sandstone aquifer the hydrological regimes of these three rivers are largely independent of rainfall; they all show only weak seasonality in discharge despite the relatively large variation in monthly precipitation (Laraque et al., 2001).

3 Sample collection and analytical techniques

130 Surface water samples (~0.3 m depth) were collected from central river channels using a standard “niskin”-type water
sampler (<http://www.tresanton.co.uk/standard.html>). We obtained 66 samples for dissolved CH₄, N₂O, O₂, NO₃⁻ and
NH₄⁺ analysis, during November 2010 (41 samples) and August 2011 (25 samples; CH₄ and N₂O only). Based on the
monthly rainfall distribution, for convenience we hereinafter refer to these as “wet season” and “dry season”
respectively. Samples were slowly decanted into a series of 125 ml glass screw top septum bottles (Sigma-Aldrich,
UK) via a silicon rubber tube. Each was over filled by at least one sample volume to avoid bubble entrainment.
135 Samples were poisoned with 25 µl 0.1 M HgCl₂ to arrest microbial activity and sealed to leave no headspace. They
were subsequently returned to Newcastle for dissolved gas analysis within several weeks of collection. Dissolved gas
samples treated in this way can be successfully stored for several months (Elkins, 1980).

140 Dissolved CH₄ and N₂O were analysed by single phase equilibration gas chromatography (Shimadzu GC 14-B), with
flame ionisation detection of CH₄ and electron capture detection of N₂O (Upstill-Goddard et al., 1996). Routine
calibration was with a mixed secondary standard (361 ppbv N₂O, 2000 ppbv CH₄) prepared by pressure dilution with
ultra-high purity N₂ (Upstill-Goddard et al., 1990). Absolute calibration was against a mixed primary standard
(10ppmv N₂O, 5ppmv CH₄) with a certified accuracy of ± 1 % (BOC Special Gases, UK). Overall analytical
precisions (1σ) for N₂O and CH₄, established via multiple analysis (n = 15) of the secondary standard, were both ±
145 1%.

Temperature, dissolved O₂ and atmospheric pressure were measured *in-situ* using a handheld multi-parameter probe
(YSI Pro-Plus: <https://www.yei.com/>). Quoted measurement accuracies are: ± 0.2 °C; ± 2 % dissolved O₂; ± 0.002 bar.
Samples for dissolved NH₄⁺ and NO₃⁻ were filtered on collection (Whatman 0.7 µm GF/F; precombusted at 550 °C for
150 8 h), directly into clean glass vials and stored acidified (pH 2) at 4 °C in the dark for several weeks. Subsequent
analysis, at Woods Hole, was by segmented flow (Astoria Analyzer; Astoria-Pacific, USA) using established methods
(U.S. Environmental Protection Agency, 1984). Analytical precisions (1σ) were ± 1% for both. Technical and
logistical issues precluded the collection of any dissolved O₂, NH₄⁺ or NO₃⁻ data during August 2011 (dry season) and
some NO₃⁻ and NH₄⁺ data during November 2010 (wet season).

155 Emission fluxes, F (mol m⁻² d⁻¹), of CH₄ and N₂O were estimated using $F = k_w L \Delta p$, where k_w is the transfer velocity
of CH₄ or N₂O (cm hr⁻¹), L is the solubility of CH₄ or N₂O (mol cm⁻³ atm⁻¹) (Wiesenburg and Guinasso, 1979; Weiss
and Price, 1980) and Δp is the corresponding water-to-air partial pressure difference. k_w values were derived from two
corresponding estimates for CO₂ in the Congo. Raymond et al. (2013) estimated a basin-wide k_w of 5.2 m d⁻¹ for CO₂,
160 using hydraulic equations involving basin slope and flow velocity. The uncertainty in this estimate is ~ ± 10%
(Raymond et al., 2013). Aufdenkampe et al. (2011) applied constant k_w values for CO₂ in streams (< 100 m wide:
3.0 m d⁻¹) and in rivers (>100 m wide: 4.2 m d⁻¹). Adjusting for the relative areas of these in the Congo basin
(Borges et al., 2015b) gives a basin-wide mean k_w ~3.9 m d⁻¹ for CO₂. We converted these estimates to k_w for CH₄ and
N₂O by multiplying by $(Sc/470.7)^{-0.5}$, where 470.7 is the Schmidt number of CO₂ in freshwater, and Sc is the Schmidt
165 number of CH₄ or N₂O ($Sc_{CH_4}=486.8$; $Sc_{N_2O}=476.9$), assuming an ambient temperature of 25°C (Wanninkhof, 1992).
The resulting k_w estimates are 5.1 and 3.9 m d⁻¹ for CH₄ and 5.2 and 4.0 m d⁻¹ for N₂O. Resulting emissions
estimates are consequently ~30% higher based on Raymond et al. (2013). Using both sets of k_w estimates enables
direct comparison with the largest study of CH₄ and N₂O fluxes for African rivers that also used this approach (Borges
et al., 2015b). While other relevant work used wind speed based k_w estimates (Koné et al., 2010; Bouillon et al., 2012)
170 the unavailability of wind speeds precludes their use here. We applied the global mean mixing ratios of CH₄ (1797
ppbv) and N₂O (323 ppbv) for the year 2010 (<http://cdiac.esd.ornl.gov/tracegases.html>).

3 Results

While CH₄ and N₂O data are available for all samples, dry season data are not available for dissolved O₂, NO₃⁻ or NH₄⁺. Wet season DIN (NO₃⁻ + NH₄⁺) is only reported for samples for which both NO₃⁻ and NH₄⁺ are available. Source data for this paper are available as supplementary material (Table S1).

3.1 Dissolved O₂ and DIN

In wet season swamp samples dissolved O₂ varied between mildly under-saturated and very strongly under-saturated (Fig. 2). The mean (36 ± 29%) and range (4-91 %) of O₂ saturation were both significantly lower than for both forest rivers (Mann-Whitney, one-tailed; *P* = 0.0001) and savannah rivers (Mann-Whitney, one-tailed; *P* = 0.002). Most forest rivers were mildly to strongly O₂ under-saturated (mean 77 ± 36%, range 14-116 %; Fig. 2) and savannah rivers were predominantly mildly under-saturated to mildly super-saturated (mean 100 ± 17%, range 70-135 %; Fig. 2).

Low wet season concentrations of dissolved inorganic nitrogen (DIN) components (Fig. 3) are consistent with agricultural, domestic and industrial DIN sources all being negligible (Clark and Decalo, 2012). The means and ranges of total DIN (NO₃⁻ + NH₄⁺) did not differ significantly between any of the three river “types” (mean savannah 6.8 ± 2.8 μmol l⁻¹; range 2.5-10.1 μmol l⁻¹; n=10; mean swamp 5.1 ± 3.1 μmol l⁻¹, range 1.5-10.2 μmol l⁻¹; n= 11; mean forest 9.4 ± 11.2.μmol l⁻¹, range 1.8-45.3 μmol l⁻¹; n=12), in contrast to the situation for dissolved O₂. Differences in NO₃⁻-N were also not significant (mean savannah 4.1 ± 2.3 μmol l⁻¹, range 0.8-6.7 μmol l⁻¹, n=11; mean swamp 3.6 ± 2.1 μmol l⁻¹, range 1.0-8.8 μmol l⁻¹, n= 16; mean forest 7.1 ± 9.2 μmol l⁻¹, range 1.2-35.1 μmol l⁻¹; n=12). There was no clear relationship between NO₃⁻ and NH₄⁺ for any of the three river types (Fig. 3a). NO₃⁻ was the dominant DIN component in 24 of the 33 samples for which both NO₃⁻ and NH₄⁺ were analysed. Considering all samples, the mean NO₃⁻ contribution to DIN was 63 ± 19%.

3.2 Dissolved CH₄ and N₂O

Table 2 summarises ranges, means and medians of riverine CH₄ and N₂O concentrations and percent saturations for the three land cover types. All samples were highly CH₄ super-saturated, concentrations spanning two orders of magnitude (11.2 - 9553 nmol L⁻¹; 440-354400% saturation). N₂O spanned a much narrower concentration range and varied from strong under-saturation to strong super-saturation (3.2-20.6 nmol L⁻¹; 47-205%). Evidently, while rivers of the ROC are strong local sources of tropospheric CH₄ they can act as both small sources and sinks for N₂O. Swamp rivers exhibited both the lowest and among the highest N₂O saturations (Table 2) but during the wet season had overall significantly lower N₂O than either forest or savannah rivers. These were always super-saturated (103-266%; Table 2) and their overall means and ranges of N₂O were not significantly different (Mann-Whitney, one-tailed: swamp vs forest and swamp vs savannah, *P* = 0:004). For CH₄, concentration means and ranges during the wet season did not differ significantly between swamp and forest rivers but they were significantly higher in both than in savannah rivers (Mann-Whitney, one-tailed: swamp vs savannah, *P* = 0:004; forest vs savannah, *P* = 0.03). In contrast, during the dry season concentration means and ranges of both CH₄ and N₂O were indistinguishable for all three land cover types. Seasonal differences in concentration means and ranges were not significant for N₂O for any of the three land cover types or for CH₄ in savannah rivers. In both swamp and forest rivers CH₄ was significantly higher during the wet season (Mann-Whitney, one-tailed: swamp *P*= 0.01; forest *P* = 0.003).

There are comparatively few measurements of CH₄ concentrations in African rivers and even fewer of N₂O. Our CH₄ data for rivers of the ROC (Table 2) are within the ranges compiled for temperate and tropical rivers (~260-128000%) (Upstill-Goddard et al. 2000; Middelburg et al. 2002). Moreover, our CH₄ and N₂O data both fall within the ranges recently reported for other rivers in sub-Saharan Africa. Studies of CH₄ alone reported 48 - 870 nmol l⁻¹ (2221 -

38719 % saturation) in three Ivory Coast rivers (Koné et al., 2010), 25 - 505 nmol l⁻¹ (850 - 21700 % saturation) in the
215 Tana river, Kenya, (Bouillon et al., 2009) and 22-71430 nmol l⁻¹ in the Congo (Borges et al., 2015a). For concurrent
measurements of CH₄ and N₂O Bouillon et al. (2012) report 74 - 280 nmol CH₄ l⁻¹ (3450 - 13200% saturation) and 6.2
- 9.6 nmol N₂O l⁻¹ (112-165% saturation) in the Oubangui, a major Congo tributary. Teodoru et al. (2015) found 7-
12127 nmol CH₄ l⁻¹ and 2.0-11.4 nmol N₂O l⁻¹ at stations along the Zambezi. Borges et al. (2015b) quote a range of
2-62966 nmol CH₄ l⁻¹ (mean: 2205 nmol l⁻¹) and 0.2-85.4 nmol N₂O l⁻¹ (mean: 9.2 nmol l⁻¹) across 12 sub-Saharan
220 river basins, including those of the Congo, Zambezi and Niger.

Considering the complete data set, CH₄ was inversely correlated with both N₂O and O₂ (Fig. 2). Highest CH₄
coincident with lowest N₂O and O₂ occurred in swamp rivers. Lowest CH₄ coincident with highest N₂O and O₂ was
observed in forest rivers. Savannah rivers were intermediate between the two (Fig. 2). Overall, log % CH₄ vs % O₂
225 showed a weak negative correlation (Fig. 2a; R² = 0.26, n = 41) while % N₂O vs % O₂ showed a weak positive
correlation (Fig. 2b; R² = 0.30, n = 41). However, for both swamp and forest rivers individually the negative
correlations between log % CH₄ and % O₂ were stronger (swamp R² = 0.38, n = 16; forest R² = 0.45, n = 13). While
there was a stronger positive correlation between % N₂O and % O₂ for swamp rivers than for the complete data set (R²
= 0.71, n = 16), the correlation for forest rivers was extremely weak (R² = 0.02, n=13). For savannah rivers we
230 found a positive correlation between log % CH₄ and % O₂ (R² = 0.23; n= 12) and a negative correlation between %
N₂O and % O₂ (R² = 0.35, n=12). N₂O co-varied positively with both NO₃⁻ and NH₄⁺ (Fig. 3). For the complete
data set the correlations were weak (N₂O vs NO₃⁻, R² = 0.28, n = 59; N₂O vs NH₄⁺, R² = 0.23, n = 40). For all three
river types individually, with the exception of N₂O vs NH₄⁺ in savannah rivers, the correlations were stronger and for
all NO₃⁻ was a stronger predictor of N₂O (N₂O vs NO₃⁻: R² swamp = 0.50, n = 29; R² forest = 0.75, n = 15; R²
235 savannah = 0.31, n = 15) than was NH₄⁺ (N₂O vs NH₄⁺: R² swamp = 0.29, n = 13; R² forest = 0.47, n = 13; R²
savannah = 0.01, n = 14).

3.3 CH₄ and N₂O emission fluxes

Table 3 summarises ranges, means and medians of CH₄ and N₂O emission fluxes using *k_w* derived from Raymond et al.
(2013) and Aufdenkampe et al. (2011). Fluxes broadly followed the distribution of concentrations. CH₄ fluxes
240 were lowest overall in savannah rivers and highest in swamp and forest rivers. N₂O fluxes were lowest in swamp
rivers and highest in savannah and forest rivers. Fluxes were always to air at all sites for CH₄ and at all savannah and
forest sites for N₂O. However, swamp rivers were predominantly a N₂O sink during the wet season (11 of 16
individual flux estimates) and predominantly a N₂O source during the dry season (10 of 16 individual flux estimates).
As far as we are aware the wet season sink for N₂O in swamp rivers is the first such reported for African rivers.

245 4 Discussion

4.1. Sources of CH₄ and N₂O

The concentrations of dissolved CH₄ and N₂O at any specified river location reflect a dynamic and complex balance.
This involves in situ production and consumption impacted by import and export mechanisms that include upstream
and downstream advection, groundwater inputs, local surface runoff and water-air exchange.

250 While dissolved O₂ was under-saturated in the majority of samples, being as low as 4% in one wet season swamp
sample, it was always detectable and indeed was super-saturated in several savannah river samples in which CH₄
saturation ranged from ~4000-10000 % (Fig. 2a). Notwithstanding that methanogenesis is an exclusively anoxic
process carried out by severely O₂-limited archaea (Bridgham et al., 2013), the existence of high CH₄ concentrations

255 in oxygenated rivers is well-known (e.g. Richey et al., 1988). It is a consequence of the diffusion of CH₄ produced in
underlying river sediments, in adjacent floodplain soils and in adjacent wetlands, into aerated river water. Previous
work showed that the CH₄ supply from groundwater to African rivers is generally comparatively low (Balagizi et al.,
2015; Borges et al., 2015b). The spatial distribution of dissolved CH₄ is more closely related to the wetland
260 distribution within the catchment, wetland water deriving principally from upland runoff (Borges et al., 2015a). In the
Congo basin floating macrophytes, both in the centre of river channels and fringing their edges, are important
additional sources of CH₄ (Borges et al., 2015a). These sources promote a unidirectional CH₄ flow, towards small and
large river channels (Borges et al., 2015a). Where these river channels are shallow and have low levels of surface
turbulence, as in the examples studied here, the CH₄ diffusion term evidently exceeds combined CH₄ losses via
oxidation and water-to-air exchange. This further promotes the accumulation of high river CH₄ concentrations.

265 In addition to methanogenesis in fully anoxic sediment and soils, CH₄ production can also occur in “anoxic
microsites” within oxic soils (e.g. Teh et al., 2005; von Fisher & Hedin, 2007). Indeed, methanogens are now
considered to be widespread in oxic soils and they are activated during flooding (Bridgman et al., 2013). Their activity
relates to soil carbon age and composition (Bridgman et al., 1998; Chanton et al., 2008) and likely involves substrate
270 competition and other interactions. Production by soil macrofauna (Kammann et al., 2009), archeal production related
to plant productivity (Updegraff et al., 2001; Dorodnikov et al., 2011) and non-microbial, direct aerobic production,
both by living plant tissue (Keppler et al., 2006; 2009) and in soils (Hurkuck et al., 2012, have all also been observed.
Although methanogenesis by photoautotroph-attached archaea has been detected in oxic lake water (Grossart et al.,
2011) this is unlikely in tributaries and wetlands of the Congo, where phytoplankton abundance is low (Descy et al.
275 2016). Additional to this variability in production mechanisms and rates, CH₄ is subject to variable and rapid aerobic
and anaerobic microbial oxidation (Meronigal et al., 2004). Oxidative CH₄ loss rates have been variously estimated
at between a few percent and >100% of the rate of methanogenesis (Bussmann, 2013; Shelley et al., 2015). Despite
such potentially high losses, water to air exchange by ebullition and by turbulent diffusion driven by wind stress,
water depth and flow velocity (Raymond and Cole, 2001) is usually considered the major CH₄ loss term, with
280 ebullition frequently considered the dominant of these two mechanisms (Stanley et al., 2016). Despite this complexity
of dissolved CH₄ cycling in rivers, it is nevertheless informative to speculate on our principal observations in the
context of potential CH₄ sources and sinks.

The first notable feature of our results is the contrasting relationship between CH₄ and O₂ in swamp and forest rivers
285 (negative) and in savannah rivers (positive) (Fig. 2a). Dissolved O₂ in rivers is primarily driven by the balance of
photosynthesis and respiration (Houser et al., 2015). It may be additionally impacted by varying contributions from
water-air exchange that under conditions of extreme turbulence may lead to super-saturations as high as 150% (Li et
al., 2010). The overall positive relationship between CH₄ and O₂ in savannah rivers (Fig 2a.) could, at least in part,
reflect high macrophyte-related productivity. This can give rise to positive relationships by direct CH₄ production
290 (Stanley et al., 2016) and by indirect production via the trapping of fine-grained organic sediments that support
methanogenesis (Sanders et al., 2007). Similar relationships were observed in Amazon floodplain lakes (Devol et al.,
1990). Offsetting this, stems and roots respire O₂ (Caraco et al., 2006). Further inspection of the data shows that the
highest dissolved O₂ saturation found in savannah rivers (134%) deviates from the general CH₄ vs O₂ trend (Fig. 2a).
This sample was collected close to an area of rapids in the Congo main stem, in the vicinity of Pool Malebo (Formerly
295 known as Stanley Pool) (Fig. 1) where other samples were also O₂ super-saturated. Intense water-air exchange in
this region via increased turbulence would tend to enhance dissolved O₂ (Li et al., 2010) while depleting dissolved
CH₄. To summarise, notwithstanding possible additional CH₄ losses via oxidation, the CH₄ vs O₂ relationship in
savannah rivers (Fig. 2a) could be explained by net macrophyte production imprinted by water-air gas exchange.
The inverse of this relationship for swamp and forest rivers (Fig. 2a) was similarly reported for the Zambezi and

300 Amazon Basins, for the latter in fast flowing waters (Teodoru et al., 2015; Richey et al., 1988; Devol et al., 1990).
Again, high gas exchange rates are plausible, especially for the small number of tropical forest samples for which O₂
was close to or in excess of 100% (Fig. 2a). For the majority of samples that were O₂ under-saturated however,
additional mechanisms must be invoked. One possibility is that these distributions largely reflect the mixing of
305 relatively well-oxygenated river waters with high CH₄, low O₂ groundwater but another possibility is that this
relationship is the aggregate of this and several of the other processes previously discussed.

A second important aspect of the overall CH₄ distributions is that swamp and forest river CH₄ was highest during the
wet season, whereas savannah samples revealed no such inter-seasonal contrast (Table 2). The constancy of CH₄ in
savannah rivers might well reflect the buffering of seasonal river discharge by the sandstone aquifer that underlies this
310 region (Laraque et al., 2001). For swamp and forest rivers a number of alternative but not mutually exclusive
possibilities might be invoked. In addition to direct and indirect macrophyte production (Stanley et al., 2016;
Sanders et al., 2007), as discussed for savannah rivers, methanogenesis following the activation of archaea during
flooding of adjacent soils (Bridgham et al., 2013) is also plausible. This is supported by the observation that swamp
and forest soils are comparatively poorly drained (Mayaux et al., 2002). In contrast, an opposing behaviour was
315 reported for three rivers of the Ivory Coast (Comoé, Bia, Tanoé). In these, overall decreases in CH₄ during the dry to
wet season transition (Koné et al., 2010) were similar to trends recorded in some temperate (European) rivers
(Middelburg et al. 2002). Koné et al. (2010) ascribed the CH₄ seasonality in Ivory Coast rivers to a combination of
the dilution of high CH₄ baseflow by low CH₄ surface runoff (e.g. Jones and Mulholland 1998a, b), higher degassing
rates during flooding (Hope et al. 2001) and/or decreased in-stream methanogenesis towards high discharge (De
320 Angelis and Scranton, 1993). Conversely, Bouillon et al. (2012) attributed relatively stable high discharge CH₄
concentrations (~100 nmol l⁻¹) in the Oubangui, a major tributary of the Congo, to terrestrial soil production in
conjunction with baseflow transport. The largest fractional CH₄ contribution from baseflow often occurs in high
elevation headwaters with high soil organic content, while progressive downstream increases in CH₄ in lowland rivers
have been linked to increasing in-stream methanogenesis (Jones and Mulholland 1998a). Assuming such processes are
325 also operative in ROC swamp and tropical forest, interpreting or predicting the direction of any seasonal CH₄ trend in
a specified river system is evidently complex.

In contrast to CH₄, natural sources of aquatic N₂O are entirely microbial, and involve several pathways. Nitrification
is a two-stage process in which NH₄⁺ is first oxidised aerobically to NO₂⁻ via hydroxylamine (NH₂OH), followed by
330 NO₂⁻ oxidation to NO₃⁻. Following the first stage, N₂O can be produced through various routes: nitrifier nitrification
(NH₂OH → N₂O), nitrifier denitrification (NO₂⁻ → NO → N₂O) and nitrification-coupled denitrification (NO₃⁻ →
NO₂⁻ → NO → N₂O) (Kool et al., 2011). Heterotrophic denitrification, in which NO₃⁻ is the terminal electron
acceptor (NO₃⁻ → NO₂⁻ → NO + N₂O → N₂), occurs in soils, sediments and waters that are anoxic, the inhibition of
denitrifier activity at very low levels of dissolved O₂ being well known (Knowles 1982). Even so, in the complete
335 absence of O₂, N₂O can be enzymatically reduced to gaseous N₂ (Wrage et al. 2001), both in sediments and in the
water column, sometimes resulting in extreme N₂O under-saturations (Nirmal Rajkumar et al., 2008).

Although we found no statistically significant differences in the means and ranges of wet or dry season N₂O
concentrations for any land cover type, higher N₂O concentrations and emissions are considered likely where soil-
340 water filled pore spaces exceed 60 % due to enhanced microbial production (Davidson, 1993). This has been
observed in African savanna during the rainy season (Castaldi et al., 2006) and throughout much of the year in humid
tropical forests (Castaldi et al., 2013). The discrepancy between these and our observations to some extent likely
reflects a complex balance between the principal sites (groundwater and in-stream) and mechanisms of N₂O cycling,
as evidenced by the variable relationships between N₂O, O₂ and DIN we observed. For example, we found both

345 positive and negative relationships between N_2O and O_2 (Fig. 2b). Sediment processes and water concentrations are
evidently closely coupled in tropical catchments (Harrison and Matson, 2003). A strong positive correlation between
 N_2O and O_2 in swamp rivers coincident with strong under-saturation of both N_2O and O_2 (Fig. 2b) is consistent with
 N_2O consumption by sediment denitrification. Positive relationships between N_2O and NO_3^- have been variously
350 interpreted to reflect nitrification (Silvennoinen et al., 2008; Beaulieu et al., 2010), or both denitrification and
nitrification (Baulch et al., 2011). Even so, for swamp rivers the stronger correlation between N_2O and NO_3^- than
between N_2O and NH_4^+ , which is often taken to indicate a sediment denitrification N_2O source (Dong et al., 2004),
supports our conclusion of a swamp river denitrification sink for N_2O . Similar N_2O vs O_2 relationships were identified
in the Amazon and Zambezi river basins (Richey et al., 1988; Teodoru et al., 2015) and in the Adyar river-estuary,
S.E. India (Nirmal Rajkumar et al., 2008). In both the Amazon and the Adyar, N_2O was undetectable in fully anoxic
355 waters (Richey et al., 1998; Nirmal Rajkumar et al., 2008). N_2O and NO_3^- were also correlated in the Oubangui
(Bouillon et al., 2012) and in several other African rivers (Borges et al., 2015b). A similar, persistent correlation in a
temperate river was ascribed to denitrification in hypoxic/anoxic sediment, favoured by the ambient low river flow
and high temperatures leading to high community respiration and low O_2 solubility (Rosamond et al., 2012). Even
though denitrification in rivers may be limited by low levels of NO_3^- (Garcia-Ruiz et al., 1998) a temperate creek was
360 a N_2O sink for combined NO_2^- and NO_3^- concentrations $< 2.7 \mu\text{mol l}^{-1}$ (Baulch et al., 2011), broadly similar to the
majority of NO_3^- concentrations we observed (Fig. 3a). By contrast, N_2O and NO_3^- were uncorrelated in the Zambezi,
for which there was also no correlation of N_2O with NH_4^+ (Teodoru et al., 2015). For savannah rivers, in which N_2O
was always super-saturated (Fig. 2b), a negative correlation between N_2O and O_2 may indicate N_2O production mainly
by nitrification. This is supported by the corresponding stronger correlation between N_2O and NH_4^+ than between N_2O
365 and NO_3^- , the opposite to what we found for swamp rivers. Although published measurements of N_2O production via
in-stream nitrification are lacking, nitrification rates may frequently exceed denitrification rates in streams and rivers
(Richardson et al., 2004; Arango et al., 2008) and nitrification rates are estimated to exceed denitrification rates two-
fold globally (Mosier et al., 1998). In addition to O_2 and DIN amount and speciation, pH and dissolved organic
carbon are important in controlling net N_2O production via nitrification and denitrification (Baulch et al., 2011). It
370 has been suggested that due to variable N_2O yields from these processes simple diagnostic relationships for N_2O
production in rivers may prove elusive (Beaulieu et al., 2008).

In summary, our data have allowed us to draw some conclusions regarding the production and cycling of CH_4 and
 N_2O in contrasting rivers of the ROC. However, for both gases an unequivocal identification of the primary controls
375 of their riverine distributions would require additional detailed measurements.

4.2 CH_4 and N_2O emissions in the wider context

As with the concentration measurements, there are few data for African rivers with which to compare our CH_4 and
 N_2O emissions estimates (Table 3). Previously published emissions estimates are listed in Table 4. For three rivers
380 of the Ivory Coast Koné et al. (2010) report 25 to 1187 $\mu\text{mol CH}_4 \text{ m}^{-2} \text{ d}^{-1}$, while for the Oubangui Bouillon et al. (2012)
found 38 to 350 $\mu\text{mol CH}_4 \text{ m}^{-2} \text{ d}^{-1}$ and 0.6 to 5.7 $\mu\text{mol N}_2\text{O} \text{ m}^{-2} \text{ d}^{-1}$. For 12 sub-Saharan African rivers Borges et al.
(2015b) give ranges of 0 to 274,600 $\text{mmol CH}_4 \text{ m}^{-2} \text{ d}^{-1}$ and -30 to 299 $\text{mmol N}_2\text{O} \text{ m}^{-2} \text{ d}^{-1}$ using k_w from Aufdenkampe
et al. (2011), and 0 to 461,967 $\text{mmol CH}_4 \text{ m}^{-2} \text{ d}^{-1}$ and -37 to 377 $\text{mmol N}_2\text{O} \text{ m}^{-2} \text{ d}^{-1}$ using k_w from Raymond et al.
(2013). For comparison, CH_4 emissions estimated for the Amazon River were 4625 to 12562 $\mu\text{mol m}^{-2} \text{ d}^{-1}$ (Bartlett
385 et al. 1990) and the range for CH_4 in temperate rivers is ~ 0 to 22000 $\mu\text{mol m}^{-2} \text{ d}^{-1}$ (De Angelis and Scranton 1993;
Lilley et al. 1996; Jones and Mulholland 1998a, b; Hope et al. 2001; Abril and Iversen 2002). Guérin et al. (2008)
reported N_2O emissions ~ 0.25 to 6.0 $\mu\text{mol m}^{-2} \text{ d}^{-1}$ for the Amazon River and floodplain while Soued et al. (2016)
found N_2O fluxes in Canadian boreal rivers to be highly variable across ecosystem types and seasons. These ranged

390 from net uptake $\sim 3.3 \mu\text{mol m}^{-2} \text{d}^{-1}$, somewhat lower than the maximum N_2O uptake we observed in swamp rivers (Table 3), to net emissions $\sim 4.8 \mu\text{mol m}^{-2} \text{d}^{-1}$.

The overall ranges of CH_4 and N_2O emissions from rivers of the ROC (33 to 48705 $\text{mmol CH}_4 \text{m}^{-2} \text{d}^{-1}$; 1 to 67 $\text{mmol N}_2\text{O m}^{-2} \text{d}^{-1}$; Table 3) fall within the ranges encompassed by these earlier estimates for African and temperate rivers. It should be acknowledged that the use of “basin-wide” values for k_w is a necessity that takes no account of spatial and temporal k_w variability, that our emissions based on k_w derived from Raymond et al. (2013) are 30% higher than those derived from Aufdenkampe et al. (2011) and that other available k_w parameterizations show five-fold variability (Barnes and Upstill-Goddard, 2011). Additionally, we did not measure CH_4 ebullition fluxes. Borges et al (2015b) report an average 20% ebullition contribution to total CH_4 emissions from the Congo and Zambezi, although their maximum estimates are considerably higher than this. For some other tropical rivers and lakes ebullition is thought to account for 30-98% of total CH_4 emissions (Melack et al., 2004; Bastviken et al., 2011; Sawakuchi et al., 2014). The uncertainties related to k_w notwithstanding, our emissions estimates for CH_4 at least, are therefore probably conservative.

5 Conclusions

Our data from the ROC support the growing consensus that river systems in Africa may be disproportionately large contributors to the global freshwater sources of tropospheric CH_4 and N_2O , as they are for CO_2 . Even so, the potential for significant sinks lends a note of caution for N_2O . The wide ranges of emissions estimates for CH_4 and N_2O now available for African rivers clearly illustrate the difficulty in deriving representative total emissions given both the comparatively small size of the available data set and the various approaches that are typically used to derive these emissions. This applies not only to African rivers, but to tropical rivers in general and indeed to freshwaters globally. At least equally important is an insufficiently mature understanding of the processes that link emissions to the environmental controls of process rates and their temporal variability, and to river catchment characteristics that include sources and seasonality of organic inputs and variability in the balance between baseflow and surface runoff. Our understanding of these interactions must improve if the system responses to future climate and land use changes are to be predicted and planned for. Lastly, the measurement of CH_4 and N_2O , data calibration and the emissions estimates for aquatic systems deriving would all benefit from agreed, standardized protocols. There are currently no internationally agreed calibration standards for CH_4 or N_2O but this is now being addressed via an international SCOR (Scientific Committee on Oceanic Research) Working Group (WG-143: <https://portal.geomar.de/web/scor-wg-143/home>), which is engaged in inter-laboratory calibration and the dissemination of high quality calibration gases. WG-143 welcomes additional interest from the wider aquatic CH_4 and N_2O research community.

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Figure 2. (a) Log percent dissolved methane saturation vs percent dissolved O₂ saturation and (b) percent dissolved nitrous oxide saturation vs percent dissolved O₂ saturation for rivers of the Republic of Congo during the wet season: open circles, savannah rivers; filled black circles, swamp forest rivers; filled grey circles, tropical forest rivers.

695 **Figure 3.** (a) Dissolved nitrate vs dissolved ammonium and (b) percent nitrous oxide saturation vs DIN (NO₃⁻ + NH₄⁺) for rivers of the Republic of Congo during the wet season: open circles, savannah rivers; filled black circles, swamp forest rivers; filled grey circles, tropical forest rivers.

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705 **Table 4.** Emissions of CH₄ and N₂O published for African rivers: (A) refers to emissions estimated using the relationship of Aufdenkampe et al (2011) and (R) refers to emissions estimated using the relationship of Raymond et al. (2013).

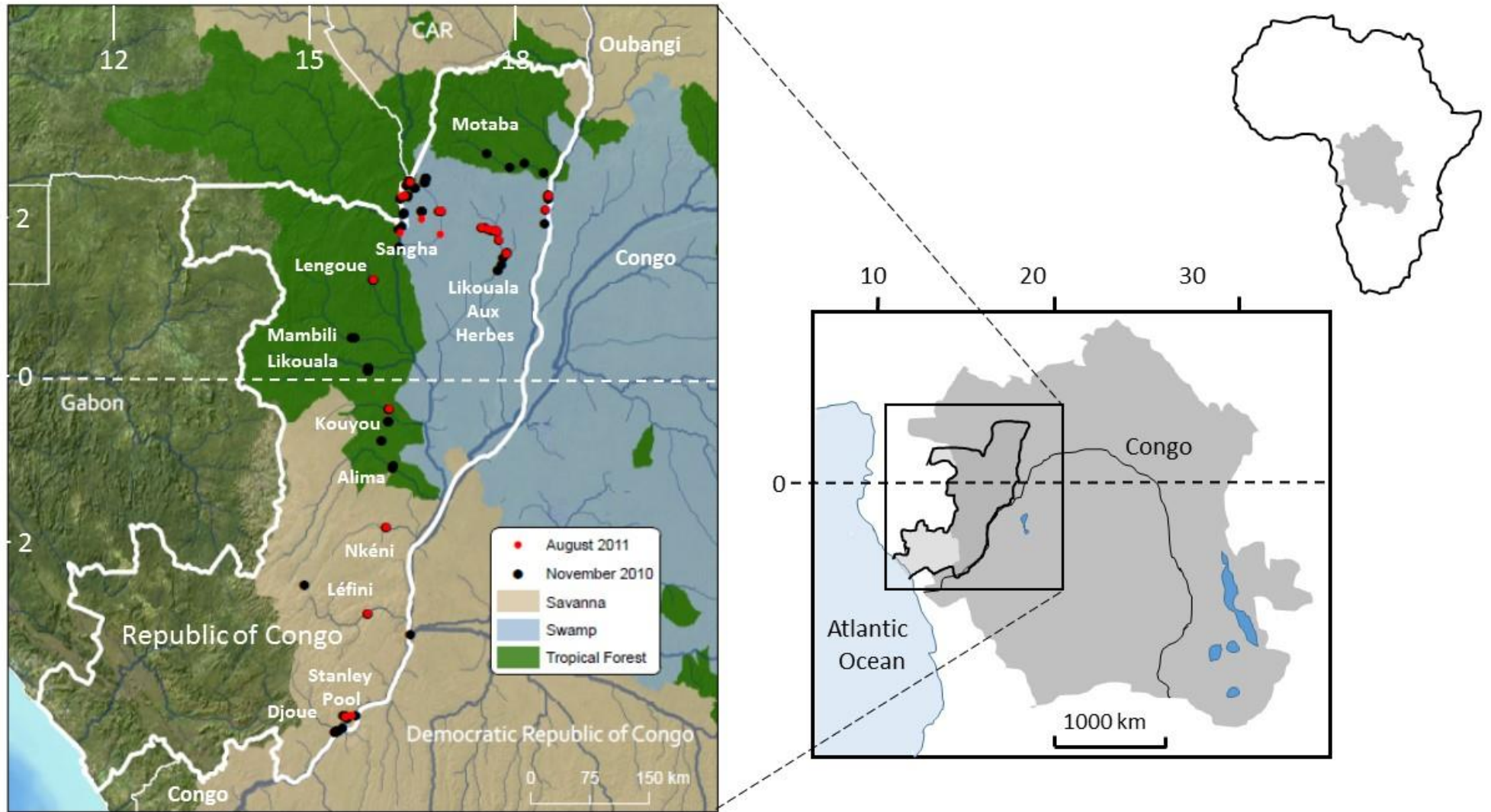


Figure 1.

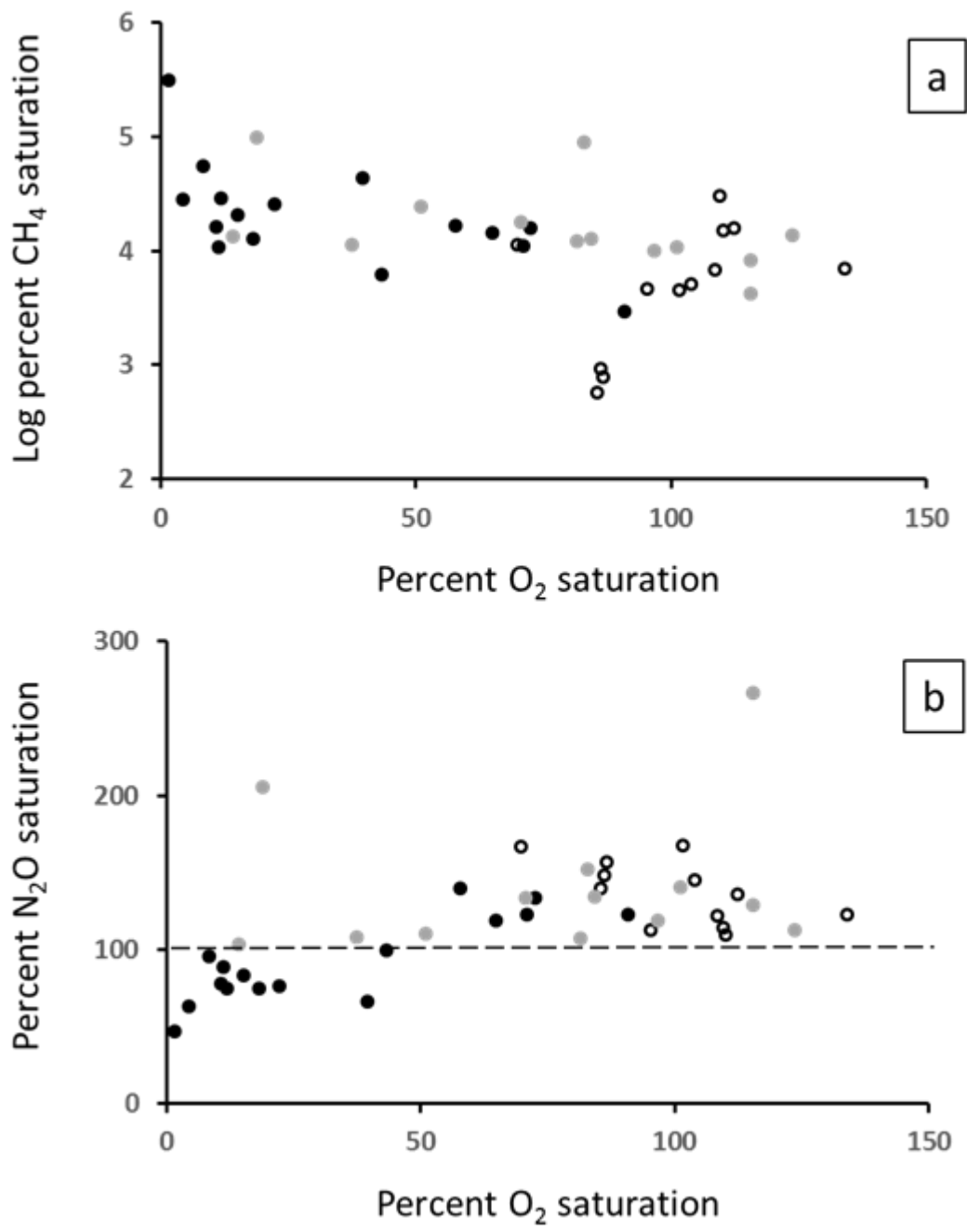


Figure 2

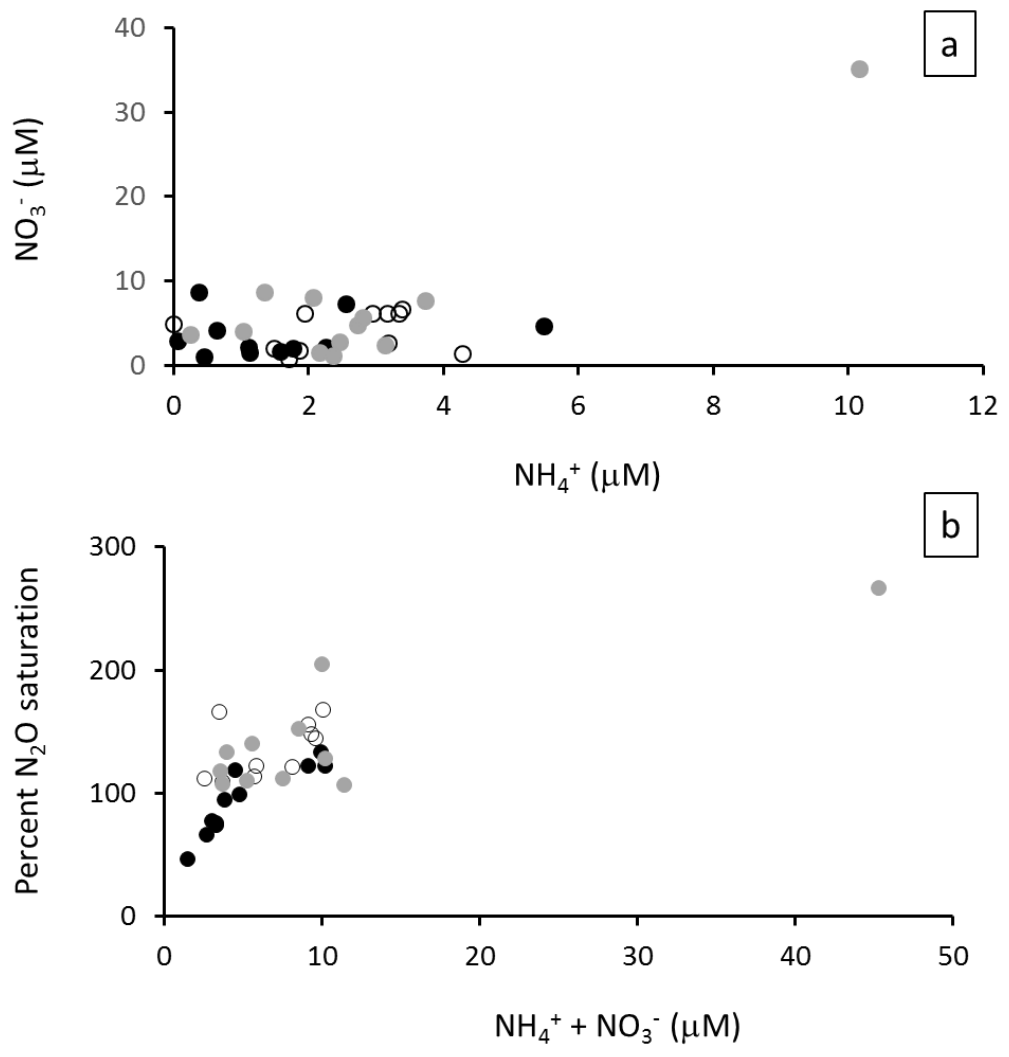


Figure 3

Table 1. Relevant physical characteristics of rivers studied in this work. All rainfall data are from Laraque et al. (2001), Djoue catchment area and discharge data are from Laraque et al. (1994) and all other data are from Laraque et al. (2009).

River/Tributary	Catchment Area km ²	Discharge m ³ s ⁻¹	Rainfall mm
Congo	3500000	40600	1528
Alima	21030	1941	1709
Nkèni	8000	261	1662
Léfini	14000	400	1615
Djoue	5740	140	1547
Likouala aux-Herbes	25000	267	1622
Sangha	211120	1941	1511
Likouala Mossaka	69800	928	1689
Kouyou	16000	191	1566
Lengoué	12125	155	
Mambili	13700	161	
Motaba	772800	4000	

Table 2. Dissolved CH₄ and N₂O in Republic of Congo rivers

Land cover	Date	CH ₄ nmol L ⁻¹			CH ₄ saturation %			N ₂ O nmol L ⁻¹			N ₂ O saturation %		
		Range	Mean	Median	Range	Mean	Median	Range	Mean	Median	Range	Mean	Median
Savannah	All	15-793	204 ± 195	150	571-29721	7442 ± 73427	5444	7.6-11.9	8.9 ± 1.1	8.6	6-167	131 ± 19	122
	Wet	15-793	229 ± 141	153	571-29721	8505 ± 4111	5936	7.6-11.9	9.1 ± 2.2	9.0	106-167	136 ± 34	137
	Dry	66-243	144 ± 70	150	1660-8717	4888 ± 2878	5444	8.0-9.2	8.4 ± 0.5	8.2	113-130	119 ± 6	117
Swamp	All	17-9553	955 ± 2150	299	659-354444	35025 ± 79578	10913	3.2-15.3	7.2 ± 2.2	7.2	47-205	101 ± 30	98
	Wet	77-8394	1052 ± 1992	456	2945-309533	38485 ± 73488	16292	3.2-10.3	6.5 ± 2.1	5.7	47-139	92 ± 28	86
	Dry	17-9553	858 ± 2359	196	659-354444	31565 ± 87529	7090	5.3-15.2	7.9 ± 2.2	7.7	78-205	111 ± 29	109
Tropical Forest	All	11-2730	551 ± 784	321	440-98582	16966 ± 22958	11302	7.6-20.6	89.6 ± 3.3	8.4	103-266	135 ± 42	28
	Wet	110-2730	691 ± 852	331	4205-98582	21205 ± 24858	12729	7.6-20.6	10.0 ± 3.7	5.7	103-266	140 ± 47	128
	Dry	11-232	94 ± 103	67	440-82183	3319 ± 3360	2310	8.0-8.5	8.3 ± 0.2	8.3	108-132	120 ± 12	120

Table 3. Emissions of CH₄ and N₂O to air from rivers in the Republic of Congo. Negative values in parentheses indicate uptake from the atmosphere.

Land cover	Date	CH ₄ emission flux ($\mu\text{mol m}^{-2} \text{d}^{-1}$)						N ₂ O emission flux ($\mu\text{mol m}^{-2} \text{d}^{-1}$)					
		Raymond et al. (2013)			Aufdenkampe et al. (2011)			Raymond et al. (2013)			Aufdenkampe et al. (2011)		
		Range	Mean	Median	Range	Mean	Median	Range	Mean	Median	Range	Mean	Median
Savannah													
	All	61-4030	1027 ± 997	752	47-3081	785 ± 762	575	3-25	11 ± 7	8	3-19	8 ± 5	6
	Wet	61-4030	1159 ± 1157	876	47-3081	887 ± 885	670	3-25	12 ± 7	13	3-19	9 ± 5	10
	Dry	321-855	595 ± 356	602	245-654	455 ± 272	460	5-11	7 ± 2	6	4-8	5 ± 2	5
Swamp													
	All	74-48705	4856 ± 10964	1512	56-37245	3713 ± 8384	1196	(-19)-41	1 ± 11	(-1)	(-15)-31	1 ± 9	(-1)
	Wet	378-42798	5349 ± 10160	2313	289-32728	4090 ± 7769	1769	(-19)-15	(-3) ± 10	(-5)	(-15)-12	(-2) ± 8	(-4)
	Dry	74-48705	4363 ± 12029	985	56-37245	3336 ± 9198	753	(-8)-41	4 ± 11	3	(-6)-31	3 ± 9	3
Tropical Forest													
	All	43-13911	2795 ± 4000	1624	33-10638	2318 ± 3059	1242	1-67	13 ± 16	9	1-51	10 ± 13	7
	Wet	549-13911	3512 ± 4387	1677	420-10638	2686 ± 3324	1282	1-67	15 ± 19	10	1-51	11 ± 14	7
	Dry	43-1167	465 ± 526	326	33-892	356 ± 403	249	3-11	7 ± 3	7	3-8	15 ± 3	5

Table 4. Emissions of CH₄ and N₂O published for African rivers: (A) refers to emissions estimated using the relationship of Aufdenkampe et al (2011) and (B) refers to emissions estimated using the relationship of Raymond et al. (2013).

River	CH ₄ emission flux ($\mu\text{mol m}^{-2} \text{d}^{-1}$)		N ₂ O emission flux ($\mu\text{mol m}^{-2} \text{d}^{-1}$)		Reference
	Range	Mean	Range	Mean	
Conoé, Ivory Coast		288 ± 107			Koné et al (2010)
Bia, Ivory Coast		155 ± 38			Koné et al (2010)
Tanoé, Ivory Coast		241 ± 91			Koné et al (2010)
Ivory Coast (all)	25-1187				Koné et al (2010)
Oubangui	38-350		0.6-5.7		Bouillon et al (2012)
Congo		14296 18534		15 (A) 19 (R)	Borges et al (2015b)
Ivory Coast		1003 1667		(A) (R)	Borges et al (2015b)
Ogooué		2115 4668		13 (A) 28 (R)	Borges et al (2015b)
Niger		502 583		4 (A) 5 (R)	Borges et al (2015b)
Zambezi		8348 13597		2 (A) 2 (R)	Borges et al (2015b)
Betsiboka		1305 3493		4 (A) 9 (R)	Borges et al (2015b)
Rianila		1923 4537		5 (A) 12 (R)	Borges et al (2015b)
Tana		568 604		6 (A) 6 (R)	Borges et al (2015b)
Athi-Galana-Sabaki		1156 1374		16 (A) 19 (R)	Borges et al (2015b)
Nyong		18019 28579		(A) (R)	Borges et al (2015b)