

Dear Editor,

Thank you for considering the second revision of our manuscript “**Phosphorus addition mitigates N₂O and CH₄ emissions in N-saturated subtropical forest, SW China**” (bg-2016-470) by Yu et al., for publication in Biogeosciences.

We have revised our manuscript (R2), based on the reviewers’ comments. In the point-to-point response, the changes refer to the line numbers in the manuscript version with marked-up changes (attached following the response).

We thank you again for handling our manuscript.

Best regards,

Longfei Yu on behalf of all coauthors

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Comments from Reviewer #2

1) Scientific Significance:

The primary objective of this study is to determine how phosphorus availability influences the efflux of N₂O and CH₄ in a nitrogen-saturated pine-dominated subtropical forest. While we have a fair understanding how the availability of nitrogen influences these processes, we have a very poor understanding how phosphorus can directly, or indirectly influence greenhouse gas production. Prior research on this subject has been inconclusive and this study will help add to the body of literature to advance our understanding of soil nutrient availability and climate forcing.

2) Scientific Quality:

Much improve from the early versions – mixed model is the way to go. My primary concern is the low replication (just three 20 x 20 m plots for each treatment) and perhaps overall discussion needs to be tempered.

R: We thank the reviewer for the critical but constructive comment. Our study explores the effect of phosphorus (P) on N cycling in the N-saturated Tieshanping (TSP) forest (Huang et al., 2015). Indeed, our field observations indicate that P application significantly reduces N₂O and CH₄ emissions from upland soils at TSP. We agree that the current field set-up has its limitation and that our findings cannot be readily scaled up. For this, further research is needed, including work at other sites. The experiment was done in triplicate, as more replicates were not feasible given the available man-power and budget. In terms of replication, our experiment, investigating GHG emissions, does not differ much from others, which also used 20 x 20 m² plots in triplicates (Muller et al., 2015; Martinson et al., 2013) or even 10 x 10 m² plots in triplicates (Zheng et al., 2016; Wang et al., 2014) for each treatment.

3) Presentation Quality:

The discussion needs minor revisions. A fair amount of the discussion is just a summary of the results with modest interpretation or explanation. While it is fine to report how these results are similar/dissimilar with other work, it needs to go beyond that and provide new insights and tieback with the present study. What do the result consistent with this study have in common, but uncommon with work that is inconsistent? More effort is needed in the discussion to explain the ‘why’ of the results. Certain interesting passages need to be developed (e.g. lines 313-315)

R: We thank the reviewer for the general suggestion. In the revised version, we now expand our explanation of the results. Major changes are made in three sections of the discussion:

1) P effect on soil NO₃⁻ concentration (line 301-310): we attribute the attenuation of soil NO₃⁻ concentration to P stimulation of N uptake by plants and soil microorganisms. Based on the reported increase of N mineralization and nitrification rates in a similar site from South China, we suggest that microbial uptake of N is less likely than the plant uptake. Although this is not supported by our measurement of tree biomass and foliar N during two years, we provide an alternative explanation, which is N uptake by understory vegetation.

2) Comparison of P effect on N₂O emission with other studies (line 324-335): In comparison with TSP, the slow response of N₂O emission to P addition at the Ecuadorian site is suggested to be a result of both low ambient N deposition and low P dose. The Chinese site (Dinghushan) receives similar inorganic N input via throughfall as TSP. But we suggest the Dinghushan site to be less N-rich than TSP, based on the lower KCl-extractable inorganic N content in soil and lower flux of N leaching at Dinghushan. This point is now clearly made in the discussion.

3) We improve our explanation of N inhibition effects on CH₄ uptake by elaborating on known CH₄/NH₄⁺ substrate competition for methane monooxygenase (line 372-373) and NO₃⁻/NO₂⁻ toxicity for methanotrophs (line 377-382). In the discussion of P effect on CH₄ uptake, we focus mainly on the direct

P effects on methanotrophs and indirect P effects by reducing inorganic N levels in soils (line 386-398). The indirect mechanism seems more likely, as support by our observation.

Line 363: I question that the study actually demonstrates that high deposition has shifted these soils to regional hotspots for N₂O and CH₄ production – this was not tested. Also, this is difficult to say without historical data or somehow experimental removing/reducing N deposition. Yes, adding P has reduced NO₃ concentration, but this result is fairly modest, especially at 20 cm.

R: We agree with the referee that historic data on the relationship between atmospheric N deposition and N concentrations in soil water as well as emission rates of N₂O and CH₄ would be needed, to ultimately prove the connection between elevated N deposition and N₂O and CH₄ emissions in the sub-tropical forest region. As such data do not exist for TSP, our conclusions are solely based on comparison with similar research sites and on manipulation experiments at TSP. For example, Huang et al (2015) showed that a doubling of N deposition at TSP caused a doubling of N leaching (as nitrate), simultaneously in increasing N₂O emission (Liu WJ, personal communication). Our manuscript adds further evidence to this, as we find that P addition causes a significant decline in NO₃⁻ concentrations in soil water in the surface horizons (O/A and AB; Fig. 1), while emission rates of N₂O and CH₄ decrease (Figs. 4 and 6). Also other studies have reported strong correlation between NO₃⁻ concentration in soil water and N₂O emission rates (Gundersen et al., 2012).

Earlier studies at the TSP site indicate chronically elevated N deposition levels, high inorganic N concentration in soil water and strong soil acidification (Zhu et al., 2013; Larssen et al., 2011; Huang et al., 2015). These factors contribute to enhanced N₂O emission and reduced CH₄ uptake in soils (Liu et al., 2010; Le Mer and Roger, 2010). Among other studies in Southern Chinese forests (Tang et al., 2006, Fang et al., 2009; Zhang et al., 2008), our study reports the highest N₂O fluxes and lowest CH₄ uptake rates (or even net emission).

In addition, our recent ¹⁵N tracer study at TSP (Yu et al., 2017), has shown that the enriched ¹⁵N signals in emitted N₂O is identical to those in soil NO₃⁻ extracted from O/A horizon rather than AB horizon. This denotes the importance of NO₃⁻ availability in surface soils to N₂O emission. Also for the CH₄ oxidation, the N inhibition effect should be most important in surface soil (Bodelier and Laanbroek, 2004), where methanotrophic activity is more active with more aerobic condition. Therefore, our observation of significant decreases in NO₃⁻ concentration from O/A and AB horizons do support that P addition migrates the N₂O and CH₄ emissions from soil.

Comments from Reviewer #3

The manuscript by Longfei Yu and others presents a replicated forest fertilization experiment in an acidified and N-saturated Masson pine-dominated forest at TieShanPing, SW China. The experiment tests the role of mineral P fertilization in regulating nitrous oxide and methane emissions (uptake). Researchers measured soil water NO₃⁻ concentrations, N₂O emissions, CH₄ emissions (uptake), forest productivity, litter fall, litter chemistry, and soil characteristics prior to and after a one-time fertilization with 79 kg P ha⁻¹ (NaH₂PO₄).

The Author's found that P fertilization results in declines in soil NO₃⁻ and suppressed emissions of N₂O, and CH₄ (not immediately, but over the long term). With P addition TSP soils switched from a CH₄ source to a sink. Elevated biomass production was not observed over the 18-month experimental period. However, understory biomass was not assessed. Based on these results, Authors hypothesized that P additions resulted in increase NO₃⁻ uptake by plants and microbes leaving less for denitrification. Also, P addition was thought to lessen the NH₄⁺ inhibition of methane oxidation.

Overall comments:

This manuscript is well written and presents a topic that is of interest generally to the readers of Biogeosciences. I have a couple of concerns that should be addressed prior to publication. First, I feel that these results and their interpretation would be easier to follow if there were a set of explicitly stated hypotheses. There is one hypothesis stated in the Abstract (that concerns the results), but not in the main body of the manuscript.

R: Following the reviewer's recommendation, we have changed our objectives to more explicit hypotheses. We now present three key hypotheses: 1) P addition stimulates tree growth; 2) P addition decreases soil inorganic N availability and 3) P addition reduces N₂O and CH₄ emissions (line 105-109).

Regarding the description of the experimental design and sampling, the description in the methods doesn't seem to reflect the data that is presented for N₂O and CH₄ (L167-179). From the methods, I gather that N₂O and CH₄ were measured a total of four times, but clearly more data points are presented. Please clarify in the text how frequently measurements were made over the entire experiment. Other clarifying points are made below in the line-by-line comments.

R: Thanks for the suggestion. To avoid confusion, we now describe the frequency of gas emission sampling together with the soil pore water sampling. In line 178-180, we add "The measurements were conducted bi-monthly in the dry and dormant season and monthly during the growing season, simultaneously with the sampling of soil pore water". In line 181, we put "also" before the description of "short-term" intensive sampling.

With the Results, at times the text is confusing because the treatment effects of P additions and the seasonal/temporal patterns are explained simultaneously. I would recommend some minor reorganizing of this information. Perhaps start with the overall seasonal patterns and then state the treatment effects or the opposite.

R: We appreciate reviewer #3's advice. We have now reorganized the results of N₂O and CH₄ fluxes, by starting with a paragraph for seasonal/temporal patterns of fluxes from Reference plots (line 243-255). The P effect on N₂O and CH₄ fluxes are then presented in another two paragraphs.

Line-by-line comments:

L16: Change GHG to green house gas

R: OK.

L18: If this is a single fertilizer event is it necessary to have the unit yr⁻¹?

R: We have revised it as “Here, we report N₂O and CH₄ emissions together with soil N and P data for a period of 18 months following a single P addition (79 kg P ha⁻¹, as NaH₂PO₄ powder)” (line 17-19).

L20: Rephrase this sentence to read “We observed a significant decline in soil water NO₃- concentrations (5 and 20 cm depths) and in soil N₂O emissions following P addition.”

L21: It is unclear if this number is the amount of reduction or if it represents the total emission. Please clarify

R: Agree. This number for emission reduction is now deleted. See line 20-22.

L23-24: The “As for N₂O” is a confusing way to begin this sentence. Can you revise to something like “P addition significantly decreased CH₄ emissions, turning TSP soils from a net source to a net sink.” I’m sure the Authors will have a more eloquent way of conveying that message.

R: It is now changed to “P addition significantly decreased CH₄ emissions and turned the soil from a net source into a net sink” (line 24-26)

L26-27: It’s my preference to put this caveat in the discussion or that it’s rephrased. The current wording suggests that you measured understory and that there was an increase in understory biomass.

R: This sentence is now rephrased, as “Within the 1.5 years after P addition, no significant increase of forest growth was observed and P stimulation of forest N uptake by understory vegetation remains to be confirmed” (line 27-30)

L48-49: ‘frequently shifting aerobic conditions’ is awkward please revise. Perhaps this is better put in terms of aerobic and anaerobic?

R: OK (line 50-53).

P4 L56: Consider changing ‘mineral’ to ‘inorganic’

R: Done.

L100: The hypothesis is stated in the abstract but not the main text. Please include in text prior to the objectives.

R: Please refer to our earlier reply to the overall comments. See also line 105-109.

L105: It’s unclear why the study site name is in quotations. L116: TSP hilltop is not intuitive. Please explain in text

R: The quotation is removed now. “TSP hilltop” refers to upland soils from the hillslopes at TSP, which has been previously documented by Zhu et al., 2013b. Here, we rephrase “at TSP hilltop” as “on the hillslopes” (line 123-124).

L121-122: Can you state over what time period the decline in growth has occurred here?

R: Wang et al. (2007) and Li et al. (2014) reported the decline in forest growth at TSP during 2001-2004 and 2004-2012, respectively. So we rephrase the sentence as “Strong soil acidification has been reported to cause severe decline in forest growth at TSP since 2001 (Li et al., 2014; Wang et al., 2007)” (line 129-131).

L128: Rather than an *, please use ×

R: OK. Change is made in line 136.

L141: Can you report the Na⁺ concentrations of the Reference plots? L157: Change (2 mm) to (2mm × 2mm)

R: We add “(0.52-1.31 mg L⁻¹)” for the reference plots (line 149). More specific values for different soil depths could be found in Table S2 of the support information.

L165: I think part of the instrument name is missing. Should this be ‘inductively coupled plasma atomic emission spectroscopy’? For all makes/models of equipment here and throughout, please add the location information.

R: OK. See line 174-175.

L171: Change to “...into 12 mL pre-evacuated glass vials... (Chromacol, UK).” L172: I would recommend splitting this into another sentence: “Vials were over pressurized to avoid contamination during sample transport.”

R: It is modified as “20 ml gas samples were injected into pre-evacuated glass vials (12 ml) crimp-sealed with butyl septa (Chromacol, UK), maintaining overpressure to avoid contamination during shipment”, as shown in line 182-184.

L173: Is ‘Mixing ratio’ what you mean or should this be ‘Fluxes of ...’ or ‘Concentrations of...’

R: Yes. Mixing ratio is the direct result that we obtain from gas analysis on gas chromatograph coupled with ECD. Mixing ratios of N₂O and CH₄ are commonly expressed as ppb or ppm, similar to “concentration”. For fluxes, we need further calculation based on the change of gas mixing ratio with time, as described later in the same paragraph.

L190: Please specify if the same trees were measured at each time point, this is critical to the interpretation of these data.

R: Yes, they had been marked since the first measurement. This is now explained (line 204).

L194: Rather than ‘sum of precipitation’ can this be termed ‘daily total precipitation’? Please provide the time period over which precipitation and temperature were measured.

R: OK (line 207-208).

L198: I gather from the methods that gas samples were collected from August 2013 forward, but only during the month of May (2, 7, 10, and 12). This doesn’t reflect all of the data points that are shown in Figures 2 and 5. I would insist that the Authors add clarity to the methods or only show data that were collected in this study.

R: Please refer to the earlier reply to the overall comments of the reviewer. Changes are made accordingly in line 178-180.

L206: It is unclear if fluxes of 'litterfall' nutrients were scaled to the biomass production. Or was litter biomass a component of the overall biomass calculation? L210-213: For tree growth, how were the 3 different time points treated? Please be specific.

R: The fluxes of litterfall were not used for tree biomass evaluation. Instead, we used allometric models (specifically for masson pine; Li et al. 2011 and Zeng et al. 2008) to estimate the tree biomass, based on measured diameters at breast height (DBH). We specify now that DBH is used for tree biomass estimation (line 205).

The data for tree growth refer to Table S3, which includes tree biomass, 500-needle weight and needle nutrient contents. The three samplings were tested for treatment effect separately with one-way ANOVA (line 223-226).

L226: The phrase 'sum of charge of dissolved base cations is unclear', at any rate, it would be more appropriate to say that charge was significantly different between fertilized and unfertilized. I am curious if the 'charge' decreases in the P treatment because of the increase of Na⁺. Can you please address?

R: We thank the reviewer for the suggestion. This phrase is changed to "the overall cationic charge" (line 240-241). The increase in total cationic charge due to Na⁺ addition should be modest, while the charge of dissolved Ca²⁺ is far more important. P addition stimulated the uptake of N, thus resulting in a decline in soil NO₃⁻ concentration. As NO₃⁻ is the major anion in soil water, the decrease of NO₃⁻ concentration leads to a decline in cation concentration (Figure S3). Thus, the observed decrease in cationic charge is a direct effect of the decline in mobile anions (line 409-411).

L232-234: Please report the block effect here. L240: Was there a significant block effect that could be reported here?

R: The block effects were clearly described in the sentences that follow, in line 247-249 for N₂O and in line 250-254 for CH₄.

L236-238: Rephrase to read: The P addition resulted in a 50% (average 3 kg N ha⁻¹ yr⁻¹) reduction of cumulative N₂O emissions (Fig. 3). Please add +/- Stderror if it is available

R: We have revised it to "The P addition resulted in a 50% (3 kg N ha⁻¹ yr⁻¹ on average) reduction of cumulative N₂O emission (Fig. 3)" (line 257-259).

L238: Change was to were.

R: "effects" is changed to "effect" (line 259).

L245: Should this unit be CH₄-C here and throughout? Also can you add +/- Stderror here?

R: The units for N₂O and CH₄ fluxes have been checked throughout the manuscript and corrected as g N₂O-N m⁻² hr⁻¹ and g CH₄-C m⁻² hr⁻¹, respectively. For annual emission, they are described as kg N₂O-N ha⁻¹ yr⁻¹ and kg CH₄-C ha⁻¹ yr⁻¹. Here, the CH₄ flux data show a skewed distribution, thus having large standard deviations. Since the treatment effect on CH₄ fluxes had been tested for significance with mixed-effect models, it is not necessary to include standard error here.

L250: What does 138 t ha⁻¹ represent? Is it an average across both years and both treatments? I'm not sure how informative that is. Based on your supplemental data, it looks as if biomass was actually lower in the P addition treatment compared to the Reference treatment.

R: Agree. This value denotes the mean for both years and in both treatments. Based on statistical results, P treatment exhibited no difference from the reference. In the text, we have now described the change as "insignificant" and deleted the value for tree biomass (line 271-272).

L252: The 500g needle weight does not need to be reported here.

R: Agree.

L253: This sentence needs to be clarified to indicate the mechanism responsible for differences in needle chemical composition. "Linked" is vague.

R: It is rephrased as "Between the two samplings in 2013 and 2014, we found differences in chemical composition of the pine needles, but the difference between the Reference and P treatment was not significant" (line 273-275).

L253: 'hardly' is a vague word, please replace. L273: Change mineral to inorganic

R: OK. "hardly" is changed to "not" now (line 276).

L275-279: This sentence is complex and confusing. Please revise, as it seems to contradict your former statement.

R: OK. We have simplified our discussion, with additional details for support. "These findings are consistent with a number of previous studies (Baral et al., 2014; Hall and Matson, 1999; Mori et al., 2014), which attributed the reduction of N₂O emissions in P-treated soils to decreased NO₃⁻ availability and thus less denitrification. The attenuation of soil NO₃⁻ by P addition at TSP may reflect stimulated N uptake by plants and/or soil microorganisms." (299-303).

L273: This paragraph is long and difficult to follow. I believe that the Authors could find a way to make it more streamlined and easier to follow.

R: See the next comment.

L283-292: I think it would be better to put your study into context of others that used similar additions. Perhaps the reference to moderate P additions is a bit of a distraction. I would recommend revising to focus on more similar studies.

R: This paragraph has now been divided into two paragraphs. The first one has been rephrased and focuses only on the key points of our experiment (296-310). The second one compares the fertilization effects of N, P and N+P on N₂O emissions among forests with different N status, involving a space-for-time comparison. While our main discussion still addresses the comparison to similar studies in Southern China, we have modified the comparison with the Ecuadorian study, and focus on different responses of N₂O emissions to P addition in forests with different N status (315-345). See more details in our response to comments from Reviewer #2.

L306: Likewise, the point of this paragraph is not entirely clear. As well, it is unclear if the referenced studies are also covering the short-term (~10day) span of time that is referenced in this manuscript.

R: To make it more concise, we have rewritten this paragraph, starting with studies that found that P addition increases N₂O emission. Then we discuss the possible mechanisms for P stimulation of N₂O emissions, and emphasize that the P effect on denitrifier activity could be rather fast (Mori et al., 2013c). However, we did not see any P effect on N₂O emission from the intensive observation shortly after P application at TSP. This may be attributed to large denitrification potentials in TSP soils, supported by a lab-incubation study (Zhu et al., 2013c). See the changes in line 346-359.

L324: Change production to 'CH₄ production' just to be clear this isn't primary production

R: OK. See line 370.

L353: Change apparently to 'may have'

R: OK. See line 407.

L355: I liked the nice flow and organization of this paragraph!

R: Much appreciated.

L375-376: Is there a citation from your previous work that you can add here? As is, these data don't provide obvious evidence for this.

R: We have added Zhu et al. (2013b) as a reference (line 429). For more details, see our reply to reviewer#2's comments.

L377: Can this statement be qualified by stating 'to overall reduce'

R: Yes, it can. See the changes made in line 430.

L379: GHG is used here and in the abstract, but is not explicitly defined. Please do so. L388: References are not alphabetized consistently.

R: For GHG, please refer to our previous response to the comment on line 16. References have been checked and corrected as requested.

Tables and Figures:

L603: Please change 'Background' to 'Ambient'

R: OK. See line 675.

Table 1. Was 5.0 mg kg⁻¹ the detection limit of the instrument for PH₂O? If so, please just indicate this in the footer of the table rather than dedicating an entire column to the < 5.0 information.

R: OK. This is now changed (line 675-681).

Table 2. Here and throughout, please be consistent that the P treatment is '+P'. Also it is unclear what the letters indicate in terms of significance. Should they not indicate significant differences among the Ref and +P treatment? Or is this across all time points? If so, the analysis should more appropriately be a repeated measures analysis.

R: The different letters indicate significant difference between the Reference and P treatment (line 682-685). This is tested by one-way ANOVA for each sampling, which has been specified in statistics (line 223-226).

Figure 4 and Figure 6: The letters indicating significance are somewhat unnecessary here. The point could be made in either the figure legend or with an asterisk centered above the two boxes. In both figures, I would recommend adding the statistical test that you used.

R: We thank the reviewer's suggestion. We would prefer keeping both the letters indicating significance and adding statistical methods to the figure captions. See line 699 and line 707.

Figure S6: Litter is spelled incorrectly in the axis title

R: OK. Revised.

1 **Phosphorus addition mitigates N₂O and CH₄ emissions in N-**
2 **saturated subtropical forest, SW China**

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10 Article type: Research Article

11 Abstract

12 Chronically elevated nitrogen (N) deposition has led to severe nutrient imbalance in forest soils.
13 Particularly in tropical and subtropical forest ecosystems, increasing N loading has aggravated
14 phosphorus (P) limitation of biomass production, and has resulted in elevated emissions of
15 nitrous oxide (N₂O) and reduced uptake of methane (CH₄), both of which are important
16 greenhouse gases. Yet, the interactions of N and P and their effects on ~~GHG-greenhouse gas~~
17 emissions remain ~~understudied~~elusive. Here, we report N₂O and CH₄ emissions together with
18 soil N and P data for a period of 18 months following a single P addition (79 kg P ha⁻¹ ~~yr⁻¹~~,
19 ~~applied~~ as NaH₂PO₄ powder) to an N-saturated, Masson pine-dominated forest soil at
20 TieShanPing (TSP), Chongqing, SW China. We observed a significant decline in both ~~in~~-NO₃⁻
21 concentrations in soil water (~~at~~-5- and 20-cm depths) and in soil N₂O emissions, following P
22 application, ~~the latter by 3 kg N ha⁻¹ yr⁻¹~~. We hypothesize that enhanced N uptake by plants ~~and~~
23 ~~soil microbes~~ in response to P addition, ~~results~~ resulted in less available NO₃⁻ for denitrification.
24 By contrast to most other forest ecosystems, TSP is a net source of CH₄. ~~As for N₂O~~, P addition
25 significantly decreased CH₄ emissions, ~~and~~ turning turned the soil from a net source into a net
26 sink. Based on our observation and previous studies in South America and China, we believe
27 that P addition relieves N-inhibition of CH₄ oxidation. Within the 1.5 years after P addition, no
28 significant increase of forest growth was observed ~~at TSP, but~~ and P stimulation of forest N
29 uptake by understory vegetation remains to be confirmed ~~we cannot exclude that understory~~
30 ~~vegetation increased~~. Our study indicates that P fertilization of N-saturated, subtropical forest
31 soils may ~~mitigates~~ N₂O and CH₄ emissions, in addition to alleviating nutrient imbalances and
32 reducing losses of N through NO₃⁻ leaching.

33 **Key Word:** N₂O and CH₄ emission, N saturation, Phosphate fertilization, soil CH₄ uptake, acid
34 forest soil.

35 **1 Introduction**

36 Anthropogenic activities have transformed the terrestrial biosphere into a net source of CH₄, N₂O
37 and CO₂, leading to increased radiative forcing (Montzka et al., 2011; Tian et al., 2016). During
38 the last decade, atmospheric concentrations of CO₂, CH₄, N₂O have increased at rates of 1.9 ppm
39 yr⁻¹, 4.8 and 0.8 ppb yr⁻¹, respectively (Hartmann et al., 2013). In China, the exponential increase
40 of reactive nitrogen (N) input into the biosphere since the 1970s has likely led to more carbon (C)
41 being sequestered in the biosphere (Cui et al., 2013; Shi et al., 2015). However, enhanced
42 emissions of N₂O and CH₄ due to chronic N pollution potentially offset the cooling effect by C
43 sequestration (Liu and Greaver, 2009; Tian et al., 2011).

44 Microbial nitrification and denitrification in soils account for about 60% of N₂O emissions
45 globally (Ciais et al., 2013; Hu et al., 2015). Although, microbial activity is often restricted in
46 low pH soils of unproductive forests, surprisingly large N₂O emissions have been reported from
47 acid, upland forest soils in South China (Zhu et al., 2013b). Reported average N₂O fluxes in
48 humid, subtropical forests range from 2.0 to 5.4 kg N₂O-N ha⁻¹ yr⁻¹ (Fang et al., 2009; Tang et al.,
49 2006; Zhu et al., 2013b), which by far exceeds global averages for temperate or tropical forest
50 ecosystems (Werner et al., 2007; Zhuang et al., 2012). This has been attributed to frequently
51 shifting shifts between aerobic and anaerobic conditions in soils ~~aeration conditions~~ during
52 monsoonal summers, promoting ~~both~~ alternating nitrification and denitrification (Zhu et al.,
53 2013b) and ~~to~~ causing large soil NO₃⁻ concentrations due to efficient cycling of deposited N in
54 acid subtropical soils (Yu et al., 2016).

55 Chronically elevated rates of N deposition (30-65 kg N ha⁻¹ yr⁻¹; Xu et al., 2015) have resulted in
56 strong nutrient imbalances in southern Chinese forests, aggravating phosphorus (P) limitation

57 (Du et al., 2016). Phosphorous deficiency in N-saturated forests restricts forest growth and thus
58 ~~constrains limits~~ its capability to retain N (Huang et al., 2015; Li et al., 2016), resulting in ample
59 amounts of ~~mineral-inorganic~~ N (NH_4^+ and NO_3^-) being present in the soil solution. Accordingly,
60 Hall & Matson (1999) observed larger N_2O emission in P-limited than in N-limited tropical
61 forests ~~after~~ one year ~~of~~ ~~after~~ repeated ~~N~~ addition ~~of N~~. Likewise, previous N manipulation
62 studies in forests of South China reported pronounced stimulation of N_2O emissions by N
63 addition (Chen et al., 2016; Wang et al., 2014; Zheng et al., 2016), supporting the idea that P
64 limitation causes forests to be more susceptible to N saturation and N_2O -N loss. In an N-limited
65 tropical montane forest in southern Ecuador, P addition alone ($10 \text{ kg P ha}^{-1} \text{ yr}^{-1}$) had no effect on
66 N_2O emissions during the first two years. However, N_2O emission was smaller when P was
67 added together with N ($50 \text{ kg N ha}^{-1} \text{ yr}^{-1}$) than in treatments with N addition alone (Martinson et
68 al., 2013). After continued fertilization for three years, also P addition alone reduced N_2O
69 emissions at these sites (Müller et al., 2015). In tropical China, with high N deposition ($\sim 36 \text{ kg}$
70 $\text{N ha}^{-1} \text{ yr}^{-1}$; Mo et al., 2008), P addition ($150 \text{ kg P ha}^{-1} \text{ yr}^{-1}$) to an old-growth forest revealed a
71 similar pattern, with no initial effect on N_2O emissions (0-2 years) but a significant longer term
72 effect (3 to 5 years) on N_2O emissions (Chen et al., 2016; Zheng et al., 2016). In a secondary
73 tropical forests in South China, Wang et al. (2014) found no effect on N_2O emissions of P alone
74 ($100 \text{ kg P ha}^{-1} \text{ yr}^{-1}$), and in treatments combining P with N ($100 \text{ kg N ha}^{-1} \text{ yr}^{-1}$), N_2O emissions
75 increased during the wet season. Meanwhile, they observed a significant increase in soil
76 microbial biomass after P addition, which is in line with previous findings in tropical forest soils
77 of South China (Liu et al., 2012). Thus, they attributed the stimulating effect of P addition on
78 N_2O emissions to the larger nitrification and denitrification potential of the increased soil

79 microbial biomass. This was also proposed by Mori et al. (2014), based on results from a short-
80 term incubation study with P addition, excluding plant roots.

81 As the sole biogenic sink for CH₄, upland soils play an important role in balancing terrestrial
82 CH₄ emissions (Ciais et al., 2013; Dutaur and Verchot, 2007). Atmospheric CH₄ uptake in soil is
83 mediated by the activity of methanotrophic bacteria, which oxidize CH₄ to CO₂ to gain energy
84 for growth. Well-drained forest and grassland soils are dominated by yet uncultured, high-
85 affinity methanotrophs residing in the upper soil layers (Le Mer and Roger, 2010). In addition to
86 edaphic factors (pH and nutrients), parameters affecting the diffusion of CH₄ into the soil (soil
87 structure, moisture, temperature) are believed to be ~~the~~ major controllers for CH₄ uptake (Smith
88 et al., 2003). A number of studies have shown that excess N affects CH₄ fluxes in forest soils
89 (Liu and Greaver, 2009; Veldkamp et al., 2013; Zhang et al., 2008b). In general, N addition
90 promotes CH₄ uptake in N-limited soils by enhancing growth and activity of methanotrophs,
91 whereas excessive N input and N saturation inhibit CH₄ oxidation on an enzymatic level by
92 substrate competition between CH₄ and NH₄⁺ (Aronson and Helliker, 2010; Bodelier and
93 Laanbroek, 2004). P addition experiments in N-enriched soils have shown positive effects on
94 CH₄ uptake (Mori et al., 2013a; Zhang et al., 2011), but the underlying mechanisms, i.e. whether
95 P addition affects the methanotrophic community ~~in soils~~ directly or alleviates the N-inhibition
96 effect on CH₄ oxidation through enhanced N uptake (Mori et al., 2013b; Veraart et al., 2015),
97 remain unresolved.

98 Subtropical forests in South China show strong signs of N saturation, with exceedingly high
99 NO₃⁻ concentrations in soil water (Larssen et al., 2011; Zhu et al., 2013b). Little is known about
100 how P addition affects N cycling and N₂O emission in these acidic, nutrient-poor soils. Likewise,
101 the importance of increased mineral-inorganic N concentrations for soil-atmosphere exchange of

102 CH₄, and how this is affected by P fertilization remain to be elucidated for soils of the subtropics.
103 Here, we assessed N₂O and CH₄ fluxes in an N-saturated subtropical forest in SW China under
104 ambient N deposition and studied the effects of P addition on emission rates, nutrient availability
105 and tree growth. ~~The objectives were i) We hypothesized that i) P addition stimulates forest~~
106 ~~growth; ii) stimulated forest growth results in increased N uptake by trees and understory~~
107 ~~vegetation, and thus decreases the soil inorganic N concentration; iii) P addition reduce~~
108 ~~soil N₂O and CH₄ emission and promotes CH₄ uptakes, ii) to investigate the effect of P~~
109 ~~addition on N₂O and CH₄ emission.~~

110 2 Materials and Methods

111 2.1 Site description

112 | The study site “TieShanPing” (TSP) is a 16.2 ha subtropical forest (29° 38' N, 106° 41' E; 450
113 | m a.s.l.), about 25 km northeast of Chongqing, SW China. TSP is a naturally regenerated,
114 | secondary mixed coniferous-broadleaf forest, which developed after clear cutting in 1962
115 | (Larssen et al., 2011). The forest stand is dominated by Masson pine (*Pinus massoniana*) and has
116 | a density of about 800 stems ha⁻¹ (Huang et al., 2015). TSP has a monsoonal climate, with mean
117 | annual precipitation of 1028 mm, and a mean annual temperature of 18.2 °C (Chen and Mulder,
118 | 2007a). Most of the precipitation (> 70%) occurs during summer periods (April to September).
119 | Soils are predominantly well-drained, loamy yellow mountain soil, classified as Haplic Acrisol
120 | (WRB 2014), with a thin O horizon (< 2 cm). In the O/A horizon, soil pH is around 3.7, and the
121 | mean C/N and N/P ratios are 17 and 16, respectively. In the AB horizon, which has a slightly
122 | higher pH, mean C/N is well above 20. The soil bulk density of the O/A horizon (~ 5 cm) is
123 | about 0.75 g cm⁻³. ~~Generally, S~~soil water-filled pore space (10 cm) ~~at TSP hilltop on the~~
124 | ~~hillslopes -generally~~ ranges from 50 to 70% (mean ~ 60%; Zhu et al., 2013b). More details on
125 | soil properties are given in Table 1.

126 | Annual ~~inorganic~~ N deposition at TSP measured in throughfall varies between 40 and 65 kg ha⁻¹
127 | (dominated by NH₄⁺; Yu et al., 2016), while the annual bulk N deposition is from 20 to 30 kg ha⁻¹
128 | (Chen and Mulder, 2007b). According to regional data, annual P deposition via throughfall is <
129 | 0.40 kg ha⁻¹ (Du et al., 2016). Strong soil acidification ~~at TSP~~ has ~~been reported to resulted~~
130 | ~~in~~cause severe decline in forest growth ~~at TSP since 2001~~ (Li et al., 2014; Wang et al., 2007),
131 | and ~~in a decreased in~~ abundance and diversity of ground vegetation (Huang et al., 2015).

132 Pronounced N saturation with strong NO_3^- leaching from the top soil has aggravated P deficiency
133 at TSP (Huang et al., 2015). The total P content in the O/A horizon is $\sim 300 \text{ mg kg}^{-1}$, while
134 ammonium lactate-extractable P is smaller than 5 mg kg^{-1} (Table 1).

135 2.2 Experimental Design

136 Three blocks, each having two $20 \text{ m} \times 20 \text{ m}$ plots, were established on well drained soils of a
137 gently sloping hillside. Adjacent plots were separated by at least 10-m buffer zone. In each block,
138 plots were randomly assigned to a Reference and a P treatment. On 4 May 2014, a single dose of
139 P fertilizer was applied as solid $\text{NaH}_2\text{PO}_4 \cdot 2\text{H}_2\text{O}$, at a rate of $79.5 \text{ kg P ha}^{-1}$. The amount of P
140 added was estimated from P adsorption isotherms (Supplementary Materials, Table S1 and
141 Figure S1), to ensure significantly increase in soil available P. To apply P fertilizer evenly, we
142 divided each plot into a $5 \text{ m} * 5 \text{ m}$ grid and broadcasted the powdered fertilizer by hand in each
143 grid cell. The P dose applied at TSP was intermediate as compared to the $10 \text{ kg P ha}^{-1} \text{ yr}^{-1}$
144 applied by Müller et al. (2015) to a mountain forest in Ecuador and the $150 \text{ kg P ha}^{-1} \text{ yr}^{-1}$ applied
145 by Zheng et al. (2016) to a subtropical forest in South China.

146 The addition of $\text{NaH}_2\text{PO}_4 \cdot 2\text{H}_2\text{O}$ at the P-treated plots also resulted in an input of 59.0 kg ha^{-1} of
147 sodium (Na). One month after the fertilizer application, Na^+ concentrations in soil water of the P
148 treatments were about 5 mg L^{-1} at 5-cm depth and 3 mg L^{-1} at 20-cm depth (Table S2). Although
149 somewhat larger than in the ~~reference-Reference~~ plots ($0.52\text{-}1.31 \text{ mg L}^{-1}$), the Na^+ concentration
150 in soil water of the P treatments are unlikely to have exerted a ~~strong-significant~~ negative impact
151 on plant and microbial activities.

152 2.3 Sample collection and analyses

153 Within each plot, three ceramic lysimeters (P80; Staatliche Porzellanmanufaktur, Berlin) were
154 installed at 5- and 20-cm soils near the plot centre in August 2013. To obtain water samples,
155 350-ml glass bottles with rubber stoppers were pre-evacuated, using a paddle pump, and
156 connected to the lysimeters for overnight sampling. Between November 2013 and October 2015,
157 we sampled soil pore water bi-monthly in the dry and dormant season and monthly during the
158 growing season. All water samples were kept frozen during storage and transport. Concentrations
159 of NH_4^+ , NO_3^- , potassium (K^+), calcium (Ca^{2+}), and magnesium (Mg^{2+}) in soil water were
160 measured at the Research Center for Eco-Environmental Sciences (RCEES), Chinese Academy
161 of Sciences, Beijing, using ion chromatography (DX-120 for cations and DX-500 for anions).

162 In August 2013, soils from the O/A (0-3 cm), AB (3-8 cm) and B (8-20 cm) horizons were
163 sampled near the lysimeters for soil analysis. Total P and plant-available P contents were
164 monitored in samples collected from the O/A horizons every six months, starting two days
165 before P addition. Soil samples were kept cold ($< 4\text{ }^\circ\text{C}$) during transport and storage. Before
166 analysis, soil samples were air dried and sieved (2 mm). Soil pH was measured in soil
167 suspensions (10 g dry soil and 50 ml deionized water) using a pH meter (PHB-4, Leici, China).
168 Total soil C and N contents were determined on dried and milled samples, using a LECO
169 elemental analyzer (TruSpec[®]CHN, USA). To measure total P, 1 g dry soil was digested with 5
170 ml of 6 M H_2SO_4 (Singh et al., 2005) and measured as ortho-phosphate by the molybdenum blue
171 method (Murphy and Riley, 1962). Ammonium lactate (0.01 M)-extractable P and H_2O -
172 extractable P (P_{Al} and $\text{P}_{\text{H}_2\text{O}}$, respectively) were measured as ortho-phosphate after extraction (1.5
173 g dry soil in 50 ml solution) (Singh et al., 2005). Ammonium oxalate (0.2 M)-extractable Fe, Al
174 and P were measured by inductively coupled plasma optical emission spectroscopy~~inductive~~

175 | ~~coupled plasma (ICP-OES7500; , Agilent, USA)~~ after extraction (1.5 g dry soil in 50 ml
176 | solution).

177 | From August 2013 onwards, we measured N₂O and CH₄ emissions in triplicate close to the
178 | lysimeters, using static chambers (Zhu et al., 2013b). The measurements were conducted bi-
179 | monthly in the dry and dormant season and monthly during the growing season, simultaneously
180 | with the sampling of soil pore water. To investigate the immediate effect of P addition on N₂O
181 | emissions, we also sampled the gas emissions once before (2 May) and three times (7, 10 and 12
182 | May) after the P application. Gas samples (~~20 ml~~) were taken 1, 5, 15 and 30 minutes after
183 | chamber deployment. and 20 ml gas samples were injected into pre-evacuated glass vials (12 ml)
184 | crimp-sealed with butyl septa (Chromacol, UK), maintaining overpressure to avoid
185 | contamination during sample transportation. Mixing ratios of N₂O and CH₄ were analyzed
186 | using a gas chromatograph (Model 7890A, Agilent, USA) at RCEES, equipped with an ECD for
187 | detection of N₂O (at 375 °C with 25 ml min⁻¹ Ar/CH₄ as make up gas), a FID for CH₄ (250 °C;
188 | 20 ml min⁻¹ N₂ as make-up gas) and a TCD for CO₂. Exchange rates between soil and
189 | atmosphere (emission/uptake) were calculated from measured concentration change in the
190 | chambers over time, applying linear or polynomial fits to the concentration data. Cumulative
191 | N₂O emissions over time were estimated by linear interpolation between measurement dates
192 | (Zhu et al., 2013b).

193 | From October 2013 onwards, litterfall was collected during the first week of every month in five
194 | replicates per plot. Litterfall collectors were made of 1 m² nylon nets (1 mm mesh size), held in
195 | place by four wooden poles 0.8 m above the ground. Fresh litter was dried at 65°C. In early
196 | November 2013 and 2014 (at the end of the growing season), we collected current-year pine
197 | needles from several branches of three trees in each plot. The collected needles were dried at

198 65 °C and the dry weight of 500 needles was determined. A subsample was dried at 80 °C and
199 finely milled prior to chemical analysis at the Chinese Academy of Forestry. Total C and N were
200 measured using an elemental analyzer (FLASH 2000; Thermo Scientific; USA). The contents of
201 K, Ca, Mg and P in the needles were determined by ICP-AES (IRIS Intrepid II; Thermo
202 Scientific; USA) after digesting 0.25 g dry weight samples with 5 ml of ultra-pure nitric acid. In
203 November 2013, and 2014, and in February of 2015, we measured the height and the diameter at
204 breast height (DBH) of 6 to 10 Masson pines (~~only those marked in November 2013; with~~ DBH >
205 5 cm) at each plot. ~~These data were~~ DBH was then used to estimate the standing biomass of
206 Masson pines based on standard allometric equations (Li et al., 2011; Zeng et al., 2008).

207 Daily average air temperature and ~~sum of daily total~~ precipitation were monitored from July 2013
208 to November 2015 by a weather station (WeatherHawk 232, USA) placed on the roof at the local
209 forest bureau, in about 1 km distance from the sampling site (Yu et al., 2016).

210 **2.4 Statistical analyses**

211 Statistical analyses were performed using R version 3.3.1 (R Core Team, 2016). All data were
212 tested for normality (Kolmogorov-Smirnov's test) and homoscedasticity (Levene's test) before
213 further analysis. If not normally distributed, the data were normalized by logarithmic
214 transformation. Considering heterogeneity among blocks, temporal variabilities of NO_3^-
215 concentrations, N_2O and CH_4 fluxes were presented separately for each block. For time series
216 data, we used linear mixed-effect (LME) models, to account for both repeated measurements and
217 within-group variance of a stratification variable (block design). LME models were applied to
218 test the effects of P addition on soil N_2O and CH_4 fluxes, NH_4^+ , NO_3^- , K^+ , Ca^{2+} and Mg^{2+}
219 concentrations in soil water, as well as litterfall weight (Koehler et al., 2009; Müller et al., 2015).

220 The analysis was based on data for plot means (the average of 3 subplot replicates) from three
221 blocks. In LME models, treatments (Reference or P addition) were considered fixed effects,
222 while sampling time and plots were treated as random effects. We then assessed the significance
223 of fixed effects through analysis of variance for LME models. One-way analysis of variance
224 (ANOVA, ~~Turkey post hoc test~~) was conducted to examine the treatment effects on soil pH,
225 nutrient contents in organic matter, tree biomass, 500-needle weight and needle nutrient content
226 for each sampling and data of tree growth. Significance levels were set to $p < 0.05$, if not
227 specified otherwise.

228 **3 Results**

229 **3.1 Nutrient concentrations in soil and soil water**

230 Addition of P resulted in a significant increase in soil P content in the O/A horizon, both as P_{AI}
231 and total P (Table 2). However, after 15 months, only P_{AI} indicated an enhanced P status, while
232 total soil P did not differ significantly from background values at the ~~reference-Reference~~ sites. P
233 addition had no significant effect on soil pH, or soil C and N content. The NO_3^- concentration in
234 soil water collected at 5 cm depth varied seasonally, with significantly greater values (30-40 mg
235 $N L^{-1}$) towards the start of the growing season in 2015 (April, Fig. S2), but not in 2014, likely
236 due to dilution by abundant precipitation in February to March 2014. Addition of P resulted in
237 significantly smaller NO_3^- concentrations in soil water at both 5- and 20-cm depths (Fig. 1b). In
238 general, the concentration of NH_4^+ in soil water was small ($< 0.5 mg L^{-1}$) and not affected by P
239 addition (Fig. 1a). At both depths, mean soil water concentrations of Mg^{2+} and Ca^{2+} were
240 significantly smaller in the P-treated than the ~~reference-Reference~~ plots, and the ~~sum-of-charge-of~~
241 ~~dissolved-baseoverall~~ cationic ~~charges~~ declined significantly in response to P addition (Fig. S3).

242 **3.2 N_2O and CH_4 fluxes: effects of P addition**

243 ~~In the Reference plots~~During the experimental period, N_2O fluxes varied seasonally (Fig. 2),
244 showing a significant relationship with daily precipitation (Fig. S4a), but not with daily mean
245 temperature (Fig. S4b). ~~In the Reference plots, m~~Mean N_2O fluxes were generally below $50 \mu g$
246 $N_2O-N m^{-2} hr^{-1}$ in the dry, cool season, but reached values of up to $600 \mu g N_2O-N m^{-2} hr^{-1}$ in the
247 growing season (Fig. 2). Cumulative N_2O emissions were estimated with seasonally averaged
248 fluxes, and they differed greatly among the three blocks (Fig. 3). ~~of which -block 2 had T~~the
249 greatest annual N_2O emission ~~was observed in the Reference plot~~ ($7.9 kg N ha^{-1}$) ~~of block 2.~~ CH_4

250 fluxes in the Reference plots also varied greatly among blocks (Fig. 5). Net-emission of CH₄ was
251 observed in summer 2013 (~ 80 µg CH₄-C m⁻² hr⁻¹) in blocks 1 and 2, whereas block 3 showed
252 CH₄ uptake. From spring 2014 until October 2015, CH₄ fluxes were less variable in all blocks,
253 with values fluctuating around zero. A longer period of net-emission was observed in block 3
254 during the dry season 2014. The fluxes did not correlate with precipitation or air temperature
255 (Figs. S5c&d).

256 Mean N₂O fluxes during the 1.5 years after P addition were significantly smaller in the P
257 treatment than in the Reference (Fig. 4). The P addition resulted in a 50% (3 kg N₂O-N ha⁻¹ yr⁻¹
258 on average) decrease-reduction in-of cumulative N₂O emission by about 3 kg N ha⁻¹ yr⁻¹ on
259 average, which is a 50% reduction (Fig. 3). No immediate effects (within days) of P addition on
260 N₂O emission was observed (Fig. S5).

261 ~~CH₄ fluxes varied greatly between blocks (Fig. 5). Net-emission of CH₄ was observed in summer~~
262 ~~2013 (~ 80 µg C m⁻² hr⁻¹) in blocks 1 and 2, whereas block 3 showed CH₄ uptake. From spring~~
263 ~~2014 until October 2015, CH₄ fluxes were less variable in all blocks, with values fluctuating~~
264 ~~around zero. A longer period of net emission was observed in block 3 during the dry season~~
265 ~~2014. The fluxes did not correlate with either precipitation or air temperature (Fig. S5c&d).~~ In
266 the 1.5 years following P addition, mean CH₄ fluxes indicated net CH₄ emission (~ +3.8 µg CH₄-
267 C m⁻² hr⁻¹) in the Reference, whereas net CH₄ uptake (~ -6.5 µg CH₄-C m⁻² hr⁻¹) was observed in
268 the P treatment (Fig. 6). The suppressing effect of P addition on CH₄ emission was significant, in
269 accordance with what was found for NO₃⁻ concentration and N₂O emission.

270 **3.3 The effect of P addition on tree growth**

271 Throughout the 2-year experimental period, we observed no significant change in tree biomass
272 ~~(138 t ha⁻¹)~~ in response to P addition (Table S3). Likewise, there was no effect of P treatment on
273 the 500-needle weight ~~(13 g on average)~~. Between the two samplings in 2013 and 2014, we
274 found differences in chemical composition of the pine needles, but ~~this effect was not linked to P~~
275 ~~addition~~ the difference between the Reference and P treatment was not significant. Also, the C/N
276 and N/P ratios of the needles (40 and 16, respectively) were ~~hardly not~~ affected by P addition.
277 Monthly litterfall varied seasonally in both Reference and P treatment (Fig. S6), but no
278 significant difference was found between the two treatments.

279 4 Discussion

280 ~~Background~~ N₂O emission rates in the Reference plots were relatively large (Fig. 2), with mean
281 values close to 100 $\mu\text{g N}_2\text{O-N m}^{-2} \text{ hr}^{-1}$ (Fig. 4). This is within the range of N₂O emission rates
282 previously reported for well-drained hillslope soils at TSP (Zhu et al., 2013b), but greater than
283 the rates reported for other forests in South China. For instance, N₂O emission rates averaged to
284 37 $\mu\text{g N}_2\text{O-N m}^{-2} \text{ hr}^{-1}$ in unmanaged sites at Dinghushan (Fang et al., 2009; Tang et al., 2006)
285 and 50 $\mu\text{g N}_2\text{O-N m}^{-2} \text{ hr}^{-1}$ in N-fertilized sites (Zhang et al., 2008a). TSP Reference plots emitted
286 on average 5.3 kg N₂O-N ha⁻¹ yr⁻¹ (Fig. 3), which is about 10% of the annual N deposition (50 kg
287 N ha⁻¹ yr⁻¹) (Huang et al., 2015). These fluxes are well above average fluxes reported for tropical
288 rainforests (Werner et al., 2007). Large N₂O emissions at TSP are likely due to the large N
289 deposition rates (Huang et al., 2015), ~~as suggested by the~~ similar trends of increasing N₂O
290 emissions with increasing N deposition rates has been reported for ~~indicated by data from~~ a wide
291 range of ecosystems (Liu et al., 2009). Also, warm-humid conditions during monsoonal summers
292 may stimulate N₂O emissions (Ju et al., 2011), as monsoonal rainstorms triggered peak fluxes
293 (Pan et al., 2003). The positive correlation between precipitation and N₂O emission peaks (Fig.
294 S4a) may indicate the importance of denitrification as the dominant N₂O source. This is
295 supported by recent ¹⁵N tracing experiments at TSP (Yu et al., 2017; Zhu et al., 2013a).

296 Addition of P caused a significant decline in soil ~~mineral-inorganic N in soil water~~
297 (predominantly NO₃⁻; Fig. 2), particularly during summers, when NO₃⁻ concentrations were
298 relatively high-large (Fig. S2). At the same time, annual N₂O emissions decreased by more than
299 50% (Figs. 3 and 4). These findings are consistent with a number of previous studies (Baral et al.,
300 2014; Hall and Matson, 1999; Mori et al., 2014), which attributed the reduction of N₂O

301 emissions in P-treated soils ~~to decreased~~ NO_3^- availability and thus less denitrification. ~~decreased~~
302 ~~mineral N content, most likely as a consequence of~~ The attenuation of soil NO_3^- by P addition at
303 ~~TSP may reflect~~ stimulated ~~plant N~~ uptake ~~by plants~~ and/or ~~soil microbial assimilation~~ organisms.
304 ~~In a similarly N-rich, tropical forest in South China, Chen et al. (2016) reported a stimulation of~~
305 ~~net N mineralization and nitrification after six years of bi-monthly P addition, despite reduced~~
306 ~~soil NO_3^- concentration. Therefore, it is likely that plant uptake plays a more important role in P-~~
307 ~~induced N retention than immobilization by soil microbes. However, during our study period of~~
308 ~~two years, we did not find significant increase of N uptake based on tree biomass and foliar N~~
309 ~~content measurements (Table S3). An alternative explanation could be that P addition stimulated~~
310 ~~of N uptake by ground vegetation, which remains to be confirmed. It is noteworthy that there was~~
311 ~~no significant correlation between N_2O emission rates and soil water NO_3^- concentration in our~~
312 ~~study (Figs. 2 and S2), suggesting that the suppressing effect of P on N_2O emissions was indirect,~~
313 ~~probably by affecting the competition for mineral N between plant roots and microbes (Zhu et al.,~~
314 ~~2016).~~

315 In contrast to our ~~1.5-year~~ study, P-addition experiments in South Ecuador (Martinson et al.,
316 2013) and South China (at Dinghushan Biosphere Reserve (Zheng et al., 2016) found no effect
317 of a single P addition on N_2O emission during the first two years after application. However,
318 significant reduction in N_2O emission was observed after three to five years ~~with of~~ continuous P
319 addition, both at the Ecuadorian and the Chinese site (Chen et al., 2016; Müller et al., 2015). For
320 the montane forest site in Ecuador, the observed delay in N_2O emission response to P addition
321 may be explained by ~~the moderate amount of P added ($10 \text{ kg P ha}^{-1} \text{ yr}^{-1}$; Martinson et al., 2013).~~
322 ~~Moreover, the experiments were conducted in a forest with~~ the relatively low ambient N
323 deposition ($\sim 10 \text{ kg N ha}^{-1} \text{ yr}^{-1}$) and small N_2O fluxes ($\sim 0.36 \text{ kg N ha}^{-1} \text{ yr}^{-1}$ in the Reference

324 plots) (Martinson et al., 2013; Müller et al., 2015). In addition, the moderate amount of P added
325 (10 kg P ha⁻¹ yr⁻¹; Martinson et al., 2013) could have resulted in an insignificant P effect in the
326 first two years. ~~By contrast, t~~ The Dinghushan site in South China receives ~~28-36~~ kg inorganic N
327 ha⁻¹ yr⁻¹ ~~through wet inorganic N deposition~~ by throughfall (Zheng et al., 2016; Chen et al., 2016;
328 Fang et al., 2008), which is similar to the inorganic N deposition at ~~our site~~ TSP (Chen and
329 Mulder, 2007b; Huang et al., 2015). However, soil KCl-extractable inorganic N (~ 40 mg N kg⁻¹;
330 Zheng et al., 2016) and NO₃⁻ leaching (~ 20 kg N ha⁻¹ yr⁻¹; Fang et al., 2008) at the Dinghushan
331 site are several-fold smaller than at our site (~ 100 mg N kg⁻¹ and ~ 50 kg N ha⁻¹ yr⁻¹,
332 respectively) (Huang et al., 2015; Zhu et al., 2013b). Also, the mean N₂O emission rates in the
333 reference plots (10 µg m⁻² h⁻¹) at Dinghushan were smaller than at TSP (> 50 µg m⁻² h⁻¹; Fig. 4).
334 These indicate that Dinghushan forest has stronger N assimilation and is thus less N-rich than
335 TSP forest. ~~They also observed larger N₂O emission rates (~ 0.88 kg N ha⁻¹ yr⁻¹ in the Reference~~
336 ~~plots) than in the Ecuadorian site. However, forests do not always display a straightforward~~
337 ~~relationship between N deposition and N₂O emissions. Manipulation experiments in the~~
338 ~~European NITREX project, for instance, revealed a much stronger correlation of N₂O emissions~~
339 ~~with soil NO₃⁻ leaching than with N deposition (Gundersen et al., 2012). Indeed, KCl extractable~~
340 ~~mineral N at the Dinghushan site (~ 40 mg kg⁻¹; Zheng et al., 2016) is several fold smaller than~~
341 ~~at our site (> 100 mg kg⁻¹; Zhu et al., 2013b), indicating that Dinghushan is less N-rich than TSP.~~
342 This ~~Therefore, we~~ suggests that the response of N₂O emission to P addition ~~might~~ may depend
343 on the N status of the soil. The fact that numerous studies found apparent suppression of N₂O
344 emission in short-term experiments (< 2 years) in N + P treatments, but not in treatments with P
345 alone, supports this idea (Müller et al., 2015; Zhang et al., 2014b; Zheng et al., 2016).

346 ~~Other studies have~~ Another study in a secondary in South China ~~reported~~ observed increased N₂O
347 emissions ~~during two years upon~~ after P addition, ~~in a secondary mixed forest~~ (Mori et al., 2013e;
348 Wang et al., 2014). ~~In an *Acacia mangium* plantation, fertilized with P, Mori et al. (2013b&e)~~
349 ~~found that N₂O emissions were stimulated in the short term but reduced in the long term.~~ While
350 suppression of N₂O emission by P has been attributed to increased plant N uptake (Mori et al.,
351 2014), increased N₂O emission ~~is~~ are generally explained by enhanced microbial ~~biomass growth~~
352 (Liu et al., 2012) and denitrification activity (Ehlers et al., 2010; He and Dijkstra, 2015). ~~P~~
353 ~~stimulation of N₂O emission by microbial denitrification should be rather fast, as indicated by~~
354 ~~Mori et al. (2013c) in a short-term (one week) incubation experiment with soils from an *Acacia*~~
355 ~~*mangium* plantation. Unlike Mori et al. (2013c), we did not find increased~~ N₂O emissions
356 ~~measured shortly within a week~~ after P addition at our site ~~in May 2014 were not different from~~
357 ~~fluxes in untreated reference plots~~ (Fig. S5). This may ~~indicate that~~ suggest that denitrifier
358 ~~community at TSP was not responsive to the P applied, probably because TSP hillslope soils~~
359 ~~have large denitrification potentials (Zhu et al., 2013c).~~ plant uptake at TSP is more important for
360 ~~the effect of P addition on N₂O emissions than changes in microbial activity, which are expected~~
361 ~~to occur more rapidly.~~

362 The Reference plots at TSP showed net CH₄ emission for extended periods ~~of the year~~ (Figs. 5
363 and 6). Also, long-term CH₄ fluxes sampled between 2012 and 2014 on hillslope ~~soilss~~ near-by
364 (Fig. S7; Zhu et al., unpublished data) showed net CH₄ emission. This is in contrast to the
365 generally reported CH₄ sink function of forested upland soils (Ciais et al., 2013; Dutaur and
366 Verchot, 2007). For example, ~~net~~ CH₄ ~~uptake rate~~ fluxes reported for well-drained, forest soils in
367 South ~~Chinese-China~~ forest range from ~~-30~~ to ~~-60~~ μg CH₄-C m⁻² hr⁻¹ (Fang et al., 2009; Tang et
368 al., 2006; Zhang et al., 2014a). Since aerated upland soils typically provide favourable conditions

369 for microbial CH₄ uptake (Le Mer and Roger, 2010), the net emission observed in our sites is
370 unlikely ~~to be~~ due to enhanced CH₄ production, but rather ~~by due to~~ suppressed CH₄ consumption.
371 One ~~reason-general explanation~~ for the net CH₄ emission at TSP could be inhibition of CH₄
372 oxidation by NH₄⁺, ~~as reported previously~~which competes with CH₄ for the active site at the
373 methane monooxygenase enzyme (Bodelier and Laanbroek, 2004; Zhang et al., 2014a). The
374 concentration of NH₄⁺ in the soil water was rather small (< 0.5 g L⁻¹; Fig. 1), which does not
375 preclude, however, that NH₄⁺ availability from the soil exchangeable pool ~~is-may have been~~ high.
376 Zhu et al. (2013b) found extraordinarily high KCL-extractable NH₄⁺ in TSP surface soils, likely
377 reflecting the large atmogenic NH₄⁺ input at the TSP site (Huang et al., 2015). On the other hand,
378 Reay and Nedwell (2004) found that NO₃⁻ inhibits methanotrophic activity in acidic soils, where
379 NH₃ is scarce. Possible mechanisms are the toxicity of denitrification intermediates (e.g. NO₂⁻;
380 Wang and Ineson, 2003) and the osmotic effect of high NO₃⁻ concentration (Hütsch et al., 1996).
381 This deduction can be supported by the high NO₃⁻ concentration in the acidic soils at TSP (Figs.
382 1 and S2).

383 P addition had a significant impact on CH₄ fluxes, changing the soil from a net source to a net
384 sink on an annual basis (Fig. 6). However, the uptake rates of CH₄ in the P treatments remained
385 smaller than those reported for forest soils in tropical China (Tang et al., 2006; Zhang et al.,
386 2008b). The stimulating effect of P addition on CH₄ uptake is consistent with previous studies
387 (Mori et al., 2013a, 2013b; Zhang et al., 2011), and has been attributed to ~~lessening the~~
388 ~~NH₄⁺ alleviating N~~ inhibition of methane oxidation. ~~Unfortunately, we did not measure KCl-~~
389 ~~extractable NH₄⁺ in our study, but a decline of available NH₄⁺, which is the substrate for~~
390 ~~nitrification, is likely as NO₃⁻ concentrations in soil water were significantly smaller with in the~~
391 ~~P-treatments (Fig. 1).~~ P addition may also result in a change of the taxonomic composition of the

392 methane oxidizing community (Mori et al., 2013a; Veraart et al., 2015). Alternatively, CH₄
393 oxidation may be stimulated by increased CH₄ diffusion into the soil, due to enhanced root
394 growth and increased soil water loss due to transpiration in P-amended plots (Zhang et al., 2011).
395 Given the strong N enrichment of TSP forest (Huang et al., 2015), it is likely that the reason for
396 the observed reduction in CH₄ emissions in response to P fertilization is due to alleviating direct
397 NH₄⁺ inhibition of methane monooxygenase (Veldkamp et al., 2013), rather than to P-
398 stimulation of methanotrophic activity (Veraart et al., 2015).

399 Shortly after fertilizer application, we observed a modest, albeit significant increase of Na⁺
400 concentration in soil water (Table S2). Other studies have documented the potential toxicity of
401 excess Na⁺ in soil water to plant and microbial activities (Rengasamy et al., 2003; Wong et al.,
402 2008). However, ~~the occurrence of~~ Na⁺ toxicity ~~at the treated plots to a degree~~ affecting N
403 turnover processes ~~in our plots~~, is unlikely, as Na⁺ concentrations in soil water, within one month
404 after application (Table S2), did not exceed 5 mg L⁻¹, ~~which is~~ far smaller than the values
405 commonly assumed to ~~cause~~ indicate toxicity (40 to 100 mg L⁻¹) (Bernstein 1975). Frequent
406 precipitation at TSP (Yu et al., 2016), both prior and following the addition of NaH₂PO₄·2H₂O
407 (Fig. 2), ~~apparently may have~~ diluted and leached Na⁺, thus preventing toxic effects.

408 P application significantly increased plant-available P in the P-limited TSP soil (Table 2).
409 Meanwhile, concentrations of leachable base cations (K⁺, Mg²⁺, Ca²⁺) in soil water decreased
410 (Fig. S3), as expected from the reduction of NO₃⁻ concentrations in the P-treatments, which
411 represent a major decline in mobile anions in the P-treated soils (Mochoge and Beese, 1986). We
412 observed no sign of stimulated forest growth or increased N uptake by ~~plants-trees~~ within the
413 relatively short period of our study (Table S3 and Fig. S6), making it difficult to link the
414 observed reduction in ~~mineral-inorganic~~ N in the soil solution (Fig. 1) to plant growth. When

415 interpreting the observed P effect on NO_3^- concentrations in soil water, several aspects need to be
416 considered. Firstly, two years of observation may be too short to detect any significant increase
417 in tree growth, due to NO_3^- uptake ~~by plants~~, given the commonly large variabilities in tree
418 biomass estimates (Alvarez-Clare et al., 2013; Huang et al., 2015). Secondly, a significant
419 proportion of the added P, and of excess N, may have been assimilated by the understory
420 ~~biomass~~vegetation, which was not assessed in this study. Previously, understory
421 ~~biomass~~vegetation has been reported to quickly respond to P addition (Fraterrigo et al., 2011).
422 Thirdly, as long-term N saturation and acidification at TSP have reduced forest health (Lu et al.,
423 2010; Wang et al., 2007), we may not expect immediate response of forest growth to P addition.
424 Large needle N/P ratios (17-22, Table S3) indicated that P limitation for tree growth was not
425 relieved 1.5 years after P addition (Li et al., 2016). Therefore, enhanced N uptake by understory
426 growth ~~and/or soil microbial biomass~~ may have been the main mechanisms responsible for
427 observed NO_3^- decline in the P-treated soil (Hall & Matson 1999).

428 ~~Overall, o~~Our study suggests that N-saturated TSP soils act as a regional hotspot for N_2O (Zhu et
429 al., 2013b) and CH_4 emissions. Within the short experimental period of 1.5 years, P fertilization
430 was shown to significantly decrease NO_3^- concentrations in soil water and to overall reduce ~~both~~
431 N_2O and CH_4 emissions. These findings provide a promising starting point for improving forest
432 management towards GHG abatement targets, taking into account the P and N status of
433 subtropical soils in the region.

434 **5 Acknowledgement**

435 Longfei Yu thanks the China Scholarship Council (CSC) for supporting his PhD study. Support
436 from the Norwegian Research Council to project 209696/E10 ‘Forest in South China: an
437 important sink for reactive nitrogen and a regional hotspot for N₂O?’ is gratefully acknowledged.
438 We thank Prof. Wang Yanhui, Prof. Duan Lei, Dr. Wang Zhangwei, Zhang Yi, Zhang Ting, Zou
439 Mingquan for their help during sample collection and data analysis. Dr. Zhu Jing is gratefully
440 acknowledged for providing unpublished data on long-term CH₄ fluxes in the TSP catchment.

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674 2012.

675 **Table 1** Background-Ambient soil properties of the experimental plots at Tieshanping (TSP).
 676 Values are means and standard deviations in parenthesis (n = 6)^φ. Soils were sampled in August
 677 2013.

	Soil Layer	pH	Total C g kg ⁻¹	Total N g kg ⁻¹	Total P mg kg ⁻¹	C/N	N/P
Block 1	O/A (0-3 cm)	3.7 (0.1)	80.7 (32.3)	4.8 (1.7)	308 (57)	17.0 (2.5)	15.5 (5.7)
	AB (3-8 cm)	3.8 (0.0)	23.9 (9.3)	1.3 (0.6)	-*	20.0 (3.0)	-
	B (8-20 cm)	3.9 (0.2)	8.6 (1.2)	< 0.05	-	-	-
Block 2	O/A (0-3 cm)	3.6 (0.1)	77.6 (13.4)	4.7 (0.8)	297 (44)	16.7 (1.3)	15.7 (2.8)
	AB (3-8 cm)	3.7 (0.1)	20.2 (5.3)	1.0 (0.3)	-	21.4 (3.3)	-
	B (8-20 cm)	3.9 (0.1)	7.1 (1.6)	< 0.05	-	-	-
Block 3	O/A (0-3 cm)	3.6 (0.1)	67.0 (15.5)	3.8 (0.8)	223 (45)	17.4 (0.6)	17.2 (3.7)
	AB (3-8 cm)	3.6 (0.1)	21.0 (7.9)	1.1 (0.5)	-	24.5 (4.6)	-
	B (8-20 cm)	3.8 (0.1)	7.2 (1.5)	< 0.05	-	-	-
	Soil Layer	P _{Al} mg kg ⁻¹	Al _{ox} mg kg ⁻¹	Fe _{ox} mg kg ⁻¹	P _{ox} mg kg ⁻¹	P _{ox} / (Al _{ox} + Fe _{ox})	
Block 1	O/A (0-3 cm)	5.8 (1.4)	1700 (513)	1933 (350)	85.8 (22.6)	0.025 (0.008)	
	AB (3-8 cm)	2.1 (0.6)	1217 (243)	1692 (493)	47.1 (22.0)	0.016 (0.007)	
	B (8-20 cm)	< 1.0	1083 (90)	1158 (249)	29.3 (28.6)	0.012 (0.011)	
Block 2	O/A (0-3 cm)	5.9 (1.0)	1500 (238)	1792 (215)	79.2 (21.5)	0.024 (0.007)	
	AB (3-8 cm)	1.6 (0.4)	925 (149)	1517 (320)	37.2 (10.7)	0.016 (0.006)	
	B (8-20 cm)	< 1.0	892 (209)	1033 (413)	16.1 (10.5)	0.009 (0.007)	
Block 3	O/A (0-3 cm)	4.1 (0.9)	1367 (180)	1667 (168)	50.7 (10.9)	0.017 (0.003)	
	AB (3-8 cm)	4.4 (4.0)	1075 (128)	1350 (150)	24.8 (8.3)	0.010 (0.002)	
	B (8-20 cm)	< 1.0	992 (130)	875 (138)	8.0 (2.0)	0.004 (0.001)	

678 P_{H2O} = Water-extractable P, P_{Al} = Ammonium lactate-extractable P,

679 Al_{ox} = Oxalate extractable Al, Fe_{ox} = Oxalate extractable Fe, P_{ox} = Oxalate extractable P.

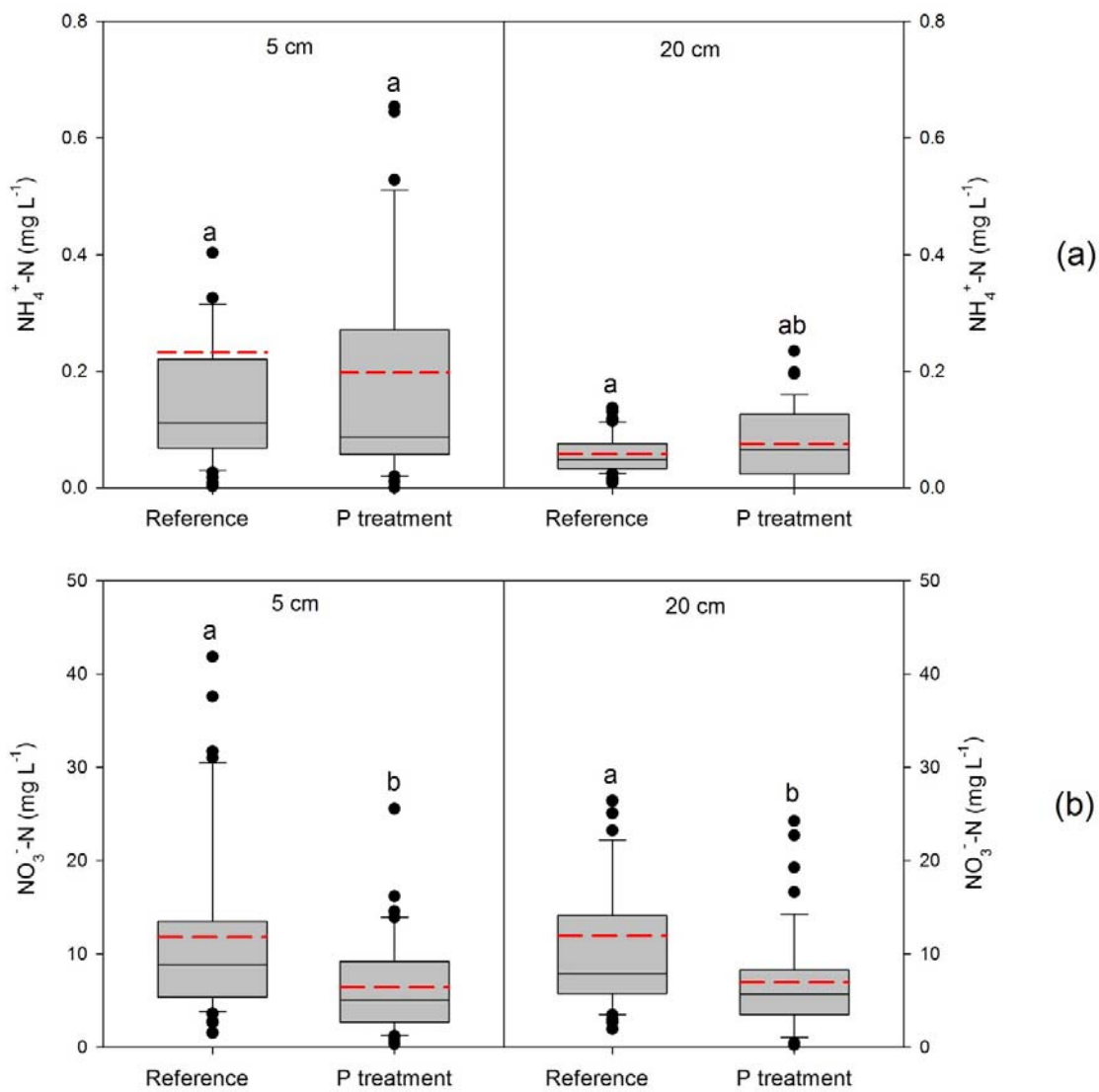
680 ^φ Water-extractable P was below a detection limit of 5 mg kg⁻¹, thus not presented in table.

681 * Data not available

682 **Table 2** Soil pH, C, N and P contents in the O/A horizon (0-3 cm) in the References (Ref) and P
 683 treatments. Values are means and standard deviations in parenthesis (n = 9). P addition was
 684 conducted on 14/05/04, after the first two sampling dates.

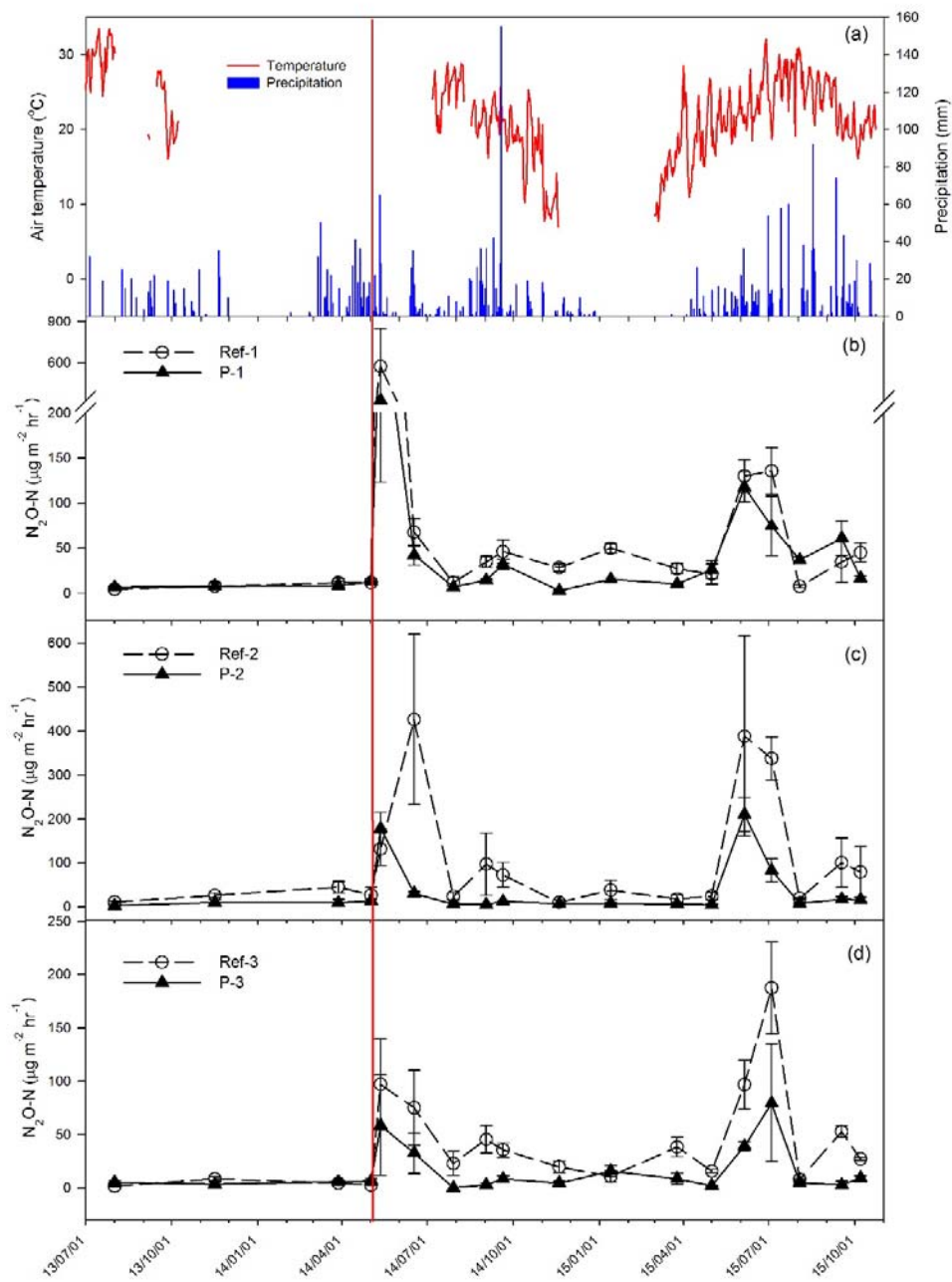
		pH	Total C g kg ⁻¹	Total N g kg ⁻¹	C/N	P _{AI} mg kg ⁻¹	Total P mg kg ⁻¹
13/08/02	Ref	3.7 (0.1) ^{ab†}	8.3 (2.3) ^{ab}	0.5 (0.1) ^{ab}	16.9 (1.1) ^b	5.4 (1.4) ^a	292 (46) ^{ab}
	P	3.6 (0.1) ^b	6.7 (2.0) ^b	0.4 (0.1) ^b	17.1 (2.1) ^{ab}	5.1 (1.3) ^a	260 (70) ^b
14/05/02	Ref	3.7 (0.1) ^a	12.2 (4.2) ^a	0.9 (0.3) ^a	13.7 (1.5) ^b	19.0 (8.0) ^a	336 (65) ^a
	P	3.8 (0.2) ^a	9.0 (3.5) ^{ab}	0.7 (0.2) ^{ab}	14.2 (2.8) ^{ab}	13.7 (5.2) ^a	270 (72) ^a
14/05/10	Ref	3.8 (0.1) ^{ab}	9.9 (2.1) ^a	0.7 (0.2) ^{ab}	14.0 (0.7) ^b	15.4 (7.0) ^b	304 (49) ^b
	P	3.9 (0.3) ^a	8.0 (1.9) ^a	0.6 (0.1) ^b	14.3 (1.3) ^{ab}	174 (114) ^a	572 (242) ^a
14/12/02	Ref	3.8 (0.1) ^a	10.5 (3.6) ^a	0.7 (0.3) ^a	14.5 (1.3) ^{ab}	14.2 (7.4) ^b	328 (102) ^b
	P	3.9 (0.2) ^a	9.5 (2.1) ^a	0.7 (0.1) ^{ab}	14.0 (0.8) ^b	66 (24) ^a	442 (106) ^{ab}
15/08/02	Ref	3.9 (0.2) ^{ab}	8.3 (2.2) ^{ab}	0.4 (0.1) ^{ab}	20.5 (2.5) ^a	13.4 (6.2) ^b	291 (61) ^a
	P	4.0 (0.2) ^a	6.5 (1.9) ^b	0.3 (0.1) ^b	19.7 (2.2) ^{ab}	57 (36) ^a	383 (136) ^a

685 † Different letters indicate significant differences between References and P treatments (p < 0.05).



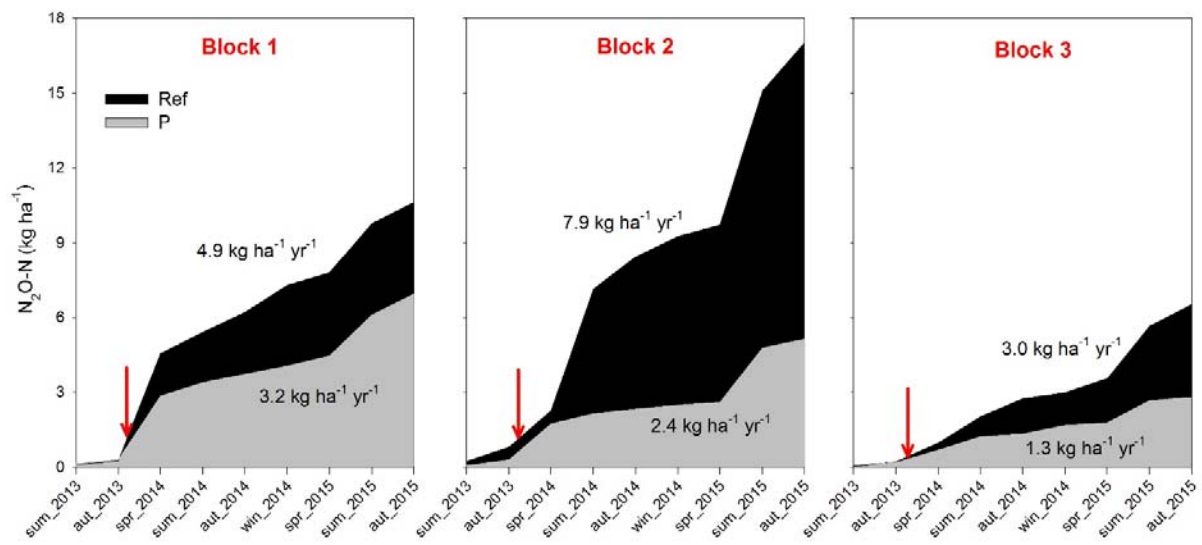
686

687 **Fig. 1** Box whisker plots of NH_4^+ (a) and NO_3^- (b) concentration in soil water at 5- and 20-cm
 688 depths in the References and P treatments, throughout 1.5 years after the P addition; red dashed
 689 lines indicate mean values; different letters indicate significant differences ($p < 0.05$).



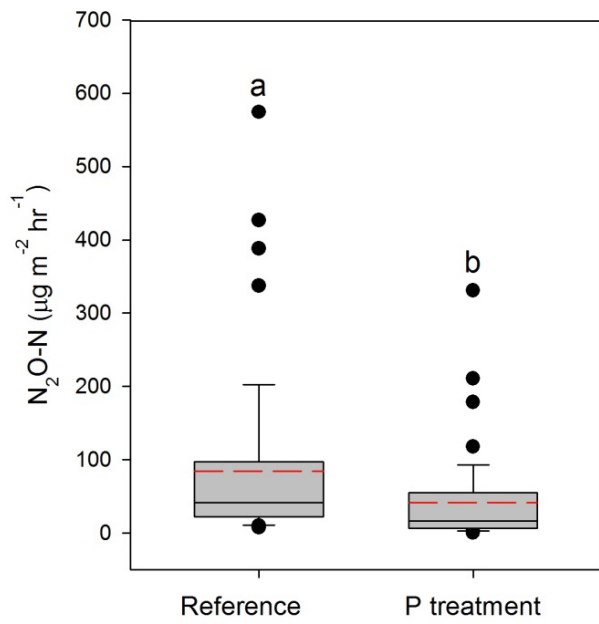
690

691 **Fig. 2** Daily mean air temperature and precipitation (a), and monthly mean N_2O fluxes ($\pm SE$) in
 692 the References (Ref) and P treatments in each of the three blocks (b-d); the red vertical line gives
 693 the date of P addition (4 May, 2014).



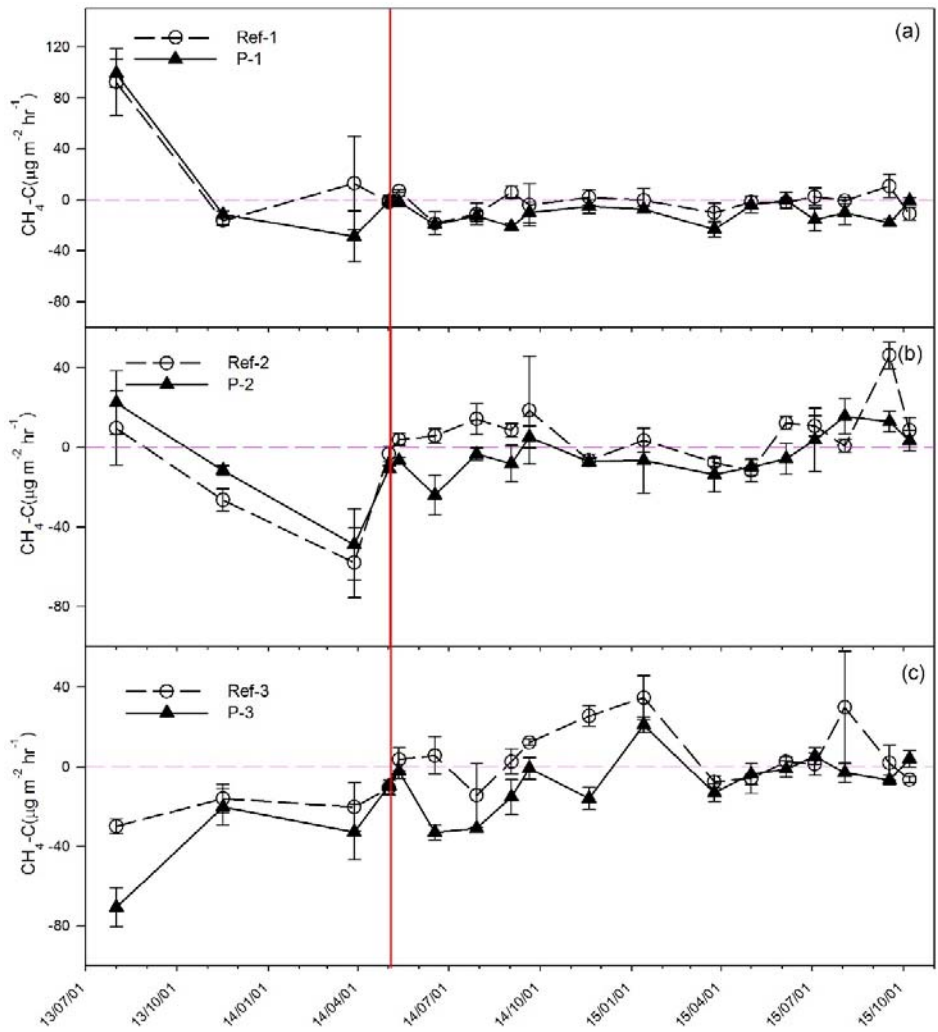
694

695 **Fig. 3** Cumulative N₂O emissions for three blocks in the References (Ref) and P treatments from
 696 summer 2013 to autumn 2015; the red arrows refer to the date of P addition (4 May, 2014).



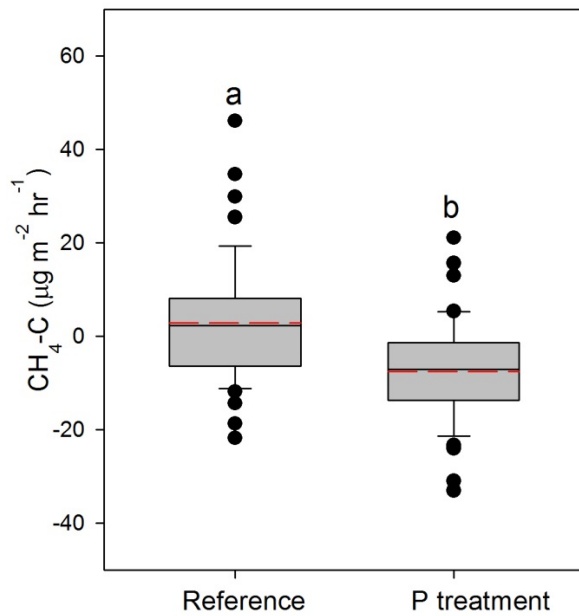
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698 **Fig. 4** Box whisker plots for N₂O fluxes in the Reference and P treatment throughout 1.5 years
 699 after the P addition; red dashed lines indicate mean values; linear mixed-effect models were used
 700 to test the P treatment effect; different letters indicate significant difference (p < 0.05).



701

702 **Fig. 5** Monthly mean CH₄ fluxes (\pm SE) in the References (Ref) and P treatments for three blocks
 703 (a-c); the horizontal broken line indicates zero flux the red vertical line refers to the date of P
 704 addition (4 May, 2014).



705

706 **Fig. 6** Box whisker plots of CH₄ fluxes in the Reference and P treatment throughout 1.5 years
 707 after the P addition; red dash lines indicate mean values; linear mixed-effect models were used to
 708 test the P treatment effect; the different letters indicate significant difference ($p < 0.05$).